

## *Supplementary material*

# *Multicolor emission from hybrid 0D-doped halide $(C_8H_{14}N_2)_2CdBr_6$ single crystal*

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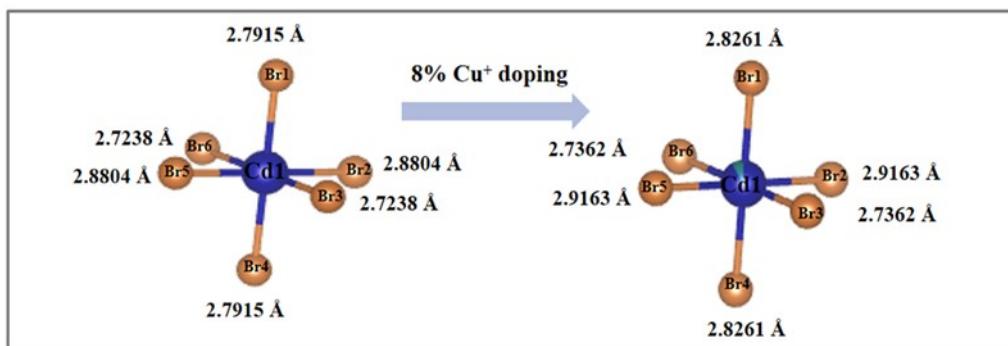
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**Table S1** Crystal data for  $C_{16}H_{28}N_4CdBr_6$

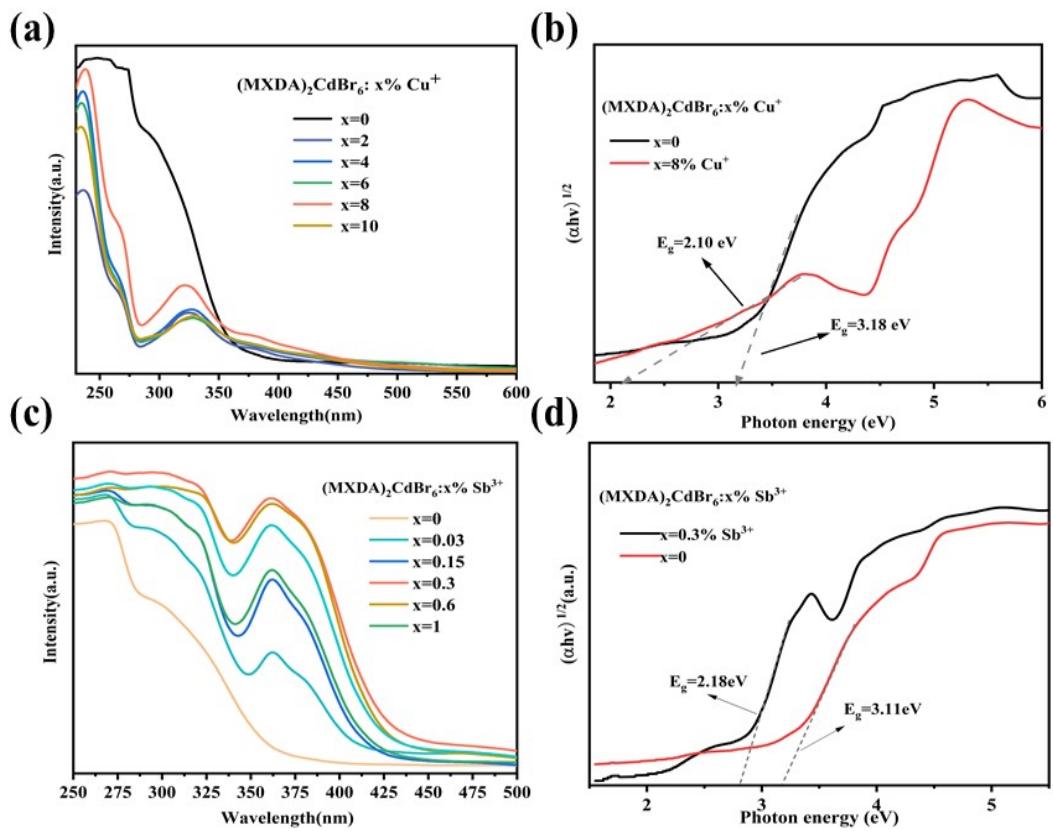
Compound	$MXDA_2CdBr_6$
Chemical formula	$C_{16}H_{28}N_4CdBr_6$
Fw	868.28
Space group	P1
a/ $\text{\AA}$	7.961(5)
b/ $\text{\AA}$	8.835(5)
c/ $\text{\AA}$	10.498(6)
$\alpha/^\circ$	100.288(18)
$\beta/^\circ$	101.344(18)
$\gamma/^\circ$	112.861(17)
Volume	640.0(6)

**Table S2** Rietveld refinement of the  $(MXDA)_2CdBr_6:M^{n+}$  samples

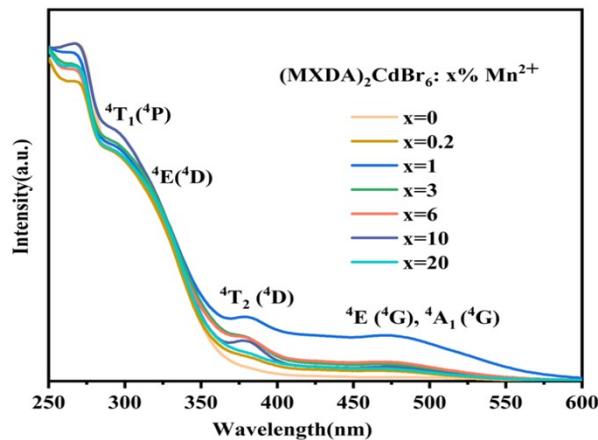
Doped ion (M <sup>n+</sup> )	Space group	A (Å)	B (Å)	C (Å)	Cell volume (Å <sup>3</sup> )	R <sub>wp</sub>	R <sub>p</sub>
M=Cu <sup>+</sup>	P 1	7.940	8.835	10.48 3	638.0	12.01 %	9.05 %
M=Mn <sup>2+</sup>	P 1	7.959	8.835	10.49 6	639.4	11.23 %	8.03 %
M=Sb <sup>3+</sup>	P 1	7.950	8.835	10.49 5	639.8	12.53 %	9.43 %



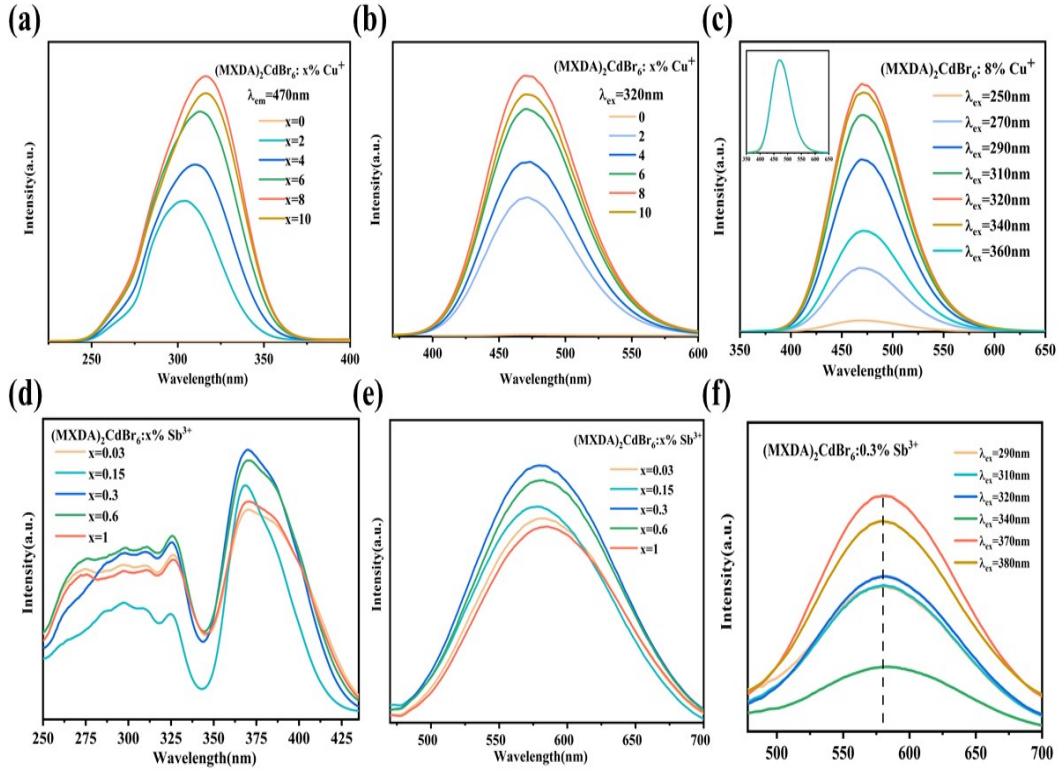
**Figure S1** Variation of Cd-Br bond lengths in [CdBr<sub>6</sub>]<sup>4-</sup>-octahedra in (MXDA)<sub>2</sub>CdBr<sub>6</sub> before and after 8%Cu<sup>+</sup> doping



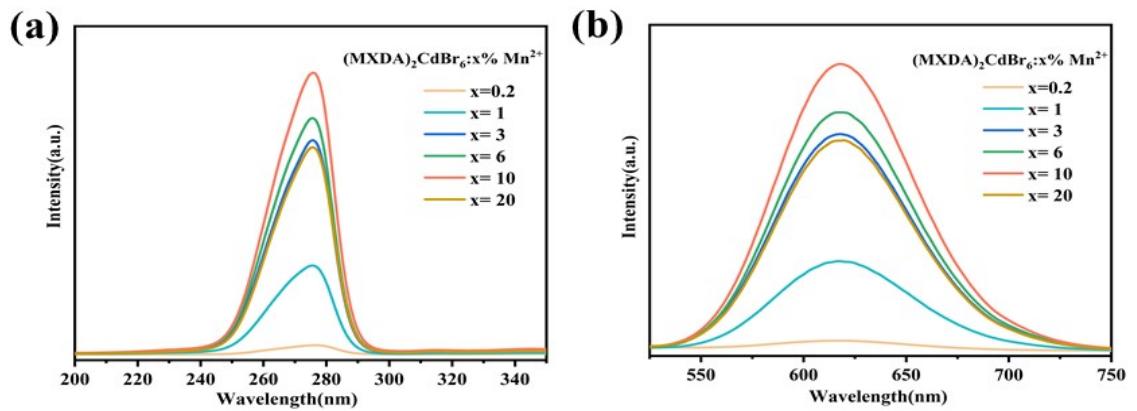
**Figure S2** (a) UV-vis absorption spectra of  $(\text{MXDA})_2\text{CdBr}_6$ : $x\%$   $\text{Cu}^+$  sample, (b) Tauc plots of  $(\text{MXDA})_2\text{CdBr}_6$  and  $(\text{MXDA})_2\text{CdBr}_6$ : $x\%$   $\text{Cu}^+$  samples, (c) UV-Vis absorption spectra of  $(\text{MXDA})_2\text{CdBr}_6$ : $x\%$   $\text{Sb}^{3+}$  ( $x=0-1$ ) sample, and (d) Tauc plots of  $(\text{MXDA})_2\text{CdBr}_6$  and  $(\text{MXDA})_2\text{CdBr}_6$ : $x\%$   $\text{Sb}^{3+}$  samples



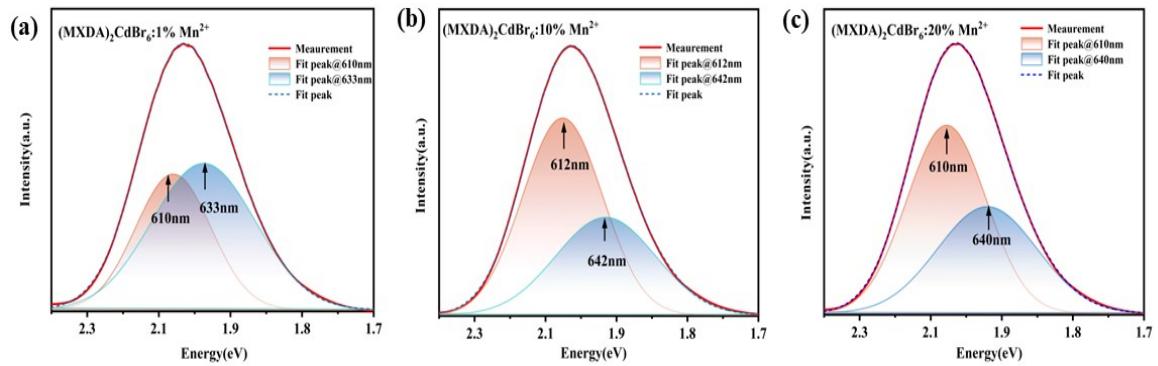
**Figure S3** UV-Vis absorption spectra of  $(\text{MXDA})_2\text{CdBr}_6$ : $x\%$   $\text{Mn}^{2+}$  samples



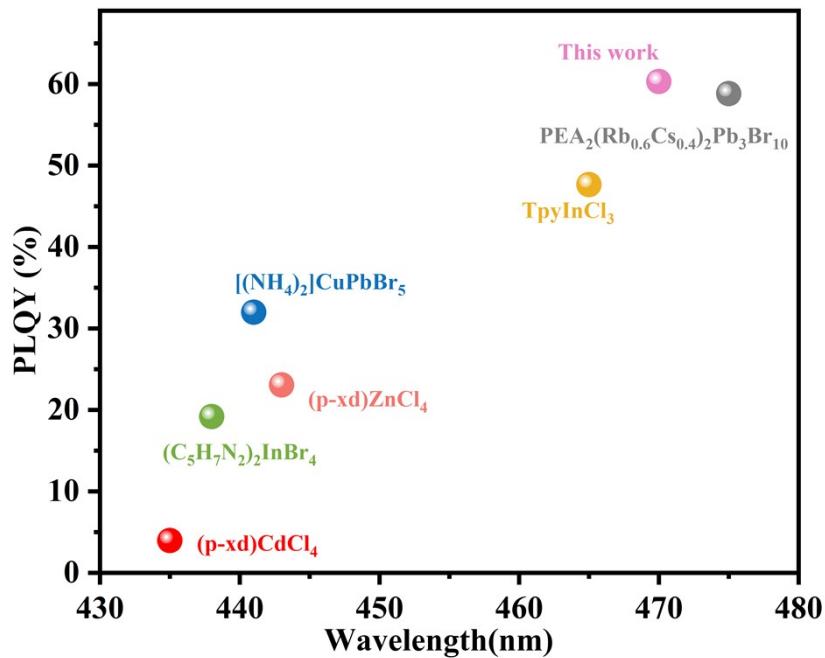
**Figure S4** (a) PLE spectra ( $\lambda_{\text{em}} = 470 \text{ nm}$ ) of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\% \text{Cu}^+$  samples, (b) PL spectra of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\% \text{Cu}^+$  samples ( $\lambda_{\text{ex}} = 370 \text{ nm}$ ), (c) variable excitation spectrum of  $(\text{MXDA})_2\text{CdBr}_6:8\% \text{Cu}^+$  ( $\lambda_{\text{em}} = 250-360 \text{ nm}$ ), (d) PLE spectrum of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\% \text{Sb}^{3+}$  sample ( $\lambda_{\text{em}} = 580 \text{ nm}$ ), (e) PL spectrum of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\% \text{Sb}^{3+}$  sample ( $\lambda_{\text{ex}} = 370 \text{ nm}$ ), and (f) variable excitation spectrum of  $(\text{MXDA})_2\text{CdBr}_6:0.3\% \text{Sb}^{3+}$  ( $\lambda_{\text{em}} = 290-380 \text{ nm}$ )



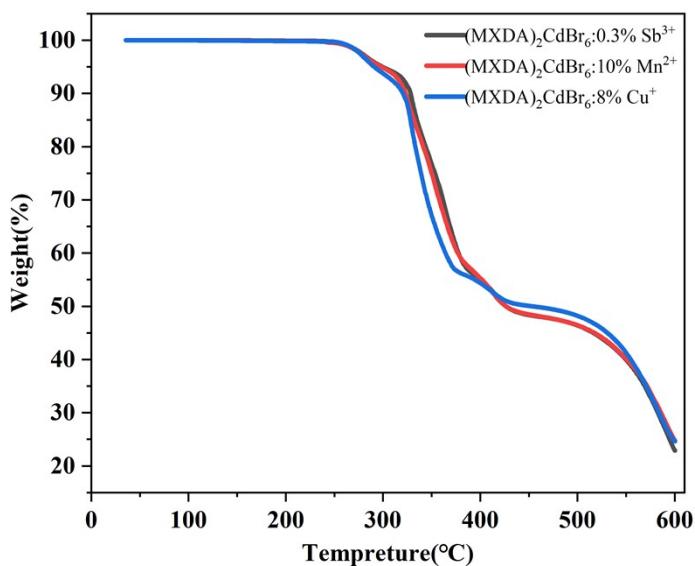
**Figure S5** (a) PLE spectra ( $\lambda_{\text{em}} = 618 \text{ nm}$ ) of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\% \text{Mn}^{2+}$  samples, and (b) PL spectra of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\% \text{Mn}^{2+}$  samples ( $\lambda_{\text{ex}} = 275 \text{ nm}$ ),



**Figure S6** Multi-peak fitting results of PL spectra of  $(\text{MXDA})_2\text{CdBr}_6:\text{x}\%\text{Mn}^{2+}$  ( $\text{x}=1, 10, 20$ )



**Figure S7** PLQY variation of blue light emission from the organic-inorganic hybrid perovskites in previous studies<sup>1–5</sup>



**Figure S8** TGA profiles of  $(\text{MXDA})_2\text{CdBr}_6:\text{M}^{\text{n}+}$  ( $\text{M}^{\text{n}+}=\text{Cu}^+, \text{Sb}^{3+}, \text{Mn}^{2+}$ )

### Reference:

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- [2] E. R. Dohner, A. Jaffe, L. R. Bradshaw and H. I. Karunadasa, J. Am. Chem. Soc., 2014, 136, 13154–13157.
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