

Supplementary Information

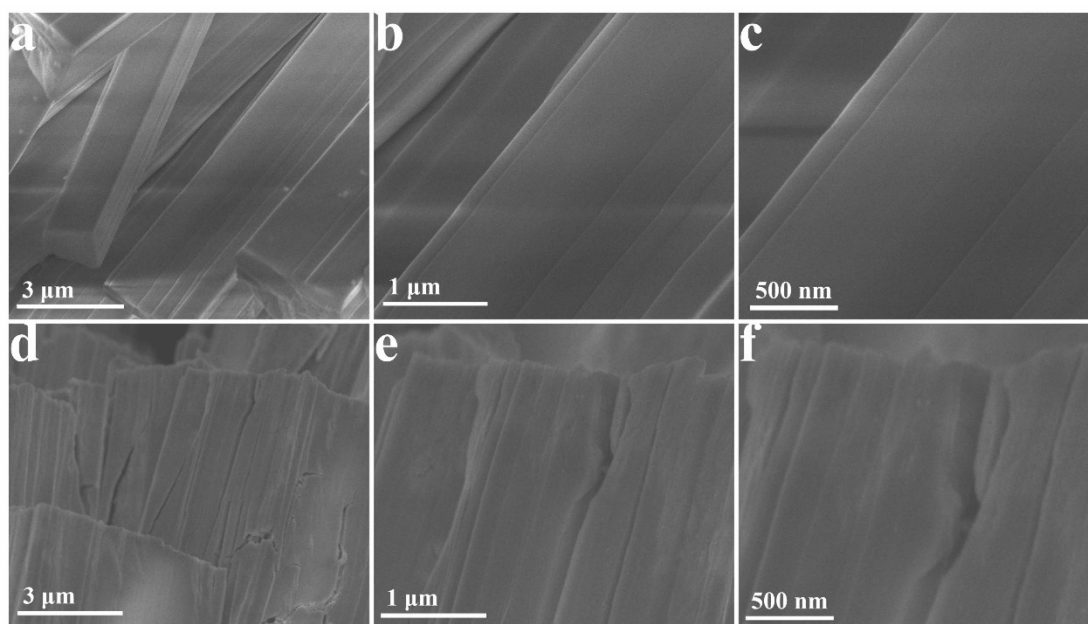


Fig. S1. SEM of (a~c) $\text{CoMoO}_4 \cdot 0.75\text{H}_2\text{O}$; (d~f) CoMoO_4 .

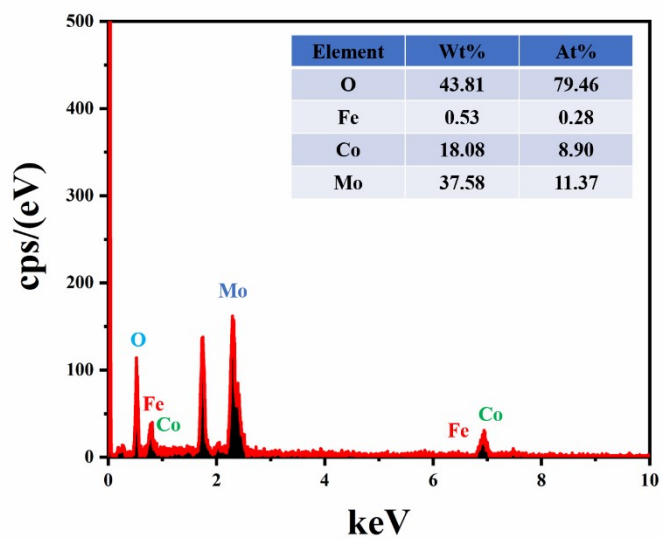


Fig. S2. EDX spectrum of Fe-CoMoO_4 .

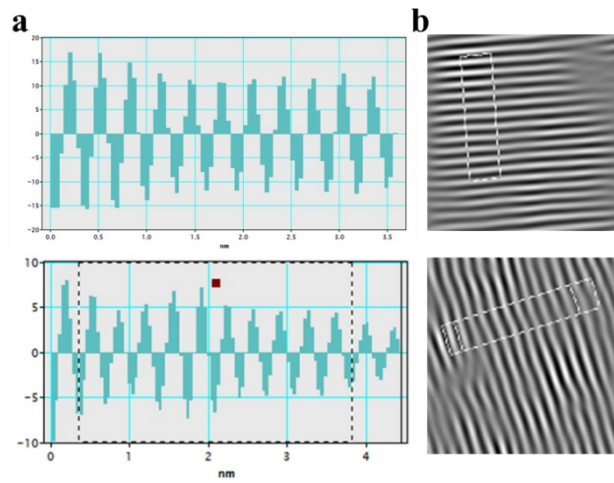


Fig. S3. HRTEM real-time profiles (a) IFFT image (b) of Fe-CoMoO₄ crystal.

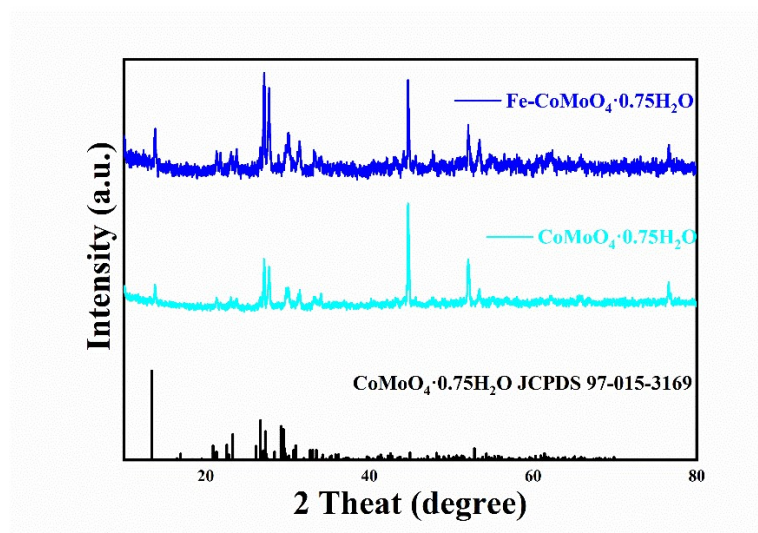


Fig. S4. XRD of Fe-CoMoO₄·0.75H₂O and CoMoO₄·0.75H₂O.

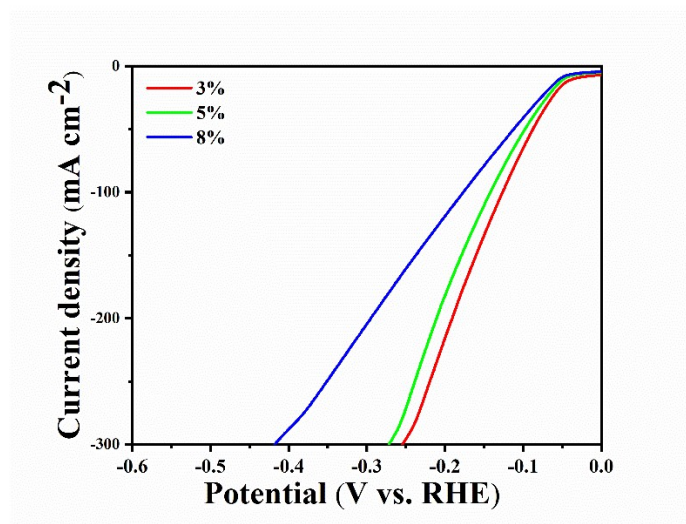


Fig. S5. The HER iR-corrected polarization curves of cobalt molybdate with different Fe doping (3, 5, 8% Fe) amounts 1 M KOH.

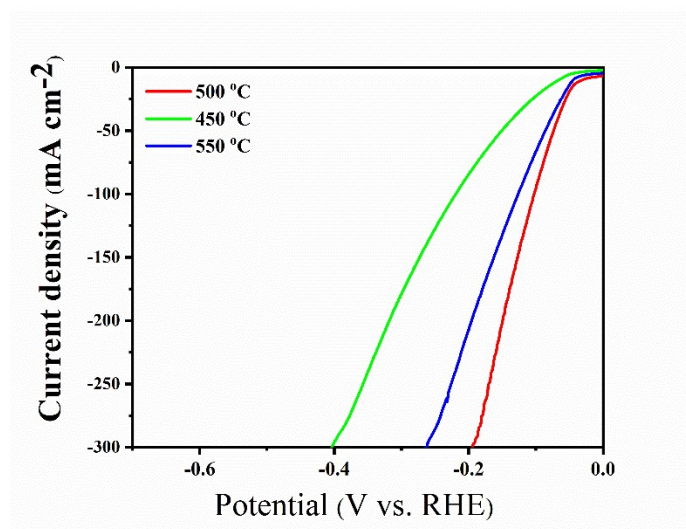


Fig. S6. The HER iR-corrected polarization curves of Fe-doped (3%) cobalt molybdate with different annealing temperatures (450, 500, or 550 °C) in 1 M KOH.

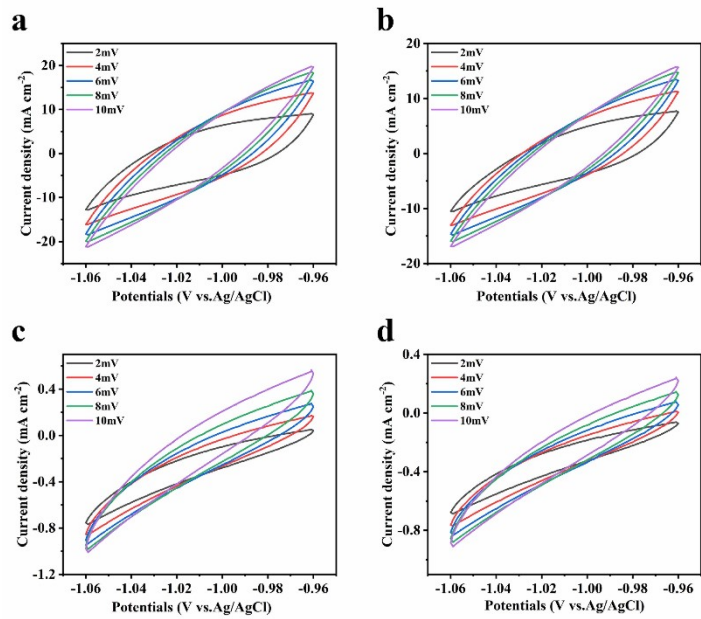


Fig. S7. Cyclic voltammograms curves of (a) Fe-CoMoO₄, (b) CoMoO₄, (c) Fe-CoMoO₄·0.75H₂O, and (d) CoMoO₄·0.75H₂O in region of -0.106~-0.96 V vs. Ag/AgCl at various scan rate.

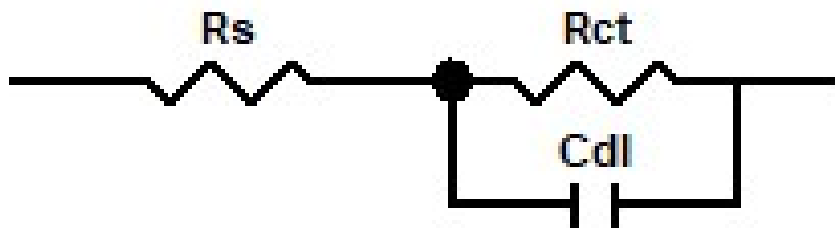


Fig. S8. Equivalent electrical circuit used to model the HER kinetics process, R_s is the solution resistance, R_{ct} is the charge transfer resistance at catalyst/electrolyte interface and C_{dl} is the double layer capacitance of the working electrode.

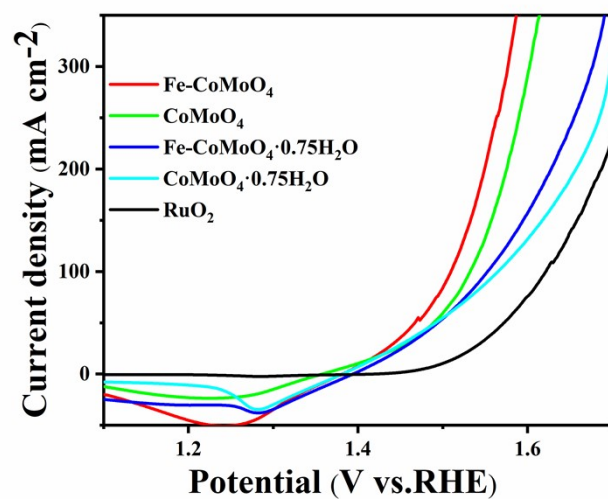


Fig. S9. OER LSV curves of Fe-CoMoO₄, CoMoO₄, Fe-CoMoO₄·0.75H₂O, CoMoO₄·0.75H₂O.

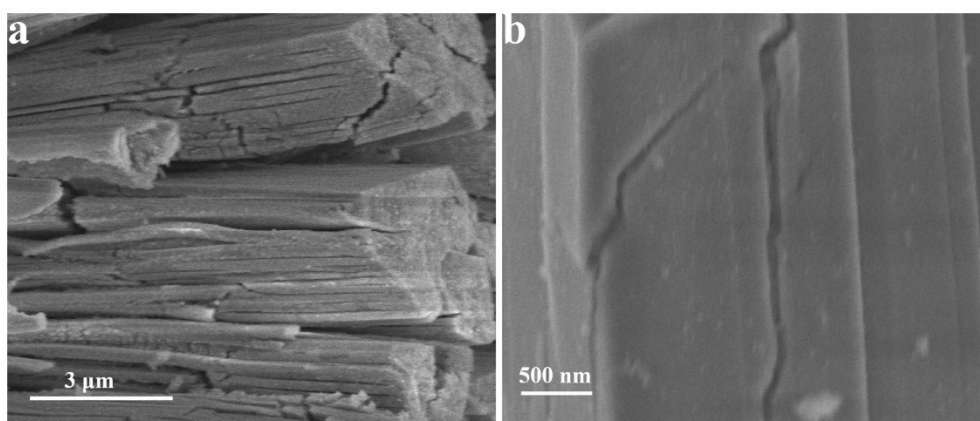


Fig. S10. SEM images of Fe-CoMoO₄ sample after stability test.

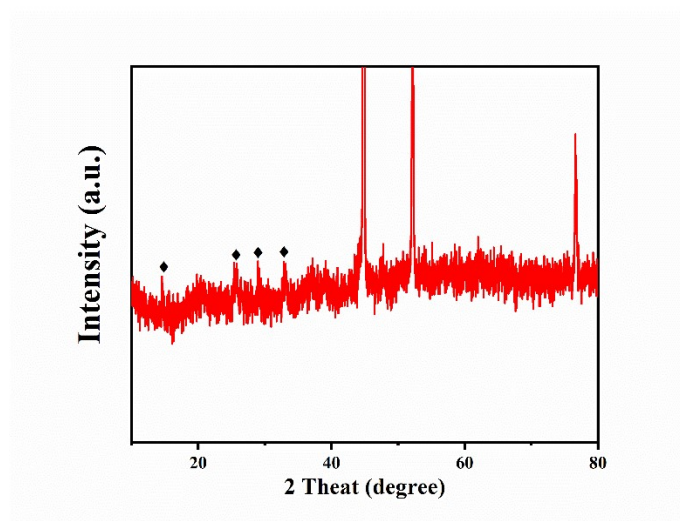


Fig. S11. XRD of Fe-CoMoO₄ sample after stability test.

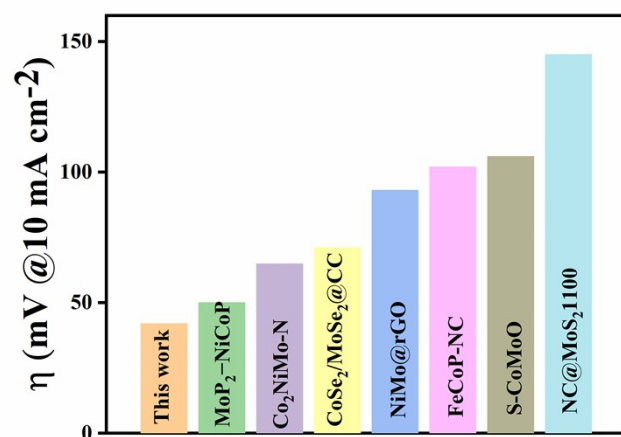


Fig. S12. Comparison of catalytic performance comparison of Fe-CoMoO₄ materials with other highly efficient transition metal-based catalysts in 1 M KOH for HER.