

Supporting information

Nonadiabatic *ab initio* chemical reaction dynamics on photoisomerization reaction of 3,5-dimethylisoxazole via the S_1 electronic state

Mizuki Kimura^a and Shinkoh Nanbu*^b

^a Graduate School of Science and Technology, Sophia University, Japan, Chiyoda, Tokyo 102-8554, Japan. Email: m-kimura-0o3@eagle.sophia.ac.jp

^b Faculty of Science and Technology, Sophia University, Chiyoda, Tokyo 102-8554, Japan

3,5-dimethylisoxazole cartesian coordinates (Å) at XMS-CASPT2

C	-1.0807968992	-0.1539346183	0.0000413499
O	-0.6956793940	1.1154687627	0.0002126342
C	1.1180432032	-0.0835435722	0.0000418655
C	0.0072335861	-0.9700101663	-0.0000731874
H	0.0119141080	-2.0458347911	-0.0002195165
N	0.7068855225	1.1861933694	0.0002178806
C	2.5800859095	-0.4061675012	-0.0000002889
H	2.8465515175	-0.9892996825	-0.8830950506
H	3.1666385010	0.5109069911	0.0001125293
H	2.8465596987	-0.9895446218	0.8829170534
C	-2.5518274005	-0.4113644303	0.0000114633
H	-3.0177176837	0.0253751514	-0.8834620528
H	-2.7411160110	-1.4837469412	-0.0001352872
H	-3.0177110631	0.0251321742	0.8836085392

3,5-dimethylisoxazole cartesian coordinates (Å) at SA4-CASSCF

C	-1.0837935534	-0.1563189470	0.0000407423
O	-0.6996231333	1.1212286737	0.0002126098
C	1.1211549246	-0.0802075405	0.0000418025
C	0.0062327251	-0.9709083766	-0.0000737954
H	0.0171859222	-2.0466641533	-0.0002198474
N	0.7127128686	1.1876898265	0.0002184199
C	2.5828337652	-0.4058281698	-0.0000008999
H	2.8459517777	-0.9904278277	-0.8830628445
H	3.1727419093	0.5091853677	0.0001148214
H	2.8459583901	-0.9906763671	0.8828830206
C	-2.5547245554	-0.4114151366	0.0000118302
H	-3.0219810679	0.0242015854	-0.8834272549
H	-2.7436120934	-1.4841907225	-0.0001328988
H	-3.0219742844	0.0239619114	0.8835722108

Table S1(a) The primary CSF for each state and the squared values of transition dipole moment (TDM) at SA4-CASSCF

Electronic State	Primary Electronic Configuration	TDM ² / a ₀ ²
S ₁	$Ryd \leftarrow (\pi_{NO} + \pi_{CCC})^*$	0.119
S ₂	$\pi_{NC}^* + \pi_{OC}^* \leftarrow (\pi_{NO} + \pi_{CCC})^*$	0.437
S ₃	$\pi_{NC}^* + \pi_{OC}^* \leftarrow (n_N + \sigma_{CO+CC})^*$	0.111

Table S1(b) The vertical excitation energy (E_h)

Electronic State	XMS-CASPT2/ cc-pVDZ + <i>sp</i>	SA4-CASSCF/ cc-pVDZ + <i>sp</i>	SAC-CI/ aug-cc-pVDZ	SAC-CI/ cc-pVTZ
S ₀	-323.01051634	-323.53623696	-323.262988	-323.375982
S ₁	-322.77050436	-323.28875461	-323.032850	-323.128486
S ₂	-322.76137757	-323.28280562	-323.024065	-323.120191
S ₃	-322.74817739	-323.26228710	-323.007678	-323.087671

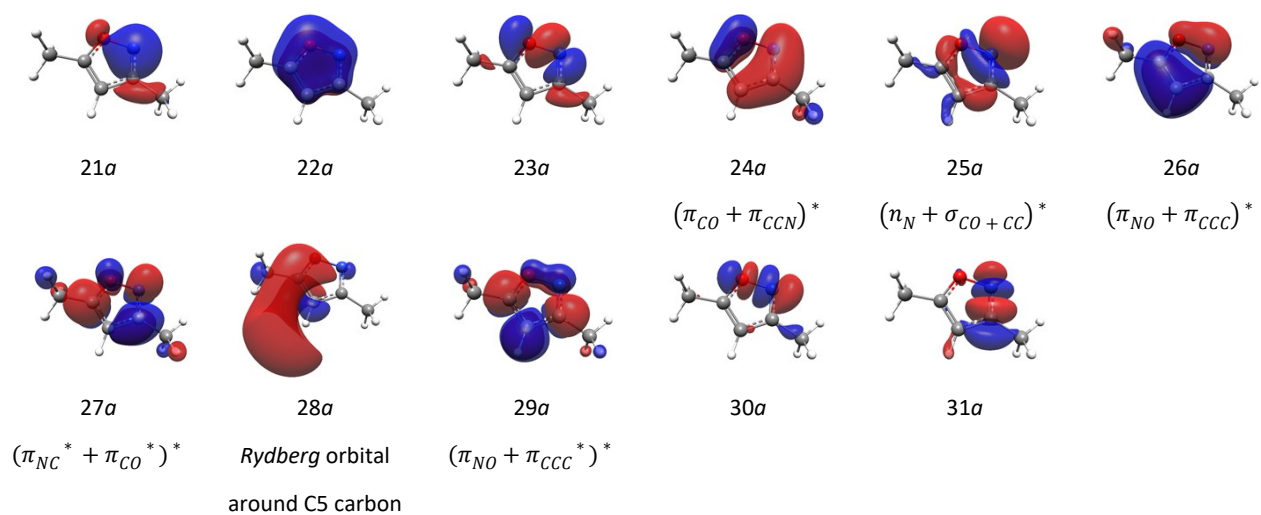


Fig. S1 The molecular orbitals (MOs) in the active space for SA4-CASSCF

Table S2(a) The coefficients of the correlation interaction (CI) for the S_0 , S_1 , S_2 , and S_3 states at XMS-CASPT2 level

Configuration State Function (CSF)	S_0	S_1	S_2	S_3
$(24a)^2(25a)^2(26a)^2(27a)^0(28a)^0(29a)^0$	-0.9387224	0.0000001	-0.0000204	-0.0433679
$(24a)^2(25a)^2(26a)^1(27a)^0(28a)^1(29a)^0$	-0.0000001	0.9269819	0.0040877	-0.0000004
$(24a)^2(25a)^2(26a)^1(27a)^1(28a)^0(29a)^0$	0.0000002	0.0041877	0.9149561	-0.0004300
$(24a)^1(25a)^2(26a)^1(27a)^1(28a)^0(29a)^0$	0.0866112	0.0000018	-0.0003280	-0.6938881
$(24a)^1(25a)^2(26a)^2(27a)^0(28a)^0(29a)^1$	0.0550695	0.0000005	0.0001656	0.3500992
$(24a)^2(25a)^2(26a)^0(27a)^1(28a)^0(29a)^1$	-0.0300139	-0.0000000	0.0001544	0.3331554

Table S2(b) The CI coefficients for each electronic state at SA4-CASSCF level

Configuration State Function (CSF)	S_0	S_1	S_2	S_3
$(24a)^2(25a)^2(26a)^2(27a)^0(28a)^0(29a)^0$	-0.9383068	0.0000001	-0.0439348	0.0000123
$(24a)^2(25a)^2(26a)^1(27a)^0(28a)^1(29a)^0$	-0.0000003	0.9267115	0.0000040	0.0034411
$(24a)^2(25a)^2(26a)^1(27a)^1(28a)^0(29a)^0$	0.0850873	0.0000037	-0.7063145	0.0001956
$(24a)^1(25a)^2(26a)^2(27a)^1(28a)^0(29a)^0$	0.0562196	0.0000008	0.3372313	-0.0000930
$(24a)^1(25a)^2(26a)^2(27a)^0(28a)^0(29a)^1$	-0.0307454	-0.0000003	0.3273773	-0.0000958
$(24a)^2(25a)^1(26a)^2(27a)^1(28a)^0(29a)^0$	0.0000003	0.0043313	0.0002571	0.9139180

Table S3 (a) The first peak of the histogram of distance in comparison by brute force of structures with non-adiabatic transitions



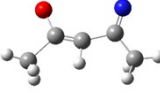
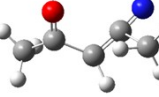



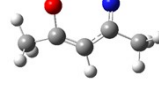
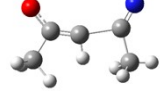

(id, Time / fs)	Distance, d	Probability, p	Geometry1	Geometry2
(021, 74.75), (495, 114.25)	3.24	0.92, 0.88		
(133, 77.75), (385, 372.00)	3.19	1.00, 0.92		
(159, 50.00), (531, 128.00)	3.24	1.00, 0.99		
(397, 55.00), (487, 512)	3.24	0.96, 0.96		
(417, 82.00), (495, 114.25)	3.15	1.00, 0.88		

Table S3 (b) The second peak of the histogram of distance in comparison by brute force of structures with non-adiabatic transitions




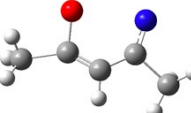

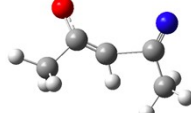
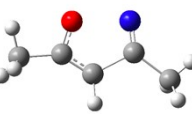
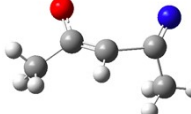
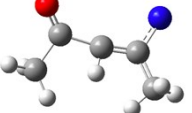
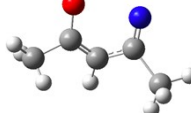
(id, Time / fs)	Distance, d	Probability, p	Geometry1	Geometry2
(105, 343.50), (257, 71.75)	6.62	0.97, 0.68		
(151, 26.00), (403, 127.75)	6.62	0.97, 0.87		
(181, 53.50), (293, 77.25)	6.61	0.88, 0.84		
(263, 216.75), (429, 138.50)	6.63	0.97, 0.92		
(443, 87.00), (595, 261.50)	6.67	0.29, 0.93		

Table S4 (a) The primary peak around $d=18.03$

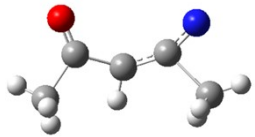

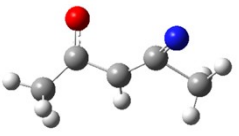

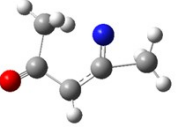
(id, Time / fs)	Distance, d	Probability, p	Geometry
(021, 132.75)	18.15	0.54	
(095, 67.75)	18.21	0.98	
(185, 320.00)	18.17	0.61	
(203, 64.00)	17.90	0.96	
(231, 413.00)	18.06	0.98	

Table S4 (b) The subsidiary peak around $d = 9.98$

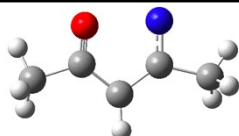
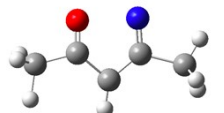


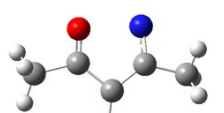
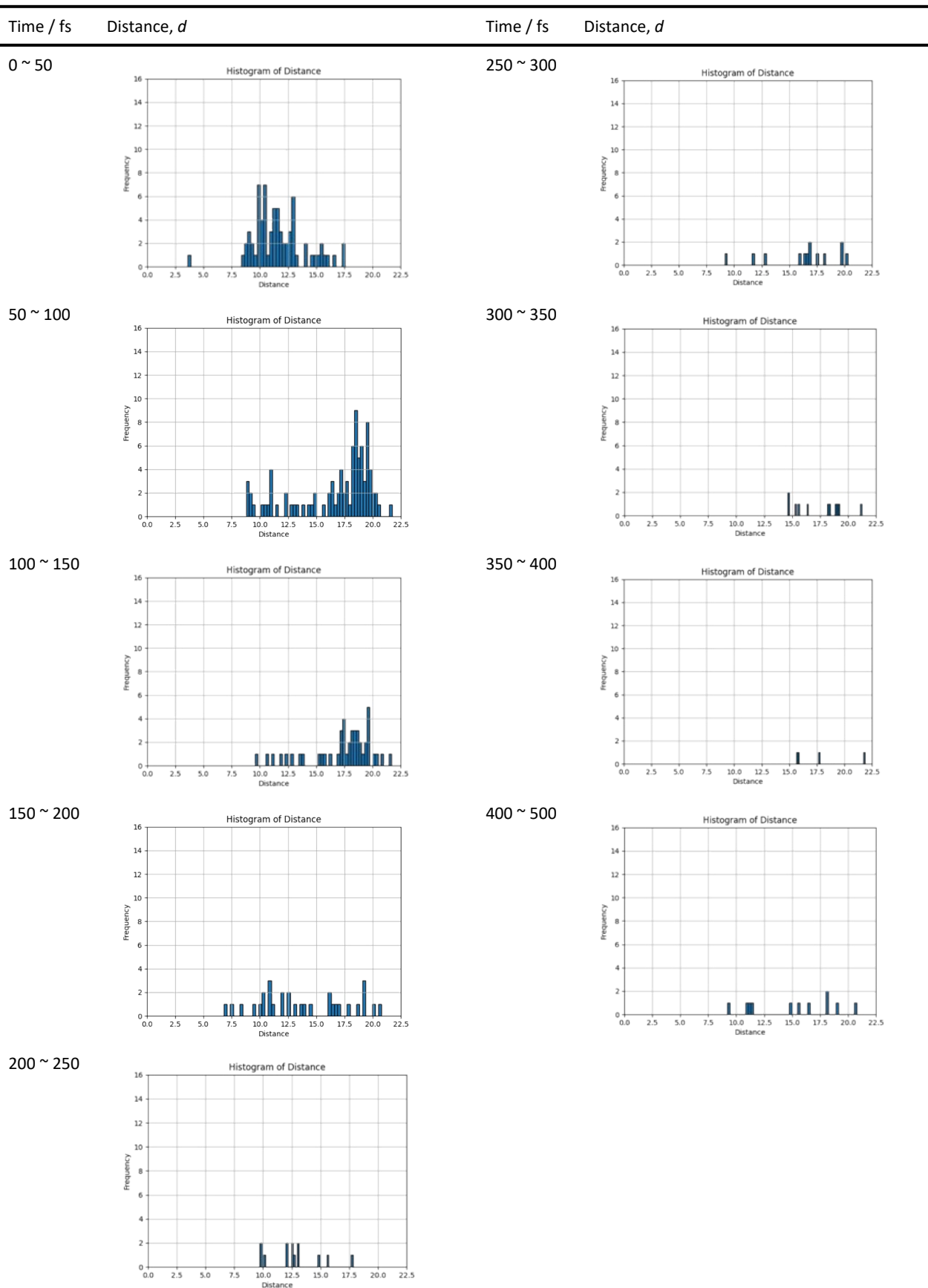
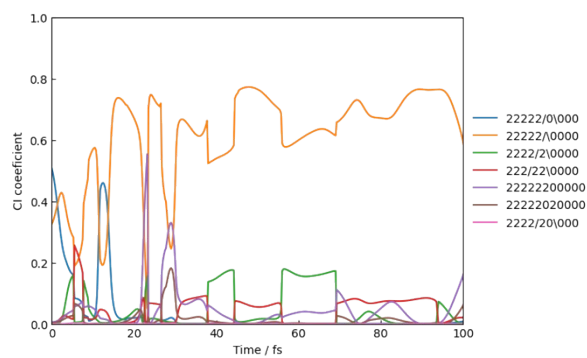
(id, Time / fs)	Distance, d	Probability, p	Geometry
(011, 24.00)	9.98	0.84	
(349, 21.25)	9.91	0.95	
(145, 25.00)	10.07	0.54	
(157, 18.00)	9.87	1.00	
(519, 19.75)	10.09	0.98	

Table S4 (c) Time-resolved histograms.



(a)



(b)

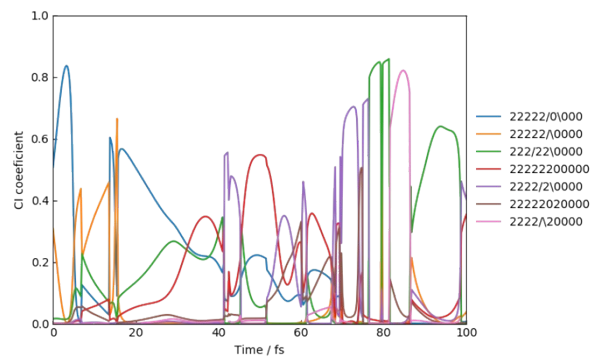


Fig. S2 (a) and (b) are time variations of the CI-coefficients between 100 fs corresponding to Figures 6(b) and 6(d).

Table S5 (a) The time variation for molecular orbitals on the trajectory producing azirine after photoexcitation

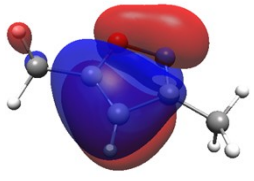
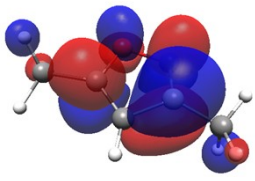
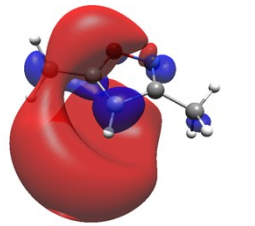
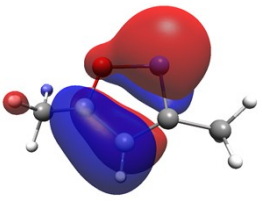
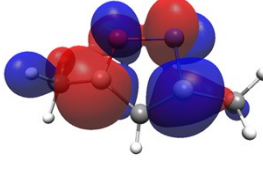
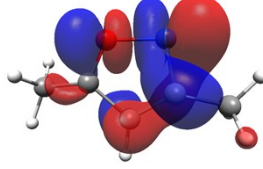
Time / fs	26a	27a	28a
0.00			
10.0			

Table S5(b) The time variation for molecular orbitals on the trajectory producing azirine

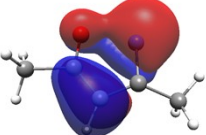
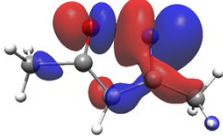
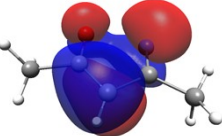
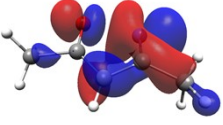
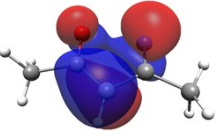
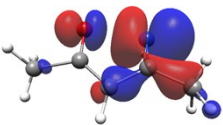
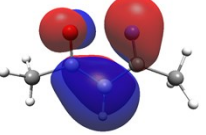
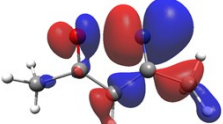
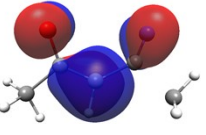
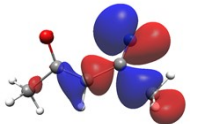
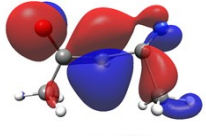
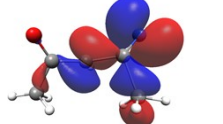
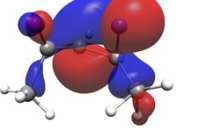

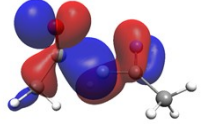
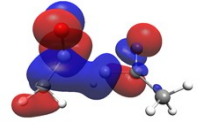
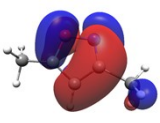
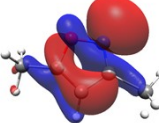
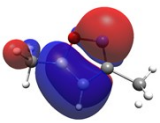
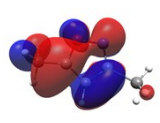
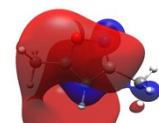
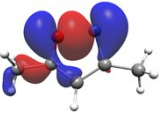
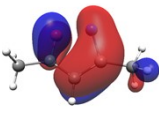
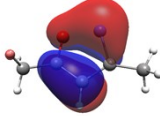
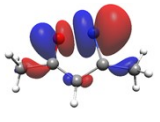
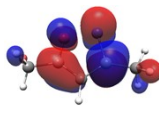
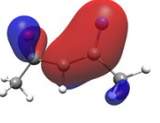
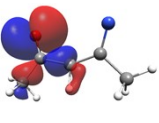
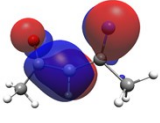
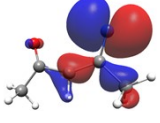
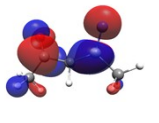
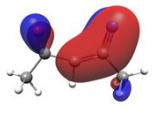
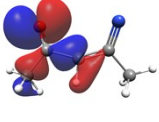
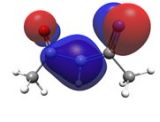
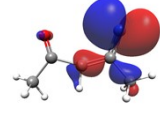
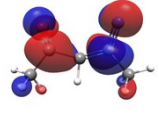
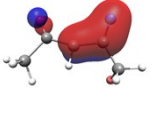
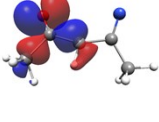
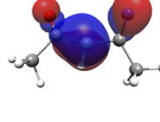
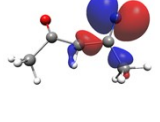
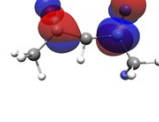
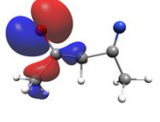
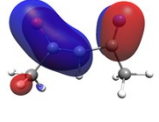
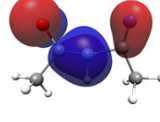
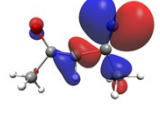
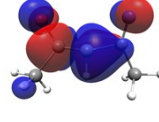
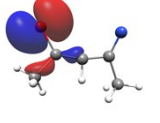
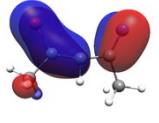
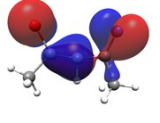
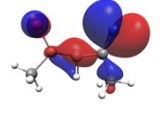
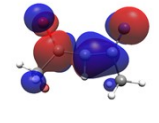
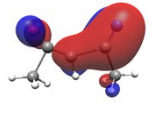
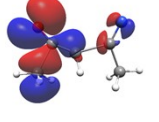
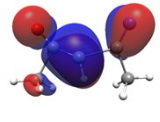
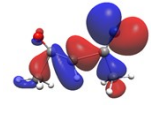
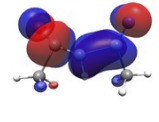
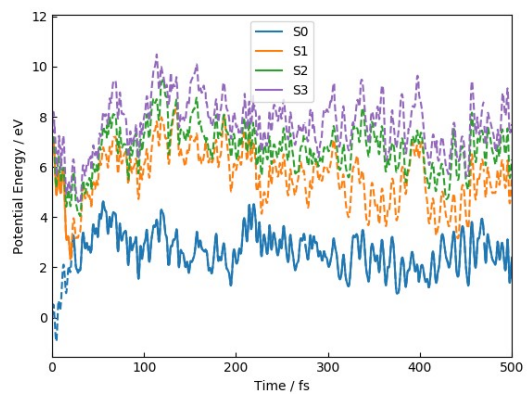
Time / fs	26a	27a
18.75		
22.50		
23.50		
25.00		
50.00		
100.00		
150.00		
250.00		

Table S5(c) The time variation for molecular orbitals on the trajectory producing azirine after photoexcitation

Time / fs	24a	25a	26a	27a	28a
0.00					
15.50					
68.25					
69.75					
74.75					
79.50					
86.50					
104.75					

(a)



(b)

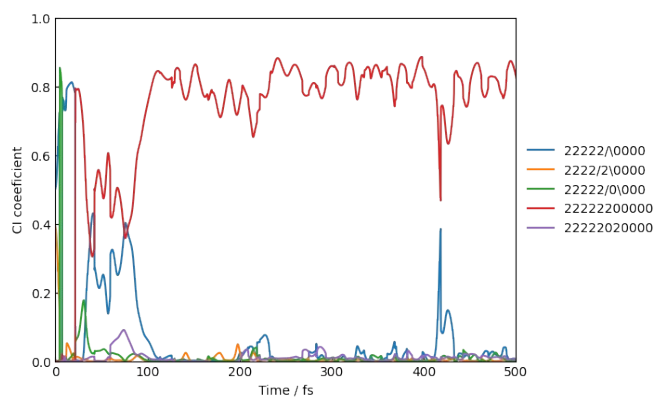


Fig. S3 The isomerization pathways to nitrile ylide product at XMS-CASPT2 level: (a) the potential energies of four electronic states over the time evolution, and (b) CI coefficients over the same time are shown. The legends used in (b) is mentioned in **Fig. 6**.

Table S6(a) The time variation for molecular orbitals on the trajectory producing nitrile ylide and oxazole

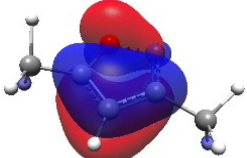
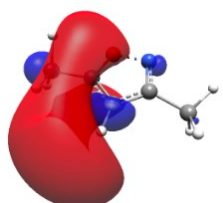
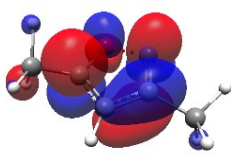
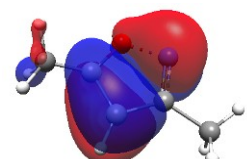
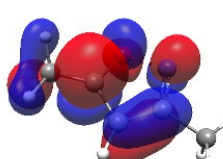
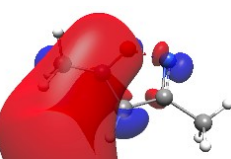
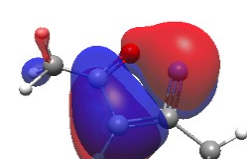
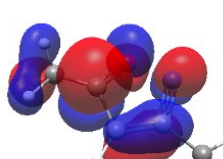
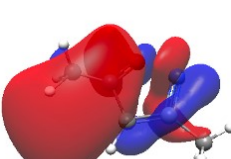
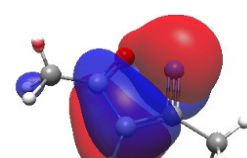
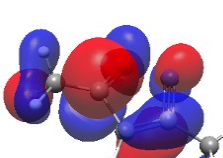
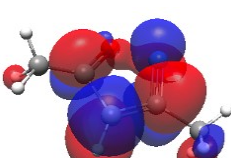
Time / fs	26a	27a	28a
0.00			
4.75			
7.25			
7.50			

Table S6(b) The time variation for molecular orbitals on the trajectory producing nitrile ylide and oxazole

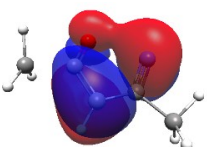
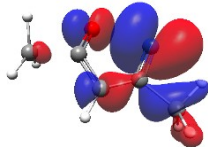
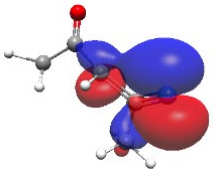
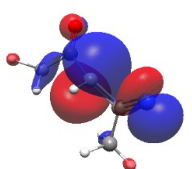
Time / fs	26a	27a
21.50		
76.00		

Table S6(c) The time variation for molecular orbitals on the trajectory producing nitrile ylide

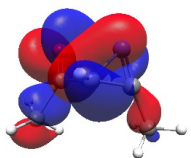
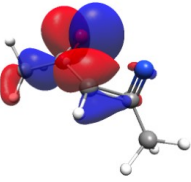
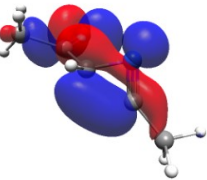
Time / fs	26a
150.00	
175.00	
200.00	

Table S6(d) The time variation for molecular orbitals on the trajectory producing oxazole

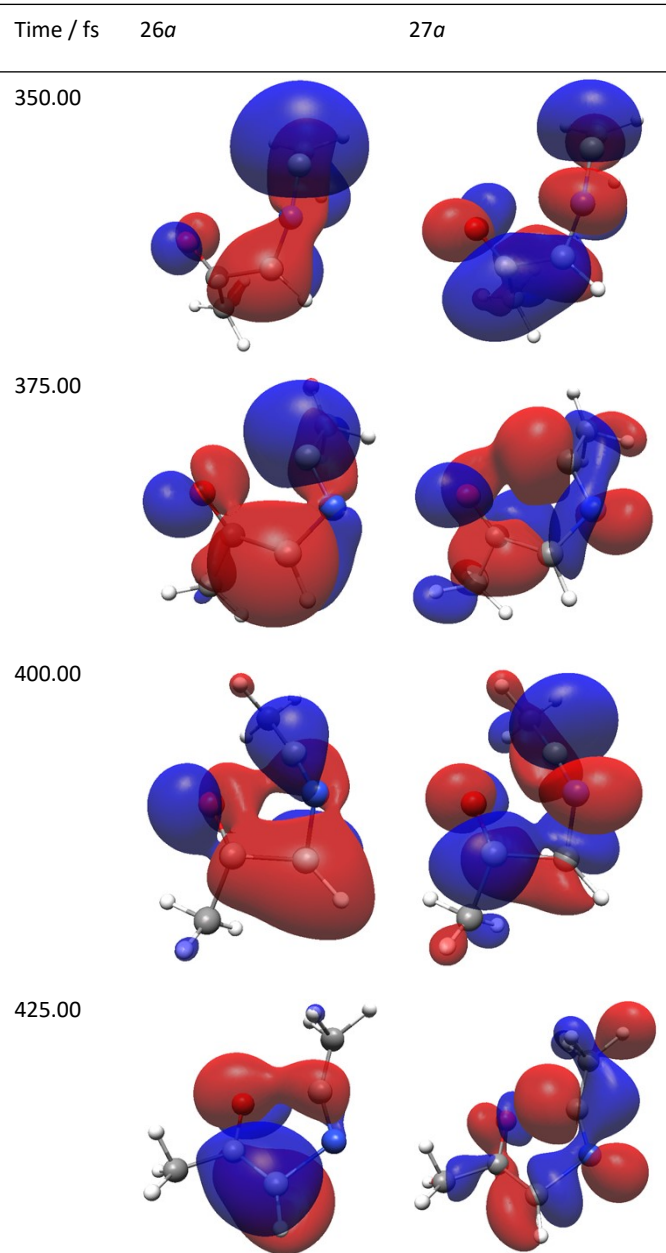


Table S7(a) The time variation for molecular orbitals on the trajectory producing ketenimine

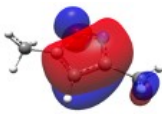
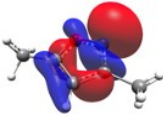
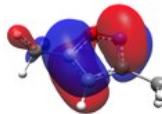
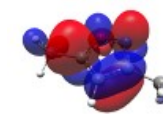

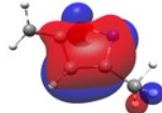
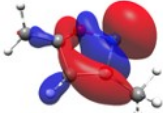
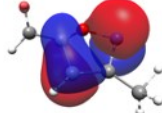
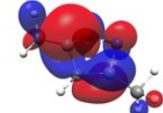
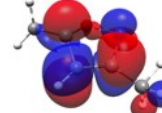
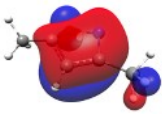
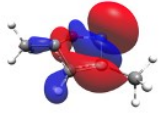
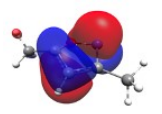
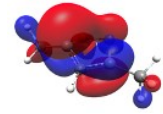
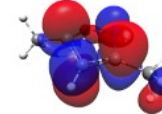

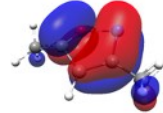
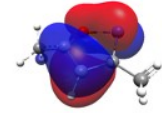
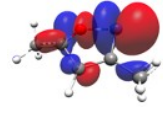
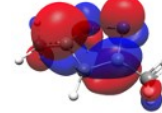
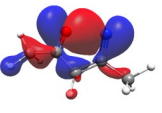
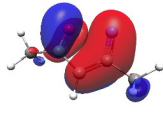
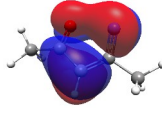
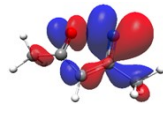

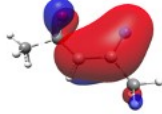
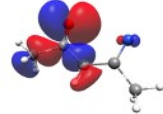
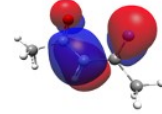
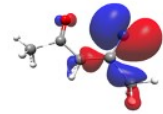
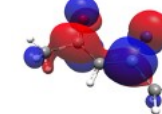
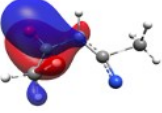
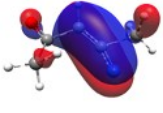
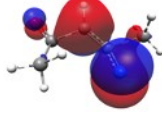
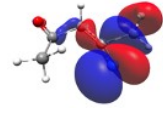
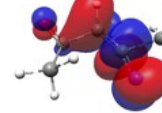
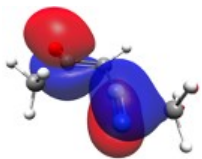
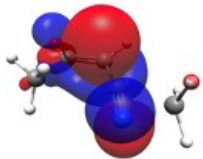
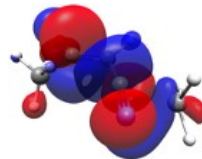

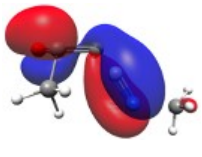
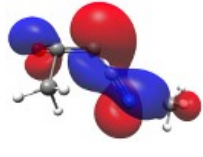

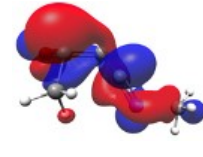
Time/ fs	24a	25a	26a	27a	28a
0.00					
16.75					
28.25					
55.75					
65.00					
92.50					
361.75					

Table S7(b) The time variation for molecular orbitals on the trajectory migrating the methyl group.

Time / fs	25a	26a	27a	28a
375.00				
400.00				

(a)



(b)



26a



27a



28a

(c)



4a



10a



15a



17a



24a

(d)



1a



2a



3a



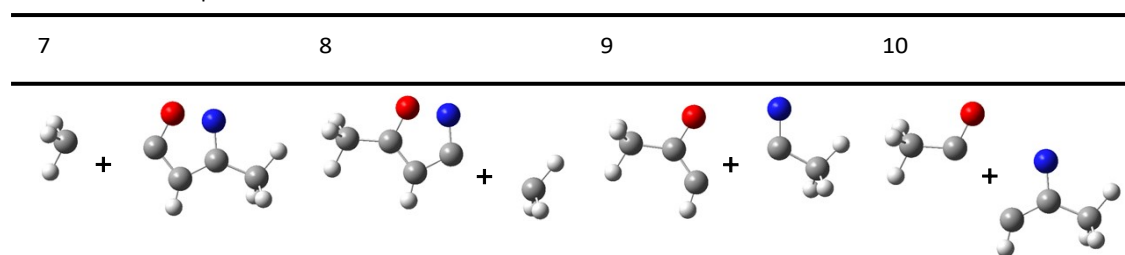
4a



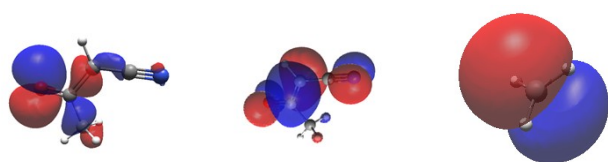
5a

Fig. S4 The methyl shift, (a) the optimized geometry of conical intersection, (b) MOs at the optimized conical intersection, (c) MOs for the methyl migration in our trajectory, and (d) MOs for methyl radical only

Table S8 The other products



(a)



25a

26a

27a

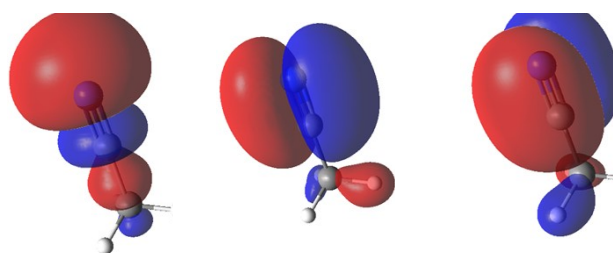
(b)



23a

26a

(c)



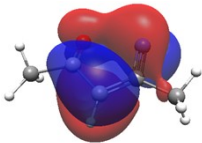
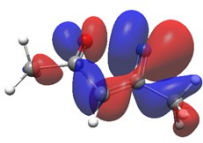
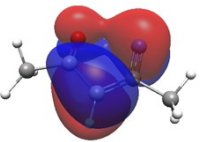
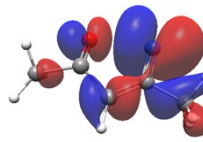
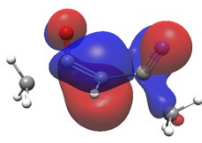
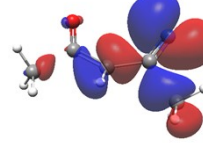
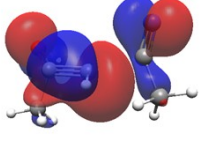
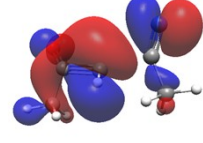
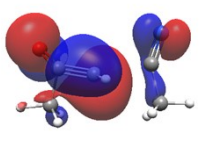
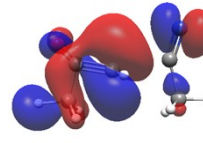
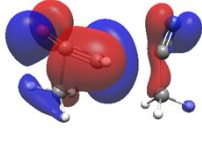
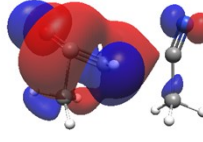
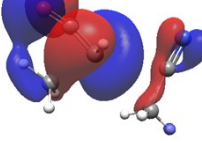
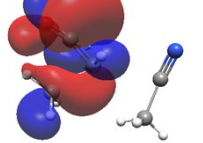
9a

10a

11a HOMO

Fig. S5 The molecular orbitals for the other products for (a) the C3-C7 bond dissociation, (b) CH₃CN + CH₃CCO, and (c) CH₃CN

Table S9 The time variation of molecular orbitals for producing acetonitrile and 1,2 -shift

Time / fs	26a	27a
24.00		
25.00		
50.00		
75.00		
81.25		
87.50		
93.75		
100.00	