Co_3O_4 (111) surfaces in contact with water: Molecular dynamics study of the surface chemistry and structure at room temperature

Tim Kox and Stephane Kenmoe*

Department of Theoretical Chemistry, University of Duisburg-Essen, Universitatetstrasse 2, D-45141, Essen, Germany, *stephane.kenmoe@uni-due.de



Figure S1: Kinetic, potential and total energies of the water covered A (top) and B-(bottom) terminations during the 20 ps simulations



Figure S2: Time evolution of the surface oxygens to water Hydrogens (Os-H) bond distances from the simulations on the A- and B-terminated Co_3O_4 (111)/H₂O interface. On the A termination, Proton transfer 1 denotes the transfer happening on the surface oxygens Os1, while Proton transfer 2 denotes the one happening on surface oxygens Os2



Figure S3: Average value of the radial distribution functions reported together with the standard deviation (sd), of the Co^{2+} to water oxygens O_w distances (left) and of the water oxygens to water hydrogens distances (right), on the A-terminated surface.



Figure S4: Average value of the radial distribution functions reported together with their standard deviation (sd), of the Co²⁺ to surface oxygens Os distances, on the clean (top left) and water covered (top right) on the A-terminated surface. Corresponding values for the surface oxygens to water hydrogens (bottom).



Figure S5: Average value of the radial distribution functions reported together with the standard deviation (sd), of the Co^{3+} to water oxygens O_w distances (left); of the Co^{2+} to water oxygens O_w distances (right) and of the water oxygens to water hydrogens distances (bottom), on the B-terminated surface.



Figure S6: Average value of the radial distribution functions reported together with their standard deviation (sd), of the Co^{3+} to surface oxygens Os distances on the clean (top left) and water covered (top right) B-terminated surface; for the Co^{2+} to surface oxygens Os distances (middle left and right). Corresponding values for the surface oxygens to water hydrogens (bottom).