Supporting information

Fluorescent Zinc(II) Thione and Selone Complexes for Light-Emitting Applications

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Figure S1: FT-IR spectrum of 1 at RT (KBr).



Figure S2: ¹H NMR spectrum of 1 in DMSO- d_6 at RT.



Figure S3: ¹³C NMR spectrum of 1 in DMSO- d_6 at RT.



Figure S4: FT-IR spectrum of 2 at RT (KBr).



Figure S5: ¹H NMR spectrum of 2 in DMSO- d_6 at RT.



Figure S6: ¹³C NMR spectrum of 2 in DMSO- d_6 at RT.



Figure S7: FT-IR spectrum of 3 at RT (KBr).



Figure S8: ¹H NMR spectrum of 3 in DMSO- d_6 at RT.



Figure S9: 13 C NMR of 3 in DMSO- d_6 at RT.



Figure S10: FT-IR spectrum of 4 at RT (KBr).



Figure S11: ¹H NMR spectrum of **4** in DMSO- d_6 at RT.



Figure S12: ¹³C NMR spectrum of 4 in DMSO- d_6 at RT.



Figure S13: Crystal structure of 1 (Hydrogen atoms are omitted for clarity).



Figure S14: Crystal structure of 2 (Hydrogen atoms are omitted for clarity).



Figure S15: Crystal structure of 3 (Hydrogen atoms are omitted for clarity).



Figure S16: Crystal structure of 4 (Hydrogen atoms are omitted for clarity).



Figure S17: PXRD pattern of 1 at RT (Experimental vs simulated).



Figure S18: PXRD pattern of 2 at RT (Experimental vs simulated).



Figure S19: PXRD pattern of 3 at RT (Experimental vs simulated).



Figure S20: PXRD pattern of 4 at RT (Experimental vs simulated).



Figure S21: Packing diagram of 1 showing intermolecular interactions.



Figure S22: Packing diagram of 2 showing intermolecular interactions.



Figure S23: Packing diagram of 3 showing intermolecular interactions.



Figure S24: Packing diagram of 4 showing intermolecular interactions.



Figure S25: Topological analysis of intramolecular S···H interactions in **1.** (I) $\rho(\mathbf{r}) = 0.0145$, $\nabla^2 \rho(\mathbf{r}) = 0.048$; (II) $\rho(\mathbf{r}) = 0.004$, $\nabla^2 \rho(\mathbf{r}) = 0.0122$; (III) $\rho(\mathbf{r}) = 0.0038$, $\nabla^2 \rho(\mathbf{r}) = 0.0114$; (IV) $\rho(\mathbf{r}) = 0.0136$, $\nabla^2 \rho(\mathbf{r}) = 0.0455$.



Figure S26: Topological analysis of intramolecular S····H interactions in **2.** (I) $\rho(\mathbf{r}) = 0.0146$, $\nabla^2 \rho(\mathbf{r}) = 0.0481$; (II) $\rho(\mathbf{r}) = 0.0146$, $\nabla^2 \rho(\mathbf{r}) = 0.0481$.



Figure S27: Topological analysis of intramolecular S····H interactions in **3.** (I) $\rho(\mathbf{r}) = 0.0146$, $\nabla^2 \rho(\mathbf{r}) = 0.0483$; (II) $\rho(\mathbf{r}) = 0.0146$, $\nabla^2 \rho(\mathbf{r}) = 0.0482$.



Figure S28: Topological analysis of intramolecular Se…H interactions in **4.** (I) $\rho(\mathbf{r}) = 0.0135$, $\nabla^2 \rho(\mathbf{r}) = 0.0404$; (II) $\rho(\mathbf{r}) = 0.0139$, $\nabla^2 \rho(\mathbf{r}) = 0.0405$.



Figure S29: Thermogravimetric analysis of complexes 1-4 from 40-790 °C at a heating range of 10 °C/min under an inert environment.



Figure S30: UV-Vis absorption spectra of L^1 and L^2 in acetonitrile at RT (C = 1 x 10⁻⁵ M).



Figure S31: (a) Emission spectra of L^1 and L^2 at RT in solution state using acetonitrile (C = 1 x 10⁻⁵ M), (b) Emission spectra of L^1 and L^2 in the crystalline state at RT, (c) Emission spectra of L^1 vs 1 in acetonitrile using the same concentration at RT, (d) Emission spectra of L^1 vs 1 in the crystalline state at RT, (e) Crystal images of L^1 and L^2 under ambient light vs under UV light.



Figure S32: UV-Vis absorption spectra of 1(a), 2(b), 3(c), and 4(d) in acetonitrile (black solid line), DMF (red solid line), and CHCl₃ (blue solid line) (C = 1 x 10⁻⁵ M).



Figure S33: Emission spectra of **1** (a), **2**(b), **3**(c), and **4**(d) in acetonitrile (black solid line), DMF (red solid line), and CHCl₃ (blue solid line).



Figure S34: Emission spectra of 1(a), 2(b), 3(c), and 4(d) in DMF (C = 10^{-5} M) with increasing water concentration from 20% to 80% showing an increase in emission intensity upon aggregation, (e) images of solution of 1 in DMF under ambient light and under UV light with increasing water percentage from 0% to 80%.



Figure S35: Photoluminescence decay profiles of 1 in crystalline state.



Figure S36: Photoluminescence decay profiles of 2 in crystalline state.



Figure S37: Photoluminescence decay profiles of 3 in crystalline state.



Figure S38: Photoluminescence decay profiles of 4 in crystalline state.



Figure S39: DFT Optimized structures of (i) 1, (ii) 2, (iii) 3, and (iv) 4.



Figure S40: DFT calculated UV-Vis spectra (gas phase) of 1(a), 2(b), 3(c), and 4(d).

Complex	$\lambda_{\text{calc.}}(\text{nm})$	Hole	Electron
1	S_1 w = 0.73 3.1724 (0.2061) 390.82 nm	8 X	B C C
	S_2 w = 0.75 3.1881 (0.064) 388.89 nm		S. AS
2	S ₁ w=0.52 3.1737 (0.1483) 390.67 nm		of the
	S_2 w = 0.52 3.1869 (0.0868) 389.04 nm		
3	S ₁ w = 0.51 3.1755 (0.1772) 390.43 nm		
	S ₂ w =0.51 3.1881 (0.064) 388.89 nm		
4	S ₁ w =0.64 3.1605 (0.1086) 392.29 nm		
	S ₂ w =0.64 3.1754 (0.126) 390.45 nm		

Figure S41: Natural transition orbitals (NTOs) of complexes 1-4 illustrate the active singlet excited states in the absorption bands. NTO pairs that contribute more than 50% to each excited state.

complexes 1	-4 compared to	o that of ligands	s L^1 and L^2 in 1	DMSO- d_6 .	

Table S1: Change in the chemical shift values of the C=S and C=Se Carbon atom in

Ligand	$C=S Carbon (\delta_c ppm)$	Complex	$C=S Carbon (\delta_c ppm)$	Shift in δ_c (ppm)
\mathbf{L}^{1}	167.37 ppm	1	160.42 ppm	6.95 ppm
	167.37 ppm	2	160.42 ppm	6.95 ppm
	167.37 ppm	3	160.44 ppm	6.93 ppm
L ²	160.76 ppm	4	154.84 ppm	6.92 ppm

Compound	τ, ns	τ1, ns	τ2, ns	RA1	RA2	decay	Emission
							nm
1	4.581	1.32	5.11	4330.37	1001.38	bi-	455
						exponential	
2	8.502	2.47	7.03	1893.38	854.30	bi-	475
						exponential	
3	18.08	6.55	21.37	1283	1379.3	bi-	535
						exponential	
4	0.95	0.37	0.99	289.62	246.45	bi-	425
						exponential	

 Table S2. Fitting parameters of fluorescence decay of 1-4. (RAx = Relative amplitude of x-th component)

Identification code	1	2	3	4
Empirical formula	$\begin{array}{c} C_{42}H_{40}Cl_2N_4S_2\\ Zn \end{array}$	$C_{42}H_{40}Br_2N_4S_2Zn$	$C_{42}H_{40}I_2N_4S_2Zn$	C _{43.5} H _{41.5} Cl ₂ N ₅ Se ₂ Zn
Formula weight	801.17	890.09	984.07	928.567
Temperature/K	273.15	273.15	273.15	298.15
Crystal system	monoclinic	triclinic	monoclinic	triclinic
Space group	C2/c	P-1	$P2_1/c$	P-1
a/Å	39.7664(15)	9.7737(15)	10.1471(17)	9.833(4)
b/Å	10.2851(4)	12.6125(19)	18.739(3)	12.903(5)
c/Å	20.4074(8)	18.399(3)	22.137(4)	18.383(8)
$\alpha/^{\circ}$	90	78.893(5)	90	78.137(14)
β/°	105.501(2)	77.020(5)	98.777(6)	76.450(14)
γ°	90	74.820(5)	90	74.168(14)
Volume/Å ³	8043.1(5)	2111.3(6)	4160.0(12)	2156.3(15)
Ζ	8	2	4	2
$\rho_{calc}g/cm^3$	1.323	1.400	1.571	1.430
μ/mm^{-1}	0.882	2.606	2.209	2.417
F(000)	3328.0	904.0	1952.0	940.6
Crystal size/mm ³	0.27 × 0.21 × 0.17	$0.28 \times 0.18 \times 0.17$	0.2 imes 0.17 imes 0.16	$0.28 \times 0.17 \times 0.17$
Dediction	MoK α (λ =	MoK α (λ =	MoK α (λ =	Mo K α (λ =
Kadiation	0.71073)	0.71073)	0.71073)	0.71073)
2⊖ range for data collection/°	4.1 to 54.302	3.806 to 54.382	3.724 to 54.702	4.32 to 53.1
	$-48 \le h \le 50, -$	$-12 \le h \le 12, -16$	$-13 \le h \le 12, -23$	$-12 \le h \le 11, -16$
Index ranges	$13 \le k \le 13, -$	\leq k \leq 16, -23 \leq l \leq	\leq k \leq 24, -28 \leq l \leq	\leq k \leq 16, -23 \leq l \leq
	$26 \le l \le 26$	23	28	22
Reflections collected	90769	42432	148885	41203
Independent	$8911 [R_{int} =$	$9377 [R_{int} =$	$9316 [R_{int} =$	$8945 [R_{int} =$
reflections	$0.0736, R_{sigma} = 0.02661$	$0.0894, R_{sigma} = 0.07871$	$0.1543, R_{sigma} = 0.06201$	$0.0518, R_{sigma} = 0.04271$
Data/restraints/parame ters	8911/0/464	9377/0/464	9316/0/464	8945/0/492
Goodness-of-fit on F ²	1.030	1.036	1.016	1.015
Final R indexes	$R_1 = 0.0388$.	$R_1 = 0.0555$,	$R_1 = 0.0471$.	$R_1 = 0.0375.$
[I>=2σ (I)]	$wR_2 = 0.0907$	$wR_2 = 0.1249$	$wR_2 = 0.0908$	$wR_2 = 0.0857$
Final R indexes [all	$R_1 = 0.0648,$	$R_1 = 0.1152,$	$R_1 = 0.0995,$	$R_1 = 0.0626,$
data]	$wR_2 = 0.1028$	$wR_2 = 0.1476$	$wR_2 = 0.1076$	$wR_2 = 0.0959$
Largest diff. peak/hole / e Å ⁻³	1.02/-0.36	1.06/-1.18	0.88/-0.95	0.69/-0.46

 Table S3. The structural parameters of complexes 1-4.

	1	2	3	4			
Bond Length [Å]							
M1-E1	2.3882(6)	2.3687(13)	2.3857(13)	2.4854(10)			
M1-E2	2.3808(7)	2.3794(13)	2.3725(13)	2.5049(8)			
M1-X1	2.2354(7)	2.3862(9)	2.5613(7)	2.2567(13)			
M1-X2	2.2433(7)	2.3869(8)	2.6113(7)	2.2541(12)			
		Bond Angle	[°]				
E1-M1-E2	109.86(2)	108.64(5)	106.49(5)	104.26(17)			
X1-M1-X2	115.84(3)	115.32(3)	117.24(2)	114.21(5)			
E1-M1-X1	111.94(3)	105.70(4)	108.75(4)	107.99(3)			
E2-M1-X2	110.48(3)	105.38(4)	104.89(4)	107.40(3)			

 Table S4. Selected bond lengths and angles of complexes 1-4.

Here, M = Zn(II)

E = S/Se

 $\mathbf{X} = \mathbf{C}\mathbf{l} \ / \ \mathbf{Br} \ / \ \mathbf{I}$

Table S5: Key parameters (HOMO/LUMO, band gap, energy level location) from DFTcalculations of 1-4.

Compound	HOMO	LUMO	HOMO-1	LUMO+1	Band gap	Singlet	Triplet
	(eV)	(eV)	(eV)	(eV)	(eV)	(eV)	(eV)
1	-5.317	-1.817	-5.324	-1.816	3.50	2.5661	2.4770
2	-5.322	-1.840	-5.345	-1.818	3.48	2.2487	2.1343
3	-5.350	-1.846	-5.352	-1.846	3.50	2.4147	2.3659
4	-5.338	-1.846	-5.343	-1.843	3.49	2.4507	2.3763



 Table S6: Excitation energies and oscillator strengths of 1.

No.	Energy	Wavelength	Osc. Strength	Symmetry	Major contribs
	(eV)	(nm)			
1	1.7510	708.09	0.0	Triplet-A	HOMO->L+1 (95%)
2	1.7565	705.87	0.0	Triplet-A	H-1->LUMO (95%)
3	3.1724	390.82	0.2061	Singlet-A	H-1->LUMO (24%),
					HOMO->L+1 (74%)
4	3.1873	388.99	0.0482	Singlet-A	H-1->LUMO (72%),
					HOMO->L+1 (22%)
5	3.2571	380.66	0.0	Triplet-A	H-12->L+1 (54%),
					HOMO->L+7 (31%)
6	3.2613	380.16	0.0	Triplet-A	H-13->LUMO (54%),
					H-1->L+6 (32%)
7	3.2839	377.56	0.0	Triplet-A	HOMO->LUMO (97%)
8	3.2839	377.55	0.0	Singlet-A	HOMO->LUMO (97%)
9	3.2935	376.45	0.0	Triplet-A	H-1->L+1 (97%)
10	3.2936	376.45	0.0	Singlet-A	H-1->L+1 (98%)
11	3.4449	359.91	0.0	Triplet-A	H-8->L+1 (80%)
12	3.4518	359.19	0.0	Triplet-A	H-9->LUMO (78%)
13	3.5308	351.15	0.0002	Singlet-A	H-2->L+1 (90%)
14	3.5395	350.28	0.0003	Singlet-A	H-2->LUMO (89%)
15	3.7990	326.36	0.001	Singlet-A	H-3->L+1 (84%)
16	3.8239	324.23	0.0005	Singlet-A	H-4->LUMO (15%),
					H-3->LUMO (74%)

 Table S7: Excitation energies and oscillator strengths of 2.

No.	Energy (eV)	Wavelength (nm)	Osc. Strength	Symmetry	Major contributors
1	1.7512	708.00	0.0	Triplet-A	H-1->LUMO (97%)
2	1.7512	707.98	0.0	Triplet-A	HOMO->L+1 (97%)
3	3.1737	390.67	0.1483	Singlet-A	H-1->LUMO (45%),
					HOMO->L+1 (53%)
4	3.1869	389.04	0.0868	Singlet-A	H-1->LUMO (53%),
					HOMO->L+1 (45%)
5	3.2306	383.78	0.0	Triplet-A	HOMO->LUMO (100%)
6	3.2307	383.77	0.0001	Singlet-A	HOMO->LUMO (100%)
7	3.2547	380.93	0.0	Triplet-A	H-14->LUMO (56%),
					H-1->L+6 (23%)
8	3.2554	380.86	0.0	Triplet-A	H-13->L+1 (56%),
					HOMO->L+7 (21%)
9	3.2752	378.55	0.0	Triplet-A	H-1->L+1 (100%)
10	3.2753	378.55	0.0	Singlet-A	H-1->L+1 (100%)
11	3.4422	360.19	0.0	Triplet-A	H-11->LUMO (83%)
12	3.4425	360.16	0.0	Triplet-A	H-10->L+1 (83%)
13	3.4796	356.32	0.0001	Singlet-A	H-2->LUMO (92%)
14	3.4941	354.84	0.0001	Singlet-A	H-2->L+1 (93%)
15	3.5934	345.03	0.0	Singlet-A	H-3->LUMO (99%)
16	3.6117	343.28	0.0	Singlet-A	H-3->L+1 (100%)

 Table S8: Excitation energies and oscillator strengths of 3.

No.	Energy (eV)	Wavelength (nm)	Osc. Strength	Symmetry	Major contributors
1	1.7519	707.72	0.0	Triplet-A	H-1->L+1 (38%), HOMO->L+1 (47%)
2	1.7519	707.71	0.0	Triplet-A	H-1->LUMO (47%), HOMO->LUMO (37%)
3	3.1755	390.43	0.1772	Singlet-A	H-1->LUMO (10%), H-1->L+1 (37%), HOMO->LUMO (40%), HOMO->L+1 (11%)
4	3.1881	388.89	0.064	Singlet-A	H-1->LUMO (41%), HOMO->L+1 (44%)
5	3.2166	385.45	0.0	Triplet-A	H-2->LUMO (90%)
6	3.2169	385.41	0.0	Triplet-A	H-2->L+1 (90%)
7	3.2177	385.32	0.0008	Singlet-A	H-2->LUMO (85%), H-2->L+1 (14%)
8	3.2177	385.32	0.0009	Singlet-A	H-2->LUMO (14%), H-2->L+1 (85%)
9	3.2576	380.60	0.0	Triplet-A	H-13->L+1 (47%)
10	3.2576	380.60	0.0	Triplet-A	H-14->LUMO (46%)
11	3.2665	379.57	0.0	Triplet-A	H-1->LUMO (36%), HOMO->LUMO (48%)
12	3.2668	379.57	0.0	Singlet-A	H-1->LUMO (40%), HOMO->LUMO (46%)
13	3.2668	379.53	0.0	Triplet-A	H-1->L+1 (46%), HOMO->L+1 (39%)
14	3.2668	379.52	0.0	Singlet-A	H-1->L+1 (50%), HOMO->L+1 (36%)
15	3.3598	369.03	0.0	Singlet-A	H-3->LUMO (86%), H-3->L+1 (14%)
16	3.3605	368.95	0.0001	Singlet-A	H-3->LUMO (14%), H-3->L+1 (86%)

 Table S9: Excitation energies and oscillator strengths of 4.

No.	Energy	Wavelength	Osc. Strength	Symmetry	Major contributors
1	(ev)	(nm)		T. 1 / A	
1	1.7496	708.66	0.0	Triplet-A	H-1->LUMO (46%),
					HOMO->LUMO (50%)
2	1.7507	708.21	0.0	Triplet-A	H-1->L+1 (50%),
					HOMO->L+1 (45%)
3	3.1605	392.29	0.1086	Singlet-A	H-1->LUMO (24%),
					H-1->L+1 (18%),
					HOMO->LUMO (40%),
					HOMO->L+1 (16%)
4	3.1754	390.45	0.126	Singlet-A	H-1->LUMO (16%),
					H-1->L+1 (26%),
					HOMO->LUMO (19%),
					HOMO->L+1 (37%)
5	3.2224	384.75	0.0	Triplet-A	H-1->LUMO (50%),
					HOMO->LUMO (49%)
6	3.2231	384.67	0.0008	Singlet-A	H-1->LUMO (60%),
					HOMO->LUMO (39%)
7	3.2248	384.47	0.0	Triplet-A	H-1->L+1 (46%),
					HOMO->L+1 (53%)
8	3.2253	384.41	0.0006	Singlet-A	H-1->L+1 (55%),
					HOMO->L+1 (45%)
9	3.2508	381.40	0.0	Triplet-A	H-12->LUMO (54%)
10	3.2541	381.01	0.0	Triplet-A	H-13->L+1 (55%)
11	3.3448	370.68	0.0	Triplet-A	H-2->LUMO (93%)
12	3.3567	369.37	0.0005	Singlet-A	H-2->LUMO (95%)
13	3.3597	369.04	0.0	Triplet-A	H-2->L+1 (92%)
14	3.3736	367.52	0.0004	Singlet-A	H-2->L+1 (95%)
15	3.6494	339.74	0.0004	Singlet-A	H-4->LUMO (17%),
					H-3->LUMO (77%)
16	3.6518	339.52	0.0006	Singlet-A	H-4->L+1 (85%),
					H-3->L+1 (11%)

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