Crystal structure validation of verinurad via proton-detected ultra-fast MAS NMR and machine learning

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Supporting Information

Raw data statement: All data and code used are available from [link to be added upon publication] under the license CC-BY-4.0 (Creative Commons Attribution-ShareAlike 4.0 International).

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1. Experimental Details

All the raw data and the associated pulse sequences and full acquisition parameters are available at: [will be added on publication].

The sample was purchased from Selleckchem, and the as received sample was packed in 0.7 and 0.4 mm rotors after being crushed with a mortar and pestle. All the experiments at 100 kHz MAS were acquired using a Bruker 0.7 mm room temperature HCN CP-MAS probe at a magnetic field of 21.1 T corresponding to a ¹H frequency of 900 MHz. The temperature was kept constant using a VT flow calibrated to compensate for the frictional heating measured externally using KBr. All the experiments at 160 kHz MAS were acquired on 18.8 T Bruker Avance Neo spectrometer corresponding to a ¹H frequency of 800 MHz using a Bruker 0.4 mm HCN CP-MAS probe. The sample temperature was regulated to 295 K using VT flow at 280 K. A States-TPPI acquisition scheme was used in all 2D experiments to obtain phase-sensitive two-dimensional spectra. All spectra were phase and baseline corrected. An exponential window function of 100 Hz in the direct dimension was applied prior to Fourier transformation of the hCH spectra. No window function was applied to the ¹H 1D MAS spectra. For the PIPNet ¹H spectrum a series of 31 spectra were recorded at spinning rates of 40 to 100 kHz MAS, and then used as input into PIPNet as described in reference 1, yielding the spectrum shown in figure 3.

Exper iment	MAS rate (kHz)	VT (K)	¹ H 90° RF amplitud e (kHz)	CP contact time, ms	recycl e delay (s)	Number of FID points	SW (kHz)	Size of real spectrum:	Nu mbe r of scan s
1D ¹ H echo	160	280	250	-	1.5	2048	50	4096	64
hCH long- range	160	280	227	4 (direct CP) & 2 (back CP)	1.5	4096 (F2) / 1024 (F1)	200 (F2) / 44 (F1)	8192 (F2) / 2048 (F1)	144
hCH short- range	100	285	299	0.25 (direct CP) & 0.125 (back CP)	1.5	4096 (F2) / 512 (F1)	200 (F2) / 50 (F1)	8192 (F2) / 1024 (F1)	200
¹ H- ¹ H BABA	160	280	250	-	1.5	4096 (F2) / 512 (F1)	91 (F2) / 40 (F1)	8192 (F2) / 1024 (F1)	64
¹³ C CP	100	280	303	2	1.5	4096	139	16384	819 2

Table S1. NMR experimental details for verinurad.



Figure S1. The aromatic region of the 2D hCH long-range spectrum of verinurad at 100 kHz MAS in blue (2 ms contact time) and at 160 kHz MAS in red (4 ms contact time).

2. Assignment

Atom label	¹³ C Chemical Shift, ppm	¹ H Chemical Shift, ppm
2	148.5	9.09
3	134.2	
4	152.1	
5	122.2	7.84
6	145.9	8.67
7	140.8	
8	129.2	8.07
9	132.8	6.63
10	109.7	
11	131.4	
12	124.0	7.86
13	129.1	7.19
14	127.4	6.10
15	127.6	7.25
16	132.5	
18	49.1	
19	25.1 / 27.0	1.12 / 1.01
20	118.1	
22	27.0 / 25.1	1.01 / 1.12
23	175.9	, , , , , , , , , , , , , , , , , , , ,
24		15.91

Table S2. Solid-state NMR assignment of ¹H and ¹³C chemical shifts of verinurad as received.

3. Probabilistic assignment

The input files used for the Bayesian probabilistic assignment (verinurad_13c.in and verinurad_C-H_mult_1.in) and the complete output data generated are given as additional Supplementary Files.

4. Chemical shift calculations

The input files used for the DFT and ShiftML2 and the complete output data generated are given as additional Supplementary Files. The optimization of the proton positions of the SXRD structure and the NMR computations were carried out using the plane-wave DFT software Quantum ESPRESSO version 6.5 with the PBE density functional, a Grimme D3 dispersion correction, wavefunction and charge density energy cut-offs of 120 and 960 Ry, respectively, and ultrasoft pseudopotentials with GIPAW reconstructure relaxation but using refined plane wave and charge density energy cut-offs of 120 and 960 Ry, respectively, a 4 x 2 x 1 Monkhorst–Pack k-point grid (with a maximum spacing of 0.05 Å⁻¹), and the ultrasoft pseudopotentials with GIPAW reconstruction from the USSP pseudopotential database v1.0.0. Conversion of σ_{calc} to δ_{calc} were performed using equation 1.

$$\delta_{calc} = \sigma_{ref} - b\sigma_{calc}$$

(1)

The rescaling parameters σ_{ref} and *b* were determined by linear regression between computed shielding and experimental shifts, permuting any ambiguities. The RMSE obtained from GIPAW NMR calculations are 0.30 ppm for ¹H and 2.48 ppm for ¹³C.

Atom label	¹³ C Chemical Shift, ppm	¹ H Chemical Shift, ppm
2	152.4	8.75
3	140.6	
4	153.5	
5	122.0	7.32
6	143.6	8.32
7	141.2	
8	134.0	7.79
9	132.1	5.95
10	105.5	
11	129.5	
12	123.5	7.92
13	125.9	7.09
14	129.5	6.33
15	126.8	7.15
16	132.6	
18	53.8	
19	25.0	1.80
20	113.7	
22	25.6	1.61
23	171.8	
24		16.69

Table S3. ¹H and ¹³C ShiftML2 predicted chemical shifts of verinurad.

Table S4. ¹H and ¹³C DFT calculated chemical shifts of verinurad.

Atom label	¹³ C Chemical Shift, ppm	¹ H Chemical Shift, ppm
2	148.4	8.49
3	136.3	
4	155.9	
5	120.5	7.31
6	143.1	8.46
7	143.0	
8	128.4	8.12
9	134.3	7.08
10	108.0	
11	130.4	
12	124.8	8.05
13	126.7	7.38
14	127.3	6.35
15	128.9	7.21
16	132.4	
18	49.6	
19	27.2	1.30
20	123.5	
22	23.2	0.90
23	169.6	
24		16.08

5. References

(1) Cordova, M.; Moutzouri, P.; Simões de Almeida, B.; Torodii, D.; Emsley, L. Pure isotropic proton NMR spectra in solids using deep learning. *Angewandte Chemie International Edition* **2023**, *62* (8), e202216607.