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## **Supplementary Information**

# Thermally Activated Delayed Fluorescence Emitters for Efficient Sensitization of Europium (III)

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Ligands			Complex		
	<sup>19</sup> F δ (ppm)	<sup>31</sup> P δ (ppm)		<sup>19</sup> F δ (ppm)	<sup>31</sup> Ρ δ (ppm)
tta	-75.7	-	Eutta <sub>3</sub> .2H <sub>2</sub> O	-82.3	-
1L	-	28.38 – 28.26 (m)	1	-79.7	-76.5
2L	-	28.37 – 28.52 (m)	2	-79.5	-76.3
3L	-	28.58 – 28.43 (m)	3	-79.5	-75.9

Table S1: Comparison of the <sup>19</sup>F and <sup>31</sup>P chemical shift values of ligands and complexes



Figure S1.  $^{19}\mathrm{F}$  NMR spectra of Eu(III) coordination polymer 1 in  $C_6D_6$  at 25  $^{\circ}\mathrm{C}$ 



Figure S3.  $^{19}\text{F}$  NMR spectra of Eu(III) coordination polymer 2 in  $C_6D_6$  at 25 °C



Figure S4.  $^{31}\text{P}$  NMR spectra of Eu(III) coordination polymer 2 in  $C_6D_6$  at 25  $^{\circ}\text{C}$ 









Figure S7. ESI-MS of Eu(III) coordination polymer 1



Figure S8. ESI-MS of Eu(III) coordination polymer 2



Figure S9. ESI-MS of Eu(III) coordination polymer 3



Figure S10. IR spectra of Eu (III) coordination polymer 1 and corresponding ligands



Figure S11. IR spectra of Eu (III) coordination polymer 2 and corresponding ligands



Figure S12. IR spectra of Eu (III) coordination polymer 3 and corresponding ligands



Figure S13. Thermo-Gravimetric Analysis (TGA) of Eu (III) coordination polymer 1, 2 and 3

### **Optical Properties**



**Figure S14**. UV-Vis absorption and luminescence spectra of TADF ligands (SFX-PO-DPA is 1L, SFX-PO-DPA-Me is 2L and SFX-PO-DPA-OMe is 3L) in toluene solution (Adapted from ref 1, Copyright, 2021, American Chemical Society).<sup>1</sup>



Figure S15. Comparison of UV-Vis absorption spectra of tta, TADF ligands (1L, 2L, and 3L) and corresponding Eu(III) coordination polymers (1, 2 and 3) in toluene solution



Figure S16. Excitation and luminescence spectra of (a) 1 (b) 2 in toluene solution ( $\lambda_{ex}$ = 340 nm and  $\lambda_{em}$ = 611 nm)



Figure S17. Luminescence decay profiles of complexes 1 and 2 in toluene solution ( $\lambda_{ex}$ =340 and  $\lambda_{em}$ =611 nm)



Figure S18. UV-Vis absorption spectra of powder samples of 1-3





Figure S19. Excitation and luminescence spectra of (a) 1 (b) 2 powder samples ( $\lambda_{ex}$ = 340 nm and  $\lambda_{em}$ = 611 nm)





Figure S20. Luminescence decay profiles of powder samples of complexes 1 and 2 ( $\lambda_{ex}$ =340 and  $\lambda_{em}$ =611 nm)



Figure S21. Absolute luminescence quantum yield measurements of solid samples of 1 ( $\lambda_{ex}$ =340)



Figure S22. Absolute luminescence quantum yield measurements of solid samples of 2 ( $\lambda_{ex}$ =340)



Figure S23. Absolute luminescence quantum yield measurements of solid samples of 3 ( $\lambda_{ex}$ =340)



Figure S24. UV-Vis absorption spectra of PMMA encapsulated films of 1-3





**Figure S25**. Excitation and luminescence spectra of PMMA encapsulated films (a) **1** and (b) **3** ( $\lambda_{ex}$ = 340 nm and  $\lambda_{em}$ = 611 nm)





Figure S26. Luminescence decay profiles of complexes 1 and 2 in PMMA films ( $\lambda_{ex}$ =340 and  $\lambda_{em}$ =611 nm)



Figure S27. Absolute luminescence quantum yield measurements of PMMA encapsulated films of 1 ( $\lambda_{ex}$ =340)



Figure S28. Absolute luminescence quantum yield measurements of PMMA encapsulated films of 2 ( $\lambda_{ex}$ =340)



Figure S29. Absolute luminescence quantum yield measurements of PMMA encapsulated films of 3 ( $\lambda_{ex}$ =340)



Figure S30. Luminescence spectra of PMMA encapsulated films of 1 at different excitations in the region 300-450 nm





Figure S31. Luminescence spectra of PMMA encapsulated films (a) 1 (b) 2 and (c) 3 ( $\lambda_{ex}$ = 400 nm)





Figure S32. Luminescence decay profiles of 1-3 in PMMA encapsulated films ( $\lambda_{ex}$ =400 and  $\lambda_{em}$ =611 nm)



Figure S33. Absolute luminescence quantum yield measurements of PMMA encapsulated films of 1 ( $\lambda_{ex}$ =400)



Figure S34. Absolute luminescence quantum yield measurements of PMMA encapsulated films of 2 ( $\lambda_{ex}$ =400)



Figure S35. Absolute luminescence quantum yield measurements of PMMA encapsulated films of 3 ( $\lambda_{ex}$ =400)

	$\tau_{Ln}(ms)^a$	$\tau_R(ms)^b$	$\Phi_{\mathrm{Ln}}(\%)^{\mathrm{c}}$	$\Phi_{tot}$ (%) <sup>d</sup>	$\eta_{\rm sen}$ (%) <sup>e</sup>	$k_{\rm r}   ({\rm s}^{-1})^{{ m f}}$	$k_{\rm nr}~({\rm s}^{-1})^{\rm g}$
1	0.52	1.14	90	38	42	0.9×10 <sup>3</sup>	1.1×10 <sup>3</sup>
2	0.56	1.17	88	36	41	0.9×10 <sup>3</sup>	$0.9 \times 10^{3}$
3	0.52	1.16	91	39	43	0.9×10 <sup>3</sup>	1.1×10 <sup>3</sup>

Table S2. Photophysical properties of 1–3 ( $\lambda_{ex}$ = 400 nm and  $\lambda_{em}$  = 611 nm).

<sup>a</sup>Lanthanide luminescence lifetime obtained from TRPL spectra. <sup>b</sup>radiative lifetime  $\tau_{R_c}$ <sup>C</sup>the intrinsic luminescence quantum yield  $\Phi_{Ln} = \tau_{Ln}/\tau_{R_r}$ , <sup>d</sup>total luminescence quantum yields ( $\Phi_{tot}$ ), <sup>e</sup>sensitization efficiency  $\eta_{sen} = \Phi_{tot} / \Phi_{Ln}$ , <sup>f</sup>radiative decay rate constant  $k_r = 1/\tau_R$  and <sup>g</sup>non-radiative decay rate constant,  $k_{nr} = (\tau_R - \tau_{Ln}) / \tau_R \tau_{Ln}$ .<sup>23</sup>



**Figure S36**. The mechanistic energy transfer pathways in the tta and TADF-ligand (1L) sensitized Eu(III) luminescence of coordination polymer 1 (energy levels are not up to the scale).



Figure S37. The mechanistic energy transfer pathways in the tta and TADF-ligand (2L) sensitized Eu(III) luminescence of coordination polymer 2 (energy levels are not up to the scale).

#### References

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