

Supplementary Information

Low-temperature Synthesis of High Entropy Amorphous Metal Oxides (HEOs) for Enhanced Oxygen Evolution Performances

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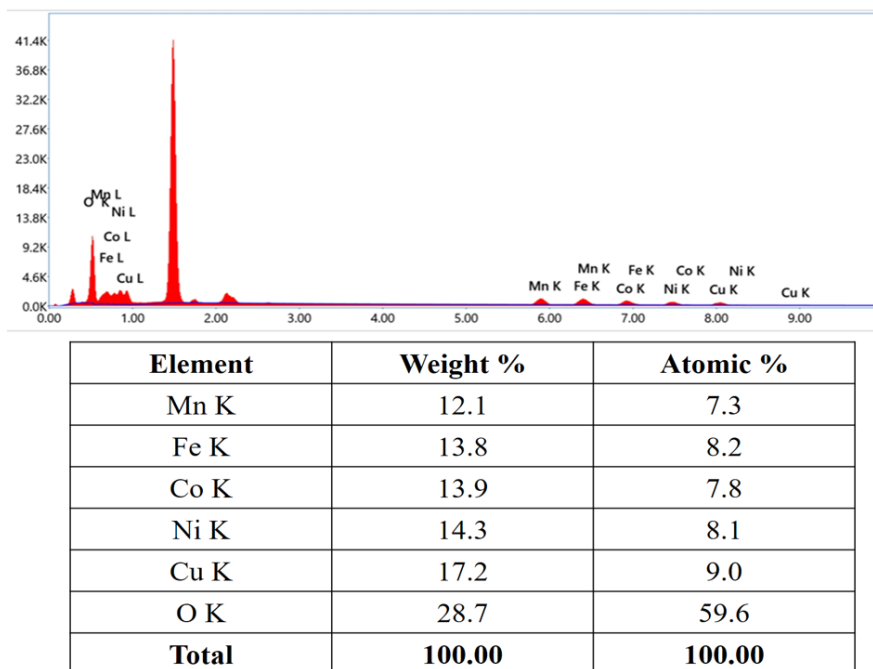


Fig. S1 EDS analysis and different elemental percentages of the as-prepared material.

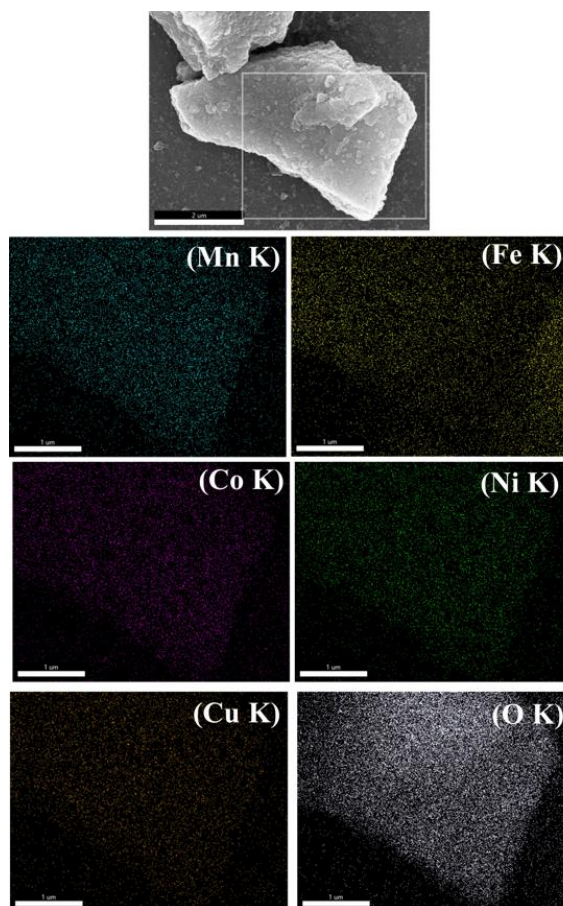


Fig. S2 Elemental mapping of HEO-110 shows the presence of Mn, Fe, Co, Ni, Cu, and O.

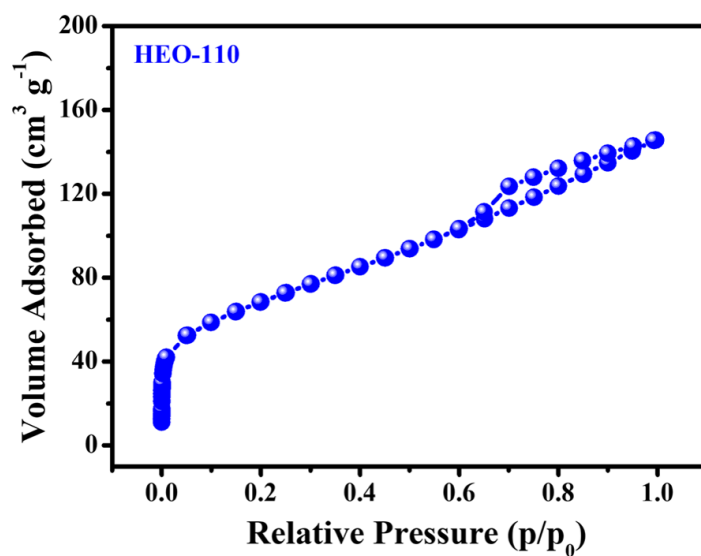


Fig. S3 BET study of high entropy amorphous material.

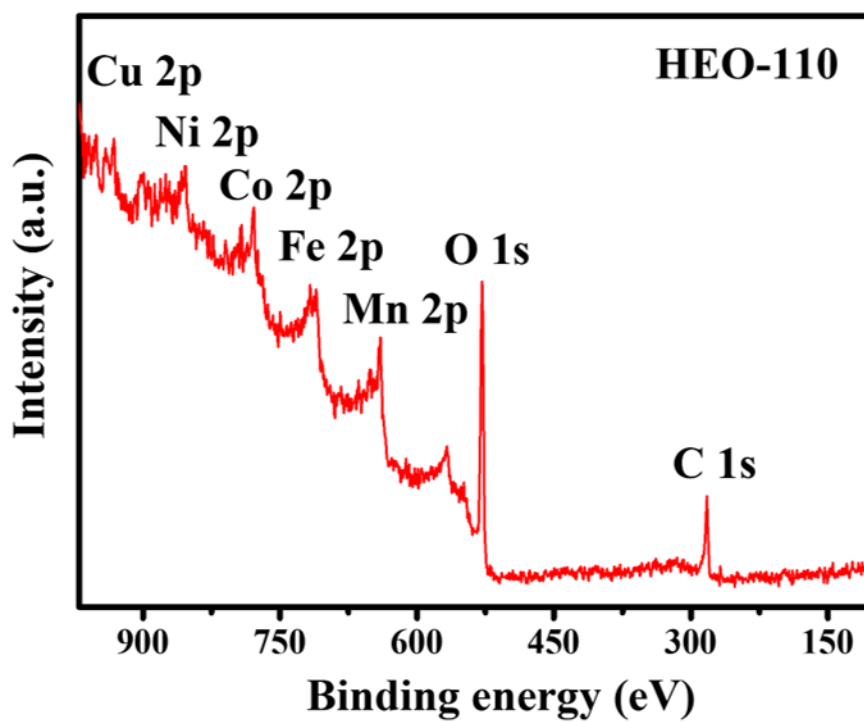


Fig. S4 Elemental survey analysis of amorphous HEO-110.

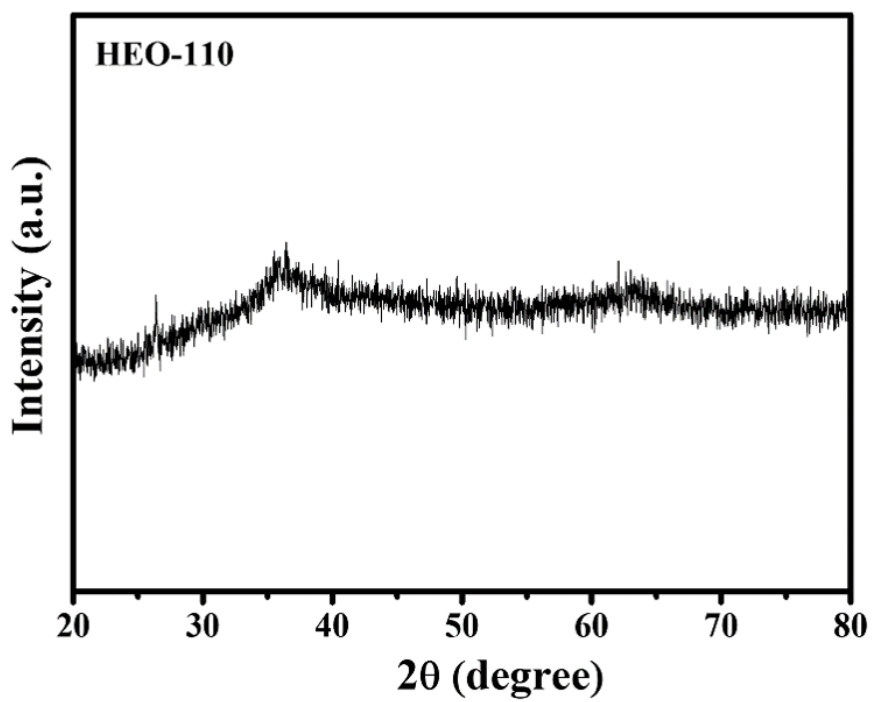


Fig. S5 Post stability PXRD of the synthesized electrode material.

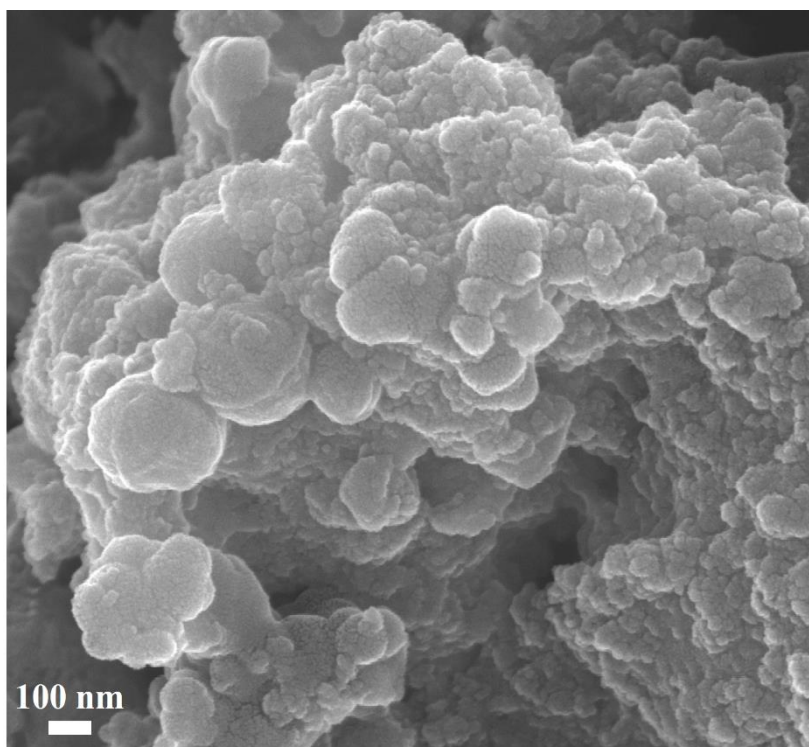


Fig. S6 FESEM image after 20 h of stability test in 1 M KOH.

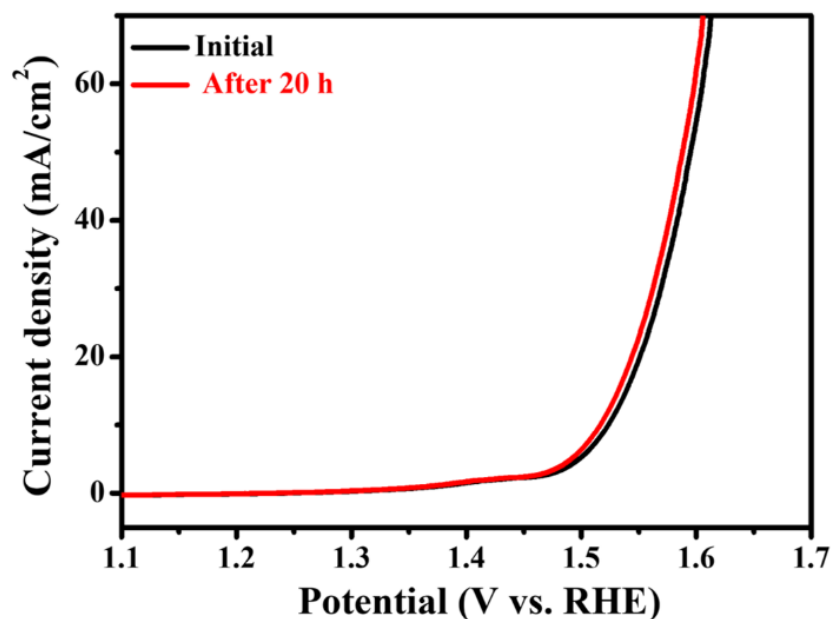


Fig. S7 LSV curves of HEO-110 before and after stability test.

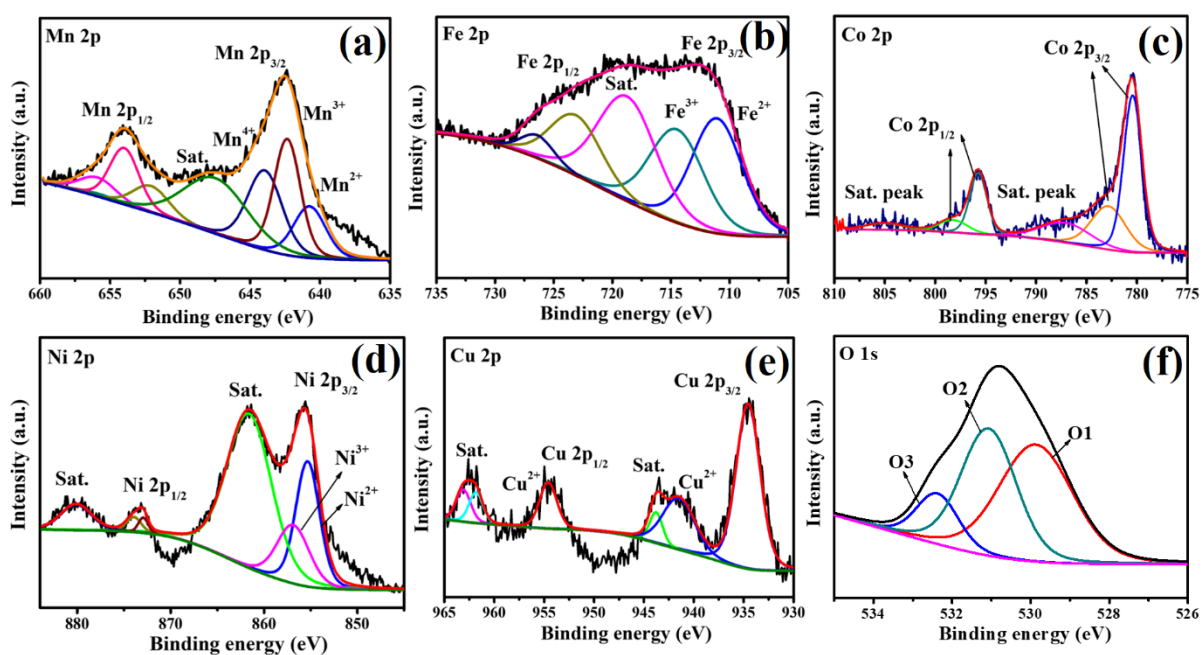


Fig. S8 Post-stability X-ray photoelectron spectra of high entropy amorphous metal oxides (HEO-110). Corresponding Mn 2p (a), Fe 2p (b), Co 2p (c), Ni 2p (d) Cu 2p (e), and O 1s (f), respectively.

Table S1 A brief literature study of high entropy electrode material for OER.

SL No.	Electrocatalyst	Electrolyte	Overpotential at 10 mA/cm ² (mV)	Tafel slope (mV/dec)	References
1	(Cr _{0.2} Mn _{0.2} Fe _{0.2} Co _{0.2} Ni _{0.2}) ₃ O ₄	1 M KOH	332	54.5	Chemical Engineering Journal, 2022, 431, 133448
2	(CoNiMnZnFe) ₃ O ₄	1 M KOH	336	47.5	Journal of Alloys and Compounds, 2021. 868, 159064
3	(Fe _{0.2} Co _{0.2} Ni _{0.2} Cr _{0.2} Mn _{0.2}) ₃ O ₄	1 M KOH	275	50.27	Chemical Engineering Journal, 2023. 460, 141675
4	(CoCuFeMnNi) ₃ O ₄	1 M KOH	400	76.7	Journal of Materials Chemistry A, 2019, 7(42), 24211-24216.
5	(Co _{0.2} Mn _{0.2} Ni _{0.2} Fe _{0.2} Zn _{0.2}) ₃ Fe ₂ O ₄	1 M KOH	326	53.6	ACS applied materials & interfaces, 2020, 12(29), 32548-32555.
6	(FeCrCoNiAl _{0.1})O _x	1 M KOH	381	60.9	Journal of Alloys and Compounds, 2021, 868,159108.

7	$\text{La}(\text{CrMnFeCo}_2\text{Ni})\text{O}_3$	1 M KOH	325	51.2	Advanced Functional Materials, 2021, 31(27), 2101632.
8	$(\text{MnFeCoNiCu})_3\text{O}_4$	1 M KOH	290	85	Our Work