

## Structurally Isomeric Ditopic 2-Mercaptobenzoxazole and 2-Hydroxybenzothiazole as Ligands for Design of 2D Sodium-Based Luminescent Coordination Polymers

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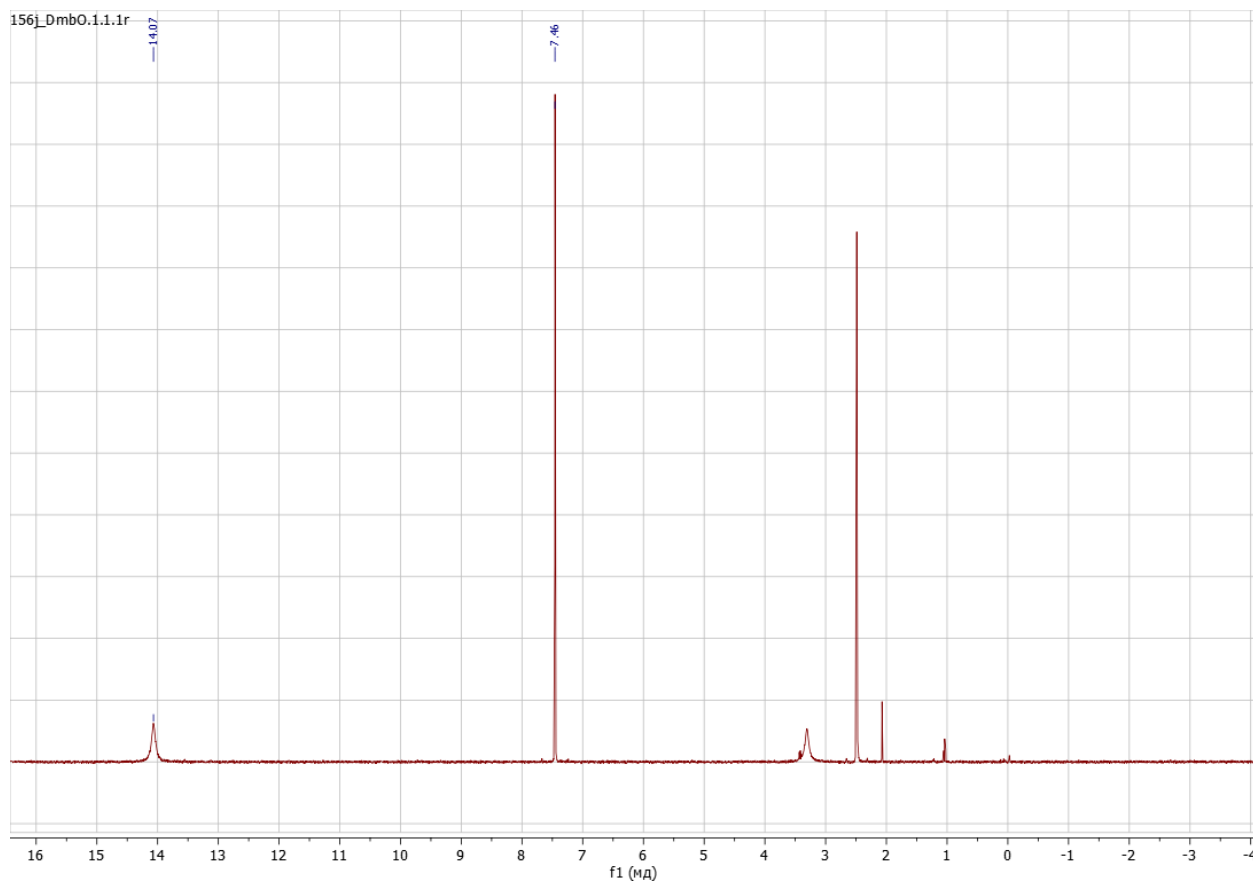


Figure S1.  $^1\text{H}$  NMR spectrum of  $\text{H}_2\text{L}^1$  in  $\text{DMSO-d}_6$ .

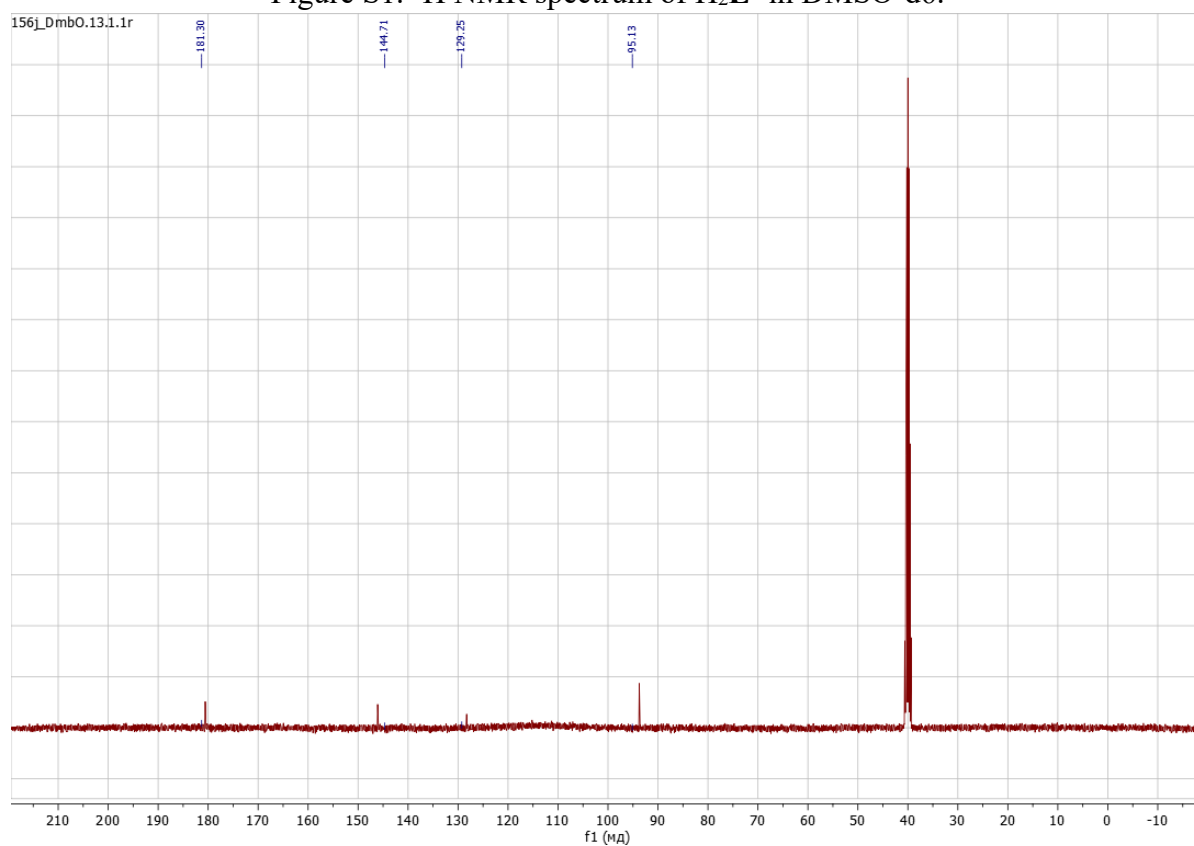


Figure S2.  $^{13}\text{C}$  NMR spectrum of  $\text{H}_2\text{L}^1$  in  $\text{DMSO-d}_6$ .

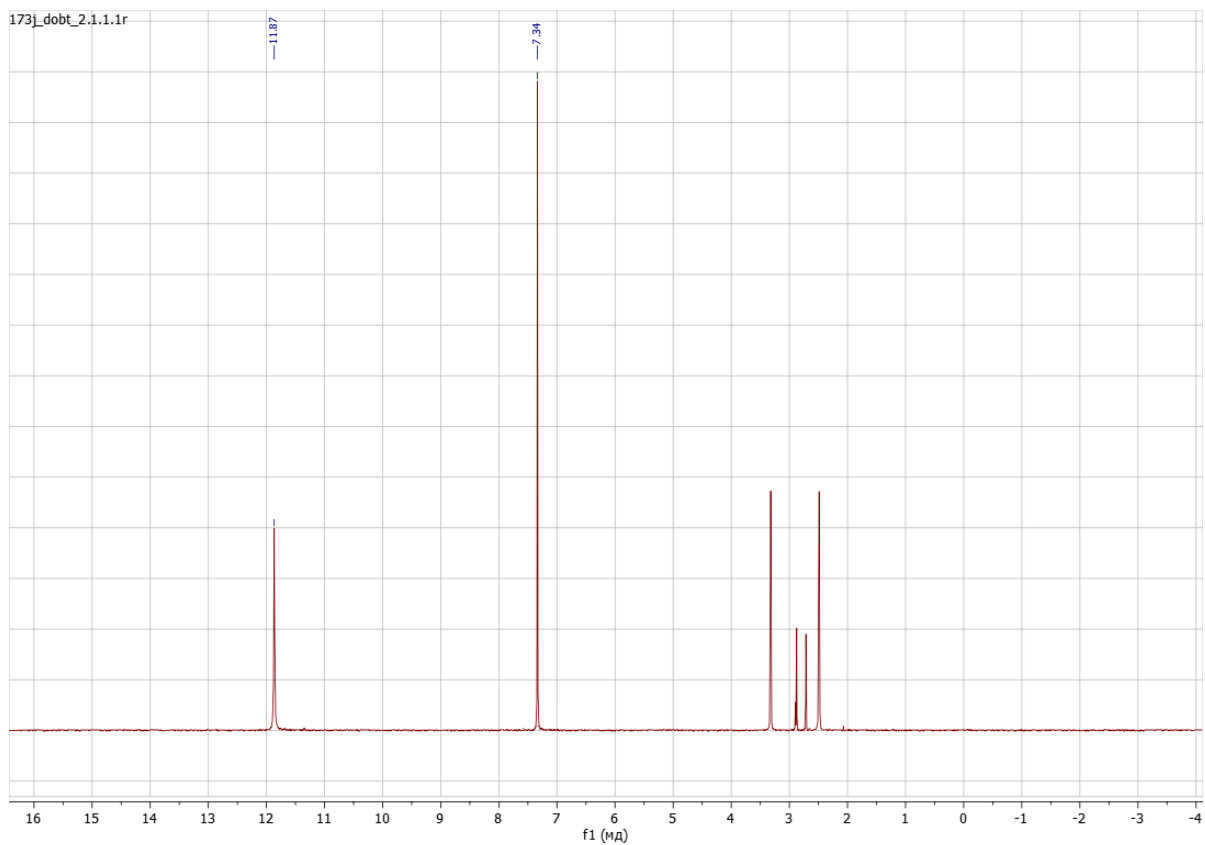


Figure S3.  $^1\text{H}$  NMR spectrum of  $\text{H}_2\text{L}^2$  in  $\text{DMSO-d}_6$ .

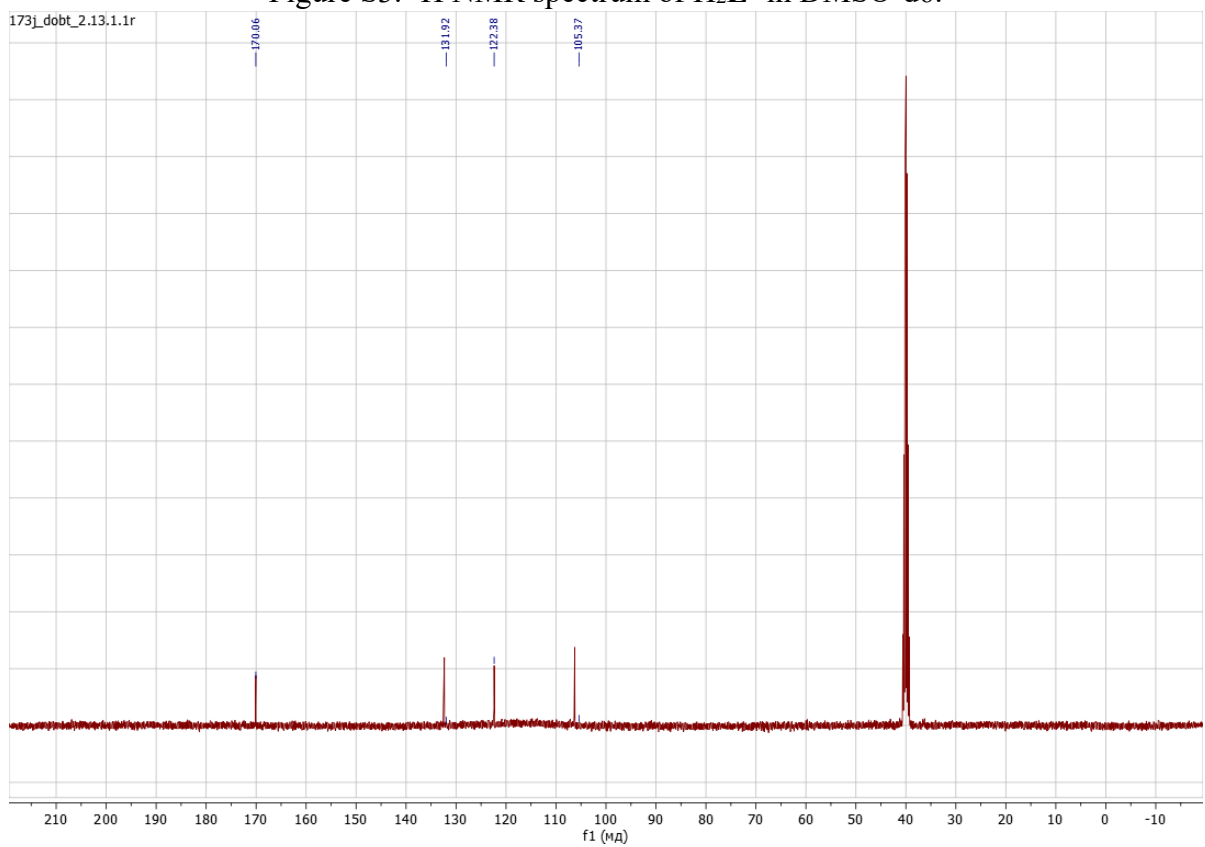


Figure S4.  $^{13}\text{C}$  NMR spectrum of  $\text{H}_2\text{L}^2$  in  $\text{DMSO-d}_6$ .

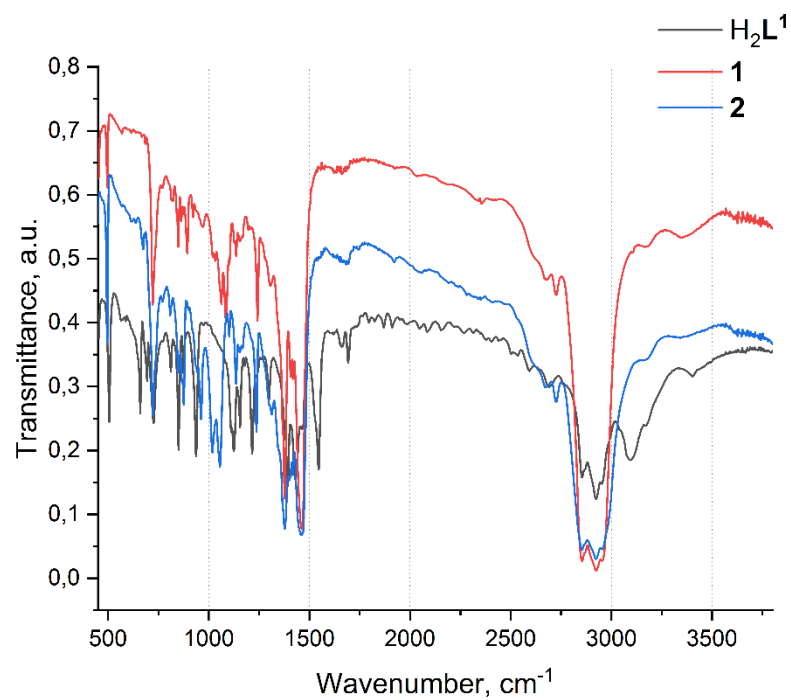


Figure S5. FTIR spectra of H<sub>2</sub>L<sup>1</sup> and CPs **1** and **2**.

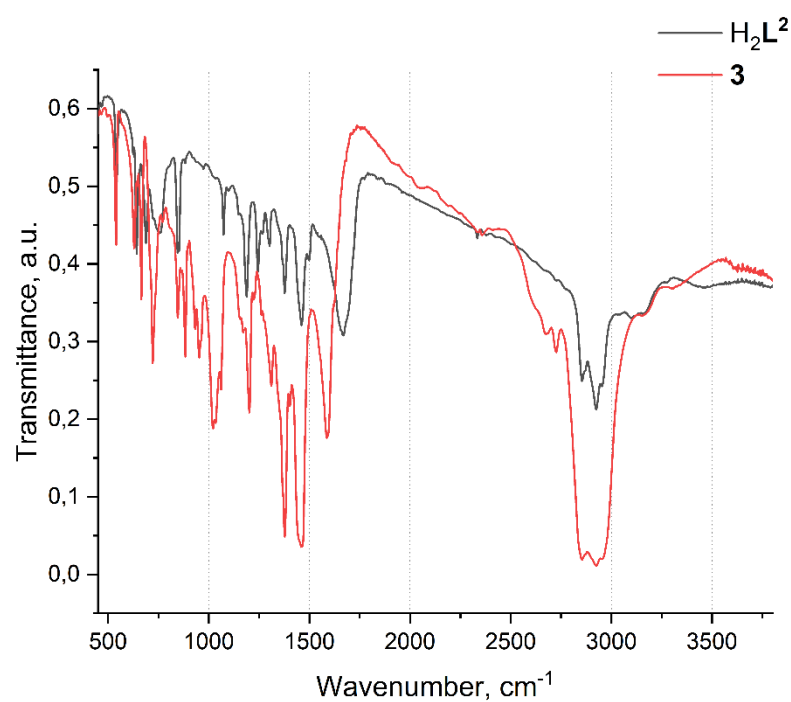


Figure S6. FTIR spectra of H<sub>2</sub>L<sup>2</sup> and CP **3**.

043\_Rogozhin1 #420-455 RT: 3.53-3.82 AV: 36 SB: 75 2.67-3.27 NL: 2.70E5  
T: + c Full ms [ 50.00-500.00]

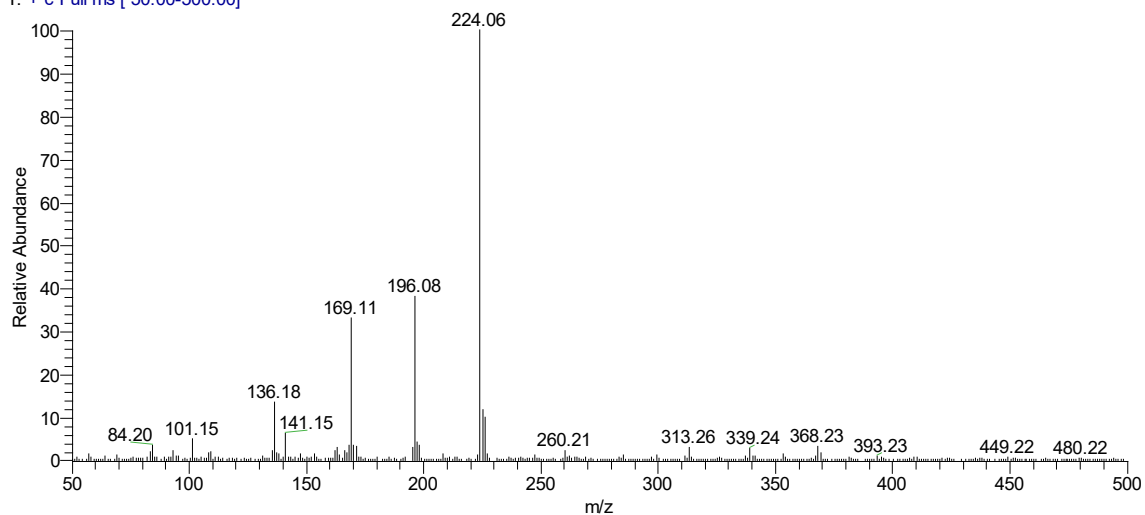


Figure S7. Mass-spectrum of H<sub>2</sub>L<sup>1</sup>.

037\_Rogozhin1 #524-565 RT: 4.33-4.67 AV: 42 NL: 2.34E5  
T: + c Full ms [ 50.00-500.00]

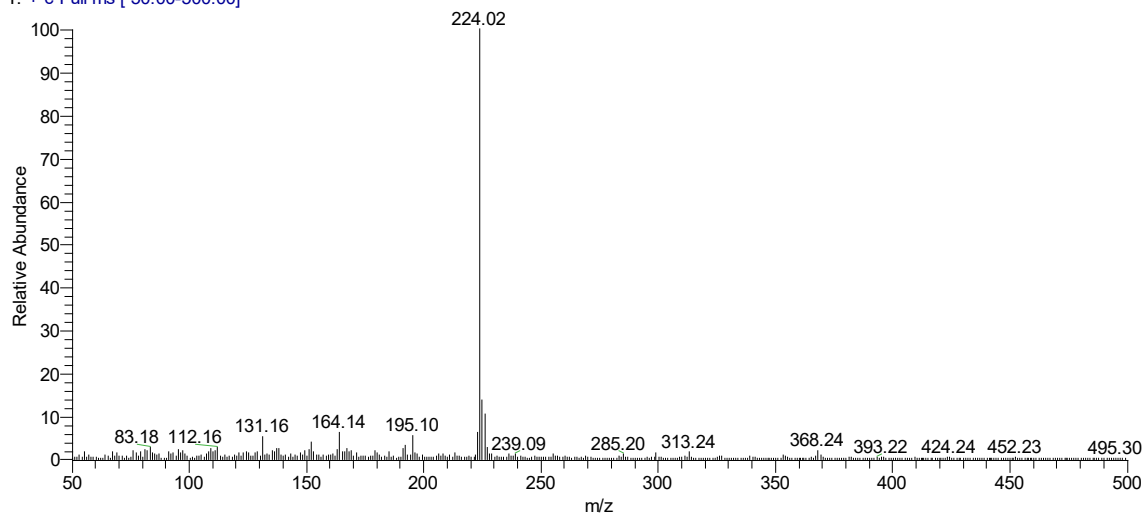


Figure S8. Mass-spectrum of H<sub>2</sub>L<sup>2</sup>.

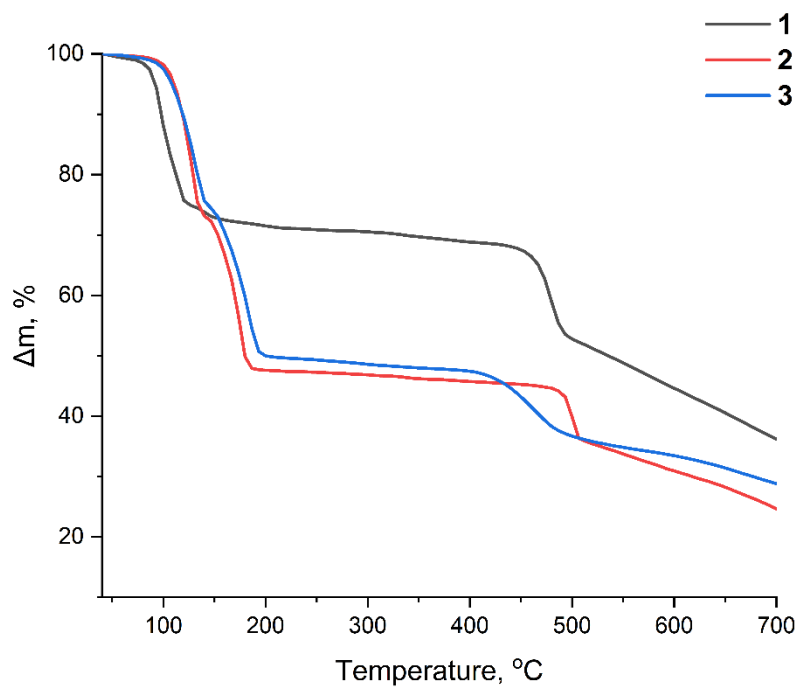


Figure S9. TG curves of **1-3**.

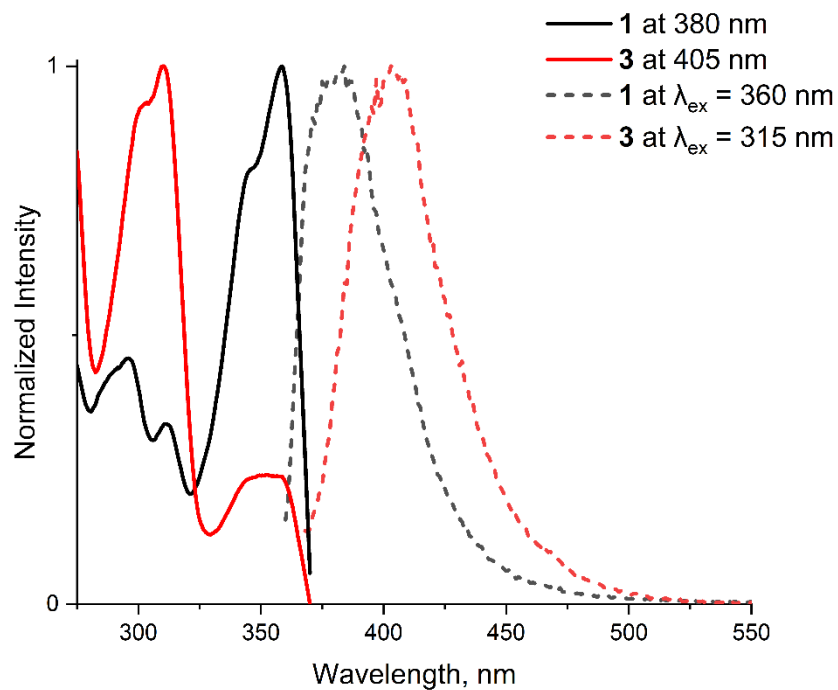


Figure S10. PL and PL excitation spectra of **1** and **3** in DME solution.

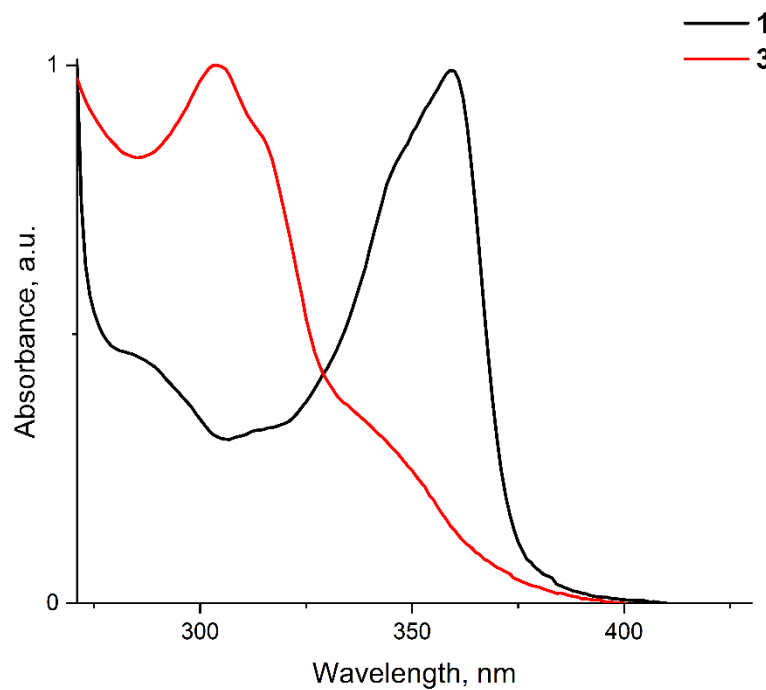


Figure S11. Absorption spectra of **1** and **3** in DME.

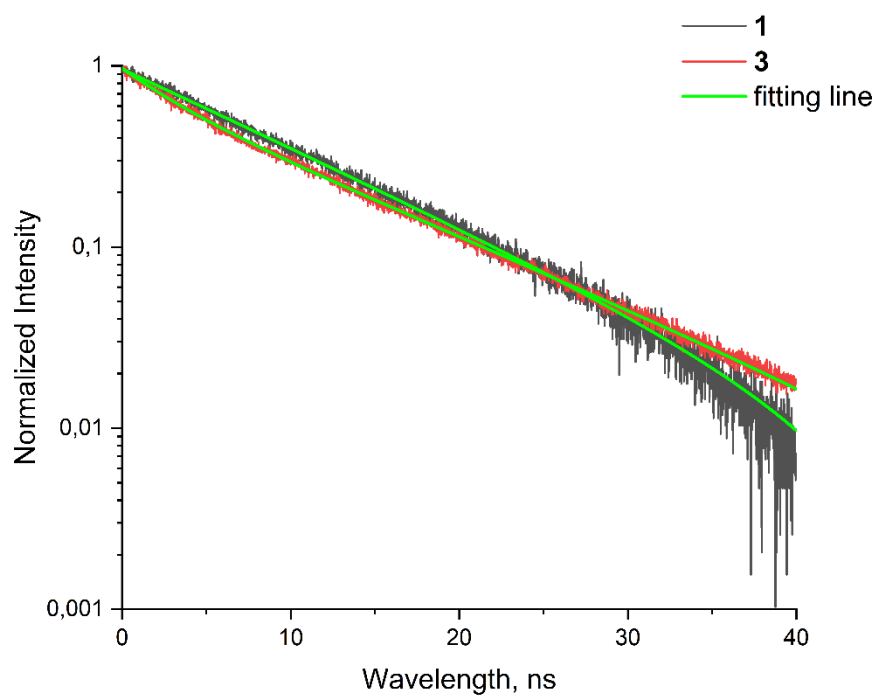


Figure S12. PL decay curves of **1** at 380 nm and **3** at 405 nm in DME solution at 298 K.

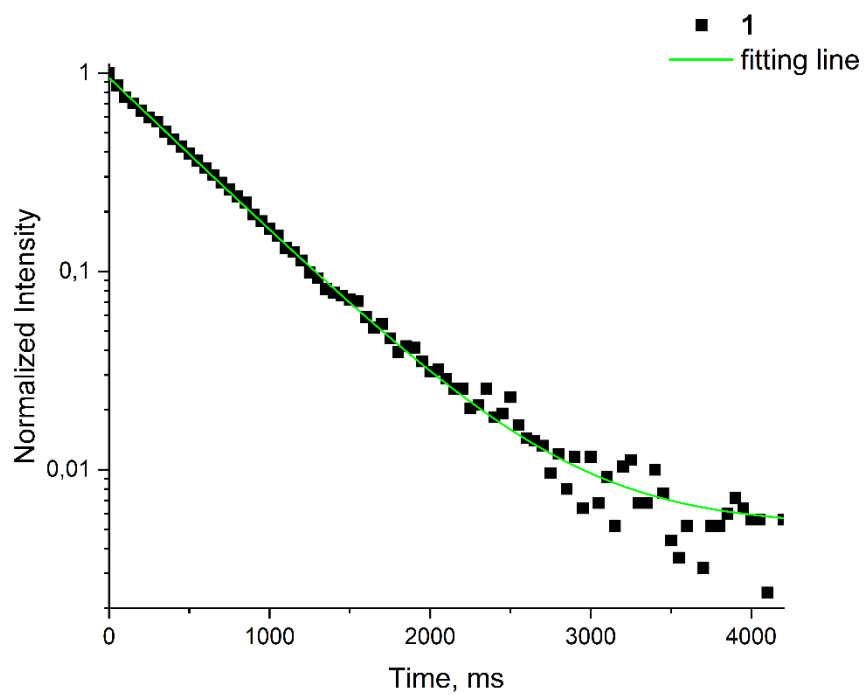


Figure S13. PL decay curve of **1** at 530 nm in DME solution at 77 K.

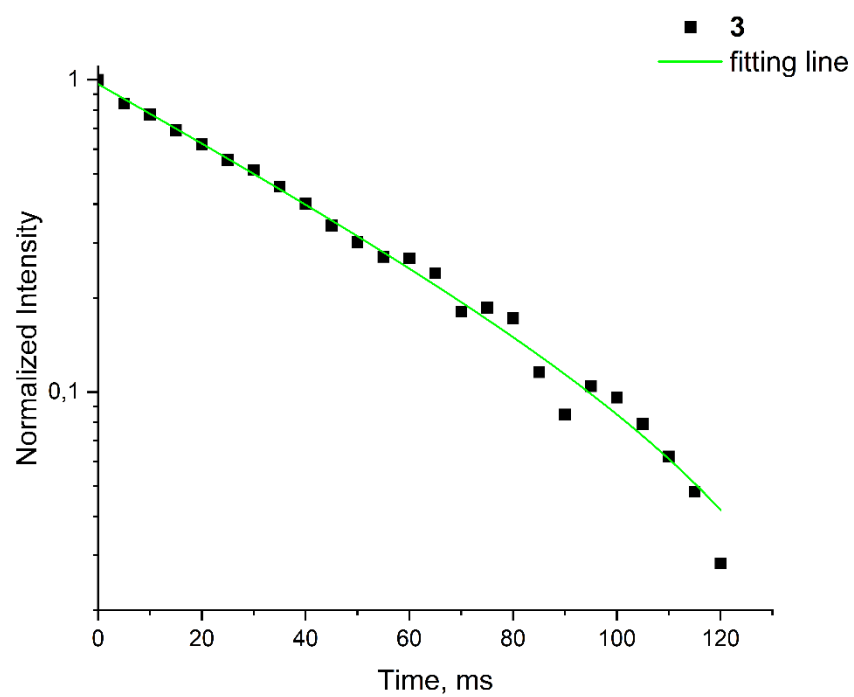


Figure S14. PL decay curve of **2** at 530 nm in DME solution at 77 K.



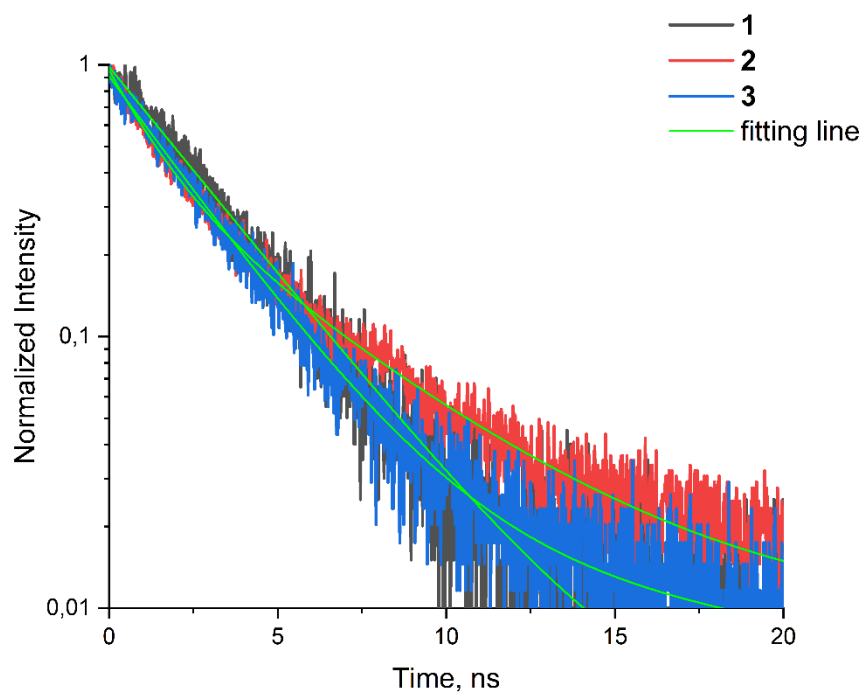


Figure S15. PL decay curves of **1-3** at 405 nm in solid state at 298 K.

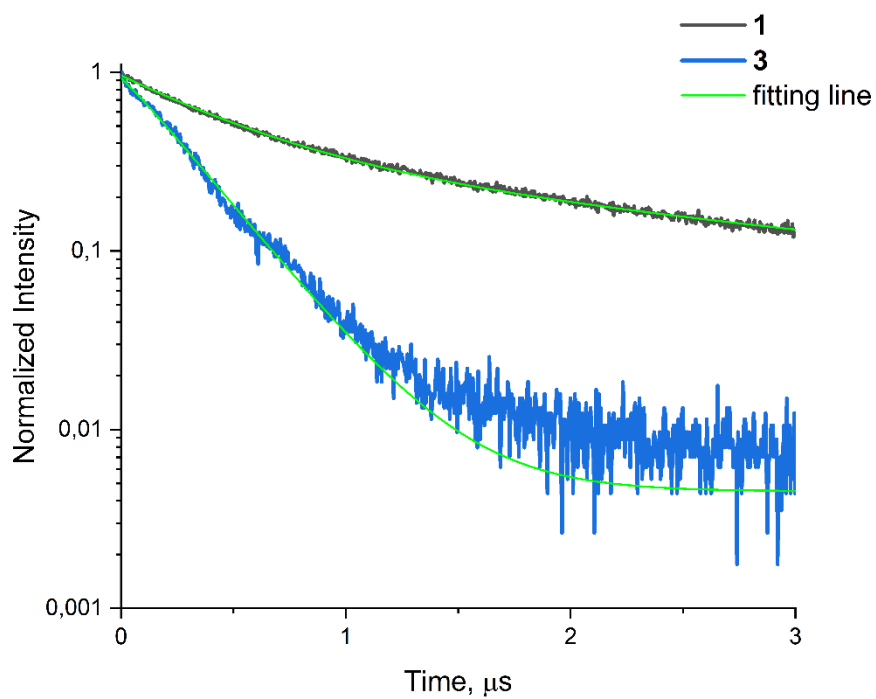


Figure S16. PL decay curves of **1-3** at 530 nm in solid state at 298 K.

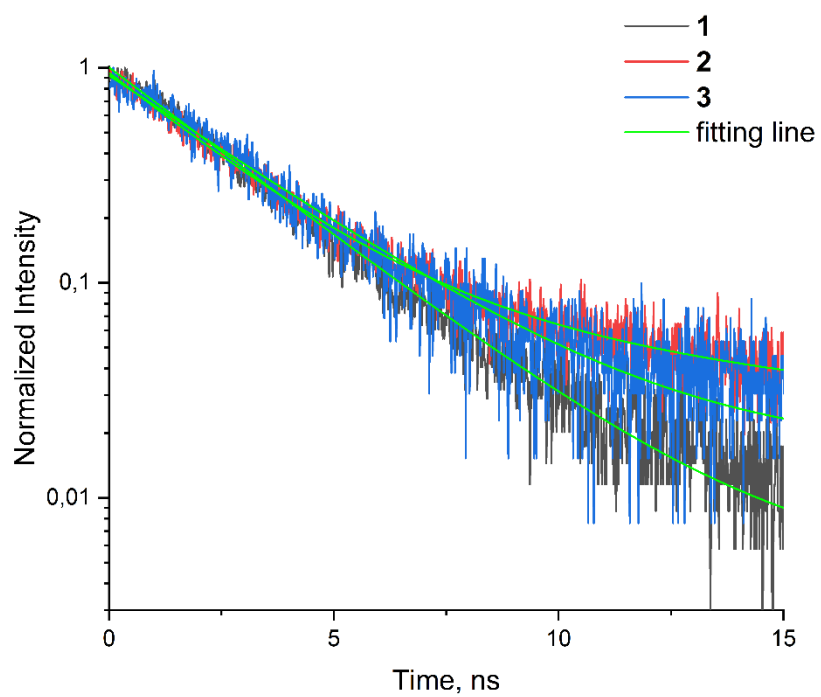


Figure S17. PL decay curves of **1-3** at 405 nm in solid state at 77 K.

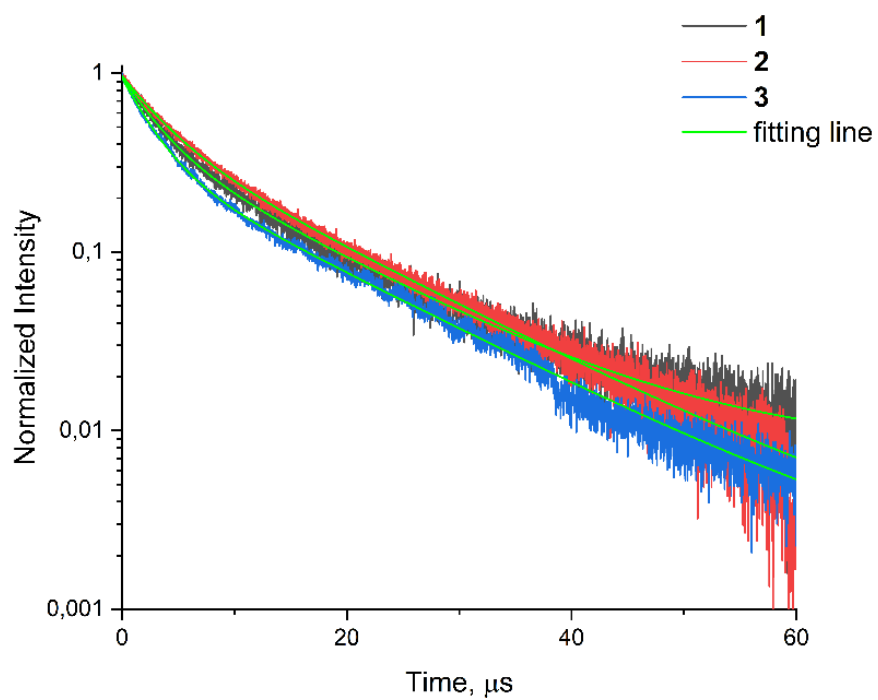


Figure S18. PL decay curves of **1-3** at 530 nm in solid state at 77 K.

Table S1. Crystal data for **1-3**.

Compound	<b>1</b>	<b>2</b>	<b>3</b>
Chemical formula	C <sub>8</sub> H <sub>11</sub> NO <sub>3</sub> SNa	C <sub>16</sub> H <sub>26</sub> N <sub>2</sub> O <sub>6</sub> S <sub>6</sub> Na <sub>2</sub>	C <sub>24</sub> H <sub>39</sub> N <sub>3</sub> Na <sub>3</sub> O <sub>9</sub> S <sub>9</sub>
$M_r$	224.23	580.73	871.09
Crystal system, space group	Monoclinic, $P2_1/c$	Monoclinic, $P2_1/n$	Monoclinic, $P2_1/c$
Temperature (K)	220	150	150
$a, b, c$ (Å)	6.2801 (5), 11.4240 (8), 14.7832 (9)	14.4519 (7), 13.6441 (7), 14.8205 (8)	9.6370 (3), 21.2068 (7), 19.2452 (7)
$\beta$ (°)	95.063 (2)	112.633 (2)	95.360 (1)
$V$ (Å <sup>3</sup> )	1056.47 (13)	2697.3 (2)	3915.9 (2)
$Z$	4	4	4
Radiation type	Mo $K\alpha$	Mo $K\alpha$	Mo $K\alpha$
$\mu$ (mm <sup>-1</sup> )	0.33	0.57	0.59
Crystal size (mm)	0.20 × 0.08 × 0.06	0.20 × 0.12 × 0.12	0.12 × 0.03 × 0.03
$T_{\min}, T_{\max}$	0.689, 0.745	0.693, 0.741	0.671, 0.745
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	10153, 2158, 1864	15929, 15929, 10678	36188, 7462, 4482
$R_{\text{int}}$	0.031	-	0.121
$(\sin \theta/\lambda)_{\text{max}}$ (Å <sup>-1</sup> )	0.625	0.716	0.611
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.030, 0.082, 1.08	0.063, 0.165, 1.05	0.055, 0.120, 1.01
No. of reflections	2158	15929	7462
No. of parameters	129	346	484
No. of restraints	-	48	36
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å <sup>-3</sup> )	0.23, -0.20	1.20, -1.22	0.70, -0.51