

Supporting information

**Broadband Blue Light Emissions of One-Dimensional
Hybrid Cu(I) Halides with Ultrahigh Anti-Water
Stability**

Na Lin,^{ab} Yun-Xia Li,^a Yu Liu,^a Kai-Qi Sun,^a Hong-Yan Zhang,^a Xiao-Wu Lei,^{a*}
Zhi-Wei Chen^{a*}

^a Research institute of Optoelectronic Functional Materials, School of Chemistry, Chemical Engineering and Materials, Jining University, Qufu, Shandong, 273155, P. R. China

^b School of Chemistry and Chemical Engineering, Qufu Normal University, Qufu, Shandong, 273165, P. R. China

Corresponding Authors: Zhi-Wei Chen, Xiao-Wu Lei

E-mail: zw_chen@jnxu.edu.cn; xwlei_jnu@163.com

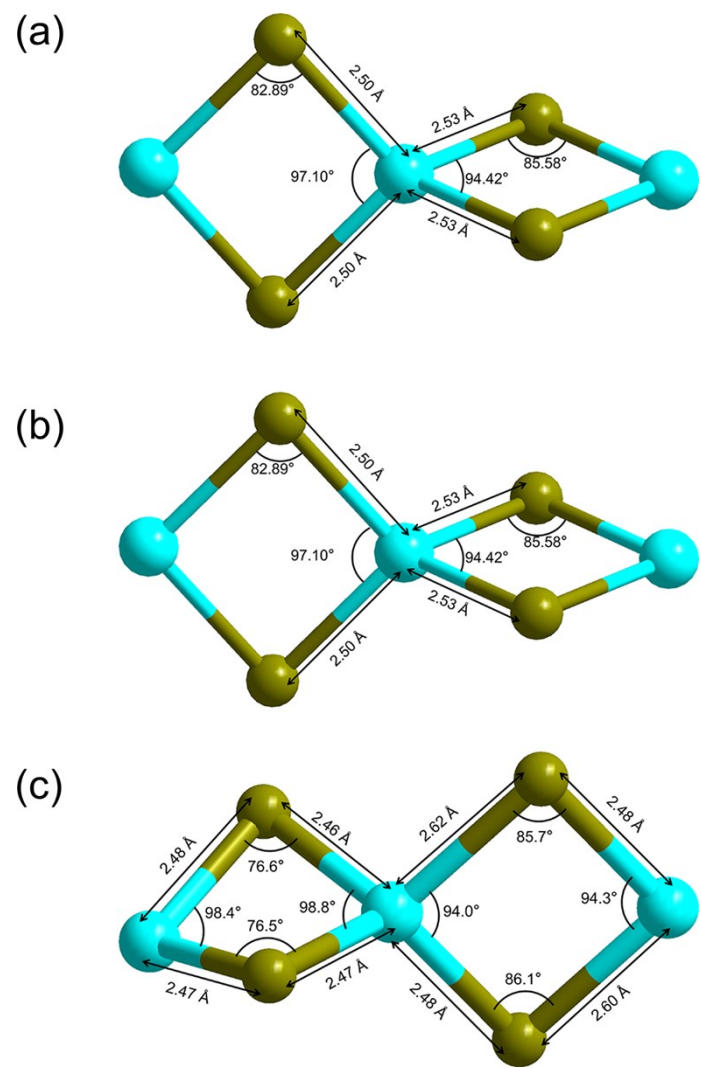


Fig. S1 The configurations of 1D $[\text{Cu}_2\text{Br}_4]^{2-}$ chain of (a) $[(\text{Me})_4\text{-Pipz}]\text{Cu}_2\text{Br}_4$, (b) $[\text{BuDA}]\text{Cu}_2\text{Br}_4$ and (c) $[\text{TMEDA}]\text{Cu}_2\text{Br}_4$.

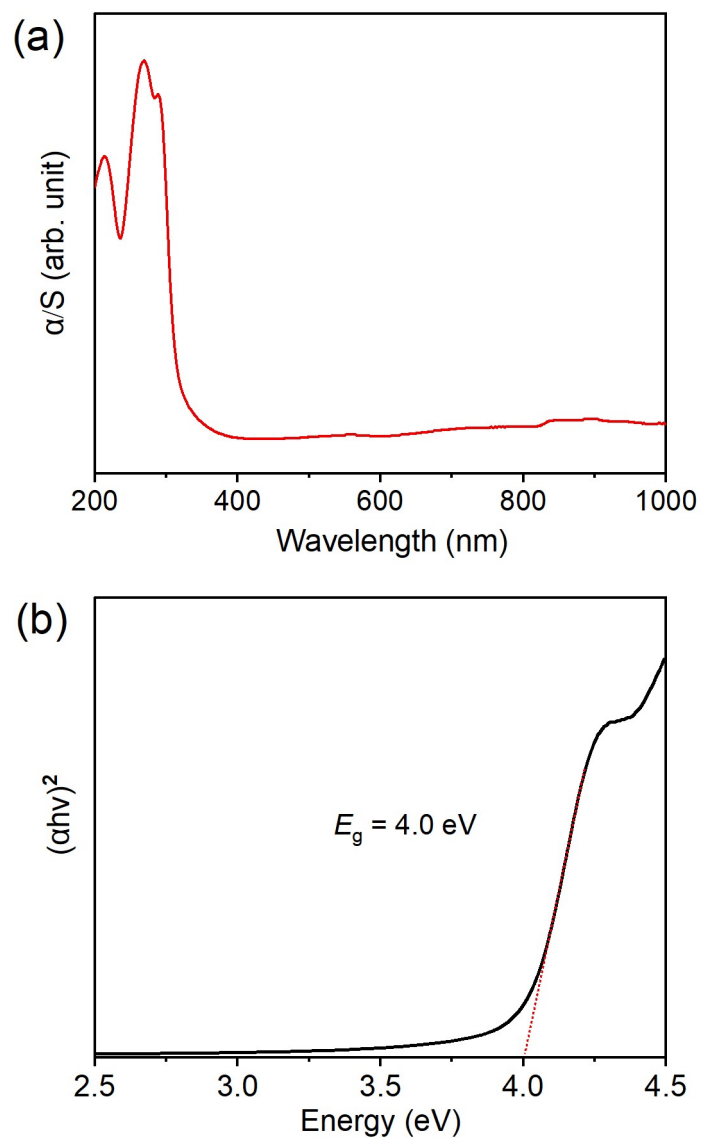


Fig. S2 The solid state UV-vis adsorption spectrum (a) and Tauc plot for band gap (b) of $[(\text{Me})_4\text{-Pipz}]\text{Cu}_2\text{Br}_4$.

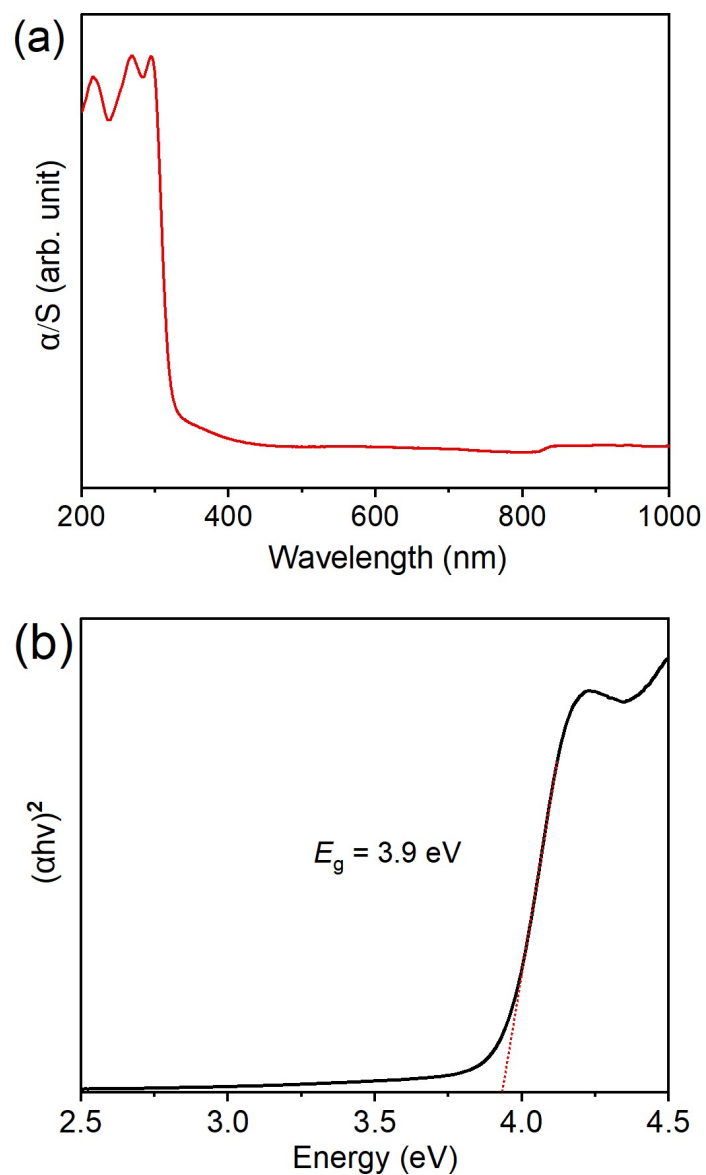


Fig. S3 The solid state UV-vis adsorption spectrum (a) and Tauc's plot for band gap (b) of [BuDA]Cu₂Br₄.

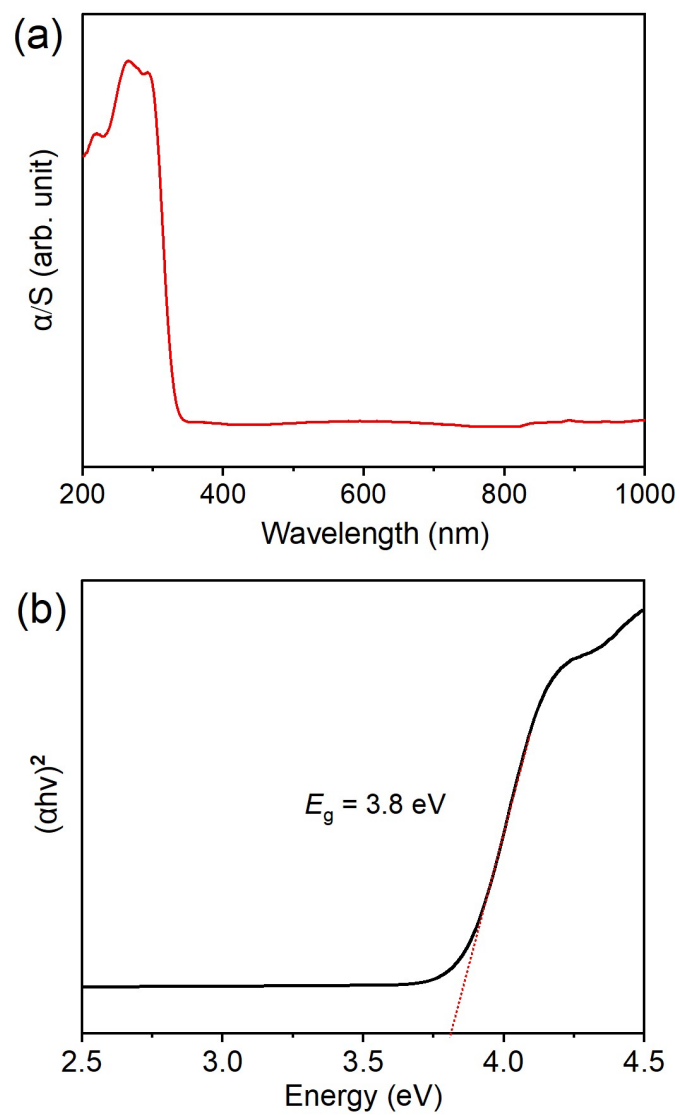


Fig. S4 The solid state UV-vis adsorption spectrum (a) and Tauc plot for band gap (b) of [TMEDA]Cu₂Br₄.

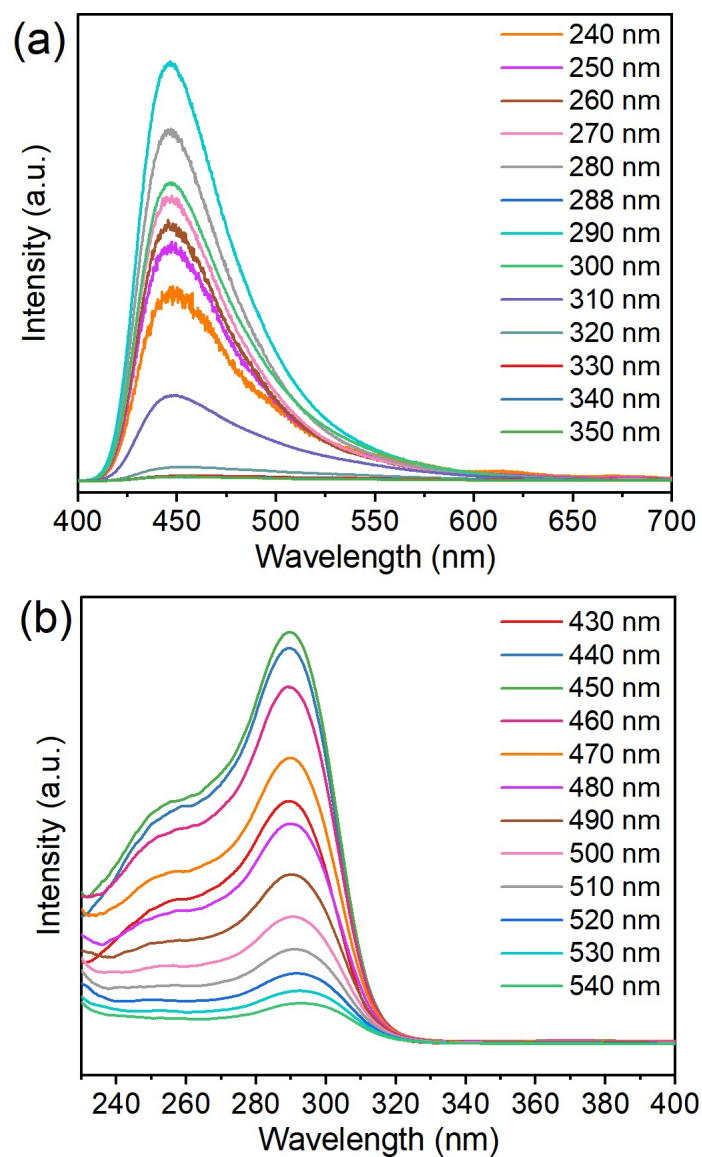


Fig. S5 (a) The excitation wavelength dependent PL emission spectra, and (b) the emission wavelength dependent PL excitation spectra of $[(\text{Me})_4\text{-Pipz}]\text{Cu}_2\text{Br}_4$.



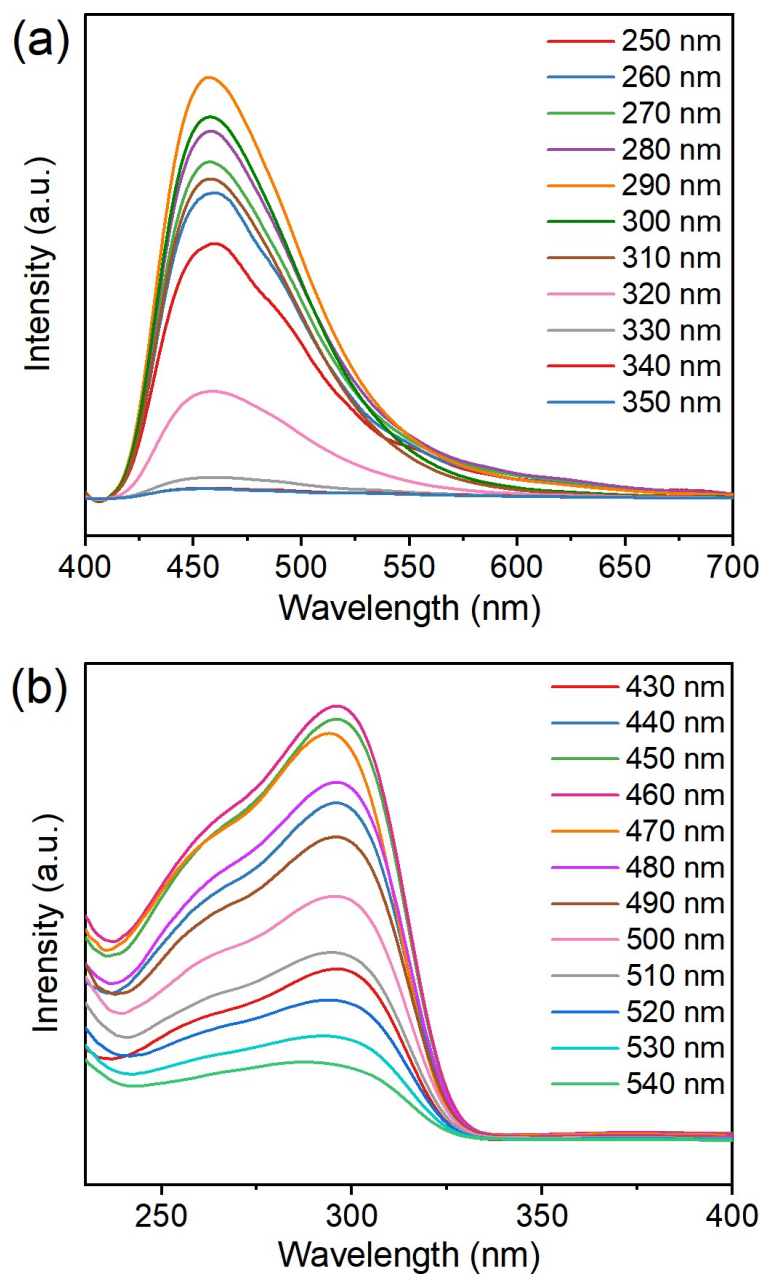


Fig. S6 (a) The excitation wavelength dependent PL spectra and (b) the emission wavelength dependent PL excitation spectra of [BuDA]Cu₂Br₄.

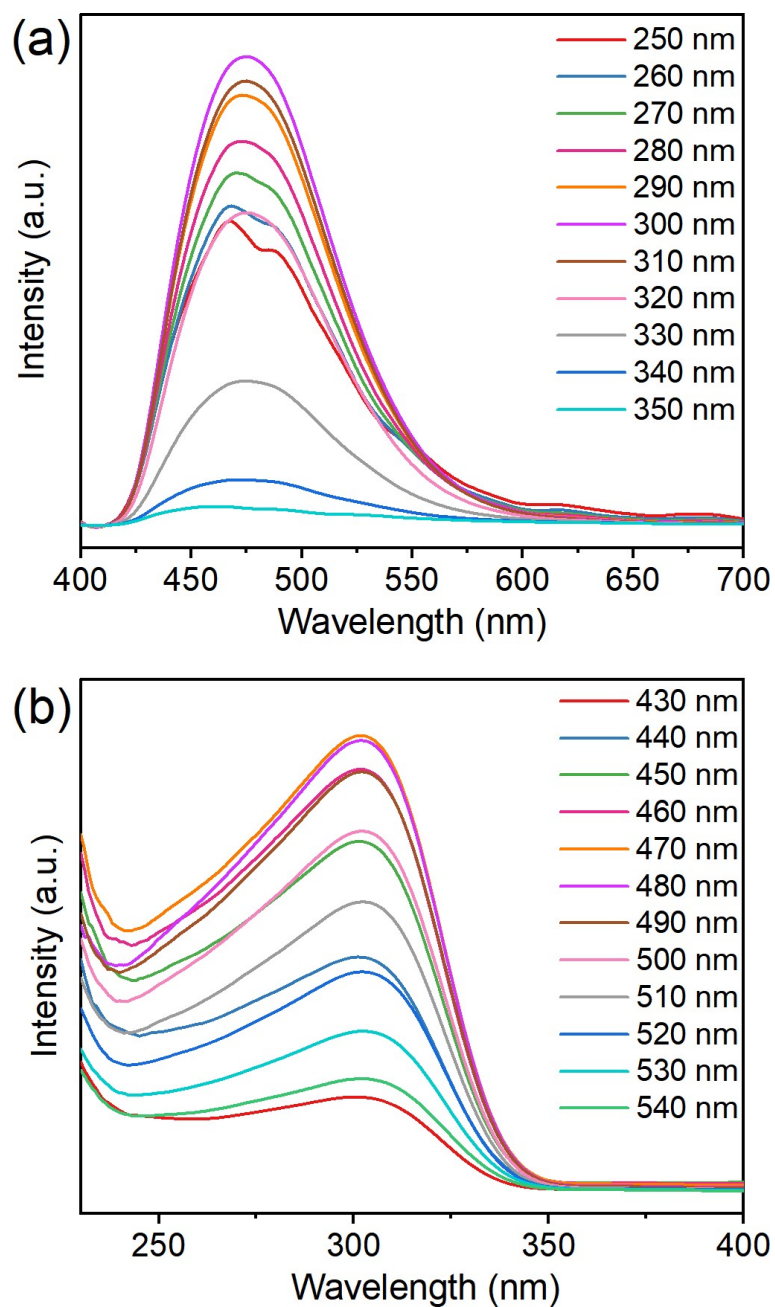


Fig. S7 (a) The excitation wavelength dependent PL spectra and (b) the emission wavelength dependent PL excitation spectra of [TMEDA]Cu₂Br₄.

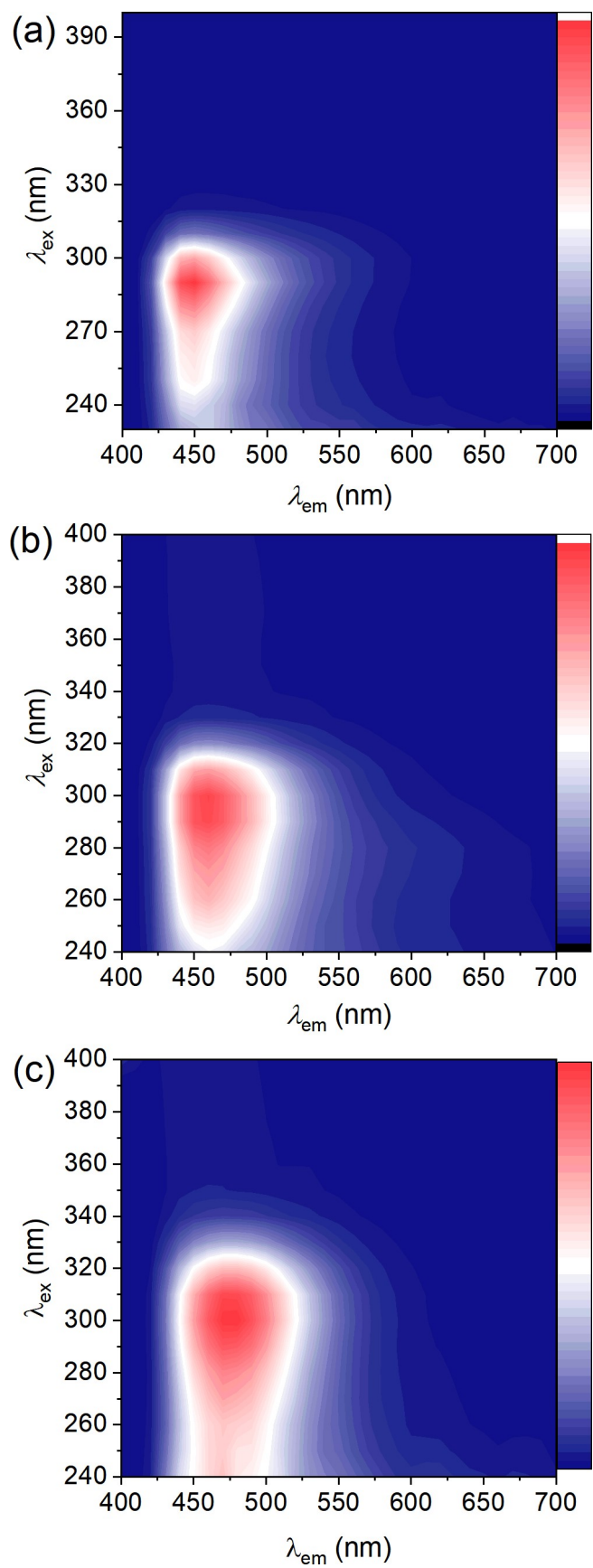


Fig. S8 3D consecutive PL excitation and emission correlation maps of (a) $[(\text{Me})_4\text{Pipz}]\text{Cu}_2\text{Br}_4$; (b) $[\text{BuDA}]\text{Cu}_2\text{Br}_4$ and (c) $[\text{TMEDA}]\text{Cu}_2\text{Br}_4$.

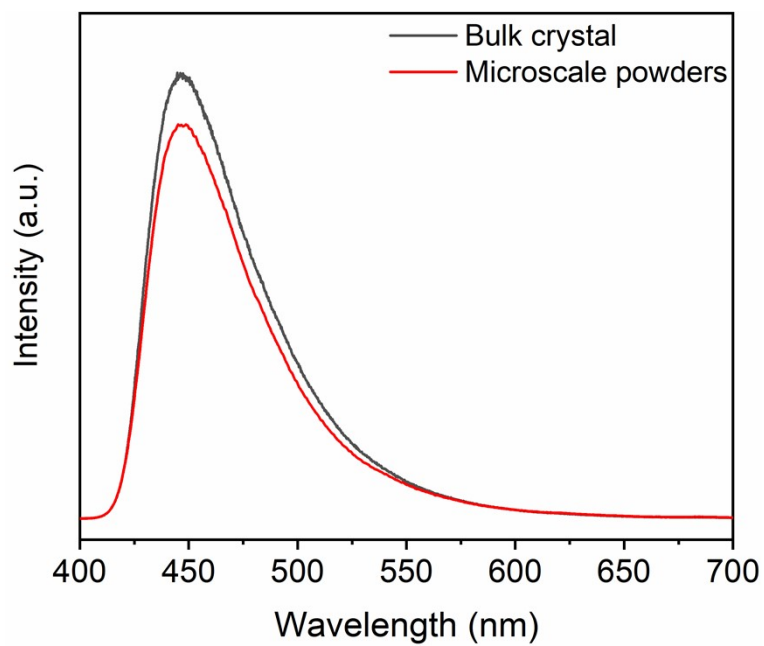


Fig. S9 Comparison of PL emission spectra of bulk crystals and microscale powders for [(Me)₄-Pipz]Cu₂Br₄.

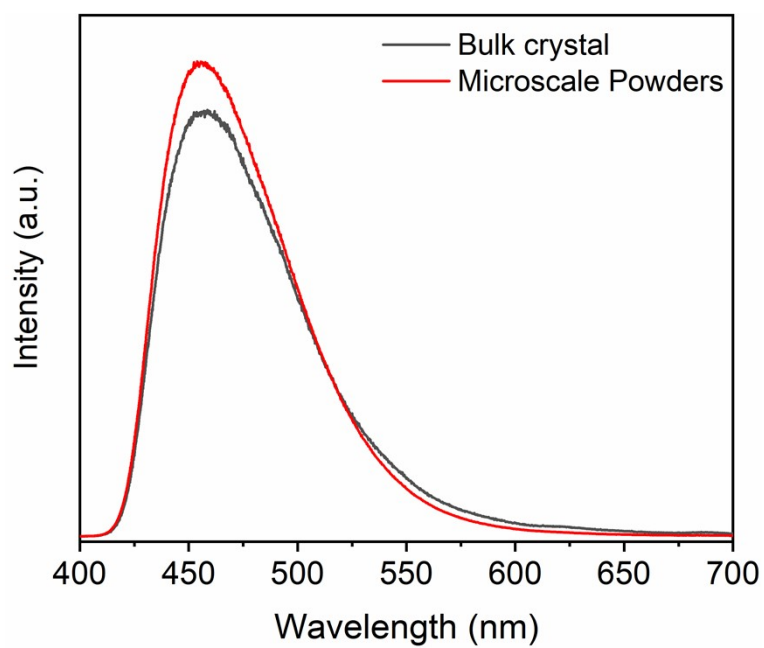


Fig. S10 Comparison of PL emission spectra of bulk crystals and microscale powders for [BuDA]Cu₂Br₄.

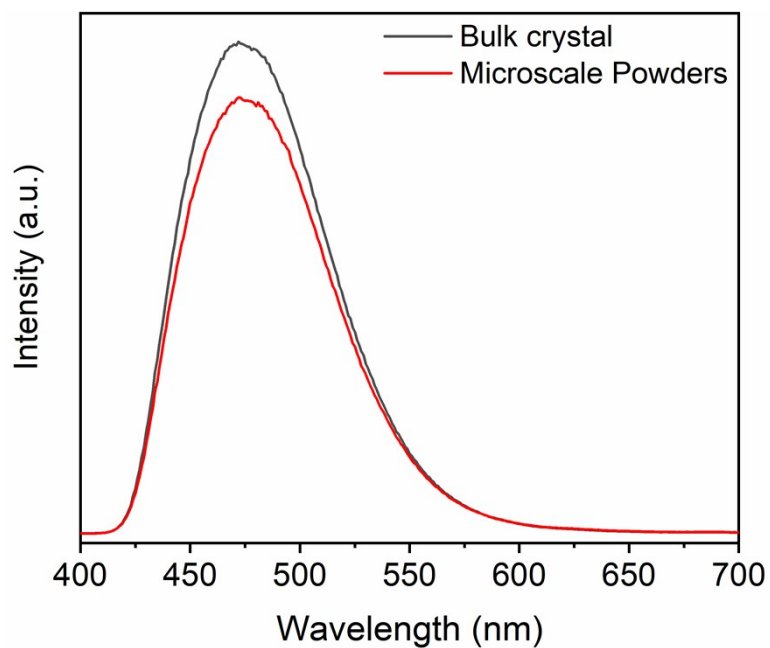


Fig. S11 Comparison of PL emission spectra of bulk crystals and microscale powders for [TMEDA]Cu₂Br₄.

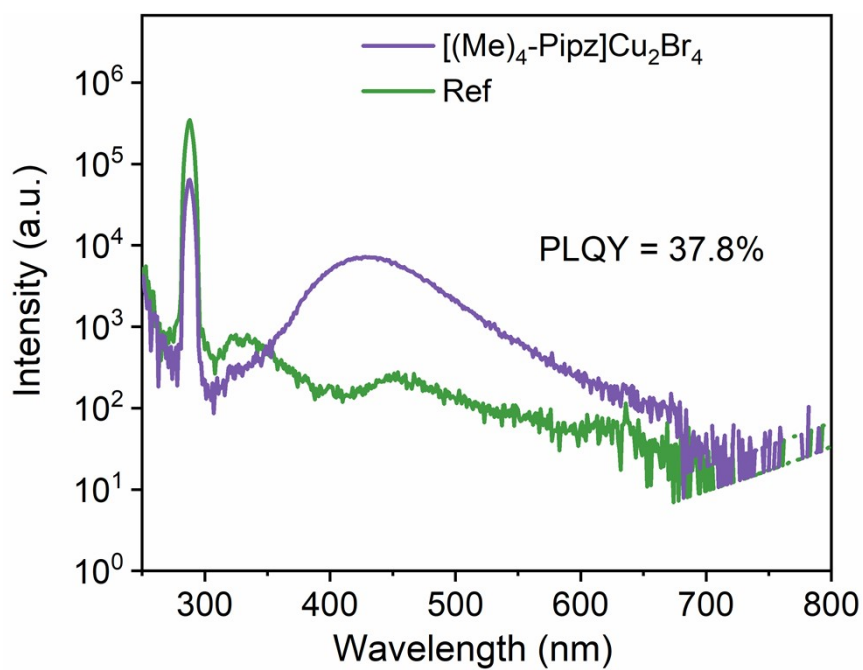


Fig. S12 The PLQY of [(Me)₄-Pipz]Cu₂Br₄.

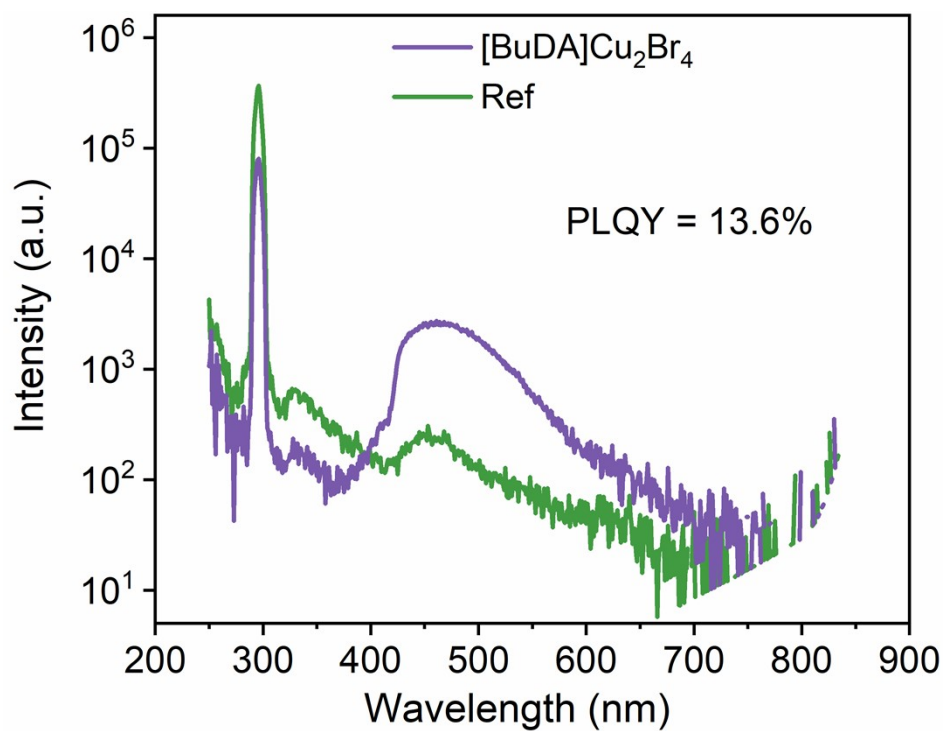


Fig. S13 The PLQY of [BuDA]Cu₂Br₄.

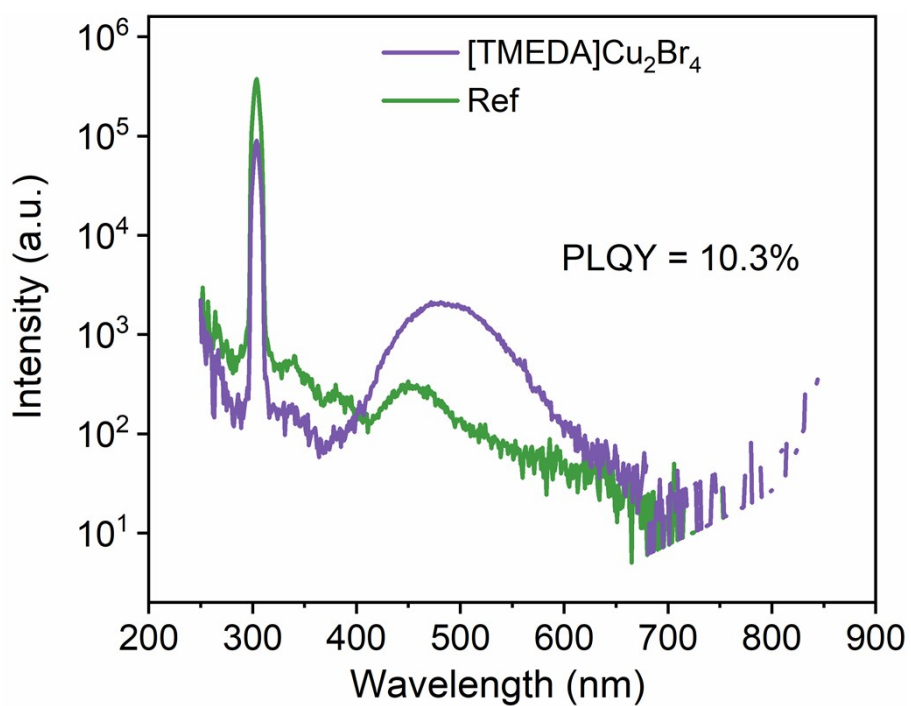


Fig. S14 The PLQY of [TMEDA]Cu₂Br₄.

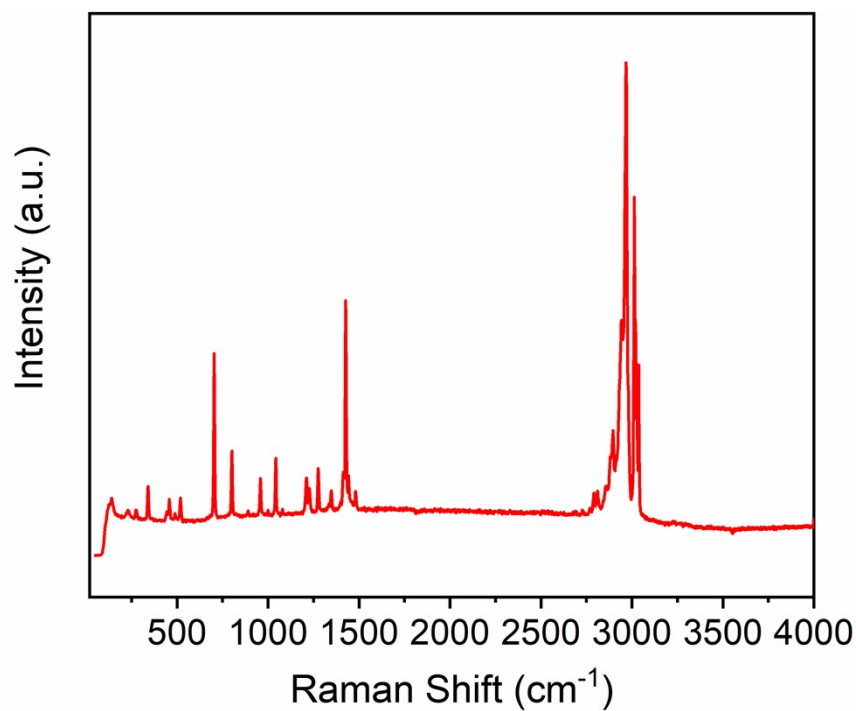


Fig. S15 Raman spectrum of [(Me)₄-Pipz]Cu₂Br₄.

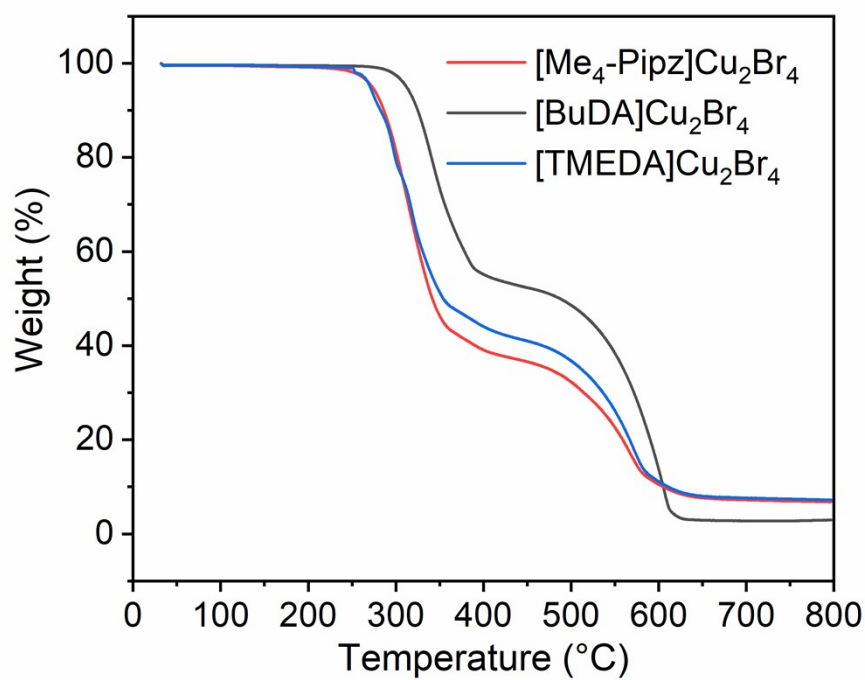


Fig. S16 The thermogravimetric analysis (TGA) curves of [(Me)₄-Pipz]Cu₂Br₄, [BuDA]Cu₂Br₄ and [TMEDA]Cu₂Br₄.

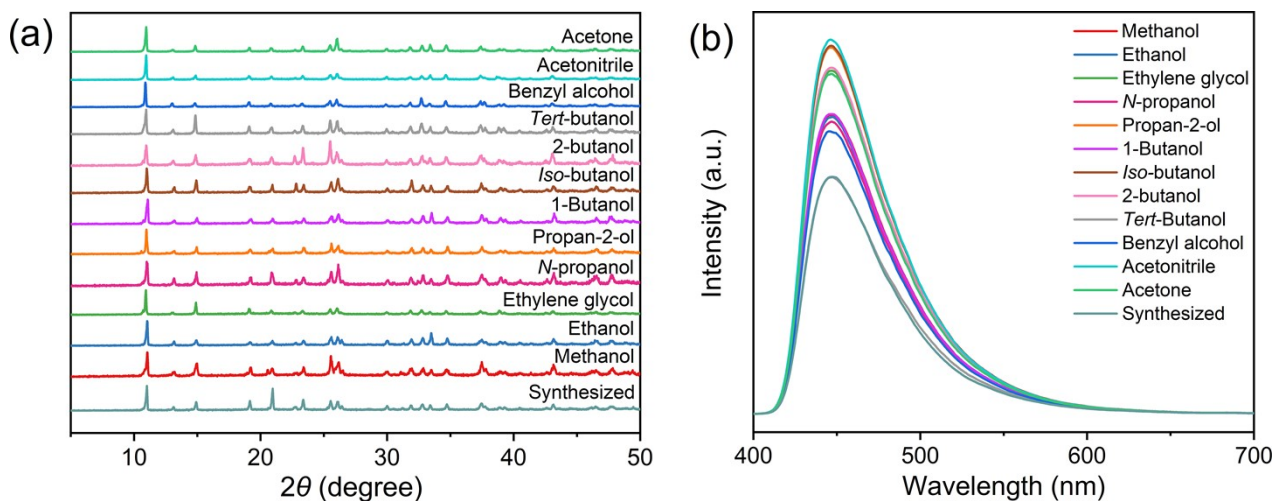


Fig. S17 PXR D patterns and PL emission spectra of [(Me)₄-Pipz]Cu₂Br₄ after immersing in various organic solvents.

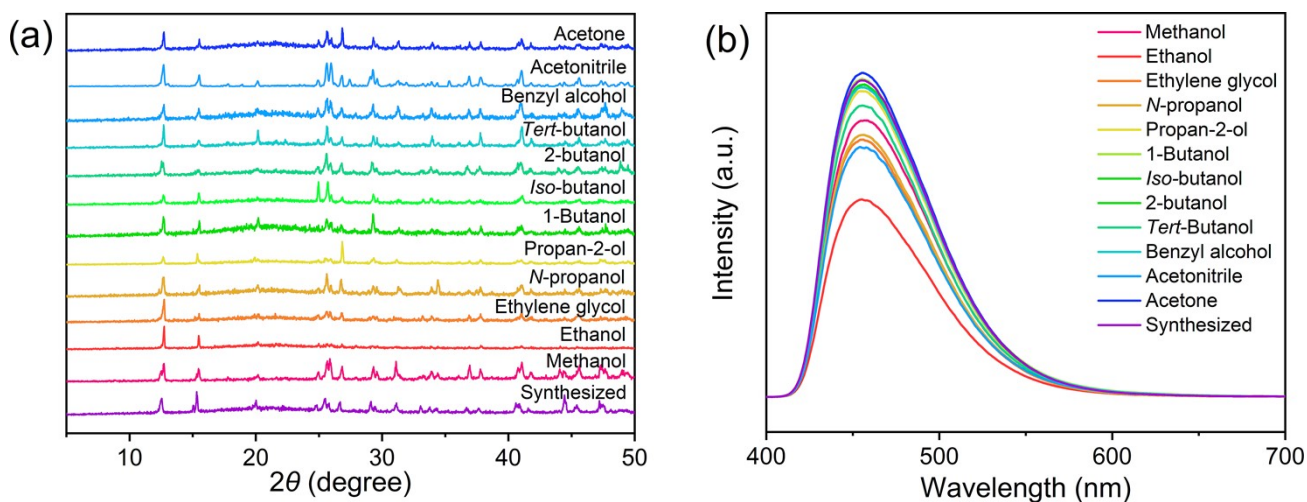


Fig. S18 PXR D patterns and PL emission spectra of [BuDA]Cu₂Br₄ after immersing in various organic solvents.

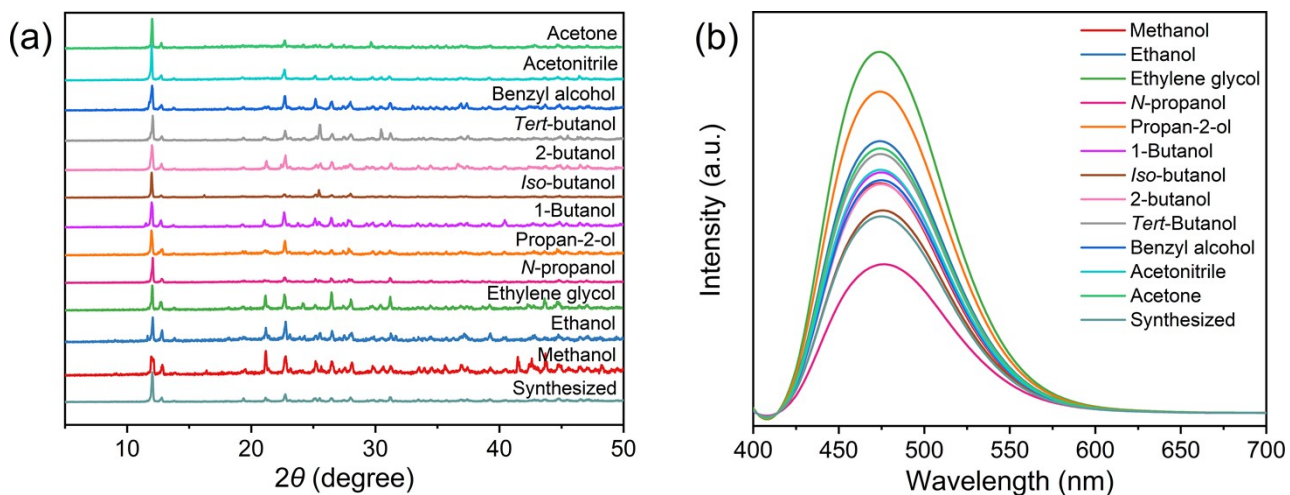


Fig. S19 PXRD patterns and PL emission spectra of [TMEDA]Cu₂Br₄ after immersing in various organic solvents.

Table S1. Crystal Data and Structural Refinements for [(Me)₄-Pipz]Cu₂Br₄.

Compound	[(Me) ₄ -Pipz]Cu ₂ Br ₄
Chemical formula	C ₈ N ₂ H ₂₀ Cu ₂ Br ₄
FW	295.49
Space group	<i>C2/m</i>
<i>a</i> /Å	13.848(4)
<i>b</i> /Å	6.7421(19)
<i>c</i> /Å	9.389(3)
α /°	90
β /°	121.275(2)
γ /°	90
<i>V</i> (Å ³)	749.2(4)
<i>Z</i>	4
<i>D</i> _{calcd} (g·cm ⁻³)	2.620
Temp (K)	296.15
μ (mm ⁻¹)	13.481
<i>F</i> (000)	560.0
Reflections collected	4263
Unique reflections	922
GOF on <i>F</i> ²	1.131
^a <i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0322/0.0834
^b <i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0361/0.0852

$${}^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|. \quad {}^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)]^{1/2}.$$

Table S2. Crystal Data and Structural Refinements for [BuDA]Cu₂Br₄.

Compound	[BuDA]Cu ₂ Br ₄
Chemical formula	C ₄ N ₂ H ₁₄ Cu ₂ Br ₄
FW	268.45
Space group	<i>P</i> 2 ₁ / <i>n</i>
<i>a</i> /Å	6.0905(3)
<i>b</i> /Å	8.8163(5)
<i>c</i> /Å	11.5467(6)
α /°	90
β /°	97.583(2)
γ /°	90
<i>V</i> (Å ³)	614.59(6)
<i>Z</i>	4
<i>D</i> _{calcd} (g·cm ⁻³)	2.901
Temp (K)	273.15
μ (mm ⁻¹)	16.419
<i>F</i> (000)	500.0
Reflections collected	6960
Unique reflections	1097
GOF on <i>F</i> ²	1.044
^a <i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0236/0.0488
^b <i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0330/0.0518

$${}^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|, \quad {}^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)]^{1/2}.$$

Table S3. Crystal Data and Structural Refinements for [TMEDA]Cu₂Br₄.

Compound	[TMEDA]Cu ₂ Br ₄
Chemical formula	C ₆ N ₂ H ₁₈ Cu ₂ Br ₄
FW	564.94
Space group	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> /Å	7.8140(7)
<i>b</i> /Å	14.6335(13)
<i>c</i> /Å	12.9812(11)
α /°	90
β /°	93.824(3)
γ /°	90
<i>V</i> (Å ³)	1481.0(2)
<i>Z</i>	4
<i>D</i> _{calcd} (g·cm ⁻³)	2.534
Temp (K)	273.15
μ (mm ⁻¹)	13.633
<i>F</i> (000)	1064.0
Reflections collected	16515
Unique reflections	2616
GOF on <i>F</i> ²	1.028
^a <i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0345/0.0656
^b <i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0682/0.0762

$${}^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|. \quad {}^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}.$$

Table S4. Hydrogen bonds data for [(Me)₄-Pipz]Cu₂Br₄.

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C1-H2...Br2 ¹	0.96	3.03	3.772(3)	135.5
C1-H3...Br2 ²	0.96	3.19	3.862(6)	128.3
C2-H4...Br1 ³	0.96	2.89	3.804(7)	159.2
C2-H5...Br1	0.96	2.83	3.765(3)	164.9
C2-H6...Br1 ⁴	0.96	2.89	3.765(3)	152.8
C3-H7...Br2 ⁵	0.97	3.18	3.808(4)	124.1
C3-H8...Br2 ⁶	0.97	2.97	3.633(4)	126.5

¹1/2+x, 1/2+y, +z; ²1-x, +y, 1-z; ³3/2-x, 1/2+y, 2-z; ⁴+x, 1+y, +z; ⁵1/2+x, -1/2+y, +z; ⁶3/2-x, -1/2+y, 2-z