## ESI (Electronic Supplementary Information)

Title:

Synergistic effect of Pd/CZO catalysts and an electric field on complete combustion of lean and humid methane at low temperatures

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Figure S1 Schematic diagram of the reaction apparatus.



Figure S2 XPS profiles of 1wt%Pd/Ce<sub>x</sub>Zr<sub>1-x</sub>O<sub>2</sub> in as-made and after reaction under the humid

conditions with the electric field (1 mA) up to 723 K.

No significant correlation was observed between the chemical state of Pd in after reaction and the

order of activity test results.



Figure S3 H<sub>2</sub>O-TPD profiles over  $1wt\%Pd/Ce_xZr_{1-x}O_2$ .



Figure S4 The results of current variation test and the energy efficiency over  $1wt\%Pd/Ce_{0.25}Zr_{0.75}O_2$ in the EF (1, 3, 5 mA) at 473 K, CH<sub>4</sub>: 2000ppm, O<sub>2</sub>: 10%, H<sub>2</sub>O: 10%, N<sub>2</sub> balance.



Figure S5 Methane conversion over 1wt%Pd/Ce<sub>x</sub>Zr<sub>1-x</sub>O<sub>2</sub> without the EF, CH<sub>4</sub>: 2000ppm, O<sub>2</sub>: 10%,

H<sub>2</sub>O: 10%, N<sub>2</sub> balance.



Figure S6 Methane conversion over  $Ce_{0.25}Zr_{0.75}O_2$  with and without the EF.



Figure S7 Methane conversion over  $1wt\%Pd/Ce_xZr_{1-x}O_2$ , (a) with the EF and (b) without the EF, CH<sub>4</sub>:

2000ppm, O<sub>2</sub>: 10%, N<sub>2</sub> balance.



Figure S8 Methane conversion over 1wt%Pd/Ce<sub>0.75</sub>Zr<sub>0.25</sub>O<sub>2</sub> in various water vapour pressures,

(a) with the EF and (b) without the EF.



Figure S9 Methane, oxygen, and water pressure dependencies on the reaction rate of methane

combustion in a kinetic range, (a) with the EF at 473 K and (b) without the EF at 673 K.

Catalyst: 60mg, total flow rate: 150 SCCM.



Figure S10 The relationship between water adsorption amount per surface area and the activity of

the methane reaction with steam over  $1wt\%Pd/Ce_{x}Zr_{1\text{-}x}O_{2}$  in the EF, 473 K,

fixed electric field power (1 W).

Support	(111) Lattice spacing	Lattice constant		
Support	/ nm	/ nm		
$Ce_{0.25}Zr_{0.75}O_2$	0.300	0.450		
$Ce_{0.5}Zr_{0.5}O_2$	0.305	0.457		
$Ce_{0.75}Zr_{0.25}O_2$	0.309	0.463		
$Ce_{0.9}Zr_{0.1}O_{2}$	0.311	0.466		

Table S1 Calculation results of the Ce-Zr-O (111) lattice spacing and the lattice constant.

Table S2 Results of methane complete oxidation in a humid condition with and without the electric field over 1wt%Pd/Ce<sub>0.25</sub>Zr<sub>0.75</sub>O<sub>2</sub> catalyst, CH<sub>4</sub>: 2000ppm, O<sub>2</sub>: 10%, H<sub>2</sub>O: 10%, N<sub>2</sub>: balance.

66	Tomporaturo / K	Voltago / kV	Methane	Select	ivity / %
CF	Temperature / K	Voltage / KV	conversion / %	СО	CO2
	473	1.93	58.9	0.7	99.3
	498	1.94	61.0	0.6	99.4
With	523	1.97	63.7	0.2	99.8
	548	1.90	68.8	0.2	99.8
(1 mA)	573	1.69	77.3	0.0	100.0
	623	1.47	92.3	0.0	100.0
	673	1.45	99.0	0.0	100.0
	723	1.40	100.0	0.0	100.0
Without	373		0.0	-	-
	423		0.0	-	-
	473		0.0	-	-
	523		0.0	-	-
	573	-	1.9	0.2	99.8
	623		17.7	0.0	100.0
	673		74.8	0.0	100.0
	723		86.5	0.0	100.0
	773		99.5	0.0	100.0

Temp.	<b>P</b> <sub>CH4</sub>	<b>P</b> <sub>02</sub>	<b>P</b> <sub>H20</sub>	r	V	$\mathbf{r} = \mathbf{k} \mathbf{P}_{CH4}{}^{\alpha} \mathbf{P}_{O2}{}^{\beta} \mathbf{P}_{H2O}{}^{\gamma}$		
/к	/ atm	/ atm	/ atm	/ mmol h <sup>-1</sup>	/ kV	α	β	γ
0.00 0.00 0.00 473	0.001			0.0433	1.14			
	0.002	0.100		0.0782	0.98			
	0.004			0.1020	0.80			
		0.075	0.100	0.0943	1.15			
		0.100		0.0782	0.98			
with EF	with EF	0.125		0.0748	1.03	0.62	-0.01	-0.16
(1 mA) 0.	0.002	0.150		0.0983	1.12			
	0.002		0.010	0.1067	1.17			
		0 100	0.050	0.0969	1.10			
		0.100	0.100	0.0782	0.98			
			0.150	0.0670	1.02			
673 w/o EF	0.001			0.0932				
	0.002	0.100	0.100	0.1704				
	0.004			0.4866				
		0.075		0.1707				
	0.002	0.010		0.1704	-	1.19	-0.04	-0.70
		0.125		0.1675				
			0.050	0.2542				
		0.100	0.100	0.1704				
			0.150	0.1156				

Table S3 Results of pressure changing tests with/without the electric field.