

Supplementary Information

Operando Spectroscopy Unravels the Nature of Deactivating Species and their Precursors in Alkene Oligomerization Catalysis

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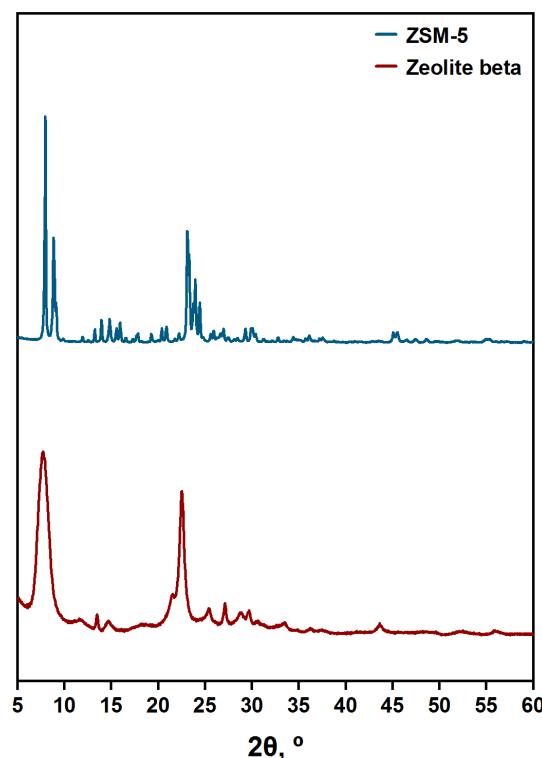


Fig. S1 XRD patterns of ZSM-5 and zeolite beta.

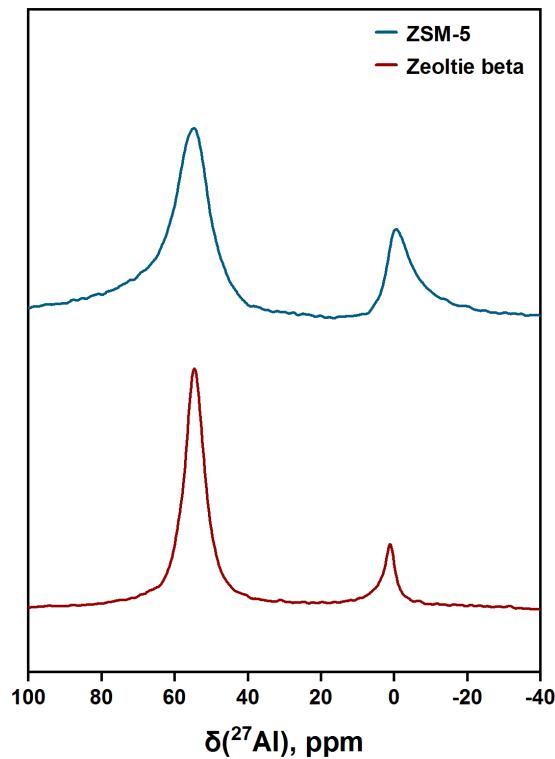


Fig. S2 ^{27}Al MAS NMR spectra of ZSM-5 and zeolite beta.

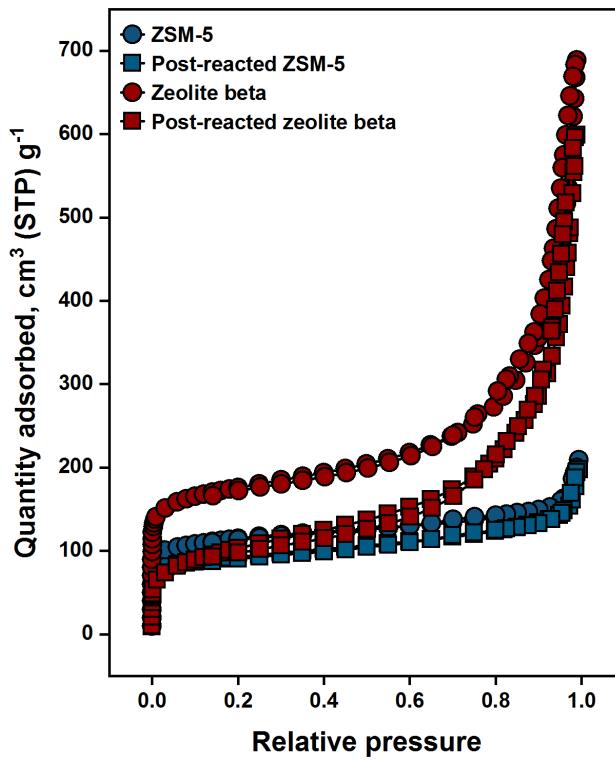


Fig. S3 Nitrogen physisorption isotherms measured on unreacted and post-reacted ZSM-5 and zeolite beta at 77 K.

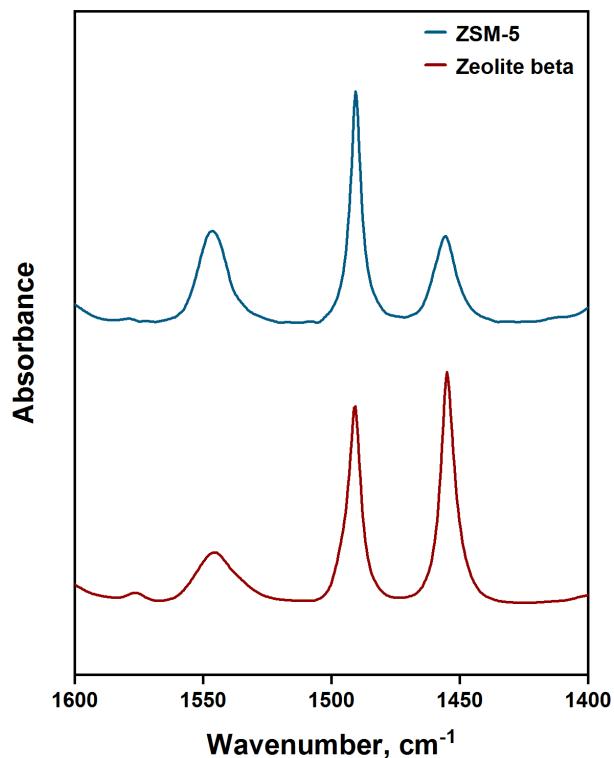


Fig. S4 Pyridine-FTIR spectra measured on ZSM-5 and zeolite beta.

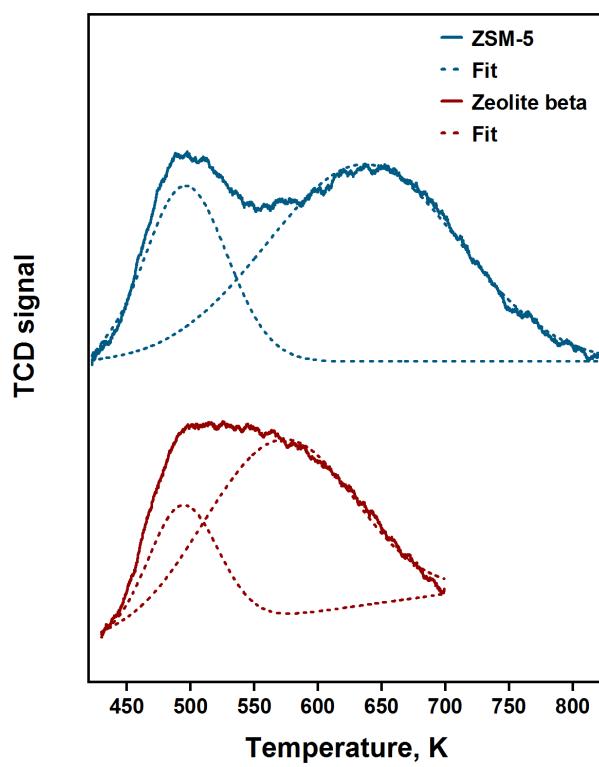


Fig. S5 NH₃-TPD profiles measured on ZSM-5 and zeolite beta.

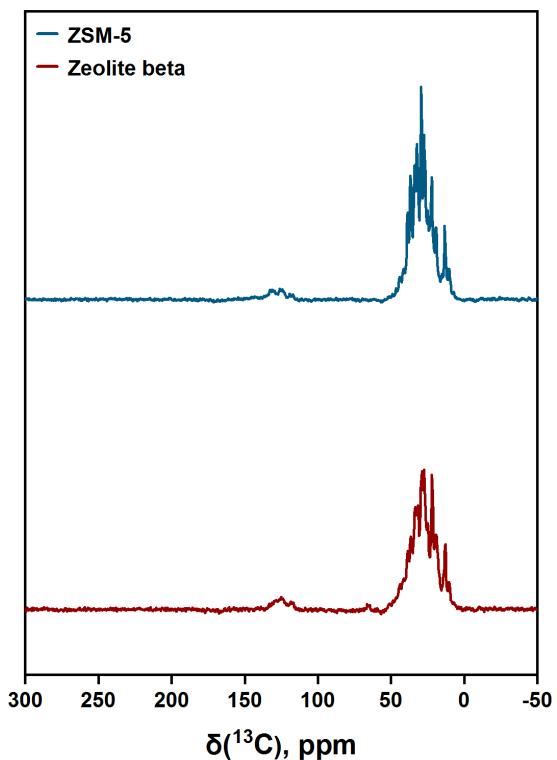


Fig. S6 INEPT-based solid-state ^{13}C NMR spectra of post-reacted ZSM-5 and zeolite beta. [523 K; 100 kPa propene pressure; 10–15 min TOS]. The spectra denote features of methyl and methylene groups (10–40 ppm) and olefinic groups (120–135 ppm).

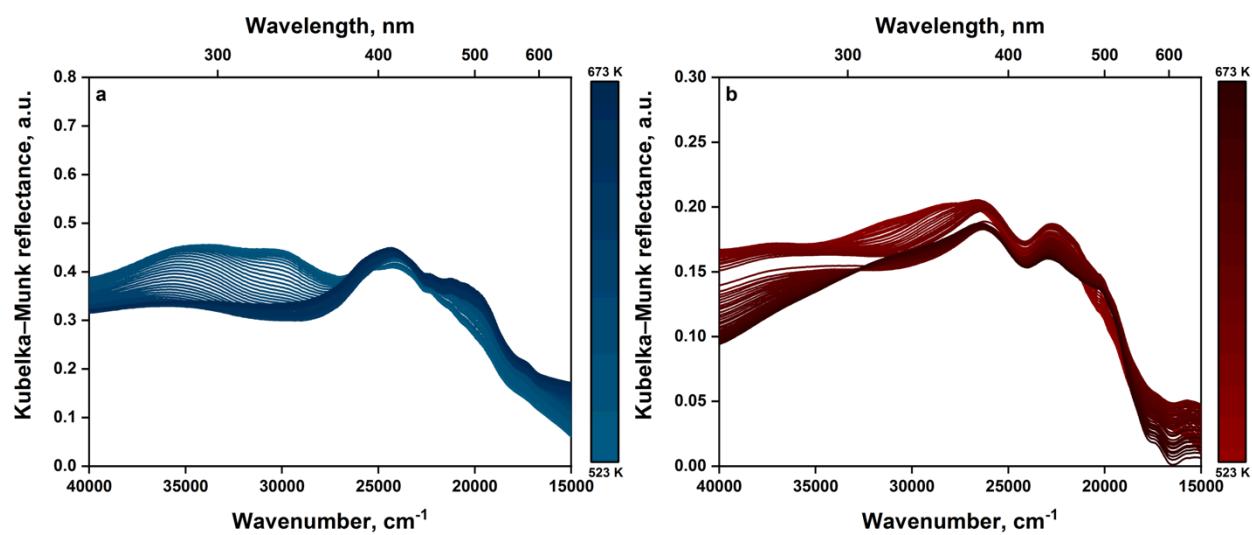


Fig. S7 Kubelka–Munk reflectance of the UV–Vis bands measured during temperature-programmed experiments on (a) ZSM-5 and (b) zeolite beta.

Table S1 Nitrogen physisorption measurements conducted on unreacted and post-reacted zeolites.

Material	$S_{\text{Micropore}},^a$ $\text{m}^2 \text{g}^{-1}$	$S_{\text{Mesopore}},^b$ $\text{m}^2 \text{g}^{-1}$	$S_{\text{BET}},^c$ $\text{m}^2 \text{g}^{-1}$	$V_{\text{Micropore}},^d$ $\text{cm}^3 \text{g}^{-1}$	$V_{\text{Mesopore}},^e$ $\text{cm}^3 \text{g}^{-1}$	$V_{\text{Total}},^f$ $\text{cm}^3 \text{g}^{-1}$
ZSM-5	269	156	425	0.11	0.14	0.25
Post-reacted ZSM-5	204	136	340	0.084	0.15	0.23
Zeolite beta	357	281	638	0.15	0.74	0.89
Post-reacted zeolite beta	37	318	355	0.021	0.62	0.64

^a Micropore area. ^b Mesopore area. ^c BET surface area. ^d Micropore volume. ^e Mesopore volume. ^f Total pore volume measured at ca. 0.95 of relative pressure.