Electronic Supplementary Information

Benzoimidazole-2-one based macrocyclic arenes: Synthesis, and Solvent-induced structural changes

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1. Materials and Methods.

All reactions were carried out with oven-dried glassware. Commercial reagents were used without further purification. Flash column chromatography was performed on 100-200 mesh silica gel. ¹H NMR, ¹³C NMR spectra were recorded on a Bruker DMX400 NMR spectrometer. Melting points were determined using WRR melting point apparatus and were uncorrected. High Resolution atmospheric-pressure chemical ionization mass spectra (APCI-MS) were determined by Bruker Daltonics. Inc, APEX II. FT-ICRMS.

2. Synthesis of host 1 and 2.

To a mixture of 1,3-Dihydro-2H-benzimidazol-2-one (268.1 mg, 2.0 mmol) and 1,4-bis(bromomethyl)benzene (523.8 mg, 2.0 mmol) in *N*,*N*-dimethylformamide (100 mL) was added caesium carbonate (1.3 g, 4.0 mmol). The mixture was stirred at 110 °C for 24 h under N₂. Then the solvent was removed in vacuo and the residue was separated by column chromatography on silica gel (eluent: 80:1 DCM/Methanol) to give macrocyclic arenes **1** and **2**.

Host 1 (80.2 mg, 17%), ¹H NMR (400 MHz, Chloroform-*d*) δ 6.89 (dd, J = 5.8, 3.1 Hz, 1H), 6.76 (dd, J = 5.8, 3.2 Hz, 1H), 5.09 (s, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 154.3, 135.6, 129.0, 127.4, 121.4, 108.2, 44.3. HRMS (APCI) m/z: [M+H]⁺ calcd for C₃₀H₂₅N₄O₂, 473.1972; found, 473.1970.

Host **2** (108.6 mg, 23%), ¹H NMR (400 MHz, Chloroform-*d*) δ 7.35 – 7.26 (m, 12H), 6.96 (dt, J = 7.7, 3.3 Hz, 6H), 6.84 (ddd, J = 13.9, 6.0, 3.2 Hz, 6H), 5.14 – 4.99 (m, 12H). ¹³C NMR (101 MHz, CDCl₃) δ 154.5, 129.2, 127. 9, 127.8, 127.7, 127.4, 121.5, 108.4, 108.3, 65.0, 44.7, 44.6. HRMS (APCI) m/z: [M+H]⁺ calcd for C₄₅H₃₇N₆O₃, 709.2922; found, 709.2921.

3. ¹H NMR and ¹³C NMR Spectral of New compounds



Figure S2.¹³C NMR spectrum (101 MHz, CDCl₃, 298K) of 1



Figure S3.¹H NMR spectrum (400 MHz, CDCl₃, 298K) of 2



Figure S4.¹³C NMR spectrum (101 MHz, CDCl₃, 298K) of 2



Figure S5. Thermogravimetric analysis of $1 \bullet CHCl_3$, $1 \bullet 2CH_2Br_2$, $2 \bullet 3CHCl_3$ and $2 \bullet CH_2Cl_2$

4. Crystal structures data of 1•CHCl₃, 1•2CH₂Br₂, 2•3CHCl₃ and 2•CH₂Cl₂

	Crystal data	a and	structure	refinement	for	1•CHCl ₃
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Identification code	1 ●CHCl ₃
Empirical formula	$C_{31}H_{25}Cl_3N_4O_2$
Formula weight	591.90
Temperature/K	296.15
Crystal system	monoclinic
Space group	C2/c
a/Å	33.696(7)
b/Å	10.737(2)
c/Å	17.724(4)
α/°	90
β/°	115.638(3)
$\gamma/^{\circ}$	90

Volume/Å ³	5781(2)	
Ζ	8	
$\rho_{calc}g/cm^3$	1.360	
µ/mm ⁻¹	0.353	
F(000)	2448.0	
Crystal size/mm ³	0.1 imes 0.1 imes 0.1	
Radiation	MoKa ($\lambda = 0.71073$)	
20 range for data collection/° 2.682 to 55.064		
Index ranges	$-43 \le h \le 43, -13 \le k \le 13, -22 \le l \le 22$	
Reflections collected	31811	
Independent reflections	6561 [$R_{int} = 0.0783, R_{sigma} = 0.0757$]	
Data/restraints/parameters	6561/0/361	
Goodness-of-fit on F ²	0.998	
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0592, wR_2 = 0.1321$	
Final R indexes [all data]	$R_1 = 0.1518, wR_2 = 0.1697$	
Largest diff. peak/hole / e Å ⁻³ 0.31/-0.28		

Crystal data and structure refinement for 1•2CH₂Br₂.

Identification code	$1 \bullet 2 CH_2 Br_2$
Empirical formula	$C_{16}H_{14}Br_2N_2O$
Formula weight	410.11
Temperature/K	296.15
Crystal system	triclinic
Space group	P-1
a/Å	9.02(2)
b/Å	9.39(2)
c/Å	10.71(3)
α/°	84.17(3)
β/°	67.13(3)
$\gamma^{/\circ}$	66.09(3)
Volume/Å ³	763(3)

Z	2	
$\rho_{calc}g/cm^3$	1.786	
μ/mm^{-1}	5.314	
F(000)	404.0	
Crystal size/mm ³	0.1 imes 0.1 imes 0.1	
Radiation	MoKa ($\lambda = 0.71073$)	
20 range for data collection/°4.136 to 52.998		
Index ranges	$-11 \le h \le 11, -11 \le k \le 11, -13 \le l \le 13$	
Reflections collected	7846	
Independent reflections	3137 [$R_{int} = 0.0875$, $R_{sigma} = 0.1122$]	
Data/restraints/parameters	3137/0/242	
Goodness-of-fit on F ²	0.879	
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0444, wR_2 = 0.0802$	
Final R indexes [all data]	$R_1 = 0.1132, wR_2 = 0.0999$	
Largest diff. peak/hole / e Å ⁻³ 0.40/-0.40		

Crystal data and structure refinement for 2•3CHCl₃.

Identification code	2 ●3CHCl ₃
Empirical formula	$C_{48}H_{39}Cl_9N_6O_3$
Formula weight	1066.90
Temperature/K	296.15
Crystal system	triclinic
Space group	P-1
a/Å	13.234(7)
b/Å	13.892(7)
c/Å	16.543(9)
α/°	109.362(7)
β/°	96.339(7)
γ/°	102.534(7)
Volume/Å ³	2746(3)

Z	2	
$\rho_{calc}g/cm^3$	1.291	
µ/mm ⁻¹	0.502	
F(000)	1092.0	
Crystal size/mm ³	$0.15 \times 0.12 \times 0.11$	
Radiation	MoKa ($\lambda = 0.71073$)	
2 Θ range for data collection/° 2.662 to 50.054		
Index ranges	$-15 \le h \le 15, -16 \le k \le 16, -19 \le l \le 19$	
Reflections collected	26432	
Independent reflections	9686 [$R_{int} = 0.0934$, $R_{sigma} = 0.1304$]	
Data/restraints/parameters	9686/513/595	
Goodness-of-fit on F ²	0.984	
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0918, wR_2 = 0.2449$	
Final R indexes [all data]	$R_1 = 0.2085, wR_2 = 0.3105$	
Largest diff. peak/hole / e Å-3 0.59/-0.48		

Crystal data and structure refinement for $2\bullet CH_2Cl_2$.

Identification code	$2 \bullet CH_2Cl_2$
Empirical formula	$C_{46}H_{38}Cl_2N_6O_3$
Formula weight	793.72
Temperature/K	296.15
Crystal system	triclinic
Space group	P-1
a/Å	12.106(6)
b/Å	13.118(6)
c/Å	13.961(7)
$\alpha/^{\circ}$	76.807(6)
β/°	75.289(6)
γ/°	84.733(7)
Volume/Å ³	2086.7(17)

Z	2		
$\rho_{calc}g/cm^3$	1.263		
µ/mm ⁻¹	0.204		
F(000)	828.0		
Crystal size/mm ³	$0.14 \times 0.11 \times 0.09$		
Radiation	MoKa ($\lambda = 0.71073$)		
2 Θ range for data collection/° 3.086 to 50.054			
Index ranges	$-14 \le h \le 14, -15 \le k \le 15, -16 \le l \le 16$		
Reflections collected	20197		
Independent reflections	7357 [$R_{int} = 0.0684, R_{sigma} = 0.1096$]		
Data/restraints/parameters	7357/468/541		
Goodness-of-fit on F ²	1.176		
Final R indexes [I>= 2σ (I)]	$R_1 = 0.1270, wR_2 = 0.3448$		
Final R indexes [all data]	$R_1 = 0.2213, wR_2 = 0.4085$		
Largest diff. peak/hole / e Å-3 1.13/-0.32			