Rationally designed Cu(II) Schiff base metal complex anchored on NiFe<sub>2</sub>O<sub>4</sub>@Chitosan: An efficient heterogeneous and magnetically retrievable hybrid nanocatalyst for the one-pot multi-component synthesis of bioactive 1-amidoalkyl-2-naphthols

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# **Supplementary Information**

# **Experimental**

# Materials and methods

FeCl<sub>3</sub>.6H<sub>2</sub>O, NiCl<sub>2</sub>.6H<sub>2</sub>O, 4-(diethylamino)-salicylaldehyde, Cu(OAc)<sub>2</sub>.H<sub>2</sub>O, ethanol, and Chitosan(CS)were acquired from SRL Chemical without undergoing further purification.

With KBr pellets, IR (infrared)spectra were obtained on a Bruker 3000 Hyperion Microscope that was set up with a Vertex 80 FT-IR system. The crystal phases were investigated using a Panalytical Xpert3 X-ray diffractometer that was outfitted with Cu-K $\alpha$  radiation ( $\lambda = 1.54$  Å). Jeol 6390LA/OXFORD XMX N SEM instrument was utilized to confirm the surface structure of the composite. With the JEOL Model 2100 F, TEM images are captured.Nova Touch LX<sub>2</sub> gas sorption analyzer from Quantachrome Instruments was used to measure the BET surface area and pore size distribution of materials. Microsense EV7 Vibrating Sample Magnetometer was used to detect the magnetic properties. X-ray photoelectron spectroscopy (XPS) was performed using a PHI 5000 VERSA Probe III (ULVAC PHI, USA). Inductively Coupled Plasma- Atomic Emission Spectroscopy (ICP-AES) was performed using ARCOS, Simultaneous ICP spectrometer.



Fig. S1. (a)  $N_2$  adsorption/desorption isotherm and (b) BJH plot for pore-size distribution o  $NiFe_2O_4@CS@CuSB$ 















Fig.S2(d). <sup>1</sup>H NMR spectra of 4d



Fig.S2(e). <sup>1</sup>H NMR spectra of 4e



Fig.S2(f). <sup>1</sup>H NMR spectra of 4f



Fig.S2(g). <sup>1</sup>H NMR spectra of 4g



Fig.S2(h). <sup>1</sup>H NMR spectra of 4h







Fig.S2(j). <sup>1</sup>H NMR spectra of 4j











Fig.S3(b). IR spectra of 4b



Fig.S3(d). IR spectra of 4d













Fig.S3(j). IR spectra of 4k



Fig.S3(k). IR spectra of 4l

## Spectral analysis of the synthesized 1-amidoalkyl-2-napthol derivatives

#### 4a. 1-(naphthalen-1-yl(phenyl)methyl)urea

Appearance- white solid, M.P: 218-220 °C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 9.71(s, 1H, OH), 7.06-7.77 (m, 14H, Ar-H, NH, NH<sub>2</sub>), 5.41 (s, 1H, CH); IR (υ, cm<sup>-1</sup>): 3448, 3227, 1630, 1586, 1497, 1444, 1157, 953, 812

#### 4b. 1-((2-hydroxyphenyl)(naphthalen-1-yl)methyl)urea

Appearance- white solid, M.P: 215-217°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 9.71(s, 2H, 2OH), 7.07-7.76 (m, 13H, Ar-H, NH, NH<sub>2</sub>), 5.40 (s, 1H, CH); IR (υ, cm<sup>-1</sup>): 3454, 3222, 1623, 1583, 1503, 1454, 1159, 957, 810

#### 4c. 1-((2-hydroxy-3-methoxyphenyl)(naphthalen-1-yl)methyl)urea

Appearance- yellow solid, M.P: 221-225°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>)  $\delta$  (ppm): 10.26 (s, 1H, OH), 9.81(s, 1H, OH), 6.61-8.00 (m, 12H, Ar-H, NH, NH<sub>2</sub>), 5.97 (s, 1H, CH); 3.82 (s, 3H, OCH<sub>3</sub>); IR ( $\upsilon$ , cm<sup>-1</sup>):3467, 3214, 1627, 1584, 1536, 1443, 1245, 1158, 1022, 822, 735

# 4d.1-(naphthalen-1-yl(4-nitrophenyl)methyl)urea

Appearance- brown, M.P: 200-202°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 10.15 (s, 1H, OH), 7.06-8.42 (m, 13H, Ar-H, NH, NH<sub>2</sub>), 5.45 (s, 1H, CH); IR (υ, cm<sup>-1</sup>):3456, 3216, 1631, 1586, 1509, 1341, 1269, 1214,1160, 1060, 806, 738

## 4e. 1-((4-chlorophenyl)(naphthalen-1-yl)methyl)urea

Appearance- yellow solid, M.P: 160-163°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 9.75 (s, 1H, OH), 7.06-8.19 (m, 13H, Ar-H, NH, NH<sub>2</sub>), 5.42 (s, 1H, CH); IR (υ, cm<sup>-1</sup>): 3446, 3334, 3231, 1589, 1504, 1459, 1379, 1267, 1160, 950, 816, 744

# 4f. 1-((4-methoxyphenyl)(naphthalen-1-yl)methyl)urea

Appearance-white, M.P: 205-208

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 9.86 (s, 1H, OH), 6.70-7.88 (m, 13H, Ar-H, NH, NH<sub>2</sub>), 5.42 (s, 1H, CH), 3.76 (s, 3H, OCH<sub>3</sub>); IR (υ, cm<sup>-1</sup>):3448, 3336, 3226, 1627, 1572, 1501, 1443, 1260, 1152, 1065, 808, 571

## 4g. 2-(naphthalen-1-yl(ureido)methyl)benzoic acid

Appearance- white, M.P: 176-179°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 10.45(s, 1H, COOH), 9.72 (s, 1H, OH), 6.97-7.83 (m, 13H, Ar-H, NH, NH<sub>2</sub>), 5.41 (s, 1H, CH);

## 4h. 1-((4-bromo-2-hydroxyphenyl)(naphthalen-1-yl)methyl)urea

Appearance- grey sticky solid, M.P: 207-209 °C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 10.23(s, 1H, OH), 9.74 (s, 1H, OH), 6.83-7.83 (m, 13H, Ar-H, NH, NH<sub>2</sub>), 5.43 (s, 1H, CH).

# 4i. 1-((2-hydroxy-4-methoxyphenyl)(naphthalen-1-yl)methyl)urea

Appearance- yellow solid, M.P: 198-200°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 9.72(s, 1H, OH), 6.62-7.77 (m, 12H, Ar-H, NH, NH<sub>2</sub>), 5.41 (s, 1H, CH), 3.90 (s, 3H, OCH<sub>3</sub>); IR (υ, cm<sup>-1</sup>):3339, 3223, 1625, 1581, 1267, 1209, 1161, 897, 803, 748

## 4j. N-(naphthalen-1-yl(phenyl)methyl)benzamide

Appearance- white, M.P: 223-226°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 10.01(s, 1H, OH), 7.29-8.53 (m, 17H, Ar-H, NH, NH<sub>2</sub>), 5.49 (s, 1H, CH); IR (υ, cm<sup>-1</sup>): 3445, 3334, 3220, 1582, 1496, 1451, 1262, 1150, 948, 814, 735

# 4k. N-((2-hydroxyphenyl)(naphthalen-1-yl)methyl)benzamide

Appearance- grey, M.P: 230-234°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 9.81 (s, 2H, 2OH), 7.05-8.00 (m, 16H, Ar-H, NH, NH<sub>2</sub>), 5.51 (s, 1H, CH); IR (υ, cm<sup>-1</sup>): 3462, 3323, 3231, 1603, 1525, 1466, 1231, 1156, 1064, 815, 732, 581

## 4l. N-((2-hydroxy-3-methoxyphenyl)(naphthalen-1-yl)methyl)benzamide

Appearance- grey, M.P: 227-230°C

<sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>) δ (ppm): 10.28(s, 1H, OH), 9.73(s, 1H, OH), 6.91-7.98 (m, 15H, Ar-H, NH, NH<sub>2</sub>), 5.46 (s, 1H, CH), 3.84 (s, 3H, OCH<sub>3</sub>); IR (υ, cm<sup>-1</sup>):3245, 3173, 1635, 1595, 1402, 1211, 946, 810, 731