

Electronic Supporting Information

pKa of Alcohols Dictate their Reactivity with Reduced Uranium-substituted Thiomolybdate Clusters

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Supporting Information Table of Contents

Table S1. Crystallographic parameters for molecular structures of complexes for $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OMe})$, $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh}^{\text{Cl}2})$, $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$, and $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$	S3
Figures S1. ^1H NMR spectrum of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OMe})$ in C_6D_6	S4
Figures S2. ^1H NMR characterization of the evolved H_2 gas from the reaction mixture of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and methanol in C_6D_6	S4
Figures S3 and S4. ^1H and ^2H NMR characterization of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OCD}_3)$	S5
Figure S5. ^2H NMR characterization of the evolved D_2 gas from the reaction mixture of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and CD_3OD in C_6H_6	S6
Figure S6. ^1H NMR spectrum of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 1.1 equivalent of phenol in C_6D_6	S6
Figure S7. ^1H NMR spectrum of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$ in C_6D_6	S7
Figure S8. ^1H NMR spectrum of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})$ in C_6D_6	S7
Figure S9. ^1H NMR spectrum of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$ obtained from the reaction of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})$ with 1 equivalent of phenol in C_6D_6	S8
Figure S10. ^1H NMR spectrum of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 2 equivalents methanol in toluene- d_8	S8
Figure S11. ^1H NMR spectrum of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 3 equivalents methanol in toluene- d_8	S9
Figure S12. ^1H NMR spectrum of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh}^{\text{Cl}2})$ in C_6D_6	S9
Figure S13. ^1H NMR spectrum of the reaction mixture of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ with 1 equiv of <i>tert</i> -butanol in C_6D_6	S10
Figure S14. ^1H NMR spectrum of the crude reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 3 equivalent of <i>tert</i> -butanol in C_6D_6	S11
Figure S15. ^1H NMR spectrum of the $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$ in C_6D_6	S11
Figure S16. ^1H NMR spectrum of the $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}^t\text{Bu})$ in C_6D_6	S12
Figure S17. Stacked ^1H NMR spectra of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 1 equivalent of <i>tert</i> -butanol with $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}^t\text{Bu})$ in C_6D_6	S12
Figure S18. ^1H NMR spectrum of the $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OC}(\text{CF}_3)_3)$ in C_6D_6	S13
Figure S19. $^{19}\text{F}\{^1\text{H}\}$ NMR spectrum of the $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OC}(\text{CF}_3)_3)$ in C_6D_6	S13

Table S1: Crystallographic parameters for molecular structures of complexes for $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OMe})$, $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh}^{\text{Cl}_2})$, $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$ and $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$.

Compound	$(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OMe})$	$(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh}^{\text{Cl}_2})$	$(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$	$(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$
Empirical formula	$\text{C}_{41}\text{H}_{63}\text{Mo}_3\text{OS}_4\text{U}$	$\text{C}_{56.50}\text{H}_{75}\text{Cl}_2\text{Mo}_3\text{O S}_4\text{U}$	$\text{C}_{58}\text{H}_{70}\text{D}_6\text{Mo}_3\text{O}_2\text{S}_4\text{U}$	$\text{C}_{42}\text{H}_{72}\text{Mo}_3\text{O}_3\text{S}_4\text{U}$
Formula weight	1226.00	1495.15	1465.31	1279.08
Temperature	100.00(10) K	100.00(10) K	100.00(10) K	100.00(10) K
Wavelength	1.54184 Å	1.54184 Å	1.54184 Å	1.54184 Å
Crystal system	monoclinic	monoclinic	orthorhombic	monoclinic
Space group	$P2_1/n$	$P2_1/c$	$P2_12_12_1$	$P2_1/n$
Unit cell dimensions	$a = 11.17593(8) \text{ \AA}$ $b = 19.26021(16) \text{ \AA}$ $c = 20.25129(16) \text{ \AA}$ $\alpha = 90^\circ$ $\beta = 99.7002(7)^\circ$ $\gamma = 90^\circ$	$a = 11.11244(9) \text{ \AA}$ $b = 20.41551(16) \text{ \AA}$ $c = 24.6678(2) \text{ \AA}$ $\alpha = 90^\circ$ $\gamma = 90^\circ$ $\beta = 97.3244(8)^\circ$	$a = 10.99181(6) \text{ \AA}$ $b = 20.28569(11) \text{ \AA}$ $c = 24.73170(15) \text{ \AA}$ $\alpha = 90^\circ$ $\beta = 90^\circ$ $\gamma = 90^\circ$	$a = 10.97728(7) \text{ \AA}$ $b = 17.85746(11) \text{ \AA}$ $c = 24.61165(16) \text{ \AA}$ $\alpha = 90^\circ$ $\beta = 96.2794(6)^\circ$ $\gamma = 90^\circ$
Volume / Å ³	4296.78(6)	5550.63(8)	5514.59(5)	4795.59(5)
Z	4	4	4	4
Crystal Size	0.103 x 0.053 x 0.027 mm ³	0.169 x 0.028 x 0.019 mm ³	0.138 x 0.038 x 0.015 mm ³	0.127 x 0.065 x 0.011 mm ³
Reflections collected	62579	92991	63887	80766
Independent reflections	9171 [R(int) = 0.0510]	11963 [R(int) = 0.0516]	11807 [R(int) = 0.0481]	10319 [R(int) = 0.0533]
Completeness ($\theta = 74.5^\circ$)	99.7%	99.9%	99.9%	99.9%
Goodness-of-fit on F^2	1.042	1.052	1.057	1.025
Final R indices [I > 2σ(I)]	R1 = 0.0457, wR2 = 0.1110	R1 = 0.0286, wR2 = 0.0704	R1 = 0.0249, wR2 = 0.0541	R1 = 0.0249, wR2 = 0.0581
Largest diff. peak and hole	3.004 and -3.149 e.Å ⁻³	4.102 and -2.074 e.Å ⁻³	1.454 and -0.874 e.Å ⁻³	1.219 and -1.055 e.Å ⁻³

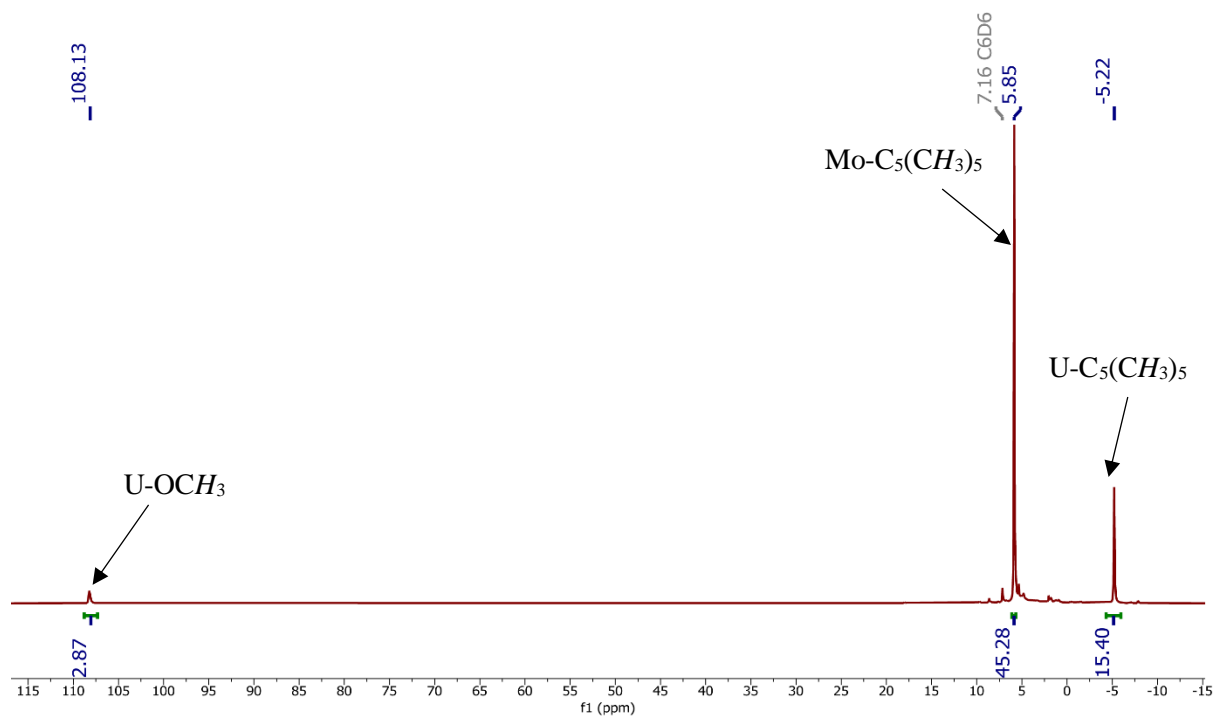


Figure S1. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OMe})$ in C_6D_6 .

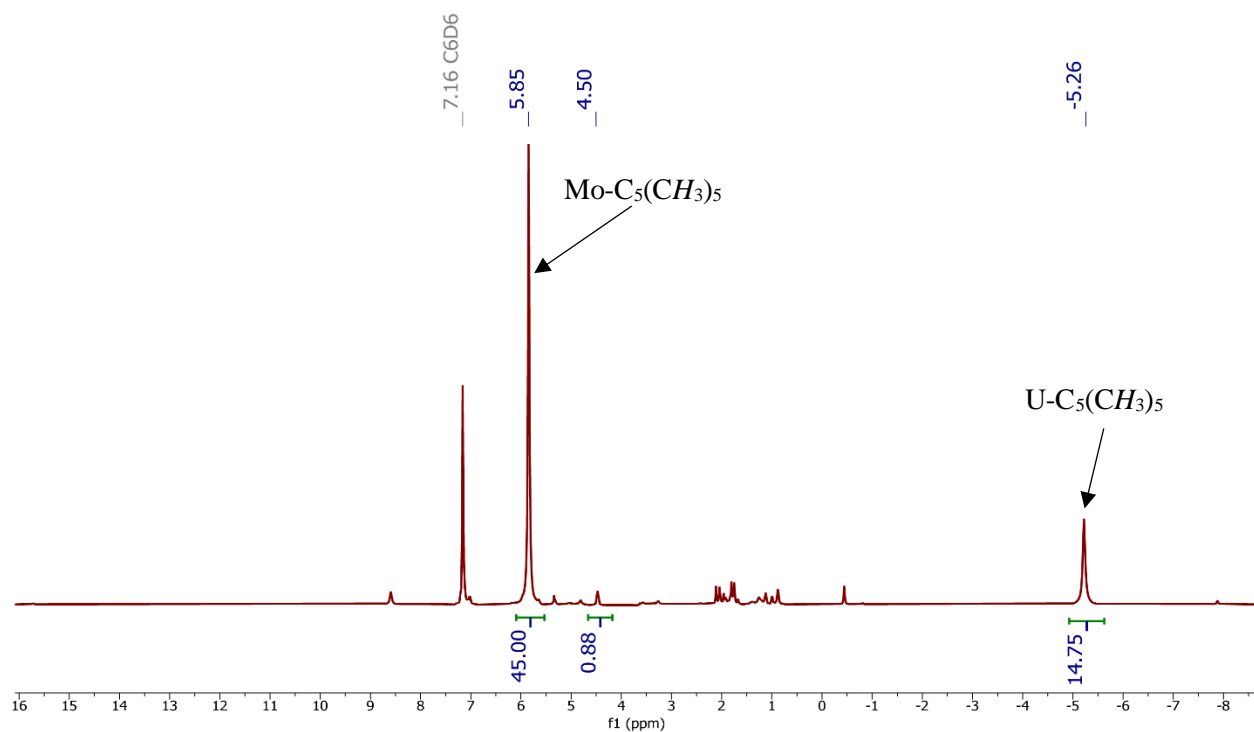


Figure S2. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and C_6D_6 solution having 1 equivalent of methanol in C_6D_6 . [Signals $\delta = 4.50$ ppm is assigned to H_2 (~88% yield).]

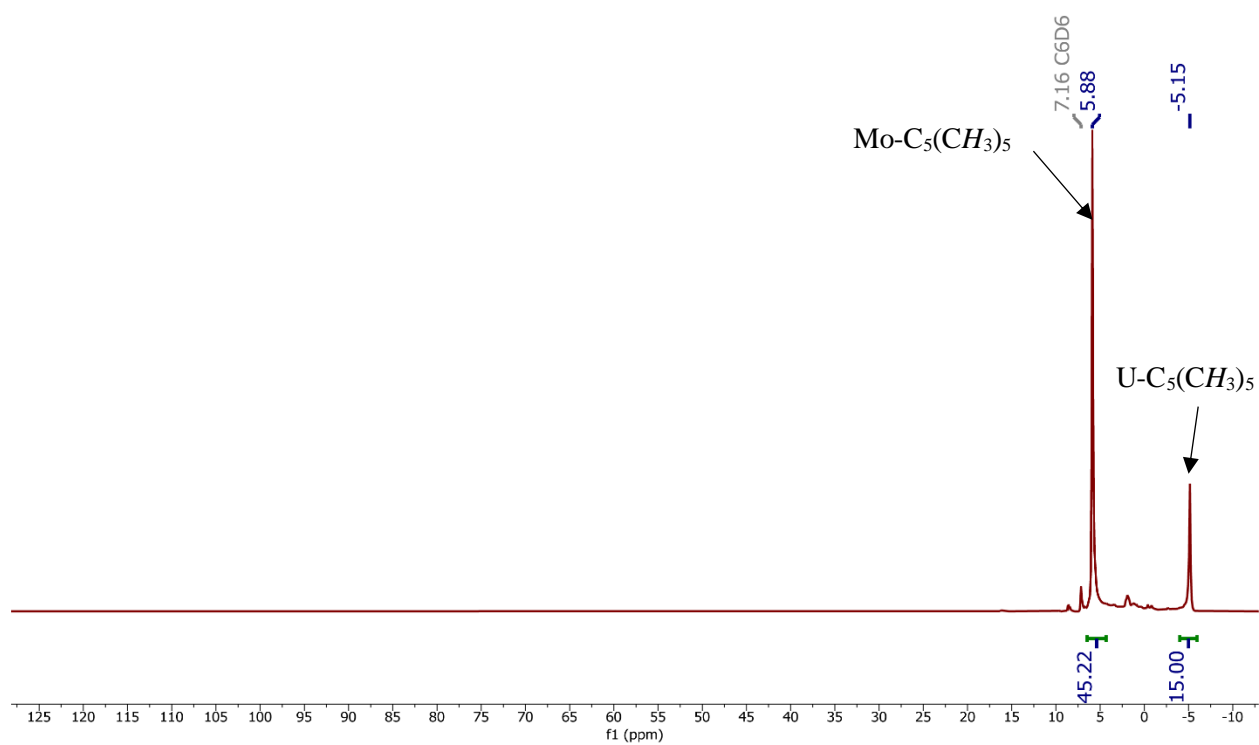


Figure S3. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OCD}_3)$ in C_6D_6 .

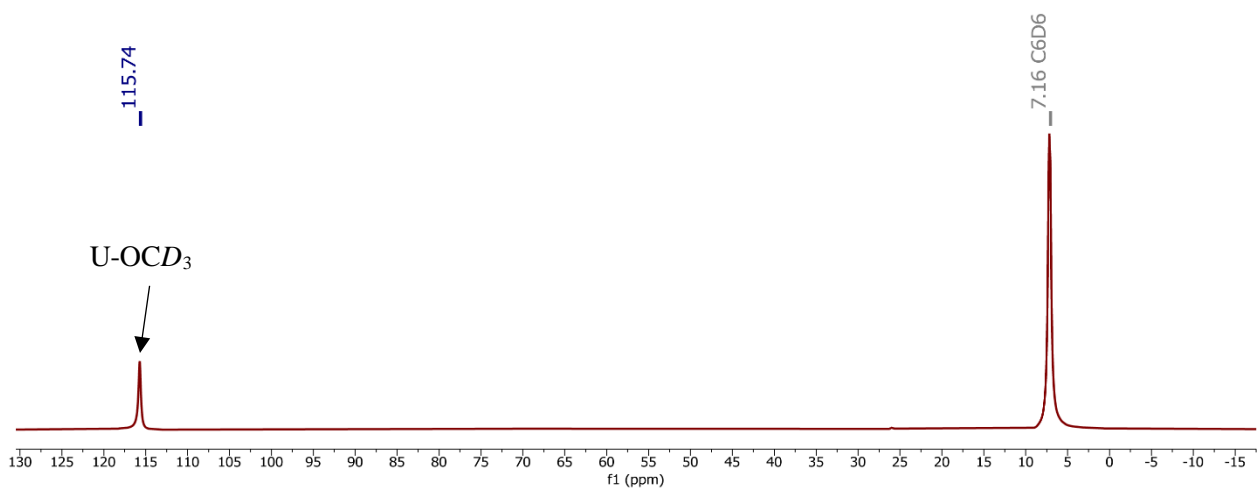


Figure S4. ^2H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OCD}_3)$ in C_6H_6 [The signal at 115.74 ppm is attributed to the $-\text{OCD}_3$].

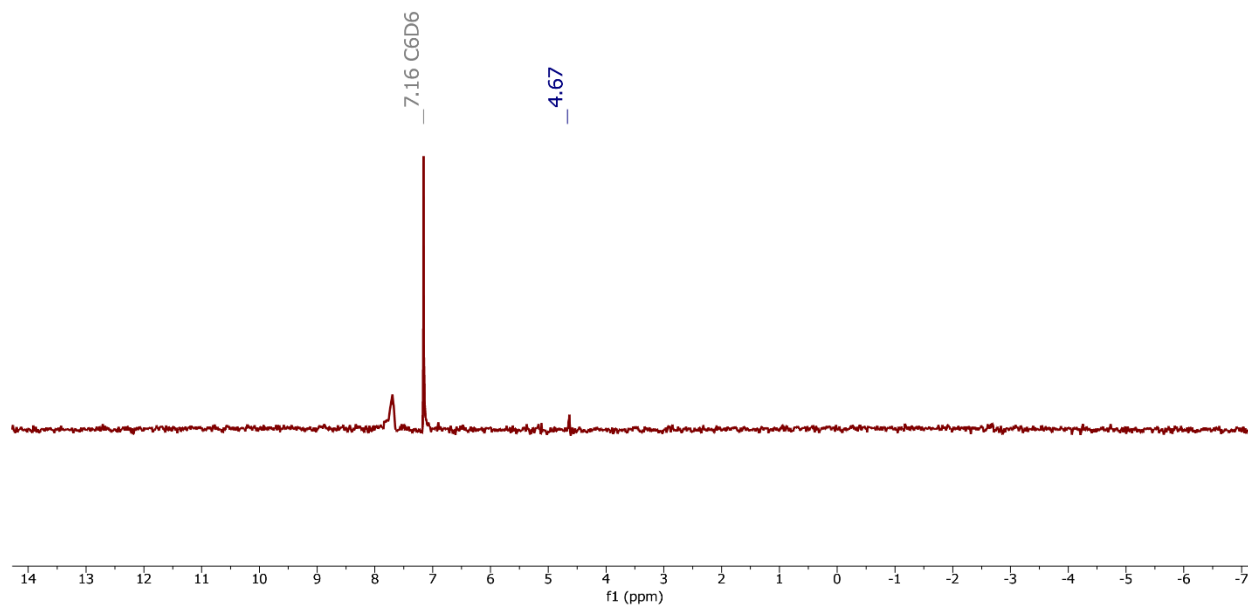


Figure S5. ^2H NMR spectrum (400 MHz) of the reaction mixture containing a C_6H_6 solution of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and a C_6H_6 solution with slightly more than 1 equivalent of CD_3OD [C_6D_6 was added as an internal reference; $\delta = 4.67$ ppm is assigned to D_2].

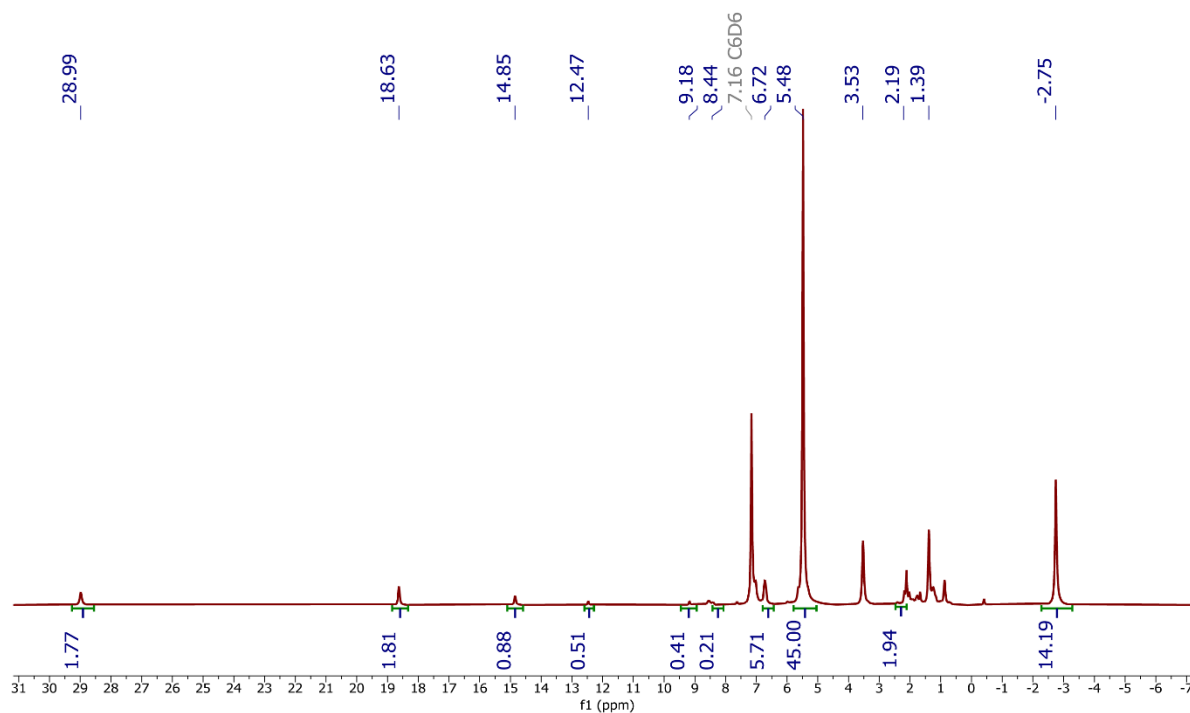


Figure S6. ^1H NMR spectrum (400 MHz) of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and ~ 1.1 equivalent of phenol in C_6D_6 . Signals at 28.99, 18.63, 14.85, 5.48 and -2.75 ppm correspond to $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})$, whereas signals at 12.47, 9.18, 8.44, 6.72 and 2.19 ppm correspond to $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$.

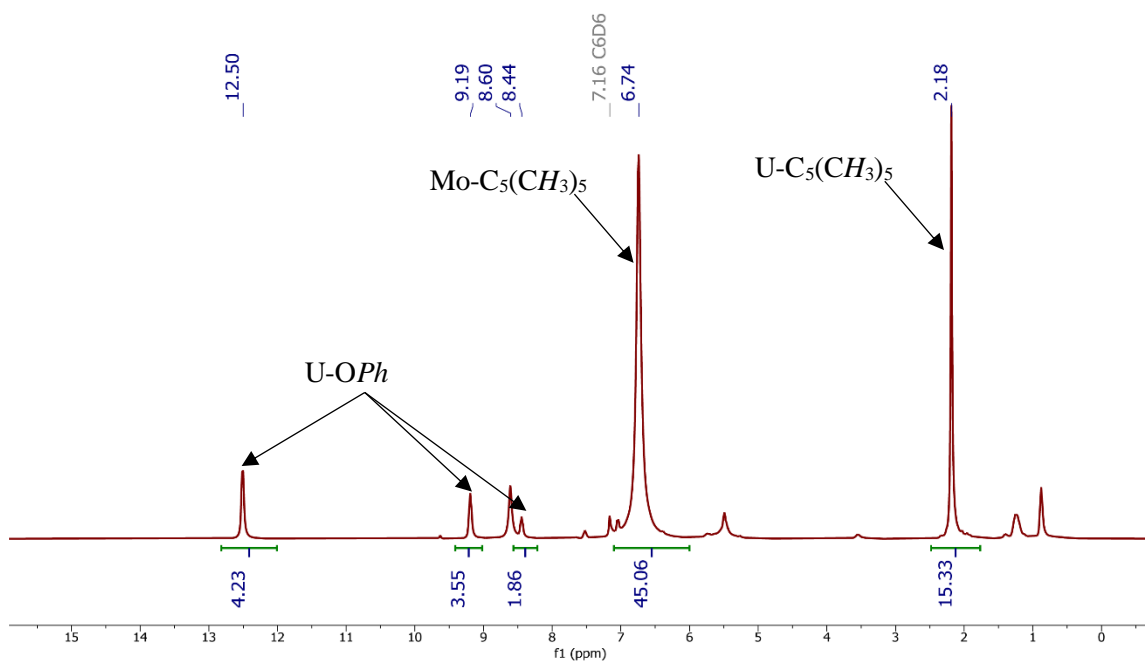


Figure S7. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$ in C_6D_6 .

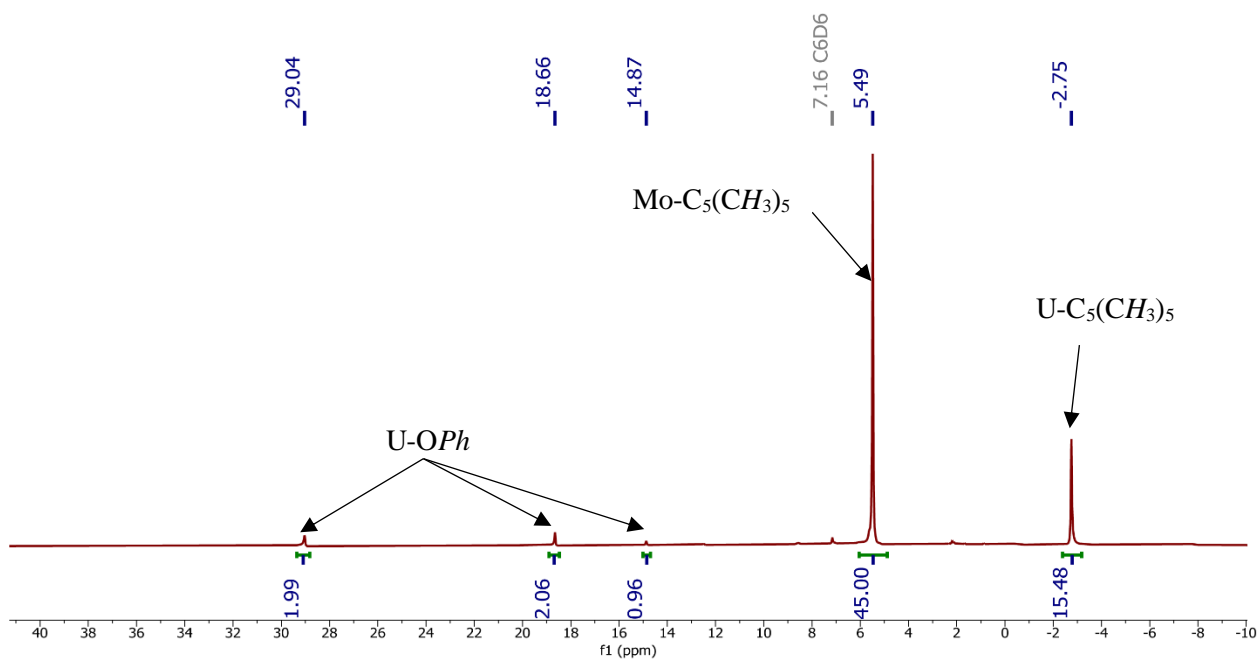


Figure S8. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})$ in C_6D_6 .

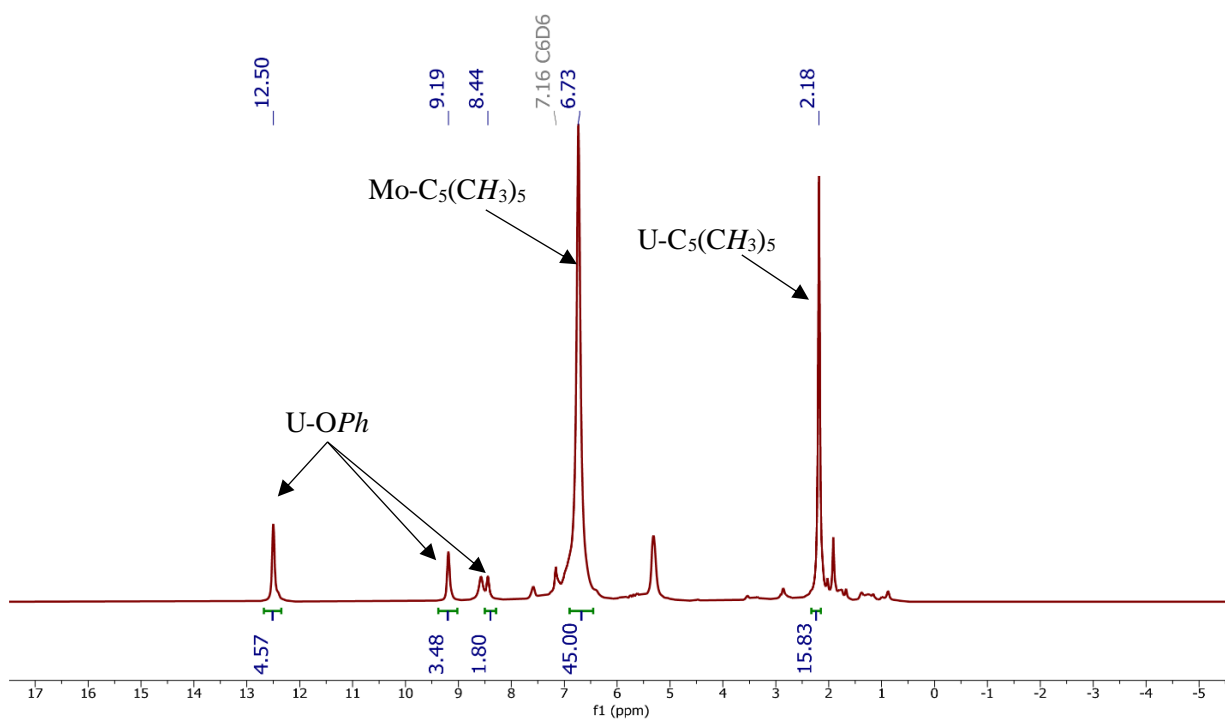


Figure S9. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})_2$ obtained from the reaction of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh})$ with 1 equivalent of phenol in C_6D_6 .

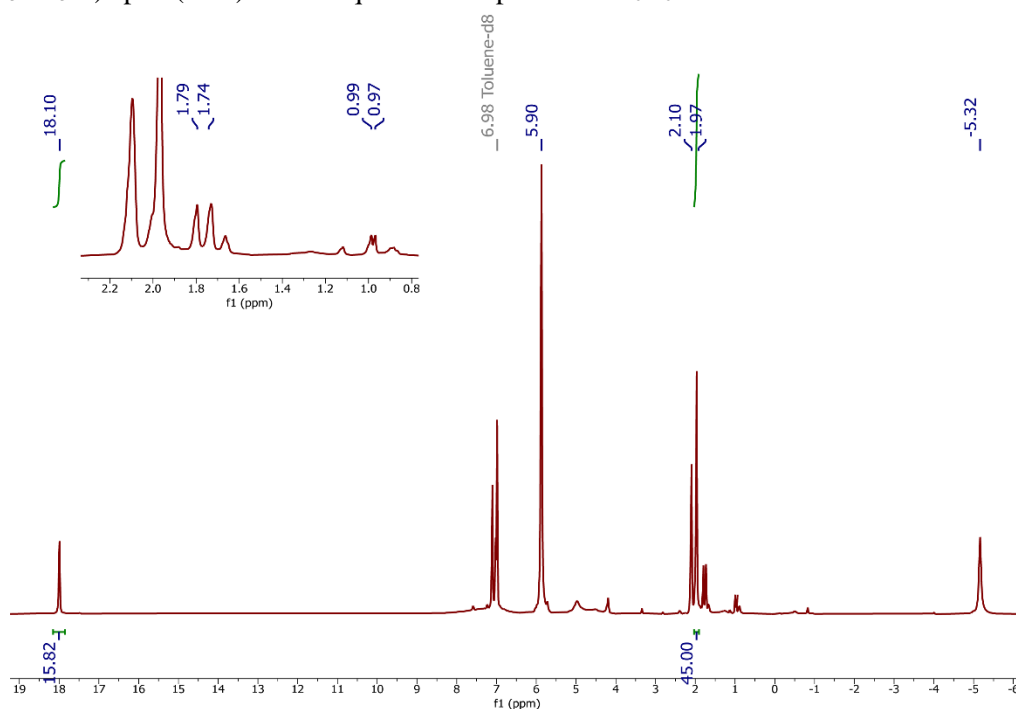


Figure S10. ^1H NMR spectrum (400 MHz) (with slow relaxation time) of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 2 equivalents methanol. (The signals at 18.10 and 1.97 ppm are assigned to $[(\text{Cp}^*_3\text{Mo}_3\text{S}_4)^+\text{U}(\text{OMe})_5^-]$ or $[(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{OMe})_5]$; signals at 5.90 and -5.32 correspond to $[(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OMe})]$; and signals at 0.98, 1.72 and 1.79 ppm assigned to Cp^*H).

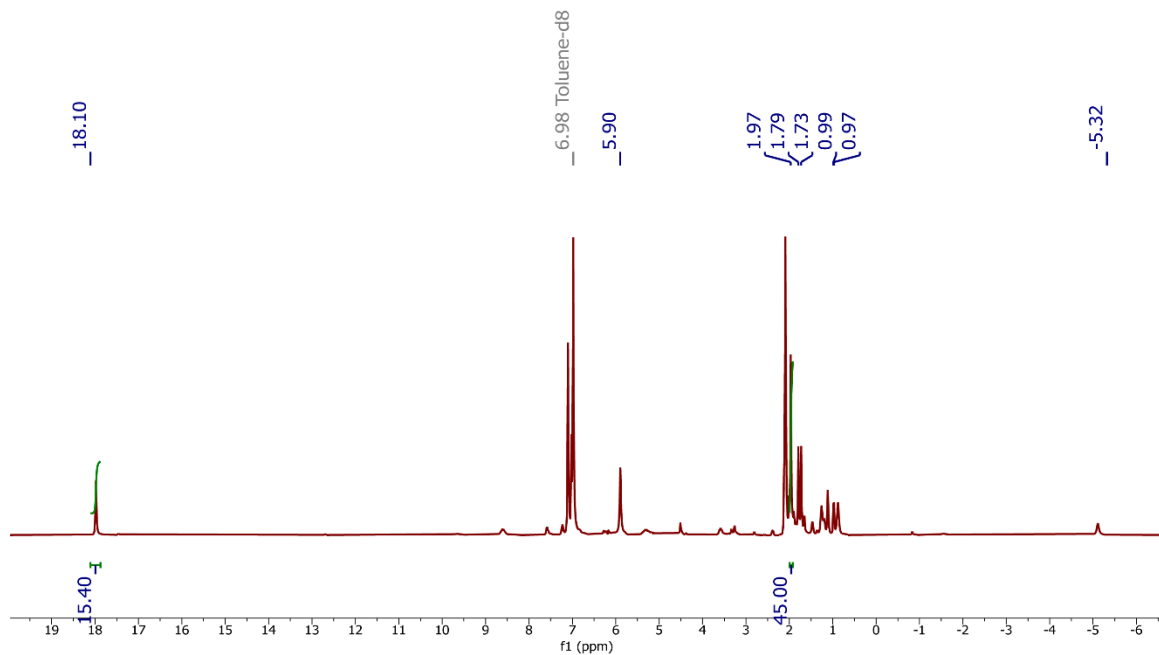


Figure S11. ^1H NMR spectrum (400 MHz) of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 3 equivalents methanol.

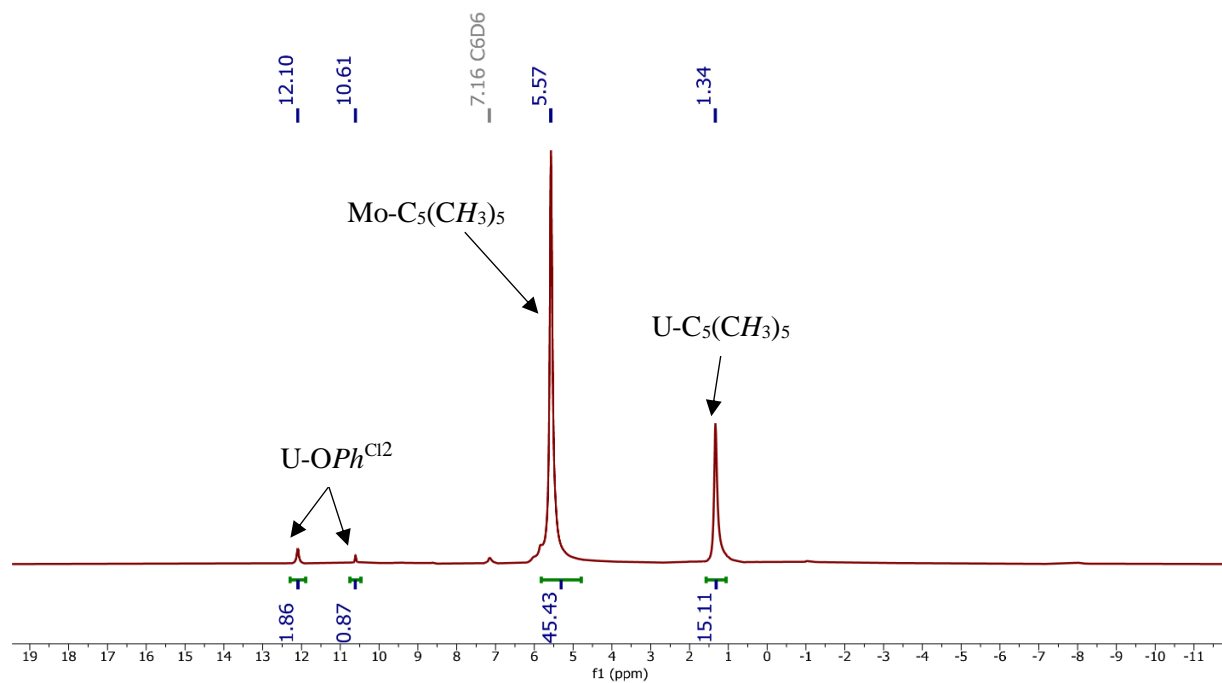


Figure S12. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OPh}^{\text{Cl}_2})$ in C_6D_6 .

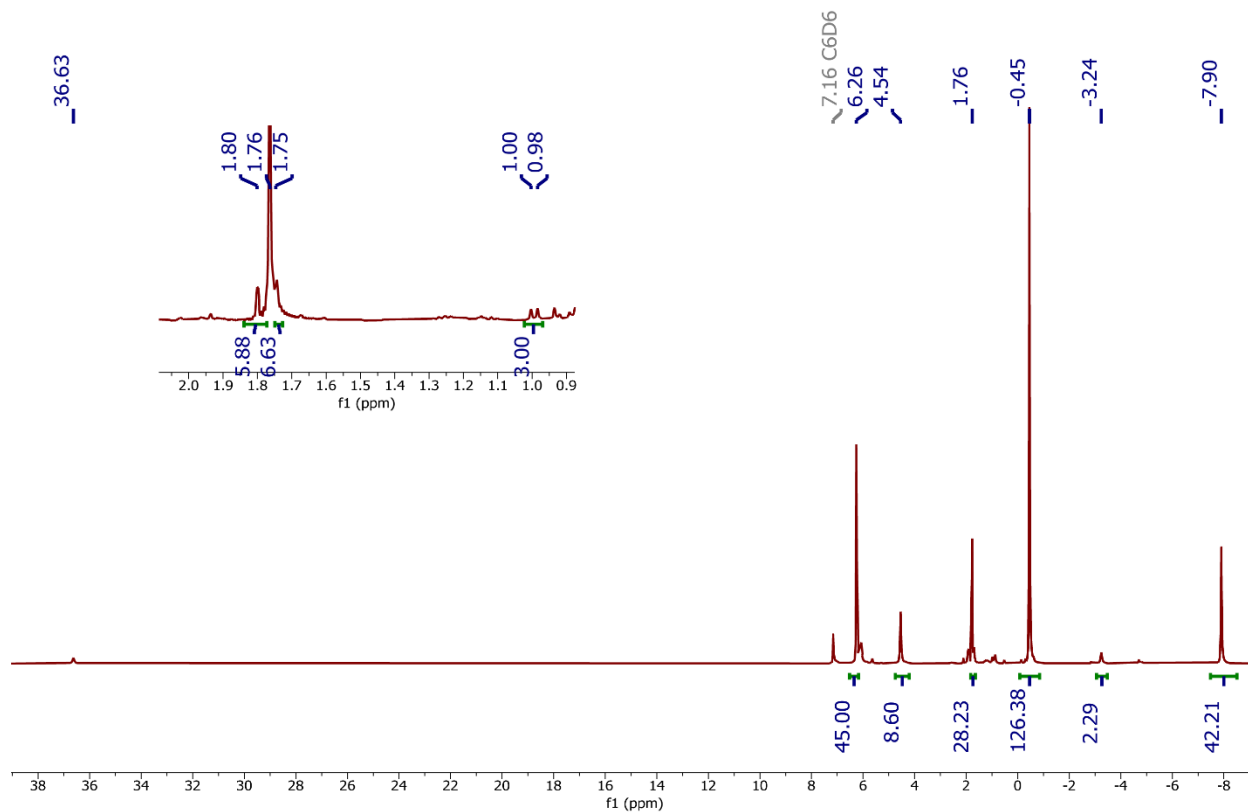


Figure S13. ^1H NMR spectrum (400 MHz) (with slow relaxation time) of the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 1 equivalent of *tert*-butanol in C_6D_6 . The signals at 6.26, and 1.76 ppm correspond to $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$; signals at 4.54, -3.24, and 36.63 ppm correspond to proposed $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}^t\text{Bu})$ ($(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3 : (\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}^t\text{Bu}) \sim 5:1$) (see figure S20 for the ^1H NMR spectrum of pure $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}^t\text{Bu})$); signals at 1.80, 1.75, and 0.99 ppm correspond Cp^*H ; signals at -0.45 and -7.90 ppm correspond to $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$.

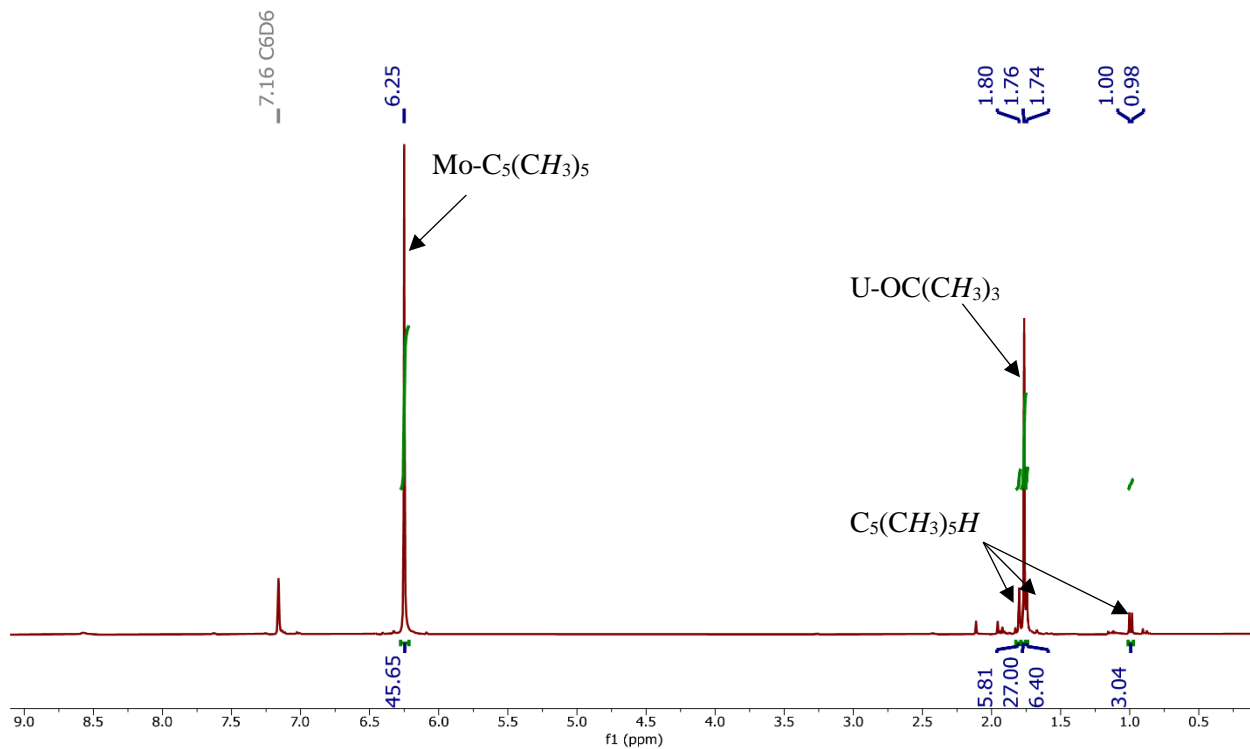


Figure S14. ^1H NMR spectrum (400 MHz) (with slow relaxation time) of the crude reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 3 equivalent of *tert*-butanol in C_6D_6 [products: $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$ and Cp^*H].

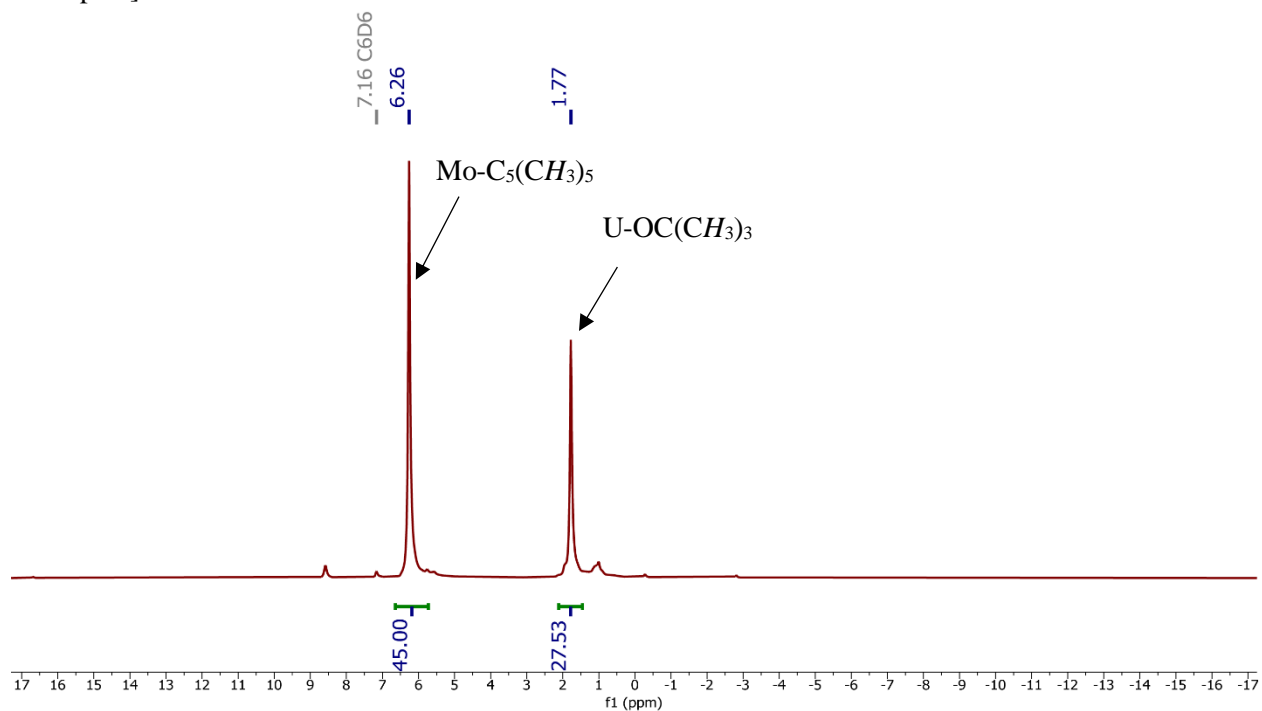


Figure S15. ^1H NMR spectrum (500 MHz) of the $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{U}(\text{O}^t\text{Bu})_3$ in C_6D_6 .

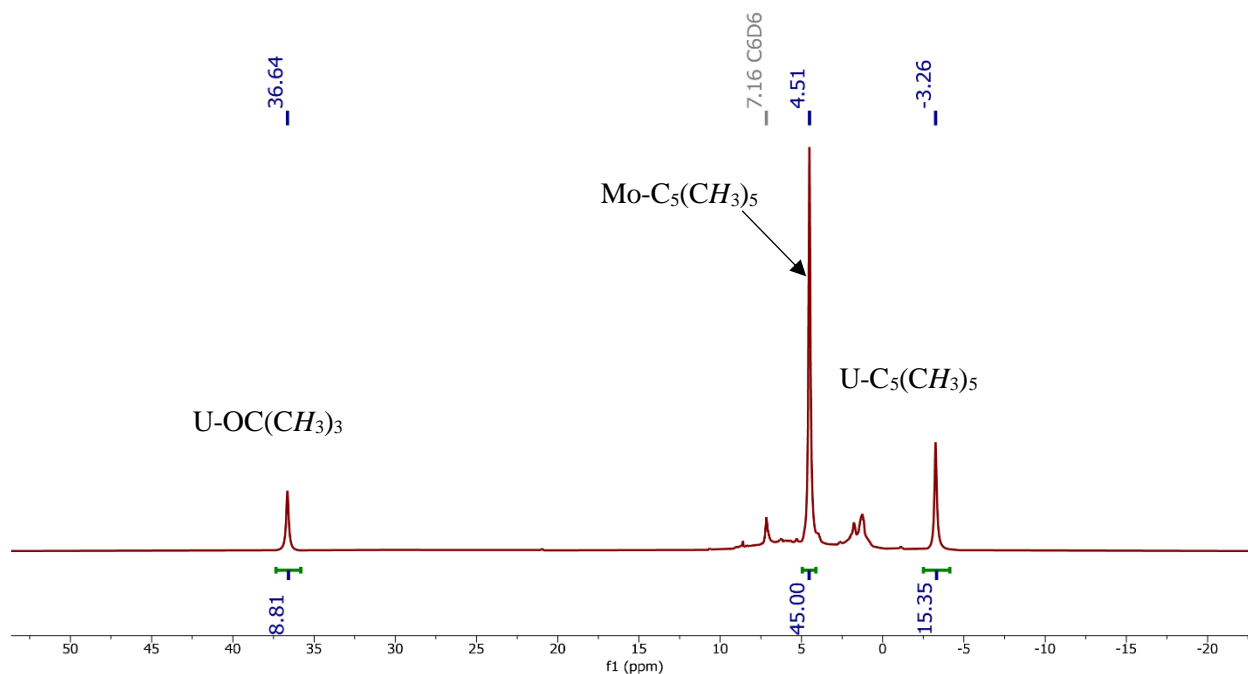


Figure S16. ^1H NMR spectrum (500 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}'\text{Bu})$ in C_6D_6 .

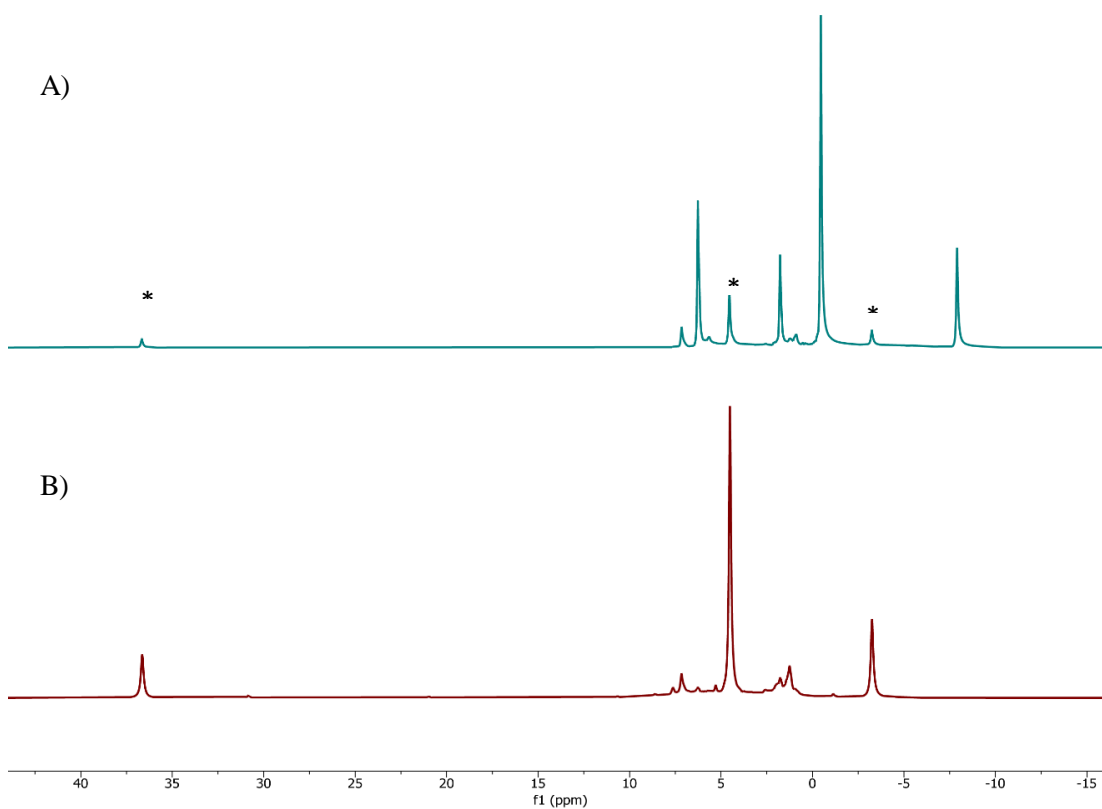


Figure S17. Stacked ^1H NMR spectra (500 MHz) of A) the reaction mixture containing $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{UCp}^*$ and 1 equivalent of *tert*-butanol and B) $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}'\text{Bu})$ in C_6D_6 [* signals represent the minor component $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{O}'\text{Bu})$ in the reaction mixture (A)].

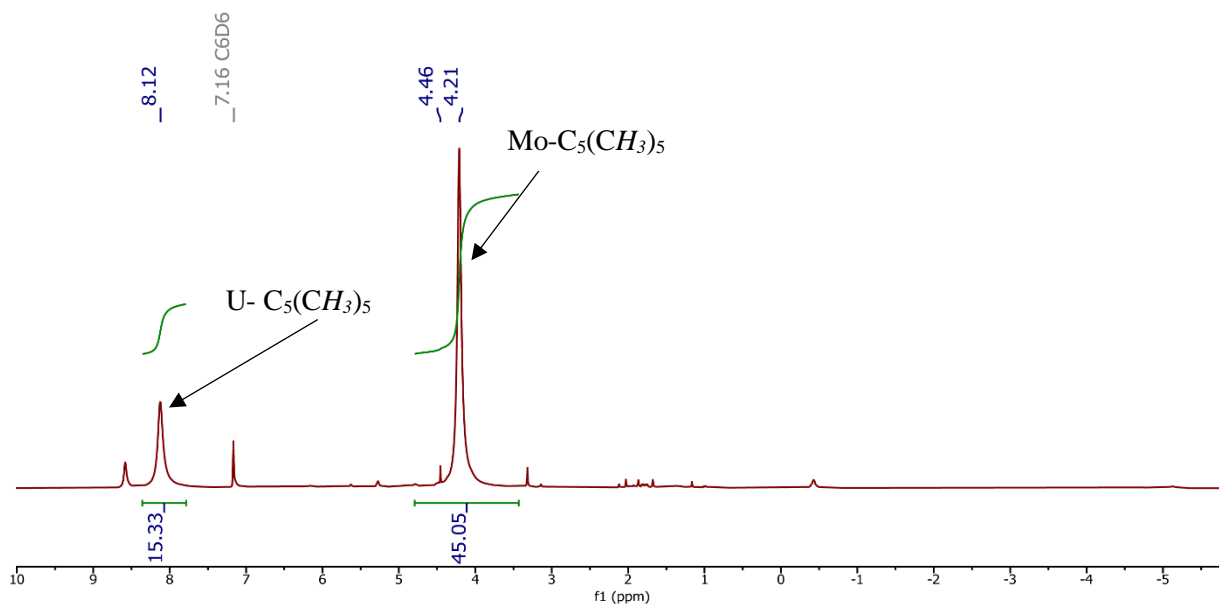


Figure S18. ^1H NMR spectrum (400 MHz) of $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OC}(\text{CF}_3)_3)$ in C_6D_6 (The signal at 4.46 ppm is assigned to H_2).

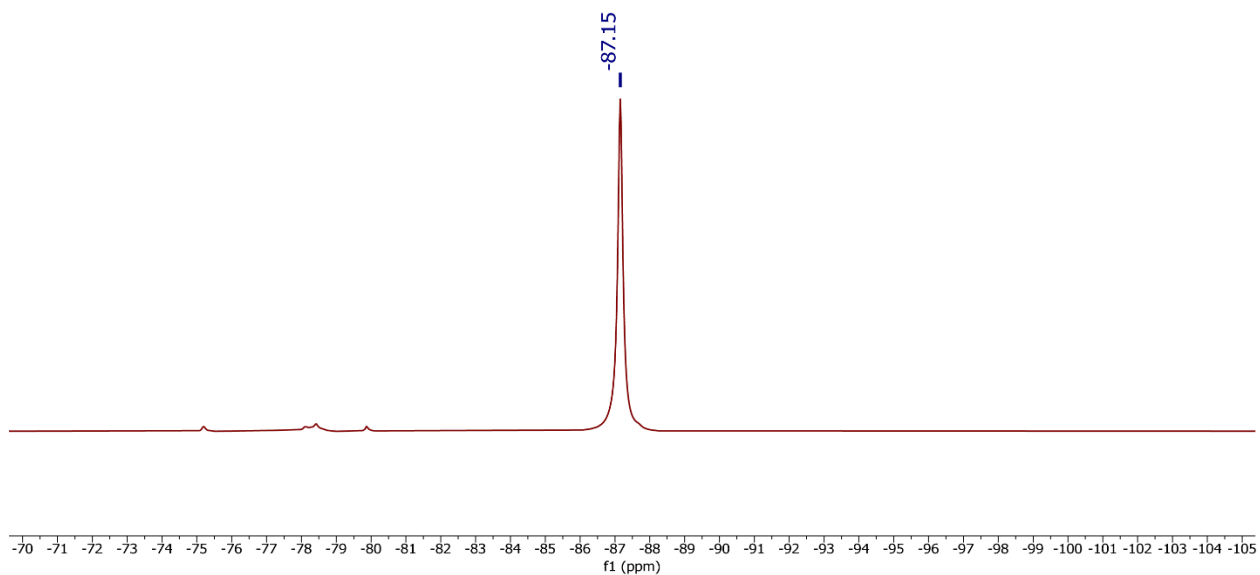


Figure S19. $^{19}\text{F}\{^1\text{H}\}$ NMR spectrum of the $(\text{Cp}^*_3\text{Mo}_3\text{S}_4)\text{Cp}^*\text{U}(\text{OC}(\text{CF}_3)_3)$ in C_6D_6 .