

Supplementary Information

Hydrogen Release Mechanism for Hydrolysis of Ammonia Borane over Pd/C₃N₄ Nanocatalyst Synthesized by Electron Beam Irradiation

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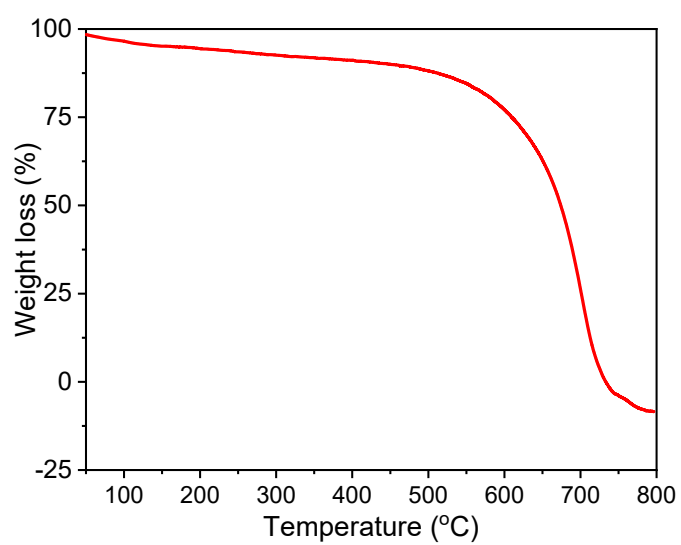
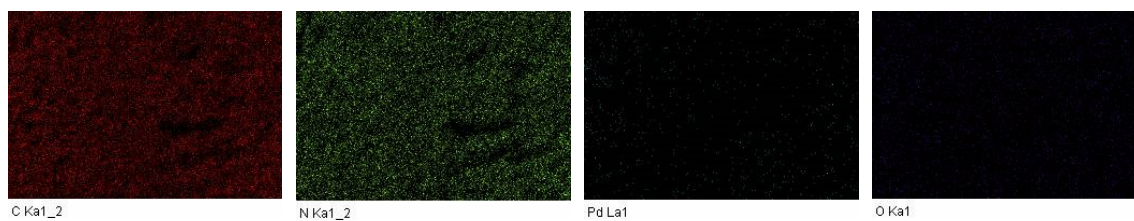
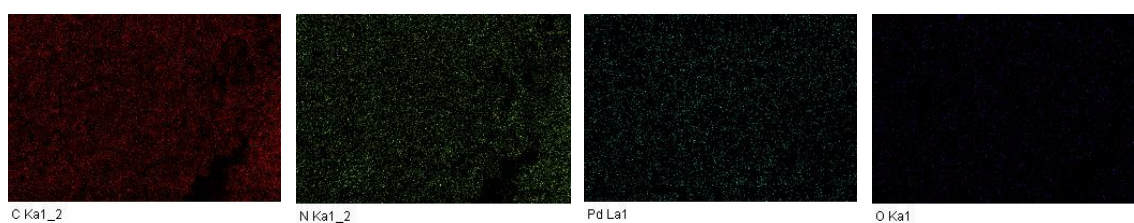


Figure S1: TGA curves of ex-C₃N₄ in air at a heating rate of 10 K min⁻¹, using Setsys Evolution, SETARAM, France

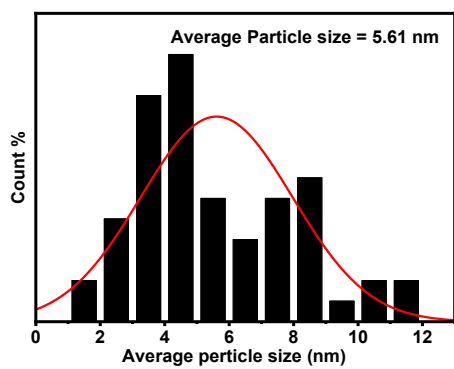


(a)

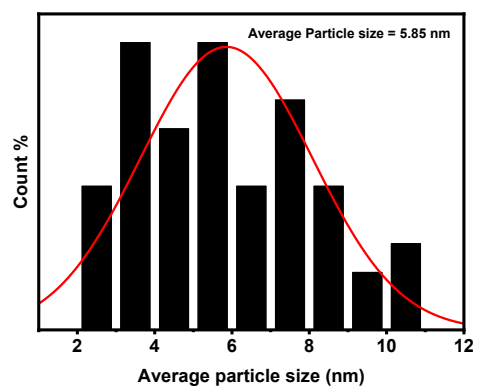


(b)

Figure S2: Elemental mapping of C, N, Pd, and O of (a) CNPd1 and (b) CNPd5



(a)



(b)

Figure S3: Particle size distribution of Pd nanoparticles in (a) CNPd1 (b) CNPd5

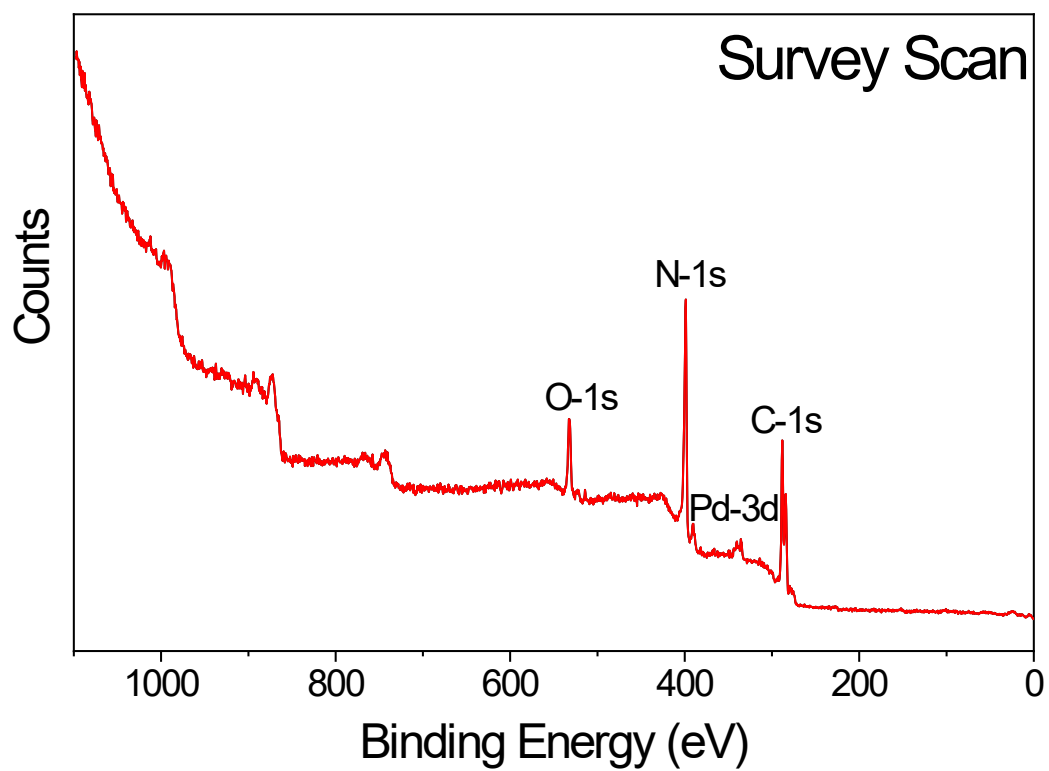
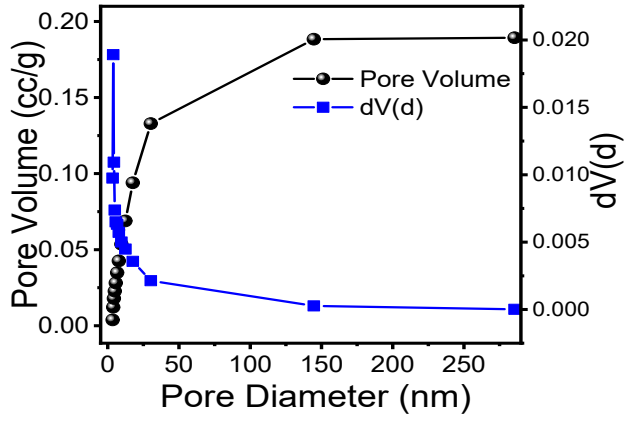
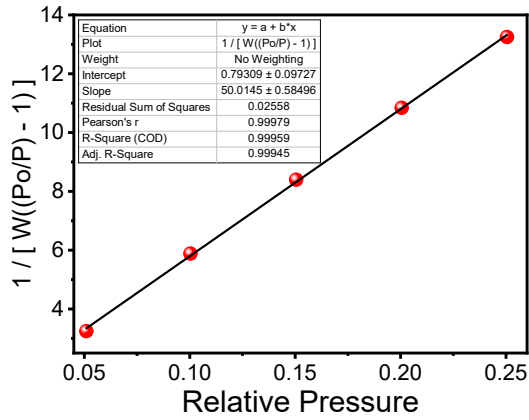
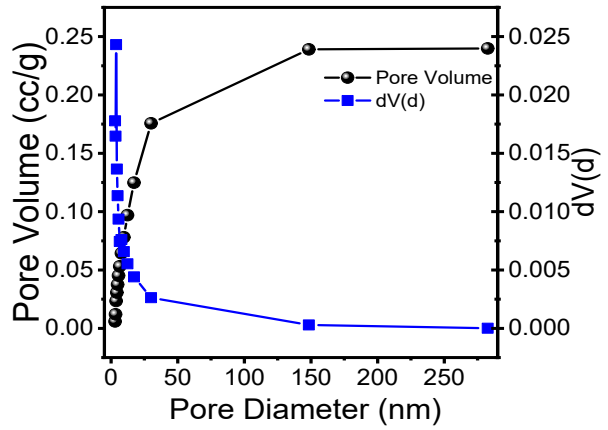
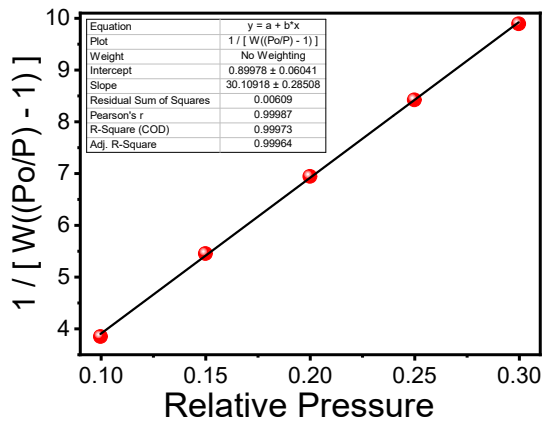


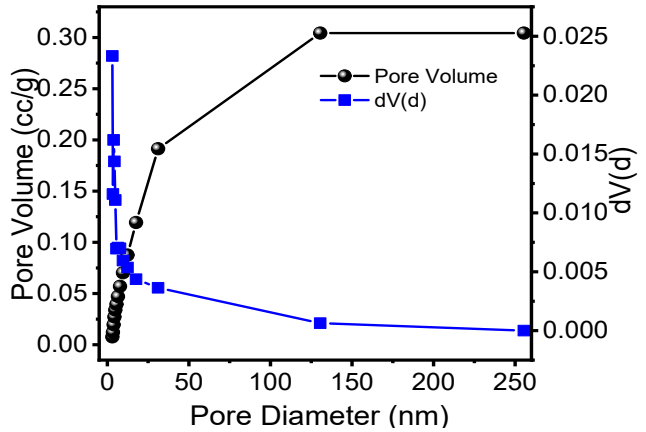
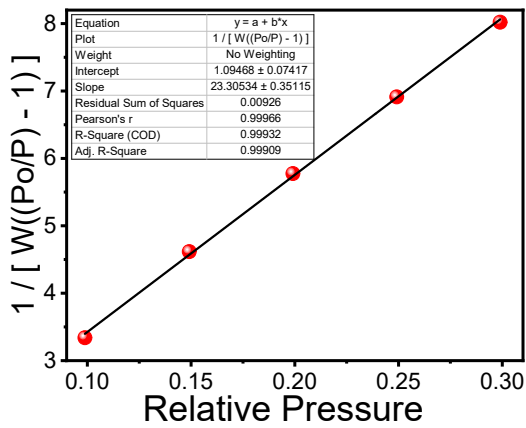
Figure S4: XPS survey scan of CNPd5



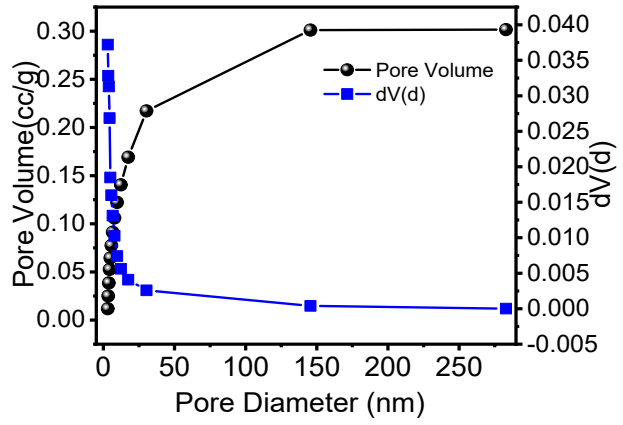
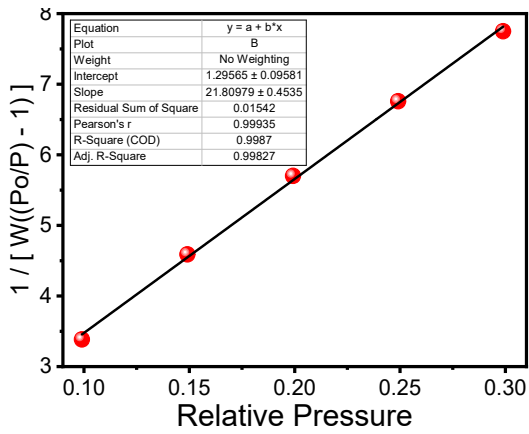
(a) C_3N_4



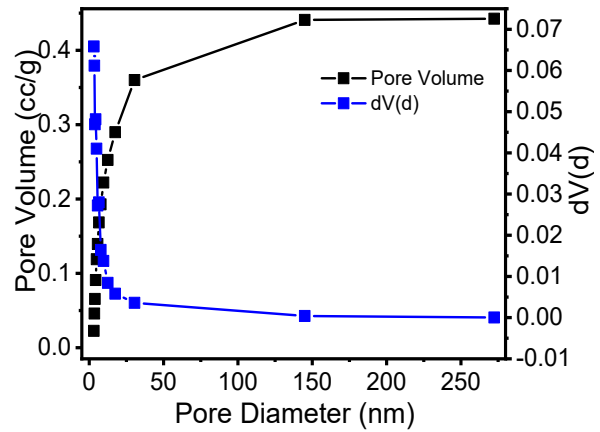
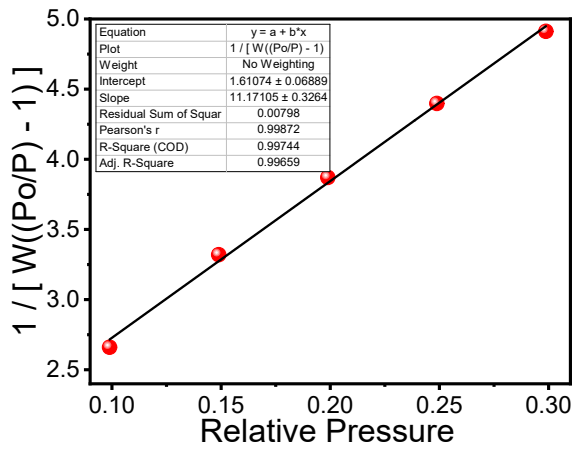
(b) Ex- C_3N_4



(c) CNPd1



(e) CNPd3



(f) CNPd5

Figure S5 BET plot for surface area analysis and BJH pore size distribution of the samples

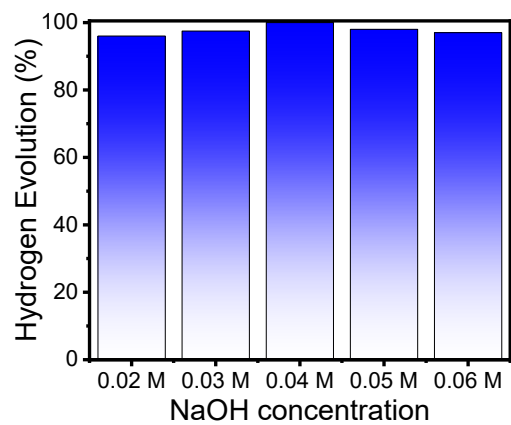


Figure S6 Optimization of the NaOH concentration for the Hydrogen evolution from the CNPd5 catalysed AB hydrolysis

Table S1: Activation energy for the ammonia borane hydrolysis reaction in presence of supported Pd catalyst (NA-Not Available)

Entry	Catalyst	Activation Energy (kJ/mol)	TOF (min ⁻¹)	Pd particle size (nm)	References
1	Pd/C ₃ N ₄	27.36	38.2	5.85	This Study
2	Pd ⁰ /CeO ₂	68	29	2-6	1
3	RGO@Pd	38	26.3	5	2
4	Pd(0)-HAP	55	NA	3.6	3
5	Pd-PVB-TiO ₂	55.9	NA	NA	4
6	Pd/Co ₃ O ₄	62	3048	2.6	5
7	Pd/PDA- CoFe ₂ O ₄	65	175	2.7	6
8	Pd/ α -LDH	20	49.5	5.6	7
9	Pd/MCN	57	125	2.7	8
10	Pd/IPCNs	29.1	122.8	2.17	9
11	Pd/Fe ₃ O ₄ @SiO ₂ -PC	28.4	47.3	2.3	10
12	Pd/CS-rGO	39	42.5	1.7	11
13	Pd/C	67.5	40	3.6	12
14	Pd(0)/g-C ₃ N ₄ -CS	35.3	25.3	1.4	13
15	Pd/CGP-GO-Fe ₃ O ₄	36.5	16.2	NA	14
16	Pd/PDA-Fe ₃ O ₄	65	14.5	2	15
17	RGO-Pd	51	6.25	4	16
18	CuFe ₂ O ₄ -NH ₂ @Pd	38.54	NA	3-5	17
19	PdCo@NCHP	36.9	470.5	NA	18

20	Ni-MoO _x	NA	85.7	3	19
21	CoCu	38.6	11.56	8.6	20
22	CuNi	23.58	2.08	3.9	21
23	RuNi	27.2	905	1.40	22
24	RuCo	22.5	139	1.56-2.94	23

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