

Supporting information

Multi-layered Heterogeneous Interfaces Created in Co_{0.85}Se@Ni₃S₄/NF to Enhance Supercapacitor Performances by Multi-step Alternating Electrodeposition

Chunyan Zhang,^a Jinkun Yang,^a Hang Li,^a Mengfei Su,^a Boru Xiong,^b Feng Gao,^b and Qingyi Lu^a

^a State Key Laboratory of Coordination Chemistry, Coordination Chemistry Institute, Collaborative Innovation Center of Advanced Microstructures, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210023, P. R. China. E-mail: qylu@nju.edu.cn

^b Department of Materials Science and Engineering, Jiangsu Key Laboratory of Artificial Functional Materials, Collaborative Innovation Center of Advanced Microstructures, College of Engineering and Applied Sciences, Nanjing University, Nanjing 210023, P. R. China. E-mail: fgao@nju.edu.cn

Synthesis of PPy

Polypyrrole (PPy) was synthesized via an electrodeposition process by galvanostatic method carried out on a CHI660E electrochemical workstation with carbon cloth as the working electrode, an Ag/AgCl, KCl (saturated) electrode as the reference electrode and a Pt sheet (1.2 cm×1.2 cm) as the counter electrode. The typical deposition was performed in a mixed solution of 0.01 mmol SDBS (sodium dodecyl benzene sulfonate) and 0.15 mmol pyrrole monomer by galvanostatic method with a potential of 1 V for 800 seconds.

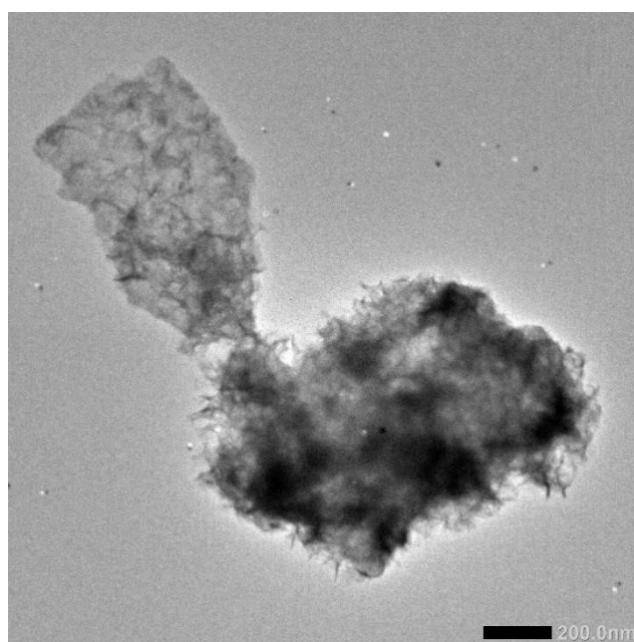


Fig. S1 The TEM image of 8L-Co_{0.85}Se@Ni₃S₄@NF

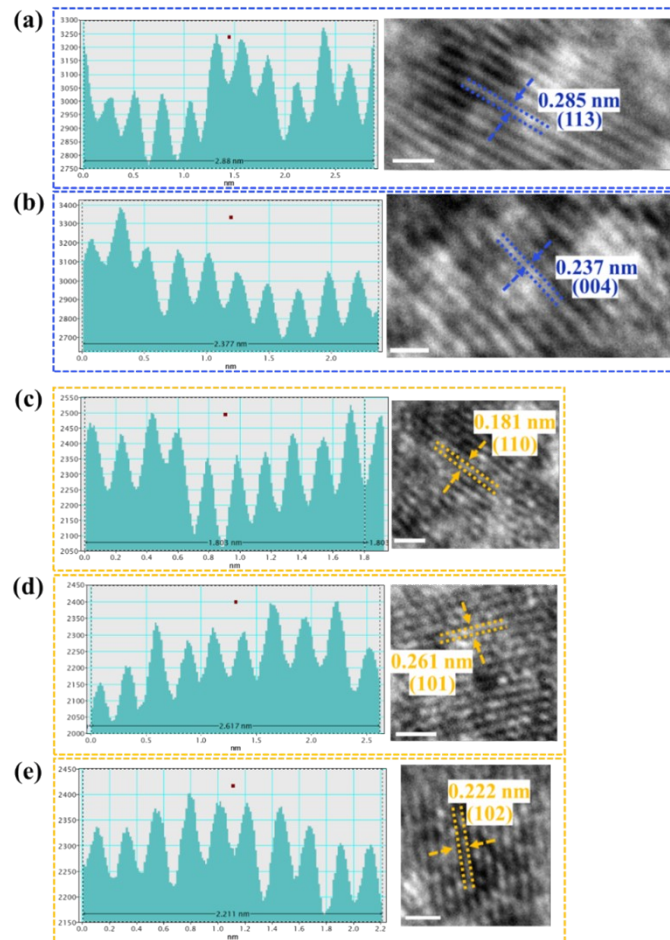


Fig. S2 HRTEM and corresponding IFFT images for Ni_3S_4 and $\text{Co}_{0.85}\text{Se}$ in Fig. 3b: (a) (113) and (b) (004) planes of Ni_3S_4 ; (c) (110), (d) (101) and (e) (102) planes of $\text{Co}_{0.85}\text{Se}$. The scale bar is 0.5 nm.

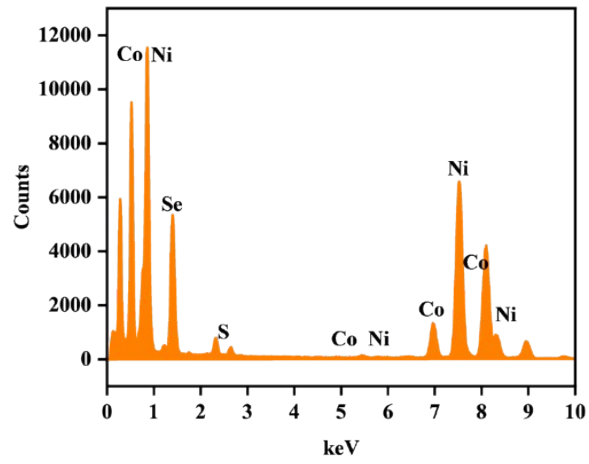


Fig. S3 EDS spectrum of 8L-Co_{0.85}Se@Ni₃S₄/NF

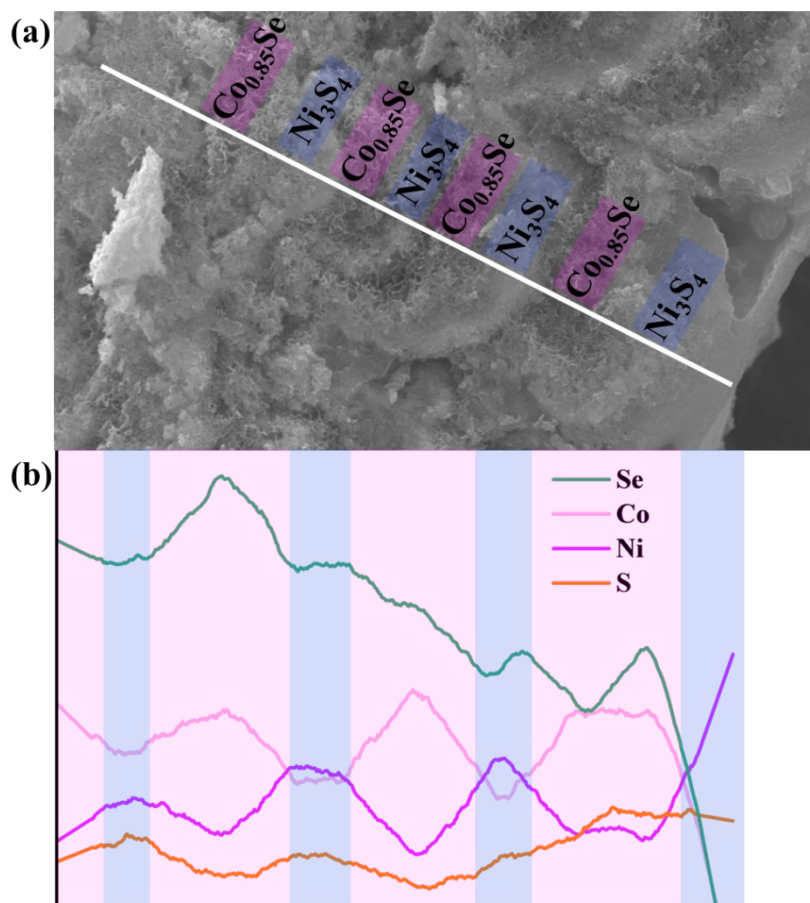


Fig. S4 SEM image and cross-section linear scanning EDS spectrum of 8L-Co_{0.85}Se@Ni₃S₄/NF.

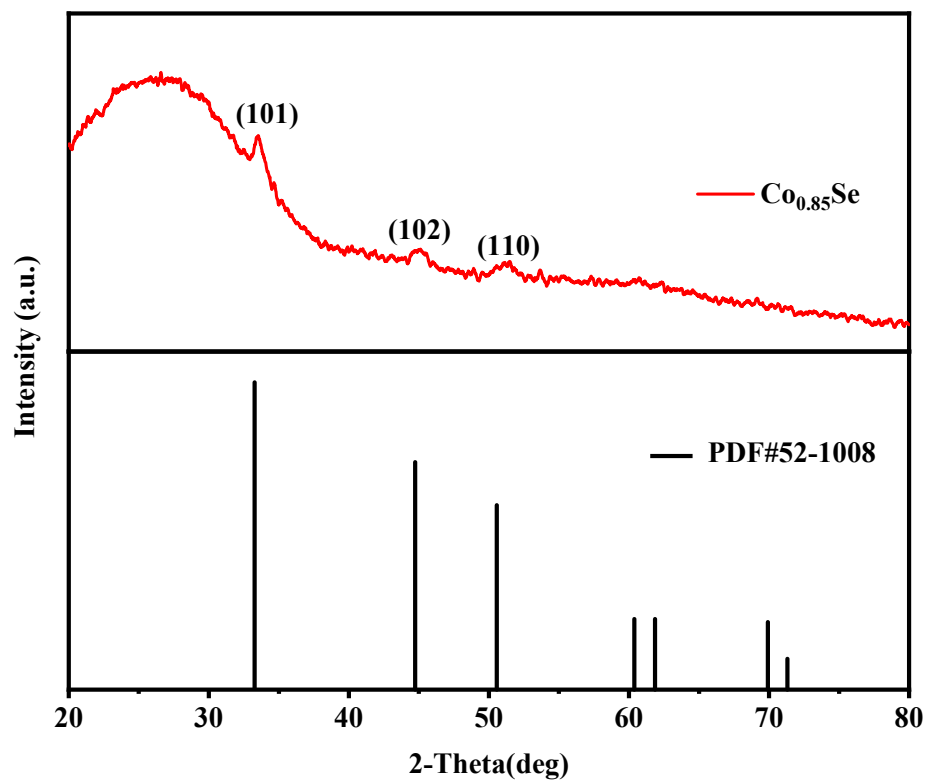


Fig. S5 XRD pattern of 1L-Co_{0.85}Se

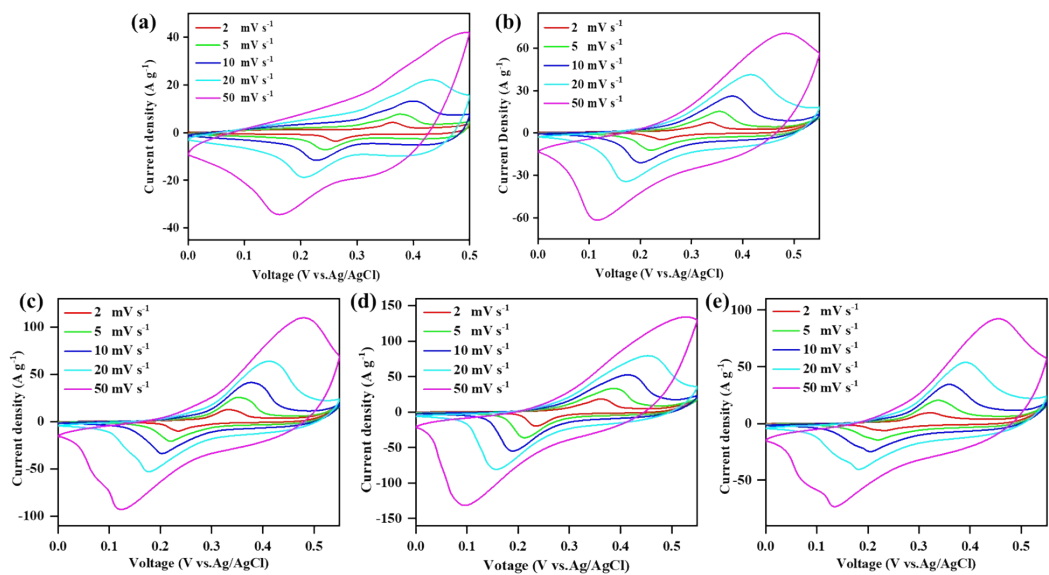


Fig. S6 CV curves of (a) 1L-Co_{0.85}Se@NF, (b) 2L-Co_{0.85}Se@Ni₃S₄/NF, (c) 4L-Co_{0.85}Se@Ni₃S₄/NF, (d) 6L-Co_{0.85}Se@Ni₃S₄/NF and (e) CoSeNiS/NF electrodes at scan rates from 2 to 50 mV s⁻¹.

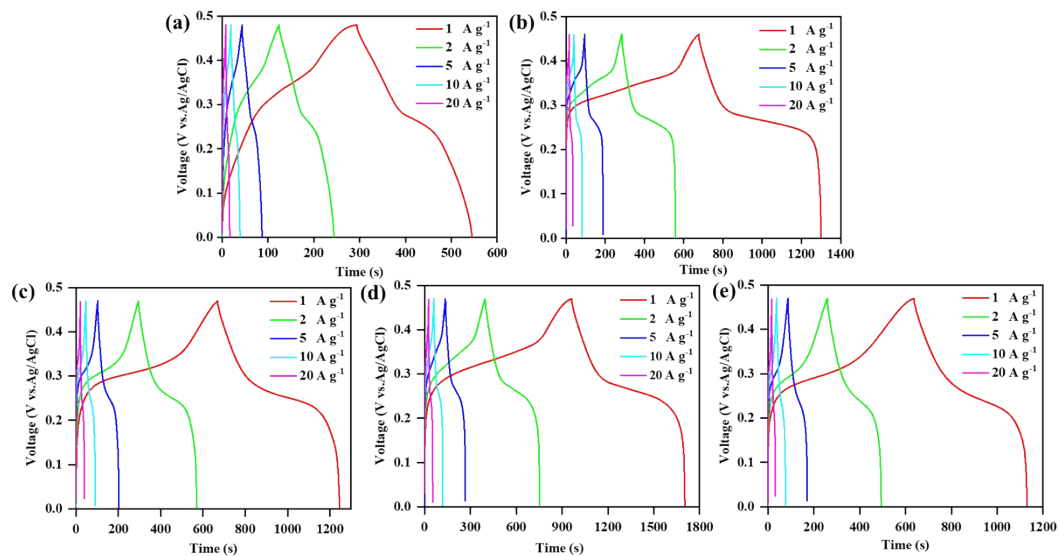


Fig. S7 GCD curves of (a) 1L-Co_{0.85}Se@NF, (b) 2L-Co_{0.85}Se@Ni₃S₄/NF, (c) 4L-Co_{0.85}Se@Ni₃S₄/NF, (d) 6L-Co_{0.85}Se@Ni₃S₄/NF and (e) CoSeNiS/NF electrodes at different current densities of 1 A g⁻¹, 2 A g⁻¹, 5 A g⁻¹, 10 A g⁻¹ and 20 A g⁻¹.

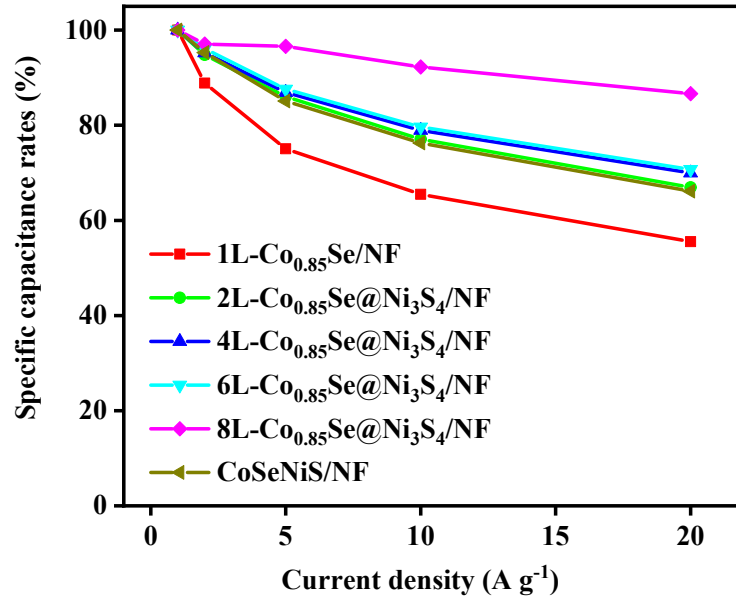


Fig. S8 Capacitance retentions rates of different electrodes at different current densities.

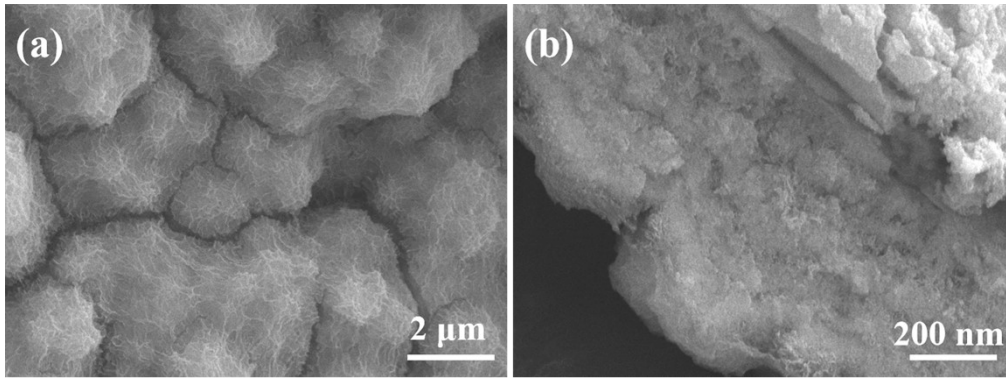


Fig. S9 SEM images of 8L-Co_{0.85}Se@Ni₃S₄@NF after cycling 5000 cycles.

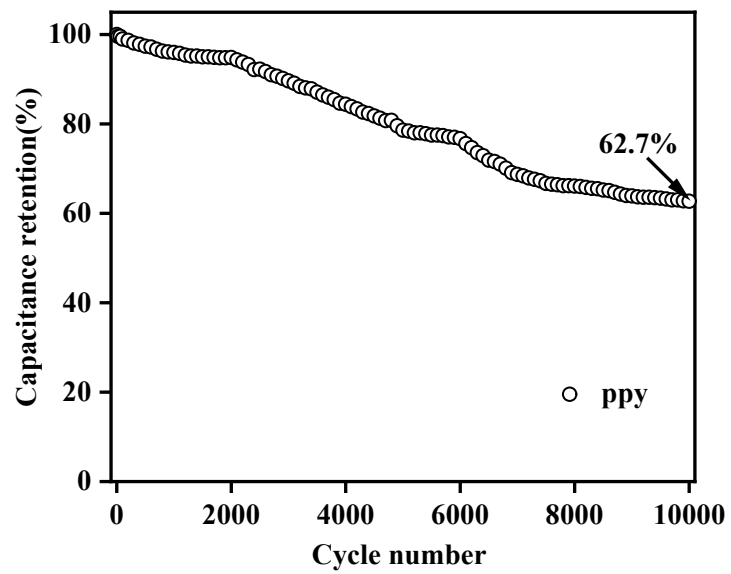


Fig. S10 Cycle retention test of ppy/NF

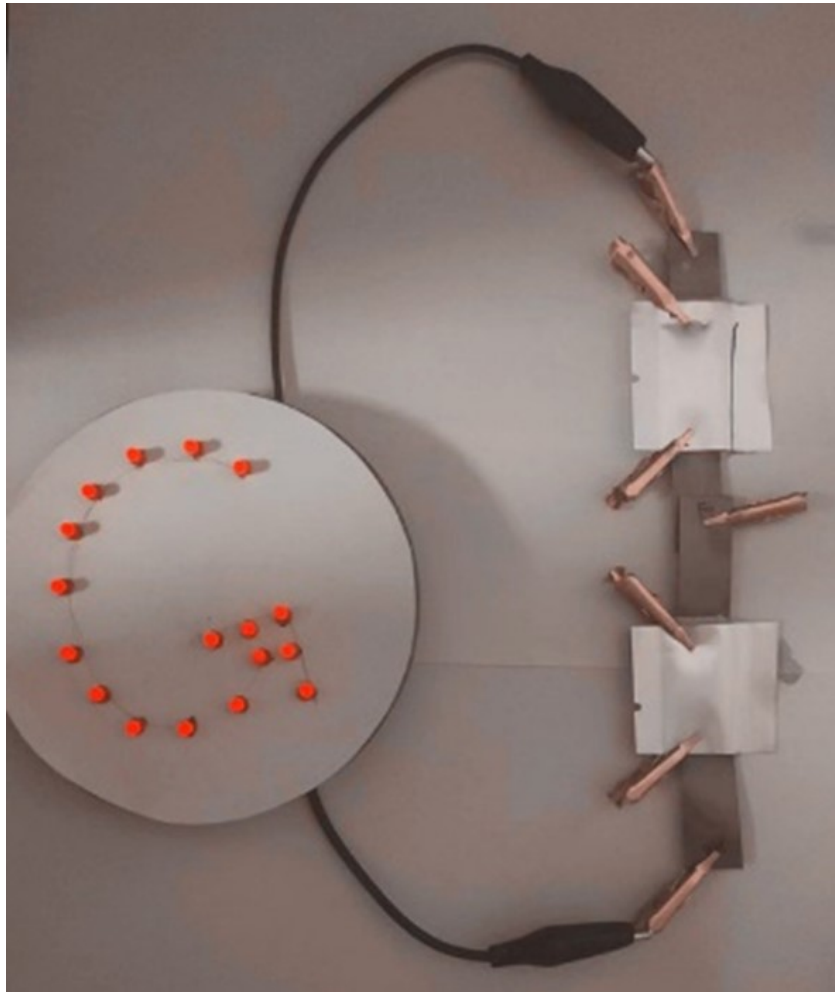


Fig. S11 Lighting 17 LED lights by two ASC devices connected in series.

Table S1 Element contents from EDS analysis of 8L-Co_{0.85}Se@Ni₃S₄/NF

Element	Weight (%)	Atomic
S	18.28	0.35
Co	22.09	0.23
Ni	24.87	0.26
Se	34.76	0.27

Table. S2 Electrochemical performances comparisons of 8L-Co_{0.85}Se@Ni₃S₄ with the recently reported CoSe-based and/or Ni₃S₄-based materials.

Materials	Specific Capacitance	References
8L-Co _{0.85} Se@Ni ₃ S ₄ /NF	1558.33 F g ⁻¹ at 1 A g ⁻¹	This work
CoFe ₂ Se ₄ @CoNi-CH	1288.89 F g ⁻¹ at 1 A g ⁻¹	<i>J. Colloid Interface Sci.</i> 2022, 621 , 149-159.
Ni ₉ S ₈ @Ni ₂ B	1555.33 F g ⁻¹ at 1 A g ⁻¹	<i>J. Colloid Interface Sci.</i> 2023, 649 , 815-825.
NiMo ₃ S ₄ /BP	830 F g ⁻¹ at 1 A g ⁻¹	<i>Small</i> 2024, 20 , 2310120
V-Ni ₃ S ₂	1448.4 F g ⁻¹ at 1 A g ⁻¹	<i>J. Colloid. Interface Sci.</i> 2023, 629 , 1049-1060.
CoSe ₂ /NiSe ₂	1302.5 F g ⁻¹ at 1 A g ⁻¹	<i>J. Electroanal. Chem.</i> 2021, 895 , 115479.
Ni _x Se _y	1025 F g ⁻¹ at 1 A g ⁻¹	<i>J. Electroanal. Chem.</i> 2021, 895 , 115479.
(Ni,Co)Se ₂ -T	1412.5 F g ⁻¹ at 1 A g ⁻¹	<i>Electrochim. Acta</i> 2021, 393 , 139049.
Ni ₃ S ₂ /Co ₉ S ₈ /C-2	1195 F g ⁻¹ at 1 A g ⁻¹	<i>Appl. Surf. Sci.</i> 2022, 574 , 151727.

Table S3 Comparison table of the other identical ASCs with Co_{0.85}Se@Ni₃S₄/NF//PPy/NF ASC

Materials	Energy density and power density	Ref
Co _{0.85} Se@Ni ₃ S ₄ /NF//PPy/NF	76.98 Wh kg ⁻¹ at 775 W kg ⁻¹	This work
NiCo ₂ S ₄ @HCs//AC	69.6 Wh kg ⁻¹ at 847 W kg ⁻¹	<i>Adv. Funct. Mater.</i> 2023, 33 , 2210238
FCNS-2//AC	50 Wh kg ⁻¹ at 1.353 kW kg ⁻¹	<i>Appl. Surf. Sci.</i> 2023, 611 , 155568
CoNiMn-S//RGO	42.1 Wh kg ⁻¹ at 750 W kg ⁻¹	<i>Chem. Eng. J.</i> 2021, 405 , 126928
Ni ₃ S _{4-x} //AC	33.1 Wh kg ⁻¹ at 1.680 kW kg ⁻¹	<i>Small</i> 2022, 18 , 2106074
NiCo ₂ S ₄ //AC	21.4 Wh kg ⁻¹ at 1.663 kW kg ⁻¹	<i>Mater. Res. Bull.</i> 2023, 157 , 112036
NiS/CNFs-2//AC	22.4 Wh kg ⁻¹ at 680 W kg ⁻¹	<i>ACS Appl. Nano Mater.</i> 2022, 5 , 6192
NiMoS ₄ /NiS ₂ //NCO	38.6 Wh kg ⁻¹ at 958.6 W kg ⁻¹	<i>Chem. Eng. J.</i> 2022, 435 , 135231
HCS//Co _{0.85} Se@CoNi ₂ S ₄ /GF	46.5 W h kg ⁻¹ at 750 W kg ⁻¹	<i>J. Mater. Chem. A</i> 2018, 6 , 15630-15639.
Ni-Co-S/GF//PPy/GF	46.5 W h kg ⁻¹ at 825.0 W kg ⁻¹	<i>Adv. Sci.</i> 2018, 5 , 1700375

Table S4 The cycle numbers corresponding to different electrochemical deposition layers

Layer numbers	Sample/ cycle numbers							
1	Co _{0.85} Se /8							
2	Co _{0.85} Se /4	Ni ₃ S ₄ /4						
4	Co _{0.85} Se /2	Ni ₃ S ₄ /2	Co _{0.85} Se /2	Ni ₃ S ₄ /2				
6	Co _{0.85} Se /2	Ni ₃ S ₄ /2	Co _{0.85} Se /1	Ni ₃ S ₄ /1	Co _{0.85} Se /1	Ni ₃ S ₄ /1		
8	Co _{0.85} Se /1	Ni ₃ S ₄ /1	Co _{0.85} Se /1	Ni ₃ S ₄ /1	Co _{0.85} Se /1	Ni ₃ S ₄ /1	Co _{0.85} Se /1	Ni ₃ S ₄ /1