## **Supporting Information**

## Theoretical investigate of Janus $Ti_2BST$ (T = O, Se) monolayers as anode materials for Na/K-ion batteries

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Table S1 The relative energies per unit cell for different configurations in eV of  $Ti_2BSO$  and  $Ti_2BSSe$  monolayers.

MBene	Ι	II	III	IV
Ti <sub>2</sub> BSO	0	0.209	0.621	0.232
Ti <sub>2</sub> BSSe	0	0.536	0.440	0.132

**Table S2** The calculated lattice constant  $a_0$  (Å), bond length l (Å), and Bader charge  $\Delta Q$  (e) transferred from Ti to S (T) atom of Ti<sub>2</sub>BSO and Ti<sub>2</sub>BSSe monolayers.

Ti <sub>2</sub> BOT	$a_0$	$l_{\rm Ti-S}$	$l_{\text{Ti-T}}$	$\Delta Q_{\rm S}$	$\Delta Q_{\mathrm{T}}$
Ti <sub>2</sub> BSO	3.130	2.384	1.998	0.418	1.207
Ti <sub>2</sub> BSSe	3.259	2.380	2.541	0.542	0.314

**Table S3** The calculated elastic constants  $C_{ij}$  (N m<sup>-1</sup>) and Young's modulus Y (N m<sup>-1</sup>) of Ti<sub>2</sub>BSO and Ti<sub>2</sub>BSSe monolayers.

Ti <sub>2</sub> BOT	$C_{11} = C_{22}$	$C_{12}$	$C_{66}$	Y
Ti <sub>2</sub> BSO	132.21	26.88	52.66	126.74
Ti <sub>2</sub> BSSe	135.99	19.13	58.43	133.29

Table S4 Diffusion constants (D) of Na and K on two surfaces of Ti<sub>2</sub>BSO and Ti<sub>2</sub>BSSe monolayers.

Surface	$D_{ m Na}~( m cm^2~s^{-1})$	$D_{\rm K}({\rm cm}^2~{\rm s}^{-1})$
Ti <sub>2</sub> BSO-S	$2.028 \times 10^{-4}$	$2.889 \times 10^{-3}$
Ti <sub>2</sub> BSO-O	$8.569 \times 10^{-7}$	7.048 ×10 <sup>-5</sup>
Ti <sub>2</sub> BSSe-S	$1.169 \times 10^{-4}$	$7.778 \times 10^{-4}$
Ti <sub>2</sub> BSSe-Se	$1.933 \times 10^{-4}$	$1.237 \times 10^{-3}$



Fig. S1 Side and top views of K adsorbed (a)-(b) Ti<sub>2</sub>BSO and (c)-(d) Ti<sub>2</sub>BSSe monolayers.



**Fig. S2** The variations of free energy for (a)  $Na_4Ti_2BSO$  and (b)  $Na_4Ti_2BSSe$  (c)  $K_{3.11}Ti_2BSO$  (d)  $K_{3.11}Ti_2BSSe$  monolayers at 300 K during the AIMD simulations (insets are the final structures after simulations).