

Supplementary Material

for

Unraveling the Orientation of an Enzyme onto Metal–Organic Framework

Zhiyong Xu, Jian Zhou*

School of Chemistry and Chemical Engineering, Guangdong Provincial Key Lab for

Green Chemical Product Technology, South China University of Technology,

Guangzhou 510640, P. R. China.

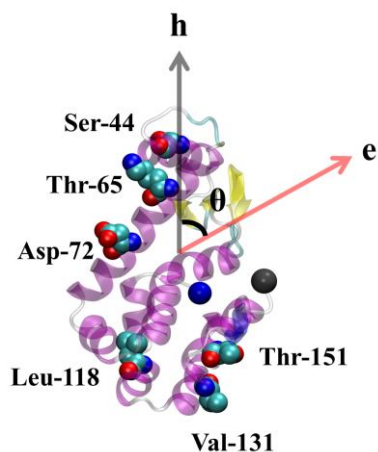


Fig. S1 The crystal structure of T4L, 6 marker sites consistent with the experimental reports are shown in vdW spheres; N-terminal and C-terminal are displayed in blue and black beads, respectively. The grey vector h and pink vector e are the unit vectors in the direction of ZIF-8 surface normal and the electrostatic dipole of the T4L

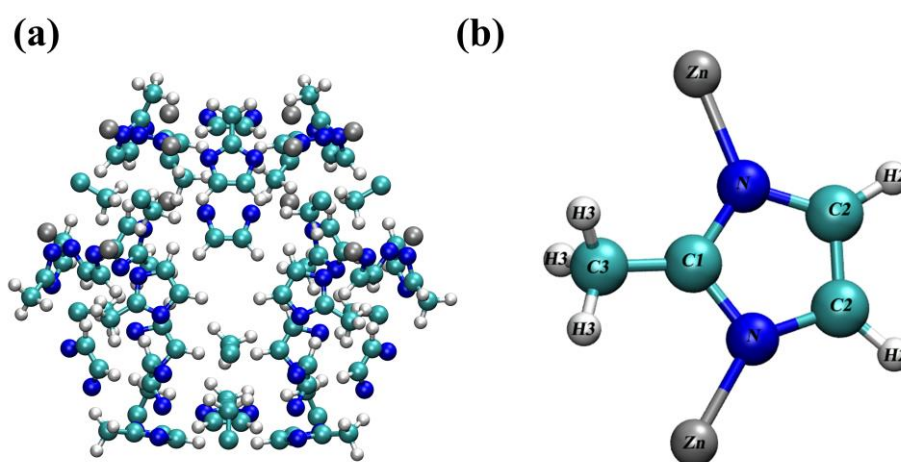


Fig. S2 The unit cell of ZIF-8 (a) and the cluster models for the 2-methylimidazolate (mIm) linker (b)

Table S1. The types and charges of atoms in ZIF-8.

Atom type	Charge	ϵ (kJ/mol)	σ (nm)
Zn	0.7599	0.196	0.196
N	-0.4312	0.325	0.325
C1	0.4280	0.340	0.340
C2	-0.0571	0.340	0.340
C3	-0.4676	0.340	0.340
H2	0.1083	0.251	0.251
H3	0.1399	0.265	0.265

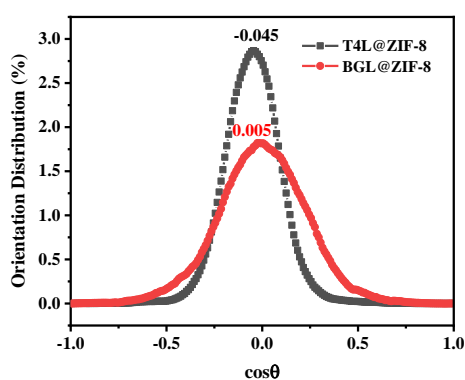


Fig. S3 Adsorption orientation distributions of T4L and BGL on ZIF-8 by MD simulations

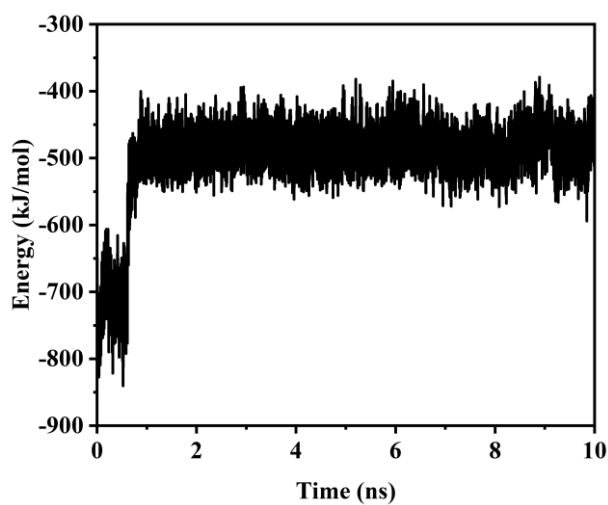


Fig. S4 Interaction energy between BGL and ZIF-8 in a random umbrella window

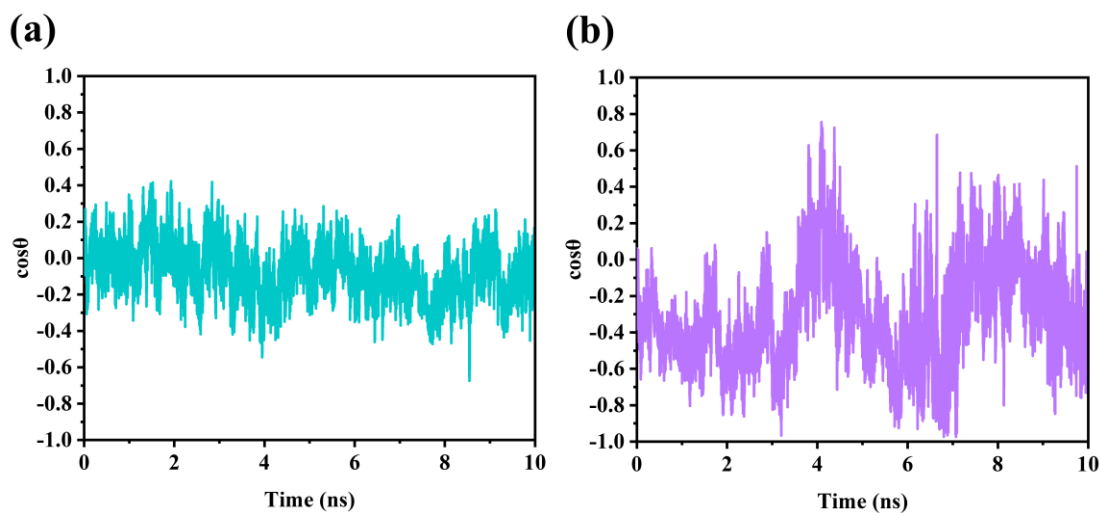


Fig. S5 Time evolutions of $\cos\theta$ between BGL and ZIF-8 among two representative umbrella windows, the COM distance between BGL and ZIF-8 is about 35 Å (a) and about 70 Å (b)

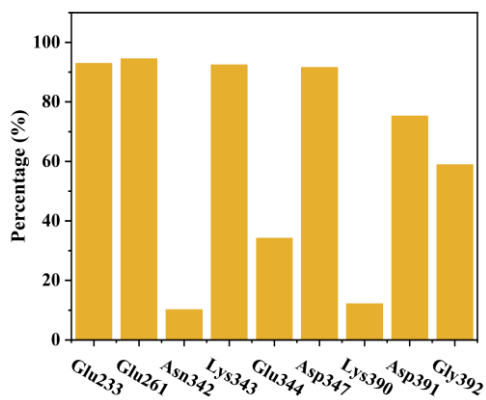
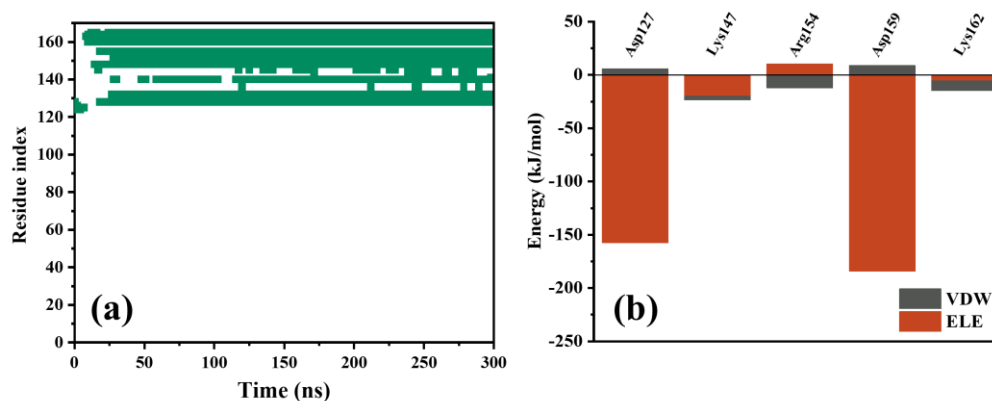


Fig. S6 Proportion of the trajectory each of the key binding residues is in contact with the ZIF-8 surface

Table S2. Adsorption orientations of T4L and BGL obtained by different methods

Proteins \ Methods	T4L	BGL
ProtPOS	-0.047	0.199
MD	-0.045	0.005

**Fig. S7** Residue contact maps of T4L adsorbed on the ZIF-8 surface (a); averaged interaction energy of an individual key residue with the ZIF-8 surface (b)

Residues Asp127, Lys147, Arg154, Asp159 and Lys162 are the key binding sites for T4L on the ZIF-8 surface. After about 30 ns simulation time, they are all stably adsorbed on the ZIF-8 surface. Through negatively charged carboxyl groups, residues Asp127 and Asp159 interact with Zn^{2+} of ZIF-8 (Figure S5a), contribute strong electrostatic energy around -157.35 and -184.26 kJ/mol, respectively; while the positively charged residues Lys147, Arg154 and Lys162 bind with the ZIF-8 surface through relatively weak vdW interactions, with their interaction energies less than -12 kJ/mol. Hence, electrostatic interaction is the dominant driving force for T4L adsorbed on the ZIF-8.

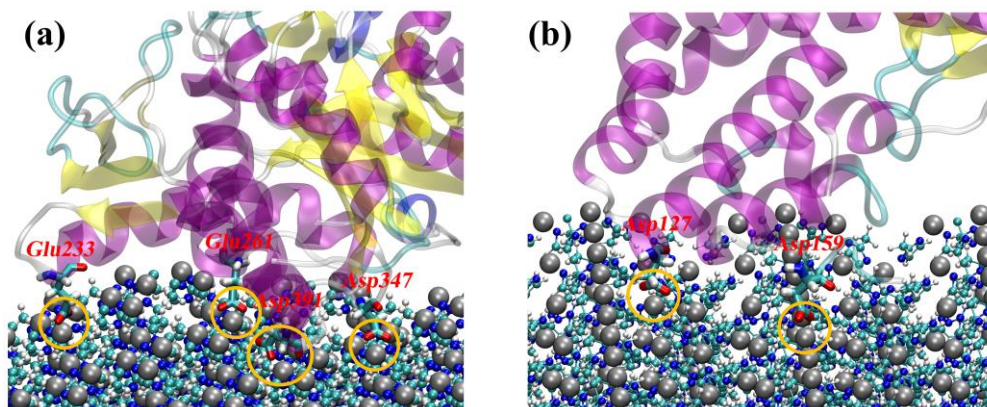


Fig. S8 The binding between deprotonated carboxyl group of acidic amino acids and the Zn^{2+} of ZIF-8, BGL@ZIF-8 system (a) and T4L@ZIF-8 system (b)

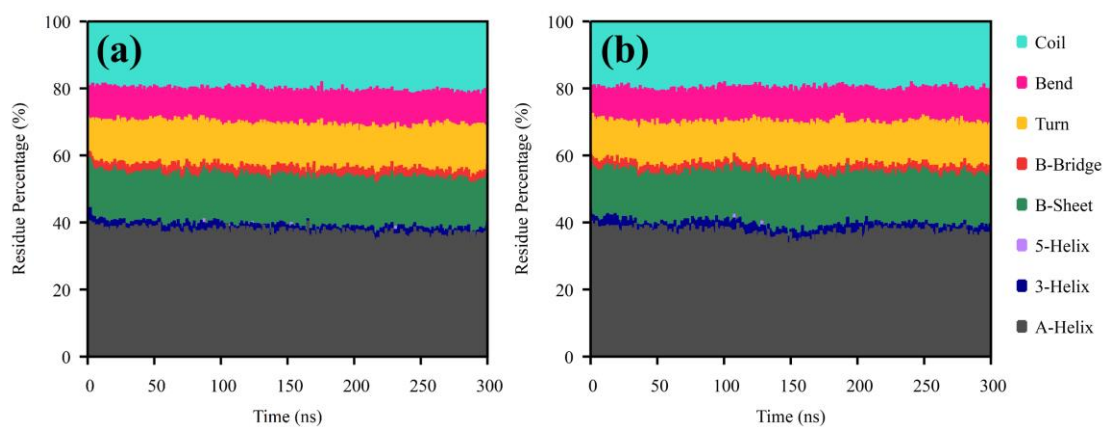


Fig. S9 Average contents of each secondary structure of the BGL for different systems, bulk solution (a) and BGL@ZIF-8 (b)