

Supporting Information

Pharmaceutical Salts to Improve Diffusion Permeability of a BCS class III β -blocker Drug Atenolol

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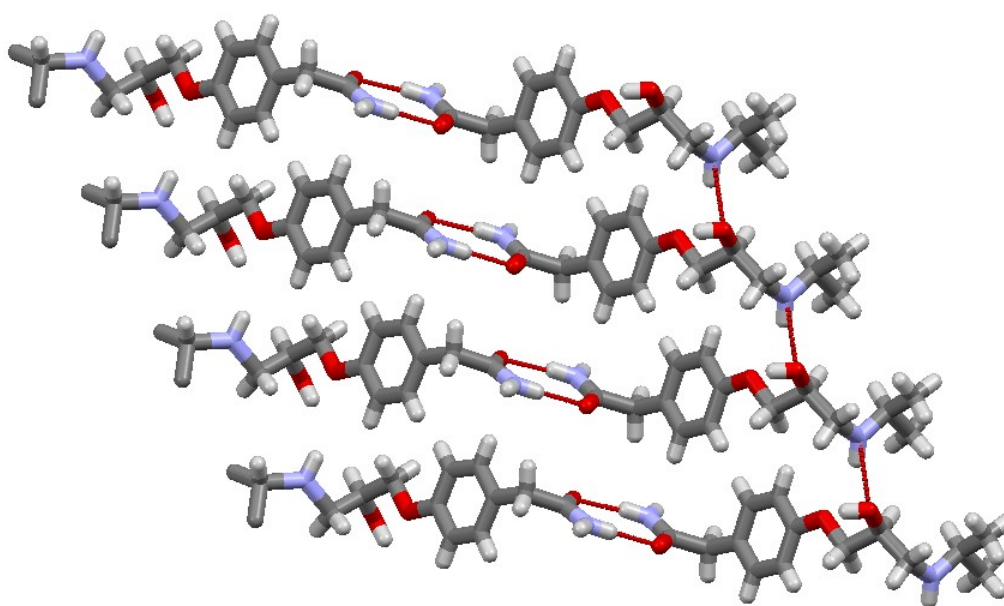
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Table S1. Hydrogen bond geometry (Å, °)

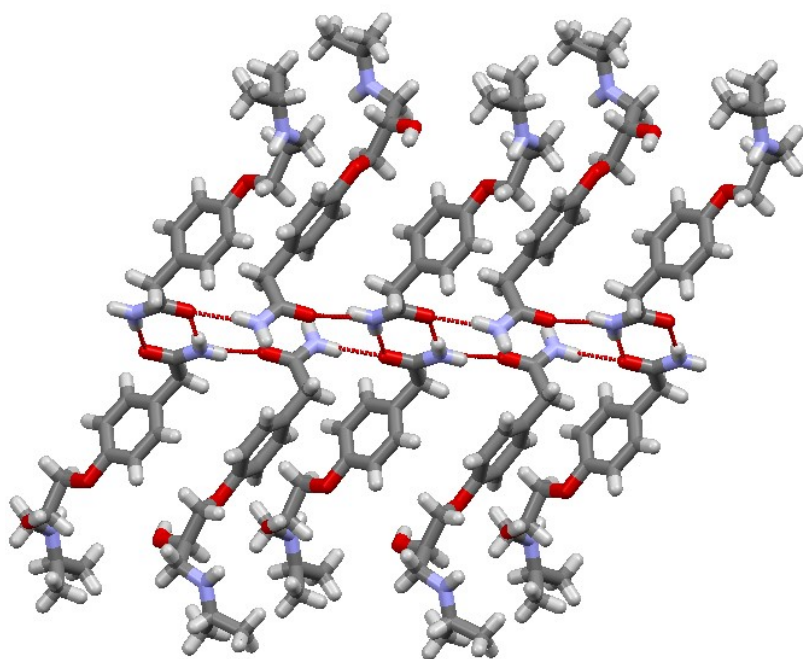
	D-H...A	D-H/ Å	H...A/Å	D...A/Å	D-H...A/°	Symmetry code
ATL-FUM hydrate	N1-H1A...O4	0.86	2.04	2.8269(1)	151	x,3/2-y,1/2+z
	N1-H1B...O1	0.86	2.25	3.0165(1)	149	-x,-1/2+y,1/2-z
	N2-H2A...O4	0.90	1.83	2.7338(1)	179	1-x,1-y,-z
	N2-H2B...O3	0.90	1.96	2.8280(1)	160	1-x,1-y,-z
	O3-H3...O5	0.84	1.78	2.6224(1)	174	1-x,1-y,-z
	O6-H6A...O5	0.84	2.06	2.8942(1)	177	-
	O6-H6B...O1	0.87	2.01	2.8082	176	x,-1+y,z
	C4-H4...O5	0.92	2.52	3.3875(1)	155	x,1+y,z
ATLA DP hydrate	N1-H1A...O4	0.86	2.23	3.0801(3)	170	1-x,1-y,-z
	N1-H1B...O3	0.86	2.17	2.9958(3)	162	-1+x,y,Z
	N2-H2A...O14	0.89	1.94	2.7738(2)	156	2-x,1/2+y,1/2-z
	N2-H2B...O15	0.89	1.90	2.7575(2)	162	-
	O3-H3...O13	0.82	1.85	2.6626(2)	173	--
	N3-H3A...O10	0.86	2.24	3.0802(3)	164	2-x,1-y,-z
	N3-H3B...O9	0.86	2.18	2.9943(3)	158	-
	N4-H4A...O20	0.89	1.94	2.7920(2)	160	1+x,y,z
	N4-H4B...O17	0.89	1.87	2.7428(2)	168	-
	N5-H5A...O6	0.86	2.14	2.9633(3)	160	2-x,1-y,-z
	N5-H5B...O1	0.86	2.04	2.8829(3)	168	-
	O6-H6...O18	0.82	1.80	2.6202(2)	173	1+x,y,z
	N6-H6A...O13	0.89	1.90	2.7569(2)	162	2-x,1/2+y,1/2-
	N6-H6B...O16	0.89	1.95	2.7843(2)	156	2-x,1/2+y,1/2-
	N7-H7A...O4	0.86	2.15	2.9685(3)	160	1-x,2-y,-z
	N7-H7B...O12	0.86	2.02	2.8506(3)	163	-1+x,y,z
	N8-H8A...O18	0.89	1.89	2.7617(2)	168	1-x,1/2+y,1/2-
	N8-H8B...O19	0.89	1.90	2.7548(2)	161	2-x,1/2+y,1/2-
	O9-H9...O15	0.82	1.82	2.6339(2)	172	2-x,1/2+y,1/2-
	O12-H12A...O20	0.82	1.90	2.7178(2)	176	2-x,1/2+y,1/2-
	O21-H21A...O14	0.85	2.06	2.7595(2)	139	-
	O21-H21B...O24	0.85	1.96	2.7762(2)	161	1+x,y,z
	O22-H22A...O22	0.85	1.93	2.7782(2)	175	2-x,-1/2+y,1/2
	O22-H22B...O24	0.85	1.89	2.7384(2)	172	-
O23-H23C...O21	0.85	2.07	2.7722(2)	140	-	
O24-H24A...O17	0.85	2.00	2.7838(2)	154	-	
O24-H24B...O23	0.85	1.89	2.7340(2)	170	-	

Table S2. Calculation of Distribution Coefficients of ATL and its salts

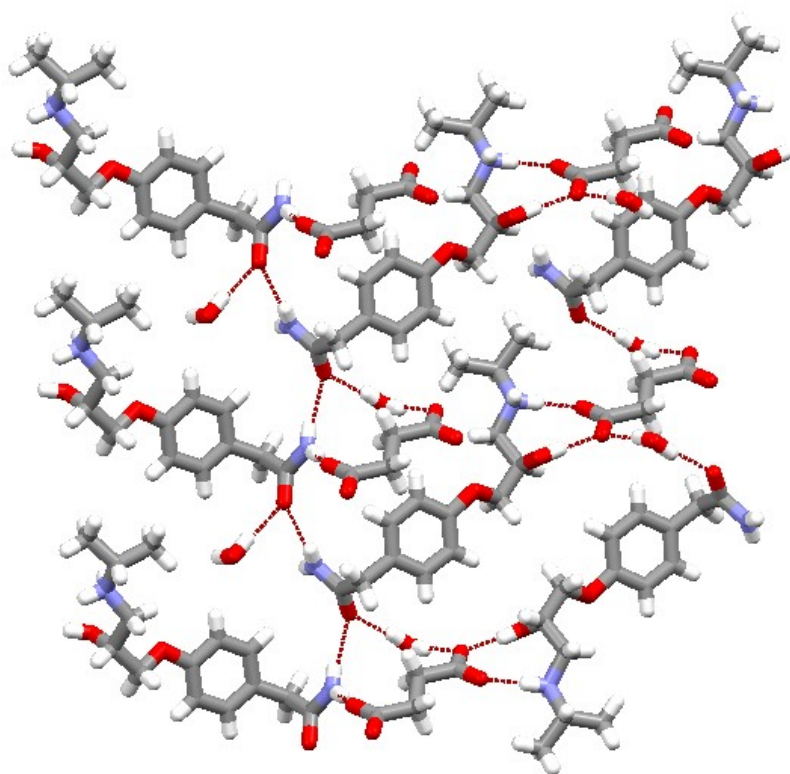
Salts	C_w (mg/mL)	C_0 (mg/mL)	C_E (mg/mL)	$C_{\text{initial}} - C_E$ (mg/mL)	$\text{Log}(C_0/C_w)$	$\text{Log}((C_{\text{initial}} - C_E)/C_E)$
ATL	1.035	0.3442	0.9041	0.1309	-0.47	-0.83
ATL-ADP	0.8869	0.3945	0.6590	0.2279	-0.351	-0.461
ATL-FUM	0.8150	0.5426	0.6647	0.1503	-0.176	-0.645
ATL-GLU	0.8084	0.4813	0.5646	0.2438	-0.22	-0.364
ATL-MAL	0.7833	0.3817	0.4954	0.2879	-0.312	-0.23
ATL-OXA	0.7786	0.4165	0.4915	0.2865	-0.271	-0.234
ATL-PIM	0.7291	0.4134	0.4734	0.2557	-0.251	-0.267



(a)

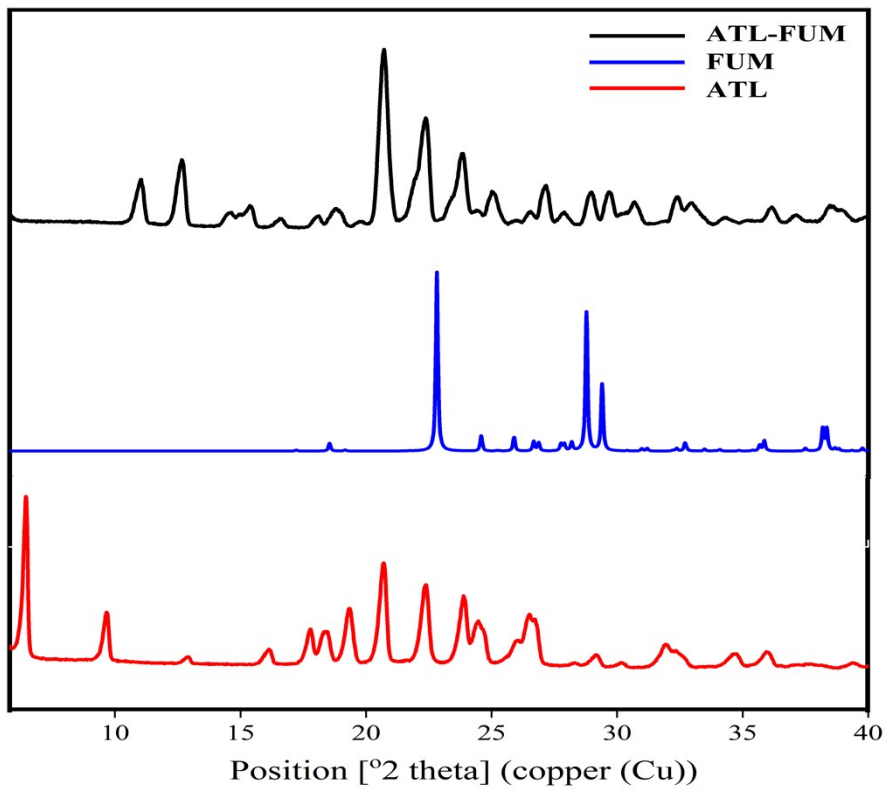


(b)

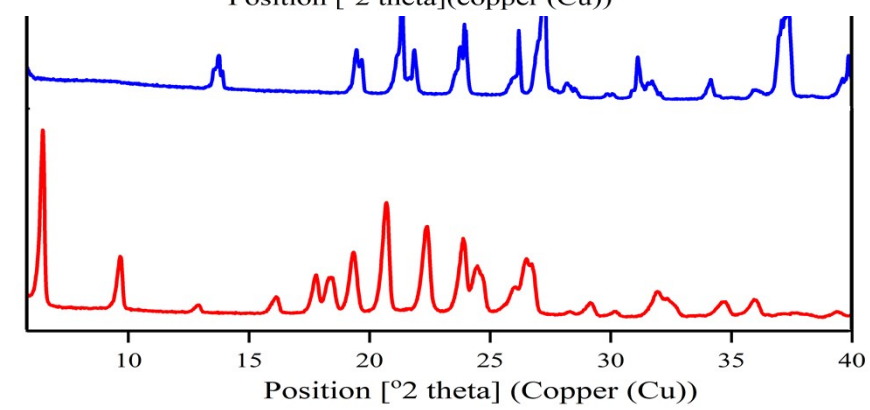
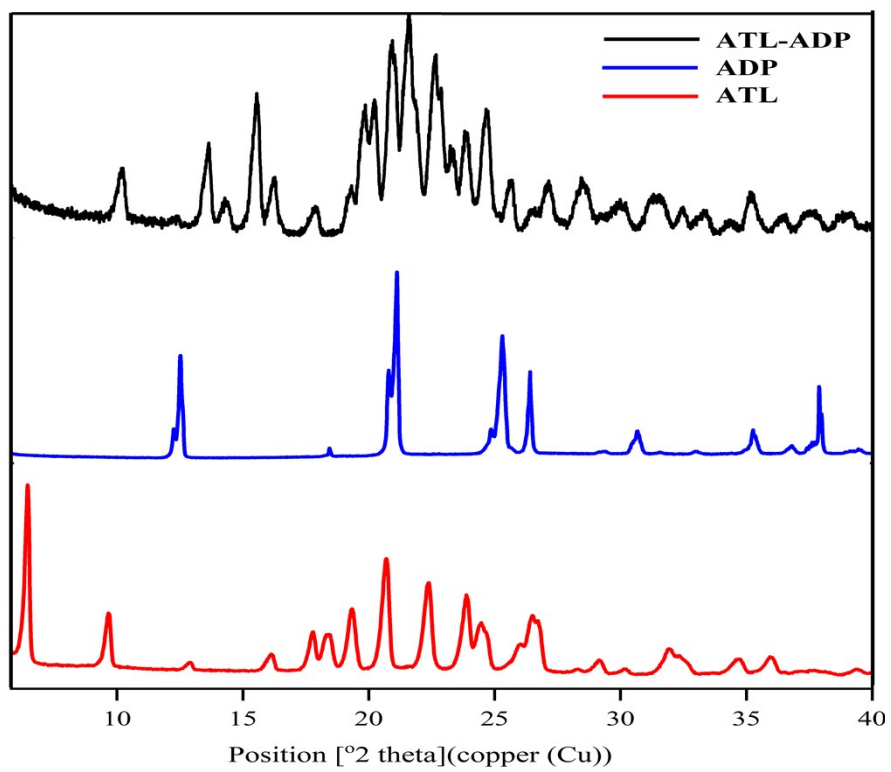


(c)

Figure S1. Hydrogen bonded amide dimer in a) S-ATL and b) RS ATL. c) ATL-succinate salt hydrate represents proton transfer from succinic acid to ATL, while amide fractions involved in catemer hydrogen bonds.

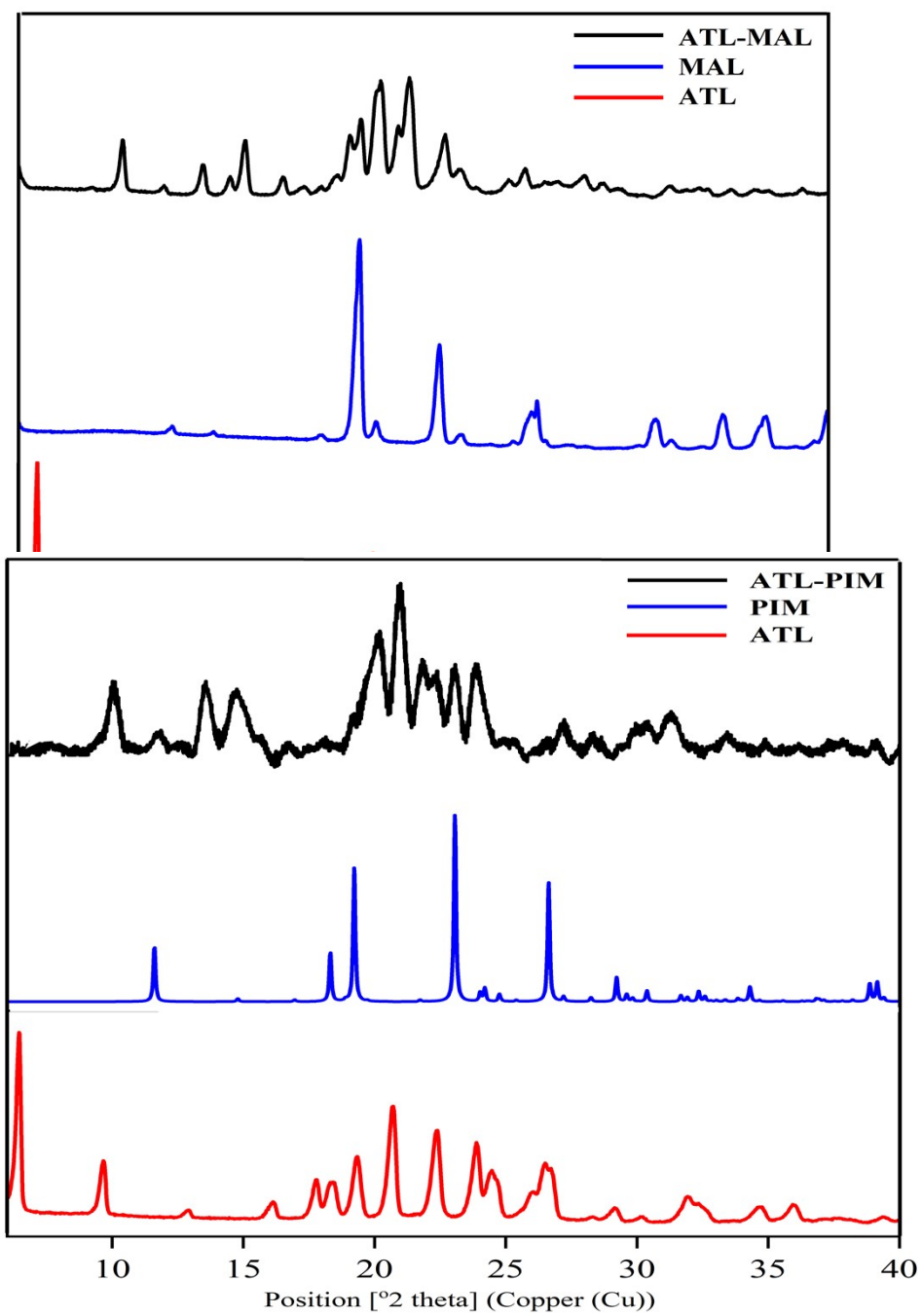


(a)

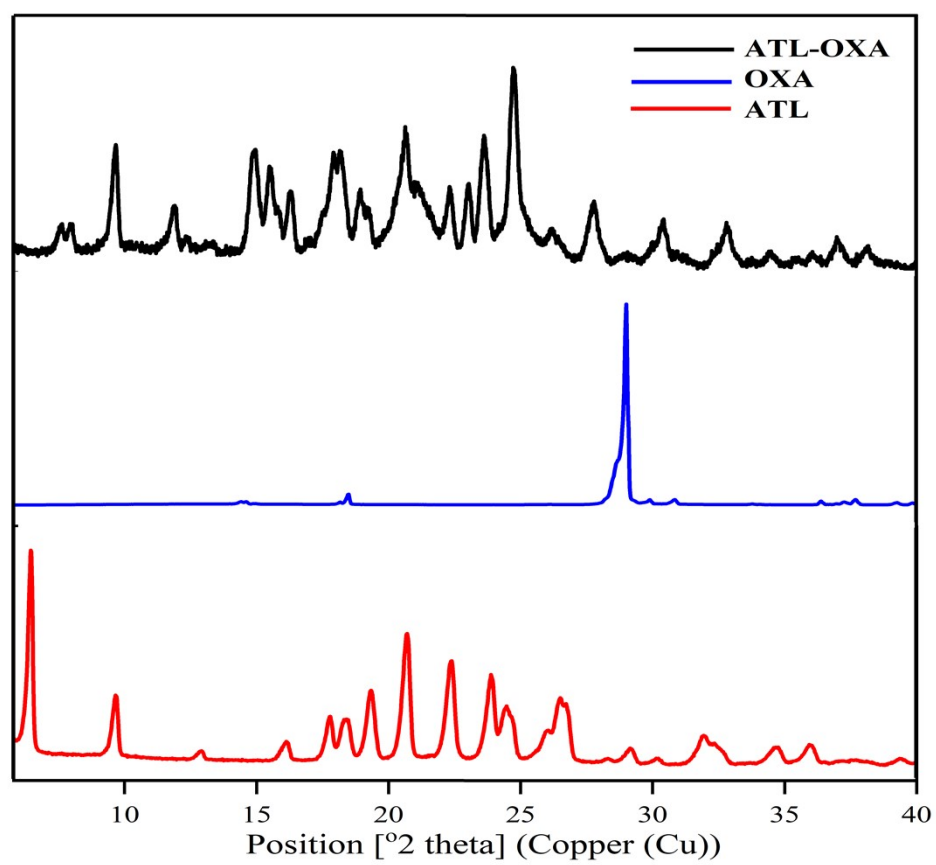


(c)

(c)



(e)



(f)

Figure S2. PXRd comparison of (a-f) ATL salts with that of ATL and salt former confirm their distinct phase

(a)

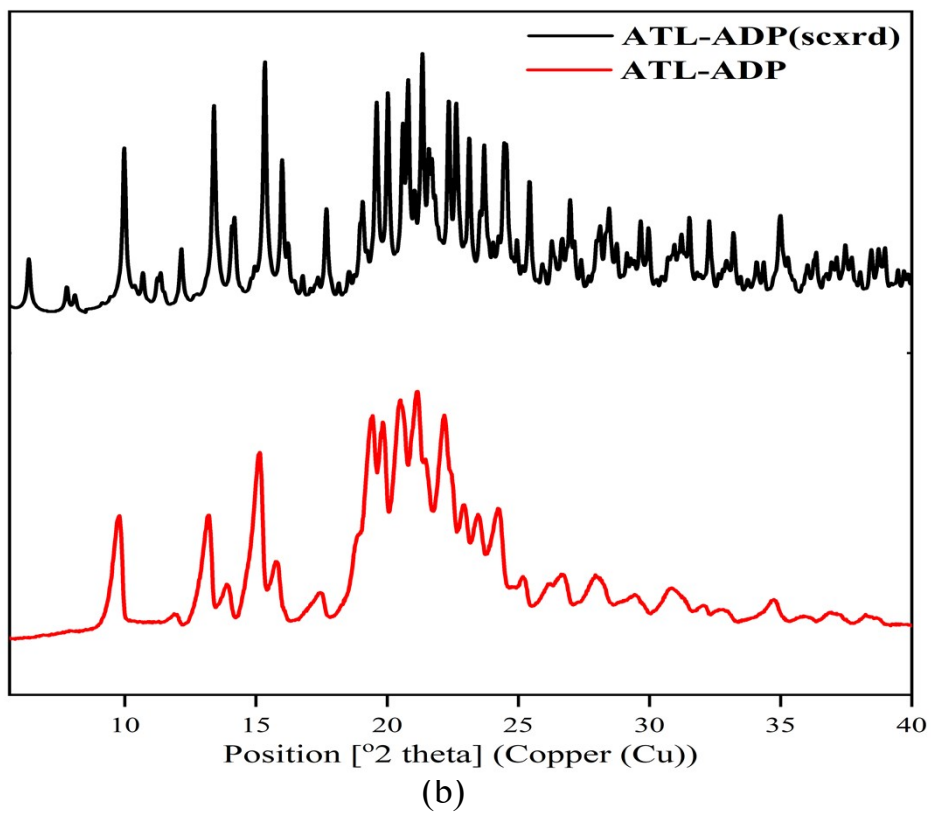
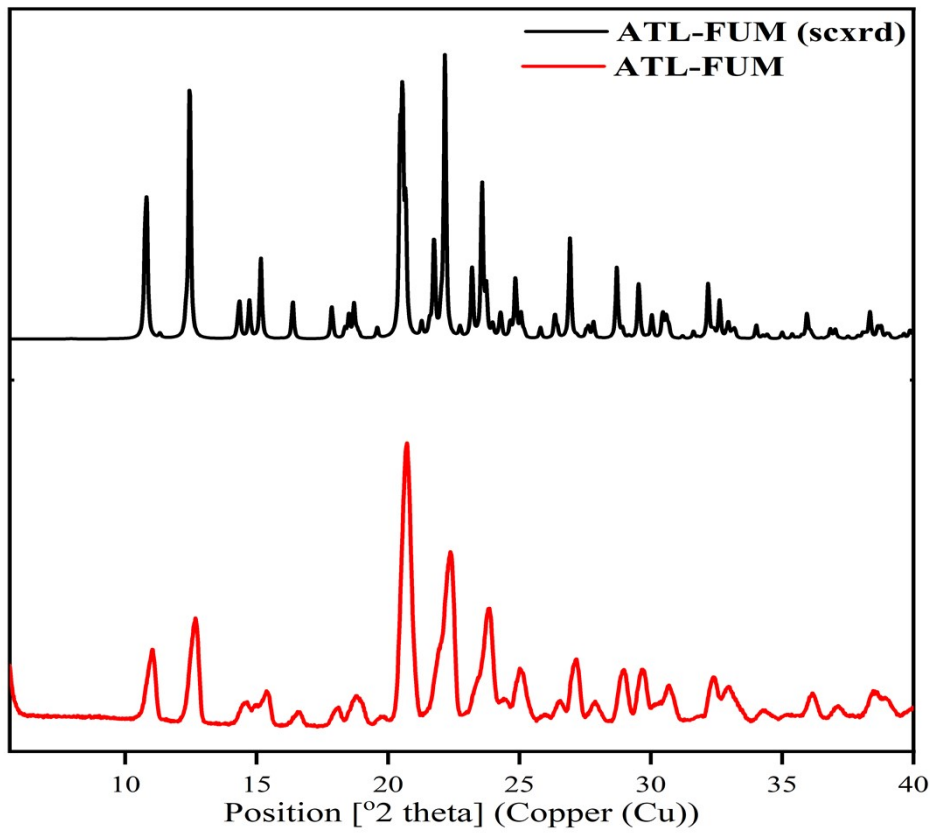
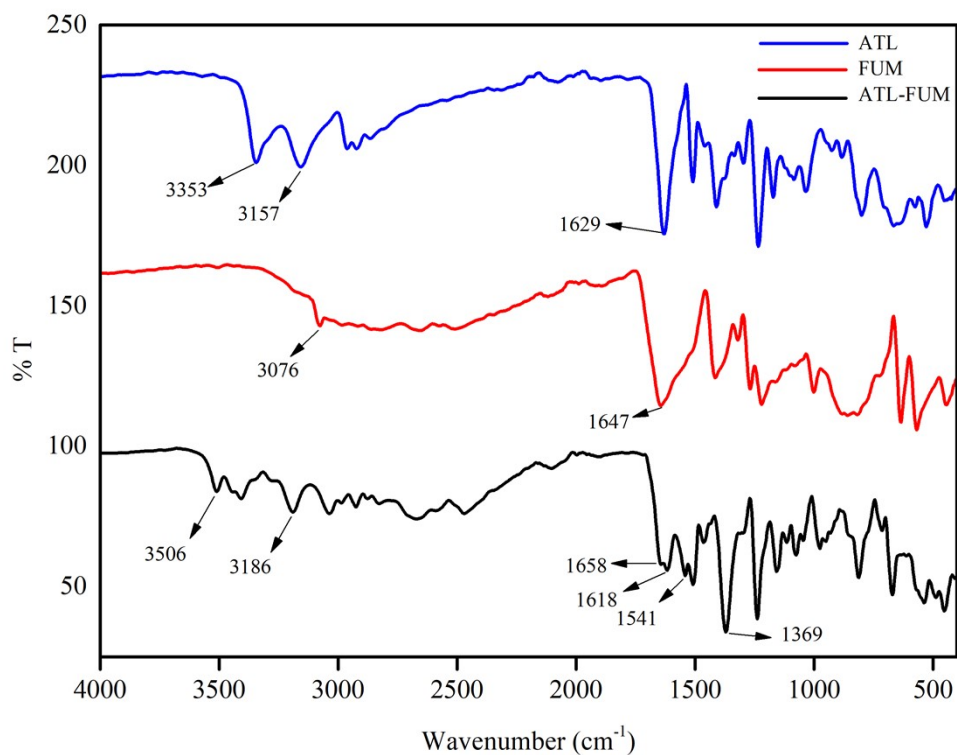
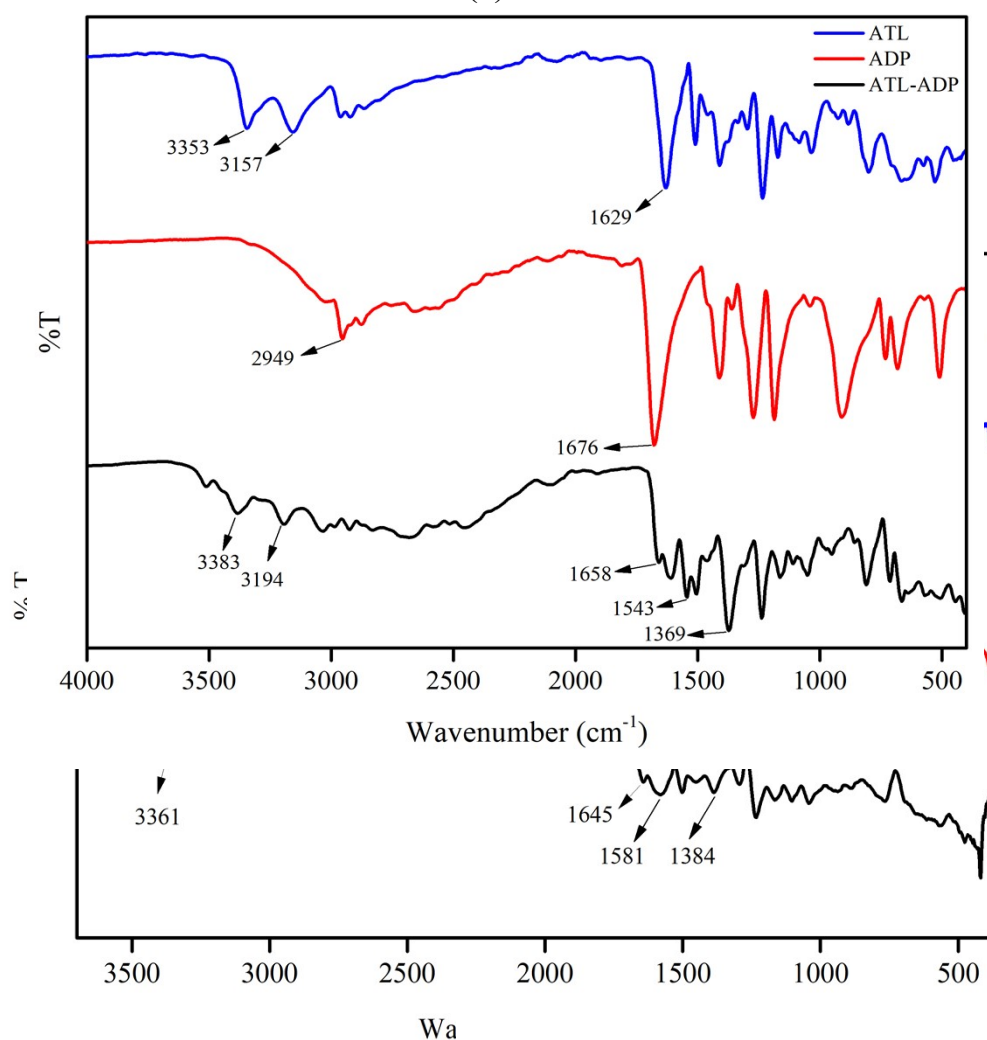


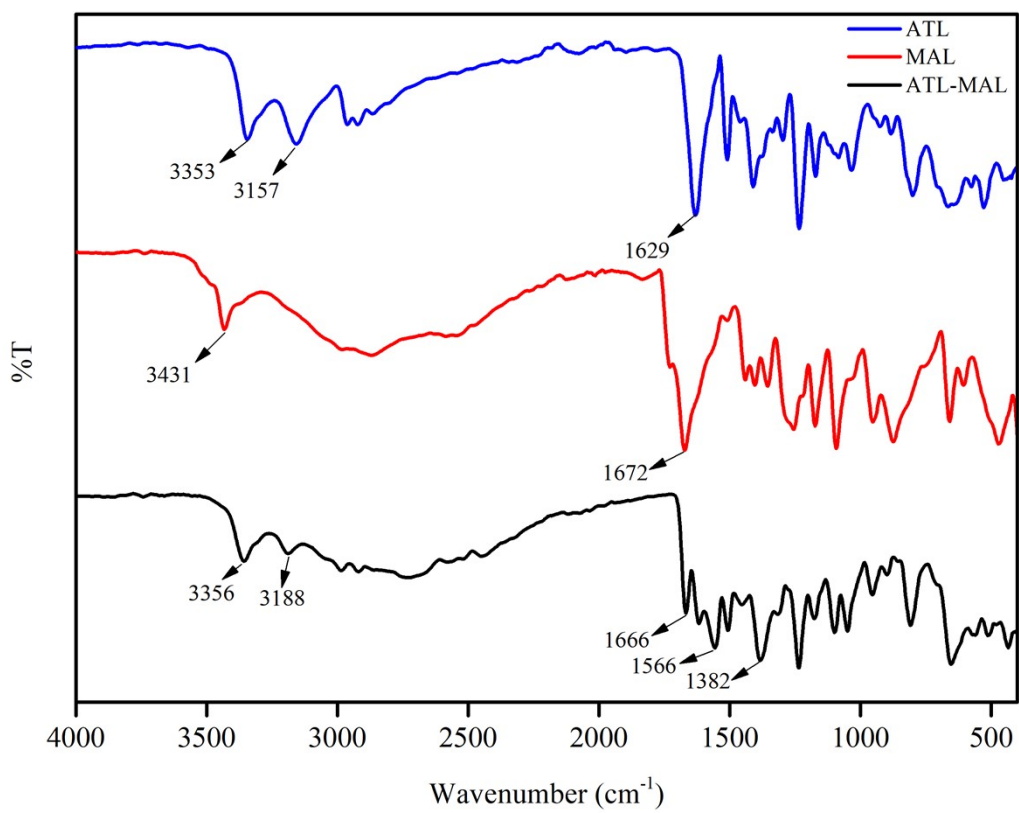
Figure S3. Comparison of PXRD pattern of a) ATL-FUM, b) ATL-ADP with the calculated X-ray pattern from their crystal structures confirmed purity of the bulk phase. Note, slight deviation of ATL-ADP hydrate in the XRD patterns compared to its crystal structure pattern indicates instability of the phase.



(a)



(c)



(d)

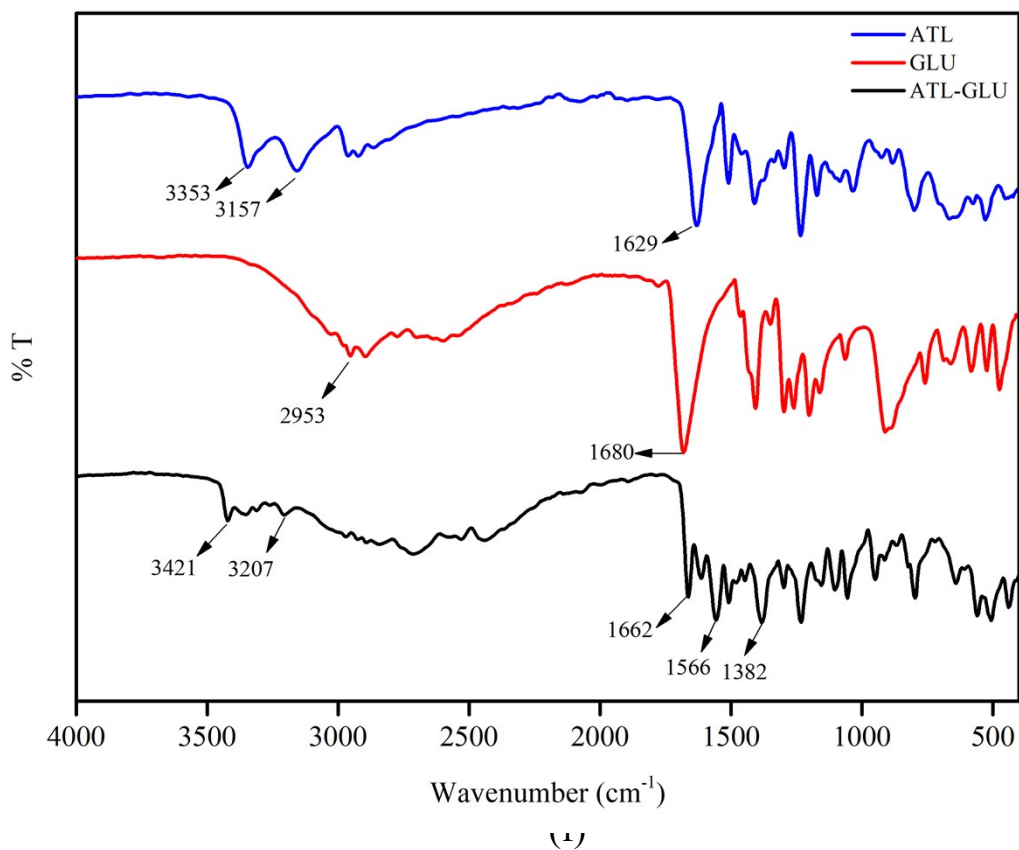
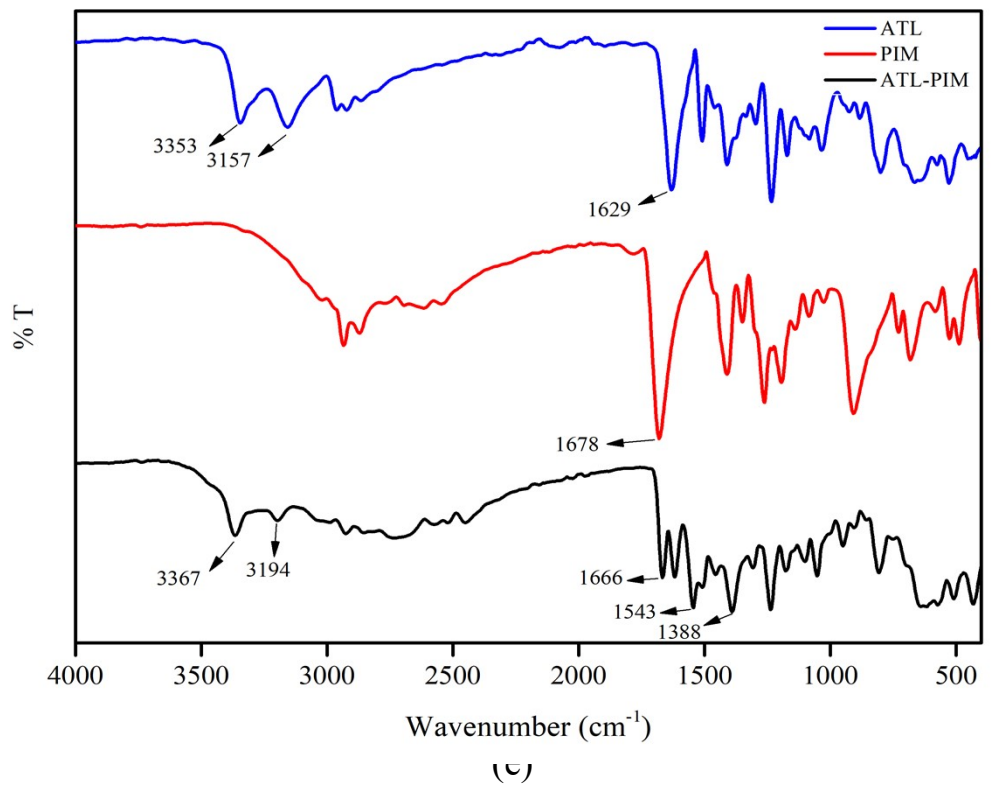
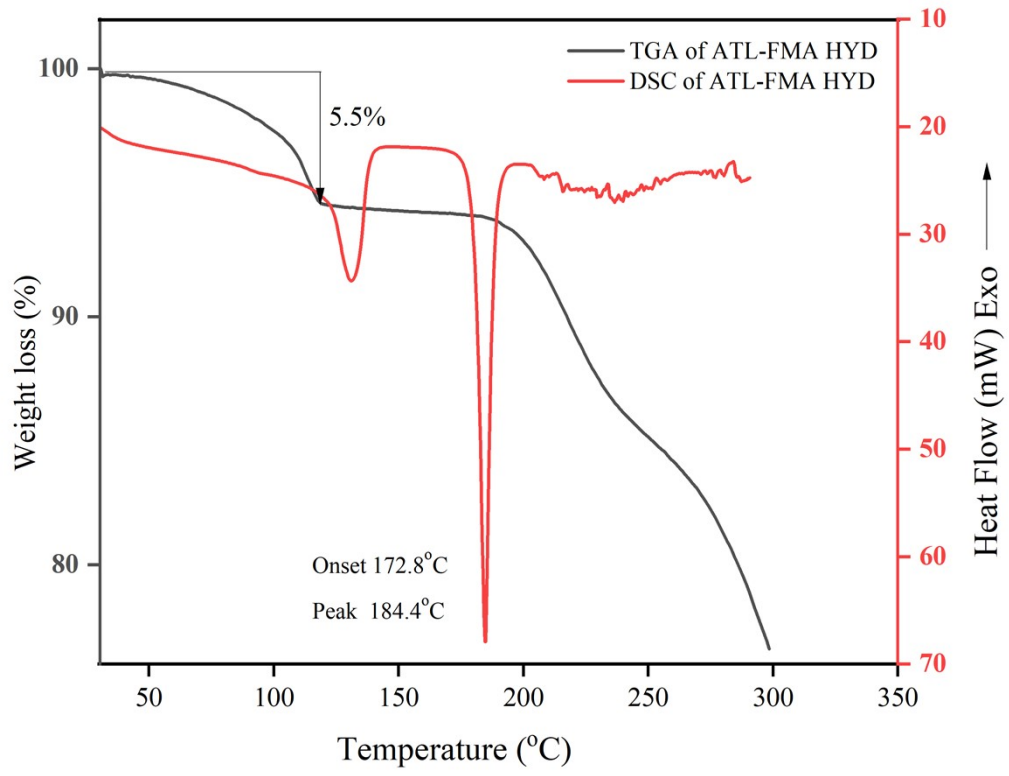
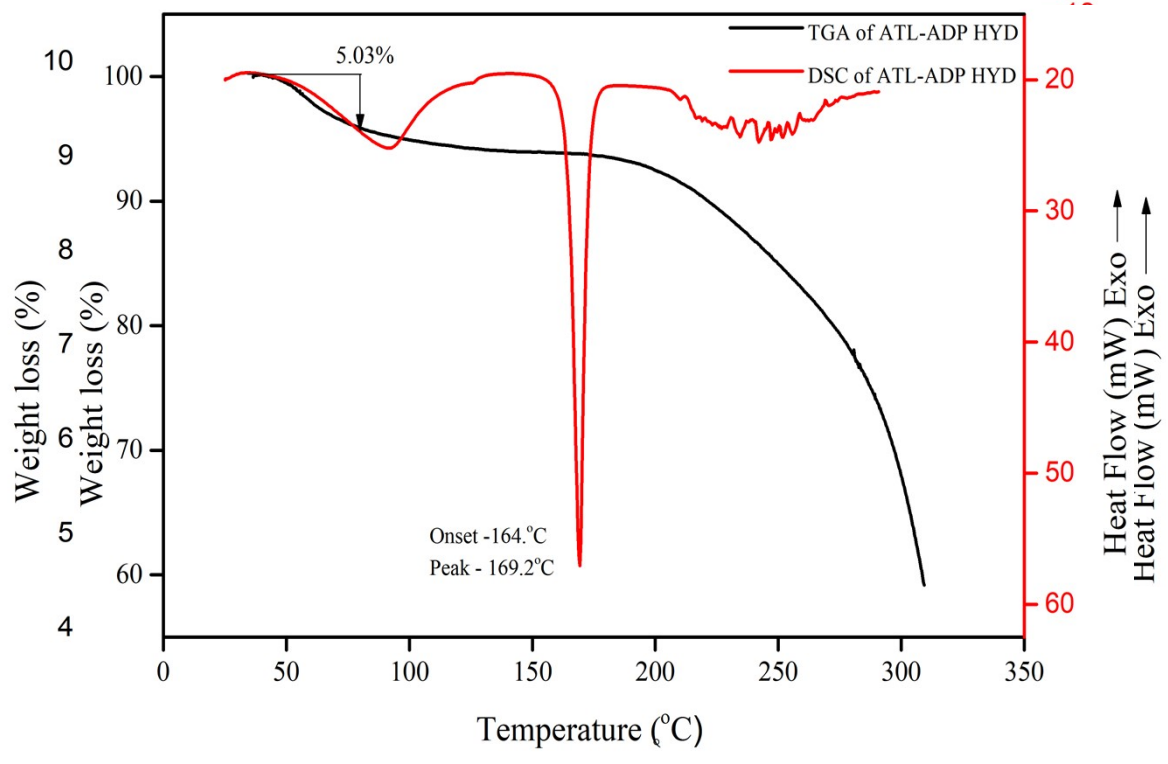


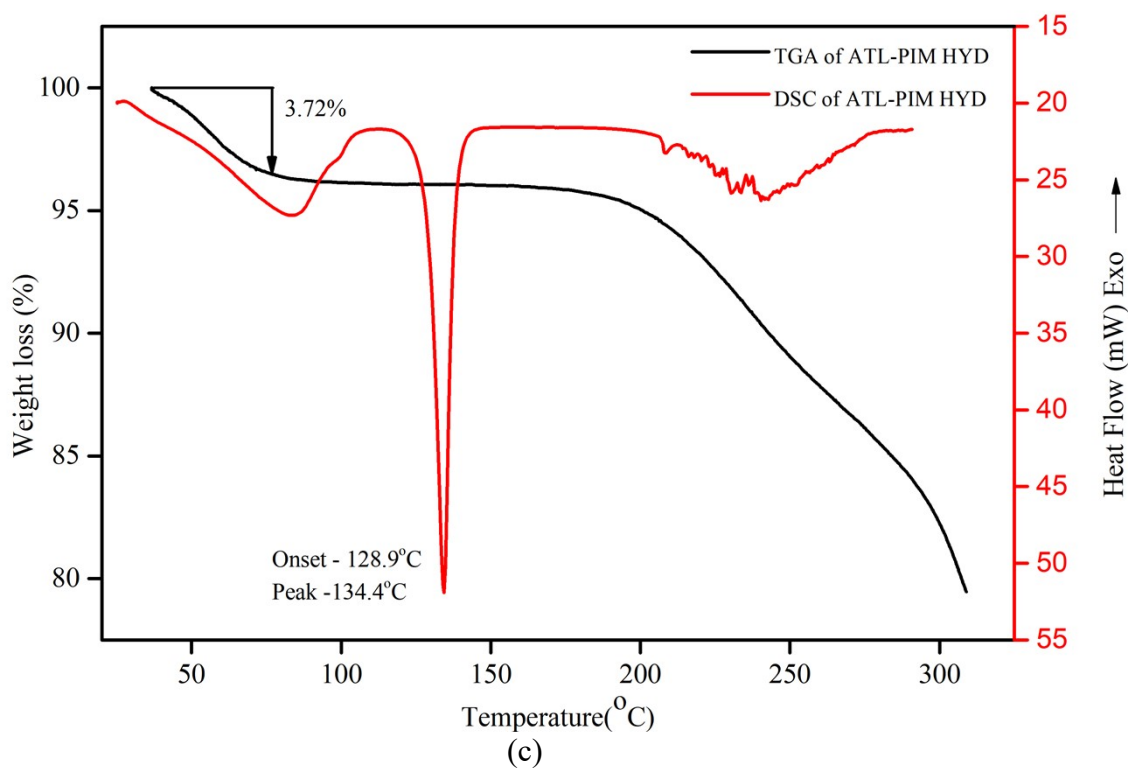
Figure S4. Vibrational frequencies comparison of ATL salts with ATL and salt formers (a-f).



(a)

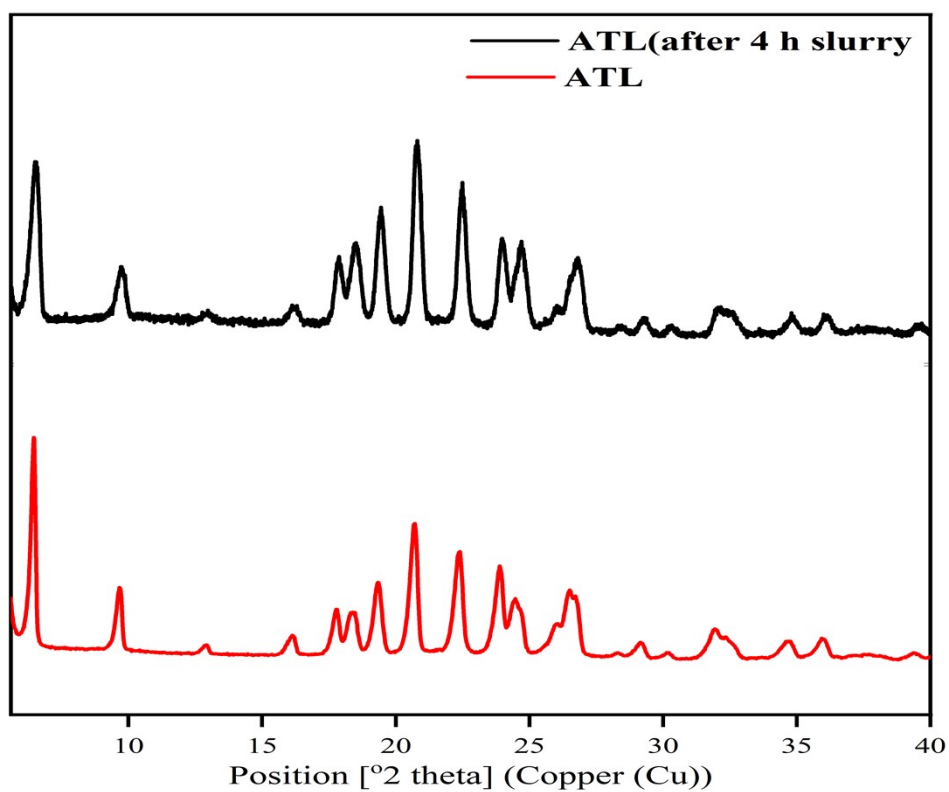


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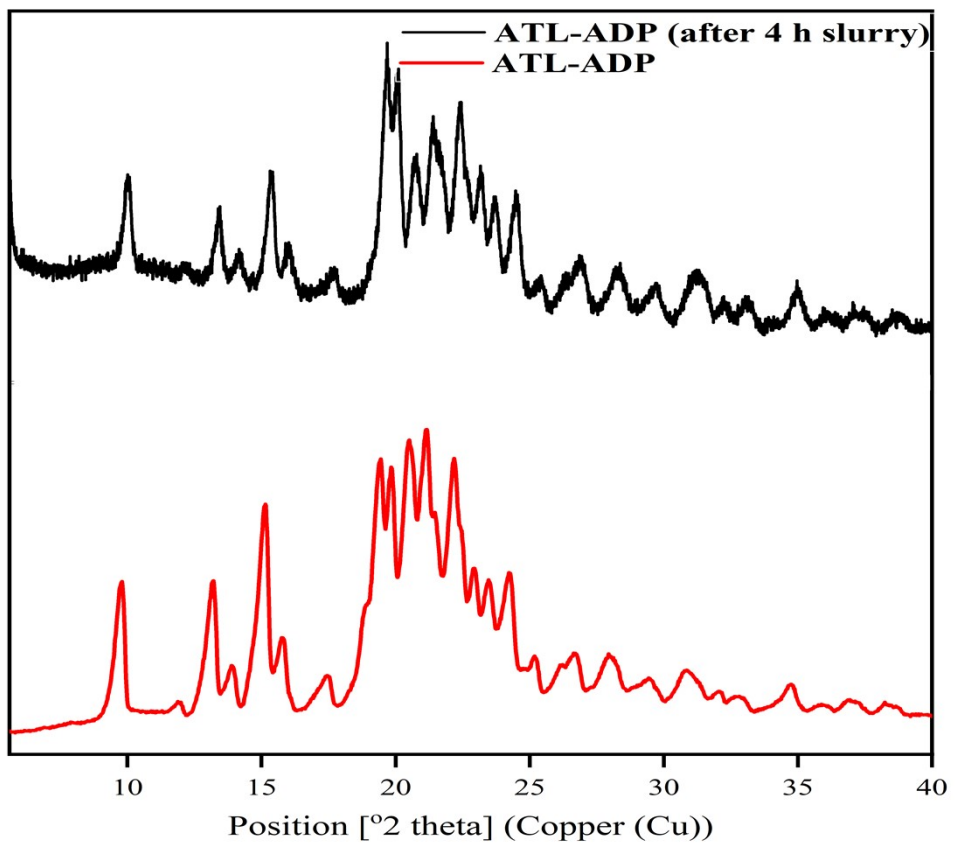


(d)

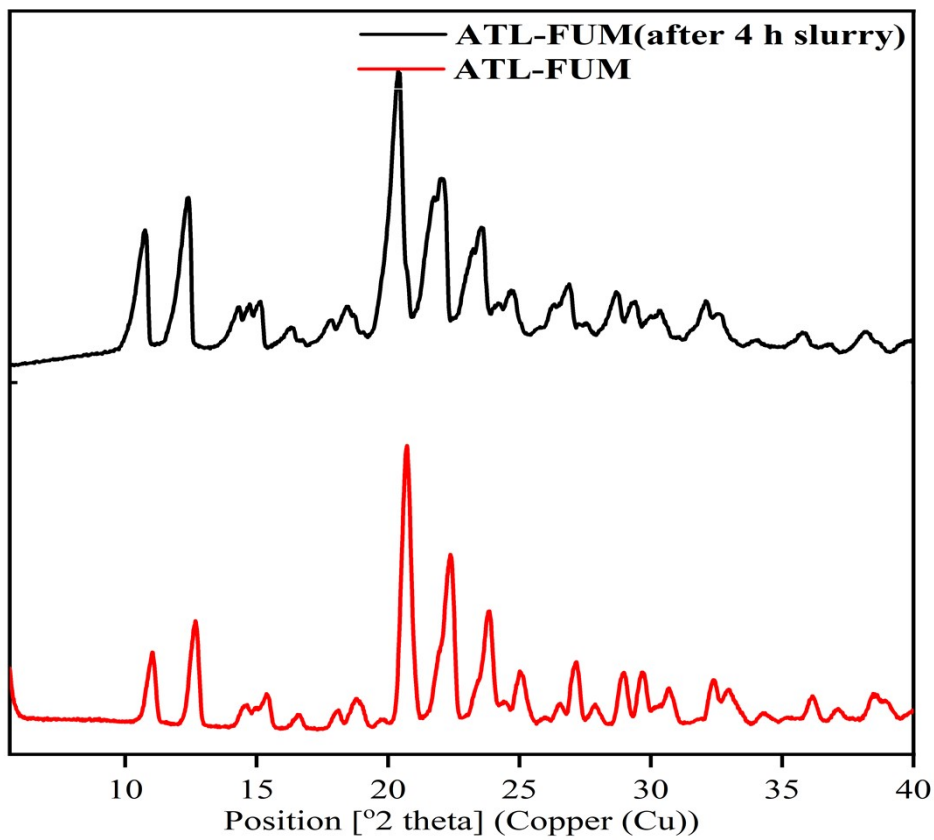
Figure S5. DSC -TGA thermograms of a) ATL-FUM, b) ATL-ADP, c) ATL-OXA, d) ATL-PIM salt hydrates.



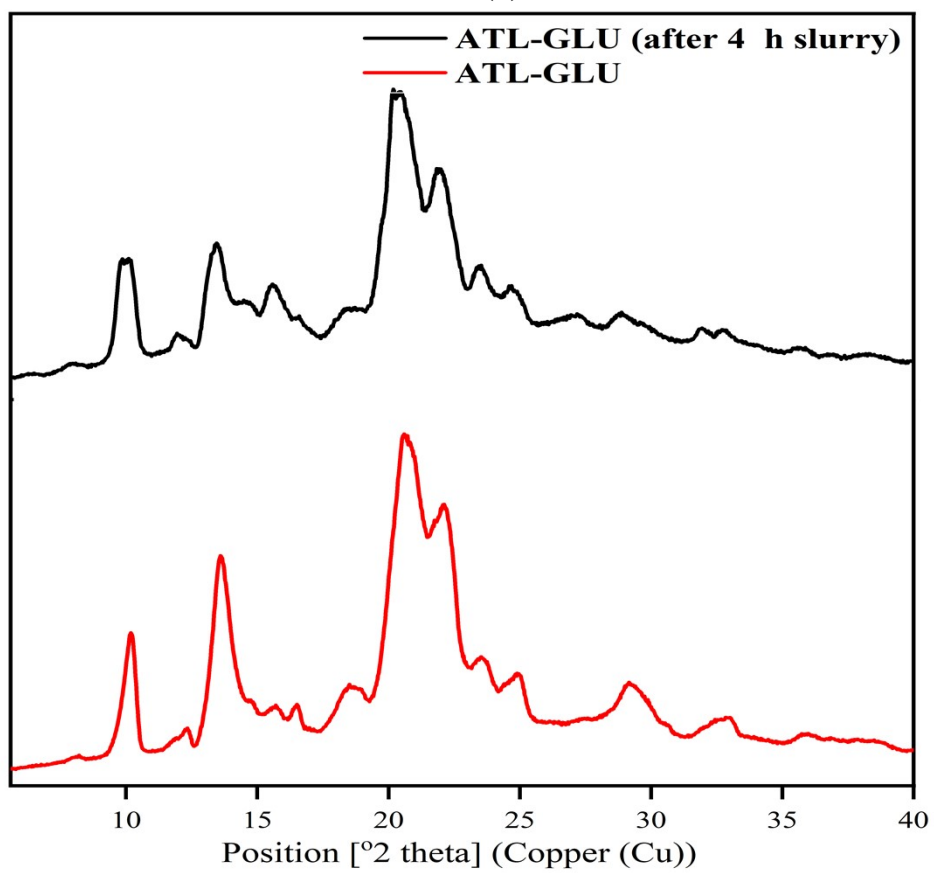
(a)



(b)



(c)



(d)

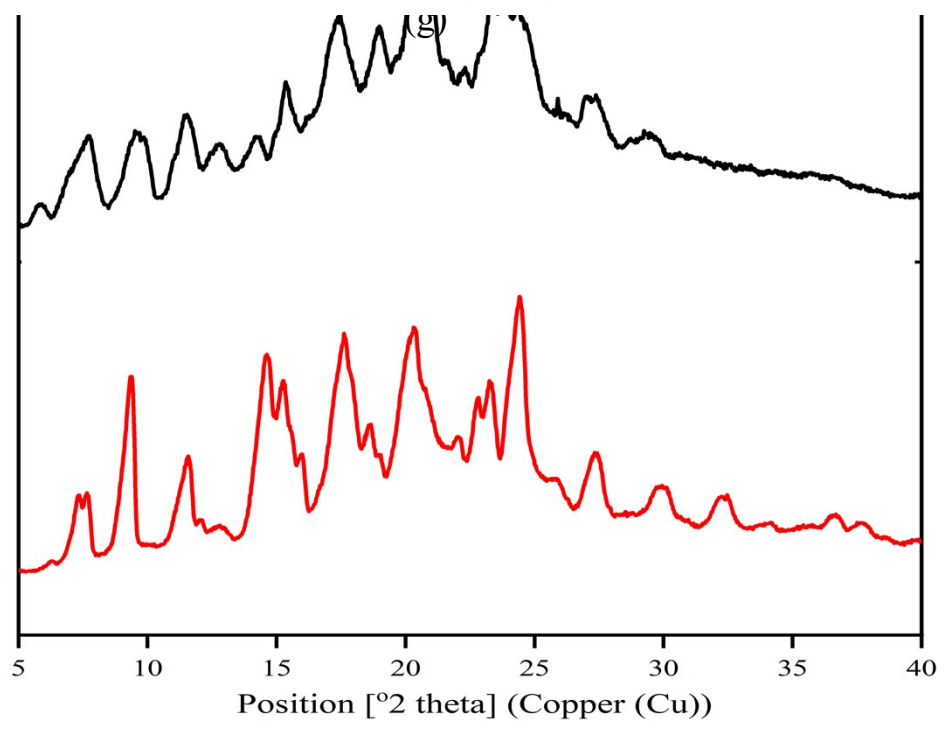
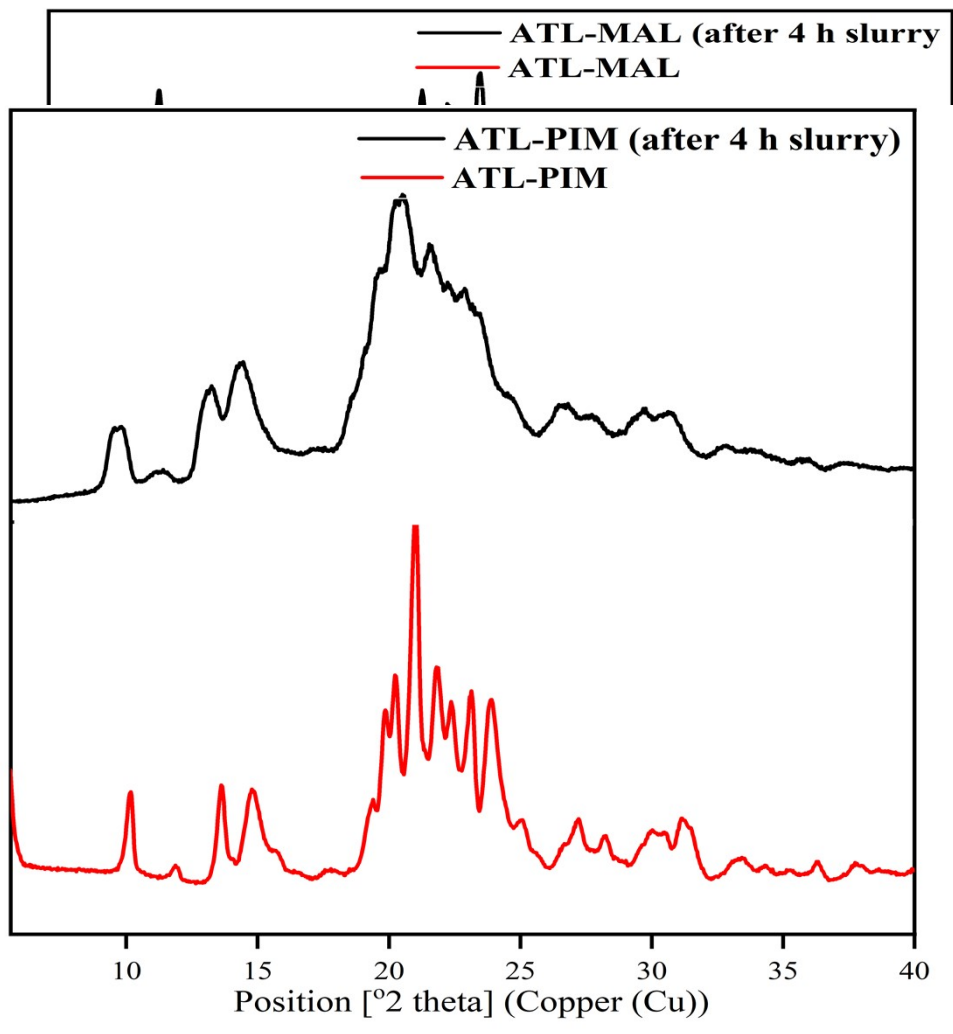
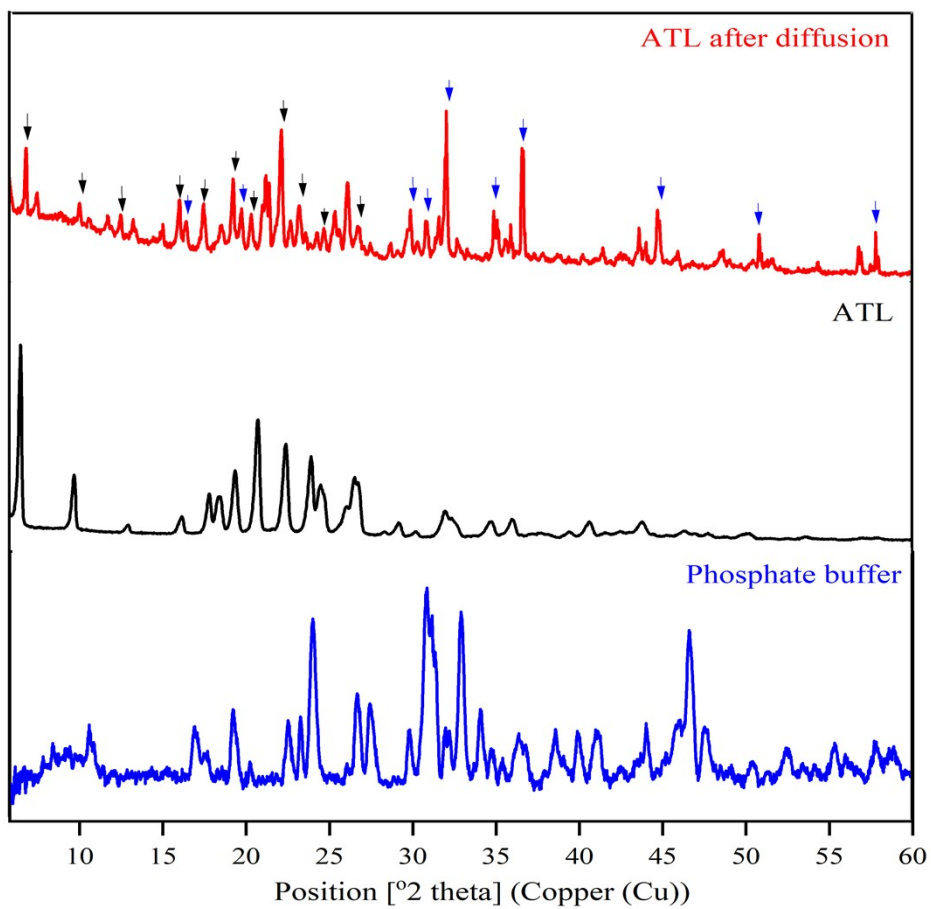


Figure
PXRD

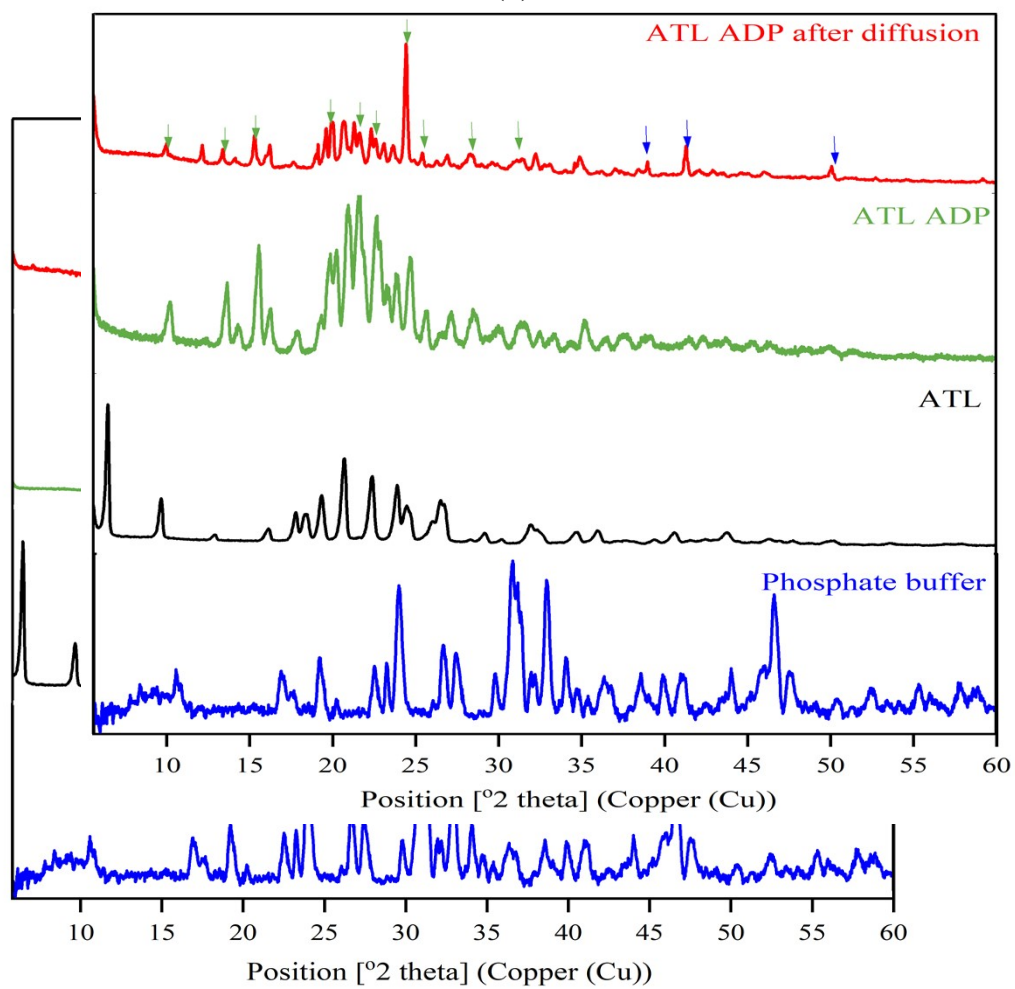
S6.

(f)

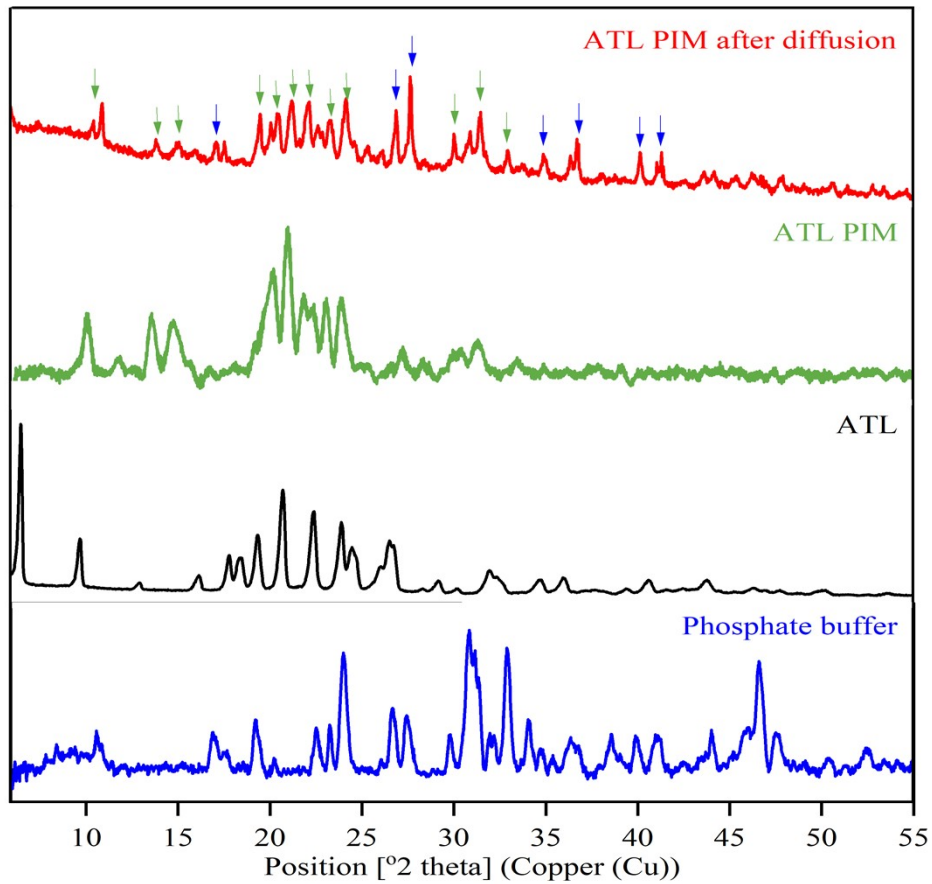
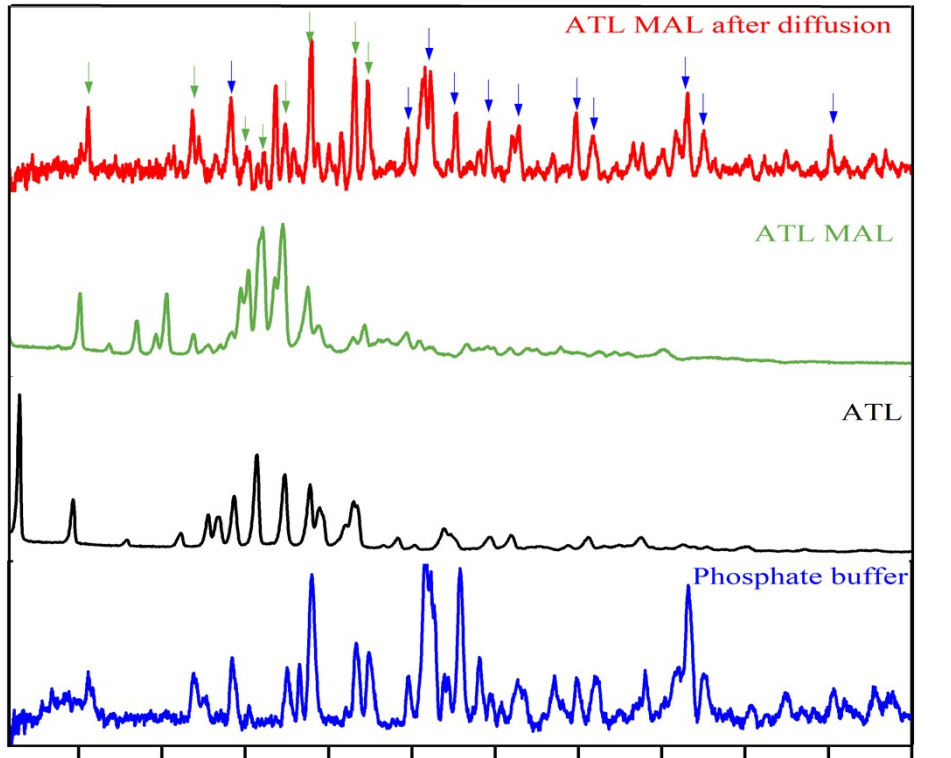
comparison of ATL salts after solubility experiment confirmed that except ATL-OXA salt, others were stable in the aqueous phase (a-g).



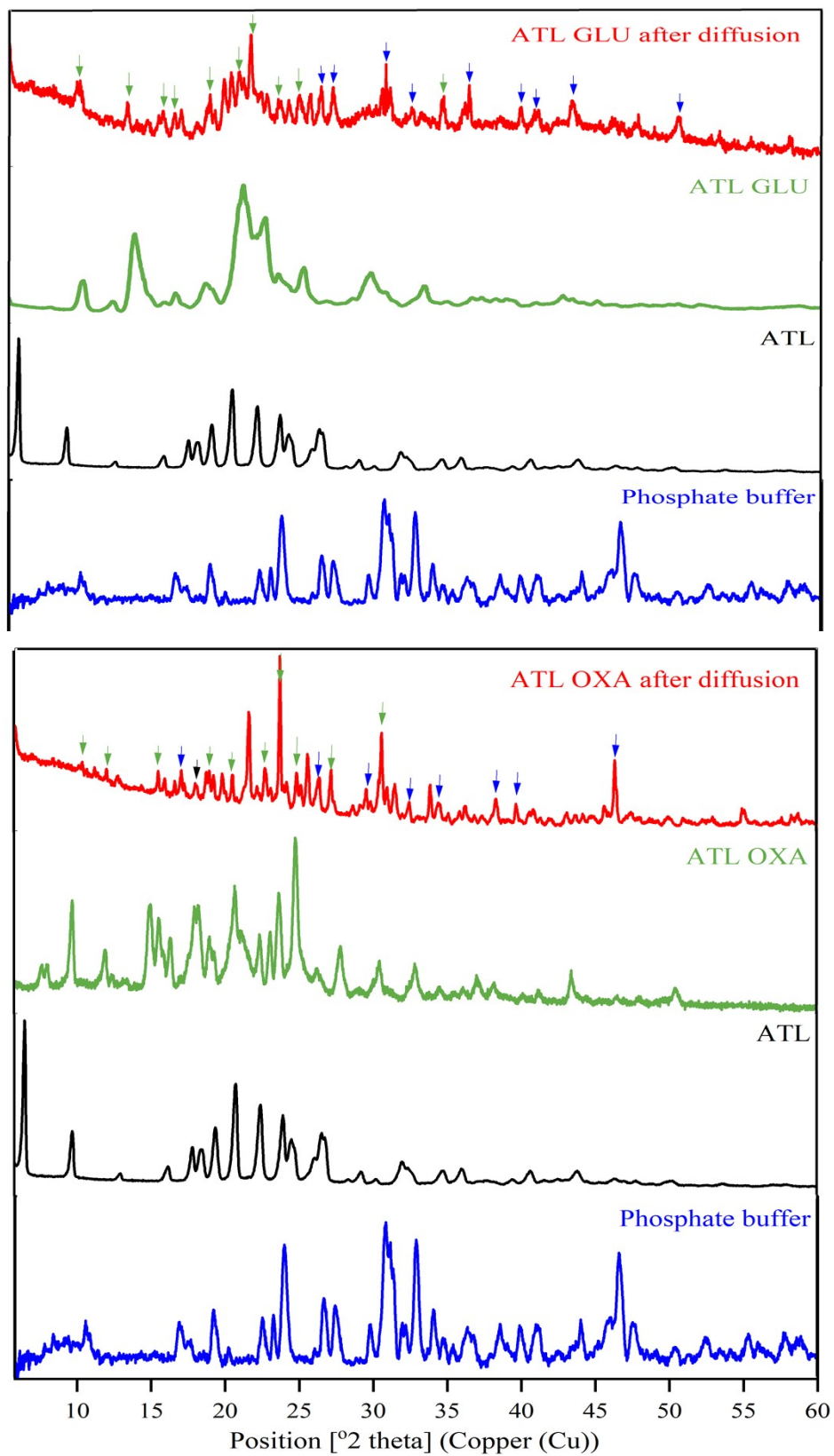
(a)



(c)

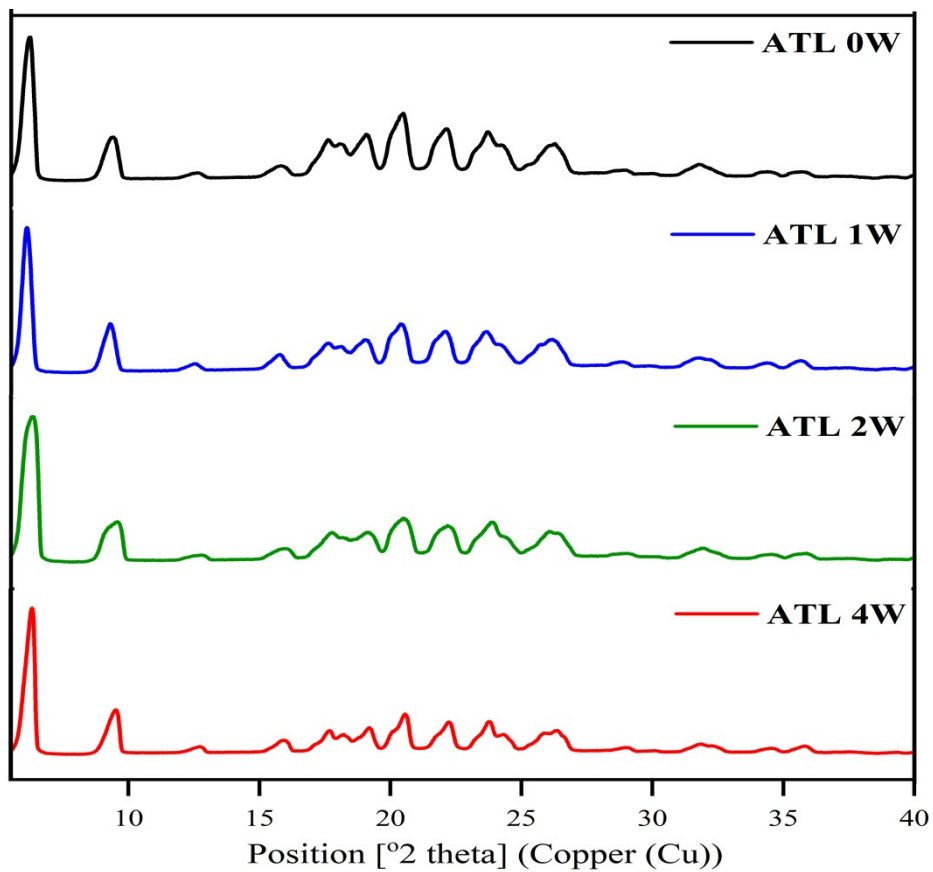


(e)

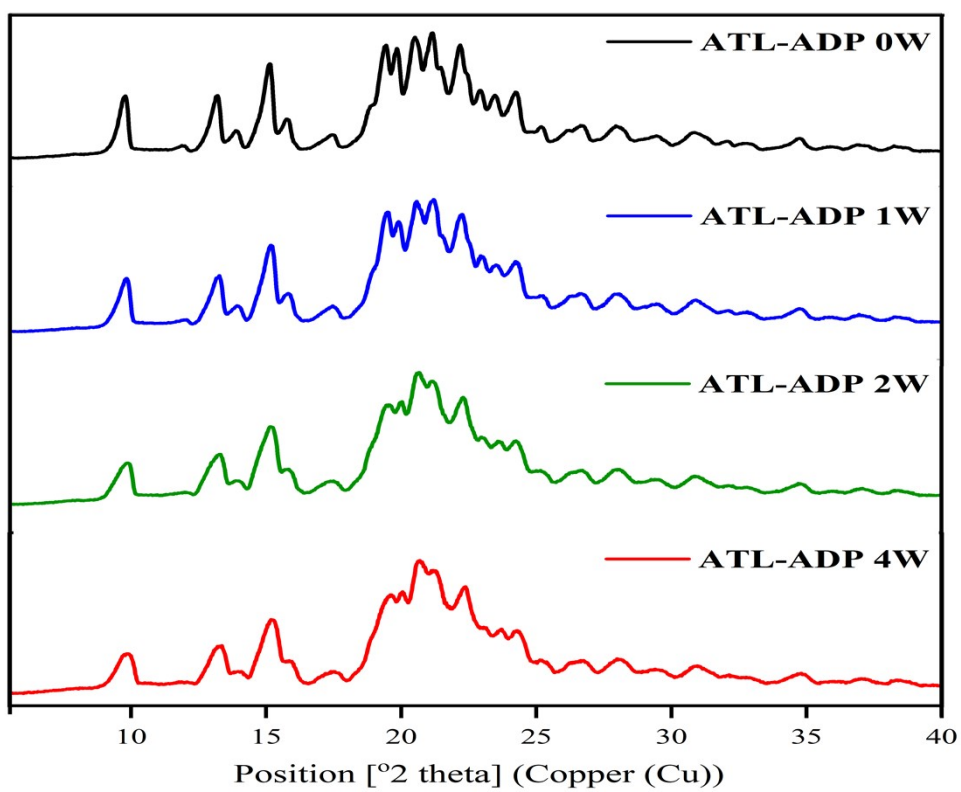


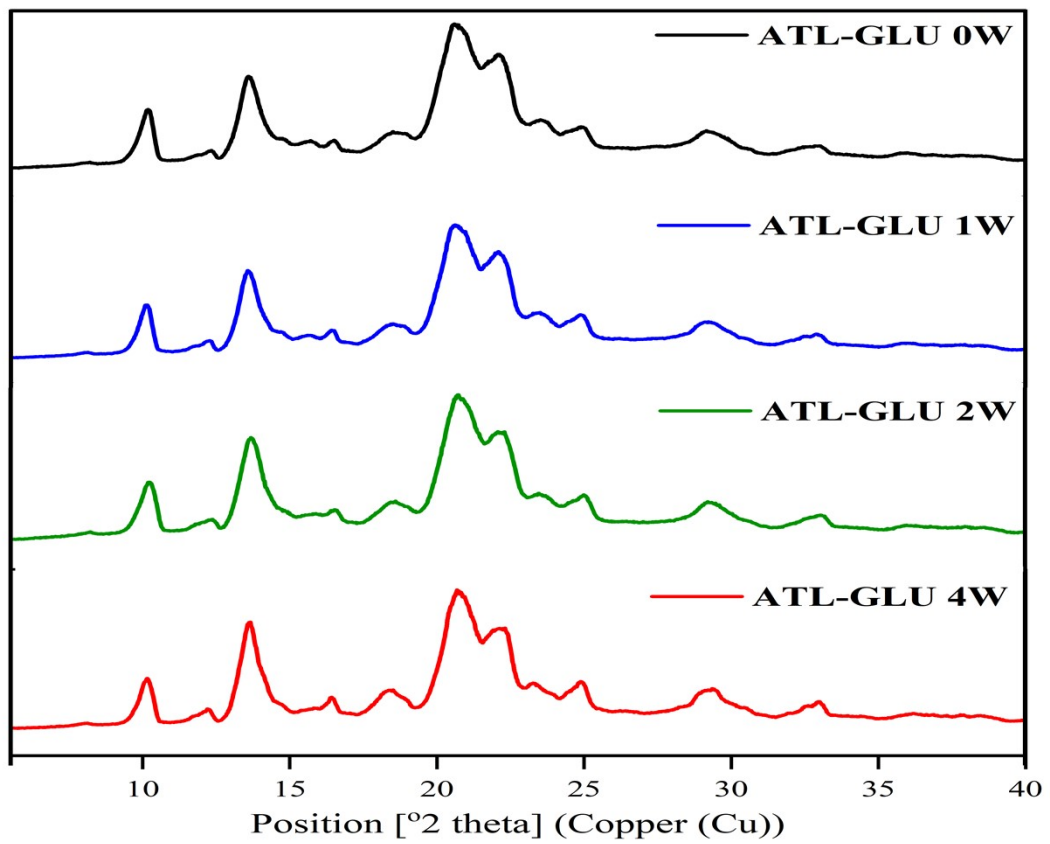
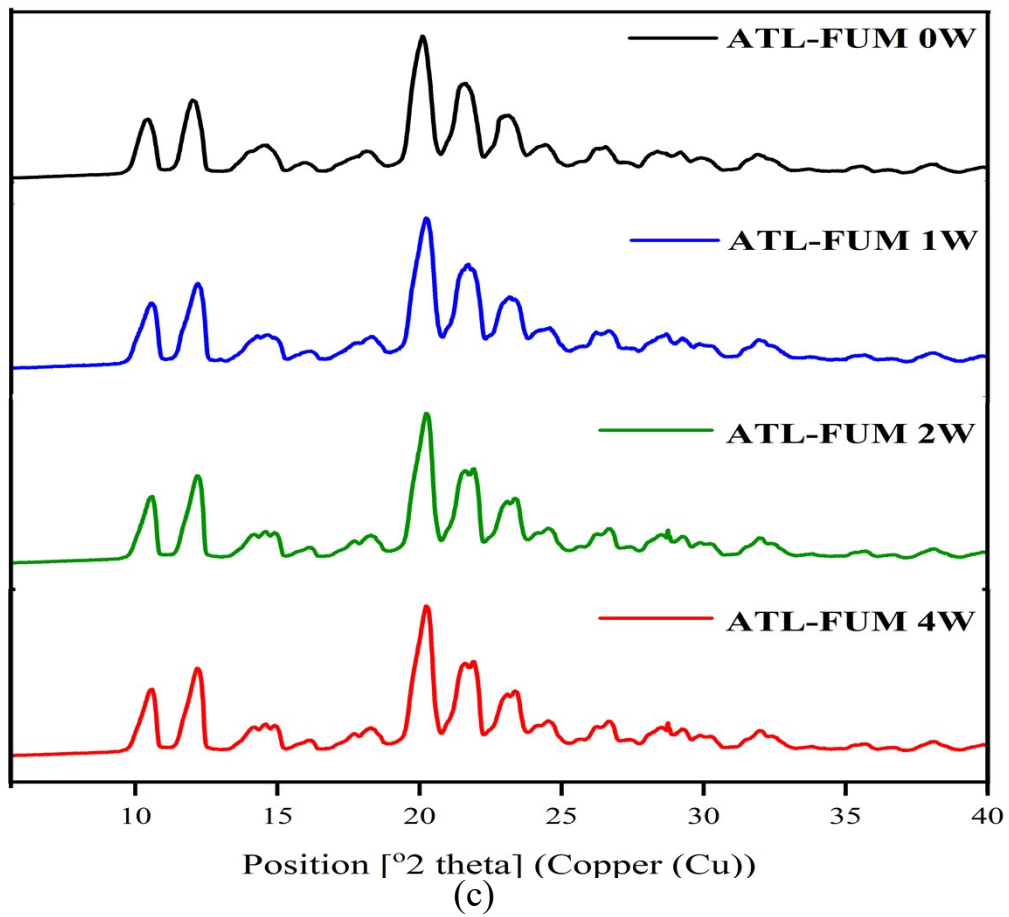
(g)

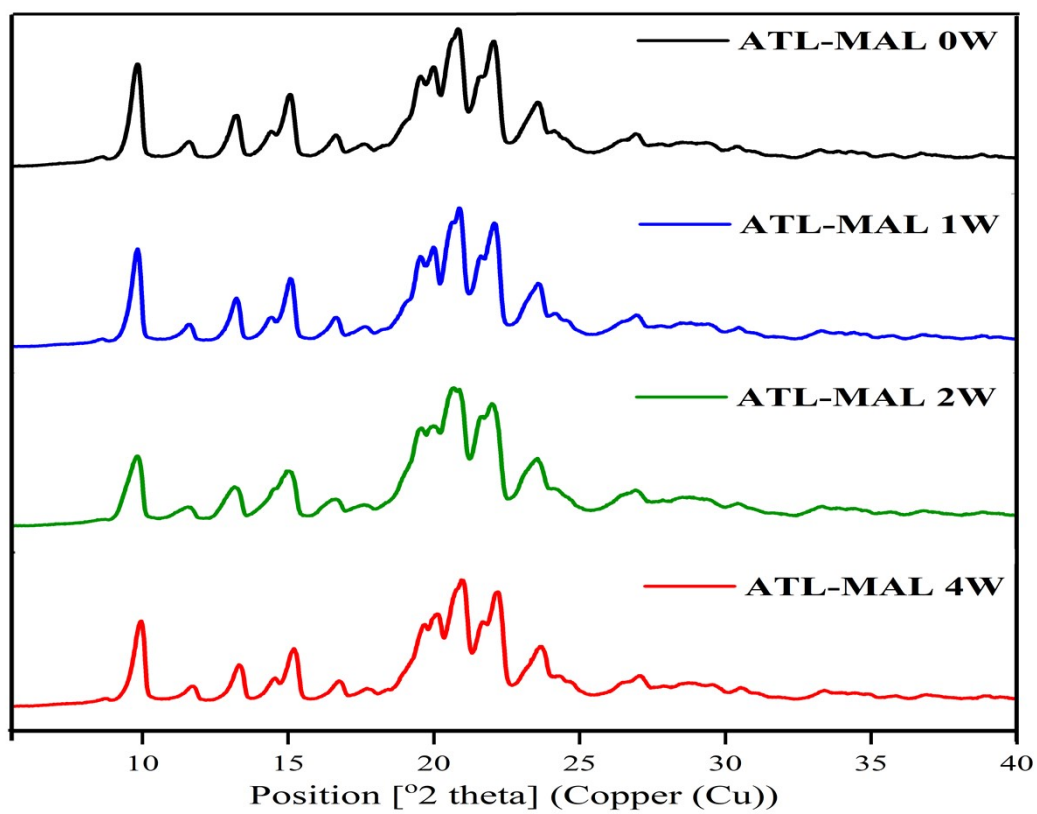
Figure S7. PXRD comparison of ATL and its salts (a-g) after 6h diffusion experiment. Note, additional diffraction peaks correspond to the phosphate buffer salts which appear along with the native drug.(green arrow corresponds to the salt peaks and blue arrow corresponds to the buffer peaks)



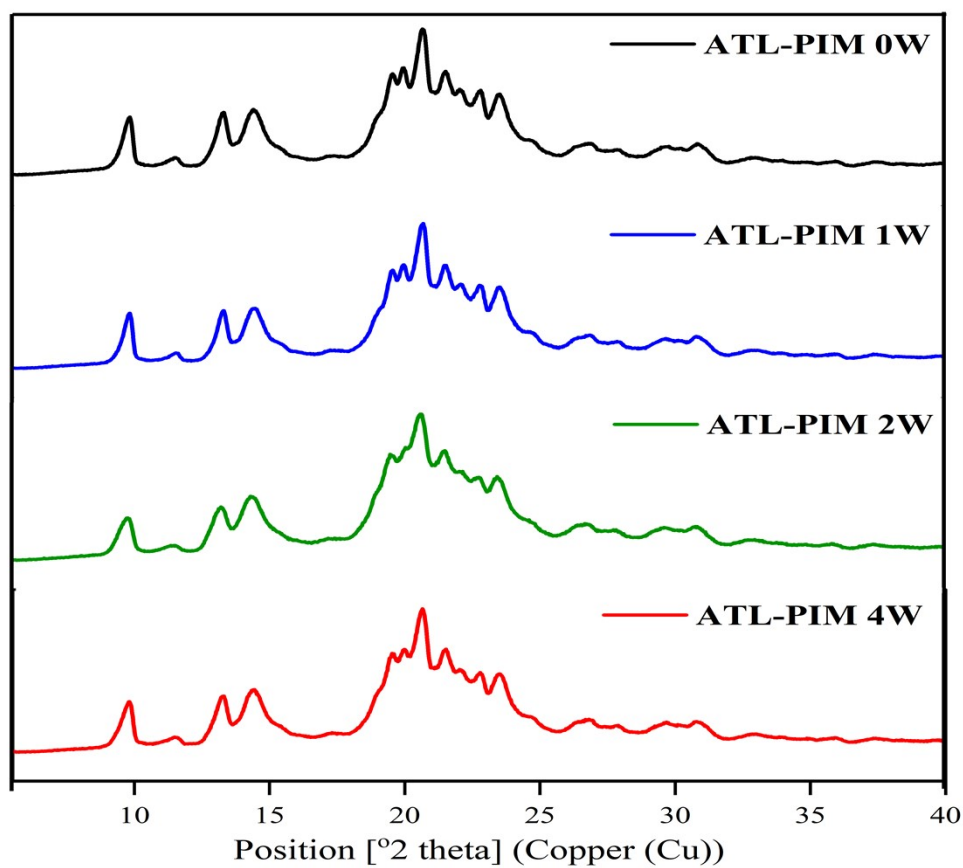
(a)







(e)



(f)

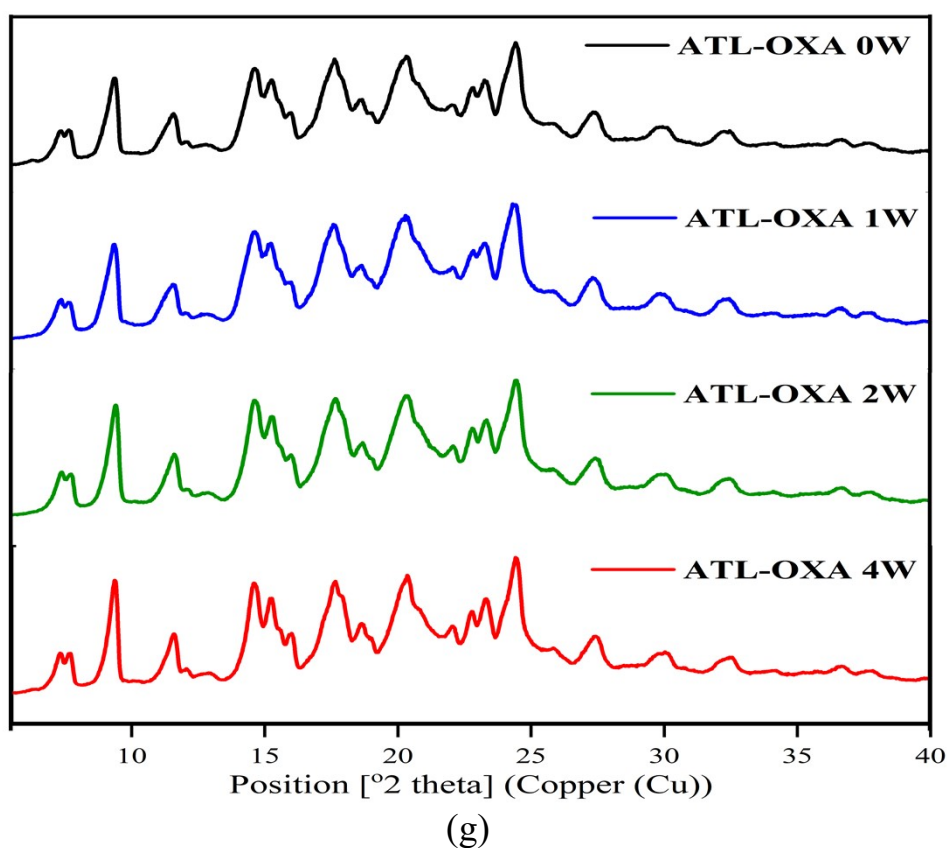
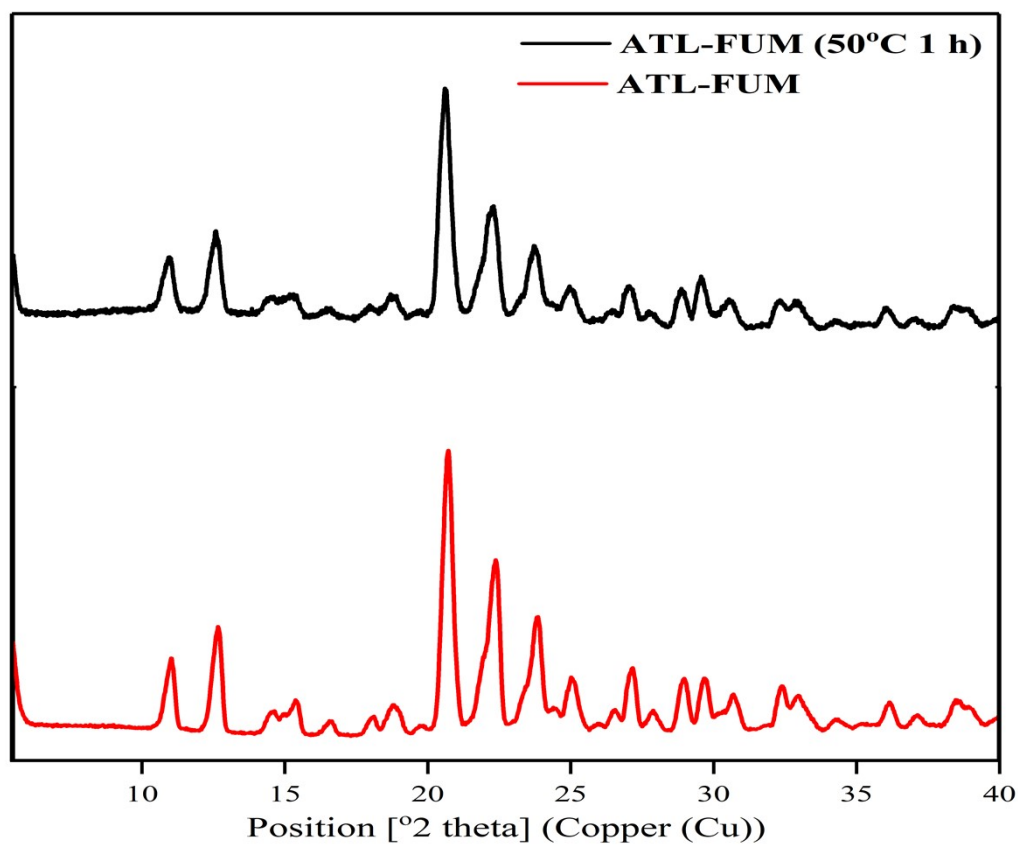
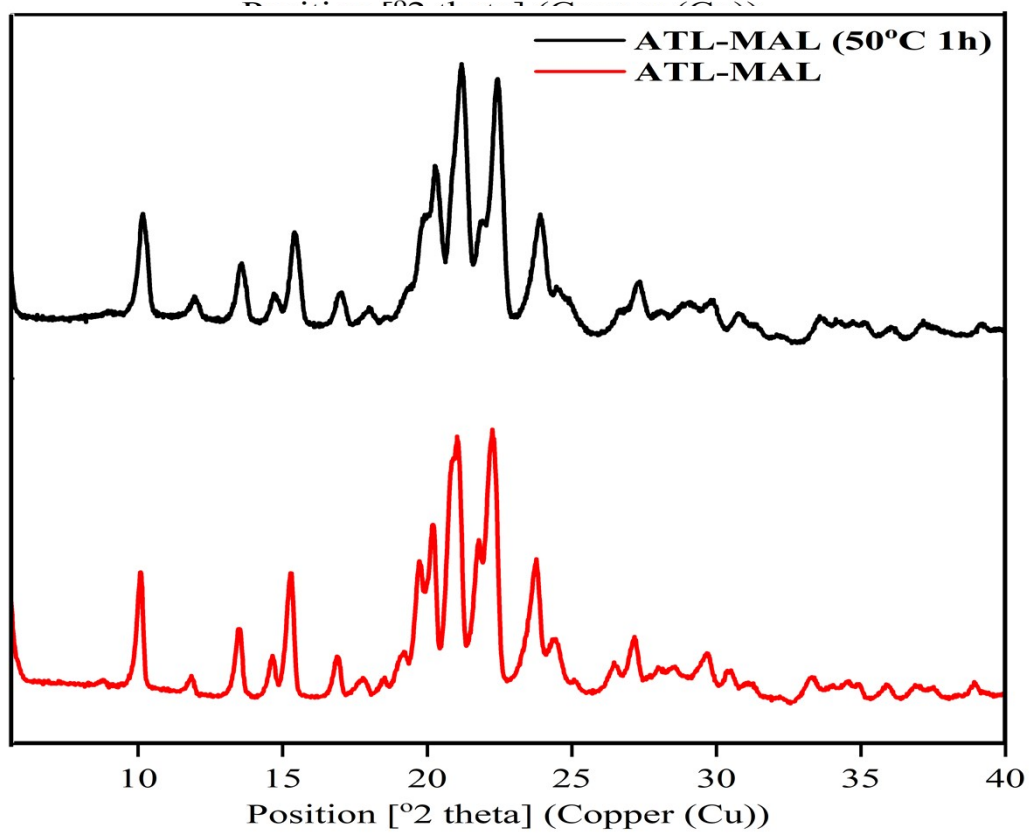
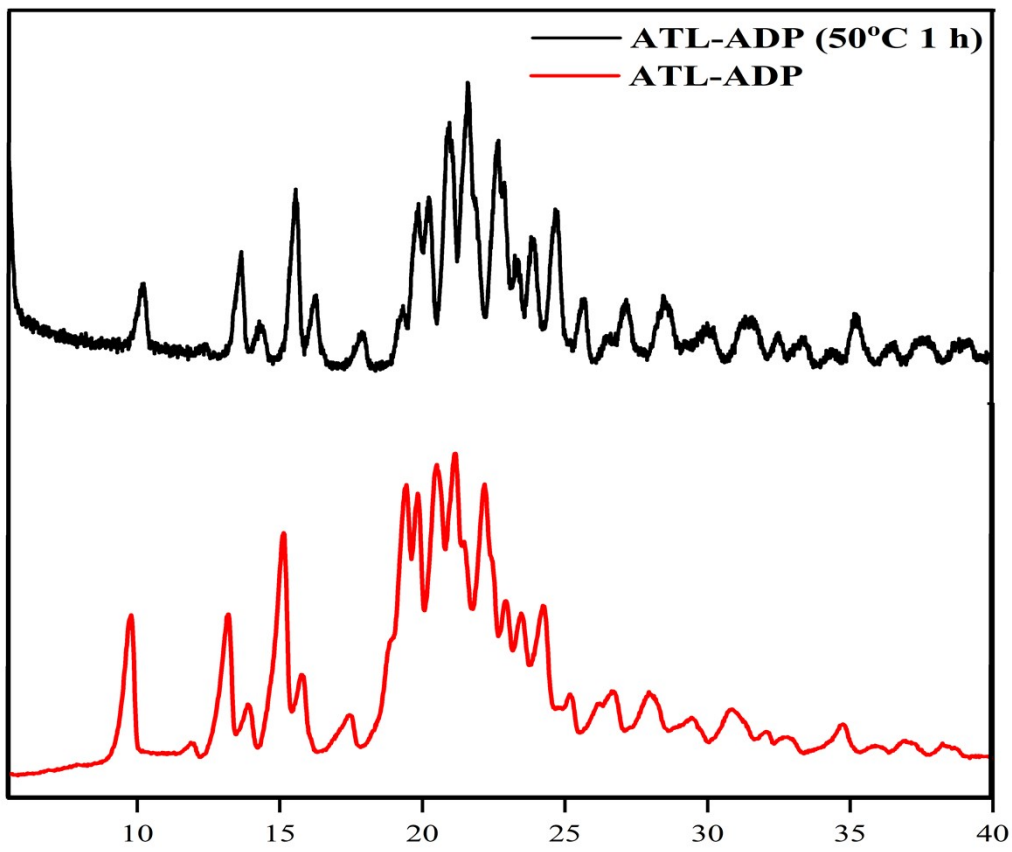
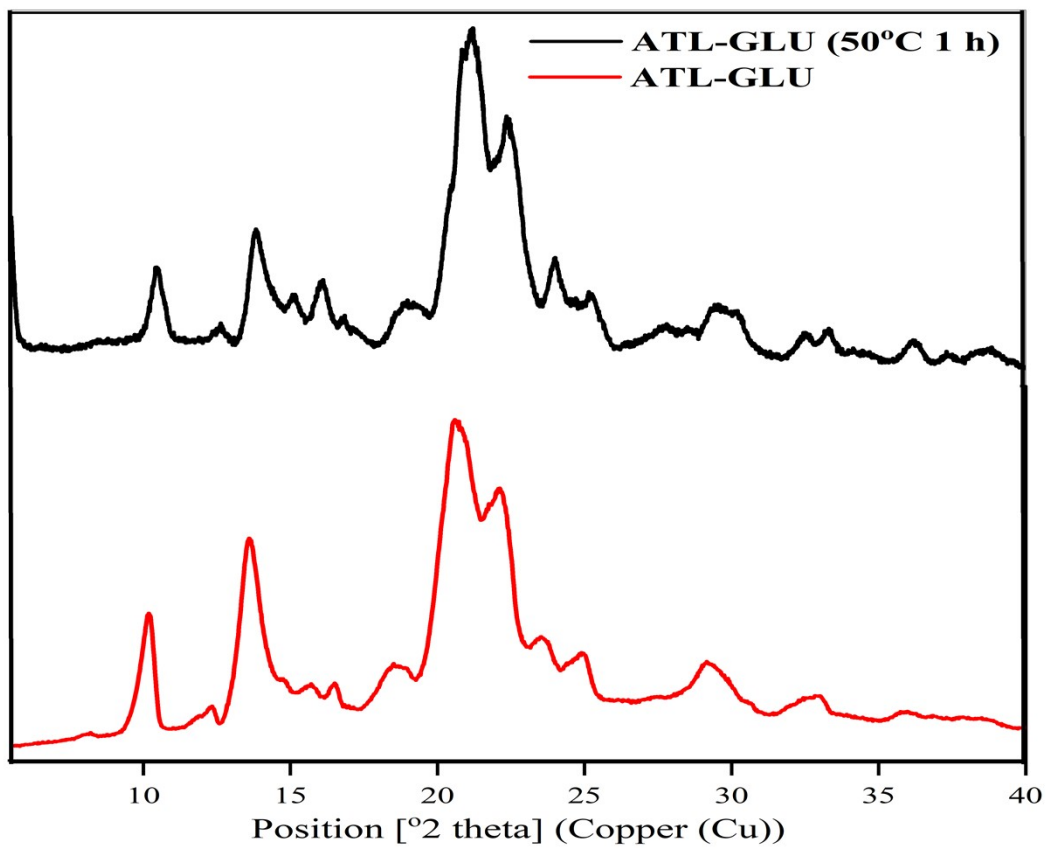


Figure S8. PXR D comparison of ATL and its salts exposed to 35 ± 5 °C and $75\pm 5\%$ relative

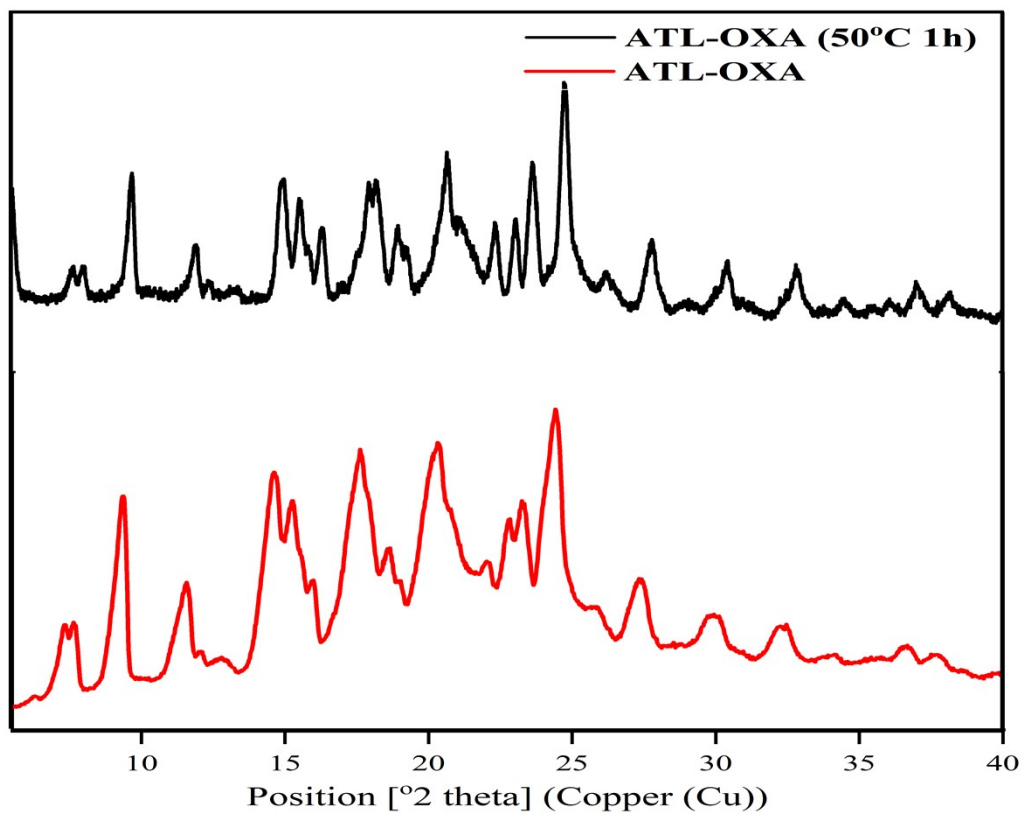
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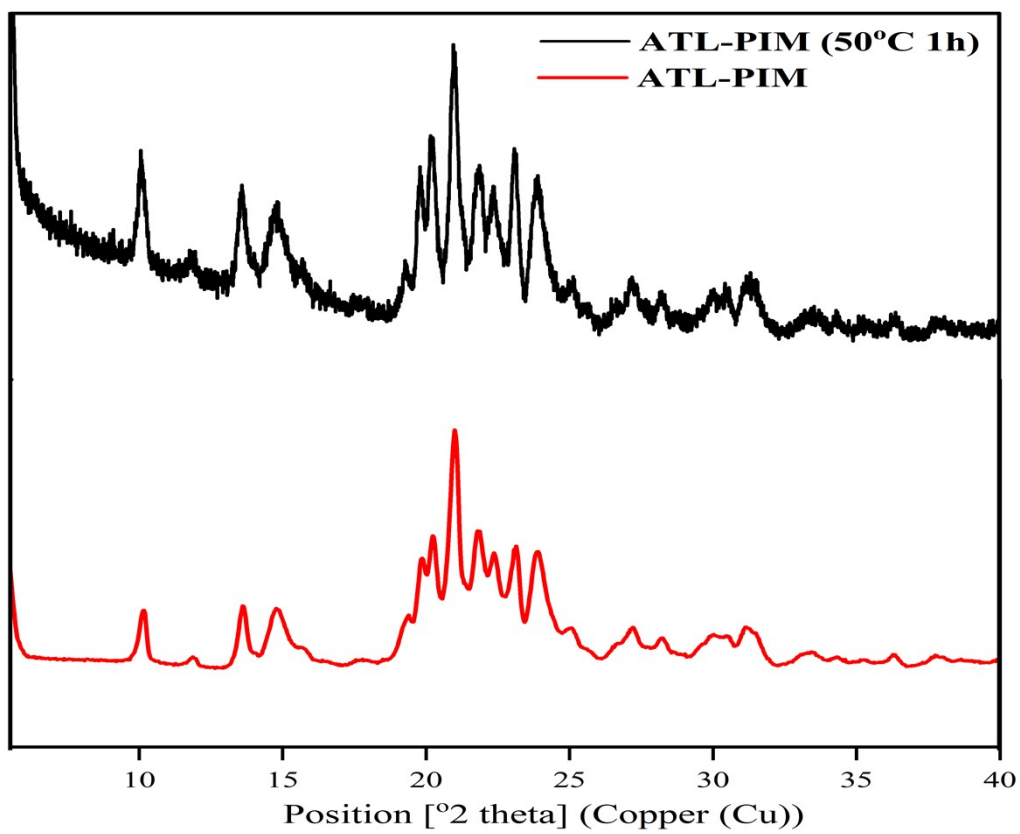




(d)



(e)



(f)

Figure S9. PXRD comparison of ATL salts at 50 °C for 1h confirms their thermal stability (a-f).