Electronic Supplementary Material (ESI) for Dalton Transactions.
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Supporting information

$Co^{(II)}$-doped hybrid $Zn^{(II)}$ tetraborate complexes, $[Zn_xCo_{(1-x)}(1,3\text{-dap})B_4O_7]$(1,3-dap=1,3-diaminopropane): BET analysis and $N_2/H_2O/D_2O$ adsorption studies.

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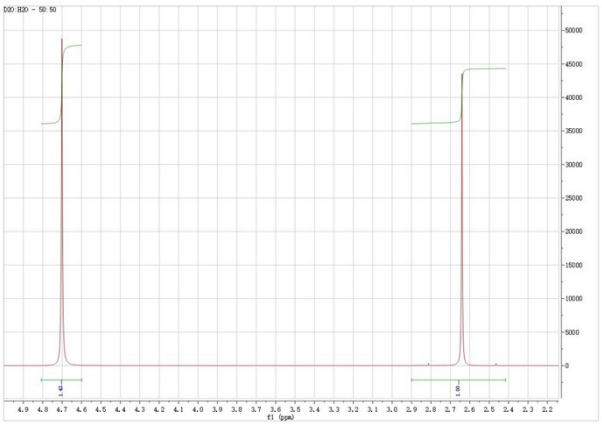
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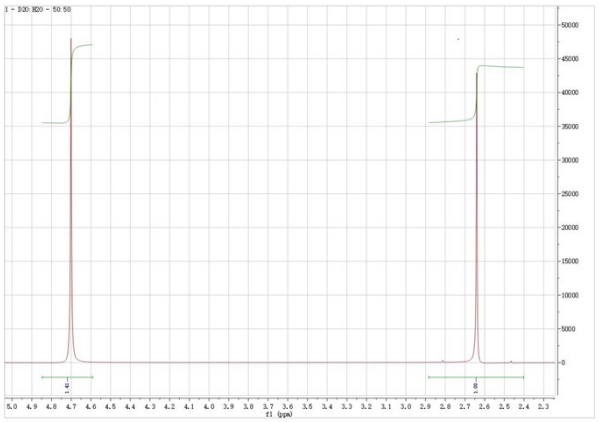
f1 (ppm)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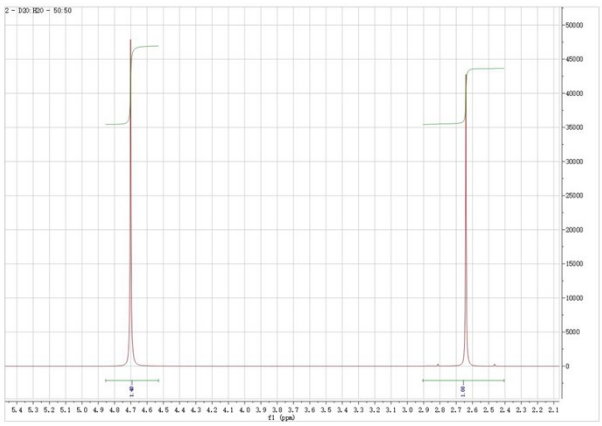
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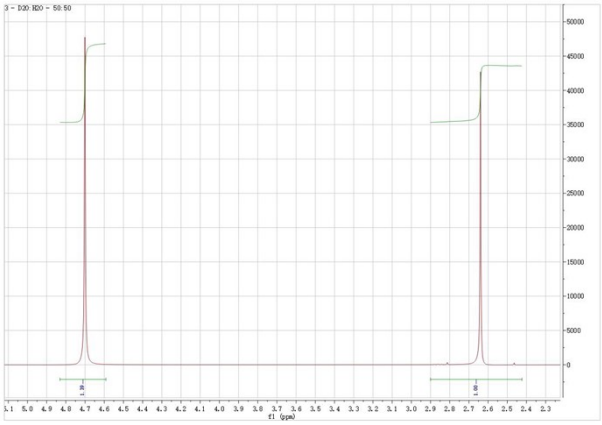
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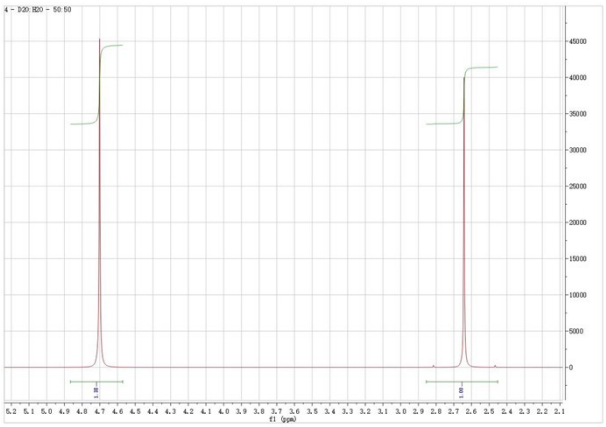
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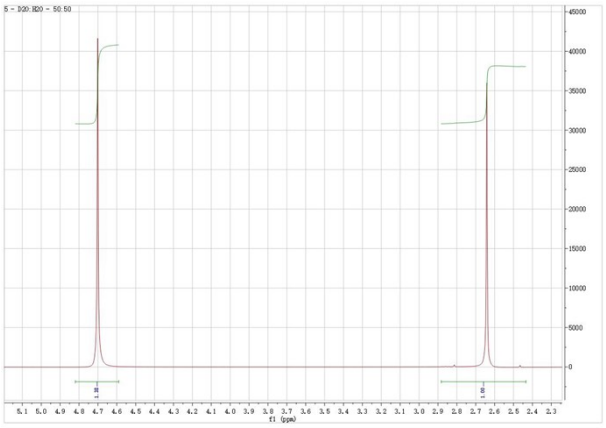
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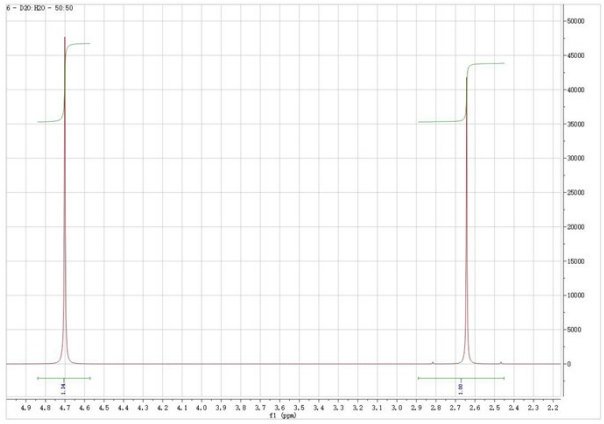
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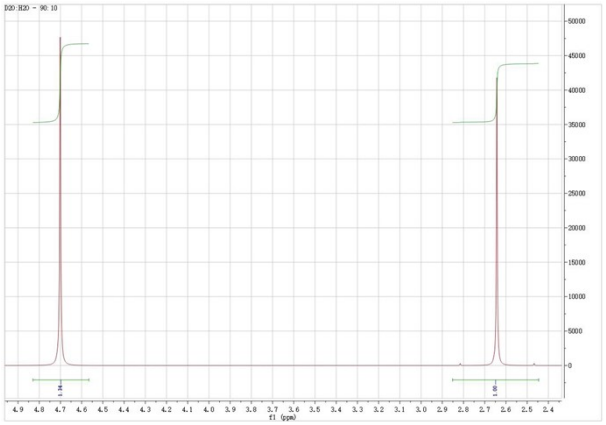
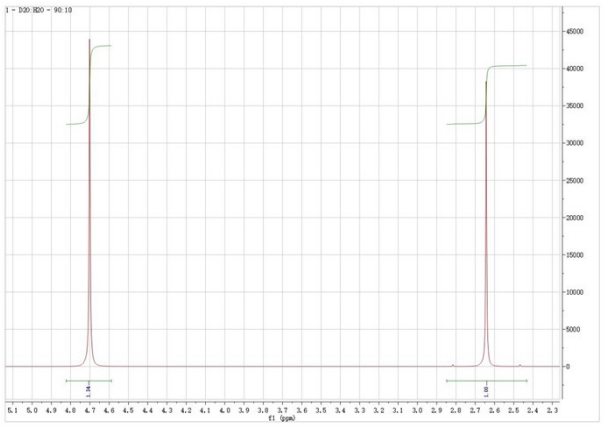
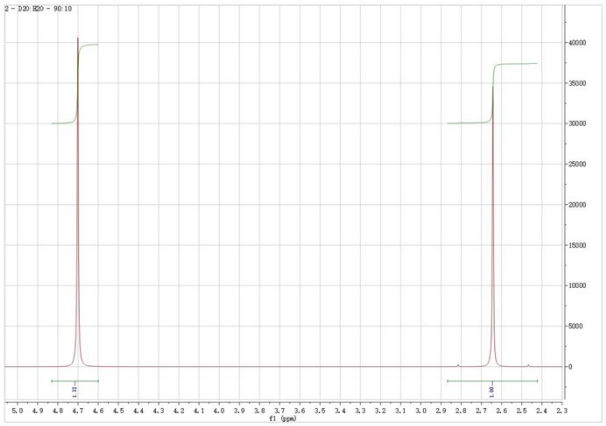
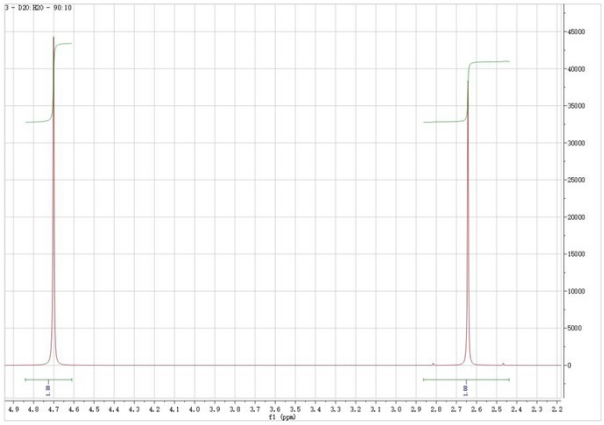
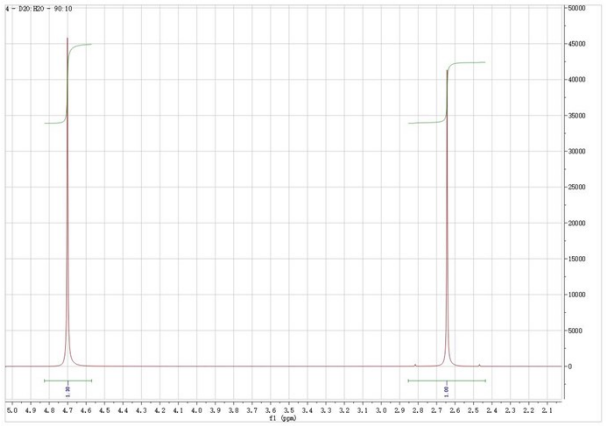
f1 (ppm)


D2O:H2O = 90:10
f1 (ppm)


1 - D2O:H2O - 90:10
f1 (ppm)


2 - D2O:H2O - 90:10
f1 (ppm)


3 - D2O:H2O - 90:10
f1 (ppm)


4 - D2O:H2O - 90:10
f1 (ppm)

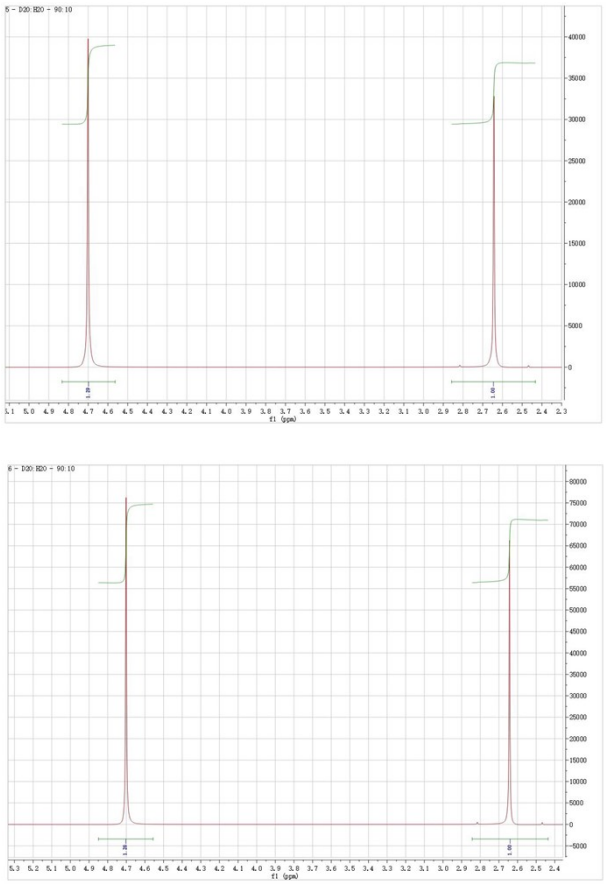


Figure S1. ^{1}H NMR: the relative integral area of H_2O, relative to an internal standard, after adsorption of synthetic D_2O/H_2O mixtures by the hybrid borates**1-6.**

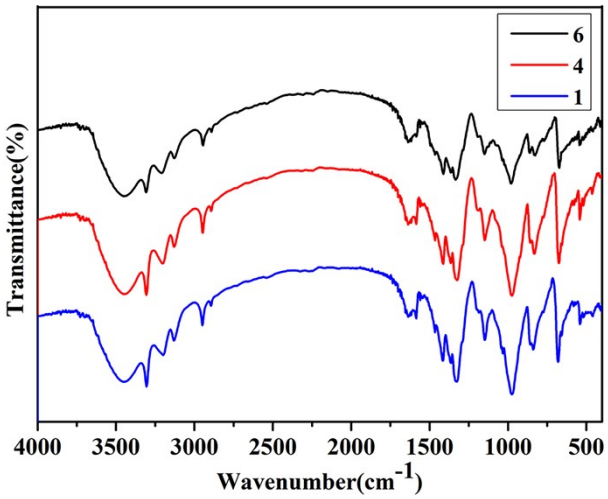


Figure S2. FT-IR spectra of **1, 4** and **6.**

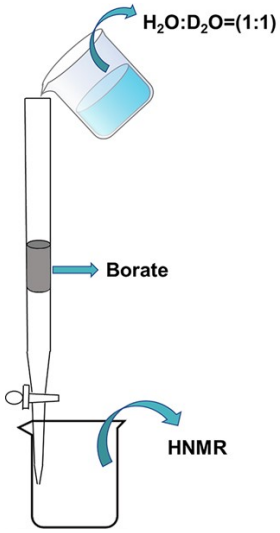


Figure S3. Schematic diagrams of the experimental device.