Co(II)-Cu(II) Mixed Oxide Catalyst for Single Step Synthesis of 2,4,5-Triaryle-1H-Imidazole Derivatives Under Microwave Irradiation

Manohar K. Jopale^{a,i}, Zhiyong Zheng^b, Soowon Choi^b, Bharat N. Shelke^{c,i}, Nilesh Mharsale^a, Shripad Patil^d, Sonali S. Chine^e, Vilas Gade^f, Shital V. Kahane^g, Ravindra N. Bulakhe^{b*}, Ji Man Kim^{b*}, Amol H. Kategaonkar^{h,i*}

^aDepartment of Chemistry, M.V.P. Samaj's Arts, Commerce & Science College Jawahar Road, Tryambakeshwar, Nashik 422 212, Maharashtra, India

^bDepartment of Chemistry, Sungkyunkwan University, Suwon, 16419, Republic of Korea

^dDepartment of Chemistry, Rayat Shikshan Sanstha's, Dada Patil Mahavidyalaya, Karjat, Ahmednagar 414 401 Maharashtra, India

^eDepartment of Engineering Sciences. & Humanities, Sanjivani College of Engineering, Kopargaon (S.P.P.U. Pune) Ahmednagar 423 601 Maharashtra, India

^fDepartment of Chemistry, Rayat Shikshan Sanstha's, Arts, Science and Commerce College, Mokhada, Palghar 401 604, Maharashtra, India

^g Dr. Vishwanath Karad MIT World Peace University, Kothrud, Pune

^hDepartment of Chemistry, M.V.P. Samaj's K.R.T. Arts, B.H. Commerce and A.M. Science (KTHM) College, Nashik-422 002, Maharashtra, India.

ⁱDepartment of Chemistry, M.V.P. Samaj's G.M.D. Arts, B. W. Commerce & Science College Sinnar, Nashik 422 103, Maharashtra, India

*Corresponding Author E-mail id: bulakhe@skku.edu, jimankim@skku.edu, amol.kategaonkar@gmail.com

FTIR:

FTIR spectrum shown in figure 1, (Model FT/IR- 4600 in solid phase) from 400–4000 cm⁻¹ for synthesized catalyst material. Characterization shows the absorption intensities at 3666, 3445, 1604 and 1399 cm⁻¹ are due to chemisorbed water molecules. Frequency 2929, 2852, 2358 and 2315 cm⁻¹ are due to asymmetric and symmetric stretching of catalyst material, 1017 cm⁻¹ shows stretching vibration of Co=O and Cu=OThe band at 659 and 581 cm⁻¹ are due to Co-O and Cu-O bending vibrations.



Figure: 1. FTIR spectra of Co-Cu oxide mixed composite.

^cDepartment of Chemistry, M.V.P. Samaj's SVKT, Arts, Commerce & Science College, Deolali Camp, Nashik 422 401, Maharashtra, India

TOF-MS Spectra:



Figure 2. TOF MS ES spectra of Co-Cu oxide mixed phase a) TOF MS ES+ b) TOF MS ES-

Catalyst preparation:



Figure 3. Preparation of Co-Cu oxide mixed phase **a**. Euphorbia neriifolia latex **b**. Latex + $CuCl_2.2H_2O$ **c**. Latex + $CuCl_2.2H_2O$ + $CoCl_2.6H_2O$ **d**. gel of Co-Cu precursor **e**. sol of Co-Cu precursor

Spectral Data:



4e.¹H-NMR spectrum of 2-(4-fluorophenyl)-4,5-diphenyl-1H-imidazole

4e.¹³C-NMR spectrum of 2-(4-fluorophenyl)-4,5-diphenyl-1H-imidazole







4f. ¹³C-NMR spectrum of 2-(2-chlorophenyl)-4,5-diphenyl-1H-imidazole



4g.¹H-NMR spectrum of 2-(4-chlorophenyl)-4,5-diphenyl-1H-imidazole



4g. ¹³C-NMR spectrum of 2-(4-chlorophenyl)-4,5-diphenyl-1H-imidazole





4h.1H-NMR spectrum of N,N-dimethyl-4-(4,5-diphenyl-1H-imidazol-2-yl)benzenamine

4h. ¹³C-NMR spectrum of N,N-dimethyl-4-(4,5-diphenyl-1H-imidazol-2-yl)benzenamine



4j.¹H-NMR spectrum of 4,5-diphenyl-2-p-tolyl-1H-imidazole



4j. ¹³C-NMR spectrum of 4,5-diphenyl-2-p-tolyl-1H-imidazole



4m.¹H-NMR spectrum of 3-(4,5-diphenyl-1H-imidazol-2-yl)-1H-indole



4m. ¹³C-NMR spectrum of 3-(4,5-diphenyl-1H-imidazol-2-yl)-1H-indole



40.1H-NMR spectrum of 4-(4,5-diphenyl-1H-imidazol-2-yl)phenol

40. ¹³C-NMR spectrum of 4-(4,5-diphenyl-1H-imidazol-2-yl)phenol

4p.1H-NMR spectrum of 2-(4,5-diphenyl-1H-imidazol-2-yl)pyridine

4p. ¹³C-NMR spectrum of 2-(4,5-diphenyl-1H-imidazol-2-yl)pyridine

