

Figure S1 (a) TEM image, line scan elemental analysis spectrum of (b) Ti, (c) Fe, (d) O elements in MIL-100(Fe)/TiO₂.

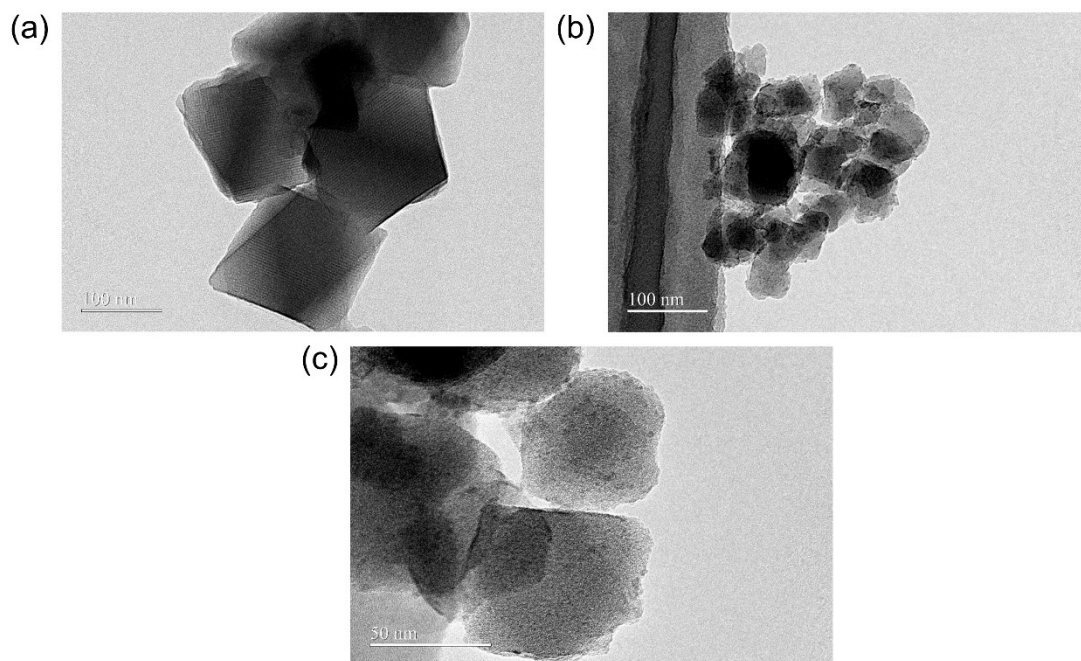


Figure S2 TEM images of MIL-100(Fe)/TiO₂ synthesized at (a) 140 °C and (b and c) 280 °C.

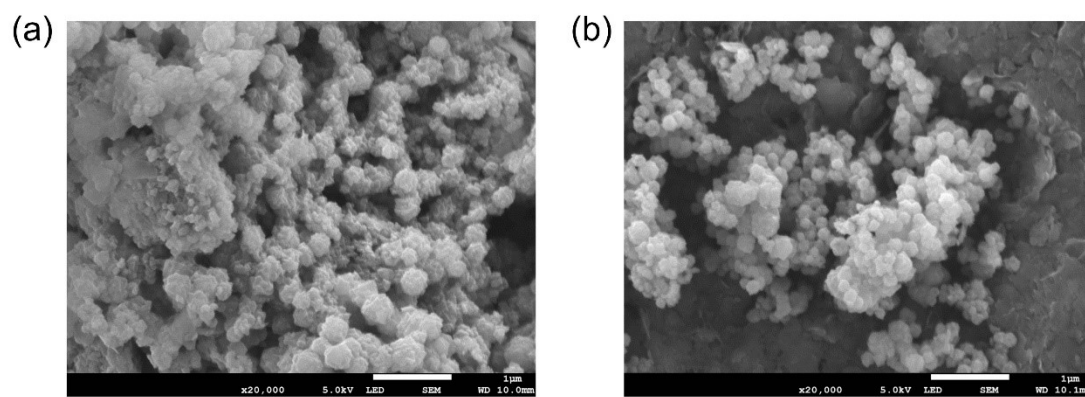


Figure S3 SEM images of MIL-100(Fe)/TiO₂/CoO_x after (a) 4 and (b) 18 hours of photocatalytic reaction.

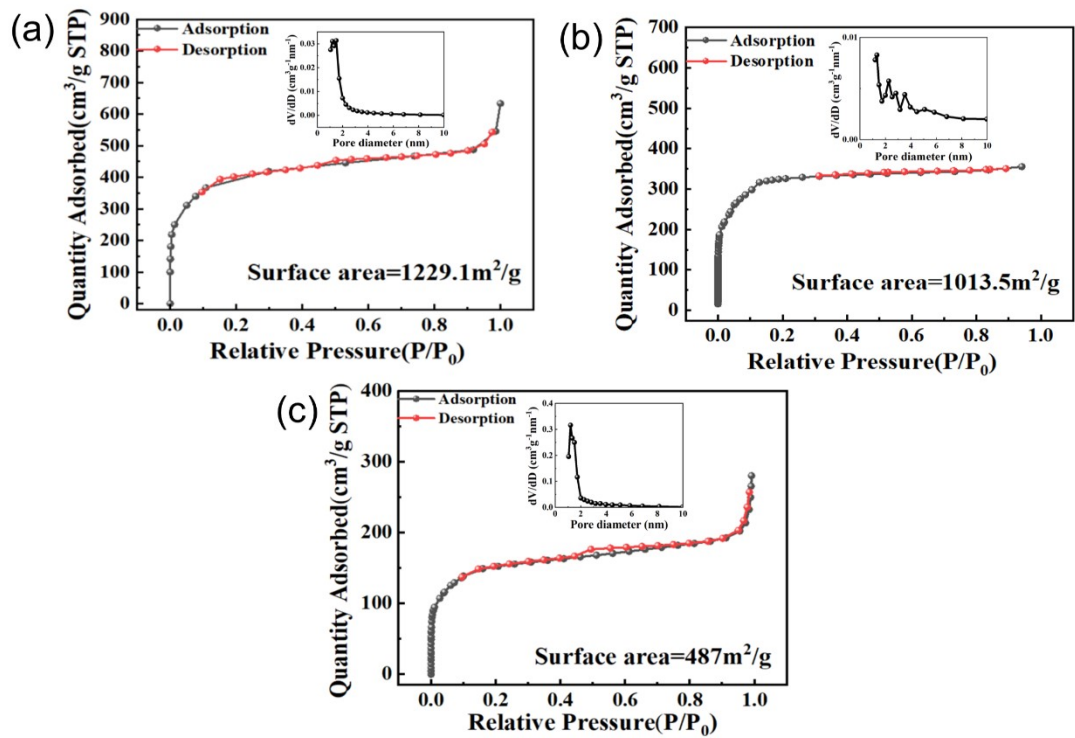


Figure S4. N_2 adsorption-desorption isotherms of (a) MIL-100(Fe), (b) MIL-100(Fe)/TiO₂ (c) MIL-100(Fe)/TiO₂/CoO_x samples.

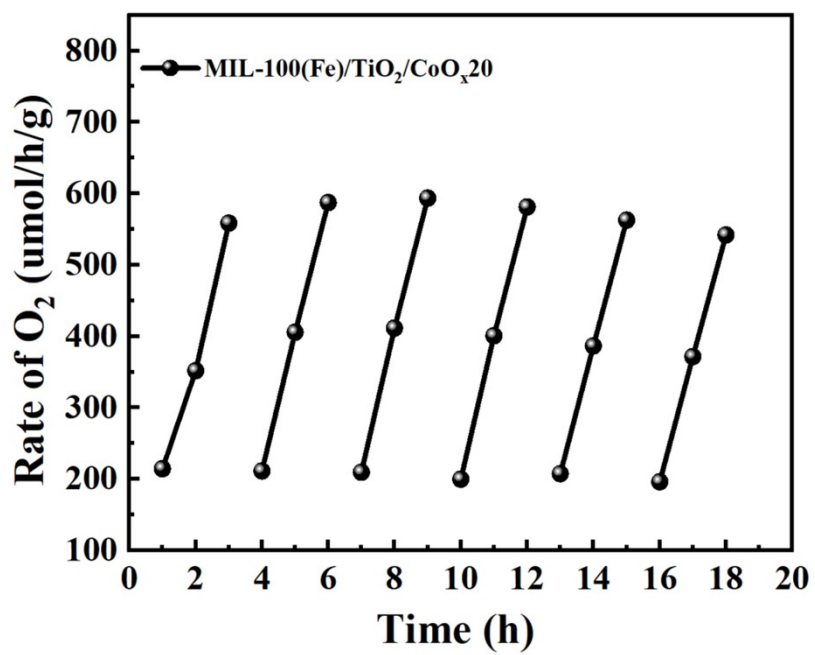


Figure S5 Reusability results of the O₂ production for the MIL-100(Fe)/TiO₂/CoO_x20 sample.

Table S1 Comparison of photocatalytic oxygen evolution performance of MIL-100(Fe)/TiO₂/CoO_x with other reported photocatalysts.

Photocatalysts	Condition	OER ($\mu\text{mol h}^{-1} \text{g}^{-1}$)	Refs
MIL100(Fe)/TiO ₂ /CoO _x	10 mg, 0.1 M FeCl ₃ , 300 W Xe lamp as the light source	558.3	This work
Ov-WO ₃ -Pt	50 mg, 0.1 M Na ₂ S ₂ O ₈ , pH 10.5 adjusted by NaOH	683	[1]
CN/ α -ZnO	under simulated sunlight irradiation	532.4	[2]
Ag ₃ PO ₄ /TiO ₂	50 mg catalyst in 50 mL solution under illumination by a 300W Xe-lamp	537	[3]
FeOOH/BiOBr	300 W simulated sunlight using AgNO ₃ as the sacrificial agent	284.34	[4]
TiO ₂ /C HNTs	under simulated solar light irradiation	539	[5]
5% CoO _x / α -Fe ₂ O ₃	visible light ($\lambda > 400 \text{ nm}$) using AgNO ₃ as the electron scavenger	195.19	[6]
Cu/ZIS/TiO ₂	300 W Xe lamp as the light source	292.75	[7]
r-TiO ₂	5 mg of catalyst in 30 mL of 0.02 M AgNO ₃ solution under a 300 W Xe lamp	71.9	[8]
Ni SA-NG/SrTiO ₃ (Al)/CoO _x	Full arc irradiation	230	[9]
RuO _x /CoO _x /Ta ₃ N ₅	50 mg, 0.1 M Na ₂ S ₂ O ₈ , under visible light irradiation ($420 < \lambda < 800 \text{ nm}$)	159	[10]

Reference

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