Study on the Growth Mechanism of Monolayer and Few-Layer Hexagonal Boron

Nitride Films on Copper Foil

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Fig. S1 Initial and final structures of H atom bond to (a, b) N and (c, d) B atoms at the ZZ edge of h-BN on Cu(111). Initial and final structures of H atom bond to (e, f) B and (g, h) N atoms at the AC edge of h-BN on Cu(111). The green, silvery, white and blue balls represent the B, N, H and Cu atoms, respectively. The red dashed circles represent the diffusion process of the marked atoms.



Fig. S2 (a-d) Schematic diagram for H atom desorption from the B and N atoms for BN-ZZ-H and BN-AC-H. The green, silvery, white and blue balls represent the B, N, H and Cu atoms, respectively. The red dashed circles and yellow arrows represent the diffusion process of the marked atoms.



Fig.S3 (a-h) Diagram for the diffusion of B and N atoms along subsurface for BN-ZZ, BN-AC, BN-ZZ-H and BN-AC-H. (i-l) Diagram for the diffusion of B and N atoms from subsurface to surface for BN-ZZ and BN-ZZ-H. The green, silvery, white and blue balls represent the B, N, H and Cu atoms, respectively. The red dashed circles and yellow arrows represent the diffusion process of the marked atoms.

Table S1 Growth of h-BN on copper substrate under different conditions

Ref	Substrate	Pressure regime	Growth temperatur e	Precursor	Atmospher e	Results	Torr
1	Polycrystalli ne Cu foil	Atmospheri c pressure	1000° C	Ammonia- borane 120– 130 °C	Ar: H ₂ =170: 30 sccm	2–5 layer film	760
2	Cu foil	Atmospheri c pressure	1000 °C	Borazine	H ₂ =2000sccm	2 to ~20 nm thick film	760
3	Polycrystalli ne Cu foil	Atmospheri c pressure	1000 °C	Ammonia- borane, 110– 130℃	Ar: H ₂ = 80: 20 sccm	6–8 layer film	760
4	Cu foil	~500 Torr	950 ℃	Ammonia- borane, 130 $^\circ\!\!\mathbb{C}$	Ar: H ₂ = 300: 50 sccm	5-layer	500
5	Cu foil	Atmospheri c pressure	1050° C	Ammonia- borane, 5 mg, 60℃	Ar: H ₂ =170: 30 sccm	1–2.2 nm thick film	760
6	Cu (solid and molten Cu on W)	Atmospheri c pressure	1000° C 1100° C	Ammonia- borane, 110 $^\circ\!\!\mathbb{C}$	Ar: H ₂ =90: 10 sccm	1–10 layers	760
7	Polycrystalli ne Cu foil	Atmospheri c pressure	1030° C	Ammonia- borane, 60– 90℃	N ₂ : H ₂ = 180: 20 sccm	Monolayer film	760
8	Cu foil	Atmospheri c pressure	1020° C	Ammonia- borane, 8.5mg, 80–100 ℃	Ar: H ₂ = 95: 5 sccm	~3 nm thick film	760
9	Cu foil	Atmospheri c pressure	1030° C	Ammonia- borane, 80 $^\circ\!$	20-mTorr air and 10- sccm H ₂	~3 nm thick film	760
10	Cu foil	Low pressure	1000° C	Ammonia- borane, 55– 120℃	Ar: H ₂ = 50: 50 sccm	Monolayer film	0.5
11	Cu foil	∼10 ^{–6} mbar	≈ 950– 1000°C	Borazine, 1 × 10^{-4} to 5 × 10^{-3} mbar	H ₂ (~0.2 mbar)	Monolayer films	7.5*10 ⁻⁷
12	Cu foil	350 mTorr	1000° ℃	Ammonia- borane, 60– 90℃	H ₂ , 10 sccm (350 mTorr)	Monolayer film and islands	0.35
13	Polycrystalli ne Cu foil	30–40 Pa	1000 ℃	Ammonia- borane, 90– 100 °C	Ar, 40 sccm	Monolayer film	0.3
14	Cu foil	400 mTorr	1050%	Ammonia-	H ₂ ,100	Monolayer	0.4

				borane, 130℃	sccm	film	
15	Cu foil	30–40 Pa	1000 °C	Ammonia- borane, 50, 70, 90, and 110℃	Different ratio of Ar and H ₂	Monolayer film and islands	0.3
16	Polycrystalli ne Cu foil	UHV, base pressure < 10 ⁻⁷ mbar	1000 ℃	Ammonia- borane, 15 mg	H ₂ , 10 sccm	Monolayer islands	7.5*10 ⁻⁸
17	Cu foil	Low pressure	1050 ℃	Ammonia- borane, 100 $^\circ\!\!\mathbb{C}$	H ₂ , 15 sccm	Monolayer film	0.5
18	Polycrystalli ne Cu foil	Low- pressure	1050° C	Ammonia- borane, 2– 3mg, 75–85℃	H ₂ , 40 sccm	Monolayer islands	0.5
19	Cu foi	Low- pressure	1050° C	Borazine	Ar: H ₂ = 70: 100 sccm	Monolayer film	0.5
20	single- crystal Cu (110) foil	low pressure (about 200 Pa)	1,035 ℃	Ammonia borane 65 $^\circ\!$	Ar: H ₂ = 5: 45 sccm	single- crystal monolayer	1.5
21	Cu (111) films deposited on sapphire	low- pressure 5.0 torr	1,050 °C	Ammonia borane, roughly 60 mg, 85 ℃	H ₂ , 30 sccm	single- crystal monolayer	5

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