## Supplementary Information for:

## Exploring Electrochemical Performance of Layered Bi<sub>2</sub>Se<sub>3</sub> Hexagonal Platelets as Anode Material for Lithium-Ion Batteries

Shaik M. Abzal<sup>a</sup><sup>‡</sup>, Sumit Khatua<sup>a</sup><sup>‡</sup>, Kurapati Kalyan<sup>a</sup>, Sai Lakshmi Janga<sup>a</sup>, Rajkumar Patel<sup>b,\*</sup>, L.N. Patro<sup>a,\*</sup>, Jatis Kumar Dash<sup>a,\*</sup>

<sup>a</sup> Department of Physics, SRM University-AP, Amaravati 522240, India.

<sup>b</sup> Energy and Environmental Science and Engineering (EESE), Integrated Science and Engineering Division (ISED), Underwood International College, Yonsei University, Yeonsugu, Incheon 21983, South Korea.

‡ These authors have equally contributed to this work.

\*Corresponding Authors

E-mail: [JKD: jatis.d@srmap.edu.in, jatiskumar@gmail.com],

[LNP: <u>laxminarayana.p@srmap.edu.in</u>], [RP: <u>rajkumar@yonsei.ac.kr</u>]



Fig S1. XRD patterns of Bi<sub>2</sub>Se<sub>3</sub> at various synthesis temperatures.



Fig S2. XRD pattern of Bi<sub>2</sub>Se<sub>3</sub> with precursor material, Bi<sub>2</sub>O<sub>3</sub> and Se.

S.no	Peak position (20 °)	FWHM (rad)	Crystallite Size (nm)
1.	9.26	0.0052	26.3
2.	18.62	0.0055	25.2
3.	25.06	0.0032	43.5
4.	29.4	0.0044	32.1
5.	43.7	0.0041	35.6
Average crystallite size			32.5

Table S1. The Average crystallite size of the  $Bi_2Se_3$ .

The average crystallite size of the hexagonal Bi<sub>2</sub>Se<sub>3</sub> is calculated by using the Debye-Scherrer equation,  $D = \frac{k\lambda}{\beta \cos \theta}$ , where *k* represents the Scherrer constant (*k* = 0.9),  $\lambda$  denotes the wavelength of X-rays used (1.54 Å),  $\beta$  refers to the full width at half maximum (FWHM) of the diffraction peak, and  $\theta$  is the Bragg's angle.

Table S2. The morphology of  $Bi_2Se_3$  reported by different synthesis methods.

Synthesis methods	Morphology	Reference
Template synthesis	Micropillars	1

Chemical vapor deposition	Triangular or	2
	hexagonal shape	
Chemical vapor deposition	Nano wire	3
	&Nano ribbons	
Microwave	Nano Sheets	4
assisted		
One-pot reaction	Nano dots	5
Chemical vapor deposition Microwave assisted One-pot reaction	hexagonal shape Nano wire &Nano ribbons Nano Sheets Nano dots	3 4 5

## TEM EDX:



Fig S3. TEM (EDAX) *Energy-dispersive X-ray analysis* of Bi<sub>2</sub>Se<sub>3</sub>.



**Fig S4.** XRD pattern of (a) pure  $Bi_2Se_3$ , Cu sheet,  $Bi_2Se_3$  coated on Cu sheet. (b)  $Bi_2Se_3$  anode materials after 1<sup>st</sup>, 2<sup>nd</sup>, and 25<sup>th</sup> cycle (3V charging) coated on copper sheet.

Figure S4 (b) shows the XRD pattern of  $Bi_2Se_3$  anode materials after 1<sup>st</sup>, 2<sup>nd</sup>, and 25<sup>th</sup> cycle (3V charging) coated on copper sheet. As shown in figure, the sharp peak at 43.5, 50.7, and 74.2° correspond to the peak due to copper sheet.



**Fig S5.** Surface morphology of  $Bi_2Se_3$  after (a)  $1^{st}$  cycle Cu sheet,  $Bi_2Se_3$  coated on Cu sheet. (b),  $2^{nd}$  cycle, (c)  $25^{th}$  cycle.

## **Reference:**

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