## **Supplementary Information**

## Enhancement of Hydrogen Production via Methanol Steam Reforming Using Ni-based Catalyst Supported by Spongy Mesoporous Alumina

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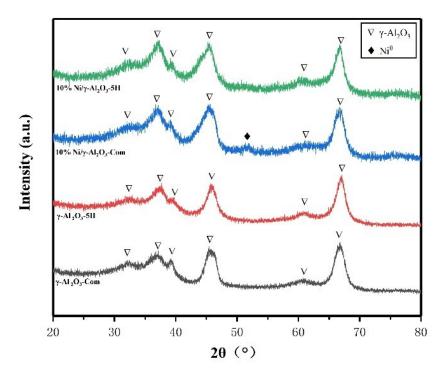


Fig. S1. XRD spectrum of the  $\gamma\text{-}Al_2O_3$  supports and  $Ni/\gamma\text{-}Al_2O_3$  catalysts.

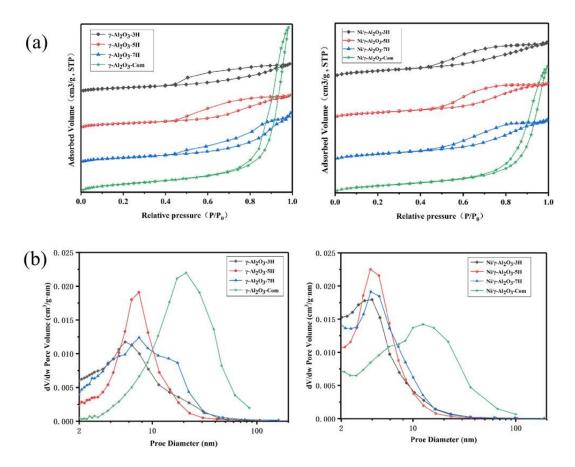


Fig. S2. The N<sub>2</sub> adsorption-desorption isotherms and the corresponding pore size distributions of  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-3H/5H/7H/Com supports and Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-3H/5H/7H/Com catalysts.

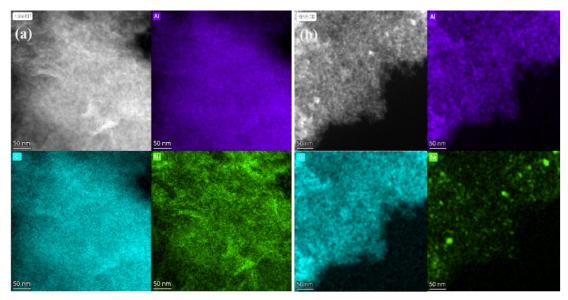


Fig. S3. TEM-EDS images of the (a,b) Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-5H and (c,d) Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-Com catalysts.

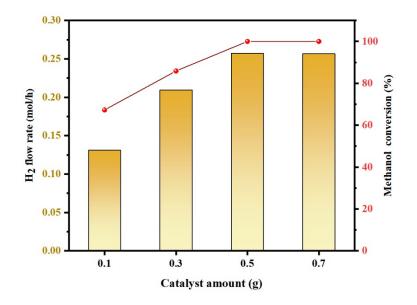


Fig. S4. The effect of catalyst amounts on the methanol steam reforming to hydrogen was examined at the temperature of 450 °C over  $Ni/\gamma$ -Al<sub>2</sub>O<sub>3</sub>-5H catalyst.

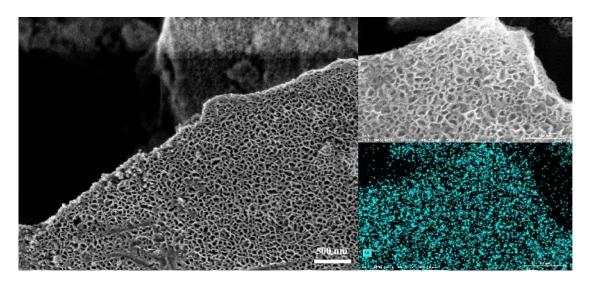


Fig. S5. SEM and EDS images of the spent  $Ni/\gamma\text{-}Al_2O_3\text{-}5H$  catalyst.

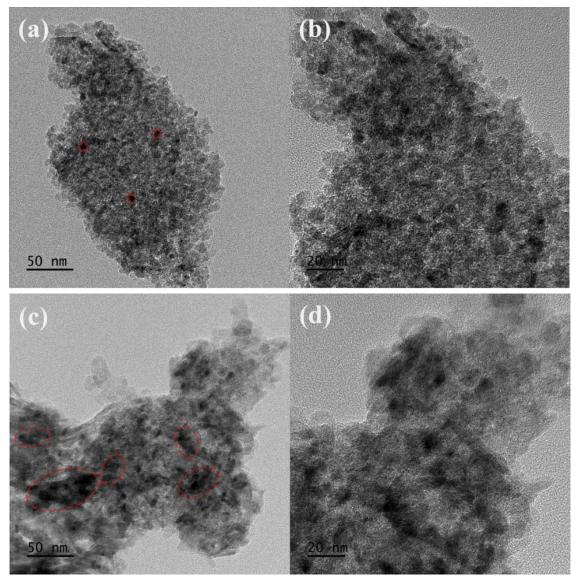


Fig. S6. TEM images of the (a,b) spent Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-5H and (c,d) spent Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-Com catalysts

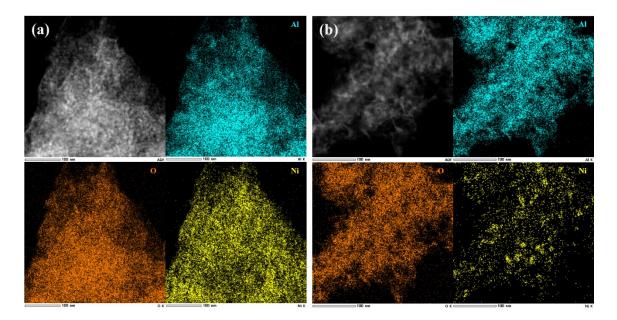


Fig. S7. TEM-EDS images of the (a,b) spent Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-5H and (c,d) spent Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-Com

catalysts.

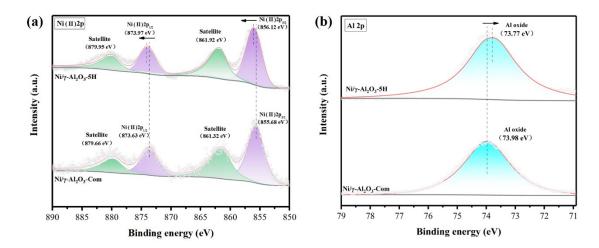


Fig. S8. XPS spectra (a) Ni 2p and (b) Al 2p of the spent Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-5H and spent Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-Com

catalysts.