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Supplementary Information

Approach to Anionic Alkaline Earth Metal-Methanesulfonates Bearing Photo-responsive Methylviologen

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Table S1. Crystallographic data of 1–4.

	1	2	3	4
Empirical formula CCDC no.	$\begin{array}{c} C_{18}H_{40}Mg_2N_2O_{22}S_6\\ 2358180 \end{array}$	$\begin{array}{c} C_{18}H_{40}Ca_2N_2O_{22}S_6\\ 2362824 \end{array}$	C ₉ H ₂₄ MgNO ₁₃ S ₃ 2358181	C ₇ H ₂₁ CaNO ₁₀ S ₃ 2362825
Formula weight	877.50	909.04	474.78	415.51
Temperature (k)	298(2)	298(2)	294(2)	296(2)
Wavelength (Å)	0.71073	0.71073	0.71073	0.71073
Crystal system	Triclinic	Triclinic	Triclinic	Monoclinic
Space group	$P\overline{1}$	$P\overline{1}$	$P\overline{1}$	P_{21}/c
a [Å]	5.7456(8)	5.8064(6)	5.7797(4)	5.80060(10)
b [Å]	9.3568(12)	9.5980(11)	9.7473(7)	31.3766(9)
c [Å]	16.623(2)	16.929(2)	17.7172(12)	9.1840(3)
α [°]	88.264(4)	88.105(4)	96.331(2)	90
β [°]	85.531(4)	84.771(4)	93.814(2)	93.5180(10)
γ [°]	79.245(4)	79.178(4)	100.767(2)	90
V [Å ³]	875.2(2)	922.67(18)	970.69(12)	1668.37(8)
Z	1	1	2	4
d_{calcd} [Mg/m ³]	1.665	1.636	1.624	1.654
$\mu [\mathrm{mm}^{-1}]$	0.515	0.732	0.478	0.797
F (000)	458	474	498	872
Crystal size [mm ³]	0.063 x 0.061 x	0.072 x 0.069 x	0.078 x 0.075 x	0.073 x 0.071 x
	0.058	0.066	0.071	0.068
Theta range for	2.216 to 28.314	2.161 to 28.356	2.144 to 28.318	2.315 to 28.272
data collection (°)				
Reflections collected	33995	33514	33196	39524
Independent	4350 [R(int) =	4603 [R(int) =	4827 [R(int) =	4130 [R(int) =
reflections	0.0482]	0.0381]	0.0493]	0.1301]
Goodness-of-fit on F^2	1.042	1.055	1.568	1.006
Completeness to	100.0	100	100	100
theta = 25.242° (%)				
Max. and min.	0.6973 and 0.7457	0.7100 and	0.6873 and	0.6369 and
transmission		0.7457	0.7457	0.7457
Data / restraints / parameters	4350 / 0 / 246	4603 / 0 / 246	4827 / 0 / 281	4103 / 0 / 213
Final R indices	R1 = 0.0334, wR2	R1 = 0.0346, wR2	R1 = 0.0563,	R1 = 0.0499,
[I>2sigma(I)]	= 0.0778	= 0.0938	wR2 = 0.1423	wR2 = 0.1333
R indices (all data)	R1 = 0.0452, wR2	R1 = 0.0374, wR2	R1 = 0.0579,	R1 = 0.0516,
	= 0.0841	= 0.0956	wR2 = 0.1439	wR2 = 0.1355
Largest diff. peak and hole (e.Å ⁻³)	0.341 and -0.373	0.493 and -0.722	1.103 and 0.963	0.957 and - 0.812

Bond lengths [Å]		Bond angles [°]	
Mg(1)-O(1)	2.0812(14)	O(1)-Mg(1)-O(5)#1	89.22(6)
Mg(1)-O(4)	2.0648(15)	O(4)-Mg(1)-O(1)	175.32(7)
Mg(1)-O(5)	2.0936(16)	O(4)-Mg(1)-O(7)	94.09(7)
Mg(1)-O(7)	2.0671(15)	O(7)-Mg(1)-O(1)	85.24(6)
Mg(1)-O(10)	2.0438(16)	O(7)-Mg(1)-O(5)#1	173.40(7)
Mg(1)-O(11)	2.0542(16)	O(10)-Mg(1)-O(1)	95.75(7)
		O(10)-Mg(1)-O(4)	88.88(7)
		O(10)-Mg(1)-O(11)	172.19(7)
		O(11)-Mg(1)-O(1)	90.95(6)
		O(11)-Mg(1)-O(4)	84.48(7)

Table S2. Selected bond lengths [Å] and angles $[\circ]$ for 1.

Selected Hydrogen bond lengths [Å] and angles [°].

D-HA	d(D-H)	d(HA)	<dha< th=""><th>d(DA)</th></dha<>	d(DA)
O10-H10BO6	0.741	2.16	166.53	2.886
O10-H10AO9	0.812	1.938	166.6	2.735
O11-H11AO3	0.794	1.962	166.99	2.741
O11-H11BO6	0.829	2.256	159.87	3.047
С5-Н5О8	0.93	2.352	170.61	3.273
С7-Н7О8	0.93	2.3	172.85	3.225
C4-H4O2	0.93	2.248	159.62	3.136

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+1,-z+1 #2 -x+1,-y+2,-z+2

Bond lengths [Å]		Bond angles [°]	
Ca(1)-O(1)	2.3245(11)	O(1)-Ca(1)-O(5)#1	87.73(5)
Ca(1)-O(4)	2.2989(12)	O(1)-Ca(1)-O(10)	98.20(6)
Ca(1)-O(5)#1	2.3321(15)	O(1)-Ca(1)-O(11)	89.23(5)
Ca(1)-O(7)	2.2940(13)	O(4)-Ca(1)-O(1)	84.15(5)
Ca(1)-O(10)	2.3253(14)	O(4)-Ca(1)-O(5)#1	94.88(6)
Ca(1)-O(11)	2.3355(14)	O(4)-Ca(1)-O(10)	88.20(5)
		O(4)-Ca(1)-O(11)	95.82(6)
		O(7)-Ca(1)-O(1)	174.66(5)
		O(7)-Ca(1)-O(4)	93.51(6)
		O(7)-Ca(1)-O(10)	86.50(6)
		O(7)-Ca(1)-O(11)	86.22(6)
		O(10)-Ca(1)-O(11)	171.89(6)
		O(7)-Ca(1)-O(1) $O(7)-Ca(1)-O(4)$ $O(7)-Ca(1)-O(10)$ $O(7)-Ca(1)-O(11)$ $O(10)-Ca(1)-O(11)$	174.66(5) 93.51(6) 86.50(6) 86.22(6) 171.89(6)

Table S3. Selected bond lengths [Å] and angles [°] for 2.

Hydrogen bond lengths [Å] and angles [°] for 2.

D-HA	d(D-H)	d(HA)	<dha< th=""><th>d(DA)</th></dha<>	d(DA)
O10-H10BO6	0.933	1.959	173.58	2.888
O10-H10AO9	0.767	2.001	166.93	2.754
O11-H11AO3	0.872	1.893	175.9	2.764
O11-H11BO6	0.782	2.391	159.86	3.137
C4-H4O2	0.93	2.269	161.65	3.165
С5-НАО8	0.93	2.309	171.7	3.232
С7-Н7О8	0.93	2.374	176.93	3.303

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,-z+1 #2 -x,-y+2,-z+2

Bond lengths [Å]		Bond angles [°]	
Mg(1)-O(1)	2.0551(13)	O(1)-Mg(1)-O(6)	88.79(5)
Mg(1)-O(6)	2.0656(12)	O(1)-Mg(1)-O(10)	89.84(6)
Mg(1)-O(10)	2.0759(12)	O(1)-Mg(1)-O(11)	94.19(6)
Mg(1)-O(11)	2.0740(13)	O(1)-Mg(1)-O(12)	178.88(6)
Mg(1)-O(12)	2.0947(13)	O(6)-Mg(1)-O(10)	91.35(5)
Mg(1)-O(13)	2.0406(13)	O(6)-Mg(1)-O(11)	177.02(6)
		O(10)-Mg(1)-O(12)	91.20(6)
		O(11)-Mg(1)-O(10)	88.57(6)
		O(11)-Mg(1)-O(12)	85.43(6)
		O(13)-Mg(1)-O(10)	175.60(6)

Table S4. Selected bond lengths [Å] and angles [°] for 3.

Hydrogen bond lengths [Å] and angles [°].

D-HA	d(D-H)	d(HA)	<dha< td=""><td>d(DA)</td></dha<>	d(DA)
O10-H10BO3	0.829	1.933	167.92	2.749
O10-H10AO7	0.76	2.011	166.76	2.756
O11-H11AO7	0.751	2.033	176.25	2.782
O11-H11BO8	0.737	2.123	155.02	2.808
O12-H12BO8	0.743	2.138	162.20	2.855
O12-H12AO9	0.769	2.061	176.08	2.829
O13-H13BO5	0.847	1.911	159.77	2.722
O13-H13AO10	0.79	2.089	161.64	2.85
С8-Н8О2	0.93	2.281	171.99	3.204

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,-z+1 #2 -x,-y+2,-z+2

Bond lengths [Å]		Bond angles [°]	
Ca(1)-O(1)	2.3193(14)	O(1)-Ca(1)-O(2)#1	88.96(6)
Ca(1)-O(2)#1	2.3307(14)	O(1)-Ca(1)-O(8)#2	94.61(6)
Ca(1)-O(4)	2.2905(16)	O(1)-Ca(1)-O(10)	83.05(6)
Ca(1)-O(7)	2.2979(15)	O(2)#1-Ca(1)-O(8)#2	85.26(6)
Ca(1)-O(8)#2	2.3327(14)	O(4)-Ca(1)-O(1)	89.17(6)
Ca(1)-O(10)	2.3688(16)	O(4)-Ca(1)-O(7)	88.32(7)
		O(4)-Ca(1)-O(8)#2	93.55(6)
		O(4)-Ca(1)-O(10)	96.99(7)
		O(7)-Ca(1)-O(1)	167.82(6)
		O(7)-Ca(1)-O(10)	85.42(7)
		O(7)-Ca(1)-O(8)#2	97.44(6)
		O(8)#2-Ca(1)-O(10)	169.16(6)

Table S5. Selected bond lengths $[{\rm \AA}]$ and angles $[^{\circ}]$ for 4.

Hydrogen bond lengths [Å] and angles [°].

D-HA	d(D-H)	d(HA)	<dha< th=""><th>d(DA)</th></dha<>	d(DA)
O10-H10AO3	0.723	2.205	169.73	2.919
O10-H10BO3	0.742	2.322	168.77	3.053

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,-z+1 #2 -x,-y+2,-z+2



Figure S1. Powder X-Ray Diffraction pattern of 1-4: black(simulated), red (primitive sample).

Figure S2. ¹H NMR spectrum of $[MV{Mg(OSO_2Me)_3(H_2O)_2}_2]$ (1) in D₂O.



Figure S3. ¹H NMR spectrum of [MV{Ca(OSO₂Me)₃(H₂O)₂}₂] (2) in D₂O.



Figure S4. ¹H NMR spectrum of $[\{Mg(OSO_2Me)_2(H_2O)_4\}MV \cdot 2OSO_2Me]$ (3) in D₂O.



Figure S5. ¹H NMR spectrum of $[Me_4\text{-pip}{Ca_2(OSO_2Me)_6(H_2O)_2}]_n$ (4) in D₂O.



Figure S6. FT-IR spectrum of 1-4.



FT-IR spectrum of 1(red), 2(black) and 3(green).



FT-IR spectrum of 4.

Figure S7. TGA profiles of 1-4.











All hydrogen atoms except for water are omitted for clarity.

Figure S10. Extended structure of 4 along the *bc*-plane.







Figure S12. UV-vis diffused reflectance spectra of 2 and 3 under ambient light and Hg lamp.



Figure S13. UV-vis spectrum of 2 in methanol.

