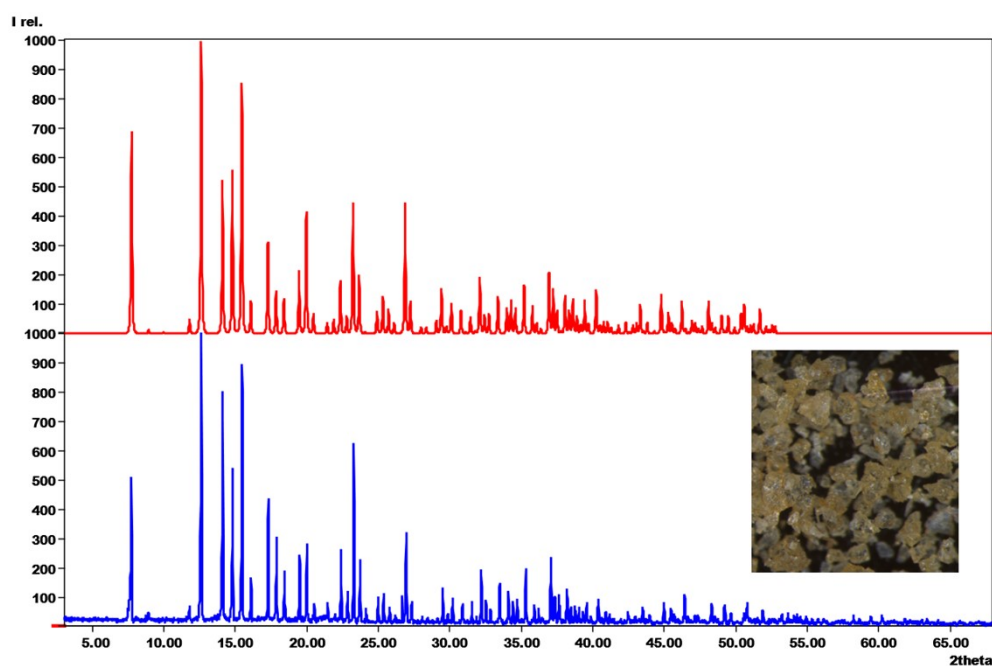


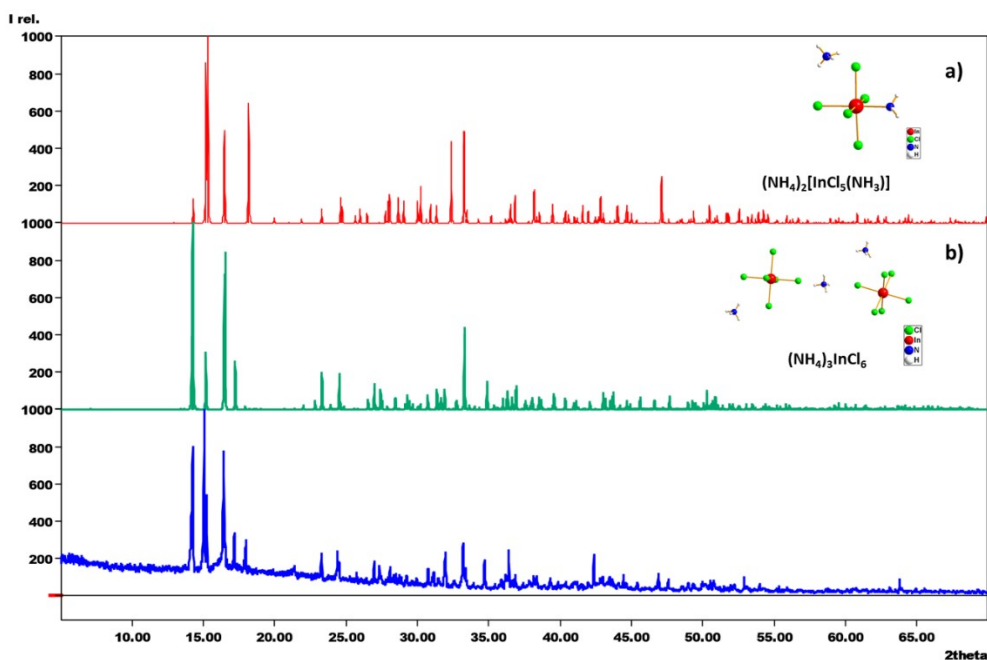
## Thermal Deprotonation and Condensation of Melamine in the Presence of Indium(III)chloride

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### Supporting Information



**Figure SI 1.** XRD pattern of the synthesized  $(\text{NH}_4)[(\text{InCl}_2)_3(\text{C}_{12}\text{N}_{20}\text{H}_8)] \cdot \frac{2}{3}[\text{InCl}_3(\text{NH}_3)]$  (bottom), Calculated pattern from single crystal measurement (top) (CCDC code: 2333063).



**Figure S1.2.** XRD pattern of the side-phases sublimated on the top part of ampule synthesized along with the main phase of  $(\text{NH}_4)[(\text{InCl}_2)_3(\text{C}_{12}\text{N}_{20}\text{H}_8)] \cdot \frac{2}{3}[\text{InCl}_3(\text{NH}_3)]$  (bottom). **(a)** Calculated pattern from single crystal measurement of  $(\text{NH}_4)_2[\text{InCl}_5(\text{NH}_3)]$  (CCDC code: 2301094) **(b)** Calculated pattern from single crystal measurement of  $(\text{NH}_4)_3\text{InCl}_6$  (CCDC code: 2334831).

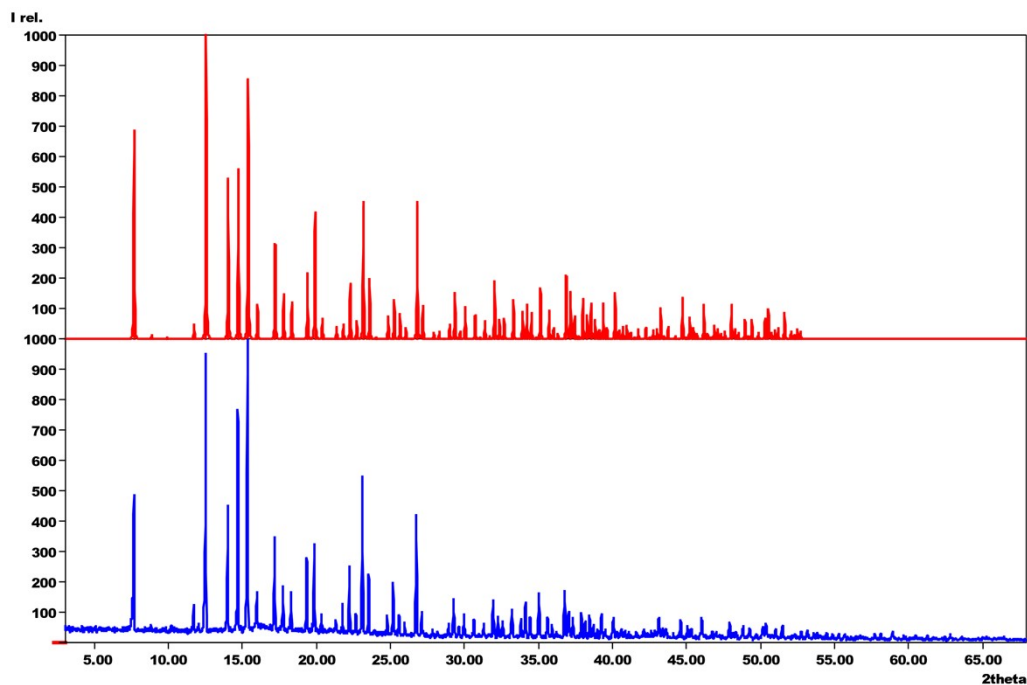
**Table S1.** Crystal structure data of  $(\text{NH}_4)_2[\text{InCl}_5(\text{NH}_3)]$ , and  $(\text{NH}_4)_3\text{InCl}_6$  as side phases of the reaction.

Empirical Formula	$(\text{NH}_4)_2[\text{InCl}_5(\text{NH}_3)]$	$(\text{NH}_4)_3\text{InCl}_6$
CCDC code	2301094	2334831
Formula weight (g/mol)	345.19	381.65
Wavelength (Cu-K $\alpha$ ) (Å)	1.54184	1.54184
Crystal system	orthorhombic	monoclinic
Space group	<i>Pnma</i>	<i>P2<sub>1</sub>/c</i>
Unit cell dimensions (Å)	<i>a</i> /Å 13.7070(2) <i>b</i> /Å 10.76210(10) <i>c</i> /Å 6.93960(10)	<i>a</i> /Å 13.1270(3) <i>b</i> /Å 7.78800(10) <i>c</i> /Å 12.2894(2)
Volume (Å <sup>3</sup> )	1023.70(2)	1192.96(4)
Z	4	4
Density (calculated) (g/cm <sup>3</sup> )	2.240	2.125
Absorption coefficient (mm <sup>-1</sup> )	29.980	27.819
Final R indices ( $I > 2\sigma(I)$ ) a)	$R_1 = 0.0133$ , $wR_2 = 0.0364$	$R_1 = 0.0148$ , $wR_2 = 0.0371$

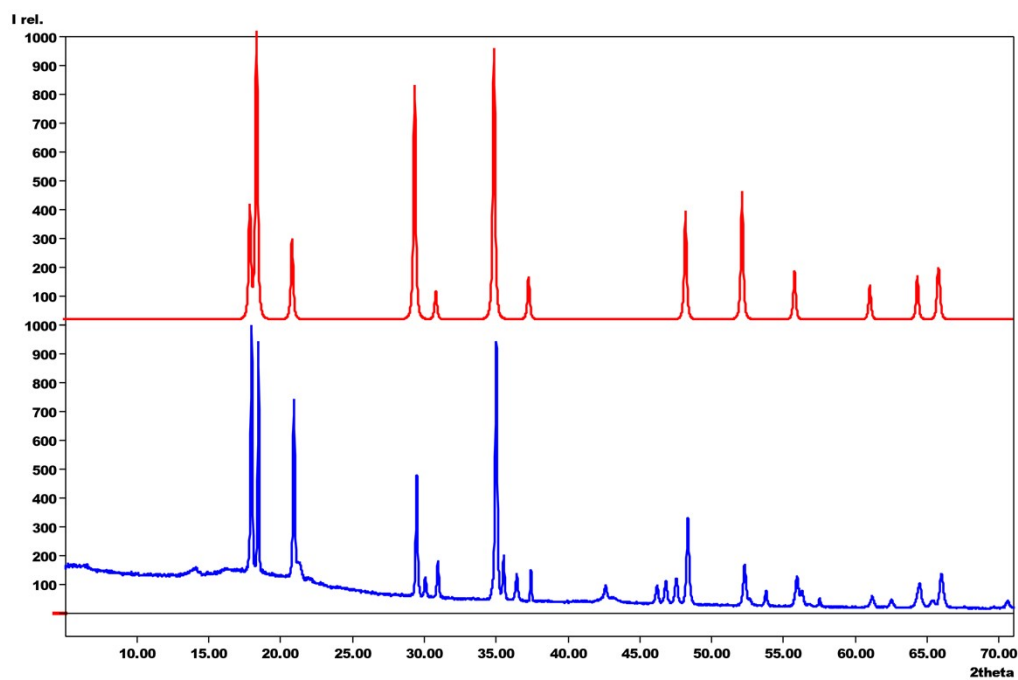
R indices (all data)	$R_1 = 0.0133$ , $wR_2 = 0.0364$	$R_1 = 0.0149$ , $wR_2 = 0.0372$
GOOF	1.264	1.211

**Table S2.** Vibrational frequencies (in  $\text{cm}^{-1}$ ) for In-Ring compared to those of Melamine, Melem, and  $(\text{NH}_4)[(\text{InCl}_2)_3(\text{C}_{12}\text{N}_{20}\text{H}_8)] \cdot \frac{2}{3}[\text{InCl}_3(\text{NH}_3)]$

	Melamine	Melem	$(\text{NH}_4)[(\text{InCl}_2)_3(\text{C}_{12}\text{N}_{20}\text{H}_8)] \cdot \frac{2}{3}[\text{InCl}_3(\text{NH}_3)]$
Ring-sextant out-of-plane bending	813	804	792
CNC bending	1193	1306	1284
Side-chain CN breathing	1434	1470	1380
	1440		1434
	1550		1504
$\text{NH}_2$ bending	1652	1612	1647
NH stretching	3128	3119	3200
	3334	3325	3348
	3421	3424	3452
	3469	3487	



**Figure SI 3.** XRD pattern of the synthesized  $(\text{NH}_4)[(\text{InCl}_2)_3(\text{C}_{12}\text{N}_{20}\text{H}_8)] \cdot \frac{2}{3}[\text{InCl}_3(\text{NH}_3)]$  after two months in ambient condition (bottom), Calculated pattern from single crystal measurement (top) (CCDC code: 2333063).



**Figure SI 4.** The XRD pattern taken from decomposed material at  $700^\circ\text{C}$  in TGA analysis(bottom), Calculated pattern from single crystal measurement of  $\text{In}_{2.24}(\text{NCN})_3$ (top)