

Supplementary Information

Investigating the vital influences of multi-factor synergies on the nitrate formations on the surface of nano MgO particles

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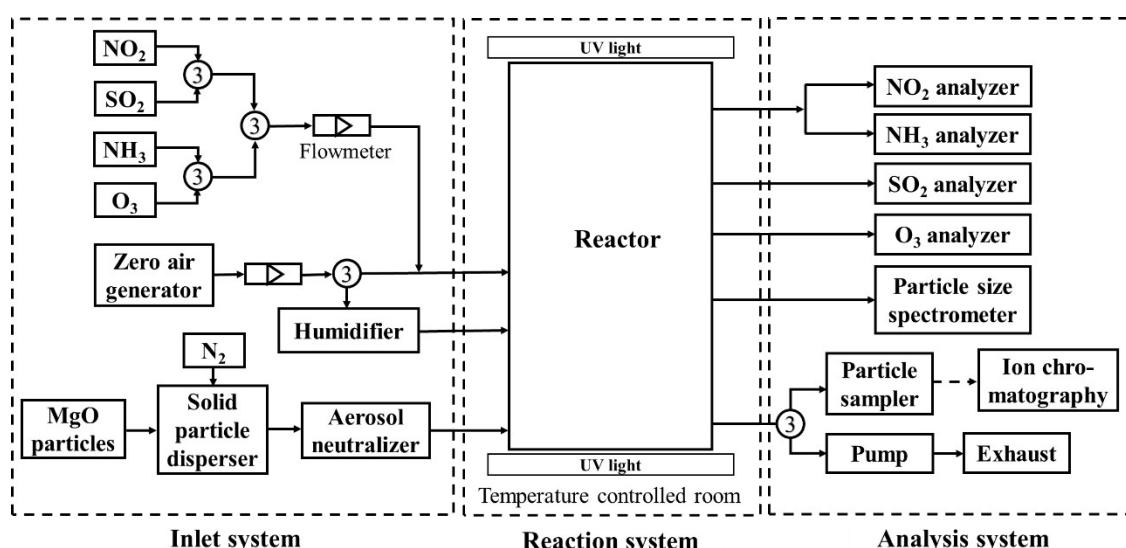


Fig.S1 Schematic of smog chamber set up at BJUT, 3: three-way valve; □: flowmeter.

Table S1 Characteristics of instruments used in the smog chamber experiments

Instrument	Producer and Model	Functions	Performance Index
Zero-air generator	Thermo Fisher Science, Model-111	Provides zero air	Maximum flow 20L/min; HC <0.1ppm; NO<0.5 ppb; NO ₂ <0.5ppb; O ₃ <0.8ppb; SO ₂ <0.5ppb; NH ₃ <0.5ppb; RH<5%; PM<1 cm ⁻³
O ₃ generator	Thermo Fisher Scientific, Model-49i IPS	Provides O ₃	Flow: 4L/min; Range of 0-1000ppb
Solid particle Disperser	PALAS, RBG-1000	Produces dry nanoscale particles	Particle generation rate of 616mg/h
Aerosol neutralizer	SIMCO, Model-6110A	Neutralize electrical properties of particles	Dissipation time < 4 seconds Ion balance < ± 25V
SO ₂ analyzer	Thermo Fisher Scientific, Model-43i	Determinate concentration of SO ₂	Detection range 0-50ppm; Accuracy of ±1.0ppb
NO-NO ₂ -NO _x analyzer	Thermo Fisher Scientific, Model-42i	Determinate concentration of NO-NO ₂ -NO _x	Detection range 0-50ppm; Accuracy of ±1.0ppb
O ₃ analyzer	Thermo Fisher Scientific, Model-49i	Determinate concentration of O ₃	Detection range 0-50ppm; Accuracy of ±1.0ppb
NH ₃ analyzer	Thermo Fisher Scientific, Model-17i	Determinate concentration of NH ₃	Detection range 0-50ppm; Accuracy of ±1.0ppb
Multi-Gas Calibrator	Thermo Fisher Scientific, Model-146i	Calibration gas analyzer	Provides accurate concentrations of SO ₂ , NO ₂ and NH ₃ or other gases
Particle size spectrometer	TOPAS, Model LAP322	Measure the size distribution of PM	Detection range of 200nm-5μm
Particle sampler	MSP, Model 125R NanoMoudi- II TM	Membrane sampling PM	Collect PM in 13 levels
Ion chromatography	Dionex™ ICS-2100; Thermo Fisher Scientific, Aquion™	Measurement of water soluble anions and cations	Digital signal range: 0 to 15000 μS

Table S2 The initial conditions of NO₂ heterogeneous reactions on the surface of nano MgO aerosol particles in the different reaction systems in this study period

Exp. No.	Reaction System	React time (min)	Initial reaction conditions							
			NO ₂ (ppb)	MgO Particle ($\mu\text{g m}^{-3}$)	T (°C)	RH (%)	Lamp (number)	O ₃ (ppb)	NH ₃ (ppb)	SO ₂ (ppb)
1		300	320.66	365.94	20.0°C	17.8%				
2		300	305.65	346.62	22.4°C	38.1%		51.54		
3		300	312.61	359.92	25.7°C	36.6%		91.12		
4		300	320.37	350.59	21.7°C	30.1%		190.47		
5		300	320.65	372.51	20.9°C	33.3%			104.35	
6		300	322.32	389.88	21.3°C	32.0%			173.05	
7		300	321.27	386.96	22.8°C	33.6%			327.11	
8	NO ₂ -MgO-Dark	300	338.23	384.69	20.2°C	28.9%				78.70
9	(Night)	300	332.83	364.47	21.3°C	32.9%				167.10
10		300	331.09	385.04	23.3°C	30.2%				281.66
11		300	330.50	392.85	22.1°C	24.4%		92.91	74.98	
12		300	327.19	386.08	21.6°C	22.0%		100.71	249.19	
13		300	327.49	372.41	20.4°C	22.7%		107.86	334.25	
14		300	321.82	378.22	22.7°C	23.5%		93.20		86.64
15		300	333.18	381.40	21.7°C	26.4%		92.39		188.47
16		300	324.29	372.12	22.3°C	25.6%		95.04		288.21
17		300	328.44	366.08	19.5°C	20.6%	20			
18		300	312.69	366.98	16.6°C	34.4%	40			
19		300	326.10	362.52	30.7°C	25.5%	40	94.73		
20		300	328.97	354.82	32.6°C	26.9%	40	169.74		
21	NO ₂ -MgO-hv (Day)	300	348.77	362.64	30.8°C	26.3%	40	286.22		
22		300	330.42	383.90	21.2°C	29.7%	40		128.92	
23		300	331.74	373.02	19.9°C	22.0%	40		211.37	
24		300	346.82	379.62	19.7°C	25.7%	40		317.60	
25		300	339.47	373.65	20.3°C	26.5%	40			75.52
26		300	333.62	388.86	19.1°C	22.0%	40			160.52
27		300	333.30	398.52	19.2°C	23.4%	40			260.86