

Variable-valence element doping mediated photogenerated electron trapping for selective oxidation reactions

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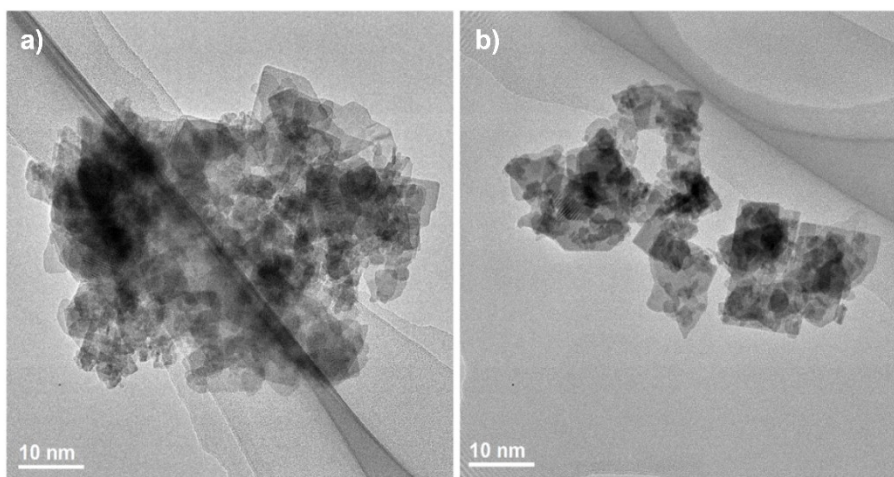


Fig. S1 TEM images of (a) Bi₂WO₆ and (b) Cu-Bi₂WO₆.

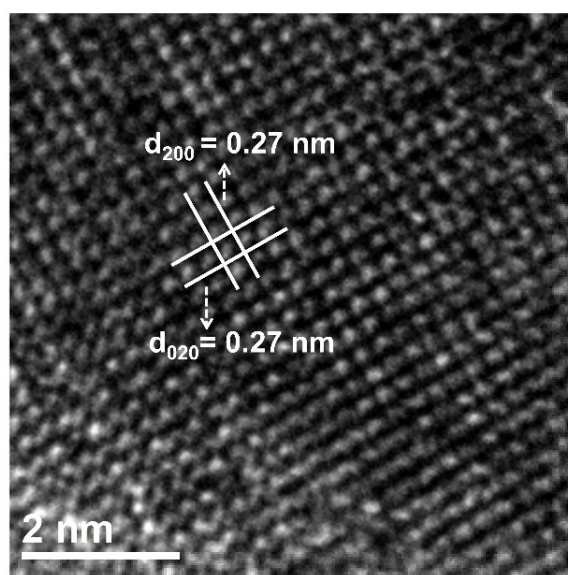


Fig. S2 HRTEM image of Bi₂WO₆.

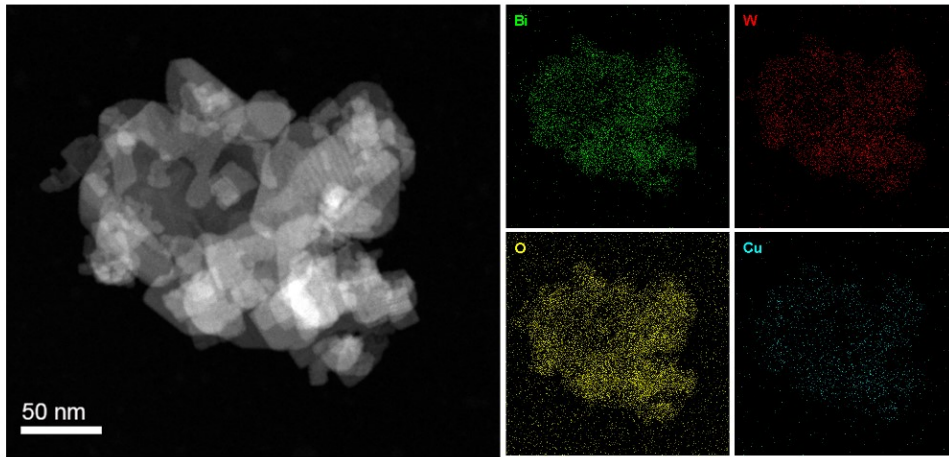


Fig. S3 HAADF-STEM images and the corresponding EDS mapping images of Cu-Bi₂WO₆.

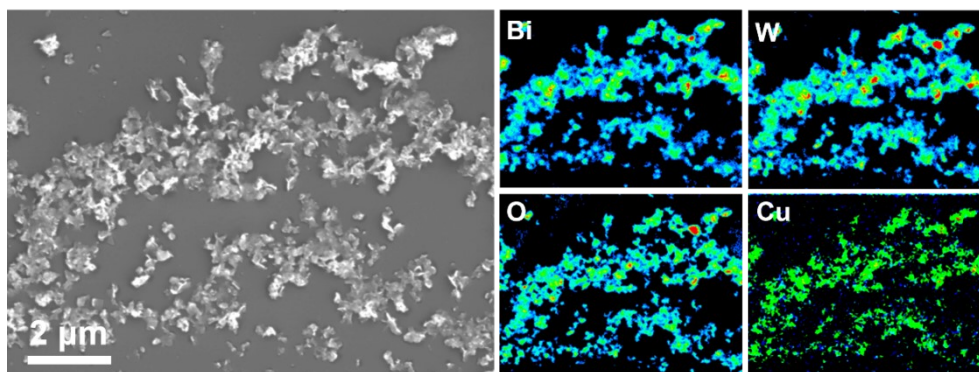


Fig. S4 EPMA analysis of the relevant elements of Cu-Bi₂WO₆.

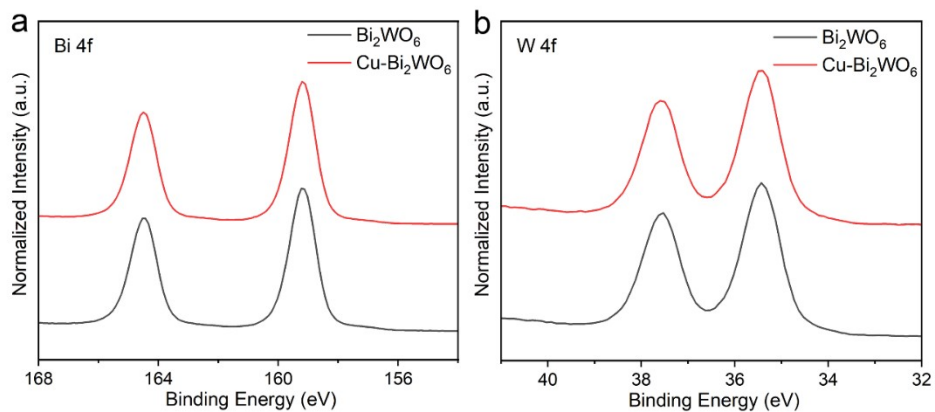


Fig. S5 XPS spectra of Bi_2WO_6 and $\text{Cu-Bi}_2\text{WO}_6$. (a) Bi 4f, (b) W 4f.

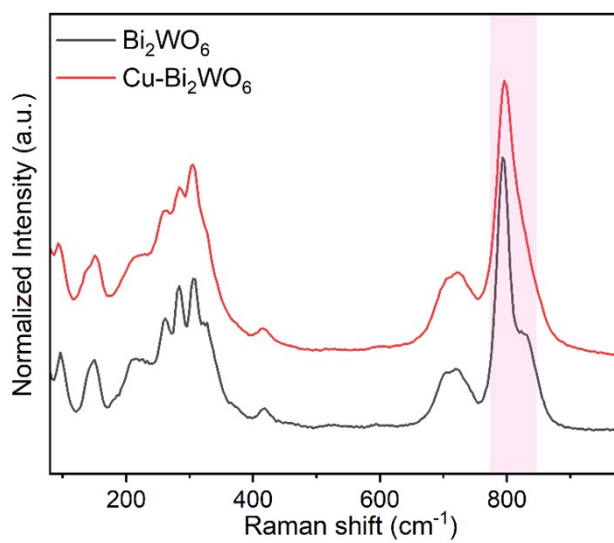


Fig. S6 Raman spectra of Bi_2WO_6 and $\text{Cu-Bi}_2\text{WO}_6$.

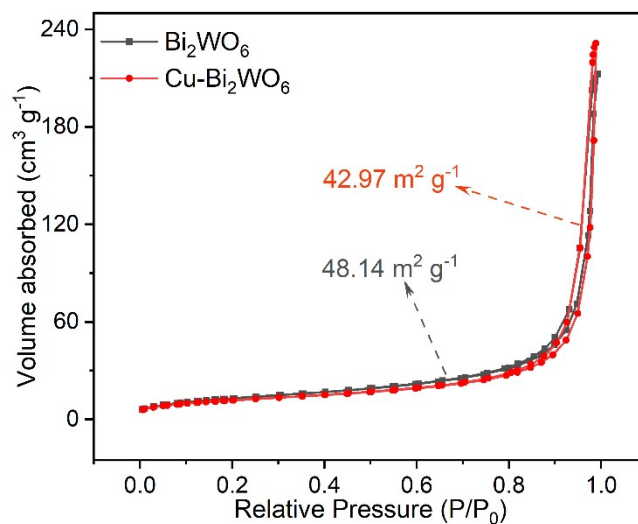


Fig. S7 N₂ adsorption/desorption isotherms of Bi₂WO₆ and Cu-Bi₂WO₆.

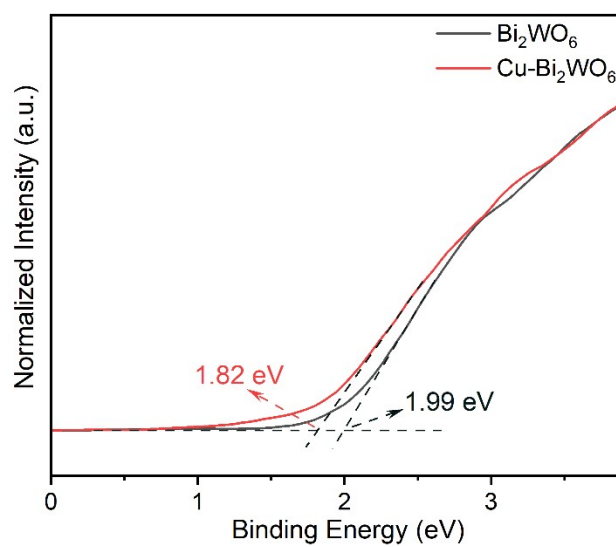


Fig. S8 XPS valence band spectra of Bi₂WO₆ and Cu-Bi₂WO₆.

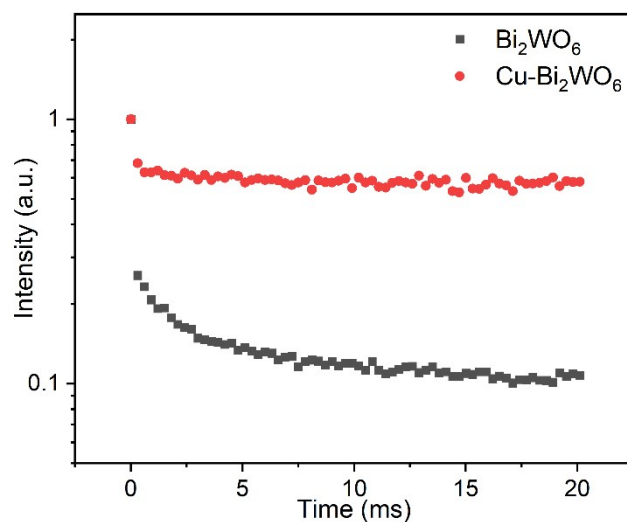


Fig. S9 Low-temperature (10 K) time-resolved phosphorescence (PH) spectra of Bi_2WO_6 and $\text{Cu-Bi}_2\text{WO}_6$.

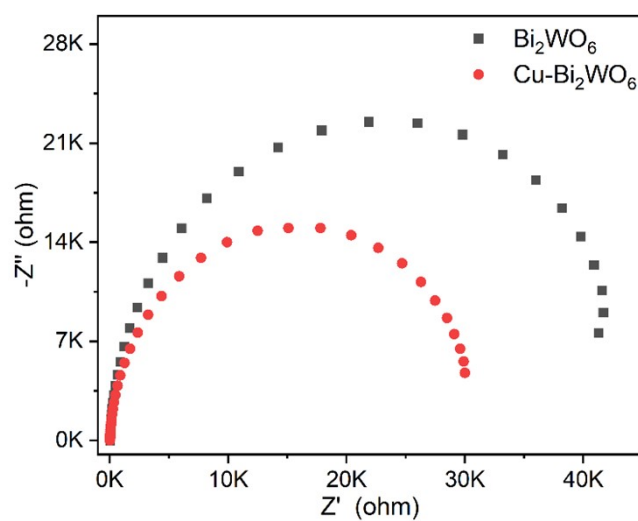


Fig. S10 Electrochemical impedance spectroscopy of Bi_2WO_6 and $\text{Cu-Bi}_2\text{WO}_6$.

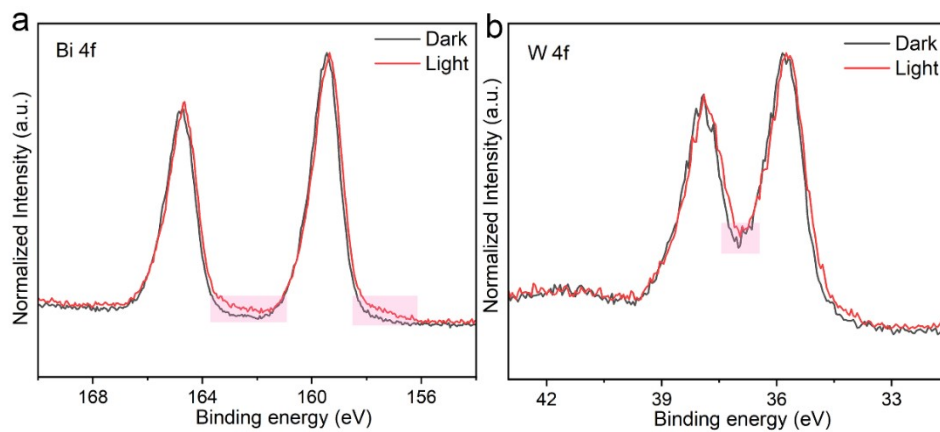


Fig. S11 *In-situ* high resolution XPS spectra of Bi 4f and W 4f of Cu-Bi₂WO₆.

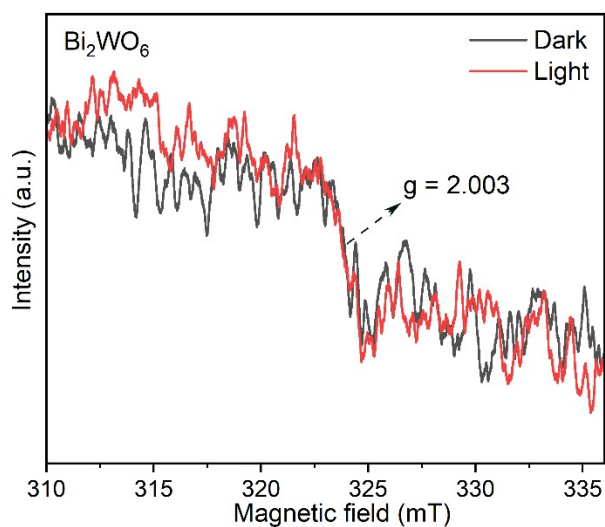


Fig. S12 *In-situ* ESR spectra of Bi₂WO₆ under dark and illumination conditions.

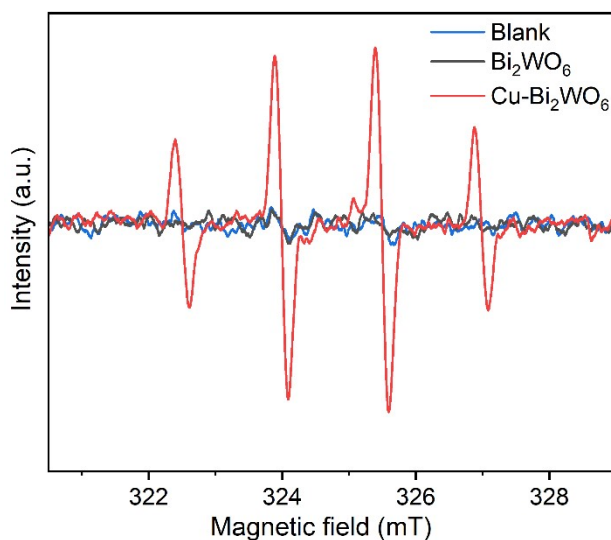


Fig. S13 ESR spectra of different samples in the presence of DMPO in water.

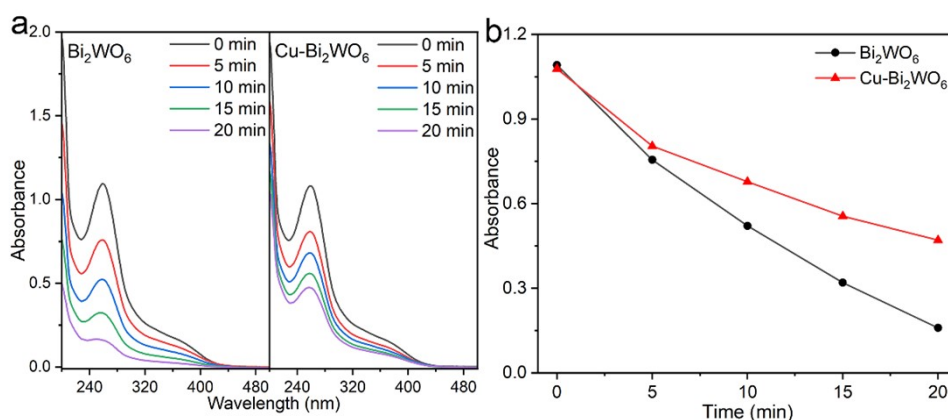


Fig. S14 (a) Time-dependent absorption spectra of NBT reduction for Bi₂WO₆ and Cu-Bi₂WO₆. (b) Evolutions of the absorbance peak monitored at 260 nm.

The $\bullet\text{O}_2^-$ can be detected by nitro blue tetrazolium (NBT), which could be reduced by $\bullet\text{O}_2^-$, resulting in decrease of absorbance spectra.¹ The time-dependent absorption curve shows that Cu-Bi₂WO₆ has a weaker ability to generate $\bullet\text{O}_2^-$ due to the trapping of photogenerated electrons at the Cu sites.

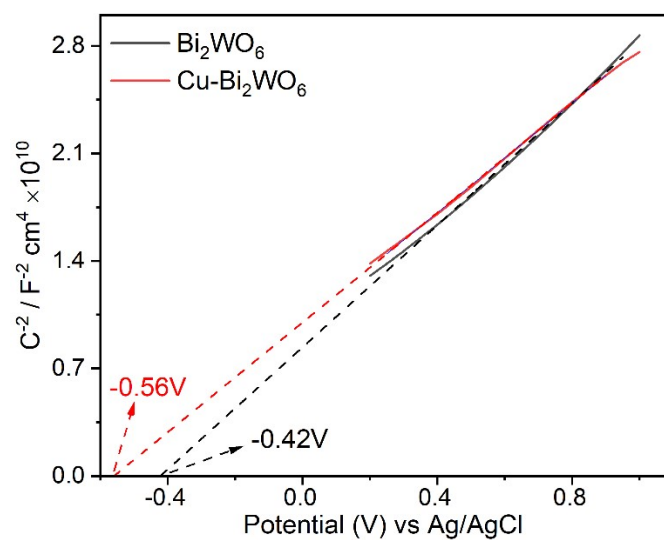


Fig. S15 Mott-Schottky plots of Bi_2WO_6 and $\text{Cu-Bi}_2\text{WO}_6$.

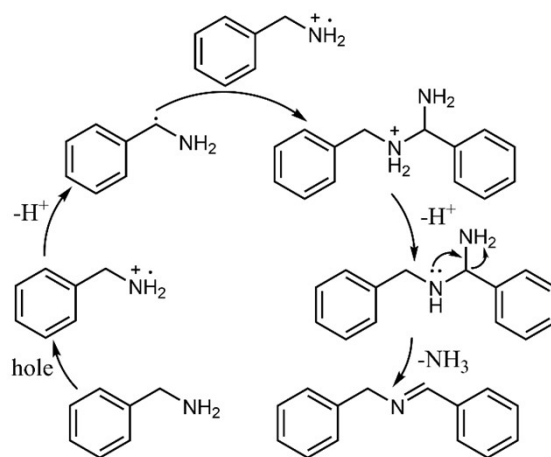


Fig. S16 Schematic diagram of hole-mediated oxidative coupling of amines to imines.²

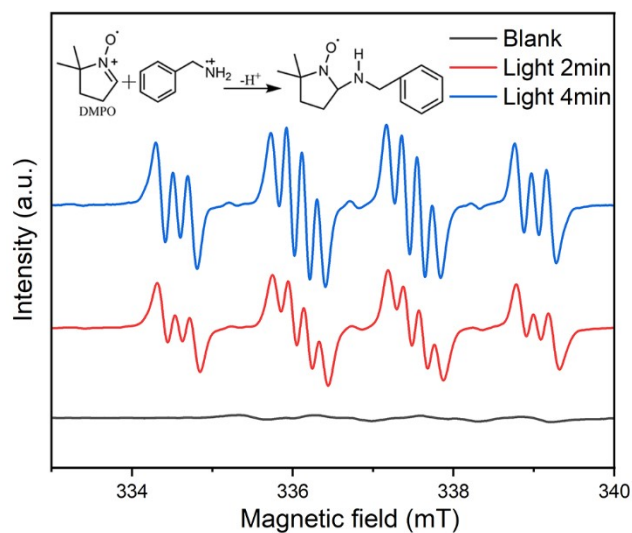


Fig. S17 *In-situ* ESR spectra of DMPO–C₆H₅CH₂NH₂⁺ for Cu-Bi₂WO₆.

In-situ ESR spectroscopy measurements utilizing 5,5-dimethyl-1-pyrroline N-oxide (DMPO) as a spin-trapping agent provide direct evidence of C₆H₅CH₂NH₂⁺ generation. Under irradiation, a specific signal attributed to a DMPO–C₆H₅CH₂NH₂⁺ spin adduct was clearly observed for Cu-Bi₂WO₆, and the signal intensity increased gradually with irradiation time.

References

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