

Supplementary Information

Separation Behavior of Hydrogen Isotopes via Water Pervaporation using Proton Conductive Membranes

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ESI 1. Through-plane proton conductivity

The proton conductivity of the fully hydrated membrane was evaluated using a ZIVE SP2 instrument (WonA Tech, South Korea) through electrochemical impedance analysis. This analysis was conducted using a custom-made through-plane conductivity cell equipped with platinum electrodes (active area: $1 \times 1 \text{ cm}^2$). The fully hydrated membrane was placed between platinum electrodes held by a Teflon holder, and the entire assembly was then immersed in deionized water. The membrane's resistance was measured by identifying the intercept of the high-frequency impedance with the real axis at $25 \text{ }^\circ\text{C}$. Specific conductivity (σ) was calculated based on the membrane's thickness (L) and the measurement area (A) using the following formula.

$$\sigma (\text{S}\cdot\text{cm}^{-1}) = L/(R\cdot A)$$

ESI 2. Tensile test

The tensile test of the membrane was conducted using a universal testing machine (Qmesys, South Korea) at ambient conditions. The membrane was prepared by air drying before the measurements. Both ends of the membrane specimen ($40 \times 6 \text{ mm}^2$) were bonded with cardboard end tabs ($10 \times 6 \text{ mm}^2$). The displacement was adjusted at a rate of 10 mm/min to obtain the stress-strain curves.

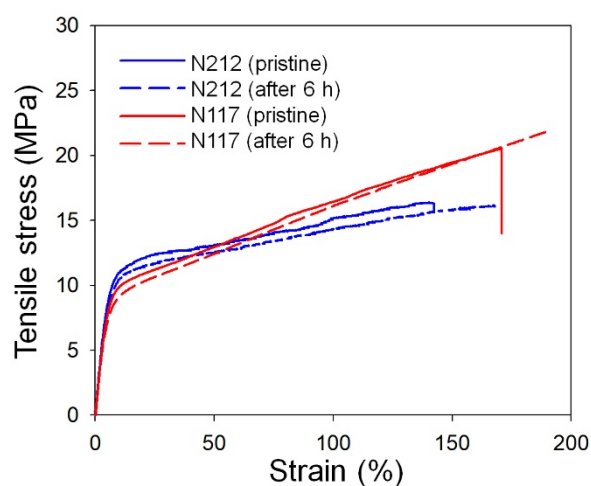


Fig. S1. Tensile stress-strain curves of dried N212 and N117 membranes measured at ambient conditions before and after the water pervaporation test, conducted for 6 hours at 35°C.