

Supporting Information for:

Development of Thiacrown Ligands for Encapsulation of Mercury-197m/g into Radiopharmaceuticals

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NMR Characterization

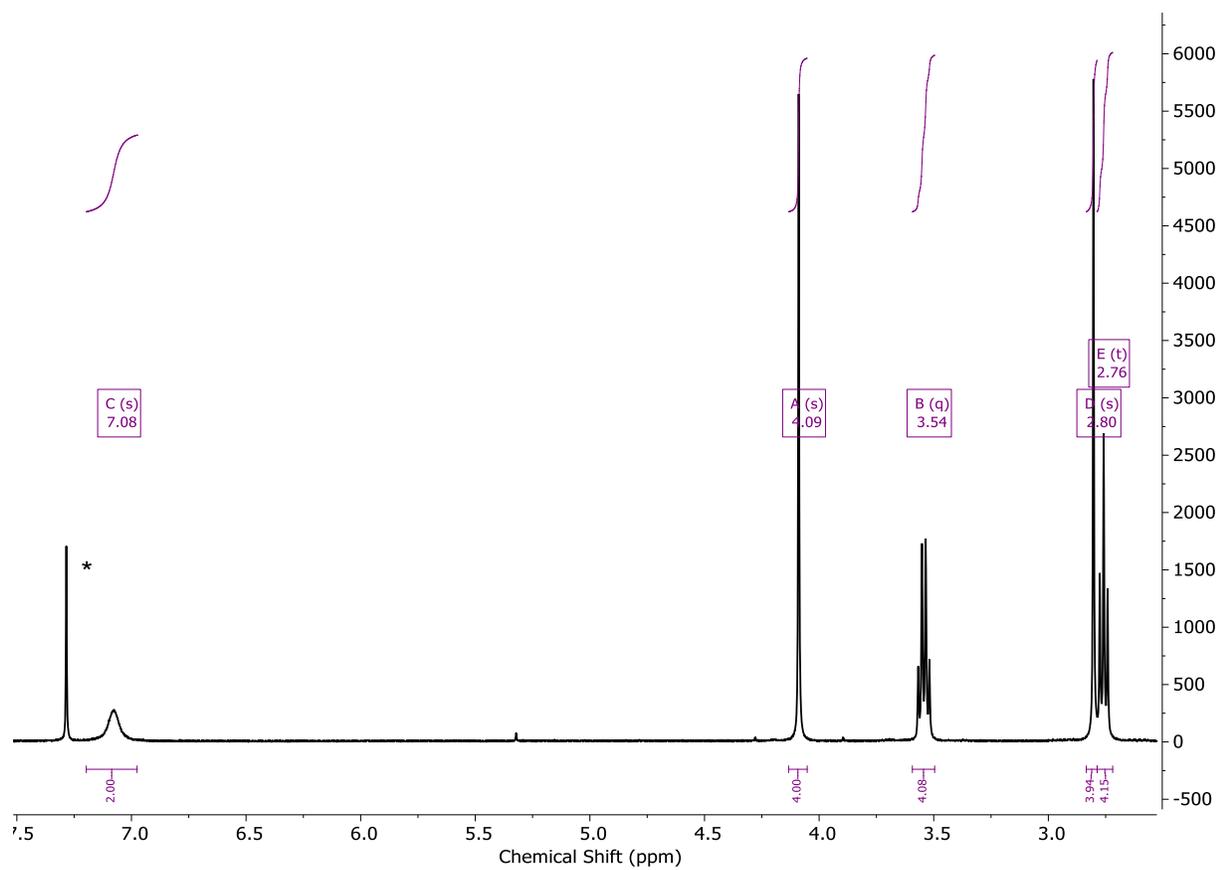


Figure S1. ¹H NMR (400 MHz, CDCl₃*, 25 °C): compound 1.

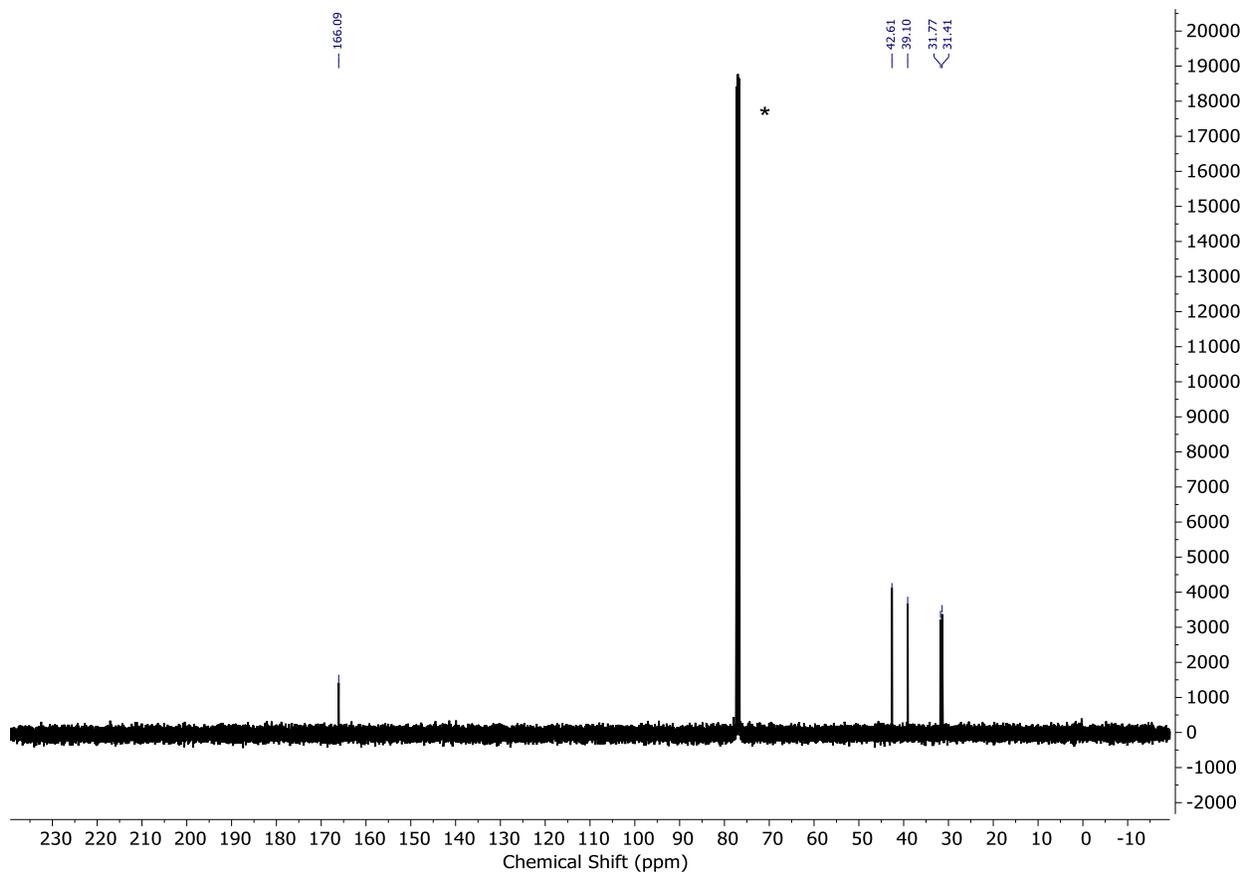


Figure S2. ¹H NMR (400 MHz, CDCl₃*, 25 °C): compound **1**.

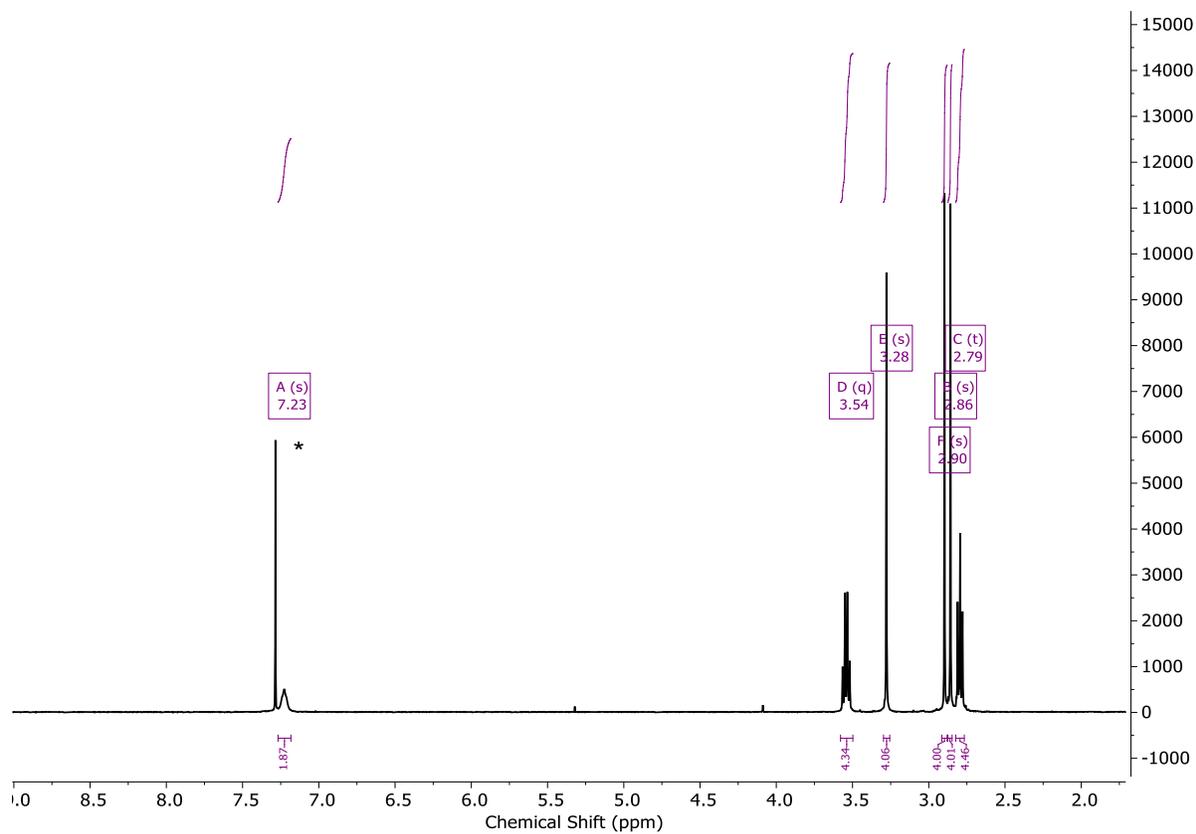


Figure S3. ¹H NMR (400 MHz, CDCl₃*, 25 °C): compound 2.

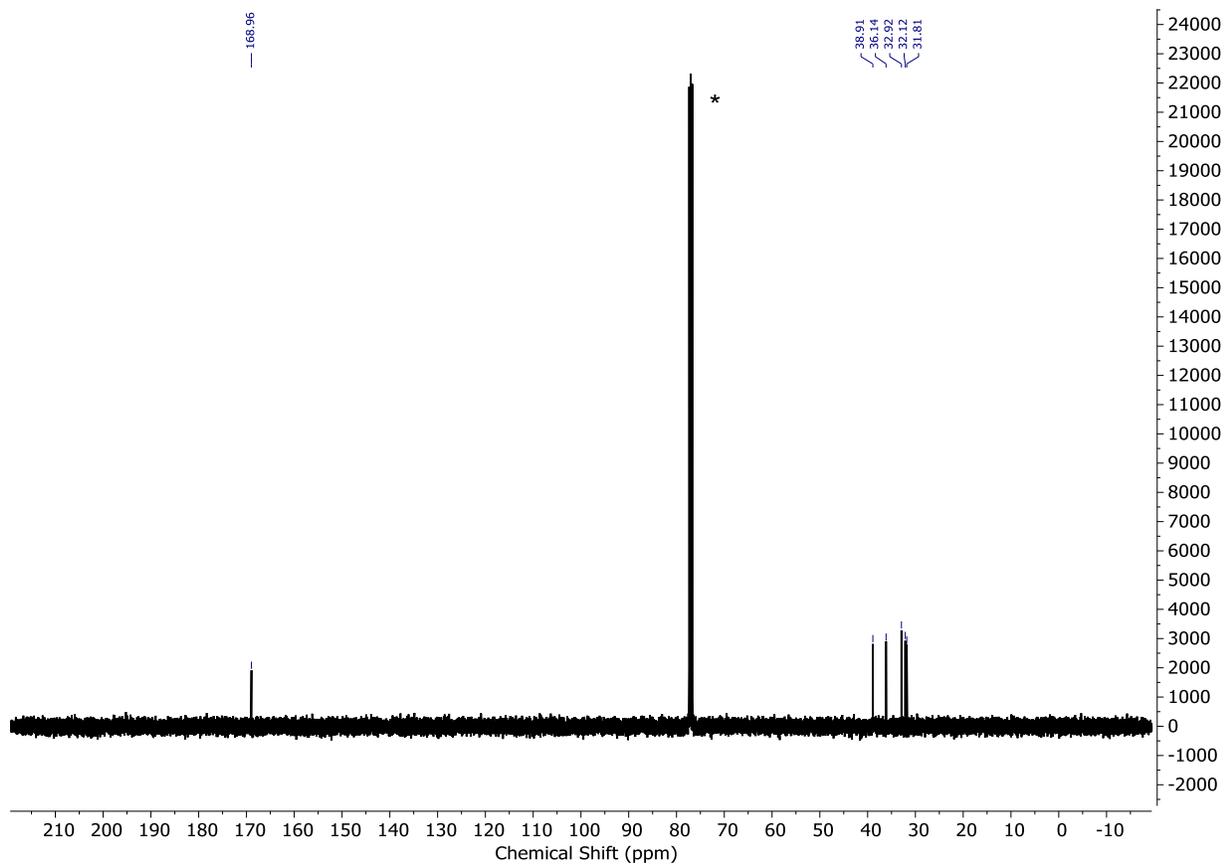


Figure S4. $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3^* , 25 °C): compound 2.

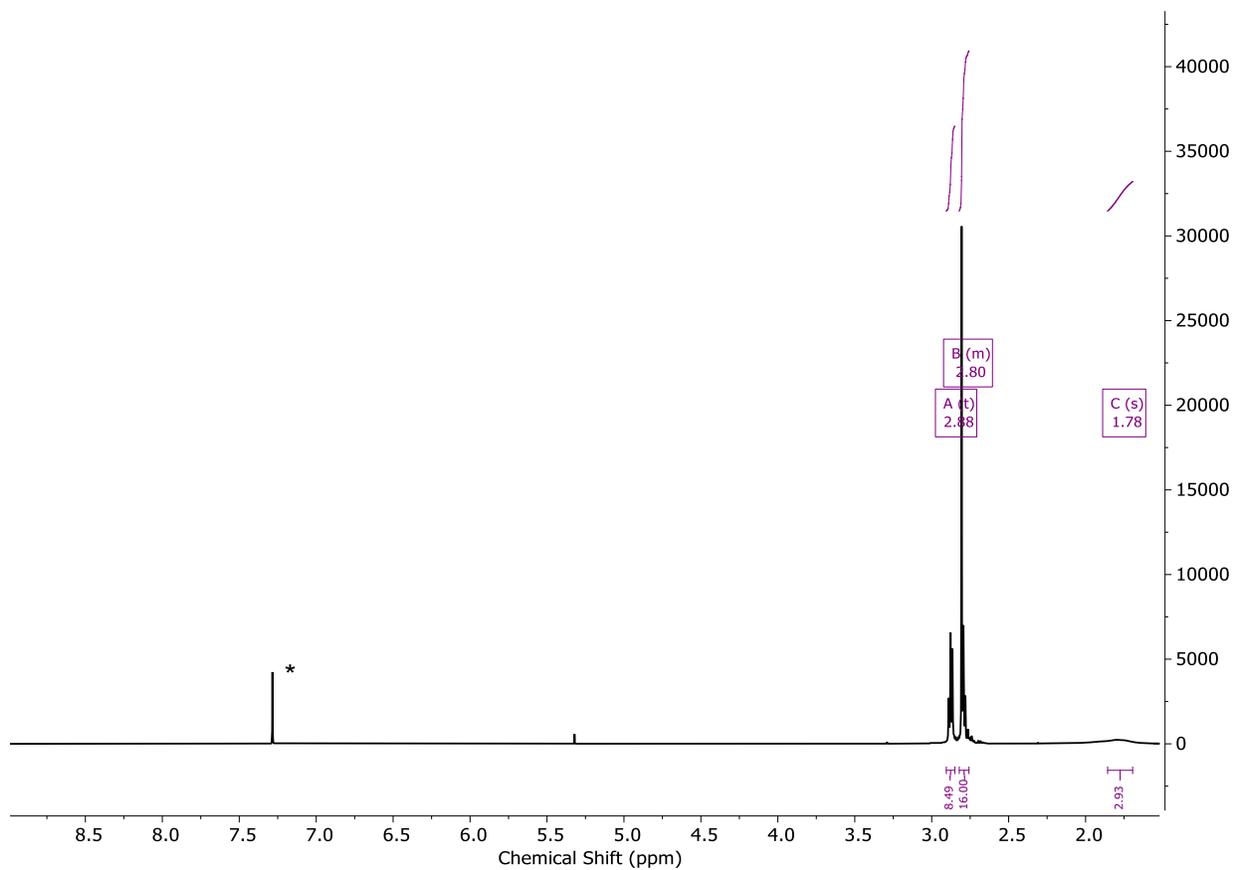


Figure S5. ¹H NMR (400 MHz, CDCl₃*, 25 °C): compound **3**.

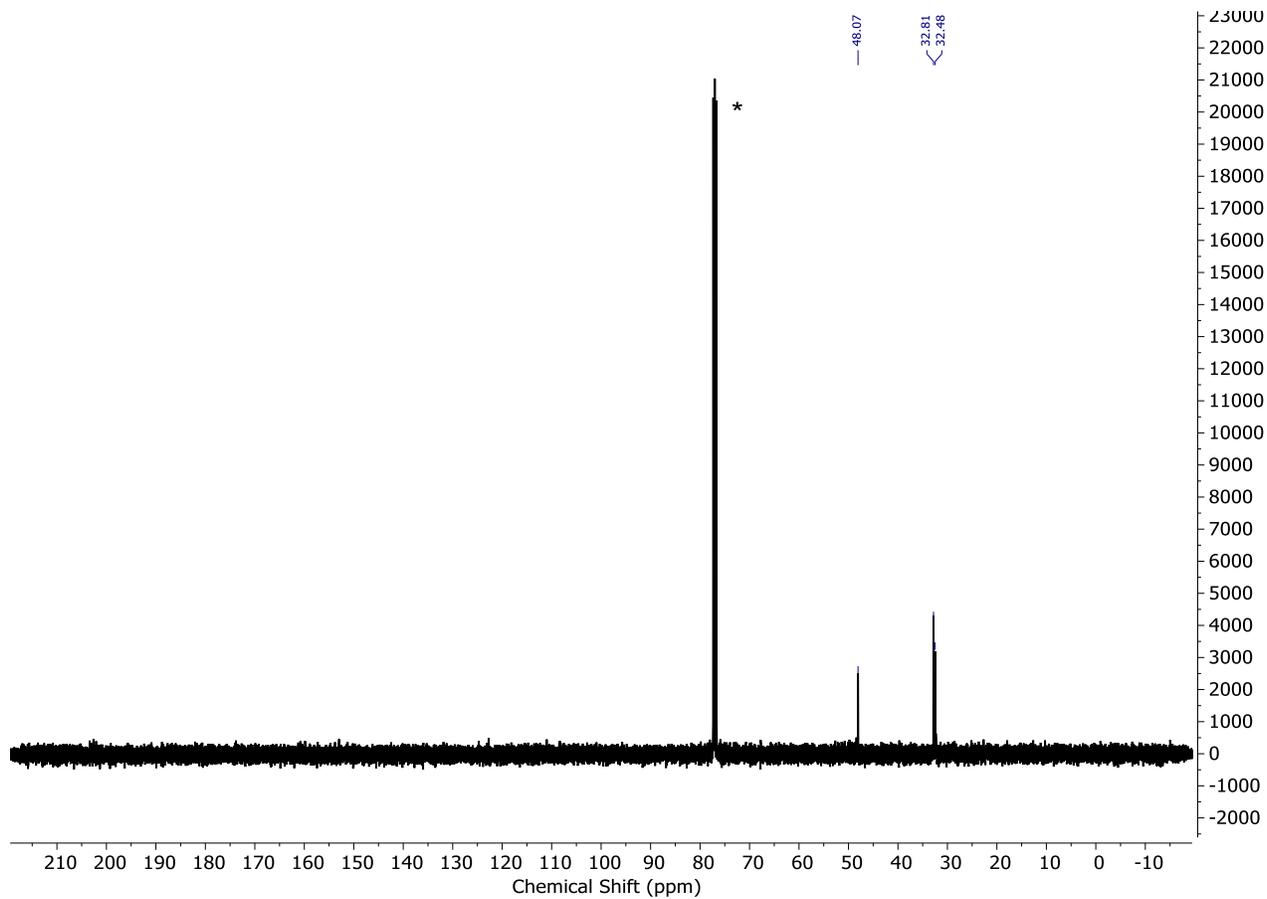


Figure S6. $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, CDCl_3^* , 25 °C): compound **3**.

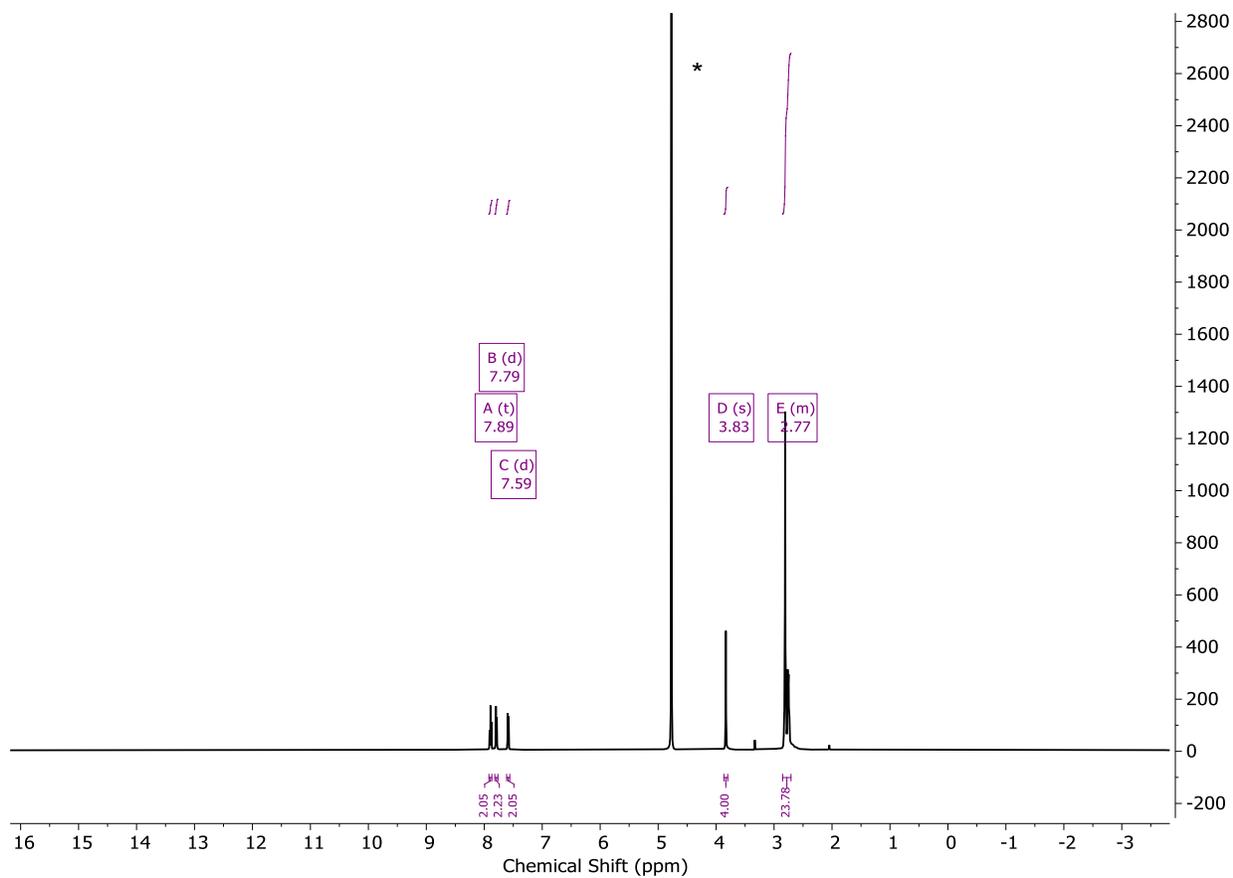


Figure S7. ^1H NMR (500 MHz, D_2O^* , 25 $^\circ\text{C}$): $\text{N}_2\text{S}_4\text{-H}_2\text{Pa}$.

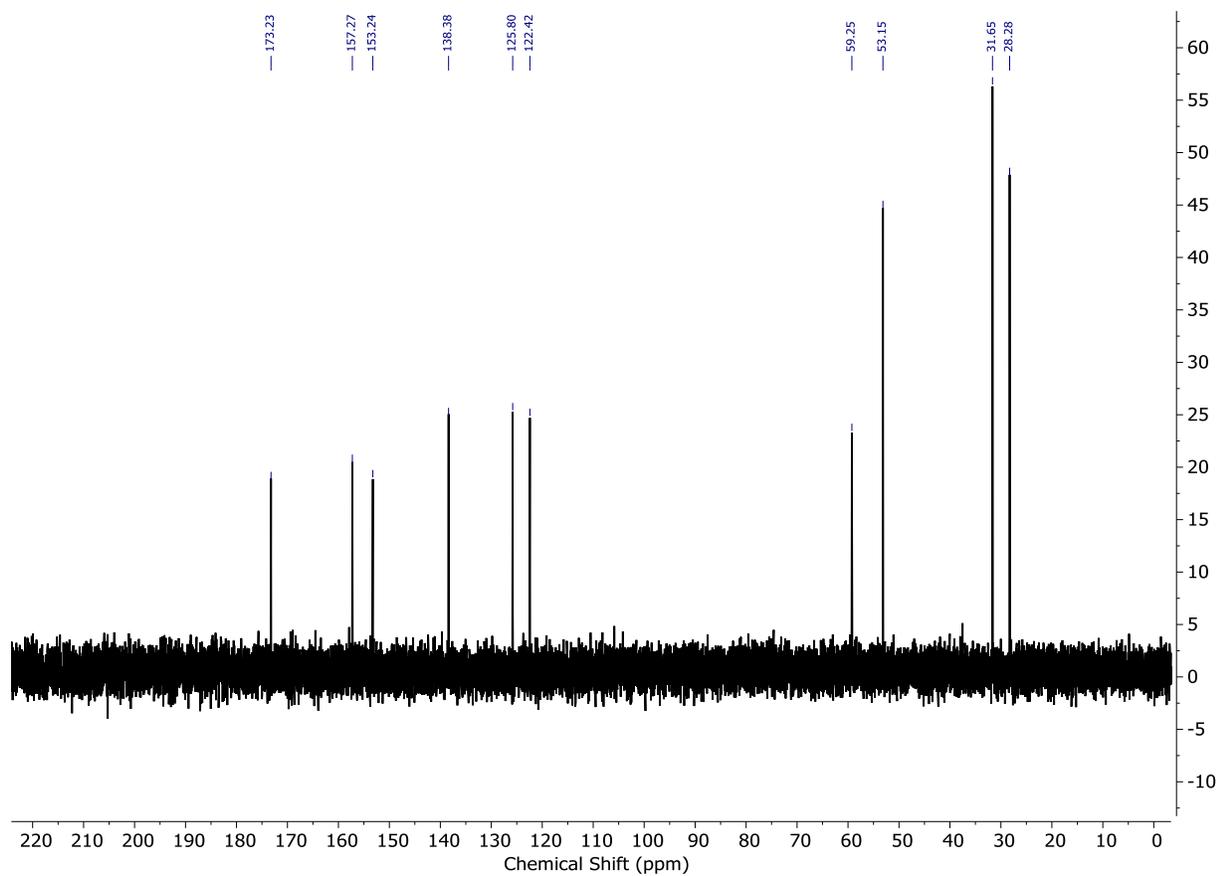


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, D_2O , $25\text{ }^\circ\text{C}$): $\text{N}_2\text{S}_4\text{-H}_2\text{Pa}$.

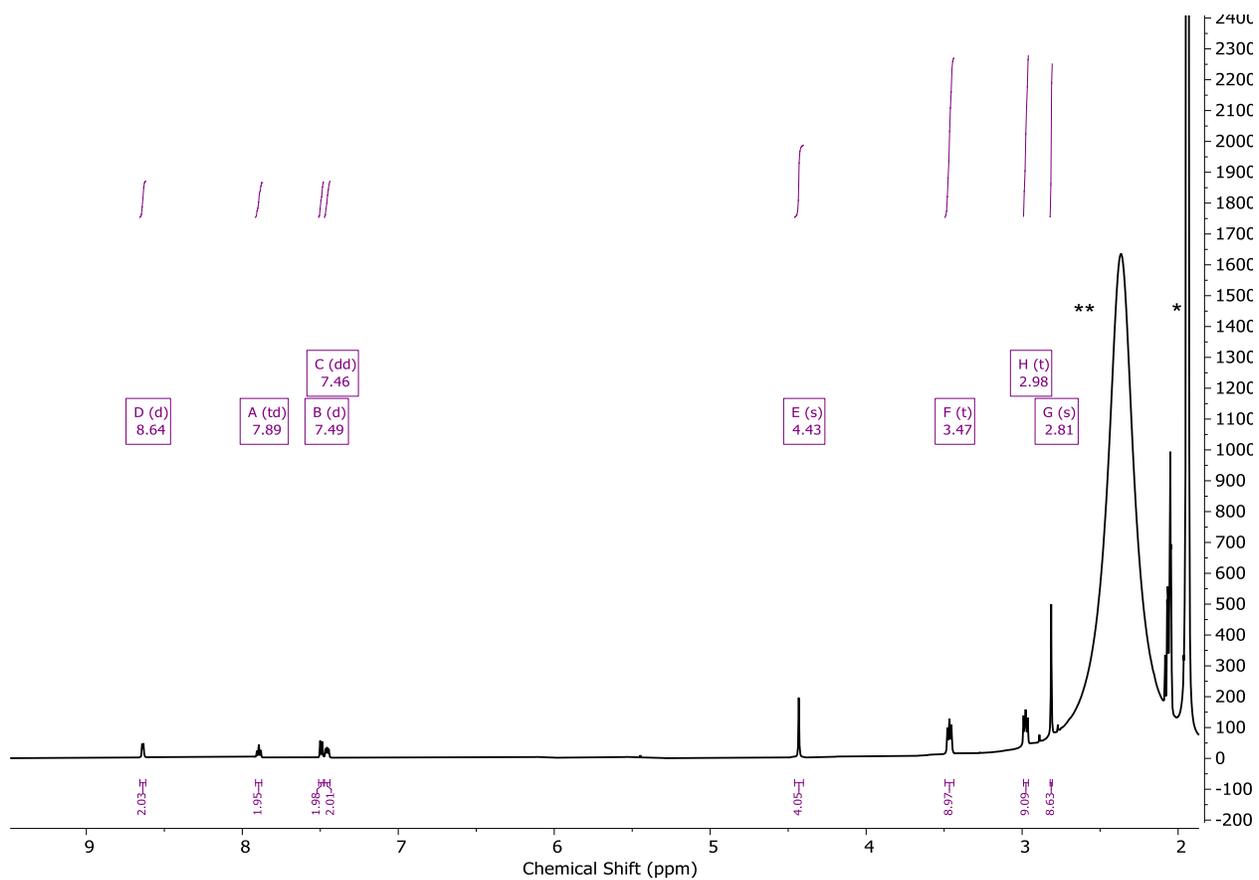


Figure S9. ^1H NMR (600 MHz, CD_3CN^* , $25\text{ }^\circ\text{C}$): $\text{N}_2\text{S}_4\text{-Py}$. (** H_2O)

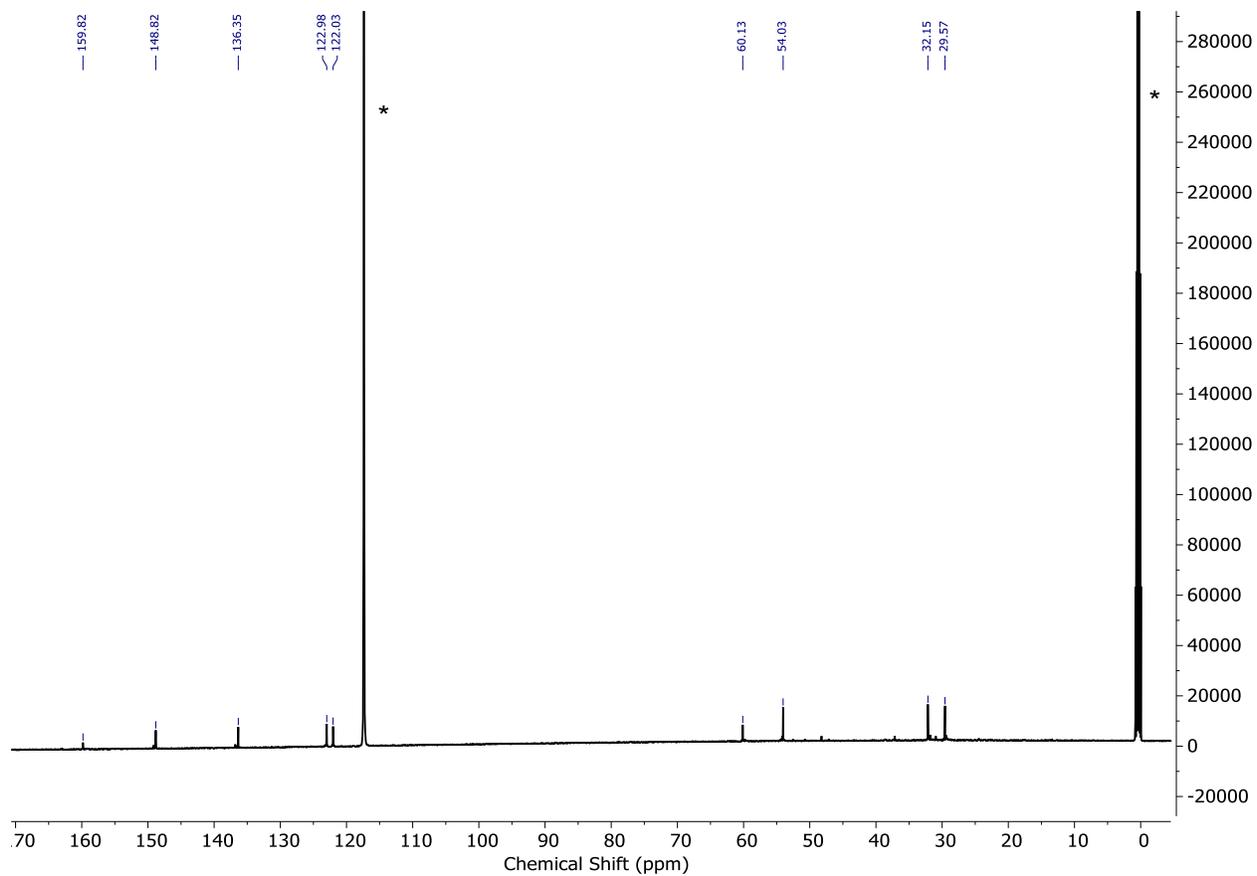


Figure S10. $^{13}\text{C}\{^1\text{H}\}$ NMR (151 MHz, CD_3CN^* , 25 °C): $\text{N}_2\text{S}_4\text{-Py}$.

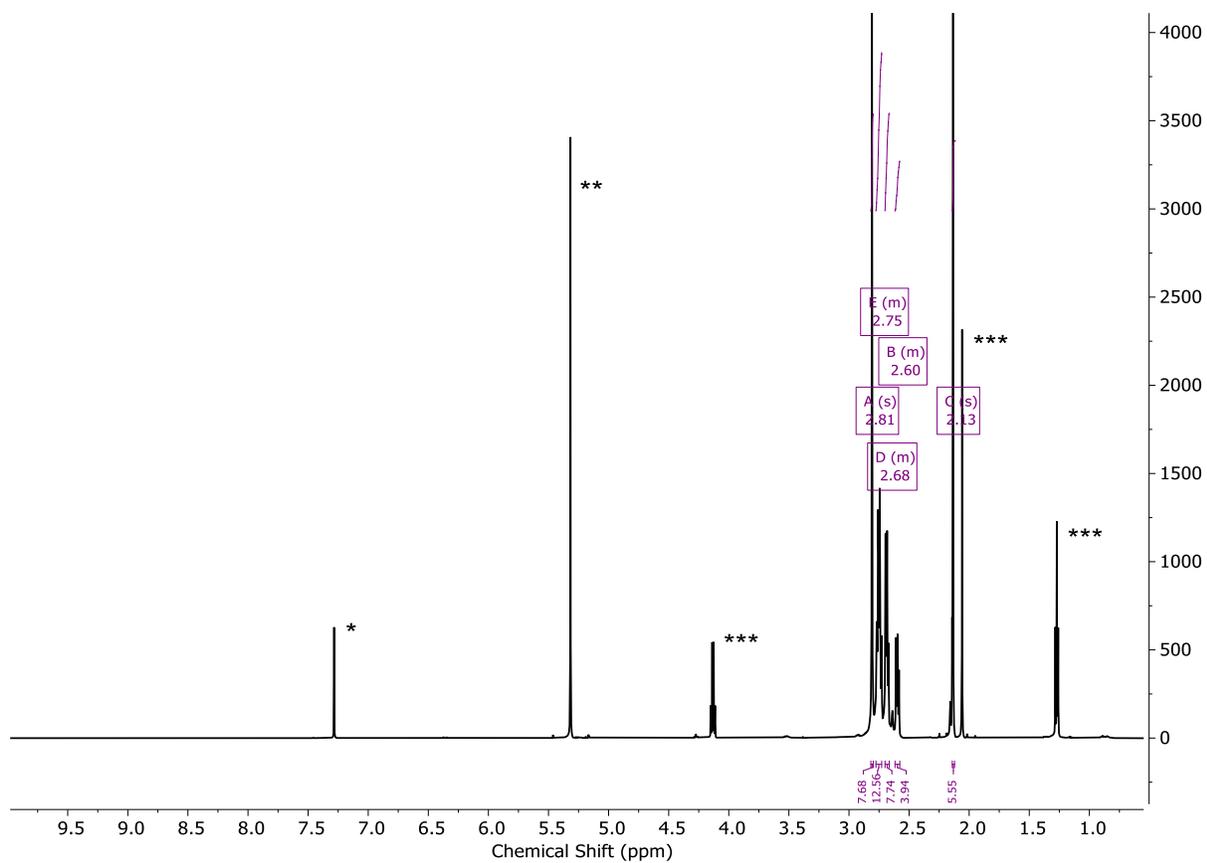


Figure S11. ¹H NMR (600 MHz, CDCl₃^{*}, 25 °C): N₂S₄-Thio. (**CH₂Cl₂, ***ethyl acetate)

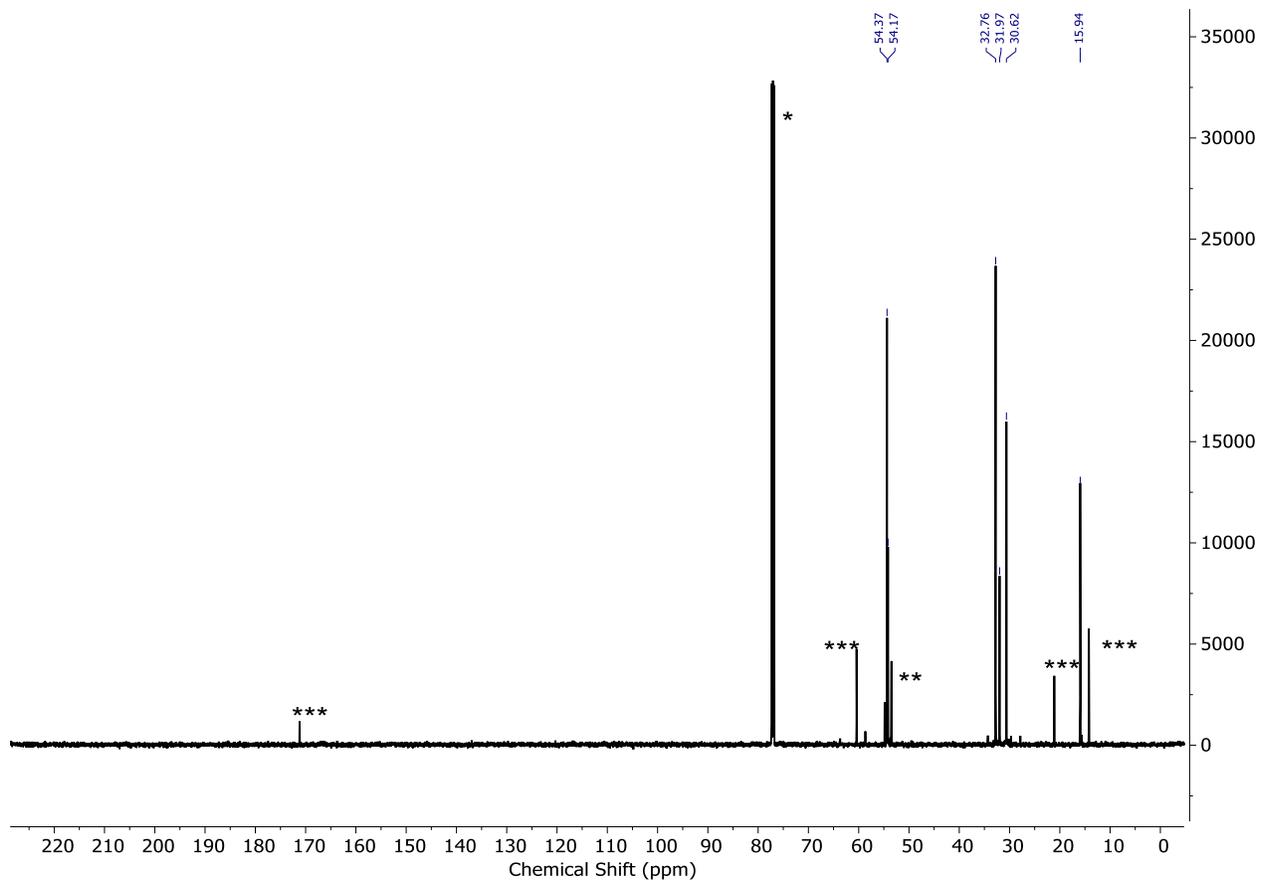


Figure S12. $^{13}\text{C}\{^1\text{H}\}$ NMR (151 MHz, CDCl_3 , 25 °C): $\text{N}_2\text{S}_4\text{-Thio}$. (** CH_2Cl_2 , ***ethyl acetate)

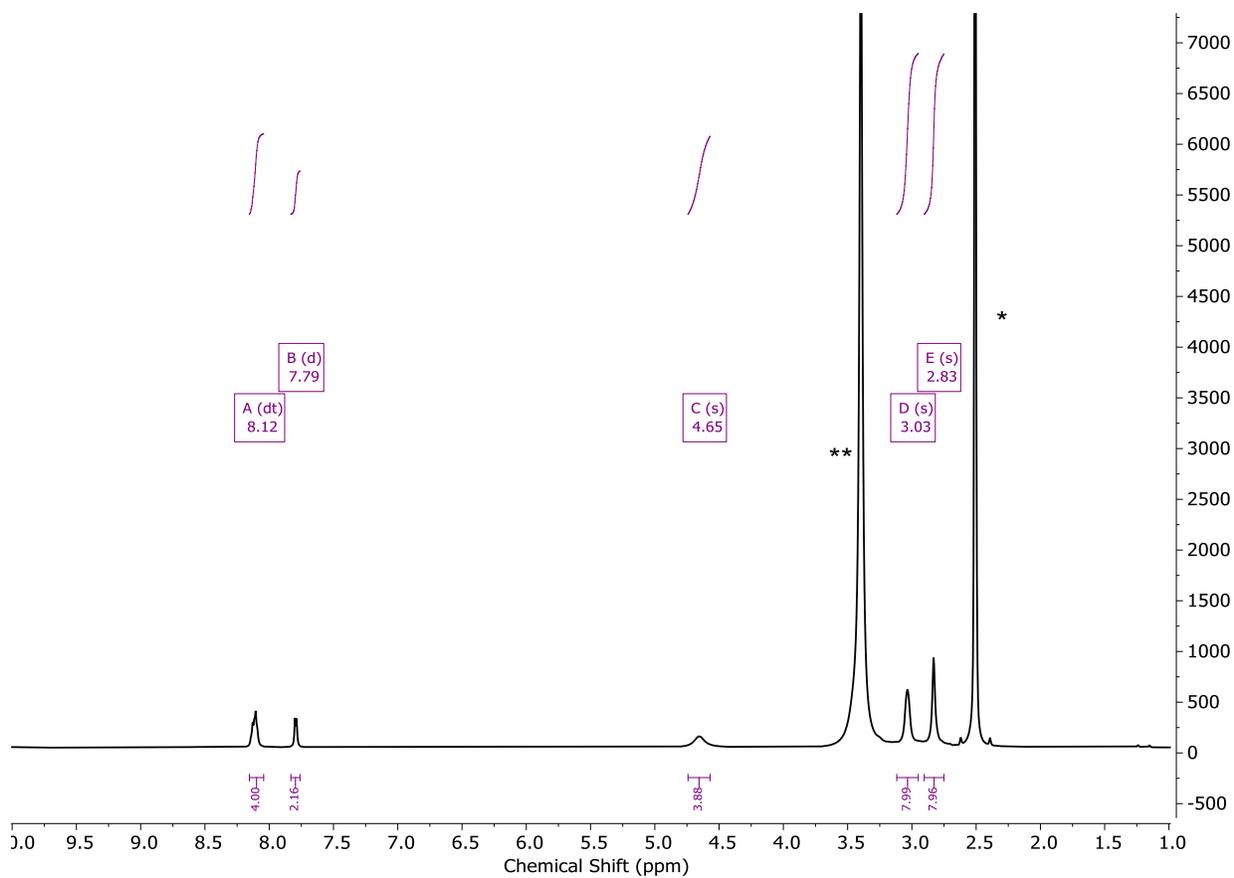


Figure S13. ^1H NMR (600 MHz, $\text{DMSO}-d_6^*$, 25 °C): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Pa})]$ at 24 h. (** H_2O)

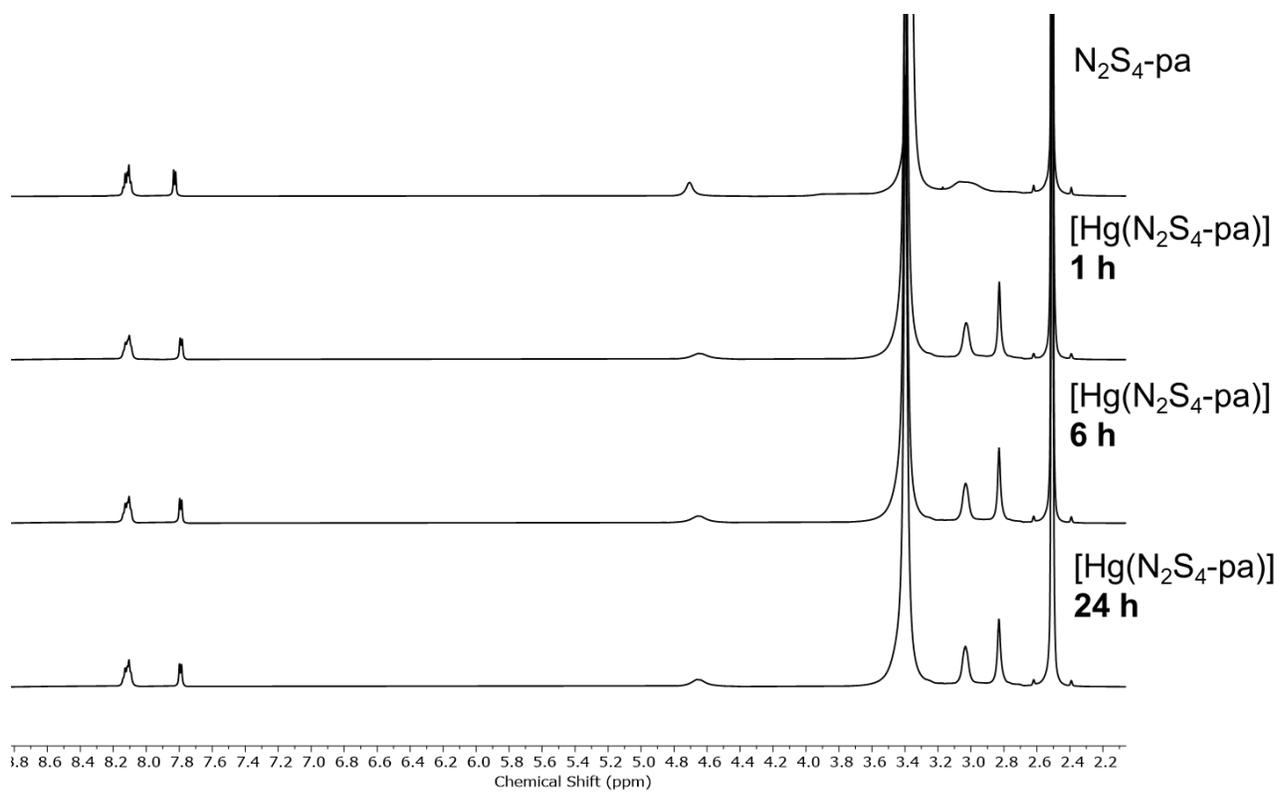


Figure S14. ^1H NMR (600 MHz, $\text{DMSO-}d_6$, 25 $^\circ\text{C}$): $\text{N}_2\text{S}_4\text{-Pa}$ and $[\text{natHg}(\text{N}_2\text{S}_4\text{-Pa})]$ formation over time.

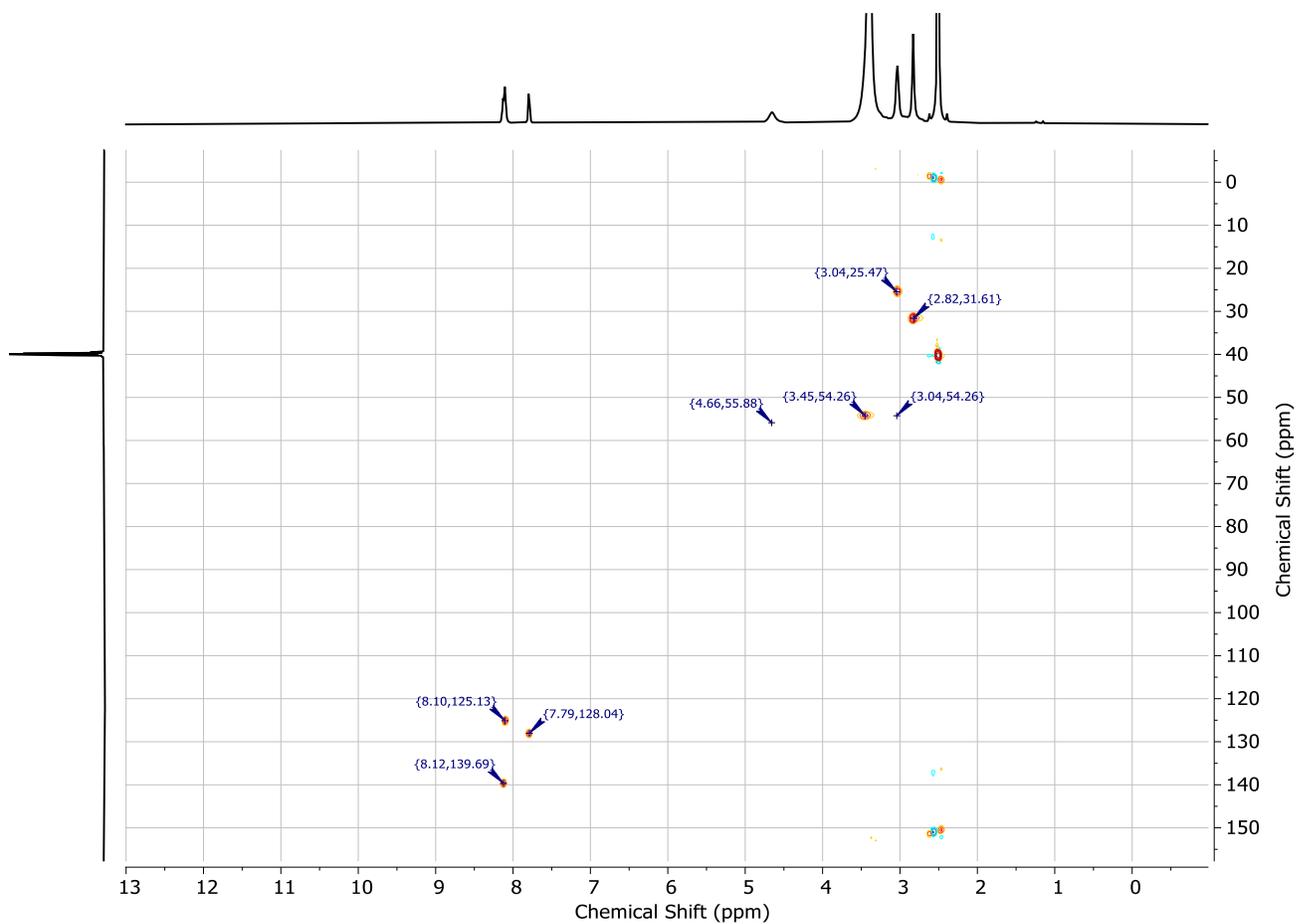


Figure S15. ^1H - ^{13}C HSQC NMR(600 MHz- 161 MHz, $\text{DMSO-}d_6$, $25\text{ }^\circ\text{C}$): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Pa})]$.

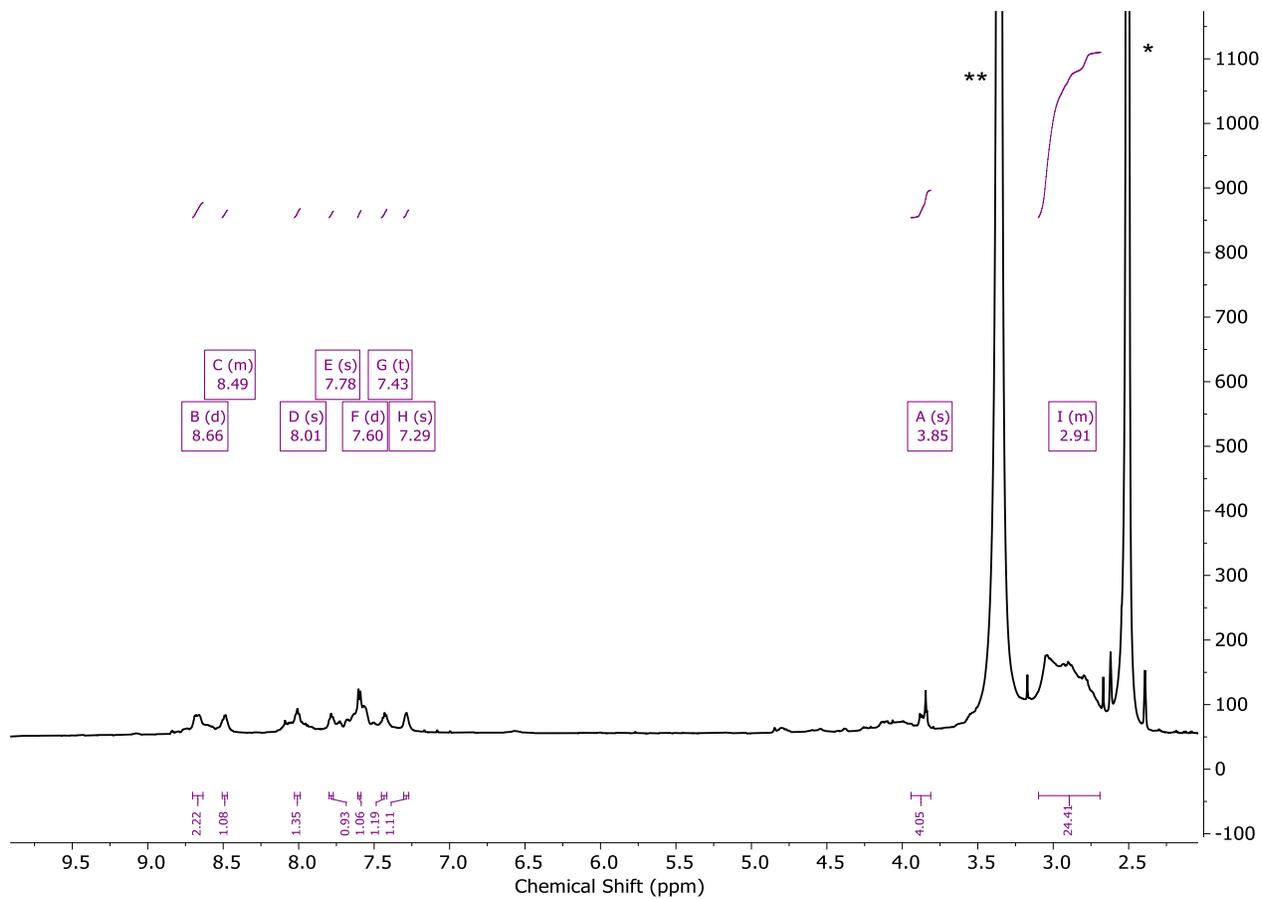


Figure S16. ^1H NMR (600 MHz, $\text{DMSO-}d_6^*$, 25 °C): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Py})]^{2+}$ at 24 h. (** H_2O)

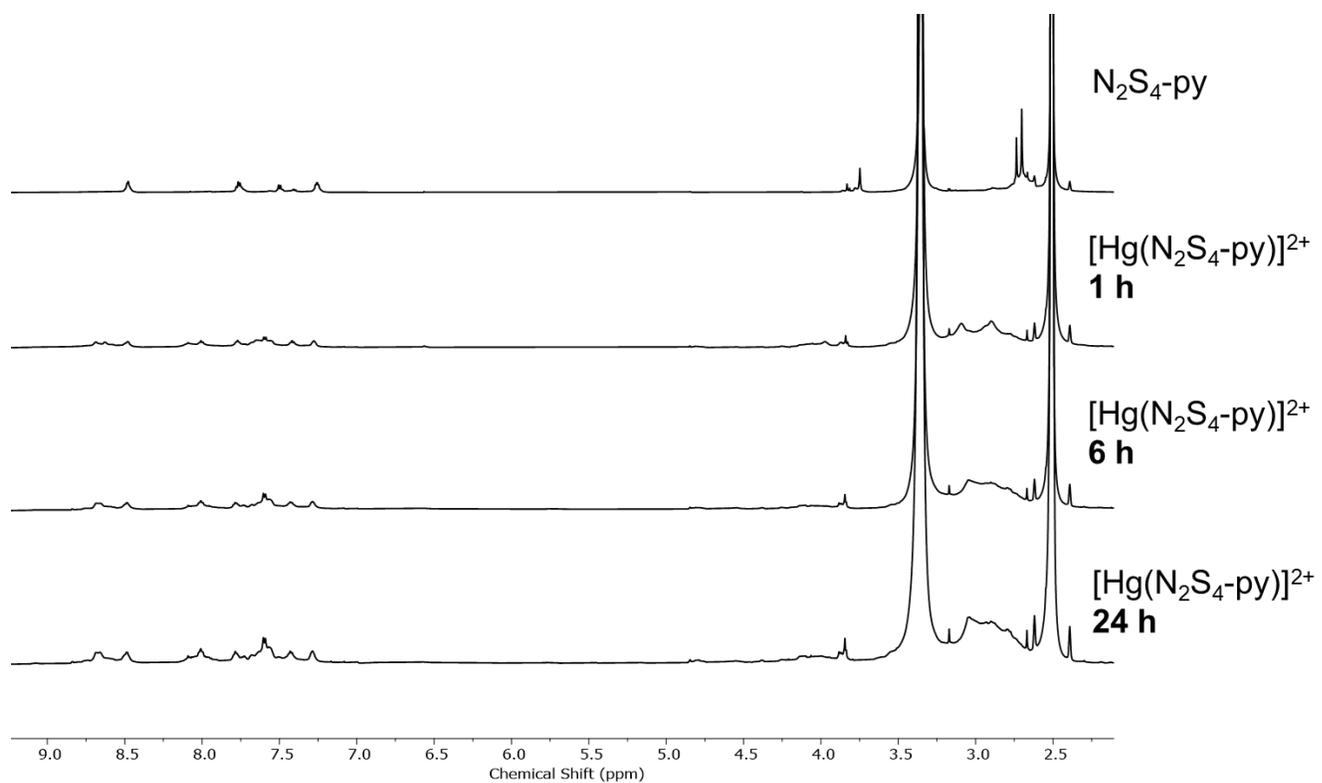


Figure S17. ^1H NMR (600 MHz, $\text{DMSO-}d_6$, 25 $^\circ\text{C}$): $\text{N}_2\text{S}_4\text{-Py}$ and $[\text{natHg}(\text{N}_2\text{S}_4\text{-Py})]^{2+}$ formation over time.

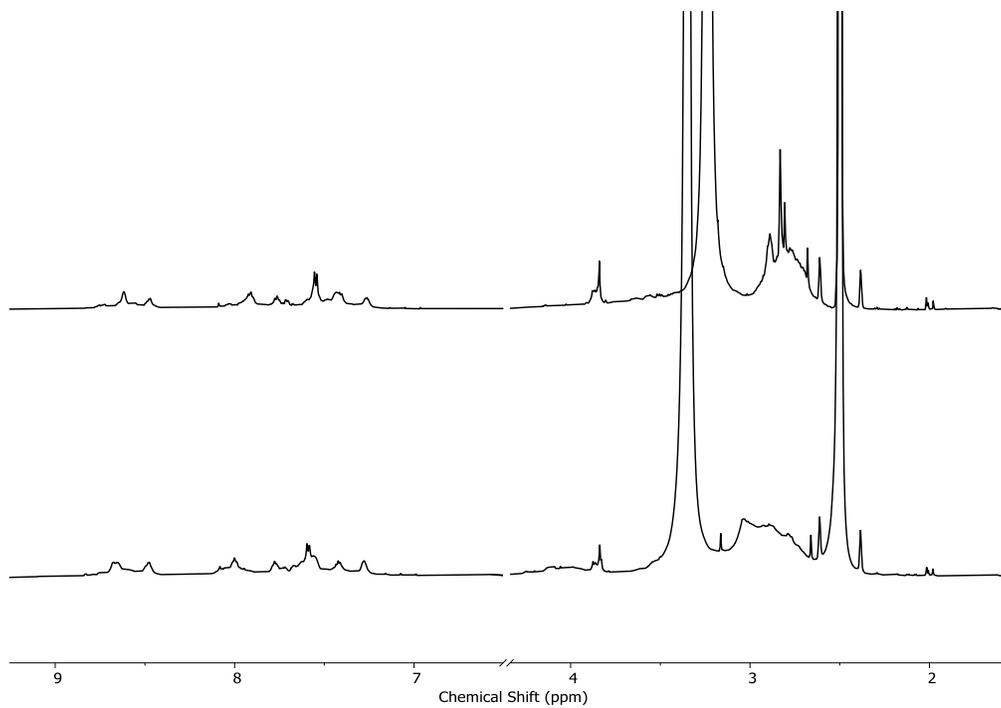


Figure S18. ^1H NMR (600 MHz, $\text{DMSO-}d_6$): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Py})]^{2+}$ at 45 °C (top) and 25 °C (bottom).

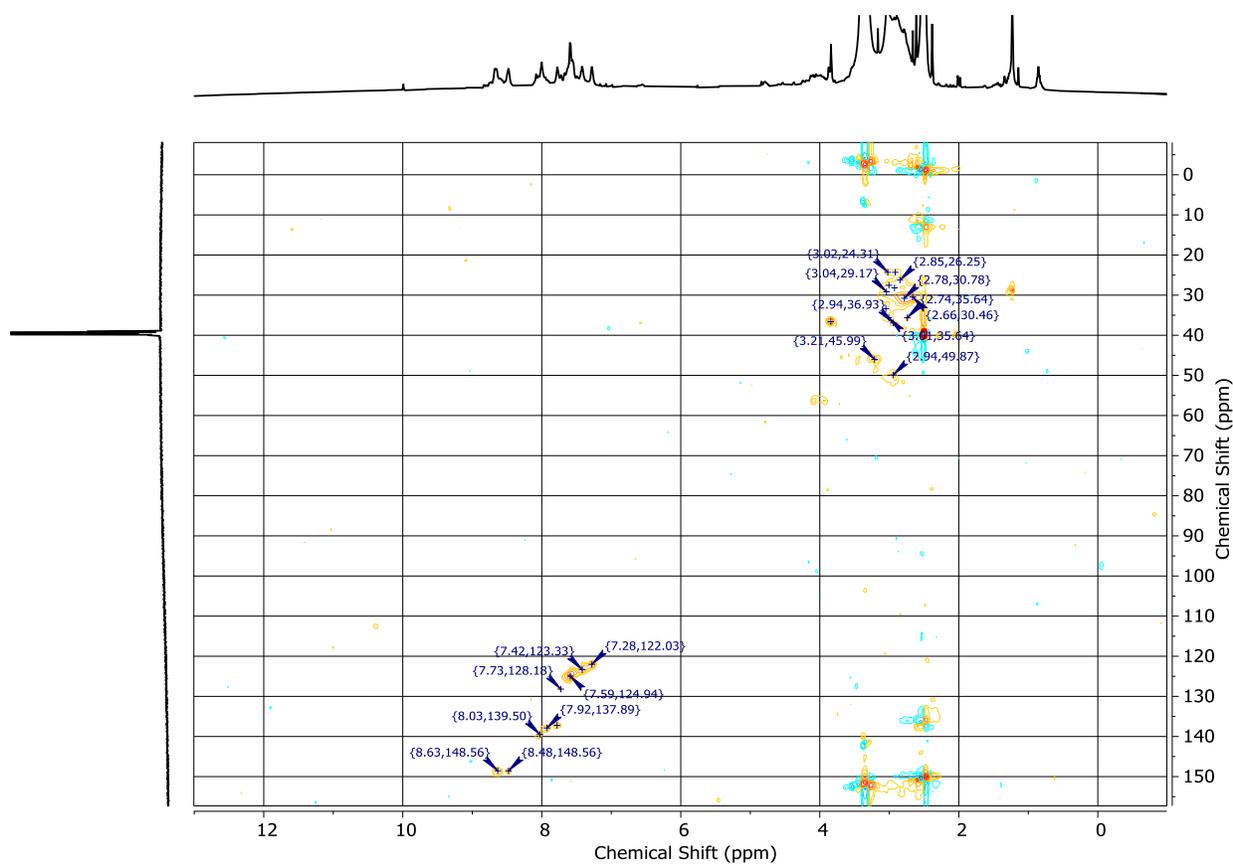


Figure S19. ^1H - ^{13}C HSQC NMR(600 MHz- 161 MHz, $\text{DMSO-}d_6$, $25\text{ }^\circ\text{C}$): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Py})]^{2+}$.

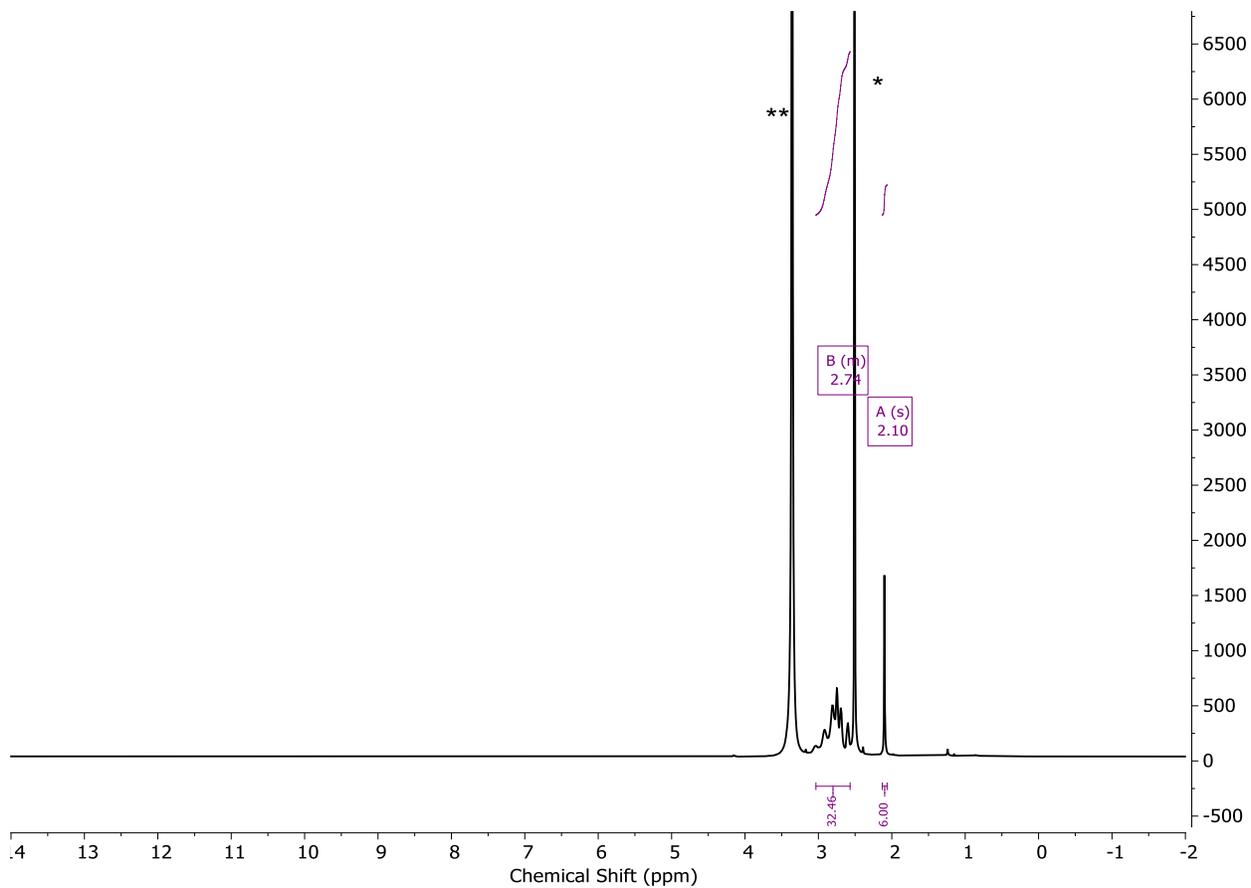


Figure S20. ^1H NMR (600 MHz, $\text{DMSO-}d_6^*$, 25 °C): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Thio})]^{2+}$ at 24 h. (** H_2O)

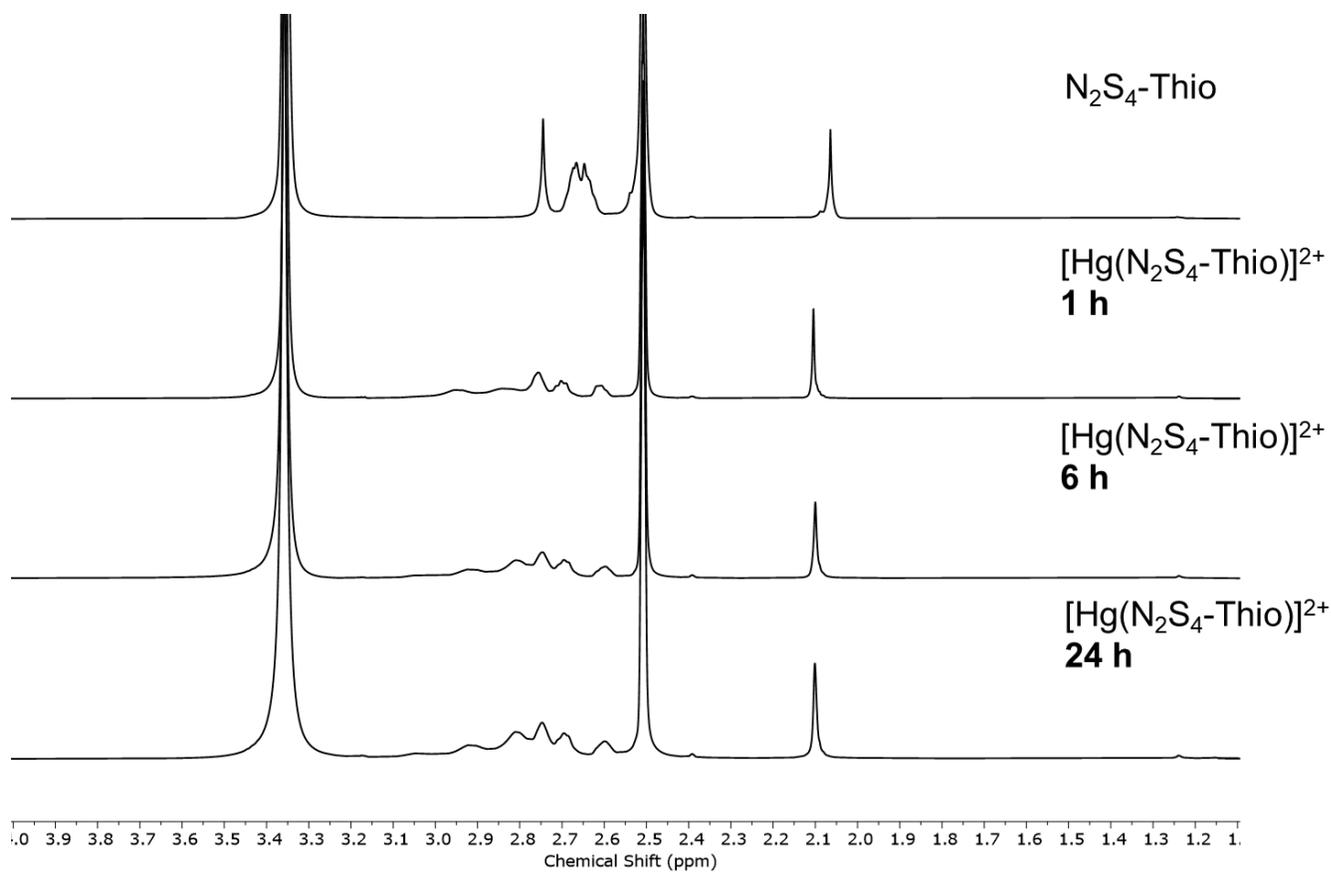


Figure S21. ^1H NMR (600 MHz, $\text{DMSO-}d_6$, 25 $^\circ\text{C}$): $\text{N}_2\text{S}_4\text{-Thio}$ and $[\text{natHg}(\text{N}_2\text{S}_4\text{-Thio})]^{2+}$ formation over time.

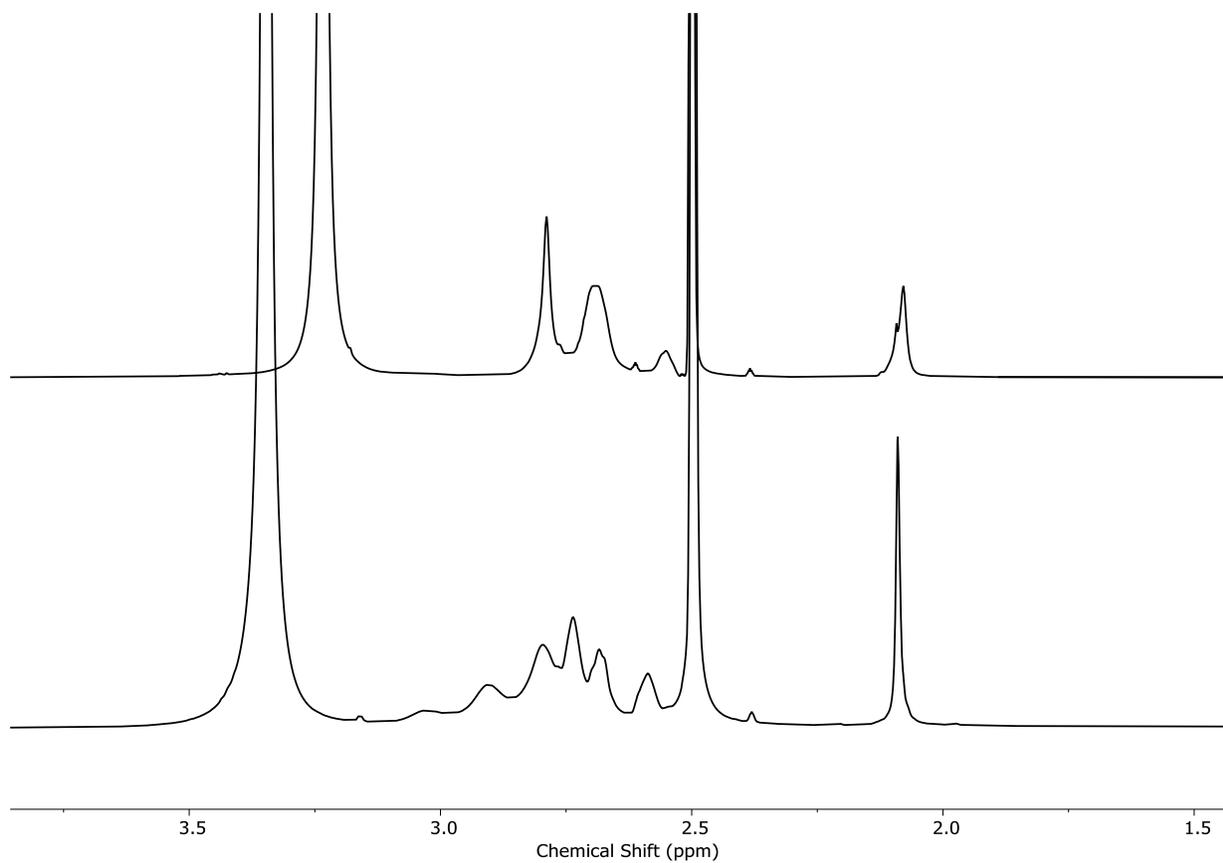


Figure S22. ¹H NMR (600 MHz, DMSO-*d*₆): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Thio})]^{2+}$ at 45 °C (top) and 25 °C (bottom).

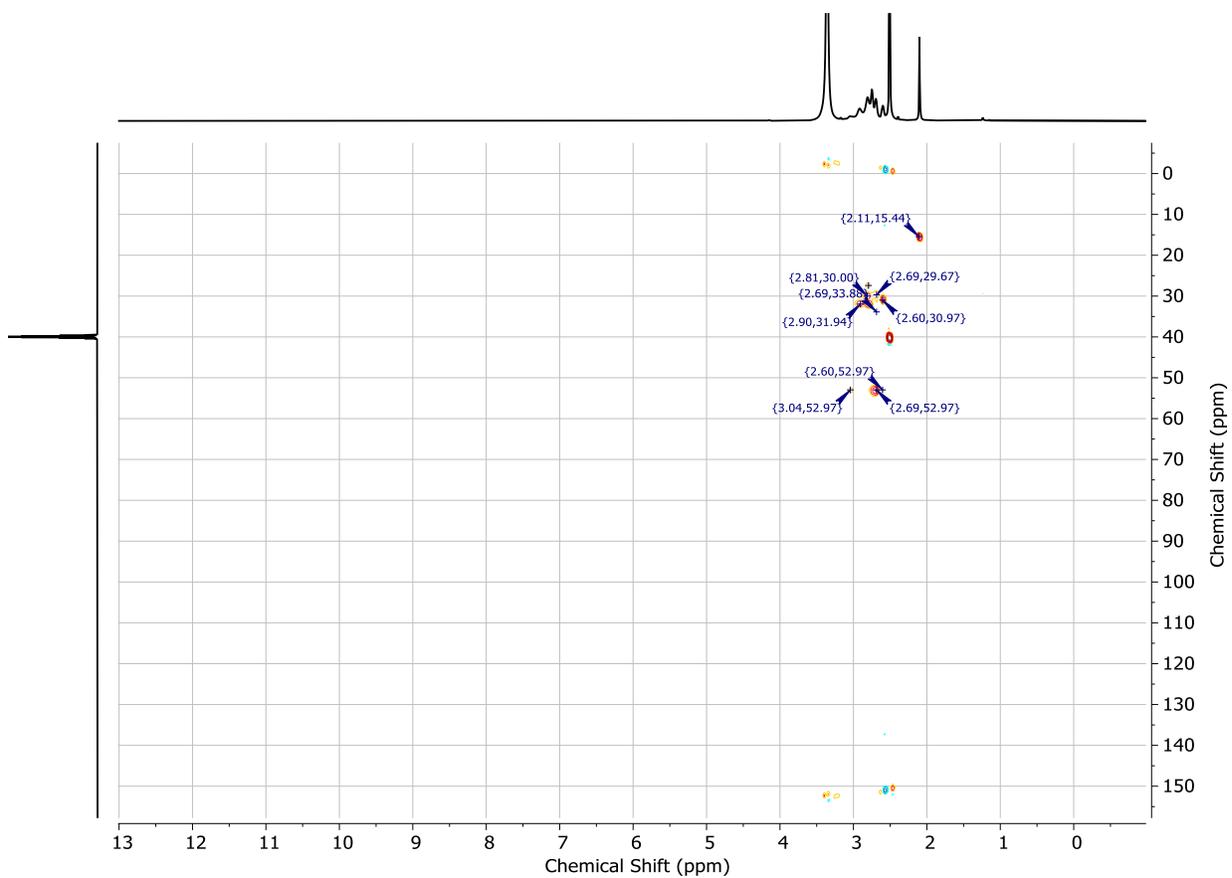


Figure S23. ^1H - ^{13}C HSQC NMR(600 MHz- 161 MHz, $\text{DMSO-}d_6$, $25\text{ }^\circ\text{C}$): $[\text{natHg}(\text{N}_2\text{S}_4\text{-Thio})]^{2+}$.

Mass Spectrometry

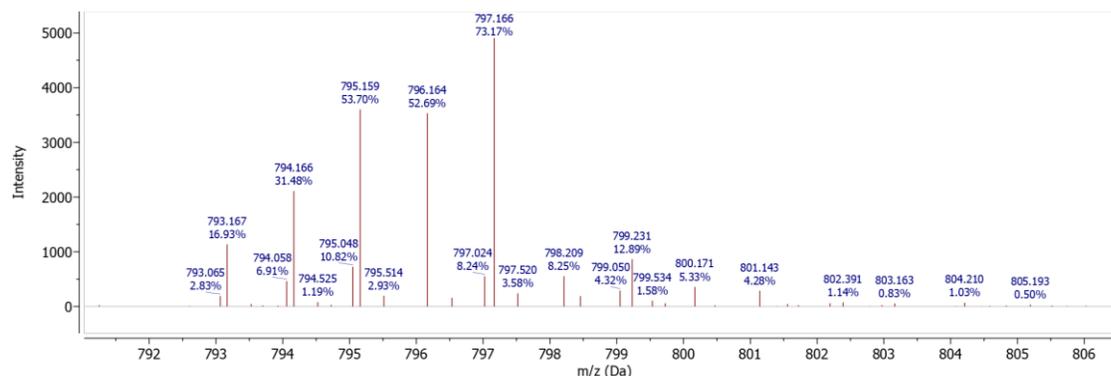


Figure S24. HRMS for [^{nat}Hg(N₂S₄-Pa)], ESI-HRMS *m/z* calcd. for [C₂₆H₃₄N₄O₄S₄Hg + H]⁺ 797.125; found 797.166 [M+H]⁺.

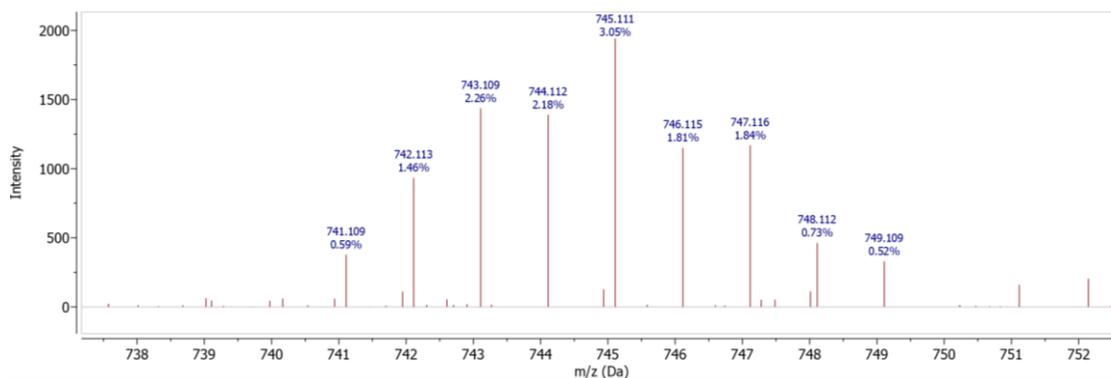


Figure S25. HRMS for [^{nat}Hg(N₂S₄-Py)]²⁺, ESI-HRMS *m/z* calcd. for [(C₂₄H₃₆N₄S₄Hg)²⁺ + Cl]⁺ 745.121; found 745.111 [M+Cl]⁺.

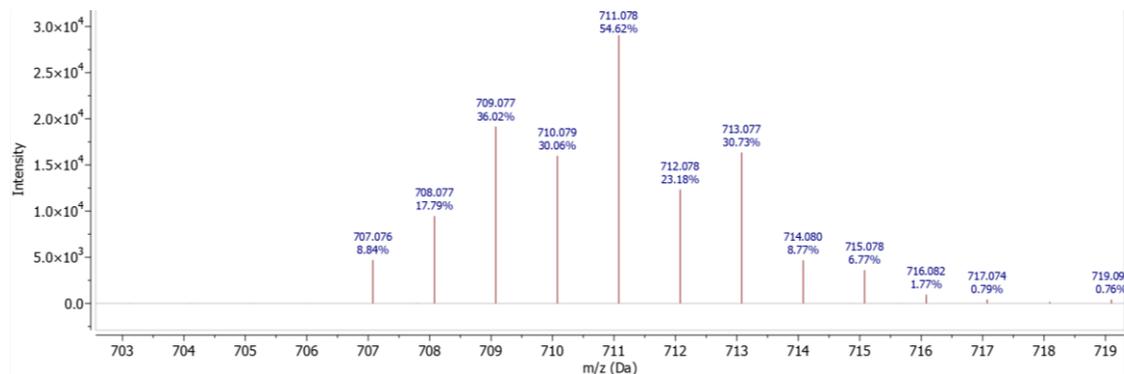


Figure S26. HRMS for [^{nat}Hg(N₂S₄-Thio)]²⁺ ESI-HRMS *m/z* calcd. for [(C₁₈H₃₈N₂S₆Hg)²⁺ + Cl]⁺ 711.075; found 711.078 [M+Cl]⁺.

Density Functional Theory

ϵ_0 =Electronic energy

ϵ_{ZPE} =Zero point energy correction

G_{corr} = Thermal free energy correction

ΔG_{solv} = Change in energy as a result of solvation

$$\begin{aligned} G_{gas} &= \epsilon_0 + \epsilon_{ZPE} + G_{corr} \\ G_{soln} &= G_{gas} + \Delta G_{solv} \end{aligned} \quad (1)0 \text{ h}$$

Table S1. Calculated ΔG_{gas} , ΔG_{solv} , G_{soln} and ΔG_{soln} energies for all investigated conformers of the $[\text{Hg}(\text{N}_2\text{S}_4\text{-Py})]^{2+}$ complex. The lowest energy conformation is highlighted in bold.

Conformer	Sidearm Orientation	Backbone	Relative G_{gas} (kJ/mol)	ΔG_{solv} (Hartree)	G_{soln} (Hartree)	Relative G_{soln} (kJ/mol)
1	cis	$\lambda, \lambda, \delta, \lambda, \lambda, \delta$	81.3	-0.22147	-2901.33939	21.1
2	trans	$\delta, \lambda, \lambda, \delta, \lambda, \delta$	14.3	-0.19454	-2901.33798	24.8
3	cis	$\delta, \lambda, \delta, \lambda, \lambda, \delta$	39.4	-0.20408	-2901.33799	24.8
4	trans	$\lambda, \lambda, \delta, \delta, \lambda, \delta$	2.2	-0.19704	-2901.34511	6.1
5	cis	$\delta, \lambda, \delta, \lambda, \lambda, \lambda$	0.0	-0.19632	-2901.34522	5.8
6	trans	$\lambda, \lambda, \lambda, \delta, \lambda, \delta$	2.3	-0.19762	-2901.34564	4.7
7	cis	$\delta, \lambda, \delta, \delta, \delta, \delta$	43.6	-0.20984	-2901.34213	13.9
8	trans	$\delta, \lambda, \lambda, \delta, \delta, \delta$	42.2	-0.20428	-2901.33711	27.1
9	cis	$\delta, \lambda, \delta, \lambda, \delta, \delta$	20.9	-0.20647	-2901.34743	0.0
10	trans	$\delta, \delta, \delta, \delta, \delta, \delta$	24.1	-0.20407	-2901.34381	9.5
11	cis	$\delta, \delta, \delta, \lambda, \delta, \delta$	39.7	-0.20301	-2901.33680	27.9
12	cis	$\delta, \delta, \delta, \lambda, \lambda, \delta$	32.8	-0.20782	-2901.34425	8.3
13	trans	$\delta, \lambda, \lambda, \delta, \delta, \lambda$	27.7	-0.20466	-2901.34300	11.6

Table S2. Calculated ΔG_{gas} , ΔG_{solv} , G_{soln} and ΔG_{soln} energies for all investigated conformers of the $[\text{Hg}(\text{N}_2\text{S}_4\text{-Thio})]^{2+}$ complex. The lowest energy conformation is highlighted in bold.

Conformer	Sidearm Orientation	Backbone	Relative G_{gas} (kJ/mol)	ΔG_{solv} (Hartree)	G_{soln} (Hartree)	Relative G_{soln} (kJ/mol)
1	cis	$\delta, \lambda, \delta, \lambda, \delta, \lambda$	45.2	-0.21420	-3360.75521	45.7
2	trans	$\lambda, \delta, \lambda, \lambda, \lambda, \lambda$	61.1	-0.21687	-3360.75183	54.6
3	cis	$\lambda, \delta, \delta, \lambda, \lambda, \delta$	43.4	-0.22067	-3360.76237	26.9
4	trans	$\lambda, \delta, \delta, \lambda, \lambda, \lambda$	19.7	-0.21606	-3360.76678	15.4
5	cis	$\lambda, \delta, \lambda, \delta, \lambda, \delta$	19.0	-0.21819	-3360.76918	9.1
6	trans	$\lambda, \delta, \lambda, \lambda, \delta, \delta$	24.0	-0.20793	-3360.75701	41.0
7	cis	$\delta, \lambda, \delta, \lambda, \delta, \lambda$	9.8	-0.21816	-3360.77263	0.0
8	cis	$\lambda, \lambda, \lambda, \delta, \lambda, \delta$	37.7	-0.22318	-3360.76702	14.7
9	trans	$\lambda, \lambda, \lambda, \lambda, \lambda, \lambda$	61.1	-0.21687	-3360.75183	54.6
10	cis	$\lambda, \lambda, \delta, \lambda, \lambda, \delta$	19.0	-0.21488	-3360.76585	17.8
11	cis	$\lambda, \delta, \delta, \delta, \lambda, \delta$	0.0	-0.20560	-3360.76381	23.2

Table S3. Calculated ΔG_{gas} , ΔG_{solv} , G_{soln} and ΔG_{soln} energies for all investigated conformers of the $[\text{Hg}(\text{N}_2\text{S}_4\text{-Pa})]$ complex. The lowest energy conformation is highlighted in bold.

Conformer	Sidearm Orientation	Backbone	Relative ΔG_{gas} (kJ/mol)	ΔG_{solv} (Hartree)	G_{soln} (Hartree)	Relative ΔG_{soln} (kJ/mol)
1	Cis	$\delta, \delta, \lambda, \delta, \lambda, \lambda$	67.9	-0.03788	-3277.69609	58.2
2	Trans	$\delta, \delta, \delta, \lambda, \delta, \lambda$	47.1	-0.04202	-3277.70814	26.6
3	Cis	$\lambda, \delta, \delta, \delta, \lambda, \delta$	33.6	-0.04562	-3277.71688	3.6
4	Trans	$\delta, \lambda, \delta, \lambda, \delta, \lambda$	76.6	-0.03391	-3277.68878	77.4
5	Cis	$\delta, \delta, \delta, \lambda, \delta, \lambda$	32.8	-0.04081	-3277.71239	15.4
6	Trans	$\lambda, \lambda, \delta, \lambda, \delta, \lambda$	123.8	-0.04292	-3277.67984	100.9
7	Cis	$\delta, \delta, \lambda, \delta, \delta, \delta$	0.0	-0.03421	-3277.71827	0.0
8	Trans	$\delta, \lambda, \delta, \delta, \lambda, \lambda$	54.4	-0.02791	-3277.69124	71.0
9	Cis	$\lambda, \lambda, \delta, \delta, \delta, \lambda$	0.6	-0.03383	-3277.71767	1.6
10	Trans	$\delta, \delta, \delta, \lambda, \lambda, \lambda$	35.7	-0.03808	-3277.70854	25.5
11	Cis	$\delta, \delta, \lambda, \delta, \delta, \lambda$	37.9	-0.04314	-3277.71278	14.4

Table S4. Selected bond angles and distances for conformers 5 ($\Delta G_{\text{gas}} = 0$ kJ/mol) and 9 ($\Delta G_{\text{soln}} = 0$ kJ/mol) of the $[\text{Hg}(\text{N}_2\text{S}_4\text{-Py})]^{2+}$ complex.

Bond Length (Å)			Bond Angle (°)		
	Conformer 5	Conformer 9		Conformer 5	Conformer 9
S1-Hg	2.909	2.724	S1-Hg-S2	75.6	78.3
S2-Hg	2.823	2.925	S2-Hg-N4	62.7	70.1
S3-Hg	2.841	2.773	N4-Hg-S3	66.1	75.9
S4-Hg	2.852	4.321	S3-Hg-S4	77.6	56.8
N1-Hg	2.985	2.934	S4-Hg-N1	67.7	53.6
N2-Hg	2.501	2.436	N1-Hg-S1	63.7	71.4
N3-Hg	2.525	2.539			
N4-Hg	3.000	2.693			

Table S5. Selected bond angles and distances for conformers 7 ($\Delta G_{\text{gas}} = 0$ kJ/mol) and 11 ($\Delta G_{\text{soln}} = 0$ kJ/mol) of the $[\text{Hg}(\text{N}_2\text{S}_4\text{-Thio})]^{2+}$ complex.

Bond Length (Å)			Bond Angle (°)		
	Conformer 11	Conformer 7		Conformer 11	Conformer 7
S1-Hg	2.939	2.941	S1-Hg-S2	71.5	76.3
S2-Hg	2.857	2.767	S2-Hg-N2	67.8	66.4
S3-Hg	2.861	2.596	N2-Hg-S3	69.1	62.2
S4-Hg	2.696	3.898	S3-Hg-S4	78.5	56.3
S5-Hg	2.961	2.658	S4-Hg-N1	57.4	50.6
S6-Hg	2.698	2.787	N1-Hg-S1	54.9	56.4
N1-Hg	3.755	3.858			
N2-Hg	3.031	3.166			

Table S6. Selected bond angles and distances for conformer 11, 7 and 3 ($\Delta G_{\text{gas}} = 0$ kJ/mol) and 9 ($\Delta G_{\text{soln}} = 0$ kJ/mol) of the $[\text{natHg}][\text{Hg}(\text{N}_2\text{S}_4\text{-Pa})]$ complex.

Bond Length (Å)			Bond Length (Å)				
	Conf. 11	Conf. 7	Conf. 3	Conf. 11	Conf. 7	Conf 3.	
S1-Hg	2.659	3.905	2.737	O1-Hg	2.471	2.445	2.378
S2-Hg	3.693	2.756	4.022	O2-Hg	2.380	2.431	2.413
S3-Hg	3.408	3.994	4.753	N2-Hg	2.524	2.482	2.444
S4-Hg	4.218	2.731	2.737	N3-Hg	2.450	2.542	2.417
N1-Hg	3.052	3.111	2.938	N4-Hg	3.070	3.238	3.684

Radiolabeling and Stability

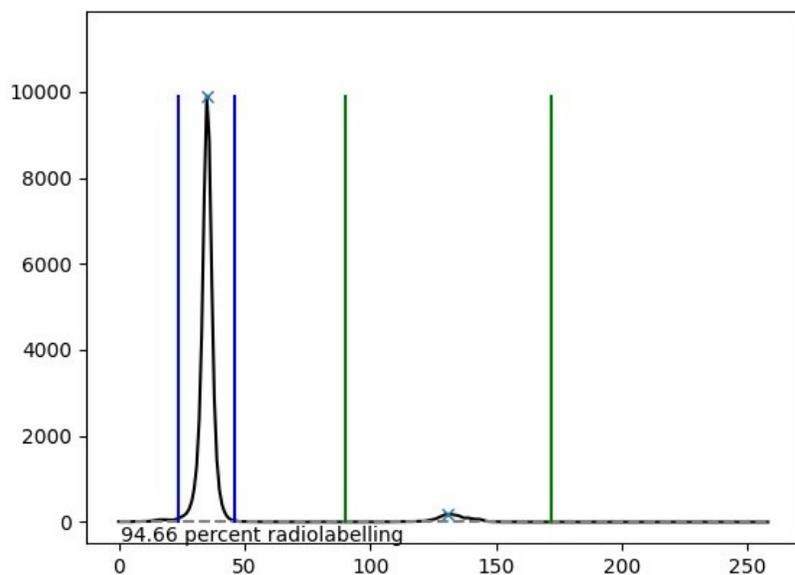


Figure S27. iTLC of $\text{N}_2\text{S}_4\text{-Thio}$ at 10^{-4} M, showing 94.7% RCY at 80 °C.

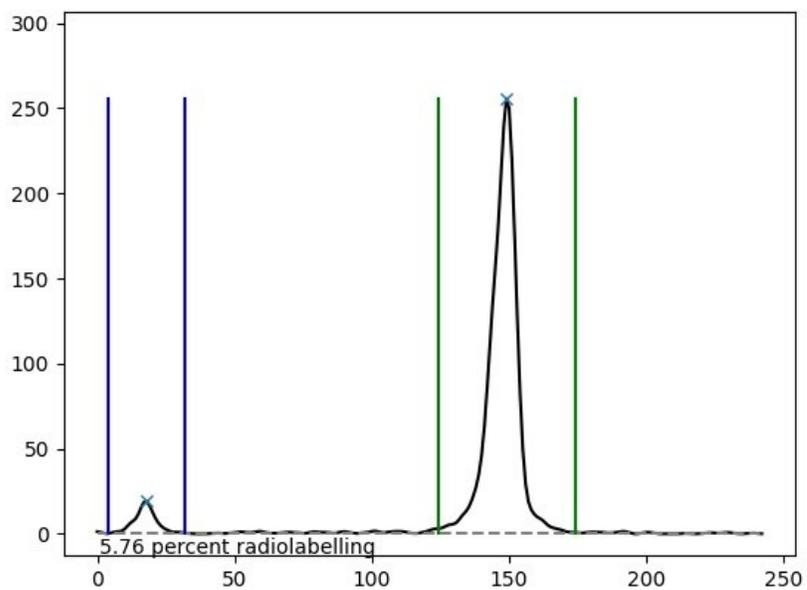


Figure S28. iTLC of N_2S_4 -Thio at 10^{-6} M, showing 5.8 % RCY at 80 °C.

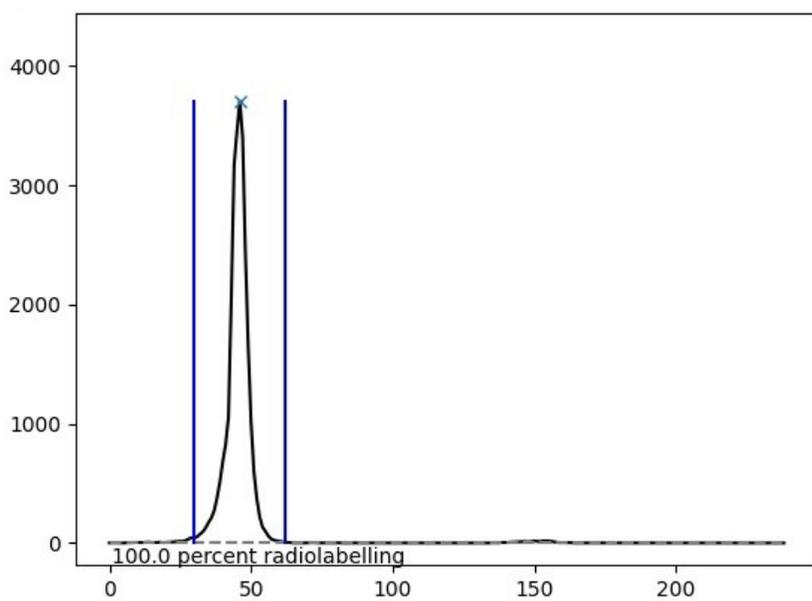


Figure S29. iTLC of N_2S_4 -Py at 10^{-4} M, showing 100 % RCY at 80 °C.

Table S7. Average RCYs (%) of [^{197m}gHg]Hg-Complexes at 80 °C after 60 min, in 1 M NH₄OAc buffer (pH 7).

%RCY	Concentration (M)			
	10 ⁻⁴	10 ⁻⁵	10 ⁻⁶	10 ⁻⁷
[^{197m} gHg][Hg(N ₂ S ₄ -Py)] ²⁺	100.0 ± 0.0%	86.2 ± 1.9 %	17.3 ± 6.6 %	6.3 ± 2.4 %
[^{197m} gHg][Hg(N ₂ S ₄ -Thio)] ²⁺	96.8 ± 0.5%	65.0 ± 4.5 %	11.9 ± 9.0 %	6.4 ± 1.0 %
[^{197m} gHg][Hg(N ₂ S ₄ -Pa)] ²⁺	26.0 ± 1.3%	-	-	-
[^{197m} gHg][Hg(NS ₄ -BA)] ^{2+*}	100.0 ± 0.0%	97.0 ± 0.3 %	31.3 ± 2.0 %	-

*[^{197m}gHg][Hg(NS₄-BA)]²⁺ radiolabeling data is reproduced from Randhawa *et al.* after 1 h.¹

Table S8. Average RCYs (%) of [^{197m}gHg]Hg-Complexes at 25 °C after 60 min, in 1 M NH₄OAc buffer (pH 7).

%RCY	Concentration (M)			
	10 ⁻⁴	10 ⁻⁵	10 ⁻⁶	10 ⁻⁷
[^{197m} gHg][Hg(N ₂ S ₄ -Py)] ²⁺	37.0 ± 3.3 %	-	-	-
[^{197m} gHg][Hg(N ₂ S ₄ -Thio)] ²⁺	95.5 ± 1.8 %	12.3 ± 2.6 %	5.3 ± 4.0 %	-
[^{197m} gHg][Hg(N ₂ S ₄ -Pa)] ²⁺	33.1 ± 0.2 %	-	-	-

Table S9. Kinetic inertness of [^{197m}gHg]Hg-Complexes at 37 °C against human serum. (n=3)

^{197m} gHg-complex % intact	Time point (h)				
	1	3	16	24	72
[^{197m} gHg][Hg(N ₂ S ₄ -Py)] ²⁺	71. ± 2.2 %	57.2 ± 8.2 %	58.9 ± 8.7 %	-	65.1 ± 1.5 %
[^{197m} gHg][Hg(N ₂ S ₄ -Thio)] ²⁺	75.5 ± 0.8 %	67.8 ± 5.0 %	70.8 ± 0.1 %	-	60.3 ± 2.1 %
[^{197m} gHg][Hg(NS ₄ -BA)] ^{2+*}	84.7 ± 3.4 %	74.1 ± 7.0 %	-	-	-

*[^{197m}gHg][Hg(NS₄-BA)]²⁺ radiolabeling data is reproduced from Randhawa *et al.* after 1 h.¹

Table S10. Kinetic inertness of [^{197m/g}Hg]Hg-Complexes at 37 °C against GSH. (n=3)

^{197m/g} Hg-complex % intact	Time point (h)				
	1	3	10	24	72
[^{197m/g} Hg][Hg(N ₂ S ₄ -Py)] ²⁺	85.8 ± 0.4 %	72.6 ± 0.4 %	-	32.0 ± 5.1%	-
[^{197m/g} Hg][Hg(N ₂ S ₄ -Thio)] ²⁺	77.8 ± 0.4 %	75.1 ± 3.0 %	-	43.7 ± 3.2%	-
[^{197m/g} Hg][Hg(NS ₄ -BA)] ^{2+*}	100.0 ± 0.0 %	-	97.3 ± 0.3 %	-	92.3 ± 0.5 %

*[^{197m/g}Hg][Hg(NS₄-BA)]²⁺ radiolabeling data is reproduced from Randhawa *et al.* after 1 h.¹

Table S11. Kinetic inertness of [^{197m/g}Hg]Hg-Complexes at 25 °C against stable biologically relevant metals (ZnCl₂, FeCl₃, CuCl₂, MgCl₂ and CoCl₂). (n=3)

^{197m/g} Hg-complex % intact	Time point (h)		
	24	48	85
[^{197m/g} Hg][Hg(N ₂ S ₄ -Py)] ²⁺	90.8 ± 8.7 %	83.9 ± 5.0 %	73.0 ± 7.0 %
[^{197m/g} Hg][Hg(N ₂ S ₄ -Thio)] ²⁺	86.2 ± 1.7 %	86.0 ± 2.6 %	86.9 ± 3.2 %

References

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