## Electronic Supplementary Information to

## Photoprocessing of cationic triazacoronene: Dissociation

 characteristics of polycyclic nitrogen heterocycles in interstellar environments[^0]This Electronic Supplementary Information contains further information to the main manuscript. First, a description on the synthethic procedures to obtain TAC is given. Next, additional figures are displayed in order of their reference in the main manuscript.

## Synthesis of 1,5,9-Triazacoronene

## Experimental

Chemical reagents were purchased from commercial suppliers and used without further purification. Flash column chromatography was performed on silica gel $60 \AA(0.04-0.063 \mathrm{~mm}$, Screening Devices B.V.). Thin Layer Chromatography (TLC) was performed on TLC Silica gel 60 plates (Kieselgel F254, Merck). NMR spectra was recorded on a Bruker AV-300 NMR or a Bruker AV-500 NMR instrument. Chemical shifts are given in ppm ( $\delta$ ) relative to the solvent signals. Mass spectrometry was performed on a Thermo Scientific LCQ Fleet spectrometer equipped with an electrospray ion source in positive ion mode. Elemental analysis was performed by Mikroanalytisches Labor Kolbe (c/o Fraunhofer Institut UMSICHT).

## Synthesis of 1,5,9-trinitrotriphenylene

To a solution of 2,3-dichloronitrobenzene ( $17.7 \mathrm{~g}, 92.2 \mathrm{mmol}, 1 \mathrm{eq}$ ) in DMF ( 142 ml ) was added copper powder ( $35.1 \mathrm{~g}, 553 \mathrm{mmol}, 6 \mathrm{eq}$ ). The mixture was stirred at reflux for 19 h under $\mathrm{N}_{2}$ atmosphere. After cooling to $120^{\circ} \mathrm{C}$, the hot solution was filtered over celite and the product was extracted from the celite with DMF $(3 \times 25 \mathrm{ml})$. The filtrate was then slowly poured into $6 \%$ ammonium solution ( 720 ml ) (diluted from $18 \% \mathrm{NH} 3 \cdot \mathrm{H}_{2} \mathrm{O}(240 \mathrm{ml})$ ) with vigorous stirring. The obtained black precipitate was collected by filtration and sequentially washed with diluted ammonium solution ( 48 ml ) and water ( 48 ml ). The obtained residue was then concentrated onto celite and purified using silica column chromatography (pentane : DCM, $2: 1 \rightarrow 1: 1$ ) to obtain a yellow solid ( $4.74 \mathrm{~g}, 13.0 \mathrm{mmol}, 42 \%$ ).
${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.21(\mathrm{dd}, \mathrm{J}=8.3,1.2 \mathrm{~Hz}, 3 \mathrm{H}), 7.82(\mathrm{dd}, \mathrm{J}=8.1,1.2 \mathrm{~Hz}, 3 \mathrm{H})$,
$7.60(\mathrm{t}, \mathrm{J}=8.2 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 148.51,135.26,135.00,130.70,130.26,123.70$.



## Synthesis of triphenylene-1,5,9-triamine

1,5,9-trinitrotriphenylene ( $3.0 \mathrm{~g}, 8.3 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in $\mathrm{THF} / \mathrm{MeOH}(4: 1, \mathrm{v} / \mathrm{v}$, $80 \mathrm{ml}) . \mathrm{Ni}(\mathrm{OAc})_{2} \cdot 4 \mathrm{H}_{2} \mathrm{O}(2.0 \mathrm{~g}, 8.3 \mathrm{mmol}, 1 \mathrm{eq})$ was added and the mixture was stirred for 5 min . Then, $\mathrm{NaBH}_{4}(4.7 \mathrm{~g}, 124 \mathrm{mmol}, 15 \mathrm{eq})$ was added in portions to the reaction mixture over 20 min and the obtained black mixture was stirred for an additional 15 min . Then, water was added ( 40 ml ) and the mixture was filtered. The obtained residue was additionally washed with $\mathrm{DCM} / \mathrm{MeOH}(200 \mathrm{ml})$, and the combined filtrate was evaporated under reduced pressure. Purification using silica column chromatography ( $0.1 \% \mathrm{MeOH}$ in DCM) yielded a light-brown solid ( $1.2 \mathrm{~g}, 4.4 \mathrm{mmol}, 52 \%$ ).
${ }^{1} \mathrm{H}_{\mathrm{NMR}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.31(\mathrm{dd}, \mathrm{J}=8.1,1.2 \mathrm{~Hz}, 3 \mathrm{H}), 7.25(\mathrm{t}, \mathrm{J}=7.9 \mathrm{~Hz}, 3 \mathrm{H}), 6.82$ (dd, J = 7.8, 1.2 Hz, 3H), 4.31 ( $\mathrm{s}, 6 \mathrm{H}$ ).
${ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 144.56,132.23,126.70,119.42,115.70,115.43$.

ESI-MS: m/z calcd for $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{3}[\mathrm{M}+\mathrm{H}]^{+}$274.13, found 274.08.



## Synthesis of $\mathbf{N}, \mathbf{N}^{\prime}, \mathbf{N}^{\prime \prime}$-(triphenylene-1,5,9-triyl)triformamide

Formic acid ( $0.37 \mathrm{ml}, 9.7 \mathrm{mmol}, 5.3 \mathrm{eq}$ ) was added to a solution of acetic anhydride ( 0.73 ml , $7.7 \mathrm{mmol}, 4.2 \mathrm{eq}$ ) at $0^{\circ} \mathrm{C}$. The mixture was heated to $50^{\circ} \mathrm{C}$ and stirred for 2 h . After cooling the prepared mixture back down to $0{ }^{\circ} \mathrm{C}$, a solution of the triphenylene-1,5,9-triamine ( $500 \mathrm{mg}, 1.8 \mathrm{mmol}, 1 \mathrm{eq}$ ) in THF ( 10 ml ) was added dropwise over 15 min . Then, the reaction mixture was allowed to warm to room temperature and stirred for 18 h . The resulting precipitate was filtered, rinsed with ethyl acetate, and dried to afford pure product as a gray solid ( $591 \mathrm{mg}, 1.7 \mathrm{mmol}, 90 \%$ ).
${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz}$, DMSO-d6) $\delta 10.43-10.29(\mathrm{~m}, 3 \mathrm{H}), 8.70-8.55(\mathrm{~m}, 3 \mathrm{H}), 8.43-8.32$
$(\mathrm{m}, 3 \mathrm{H}), 7.69-7.59(\mathrm{~m}, 2 \mathrm{H}), 7.59-7.48(\mathrm{~m}, 3 \mathrm{H}), 7.45(\mathrm{t}, \mathrm{J}=7.6 \mathrm{~Hz}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO-d6) $\delta 164.04,160.38$, 133.90, 132.74, 132.67, 130.24, 126.88, 126.67, 126.50, 125.78, 125.74, 125.58, 125.49, 124.18, 124.03, 123.85, 123.70.

ESI-MS: calcd for $\mathrm{C}_{21} \mathrm{H}_{16} \mathrm{~N}_{3} \mathrm{O}_{3}[\mathrm{M}+\mathrm{H}]^{+} 358.12$, found 358.08; $\mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{4} \mathrm{O}_{3}\left[\mathrm{M}+\mathrm{NH}_{4}\right]^{+} 375.15$, found 375.08.



## Synthesis of 1,5,9-triazocoronene

A microwave vial was charged with $\mathrm{N}, \mathrm{N}^{\prime}, \mathrm{N}^{\prime \prime}$-(Triphenylene-1,5,9-triyl)triformamide ( 0.50 g , $1.4 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{NaCl}(4.1 \mathrm{~g}, 31 \mathrm{mmol}, 22 \mathrm{eq})$ and $\mathrm{AlCl}_{3}(4.1 \mathrm{~g}, 94 \mathrm{mmol}, 67 \mathrm{eq})$ and purged with Ar. The mixture was heated to $230{ }^{\circ} \mathrm{C}$ and stirred for 4 h . After cooling to RT, the solids were stirred overnight in $15 \%$ aqueous $\mathrm{NaOH}(15 \mathrm{ml})$. The obtained yellow slurry was filtered over celite and washed with water $(50 \mathrm{ml})$. The product dispersed in celite was dissolved in copious amounts of $\mathrm{DCM} / \mathrm{MeOH}$ 10:1 until the filtrate was no longer yellow. The filtrate was evaporated under reduced pressure and washed with hot $\mathrm{DCM} / \mathrm{MeOH}$ 10:1 $(20 \mathrm{ml})$. The residue was collected by centrifugation to obtain product as brown-yellow solid ( $120 \mathrm{mg}, 0.40 \mathrm{mmol}, 23 \%$ ).
${ }^{1} \mathrm{H} \operatorname{NMR}(500 \mathrm{MHz}, \mathrm{TFA}) \delta 11.38(\mathrm{~s}, 3 \mathrm{H}), 10.12(\mathrm{~d}, \mathrm{~J}=8.9 \mathrm{~Hz}, 3 \mathrm{H}), 9.91(\mathrm{~d}, \mathrm{~J}=8.9 \mathrm{~Hz}$, $3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (126 MHz, TFA) $\delta 155.15,141.04,138.61,129.89,126.93,125.17,118.36$.
ESI-MS: $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{21} \mathrm{H}_{10} \mathrm{~N}_{3}[\mathrm{M}+\mathrm{H}]+304.09$, found 304.17.
Anal. Calcd for $\mathrm{C}_{21} \mathrm{H}_{9} \mathrm{~N}_{3}$ : C, 83.16; H, 2.99; N, 13.85. Found: C, 83.09; H, 3.02; N, 13.81 .




Figure S1: Top: Example of experimental data and fitted peaks used for integration. Bottom: Residuals between experiment and fit.


Figure S2: Mass spectrum of TAC ${ }^{+}$dissociation with a mix of $5 \%$ acetylene instead of pure helium. The water addition product ( $\mathrm{m} / \mathrm{z} 320$ ) is almost completely replaced by the acetylene addition product ( $m / z 328$ ).


Figure S3: Zoomed in view of the hydrogen loss region from TAC ${ }^{+}$. No significant hydrogen losses are observed.


Figure S4: Dissociation mass spectra for $m / z 320$. No dissociation is observed.


Figure S5: Integrated Intensities of all mass peaks in the $\Gamma_{9}$ region.


Figure S6: Integrated Intensities of all mass peaks in the $\Gamma_{10}$ region.


Figure S7: Integrated Intensities of all mass peaks in the $\Gamma_{11}$ region.


Figure S8: Integrated Intensities of all mass peaks in the $\Gamma_{12}$ region.


Figure S9: Integrated mass peaks for the $\Gamma_{10}$ region for Cor $^{+}$(top) and TAC $^{+}$(bottom).


Figure S10: Integrated mass peaks for the $\Gamma_{11}$ region for Cor $^{+}$(top) and TAC ${ }^{+}$(bottom).


Figure S11: Integrated mass peaks for the $\Gamma_{12}$ region for Cor $^{+}$(top) and TAC ${ }^{+}$(bottom).


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