

Supplementary information

Aspects in Cell Design for H₂O₂ Electrosynthesis and Its Integration in Tandem Systems

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Table S1. Performance comparison of GDE reactor

GDE reactor	anodic reaction	co-generation of electricity	H₂O₂ production rate (mmol cm⁻² h⁻¹)	H₂O₂ Faradaic efficiency	stability	Ref
flow-cell	OER	No	3.3~6.1	85~95%	100~400 h	1-3
membrane-free flow-cell	OER	No	-	>90%	2 h	4
PEMFC	HOR	Yes	0.36~2.3	20~70%	80~120 h	5-7
AEMFC	HOR	Yes				n/a
SE-cell (CEM and AEM)	HOR	Yes	3.4~4.35	80~95%	100~500 h	1, 8, 9
SE-cell (CEM)	HOR	Yes	6.53	>90%	50 h	10
SE-cell (AEM)	HOR	Yes				n/a
Zn-Air battery	Zinc oxidation	Yes	0.593	85~92.5%	4~8 h	11, 12

Table S2. Summary of electrocatalyst performances in RRDE and GDE reactor from 2022 to 2024

(RRDE performances)

catalyst type	electrocatalyst	electrolyte	onset potential (V vs. RHE)	H₂O₂ selectivity	stability	Ref
metal-free carbon-based catalysts	N-C (2:3)	-	-	-	-	10
	PD/N-C	0.1 M HClO ₄	0.6	>90%@0.2~0.5 V	10 h@0.3 V	13
	BP2000	-	-	-	-	9
	SCN	0.1 M HClO ₄	0.59	>70%@0~0.4 V	28 h@0.2 V	14
	SCN	0.1 M Na ₂ SO ₄	0.5	>90%@ 0.1~0.4 V	28 h@0.2 V	
	MeCN	0.1 M KOH	0.85	>90%@0.2~0.8 V	9 h@0.5 V	15
	MHCS _{0.5}	0.1 M KOH	0.85	>90%@0.2~0.8 V	12 h@0.5 V	16
	MHCS _{0.6}	0.1 M PBS	0.6	>85%@0.1~0.5 V	12 h@0.5 V	
	BN-C/rGO	0.1 M KOH	0.78	>80%@0.35~0.75 V	12 h@0.4 V	17
	PBT	0.5 M KOH	0.77	>90%@0.45~0.75 V	12 h@0.6 V	18
	BS-C-3	0.1 M KOH	0.80	>90%@0.3~0.7 V	12 h@0.5 V	19
	CB-PDA-A	0.1 M KOH	0.83	>95%@0.4~0.7 V	12 h@0.5 V	20
	O-CDs-3	0.1 M KOH	0.7	>75%@0.2~0.5 V	-	21
	B/N-onion carbon	-	-	-	-	22
	N,S-TCNTs	0.1 M KOH	0.78	>90%@0.2~0.8 V	12 h@0.5 V	1
	BC-C-1	0.1 M KOH	0.79	>80%@0.45~0.75 V	10 h@0.4 V	23
	NBO-G/CNTs	0.1 M KOH	0.75	>80%@0.2~0.75 V	12 h@0.4 V	24
	E-BPC	0.1 M KOH	0.8	>80%@0.3~0.7 V	10 h@0.6 V	25
	VG array	-	-	-	-	26

	CNB-ZIL	0.1 M KOH	0.8	>70%@0~0.7 V	12 h@0.3 V	27
	CB	-	-	-	-	28
	g-C ₃ N ₄ /CNT	0.1 M KOH	0.8	>90%@0.1~0.6 V	10 h@0.5 V	29
	CB	0.2 Mm DTAB + 0.1 M KOH	0.7	>80%@0.1~0.6 V	-	30
	N,O-CNS _{0.5}	0.1 M K ₂ SO ₄	0.65	>80%@0.1~0.6 V	12 h@0.3 V	31
	CBNO	0.1 M Na ₂ SO ₄	0.55	>80%@0~0.6 V	10 h@0.1 V	32
	HCB(N/O)	0.1 M H ₂ SO ₄	0.35	>90%@0~0.3 V	12 h@0.25 V	33
	O-DG-30	0.1 M KOH	0.81	~80%@0.3~0.7 V	-	34
	COPN-3	0.1 M KOH	0.65	>90%@0~0.6 V	90 h@0.4 V	35
	N-CBMC-500	0.1 M KOH	0.7	>90%@0~0.6 V	10 h@0.4 V	36
	o-CNT-8	0.1 M KOH	0.795	>90%@0.4~0.7 V	-	37
	C-0.1M80	0.1 M KOH	0.795	>80%@0.2~0.6 V	-	38
non-precious metal catalysts	CoSe _{2-x} -1	0.1 M HClO ₄	0.73	>80%@0.2~0.6 V	-	39
	CoIn-N-C	0.1 M HClO ₄	0.78	>80%@0.4~0.7 V	-	40
	CoPc-OCNT	-	-	-	-	41
	p-Co-N-C	0.5 M H ₂ SO ₄	0.65	>80%@0~0.6 V	-	42
	CoN ₄ /VG	-	-	-	-	43
	Co-N-C/Ti ₃ C ₂ T _x	0.5 M H ₂ SO ₄	0.72	>60%@0.2~0.7 V	-	44
	Co-N SAC _{Dp}	0.1 M HClO ₄	0.65	>90%@0.1~0.6 V	5.5 h@0.25 V	45
	Co ₂ -DAC	0.1 M HClO ₄	0.72	>90%@0.1~0.6 V	12 h@0.3 V	2
	CoTPP@RGO-160	0.5 M H ₂ SO ₄	0.6	>90%@0.1~0.4 V	8 h@0.1 V	46
	ZnO@ZnO ₂	0.1 M K ₂ SO ₄	0.5	~100%@0~0.4 V	40 h@0.1 V	47
	h-SnO ₂	0.1 M Na ₂ SO ₄	0.48	>95%@0~0.4 V	-	48
	ZnO/rGO	-	-	-	-	49

	P-hcp Ni	0.1 M Na ₂ SO ₄	0.42	>95%@0~0.2 V	9 h@0.2 V	50
	ZnO-v	0.6 M K ₂ SO ₄	0.43	>80%@0.1~0.4 V	-	51
	L-ZnO	0.6 M K ₂ SO ₄	0.35	>80%@0.05~0.35 V	-	52
	Co ₁ -NBC	0.1 M HClO ₄	0.724	~90%@0.2~0.6 V	10 h@0.35 V	53
	CoN ₄ C-N	0.1 M HClO ₄	0.72	>80%@0.1~0.6 V	-	54
	Co-N ₃ O	0.1 M KOH	0.8	~90%@0.1~0.6 V	-	55
	Ni-N ₃ S	0.1 M KOH	0.8	~90%@0.1~0.6 V	-	55
	TiO ₂ @Ti ₂ O ₃	0.1 M KOH	0.78	>80%@0.1~0.6 V	12 h@0.2 V	56
	CoPorF/CNT	0.1 M HClO ₄	0.62	~90%@0.2~0.5 V	-	57
	Mn CD/C	0.1 M KOH	0.786	>90%@0.2~0.7 V	10 h@0.5 V	58
	F/Co-N-G	0.1 M KOH	0.81	>80%@0.1~0.8 V	10 h@0.1 V	59
	E-Niln	0.1 M KOH	0.72	>90%@0.1~0.6 V	24 h@0.55 V	60
	CoPc-S-COF	0.1 M KOH	0.8	>80%@0.2~0.7 V	10 h@0.52 V	61
	Pb SA/OSC	0.1 M KOH	0.76	>90%@0.3~0.7 V	-	62
Precious metal catalysts	a-PdSe ₂ NP/C	0.1 M HClO ₄	0.57	>90%@0~0.3 V	10 h@0 V	63
	Pt-N-CNT-Gly	0.1 M HClO ₄	0.45	>80%@0.05~0.45 V	14 h@0.3 V	64
	Se ₂ -Pt	0.1 M HClO ₄	0.7	>95%@0~0.6 V	15 h@0 V	3
	PdSe ₂	0.05 M NaPi	0.6	>60%@0~0.6 V	-	65
	Pd/MCS-8	0.1 M K ₂ SO ₄	0.65	>90%@0.1~0.6 V	12 h@0.3 V	66
	Pd SNCs/NiTe ₂	0.1 M KOH	0.81	>80%@0.3~0.6 V	20 h@0.4 V	67

Table S3. Summary of electrocatalyst performances in RRDE and GDE reactor from 2022 to 2024

(Reactor performances)

catalyst type	electrocatalyst	GDE reactor	electrolyte	current density (mA cm⁻²)	H₂O₂ production rate (mmol cm⁻² h⁻¹)	H₂O₂ Faradaic efficiency	stability	Ref
metal-free carbon-based catalysts	N-C (2:3)	CEM-SE-cell	pure water	389@2.35 V	6.53@2.35 V	>90%	50 h@70 mA cm ⁻²	10
	PD/N-C	flow-cell	0.1 M HClO ₄	38@-0.6 V	-	~98%	144 h@-0.6 V	13
	BP2000	flow-cell	0.1 M H ₂ SO ₄ + 0.05 M Na ₂ SO ₄	400@-0.9 V	6.5@-0.9 V	>80%	20 h@200 mA cm ⁻²	
	BP2000	MEA(ORR-OER)	0.1 M H ₂ SO ₄ + 0.05 M Na ₂ SO ₄	70 @1.7 V	-	~10%	-	9
	BP2000	double-PEM SE	0.03 M Na ₂ SO ₄	50@1.85 V	-	~90%	500 h@50 mA cm ⁻²	
	SCN	flow-cell	0.1 M HClO ₄	100@-0.8 V	1.3@100 mA cm ⁻²	~70%	168 h@100 mA cm ⁻²	14
	SCN	flow-cell	1 M Na ₂ SO ₄	100@0 V	1.65@100 mA cm ⁻²	~90%	168 h@100 mA cm ⁻²	
	MeCN	flow-cell	0.1 M KOH	200@0.2 V	3.6@0.2 V	>90%	24 h@0.2 V	15
	MHCS _{0.5}	flow-cell	0.1 M KOH	250@0.1 V	4.3@0.1 V	>90%	16 h@0.1 V	16
	MHCS _{0.6}	flow-cell	0.1 M PBS	175@0.1 V	3.16@0.1 V	>90%	16 h@0.1 V	
	BN-C/rGO	flow-cell	0.1 M KOH	50@0.2 V	0.96@0.2 V	70~100%	8 h@0.2 V	17
PBT	flow-cell	1.0 M KOH	100@0.4 V	1.88@100 mA cm ⁻²	>80%	50 h@100 mA cm ⁻²	18	

BS-C-3	flow-cell	0.1 M KOH	50@0.2 V	0.75@50 mA cm ⁻²	>90%	11 h@0.2 V	19
CB-PDA-A	flow-cell	0.1 M KOH	60@1.5 V	0.54@1.5 V	>90%	250 h@1.5 V	20
O-CDs-3	flow-cell	0.1 M KOH + 10 Mm EDTA	-	-	>70%	10 h@10 mA cm ⁻²	21
B/N-onion carbon	flow-cell	1 M KOH	100@0.4 V	1.75@0.4 V	>95%	12 h@0.2 V	22
N,S-TCNTs	flow-cell	1 M KOH	300@0.5 V	6.07@350 mA cm ⁻²	~95%	200 h@32 mA cm ⁻²	1
N,S-TCNTs	SE cell	pure water	300@2.7 V	4.35@300 mA cm ⁻²	~70%	300 h@20 mA cm ⁻²	
BC-C-1	flow-cell	1 M KOH	50@0.3 V	0.76@0.3 V	>75%	12 h@0.3 V	23
NBO-G/CNTs	flow-cell	1 M KOH	50@0.2 V	0.71@0.2 V	80~50%	12 h@0.2 V	24
E-BPC	flow-cell	1 M KOH	300@0.67 V	4.7@300 mA cm ⁻²	>80%	100 h@100 mA cm ⁻²	25
E-BPC	flow-cell	1 M Na ₂ SO ₄	300@0.27 V	4.76@300 mA cm ⁻²	>80%	24 h@100 mA cm ⁻²	
VG array	flow-cell	0.1 M KOH	100@0.4 V	1.80@100 mA cm ⁻²	~90%	16 h@100 mA cm ⁻²	26
CNB-ZIL	flow-cell	0.1 M KOH	50@1.4 V	-	~80%	9 h@1.4 V	27
CB	flow-cell	0.1 M NaClO ₄	-	0.57@28 mA cm ⁻²	>80%		28
g-C ₃ N ₄ /CNT	flow-cell	1 M KOH	100@0.37 V	-	>90%	8 h@100 mA cm ⁻²	29
CB	flow-cell	0.2 Mm DTAB + 0.1 M KOH	10@0.2 V	-	~90%	30 h@0.2 V	30
N,O-CNS _{0.5}	flow-cell	0.1 M K ₂ SO ₄	100@0.2 V	1.67@0.2 V	>90%	24 h@0.2 V	31
CBNO	flow-cell	1 M Na ₂ SO ₄	100@-0.4 V	1.34@100 mA cm ⁻²	>90%	24 h@100 mA cm ⁻²	32

	HCB(N/O)	flow-cell	0.1 M H ₂ SO ₄	31@-0.5 V	-	~90%	80 h@-0.5 V	33
	O-DG-30	flow-cell	0.1 M KOH	35@0.50 V	0.59@0.50 V	>95%	10 h@0.5 V	34
			1 M KOH	-	1.76@130 mA cm ⁻²	~70%	-	
	COPN-3	flow-cell	0.5 M Na ₂ SO ₄	130@-2 V	2.57@130 mA cm ⁻²	~100%	70 h@130 mA cm ⁻²	35
			0.5 M H ₂ SO ₄	-	0.66@130 mA cm ⁻²	~25%	-	
			1 M KOH	-	0.42@-1.2 V	~60%	-	
	N-CBMC-500	flow-cell	0.5 M Na ₂ SO ₄	-	0.49@-0.65 V	~70%	-	36
			0.5 M H ₂ SO ₄	-	0.24@-0.9 V	~30%	-	
	o-CNT-8	SE-cell	pure water	93@4 V	1.59@4 V	~92%	100 h@4 V	37
	C-0.1M80	SE-cell (HOR-2e ⁻ ORR)	pure water	100@1 V	1.79@100 mA cm ⁻²	95.8%	200 h@100 mA cm ⁻²	38
	CoSe _{2-x} -1	flow-cell	0.1 M HClO ₄	10@0 V	-	70%	32 h@0 V	39
	CoIn-N-C	flow-cell	0.1 M HClO ₄	100@0.2 V	1.94@100 mA cm ⁻²	~80%	4 h@100 mA cm ⁻²	40
	CoPc-OCNT	flow-cell	0.3 M K ₂ SO ₄ (pH=1.5)	-	5.75@300 mA cm ⁻²	~100%	-	41
non-precious metal catalysts	p-Co-N-C	flow-cell	0.5 M H ₂ SO ₄	35@0 V	-	>80%	4 h@0.1 V	42
	CoN ₄ /VG	flow-cell	0.1 M HClO ₄	24@1.8 V	0.304@1.8 V	~70%	6 h@1.8 V	43
	Co-N-C/Ti ₃ C ₂ T _x	flow-cell	0.5 M H ₂ SO ₄	75@0.1 V	2.8@0.1 V	~75	20 h@0.1 V	44
	Co-N SAC _{Dp}	flow-cell	0.1 M HClO ₄	50@0 V	0.78@50 mA cm ⁻²	>70%	90 h@0.1 V	45
				50@0 V	0.85@50 mA cm ⁻²	~90%	100 h@50 mA cm ⁻²	
	Co ₂ -DAC	flow-cell	0.1 M HClO ₄	200@-0.8 V	3.3@200 mA cm ⁻²	~85%	100 h@200 mA cm ⁻²	2

CoTPP@RGO-160	flow-cell	0.5 M H ₂ SO ₄	200@-0.6 V	3.1@200 mA cm ⁻²	>80%	16 h@200 mA cm ⁻²	46
	PEMWE (OER-2e ⁻ ORR)	pure water	400@2.1 V	-	-	225 h@ 400 mA cm ⁻²	
ZnO@ZnO ₂	flow-cell	0.1 M K ₂ SO ₄	125@0.1 V	2.2@0.1V	~95%	50 h@0.1 V	47
h-SnO ₂	flow-cell	1M Na ₂ SO ₄	124@0 V	2.06@0 V	~90%	40 h@120 mA cm ⁻²	48
h-SnO ₂	flow-cell	pure water	100@0 V	-	-	20 h@0 V	
ZnO/rGO	flow-cell	0.6 M K ₂ SO ₄	500@-0.8 V	9.7@-0.8 V	~90%	24 h@200 mA cm ⁻²	49
ZnO/rGO	tandem-parallel flow-cell	0.6 M K ₂ SO ₄	455@-0.8 V	8.73@-0.8 V	~90%	24 h@100 mA cm ⁻²	
P-hcp Ni	flow-cell	pure water	150@-1 V	2.42@150 mA cm ⁻²	~90%	120 h@-1 V	50
		sea water	180@-1 V	2.5 @-1 V	~90%	3 h@-1 V	
		pure water	150@-0.6 V	2.44@150 mA cm ⁻²	~90%	130 h@150 mA cm ⁻²	
ZnO-v	flow-cell	0.6 M K ₂ SO ₄	-	18.3@1000 mA cm ⁻²	>95%	20 h@200 mA cm ⁻²	51
L-ZnO	flow-cell	0.6 M K ₂ SO ₄	-	18.2@1000 mA cm ⁻²	~95%	20 h@200 mA cm ⁻²	52
Co ₁ -NBC	flow-cell	0.5 M H ₂ SO ₄	350@-0.5 V	5.96@350 mA cm ⁻²	~86%	60 h@200 mA cm ⁻²	53
CoN ₄ C-N	flow-cell	0.1 M HClO ₄	10@0.358 V	0.096@0.358 V	~82%	200 h@0.358 V	54
Co-N ₃ O	flow-cell	1 M KOH	110@0.1 V	1.42@0.1 V	~90%	12 h@0.1 V	55

	Ni-N ₃ S	flow-cell	1 M KOH	110@0.2 V	1.75@0.2 V	~90%	12 h@0.2 V	
	TiO ₂ @Ti ₂ O ₃	SE-cell	pure water	200@2.5 V	-	>80%	100 h@2.28 V	56
	CoPorF/CNT	SE-cell	pure water	50@2.7 V	5.38@3.02 V	~80%	48 h@50 mA cm ⁻²	57
	Mn CD/C	flow-cell	1 M KOH	200@0.5 V	-	~80%	50 h@200 mA cm ⁻²	58
	F/Co-N-G	flow-cell	1 M KOH	70@0.3 V	1.21@0.3 V	~70%	80 h@0.3 V	59
	E-Niln	flow-cell	0.1 M KOH	200@0.4 V	2.4@0.4 V	~90%	20 h@200 mA cm ⁻²	60
	CoPc-S-COF	flow-cell	1 M KOH	125@0.67 V	-	~90%	20 h@125 mA cm ⁻²	61
	Pb SA/OSC	flow-cell	1 M KOH	-	6.9@400 mA cm ⁻²	93%	100 h@50 mA cm ⁻²	62
Precious metal catalysts	a-PdSe ₂ NP/C	flow-cell	0.1 M HClO ₄	50@-0.5 V	0.70@0 V	>80%	2 h@50 mA cm ⁻²	63
	Pt-N-CNT-Gly	flow-cell	0.05 M H ₂ SO ₄ + 0.05 M Na ₂ SO ₄	100@-0.2 V	-	~85%	30 h@100 mA cm ⁻²	64
	Se ₂ -Pt	flow-cell	0.1 M HClO ₄	250@3 V	4.16@250 mA cm ⁻²	~90%	400 h@250 mA cm ⁻²	3
	PdSe ₂	flow-cell	0.5 M NaPi	5.5@0.3 V	0.00356@5.5 mA cm ⁻²	~30%	3 h@5.5 mA cm ⁻²	65
	Pd/MCS-8	flow-cell	0.5 M K ₂ SO ₄	250@0.2 V	3.94@0.2 V	>86%	24 h@0.2 V	66
	Pd SNCs/NiTe ₂	flow-cell	0.1 M KOH	100@0 V	1.75@100 mA cm ⁻²	95%	10 h@0 V	67

Table S4. Summary of 2e⁻ ORR integrated tandem systems from 2022 to 2024

electro-catalyst	thermo-catalyst	reactor	tandem mode	electrolyte	H ₂ O ₂ Faradaic efficiency	tandem application	performance	stability	Ref
CNT-NH ₂	TS-1	SE-cell	ex-cell	pure water	~80%@200 mA cm ⁻²	oxidation of ethylene	2.2 mmol h ⁻¹ and 70% FE of ethylene glycol	100 h@30 mA cm ⁻²	68
carbon nanofibres	-	SE-cell	ex-cell	pure water	-	oxidation of nitrogen	5.84 nmol s ⁻¹ and 25% FE of nitrates	-	69
nitric-acid-treated carbon black	TS-1	SE-cell	ex-cell	pure water	98.1%@-147.1 mA cm ⁻²	synthesize of glycine	3.61 mmol g ⁻¹ h ⁻¹ of glycine	100 h@10 mA cm ⁻²	70

F-CB	TS-1	SE-cell	in-cell	0.1 M K ₂ SO ₄	~80%@37.5 mA cm ⁻²	oxidation of phenol	30.25% FE and 94.68% selectivity of catechol and hydroquinone	18 h@37.5 mA cm ⁻²	71
C-acid	MIL-53 (Al, Fe)	PEMFC	in-cell	pure water	~80%@-0.1~0 V	oxidation of propane	2.65 μmol h ⁻¹ cm ⁻² of C ₃ products	10 h@-0.2 V	72
C-O	TS-1	flow-cell	in-cell	0.1 M H ₂ SO ₄	-	oxidation of propylene	4.28 mmol g ⁻¹ h ⁻¹ and 79.7% selectivity of propylene glycol	-	73
a-KB	-	H-cell	in-cell	0.05 M H ₂ SO ₄	~90%@0.3~0 V	oxidation of methane	80.7% selectivity of formic acid	6 h@0 V	74
MC-3	TS-1	flow-cell	in-cell	0.1M NaPi (pH=10)	~90%@0.4~0.6 V	oxidation of ethylene	51.48 mmol g ⁻¹ h ⁻¹ of ethylene glycol	50 h@-4 mA cm ⁻²	75
CB	TS-1	SE-cell	in-cell	pure water	>80%@50 mA cm ⁻²	oxidation of ethylene	99.7% selectivity of ethylene glycol	200 h@10 mA cm ⁻²	76
PDC@VG	-	flow-cell	in-cell	0.5 M H ₂ SO ₄ + 0.5 M C ₂ H ₄ O ₂	>80%@0.2~0.4 V	oxidation of acetic acid	58.5 μmol cm ⁻² h ⁻¹ of peracetic acid	7 h@-1.6 V	77
CB	-	flow-cell	in-cell	1 mol L ⁻¹ HCOOH/CHO OK + 1 wt% H ₃ PO ₄	-	oxidation of formic acid	24% conversion of performic acid	6 h@100 mA cm ⁻²	78

CB-3	Ti-MOR	flow-cell	in-cell	1 M NH ₄ HCO ₃	~80%@0.4~0.7 V	ammoxidations of cyclohexanone	96% FE and 99% selectivity of cyclohexanone oxime	45 h@0.1 V	79
Co-N-C	TS-1	H-cell	in-cell	0.5 M H ₂ SO ₄	>90%@0.1~0.3 V	oxidation of phenol	30.05% FE and 99.45% selectivity of bisphenol	-	80
Mg ₃ (HHT P) ₂	TS-1	flow-cell	in-cell	pure water	~90%@1.6~2.6 V	oxidation of propylene	99% selectivity of propylene oxide	8 h@20 mA cm ⁻²	81
FeSAs/AC s-BCC	-	H-cell	in-cell	0.1 M Na ₂ SO ₄ (pH~2.43) + 10 mM EG + 1 mM Fe ²⁺	>90%@0.2~0.7 V	oxidation of ethylene glycol	86.2% selectivity of formic acid	-	82
Co-O-C	-	flow-cell	in-cell	0.1 M KHCO ₃ + 0.05 M KNO ₃	>70%@0.3~0.6 V	oxidation of urea	4648.2 μg h ⁻¹ mg ⁻¹ and 53% FE of urea peroxide	-	83
Mn-NO- C _H	-	flow-cell	in-cell	1 M KOH	>80%@0~0.6 V	hydration of benzonitrile	100% conversion of benzonitrile	55 h@0.1 V	84
Ag foil	-	H-cell	in-cell	0.1 M HClO ₄ + 2 mM Fe ²⁺	~90%@0.2 V	oxidation of methane	11.5 mmol h ⁻¹ g _{Fe} ⁻¹ and 81.4% FE of formic acid	8 h@0.2 V	85
CoPorF/C NT	-	flow-cell	in-cell	pure water	>90%@0.2~0.5 V	hydroxylation of phenol	94.7% selectivity of dihydroxy benzene	48 h@ 50 mA cm ⁻²	57
B/N-onion carbon	-	flow-cell	in-cell	1 M KOH	>80%@0.2~0.7 V	synthesize of sodium peroxyborate and dibenzoyl peroxide	-	10 h@0.7 V	86
c- NiSe ₂ /CFP	-	H-cell	in-cell	0.1 M NaHSO ₄ /Na ₂ S	>60%@0~0.6 V	oxidation of glycerol	55% conversion of C ₃ products	37 h@0.6 V	87

O ₄ (pH ~2.8)									
T-NiOC	-	flow-cell	in-cell	0.4 M NaOH + saturated Ca(OH) ₂	-	synthesis of alkaline- earth metal peroxides	99% selectivity of calcium peroxide	1000 h@50 mA cm ⁻²	88
graphite rod	-	H-cell	in-cell	0.25 M LiSO ₄ /H ₂ SO ₄ (pH=3) with 1 Mm Fe ²⁺	-	oxidation of nitrogen	141.83 μmol h ⁻¹ g ⁻¹ and 25.37% FE of nitrates	10 h@0 V	89

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