

## Supporting Information

# Tunable Benzamidinate Zinc Complexes: Coordination Modes and Catalytic Activity in the Ring-Opening Polymerization of L-lactide

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## Crystal structure data

**Table S1.** Crystallographic data and refinement details for complexes **1-4** and **7**

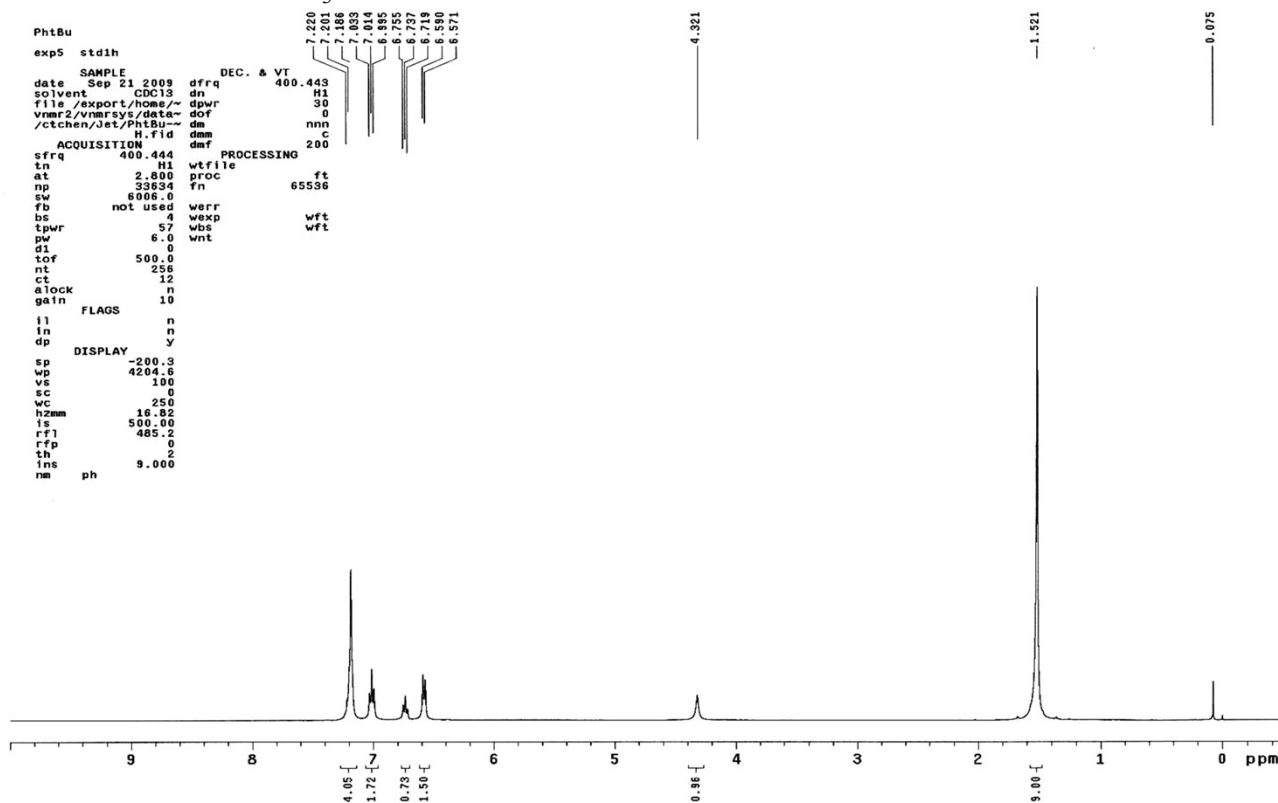
	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>7</b>
Formula	C <sub>38</sub> H <sub>48</sub> N <sub>4</sub> Zn <sub>2</sub>	C <sub>36</sub> H <sub>44</sub> N <sub>4</sub> O <sub>2</sub> Zn <sub>2</sub>	C <sub>38</sub> H <sub>50</sub> N <sub>6</sub> Zn <sub>2</sub>	C <sub>68</sub> H <sub>76</sub> N <sub>8</sub> Zn <sub>2</sub>	C <sub>38</sub> H <sub>32</sub> N <sub>6</sub> Zn
Fw	691.54	695.49	721.58	1136.11	638.07
T, K	100(2)	100(2)	100(2)	100(2)	100(2)
Crystal system	Triclinic	Monoclinic	Monoclinic	Monoclinic	Orthorhombic
Space group	<i>P</i> -1	<i>C</i> <sub>2</sub> / <i>c</i>	<i>P</i> 2(1)/ <i>n</i>	<i>C</i> <sub>2</sub> / <i>c</i>	<i>P</i> <sub>bca</sub>
<i>a</i> , Å	9.4824(4)	12.7899(4)	8.4604(2)	14.8451(4)	13.3353(3)
<i>b</i> , Å	10.0818(5)	12.0267(3)	9.6640(2)	20.7782(5)	18.1750(4)
<i>c</i> , Å	10.7058(4)	22.7927(6)	22.0213(5)	20.2976(5)	26.2092(6)
<i>α</i> °	103.486(4)	90	90	90	90
<i>β</i> °	104.402(4)	92.379(2)	98.548(2)	103.280(3)	90
<i>γ</i> °	109.408(4)	90	90	90	90
<i>V</i> , Å <sup>3</sup>	877.31(7)	3502.96(17)	1780.49(7)	6093.5(3)	6352.3(2)
<b>Z</b>	1	4	2	4	8
$\rho_{\text{calc}}$ , Mg/m <sup>3</sup>	1.309	1.319	1.346	1.238	1.334
$\mu$ (Mo K $\alpha$ ), mm <sup>-1</sup>	1.398	1.405	1.382	0.834	0.810
Reflections collected	7035	15350	13549	15464	36959
No. of parameters	199	199	221	352	406
Indep. reflns	3973	4300	4231	7071	7946
( <i>R</i> <sub>int</sub> )	(0.0146)	(0.0237)	(0.0171)	(0.0391)	(0.0235)
Final R indices	0.0241,	0.0274,	0.0338,	0.0455,	0.0276,
<i>R</i> <sub>1</sub> <sup>a</sup> , <i>wR</i> <sub>2</sub> <sup>a</sup>	0.0738	0.0698	0.1230	0.0699	0.0738
<i>R</i> indices (all data)	0.0264, 0.0748	0.0379, 0.0716	0.0414, 0.1267	0.0854, 0.0731	0.0433, 0.0779
GoF <sup>b</sup>	1.040	1.017	1.024	1.030	1.034

<sup>a</sup>  $P_1 = [\sum(|\Phi_0| - |\Phi_\chi|) / \sum |\Phi_0|]$ ;  $\omega P_2 = [\sum \omega(\Phi_0^2 - \Phi_\chi^2) / \sum \omega(\Phi_0^2)]^{1/2}$ ,  $\omega = 0.10$ .

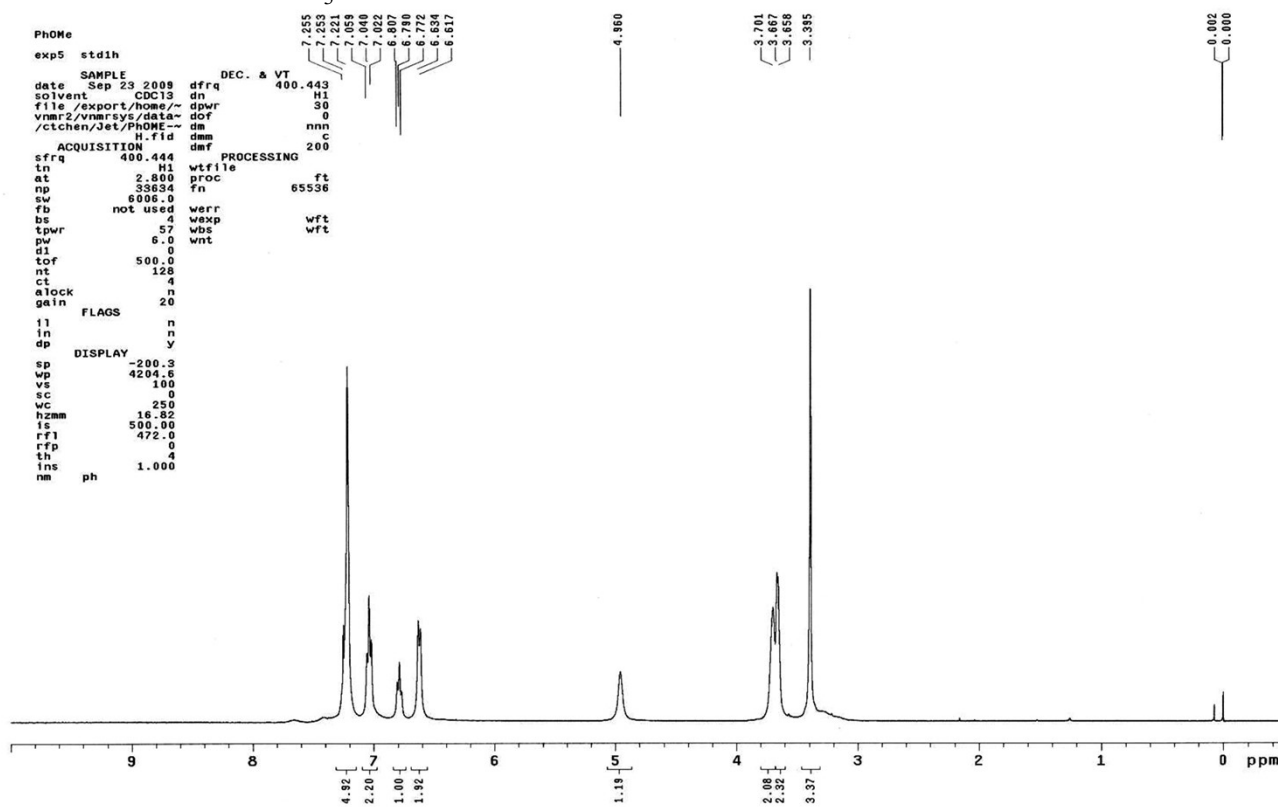
<sup>b</sup> GoF =  $[\sum w(F_0^2 - F_c^2)^2 / (N_{\text{reflns}} - N_{\text{params}})]^{1/2}$

## Spectra of ligand precursors L1a-L1b

### <sup>1</sup>H NMR of L1a in CDCl<sub>3</sub>



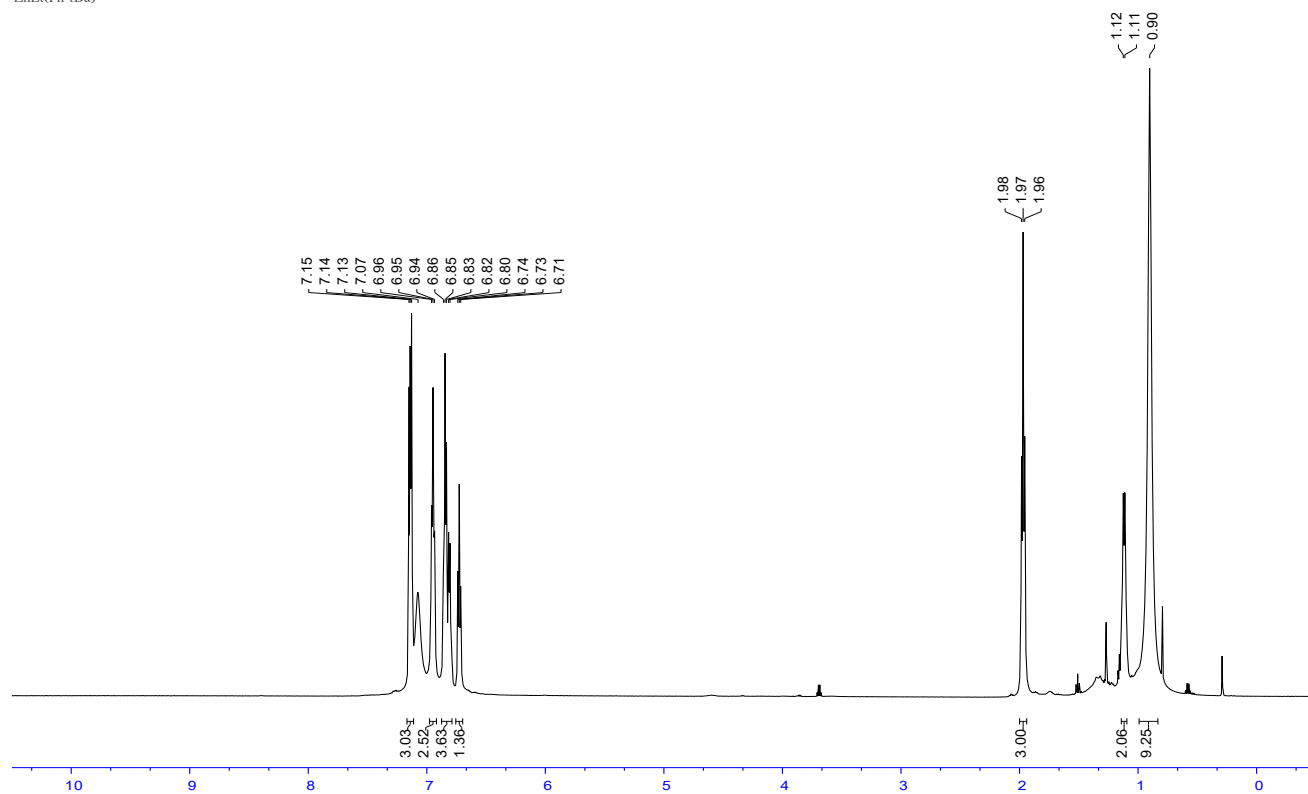
### <sup>1</sup>H NMR of L1b in CDCl<sub>3</sub>



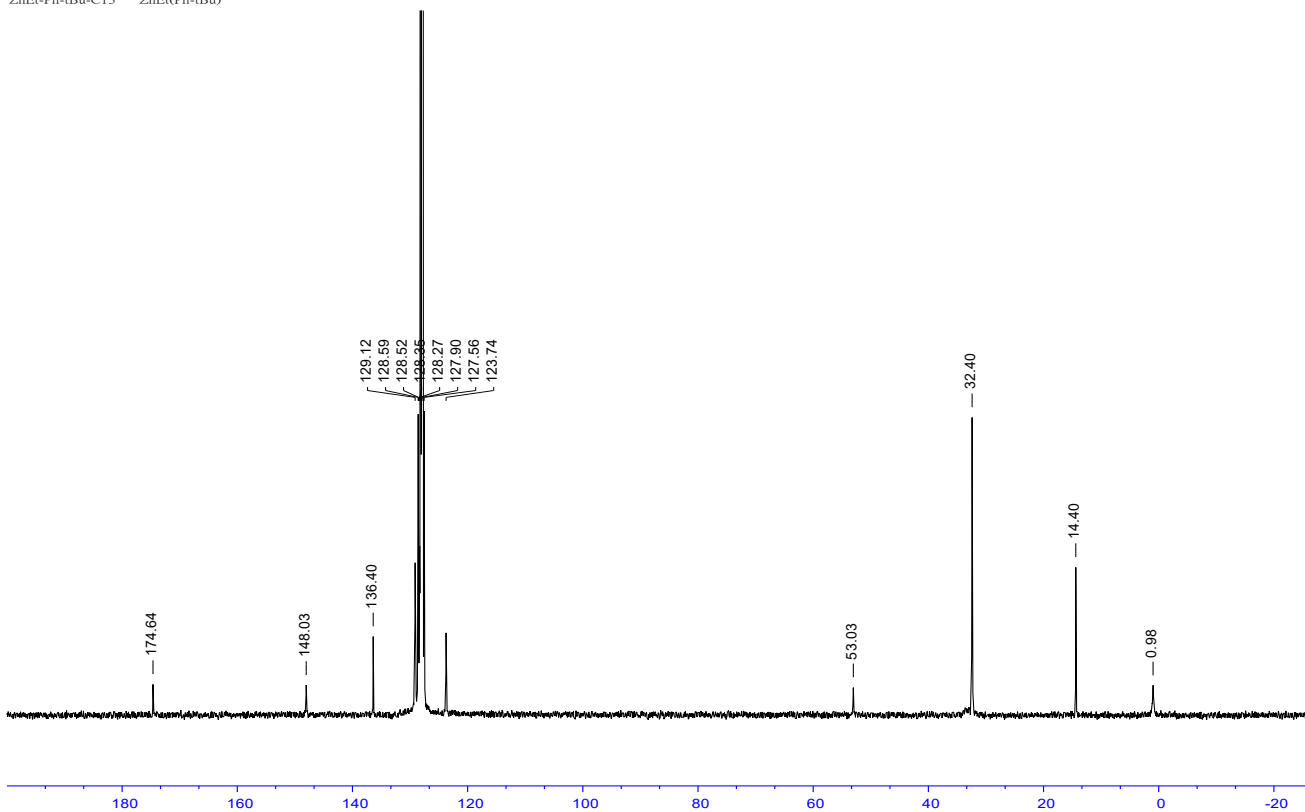
# Spectra of complexes 1-7

$^1\text{H}$  NMR of complex 1 in  $\text{C}_6\text{D}_6$

ZnEt(Ph-tBu-H)  
ZnEt(Ph-tBu)

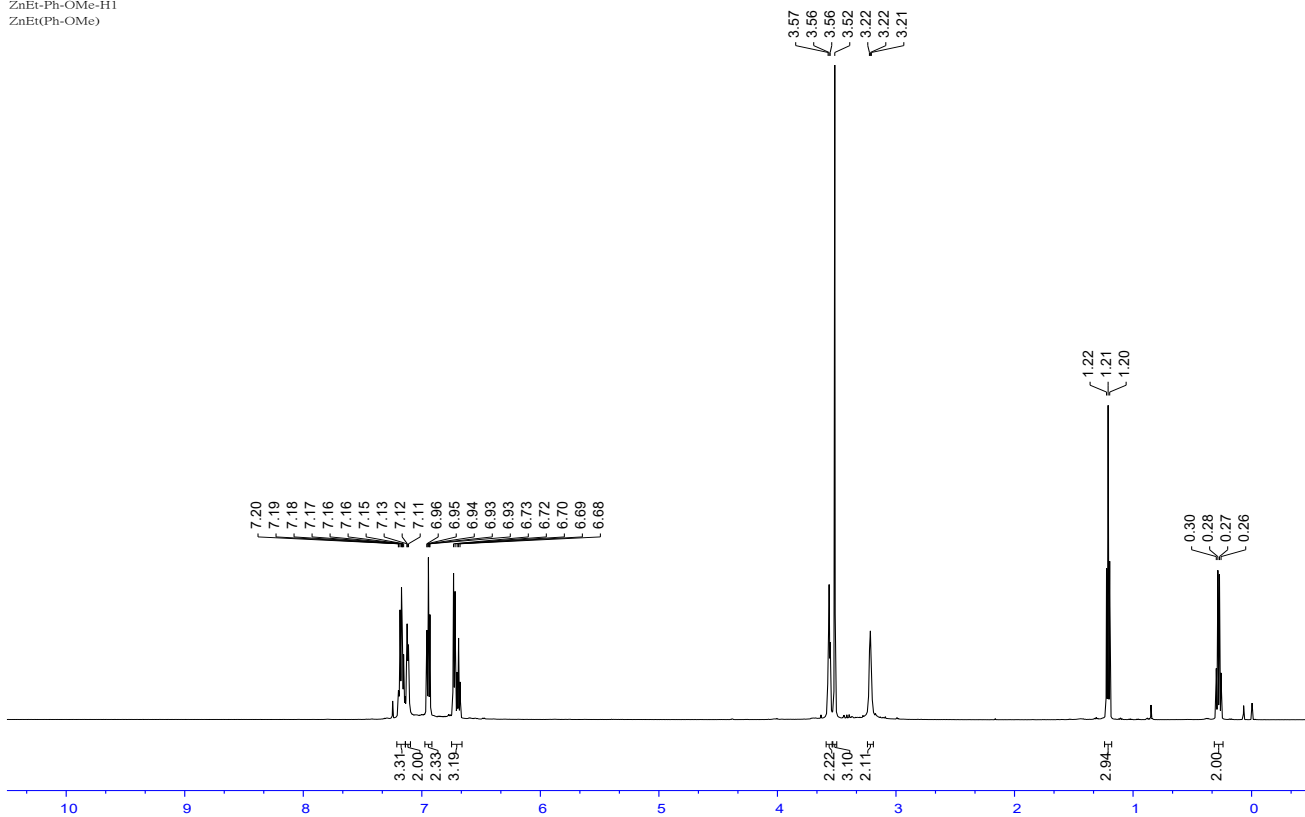


$^{13}\text{C}\{^1\text{H}\}$  NMR of complex 1 in  $\text{C}_6\text{D}_6$



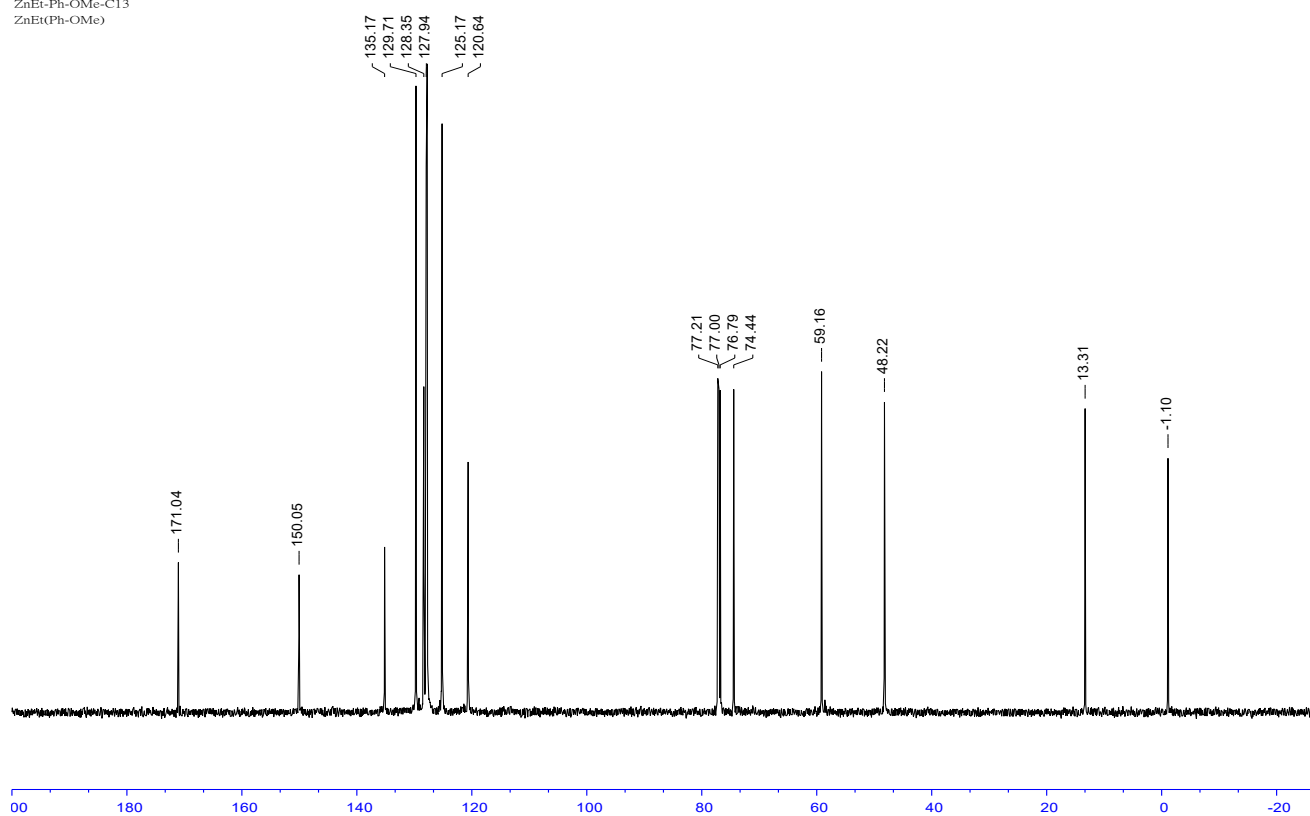
### <sup>1</sup>H NMR of complex **2** in CDCl<sub>3</sub>

ZnEt-Ph-OMe-H1  
ZnEt(Ph-OMe)



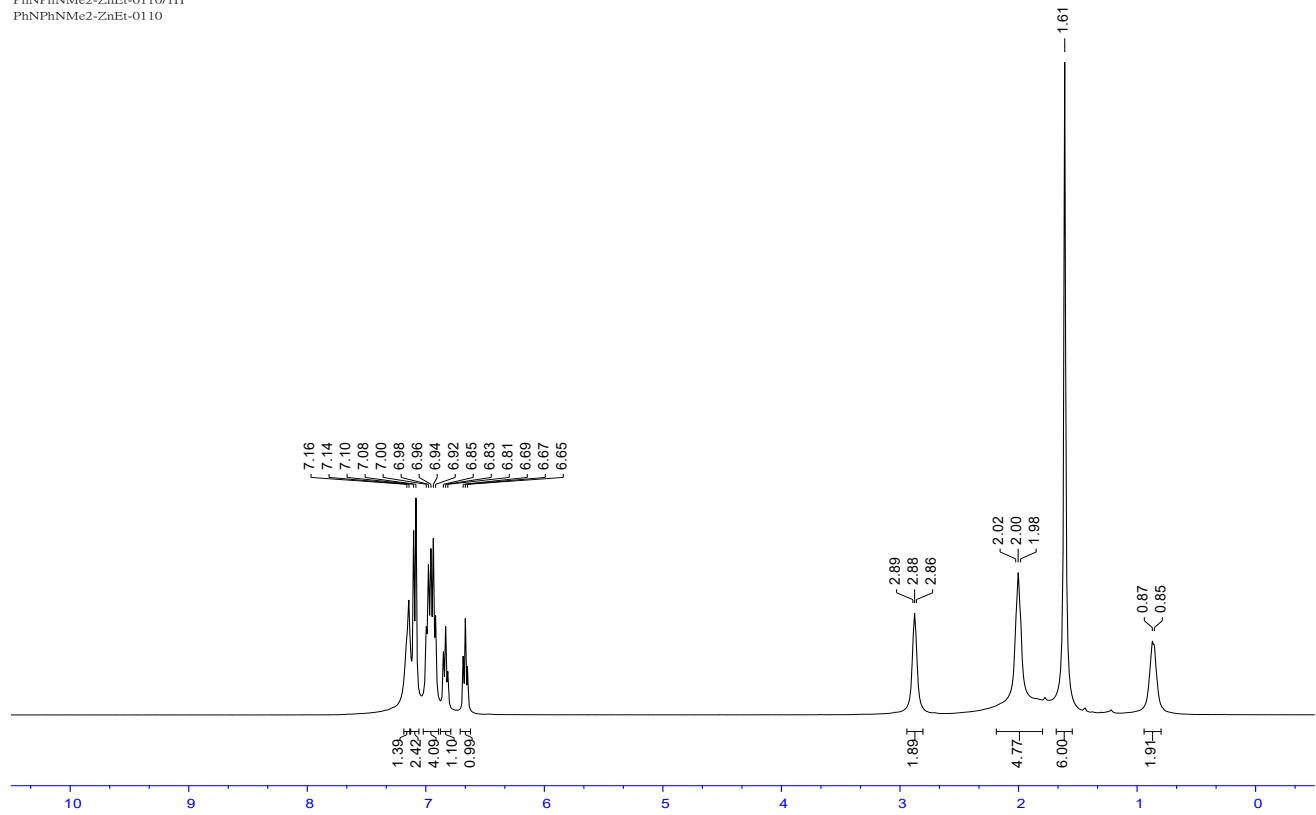
$^{13}\text{C}\{^1\text{H}\}$  NMR of complex **2** in  $\text{CDCl}_3$

ZnEt-Ph-OMe- $\text{C}^{13}$   
ZnEt(Ph-OMe)



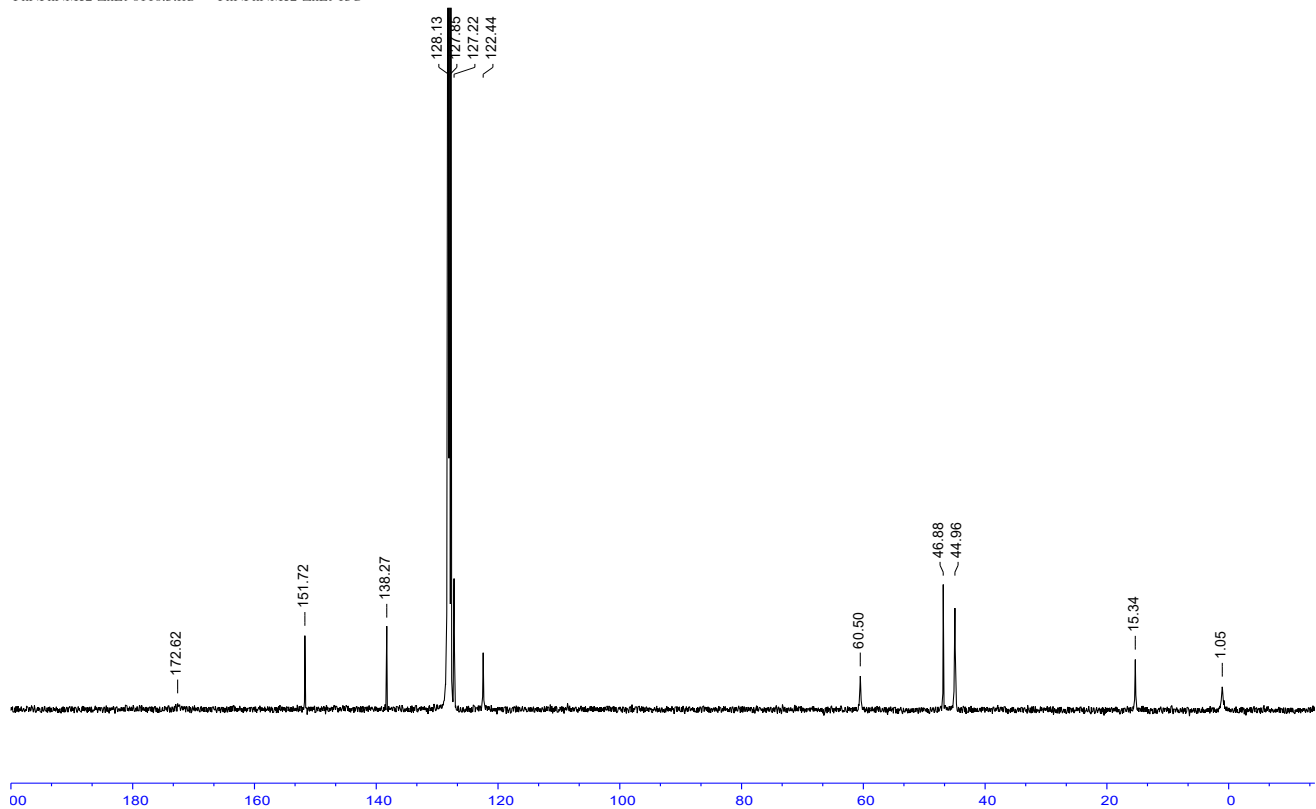
### $^1\text{H}$ NMR of complex **3** in $\text{C}_6\text{D}_6$

PhNPhNMe2-ZnEt-0110/1H  
PhNPhNMe2-ZnEt-0110



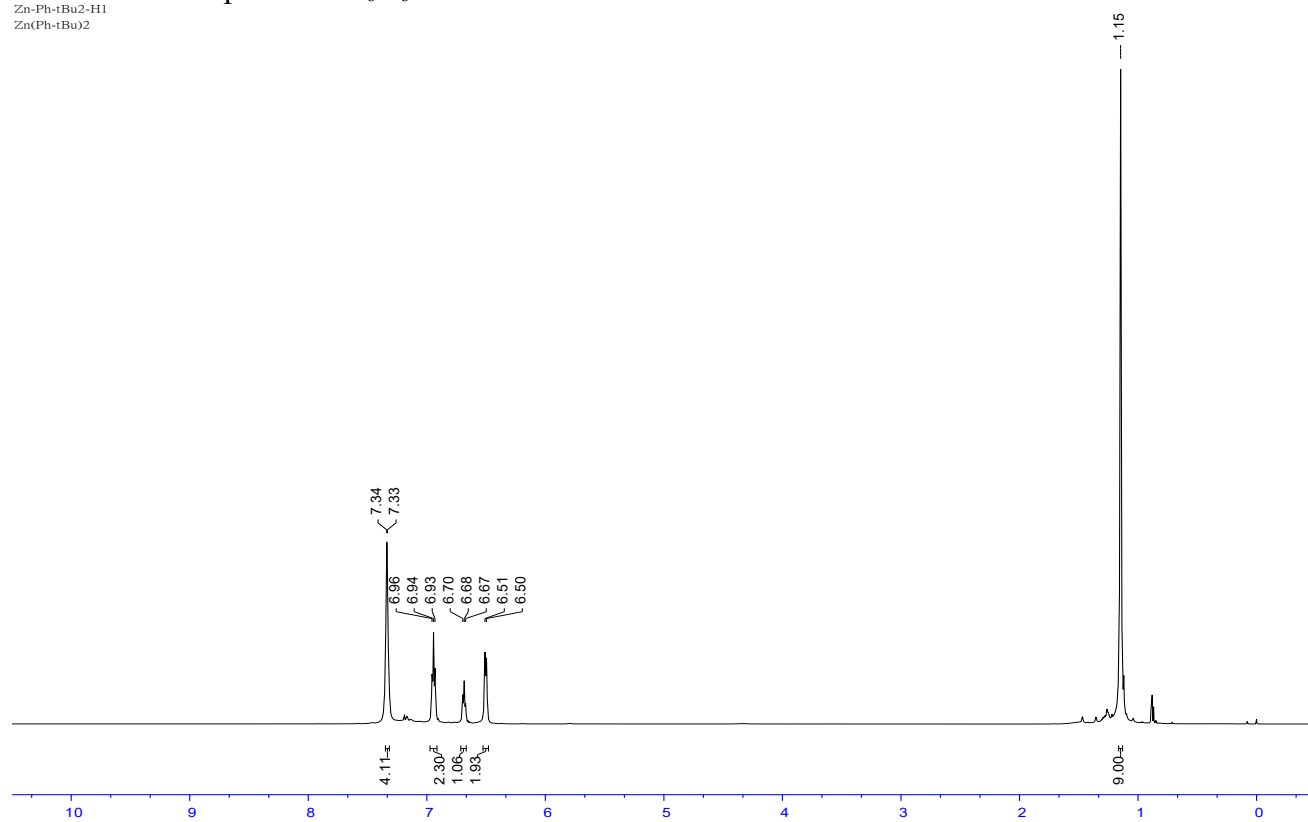
### $^{13}\text{C}\{^1\text{H}\}$ NMR of complex **3** in $\text{C}_6\text{D}_6$

PhNPhNMe2-ZnEt-0110.3.fid — PhNPhNMe2-ZnEt-13C



$^1\text{H}$  NMR of complex **4** in  $\text{C}_6\text{D}_6$

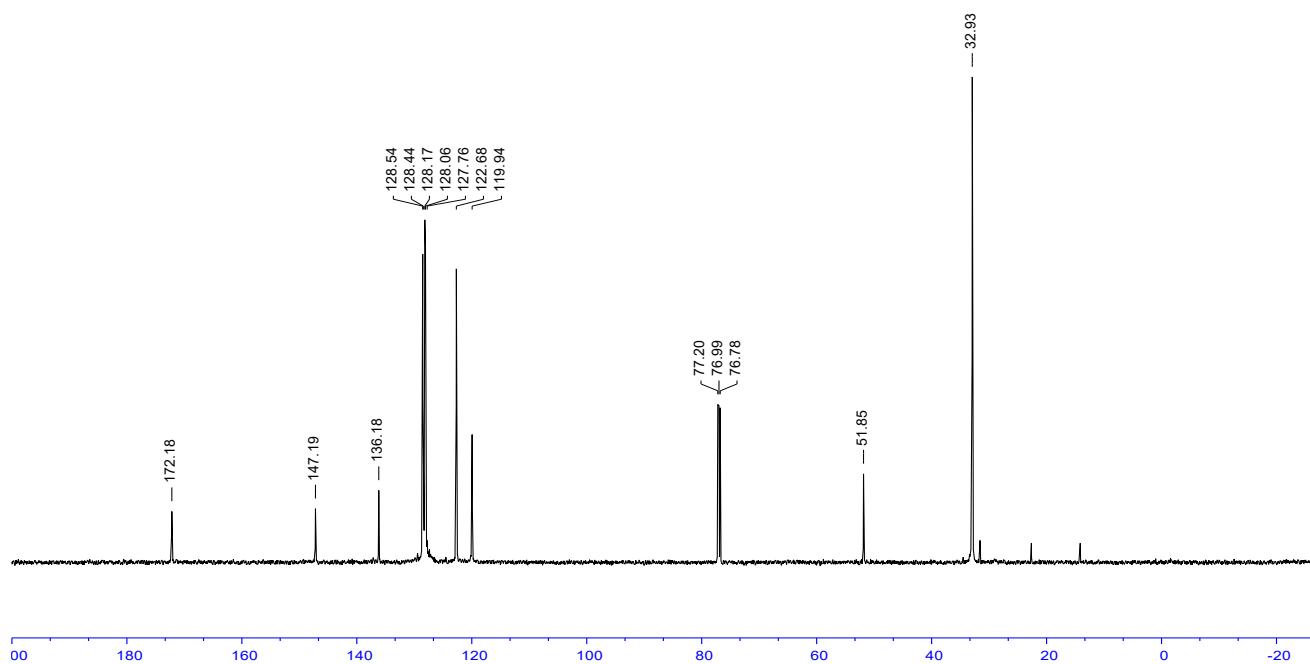
Zn-Ph-tBu2-H1  
Zn(Ph-tBu)2



$^{13}\text{C}\{^1\text{H}\}$  NMR of complex **4** in  $\text{C}_6\text{D}_6$

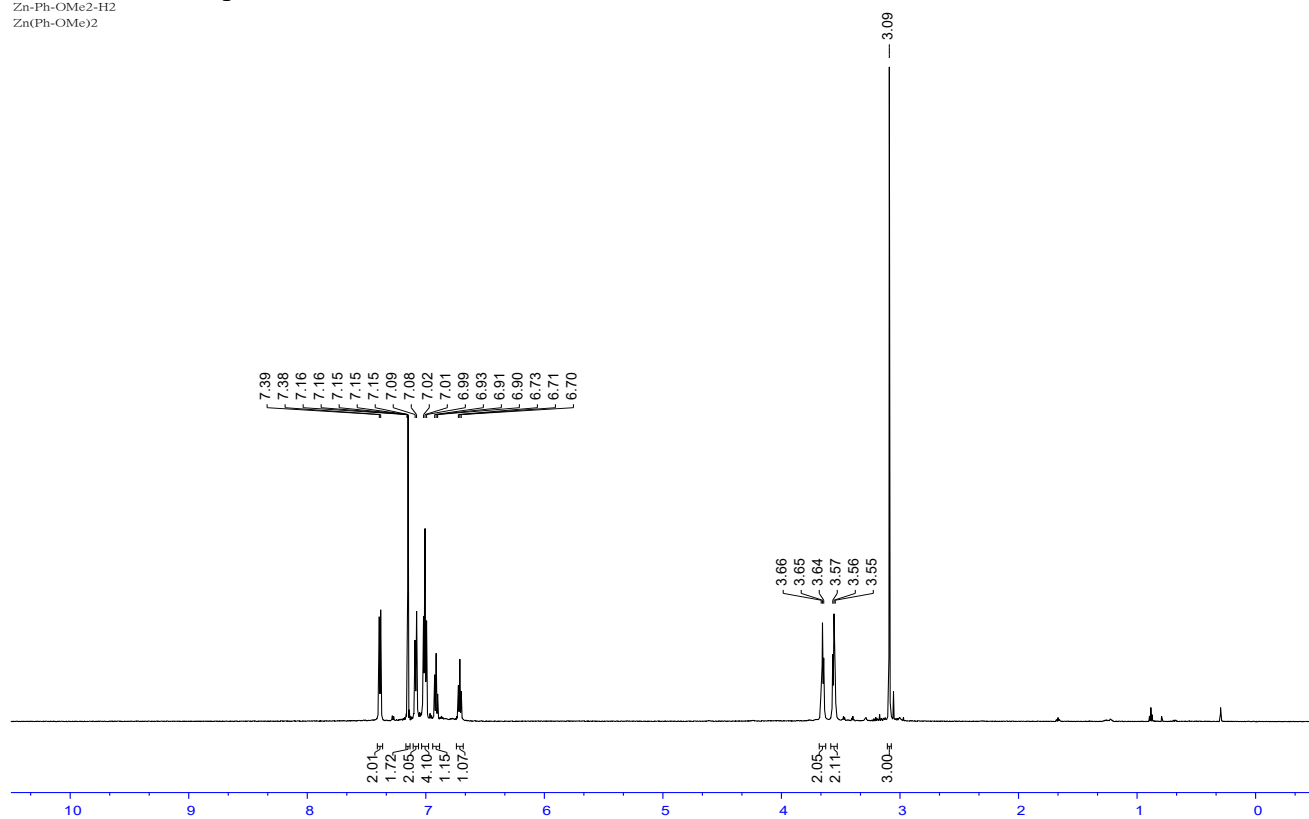


Zn-Ph-tBu2-C13  
Zn(Ph-tBu)2



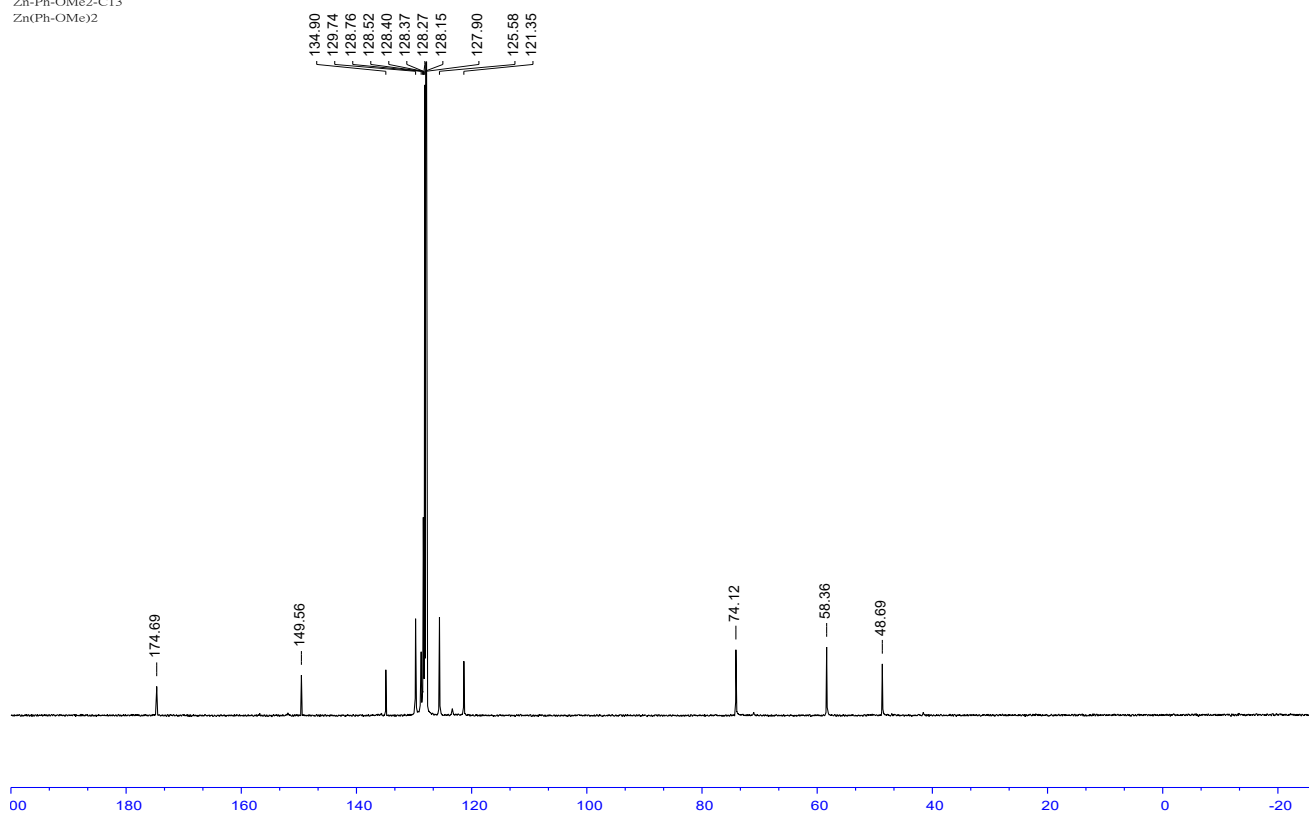
### $^1\text{H}$ NMR of complex 5 in $\text{C}_6\text{D}_6$

Zn-Ph-OMe2-H2  
Zn(Ph-OMe) $_2$



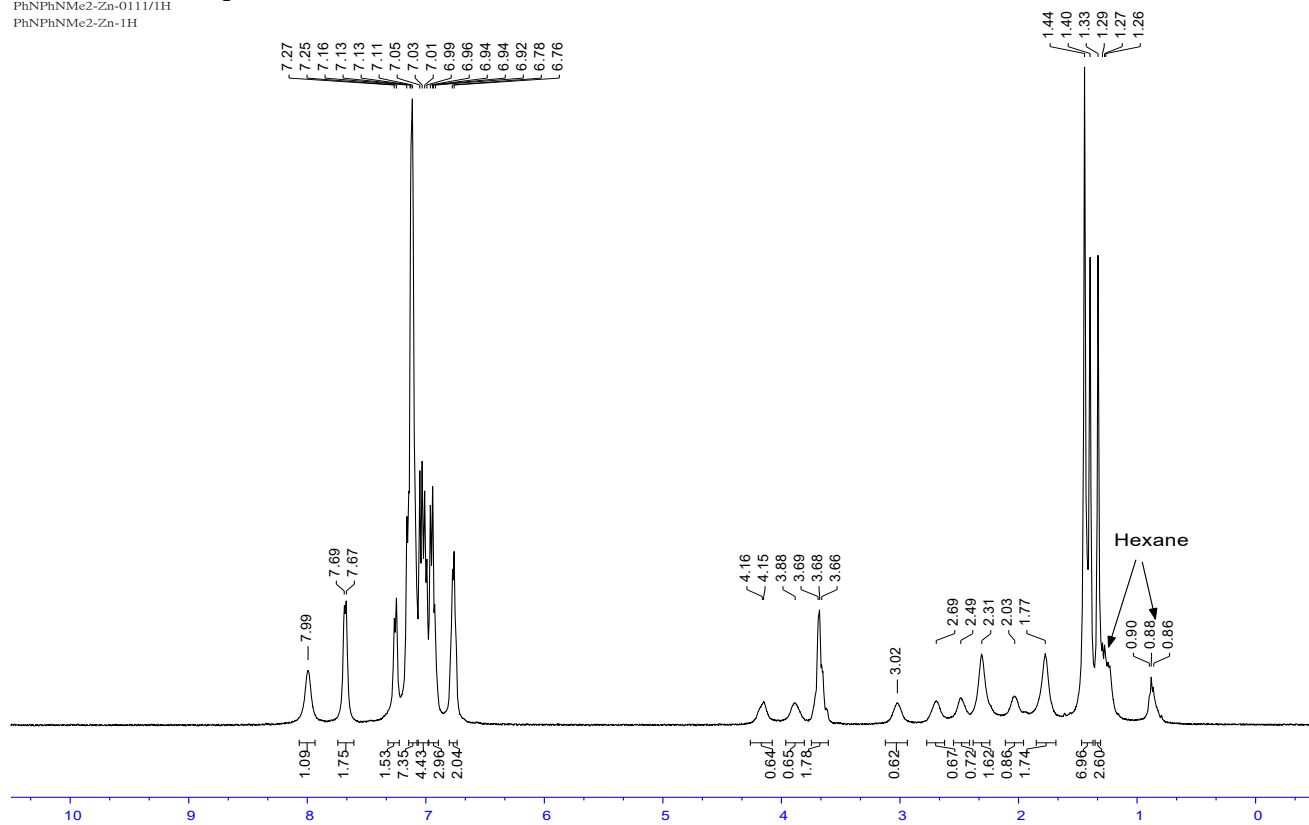
$^{13}\text{C}\{^1\text{H}\}$  NMR of complex **5** in  $\text{C}_6\text{D}_6$

Zn-Ph-OMe2-C13  
Zn(Ph-OMe)2

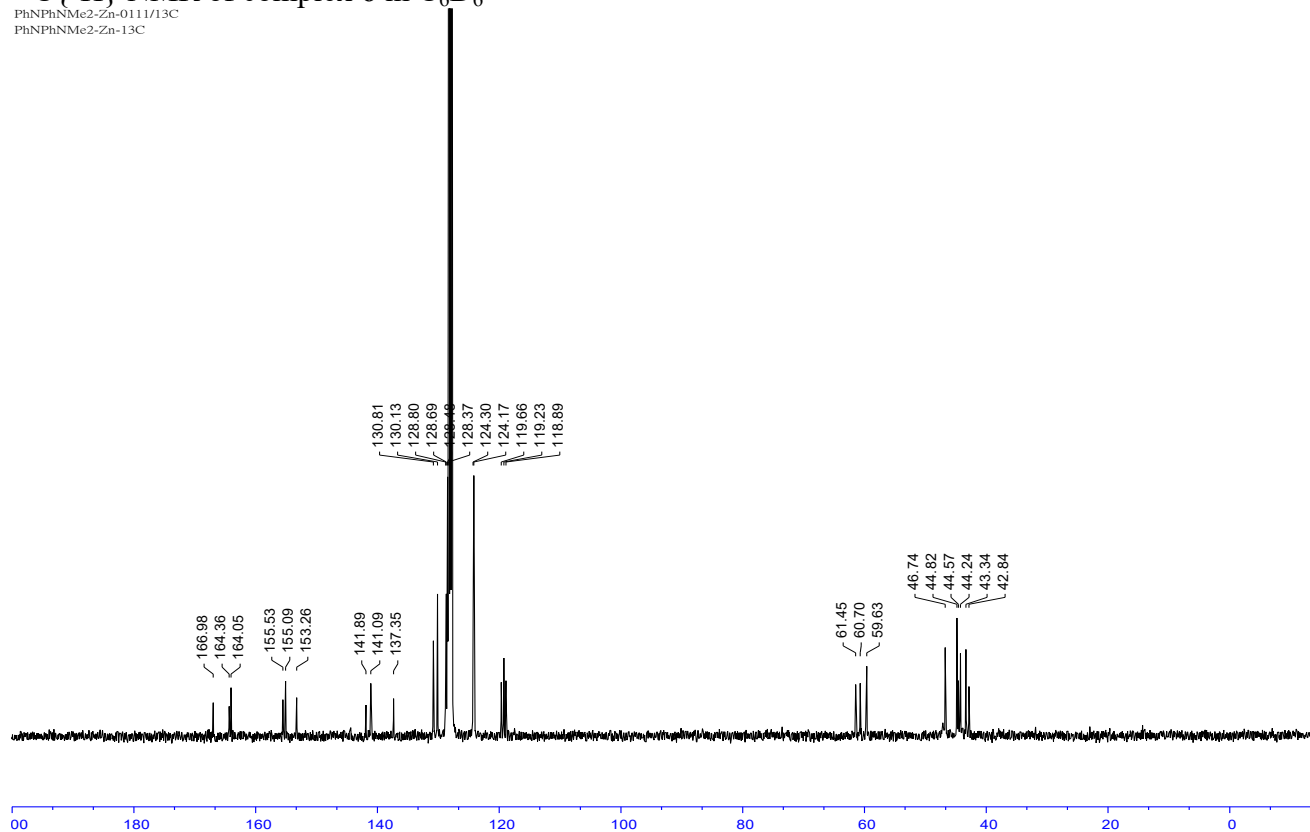


$^1\text{H}$  NMR of complex **6** in  $\text{C}_6\text{D}_6$

PhNPhNMe2-Zn-0111/1H  
PhNPhNMe2-Zn-1H

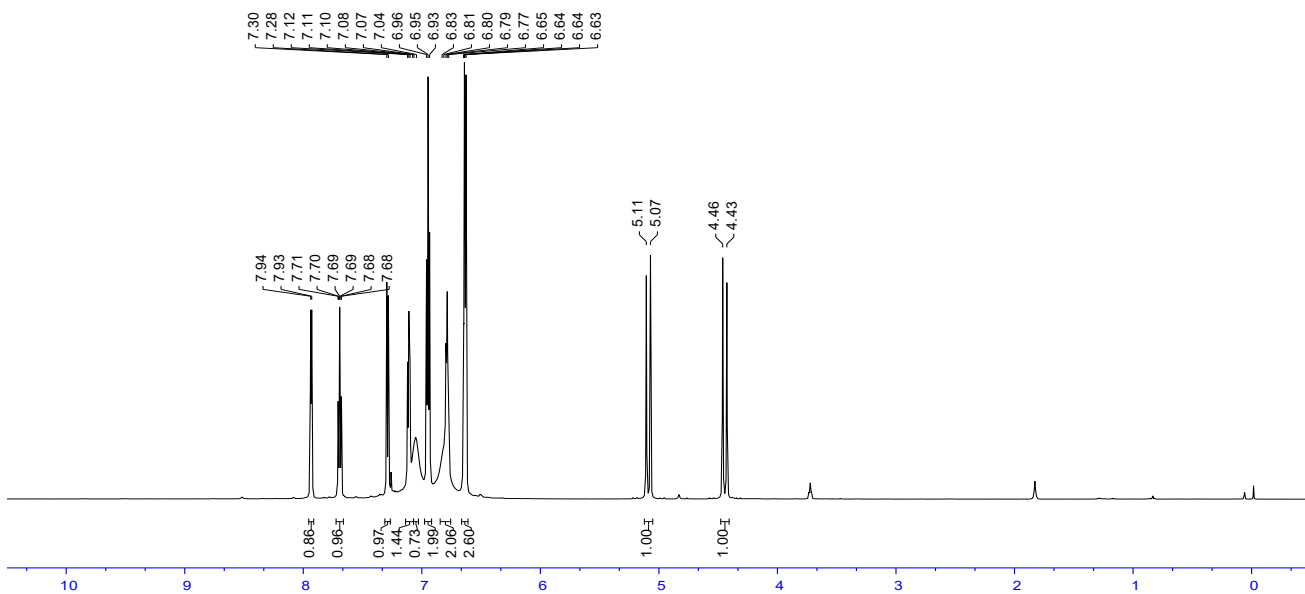


$^{13}\text{C}\{^1\text{H}\}$  NMR of complex **6** in  $\text{C}_6\text{D}_6$   
PhNPhNMe<sub>2</sub>-Zn-0111/13C  
PhNPhNMe<sub>2</sub>-Zn-13C

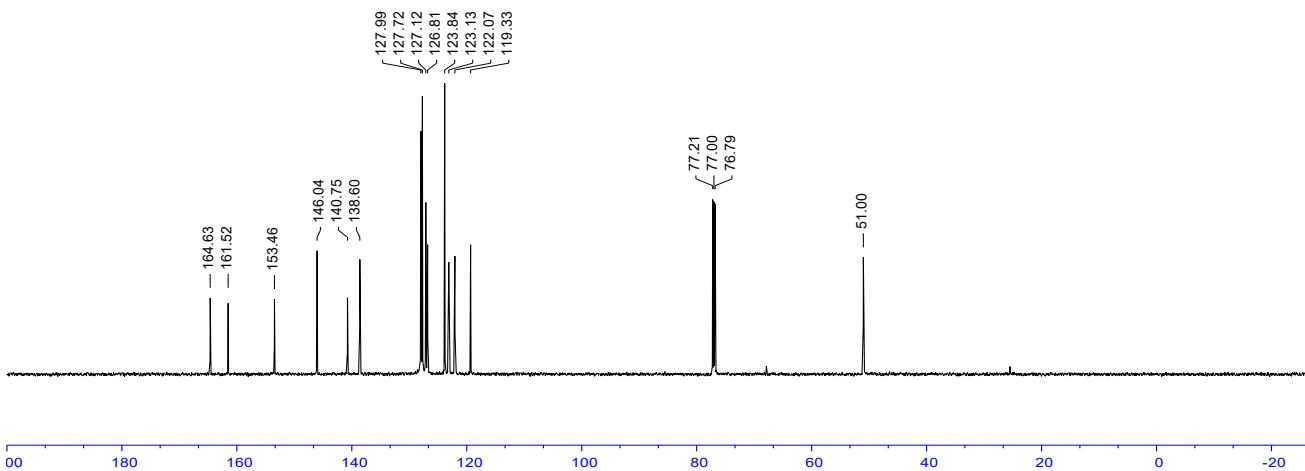


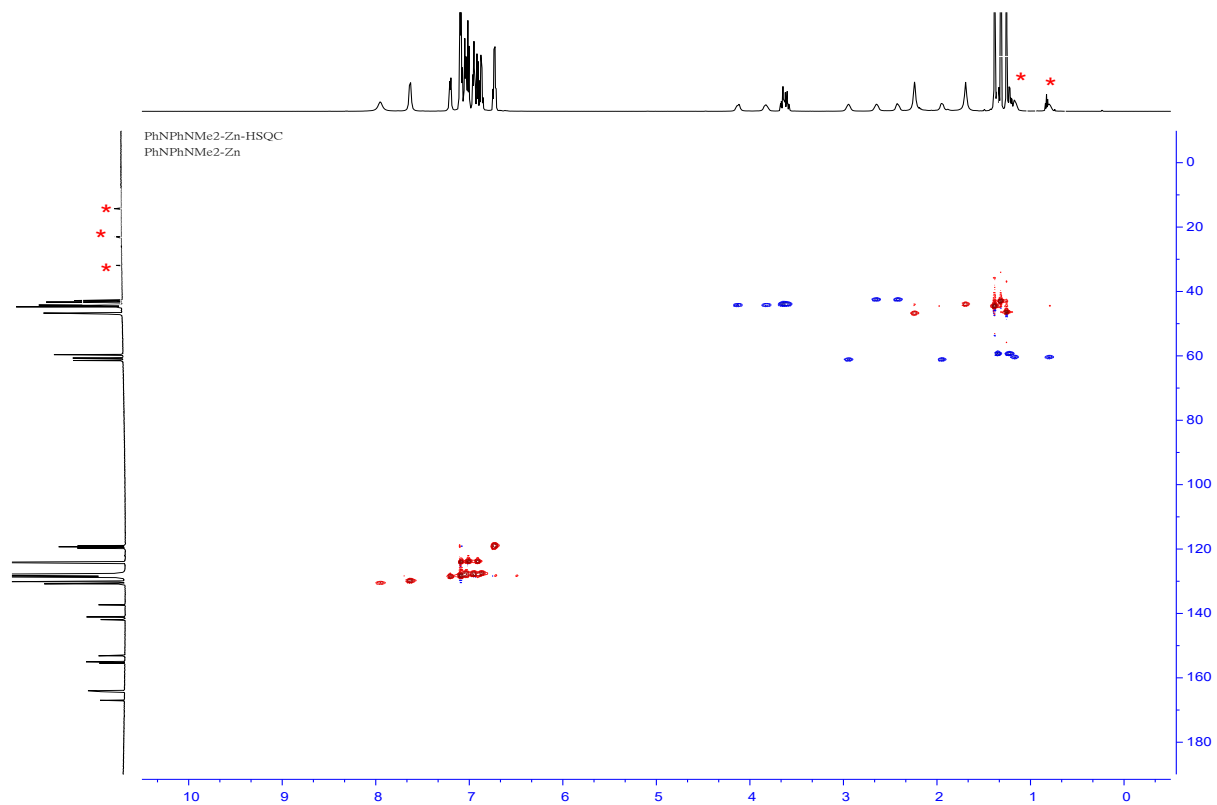
$^1\text{H}$  NMR of complex **7** in  $\text{CDCl}_3$

Zn-Ph-Py2-H1  
Zn(Ph-Py)2

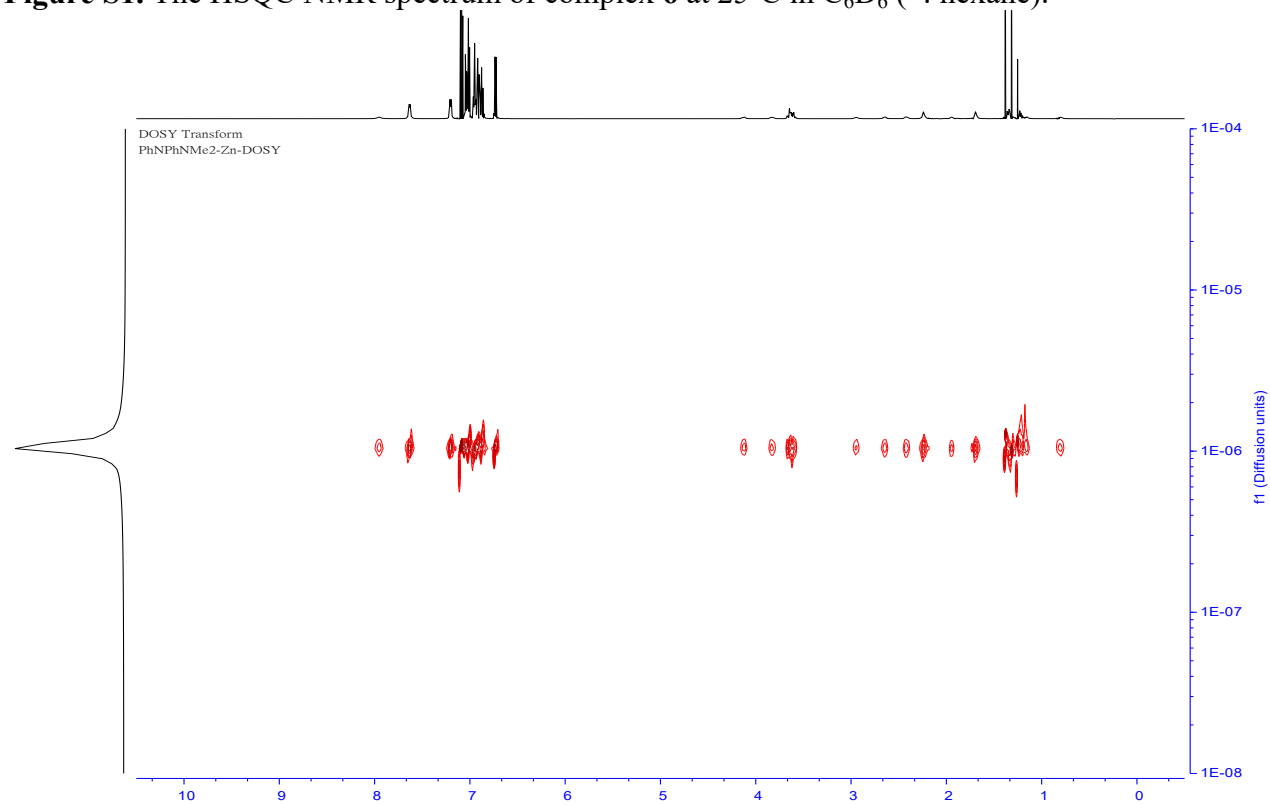


$^{13}\text{C}\{^1\text{H}\}$  NMR of complex **7** in CDCl<sub>3</sub>  
Zn-Ph-Py2-C15  
Zn(Ph-Py)2



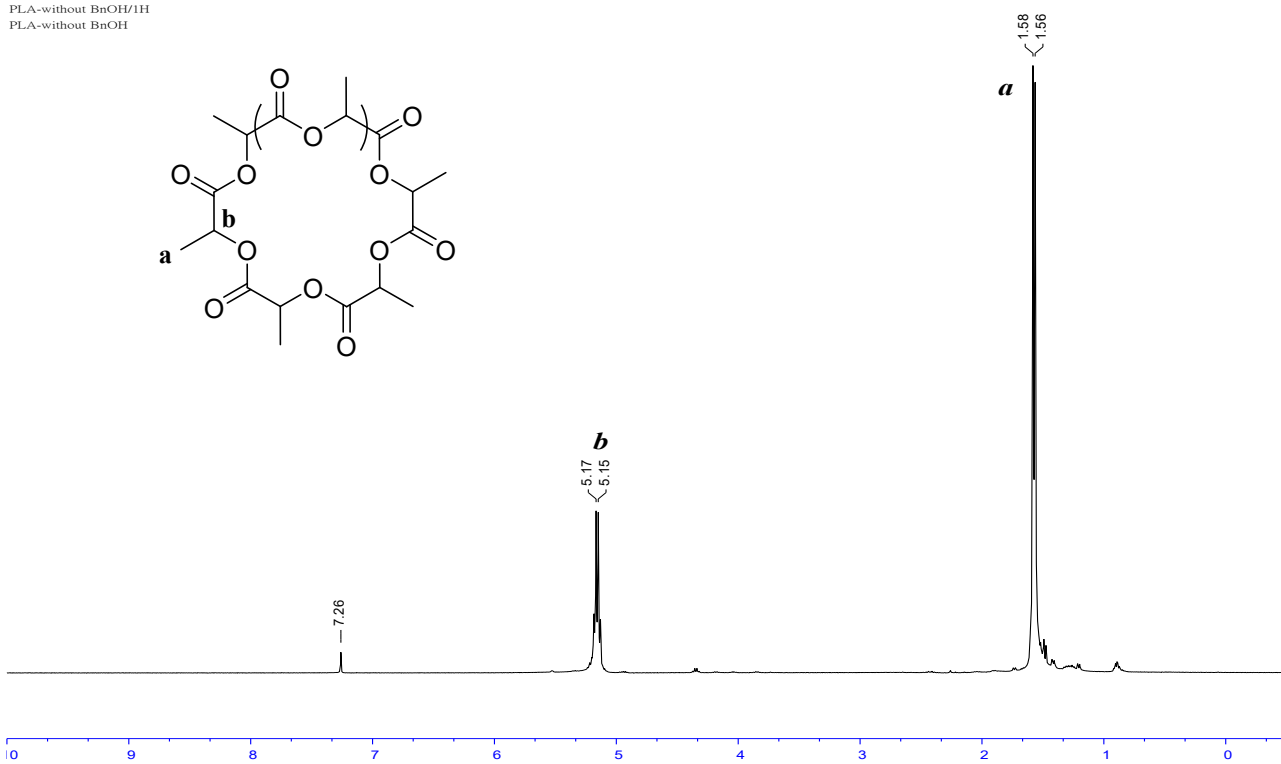


**Figure S1.** The HSQC NMR spectrum of complex **6** at 25°C in C<sub>6</sub>D<sub>6</sub> (\*: hexane).



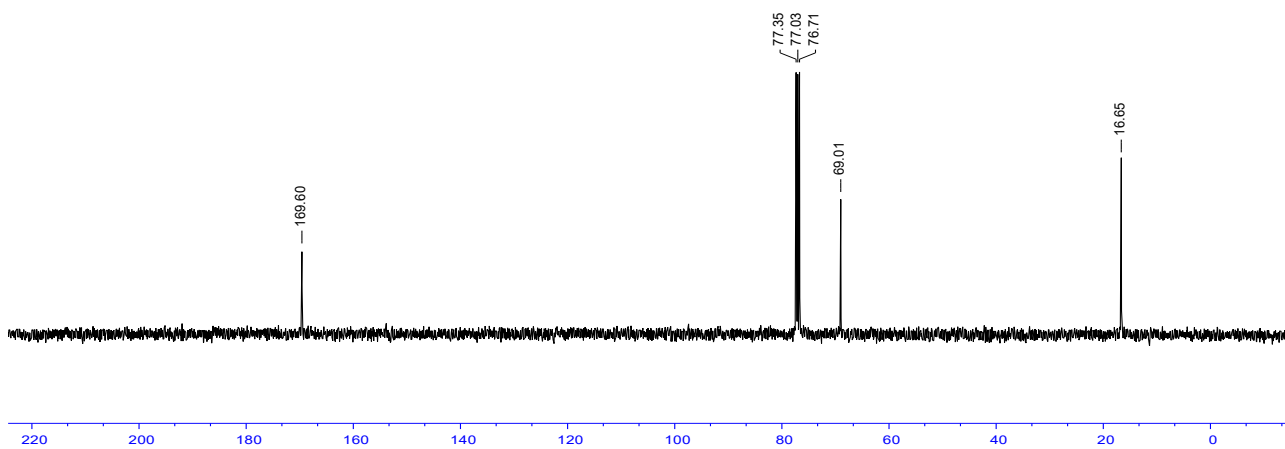
**Figure S2.** The <sup>1</sup>H DOSY NMR spectrum of complex **6** obtained at 25°C in C<sub>6</sub>D<sub>6</sub>.

PLA-without BnOH/1H  
PLA-without BnOH

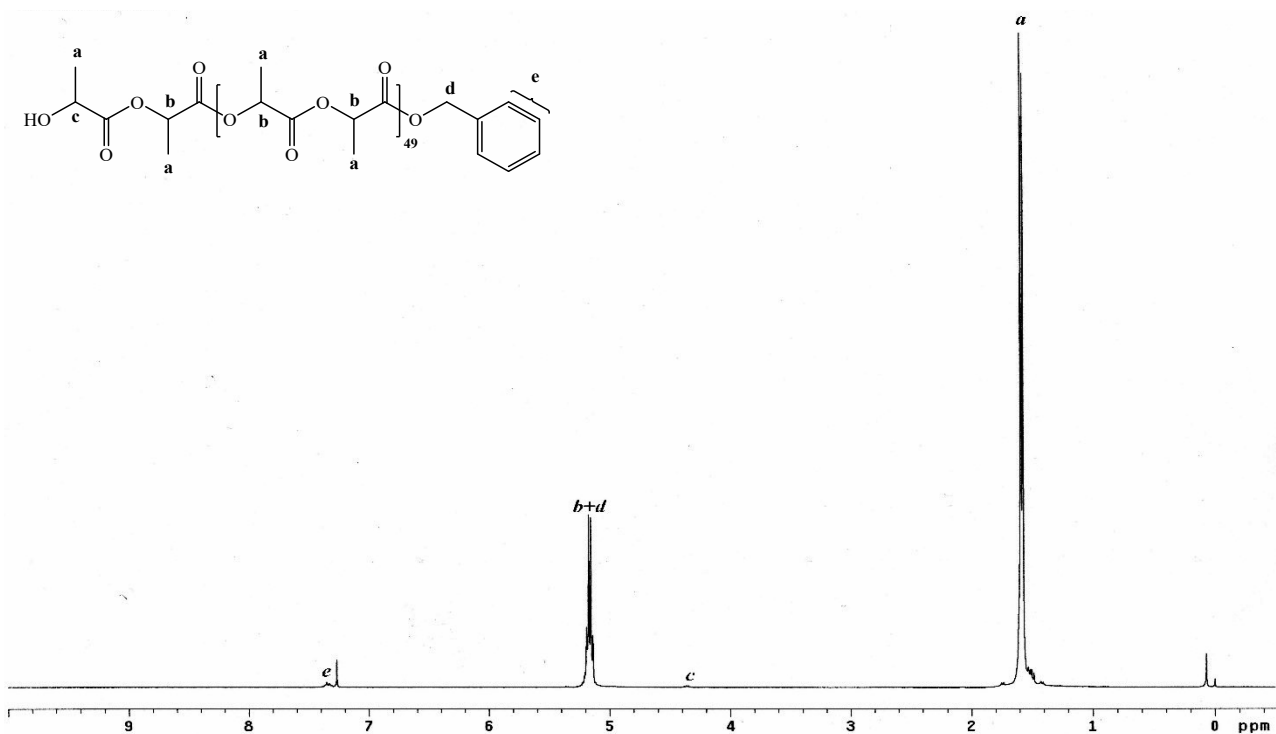


**Figure S3.** The  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ ) of PLA-50 catalyzed by **7** in toluene at  $25^\circ\text{C}$ .

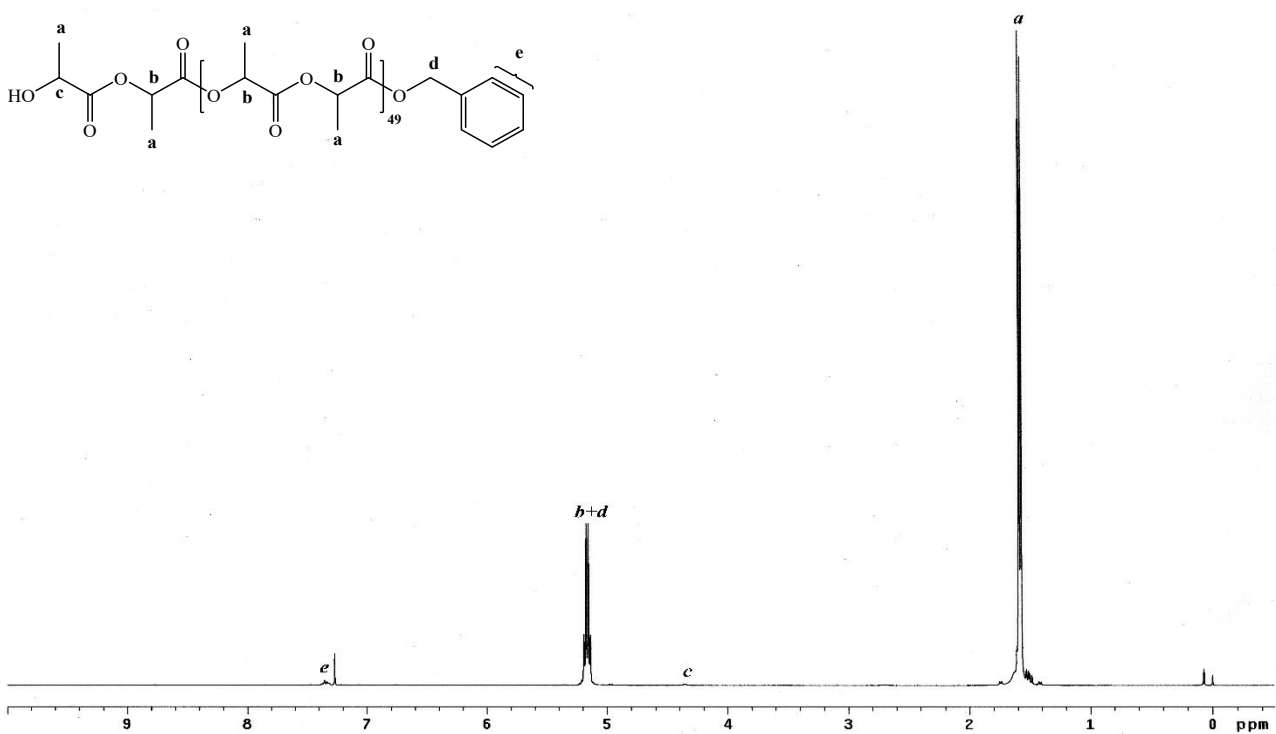
PLA-without BnOH.7.fid  
PLA-without BnOH-13C



**Figure S4.** The  $^{13}\text{C}\{^1\text{H}\}$  NMR spectrum ( $\text{CDCl}_3$ ) of PLA-50 catalyzed by **7** in toluene at  $25^\circ\text{C}$ .



**Figure S5.** The <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of PLA-50 catalyzed by **2**/BnOH in toluene at 50°C.

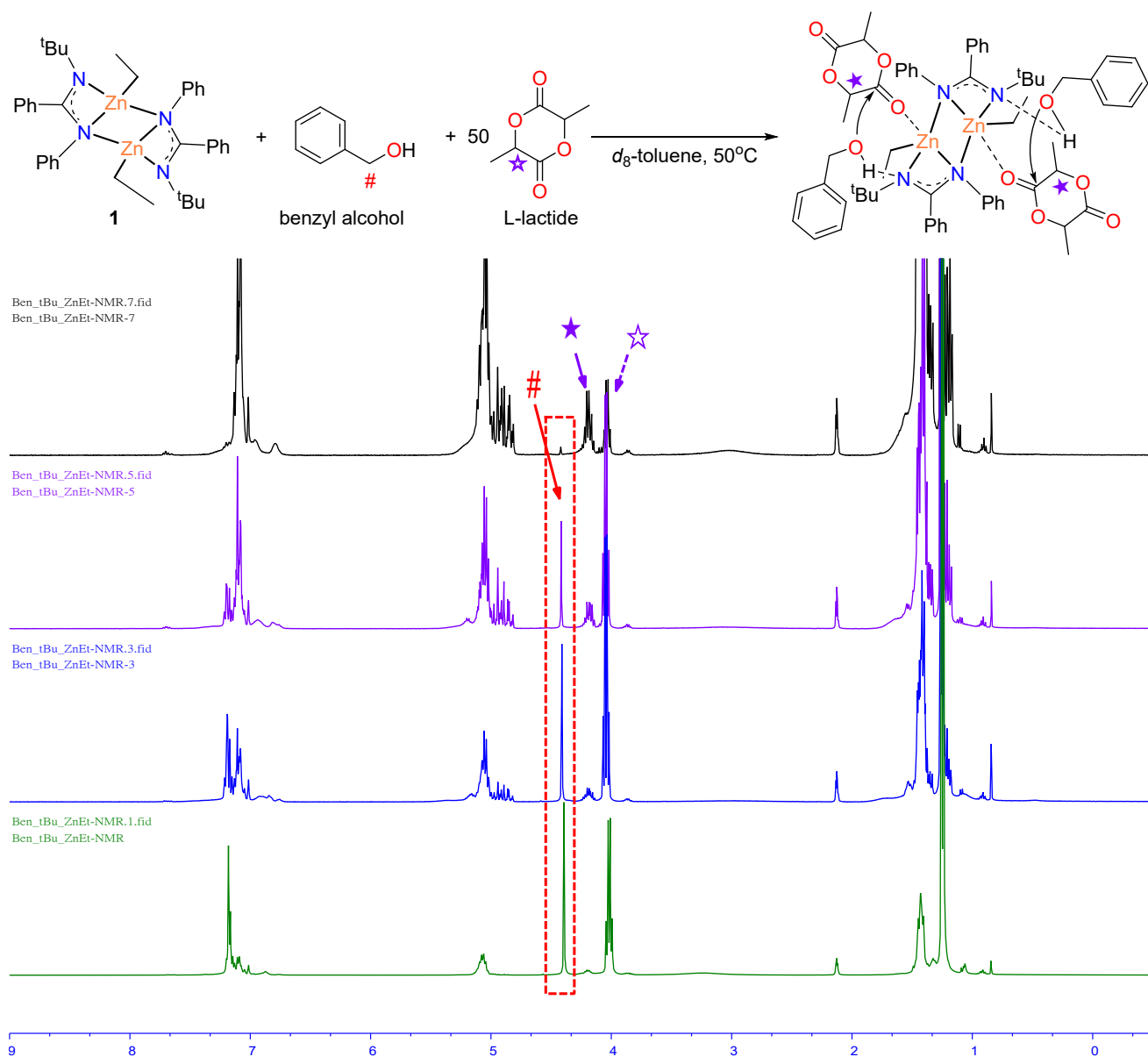


**Figure S6.** The <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of PLA-50 catalyzed by **7**/BnOH in toluene at 25°C.

## $^1\text{H}$ NMR study of the Mechanism in the ROP of L-lactide

\*\*\* In a typical experiment, a J. Young NMR tube was charged with a solution of complex (0.0125 mmol for **1**; 0.025 mmol for **7**), benzyl alcohol (0.025 mmol, 2.6  $\mu\text{L}$ ), and L-lactide (1.25 mmol, 0.18 g) in 0.8 mL of  $d_8$ -toluene. The tube was then transferred to the spectrometer, and  $^1\text{H}$  NMR spectra were recorded at the reaction temperature at different time. The internal standard was referenced to  $d_8$ -toluene (2.35 ppm).

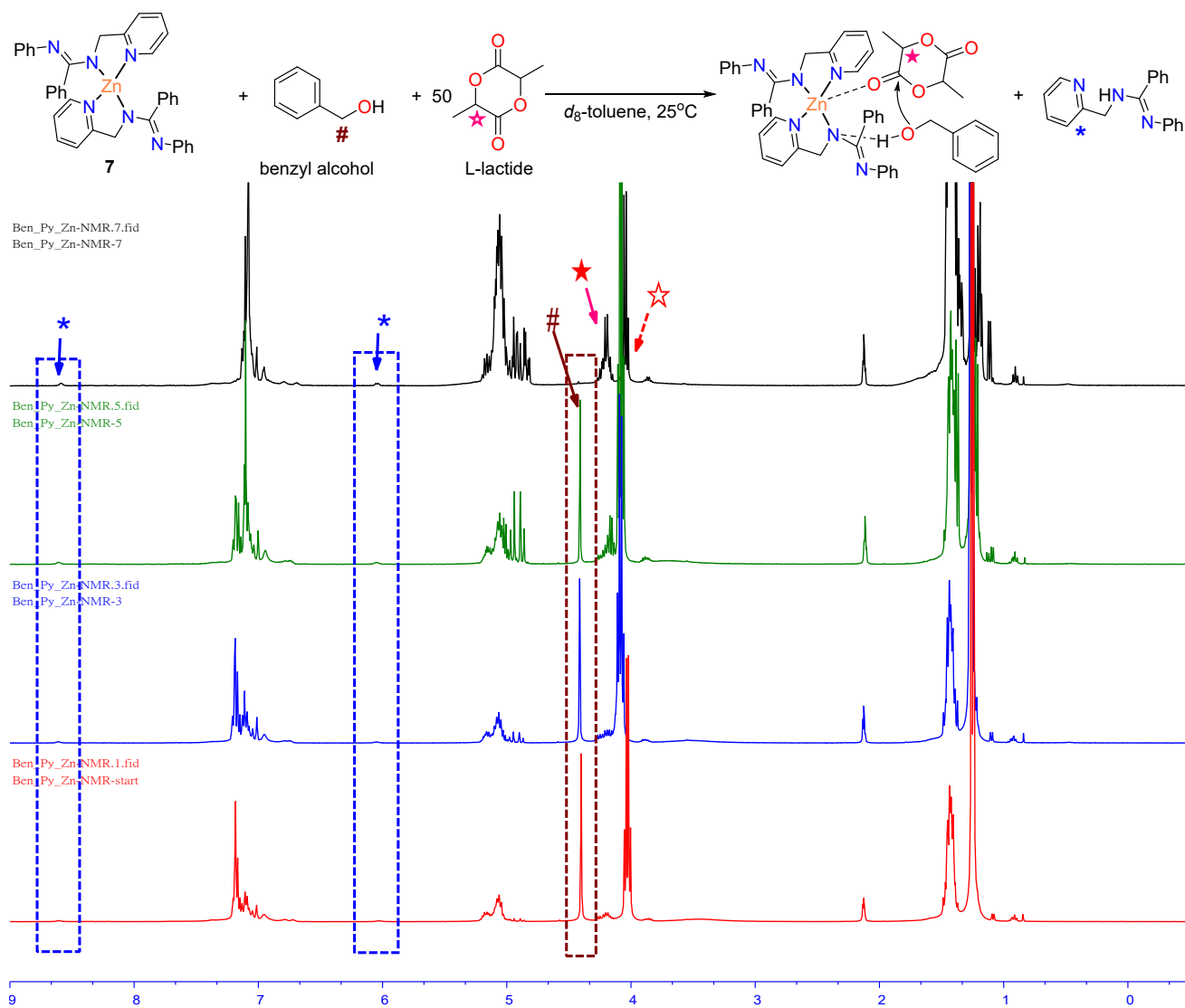
For complex **1**



**Figure S7.**  $^1\text{H}$  NMR spectra of complex **1**/BnOH/L-LA at different time at 50°C in  $d_8$ -toluene (#benzyl alcohol; ★activated L-lactide).

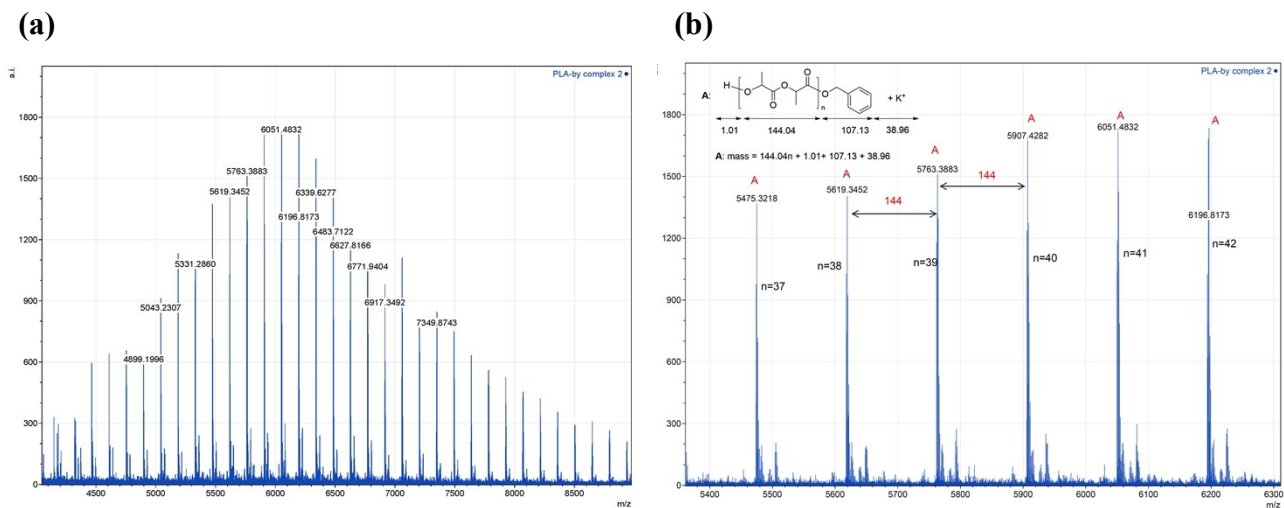


For complex **7**

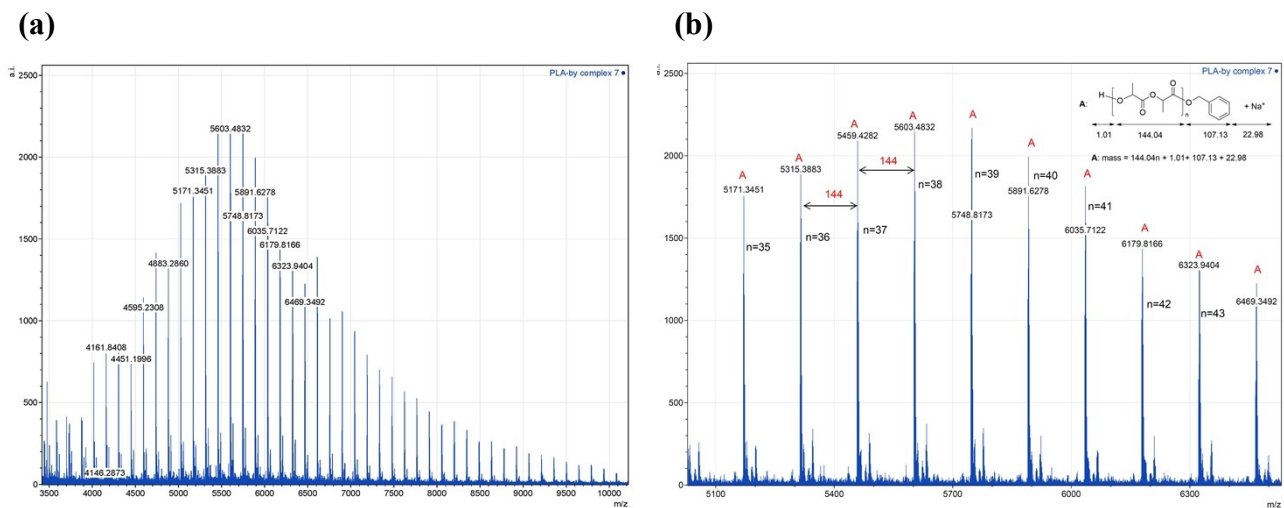


**Figure S8.**  $^1\text{H}$  NMR spectra of complex **7**/BnOH/L-LA at different time at 25°C in  $d_8$ -toluene (\*ligand precursor **L1d**; #benzyl alcohol; ★activated L-lactide).

## MALDI-TOF mass analysis of prepared PLAs



**Figure S9.** (a)-(b) MALDI-TOF spectrum of PLA initiated by L-LA/complex 2/BnOH (Table 2, entry 11).



**Figure S10.** (a)-(b) MALDI-TOF spectrum of PLA initiated by L-LA/complex 7/BnOH (Table 2, entry 1).