

Electronic supplementary information (ESI)

Complex Isomerism Influencing the Texture Properties of Organometallic [Cu(salen)] Porous Polymers: Paramagnetic Solid-State NMR Characterization and Heterogeneous Catalysis

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L1:

(1*S*,2*S*)-*N,N'*-bis(5-ethynylsalicylidene)-1,2-diaminocyclohexane

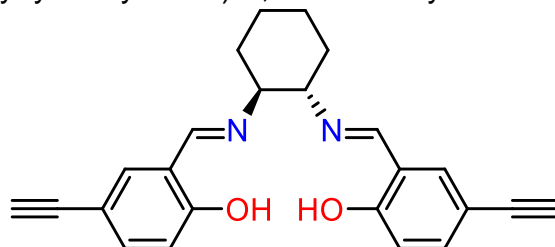


Figure S1: Structure of the ligand L1

¹H NMR (400 MHz, CD₂Cl₂) δ 13.52 (s, 2H), 8.25 (s, 2H), 7.37 (d, *J* = 2.1 Hz, 2H), 7.34 (s, 2H), 6.82 (d, *J* = 8.7 Hz, 2H), 3.38 – 3.29 (m, 2H), 3.00 (s, 2H), 2.00 – 1.81 (m, 2H), 1.79 – 1.65 (m, 2H), 1.57 – 1.43 (m, 4H).

¹³C NMR (101 MHz, CD₂Cl₂) δ 164.36, 162.08, 136.13, 135.73, 118.95, 117.57, 112.46, 83.27, 75.88, 72.86, 33.30, 24.52.

HR-MS ESI, measured (calculated) *m/z* of M+H adduct: 371.174411 (371.175404), C₂₄H₂₃N₂O₂

L2:

trans-N,N'-bis(5-ethynylsalicylidene)-1,2-diaminocyclohexane

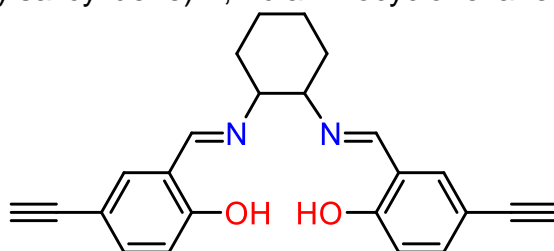


Figure S2: Structure of the ligand **L2**

^1H NMR (400 MHz, CD_2Cl_2) δ 13.52 (s, 2H), 8.25 (s, 2H), 7.37 (d, $J = 2.1$ Hz, 2H), 7.34 (s, 2H), 6.82 (d, $J = 8.8$ Hz, 2H), 3.40 – 3.33 (m, 2H), 3.00 (s, 2H), 1.98 – 1.85 (m, 2H), 1.79 – 1.65 (m, 2H), 1.56 – 1.44 (m, 4H).

^{13}C NMR (101 MHz, CD_2Cl_2) δ 164.36, 162.08, 136.13, 135.73, 118.95, 117.57, 112.46, 83.27, 75.88, 72.85, 33.30, 24.52.

HR-MS ESI, measured (calculated) m/z of $\text{M}+\text{H}$ adduct: 371.174846 (371.175404), $\text{C}_{24}\text{H}_{23}\text{N}_2\text{O}_2$

L3:

cis-N,N'-bis(5-ethynylsalicylidene)-1,2-diaminocyclohexane

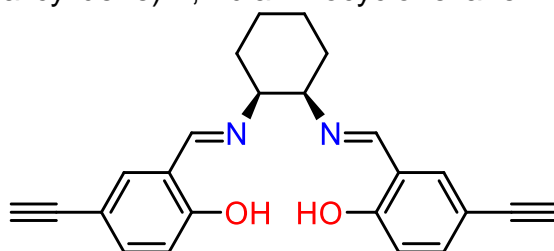


Figure S3: Structure of the ligand **L3**

^1H NMR (400 MHz, CD_2Cl_2) δ 13.76 (s, 2H), 8.33 (s, 2H), 7.46 – 7.37 (m, 4H), 6.86 (d, $J = 8.4$ Hz, 2H), 3.66 – 3.58 (m, 2H), 3.03 (s, 2H), 2.02 – 1.73 (m, 4H), 1.67 – 1.51 (m, 4H).

^{13}C NMR (101 MHz, CD_2Cl_2) δ 164.00, 162.40, 136.23, 135.74, 119.10, 117.77, 112.37, 83.38, 75.88, 69.76, 31.10, 22.74.

HR-MS ESI, measured (calculated) m/z of $\text{M}+\text{H}$ adduct: 371.174930 (371.175404), $\text{C}_{24}\text{H}_{23}\text{N}_2\text{O}_2$

L1

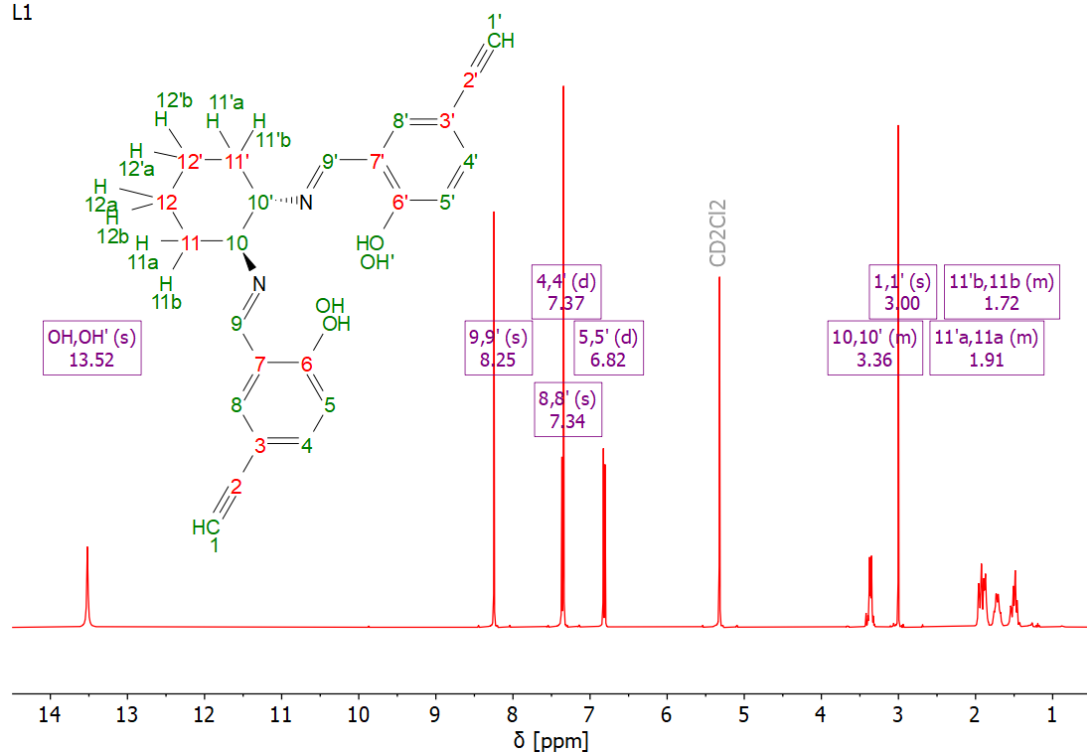


Figure S4: ¹H NMR spectrum of L1 ligand

L1

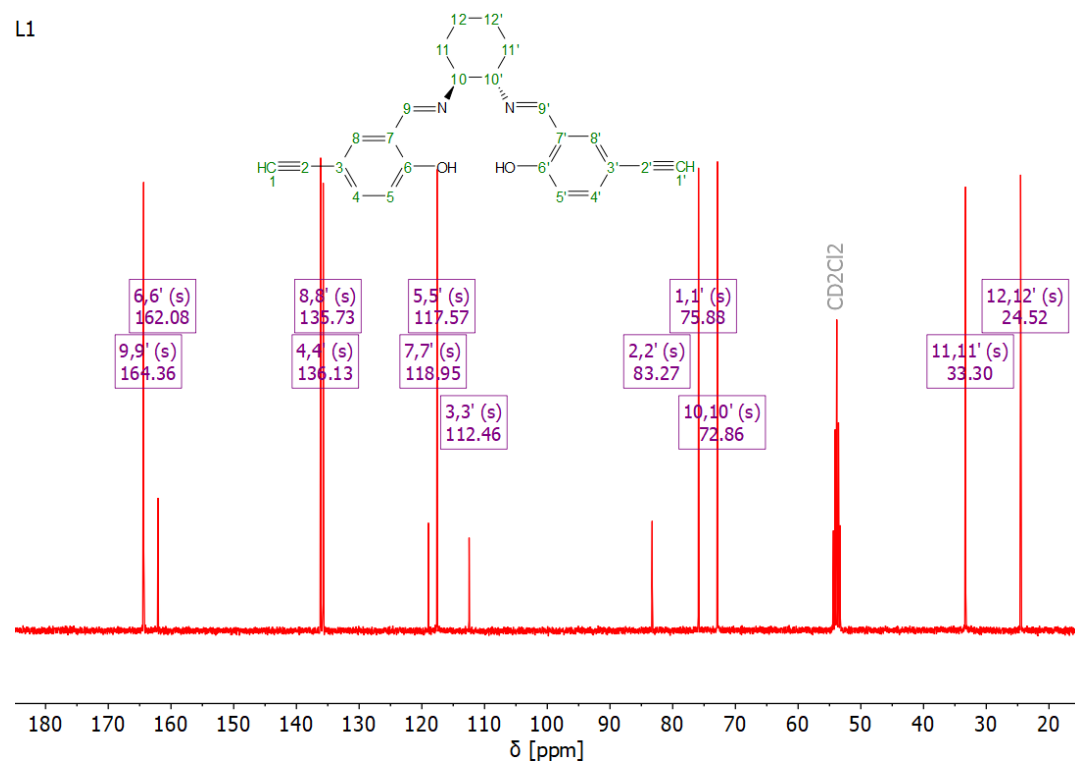


Figure S5: ¹³C{¹H} NMR spectrum of L1 ligand

L2

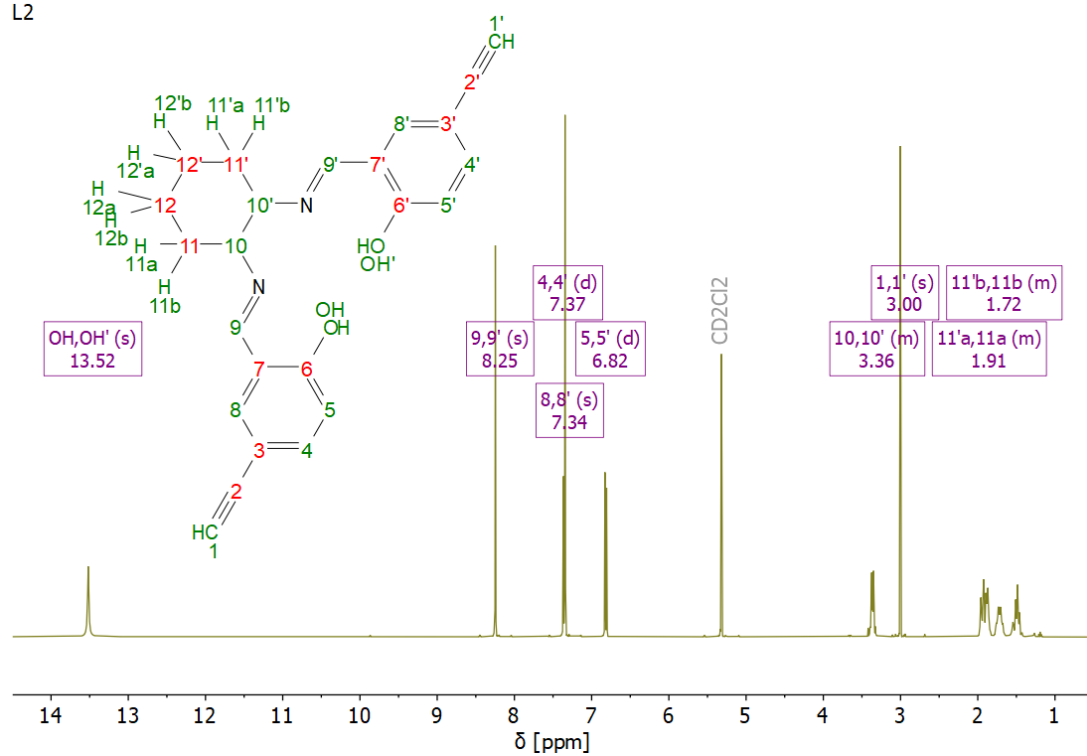


Figure S6: ^1H NMR spectrum of L2 ligand

L2

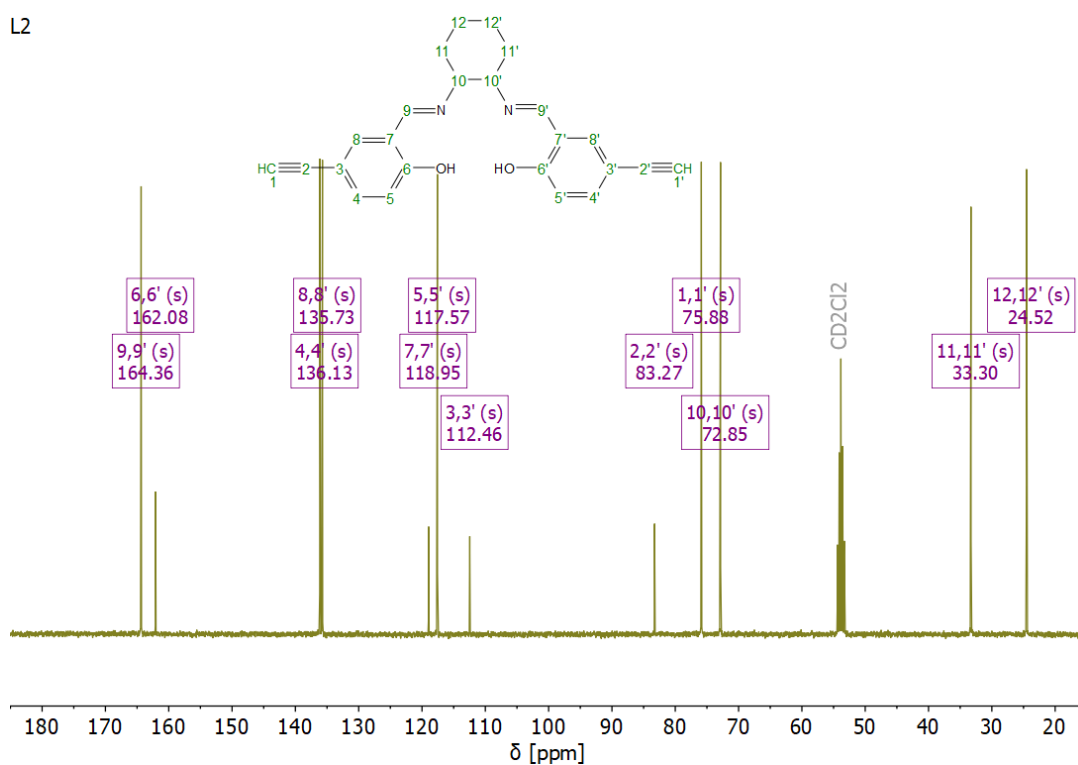
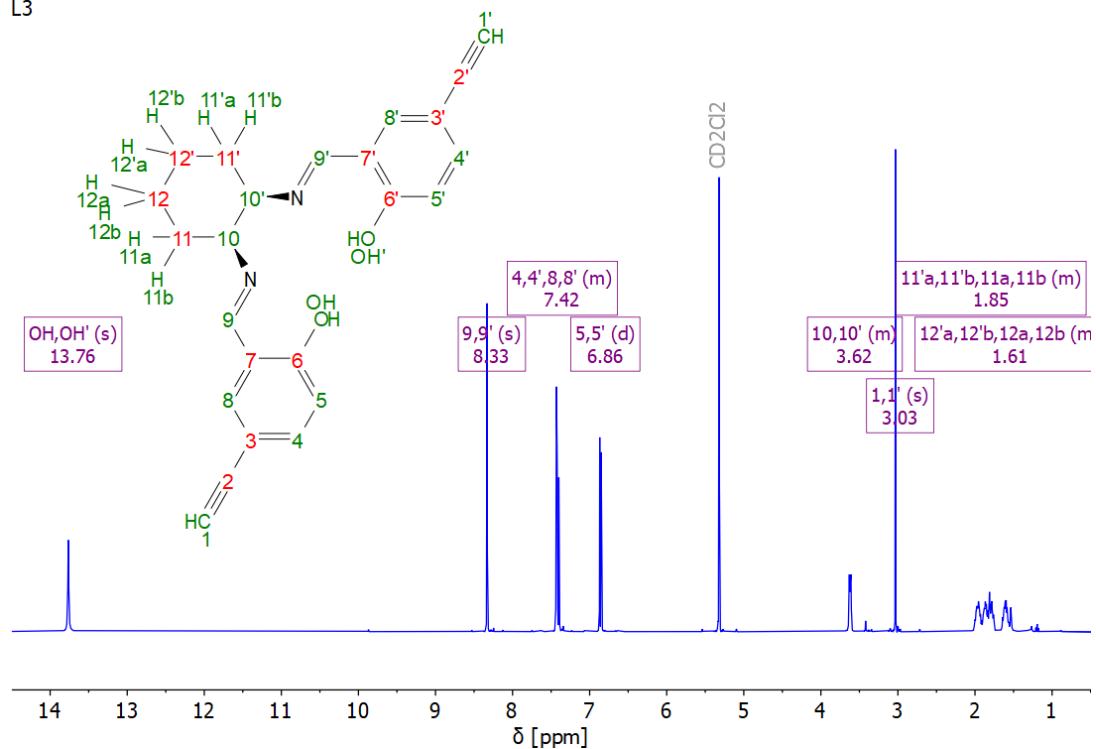
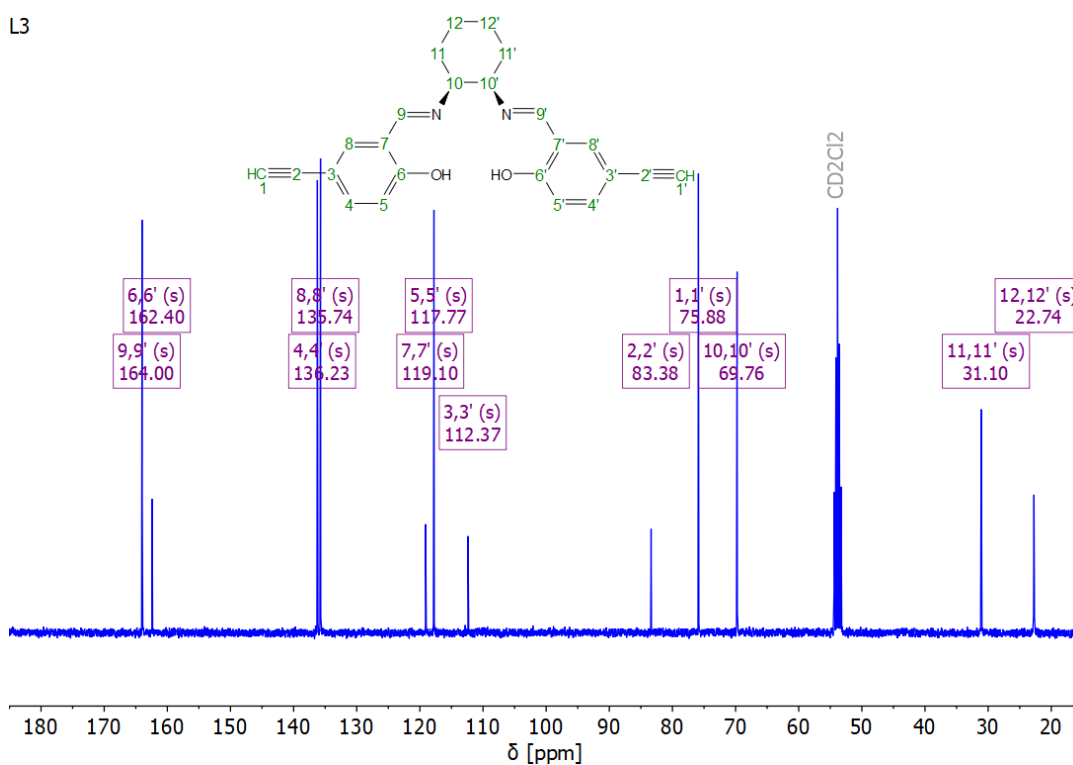


Figure S7: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of L2 ligand

L3

Figure S8: ^1H NMR spectrum of L3 ligand

L3

Figure S9: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of L3 ligand

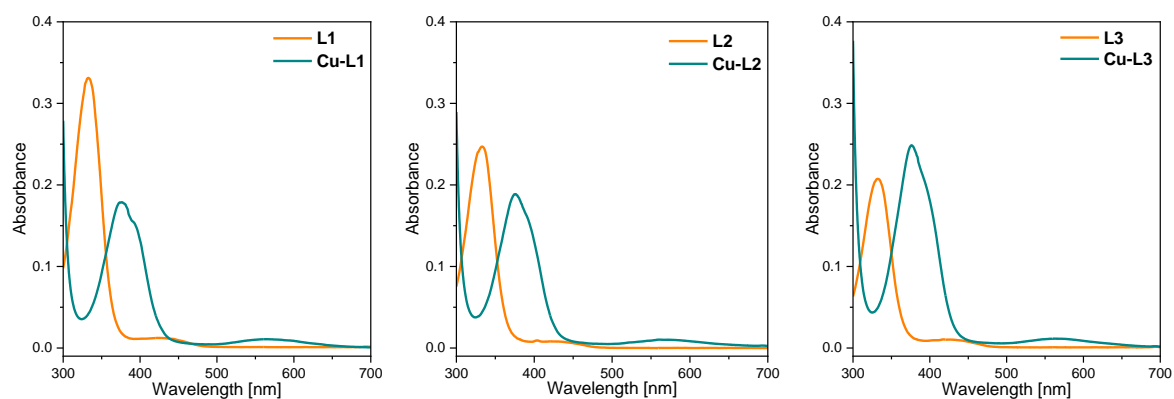


Figure S10: UV/VIS spectra of ligands (**L1**, **L2** and **L3**) and complexes (**Cu-L1**, **Cu-L2** and **Cu-L3**)

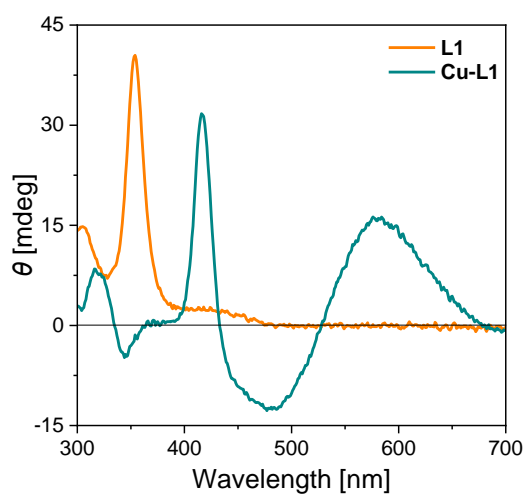


Figure S11: Circular dichroism (CD) spectra of ligand **L1** and complex **Cu-L1**.

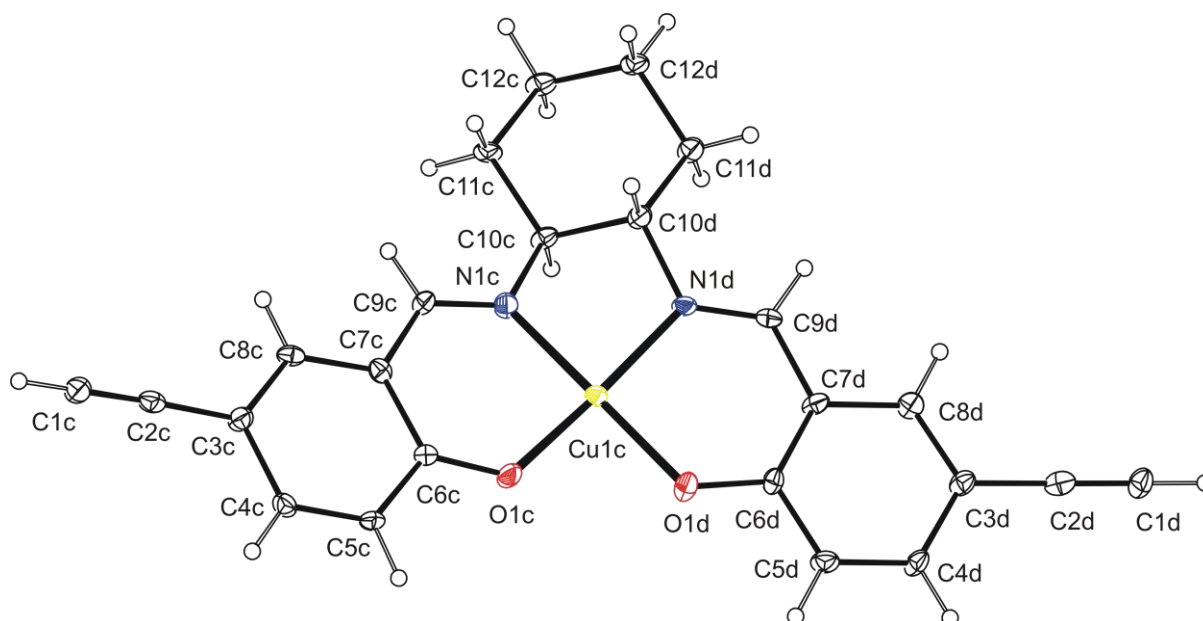


Figure S12: View on the one of two symmetrically independent molecule of **Cu-L1** with the atom numbering schema, the displacement ellipsoids are drawn at 50% probability level.

Crystal data for **Cu-L1**: $C_{24}H_{20}CuN_2O_2 \cdot C_3H_7NO$, $M_r = 505.05$; Monoclinic, $P2_1$, (No 4), $a = 8.6380$ (3) Å, $b = 31.0524$ (10) Å, $c = 8.6749$ (3) Å, $\beta = 90.667$ (2), $V = 2326.72$

(14) \AA^3 , $Z = 4$, $D_x = 1.442 \text{ Mg m}^{-3}$, brown plate of dimensions $0.28 \times 0.16 \times 0.03 \text{ mm}$, multi-scan absorption correction ($\mu = 1.61 \text{ mm}^{-1}$) $T_{\min} = 0.79$, $T_{\max} = 0.96$; a total of 24115 measured reflections ($\theta_{\max} = 68.4^\circ$), from which 8359 were unique ($R_{\text{int}} = 0.050$) and 7893 observed according to the $I > 2\sigma(I)$ criterion. The refinement converged ($\Delta/\sigma_{\max} = 0.001$) to $R = 0.061$ for observed reflections and $wR(F^2) = 0.143$, $GOF = 1.16$ for 617 parameters and all 8359 reflections. The final difference map displayed no peaks of chemical significance ($\Delta\rho_{\max} = 0.60$, $\Delta\rho_{\min} -0.88 \text{ e.\AA}^{-3}$).

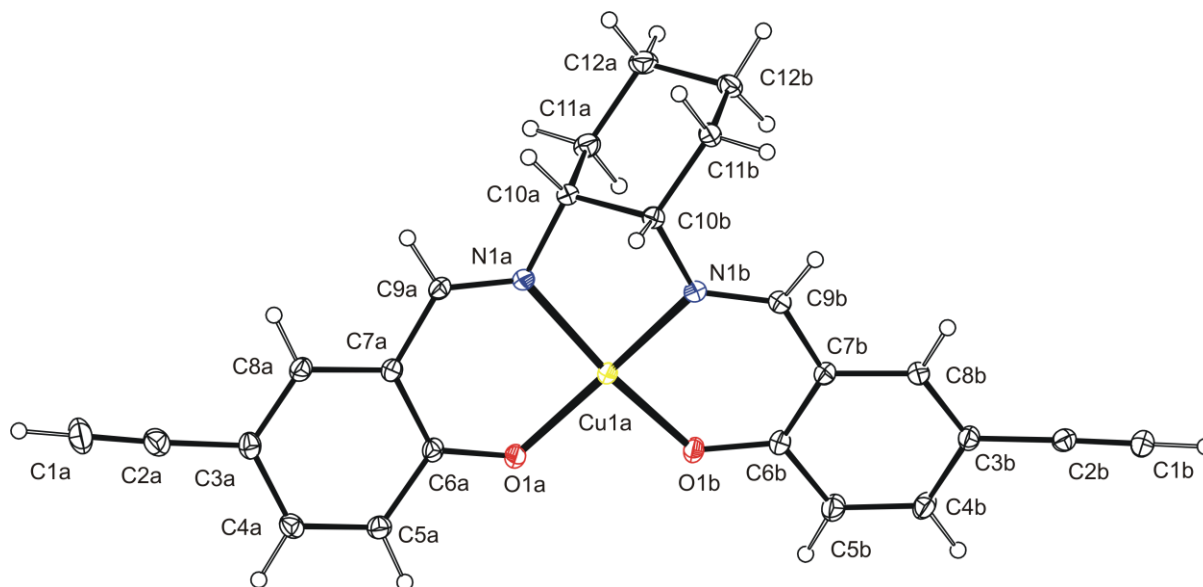


Figure S13: View on molecule of **Cu-L3** with the atom numbering schema, the displacement ellipsoids are drawn at 50% probability level.

Crystal data for **Cu-L3**: $\text{C}_{24}\text{H}_{20}\text{CuN}_2\text{O}_2$, $M_r = 431.96$; Monoclinic, $P2_1/n$, (No 14), $a = 14.1633 (10) \text{ \AA}$, $b = 9.0797 (7) \text{ \AA}$, $c = 16.0469 (11) \text{ \AA}$, $\beta = 112.615 (2)$, $V = 1904.9 (2) \text{ \AA}^3$, $Z = 4$, $D_x = 1.506 \text{ Mg m}^{-3}$, green prism of dimensions $0.25 \times 0.10 \times 0.07 \text{ mm}$, multi-scan absorption correction ($\mu = 1.81 \text{ mm}^{-1}$) $T_{\min} = 0.77$, $T_{\max} = 0.89$; a total of 57728 measured reflections ($\theta_{\max} = 77.5^\circ$), from which 4036 were unique ($R_{\text{int}} = 0.031$) and 3854 observed according to the $I > 2\sigma(I)$ criterion. The refinement converged ($\Delta/\sigma_{\max} = 0.001$) to $R = 0.032$ for observed reflections and $wR(F^2) = 0.088$, $GOF = 1.04$ for 262 parameters and all 4036 reflections. The final difference map displayed no peaks of chemical significance ($\Delta\rho_{\max} = 0.67$, $\Delta\rho_{\min} -0.70 \text{ e.\AA}^{-3}$).

X-ray crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (CCDC) under deposition number 2333975 and 2333976 for **Cu-L1** and **Cu-L3**, respectively and can be obtained free of charge from the Centre via its website (<https://www.ccdc.cam.ac.uk/structures/>).

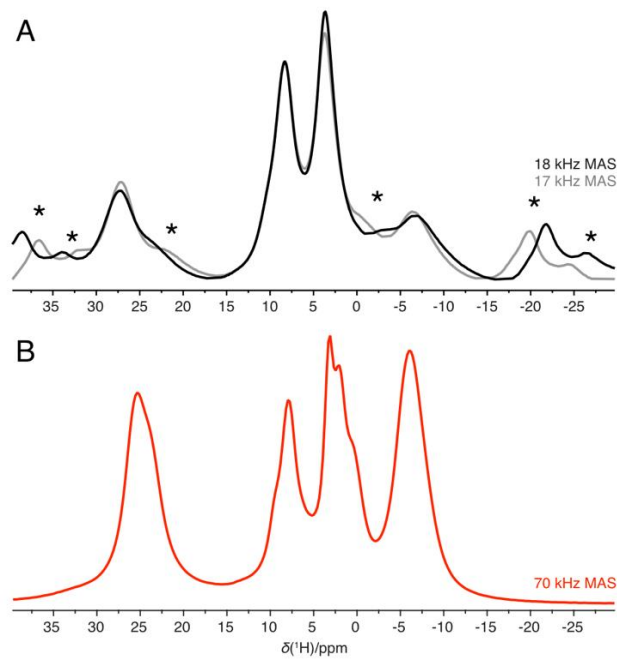


Figure S14: Comparison of slow and fast MAS ^1H NMR spectra of **Cu-L1**.

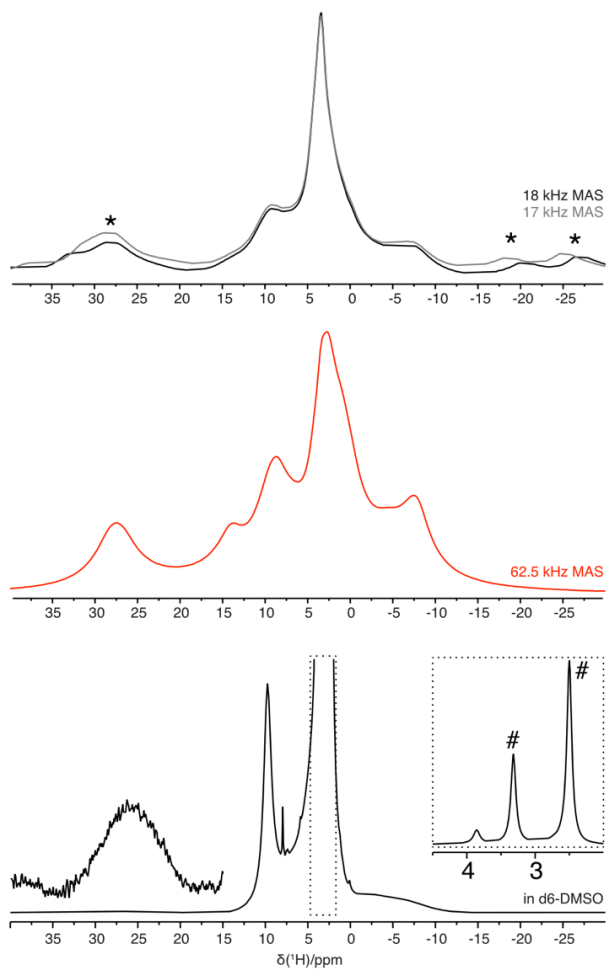


Figure S15: Comparison of slow and fast MAS ^1H NMR spectra of **Cu-L3** and solution ^1H NMR spectra of **Cu-L3**.

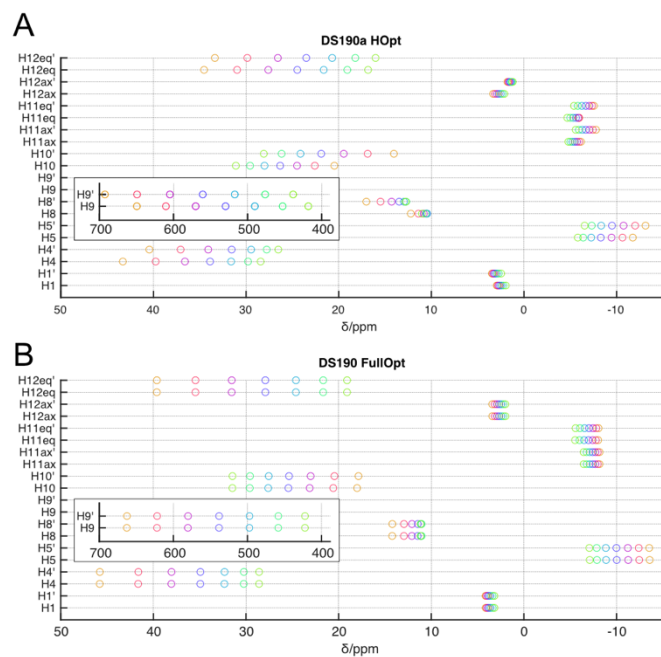


Figure S16: ^1H NMR shifts of **Cu-L1** predicted by calculations for X-ray based geometry with refined ^1H positions (A) and fully optimized geometry in COSMO/ CHCl_3 solvent model (B) calculated for range of HFX admixture (from orange 10% for green 40% in 5% steps) (298 K).

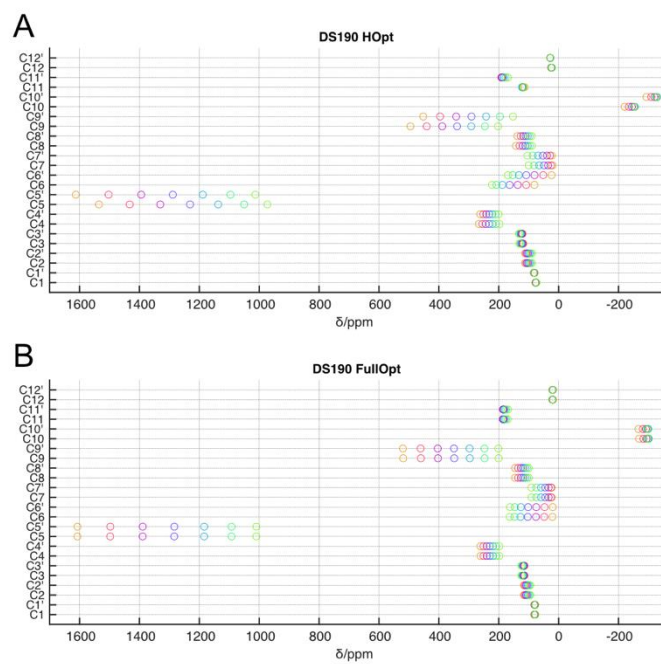


Figure S17: ^{13}C NMR shifts of **Cu-L1** predicted by calculations for X-ray based geometry with refined ^1H positions (A) and fully optimized geometry in COSMO/ CHCl_3 solvent model (B) calculated for range of HFX admixture (from orange 10% for green 40% in 5% steps) (298 K).

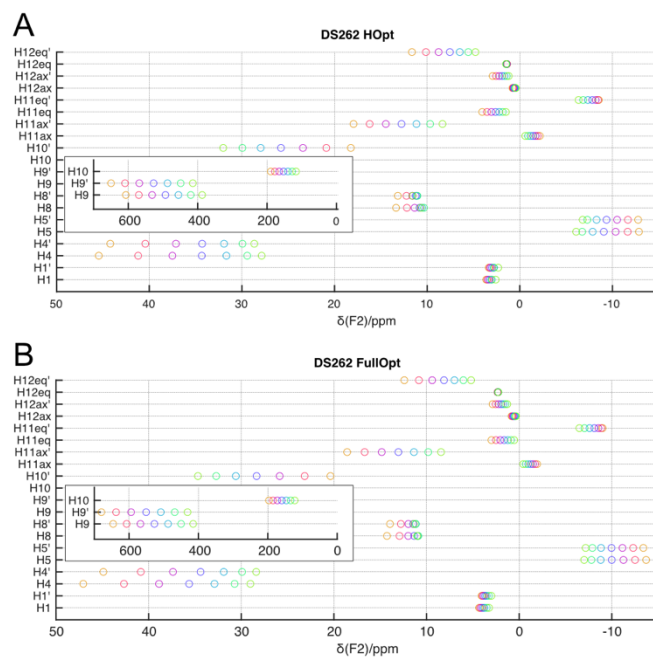


Figure S18: ^1H NMR shifts of **Cu-L3** predicted by calculations for X-ray based geometry with refined ^1H positions (A) and fully optimized geometry in COSMO/ CHCl_3 solvent model (B) calculated for range of HFx admixture (from orange 10% for green 40% in 5% steps) (298 K).

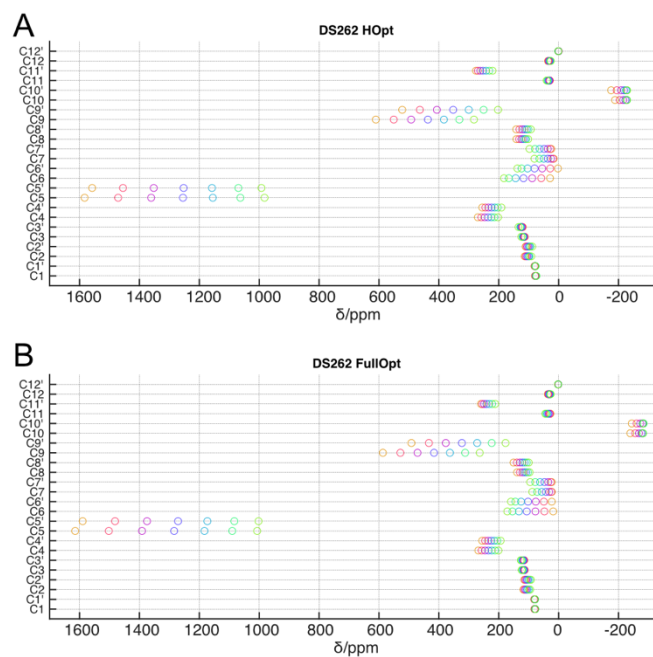


Figure S19: ^{13}C NMR shifts of **Cu-L3** predicted by calculations for X-ray based geometry with refined ^1H positions (A) and fully optimized geometry in COSMO/ CHCl_3 solvent model (B) calculated for range of HFx admixture (from orange 10% for green 40% in 5% steps) (298 K).

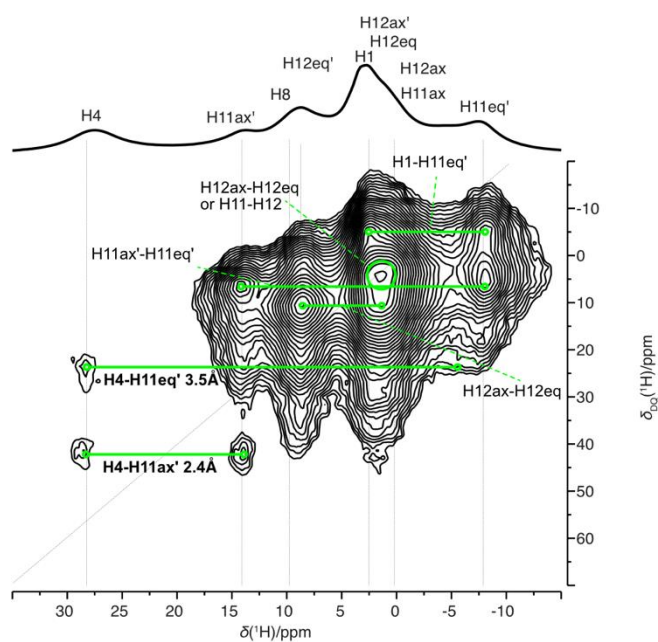


Figure S20: Homonuclear BaBa16 ^1H single-double quantum (SQ-DQ) correlation of **Cu-L3**.

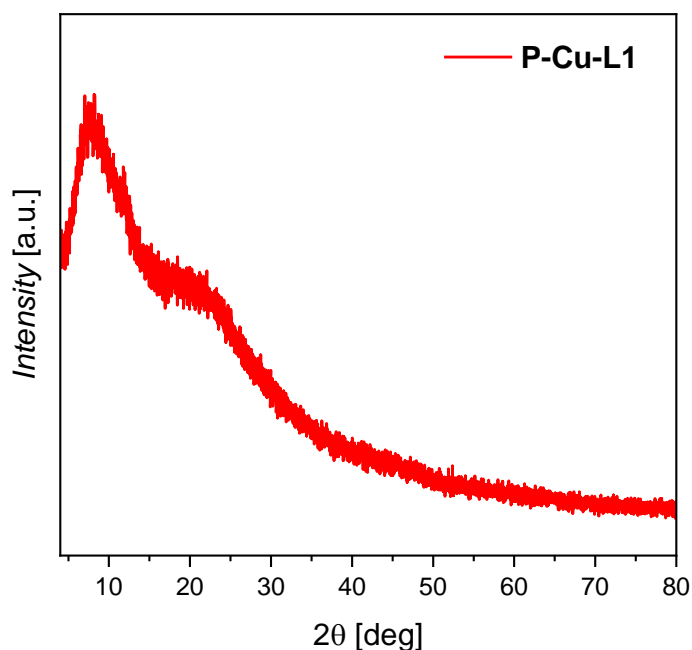


Figure S21: Powder X-Ray Diffraction (pXRD) pattern of network **P-Cu-L1**.

Powder X-ray Diffraction (XRD) pattern was collected using a high-resolution Explorer diffractometer (GNR Analytical Instruments, Italy) equipped with a one-dimensional Mythen 1K silicon strip detector (Dectris, Switzerland). A Cu X-Ray tube (wavelength $\lambda = 1.54 \text{ \AA}$) operated at 40 kV and 30 mA and monochromatized with Ni foil (β filter) was used. Measurements were performed in the 2θ range of $5 - 80^\circ$ with a 0.1° step and 15 s exposure time at each step.

Table S1: Observed ^1H and ^{13}C NMR shift from HSQC-TEDOR experiment of **Cu-L1** a related calculated value for 10%, 25% and 40% of Hartee-Fock exchange admixture (at 298 K) and for various temperature 298 K and 323 K (using 25% HFX), as well as decomposition of calculated NMR shift (298 K, 25% HFX) to orbital shift ($\delta_{dia,K} = \sigma_{ref} - \sigma_{dia,K}$), Fermi contact (σ_{FC}) and spin-dipolar (σ_{SD}) shielding. All values listed in ppm, for atom label see Scheme 1.

| | δ_{exp} | $\delta_{calc}^{10\%}$ | $\delta_{calc}^{25\%}$ | $\delta_{calc}^{40\%}$ | δ_{calc}^{298K} | δ_{calc}^{323K} | $\delta_{dia,K}$ | σ_{FC} | σ_{SD} |
|-------|----------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------|---------------|---------------|
| H1 | 3.9 | 4.13 | 3.77 | 3.18 | 3.77 | 3.70 | 2.81 | -1.01 | 0.05 |
| H4 | 24.5 | 45.79 | 34.93 | 28.59 | 34.93 | 32.82 | 7.76 | -27.36 | 0.20 |
| H5 | ND | -13.55 | -10.00 | -7.08 | -10.00 | -8.68 | 7.14 | 16.51 | 0.63 |
| H8 | 8.9 | 14.21 | 11.47 | 11.15 | 11.47 | 11.16 | 7.54 | -4.24 | 0.31 |
| H9 | ND | 662.79 | 538.35 | 422.32 | 538.35 | 497.34 | 8.31 | -530.81 | 0.77 |
| H10 | ND | 17.93 | 25.39 | 31.47 | 25.39 | 23.67 | 3.14 | -22.44 | 0.19 |
| H11ax | -5.8 | -8.16 | -7.43 | -6.45 | -7.43 | -6.76 | 1.24 | 8.34 | 0.32 |
| H11eq | -5.8 | -8.04 | -7.00 | -5.55 | -7.00 | -6.27 | 2.48 | 9.02 | 0.46 |
| H12ax | 1.6 | 3.42 | 2.66 | 2.00 | 2.66 | 2.56 | 1.36 | -1.48 | 0.18 |
| H12eq | 25.0 | 39.63 | 27.91 | 19.09 | 27.91 | 25.91 | 2.02 | -26.09 | 0.19 |
| C1 | 80.4 | 81.5 | 78.8 | 77.8 | 78.8 | 79.0 | 82.08 | 3.14 | 0.17 |
| C2 | Q | 115.9 | 105.9 | 94.4 | 105.9 | 104.7 | 90.16 | -15.85 | 0.10 |
| C3 | Q | 114.1 | 117.4 | 124.3 | 117.4 | 117.2 | 114.34 | -3.44 | 0.38 |
| C4 | 214 | 261.0 | 229.8 | 198.2 | 229.8 | 223.6 | 149.74 | -81.55 | 1.52 |
| C5 | ND | 1607.0 | 1283.6 | 1009.5 | 1283.6 | 1194.4 | 131.63 | -1153.34 | 1.41 |
| C6 | Q | 20.2 | 102.4 | 162.9 | 102.4 | 108.5 | 180.85 | 74.40 | 4.04 |
| C7 | Q | 23.2 | 44.8 | 90.8 | 44.8 | 51.0 | 125.66 | 78.53 | 2.37 |
| C8 | 99 | 145.7 | 118.4 | 99.9 | 118.4 | 120.9 | 150.85 | 31.31 | 1.12 |
| C9 | NA | 519.6 | 349.0 | 200.9 | 349.0 | 335.3 | 170.89 | -184.00 | 5.84 |
| C10 | NA | -267.7 | -298.4 | -296.2 | -298.4 | -269.7 | 71.59 | 367.18 | 2.77 |
| C11 | 162 | 184.0 | 183.7 | 169.4 | 183.7 | 171.9 | 31.57 | -153.36 | 1.27 |
| C12 | 24 | 19.2 | 20.7 | 21.3 | 20.7 | 21.2 | 27.85 | 6.76 | 0.40 |

ND – not detected signal, Q - quaternary carbon

Table S2: Observed ^1H and ^{13}C NMR shift from HSQC-TEDOR experiment of **Cu-L3** a related calculated value for 10%, 25% and 40% of Hartee-Fock exchange admixture (at 298 K) and for various temperature 298 K and 323 K (using 25% HFX), as well as decomposition of calculated NMR shift (298 K, 25% HFX) to orbital shift ($\delta_{dia,K} = \sigma_{ref} - \sigma_{dia,K}$), Fermi contact (σ_{FC}) and spin-dipolar (σ_{SD}) shielding. All values listed in ppm, for atom label see Scheme 1.

| Values in ppm: | δ_{exp} | $\delta_{calc}^{10\%}$ | $\delta_{calc}^{25\%}$ | $\delta_{calc}^{40\%}$ | δ_{calc}^{298K} | δ_{calc}^{323K} | $\delta_{dia,K}$ | σ_{FC} | σ_{SD} |
|----------------|----------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------|---------------|---------------|
| H1 | 2.7 | 4.32 | 3.90 | 3.26 | 3.90 | 3.82 | 2.81 | -1.14 | 0.05 |
| H1' | 2.7 | 4.05 | 3.67 | 3.01 | 3.67 | 3.61 | 2.82 | -0.90 | 0.05 |
| H4 | 27.5 | 47.06 | 35.63 | 29.02 | 35.63 | 33.47 | 7.76 | -28.06 | 0.19 |
| H4' | 28.5 | 44.89 | 34.39 | 28.39 | 34.39 | 32.33 | 7.77 | -26.83 | 0.21 |
| H5 | ND | -13.70 | -9.99 | -7.01 | -9.99 | -8.67 | 7.13 | 16.50 | 0.62 |
| H5' | ND | -13.41 | -9.97 | -7.17 | -9.97 | -8.64 | 7.15 | 16.46 | 0.65 |
| H8 | 9.5 | 14.27 | 11.34 | 11.02 | 11.34 | 11.04 | 7.52 | -4.12 | 0.30 |
| H8' | 9.5 | 13.95 | 11.44 | 11.36 | 11.44 | 11.14 | 7.55 | -4.22 | 0.33 |
| H9 | ND | 645.79 | 526.92 | 415.16 | 526.92 | 486.79 | 8.22 | -519.47 | 0.76 |
| H9' | ND | 679.57 | 550.72 | 431.41 | 550.72 | 508.77 | 8.37 | -543.14 | 0.79 |
| H10 | ND | 196.70 | 159.64 | 122.54 | 159.64 | 147.53 | 3.15 | -157.47 | 0.98 |
| H10' | ND | 20.38 | 28.35 | 34.68 | 28.35 | 26.45 | 3.84 | -24.76 | 0.26 |
| H11ax | -8.0 | -1.93 | -1.25 | -0.43 | -1.25 | -0.99 | 2.08 | 4.48 | -1.15 |
| H11ax' | 13.6 | 18.57 | 13.04 | 8.44 | 13.04 | 12.15 | 1.55 | -11.91 | 0.42 |
| H11eq | 0.8 | 2.99 | 1.62 | 0.54 | 1.62 | 1.62 | 1.61 | 0.26 | -0.26 |
| H11eq' | 0.8 | -9.01 | -8.11 | -6.50 | -8.11 | -7.29 | 2.58 | 10.19 | 0.51 |
| H12ax | 0.8 | 0.83 | 0.57 | 0.35 | 0.57 | 0.63 | 1.36 | 0.67 | 0.12 |
| H12ax' | 1.1 | 2.88 | 1.94 | 1.29 | 1.94 | 1.89 | 1.30 | -0.72 | 0.07 |
| H12eq | 0.8 | 2.34 | 2.29 | 2.26 | 2.29 | 2.25 | 1.82 | -0.37 | -0.09 |
| H12eq' | 8.5 | 12.40 | 8.12 | 5.19 | 8.12 | 7.60 | 1.41 | -6.83 | 0.12 |
| C1 | 85.3 | 81.2 | 78.5 | 77.7 | 78.5 | 78.8 | 82.2 | 3.5 | 0.2 |

| | | | | | | | | | |
|------|------|--------|--------|--------|--------|--------|-------|---------|------|
| C1' | 85.3 | 81.9 | 79.4 | 78.9 | 79.4 | 79.6 | 82.2 | 2.7 | 0.2 |
| C2 | Q | 117.7 | 107.2 | 95.2 | 107.2 | 105.9 | 90.1 | -17.2 | 0.1 |
| C2' | Q | 114.7 | 104.7 | 92.6 | 104.7 | 103.5 | 90.1 | -14.7 | 0.1 |
| C3 | Q | 113.3 | 116.6 | 123.7 | 116.6 | 116.4 | 114.4 | -2.6 | 0.4 |
| C3' | Q | 114.9 | 118.7 | 126.5 | 118.7 | 118.3 | 114.5 | -4.5 | 0.4 |
| C4 | 214 | 268.0 | 234.5 | 201.2 | 234.5 | 227.9 | 149.7 | -86.3 | 1.5 |
| C4' | 202 | 255.3 | 225.0 | 193.7 | 225.0 | 219.2 | 149.9 | -76.7 | 1.6 |
| C5 | ND | 1614.5 | 1284.1 | 1006.8 | 1284.1 | 1194.9 | 131.6 | -1153.8 | 1.3 |
| C5' | ND | 1589.5 | 1271.0 | 1002.2 | 1271.0 | 1182.9 | 131.6 | -1140.9 | 1.5 |
| C6 | Q | 18.2 | 106.4 | 171.3 | 106.4 | 112.2 | 181.2 | 71.0 | 3.8 |
| C6' | Q | 22.9 | 102.1 | 159.1 | 102.1 | 108.2 | 180.6 | 74.2 | 4.2 |
| C7 | Q | 22.8 | 42.6 | 87.9 | 42.6 | 49.0 | 125.7 | 80.9 | 2.3 |
| C7' | Q | 23.1 | 47.2 | 95.0 | 47.2 | 53.2 | 125.8 | 76.3 | 2.4 |
| C8 | 101 | 138.5 | 113.9 | 97.0 | 113.9 | 116.7 | 150.2 | 35.2 | 1.1 |
| C8' | 101 | 150.6 | 120.8 | 100.0 | 120.8 | 123.1 | 150.8 | 28.9 | 1.2 |
| C9 | ND | 586.9 | 416.1 | 263.5 | 416.1 | 397.3 | 173.0 | -248.7 | 5.6 |
| C9' | ND | 491.7 | 323.1 | 177.3 | 323.1 | 311.5 | 173.9 | -155.3 | 6.1 |
| C10 | ND | -239.3 | -276.5 | -281.1 | -276.5 | -249.3 | 75.1 | 348.8 | 2.8 |
| C10' | ND | -244.2 | -279.8 | -281.8 | -279.8 | -253.1 | 65.5 | 342.6 | 2.8 |
| C11 | 33.1 | 27.4 | 34.6 | 44.5 | 34.6 | 34.7 | 35.6 | 1.3 | -0.3 |
| C11' | ~240 | 259.3 | 241.0 | 211.8 | 241.0 | 224.4 | 26.5 | -215.6 | 1.1 |
| C12 | 33.1 | 36.3 | 31.1 | 26.6 | 31.1 | 30.9 | 27.9 | -3.2 | 0.0 |
| C12' | 7.8 | 1.0 | 1.5 | 1.9 | 1.5 | 2.9 | 20.3 | 18.6 | 0.2 |

ND – not detected signal, Q - quaternary carbon

Table S3: Results achieved in repeating application of **P-Cu-L1** as heterogeneous catalyst of styrene oxidation.

| | Styrene conversion [%] | Selectivity [%] | |
|-----------------------|------------------------|-----------------|--------------|
| | | Styrene oxide | Benzaldehyde |
| 1 st cycle | 80 | 41 | 40 |
| 2 nd cycle | 88 | 30 | 53 |
| 3 rd cycle | 96 | 18 | 51 |