

Supplementary Information

Enhanced NH₃ and NO Sensing Performance of Ti₃C₂O₂ MXene by biaxial strain: Insights from First-Principles Calculations

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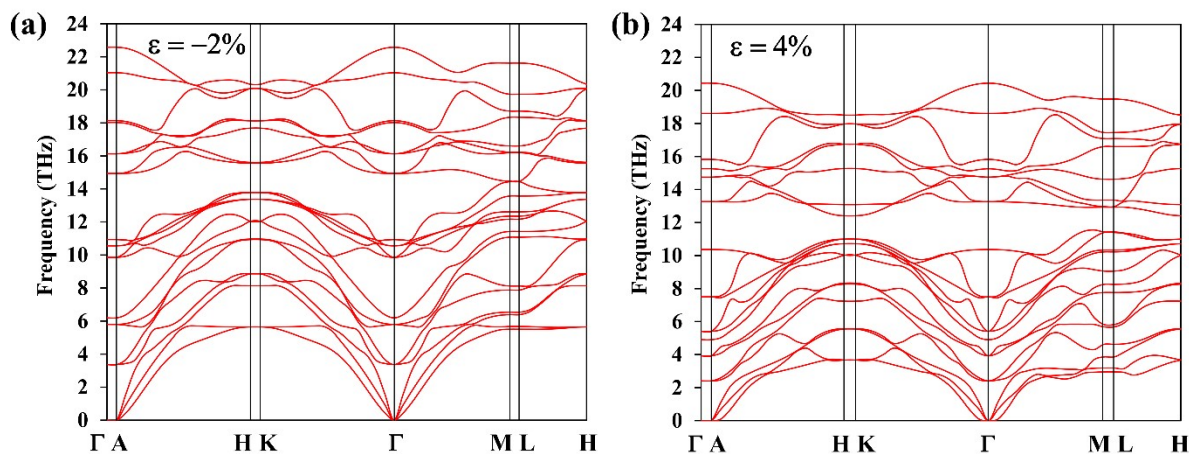


Fig. 1S Phonon dispersion curves of Ti₃C₂O₂ at strains of -2% and 4% represented by panel (a), and (b), respectively.

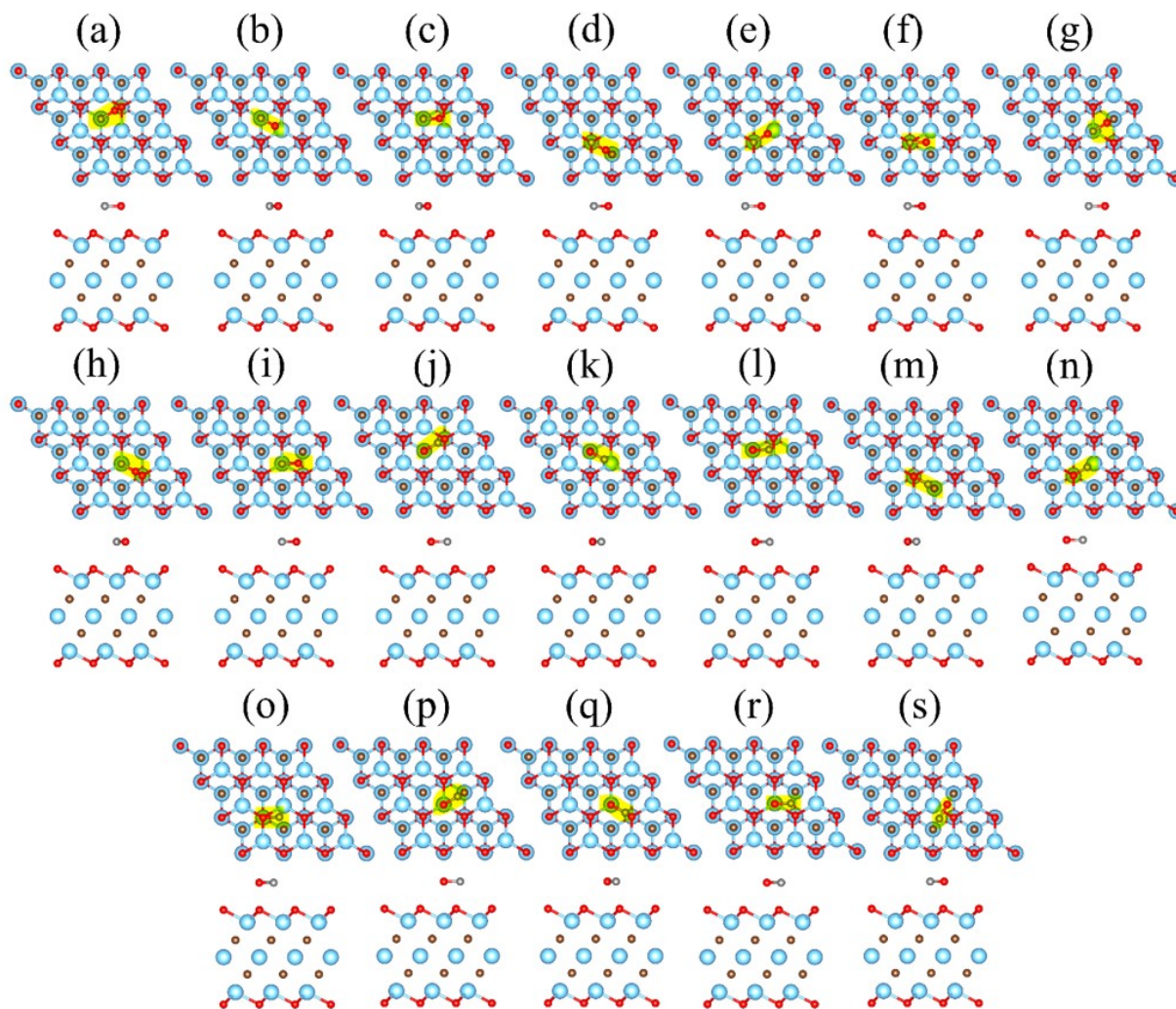


Fig. 2S Top and side views of the possible adsorption sites of CO and NO on $\text{Ti}_3\text{C}_2\text{O}_2$. The azure, brown, red, and grey balls represent the Ti, C, O, and C or N of gases, respectively. The yellow-highlighted area in the top view clearly indicates the position of the gases.

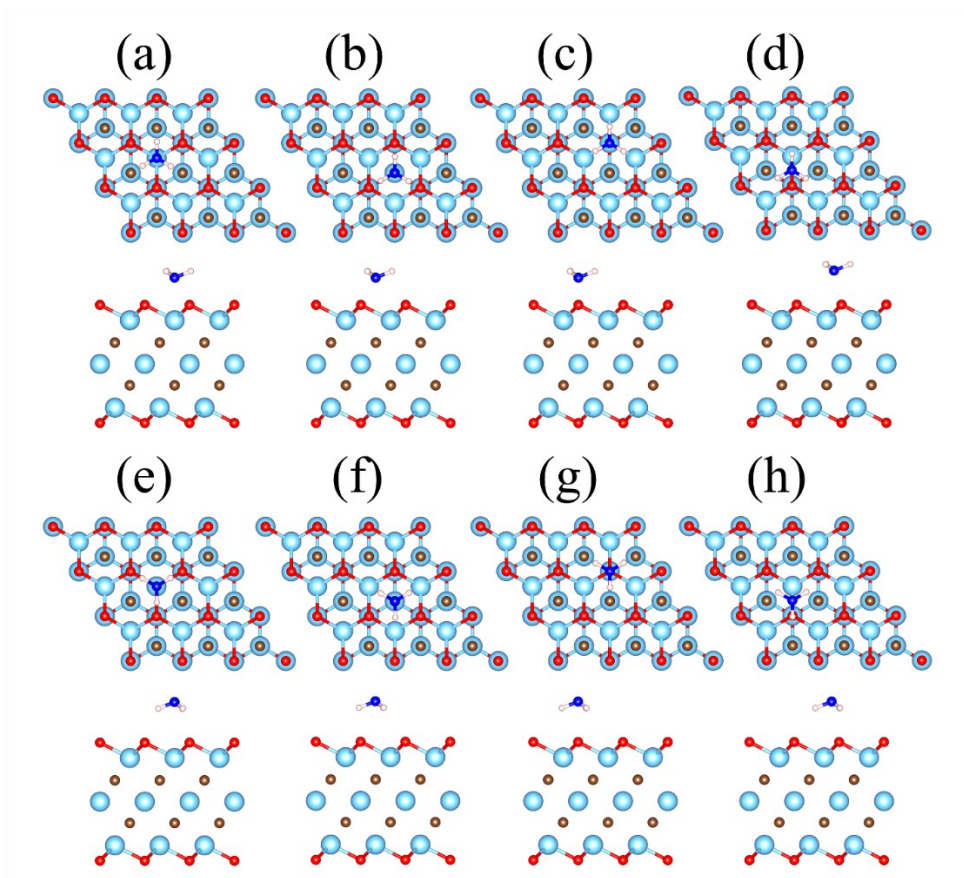


Fig. 3S Top and side views of the possible adsorption sites of NH_3 on $\text{Ti}_3\text{C}_2\text{O}_2$. adsorption site configuration on $\text{Ti}_3\text{C}_2\text{O}_2$ surface. The azure, brown, red, blue, and pink balls represent the Ti, C, O, N, and H, respectively.

Table 1S The Mulliken charge analysis of CO, NH₃, and NO before and after adsorption on Ti₃C₂O₂.

	Gases	Species	States		Charge (e)
			s	p	
Before	CO	C	1.68	1.90	0.42
		O	1.84	4.57	-0.42
	NH₃	N	1.75	4.49	-1.23
		H	0.59	-	0.41
		H	0.59	-	0.41
	NO	N	1.80	3.07	0.13
		O	1.84	4.29	-0.13
	After	CO	C	1.66	1.95
O			1.84	4.53	-0.37
NH₃		N	1.68	4.32	-1.00
		H	0.65	-	0.35
		H	0.65	-	0.35
NO		N	1.79	2.97	0.24
		O	1.84	4.20	-0.05

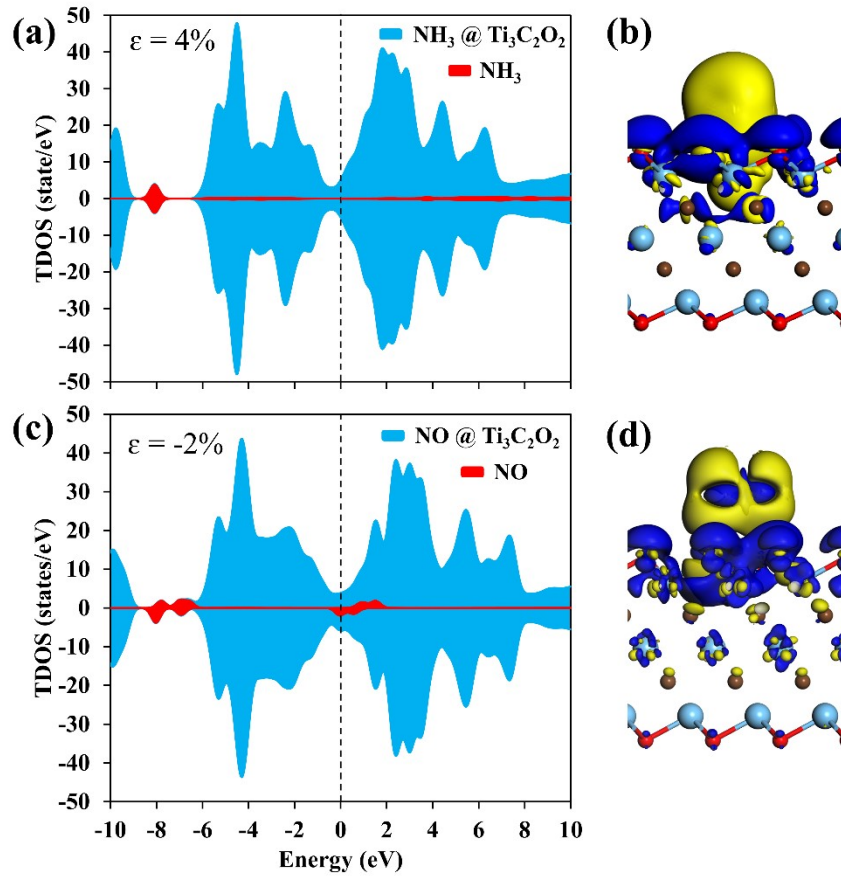


Fig. 4S Electronic density of state (TDOS) and electron differences density (EDD) of (a, b) NH₃ adsorbed on Ti₃C₂O₂ at 4% strain and (c, d) NO adsorbed on Ti₃C₂O₂ at -2% strain.

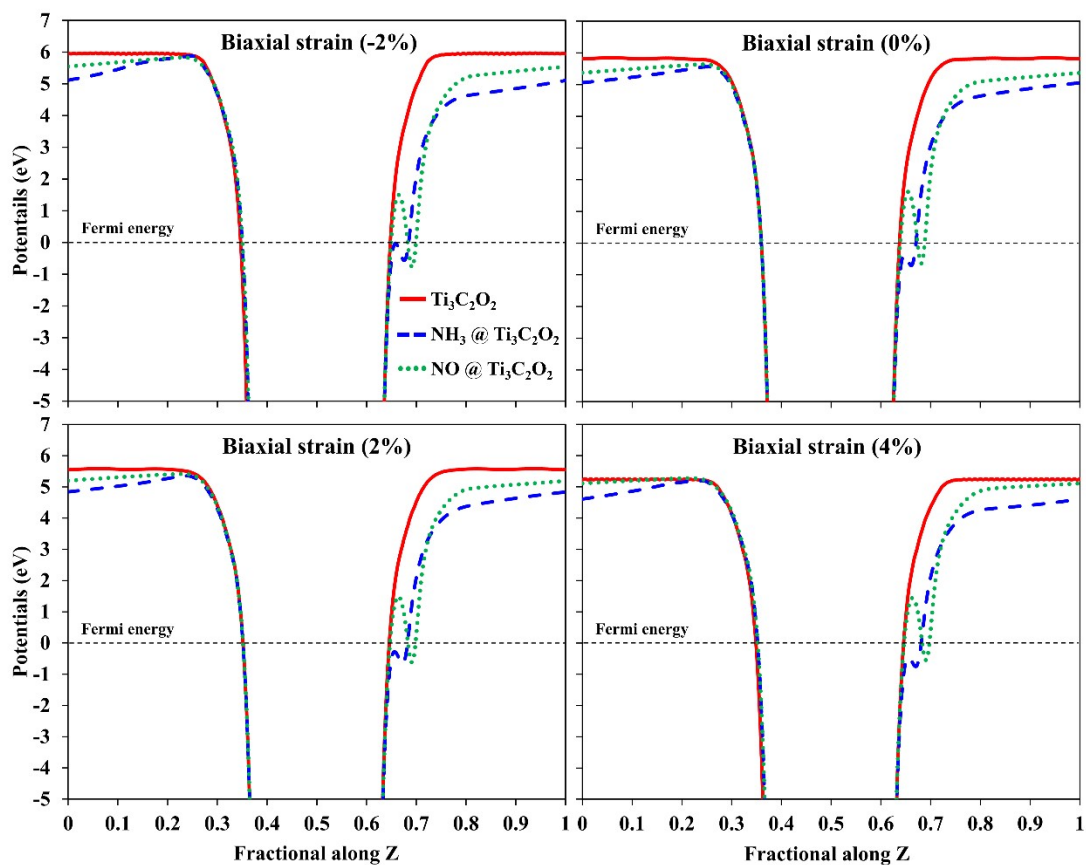


Fig. 5S Electrostatic potentials of toxic gas molecules adsorption on $\text{Ti}_3\text{C}_2\text{O}_2$ under biaxial strain -2% to 4%.