Supplementary Materials

## Coordination Polymers Constructed from Bis-pyridyl-bis-amide and 1,4,5,8-Naphthalenetetracarboxylic Acid: Ligand Transformation and Metal Sensing

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Fig. S1. PXRD patterns of complex 1.



Fig. S2. PXRD patterns of complex 2.



Fig. S3. PXRD patterns of complex 3.



Fig. S4. PXRD patterns of complex 4.



Fig. S5. PXRD patterns of complex 5a.



Fig. S6. PXRD patterns of complex 5b.



Fig. S7. PXRD patterns of complex 6.



Fig. S8. PXRD patterns of complex 7.



Fig. S9. TGA curve of complex 1.



Fig. S10. TGA curve of complex 2.



Fig. S11. TGA curve of complex 3.



Fig. S12. TGA curve of complex 4.



Fig. S13. TGA curve of complex 5a.



Fig. S14. TGA curve of complex 5b.



Fig. S15. TGA curve of complex 6.



Fig. S16. TGA curve of complex 7.







Fig. S18. PXRD patterns of complex 2 in different solvents





Fig. S19. PXRD patterns of complex 3 in different solvents





Fig. S20. PXRD patterns of complex 6 in different solvents



Fig. S21. PXRD patterns of complex 7 in different solvents



THF

Hexane

50

Experiment of 7 Simulated of 7

45



M

10

15

5

Munnethur

25

2 theta

30

35

40

20



Fig. S23. The emission and excitation spectra of  $L^4$  and  $H_4NTC$ .



Fig. S24. The emission and excitation spectra of complexes 1 and 2.



Fig. S25. The emission and excitation spectra of complexes 3 and 6.



Fig. S26. The emission and excitation spectra of complex 7.



Fig. S27. Emission spectra and the bar charts showing the percentages of quenching of complex 1 for triplicate experiments.



Fig. S28. Emission spectra and the bar charts showing the percentages of quenching of complex 2 for triplicate experiments.



Fig. S29. Emission spectra and the bar charts showing the percentages of quenching of complex 3 for triplicate experiments.



Fig. S30. Emission spectra and the bar charts showing the percentages of quenching of complex 6 for triplicate experiments.



**Fig. S31.** Emission spectra and the bar charts showing the percentages of quenching of complex 7 for triplicate experiments.



Fig. S32. PXRD patterns of complex 7 in different cationic solutions.



Fig. S33. A drawing showing the overlap of the excitation spectrum of complex 7 and the UV-vis spectrum of  $Fe^{3+}$ .



Fig. S34. The EDX data of complex 7 in  $Fe^{3+}$ .





The element percentage analysis of point scan.

Element	Weight%	Atomic%
C K	43.13	60.46
O K	31.57	33.23
Fe L	16.60	5.00
Cd L	8.70	1.30
Totals	100.00	100.00











**Fig. S37.** Binding energies fitting analysis for O1s electrons of complex 7 (a) before and (b) after immersing in Fe<sup>3+</sup>.



(a)



(b)

Fig. S38. The emission spectra of complex 7 in  $Fe^{3+}$  with different concentrations.



Fig. S39. PXRD patterns of complex 7 in  $Fe^{3+}$  solution for five cycles.







Fig. S41. Pore-size distribution curve for complex 1.



Fig. S42.  $N_2$  adsorption-desorption isotherms for complex 2.



Fig. S43. Pore-size distribution curve for complex 2.



Fig. S44.  $N_2$  adsorption-desorption isotherms for complex 3.



Fig. S45. Pore-size distribution curve for complex 3.



Fig. S46.  $N_2$  adsorption-desorption isotherms for complex 6.



Fig. S47. Pore-size distribution curve for complex 6.



Fig. S48. N<sub>2</sub> adsorption-desorption isotherms for complex 7.



Fig. S49. Pore-size distribution curve for complex 7.

