Supplementary Information

Evolution of Silicate Coordination in Architected Amorphous and Crystalline Magnesium Silicates during Carbon Mineralization

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Mg-	Unreacte	H ₂ O,	NaHCO ₃ ,	H ₂ O, No-	NaHCO ₃ ,
silicates	d	Stirring	Stirring	stirring	No-stirring
AC	13.05 µm	99.15 µm	31.95 µm	14.03 μm	9.61 μm
HC	25.04 µm	15.78 μm	8.01 µm	14.32 μm	7.99 μm

Table S1. Mean particle sizes of the unreacted and carbonate-bearing silicate sorbents.



Figure S1. The carbon mineralization extent of MgO without stirring at 200 °C, 20 atm.



Figure S2. The deconvoluted Si2p spectrum of the (a-1) AC Mg-silicate, (a-2) carbonate-bearing AC Mg-silicate with DI-water, (a-3) carbonate-bearing AC Mg-silicate with NaHCO₃ and (b-1) HC Mg-silicate, (b-2) carbonate-bearing HC Mg-silicate with DI-water, (b-3) carbonate-bearing HC Mg-silicate with NaHCO₃, in stirring mode at 200 °C, 20 atm determined by X-ray photoelectron spectroscopy.



Figure S3. The deconvoluted Si2p spectrum of the (a) carbonate-bearing AC Mgsilicate with DI-water, (b) carbonate-bearing AC Mg-silicate with NaHCO₃, (c) carbonate-bearing HC Mg-silicate with DI-water, and (d) carbonate-bearing HC Mgsilicate with NaHCO₃, in no-stirring mode at 200 °C, 20 atm determined by X-ray photoelectron spectroscopy.



Figure S4. The concentration of Mg ions in the collected aqueous samples determined by Inductively Coupled Plasma (ICP) analysis.



Figure S5. Structural arrangement of (a) carbonate-bearing AC Mg-silicate with DIwater, (b) carbonate-bearing AC Mg-silicate with NaHCO₃, (c) carbonate-bearing HC Mg-silicate with DI-water, and (d) carbonate-bearing HC Mg-silicate with NaHCO₃ in no-stirring mode at 200 °C, 20 atm determined by X-ray diffraction measurements.



Figure S6. Morphologies determined using Scanning Electron Microscopy (SEM) of: (a) carbonate-bearing AC Mg-silicate with DI-water, (b) carbonate-bearing AC Mgsilicate with NaHCO₃, (c) carbonate-bearing HC Mg-silicate with DI-water, and (d) carbonate-bearing HC Mg-silicate with NaHCO₃ in no-stirring mode. Experiments are conducted at 200 °C, pCO₂ of 20 atm for a reaction time of 3 hours.