

# Supporting Information

## Eggshell membrane-derived metal sulfide catalysts for seawater splitting

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**Table S1**

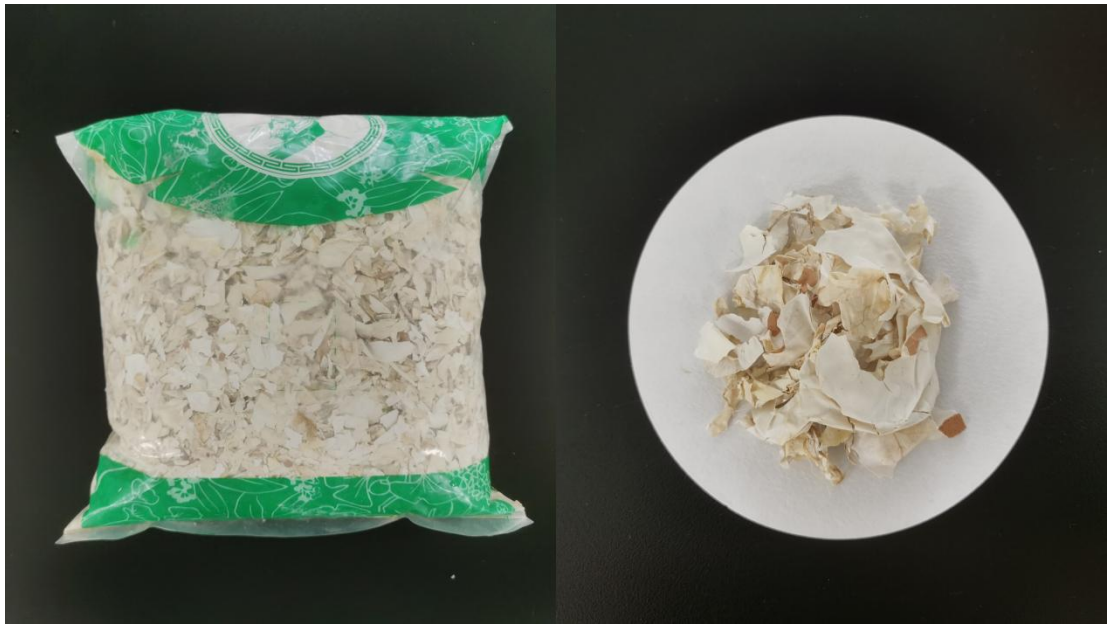
Composition of feedstock and pyrolysis temperature for sample syntheses.

Samples	Chemicals				Hydrothermal temperature /°C	Calcination temperature /°C
	Co(NO <sub>3</sub> ) <sub>2</sub> ·6H <sub>2</sub> O /mmol	Fe(NO <sub>3</sub> ) <sub>3</sub> ·9H <sub>2</sub> O /mmol	Ni(NO <sub>3</sub> ) <sub>2</sub> ·6H <sub>2</sub> O /mmol	ESM /g		
<b>Co<sub>9</sub>S<sub>8</sub>/ESM-T</b>	Co <sub>9</sub> S <sub>8</sub> /ESM-200					-
	Co <sub>9</sub> S <sub>8</sub> /ESM-800	0.9	-	-	200	800
	Co <sub>9</sub> S <sub>8</sub> /ESM-900					900
	Co <sub>9</sub> S <sub>8</sub> /ESM-1000					1000
<b>Co<sub>8</sub>FeS<sub>8</sub>/ESM-T</b>	Co <sub>8</sub> FeS <sub>8</sub> /ESM-200					-
	Co <sub>8</sub> FeS <sub>8</sub> /ESM-800	0.8	0.1	-	200	800
	Co <sub>8</sub> FeS <sub>8</sub> /ESM-900					900
	Co <sub>8</sub> FeS <sub>8</sub> /ESM-1000					1000
<b>FeS<sub>x</sub>/ESM-T</b>	FeS <sub>x</sub> /ESM-200					-
	FeS <sub>x</sub> /ESM-800	-	0.9	-	200	800
	FeS <sub>x</sub> /ESM-900					900
	FeS <sub>x</sub> /ESM-1000					1000
<b>Ni<sub>9</sub>S<sub>8</sub>/ESM-T</b>	Ni <sub>9</sub> S <sub>8</sub> /ESM-900	-	-	0.9	200	900
<b>Ni<sub>8</sub>FeS<sub>8</sub>/ESM-T</b>	Ni <sub>8</sub> FeS <sub>8</sub> /ESM-900	-	0.1	0.8	200	900

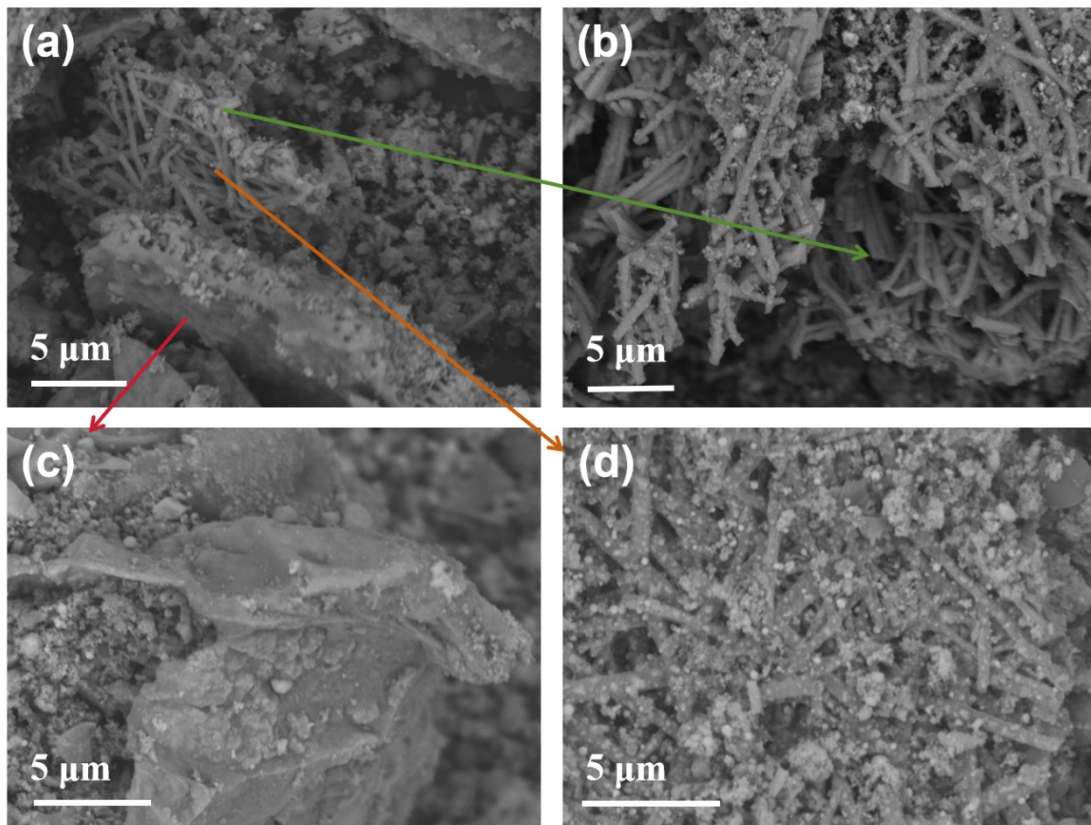
**Table S2**

Comparison of the OER performance of the Co-based electrocatalysts.

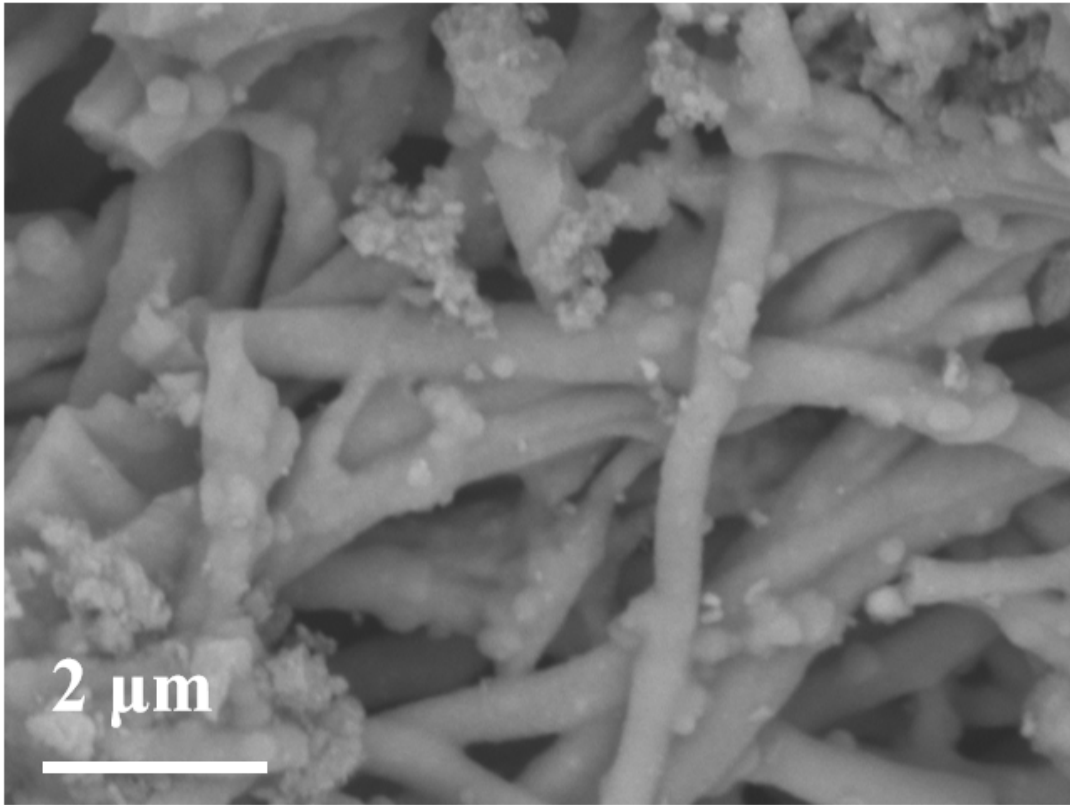
Electrocatalyst	Overpotential (mV) (j=10 mA cm <sup>-2</sup> )	Tafel slope (mV dec <sup>-1</sup> )	Electrolyte	Substrate	Loading (μg cm <sup>-2</sup> )	Reference
Co <sub>8</sub> FeS <sub>8</sub> /ESM-900	270	51.04	1 M KOH	GCE	120	This work
CoPS	280.7	87	1 M KOH	NF	769	R1
NiCoSe S/BP	285	116	1 M KOH	GCE	160	R2
Fe-Co-Ni-S <sub>x</sub> /NF	280	86	1 M KOH	NF	-	R3
N/O-dual doped carbon coated CoNPs (EK-b)	378	60	1 M KOH	CC	205	R4
Co <sub>4</sub> Ni <sub>1</sub> S/CC	296	52	1 M KOH	CC	520	R5
CoS/MoS <sub>2</sub>	281	79	1 M KOH	GCE	132	R6
FeCoNiP@P-rGO	354	155	1 M KOH	-	-	R7
Co@S-MoO <sub>x</sub> NSs	274	62	1 M KOH	NF	-	R8
Co-Ni-S	368	86	1 M KOH	CC	1000	R9
Co@CoMoO <sub>x</sub> -α-CrOOH	278	67.9	1 M KOH	NF	-	R10
CNO@NSG	287.4	66.3	1 M KOH	NF	100	R11
Ni-Co/Ni-Co-O-P@CS	310	151	1 M KOH	-	-	R12
P, S-Co <sub>x</sub> O <sub>y</sub> /Cu@CuS NWs	280	73.9	1 M KOH	CF	184	R13



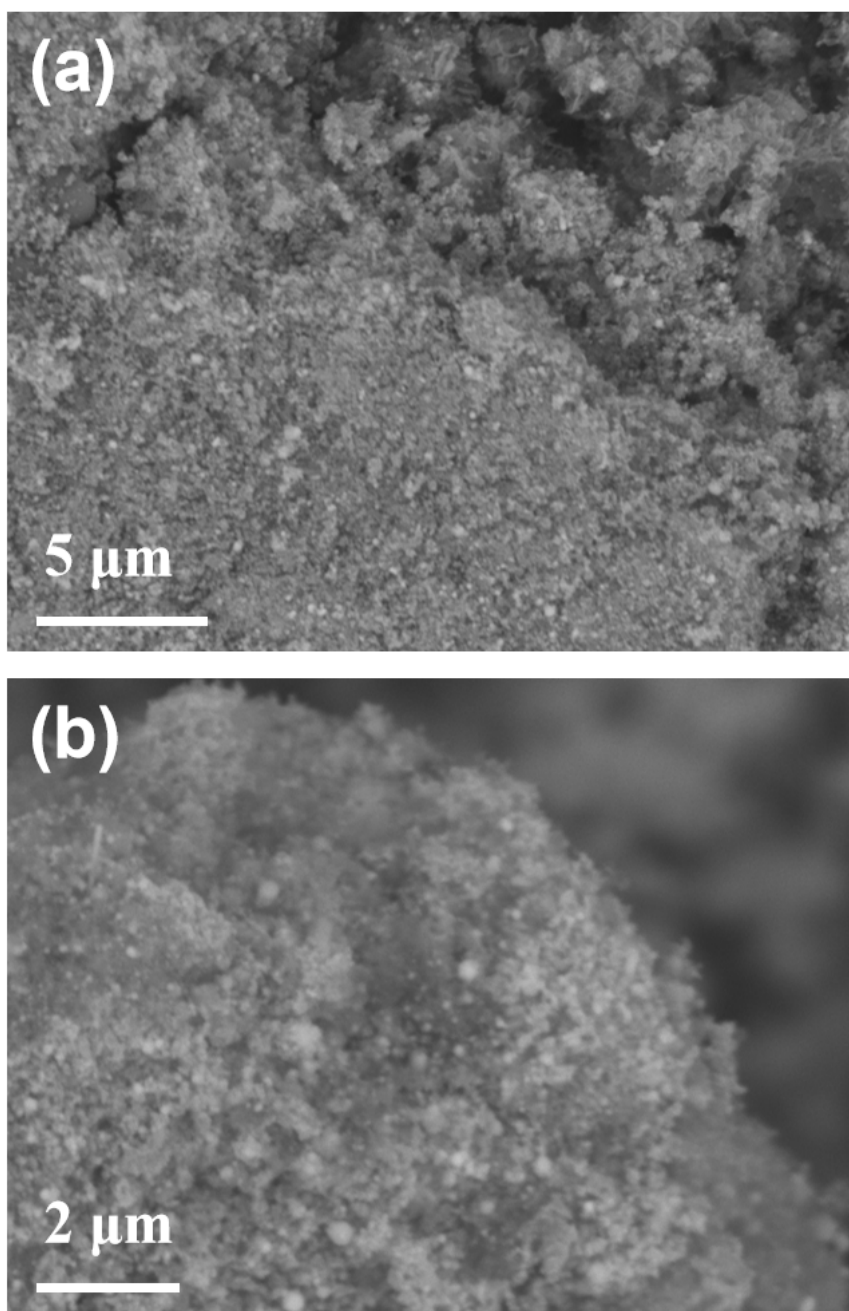
**Fig. S1** Digital photos of eggshell membranes.



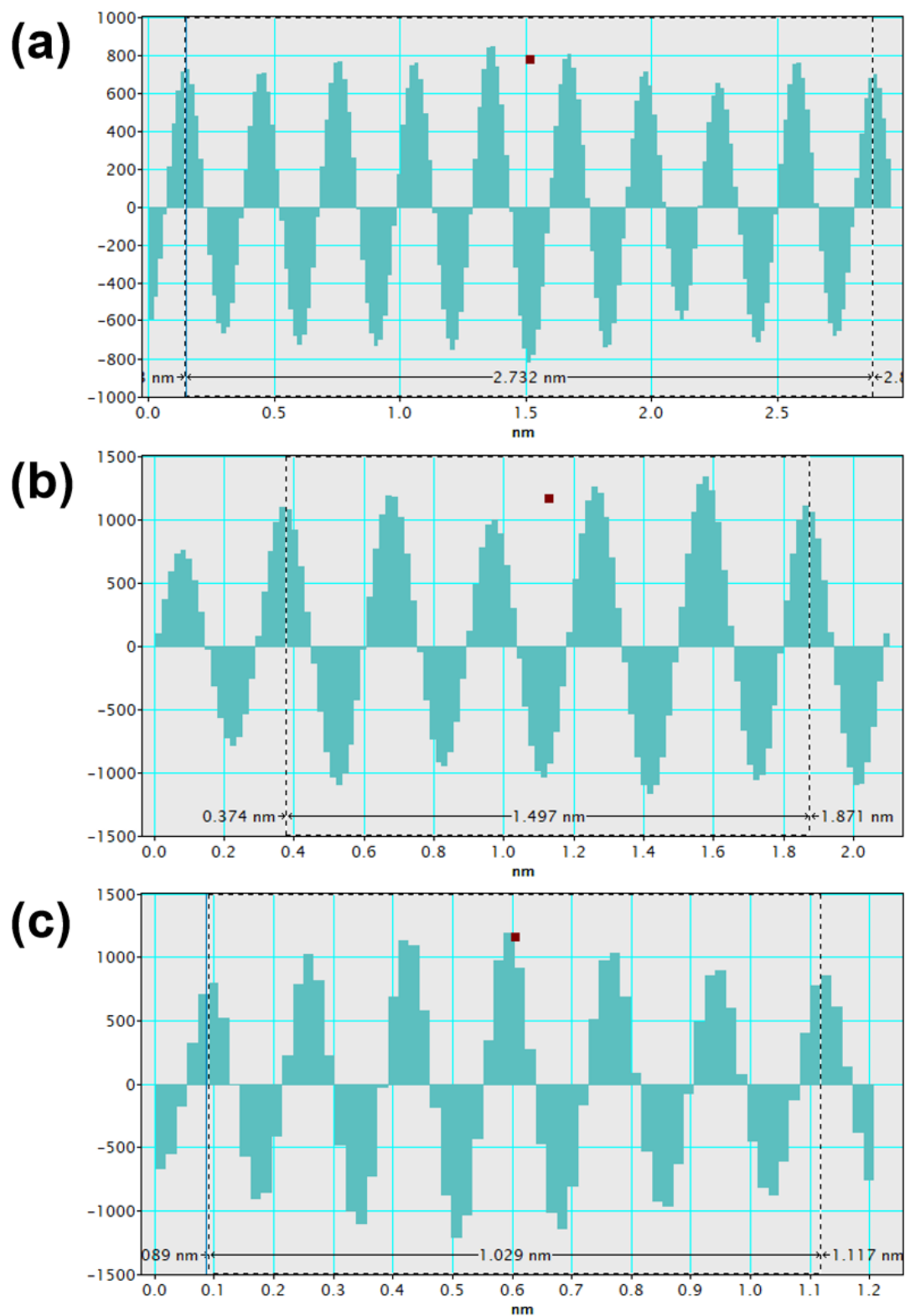
**Fig. S2** FESEM images of (a) individual ESM fragments, (b) the inner shell membrane, (c) the limiting shell membrane, and (d) the outer membrane after carbonization.



**Fig. S3** FESEM image of  $\text{Co}_9\text{S}_8/\text{ESM-900}$ .

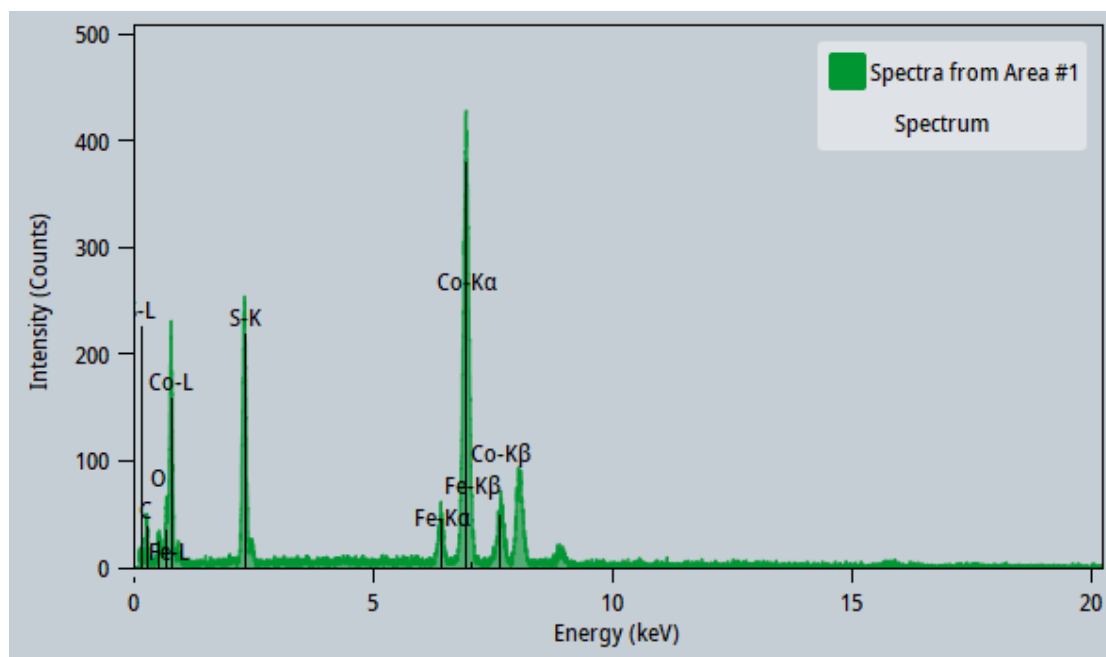


**Fig. S4** SEM images of FeS<sub>x</sub>/ESM-900.

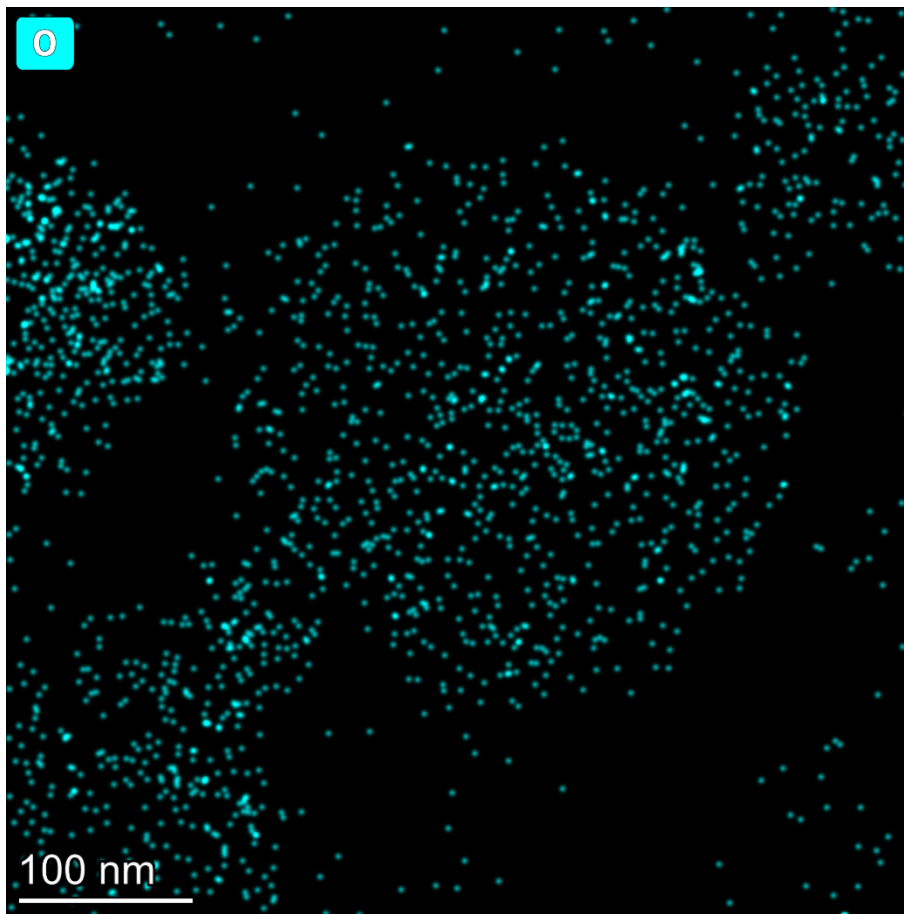


**Fig. S5** Intensity profiles within (a) region 1 and (b-c) region 2 shown in Fig. 1.

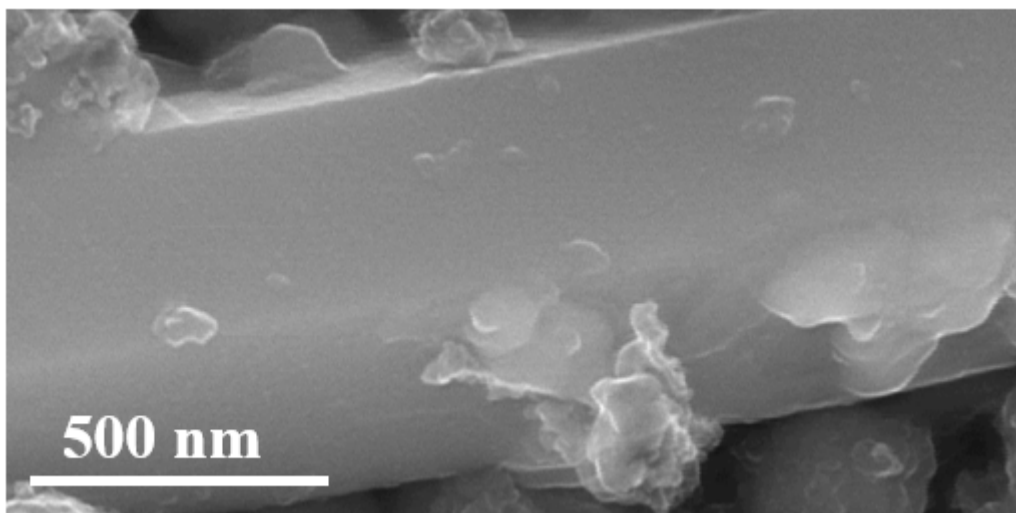




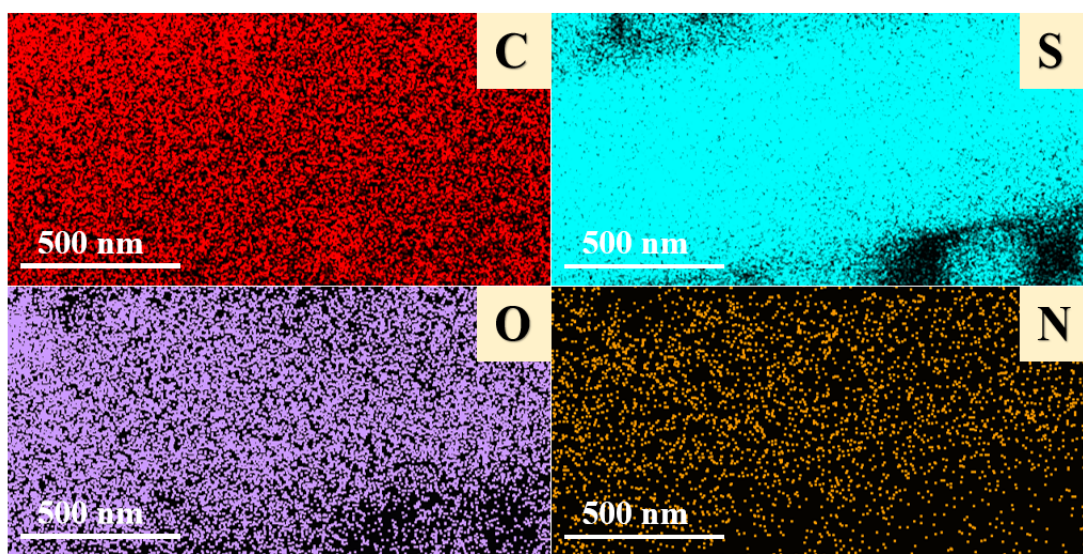
**Fig. S6** EDS spectrum of  $\text{Co}_8\text{FeS}_8/\text{ESM-900}$ .



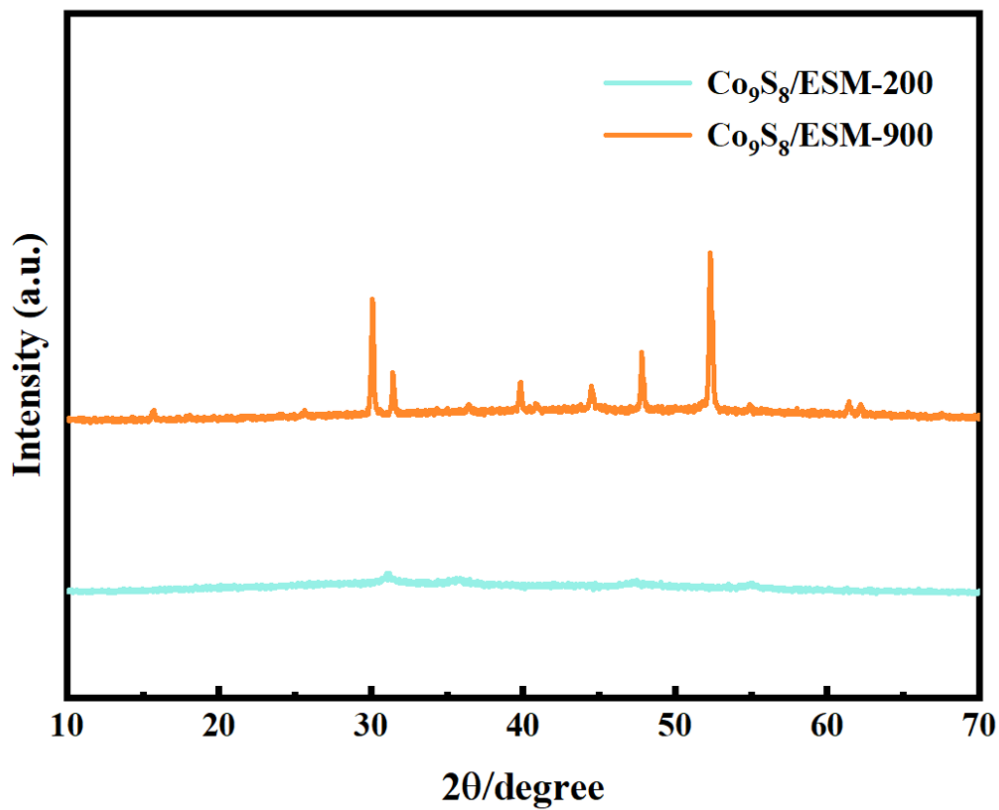
**Fig. S7** EDS element mappings of oxygen.



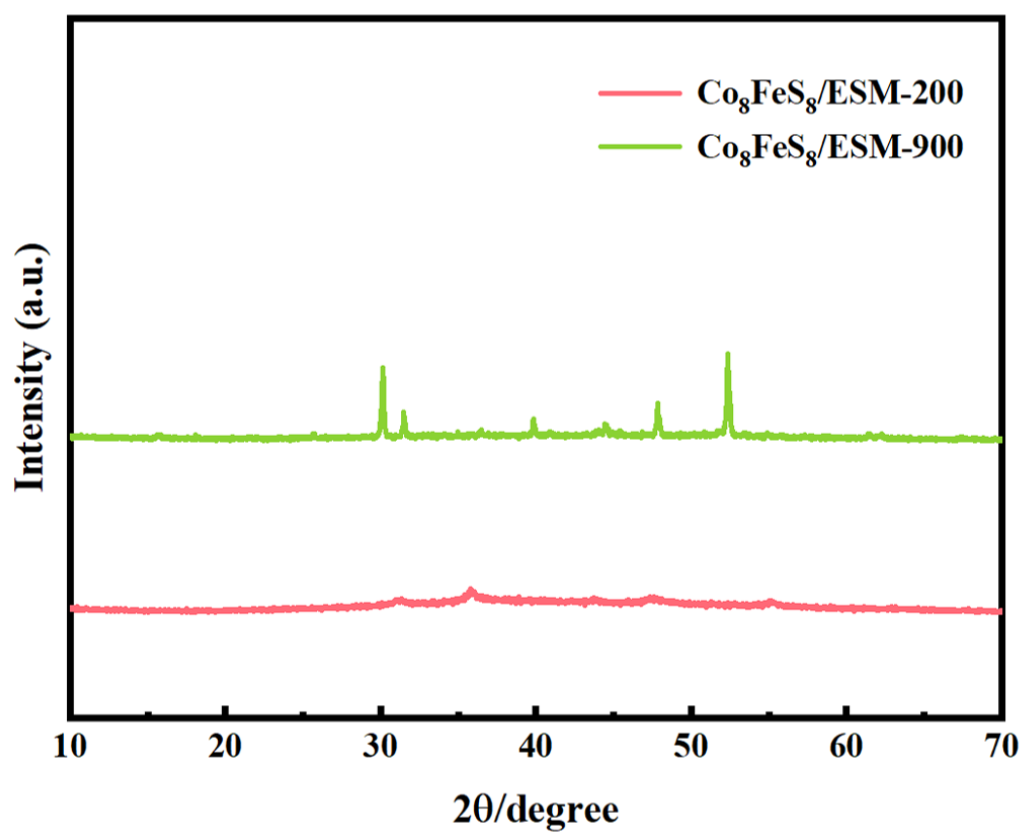
**Fig. S8** SEM images of the support.



**Fig. S9** EDS element mapping images of the support.



**Fig. S10** XRD patterns of  $\text{Co}_9\text{S}_8/\text{ESM-200}$  and  $\text{Co}_9\text{S}_8/\text{ESM-900}$ .



**Fig. S11** XRD patterns of  $\text{Co}_8\text{FeS}_8/\text{ESM-200}$  and  $\text{Co}_8\text{FeS}_8/\text{ESM-900}$ .

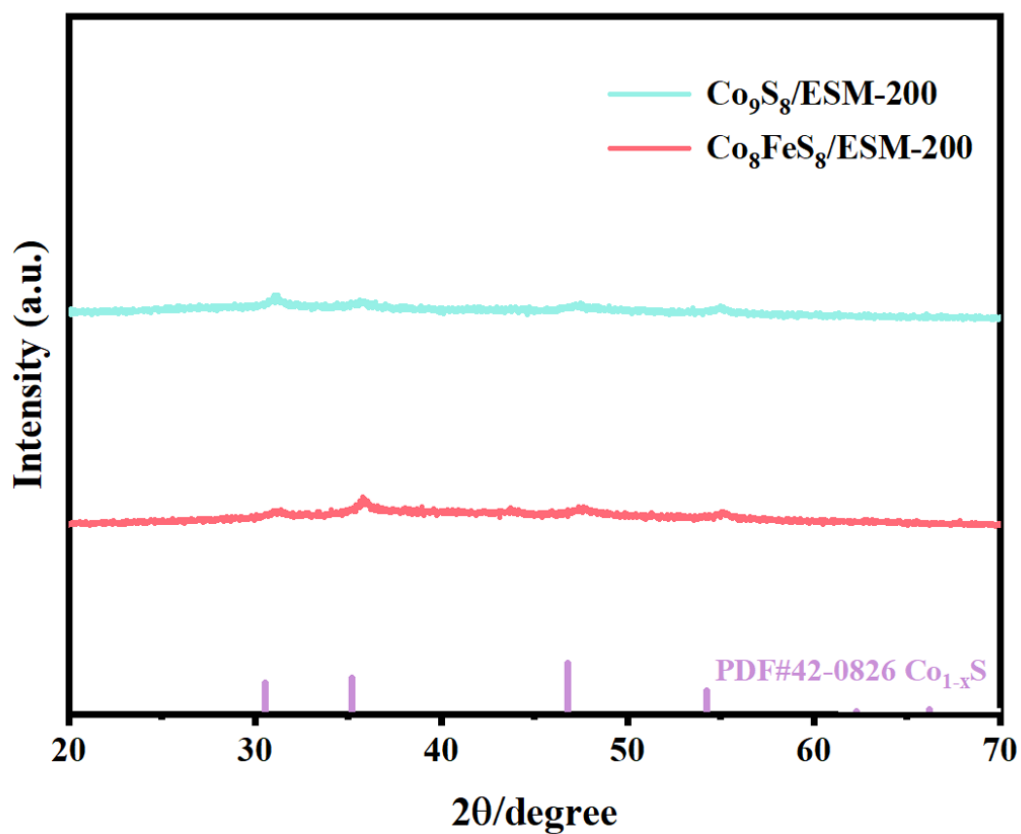
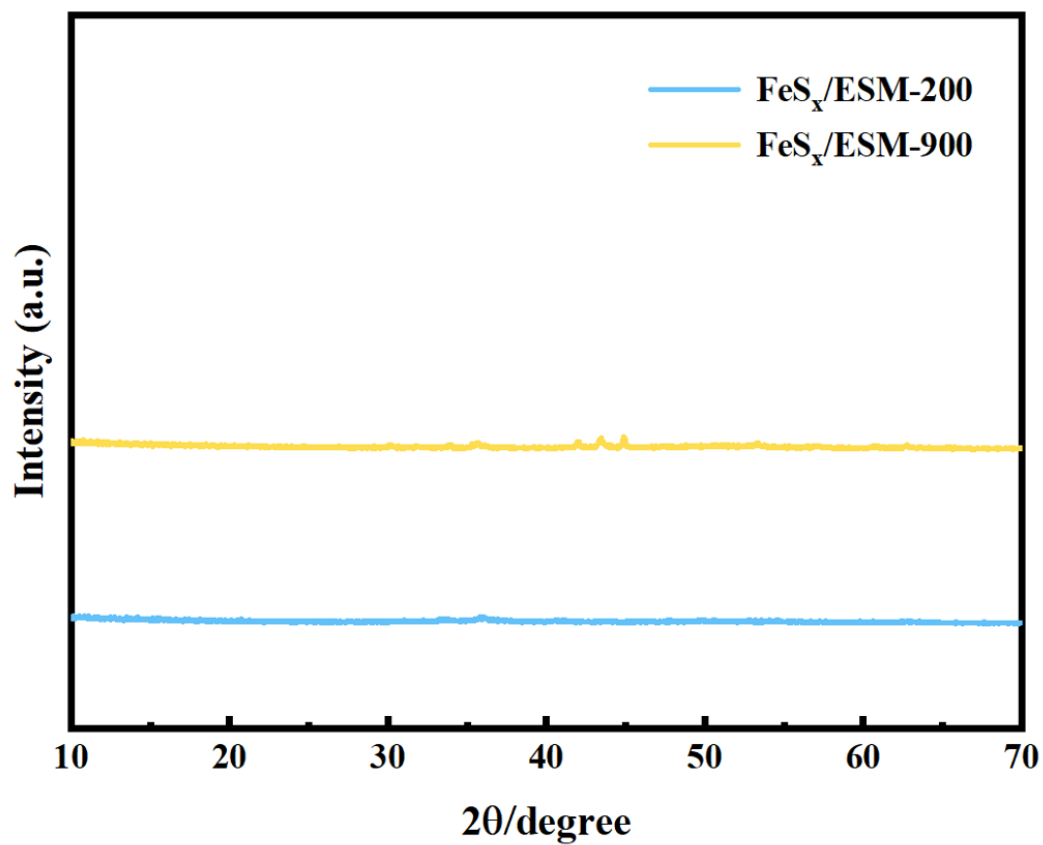
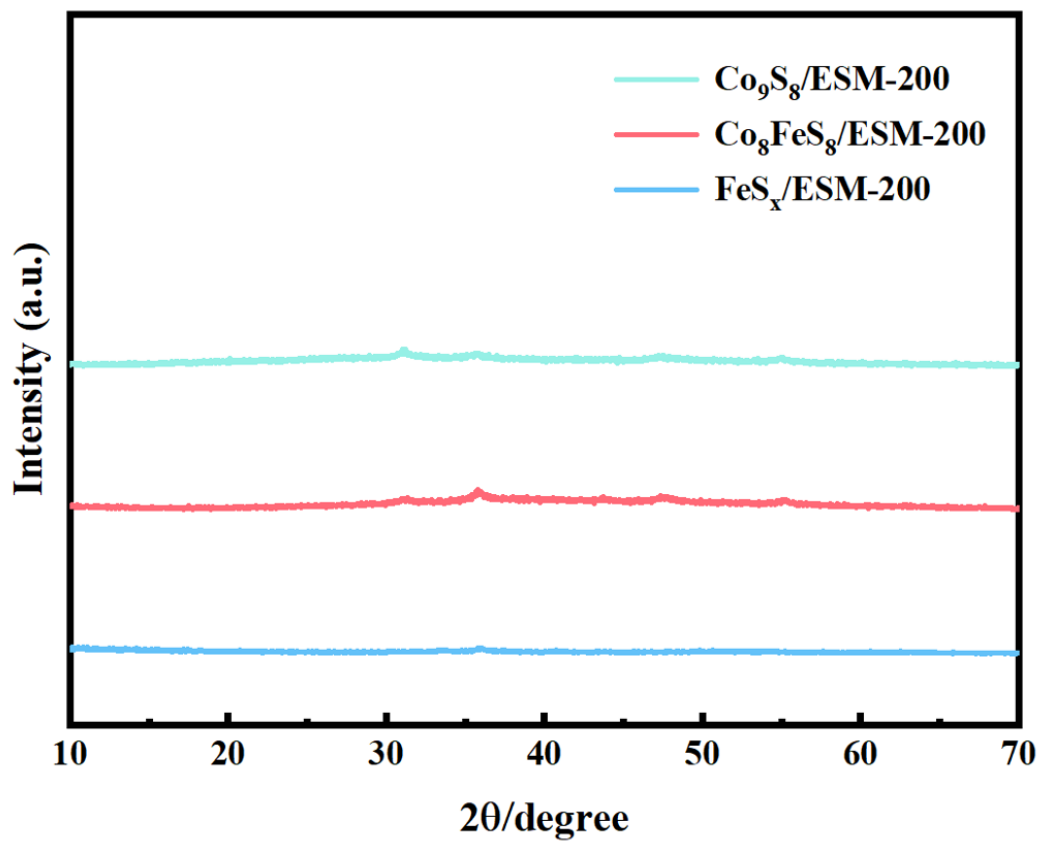


Fig. S12 XRD patterns of  $\text{Co}_9\text{S}_8/\text{ESM-200}$  and  $\text{Co}_8\text{FeS}_8/\text{ESM-200}$ .



**Fig. S13** XRD patterns of FeS<sub>x</sub>/ESM-200 and FeS<sub>x</sub>/ESM-900.





**Fig. S14** XRD patterns of Co<sub>9</sub>S<sub>8</sub>/ESM-200, Co<sub>8</sub>FeS<sub>8</sub>/ESM-200 and FeS<sub>x</sub>/ESM-200.

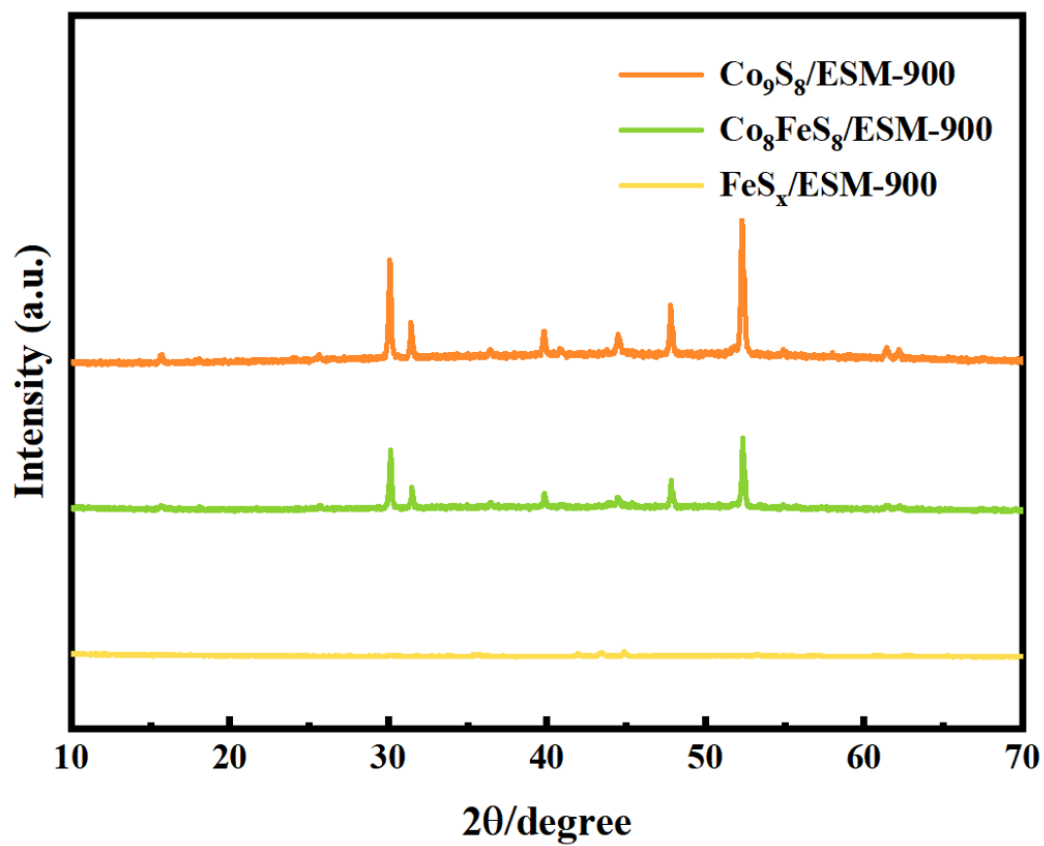
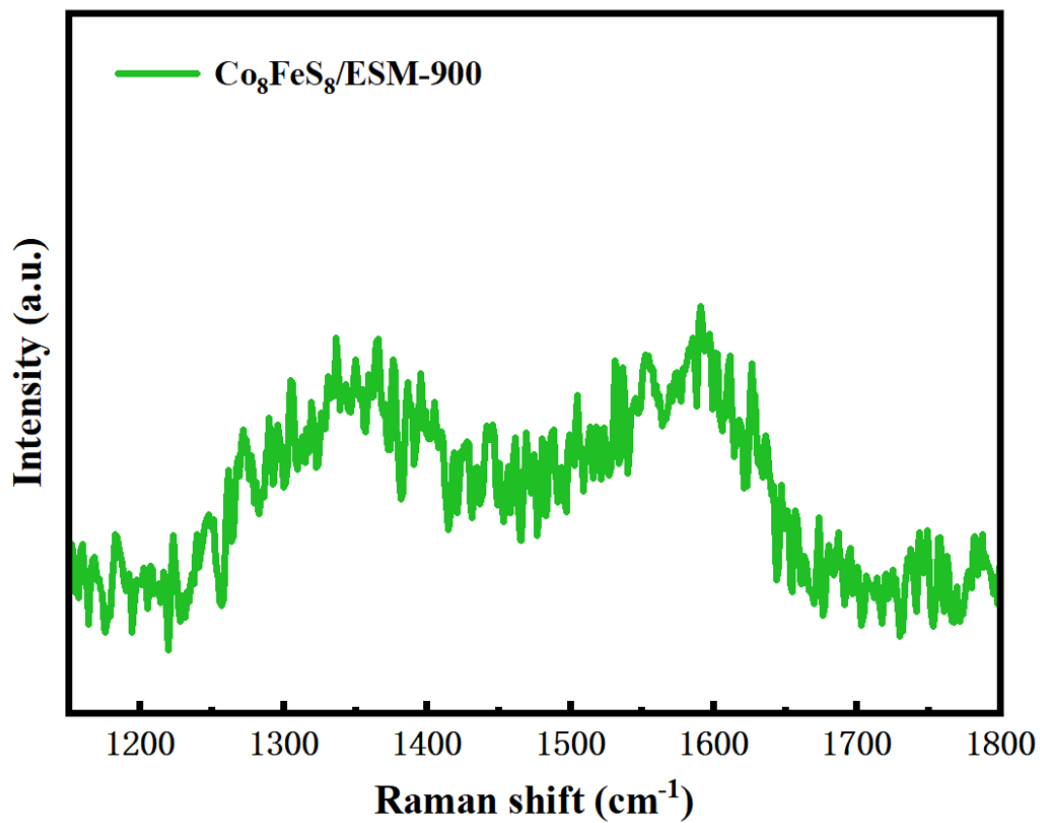


Fig. S15 XRD patterns of Co<sub>9</sub>S<sub>8</sub>/ESM-900, Co<sub>8</sub>FeS<sub>8</sub>/ESM-900 and FeS<sub>x</sub>/ESM-900.



**Fig. S16** Raman spectrum of  $\text{Co}_8\text{FeS}_8/\text{ESM-900}$ .

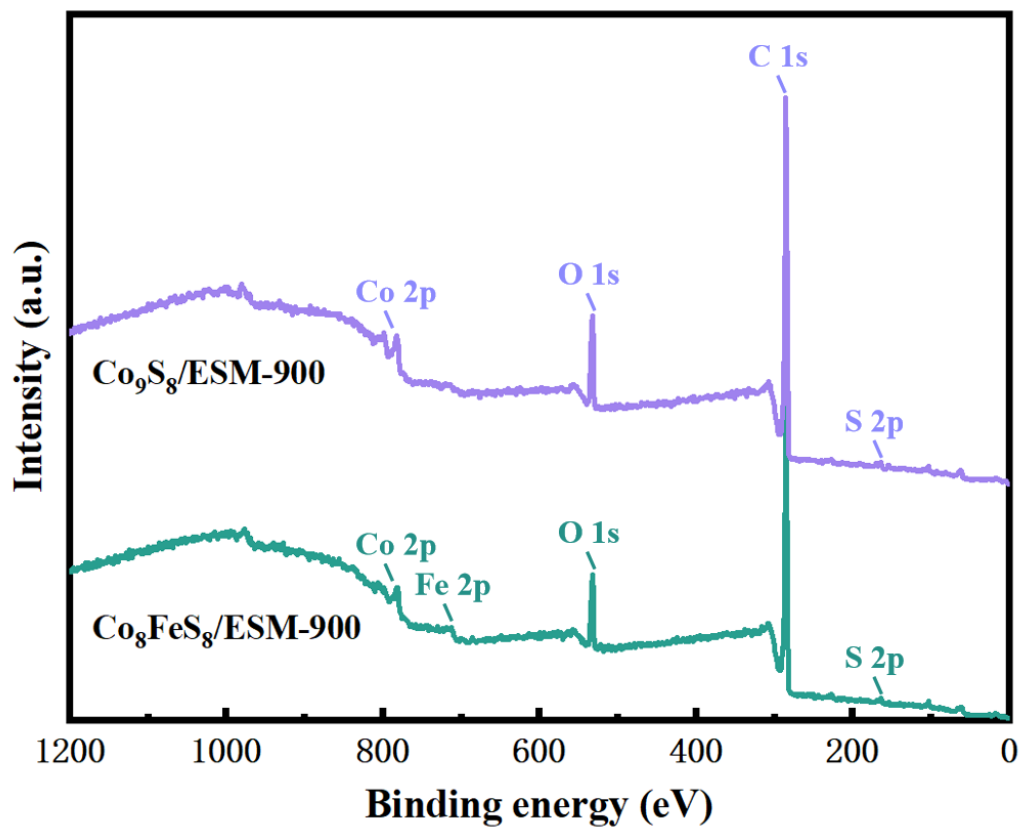


Fig. S17 Surveys spectra of  $\text{Co}_9\text{S}_8/\text{ESM-900}$  and  $\text{Co}_8\text{FeS}_8/\text{ESM-900}$ .

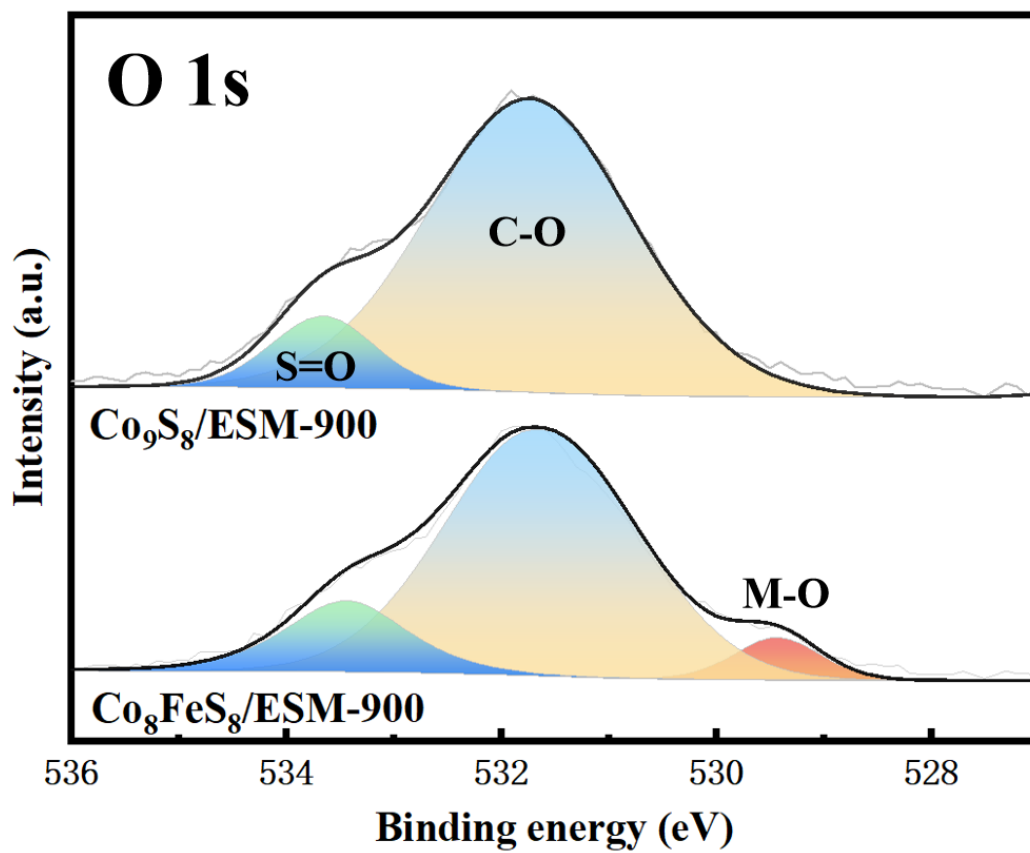
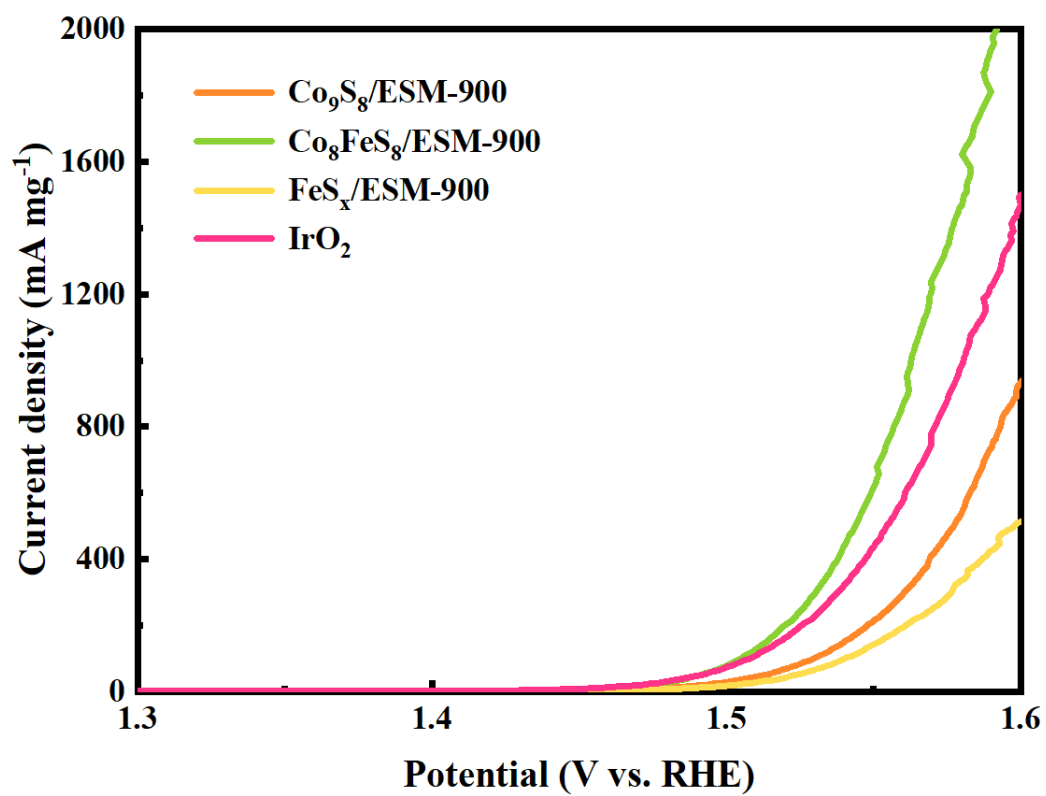
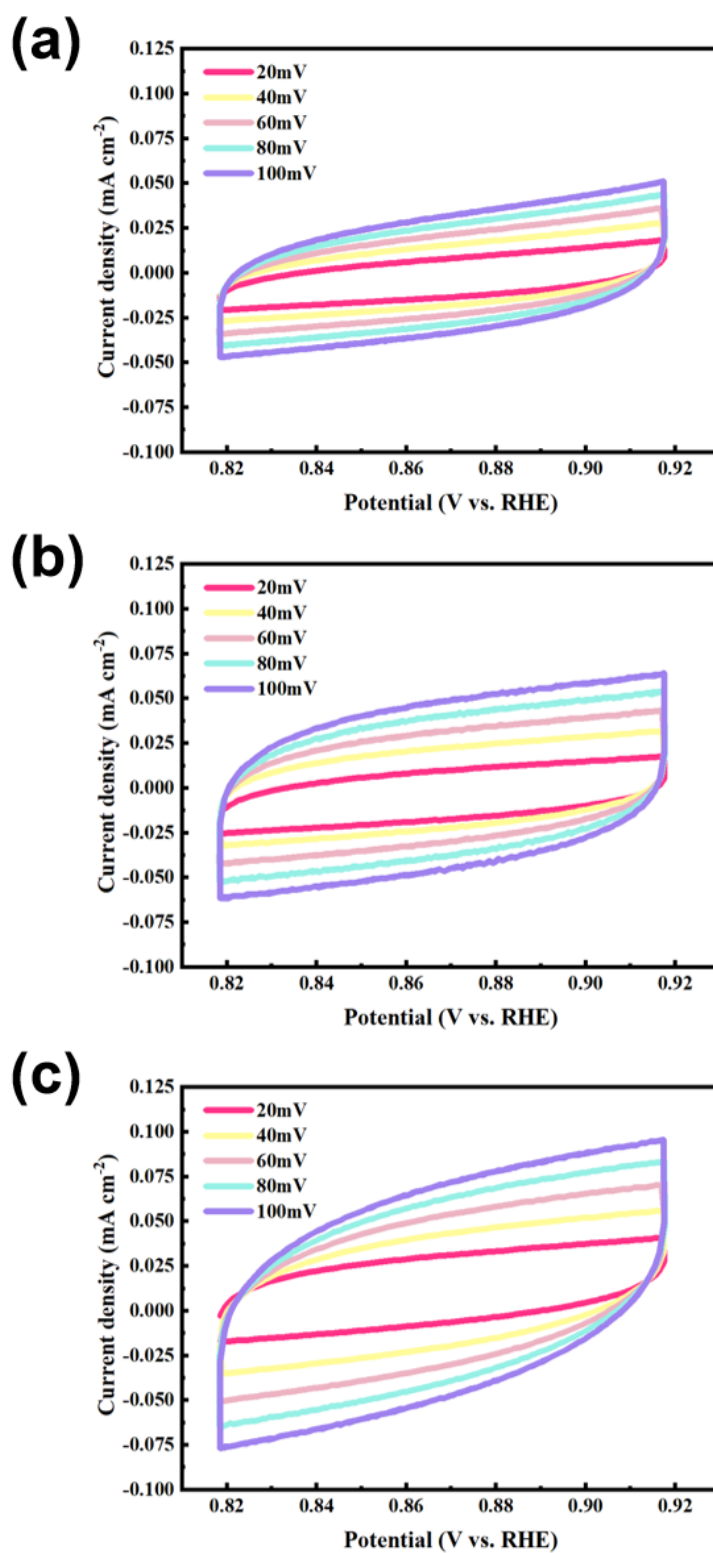


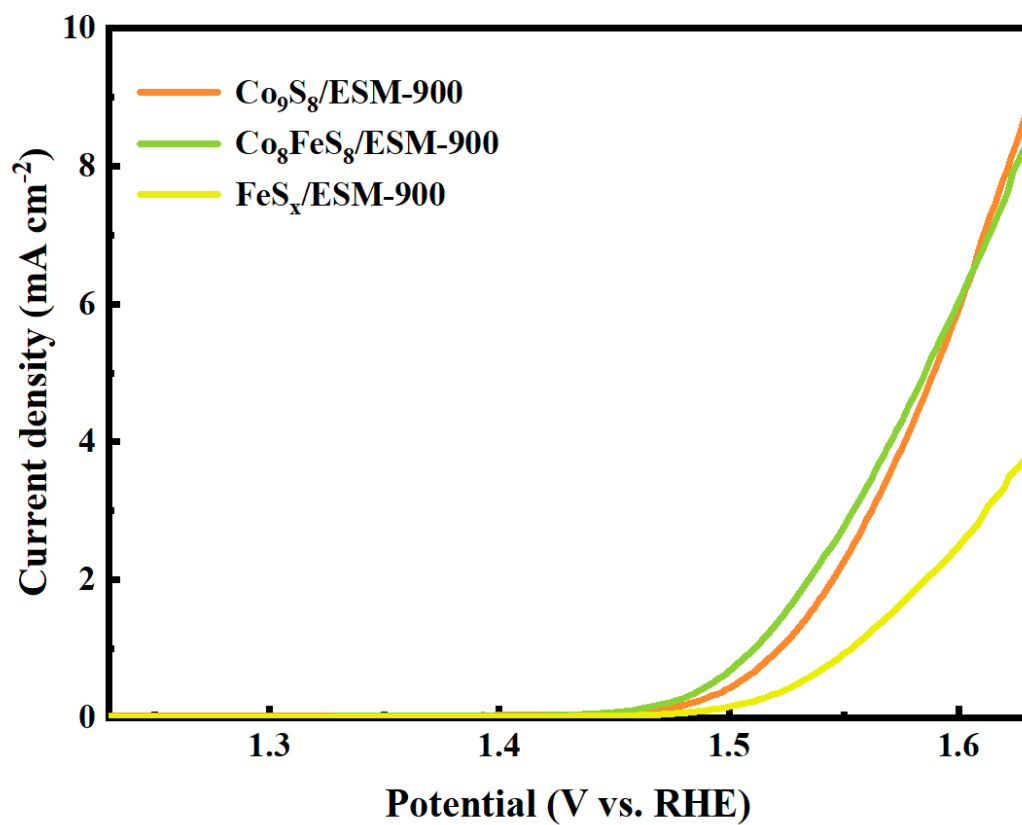
Fig. S18 XPS spectra of O1s.



**Fig. S19** Mass-normalized LSV curves of Co<sub>9</sub>S<sub>8</sub>/ESM-900, Co<sub>8</sub>FeS<sub>8</sub>/ESM-900, FeS<sub>x</sub>/ESM-900 and IrO<sub>2</sub> recorded in 1 M KOH.



**Fig. S20** CVs of (a) Co<sub>9</sub>S<sub>8</sub>/ESM-900, (b) Co<sub>8</sub>FeS<sub>8</sub>/ESM-900, and (c) FeS<sub>x</sub>/ESM-900 recorded with varying scan rates.



**Fig. S21** ECSA-normalized LSV curves of  $\text{Co}_9\text{S}_8/\text{ESM-900}$ ,  $\text{Co}_8\text{FeS}_8/\text{ESM-900}$  and  $\text{FeS}_x/\text{ESM-900}$  recorded in 1 M KOH.



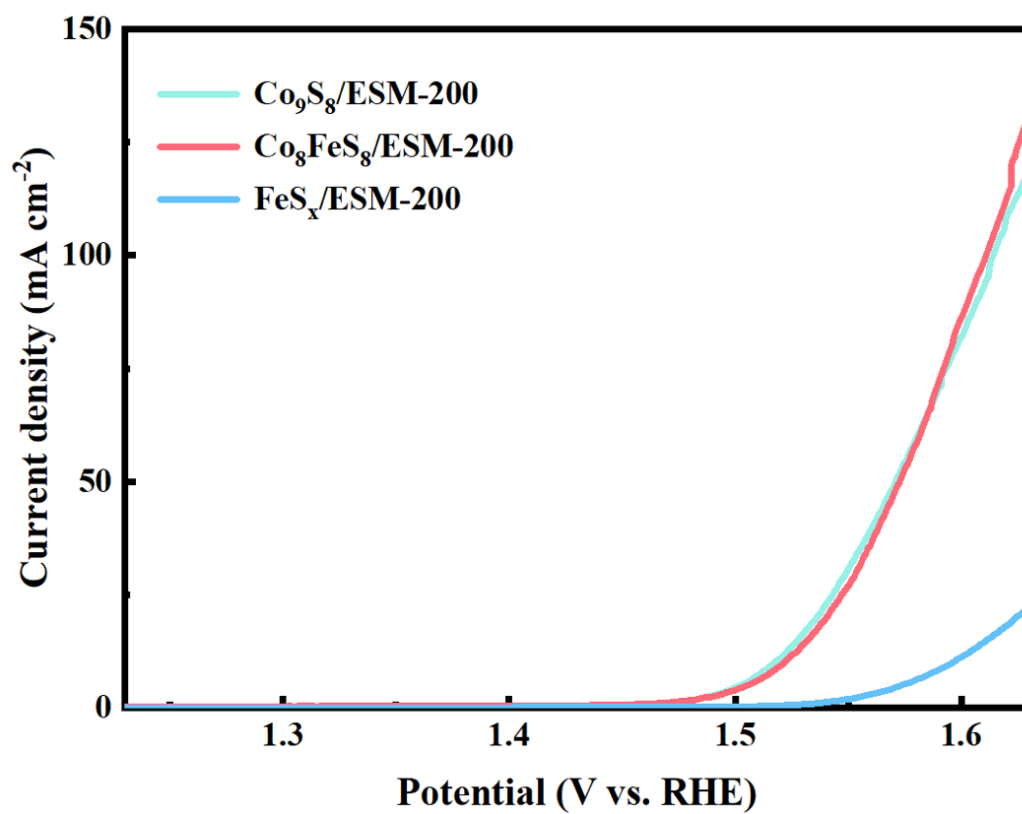
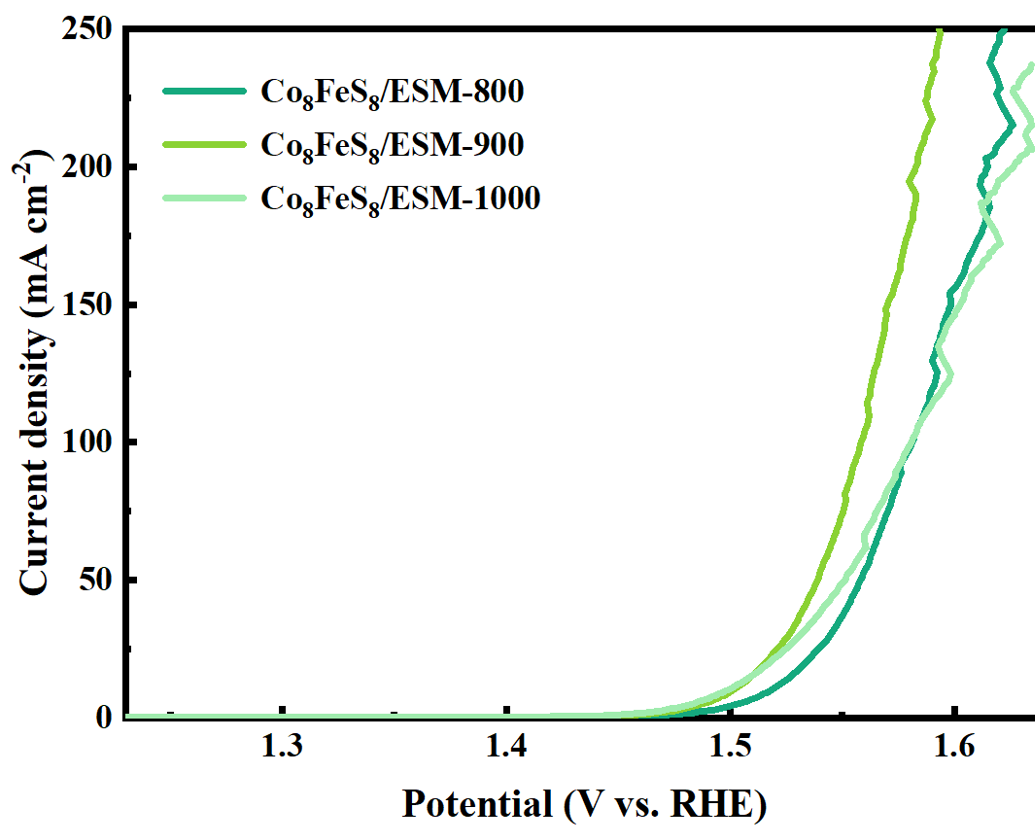
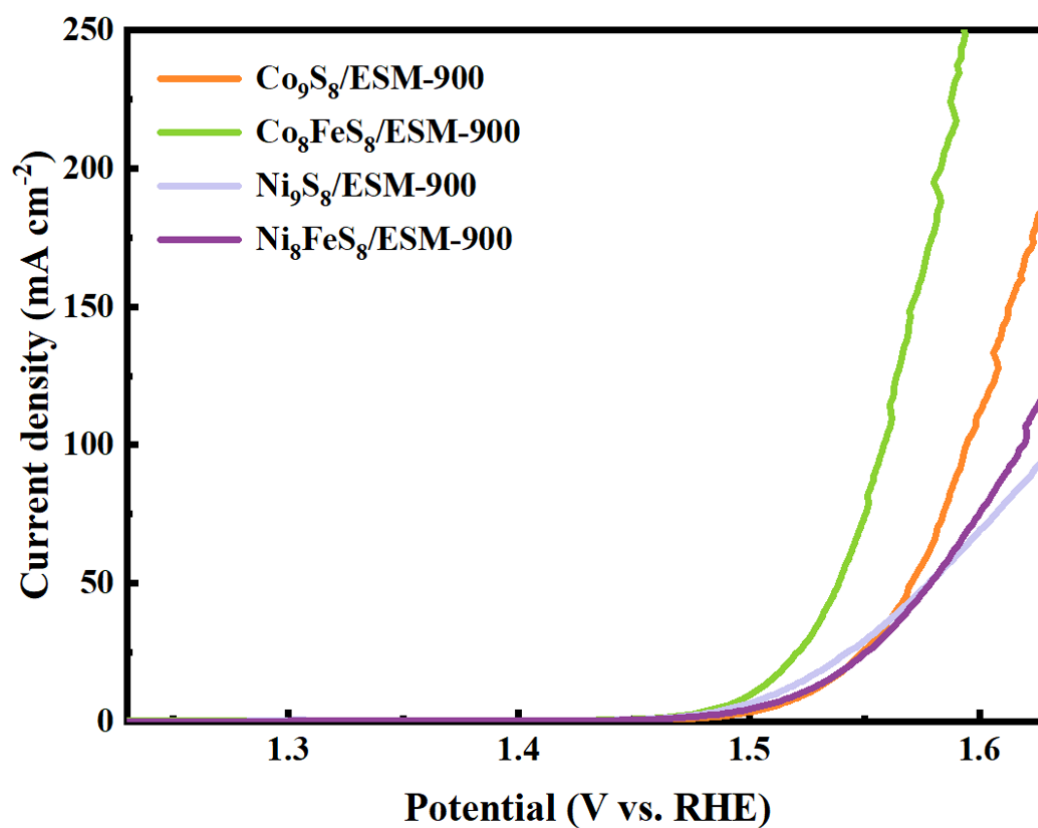


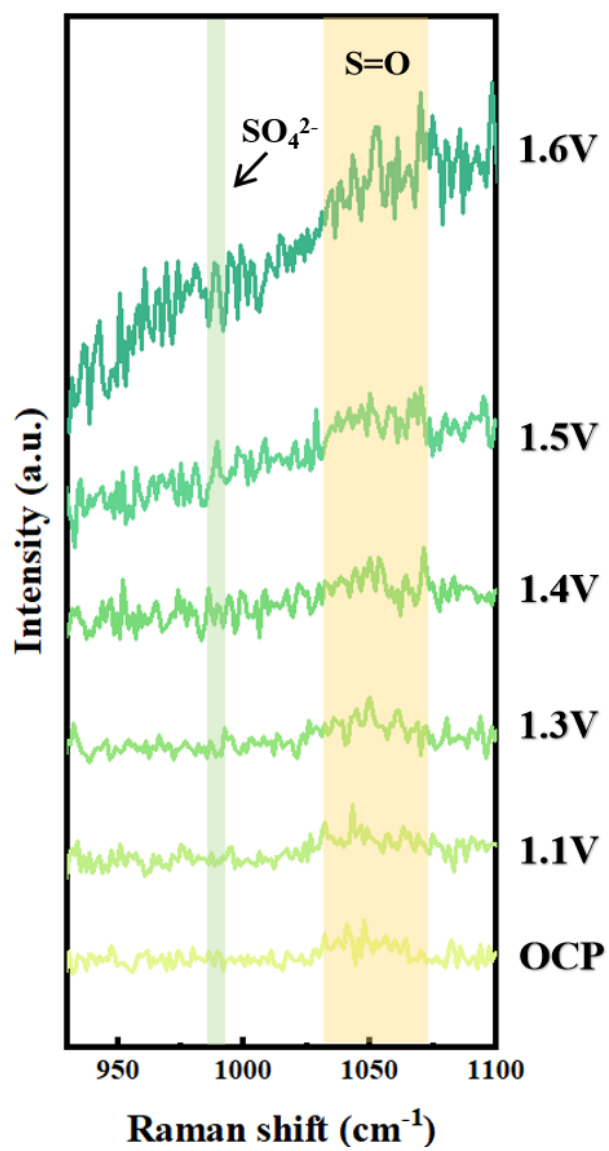
Fig. S22 LSV curves of  $\text{Co}_9\text{S}_8/\text{ESM-200}$ ,  $\text{Co}_8\text{FeS}_8/\text{ESM-200}$  and  $\text{FeS}_x/\text{ESM-200}$  recorded in 1 M KOH.



**Fig. S23** LSV curves of  $\text{Co}_8\text{FeS}_8/\text{ESM-800}$ ,  $\text{Co}_8\text{FeS}_8/\text{ESM-900}$  and  $\text{Co}_8\text{FeS}_8/\text{ESM-1000}$  recorded in 1 M KOH.



**Fig. S24** LSV curves of Co<sub>9</sub>S<sub>8</sub>/ESM-900, Co<sub>8</sub>FeS<sub>8</sub>/ESM-900, Ni<sub>9</sub>S<sub>8</sub>/ESM-900 and Ni<sub>8</sub>FeS<sub>8</sub>/ESM-900 recorded in 1 M KOH.



**Fig. S25** In situ Raman spectroscopy of Co<sub>8</sub>FeS<sub>8</sub>/ESM-900.

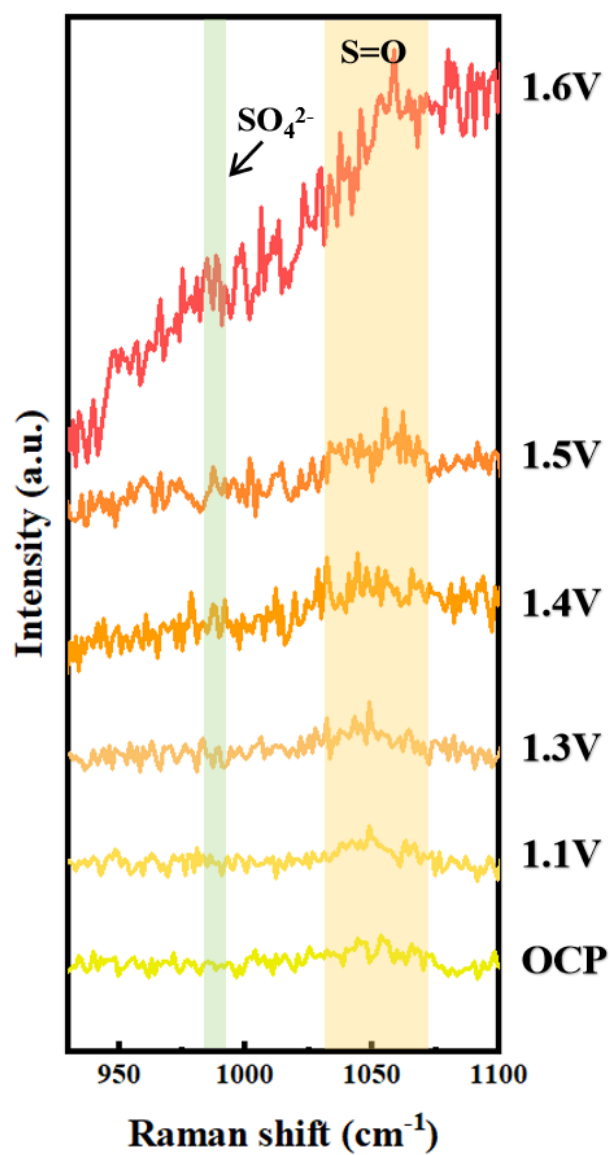


Fig. S26 In situ Raman spectroscopy of Co<sub>9</sub>S<sub>8</sub>/ESM-900.

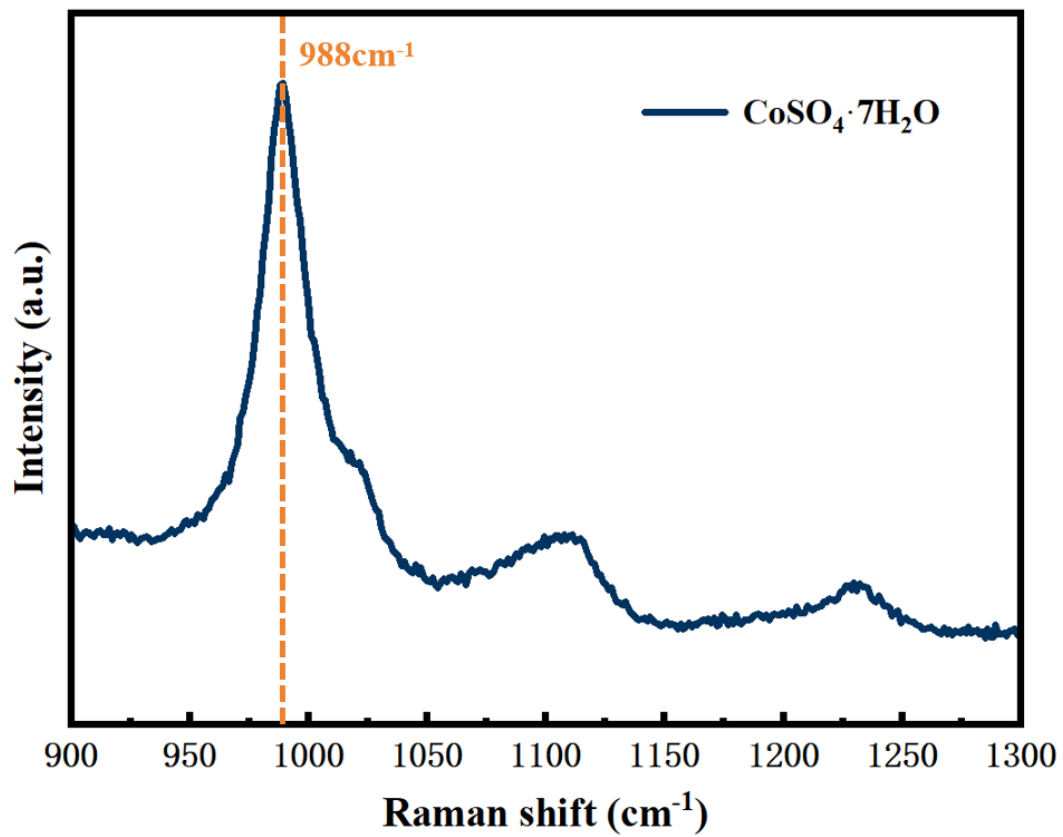
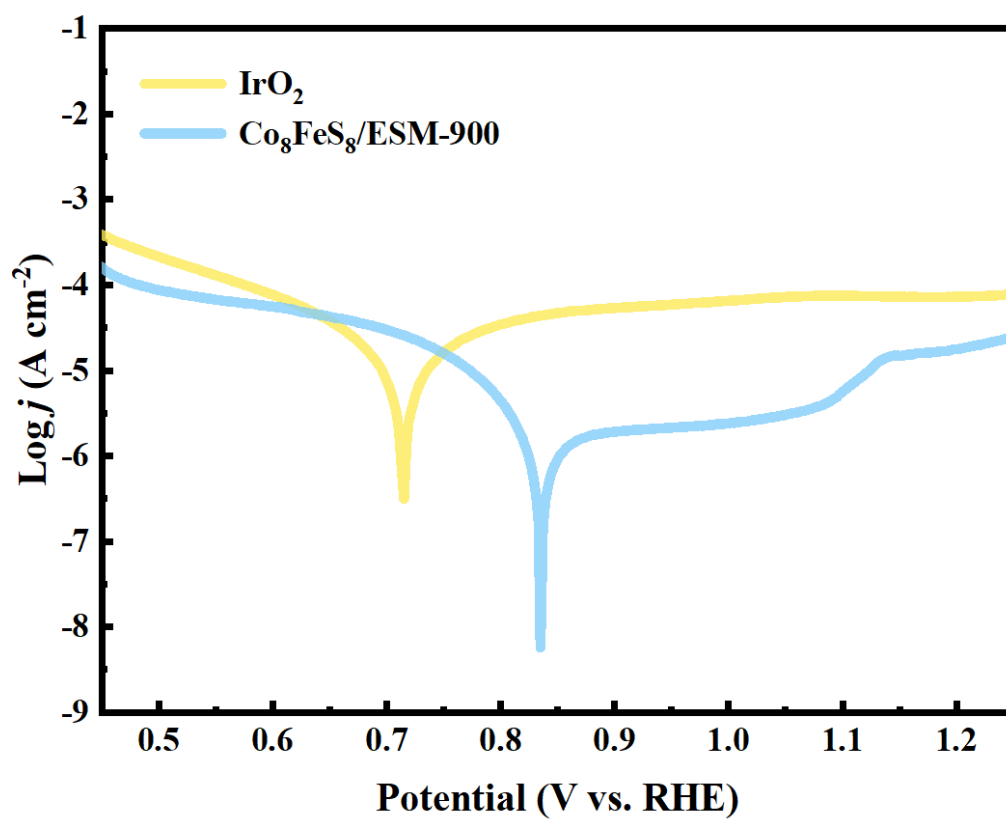


Fig. S27 Raman spectroscopy of  $\text{CoSO}_4 \cdot 7\text{H}_2\text{O}$ .

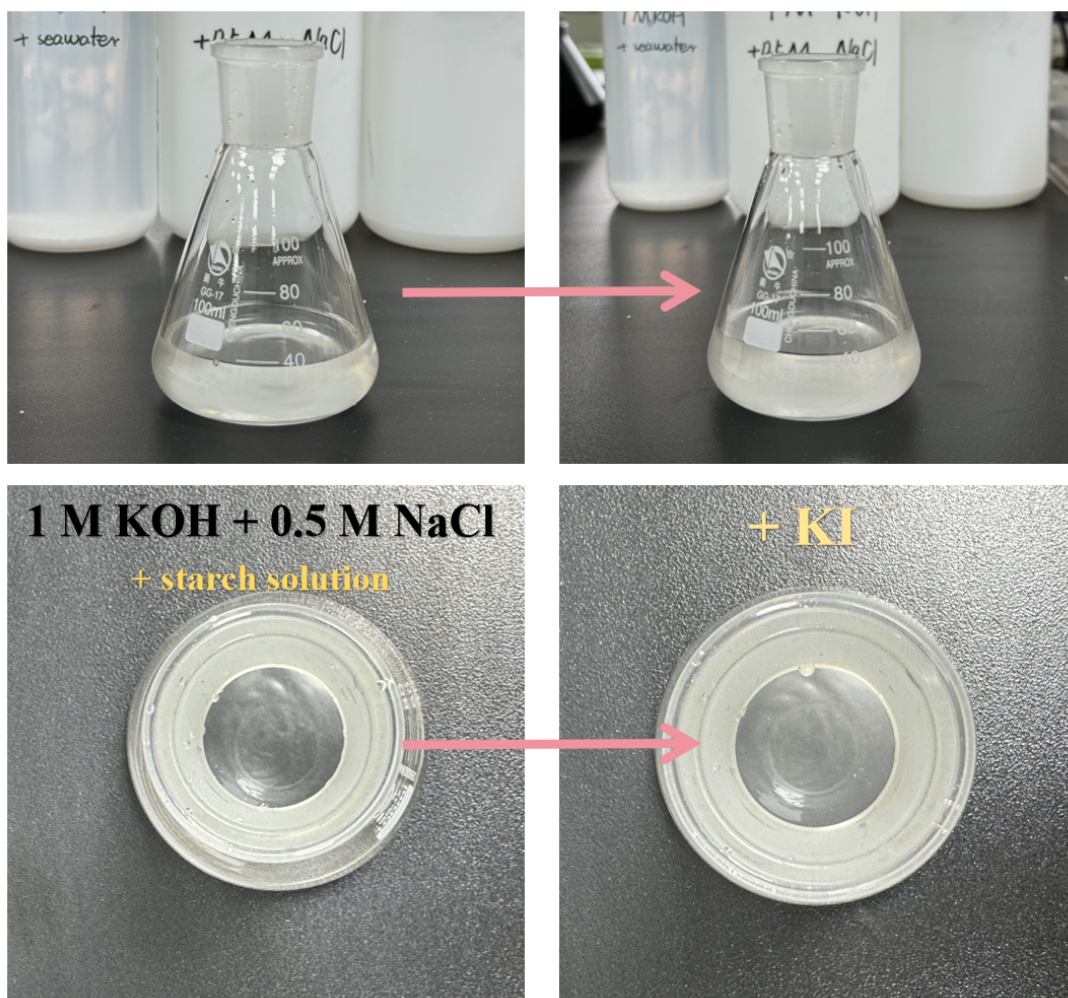


**Fig. S28** Schematic diagram of the apparatus for the Faraday efficiency test.

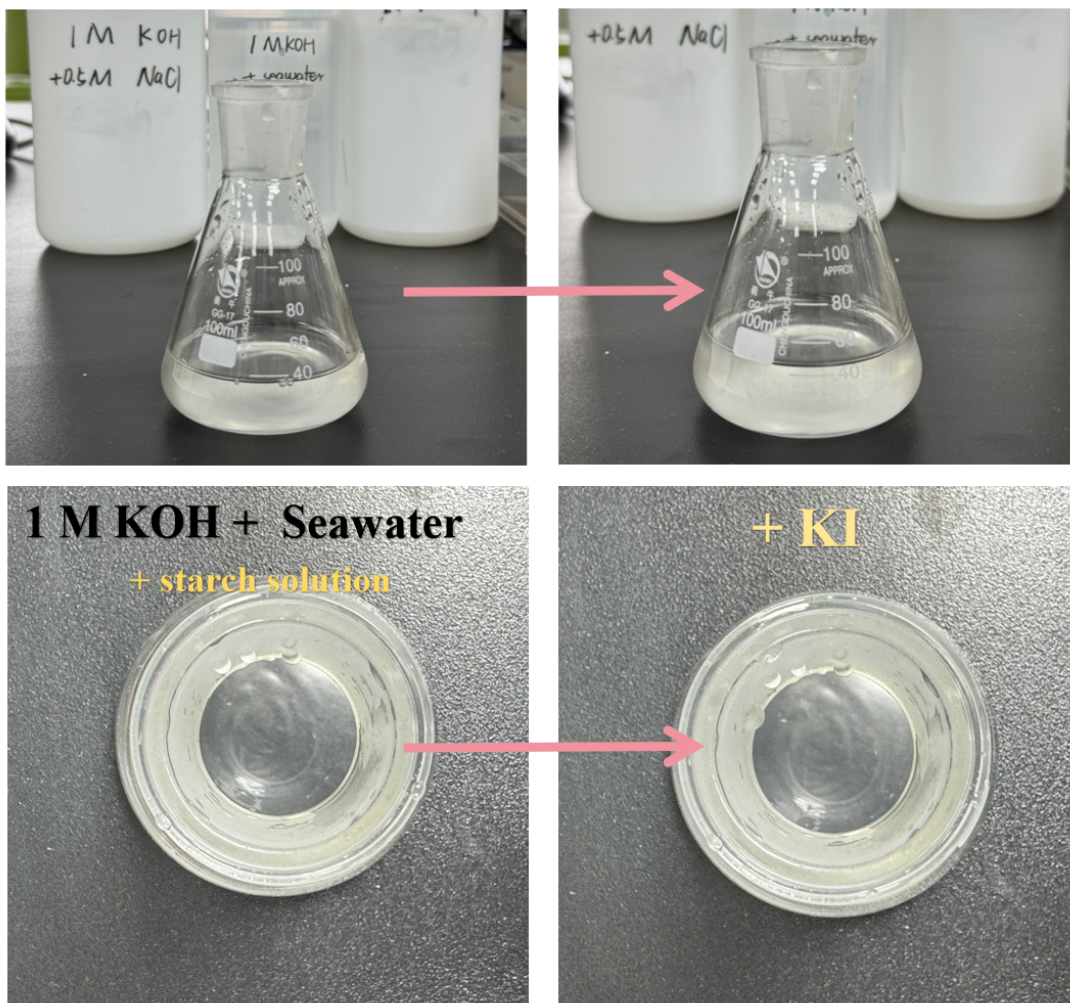


**Fig. S29** Corrosion polarization curves of Co<sub>8</sub>FeS<sub>8</sub>/ESM-900 and IrO<sub>2</sub> in 1 M KOH + seawater.





**Fig. S30** Digital photos of illustrating the determination of  $\text{ClO}^-$  in simulated seawater.



**Fig. S31** Digital photos of illustrating the determination of  $\text{ClO}^-$  in natural seawater.



**Fig. S32** Schematic diagram of illustrating the determination of ClO<sup>-</sup>.

## References

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