

## Electronic Supplementary Information

### Crystal-to-Crystal Polymerisation of Monosubstituted [PW<sub>11</sub>O<sub>39</sub>Cu(H<sub>2</sub>O)]<sup>5-</sup> Keggin-Type Anions

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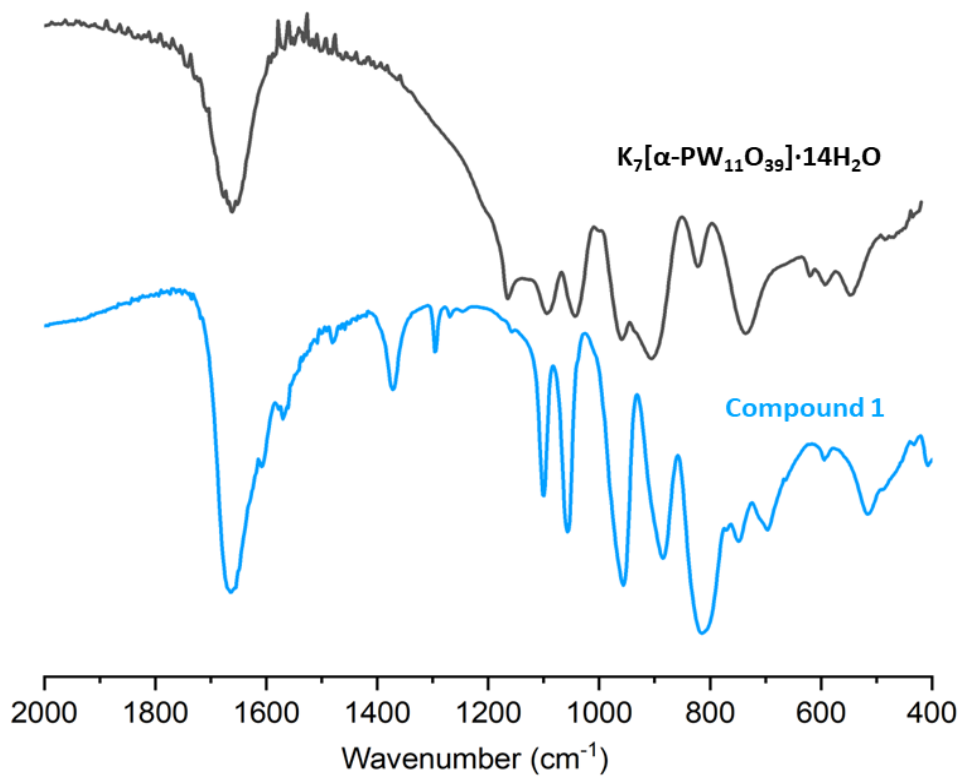
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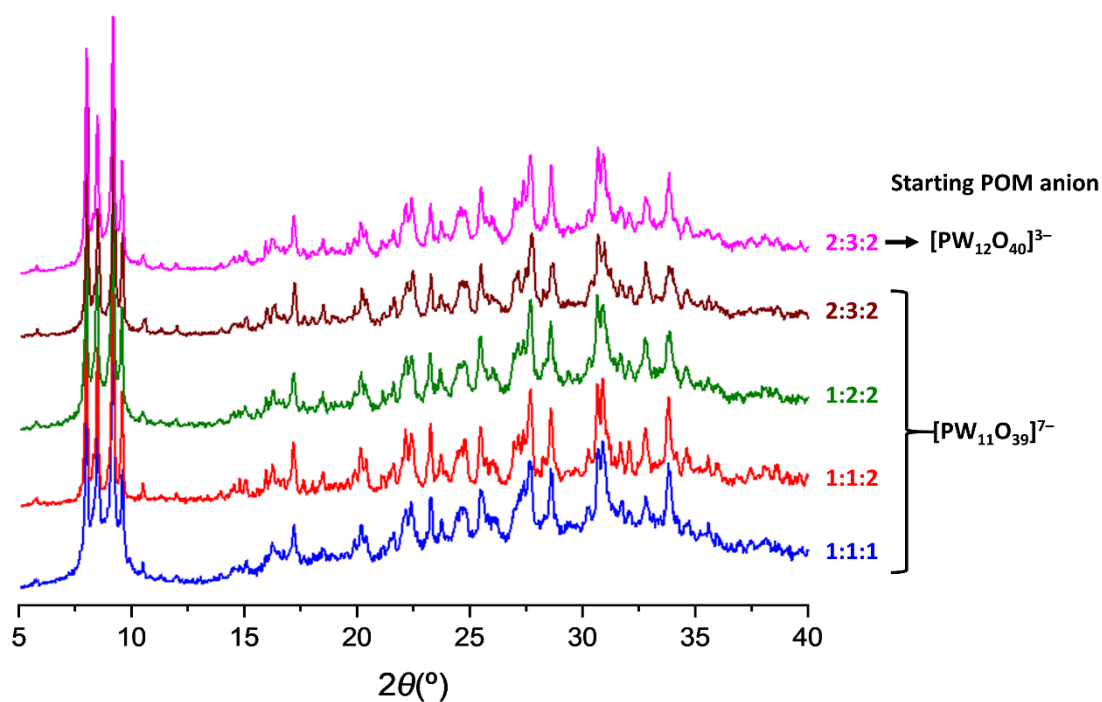
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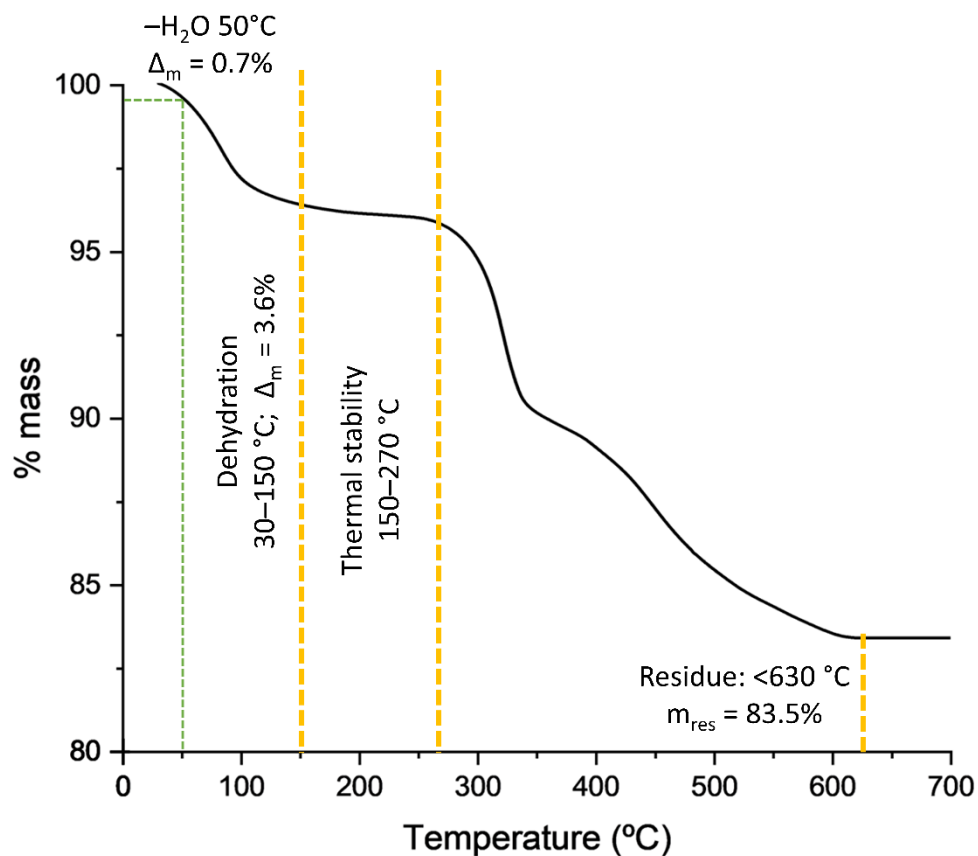
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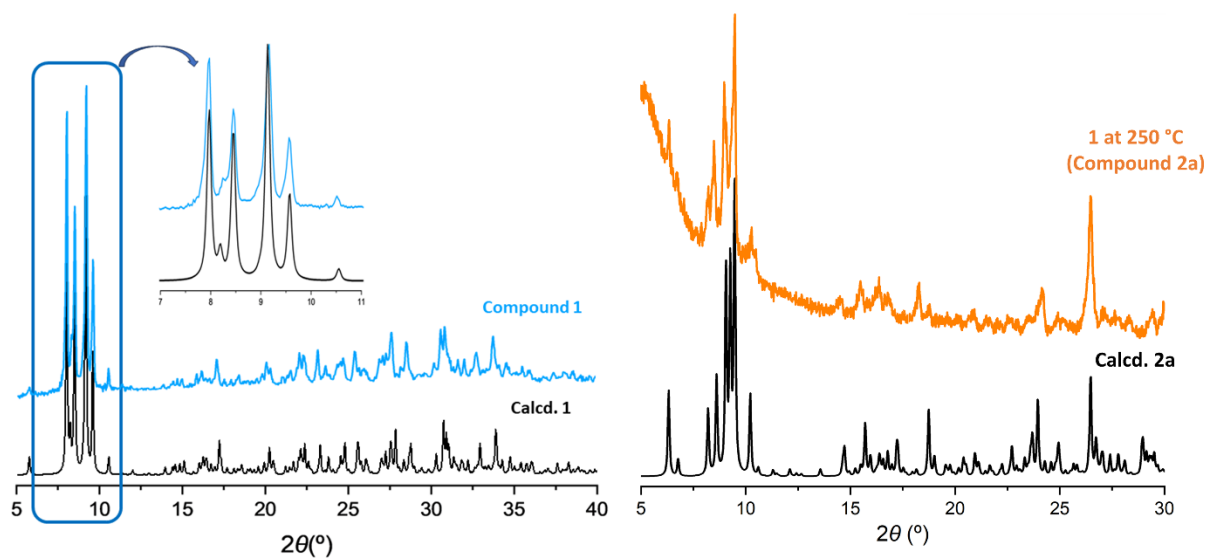
**Figure S1.** FTIR spectrum of **1** (blue) compared to that of the  $\text{K}_7[\alpha\text{-PW}_{11}\text{O}_{39}] \cdot 14\text{H}_2\text{O}$  precursor.



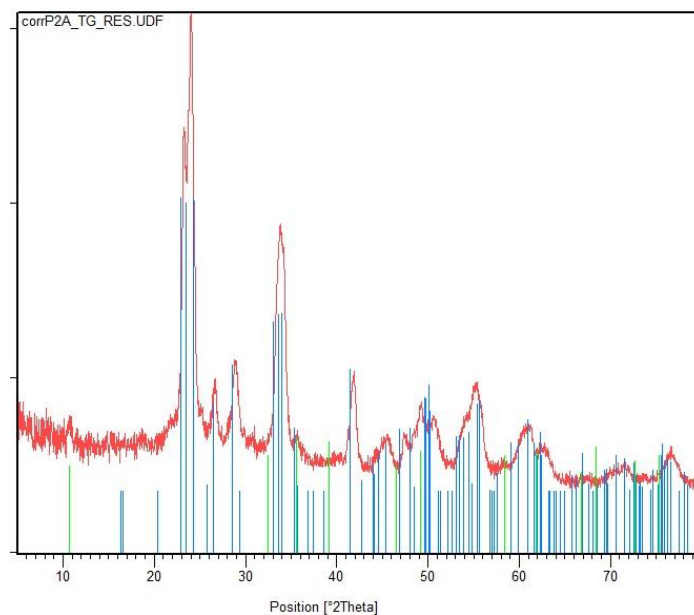
**Figure S2.** PXRD diffraction patterns for the syntheses carried out for different POM: $\text{Cu}^{2+}$ :pic ratios.



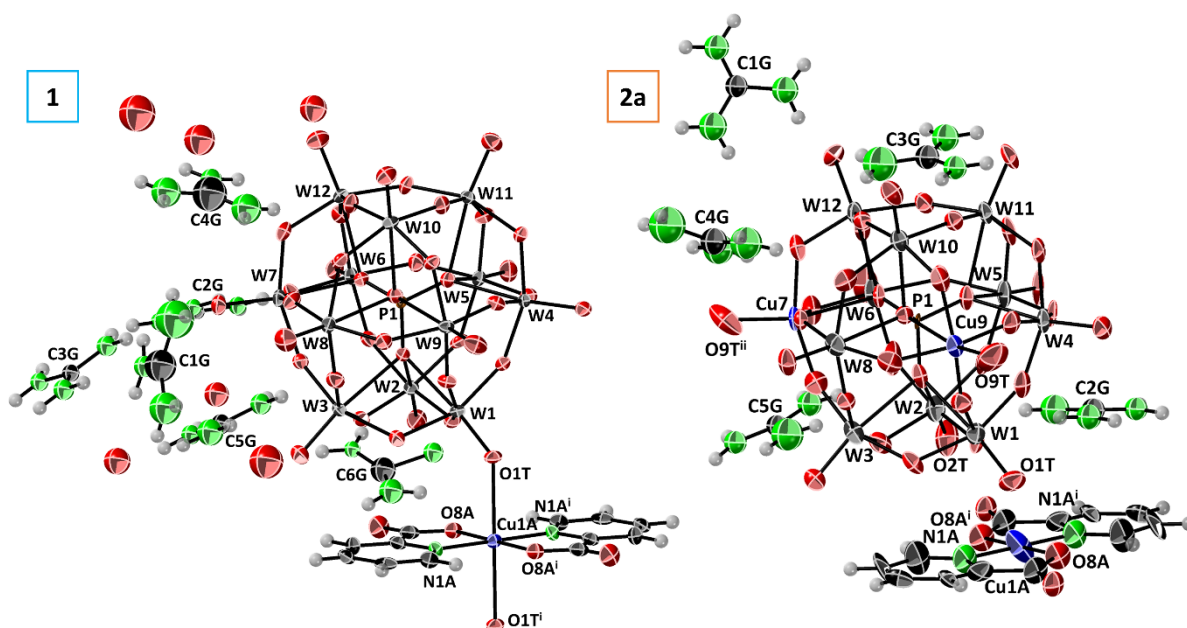
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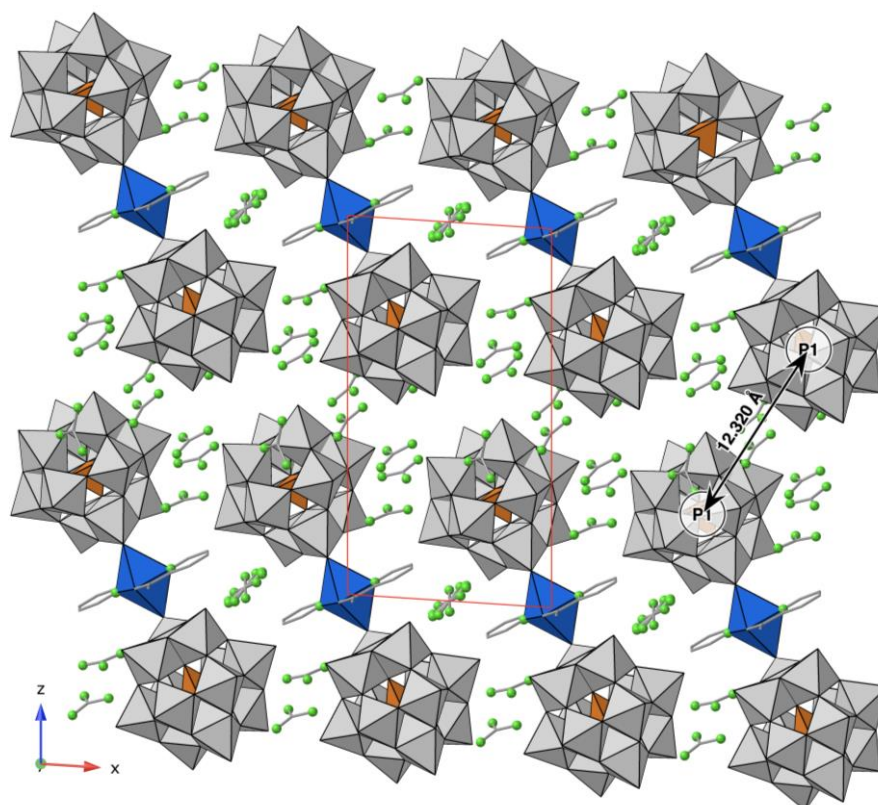
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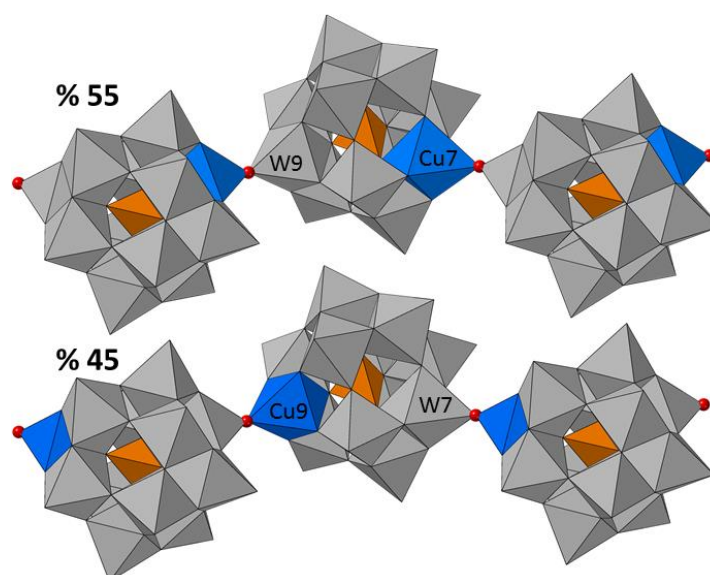
**Figure S5.** Identification of the final residue from the thermal decomposition of **1** by PXRD analyses. Blue lines correspond to diffraction maxima from monoclinic *Pmb* WO<sub>3</sub> (PDF: 01-071-0131) and green lines to monoclinic *C2/c* CuO (PDF: 00-002-1041).



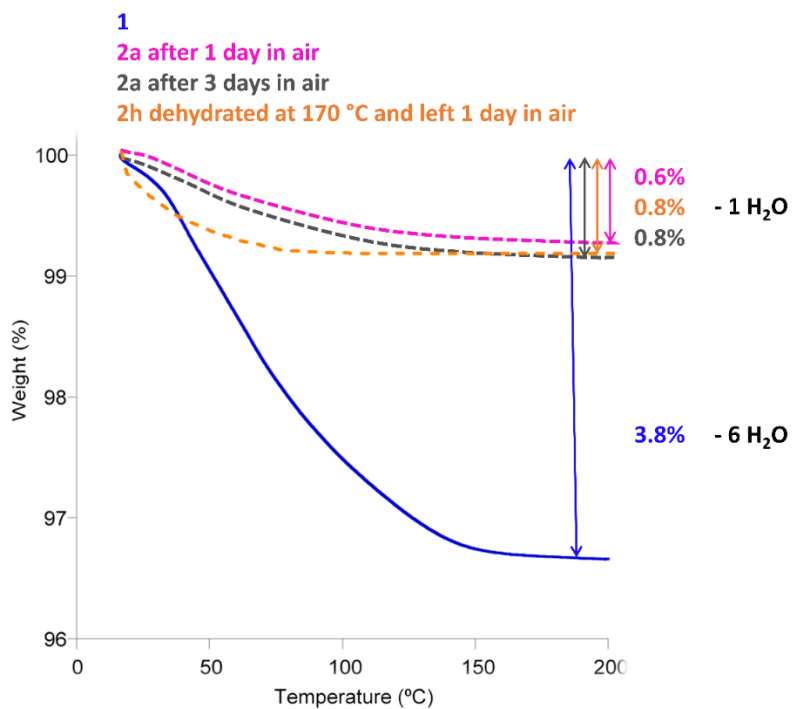
**Figure S6.** ORTEP representation of **1** and **2a** showing 50% probability ellipsoids and atom labelling scheme. Colour code: Mo, grey; P, dark red; Cu, blue; O, red; N, green; C, black; H, light grey. The disordered Cu<sup>II</sup> atom in the Keggin anion has been omitted for **1**, but has been depicted on the addenda metal positions 7 and 9 for **2a**. Symmetry codes: i)  $-x, -y, 1-z$ ; ii)  $1-x, \frac{1}{2}+y, \frac{1}{2}-z$ .



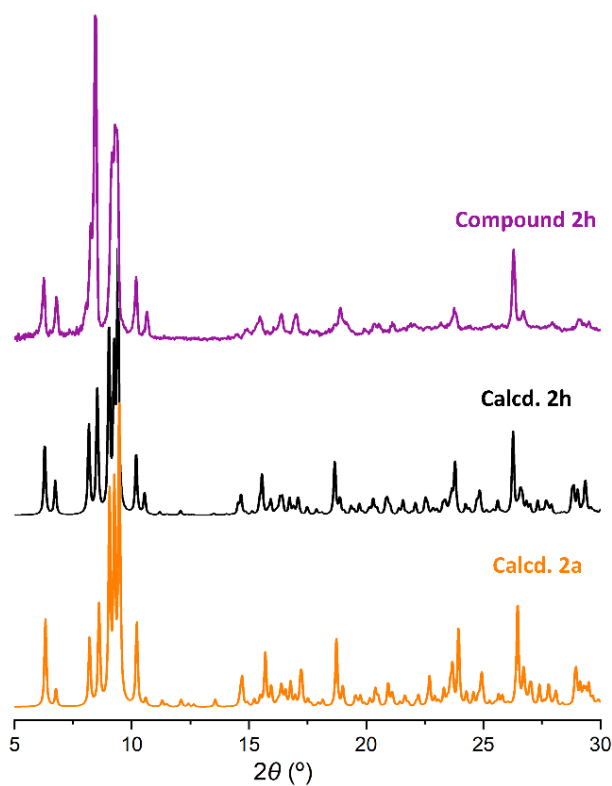
**Figure S7.** View of the crystal packing of **1** along the crystallographic y axis.



**Figure S8.** Polyhedral representation of  $\{PW_{11}O_{39}Cu\}_n$  chains in **2a**.

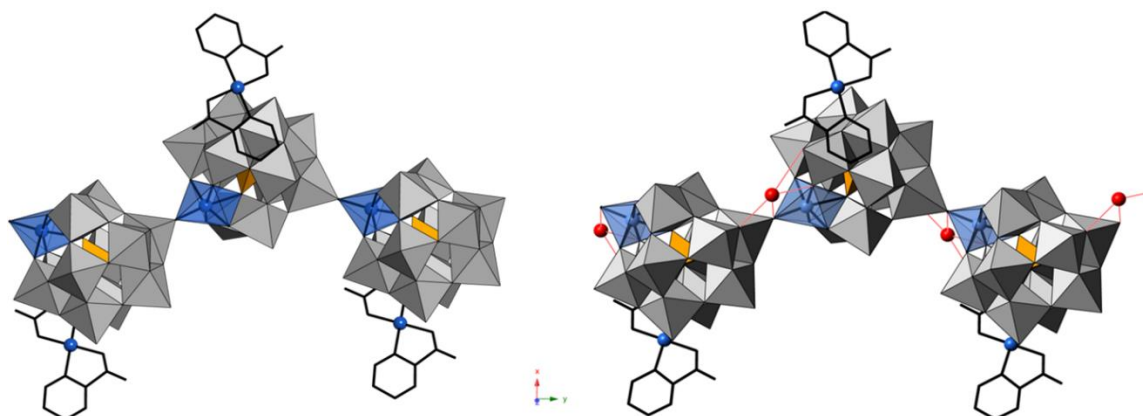


**Figure S9.** TGA curve for the dehydration of **1** (blue) compared to those of samples of **2a** left to hydrate in open air for 1 (pink) or 3 days (grey), as well as that of **2h** dehydrated at 170 °C and exposed to ambient moisture for 1 day (orange).

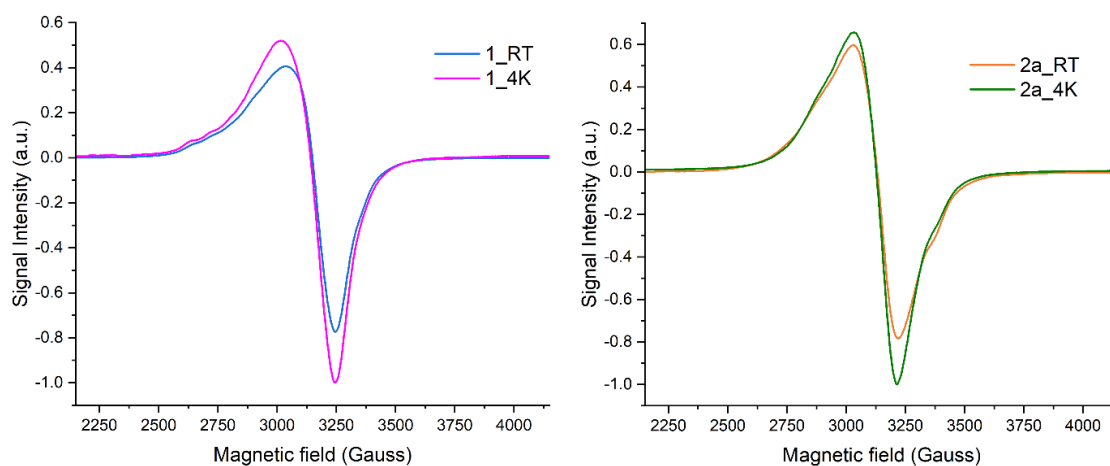


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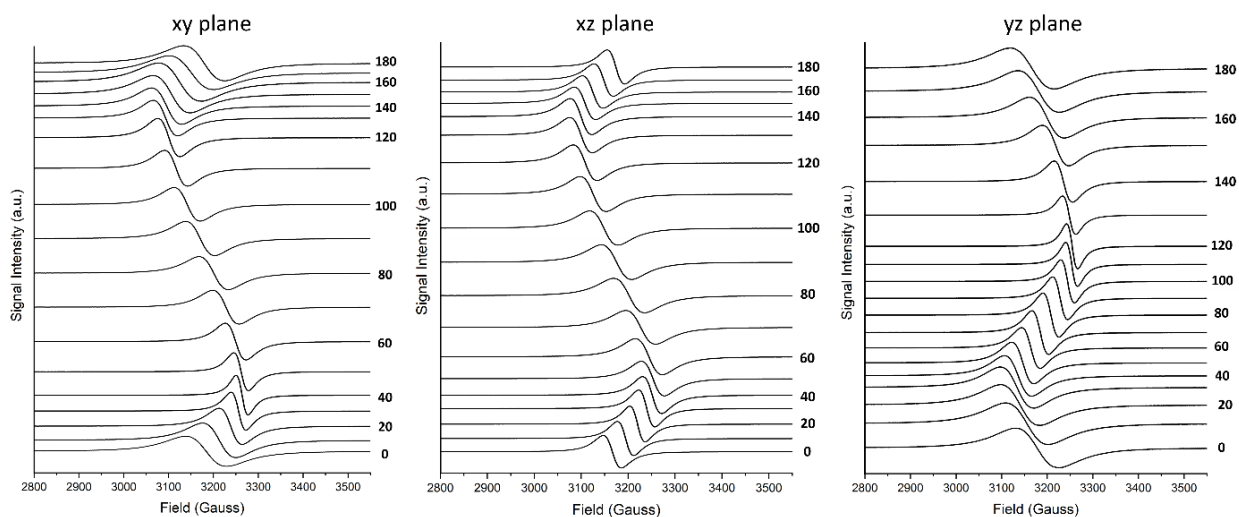




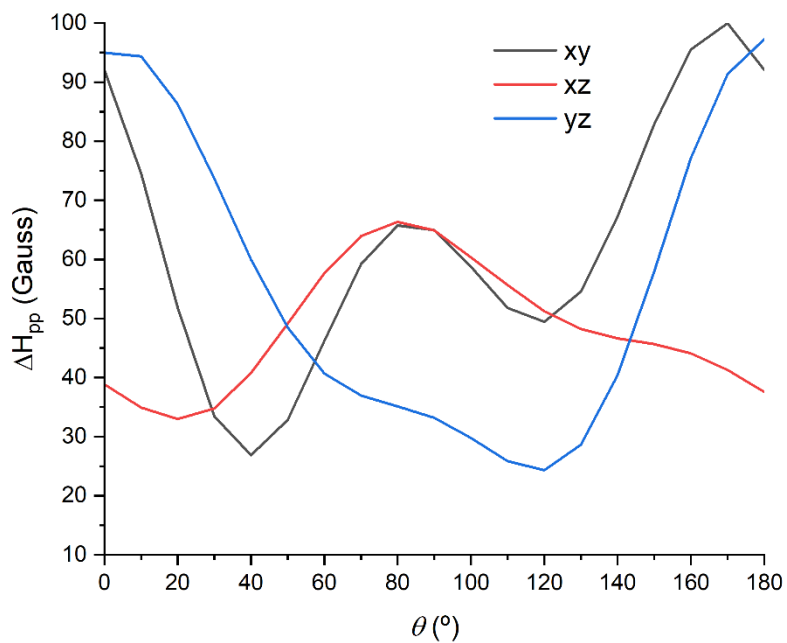
**Figure S11.** Structural comparison of the polymeric entities in **2a** (left) and **2h** (right). Hydrogen-bonding interactions involving water molecules are depicted as red lines.



**Figure S12.** X-band EPR spectra at room temperature and 4K for **1** (left) and **2a** (right).



**Figure S13.** Evolution of the X-band EPR single-crystal spectra of **1** for the three perpendicular xy, xz and yz planes. The spectra were recorded for different angles every 10°.



**Figure S14.** Angular variation of the linewidth in X-band EPR single-crystal spectra of **1** for the xy, xz and yz planes.

**Equation 1.** Set of equations employed to fit the angular variation of the EPR signal in three mutually perpendicular planes. The principal g values were calculated by diagonalization of the obtained  $g^2$  tensor.

$$g^2_{(xy)} = g_{xx}^2 \cos^2(\theta) + 2 g_{xy} \cos(\theta) \sin(\theta) + g_{yy}^2 \sin^2(\theta)$$

$$g^2_{(xz)} = g_{xx}^2 \cos^2(\theta) + 2 g_{xz} \cos(\theta) \sin(\theta) + g_{zz}^2 \sin^2(\theta)$$

$$g^2_{(yz)} = g_{yy}^2 \cos^2(\theta) + 2 g_{yz} \cos(\theta) \sin(\theta) + g_{zz}^2 \sin^2(\theta)$$

**Table S1.** Copper population factors for the addenda metal positions within the inorganic  $[PW_{11}O_{39}Cu(H_2O)]^{5-}$  building-block in **1**, **2a** and **2h**.

	W1	W2	W3	W4	W5	W6	W7	W8	W9	W10	W11	W12
<b>1</b>	–	4.8	7.9	9.0	7.9	3.2	10.6	18.6	18.5	4.2	6.9	8.4
<b>2a</b>	–	–	–	–	–	–	44.9	–	55.1	–	–	–
<b>2h</b>	–	–	–	–	–	–	45.0	–	55.0	–	–	–

**Table S2.** Cu–O and Cu–N bond lengths (Å) and O–Cu–N and O–Cu–O angles (°) in the metalorganic units of **1**, **2a** and **2h**.

<b>1</b>		<b>2a</b>		<b>2h</b>	
Cu1A–O1T	2.507(5)	Cu1A···O2T	3.79(3)	Cu1A–O2T	3.99(3)
Cu1A–O1T <sup>i</sup>	2.507(5)	Cu1A···O2T <sup>i</sup>	3.79(3)	Cu1A–O2T <sup>i</sup>	3.99(3)
Cu1A–O8A	1.945(5)	Cu1A–O8A	1.91(2)	Cu1A–O8A	1.89(3)
Cu1A–O8A <sup>i</sup>	1.945(5)	Cu1A–O8A <sup>i</sup>	1.91(2)	Cu1A–O8A <sup>i</sup>	1.89(3)
Cu1A–N1A	1.951(7)	Cu1A–N1A	1.99(4)	Cu1A–N1A	1.87(4)
Cu1A–N1A <sup>i</sup>	1.951(7)	Cu1A–N1A <sup>i</sup>	1.99(4)	Cu1A–N1A <sup>i</sup>	1.87(4)

Symmetry code: *i*) 1 -x, 1-y, 1-z.

**Table S3.** Geometrical parameters of the intermolecular N–H···O and C–H···O hydrogen bonds in **1**, **2a** and **2h**.

Donor–H···Acceptor	D–H	H···A	D···A	D–H···A
<b>1</b>				
N1G–H1GA···O3W <sup>i</sup>	0.88	1.93	2.743(18)	153
N1G–H1GB···O1W	0.88	2.03	2.889(17)	167
N2G–H2GA···O910 <sup>ii</sup>	0.88	2.42	3.127(11)	138
N2G–H2GB···O2W <sup>iii</sup>	0.88	2.19	3.065(14)	174
N3G–H3GB···O2T	0.88	2.48	3.278(18)	152
N4G–H4GA···O910 <sup>iv</sup>	0.88	2.05	2.920(9)	170
N4G–H4GB···O67	0.88	2.18	3.030(91)	163
N5G–H5GA···O12 <sup>ii</sup>	0.88	2.49	3.326(9)	159
N5G–H5GB···O810 <sup>iv</sup>	0.88	2.16	3.030(9)	168
N10G–H10B···O712 <sup>v</sup>	0.88	2.24	2.926(16)	135
N10G–H10A···O5W	0.88	2.14	2.85(2)	137
N6G–H6GA···O14 <sup>ii</sup>	0.88	2.17	2.953(9)	148
N6G–H6GB···O6T	0.88	2.25	3.043(9)	150
N7G–H7GA···O9A	0.88	2.36	3.024(10)	133
N13G–H13A···O4T <sup>ii</sup>	0.88	2.19	3.040(10)	163
N13G–H13B···O37	0.88	2.29	3.096(11)	151
N14G–H14A···O45 <sup>ii</sup>	0.88	2.05	2.912(10)	167
N14G–H14B···O11T <sup>i</sup>	0.88	2.35	3.077(11)	140
N8G–H8GB···O9A	0.88	2.08	2.820(9)	141
N15G–H15A···O2W <sup>ii</sup>	0.88	2.15	3.011(14)	165
N15G–H15B···O3T	0.88	2.45	3.194(11)	143
N8G–H8GB···O56 <sup>i</sup>	0.88	2.49	3.103(11)	127
N9G–H9GA···O411 <sup>ii</sup>	0.88	2.04	2.906(8)	168
N9G–H9GB···O511 <sup>i</sup>	0.88	2.12	2.961(8)	160
C3A–H3A···O89 <sup>vi</sup>	0.95	2.28	3.212(11)	166
C5A–H5A···O612 <sup>i</sup>	0.95	2.33	3.043(11)	131

**Table S3 (continuation).** Geometrical parameters of the intermolecular N–H...O and C–H...O type hydrogen bonds in **1**, **2a** and **2h**.

<b>2a</b>				
N1G–H1GA...O9A <sup>vii</sup>	0.88	2.12	2.87(3)	143
N2G–H2GA...O910 <sup>viii</sup>	0.88	2.07	2.89(3)	155
N2G–H2GB...O12T	0.88	2.07	2.85(3)	148
N3G–H3GA...O9A <sup>vii</sup>	0.88	2.11	2.87(3)	144
N3G–H3GB...O49 <sup>viii</sup>	0.88	2.46	3.17(3)	138
N4G–H4GA...O67 <sup>ix</sup>	0.88	2.12	2.95(3)	159
N4G–H4GB...O89 <sup>x</sup>	0.88	2.08	2.95(3)	168
N5G–H5GA...O8T <sup>x</sup>	0.88	1.88	2.75(3)	172
N5G–H5GB...O2T	0.88	2.31	3.11(4)	151
N5G–H5GB...O25	0.88	2.25	2.93(3)	134
N10G–H10A...O78	0.88	2.33	3.07(3)	142
N6G–H6GA...O6T <sup>ix</sup>	0.88	1.93	2.80(3)	171
N11G–H11A...O6T <sup>xi</sup>	0.87	2.51	3.13(4)	129
N11G–H11B...O1T <sup>vii</sup>	0.88	2.26	3.05(4)	149
N6G–H6GB...O14	0.89	2.58	3.27(3)	135
N12G–H12A...O810	0.88	2.15	3.01(4)	168
N12G–H12B...O5T <sup>xi</sup>	0.88	2.21	2.96(4)	144
N7G–H7GA...O9A <sup>xii</sup>	0.88	1.97	2.81(3)	159
N13G–H13B...O45 <sup>xii</sup>	0.88	2.05	2.91(3)	165
N7G–H7GB...O411 <sup>xiii</sup>	0.88	1.96	2.80(3)	161
N14G–H14A...O10T <sup>vii</sup>	0.88	2.29	3.09(3)	152
N14G–H14B...O5T <sup>xii</sup>	0.88	2.03	2.90(3)	171
N8G–H8GA...O8A <sup>xii</sup>	0.87	2.36	3.20(4)	161
N15G–H15A...O23	0.88	2.19	2.98(4)	150
N15G–H15B...O12T <sup>i</sup>	0.88	2.18	2.91(4)	139
N8G–H8GB...O612	0.88	2.13	2.95(4)	154
<b>2h</b>				
N1G–H1GA...O9 <sup>xiv</sup>	0.88	1.99	2.78(5)	149
N1G–H1GB...O1W <sup>xiv</sup>	0.88	1.91	2.70(6)	150
N2–H2A...O45 <sup>xii</sup>	0.88	2.05	2.92(4)	172
N2G–H2GA...O12T <sup>viii</sup>	0.87	2.22	2.98(5)	147
N3–H3A...O10T <sup>vii</sup>	0.87	2.27	3.11(3)	161
N3–H3B...O5T <sup>xii</sup>	0.88	2.21	2.97(4)	160
N2G–H2GB...O910	0.88	2.12	2.94(4)	154
N3G–H3GA...O9 <sup>xiv</sup>	0.88	2.35	3.04(5)	136
N3G–H3GB...O49	0.88	2.32	3.06(4)	141
N4G–H4GA...O89 <sup>xv</sup>	0.88	2.08	2.94(3)	169
N4G–H4GB...O67 <sup>ix</sup>	0.88	2.14	2.98(3)	160
N5G–H5GA...O2T	0.88	2.47	3.25(5)	150
N5G–H5GA...O25	0.88	2.25	2.95(4)	137
N5G–H5GB...O8T <sup>xv</sup>	0.88	1.89	2.77(4)	172
N10G–H10A...O8T <sup>xv</sup>	0.87	2.36	3.05(10)	136
N10G–H10B...O4T <sup>xii</sup>	0.89	2.50	3.23(9)	139
N6G–H6GA...O14	0.88	2.59	3.26(4)	134
N11G–H11A...O11T <sup>xiii</sup>	0.87	2.12	2.98(8)	170
N6G–H6GB...O6T <sup>ix</sup>	0.88	2.01	2.87(4)	168
N7G–H7GA...O411 <sup>xiii</sup>	0.88	2.01	2.84(4)	156
N13G–H13A...O4T <sup>xii</sup>	0.88	2.31	3.11(8)	151
N7G–H7GA...O9 <sup>xii</sup>	0.88	1.99	2.83(4)	160
N14G–H14B...O6T	0.87	2.38	3.13(8)	145
N8G–H8GA...O612	0.88	2.15	3.00(5)	163
N15G–H15A...O12T <sup>i</sup>	0.88	2.36	3.01(5)	132
N15G–H15B...O23	0.88	2.33	3.08(4)	143
N8G–H8GB...O8 <sup>xii</sup>	0.87	2.34	3.20(5)	167

Symmetry codes: i)  $x, \frac{1}{2}-y, \frac{1}{2}+z$ ; ii)  $1-x, -\frac{1}{2}+y, \frac{1}{2}-z$ ; iii)  $-x, -1/2+y, \frac{1}{2}-z$ ; iv)  $2-x, -1/2+y, \frac{1}{2}-z$ ; v)  $-1+x, y, z$ ; vi)  $2-x, 1-y, 1-z$ ; vii)  $1-x, \frac{1}{2}+y, \frac{1}{2}-z$ ; viii)  $1-x, -y, -z$ ; ix)  $-x, -1/2+y, \frac{1}{2}-z$ ; x)  $-1+x, y, z$ ; xi)  $1+x, y, z$ ; xii)  $-x, \frac{1}{2}+y, \frac{1}{2}-z$ ; xiii)  $-x, -y, -z$ ; xiv)  $x, -1/2-y, -1/2+z$ ; xv)  $1+x, y, z$ .

**Table S4** Distances (Å) from the O atoms of W1-W2-W3 trimers to picolinate ring centroids in **2a** and **2h** ( $C_{g_{pic}}$  = N1A, C2A, C3A, C4A, C5A, C6A).

	<b>2a</b>	<b>2h</b>
O1T... $C_{g_{pic}}$	3.70(2)	3.63(3)
O2T... $C_{g_{pic}}$	4.97(2)	5.16(3)
O3T... $C_{g_{pic}}$	4.13(3)	4.17(2)
O12... $C_{g_{pic}}$	3.79(2)	3.87(2)
O13... $C_{g_{pic}}$	3.23(2)	3.26(2)
O23... $C_{g_{pic}}$	3.98(2)	4.13(2)

**Table S5.** Distances (Å) from the O atoms of W1-W2-W3 trimers to picolinate ring planes (N1A, C2A, C3A, C4A, C5A, C6A) in **2a** and **2h**

	<b>2a</b>	<b>2h</b>
O1T...plane	2.724	2.864
O2T...plane	2.676	2.728
O3T...plane	2.613	2.475
O12...plane	3.212	3.357
O13...plane	3.159	3.170
O23...plane	3.109	3.100