Supplementary Information (SI) for Physical Chemistry Chemical Physics. This journal is © the Owner Societies 2024

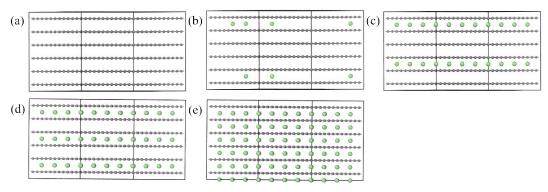


Figure S1 (a)  $\cdot$  (b)  $\cdot$  (c)  $\cdot$  (d)and (e) are respectively the simulated initial models of Li<sub>x</sub>C<sub>6</sub> (X=0.02-1).

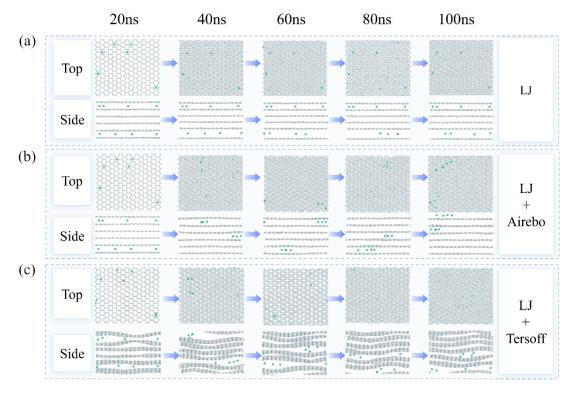


Figure S2 MD simulations of the top and side views of the  $\text{Li}_{0.02}\text{C}_6$  model were carried out within 100ns. During the simulations, (a) the LJ potential function, (b) the Airebo potential function and (c) the Tersoff potential function describe the interaction of C-C bonds between graphite layers. where grey represents carbon atoms and green represents lithium atoms.

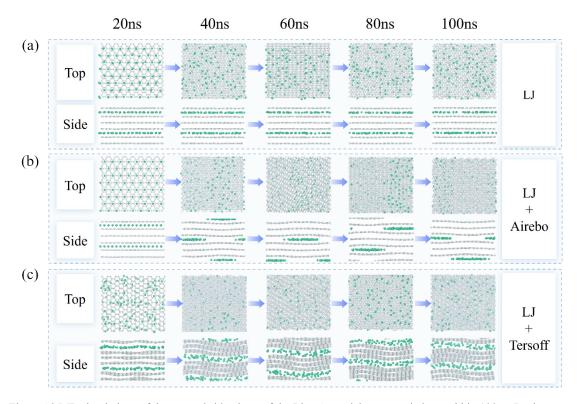


Figure S3 MD simulations of the top and side views of the  $Li_{0.33}C_6$  model were carried out within 100ns. During the simulations, (a) the LJ potential function, (b) the Airebo potential function and (c) the Tersoff potential function describe the interaction of C-C bonds between graphite layers. where grey represents carbon atoms and green represents lithium atoms.

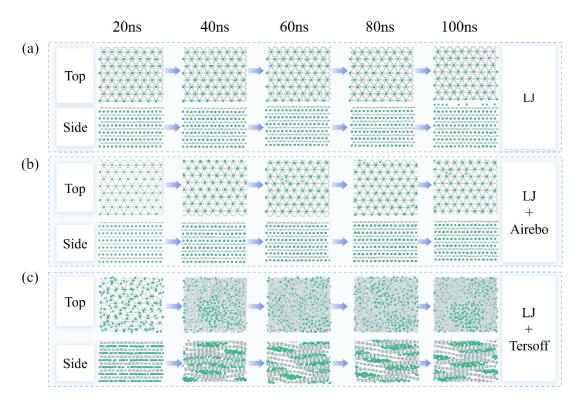


Figure S4 MD simulations of the top and side views of the  $Li_{0.5}C_6$  model were carried out within 100ns. During the simulations, (a) the LJ potential function, (b) the Airebo potential function and (c) the Tersoff potential function

describe the interaction of C-C bonds between graphite layers. where grey represents carbon atoms and green represents lithium atoms.

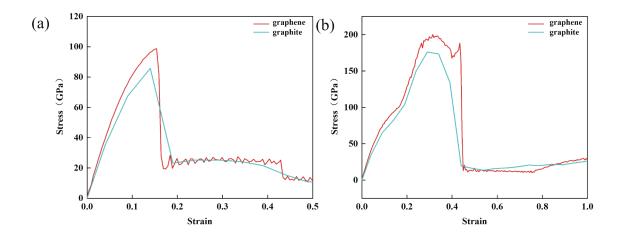


Figure S5 The stress-strain curves of graphene with different layers: (a) Represents the stress-strain curves of monolayer graphene and six-layer graphite obtained using the AIREBO potential function. (b) Represents the stress-strain curves of monolayer graphene and six-layer graphite obtained using the Tersoff potential function.

Table S1

Diffusion Coefficients of Lithium ions (cm²/sec)				
	This Work		$MC^1$	$\mathrm{DFT^2}$
	Airebo	Tersoff		
$Li_{0.02}C_6$	$1.3955 \times 10^{-6}$	$4.95167 \times 10^{-6}$	/	/
$Li_{0.2}C_6$	/	/	$1.43 \times 10^{-8}$	$7.04 \times 10^{-6}$
$Li_{0.33}C_6$	$6.66 \times 10^{-7}$	$3.63833 \times 10^{-6}$	$1.40 \times 10^{-8} \sim$	$6.71 \times 10^{-6}$
$Li_{0.5}C_6$	$3.26075 \times 10^{-8}$	$3.22202 \times 10^{-6}$	$7.74 \times 10^{-12} \sim 7.59 \times 10^{-9}$	$6.66 \times 10^{-6}$
$LiC_6$	$3.98 \times 10^{-9}$	$8.53 \times 10^{-7}$	$7.70 \times 10^{-10}$	$6.83 \times 10^{-6}$

<sup>&</sup>lt;sup>1</sup> Persson K, Sethuraman V A, Hardwick L J, et al. Lithium diffusion in graphitic carbon[J]. The journal of physical chemistry letters, 2010, 1(8): 1176-1180.

<sup>&</sup>lt;sup>2</sup> Kresse, G. Furthmuller, Efficiency of Ab-Initio Total Energy Calculations for Metals and Semiconductors Using a PlaneWave Basis Set. J. Comput. Mater. Sci. 1996, 6, 15–50.