

Supplementary Information

Covalent organic frameworks as superior adsorbents for the removal of toxic substances

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Table S1 Properties and pathogenesis of typical toxicants

Type of toxicants	Toxicants	Physical and chemical properties	Toxicity endpoint or target sites	Toxic mechanism	Ref.
	Urea	<ul style="list-style-type: none">- An organic compound that amide exhibits two $-NH_2$ groups- Water soluble- Four hydrogen donor sites	<ul style="list-style-type: none">- Chronic kidney disease- Uremia- Cardiovascular disease- Depression- Diabetes mellitus	<ul style="list-style-type: none">- Cause excessive formation of neutrophil extracellular traps, leading to endothelial cell dysfunction- Induce mitochondrial reactive oxygen species (ROS), thereby inhibiting GAPDH, inactivating the anti-atherogenic enzyme PGI2 synthase and causing ER stress- Urea or cyanate carbamylates mTOR to inhibit the mTORC1-S6K dependent dendritic protein synthesis, causing depression-like behavior	1-4
	Creatinine (CR)	<ul style="list-style-type: none">- Nitrogen-containing small molecule toxin- Water soluble- Two hydrogen donor sites- Namely trioxopurine	<ul style="list-style-type: none">- Uremia- Renal failure	<ul style="list-style-type: none">- Decrease glomerular filtration rate- Accumulation in the body, accelerate the decline of renal function	1
	Uric acid (UA)	<ul style="list-style-type: none">- Water soluble- Alcoholic form is weakly acidic	<ul style="list-style-type: none">- Uremia- Chronic kidney disease- Gout	<ul style="list-style-type: none">- When the level of UA in the blood is high, UA crystals form, which accumulate in the joints and kidney, causing arthritis, gout and uremia- Increase oxidative stress and endothelial dysfunction	1,5

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Indoxyl sulfate (IS)	<ul style="list-style-type: none"> - High protein-binding properties - 32% of IS can be cleared by dialysis - Fold change (M/N): 43.2 	<ul style="list-style-type: none"> - Nephrotoxicity - Cardiovascular toxicity - Disordered bone metabolism - Intestinal barrier injury 	<ul style="list-style-type: none"> - ROS production and induction of oxidative stress and renal interstitial fibrosis/inflammation - Increase transcription factors expression, promote thrombosis and cause damage to cardiovascular system - IS-induced oxidative stress in osteoblasts initiates the dysregulation of bone remodeling process in early CKD - IS suppressed transepithelial electrical resistance and the expressions of tight junction-related genes in intestinal epithelial cells, inhibited the expression of dynamin-related protein 1 (DRP1) and mitophagic flux - Regulate apoptosis-related genes, induce neuronal apoptosis 	6-9
Amyloid- β peptide ($A\beta$)	<ul style="list-style-type: none"> - The length of the $A\beta$ peptide varies from 38-43 amino acids - The predominant form of $A\beta$ peptide is $A\beta_{40}$ and $A\beta_{42}$ 	<ul style="list-style-type: none"> - Neurotoxicity - Alzheimer's disease (AD) 	<ul style="list-style-type: none"> - Induce inflammatory cascade response by activating microglia - Induce abnormal elevations in glutamate levels, disrupt intracellular calcium balance, impair mitochondria dysfunction, and trigger oxidative stress, leading to protein and lipid oxidation 	10,11
Aflatoxins (AFs)	<ul style="list-style-type: none"> - Derivatives of difuran-coumarin - Almost insoluble in water, but soluble in most organic solvents - High thermal stability 	<ul style="list-style-type: none"> - Hepatotoxicity - Immunotoxicity - Carcinogenicity - Teratogenic and mutagenic effect - Genotoxicity - Complications associated with malnutrition - Developmental delay 	<ul style="list-style-type: none"> - AFs are metabolized under the action of cytochrome P450 microsomal enzymes, especially AFB_1 is converted into aflatoxin-8,9-epoxide that combines with DNA and albumin in serum to form adducts, ultimately leading to DNA damage - Modulation of cytokine expression - Alter intestine integrity 	12-14

	Patulin (PAT)	<ul style="list-style-type: none"> - An unsaturated heterocyclic lactone mycotoxin 	<ul style="list-style-type: none"> - Genotoxicity - Neurotoxicity - Carcinogenicity - Teratogenicity - Mutagenicity - Immunotoxicity - Hepatotoxicity - Nephrotoxicity - Intestinal toxicity 	<ul style="list-style-type: none"> - Cause damage to protein activity or function through the reaction of electrophilic groups on PAT with sulfhydryl groups on protein molecules or enzyme active sites - Inhibit a variety of enzyme activities - Induce DNA strand cross-linking, DNA double-strand breakage and oxidative damage 	15-18
	Microcystins (MCs)	<ul style="list-style-type: none"> - Monocyclic heptapeptide substances - Molecular structure contains carboxyl, amino, and amide groups 	<ul style="list-style-type: none"> - Acute hepatotoxicity - Liver cancer - Neurotoxicity - Multi-organ toxicity 	<ul style="list-style-type: none"> - Inhibit protein phosphatase 1 and 2A - Inhibit hyperphosphorylation caused by protein phosphatase, leading to severe liver injury - Oxidative stress, etc - Cross blood-brain barrier with the transport of Oatps/OATPs, causing neurostructural, functional, and behavioral changes 	19-22
Marine toxins	Okadaic acid (OA)	<ul style="list-style-type: none"> - A polyketide and polyether derivative of a C₃₈ fatty acid - A lipophilic toxin (LogP = 4.21) with hydroxyl, polycyclic ether, and carboxyl groups 	<ul style="list-style-type: none"> - Diarrhetic shellfish poisoning - Neurotoxicity - Embryotoxicity - Immunotoxicity - Genotoxicity 	<ul style="list-style-type: none"> - Induce secretory diarrhea by disrupting the electrolyte balance of the intestinal epithelial barrier - Cause protein hyperphosphorylation to disrupt cellular biological function - Inhibition of serine/threonine protein phosphatases (PPs), particularly of PP1 and PP2A 	23-25

Phytotoxins	Aristolochic acids (AAs)	<ul style="list-style-type: none"> - One kind of nitrophenanthrene carboxylic acids - Mainly exists in the form of anions in nature 	<ul style="list-style-type: none"> - Strong carcinogenicity - Nephrotoxicity - Ovarian dysfunction 	<ul style="list-style-type: none"> - Enzymatic metabolization of AA leading to the formation of DNA adducts and induction of Tp53 mutations - Induce oxidative/nitrosative reactive stress and mitochondrial dysfunction - Induction of apoptosis and inflammatory response - Trigger ovarian dysfunction by activating NLRP3 inflammasome and affecting mitochondrial homeostasis 	26-28
	Hg(II)	<ul style="list-style-type: none"> - The most common form of Hg in the aqueous phase - More stable - Highly toxic - Unable to be biodegraded 	<ul style="list-style-type: none"> - Gastrointestinal ulcer - Central nervous system injury - Renal injury - Hepatotoxicity 	<ul style="list-style-type: none"> - ROS production, lipid peroxidation, induce oxidative stress - Connect to the sulfhydryl groups on metallothionein, erythrocytes, or glutathione - Promote depolarization of inner mitochondrial membrane and enhanced production of H₂O₂ - Enzyme inhibition - Reduction of aquaporin mRNA 	29,30
	Pb(II)	<ul style="list-style-type: none"> - A main existing form of Pb - Bioaccumulation and non-biodegradability - Highly toxic 	<ul style="list-style-type: none"> - Carcinogenicity - Neurological diseases - Multi-organ damage, such as brain, heart, lungs, liver, blood, digestive, kidney reproductive and immune systems - Genotoxicity 	<ul style="list-style-type: none"> - Cause DNA damage, chromosomal abnormalities - Induce posttranslational alterations to protein structure - ROS oxidative damage, increasing the levels of lipid peroxidation - Activation of glial cells, increasing inflammatory cytokines (IL-1β, TNF-α and IL-6) and serum ET-1, NO and EPO - Inhibit heme biosynthesis 	29-31

As(III) As(V)	<ul style="list-style-type: none">- As(III) and As(V) are the main inorganic As species- As(III) has higher fluidity and toxicity, more than 60 times that of As(V)- As(V) is the thermodynamically stable form in oxygenated aqueous environments, while As(III) is generally stable under moderately reducing conditions	<ul style="list-style-type: none">- Cardiovascular disease- Gastrointestinal disorders- Central nervous system injury- Renal insufficiency- Hepatotoxicity- Carcinogenicity- Skin damage	<ul style="list-style-type: none">- Thiol binding (GSH conjugation)- Cause oxidative stress, inflammation, DNA damage, mitochondrial dysfunction- Neurotransmitter homeostasis and molecular alterations- Decreased the activity of serum cholinesterase, mitochondrial enzymes and uncoupling of oxidative phosphorylation	29,30,32,3 3
Cd(II)	<ul style="list-style-type: none">- Cd exists as Cd(II) in the natural environment- Highly soluble- Highly toxic- Bioaccumulation and non-biodegradability	<ul style="list-style-type: none">- Testicular damage- Fulminant hepatitis- Pulmonary edema- Hemorrhage- Kidney and liver damage- Brain and bone toxicity- Autoimmune- Genotoxicity and cancer	<ul style="list-style-type: none">- Cause apoptosis- ROS generation and oxidative stress development- Aberrant gene expression- Endoplasmic reticulum stress- Disturb the DNA synthesis and DNA repair by the proteotoxicity and DNA damage induced by oxidative stress- Alter phosphorylation cascades	29,30,34

Cr(VI)	<ul style="list-style-type: none"> - A strong oxidant with a high positive charge and a small radius - Mainly in the form of anions such as CrO_4^{2-}, $\text{Cr}_2\text{O}_7^{2-}$, and HCrO_4^- 	<ul style="list-style-type: none"> - Renal insufficiency - Gastrointestinal disorders - Skin disorders - Increase the incidence of many cancers - Respiratory, immune, cardiovascular, nervous and reproductive system damage - Genotoxicity 	<ul style="list-style-type: none"> - ROS generation and cause oxidative stress - DNA adduct formation to induce DNA damage - Dysregulated cell proliferation - Induce free radical production and result in DNA damage - Lead to the formation of DNA-protein cross-links and Cr-DNA adducts, block DNA transcription and replication 	29,30,33,3 4
Neonicotinoid insecticides (NEOs)	<ul style="list-style-type: none"> - Obtained by functional group transformation on the basis of nicotine molecular structure - High water solubility - Has hydrogen bond receptor and hydrogen bond donor 	<ul style="list-style-type: none"> - Neurotoxicity - Cardiovascular effects - Respiratory failure - Sedation and seizures - Rhabdomyolysis 	<ul style="list-style-type: none"> - Bind to nicotinic acetylcholine receptor, causing nicotinic symptoms and severe neurological damage 	35-37
Triazole fungicides	<ul style="list-style-type: none"> - Nitrogen-containing heterocyclic compounds 	<ul style="list-style-type: none"> - Reproductive and developmental toxicity - Mutagenicity - Carcinogenicity - Endocrine disrupting effects - Liver toxicity 	<ul style="list-style-type: none"> - Accumulate in the liver, cause liver functional abnormalities and metabolic disorders - Interfere with lipid metabolism by binding to nuclear receptors (peroxisome proliferator-activated receptors and retinoid X receptor) - Interfere with estrogen-induced signal transduction 	38,39

Organophosphorus pesticides (OPPs)	<ul style="list-style-type: none"> - Most of them are phosphate or phosphorothioate esters - R₁ and R₂ in their structural formulas are mostly methoxy (CH₃O⁻) or ethoxy (C₂H₅O⁻) 	<ul style="list-style-type: none"> - Various neurological disorders, such as AD, Parkinson's disease, and other neurotoxicity - Metabolic diseases, e.g. diabetes, hyperlipidemia - Hepatic toxicity - Renal insufficiency - Cardiopulmonary toxicity - Immunotoxicity - Genotoxicity - Endocrine toxicity - Neurotoxicity - Hepatotoxicity - Nephrotoxicity 	<ul style="list-style-type: none"> - Irreversibly bind to acetylcholinesterase - Cause oxidative stress, reduce the membrane potential of mitochondria, leading to mitochondrial DNA mutations - Inflammation response 	40,41
Fipronil	<ul style="list-style-type: none"> - Phenylpyrazole chemicals 	<ul style="list-style-type: none"> - Reproductive toxicity - Neurodegenerative effects - Endocrine system changes - Carcinogenicity 	<ul style="list-style-type: none"> - Interfere with GABA-gated channels and disrupt normal neuronal influx by targeting GABA-gated chloride channels - Inhibit human α_1, $\alpha_1\beta$, α_2, and α_3 glycine receptor subtypes 	42,43
Trifluralin	<ul style="list-style-type: none"> - Fluorinated dinitroaniline - Low water solubility (lipophilic) 	<ul style="list-style-type: none"> - Cytotoxicity - Genotoxicity 	<ul style="list-style-type: none"> - Inhibition of Ca²⁺ transport mechanisms at the plasmatic membrane - Activate the oxidative stress pathway and chromosomal damage - Mild mitochondrial dysfunction, mitophagy, or inhibition of proliferation - Alter the expression levels of cancer-related genes 	44,45

	Iodine (I)	<ul style="list-style-type: none"> - Strong volatility and high diffusion coefficient - ^{129}I and ^{131}I are the two main radionuclides of iodine - Form polyiodides in an aqueous medium - The most common are the triiodide ions 	<ul style="list-style-type: none"> - Thyroid cancer - Unconsciousness 	<ul style="list-style-type: none"> - Accumulate in the thyroid gland and interfere metabolic system 	46,47
Radionuclides	Uranium (U)	<ul style="list-style-type: none"> - U(VI) has strong fluidity, high radioactivity and high toxicity - U(IV) is less soluble and relatively immobile 	<ul style="list-style-type: none"> - Severe kidney injury - Pulmonary toxicity - Other organs damage, mainly hepatic, hematopoietic, neurological, and reproductive effects 	<ul style="list-style-type: none"> - Oxidative stress - Genetic damage - Protein injury - Cell apoptosis - Inflammation - Metabolic disorders 	48,49
	Technetium (Tc)	<ul style="list-style-type: none"> - Exist in the form of pertechnetate anion (TcO_4^-) - High water solubility - Extremely high mobility 	<ul style="list-style-type: none"> - Radiation damage 	<ul style="list-style-type: none"> - Radiation causes ionization and stimulation of intracellular molecules, disrupting the normal functioning of biological organisms - Cause ionization or excitation of biological macromolecules and proteins, resulting in structural changes and loss of activity 	50,51

Others	Chlorination disinfection by-products (Cl-DBPs)	<ul style="list-style-type: none"> - Include aliphatic and aromatic DBPs - Aromatic DBPs are more toxic than many aliphatic DBPs 	<ul style="list-style-type: none"> - Colorectal cancer - Bladder, stomach, lymphoma, kidney, pancreas, breast, and oesophagus cancer - Birth defects - Stillbirths and spontaneous abortions - Genotoxicity - Cytotoxicity - Teratogenicity and endocrine disruption - Neurotoxicity - Harmful endocrine effects: 	<ul style="list-style-type: none"> - Modulate various signaling pathways, leading to endocrine disorders and DNA damage, mimicking or antagonizing endogenous hormones 	52-54
	Perfluoroalkyl and polyfluoroalkyl substances (PFASs)	<ul style="list-style-type: none"> - Perfluoroalkyl is fully fluorine - Polyfluoroalkyl substances have a non-fluorine atom attached to at least one carbon atom 	<ul style="list-style-type: none"> - Thyroid disease - Hepatotoxicity - Immunotoxicity - Neurotoxicity - Reproduction and development toxicity - Carcinotoxicity 	<ul style="list-style-type: none"> - Disturb steroidogenesis - Interfere with steroid hormone receptors - PFAS is proved to be both PPARα and CAR agonist, causing toxicity - Changes in lipid metabolism - Oxidative stress - Result in altered androgen and insulin-like factor 3 (INSL3) dependent processes 	55,56

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