

## Electronic supplementary information

### **Mo<sub>2</sub>C coated with Ni nanoparticles as cathode catalyst towards efficient hydrogen evolution reaction: an experimental and computational investigation**

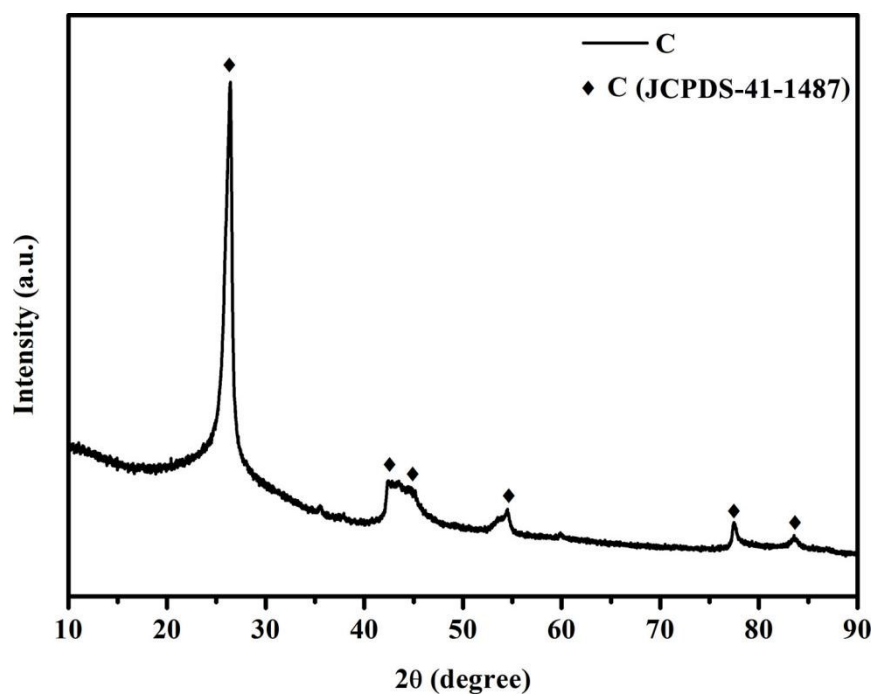
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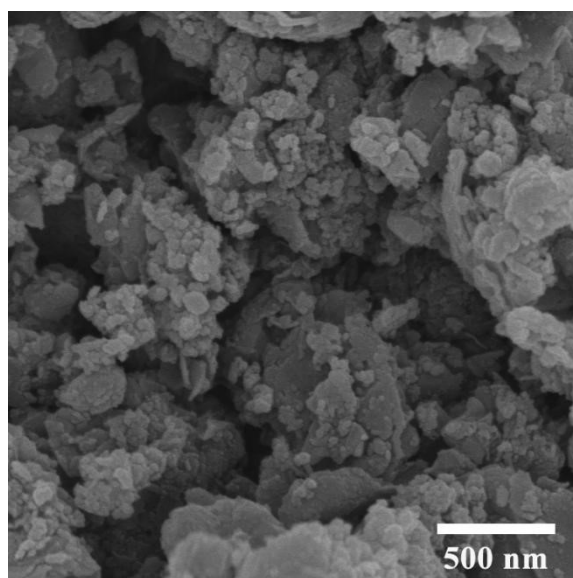
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**Fig. S1** XRD pattern of fabricated C material.



**Fig. S2** SEM image of synthesized alkynyl carbon.

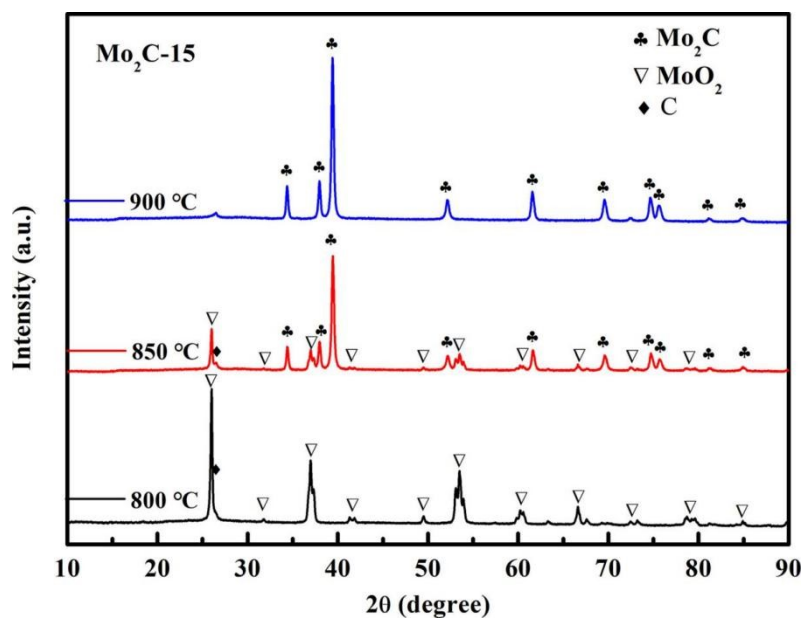


Fig. S3 XRD patterns of Mo<sub>2</sub>C-15 prepared at different temperature.

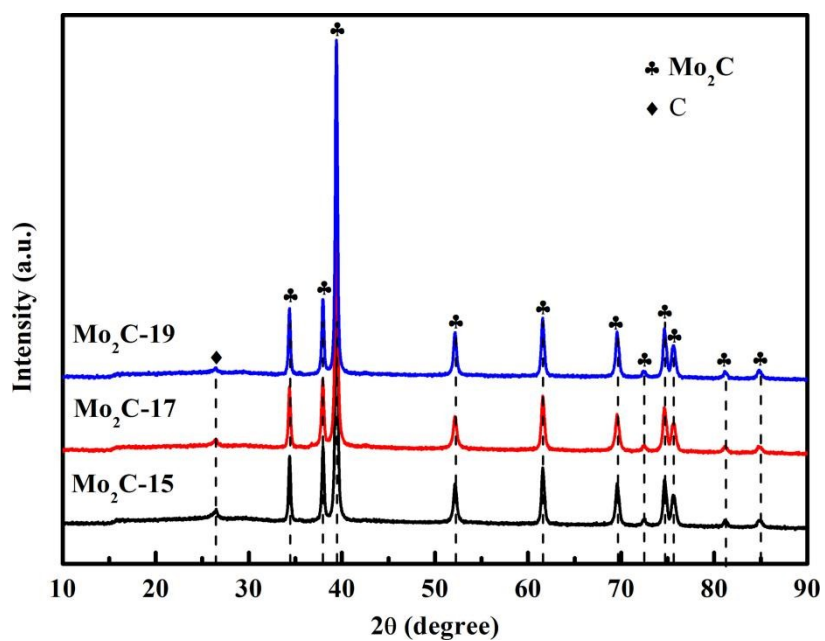
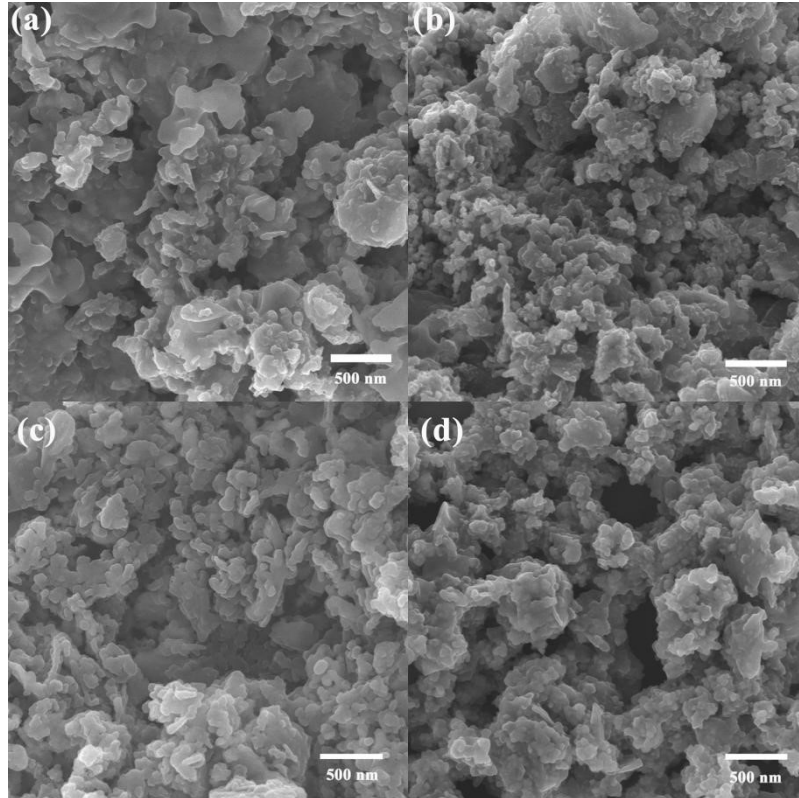
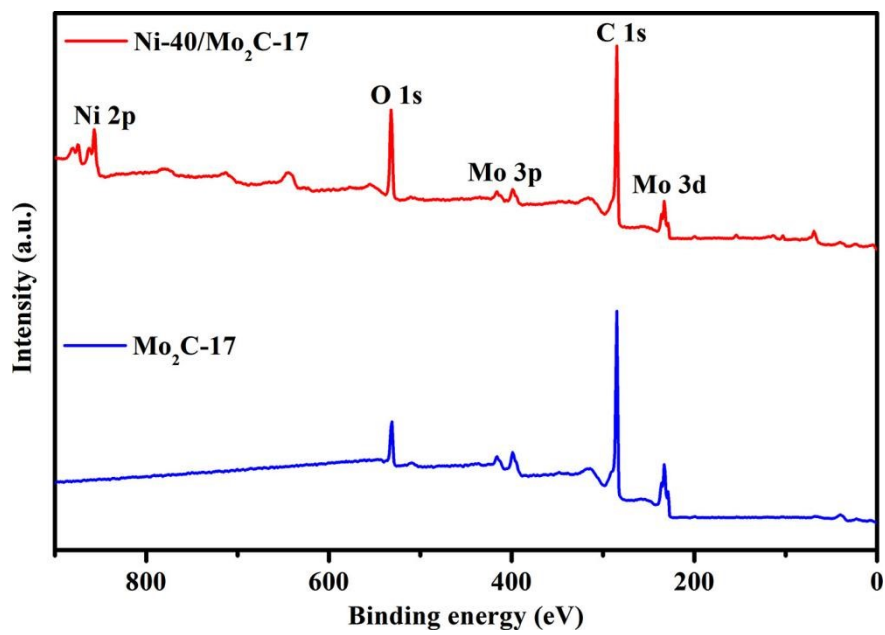


Fig. S4 XRD patterns of Mo<sub>2</sub>C-15, Mo<sub>2</sub>C-17, Mo<sub>2</sub>C-19 samples.



**Fig. S5** (a) SEM image of Ni-30/Mo<sub>2</sub>C-17. (b) SEM image of Ni-50/Mo<sub>2</sub>C-17. (c) SEM image of Mo<sub>2</sub>C-17. (d) SEM image of Ni-40/C.



**Fig. S6** XPS full scans of Mo<sub>2</sub>C-17 and Ni-40/Mo<sub>2</sub>C-17.

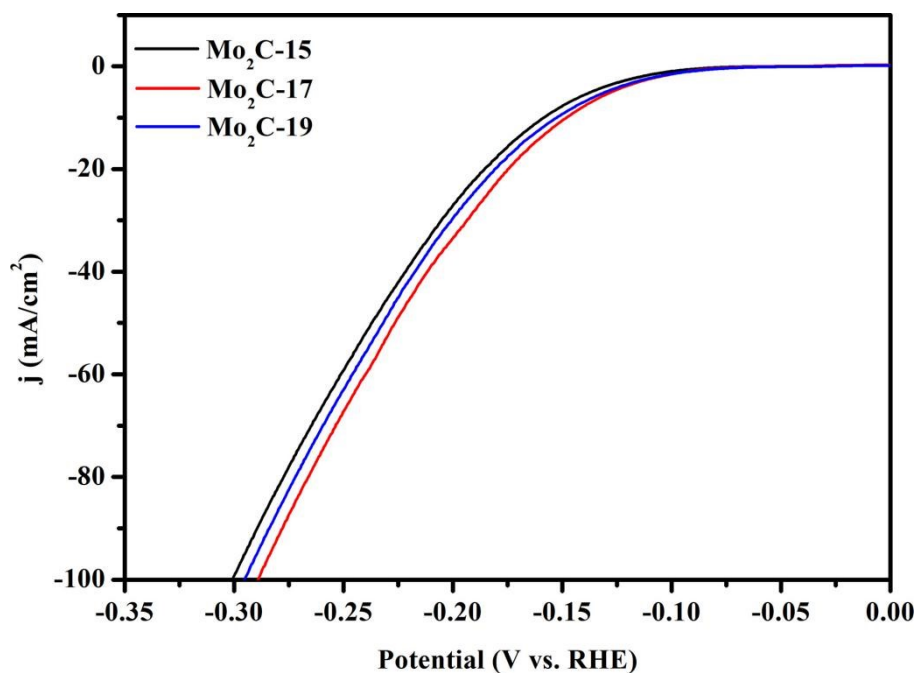


Fig. S7 Polarization curves of Mo<sub>2</sub>C prepared by molybdenum precursor with different content.

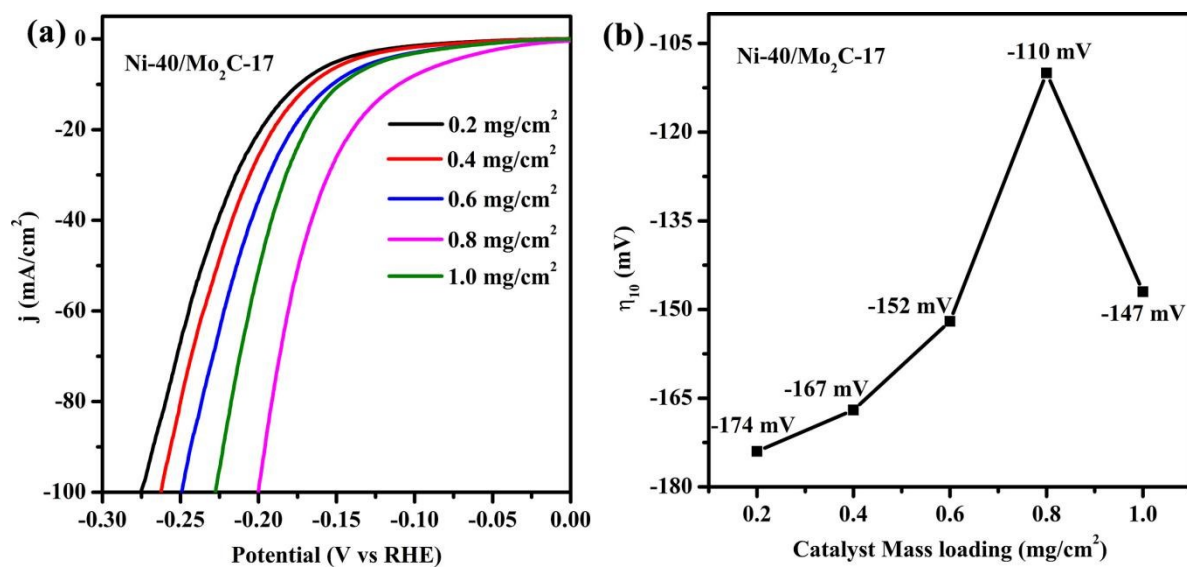
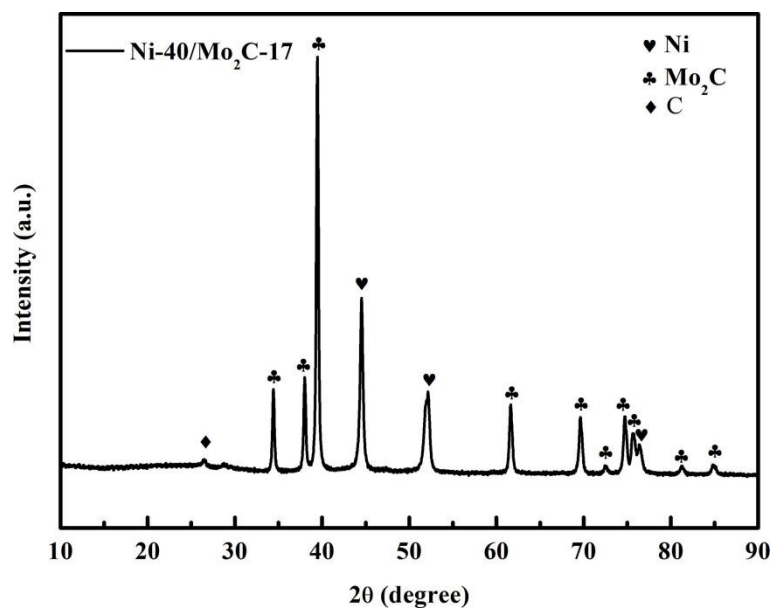
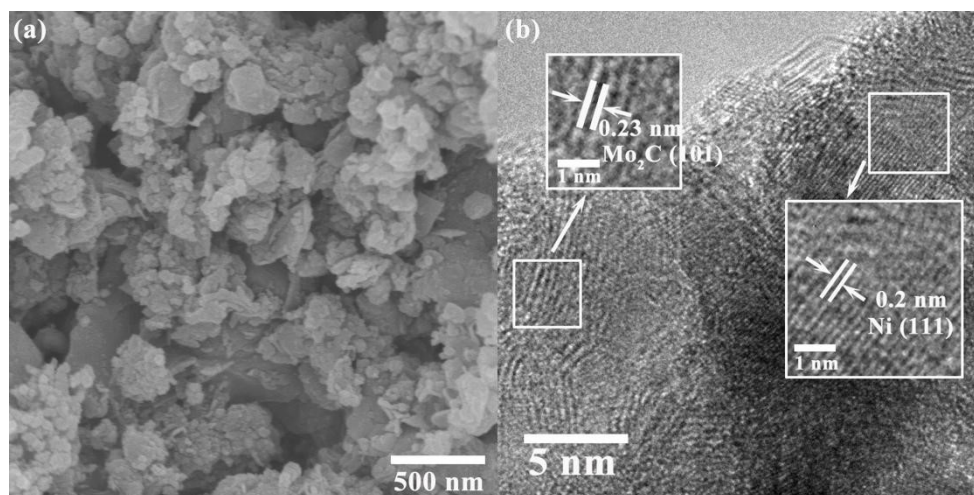


Fig. S8 (a) Polarization curves of Ni-40/Mo<sub>2</sub>C-17 under different mass loading, (b) the relationship between mass loading and  $\eta_{10}$  in 1 M KOH.

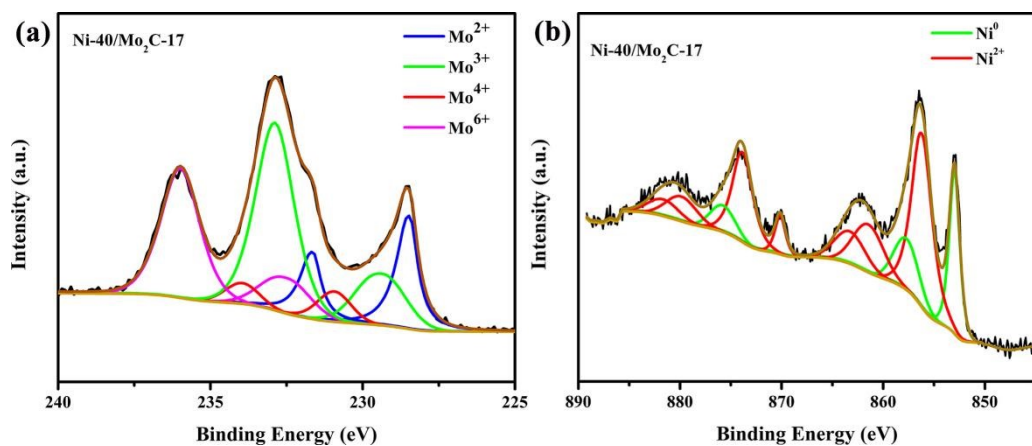


**Fig. S9** XRD characterization of Ni-40/Mo<sub>2</sub>C-17 catalyst after long-term stability test.

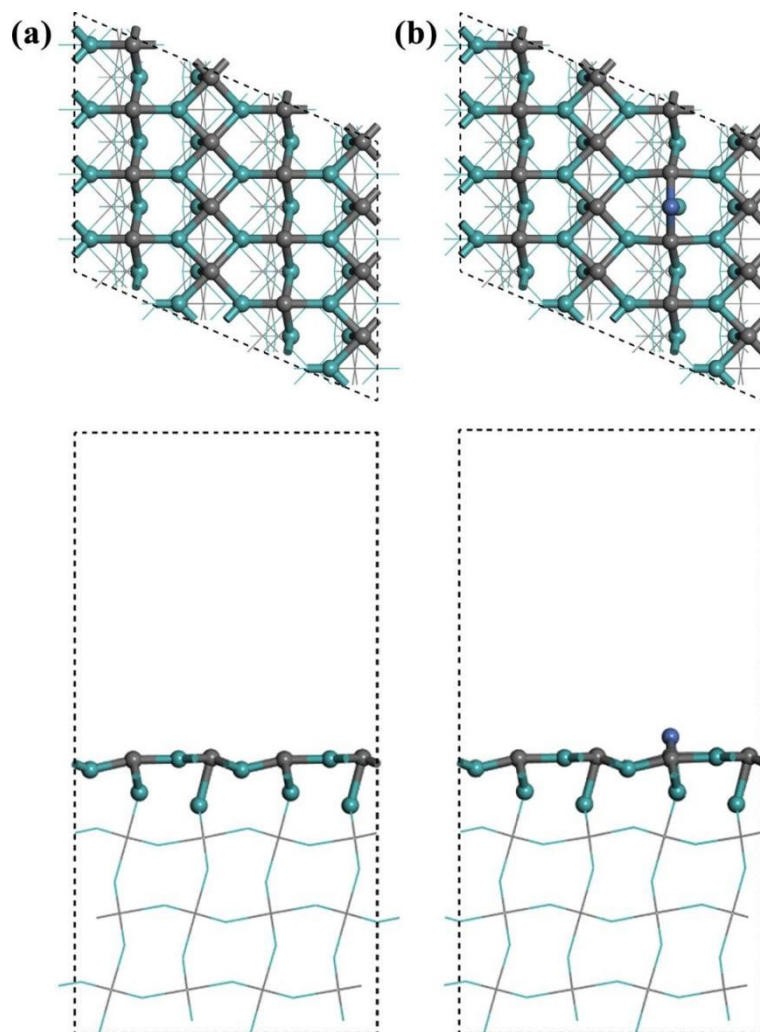


**Fig. S10** (a) SEM image of Ni-40/Mo<sub>2</sub>C-17 after long-term stability test. (b) HRTEM image of Ni-40/Mo<sub>2</sub>C-17 after long-term stability test.





**Fig. S11** (a) Mo  $3d$  XPS spectra of Ni-40/Mo<sub>2</sub>C-17 after long-term stability test. (b) Ni  $2p$  XPS spectra of Ni-40/Mo<sub>2</sub>C-17 after long-term stability test.



**Fig. S12** The optimized structures of (a) clean and (b) Ni doped Mo<sub>2</sub>C (101) surface.

**Table S1** Comparison of the HER performance for Ni-40/Mo<sub>2</sub>C-17 catalyst with other Mo<sub>2</sub>C-based or Ni-based catalysts in alkaline electrolytes (1 M KOH).

Catalyst	Loading density (mg cm <sup>-2</sup> )	$\eta_{10}$ (mV)	Reference
Mo <sub>2</sub> C-Ni	0.285	372	<i>New J. Chem.</i> 2017, 41, 12956
Ni/Mo <sub>2</sub> C-PC	0.5	179	<i>Chem. Sci.</i> 2017, 8, 968
Ni/Mo <sub>2</sub> C(1:2)- NCNFs	1.4	143	<i>Adv. Energy Mater.</i> 2019, 9, 1803185
Mo <sub>2</sub> C	0.8	190	<i>Angew. Chem. Int. Ed.</i> 2012, 51, 1270
Mo <sub>2</sub> C	0.009	270	<i>J. Am. Chem. Soc.</i> 2015, 137, 7035
L-Mo <sub>2</sub> C	N/A	95.8	<i>ACS Appl. Mater. Interfaces</i> , 2018, <b>10</b> , 40500
Mo <sub>2</sub> C-NCNT	3	257	<i>J. Mater. Chem. A</i> 2015, 3, 5783
Mo <sub>2</sub> C/carbon sheets	0.4	178	<i>ChemSusChem</i> 2017, 10, 3540
NiO/Ni-CNT	0.28	80	<i>Nat. Commun.</i> 2014, 5, 4695
NiFe LDH/NF	N/A	210	<i>Science</i> 2014, 345, 1593
NiFe LDH/rGO	0.25	560	<i>ACS Nano</i> 2015, 89, 1977
Ni-40/Mo <sub>2</sub> C-17	0.8	110	This work