

F-SiO₂ Eables PLA-based Superhydrophobic Nanofiber Membrane with Highly Efficient Membrane Distillation

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Information S1. Materials

Dioxane (DX) was offered by Sinopharm Chemical Reagent Co Ltd (Shanghai, China). Polylactic acid (PLA) (4032D) was offered by NatureWorks (US). N-methyl pyrrolidone (NMP) was purchased from Tianjin Beichen Fangzheng Chemical Reagent Factory (Tianjin, China). Polydimethylsiloxane (PDMS) and its curing agent were supplied by Dow Corning Co., Ltd (US). Tetrahydrofuran (THF), silane (FTOS), silicon dioxide (25 mm), n-octane, tetraethyl orthosilicate (TEOS), trichloro (1H,1H,2H,2H-tridecafluoro-n-octyl) and trimethoxy (1H, 1H, 2H, 2H-heptadecafluorodecyl) silane were all obtained from Aladdin Biochemical Technology Co., Ltd (Shanghai, China).

Information S2. Calculation of membrane distillation flux performance.

$$L = \Delta V / (A \times \Delta t) \quad \backslash * \text{MERGEFORMAT (1.1)}$$

where L represents the flux ($\text{L m}^{-2} \text{h}^{-1}$) and ΔV , A and Δt represent the distillate volume variation over a predetermined time, the effective area of the composite membrane (m^2) and filtration time (h), respectively. The A is 11.63 cm^2 . The cross-flow velocity on the membrane surface was 10 mL/min , and the downstream freshwater temperature was controlled at $10 \pm 2^\circ\text{C}$ by the low-temperature thermostat.

Information S3. Calculation of the salt rejection of the membrane.

$$R = \left(1 - \frac{C_p}{C_f}\right) \times 100\% \quad \backslash * \text{MERGEFORMAT (1.2)}$$

where C_f (g L^{-1}) and C_p (g L^{-1}) represent the conductivity of the feed solution and permeate water, respectively.

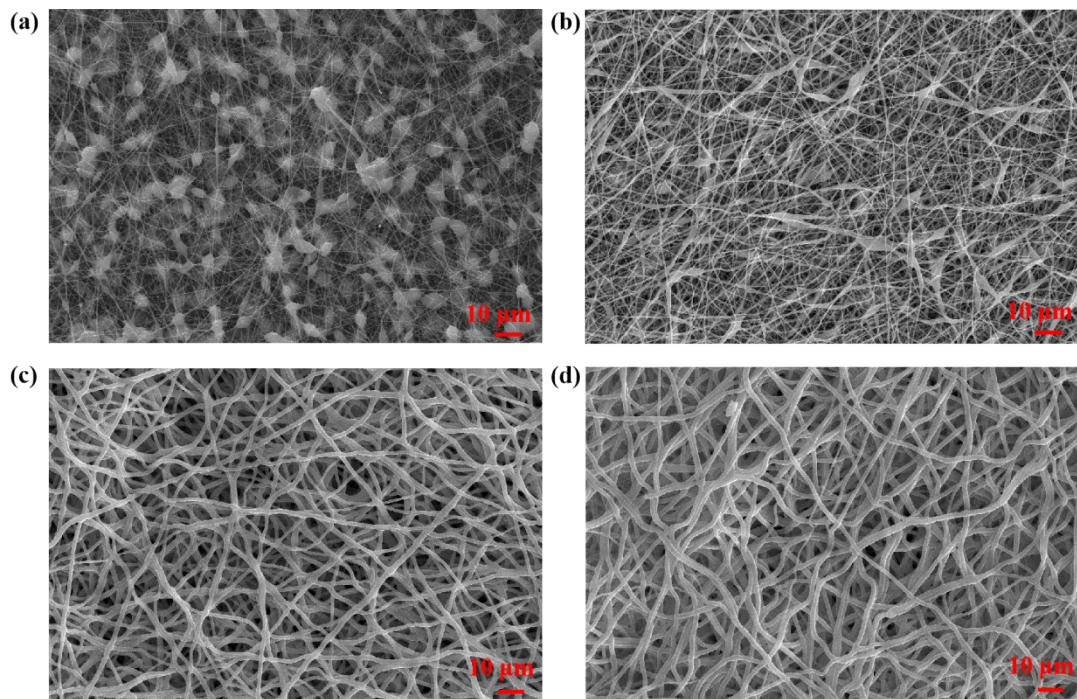


Fig. S1. Membrane surface morphology of desalination membranes with different concentrations of PLA: (a) 10%; (b) 12%; (c) 15%; (d) 17%.

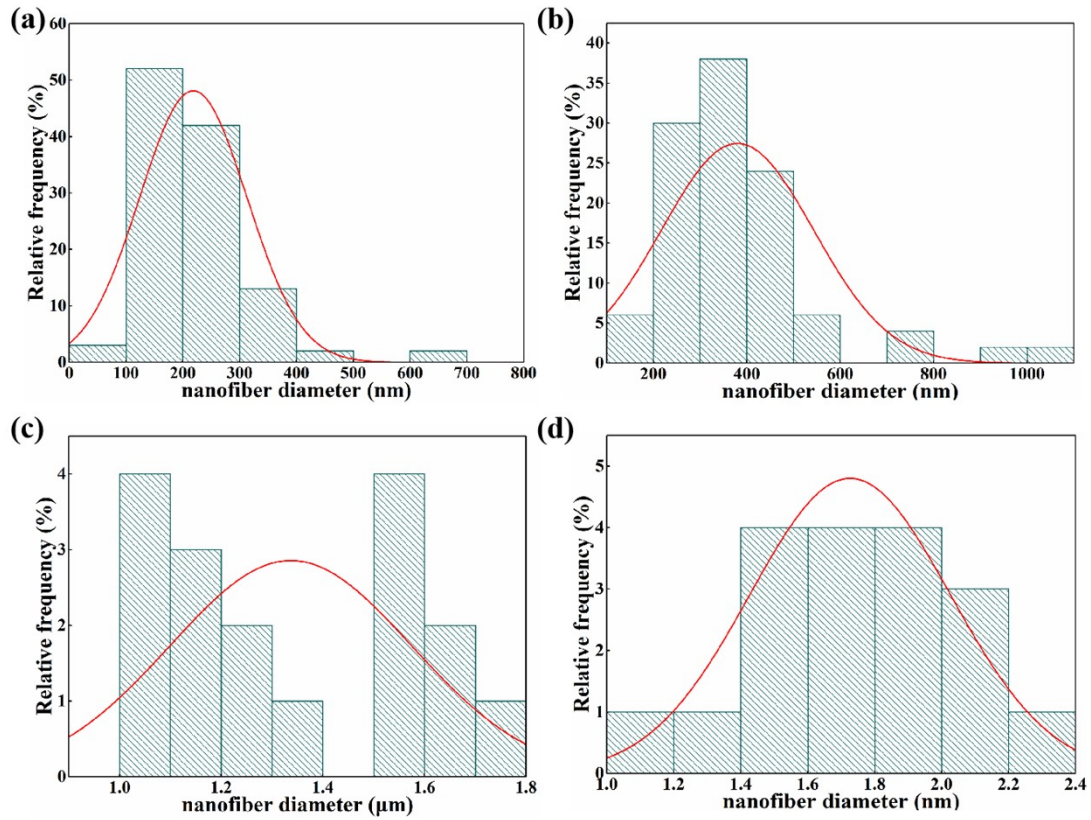


Fig. S2. Nanofibre diameters of desalination membranes with different concentrations of PLA: (a) 10%; (b) 12%; (c) 15%; (d) 17%.

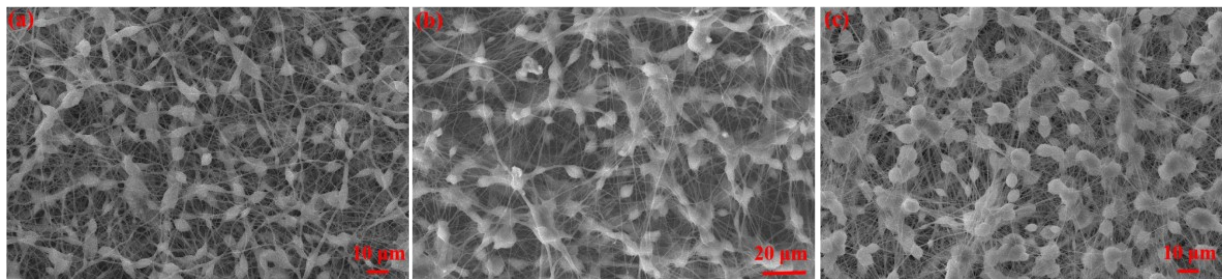


Fig. S3. Membrane surface morphology of desalination membranes with different concentrations of PDMS: (a) 8%; (b) 10%; (c) 12%.

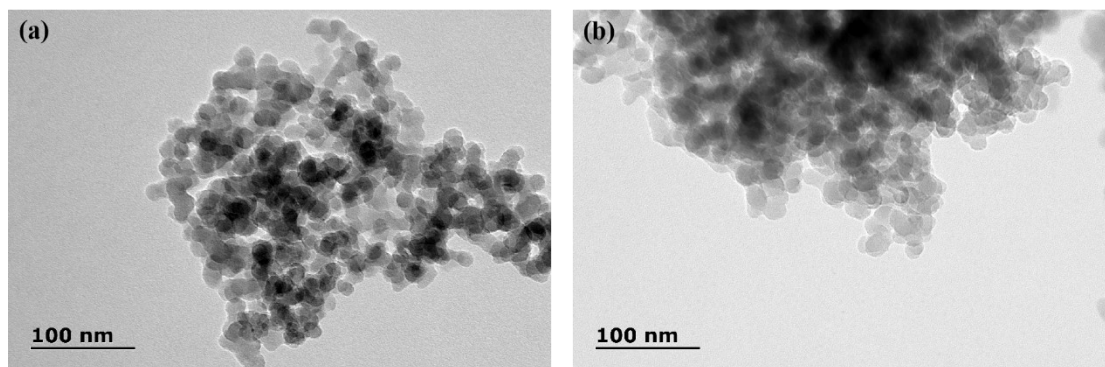


Fig. S4. TEM images of (a) SiO₂, (2)F-SiO₂.

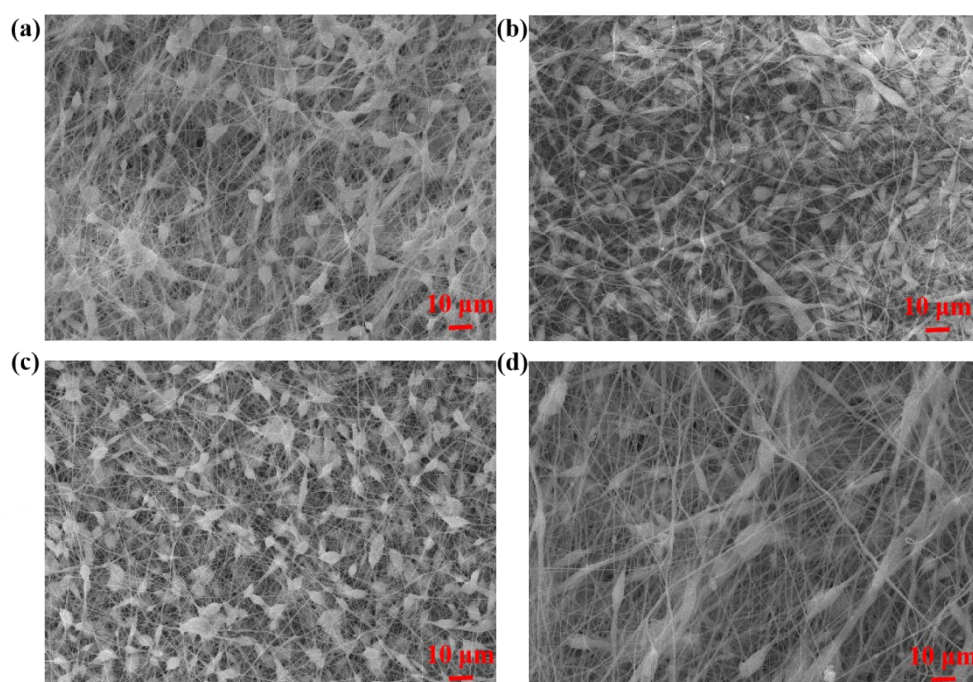


Fig. S5. Membrane surface morphology of desalination membranes with different concentrations

of F-SiO₂ nanoparticles: (a) 0.05%; (b) 0.1%; (c) 0.2%; (d) 0.5%.

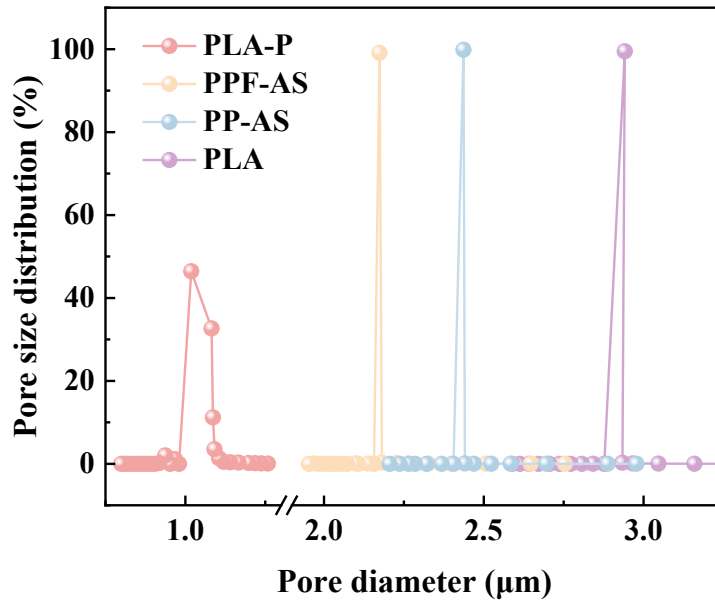


Fig. S6. Pore size distributions of PLA, PLA-P, PP-AS and PPF-AS membranes.

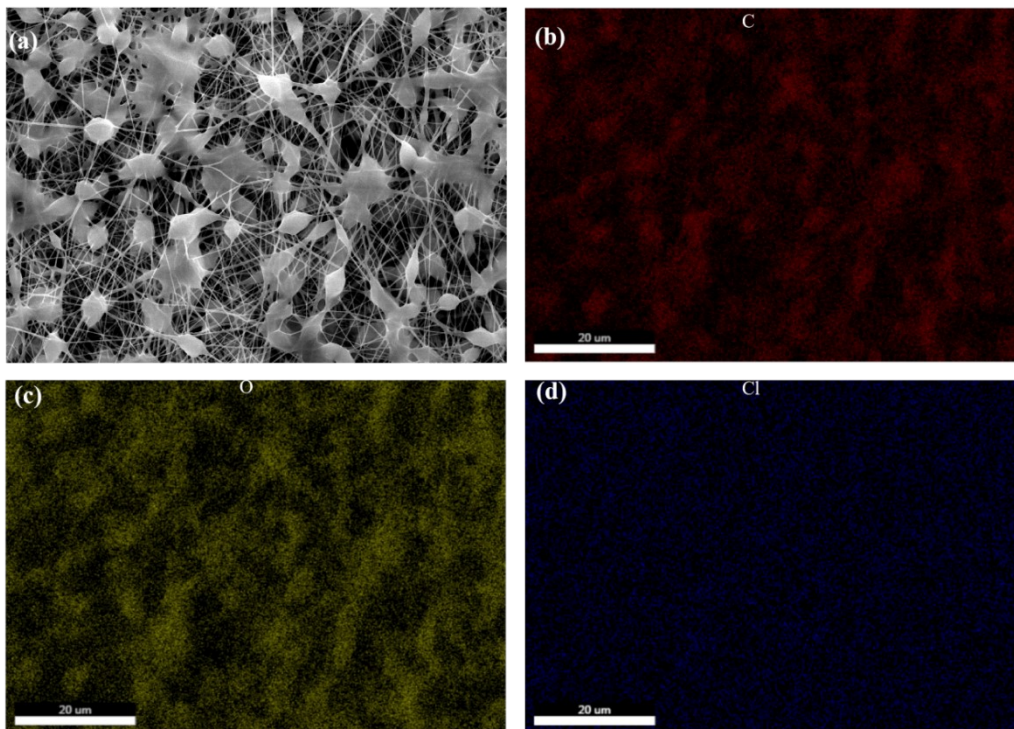


Fig. S7. EDS images of the PPF-AS membrane: (a) Range of distribution, (b) C element, (c) O element, (d) Cl element.