

Support Information

First-principles study of photocatalytic performance of bismuth bromide
oxide-based catalysts

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In this study, we used a first-principles-based computational approach to calculate the catalytic performance of $\text{Bi}_x\text{O}_y\text{Br}_z$, specifically using the Vienna Ab initio Computational Simulation Package (VASP) combined with Density Functional Theory (DFT). For the treatment of the interaction of electron exchange and correlation coefficients, we have chosen the generalized gradient approximation (GGA-PBE) proposed by Perdew, Burke and Enzel. In the framework of the projector, we applied the method of augmented waves (PAW) and set the cutoff energy to 500 eV, choosing a plane wave basis set. In addition, we ensure that the energy convergence criterion reaches 1×10^{-4} eV/atom and that the residual force on any atom is less than 0.03 eV to guarantee the accuracy and reliability of the computational results.

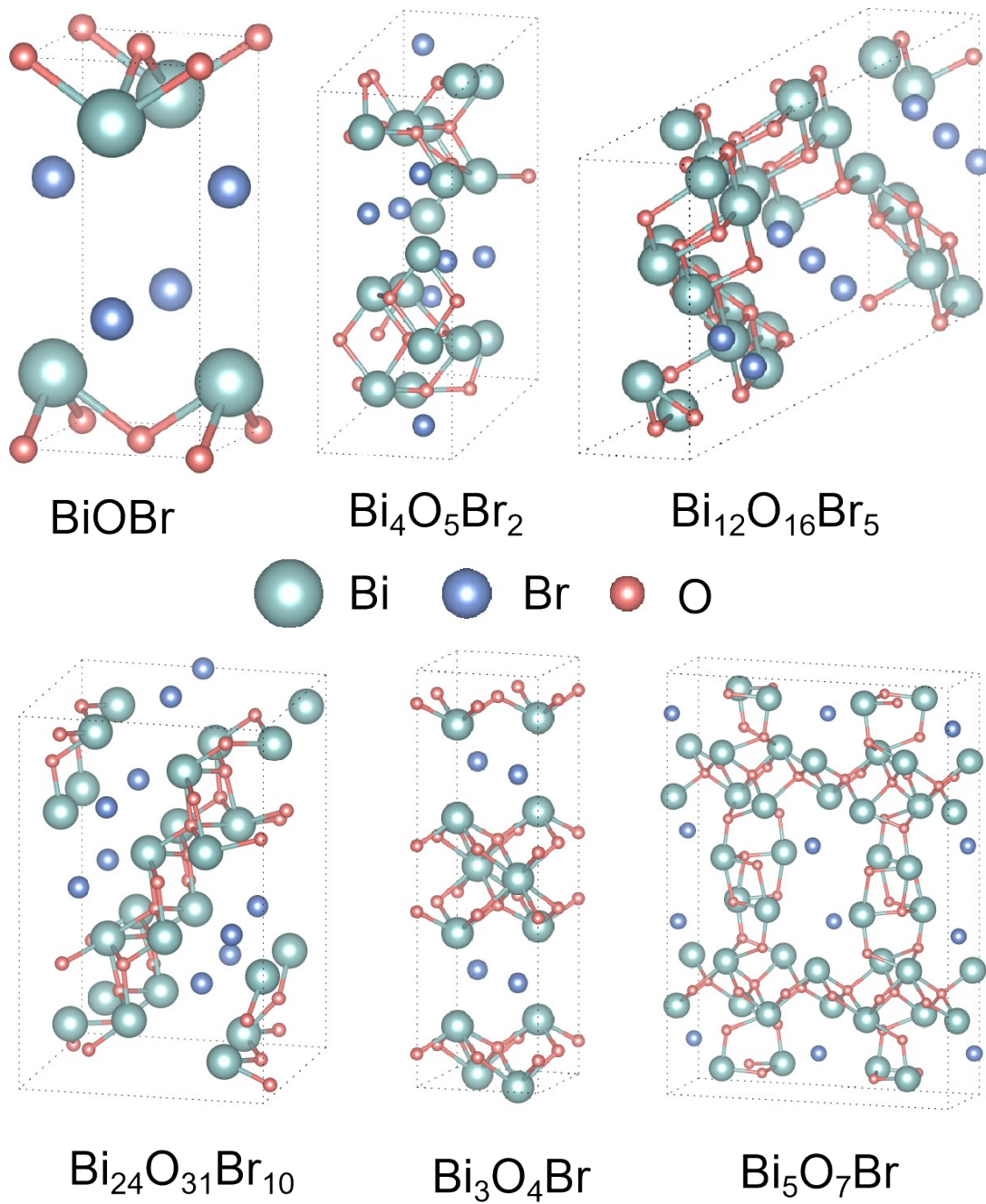


Figure S1 Initial cell structures of BiOBr , Bi_2OBr_2 , $\text{Bi}_3\text{O}_4\text{Br}$, $\text{Bi}_4\text{O}_5\text{Br}_2$, $\text{Bi}_5\text{O}_7\text{Br}$, $\text{Bi}_{12}\text{O}_{16}\text{Br}_5$, $\text{Bi}_{24}\text{O}_{31}\text{Br}_{10}$.

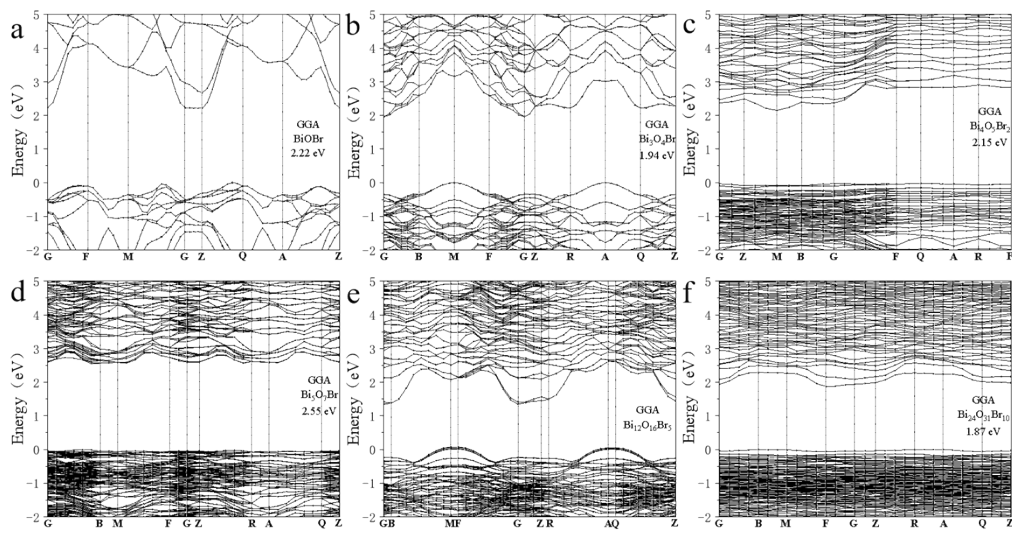


Figure S2 GGA-PBE functional calculated energy bands.

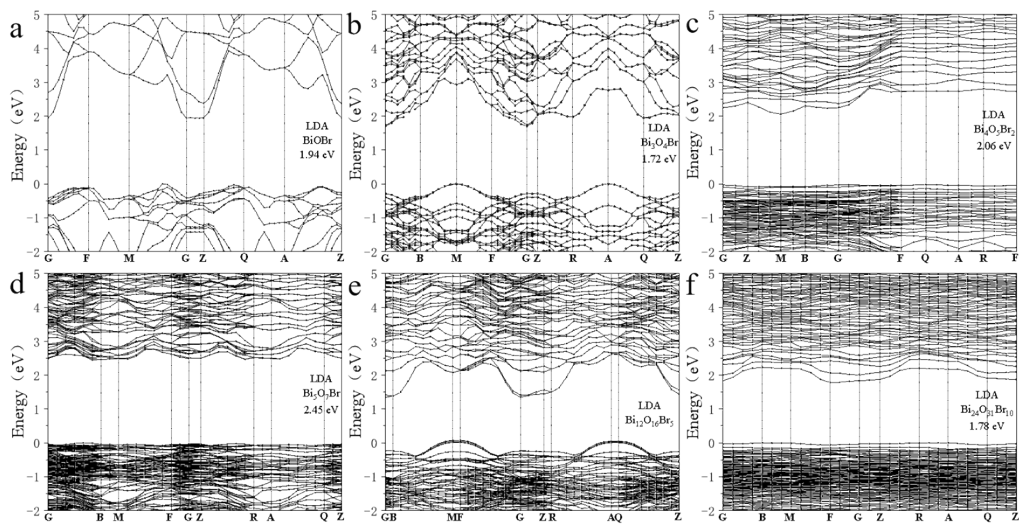


Figure S3 LDA CA-PZ functional calculated energy bands.

Table S1 The calculated lattice parameters and relative errors (%) using first-principles compared with the experimental data.

Lattice	a/Å	b/Å	c/Å	V/Å³	Error(%)
Parameters					
BiOBr¹	3.794	3.923	8.092	124.535	3.6~3.9
Bi₂OBr₂	-	-	-	-	-
Bi₃O₄Br²	5.705	5.693	18.949	615.501	0.3~1.9
Bi₄O₅Br₂³	10.782	5.605	14.539	870.6	0.4~2.6
Bi₅O₇Br⁴	5.296	16.087	23.022	1961.6	2.3~8.6
Bi₁₂O₁₆Br₅	-	-	-	-	-
Bi₂₄O₃₁Br₁₀⁵	4.008	10.13	29.97	1217.5	0.5~7.6

Table S2 Band gap and UV-vis absorption of BiOBr.

Sample	Gap(eV)	UV-vis(nm)	Ref
Ag/BiOBr	2.95	420	6
BiOBr/BiOCl	2.94	421	7
IBP-BiOBr	2.85	435	8
BiOBr/lignin-biocha composites	2.65	442	9
BiOBr-OV	2.58	445	10
BiOBr/BiOCl/C	~2.68	~460	11
TiO ₂ -BiOBr	~3.16	~392	12
BiOBr/Bi	2.85	435	13
BiOBr/Bi ₁₂ O ₁₅ C ₁₆	2.59	~479	14
BiOBr(MP)	2.68	430	15
BiOBr(NS)	2.73	426	15
C/BiOBr	~2.72	~456	16
Bi/BiOBr-OV	2.59	425	17
GO/CDots/BiOBr	2.86	435	18
BiOBr-based	2.78	~446	19
Bi/BiOBr/AgBr	2.7	~459	20
Ag/AgBr/BiOBr	2.65	~468	21
BiOBr-OV microspheres	2.34	~530	22
BiOBr/Ti ₃ C ₂	2.88	430	23
Ag-BiOBr-rGO	2.92	425	24

MWCNT/BiOBr	2.8	~443	25
BiOBr/BiPO₄	2.52	~492	26
Bi/BiOBr	2.7	~459	27
BiOBr	2.75	~451	28
AgBr-Ag-BiOBr	2.8	~443	29
BN/BiOBr	2.78	~446	30
BiOBr-square	2.68	460	31
BiOBr-circle	2.82	440	31
BiOBr/CQDs	~2.82	~440	32
BiOBr-(001)	2.7	~440	33
BiOBr-(010)	2.73	~430	33
up-conversion phosphors/BiOBr	2.81	~441	34
BiOBr/UiO	2.64	443	35
CdS/BiOBr	2.77	447	36
BiOBr/Bi₂WO₆	2.57	~482	37
OVs-BiOBr-30	2.89	~429	38
MIL-125(Ti)/BiOBr	2.66	426	39
BiOBr/ZnWO₄	2.87	450	40
CaCO₃-BiOBr	2.7	416	41
BN/BiOBr-1	2.86	~434	42
Fe-MOF- BiOBr	3.06	~405	43
BiOBr-Bi₂S₃	2.87	420	44

Cu-BiOBr	2.9	~428	45
BN/BiOBr-2	2.8	443	46
2D/2D BiOBr/CDs/g-C₃N₄	2.64	460	47
BiOBr/La₂Ti₂O₇	2.78	430	48
Ag@BiPO₄/BiOBr/BiFeO₃	2.5	450	49
CoAl-LDH/BiOBr	2.69	440	50
Bi₃O₄Br-BiOBr	2.7	430	51
Bi₅O₇Br/BiOBr	2.8	442	52
BiOBr@Bi₄O₅Br₂@Bi₅O₇Br	2.86	~434	53

Table S3 Band gap and UV-vis absorption of Bi₄O₅Br₂.

Sample	Gap(eV)	UV-vis(nm)	Ref
Bi₄O₅Br₂/Bi₂₄O₃₁Br₁₀	2.53	499	54
BiOBr@Bi₄O₅Br₂@Bi₅O₇Br	2.28	544	53
Bi₄O₅Br₂-UN	2.64	469	55
CDs/Bi₄O₅Br₂	2.41	455	56
Ag/AgBr/Bi₄O₅Br₂	2.4	517	57
I-Bi₄O₅Br₂	2.62	486	58
Bi₄O₅Br₂	2	~620	59
CdS/Bi₄O₅Br₂	2.31	537	60

Bi₄O₅Br₂/g-C₃N₄	2.98	~416	61
Bi₄O₅Br₂-GO	2.26	450	62
Bi₄O₅Br₂-1	2.28	~544	63
Bi₄O₅Br₂ NSV	2.42	~512	64
Bi₄O₅Br₂/g-C₃N₄-1	2.18	550	65
Bi₄O₅Br₂/ZIF-8	2.16	570	66
Bi₄O₅Br₂@GO	2.63	471	67
Bi-Bi₄O₅Br₂	2.38	~521	68
Bi₄O₅Br₂ nanosheets	2.33	~532	69
Bi₄O₅Br₂ Nano	2.37	~523	70
Bi₄O₅Br₂ BC	2.31	~537	71
AgI/MoSe₂/Bi₄O₅Br₂	2.62	474	72
Bi₄O₅Br₂@TiO₂	2.45	~506	73
Bi₄O₅Br₂-I	2.46	503	74
3D flower-sphere Bi₄O₅Br₂	2.45	492	75
WO₃/Bi₄O₅Br₂	2.27	~546	76
Bi-based Bi_xO_yBr_z(Bi₄O₅Br₂)	2.57	483	77
Bi₂S₃/Bi₄O₅Br₂	2.53	~490	78
Bi₄O₅Br₂/graphene	2.54	~488	79
Bi₄O₅Br₂/ AgBr	2.53	490	80
Bi plasmonic/Bi₄O₅Br₂	2.41	~515	81
3D Bi₄O₅Br₂	2.5	490	82

Bi₄O₅Br₂-MoS₂	2.43	~510	83
Bi₄O₅Br₂/Bi₄O₅I₂/Fe₃O₄	2.53	490	84
Bi₄O₅Br₂-Bi	2.21	470	85
Bi₄O₅Br₂/Mn_xZn_{1-x}Fe₂O₄	2.43	~510	86
Bi₄O₅Br₂ g-C₃N₅	2.62	~473	87
Fe-Bi₄O₅Br₂	2.26	500	88
CuTCPP/Bi₄O₅Br₂	2.25	~551	89
Bi₄O₅Br₂-OVs	2.44	~508	90
Bi/Bi₄O₅Br₂	2.25	~551	91
Bi₁₂O₁₇Br₂/Bi₄O₅Br₂	2.71	440	92
Bi₄O₅Br₂/α-MnS	2.29	~541	93
Bi₄O₅Br₂/NH₂-MIL-125(Ti)	2.25	500	94

Table S4 Band gap and UV-vis absorption of Bi₂₄O₃₁Br₁₀.

Sample	Gap(eV)	UV- vis(nm)	Ref
Bi₃O₄Br-Bi₂₄O₃₁Br₁₀	2.07	550	95
Bi₄O₅Br₂/Bi₂₄O₃₁Br₁₀	2.5	514	54
Bi₂₄O₃₁Br₁₀ NBs	2.4	~517	96
Bi₂₄O₃₁Br₁₀ NSs	2.47	~502	96

AgI-Bi₂₄O₃₁Br₁₀	2.53	491	97
Bi₂₄O₃₁Br₁₀(OH)δ	~2.5	~496	98
One-pot Bi₂₄O₃₁Br₁₀	2.64	~470	99
CdS-Bi₂₄O₃₁Br₁₀	2.6	473	100
Bi₂₄O₃₁Br₁₀ nanoflakes	2.51	~494	101
BiOBr/Bi₂₄O₃₁Br₁₀/TiO₂	2.64	470	102
BOB-S	2.3	~539	103
BOB-M	2.33	~532	103
BOB-L	2.35	~528	103
BUC-21/Bi₂₄O₃₁Br₁₀	2.65	~468	104
Bi₂₄O₃₁Br₁₀-nanobelts	2.45	~506	105
Bi₂₄O₃₁Br₁₀-nanosheets	2.53	~490	105
Bi₂₄O₃₁Br₁₀/SrTiO₃	2.79	465	106
Bi₄O₅Br₂/Bi₂₄O₃₁Br₁₀/Bi₂SiO₅	2.79	444	107
Bi₂₄O₃₁Br₁₀ PSQDs	2.83	~438	108
VBi-Bi₂₄O₃₁Br₁₀	2.87	~432	109
Bi₂₄O₃₁Cl_xBr_{10-x}(Bi₂₄O₃₁Br₁₀)	2.68	~463	110
Bi₂₄O₃₁Br₁₀-pH	2.4	500	111

Table S5 Band gap and UV-vis absorption of Bi₃O₄Br.

Sample	Gap(eV)	UV-vis(nm)	Ref
AgBr/Bi ₃ O ₄ Br	2.55	~486	112
CQD-Bi ₃ O ₄ Br	2.68	463	113
Bi ₃ O ₄ Br/BiOCl	2.42	512	114
AgI/Bi ₃ O ₄ Br	2.4	550	115
bulk Bi ₃ O ₄ Br	2.42	~512	116
Bi ₃ O ₄ Br- Bi ₂₄ O ₃₁ Br ₁₀	2.25	600	95
Bi ₃ O ₄ Br-BiOBr	2.4	467	51
Bi ₃ O ₄ Br nanoplates	2.64	470	117
Bi ₃ O ₄ Br/CuBi ₂ O ₄	2.83	~438	118
α -Bi ₂ O ₃ /Bi ₃ O ₄ Br	2.65	~468	119
Bi ₃ O ₄ Br/NH ₂ -MIL-125(Ti)	2.5	476	120
C-Bi ₃ O ₄ Br	2.31	537	121
Bi ₃ O ₄ Br/Bi ₂ O ₃	2.58	~481	122
C-Bi ₃ O ₄ Br-1	2.5	538	123

Table S6 Band gap and UV-vis absorption of Bi₅O₇Br.

Sample	Gap(eV)	UV-vis(nm)	Ref
Bi-Bi ₅ O ₇ Br	2.38	521	124
Bi ₅ O ₇ Br/BiOBr	2.38	507	52
Bi ₂ MoO ₆ /Bi ₅ O ₇ Br	2.31	~537	125

Bi₅O₇Br	3.13	396	126
5-Bi₅O₇Br-NT	2.89	~429	127
Bi₅O₇Br_{0.5}I_{0.5}(Bi₅O₇I, Bi₅O₇Br)	2.21	441	128
Bi₅O₇Br/NiFe-LDH	2.65	520	129
Bi₅O₇Br nanowires	2.98	~416	130
Ag/Bi₅O₇Br	2.22	517	131
BiOBr@Bi₄O₅Br₂@Bi₅O₇Br	2.64	470	53

The formation energy is calculated using the following formula:

$$E_{\text{formation}} = E_{\text{compound}} - \sum n_i \mu_i$$

where E_{compound} is the total energy of the compound, μ_i is the chemical potential of the constituent element, and n_i is the number of elements in the compound. The chemical potential μ_i is the energy obtained from the element's monomers or its stable phase, depending on the conditions of the calculation.

Table S7 The formation energy of Bi_xO_yBr_z.

	Number of atoms	Formation Energy (eV / atom)	Method
BiOBr	6	-1.55	Meta-GGA

Bi₂OBr₂	20	-0.46	Meta-GGA
Bi₃O₄Br	32	-1.53	Meta-GGA
Bi₄O₅Br₂	44	-1.55	Meta-GGA
Bi₅O₇Br	104	-1.53	Meta-GGA
Bi₁₂O₁₆Br₅	66	-1.51	Meta-GGA
Bi₂₄O₃₁Br₁₀	65	-1.51	Meta-GGA

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