

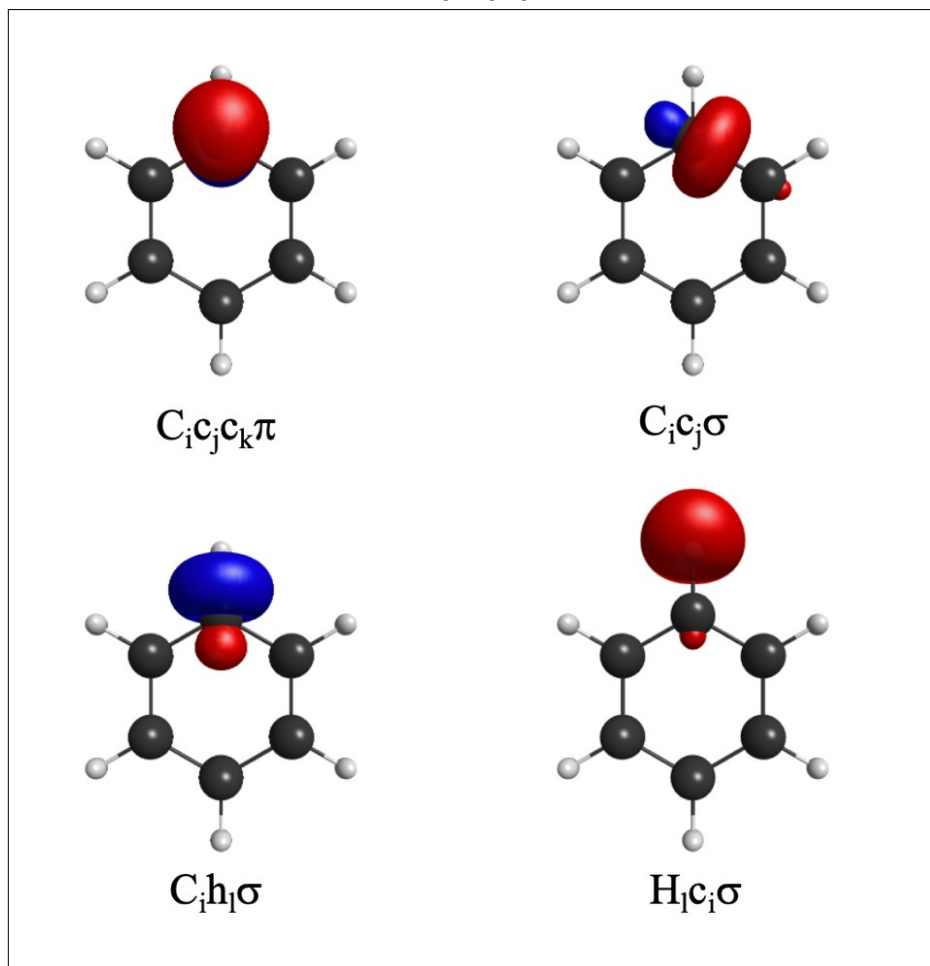
## Analysis of Bonding Motifs in Unusual Molecules: Planar Hexacoordinated Carbon and Infitene

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### Supporting Information

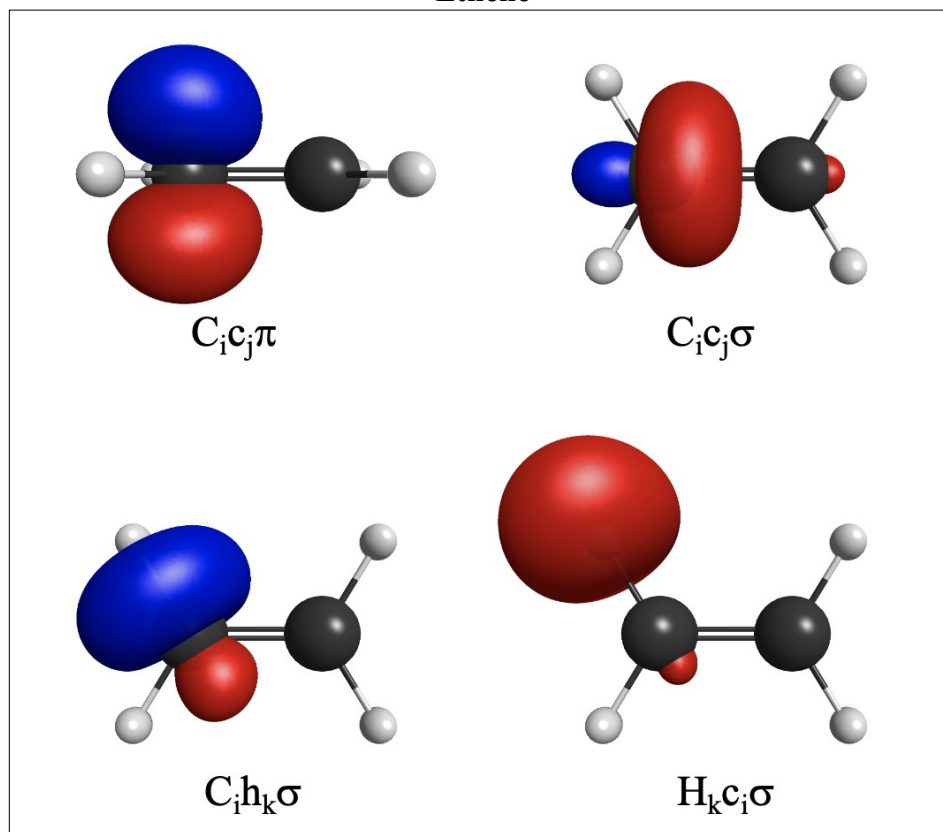
Optimized geometries of all relevant molecules are given in .xyz format in a .zip folder.

#### Benzene



SI Figure SI:1. Symmetric unique  $\sigma$  and  $\pi$ - QAOs in benzene. Subscripted letters,  $i, j, k$ , and  $l$  correspond to different atoms. The structure of benzene was optimized at the RHF/6-31G(d) level of theory. Bond lengths for CC = 1.39 Å and CH = 1.08 Å.

### Ethene



SI Figure SI:2. Symmetric unique  $\sigma$  and  $\pi$ - QUAOs in ethene. Subscripted letters,  $i, j$ , and  $k$  correspond to different atoms. The structure of ethene was optimized at the RHF/6-31G(d) level of theory. Bond lengths for CC = 1.34 Å and CH= 1.09 Å.

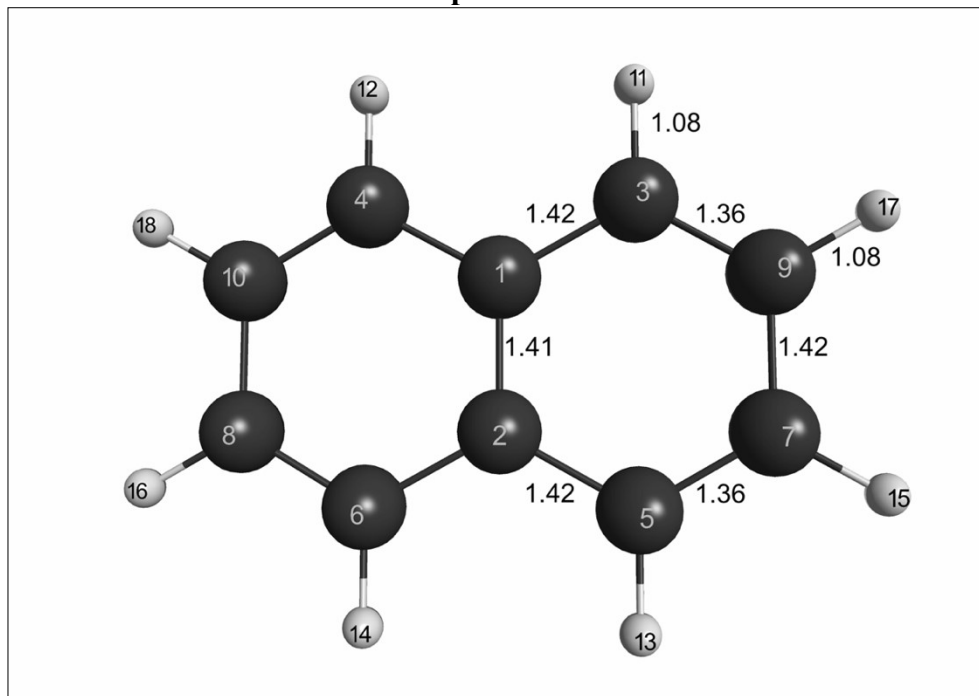
SI Table SI:1: Symmetric unique  $\sigma$  and  $\pi$ - Bonding interactions in ethene. Orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol) are shown for non-repetitive interactions. The structure of ethene was optimized at the RHF/6-31G(d) level of theory. Bond lengths for CC = 1.34 Å and CH= 1.09 Å.

Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
$C_i c_j \sigma$	1.00	$C_j c_i \sigma$	1.00	2.00	0.99	-53.3
$H_k c_j \sigma$	0.86	$C_j h_k \sigma$	1.14	2.00	0.98	-38.0
$C_i c_j \pi$	1.00	$C_j c_i \pi$	1.00	2.00	1.00	-23.0

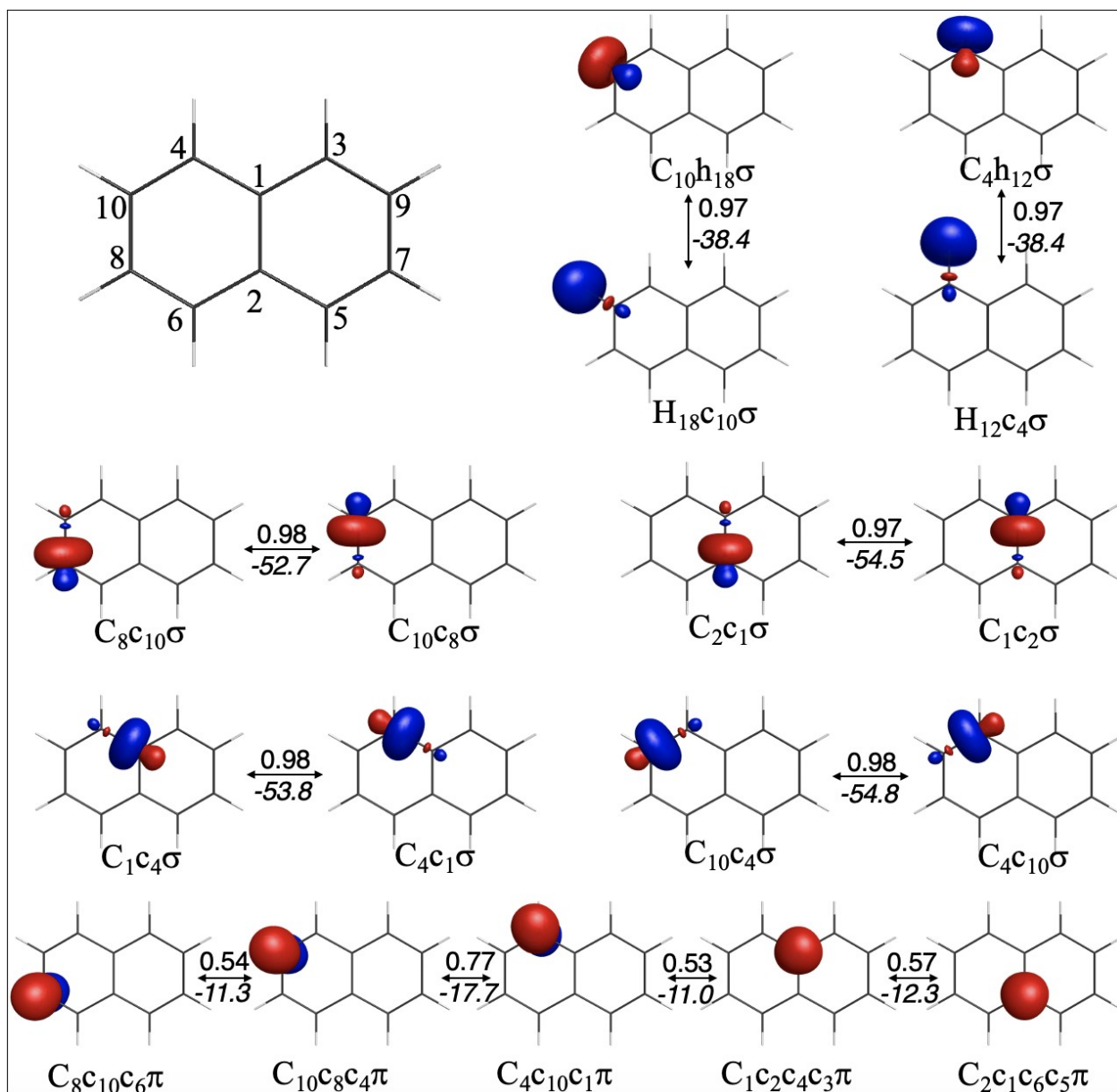
SI Table SI:2: Orbital hybridization fraction, s- and p-character, in ethene. SI Figure X3. Non-repetitive  $\sigma$  and  $\pi$ - bonding interactions in naphthalene. Bond orders (above arrow), and KBOs (below arrow) in kcal/mol are shown between pairs of bonding QUAOs.

Orbital	s-fraction	p-fraction
$C_i c_j \sigma$	0.34	0.66
$C_i h_k \sigma$	0.27	0.73
$C_i c_j \pi$	0.00	1.00
$H_k c_i \sigma$	1.00	0.00

### Naphthalene



SI Figure SI:3: Optimized structure of naphthalene at the RHF/6-31G(d) level of theory. Carbon-carbon and carbon-hydrogen bond lengths are shown in angstrom.

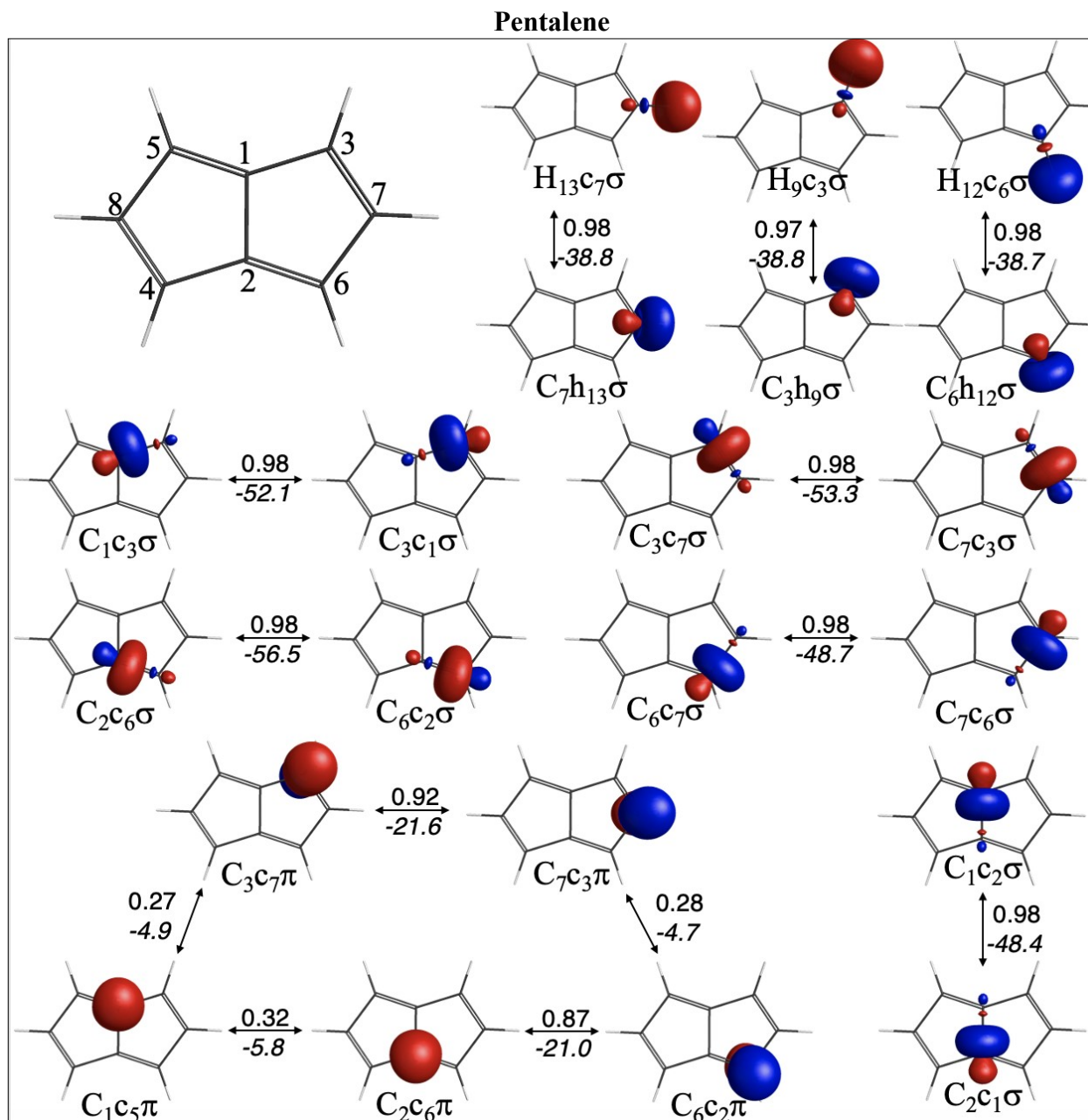


SI Figure SI:4. Non-repetitive  $\sigma$  and  $\pi$ -bonding interactions in naphthalene. Bond orders (above arrow), and KBOs (below arrow) in kcal/mol are shown between pairs of bonding QUAOs.

SI Table SI:3:  $\sigma$  and  $\pi$ -Bonding interactions in naphthalene. Orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol) are shown for non-repetitive interactions. Numbers display in table reference the atom numbered on SI Figure X3.

Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
$C_8c_6\sigma$	1.00	$C_6c_8\sigma$	1.00	2.00	0.98	-54.8
$C_1c_2\sigma$	1.00	$C_2c_1\sigma$	1.00	2.00	0.97	-54.5
$C_1c_4\sigma$	1.00	$C_4c_1\sigma$	1.00	2.00	0.98	-53.8
$C_8c_{10}\sigma$	1.00	$C_{10}c_8\sigma$	1.00	2.00	0.98	-52.7
$H_{12}c_4\sigma$	0.86	$C_4h_{12}\sigma$	1.14	2.00	0.97	-38.4

$C_4c_{10}c_1\pi$	1.00	$C_{10}c_8c_4\pi$	1.00	2.00	0.77	-17.7
$C_2c_1c_6c_5\pi$	1.00	$C_1c_2c_4c_3\pi$	1.00	2.00	0.57	-12.3
$C_8c_{10}c_6\pi$	1.00	$C_{10}c_8c_4\pi$	1.00	2.00	0.54	-11.3
$C_4c_{10}c_1\pi$	1.00	$C_1c_2c_4c_3\pi$	1.00	2.00	0.53	-11.0

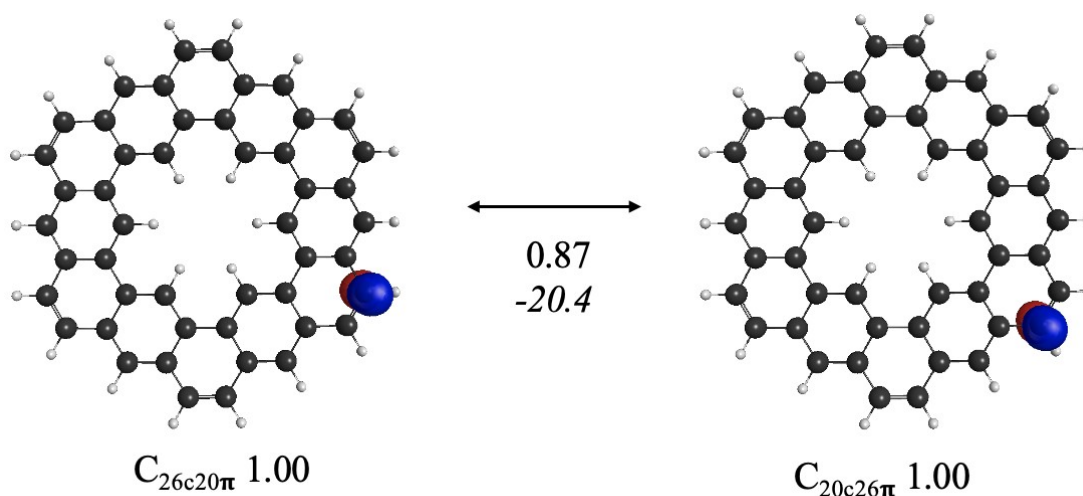


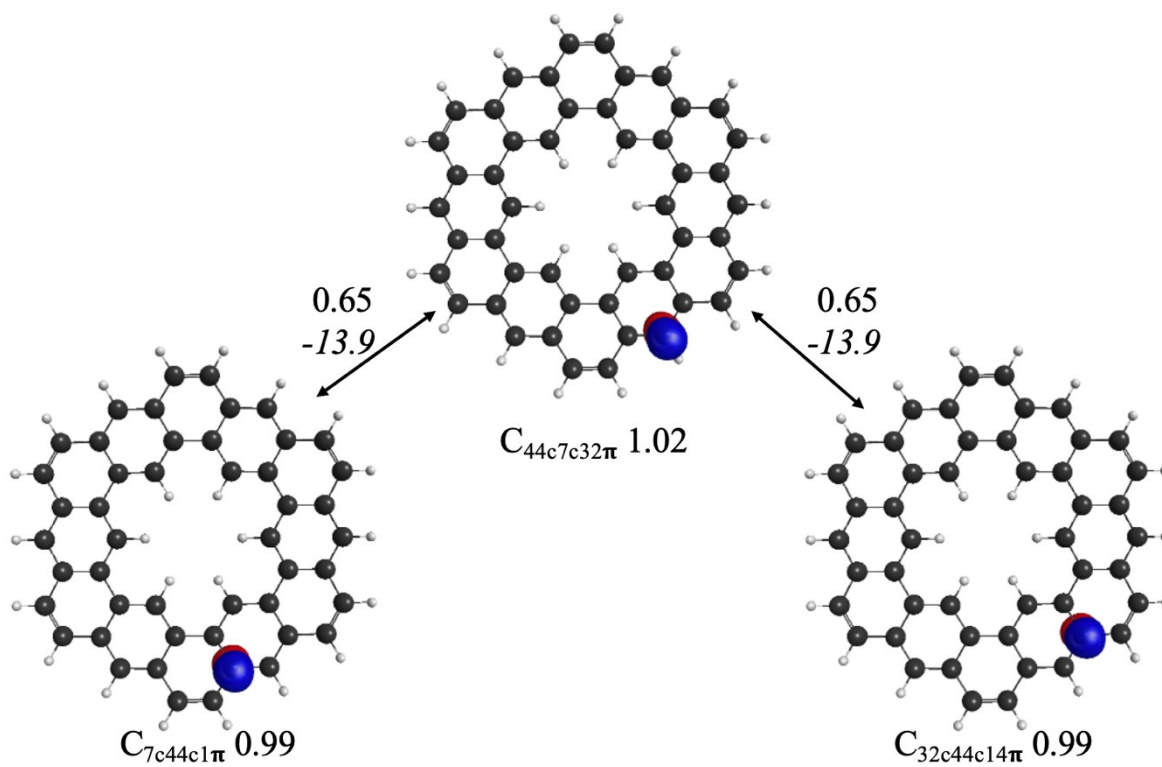
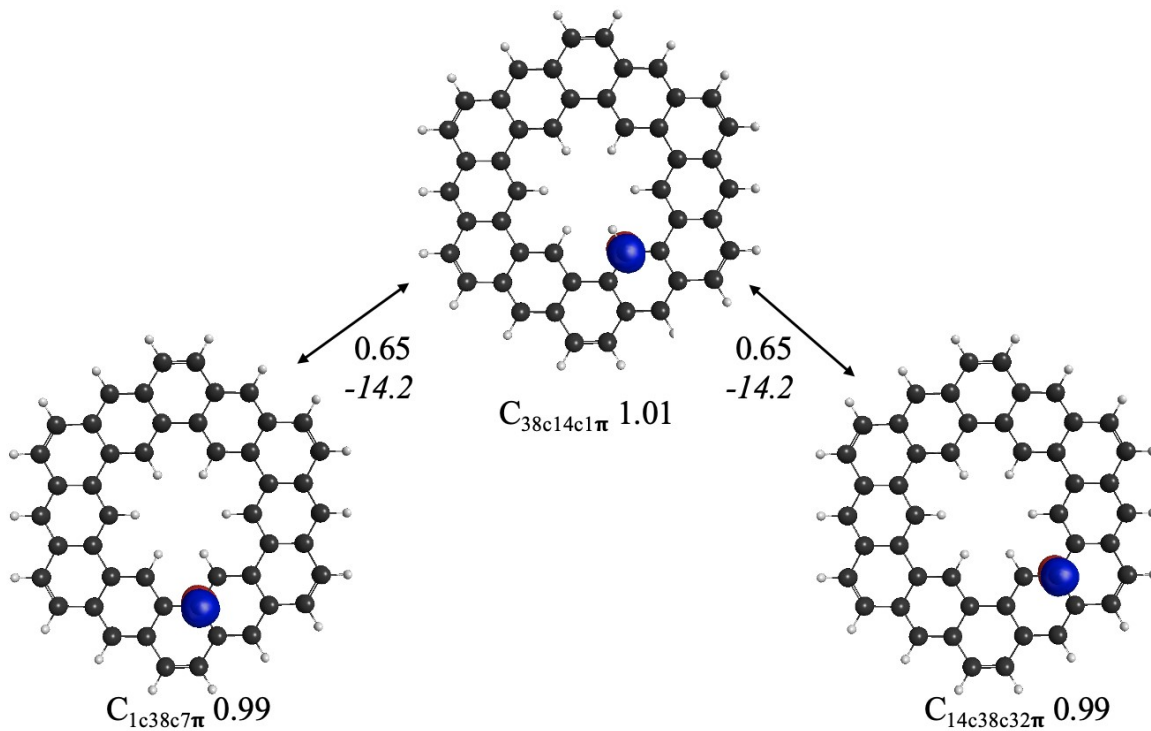
SI Figure SI:5. Non-repetitive  $\sigma$  and  $\pi$ -bonding interactions in pentalene. Bond orders (above arrow), and KBOs (below arrow) in kcal/mol are shown between pairs of bonding QUAOs.

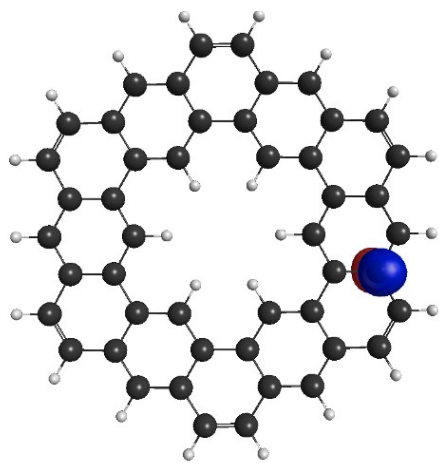
SI Table SI:4:  $\sigma$  and  $\pi$ - Bonding interactions in pentalene. Orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol) are shown for non-repetitive interactions. Numbers display in table reference the atom numbered in Figure 3.

Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
$C_6c_2\sigma$	1.00	$C_2c_6\sigma$	1.00	2.00	0.98	-56.5
$C_3c_7\sigma$	1.00	$C_7c_3\sigma$	0.99	1.99	0.98	-53.3
$C_1c_3\sigma$	1.01	$C_3c_1\sigma$	0.99	2.00	0.98	-52.1
$C_7c_6\sigma$	0.98	$C_6c_7\sigma$	1.01	1.99	0.98	-48.7
$C_1c_2\sigma$	0.99	$C_2c_1\sigma$	0.99	1.98	0.98	-48.4
$H_{13}c_7\sigma$	0.86	$C_7h_7\sigma$	1.15	2.01	0.98	-38.8
$H_9c_3\sigma$	0.85	$C_3h_9\sigma$	1.16	2.01	0.97	-38.8
$H_{12}c_6\sigma$	0.85	$C_6h_{12}\sigma$	1.16	2.01	0.98	-38.7
$C_7c_3\pi$	1.03	$C_3c_7\pi$	1.00	2.03	0.92	-21.6
$C_2c_6\pi$	1.03	$C_6c_2\pi$	0.94	1.97	0.87	-21.0
$C_1c_5\pi$	1.03	$C_2c_6\pi$	1.03	2.06	0.32	-5.8
$C_3c_7\pi$	1.00	$C_1c_5\pi$	1.03	2.03	0.27	-4.9
$C_7c_3\pi$	1.03	$C_6c_2\pi$	0.94	1.97	0.28	-4.7

Kekulene

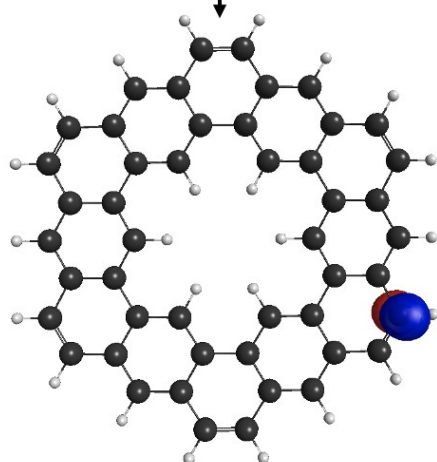




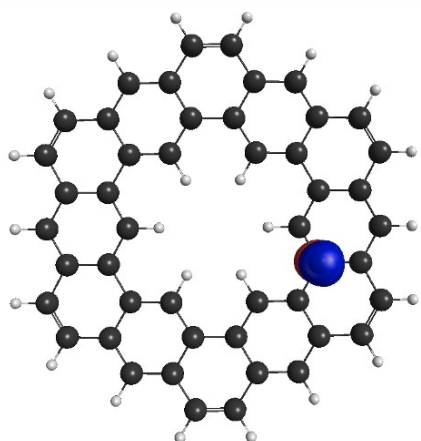


$C_{8c45c2\pi}$  0.99

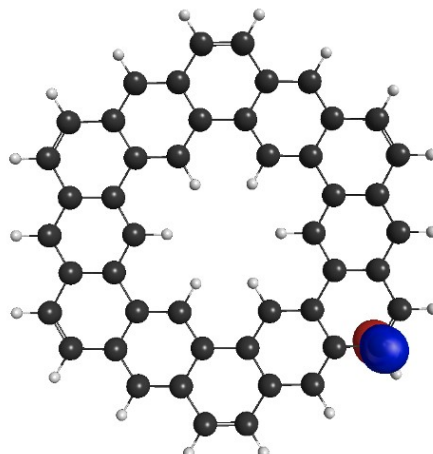
0.38  
-7.2



$C_{20c26\pi}$  1.00

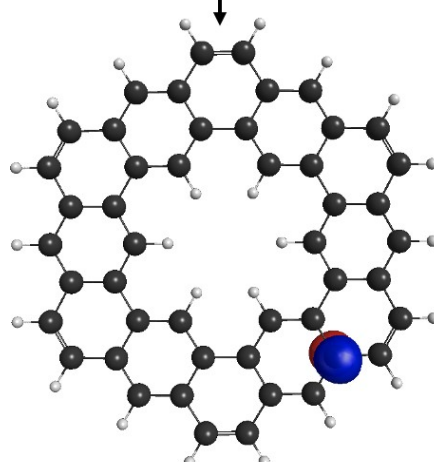


$C_{2c39c8\pi}$  0.99

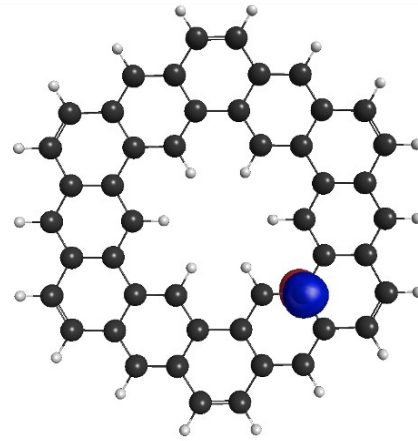


$C_{26c20\pi}$  1.00

0.38  
-7.2



$C_{32c44c14\pi}$  0.99

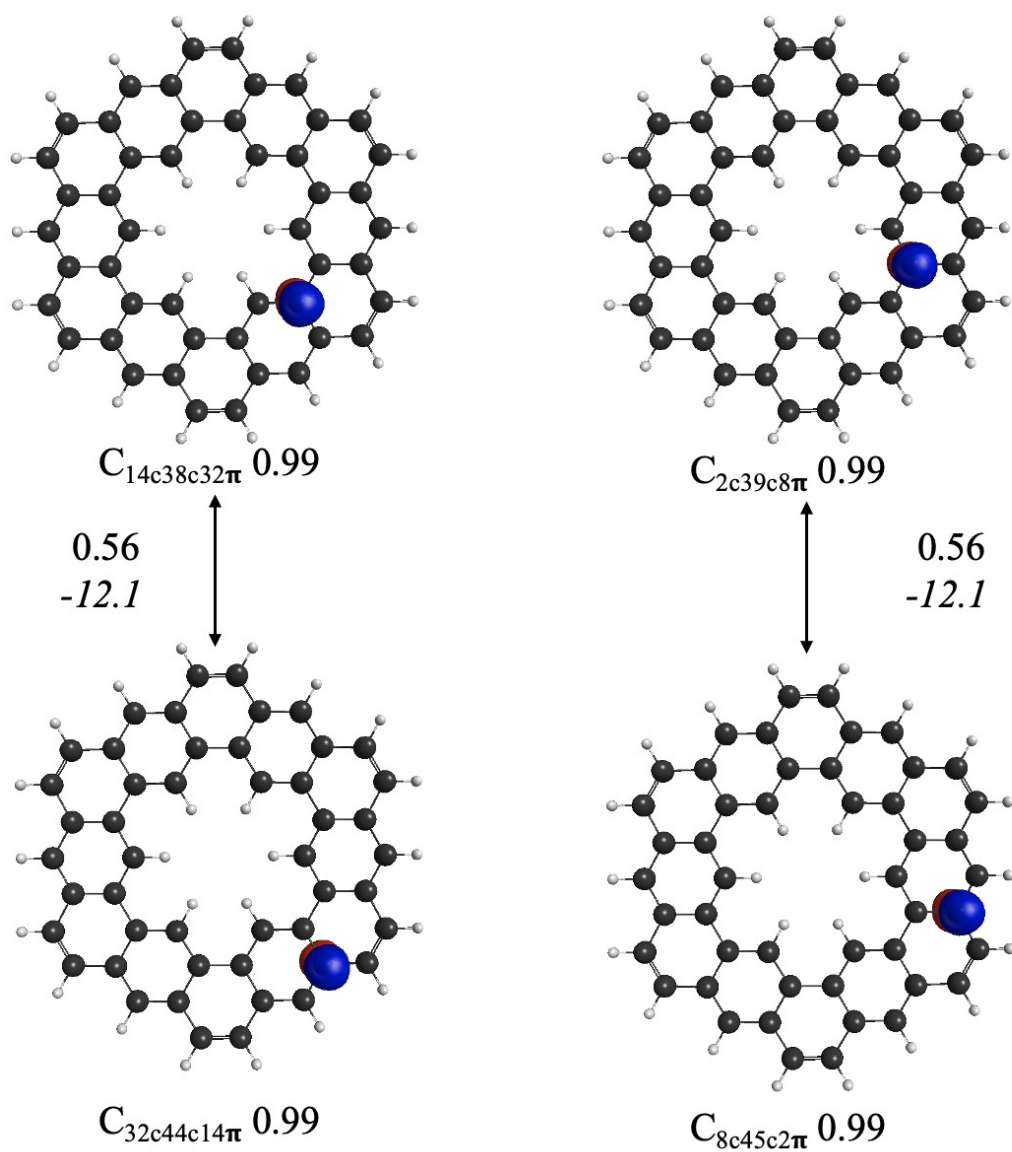


$C_{14c38c32\pi}$  0.99



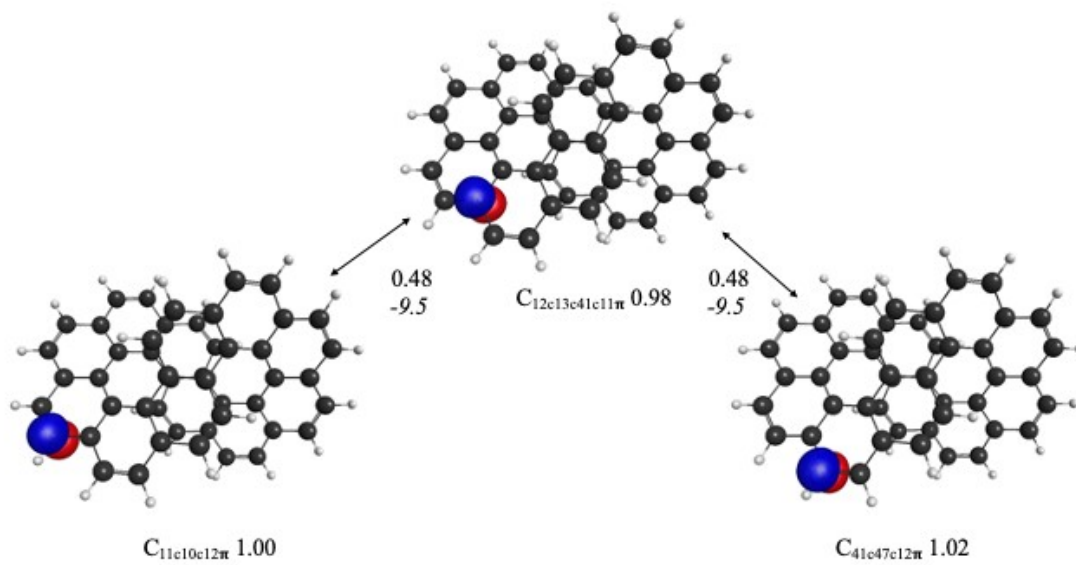
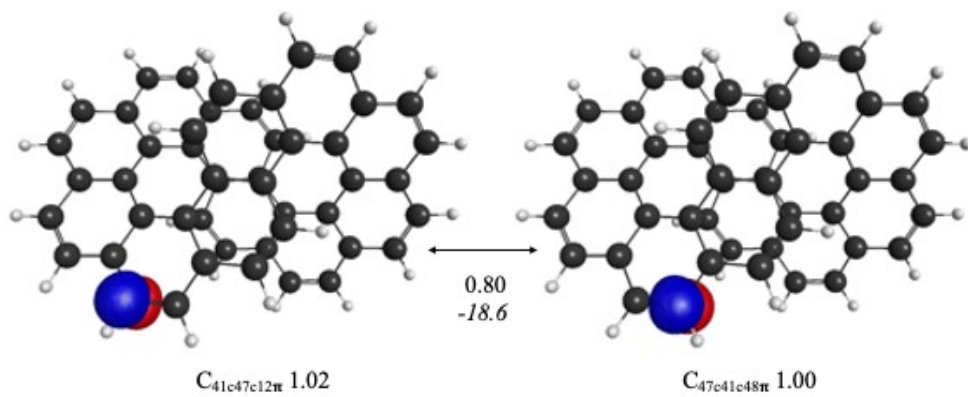
0.36  
-6.9

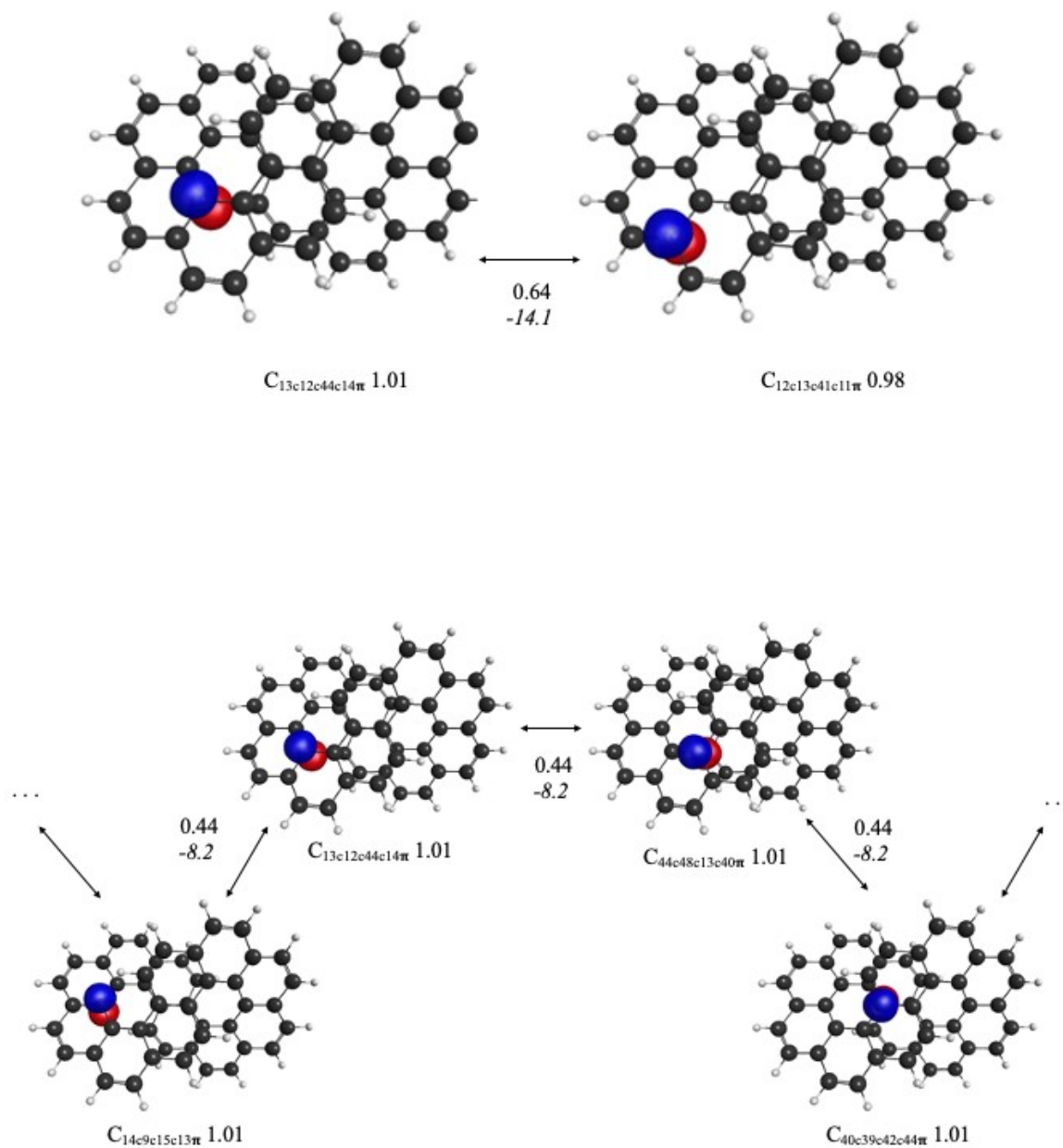




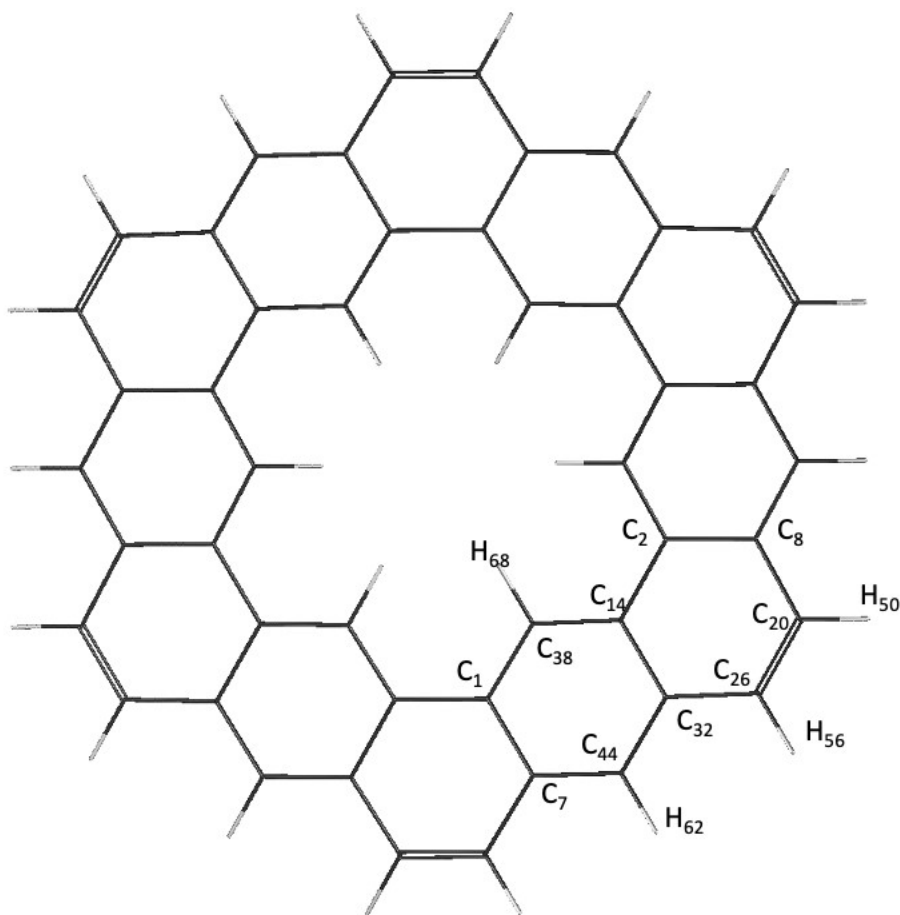
SI Figure SI:6. Non-repetitive  $\sigma$  and  $\pi$ - bonding interactions in kekulene. Bond orders (below arrow), and KBOs (in italics) in kcal/mol are shown between pairs of bonding QUAOs. Each QUAO label is given below each QUAO representation along with the QUAO population.

# Infinitone





SI Figure SI:7. Non-repetitive  $\sigma$  and  $\pi$ - bonding interactions in infinitene. Bond orders (below arrow), and KBOs (in italics) in kcal/mol are shown between pairs of bonding QUAOs. Each QUAO label is given below each QUAO representation along with the QUAO population.



SI Figure SI:8: Atom labels for non-repetitive and symmetric portion of kekulene. The atom numbers shown here were generated by GAMESS

SI Table SI:5: C-H  $\sigma$ -bonding QUAOs for non-repetitive and symmetric portion of kekulene with orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol). The atom numbers shown here were generated by GAMESS.

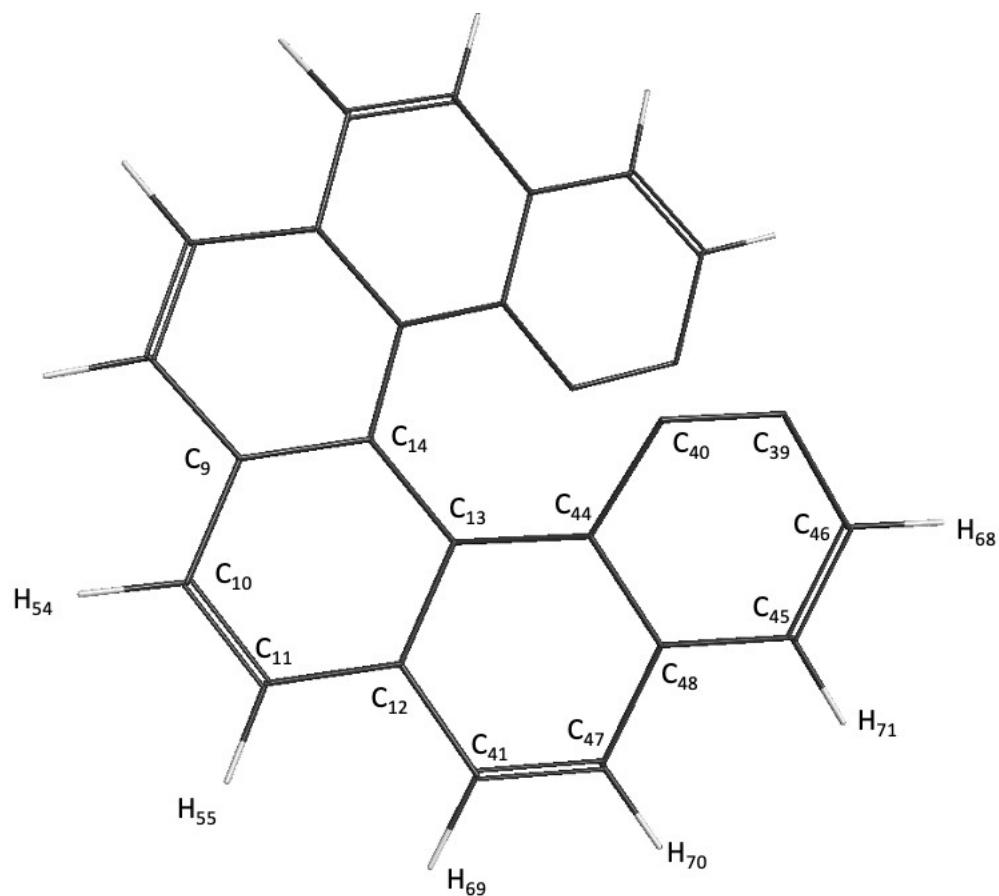
Kekulene C-H Sigma Bonds						
Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
H <sub>68</sub> C <sub>38</sub> $\sigma$	0.87	C <sub>38</sub> H <sub>68</sub> $\sigma$	1.13	2.00	-0.97	-41.68
H <sub>62</sub> C <sub>44</sub> $\sigma$	0.87	C <sub>44</sub> H <sub>62</sub> $\sigma$	1.13	2.00	0.98	-39.08
H <sub>50</sub> C <sub>20</sub> $\sigma$	0.86	C <sub>20</sub> H <sub>50</sub> $\sigma$	1.14	2.00	-0.97	-38.94
H <sub>56</sub> C <sub>26</sub> $\sigma$	0.86	C <sub>26</sub> H <sub>56</sub> $\sigma$	1.14	2.00	-0.97	-38.94

SI Table SI:6: C-C  $\pi$ -bonding QAOs for non-repetitive and symmetric portion of kekulene with orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol). The atom numbers shown here were generated by GAMESS.

Kekulene C-C Pi Bonds						
Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
C <sub>26</sub> c <sub>20</sub> $\pi$	1.00	C <sub>20</sub> c <sub>26</sub> $\pi$	1.00	2.01	0.87	-20.4
C <sub>38</sub> c <sub>14</sub> c <sub>1</sub> $\pi$	1.01	C <sub>14</sub> c <sub>38</sub> c <sub>32</sub> $\pi$	0.99	2.00	0.65	-14.2
C <sub>38</sub> c <sub>14</sub> c <sub>1</sub> $\pi$	1.01	C <sub>1</sub> c <sub>38</sub> c <sub>7</sub> $\pi$	0.99	2.00	0.65	-14.2
C <sub>44</sub> c <sub>7</sub> c <sub>32</sub> $\pi$	1.02	C <sub>7</sub> c <sub>44</sub> c <sub>1</sub> $\pi$	0.99	2.01	-0.64	-13.9
C <sub>32</sub> c <sub>44</sub> c <sub>14</sub> $\pi$	0.99	C <sub>14</sub> c <sub>38</sub> c <sub>32</sub> $\pi$	0.99	1.98	0.56	-12.1
C <sub>8</sub> c <sub>45</sub> c <sub>2</sub> $\pi$	0.99	C <sub>2</sub> c <sub>39</sub> c <sub>8</sub> $\pi$	0.99	1.98	0.56	-12.1
C <sub>20</sub> c <sub>26</sub> $\pi$	1.00	C <sub>8</sub> c <sub>45</sub> c <sub>2</sub> $\pi$	0.99	1.99	0.38	-7.2
C <sub>14</sub> c <sub>32</sub> c <sub>38</sub> $\pi$	0.99	C <sub>2</sub> c <sub>39</sub> c <sub>8</sub> $\pi$	0.99	1.99	0.36	-6.9

SI Table SI:7: C-C  $\sigma$ -bonding QAOs for symmetric quarter of kekulene with orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol). The atom numbers shown here were generated by GAMESS.

Kekulene C-C sigma Bonds						
Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
C <sub>26</sub> c <sub>20</sub> $\sigma$	1.00	C <sub>20</sub> c <sub>26</sub> $\sigma$	1.00	2.00	0.99	-55.7
C <sub>14</sub> c <sub>38</sub> $\sigma$	1.00	C <sub>38</sub> c <sub>14</sub> $\sigma$	1.00	2.00	0.98	-55.4
C <sub>38</sub> c <sub>1</sub> $\sigma$	1.00	C <sub>1</sub> c <sub>38</sub> $\sigma$	1.00	2.00	0.98	-55.4
C <sub>44</sub> c <sub>32</sub> $\sigma$	1.00	C <sub>32</sub> c <sub>44</sub> $\sigma$	1.00	2.00	0.98	-54.9
C <sub>7</sub> c <sub>44</sub> $\sigma$	1.00	C <sub>44</sub> c <sub>7</sub> $\sigma$	1.00	2.00	0.98	-54.9
C <sub>1</sub> c <sub>7</sub> $\sigma$	1.00	C <sub>7</sub> c <sub>1</sub> $\sigma$	1.00	2.00	0.97	-54.3
C <sub>32</sub> c <sub>14</sub> $\sigma$	1.00	C <sub>14</sub> c <sub>32</sub> $\sigma$	1.00	2.00	0.97	-54.3
C <sub>8</sub> c <sub>2</sub> $\sigma$	1.00	C <sub>2</sub> c <sub>8</sub> $\sigma$	1.00	2.00	0.97	-54.3
C <sub>2</sub> c <sub>14</sub> $\sigma$	1.00	C <sub>14</sub> c <sub>2</sub> $\sigma$	1.00	2.00	0.98	-53.5
C <sub>32</sub> c <sub>26</sub> $\sigma$	1.00	C <sub>26</sub> c <sub>32</sub> $\sigma$	1.00	2.00	0.98	-52.5
C <sub>20</sub> c <sub>8</sub> $\sigma$	1.00	C <sub>8</sub> c <sub>20</sub> $\sigma$	1.00	2.00	0.98	-52.5



SI Figure SI:9: Atom labels for symmetric quarter of infinitene. The atom numbers shown here were generated by GAMESS.

SI Table SI:8: C-H  $\sigma$ -bonding QUAOs for symmetric quarter of infinitene with orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol). The atom numbers shown here were generated by GAMESS.

Infinitene C-H Sigma Bonds						
Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
H <sub>54</sub> c <sub>10</sub> $\sigma$	0.86	C <sub>10</sub> h <sub>54</sub> $\sigma$	1.14	2.00	-0.97	-38.19
H <sub>55</sub> c <sub>11</sub> $\sigma$	0.86	C <sub>11</sub> h <sub>55</sub> $\sigma$	1.14	2.00	-0.97	-38.20
H <sub>68</sub> c <sub>46</sub> $\sigma$	0.85	C <sub>46</sub> h <sub>68</sub> $\sigma$	1.15	2.00	0.97	-38.22
H <sub>69</sub> c <sub>41</sub> $\sigma$	0.86	C <sub>41</sub> h <sub>69</sub> $\sigma$	1.14	2.00	0.97	-38.19
H <sub>70</sub> c <sub>47</sub> $\sigma$	0.86	C <sub>47</sub> h <sub>70</sub> $\sigma$	1.14	2.00	-0.97	-38.20
H <sub>71</sub> c <sub>45</sub> $\sigma$	0.86	C <sub>45</sub> h <sub>71</sub> $\sigma$	1.14	2.00	0.97	-38.18

SI Table SI:9: C-C  $\pi$ -bonding QUAOs for symmetric quarter of infinitene with orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol). The atom numbers shown here were generated by GAMESS.

Infinitene C-C Pi Bonds						
Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
C <sub>11</sub> c <sub>10</sub> c <sub>12</sub> $\pi$	1.00	C <sub>10</sub> c <sub>11</sub> c <sub>9</sub> $\pi$	1.01	2.01	0.81	-18.7
C <sub>46</sub> c <sub>45</sub> c <sub>39</sub> $\pi$	0.99	C <sub>45</sub> c <sub>46</sub> c <sub>48</sub> $\pi$	1.02	2.01	0.80	-18.6
C <sub>47</sub> c <sub>41</sub> c <sub>48</sub> $\pi$	1.00	C <sub>41</sub> c <sub>47</sub> c <sub>12</sub> $\pi$	1.02	2.01	-0.80	-18.5
C <sub>14</sub> c <sub>9</sub> c <sub>13</sub> c <sub>15</sub> $\pi$	1.01	C <sub>9</sub> c <sub>14</sub> c <sub>8</sub> c <sub>10</sub> $\pi$	0.98	2.00	0.65	-14.3
C <sub>48</sub> c <sub>44</sub> c <sub>47</sub> c <sub>45</sub> $\pi$	0.98	C <sub>44</sub> c <sub>48</sub> c <sub>13</sub> c <sub>40</sub> $\pi$	1.01	1.99	-0.64	-14.1
C <sub>40</sub> c <sub>39</sub> c <sub>42</sub> c <sub>44</sub> $\pi$	0.99	C <sub>39</sub> c <sub>40</sub> c <sub>46</sub> c <sub>34</sub> $\pi$	0.99	1.97	0.65	-14
C <sub>47</sub> c <sub>41</sub> c <sub>48</sub> $\pi$	1.00	C <sub>48</sub> c <sub>44</sub> c <sub>47</sub> c <sub>45</sub> $\pi$	0.98	1.98	0.48	-9.7
C <sub>41</sub> c <sub>47</sub> c <sub>12</sub> $\pi$	1.02	C <sub>12</sub> c <sub>13</sub> c <sub>41</sub> c <sub>11</sub> $\pi$	0.98	2.00	0.48	-9.6
C <sub>46</sub> c <sub>45</sub> c <sub>39</sub> $\pi$	0.99	C <sub>39</sub> c <sub>40</sub> c <sub>46</sub> c <sub>34</sub> $\pi$	0.99	1.98	0.48	-9.5
C <sub>11</sub> c <sub>10</sub> c <sub>12</sub> $\pi$	1.00	C <sub>12</sub> c <sub>13</sub> c <sub>41</sub> c <sub>11</sub> $\pi$	0.98	1.99	0.47	-9.4
C <sub>10</sub> c <sub>11</sub> c <sub>9</sub> $\pi$	1.01	C <sub>9</sub> c <sub>14</sub> c <sub>8</sub> c <sub>10</sub> $\pi$	0.98	1.99	-0.47	-9.4
C <sub>45</sub> c <sub>46</sub> c <sub>48</sub> $\pi$	1.02	C <sub>48</sub> c <sub>44</sub> c <sub>47</sub> c <sub>45</sub> $\pi$	0.98	2.01	-0.47	-9.3
C <sub>44</sub> c <sub>48</sub> c <sub>13</sub> c <sub>40</sub> $\pi$	1.01	C <sub>13</sub> c <sub>12</sub> c <sub>44</sub> c <sub>14</sub> $\pi$	1.01	2.01	0.46	-8.7
C <sub>44</sub> c <sub>48</sub> c <sub>13</sub> c <sub>40</sub> $\pi$	1.01	C <sub>40</sub> c <sub>39</sub> c <sub>42</sub> c <sub>44</sub> $\pi$	0.99	1.99	0.44	-8.3
C <sub>14</sub> c <sub>9</sub> c <sub>13</sub> c <sub>15</sub> $\pi$	1.01	C <sub>13</sub> c <sub>12</sub> c <sub>44</sub> c <sub>14</sub> $\pi$	1.01	2.02	-0.44	-8.0

SI Table SI:10: C-C  $\sigma$ -bonding QUAOs for symmetric quarter of infinitene with orbital occupations, bond occupation totals, bond orders, and KBOs (kcal/mol). The atom numbers shown here were generated by GAMESS.

Infinitene C-C sigma Bonds						
Orbital I	Occupation I	Orbital J	Occupation J	Bond Total	BO	KBO (kcal/mol)
C <sub>10</sub> c <sub>11</sub> $\sigma$	1.00	C <sub>11</sub> c <sub>10</sub> $\sigma$	1.00	2.00	0.98	-55.8
C <sub>41</sub> c <sub>47</sub> $\sigma$	1.00	C <sub>47</sub> c <sub>41</sub> $\sigma$	1.00	2.00	0.98	-55.6
C <sub>45</sub> c <sub>46</sub> $\sigma$	1.00	C <sub>46</sub> c <sub>45</sub> $\sigma$	1.00	2.00	0.98	-55.6
C <sub>48</sub> c <sub>44</sub> $\sigma$	1.00	C <sub>44</sub> c <sub>48</sub> $\sigma$	1.00	2.00	0.98	-54.7
C <sub>39</sub> c <sub>40</sub> $\sigma$	1.00	C <sub>40</sub> c <sub>39</sub> $\sigma$	1.00	2.00	0.98	-54.7
C <sub>12</sub> c <sub>13</sub> $\sigma$	1.00	C <sub>13</sub> c <sub>12</sub> $\sigma$	1.00	2.00	0.98	-54.4
C <sub>9</sub> c <sub>14</sub> $\sigma$	1.00	C <sub>14</sub> c <sub>9</sub> $\sigma$	1.00	2.00	0.98	-54.3
C <sub>40</sub> c <sub>44</sub> $\sigma$	1.00	C <sub>44</sub> c <sub>40</sub> $\sigma$	1.00	2.00	0.98	-53.9
C <sub>44</sub> c <sub>13</sub> $\sigma$	1.00	C <sub>13</sub> c <sub>44</sub> $\sigma$	1.01	2.00	0.98	-53.7
C <sub>47</sub> c <sub>48</sub> $\sigma$	1.00	C <sub>48</sub> c <sub>47</sub> $\sigma$	1.00	2.00	0.98	-53.3
C <sub>48</sub> c <sub>45</sub> $\sigma$	1.01	C <sub>45</sub> c <sub>48</sub> $\sigma$	0.99	2.00	0.98	-53.1
C <sub>46</sub> c <sub>39</sub> $\sigma$	0.99	C <sub>39</sub> c <sub>46</sub> $\sigma$	1.01	2.00	0.98	-53.1
C <sub>11</sub> c <sub>12</sub> $\sigma$	1.00	C <sub>12</sub> c <sub>11</sub> $\sigma$	1.00	2.00	0.98	-53.0
C <sub>13</sub> c <sub>14</sub> $\sigma$	1.00	C <sub>14</sub> c <sub>13</sub> $\sigma$	1.00	2.00	0.98	-53.0
C <sub>12</sub> c <sub>41</sub> $\sigma$	1.00	C <sub>41</sub> c <sub>12</sub> $\sigma$	0.99	2.00	0.98	-53.0
C <sub>9</sub> c <sub>10</sub> $\sigma$	1.00	C <sub>10</sub> c <sub>9</sub> $\sigma$	1.00	2.00	0.98	-52.9