

Supporting Information

Radical Cascade Synthesis of γ -Amino Acids or γ -Lactams via Carboxyl-Mediated Intramolecular C–H Amination

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1 Supplementary Methods

1.1 General Considerations

Unless otherwise noted, all reactions for substrates preparation were conducted in a flame-dried glass tube under an Ar atmosphere using anhydrous solvents. Commercially available chemicals were obtained from Adamas-beta, Energy Chemical, Bidepharm, Leyan, TCI or J&K Scientific and used as received unless otherwise stated. Anhydrous ethyl acetate (EA) was purchased from Energy Chemical. Flash column chromatography was performed over silica gel (300-400 mesh). Organic solutions were concentrated under reduced pressure on an IKA MVP 10 rotary evaporating using a temperature-controlled water bath.

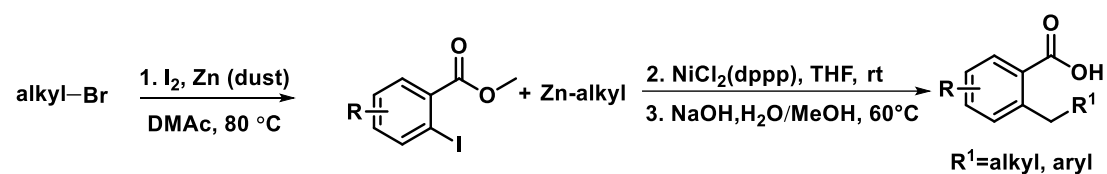
^1H , ^{19}F , and ^{13}C NMR spectra were recorded on a Bruker AVANCE 600 MHz or a Bruker AVANCE 400 MHz spectrometer. Chemical shifts (δ) for ^1H and ^{13}C NMR spectra are given in ppm relative to TMS. The residual solvent signals were used as references for ^1H and ^{13}C NMR spectra and the chemical shifts converted to the TMS scale (CDCl_3 referenced at 7.26 ppm and 77.16 ppm, respectively; methanol- d_4 referenced at 3.31 and 49.00 ppm, respectively; DMSO- d_6 referenced at 2.50 and 39.52 ppm, respectively; D_2O referenced at 4.79). ^1H and ^{19}F NMR are reported as follows: chemical shift (δ ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, p = pentet, h = hextet, hept = heptet, m = multiplet, b = broad), coupling constant (Hz), and integration. Data for ^{13}C NMR are reported in terms of chemical shift. High-resolution mass spectra (HRMS) were obtained from the Lanzhou University Analysis and Testing Center on Thermo Scientific Q Exactive Orbitrap LC-MS/MS System.

2 Supplementary Discussion

2.1 Preparation of substrates

Substrates s1-s100 were prepared according to the previously reported literature procedures.

General synthetic methods for primary alkyl-benzoic acids



Step 1. Preparation of the Organozinc Reagents

Following the general procedure with slight modification.¹ A 25 mL round-bottomed flask was charged with zinc powder (1.5 equiv.) and heated to 70 °C under high vacuum for 30 min. After back-filling with argon, I₂ (0.05 equiv.) and DMAc (1.5 M) were added, and the resulting heterogenous red mixture was allowed to stir until the red color of the iodine had faded. Then, the alkyl halide (1 equiv.) was added. The colorless reaction mixture was allowed to stir for 12 h at 70 °C, then the mixture was allowed to cool to room temperature. The gray solution was passed through dry celite and stored under argon.

Step 2. Negishi coupling of primary alkyl zinc reagents.

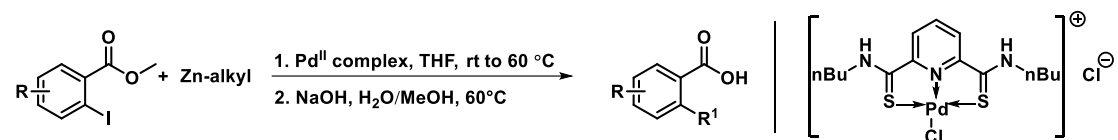
Following the general procedure with slight modification.^{2,3} A 100 mL Schlenk tube was charged with NiCl₂(dppp) (406 mg, 0.75 mmol) and a stir bar in the air. The tube was sealed with a septum and backfilled with Ar. A solution of the methyl 2-iodobenzoate (15 mmol) in dry THF (10 mL) was added via a syringe. After stirring for 3-4 min at rt, the corresponding alkylzinc bromide (0.5 M THF solution, 30 mmol) was added, and the reaction mixture was stirred at room temperature for the indicated time and the corresponding Negishi coupling products in 23-90% yield after 18 hours. Besides the unconverted starting materials, side products included the biaryl coupling adduct and the dehalogenated esters. The reaction mixture was then transferred with EtOAc (40 mL) to a separatory funnel containing water (40 mL), the product was extracted with EtOAc (3×30 mL) and the combined organic extracts were washed with water (60 mL) and brine (80 mL). After drying (anhydrous Na₂SO₄), the solution was filtered and concentrated, and the residue was purified by

silica gel column chromatography (petroleum ether: EtOAc = 20:1).

Step 3. General procedure for ester hydrolysis

Following the general procedure with slight modification.⁴ To a solution of methyl 2-alkyl benzoate (10 mmol) in MeOH (20 mL) was added 20 mL H₂O and sodium hydroxide (5 equiv.), and the resulting mixture was stirred at 60 °C for 4 hours before adjusted pH to 2 with 1M HCl. The reaction mixture was extracted with EtOAc (3 × 10 mL). The organic phase was combined and dried over anhydrous Na₂SO₄, filtered, and evaporated to dryness *in vacuo*, affording the compound as a solid. Further crystallization from EtOAc–petroleum ether gave pure 2-alkylbenzoic acid.

General synthetic methods for secondary alkyl-benzoic acids (S52-S63)

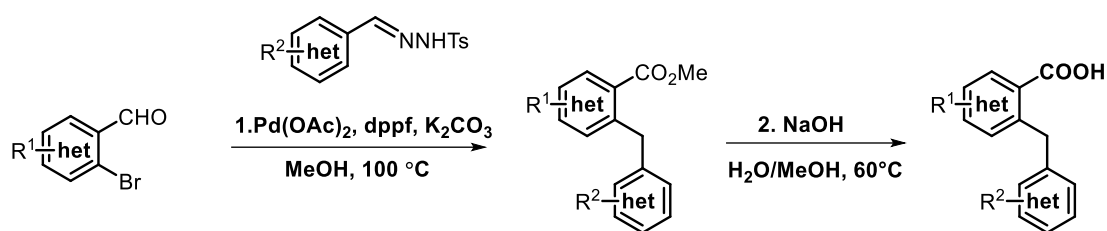


Step 1. Negishi coupling of secondary alkyl zinc reagents.

Following the general procedure with slight modification.⁵ Zinc chloride (24.0 mmol) was added to a Schlenk tube, which was sealed with a rubber septum and transferred out. After THF (12.0 mL) was injected into the tube, alkylmagnesium chloride (1.0 M, 24 mL) was added dropwise. Then the mixture was stirred for 1 h at r.t., and then pincer thioamide Pd^{II} complex (0.1 mol%) and 24 mmol methyl 2-iodobenzoate derivatives were added. The resultant mixture was stirred at 60 °C. After the reaction was completed, the mixture was cooled to 0 °C, quenched with saturated NH₄Cl and extracted with ethyl acetate three times. The combined organic phase was washed with brine, and dried over Na₂SO₄. Then, it was concentrated *in vacuo*, and the resulting crude product was purified by silica gel column chromatography (petroleum ether: EtOAc = 20:1).

Step 2. To a solution of methyl 2-alkylbenzoate (10 mmol) in MeOH (20 mL) was added 20 mL H₂O and NaOH (5 equiv.) and the resulting mixture stirred at 60 °C for 4 hours before adjusted pH to 2 with HCl (1N). The reaction mixture was extracted with ethyl acetate three times. The organic phase was combined and dried over anhydrous Na₂SO₄, filtered, and evaporated to dryness *in vacuo*, further crystallization from EtOAc/petroleum ether gave pure products, as white solids.

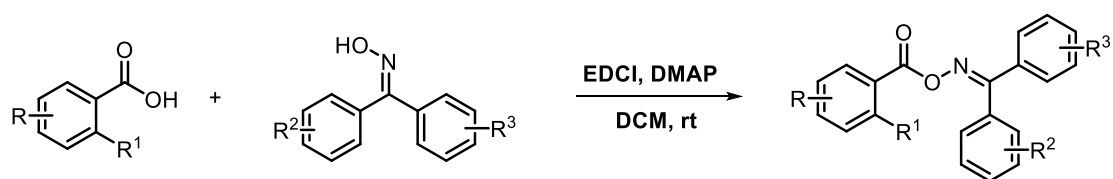
Synthesis of substrate S17, S18



Step 1. Following the general procedure with slight modification.⁶ An oven-dried Schlenk tube was charged with *N*-tosylhydrazone (15 mmol, 1.5 equiv.), K_2CO_3 (30 mmol, 3.0 equiv.), $Pd(OAc)_2$ (5 mol%), *dppf* (7.5 mol%). The tube was evacuated and backfilled with argon, and this procedure was repeated three times. To this mixture was added MeOH (100 mL) and 2-bromobenzaldehyde (10 mmol, 1.0 equiv.). The reaction was stirring at the indicated temperature. After the reaction was completed, the reaction mixture was quenched with water (40 mL) and extracted with ethyl acetate three times, dried over Na_2SO_4 . The combined filtrate was concentrated, and the residue was purified by silica gel column chromatography to give the pure products.

Step 2. To a solution of methyl 2-alkyl benzoates (10 mmol) in MeOH (20 mL) was added 20 mL H_2O and NaOH (5 equiv.), and the resulting mixture was stirred at 60 °C for 4 hours before adjusted pH to 2 with HCl (1N). The reaction mixture was extracted with ethyl acetate three times. The organic phase was combined and dried over anhydrous Na_2SO_4 , filtered, and evaporated to dryness *in vacuo*, further crystallization from EtOAc/petroleum ether gave pure products as white solids.

General synthetic methods for substrates S1-S63

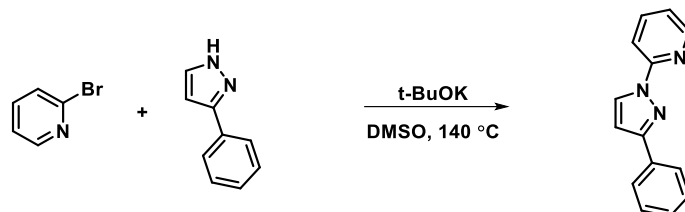


Oxime esters were prepared following the reported literature procedure.⁷ To a solution of oximes (10.0 mmol) and carboxylic acid (10.0 mmol) in CH_2Cl_2 (30 mL), DMAP (10 mol%) and EDCI (25 mmol) was added. The mixture was stirred at room temperature under an inert atmosphere until the reaction was complete, as observed from TLC monitoring. The mixture was diluted with distilled water (25 mL) and the CH_2Cl_2 layer was separated, dried over anhydrous Na_2SO_4 , and concentrated. The crude mass was treated with pentane (10 mL) and sonicated for 15 minutes. The resultant solid was filtered and dried under vacuum to obtain the pure oxime esters in most cases. In some cases,

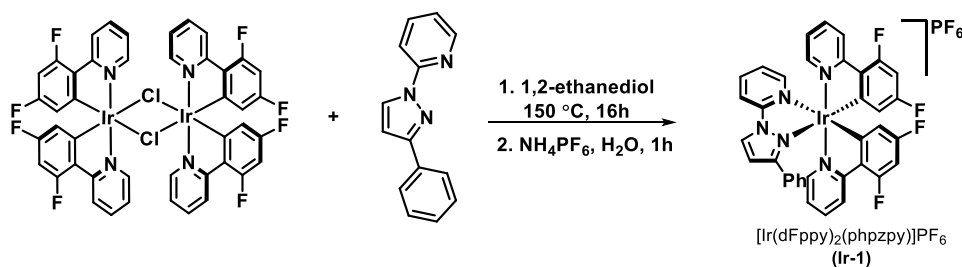
the final compound was purified by flash column chromatography using pentane/dichloromethane (DCM) or pentane/ethyl acetate (EtOAc) as eluent.

2.2 Synthesis of Ir-1, Ir-4

Synthesis of 2-(3-Phenyl-1*H*-pyrazol-1-yl)pyridine (phpzpy).

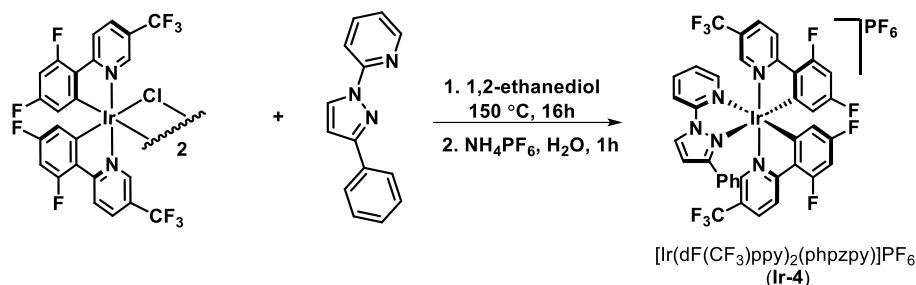


The ancillary ligand, phpzpy, was synthesized by using the reported procedure.^{8,9} 3-Phenyl-1*H*-pyrazole (15 mmol, 1.0 equiv.) and potassium *tert*-butoxide (17 mmol, 1.13 equiv.) were dissolved at room temperature in dry DMSO (15 mL). To the basic solution, 2-bromopyridine (16 mmol, 1.1 equiv.) was added slowly under constant stirring. The reaction mixture was refluxed at 140 °C for 12 h under nitrogen and then cooled to room temperature, after which it was extracted with water and ether. The organic layer was washed with water and ether to remove DMSO as well as excess potassium *tert*-butoxide. The organic layer was then isolated, dried over anhydrous Na₂SO₄, and filtered. The solvent was then removed under reduced pressure. The crude product was purified by column chromatography on silica gel with hexane/ethyl acetate (20:1) as the eluent, to yield colorless oil that crystallized into colorless solid crystals after a while.



A solution of [Ir(dFppy)₂Cl]₂ (2 mmol, 1.0 equiv.) and phpzpy (2.3 equiv.) in 1,2-ethanediol (150 mL) The reaction mixture was heated to reflux at 150 °C for 16 h under nitrogen atmosphere with constant stirring and then cooled to room temperature. A concentrated solution of NH₄PF₆ (60 equiv.) in deionized water (100 mL) was added slowly to the reaction mixture under stirring for 1 h, resulting in a yellow suspension. The suspension was filtered, and the resultant precipitate was

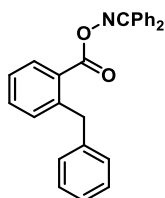
washed with plenty of deionized water. The yellow precipitate obtained was dried under vacuum at 100 °C for 12 h. The crude material was subsequently crystallized from dichloromethane/hexane, yielding a yellow powder.^{8,10}



The same method as **Ir-1**, $[\text{Ir}(\text{dF}(\text{CF}_3)\text{ppy})_2\text{Cl}]_2$ (1.12 g, 0.75 mmol) and 2-(3-phenyl-1*H*-pyrazol-1-yl)pyridine (0.35 g, 1.6 mmol) in 1,2-ethanediol (50 mL) under nitrogen. Subsequently it crystallized from dichloromethane/hexane, yielding a bright light-yellow powder.

2.3 Characterization data of substrates

diphenylmethanone *O*-(2-benzylbenzoyl) oxime (**S1**)



Chemical Formula: C₂₇H₂₁NO₂

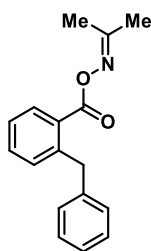
S1 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.70 – 7.61 (m, 2H), 7.54 (dd, *J* = 8.1, 1.5 Hz, 1H), 7.51 – 7.40 (m, 4H), 7.40 – 7.29 (m, 5H), 7.28 – 7.21 (m, 2H), 7.21 – 7.10 (m, 5H), 4.29 (s, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 165.6, 164.7, 143.1, 140.6, 134.6, 132.9, 132.2, 131.4, 131.0, 130.5, 129.6, 129.2, 129.1, 128.6, 128.4, 128.3, 128.3, 128.2, 126.1, 126.0, 39.2.

HRMS (ESI): *m/z* calculated for [C₂₇H₂₁NO₂Na]⁺ [*M* + Na]⁺: 414.1465, found: 414.1460.

propan-2-one *O*-(2-benzylbenzoyl) oxime (**S1-1**)



Chemical Formula: C₁₇H₁₇NO₂

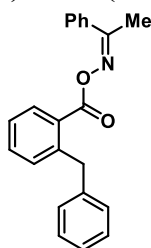
S1-1 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.85 (dd, *J* = 7.8, 1.5 Hz, 1H), 7.43 (td, *J* = 7.6, 1.5 Hz, 1H), 7.36 – 7.07 (m, 7H), 4.38 (s, 2H), 2.08 (s, 3H), 1.94 (s, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 165.1, 164.6, 142.3, 140.5, 132.0, 131.4, 130.1, 129.0, 128.9, 128.3, 126.1, 125.9, 39.2, 22.0, 17.1.

HRMS (ESI): *m/z* calculated for [C₁₇H₁₇NO₂Na]⁺ [M + Na]⁺: 290.1152, found: 290.1152.

1-phenylethan-1-one *O*-(2-benzylbenzoyl) oxime (S1-2)



Chemical Formula: C₂₂H₁₉NO₂

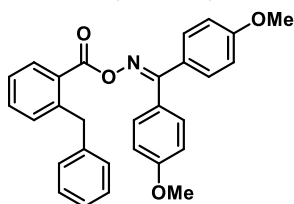
S1-2 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.93 (dd, *J* = 7.8, 1.5 Hz, 1H), 7.82 – 7.73 (m, 2H), 7.47 – 7.34 (m, 4H), 7.33 – 7.09 (m, 7H), 4.42 (s, 2H), 2.31 (s, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 164.8, 163.4, 142.5, 140.5, 134.7, 132.2, 131.5, 130.6, 130.2, 129.0, 128.8, 128.5, 128.3, 127.0, 126.2, 125.9, 39.2, 14.6.

HRMS (ESI): *m/z* calculated for [C₂₂H₁₉NO₂Na]⁺ [M + Na]⁺: 352.1308, found: 352.1308.

bis(4-methoxyphenyl)methanone *O*-(2-benzylbenzoyl) oxime (S1-3)



Chemical Formula: C₂₉H₂₅NO₄

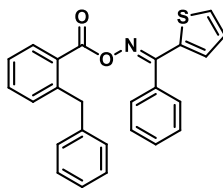
S1-3 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.68 – 7.54 (m, 3H), 7.36 (td, *J* = 7.6, 1.5 Hz, 1H), 7.31 – 7.20 (m, 4H), 7.20 – 7.07 (m, 5H), 6.97 – 6.83 (m, 4H), 4.29 (s, 2H), 3.82 (s, 6H).

¹³C NMR (101 MHz, CDCl₃) δ 165.2, 165.1, 161.9, 160.5, 143.0, 140.7, 132.2, 131.4, 130.9, 130.8, 130.5, 129.3, 128.6, 128.4, 127.5, 126.2, 126.0, 125.2, 113.8, 113.6, 55.4, 55.3, 39.2.

HRMS (ESI): *m/z* calculated for [C₂₉H₂₅NO₄Na]⁺ [M + Na]⁺: 474.1676, found: 474.1676.

phenyl(thiophen-2-yl)methanone *O*-(2-benzylbenzoyl) oxime (S1-4)



Chemical Formula: $C_{25}H_{19}NO_2S$

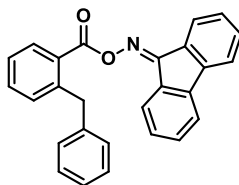
S1-4 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 8.07 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.69 – 7.59 (m, 2H), 7.54 – 7.31 (m, 6H), 7.31 – 7.20 (m, 4H), 7.20 – 6.96 (m, 4H), 4.35 (d, $J = 69.3$ Hz, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) major-**S1-4**: δ 164.7, 161.2, 158.1, 143.3, 140.6, 137.9, 135.4, 133.0, 132.4, 132.3, 131.5, 130.8, 130.2, 129.9, 129.3, 128.4, 128.4, 127.3, 126.3, 126.1, 39.3. minor-**S1-4**: δ 164.4, 143.1, 135.4, 132.7, 132.1, 131.5, 131.4, 130.5, 130.3, 129.8, 129.2, 128.6, 128.3, 128.1, 126.3, 126.1, 126.1, 126.0, 39.1.

HRMS (ESI): m/z calculated for $[C_{25}H_{19}NO_2SNa]^+$ $[M + Na]^+$: 420.1029, found: 420.1030.

9H-fluoren-9-one O-(2-benzylbenzoyl) oxime (S1-5)



Chemical Formula: $C_{27}H_{19}NO_2$

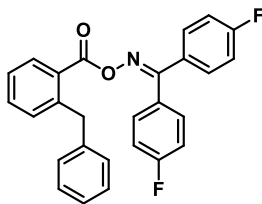
S1-5 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 8.10 (d, $J = 7.7$ Hz, 1H), 7.99 (dd, $J = 7.6, 1.8$ Hz, 2H), 7.58 (dd, $J = 12.8, 7.5$ Hz, 2H), 7.51 (td, $J = 7.6, 1.5$ Hz, 1H), 7.47 – 7.35 (m, 3H), 7.35 – 7.28 (m, 2H), 7.27 – 7.19 (m, 5H), 7.14 (h, $J = 4.3$ Hz, 1H), 4.45 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 164.9, 158.9, 143.2, 142.7, 141.2, 140.5, 134.5, 132.6, 132.5, 131.8, 131.7, 130.2, 130.1, 130.1, 129.2, 128.6, 128.5, 126.5, 126.2, 123.5, 120.4, 120.2, 39.2.

HRMS (ESI): m/z calculated for $[C_{27}H_{19}NO_2Na]^+$ $[M + Na]^+$: 412.1308, found: 412.1310.

bis(4-fluorophenyl)methanone O-(2-benzylbenzoyl) oxime (S1-6)



Chemical Formula: $C_{27}H_{19}F_2NO_2$

S1-6 was obtained as a white solid.

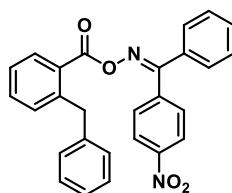
1H NMR (400 MHz, $CDCl_3$) δ 7.63 (dd, $J = 8.7, 5.5$ Hz, 2H), 7.59 – 7.51 (m, 1H), 7.44 – 7.35 (m, 1H), 7.35 – 7.22 (m, 4H), 7.22 – 7.03 (m, 9H), 4.31 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.2 (d, $J = 137.36$ Hz), 164.6, 163.6, 162.7 (d, $J = 135.34$ Hz), 143.2, 140.54, 132.5, 131.6, 131.2, 131.2, 131.0, 130.9, 130.7 (d, $J = 3.03$ Hz), 130.4, 129.2, 128.5 (d, $J = 3.03$ Hz), 128.4, 128.0, 126.2 (d, $J = 11.11$ Hz), 115.8 (d, $J = 3.03$ Hz), 115.6 (d, $J = 2.02$ Hz), 39.3.

^{19}F NMR (376 MHz, CDCl_3) δ -108.5, -109.8.

HRMS (ESI): m/z calculated for $[\text{C}_{27}\text{H}_{19}\text{F}_2\text{NO}_2\text{Na}]^+$ $[\text{M} + \text{Na}]^+$: 450.1276, found: 450.1274.

(4-nitrophenyl)(phenyl)methanone *O*-(2-benzylbenzoyl) oxime (S1-7)



Chemical Formula: $\text{C}_{27}\text{H}_{20}\text{N}_2\text{O}_4$

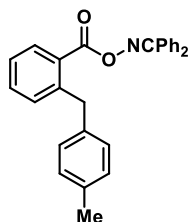
S1-7 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 8.32 – 8.19 (m, 2H), 7.88 – 7.80 (m, 1H), 7.64 – 7.57 (m, 2H), 7.54 – 7.44 (m, 4H), 7.44 – 7.36 (m, 2H), 7.30 – 7.05 (m, 7H), 4.29 (d, J = 15.4 Hz, 2H).

^{13}C NMR (101 MHz, CDCl_3) major-S1-7: δ 164.3, 163.5, 148.3, 143.0, 140.4, 139.3, 133.3, 132.6, 131.6, 130.2, 130.0, 129.6, 129.1, 128.8, 128.7, 128.4, 127.7, 126.3, 126.1, 123.7, 39.2. minor-S1-7: δ 164.2, 149.2, 143.4, 140.7, 131.7, 130.5, 130.3, 129.1, 128.7, 128.6, 127.5, 126.2, 126.0, 123.6, 39.3.

HRMS (ESI): m/z calculated for $[\text{C}_{27}\text{H}_{20}\text{N}_2\text{O}_4\text{Na}]^+$ $[\text{M} + \text{Na}]^+$: 459.1315, found: 459.1314.

diphenylmethanone *O*-(2-(4-methylbenzyl)benzoyl) oxime (S2)



Chemical Formula: $\text{C}_{28}\text{H}_{23}\text{NO}_2$

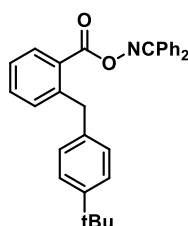
S2 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.72 – 7.65 (m, 2H), 7.56 (dd, J = 8.2, 1.5 Hz, 1H), 7.51 – 7.44 (m, 4H), 7.44 – 7.32 (m, 5H), 7.18 – 7.12 (m, 2H), 7.12 – 7.02 (m, 4H), 4.27 (s, 2H), 2.32 (s, 3H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.6, 164.8, 143.5, 137.6, 135.4, 134.7, 133.0, 132.2, 131.3, 131.0, 130.5, 129.6, 129.1, 129.1, 129.1, 128.7, 128.5, 128.5, 128.3, 128.2, 126.1, 38.8, 21.1.

HRMS (ESI): m/z calculated for $[\text{C}_{28}\text{H}_{23}\text{NO}_2\text{Na}]^+$ $[\text{M} + \text{Na}]^+$: 428.1621, found: 428.1621.

diphenylmethanone *O*-(2-(4-(tert-butyl)benzyl)benzoyl) oxime (S3)



Chemical Formula: $\text{C}_{31}\text{H}_{29}\text{NO}_2$

S3 was obtained as a white solid.

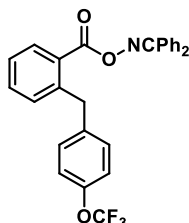
^1H NMR (400 MHz, CDCl_3) δ 7.68 – 7.61 (m, 2H), 7.52 (dd, J = 7.8, 1.5 Hz, 1H), 7.49 – 7.41 (m,

4H), 7.36 (dddd, $J = 14.3, 7.9, 3.5, 1.6$ Hz, 5H), 7.30 – 7.23 (m, 2H), 7.19 – 7.10 (m, 2H), 7.10 – 7.02 (m, 2H), 4.26 (s, 2H), 1.29 (s, 9H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.6, 164.8, 148.7, 143.4, 137.5, 134.7, 133.0, 132.2, 131.4, 131.0, 130.4, 129.6, 129.1, 128.8, 128.7, 128.4, 128.3, 128.2, 126.0, 125.3, 38.7, 34.4, 31.4.

HRMS (ESI): m/z calculated for $[\text{C}_{31}\text{H}_{30}\text{NO}_2]^+$ $[\text{M} + \text{H}]^+$: 448.2271, found: 448.2269.

diphenylmethanone *O*-(2-(4-(trifluoromethoxy)benzyl)benzoyl) oxime (S4)



Chemical Formula: $\text{C}_{28}\text{H}_{20}\text{F}_3\text{NO}_3$

S4 was obtained as a white solid.

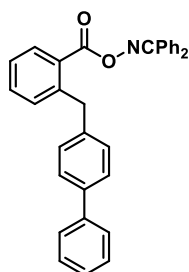
^1H NMR (400 MHz, CDCl_3) δ 7.72 – 7.61 (m, 2H), 7.56 (d, $J = 7.1$ Hz, 1H), 7.52 – 7.29 (m, 10H), 7.23 – 7.02 (m, 7H), 4.32 (s, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.8, 164.5, 147.5, 147.5, 142.5, 139.5, 134.6, 132.9, 132.5, 131.6, 131.1, 130.7, 130.3, 129.7, 129.1, 128.6, 128.5, 128.3, 128.1, 126.5, 120.8, 120.5 (q, $J = 257.55$ Hz), 38.6.

^{19}F NMR (376 MHz, CDCl_3) δ -57.8.

HRMS (ESI): m/z calculated for $[\text{C}_{28}\text{H}_{21}\text{F}_3\text{NO}_3]^+$ $[\text{M} + \text{H}]^+$: 476.1468, found: 476.1466.

diphenylmethanone *O*-(2-([1,1'-biphenyl]-4-ylmethyl)benzoyl) oxime (S5)



Chemical Formula: $\text{C}_{33}\text{H}_{25}\text{NO}_2$

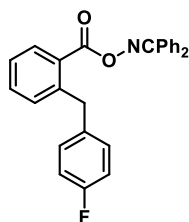
S5 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.70 – 7.62 (m, 2H), 7.60 – 7.51 (m, 3H), 7.51 – 7.26 (m, 14H), 7.26 – 7.10 (m, 4H), 4.34 (s, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.6, 164.7, 143.0, 141.0, 139.7, 138.8, 134.6, 132.9, 132.4, 131.5, 131.0, 130.5, 129.6, 129.5, 129.1, 128.7, 128.6, 128.4, 128.3, 128.1, 127.1, 127.0, 127.0, 126.2, 38.9.

HRMS (ESI): m/z calculated for $[\text{C}_{33}\text{H}_{26}\text{NO}_2]^+$ $[\text{M} + \text{H}]^+$: 468.1958, found: 468.1957.

diphenylmethanone *O*-(2-(4-fluorobenzyl)benzoyl) oxime (S6)



Chemical Formula: $C_{27}H_{20}FNO_2$

S6 was obtained as a white solid.

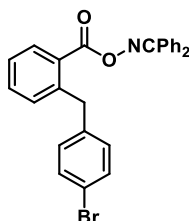
1H NMR (400 MHz, $CDCl_3$) δ 7.73 – 7.64 (m, 2H), 7.62 – 7.53 (m, 1H), 7.54 – 7.31 (m, 10H), 7.21 – 7.06 (m, 4H), 7.01 – 6.88 (m, 2H), 4.29 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.1 (d, $J = 109.08$ Hz), 161.3 (d, $J = 244.42$ Hz), 142.9, 136.2 (d, $J = 3.03$ Hz), 134.5, 132.8, 132.3, 131.3, 131.0, 130.5 (d, $J = 3.03$ Hz), 130.4, 129.6, 129.0, 128.6, 128.4, 128.3, 128.0, 126.3, 115.0 (d, $J = 21.21$ Hz), 38.4.

^{19}F NMR (376 MHz, $CDCl_3$) δ -117.5.

HRMS (ESI): m/z calculated for $[C_{27}H_{21}FNO_2]^+ [M + H]^+$: 410.1551, found: 410.1549.

diphenylmethanone *O*-(2-(4-bromobenzyl)benzoyl) oxime (S7)



Chemical Formula: $C_{27}H_{20}BrNO_2$

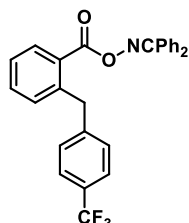
S7 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.69 – 7.61 (m, 2H), 7.55 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.51 – 7.43 (m, 4H), 7.36 (dddd, $J = 18.3, 7.9, 4.0, 2.1$ Hz, 7H), 7.21 – 7.12 (m, 2H), 7.07 – 6.97 (m, 2H), 4.25 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.8, 164.6, 142.5, 139.7, 134.6, 132.9, 132.5, 131.5, 131.4, 131.1, 130.9, 130.7, 129.7, 129.1, 128.7, 128.5, 128.4, 128.0, 126.5, 119.9, 38.7.

HRMS (ESI): m/z calculated for $[C_{27}H_{20}BrNO_2Na]^+ [M + Na]^+$: 492.0570, found: 492.0569.

diphenylmethanone *O*-(2-(4-(trifluoromethyl)benzyl)benzoyl) oxime (S8)



Chemical Formula: $C_{28}H_{20}F_3NO_2$

S8 was obtained as a white solid.

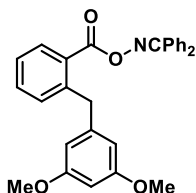
1H NMR (400 MHz, $CDCl_3$) δ 7.64 (dt, $J = 7.1, 1.4$ Hz, 2H), 7.56 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.52 – 7.27 (m, 11H), 7.24 (d, $J = 8.0$ Hz, 2H), 7.21 – 7.11 (m, 2H), 4.36 (s, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.9, 164.5, 144.9 (d, $J = 1.21$ Hz), 142.0, 134.6, 132.9, 132.6, 131.7, 131.1, 130.8, 129.7, 129.3, 129.1, 128.6, 128.5, 128.4, 128.1, 128.3 (q, $J = 32.32$ Hz), 126.7, 125.2 (q, $J = 3.03$ Hz), 124.4 (q, $J = 273.71$ Hz), 39.2.

^{19}F NMR (376 MHz, CDCl_3) δ -62.3.

HRMS (ESI): m/z calculated for $[\text{C}_{28}\text{H}_{21}\text{F}_3\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 460.1519, found: 460.1517.

diphenylmethanone *O*-(2-(3,5-dimethoxybenzyl)benzoyl) oxime (S9)



Chemical Formula: $\text{C}_{29}\text{H}_{25}\text{NO}_4$

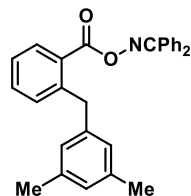
S9 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.65 (dt, $J = 8.6, 1.5$ Hz, 2H), 7.52 (dt, $J = 7.9, 1.9$ Hz, 1H), 7.49 – 7.27 (m, 10H), 7.12 (ddd, $J = 13.0, 6.8, 2.9$ Hz, 2H), 6.30 (q, $J = 2.1$ Hz, 3H), 4.22 (s, 2H), 3.70 (s, 6H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.7, 164.9, 160.8, 143.0, 142.7, 134.7, 133.0, 132.3, 131.4, 131.1, 130.4, 129.7, 129.4, 129.3, 129.1, 129.0, 128.7, 128.5, 128.4, 128.3, 128.2, 127.9, 126.2, 107.4, 98.2, 55.3, 39.4.

HRMS (ESI): m/z calculated for $[\text{C}_{29}\text{H}_{26}\text{NO}_4]^+ [\text{M} + \text{H}]^+$: 452.1856, found: 452.1856.

diphenylmethanone *O*-(2-(3,5-dimethylbenzyl)benzoyl) oxime (S10)



Chemical Formula: $\text{C}_{29}\text{H}_{26}\text{NO}_2$

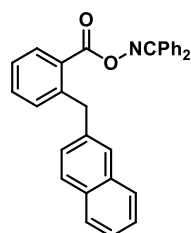
S10 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.77 – 7.62 (m, 2H), 7.58 – 7.51 (m, 1H), 7.51 – 7.31 (m, 9H), 7.15 (dt, $J = 7.6, 3.6$ Hz, 2H), 6.84 (s, 1H), 6.78 (s, 2H), 4.23 (s, 2H), 2.26 (s, 6H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.5, 164.8, 143.4, 140.4, 137.8, 134.7, 132.9, 132.2, 131.4, 131.0, 130.4, 129.6, 129.1, 128.7, 128.4, 128.3, 128.1, 127.7, 127.1, 126.0, 38.9, 21.3.

HRMS (ESI): m/z calculated for $[\text{C}_{29}\text{H}_{26}\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 420.1958, found: 420.1960.

diphenylmethanone *O*-(2-(naphthalen-2-ylmethyl)benzoyl) oxime (S11)



Chemical Formula: $\text{C}_{31}\text{H}_{23}\text{NO}_2$

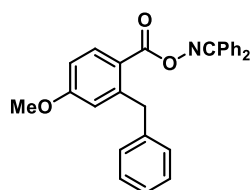
S11 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.82 – 7.76 (m, 1H), 7.72 (d, *J* = 7.9 Hz, 2H), 7.68 – 7.60 (m, 2H), 7.60 – 7.52 (m, 2H), 7.51 – 7.33 (m, 11H), 7.33 – 7.21 (m, 4H), 7.16 (t, *J* = 7.6 Hz, 2H), 4.44 (s, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 165.7, 164.9, 143.0, 138.2, 134.6, 133.6, 133.0, 132.3, 132.1, 131.5, 131.0, 130.6, 129.6, 129.2, 129.1, 128.8, 128.6, 128.5, 128.4, 128.3, 128.0, 127.9, 127.7, 127.6, 127.4, 126.2, 125.89, 125.3, 39.4.

HRMS (ESI): *m/z* calculated for [C₃₁H₂₃NO₂Na]⁺ [*M* + Na]⁺: 464.1621, found: 464.1618.

diphenylmethanone *O*-(2-benzyl-4-methoxybenzoyl) oxime (S12)



Chemical Formula: C₂₈H₂₃NO₃

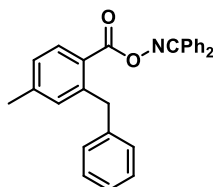
S12 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.73 – 7.60 (m, 2H), 7.54 (d, *J* = 9.5 Hz, 1H), 7.48 – 7.41 (m, 4H), 7.40 – 7.31 (m, 4H), 7.25 (ddd, *J* = 8.1, 6.4, 1.1 Hz, 2H), 7.16 (td, *J* = 7.4, 1.6 Hz, 3H), 6.69 – 6.57 (m, 2H), 4.32 (s, 2H), 3.72 (s, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 165.1, 164.1, 162.6, 146.3, 140.5, 134.8, 133.2, 133.0, 130.9, 129.5, 129.2, 129.1, 128.7, 128.4, 128.4, 128.3, 126.0, 120.0, 117.2, 111.1, 55.3, 39.5.

HRMS (ESI): *m/z* calculated for [C₂₈H₂₃NO₃Na]⁺ [*M* + Na]⁺: 482.1338, found: 482.1335.

diphenylmethanone *O*-(2-benzyl-4-methylbenzoyl) oxime (S13)



Chemical Formula: C₂₈H₂₃NO₂

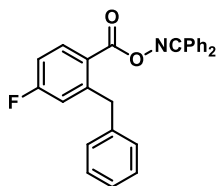
S13 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.69 – 7.61 (m, 2H), 7.48 – 7.39 (m, 5H), 7.39 – 7.28 (m, 4H), 7.28 – 7.19 (m, 2H), 7.15 (td, *J* = 7.2, 1.6 Hz, 3H), 6.99 – 6.90 (m, 2H), 4.28 (s, 2H), 2.25 (s, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 165.4, 164.7, 143.4, 143.0, 140.9, 134.8, 133.1, 132.3, 131.0, 130.8, 129.6, 129.2, 129.1, 128.7, 128.4, 128.3, 127.0, 126.0, 125.2, 39.2, 21.6.

HRMS (ESI): *m/z* calculated for [C₂₈H₂₄NO₂]⁺ [*M* + H]⁺: 406.1802, found: 406.1779.

diphenylmethanone *O*-(2-benzyl-4-fluorobenzoyl) oxime (S14)



Chemical Formula: $C_{27}H_{20}FNO_2$

S14 was obtained as a white solid.

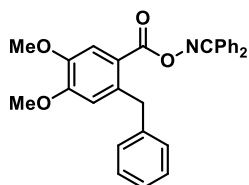
1H NMR (400 MHz, $CDCl_3$) δ 7.69 – 7.62 (m, 2H), 7.56 (dd, $J = 8.6, 5.9$ Hz, 1H), 7.46 (qd, $J = 4.5, 1.5$ Hz, 4H), 7.42 – 7.31 (m, 5H), 7.30 – 7.23 (m, 2H), 7.23 – 7.17 (m, 1H), 7.14 (dd, $J = 6.9, 1.8$ Hz, 2H), 6.85 – 6.75 (m, 2H), 4.29 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) major-**S14**: δ 165.8, 163.8, 147.2(d, $J = 8.08$ Hz), 139.7, 134.5, 133.1(d, $J = 9.09$ Hz), 133.0, 131.2, 129.7, 129.3, 129.1, 128.6, 128.6, 128.5, 128.4, 126.4, 124.1, 118.2 (d, $J = 22.22$ Hz), 113.3 (d, $J = 22.22$ Hz), 39.3. minor-**S14**: 166.2, 163.7, 131.2, 129.1, 128.5, 128.5, 124.0, 39.0.

^{19}F NMR (376 MHz, $CDCl_3$) δ -106.0.

HRMS (ESI): m/z calculated for $[C_{27}H_{20}FNO_2Na]^+$ [$M + Na$] $^+$: 432.1370, found: 432.1368.

diphenylmethanone *O*-(2-benzyl-4,5-dimethoxybenzoyl) oxime (**S15**)



Chemical Formula: $C_{29}H_{25}NO_4$

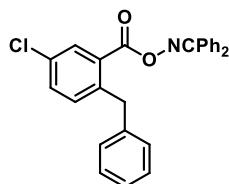
S15 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.71 – 7.61 (m, 2H), 7.53 – 7.41 (m, 4H), 7.41 – 7.33 (m, 4H), 7.30 – 7.20 (m, 2H), 7.20 – 7.12 (m, 3H), 6.99 (s, 1H), 6.62 (s, 1H), 4.37 (s, 2H), 3.78 (s, 3H), 3.58 (s, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 164.9, 163.4, 152.2, 146.6, 141.0, 138.5, 134.5, 133.6, 131.0, 129.3, 129.0, 128.9, 128.5, 128.4, 128.4, 128.3, 125.9, 123.8, 118.9, 114.1, 112.8, 55.8, 55.8, 39.0.

HRMS (ESI): m/z calculated for $[C_{29}H_{26}NO_4]^+$ [$M + H$] $^+$: 474.1676, found: 474.1676.

diphenylmethanone *O*-(2-benzyl-5-chlorobenzoyl) oxime (**S16**)



Chemical Formula: $C_{27}H_{20}ClNO_2$

S16 was obtained as a white solid.

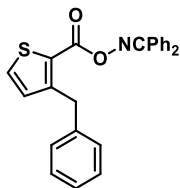
1H NMR (400 MHz, $CDCl_3$) δ 7.68 – 7.64 (m, 2H), 7.50 – 7.44 (m, 5H), 7.41 – 7.35 (m, 2H), 7.35 – 7.29 (m, 3H), 7.28 – 7.21 (m, 2H), 7.21 – 7.14 (m, 1H), 7.13 – 7.08 (m, 2H), 7.06 (d, $J = 8.3$ Hz, 1H), 4.25 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 166.2, 163.5, 141.7, 140.1, 134.4, 132.8, 132.2, 132.0, 131.2,

130.5, 129.8, 129.6, 129.12, 129.1, 128.6, 128.5, 128.5, 128.5, 126.2, 38.6.

HRMS (ESI): m/z calculated for $[C_{27}H_{21}ClNO_2]^+$ $[M + H]^+$: 426.1255, found: 426.1256.

diphenylmethanone *O*-(3-benzylthiophene-2-carbonyl) oxime (S17)



Chemical Formula: $C_{25}H_{19}NO_2S$

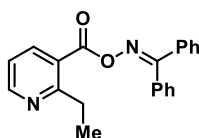
S17 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.71 – 7.62 (m, 2H), 7.51 – 7.42 (m, 4H), 7.42 – 7.33 (m, 4H), 7.30 (d, $J = 5.1$ Hz, 1H), 7.28 – 7.22 (m, 2H), 7.22 – 7.14 (m, 3H), 6.79 (d, $J = 5.1$ Hz, 1H), 4.27 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.2, 160.0, 150.6, 140.0, 134.6, 132.8, 131.2, 131.0, 129.6, 129.1, 129.0, 128.9, 128.5, 128.5, 128.3, 126.2, 124.4, 35.3.

HRMS (ESI): m/z calculated for $[C_{25}H_{20}NO_2S]^+$ $[M + H]^+$: 398.1209, found: 398.1206.

diphenylmethanone *O*-(2-ethylnicotinoyl) oxime (S18)



Chemical Formula: $C_{21}H_{18}N_2O_2$

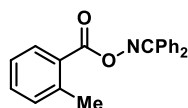
S18 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 8.58 (dd, $J = 4.8, 1.8$ Hz, 1H), 7.75 (dd, $J = 7.9, 1.8$ Hz, 1H), 7.69 – 7.60 (m, 2H), 7.51 – 7.41 (m, 4H), 7.41 – 7.30 (m, 4H), 7.06 (dd, $J = 7.9, 4.8$ Hz, 1H), 3.03 (q, $J = 7.5$ Hz, 2H), 1.19 (t, $J = 7.5$ Hz, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 166.0, 164.8, 163.8, 152.0, 138.1, 134.2, 132.7, 131.2, 129.6, 128.9, 128.4, 128.4, 128.3, 123.7, 120.6, 30.0, 13.7.

HRMS (ESI): m/z calculated for $[C_{21}H_{19}N_2O_2]^+$ $[M + H]^+$: 331.1441, found: 331.1438.

diphenylmethanone *O*-(2-methylbenzoyl) oxime (S19)



Chemical Formula: $C_{21}H_{17}NO_2$

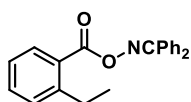
S19 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.77 – 7.63 (m, 2H), 7.57 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.53 – 7.30 (m, 9H), 7.25 – 7.06 (m, 2H), 2.49 (s, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.4, 164.7, 140.8, 134.6, 133.1, 132.2, 131.7, 131.0, 130.4, 129.5, 129.0, 128.6, 128.4, 128.3, 128.0, 125.6, 21.5.

HRMS (ESI): m/z calculated for $[C_{21}H_{18}NO_2]^+$ $[M + H]^+$: 316.1332, found: 316.1329.

diphenylmethanone O-(2-ethylbenzoyl) oxime (S20)



Chemical Formula: C₂₂H₁₉NO₂

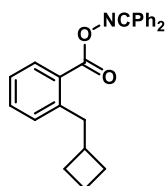
S20 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.73 – 7.63 (m, 2H), 7.55 – 7.42 (m, 5H), 7.42 – 7.31 (m, 5H), 7.23 (d, *J* = 7.9 Hz, 1H), 7.17 – 7.04 (m, 1H), 2.89 (q, *J* = 7.5 Hz, 2H), 1.16 (t, *J* = 7.5 Hz, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 165.5, 164.8, 146.8, 134.7, 133.1, 132.3, 131.0, 130.4, 130.3, 129.6, 129.1, 128.6, 128.4, 128.4, 127.7, 125.7, 27.4, 15.8.

HRMS (ESI): *m/z* calculated for [C₂₂H₂₀NO₂]⁺ [M + H]⁺: 330.1489, found: 330.1490.

diphenylmethanone O-(2-(cyclobutylmethyl)benzoyl) oxime (S21)



Chemical Formula: C₂₅H₂₃NO₂

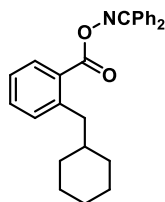
S21 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.80 – 7.64 (m, 2H), 7.57 – 7.31 (m, 10H), 7.25 – 7.07 (m, 2H), 3.02 (d, *J* = 7.5 Hz, 2H), 2.61 (p, *J* = 7.8 Hz, 1H), 1.97 (tt, *J* = 10.8, 4.9 Hz, 2H), 1.87 – 1.60 (m, 4H).

¹³C NMR (101 MHz, CDCl₃) δ 165.3, 164.8, 143.3, 134.6, 132.8, 131.8, 130.9, 130.7, 130.2, 129.5, 128.9, 128.5, 128.3, 128.2, 127.8, 125.6, 40.5, 36.8, 28.1, 18.2.

HRMS (ESI): *m/z* calculated for [C₂₅H₂₄NO₂]⁺ [M + H]⁺: 370.1802, found: 370.1800.

diphenylmethanone O-(2-(cyclohexylmethyl)benzoyl) oxime (S22)



Chemical Formula: C₂₇H₂₇NO₂

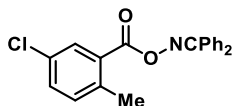
S22 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.75 – 7.66 (m, 2H), 7.56 – 7.45 (m, 5H), 7.45 – 7.34 (m, 5H), 7.18 (dd, *J* = 7.7, 1.3 Hz, 1H), 7.11 (td, *J* = 7.6, 1.3 Hz, 1H), 2.83 (d, *J* = 7.0 Hz, 2H), 1.72 – 1.53 (m, 6H), 1.24 – 1.09 (m, 3H), 0.95 (ddt, *J* = 17.8, 9.6, 4.9 Hz, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 165.4, 164.7, 144.0, 134.7, 132.9, 132.0, 131.6, 130.9, 130.3, 129.6, 129.1, 128.7, 128.4, 128.3, 128.1, 125.6, 41.7, 39.4, 33.1, 26.5, 26.3.

HRMS (ESI): *m/z* calculated for [C₂₇H₂₈NO₂]⁺ [M + H]⁺: 398.2115, found: 398.2115.

diphenylmethanone O-(5-chloro-2-methylbenzoyl) oxime (S23)



Chemical Formula: $C_{21}H_{16}ClNO_2$

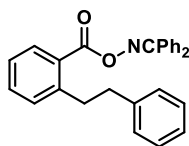
S23 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.72 – 7.65 (m, 2H), 7.56 – 7.43 (m, 5H), 7.43 – 7.34 (m, 4H), 7.31 (dd, J = 8.2, 2.4 Hz, 1H), 7.12 (d, J = 8.3 Hz, 1H), 2.43 (s, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 166.0, 163.5, 139.3, 134.3, 133.1, 132.9, 132.2, 131.4, 131.2, 130.4, 129.8, 129.3, 129.1, 128.5, 128.5, 21.0.

HRMS (ESI): m/z calculated for $[C_{21}H_{16}ClNO_2Na]^+ [M + Na]^+$: 372.0762, found: 372.0761.

diphenylmethanone *O*-(2-phenethylbenzoyl) oxime (S24)



Chemical Formula: $C_{28}H_{23}NO_2$

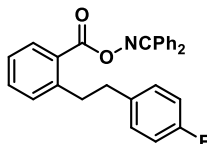
S24 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.74 – 7.62 (m, 2H), 7.57 – 7.31 (m, 11H), 7.31 – 7.09 (m, 8H), 3.27 – 3.14 (m, 2H), 2.96 – 2.80 (m, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.6, 164.4, 144.6, 141.9, 134.7, 132.9, 132.3, 131.4, 131.0, 130.6, 129.6, 129.1, 128.7, 128.4, 128.3, 128.2, 127.6, 126.0, 125.8, 38.0, 36.8.

HRMS (ESI): m/z calculated for $[C_{28}H_{23}NO_2Na]^+ [M + Na]^+$: 428.1621, found: 428.1617.

diphenylmethanone *O*-(2-(4-fluorophenethyl)benzoyl) oxime (S25)



Chemical Formula: $C_{28}H_{22}FNO_2$

S25 was obtained as a white solid.

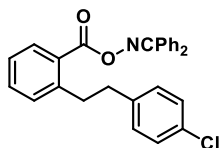
1H NMR (400 MHz, $CDCl_3$) δ 7.71 – 7.67 (m, 2H), 7.53 – 7.42 (m, 5H), 7.42 – 7.34 (m, 5H), 7.20 – 7.10 (m, 4H), 6.97 – 6.89 (m, 2H), 3.24 – 3.13 (m, 2H), 2.95 – 2.79 (m, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) major-**S24**: δ 165.7, 164.4, 144.4, 137.5 (d, J = 2.02 Hz), 134.7, 133.0, 132.4, 131.4, 131.1, 130.6, 130.1, 129.7, 129.1, 128.7, 128.5, 128.4, 127.6, 126.1, 114.9 (d, J = 14.14 Hz), 37.1, 36.9. minor-**S24**: 162.1, 160.5, 132.2, 130.0, 129.2, 128.9, 128.6, 128.3, 128.3, 125.9.

^{19}F NMR (376 MHz, $CDCl_3$) δ -117.8.

HRMS (ESI): m/z calculated for $[C_{28}H_{23}FNO_2]^+ [M + H]^+$: 452.1856, found: 452.1856.

diphenylmethanone *O*-(2-(4-chlorophenethyl)benzoyl) oxime (S26)



Chemical Formula: $C_{28}H_{22}ClNO_2$

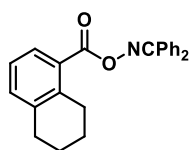
S26 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.74 – 7.67 (m, 2H), 7.56 – 7.34 (m, 11H), 7.28 – 7.20 (m, 2H), 7.20 – 7.09 (m, 4H), 3.25 – 3.15 (m, 2H), 2.94 – 2.80 (m, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.7, 164.3, 144.3, 140.3, 134.6, 132.9, 132.4, 131.5, 131.4, 131.1, 130.6, 130.1, 129.6, 129.1, 128.7, 128.5, 128.3, 128.3, 127.5, 126.2, 37.3, 36.7.

HRMS (ESI): m/z calculated for $[C_{28}H_{23}ClNO_2]^+ [M + H]^+$: 440.1412, found: 440.1413.

diphenylmethanone *O*-(5,6,7,8-tetrahydronaphthalene-1-carbonyl) oxime (**S27**)



Chemical Formula: $C_{24}H_{21}NO_2$

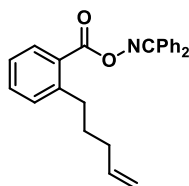
S27 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.70 (dt, $J = 7.1, 1.4$ Hz, 2H), 7.55 – 7.44 (m, 4H), 7.44 – 7.33 (m, 5H), 7.24 – 7.12 (m, 1H), 7.04 (t, $J = 7.7$ Hz, 1H), 3.03 – 2.87 (m, 2H), 2.87 – 2.71 (m, 2H), 1.73 (p, $J = 3.4$ Hz, 4H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.2, 165.1, 139.1, 138.4, 134.6, 133.4, 133.0, 130.9, 129.4, 128.9, 128.5, 128.4, 128.3, 128.2, 127.8, 124.8, 30.1, 27.5, 22.9, 22.2.

HRMS (ESI): m/z calculated for $[C_{24}H_{22}NO_2]^+ [M + H]^+$: 356.1645, found: 356.1644.

diphenylmethanone *O*-(2-(pent-4-en-1-yl)benzoyl) oxime (**S28**)



Chemical Formula: $C_{25}H_{23}NO_2$

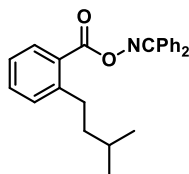
S28 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.68 (dt, $J = 8.7, 1.7$ Hz, 2H), 7.53 – 7.29 (m, 11H), 7.28 – 7.18 (m, 1H), 7.10 (tt, $J = 7.6, 1.6$ Hz, 1H), 5.82 (ddtd, $J = 16.9, 10.2, 6.6, 1.7$ Hz, 1H), 5.10 – 4.88 (m, 2H), 2.92 (td, $J = 7.7, 1.6$ Hz, 2H), 2.19 – 2.01 (m, 2H), 1.76 – 1.59 (m, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.5, 164.6, 145.3, 138.7, 134.7, 133.0, 132.2, 131.1, 131.0, 130.5, 129.6, 129.1, 128.7, 128.5, 128.3, 127.8, 125.8, 114.6, 33.9, 33.8, 30.8.

HRMS (ESI): m/z calculated for $[C_{25}H_{24}NO_2]^+ [M + H]^+$: 370.1802, found: 370.1801.

diphenylmethanone *O*-(2-isopentylbenzoyl) oxime (**S29**)



Chemical Formula: $C_{25}H_{25}NO_2$

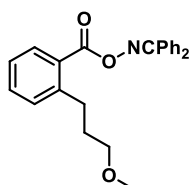
S29 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.68 (dt, $J = 8.6, 1.5$ Hz, 2H), 7.53 – 7.41 (m, 5H), 7.42 – 7.31 (m, 6H), 7.26 – 7.19 (m, 1H), 7.14 – 7.03 (m, 1H), 2.97 – 2.86 (m, 2H), 1.66 – 1.58 (m, 1H), 1.52 – 1.41 (m, 2H), 0.93 (d, $J = 6.6, 1.5$ Hz, 6H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.4, 164.5, 146.0, 134.7, 132.9, 132.1, 130.9, 130.9, 130.3, 129.6, 129.1, 128.7, 128.4, 128.3, 125.5, 40.9, 32.3, 28.2, 22.5, 22.5.

HRMS (ESI): m/z calculated for $[C_{25}H_{26}NO_2]^+ [M + H]^+$: 372.1958, found: 372.1957.

diphenylmethanone *O*-(2-(3-methoxypropyl)benzoyl) oxime (S30)



Chemical Formula: $C_{24}H_{23}NO_3$

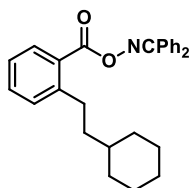
S30 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.67 (dt, $J = 7.2, 1.4$ Hz, 2H), 7.53 – 7.42 (m, 5H), 7.42 – 7.33 (m, 5H), 7.25 (dd, $J = 7.8, 1.3$ Hz, 1H), 7.11 (td, $J = 7.6, 1.3$ Hz, 1H), 3.38 (t, $J = 6.4$ Hz, 2H), 3.32 (s, 3H), 3.04 – 2.91 (m, 2H), 1.92 – 1.80 (m, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.5, 164.5, 144.9, 134.7, 132.9, 132.3, 131.3, 131.0, 130.5, 129.7, 129.1, 128.7, 128.5, 128.5, 128.4, 127.7, 125.9, 72.2, 58.5, 31.3, 30.9.

HRMS (ESI): m/z calculated for $[C_{24}H_{24}NO_3]^+ [M + H]^+$: 374.1751, found: 374.1751.

diphenylmethanone *O*-(2-(2-cyclohexylethyl)benzoyl) oxime (S31)



Chemical Formula: $C_{28}H_{29}NO_2$

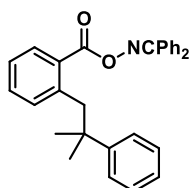
S31 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.75 – 7.63 (m, 2H), 7.54 – 7.44 (m, 5H), 7.44 – 7.32 (m, 5H), 7.23 (dd, $J = 7.7, 1.3$ Hz, 1H), 7.10 (td, $J = 7.6, 1.3$ Hz, 1H), 3.00 – 2.87 (m, 2H), 1.84 – 1.59 (m, 5H), 1.55 – 1.43 (m, 2H), 1.39 – 1.08 (m, 4H), 0.95 (qd, $J = 11.8, 3.2$ Hz, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.5, 164.6, 146.2, 134.8, 133.0, 132.2, 131.0, 131.0, 130.4, 129.6, 129.1, 128.8, 128.4, 128.3, 127.7, 125.6, 39.6, 37.9, 33.3, 31.8, 26.8, 26.4.

HRMS (ESI): m/z calculated for $[C_{28}H_{30}NO_2]^+ [M + H]^+$: 412.2271, found: 412.2271.

diphenylmethanone *O*-(2-(2-methyl-2-phenylpropyl)benzoyl) oxime (S32)



Chemical Formula: C₃₀H₂₇NO₂

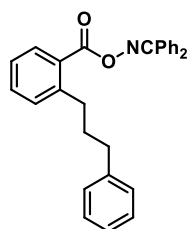
S32 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.73 – 7.63 (m, 2H), 7.50 – 7.31 (m, 9H), 7.29 – 7.19 (m, 4H), 7.19 – 7.10 (m, 2H), 7.05 (td, *J* = 7.6, 1.4 Hz, 1H), 6.68 (dd, *J* = 7.7, 1.4 Hz, 1H), 3.32 (s, 2H), 1.26 (s, 6H).

¹³C NMR (101 MHz, CDCl₃) δ 165.6, 165.4, 149.1, 140.5, 134.8, 133.0, 132.7, 131.0, 130.8, 130.1, 129.8, 129.6, 129.2, 128.8, 128.5, 128.3, 128.0, 126.3, 125.9, 125.7, 46.0, 39.6, 27.9.

HRMS (ESI): *m/z* calculated for [C₃₀H₂₈NO₂]⁺ [*M* + *H*]⁺: 434.2115, found: 434.2114.

diphenylmethanone *O*-(2-(3-phenylpropyl)benzoyl) oxime (S33)



Chemical Formula: C₂₉H₂₅NO₂

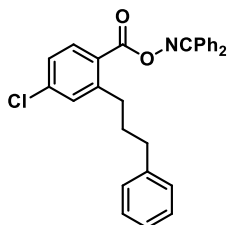
S33 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.72 – 7.64 (m, 2H), 7.50 – 7.41 (m, 5H), 7.36 (dddd, *J* = 9.2, 7.4, 5.8, 1.6 Hz, 5H), 7.28 – 7.21 (m, 2H), 7.21 – 7.07 (m, 5H), 2.99 – 2.92 (m, 2H), 2.70 – 2.61 (m, 2H), 1.96 – 1.86 (m, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 165.6, 164.7, 145.1, 142.4, 134.7, 133.0, 132.2, 131.1, 130.5, 129.6, 129.1, 128.7, 128.5, 128.4, 128.3, 127.8, 125.9, 125.7, 35.9, 34.1, 33.3.

HRMS (ESI): *m/z* calculated for [C₂₉H₂₅NO₂Na]⁺ [*M* + Na]⁺: 442.1778, found: 442.1777.

diphenylmethanone *O*-(4-chloro-2-(3-phenylpropyl)benzoyl) oxime (S34)



Chemical Formula: C₂₉H₂₄ClNO₂

S34 was obtained as a white solid.

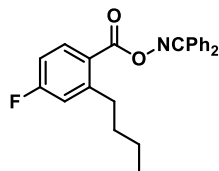
¹H NMR (400 MHz, CDCl₃) δ 7.73 – 7.65 (m, 2H), 7.52 – 7.43 (m, 4H), 7.43 – 7.34 (m, 5H), 7.31 (dd, *J* = 8.3, 2.4 Hz, 1H), 7.25 (dd, *J* = 8.2, 6.9 Hz, 2H), 7.14 (dd, *J* = 18.3, 7.8 Hz, 4H), 2.96 – 2.88 (m, 2H), 2.68 – 2.60 (m, 2H), 1.93 – 1.82 (m, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 166.1, 163.3, 143.5, 142.1, 134.4, 132.7, 132.3, 132.1, 131.5,

131.2, 130.4, 129.8, 129.1, 129.1, 128.5, 128.5, 128.4, 128.3, 125.7, 35.8, 33.4, 33.1.

HRMS (ESI): m/z calculated for $[C_{29}H_{24}ClNO_2Na]^+$ $[M + Na]^+$: 476.1388, found: 476.1382.

diphenylmethanone *O*-(2-butyl-4-fluorobenzoyl) oxime (S35)



Chemical Formula: $C_{24}H_{22}FNO_2$

S35 was obtained as a white solid.

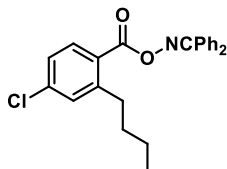
1H NMR (400 MHz, $CDCl_3$) δ 7.75 – 7.60 (m, 2H), 7.54 – 7.43 (m, 5H), 7.43 – 7.34 (m, 4H), 6.93 (dd, $J = 9.8, 2.6$ Hz, 1H), 6.78 (ddd, $J = 8.8, 7.9, 2.7$ Hz, 1H), 2.99 – 2.84 (m, 2H), 1.64 – 1.47 (m, 2H), 1.36 (h, $J = 7.3$ Hz, 2H), 0.91 (t, $J = 7.3$ Hz, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 164.7 (d, $J = 254.52$ Hz), 164.5 (d, $J = 191.9$ Hz), 163.6, 149.5 (d, $J = 9.09$ Hz), 134.6, 133.0, 132.9, 131.0, 129.6, 129.0, 128.6, 128.4, 128.3, 123.7 (d, $J = 3.03$ Hz), 117.6 (d, $J = 21.21$ Hz), 112.7 (d, $J = 21.21$ Hz), 34.1, 33.4, 22.7, 13.9.

^{19}F NMR (376 MHz, $CDCl_3$) δ -106.9.

HRMS (ESI): m/z calculated for $[C_{24}H_{23}FNO_2]^+$ $[M + H]^+$: 376.1707, found: 376.1706.

diphenylmethanone *O*-(2-butyl-4-chlorobenzoyl) oxime (S36)



Chemical Formula: $C_{24}H_{22}ClNO_2$

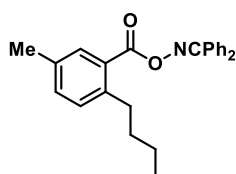
S36 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.73 – 7.63 (m, 2H), 7.53 – 7.43 (m, 4H), 7.43 – 7.32 (m, 5H), 7.22 (d, $J = 2.2$ Hz, 1H), 7.08 (dd, $J = 8.4, 2.2$ Hz, 1H), 2.98 – 2.78 (m, 2H), 1.62 – 1.47 (m, 2H), 1.36 (dq, $J = 14.6, 7.3$ Hz, 2H), 0.91 (t, $J = 7.3$ Hz, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.5, 164.7, 142.7, 135.1, 134.7, 133.1, 133.0, 131.1, 131.0, 131.0, 129.5, 129.1, 128.7, 128.4, 128.3, 127.4, 34.0, 33.6, 22.8, 20.7, 14.0.

HRMS (ESI): m/z calculated for $[C_{24}H_{22}ClNO_2Na]^+$ $[M + Na]^+$: 414.1231, found: 414.1230.

diphenylmethanone *O*-(2-butyl-5-methylbenzoyl) oxime (S37)



Chemical Formula: $C_{25}H_{25}NO_2$

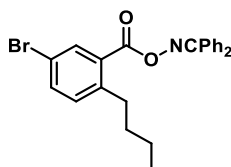
S37 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.76 – 7.67 (m, 2H), 7.55 – 7.44 (m, 4H), 7.44 – 7.35 (m, 4H), 7.28 (d, *J* = 2.0 Hz, 1H), 7.21 – 7.16 (m, 1H), 7.12 (d, *J* = 7.8 Hz, 1H), 2.93 – 2.82 (m, 2H), 2.21 (s, 3H), 1.60 – 1.47 (m, 2H), 1.35 (dq, *J* = 14.6, 7.3 Hz, 2H), 0.91 (t, *J* = 7.3 Hz, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 165.5, 164.7, 142.7, 135.1, 134.7, 133.1, 132.9, 131.1, 131.0, 131.0, 129.5, 129.1, 128.7, 128.4, 128.3, 127.4, 33.9, 33.6, 22.8, 20.7, 14.0.

HRMS (ESI): *m/z* calculated for [C₂₅H₂₆NO₂]⁺ [M + H]⁺: 372.1958, found: 372.1957.

diphenylmethanone *O*-(5-bromo-2-butylbenzoyl) oxime (S38)



Chemical Formula: C₂₄H₂₂BrNO₂

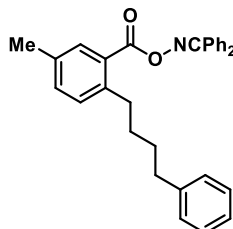
S38 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.75 – 7.65 (m, 2H), 7.56 (d, *J* = 2.3 Hz, 1H), 7.55 – 7.44 (m, 5H), 7.44 – 7.34 (m, 4H), 7.09 (d, *J* = 8.2 Hz, 1H), 2.94 – 2.77 (m, 2H), 1.52 (tt, *J* = 7.8, 6.4 Hz, 2H), 1.34 (dq, *J* = 14.6, 7.3 Hz, 2H), 0.90 (t, *J* = 7.3 Hz, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 166.1, 163.2, 144.6, 135.0, 134.4, 133.3, 132.8, 132.7, 131.2, 129.8, 129.4, 129.1, 128.6, 128.5, 128.5, 119.1, 33.7, 33.5, 22.7, 14.0.

HRMS (ESI): *m/z* calculated for [C₂₄H₂₃BrNO₂]⁺ [M + H]⁺: 436.0907, found: 436.0906.

diphenylmethanone *O*-(5-methyl-2-(4-phenylbutyl)benzoyl) oxime (S39)



Chemical Formula: C₃₁H₂₉NO₂

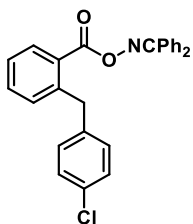
S39 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.73 – 7.66 (m, 2H), 7.53 – 7.43 (m, 4H), 7.42 – 7.35 (m, 4H), 7.25 (ddd, *J* = 8.9, 5.5, 3.3 Hz, 3H), 7.20 – 7.11 (m, 4H), 7.08 (d, *J* = 7.8 Hz, 1H), 2.88 (t, *J* = 7.3 Hz, 2H), 2.60 (t, *J* = 7.3 Hz, 2H), 2.19 (s, 3H), 1.62 (tdd, *J* = 15.9, 6.6, 2.7 Hz, 4H).

¹³C NMR (101 MHz, CDCl₃) δ 165.5, 164.6, 142.8, 142.4, 135.2, 134.6, 133.1, 133.0, 131.1, 131.0, 131.0, 129.5, 129.1, 128.7, 128.4, 128.3, 128.2, 127.4, 125.6, 35.9, 33.7, 31.5, 31.5, 20.7.

HRMS (ESI): *m/z* calculated for [C₃₁H₃₀NO₂]⁺ [M + H]⁺: 448.2271, found: 448.2271.

diphenylmethanone *O*-(2-(4-chlorobenzyl)benzoyl) oxime (S42)



Chemical Formula: $C_{27}H_{20}ClNO_2$

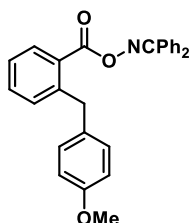
S42 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.69 – 7.63 (m, 2H), 7.55 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.51 – 7.43 (m, 4H), 7.43 – 7.36 (m, 3H), 7.36 – 7.30 (m, 2H), 7.24 – 7.12 (m, 4H), 7.10 – 7.04 (m, 2H), 4.27 (s, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.7, 164.6, 142.6, 139.1, 134.5, 132.9, 132.4, 131.7, 131.5, 131.1, 130.6, 130.5, 129.6, 129.1, 128.6, 128.4, 128.4, 128.3, 128.1, 126.4, 38.6.

HRMS (ESI): m/z calculated for $[C_{27}H_{20}ClNO_2Na]^+$ $[M + Na]^+$: 448.1075, found: 448.1072.

diphenylmethanone O-(2-(4-methoxybenzyl)benzoyl) oxime (S45)



Chemical Formula: $C_{28}H_{23}NO_3$

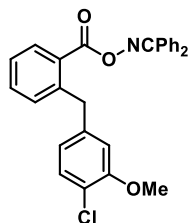
S45 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.85 (d, $J = 8.4$ Hz, 2H), 7.71 – 7.63 (m, 1H), 7.62 – 7.51 (m, 4H), 7.51 – 7.32 (m, 6H), 7.15 (ddd, $J = 7.1, 3.8, 2.6$ Hz, 2H), 7.11 – 7.04 (m, 2H), 6.85 – 6.77 (m, 2H), 4.23 (s, 2H), 3.77 (s, 3H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 178.8, 165.6, 164.8, 143.6, 134.7, 133.0, 132.3, 131.3, 131.0, 130.5, 129.6, 129.1, 128.7, 128.5, 126.1, 55.3, 38.3.

HRMS (ESI): m/z calculated for $[C_{28}H_{24}NO_3]^+$ $[M + H]^+$: 422.1751, found: 422.1754.

diphenylmethanone O-(2-(4-chloro-3-methoxybenzyl)benzoyl) oxime (S46)



Chemical Formula: $C_{28}H_{22}ClNO_3$

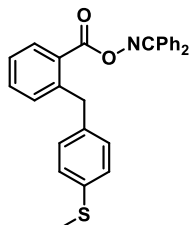
S46 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.71 – 7.61 (m, 2H), 7.59 – 7.52 (m, 1H), 7.52 – 7.29 (m, 8H), 7.24 – 7.12 (m, 3H), 7.03 (s, 1H), 6.77 (d, $J = 2.0$ Hz, 1H), 6.70 – 6.62 (m, 1H), 4.26 (s, 2H), 3.82 (s, 3H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.8, 164.7, 154.8, 142.5, 140.8, 134.6, 132.9, 131.3, 130.5, 129.9, 129.9, 129.1, 128.6, 128.5, 126.4, 121.9, 113.2, 56.1, 39.1.

HRMS (ESI): m/z calculated for $[\text{C}_{28}\text{H}_{22}\text{ClNO}_3\text{Na}]^+ [\text{M} + \text{Na}]^+$: 478.1180, found: 478.1191.

diphenylmethanone *O*-(2-(4-(methylthio)benzyl)benzoyl) oxime (S47)



Chemical Formula: $\text{C}_{28}\text{H}_{23}\text{NO}_2\text{S}$

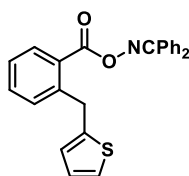
S47 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.70 – 7.64 (m, 2H), 7.55 (dd, $J = 8.2, 1.5$ Hz, 1H), 7.50 – 7.43 (m, 4H), 7.43 – 7.31 (m, 5H), 7.16 (td, $J = 6.9, 1.7$ Hz, 4H), 7.12 – 7.05 (m, 2H), 4.26 (s, 2H), 2.45 (s, 3H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.6, 164.7, 143.0, 137.7, 135.5, 134.6, 132.9, 132.3, 131.4, 131.0, 130.5, 129.7, 129.6, 129.0, 128.6, 128.4, 128.3, 128.1, 126.9, 126.2, 38.7, 16.1.

HRMS (ESI): m/z calculated for $[\text{C}_{28}\text{H}_{24}\text{NO}_2\text{S}]^+ [\text{M} + \text{H}]^+$: 460.1342, found: 460.1340.

diphenylmethanone *O*-(2-(thiophen-2-ylmethyl)benzoyl) oxime (S48)



Chemical Formula: $\text{C}_{25}\text{H}_{19}\text{NO}_2\text{S}$

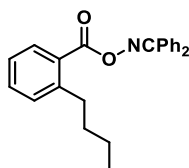
S48 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.70 – 7.62 (m, 2H), 7.54 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.49 – 7.42 (m, 4H), 7.42 – 7.30 (m, 5H), 7.27 – 7.21 (m, 1H), 7.15 (td, $J = 7.6, 1.3$ Hz, 1H), 7.10 (dd, $J = 5.1, 1.2$ Hz, 1H), 6.88 (dd, $J = 5.2, 3.5$ Hz, 1H), 6.78 (dq, $J = 3.3, 1.1$ Hz, 1H), 4.46 (s, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.6, 164.5, 143.2, 142.7, 134.6, 132.9, 132.5, 131.0, 131.0, 130.6, 129.6, 129.1, 128.6, 128.4, 128.3, 127.6, 126.7, 126.5, 125.6, 123.9, 33.6.

HRMS (ESI): m/z calculated for $[\text{C}_{25}\text{H}_{20}\text{NO}_2\text{S}]^+ [\text{M} + \text{H}]^+$: 398.1209, found: 398.1207.

diphenylmethanone *O*-(2-butylbenzoyl) oxime (S50)



Chemical Formula: $\text{C}_{24}\text{H}_{23}\text{NO}_2$

S50 was obtained as a white solid.

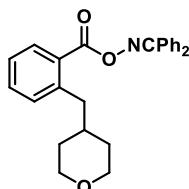
^1H NMR (400 MHz, CDCl_3) δ 7.68 (dd, $J = 8.3, 1.5$ Hz, 2H), 7.54 – 7.27 (m, 10H), 7.22 (t, $J = 7.1$

Hz, 1H), 7.09 (td, $J = 7.6, 1.4$ Hz, 1H), 2.96 – 2.83 (m, 2H), 1.61 – 1.48 (m, 2H), 1.35 (q, $J = 7.5$ Hz, 2H), 0.90 (t, $J = 7.3$ Hz, 3H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.4, 164.6, 145.6, 134.7, 132.9, 132.1, 131.0, 130.9, 130.3, 129.5, 129.0, 128.7, 128.4, 128.3, 127.7, 125.6, 34.0, 33.8, 22.7, 14.0.

HRMS (ESI): m/z calculated for $[\text{C}_{24}\text{H}_{24}\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 358.1802, found: 358.1801.

diphenylmethanone *O*-(2-((tetrahydro-2H-pyran-4-yl)methyl)benzoyl) oxime (S51)



Chemical Formula: $\text{C}_{26}\text{H}_{25}\text{NO}_3$

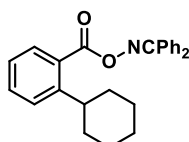
S51 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.73 – 7.61 (m, 2H), 7.55 – 7.43 (m, 5H), 7.37 (ddt, $J = 10.9, 5.1, 1.8$ Hz, 5H), 7.22 – 7.03 (m, 2H), 3.89 (ddd, $J = 11.4, 4.5, 1.8$ Hz, 2H), 3.28 (td, $J = 11.8, 2.1$ Hz, 2H), 2.88 (d, $J = 7.0$ Hz, 2H), 1.86 (ttt, $J = 11.0, 7.1, 3.8$ Hz, 1H), 1.52 (ddd, $J = 12.9, 4.0, 2.0$ Hz, 2H), 1.44 – 1.20 (m, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.5, 164.3, 142.9, 134.6, 132.8, 132.1, 131.8, 131.0, 130.5, 129.6, 129.0, 128.6, 128.4, 128.2, 127.7, 126.0, 67.9, 41.5, 36.5, 33.0.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{26}\text{NO}_3]^+ [\text{M} + \text{H}]^+$: 400.1907, found: 400.1906.

diphenylmethanone *O*-(2-cyclohexylbenzoyl) oxime (S52)



Chemical Formula: $\text{C}_{26}\text{H}_{25}\text{NO}_2$

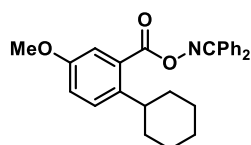
S46 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.76 – 7.63 (m, 2H), 7.55 – 7.29 (m, 11H), 7.10 (td, $J = 7.4, 1.6$ Hz, 1H), 3.34 (tt, $J = 11.3, 2.8$ Hz, 1H), 1.90 – 1.67 (m, 5H), 1.49 – 1.33 (m, 4H), 1.25 (tt, $J = 13.0, 2.8$ Hz, 1H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.5, 165.4, 149.3, 134.7, 132.7, 131.9, 130.9, 129.7, 129.5, 129.0, 128.6, 128.4, 128.2, 128.2, 126.9, 125.3, 40.0, 34.3, 26.8, 26.2.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{26}\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 384.1958, found: 384.1956.

diphenylmethanone *O*-(2-cyclohexyl-5-methoxybenzoyl) oxime (S53)



Chemical Formula: $\text{C}_{27}\text{H}_{27}\text{NO}_3$

S53 was obtained as a white solid.

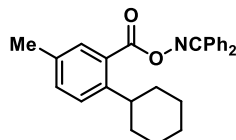
^1H NMR (400 MHz, CDCl_3) δ 7.75 – 7.62 (m, 2H), 7.53 – 7.31 (m, 8H), 7.26 (d, $J = 8.6$ Hz, 1H),

6.96 (dd, $J = 8.7, 2.9$ Hz, 1H), 6.88 (d, $J = 2.9$ Hz, 1H), 3.60 (s, 3H), 3.30 (tt, $J = 8.6, 3.3$ Hz, 1H), 1.83 – 1.65 (m, 5H), 1.48 – 1.09 (m, 5H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.5, 164.9, 156.8, 142.0, 134.6, 133.0, 131.0, 129.4, 129.0, 128.6, 128.5, 128.4, 128.3, 128.1, 119.1, 113.5, 55.2, 39.3, 34.5, 26.9, 26.3.

HRMS (ESI): m/z calculated for $[\text{C}_{27}\text{H}_{28}\text{NO}_3]^+ [\text{M} + \text{H}]^+$: 414.2064, found: 414.2062.

diphenylmethanone *O*-(2-cyclohexyl-5-methylbenzoyl) oxime (S54)



Chemical Formula: $\text{C}_{27}\text{H}_{27}\text{NO}_2$

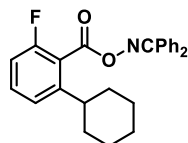
S54 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.76 – 7.61 (m, 2H), 7.45 (ddd, $J = 6.4, 2.9, 1.6$ Hz, 4H), 7.42 – 7.33 (m, 4H), 7.27 – 7.12 (m, 3H), 3.28 (tt, $J = 11.1, 2.8$ Hz, 1H), 2.20 (s, 3H), 1.83 – 1.62 (m, 5H), 1.47 – 1.09 (m, 5H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.5, 165.4, 146.5, 134.8, 134.6, 132.9, 132.7, 131.0, 130.4, 129.5, 129.0, 128.7, 128.4, 128.2, 127.9, 126.9, 39.6, 34.4, 26.9, 26.2, 20.7.

HRMS (ESI): m/z calculated for $[\text{C}_{27}\text{H}_{28}\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 398.2115, found: 398.2113.

diphenylmethanone *O*-(2-cyclohexyl-6-fluorobenzoyl) oxime (S55)



Chemical Formula: $\text{C}_{26}\text{H}_{24}\text{FNO}_2$

S55 was obtained as a white solid.

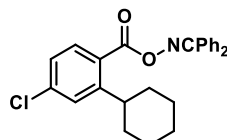
^1H NMR (400 MHz, CDCl_3) δ 7.51 (td, $J = 7.8, 4.2$ Hz, 1H), 7.40 (s, 1H), 7.17 (d, $J = 7.6$ Hz, 1H), 7.06 (t, $J = 8.8$ Hz, 1H), 1.93 – 1.81 (m, 5H), 1.65 – 1.47 (m, 4H), 1.46 – 1.33 (m, 1H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.8, 163.8, 160.6, 158.1, 148.3 (d, $J = 2.02$ Hz), 134.6, 132.4, 131.5 (d, $J = 9.09$ Hz), 131.1, 129.7, 129.2, 128.8, 128.4, 128.2, 121.9 (d, $J = 3.03$ Hz), 119.8 (d, $J = 16.16$ Hz), 112.9 (d, $J = 21.21$ Hz), 41.5, 34.1, 26.6, 26.0.

^{19}F NMR (376 MHz, CDCl_3) δ -114.2.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{25}\text{FNO}_2]^+ [\text{M} + \text{H}]^+$: 402.1864, found: 402.1862.

diphenylmethanone *O*-(4-chloro-2-cyclohexylbenzoyl) oxime (S56)



Chemical Formula: $\text{C}_{26}\text{H}_{24}\text{ClNO}_2$

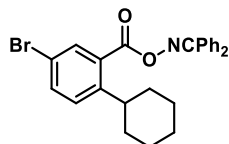
S56 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.72 – 7.63 (m, 2H), 7.52 – 7.43 (m, 4H), 7.43 – 7.29 (m, 6H), 7.07 (dd, $J = 8.4, 2.1$ Hz, 1H), 3.41 – 3.26 (m, 1H), 1.86 – 1.67 (m, 5H), 1.46 – 1.14 (m, 5H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.7, 164.5, 151.7, 138.4, 134.5, 132.7, 131.3, 131.1, 129.7, 129.1, 128.6, 128.4, 128.3, 127.5, 126.4, 125.7, 39.9, 34.2, 26.7, 26.1.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{25}\text{ClNO}_2]^+ [\text{M} + \text{H}]^+$: 418.1568, found: 418.1566.

diphenylmethanone *O*-(5-bromo-2-cyclohexylbenzoyl) oxime (S57)



Chemical Formula: $\text{C}_{26}\text{H}_{24}\text{BrNO}_2$

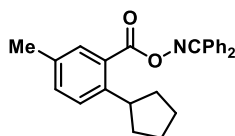
S57 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.69 (d, $J = 7.7$ Hz, 2H), 7.55 – 7.34 (m, 10H), 7.23 (d, $J = 8.4$ Hz, 1H), 3.28 (t, $J = 11.6$ Hz, 1H), 1.85 – 1.68 (m, 5H), 1.45 – 1.26 (m, 4H), 1.21 (q, $J = 12.8$ Hz, 1H).

^{13}C NMR (101 MHz, CDCl_3) δ 166.2, 163.9, 148.4, 134.8, 134.4, 132.6, 132.6, 131.2, 129.9, 129.8, 129.1, 128.8, 128.5, 128.5, 128.4, 118.8, 39.7, 34.2, 26.7, 26.1.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{25}\text{BrNO}_2]^+ [\text{M} + \text{H}]^+$: 462.1063, found: 462.1063.

diphenylmethanone *O*-(2-cyclopentyl-5-methylbenzoyl) oxime (S58)



Chemical Formula: $\text{C}_{26}\text{H}_{25}\text{NO}_2$

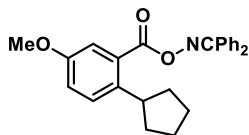
S58 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.68 (dd, $J = 7.2, 1.8$ Hz, 2H), 7.52 – 7.42 (m, 4H), 7.38 (td, $J = 6.9, 2.3$ Hz, 4H), 7.25 (d, $J = 8.7$ Hz, 1H), 7.20 (dd, $J = 6.5, 2.0$ Hz, 2H), 3.60 (tt, $J = 9.6, 7.5$ Hz, 1H), 2.21 (s, 3H), 2.01 – 1.88 (m, 2H), 1.80 – 1.54 (m, 4H), 1.54 – 1.38 (m, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.7, 165.6, 144.9, 134.8, 134.6, 133.0, 132.7, 131.0, 130.2, 129.5, 129.0, 128.8, 128.6, 128.4, 128.3, 126.8, 41.3, 34.8, 25.6, 20.6.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{26}\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 384.1958, found: 384.1958.

diphenylmethanone *O*-(2-cyclopentyl-5-methoxybenzoyl) oxime (S59)



Chemical Formula: $\text{C}_{26}\text{H}_{25}\text{NO}_3$

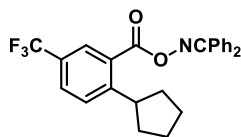
S59 was obtained as a white solid.

^1H NMR (400 MHz, CDCl_3) δ 7.74 – 7.59 (m, 2H), 7.46 (qd, $J = 5.3, 1.7$ Hz, 4H), 7.42 – 7.34 (m, 4H), 7.31 – 7.24 (m, 1H), 6.96 (dd, $J = 8.7, 2.9$ Hz, 1H), 6.88 (d, $J = 2.9$ Hz, 1H), 3.62 (s, 3H), 2.02 – 1.90 (m, 2H), 1.82 – 1.54 (m, 4H), 1.46 (ddt, $J = 16.7, 11.2, 7.8$ Hz, 2H).

^{13}C NMR (101 MHz, CDCl_3) δ 165.6, 165.2, 156.8, 140.2, 134.6, 133.1, 131.1, 129.4, 129.4, 129.0, 128.5, 128.4, 128.3, 128.1, 119.2, 113.3, 55.3, 41.0, 34.8, 25.6.

HRMS (ESI): m/z calculated for $[\text{C}_{26}\text{H}_{26}\text{NO}_3]^+ [\text{M} + \text{H}]^+$: 384.1958, found: 384.1958.

diphenylmethanone *O*-(2-cyclopentyl-5-(trifluoromethyl)benzoyl) oxime (S60)



Chemical Formula: C₂₆H₂₂F₃NO₂

S60 was obtained as a white solid.

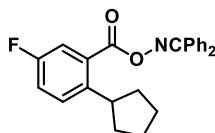
¹H NMR (400 MHz, CDCl₃) δ 7.76 – 7.56 (m, 4H), 7.56 – 7.33 (m, 9H), 3.77 (ddd, *J* = 17.2, 9.5, 7.5 Hz, 1H), 2.04 (dt, *J* = 10.9, 5.7, 2.4 Hz, 2H), 1.88 – 1.59 (m, 4H), 1.59 – 1.43 (m, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 166.4, 164.1, 152.5, 134.3, 132.6, 131.3, 129.8, 129.3, 129.1, 128.5, 128.4, 128.4, 128.2, 127.7 (q, *J* = 33.33 Hz), 127.7, 126.8 (q, *J* = 4.04 Hz), 123.6 (q, *J* = 272.7 Hz), 41.6, 34.9, 25.8.

¹⁹F NMR (376 MHz, CDCl₃) δ -62.7.

HRMS (ESI): *m/z* calculated for [C₂₆H₂₃F₃NO₂]⁺ [*M* + *H*]⁺: 438.1675, found: 438.1674.

diphenylmethanone *O*-(2-cyclopentyl-5-fluorobenzoyl) oxime (S61)



Chemical Formula: C₂₅H₂₂FNO₂

S61 was obtained as a white solid.

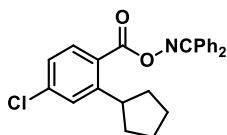
¹H NMR (400 MHz, CDCl₃) δ 7.68 (dt, *J* = 7.2, 1.4 Hz, 2H), 7.53 – 7.44 (m, 4H), 7.44 – 7.29 (m, 5H), 7.09 (ddd, *J* = 16.8, 8.5, 2.8 Hz, 2H), 3.62 (tt, *J* = 9.7, 7.5 Hz, 1H), 2.05 – 1.91 (m, 2H), 1.75 (tt, *J* = 9.5, 6.9, 4.2 Hz, 2H), 1.64 (dtd, *J* = 12.3, 8.0, 3.2 Hz, 2H), 1.53 – 1.39 (m, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 166.0, 164.5 (d, *J* = 3.03 Hz), 161.2, 158.7, 143.8 (d, *J* = 3.03 Hz), 134.4, 132.7, 131.2, 130.3 (d, *J* = 7.07 Hz), 129.7, 129.1, 128.7 (d, *J* = 7.07 Hz), 128.5, 128.5, 128.4, 119.0 (d, *J* = 21.21 Hz), 116.1 (d, *J* = 23.23 Hz), 41.1, 34.9, 25.6.

¹⁹F NMR (376 MHz, CDCl₃) δ -117.2.

HRMS (ESI): *m/z* calculated for [C₂₅H₂₃FNO₂]⁺ [*M* + *H*]⁺: 388.1707, found: 388.1707.

diphenylmethanone *O*-(4-chloro-2-cyclopentylbenzoyl) oxime (S62)



Chemical Formula: C₂₅H₂₂ClNO₂

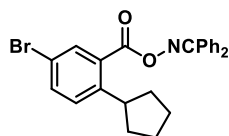
S62 was obtained as a white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.67 (dt, *J* = 7.1, 1.4 Hz, 2H), 7.54 – 7.29 (m, 10H), 7.08 (dd, *J* = 8.4, 2.1 Hz, 1H), 3.65 (tt, *J* = 9.6, 7.5 Hz, 1H), 1.99 (dq, *J* = 16.7, 5.9 Hz, 2H), 1.76 (qdt, *J* = 10.4, 7.9, 4.1 Hz, 2H), 1.63 (qd, *J* = 8.1, 3.9 Hz, 2H), 1.55 – 1.37 (m, 2H).

¹³C NMR (101 MHz, CDCl₃) δ 165.8, 164.9, 150.4, 138.3, 134.5, 132.8, 131.2, 131.2, 129.7, 129.1, 128.6, 128.5, 128.4, 127.3, 125.7, 41.6, 34.7, 25.7.

HRMS (ESI): m/z calculated for $[C_{25}H_{23}ClNO_2]^+ [M + H]^+$: 404.1412, found: 404.1408.

diphenylmethanone O-(5-bromo-2-cyclopentylbenzoyl) oxime (S63)



Chemical Formula: $C_{25}H_{22}BrNO_2$

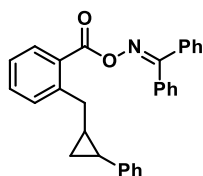
S63 was obtained as a white solid.

1H NMR (400 MHz, $CDCl_3$) δ 7.75 – 7.61 (m, 2H), 7.55 – 7.44 (m, 6H), 7.44 – 7.32 (m, 4H), 7.23 (d, $J = 8.2$ Hz, 1H), 3.60 (ddd, $J = 17.1, 9.6, 7.5$ Hz, 1H), 1.97 (dq, $J = 10.7, 6.4, 2.4$ Hz, 2H), 1.83 – 1.54 (m, 4H), 1.53 – 1.37 (m, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 166.2, 164.2, 147.1, 134.8, 134.4, 132.7, 132.4, 131.2, 130.8, 129.8, 129.1, 128.8, 128.5, 128.5, 118.7, 41.3, 34.8, 25.7.

HRMS (ESI): m/z calculated for $[C_{25}H_{23}BrNO_2]^+ [M + H]^+$: 448.0907, found: 448.0905.

diphenylmethanone O-(2-((2-phenylcyclopropyl)methyl)benzoyl) oxime (S64)



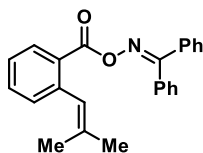
Chemical Formula: $C_{30}H_{25}NO_2$

1H NMR (400 MHz, $CDCl_3$) δ 7.73 – 7.62 (m, 2H), 7.57 – 7.28 (m, 13H), 7.27 – 7.06 (m, 4H), 7.06 – 6.93 (m, 2H), 3.01 (qd, $J = 15.1, 6.8$ Hz, 2H), 1.76 (dt, $J = 9.1, 4.9$ Hz, 1H), 1.48 – 1.30 (m, 1H), 0.91 (ddt, $J = 14.1, 8.5, 5.1$ Hz, 2H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 165.51, 164.86, 143.68, 143.40, 134.64, 132.97, 132.31, 131.01, 130.38, 129.58, 129.19, 129.05, 128.99, 128.64, 128.42, 128.33, 128.27, 128.22, 128.20, 127.90, 127.81, 125.96, 125.74, 125.26, 37.77, 23.78, 23.46, 16.03.

HRMS (ESI): m/z calculated for $[C_{30}H_{26}NO_2]^+ [M + H]^+$: 432.1958, found: 432.1958.

diphenylmethanone O-(2-(2-methylprop-1-en-1-yl)benzoyl) oxime (S65)



Chemical Formula: $C_{24}H_{21}NO_2$

1H NMR (400 MHz, $CDCl_3$) δ 7.70 – 7.65 (m, 2H), 7.56 (dd, $J = 7.9, 1.4$ Hz, 1H), 7.50 – 7.43 (m, 4H), 7.42 – 7.34 (m, 5H), 7.22 (d, $J = 7.7$ Hz, 1H), 7.16 (t, $J = 7.6$ Hz, 1H), 6.55 (s, 1H), 1.84 (d, $J = 1.5$ Hz, 3H), 1.66 (d, $J = 1.4$ Hz, 3H).

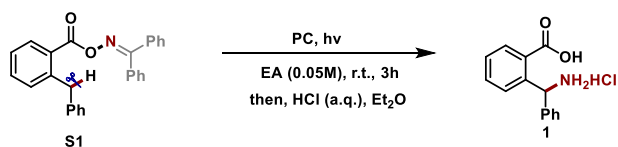
^{13}C NMR (101 MHz, $CDCl_3$) δ 165.1, 164.7, 140.4, 135.2, 134.8, 132.8, 131.6, 131.1, 131.1, 130.9, 130.0, 129.5, 129.0, 128.8, 128.3, 128.2, 128.2, 127.9, 125.9, 124.3, 26.2, 19.2.

HRMS (ESI): m/z calculated for $[C_{24}H_{22}NO_2]^+ [M + H]^+$: 356.1645, found: 356.1645.

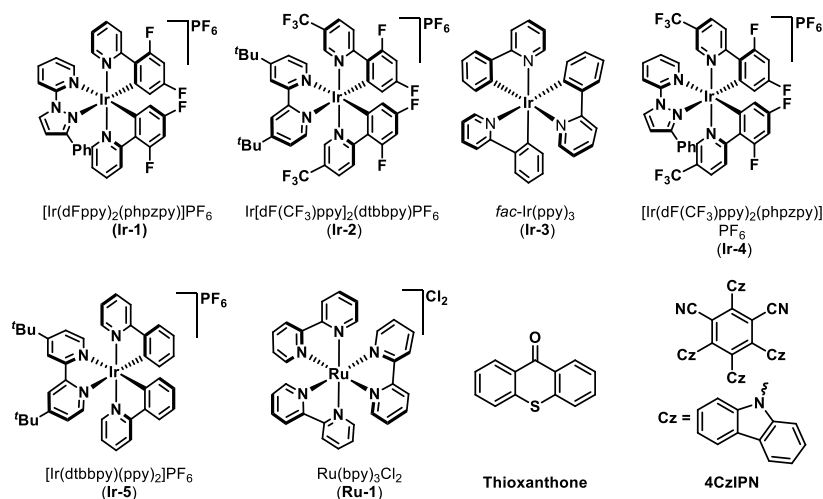
2.4 Additional Reaction Optimization

Screening reaction parameters for γ C–H amination

Supplementary Table 1: Screening of photocatalyst^a

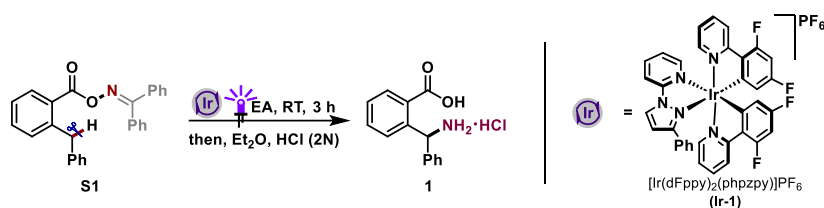


Entry	PC	Yield
1	[Ir(dFppy) ₂ (phppzpy)]PF ₆ (Ir-1)	82%
2	Ir[dF(CF ₃)ppy] ₂ (dtbbpy)PF ₆ (Ir-2)	57%
3 ^b	fac-Ir(ppy) ₃ (Ir-3)	N.D.
4	[Ir(dF(CF ₃)ppy) ₂ (phppzpy)]PF ₆ (Ir-4)	64%
5	[Ir(dtbbpy)(ppy) ₂]PF ₆ (Ir-5)	24%
6 ^b	Ru(bpy) ₃ Cl ₂ (Ru-1)	N.D.
7 ^c	Thioxanthone	73%
8 ^c	4CzIPN	58%



^aReaction conditions: **S1** (0.2 mmol, 1.0 equiv.), PC (1 mol%), EA (4 mL), irradiation by purple LEDs ($\lambda_{\text{max}} = 395$ nm) at room temperature under argon for 3 h, then the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and product **1** was obtained as a white solid. ^bThe reaction was irradiated by blue LEDs ($\lambda_{\text{max}} = 450$ nm). ^cPC (5 mol%) was used. N.D. = not detected.

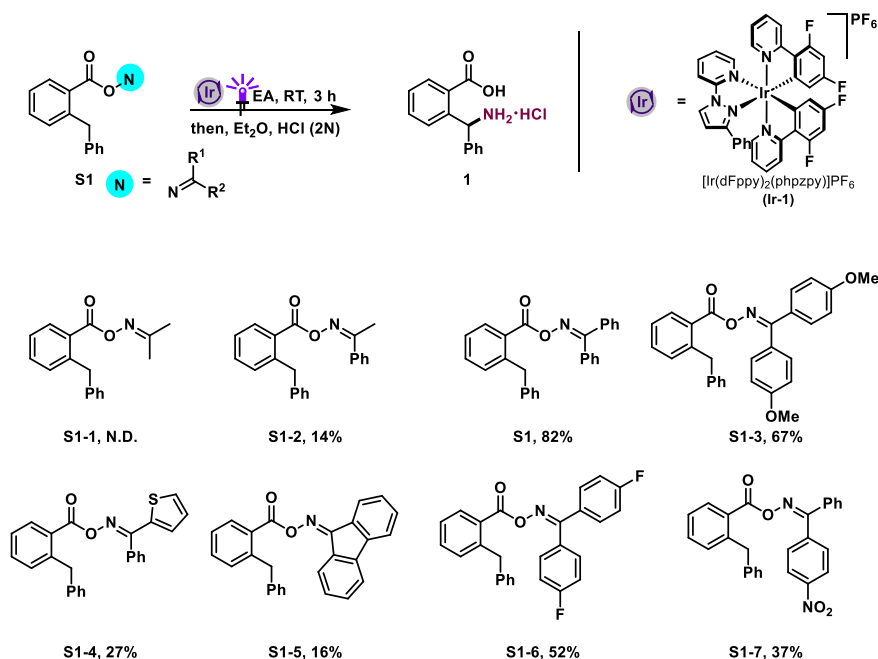
Supplementary Table 2: Screening of solvents^a



Entry	Solvent	Yield
1	EA	82%
2	CH ₃ CN	44%
3	DCM	60%
4	CH ₃ COCH ₃	45%
5	DCE	57%
6	THF	68%
7	MeOH	49%
8	Toluene	58%
9	1,4-dioxane	64%

^aReaction conditions: **S1** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phzppy)]PF₆ (1 mol%), solvent (4 mL), irradiation by purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature under argon for 3 h, then the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and product **1** was obtained as a white solid.

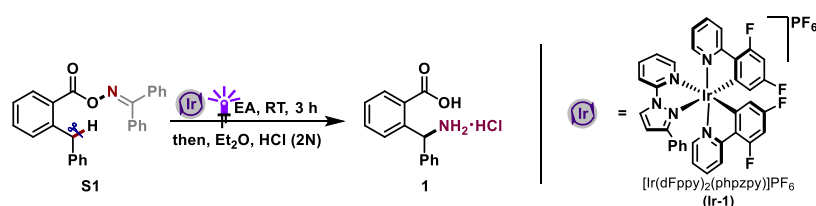
Supplementary Figure 1: Screening the effect of different activators^a



^aReaction conditions: **S1** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phzppy)]PF₆ (1 mol%), EA (4 mL),

irradiation by purple LEDs ($\lambda_{\max} = 395$ nm) at room temperature under argon for 3 h, then the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and product **1** was obtained as a white solid.

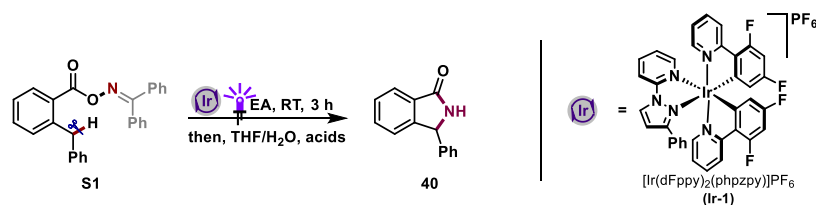
Supplementary Table 3: Screening the effect of triplet state inhibitor^a



Entry	Conditions	Yield
1	none	82%
2	no light	N.D.
3	add 3 equiv. <i>E</i> -stilbene	54%
4	in air	39%

^aReaction conditions: **S1** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), EA (4 mL), irradiation by purple LEDs ($\lambda_{\max} = 395$ nm) at room temperature under argon for 3 h, then the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and product **1** was obtained as a white solid.

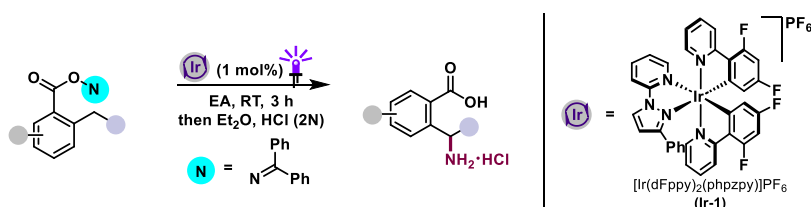
Supplementary Table 4: Screening of acids for the synthesis of γ -lactams^a



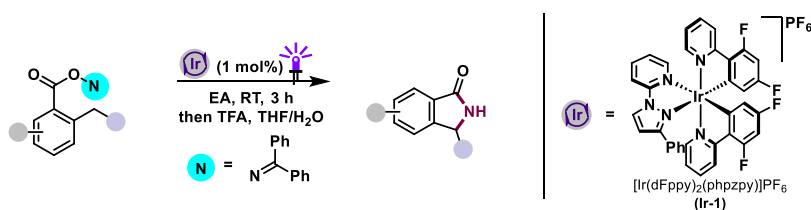
Entry	Acid	Yield
1	TFA	71%
2	HCl (2N)	41%
3	AcOH	56%
4	H ₂ SO ₄	45%
5	PTSA	37%

^aReaction conditions: **S1** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), EA (4 mL), irradiation by purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature under argon for 3 h, then the solvents were removed under reduced pressure, stirred in THF (2 mL)/H₂O (50 equiv. 180 μ L), acid (1 equiv.) for 12 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was concentrated *in vacuo*. The residue was purified by column chromatography on silica gel (eluent: *n*-hexanes/10% methanol-EtOAc solution = 2:1), and product **40** was obtained as a white solid. TFA, trifluoroacetic acid. PTSA, *p*-toluenesulfonic acid.

2.5 Experimental Procedures



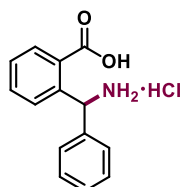
Procedure: To an oven-dried glass tube (10 mL) equipped with a magnetic stir bar was added the oxime esters (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle. The tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25-30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and the product was obtained as a white solid.



Procedure: To an oven-dried glass tube (10 mL) equipped with a magnetic stir bar were added the oxime esters (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25-30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, and stirred in THF (2 mL)/H₂O (50 equiv. 180 μL), TFA (1.0 equiv.) for 12 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was concentrated *in vacuo*. The residue was purified by column chromatography on silica gel (eluent: *n*-hexanes/10% methanol-EtOAc solution = 2:1), and the product was obtained as a white solid.

2.6 Characterization data of products

2-(amino(phenyl)methyl)benzoic acid hydrochloride (1)



Chemical Formula: C₁₄H₁₄ClNO₂

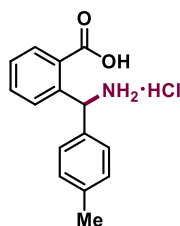
Product **1** was obtained as a white solid (43.2 mg, 82%)

¹H NMR (400 MHz, D₂O) δ 7.95 (dd, $J = 7.9, 1.4 \text{ Hz}$, 1H), 7.63 (td, $J = 7.6, 1.5 \text{ Hz}$, 1H), 7.51 (td, $J = 7.7, 1.2 \text{ Hz}$, 1H), 7.47 – 7.35 (m, 4H), 7.29 (dd, $J = 7.5, 2.2 \text{ Hz}$, 2H), 6.35 (s, 1H).

¹³C NMR (101 MHz, D₂O) δ 170.9, 136.4, 135.9, 133.1, 131.4, 130.1, 129.3, 129.0, 128.9, 128.5, 127.3, 54.9.

HRMS (ESI): m/z calculated for [C₁₄H₁₄NO₂]⁺ [M]⁺: 228.1019, found: 228.1018.

2-(amino(*p*-tolyl)methyl)benzoic acid hydrochloride (2)



Chemical Formula: C₁₅H₁₆ClNO₂

Product **2** was obtained as a white solid (41.8 mg, 75%)

$^1\text{H NMR}$ (600 MHz, D_2O) δ 7.95 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.63 (td, $J = 7.7, 1.4$ Hz, 1H), 7.56 – 7.47 (m, 1H), 7.43 (d, $J = 7.9$ Hz, 1H), 7.23 (d, $J = 7.9$ Hz, 2H), 7.16 (d, $J = 7.9$ Hz, 2H), 6.31 (s, 1H), 2.28 (s, 3H).

$^{13}\text{C NMR}$ (151 MHz, D_2O) δ 170.9, 139.3, 136.6, 133.0, 132.9, 131.4, 130.1, 129.6, 129.3, 128.3, 127.3, 54.7, 20.1.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{16}\text{NO}_2]^+ [\text{M}]^+$: 242.1176, found: 242.1175.

2-(amino(4-(tert-butyl)phenyl)methyl)benzoic acid hydrochloride (**3**)



Chemical Formula: $\text{C}_{18}\text{H}_{22}\text{ClNO}_2$

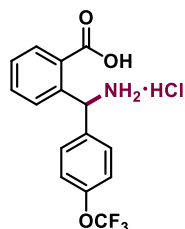
Product **3** was obtained as a white solid (43.7 mg, 68%)

$^1\text{H NMR}$ (600 MHz, D_2O) δ 7.94 (d, $J = 7.7$ Hz, 1H), 7.55 (t, $J = 7.7$ Hz, 1H), 7.46 (t, $J = 7.6$ Hz, 1H), 7.37 (dd, $J = 12.6, 8.0$ Hz, 3H), 7.18 (d, $J = 8.2$ Hz, 2H), 6.32 (s, 1H), 1.16 (s, 9H).

$^{13}\text{C NMR}$ (151 MHz, D_2O) δ 170.6, 152.3, 136.8, 133.2, 133.0, 131.6, 129.8, 129.3, 128.6, 127.1, 126.0, 54.5, 34.0, 30.4.

HRMS (ESI): m/z calculated for $[\text{C}_{18}\text{H}_{22}\text{NO}_2]^+ [\text{M}]^+$: 284.1645, found: 284.1641.

2-(amino(4-(trifluoromethoxy)phenyl)methyl)benzoic acid hydrochloride (**4**)



Chemical Formula: $\text{C}_{15}\text{H}_{13}\text{ClF}_3\text{NO}_3$

Product **4** was obtained as a white solid (42.0 mg, 60%)

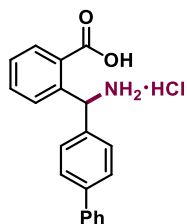
$^1\text{H NMR}$ (600 MHz, D_2O) δ 8.02 (d, $J = 7.7$ Hz, 1H), 7.71 (t, $J = 7.7$ Hz, 1H), 7.59 (t, $J = 7.6$ Hz, 1H), 7.48 (d, $J = 7.9$ Hz, 1H), 7.41 (d, $J = 8.7$ Hz, 2H), 7.37 (d, $J = 8.5$ Hz, 2H), 6.44 (s, 1H).

$^{13}\text{C NMR}$ (151 MHz, D_2O) δ 170.9, 149.0, 136.1, 134.8, 133.2, 131.6, 130.3, 129.6, 129.2, 128.5, 121.4, 121.2 (q, $J = 256.7$ Hz), 54.5.

$^{19}\text{F NMR}$ (565 MHz, D_2O) δ -56.85.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{13}\text{F}_3\text{NO}_3]^+ [\text{M}]^+$: 312.0842, found: 312.0839.

2-([1,1'-biphenyl]-4-yl(amino)methyl)benzoic acid hydrochloride (**5**)



Chemical Formula: $C_{20}H_{18}ClNO_2$

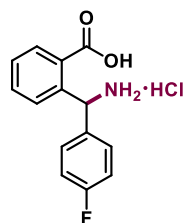
Product **5** was obtained as a white solid (49.5 mg, 73%)

1H NMR (600 MHz, MeOD) δ 8.15 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.68 (td, $J = 8.1, 1.7$ Hz, 3H), 7.64 – 7.60 (m, 2H), 7.58 – 7.50 (m, 2H), 7.49 – 7.41 (m, 4H), 7.37 – 7.32 (m, 1H), 6.65 (s, 1H).

^{13}C NMR (151 MHz, MeOD) δ 170.2, 142.9, 141.3, 139.3, 136.8, 134.1, 133.1, 131.4, 131.0, 130.4, 130.1, 130.0, 129.0, 128.8, 128.5, 128.5, 128.0, 55.6.

HRMS (ESI): m/z calculated for $[C_{20}H_{18}NO_2]^+ [M]^+$: 304.1332, found: 304.1328.

2-(amino(4-fluorophenyl)methyl)benzoic acid hydrochloride (6)



Chemical Formula: $C_{14}H_{13}ClFNO_2$

Product **6** was obtained as a white solid (33.2mg, 59%)

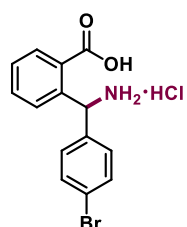
1H NMR (400 MHz, D_2O) δ 7.94 (dd, $J = 7.7, 1.5$ Hz, 1H), 7.67 (td, $J = 7.7, 1.5$ Hz, 1H), 7.55 (td, $J = 7.6, 1.2$ Hz, 1H), 7.46 (dd, $J = 7.8, 1.2$ Hz, 1H), 7.37 – 7.26 (m, 2H), 7.20 – 7.11 (m, 2H), 6.33 (s, 1H).

^{13}C NMR (101 MHz, D_2O) δ 171.4, 163.7, 161.3, 135.9, 132.7, 131.9 (d, $J = 3.03$ Hz), 131.2, 130.9, 129.5 (d, $J = 9.09$ Hz), 129.4, 128.1, 115.8 (d, $J = 22.22$ Hz), 54.6.

^{19}F NMR (376 MHz, D_2O) δ -113.49.

HRMS (ESI): m/z calculated for $[C_{14}H_{13}FNO_2]^+ [M]^+$: 246.0925, found: 246.0924.

2-(amino(4-bromophenyl)methyl)benzoic acid hydrochloride (7)



Chemical Formula: $C_{14}H_{13}BrClNO_2$

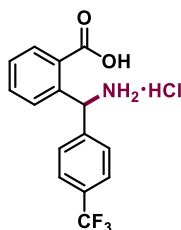
Product **7** was obtained as a white solid (46.1mg, 67%)

1H NMR (400 MHz, D_2O) δ 7.94 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.61 (td, $J = 7.7, 1.5$ Hz, 1H), 7.55 – 7.46 (m, 3H), 7.43 – 7.33 (m, 1H), 7.19 – 7.09 (m, 2H), 6.30 (s, 1H).

^{13}C NMR (101 MHz, D_2O) δ 170.6, 136.0, 135.0, 133.1, 131.9, 131.6, 129.9, 129.4, 129.1, 128.31, 122.4, 54.4.

HRMS (ESI): m/z calculated for $[\text{C}_{14}\text{H}_{13}\text{BrNO}_2]^+ [\text{M}]^+$: 306.0124, found: 306.0124.

2-(amino(4-(trifluoromethyl)phenyl)methyl)benzoic acid hydrochloride (8)



Chemical Formula: $\text{C}_{15}\text{H}_{13}\text{ClF}_3\text{NO}_2$

Product **8** was obtained as a white solid (37.7 mg, 57%)

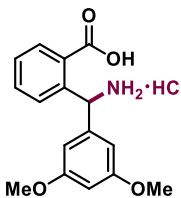
^1H NMR (600 MHz, D_2O) δ 7.78 (d, $J = 8.2$ Hz, 2H), 7.69 (td, $J = 7.7, 1.5$ Hz, 1H), 7.60 (td, $J = 7.7, 1.2$ Hz, 1H), 7.53 – 7.47 (m, 2H), 7.47 – 7.42 (m, 1H), 6.45 (s, 1H).

^{13}C NMR (151 MHz, D_2O) δ 171.1, 134.0, 135.7, 133.0, 131.5, 130.7, 130.1 (q, $J = 33.22$ Hz), 129.7, 128.8, 127.8, 125.9 (q, $J = 4.53$ Hz), 123.9 (q, $J = 271.8$ Hz), 54.8.

^{19}F NMR (565 MHz, D_2O) δ -62.6.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{13}\text{F}_3\text{NO}_2]^+ [\text{M}]^+$: 296.0893, found: 296.0890.

2-(amino(3,5-dimethoxyphenyl)methyl)benzoic acid hydrochloride (9)



Chemical Formula: $\text{C}_{16}\text{H}_{18}\text{ClNO}_4$

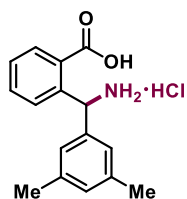
Product **9** was obtained as a white solid (41.3 mg, 64%)

^1H NMR (400 MHz, D_2O) δ 7.94 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.59 (td, $J = 7.7, 1.5$ Hz, 1H), 7.48 (td, $J = 7.6, 1.3$ Hz, 1H), 7.39 (dd, $J = 8.0, 1.2$ Hz, 1H), 6.49 (q, $J = 2.0$ Hz, 3H), 6.26 (s, 1H), 3.71 (s, 6H).

^{13}C NMR (101 MHz, D_2O) δ 171.0, 160.6, 138.4, 136.0, 133.0, 131.4, 130.4, 129.5, 128.6, 105.9, 100.3, 55.4, 54.8.

HRMS (ESI): m/z calculated for $[\text{C}_{16}\text{H}_{18}\text{NO}_4]^+ [\text{M}]^+$: 288.1230, found: 288.1229.

2-(amino(3,5-dimethylphenyl)methyl)benzoic acid hydrochloride (10)



Chemical Formula: $C_{16}H_{18}ClNO_2$

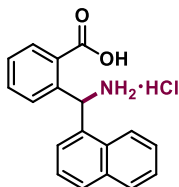
Product **10** was obtained as a white solid (45.3 mg, 78%)

1H NMR (400 MHz, D_2O) δ 7.94 (dd, $J = 7.8, 1.5$ Hz, 1H), 7.61 (td, $J = 7.7, 1.5$ Hz, 1H), 7.49 (td, $J = 7.7, 1.2$ Hz, 1H), 7.41 (dd, $J = 7.9, 1.2$ Hz, 1H), 7.02 (d, $J = 1.9$ Hz, 1H), 6.97 – 6.79 (m, 2H), 6.27 (s, 1H), 2.21 (s, 6H).

^{13}C NMR (101 MHz, D_2O) δ 170.9, 139.2, 136.5, 136.0, 133.0, 131.4, 130.1, 130.1, 129.3, 128.5, 124.8, 54.8, 20.3.

HRMS (ESI): m/z calculated for $[C_{16}H_{18}NO_2]^+ [M]^+$: 256.1332, found: 256.1334.

2-(amino(naphthalen-1-yl)methyl)benzoic acid hydrochloride (11)



Chemical Formula: $C_{18}H_{16}ClNO_2$

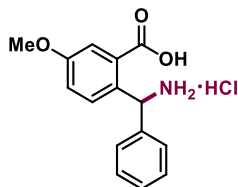
Product **11** was obtained as a white solid (40.8 mg, 65%)

1H NMR (600 MHz, D_2O) δ 8.02 (dd, $J = 7.9, 1.5$ Hz, 1H), 7.88 (dh, $J = 7.2, 3.6$ Hz, 3H), 7.84 – 7.79 (m, 1H), 7.65 (td, $J = 7.7, 1.5$ Hz, 1H), 7.61 – 7.53 (m, 3H), 7.47 (d, $J = 7.9$ Hz, 1H), 7.34 (dd, $J = 8.6, 2.0$ Hz, 1H), 6.55 (s, 1H).

^{13}C NMR (151 MHz, D_2O) δ 170.7, 136.5, 133.5, 133.2, 132.7, 132.6, 131.6, 130.0, 129.5, 128.9, 128.7, 128.1, 127.7, 127.2, 127.1, 126.2, 125.0, 55.0.

HRMS (ESI): m/z calculated for $[C_{18}H_{16}NO_2]^+ [M]^+$: 278.1176, found: 278.1175.

2-(amino(phenyl)methyl)-5-methoxybenzoic acid hydrochloride (12)



Chemical Formula: $C_{15}H_{16}ClNO_3$

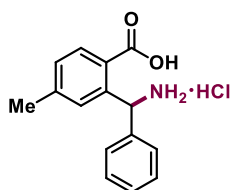
Product **12** was obtained as a white solid (41.0 mg, 70%)

1H NMR (600 MHz, D_2O) δ 8.01 (d, $J = 8.7$ Hz, 1H), 7.55 – 7.39 (m, 3H), 7.39 – 7.31 (m, 2H), 7.03 (dd, $J = 8.8, 2.5$ Hz, 1H), 6.95 (d, $J = 2.6$ Hz, 1H), 6.44 (s, 1H), 3.83 (s, 3H).

^{13}C NMR (151 MHz, D_2O) δ 170.1, 162.7, 139.4, 135.7, 134.5, 129.1, 128.9, 127.2, 121.5, 115.3, 113.5, 55.6, 54.9.

HRMS (ESI): m/z calculated for $[C_{15}H_{16}NO_3]^+ [M]^+$: 258.1125, found: 258.1123.

2-(amino(phenyl)methyl)-4-methylbenzoic acid hydrochloride (13)



Chemical Formula: C₁₅H₁₆ClNO₂

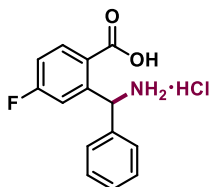
Product **13** was obtained as a white solid (37.8 mg, 68%)

¹H NMR (600 MHz, D₂O) δ 7.90 (d, *J* = 8.0 Hz, 1H), 7.49 – 7.40 (m, 3H), 7.40 – 7.31 (m, 3H), 7.29 – 7.23 (m, 1H), 6.38 (s, 1H), 2.37 (s, 3H).

¹³C NMR (151 MHz, D₂O) δ 170.9, 144.6, 136.7, 136.0, 131.7, 129.8, 129.5, 129.3, 129.0, 128.8, 127.2, 127.0, 54.9, 20.7.

HRMS (ESI): *m/z* calculated for [C₁₅H₁₆NO₂]⁺ [M]⁺: 277.0864, found: 277.0859.

2-(amino(phenyl)methyl)-4-fluorobenzoic acid hydrochloride (14)



Chemical Formula: C₁₄H₁₃ClFNO₂

Product **14** was obtained as a white solid (37.8 mg, 67%)

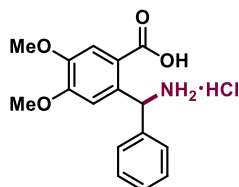
¹H NMR (400 MHz, D₂O) δ 8.02 (dd, *J* = 8.8, 5.9 Hz, 1H), 7.45 – 7.36 (m, 3H), 7.33 – 7.26 (m, 2H), 7.21 (td, *J* = 8.3, 2.6 Hz, 1H), 7.10 (dd, *J* = 10.0, 2.6 Hz, 1H), 6.38 (s, 1H).

¹³C NMR (101 MHz, D₂O) δ 169.7, 166.0, 163.4, 139.9 (d, *J* = 8.08 Hz), 135.2, 134.6 (d, *J* = 10.1 Hz), 129.1, 129.0, 127.2, 126.1 (d, *J* = 4.04 Hz), 116.3 (d, *J* = 8.08 Hz), 116.1 (d, *J* = 6.06 Hz), 54.5.

¹⁹F NMR (376 MHz, D₂O) δ -105.0.

HRMS (ESI): *m/z* calculated for [C₁₄H₁₃FNO₂]⁺ [M]⁺: 246.0925, found: 276.0924.

2-(amino(phenyl)methyl)-4,5-dimethoxybenzoic acid hydrochloride (15)



Chemical Formula: C₁₆H₁₈ClNO₄

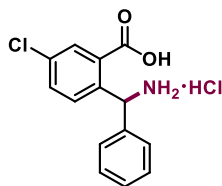
Product **15** was obtained as a white solid (58.1 mg, 90%)

¹H NMR (600 MHz, D₂O) δ 7.48 (s, 1H), 7.40 (ddd, *J* = 13.2, 8.0, 6.3 Hz, 3H), 7.32 – 7.27 (m, 2H), 6.86 (s, 1H), 6.39 (s, 1H), 3.75 (s, 3H), 3.74 (s, 3H).

¹³C NMR (151 MHz, D₂O) δ 169.9, 151.7, 147.8, 136.0, 131.4, 129.1, 128.8, 126.9, 122.3, 114.4, 111.8, 55.7, 54.6.

HRMS (ESI): *m/z* calculated for [C₁₆H₁₈NO₄]⁺ [M]⁺: 288.1230, found: 288.1227.

2-(amino(phenyl)methyl)-5-chlorobenzoic acid hydrochloride (16)



Chemical Formula: C₁₄H₁₃Cl₂NO₂

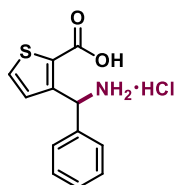
Product **16** was obtained as a white solid (36.9 mg, 62%)

¹H NMR (600 MHz, D₂O) δ 7.99 (d, *J* = 2.3 Hz, 1H), 7.64 (dd, *J* = 8.4, 2.4 Hz, 1H), 7.50 – 7.42 (m, 3H), 7.38 (d, *J* = 8.5 Hz, 1H), 7.35 – 7.29 (m, 2H), 6.32 (s, 1H).

¹³C NMR (151 MHz, D₂O) δ 170.1, 135.5, 134.8, 134.8, 132.9, 132.4, 131.1, 130.2, 129.1, 128.9, 127.2, 54.6.

HRMS (ESI): *m/z* calculated for [C₁₄H₁₃ClNO₂]⁺ [M]⁺: 262.0629, found: 262.0628.

3-(amino(phenyl)methyl)thiophene-2-carboxylic acid hydrochloride (17)



Chemical Formula: C₁₂H₁₂ClNO₂S

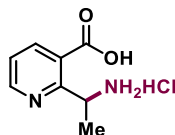
Product **17** was obtained as a white solid (36.1 mg, 67%)

¹H NMR (400 MHz, D₂O) δ 7.70 (d, *J* = 5.2 Hz, 1H), 7.47 – 7.32 (m, 5H), 7.12 (d, *J* = 5.2 Hz, 1H), 6.32 (s, 1H).

¹³C NMR (101 MHz, D₂O) δ 165.1, 142.7, 135.5, 133.1, 130.2, 129.2, 129.2, 128.4, 127.1, 52.4.

HRMS (ESI): *m/z* calculated for [C₁₂H₁₂NO₂S]⁺ [M]⁺: 234.0583, found: 234.0582.

2-(1-aminoethyl)nicotinic acid hydrochloride (18)



Chemical Formula: C₈H₁₁ClN₂O₂

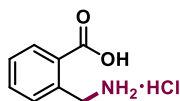
Product **18** was obtained as a white solid (36.5 mg, 90%)

¹H NMR (400 MHz, D₂O) δ 8.75 (dd, *J* = 4.9, 1.7 Hz, 1H), 8.43 (dd, *J* = 8.0, 1.7 Hz, 1H), 7.56 (dd, *J* = 8.0, 4.9 Hz, 1H), 5.45 (t, *J* = 6.7 Hz, 1H), 1.59 (d, *J* = 6.7 Hz, 3H).

¹³C NMR (101 MHz, D₂O) δ 168.5, 156.4, 151.8, 140.8, 124.9, 124.1, 49.0, 18.9.

HRMS (ESI): *m/z* calculated for [C₈H₁₁N₂O₂]⁺ [M]⁺: 167.0815, found: 167.0815.

2-(aminomethyl)benzoic acid hydrochloride (19)



Chemical Formula: C₈H₁₀ClNO₂

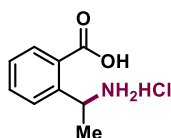
Product **19** was obtained as a white solid (25.7 mg, 68%)

¹H NMR (600 MHz, D₂O) δ 8.08 (dd, *J* = 7.8, 1.4 Hz, 1H), 7.67 (td, *J* = 7.6, 1.4 Hz, 1H), 7.58 (td, *J* = 7.7, 1.3 Hz, 1H), 7.53 (dd, *J* = 7.6, 1.3 Hz, 1H), 4.37 (s, 2H).

¹³C NMR (151 MHz, D₂O) δ 170.4, 133.6, 133.3, 132.3, 131.7, 130.0, 129.9, 42.6.

HRMS (ESI): *m/z* calculated for [C₈H₁₀NO₂]⁺ [M]⁺: 152.0706, found: 152.0705.

2-(1-aminoethyl)benzoic acid hydrochloride (20)



Chemical Formula: C₉H₁₂ClNO₂

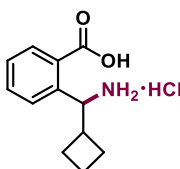
Product **20** was obtained as a white solid (37.8 mg, 94%)

¹H NMR (400 MHz, D₂O) δ 7.97 (dd, *J* = 7.8, 1.4 Hz, 1H), 7.67 (dtd, *J* = 15.9, 7.9, 1.5 Hz, 2H), 7.53 (td, *J* = 7.5, 1.5 Hz, 1H), 5.13 (q, *J* = 6.9 Hz, 1H), 1.66 (d, *J* = 6.9 Hz, 3H).

¹³C NMR (101 MHz, D₂O) δ 171.1, 137.5, 133.3, 131.2, 130.0, 129.3, 127.2, 47.7, 18.1.

HRMS (ESI): *m/z* calculated for [C₉H₁₂NO₂]⁺ [M]⁺: 166.0863, found: 166.0862.

2-(amino(cyclobutyl)methyl)benzoic acid hydrochloride (21)



Chemical Formula: C₁₂H₁₆ClNO₂

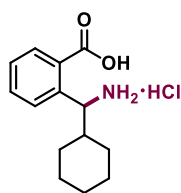
Product **21** was obtained as a white solid (42.8 mg, 89%)

¹H NMR (400 MHz, D₂O) δ 7.95 (dd, *J* = 7.9, 1.5 Hz, 1H), 7.65 (td, *J* = 7.6, 1.5 Hz, 1H), 7.52 (ddd, *J* = 11.1, 7.4, 2.2 Hz, 2H), 4.94 (d, *J* = 10.7 Hz, 1H), 3.16 – 3.01 (m, 1H), 2.25 (m, 1H), 2.03 – 1.85 (m, 2H), 1.79 (m, 2H), 1.63 – 1.48 (m, 1H).

¹³C NMR (101 MHz, D₂O) δ 171.3, 134.9, 133.1, 131.2, 131.1, 129.4, 128.3, 56.9, 36.6, 25.7, 24.6, 16.9.

HRMS (ESI): *m/z* calculated for [C₁₂H₁₆NO₂]⁺ [M]⁺: 206.1176, found: 206.1174.

2-(amino(cyclohexyl)methyl)benzoic acid hydrochloride (22)



Chemical Formula: $C_{14}H_{20}ClNO_2$

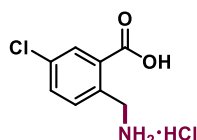
Product **22** was obtained as a white solid (48.6 mg, 90%)

1H NMR (600 MHz, D_2O) δ 7.94 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.69 (td, $J = 7.7, 1.4$ Hz, 1H), 7.61 – 7.51 (m, 2H), 4.74 (s, 1H), 2.08 – 1.93 (m, 2H), 1.82 (m, 1H), 1.60 (m, 2H), 1.29 (m, 1H), 1.25 – 1.06 (m, 4H), 0.87 (m, 1H).

^{13}C NMR (151 MHz, D_2O) δ 171.8, 136.0, 132.9, 131.5, 130.8, 129.2, 40.4, 29.1, 29.0, 25.3, 25.0, 25.0.

HRMS (ESI): m/z calculated for $[C_{14}H_{20}NO_2]^+$ $[M]^+$: 234.1489, found: 234.1488.

2-(aminomethyl)-5-chlorobenzoic acid hydrochloride (23)



Chemical Formula: $C_8H_9Cl_2NO_2$

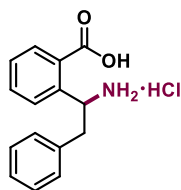
Product **23** was obtained as a white solid (32.0 mg, 72%)

1H NMR (600 MHz, D_2O) δ 8.03 (d, $J = 2.3$ Hz, 1H), 7.62 (dd, $J = 8.2, 2.3$ Hz, 1H), 7.45 (d, $J = 8.2$ Hz, 1H), 4.32 (s, 2H).

^{13}C NMR (151 MHz, D_2O) δ 169.2, 135.2, 133.7, 133.1, 132.0, 131.8, 131.4, 41.9.

HRMS (ESI): m/z calculated for $[C_8H_9ClNO_2]^+$ $[M]^+$: 186.0316, found: 186.0317.

2-(1-amino-2-phenylethyl)benzoic acid hydrochloride (24)



Chemical Formula: $C_{15}H_{16}ClNO_2$

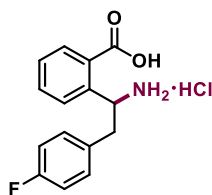
Product **24** was obtained as a white solid (48.1 mg, 87%)

1H NMR (400 MHz, D_2O) δ 7.85 (dd, $J = 7.9, 1.4$ Hz, 1H), 7.70 – 7.56 (m, 2H), 7.47 (ddd, $J = 7.7, 7.0, 1.6$ Hz, 1H), 7.32 – 7.20 (m, 3H), 7.18 – 7.08 (m, 2H), 5.40 (t, $J = 7.8$ Hz, 1H), 3.28 (qd, $J = 13.8, 7.8$ Hz, 2H).

^{13}C NMR (101 MHz, D_2O) δ 170.9, 135.9, 135.4, 133.2, 131.1, 130.2, 129.3, 129.2, 128.8, 127.6, 127.4, 53.1, 39.3.

HRMS (ESI): m/z calculated for $[C_{15}H_{16}NO_2]^+$ $[M]^+$: 242.1176, found: 242.1173.

2-(1-amino-2-(4-fluorophenyl)ethyl)benzoic acid hydrochloride (25)



Chemical Formula: $C_{15}H_{15}ClFNO_2$

Product **25** was obtained as a white solid (41.6 mg, 70%)

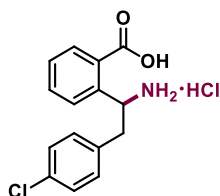
1H NMR (400 MHz, D_2O) δ 7.84 (dd, $J = 7.9, 1.4$ Hz, 1H), 7.71 – 7.53 (m, 2H), 7.48 (td, $J = 7.5, 1.4$ Hz, 1H), 7.16 – 7.04 (m, 2H), 7.03 – 6.89 (m, 2H), 5.37 (t, $J = 7.8$ Hz, 1H), 3.36 – 3.14 (m, 2H).

^{13}C NMR (101 MHz, D_2O) δ 170.9, 161.8 (d, $J = 244.42$ Hz), 135.7, 133.2 (d, $J = 4.04$ Hz), 131.2, 131.1 (d, $J = 3.03$ Hz), 131.0, 130.4, 129.2, 127.5, 115.4 (d, $J = 22.22$ Hz), 53.1, 38.6.

^{19}F NMR (376 MHz, D_2O) δ -116.1.

HRMS (ESI): m/z calculated for $[C_{15}H_{15}FNO_2]^+ [M]^+$: 260.1081, found: 260.1078.

2-(1-amino-2-(4-chlorophenyl)ethyl)benzoic acid hydrochloride (26)



Chemical Formula: $C_{15}H_{15}Cl_2NO_2$

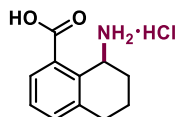
Product **26** was obtained as a white solid (43.4 mg, 70%)

1H NMR (400 MHz, D_2O) δ 7.74 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.59 – 7.45 (m, 2H), 7.38 (td, $J = 7.6, 1.4$ Hz, 1H), 7.18 – 7.08 (m, 2H), 7.01 – 6.89 (m, 2H), 5.34 – 5.19 (m, 1H), 3.23 (dd, $J = 13.7, 6.9$ Hz, 1H), 3.13 (dd, $J = 13.7, 8.9$ Hz, 1H).

^{13}C NMR (101 MHz, D_2O) δ 171.1, 135.5, 134.0, 133.1, 132.5, 131.0, 130.8, 130.7, 129.3, 128.6, 127.7, 52.9, 38.6.

HRMS (ESI): m/z calculated for $[C_{15}H_{15}ClNO_2]^+ [M]^+$: 276.0786, found: 276.0783.

8-amino-5,6,7,8-tetrahydronaphthalene-1-carboxylic acid hydrochloride (27)



Chemical Formula: $C_{11}H_{14}ClNO_2$

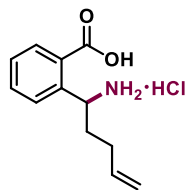
Product **27** was obtained as a white solid (44.9 mg, 99%)

1H NMR (400 MHz, D_2O) δ 7.83 (dd, $J = 7.3, 1.8$ Hz, 1H), 7.49 – 7.37 (m, 2H), 4.93 (t, $J = 3.4$ Hz, 1H), 2.98 (m, 1H), 2.84 (ddd, $J = 17.8, 11.8, 6.4$ Hz, 1H), 2.22 (m, 1H), 2.03 – 1.90 (m, 2H), 1.90 – 1.74 (m, 1H).

^{13}C NMR (101 MHz, D_2O) δ 171.3, 139.4, 134.9, 130.8, 130.4, 129.4, 129.0, 46.4, 28.4, 26.5, 15.7.

HRMS (ESI): m/z calculated for $[C_{11}H_{14}NO_2]^+ [M]^+$: 192.1019, found: 192.1018.

2-(1-aminopent-4-en-1-yl)benzoic acid hydrochloride (28)



Chemical Formula: C₁₂H₁₆ClNO₂

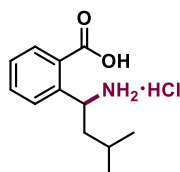
Product **28** was obtained as a white solid (43.9 mg, 91%)

¹H NMR (400 MHz, D₂O) δ 7.96 (dd, *J* = 7.9, 1.4 Hz, 1H), 7.70 (td, *J* = 7.6, 1.4 Hz, 1H), 7.63 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.55 (td, *J* = 7.6, 1.4 Hz, 1H), 5.17 – 5.04 (m, 1H), 2.00 (ddd, *J* = 13.8, 8.9, 6.4 Hz, 1H), 1.84 (ddd, *J* = 14.1, 7.8, 6.6 Hz, 1H), 1.52 – 1.36 (m, 1H), 0.87 (dd, *J* = 8.6, 6.6 Hz, 6H).

¹³C NMR (101 MHz, D₂O) δ 171.4, 136.3, 133.2, 131.1, 130.7, 129.4, 128.0, 50.2, 41.4, 24.3, 21.7, 21.1.

HRMS (ESI): *m/z* calculated for [C₁₂H₁₆NO₂]⁺ [M]⁺: 206.1176, found: 206.1176.

2-(1-amino-3-methylbutyl)benzoic acid hydrochloride (29)



Chemical Formula: C₁₂H₁₈ClNO₂

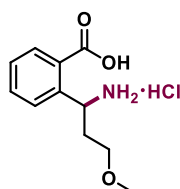
Product **29** was obtained as a white solid (45.5 mg, 93%)

¹H NMR (400 MHz, D₂O) δ 7.94 (dd, *J* = 7.9, 1.4 Hz, 1H), 7.67 (td, *J* = 7.6, 1.5 Hz, 1H), 7.63 – 7.58 (m, 1H), 7.53 (td, *J* = 7.5, 1.4 Hz, 1H), 5.05 (dd, *J* = 8.9, 6.6 Hz, 1H), 1.98 (m, 1H), 1.82 (m, 1H), 1.47 – 1.36 (m, 1H), 0.85 (dd, *J* = 8.5, 6.6 Hz, 6H).

¹³C NMR (101 MHz, D₂O) δ 171.6, 136.1, 133.0, 131.0, 130.9, 129.3, 127.9, 50.2, 41.3, 24.2, 21.6, 21.0.

HRMS (ESI): *m/z* calculated for [C₁₂H₁₈NO₂]⁺ [M]⁺: 208.1332, found: 208.1331.

2-(1-amino-3-methoxypropyl)benzoic acid hydrochloride (30)



Chemical Formula: C₁₁H₁₆ClNO₃

Product **30** was obtained as a white solid (48.3 mg, 98%)

¹H NMR (400 MHz, D₂O) δ 7.97 (dd, *J* = 7.8, 1.5 Hz, 1H), 7.69 (td, *J* = 7.7, 1.5 Hz, 1H), 7.63 – 7.58 (m, 1H), 7.54 (td, *J* = 7.6, 1.4 Hz, 1H), 5.18 (t, *J* = 7.4 Hz, 1H), 3.48 (m, 1H), 3.32 (m, 1H),

3.23 (s, 3H), 2.42 – 2.14 (m, 2H).

^{13}C NMR (101 MHz, D_2O) δ 171.0, 135.8, 133.3, 131.3, 130.3, 129.5, 127.8, 68.7, 58.0, 50.2, 32.2.

HRMS (ESI): m/z calculated for $[\text{C}_{11}\text{H}_{16}\text{NO}_3]^+ [\text{M}]^+$: 210.1125, found: 210.1124.

2-(1-amino-2-cyclohexylethyl)benzoic acid hydrochloride (31)



Chemical Formula: $\text{C}_{15}\text{H}_{22}\text{ClNO}_2$

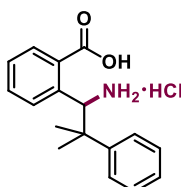
Product **31** was obtained as a white solid (45.6 mg, 80%)

^1H NMR (600 MHz, D_2O) δ 7.95 (dd, $J = 7.9, 1.4$ Hz, 1H), 7.70 (td, $J = 7.7, 1.4$ Hz, 1H), 7.66 – 7.61 (m, 1H), 7.55 (td, $J = 7.6, 1.3$ Hz, 1H), 5.12 (t, $J = 7.8$ Hz, 1H), 1.99 (ddd, $J = 13.7, 9.3, 5.8$ Hz, 1H), 1.86 (ddd, $J = 14.0, 7.9, 6.3$ Hz, 1H), 1.71 – 1.48 (m, 5H), 1.19 – 1.01 (m, 4H), 1.01 – 0.80 (m, 2H).

^{13}C NMR (151 MHz, D_2O) δ 171.6, 136.2, 133.1, 131.0, 131.0, 129.3, 127.9, 49.7, 40.1, 33.6, 32.7, 32.1, 25.8, 25.5, 25.4.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{22}\text{NO}_2]^+ [\text{M}]^+$: 248.1645, found: 248.1644.

2-(1-amino-2-methyl-2-phenylpropyl)benzoic acid hydrochloride (32)



Chemical Formula: $\text{C}_{17}\text{H}_{20}\text{ClNO}_2$

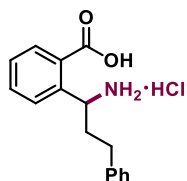
Product **32** was obtained as a white solid (53.8 mg, 88%)

^1H NMR (600 MHz, DMSO-d_6) δ 8.92 (s, 1H), 7.58 (d, $J = 7.5$ Hz, 1H), 7.43 (d, $J = 7.4$ Hz, 2H), 7.37 (q, $J = 7.6$ Hz, 3H), 7.31 (t, $J = 7.2$ Hz, 1H), 7.28 – 7.21 (m, 1H), 6.02 (d, $J = 7.7$ Hz, 1H), 4.97 (s, 1H), 3.36 (d, $J = 1.5$ Hz, 1H), 1.48 (s, 3H), 0.92 (s, 3H).

^{13}C NMR (151 MHz, DMSO-d_6) δ 169.8, 146.0, 145.2, 133.2, 130.6, 128.2, 127.9, 126.6, 126.6, 123.3, 122.5, 64.8, 41.5, 28.9, 19.7.

HRMS (ESI): m/z calculated for $[\text{C}_{17}\text{H}_{20}\text{NO}_2]^+ [\text{M}]^+$: 270.1489, found: 270.1486.

2-(1-amino-3-phenylpropyl)benzoic acid hydrochloride (33)



Chemical Formula: $C_{16}H_{18}ClNO_2$

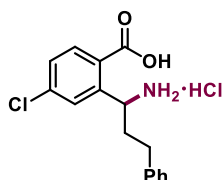
Product **33** was obtained as a white solid (52.7 mg, 91%)

1H NMR (400 MHz, D_2O) δ 7.92 (dd, $J = 7.8, 1.4$ Hz, 1H), 7.65 (td, $J = 7.6, 1.4$ Hz, 1H), 7.58 (dd, $J = 8.0, 1.3$ Hz, 1H), 7.50 (td, $J = 7.6, 1.3$ Hz, 1H), 7.26 – 7.09 (m, 3H), 7.03 – 6.91 (m, 2H), 4.87 (dd, $J = 8.7, 5.9$ Hz, 1H), 2.52 – 2.21 (m, 4H).

^{13}C NMR (101 MHz, D_2O) δ 170.8, 140.0, 135.4, 133.4, 131.3, 130.8, 129.5, 128.6, 128.3, 128.1, 126.4, 51.1, 33.4, 31.1.

HRMS (ESI): m/z calculated for $[C_{16}H_{18}NO_2]^+ [M]^+$: 256.1332, found: 256.1329.

2-(1-amino-3-phenylpropyl)-4-chlorobenzoic acid hydrochloride (34)



Chemical Formula: $C_{16}H_{17}Cl_2NO_2$

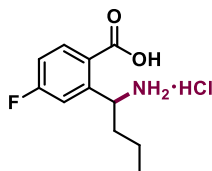
Product **34** was obtained as a white solid (47.0 mg, 72%)

1H NMR (600 MHz, D_2O) δ 7.91 (d, $J = 2.3$ Hz, 1H), 7.62 (dd, $J = 8.4, 2.3$ Hz, 1H), 7.53 (d, $J = 8.5$ Hz, 1H), 7.24 – 7.11 (m, 3H), 7.02 – 6.95 (m, 2H), 4.83 (dd, $J = 9.9, 5.4$ Hz, 1H), 2.55 (ddd, $J = 13.5, 8.3, 5.2$ Hz, 1H), 2.46 – 2.39 (m, 1H), 2.39 – 2.33 (m, 1H), 2.33 – 2.24 (m, 1H).

^{13}C NMR (151 MHz, D_2O) δ 169.4, 140.0, 134.9, 134.0, 133.1, 132.5, 131.1, 129.9, 128.7, 128.4, 126.4, 50.4, 33.1, 30.9.

HRMS (ESI): m/z calculated for $[C_{16}H_{17}ClNO_2]^+ [M]^+$: 290.0942, found: 290.0939.

2-(1-aminobutyl)-4-fluorobenzoic acid hydrochloride (35)



Chemical Formula: $C_{11}H_{15}ClFNO_2$

Product **35** was obtained as a white solid (41.4 mg, 85%)

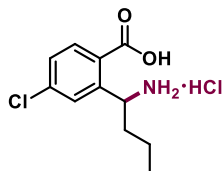
1H NMR (600 MHz, D_2O) δ 8.05 (dd, $J = 8.8, 5.9$ Hz, 1H), 7.38 (dd, $J = 10.0, 2.6$ Hz, 1H), 7.27 (ddd, $J = 8.7, 8.0, 2.6$ Hz, 1H), 5.07 (t, $J = 7.6$ Hz, 1H), 2.12 – 1.91 (m, 2H), 1.45 – 1.09 (m, 2H), 0.89 (t, $J = 7.4$ Hz, 3H).

^{13}C NMR (151 MHz, D_2O) δ 170.3, 165.7, 164.1, 139.7 (d, $J = 9.06$ Hz), 134.2 (d, $J = 9.06$ Hz), 127.0 (d, $J = 3.02$ Hz), 116.2 (d, $J = 21.14$ Hz), 115.2 (d, $J = 24.16$ Hz), 51.3, 34.3, 18.5, 12.7.

^{19}F NMR (565 MHz, D_2O) δ -105.5.

HRMS (ESI): m/z calculated for $[\text{C}_{11}\text{H}_{15}\text{FNO}_2]^+ [\text{M}]^+$: 212.1081, found: 212.1080.

2-(1-aminobutyl)-4-chlorobenzoic acid hydrochloride (36)



Chemical Formula: $\text{C}_{11}\text{H}_{15}\text{Cl}_2\text{NO}_2$

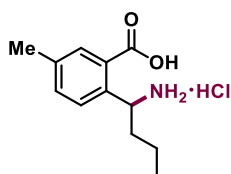
Product **36** was obtained as a white solid (46.2 mg, 87%)

^1H NMR (600 MHz, D_2O) δ 7.94 (d, J = 8.4 Hz, 1H), 7.65 (d, J = 2.1 Hz, 1H), 7.56 (dd, J = 8.4, 2.1 Hz, 1H), 5.00 (t, J = 7.6 Hz, 1H), 2.14 – 1.88 (m, 2H), 1.42 – 1.16 (m, 2H), 0.89 (t, J = 7.4 Hz, 3H).

^{13}C NMR (151 MHz, D_2O) δ 170.7, 138.5, 138.1, 132.7, 129.9, 129.4, 128.1, 51.5, 34.3, 18.5, 12.7.

HRMS (ESI): m/z calculated for $[\text{C}_{11}\text{H}_{15}\text{ClNO}_2]^+ [\text{M}]^+$: 228.0786, found: 228.0784.

2-(1-aminobutyl)-5-methylbenzoic acid hydrochloride (37)



Chemical Formula: $\text{C}_{12}\text{H}_{18}\text{ClNO}_2$

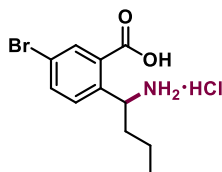
Product **37** was obtained as a white solid (32.7 mg, 67%)

^1H NMR (400 MHz, D_2O) δ 7.72 (d, J = 1.7 Hz, 1H), 7.47 – 7.37 (m, 2H), 4.86 (dd, J = 8.6, 6.8 Hz, 1H), 2.30 (s, 3H), 2.01 – 1.83 (m, 2H), 1.28 – 1.06 (m, 2H), 0.80 (td, J = 7.4, 3.6 Hz, 3H).

^{13}C NMR (101 MHz, D_2O) δ 171.6, 139.8, 133.5, 133.0, 131.4, 130.7, 127.7, 51.4, 34.4, 19.9, 18.5, 12.6.

HRMS (ESI): m/z calculated for $[\text{C}_{12}\text{H}_{18}\text{NO}_2]^+ [\text{M}]^+$: 208.1332, found: 208.1330.

2-(1-aminobutyl)-5-bromobenzoic acid hydrochloride (38)



Chemical Formula: $\text{C}_{11}\text{H}_{15}\text{BrClNO}_2$

Product **38** was obtained as a white solid (45.9 mg, 74%)

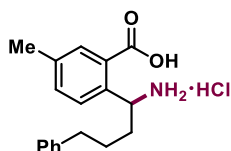
^1H NMR (600 MHz, D_2O) δ 8.10 (d, J = 2.2 Hz, 1H), 7.82 (dd, J = 8.4, 2.2 Hz, 1H), 7.50 (d, J =

8.5 Hz, 1H), 4.97 (t, $J = 7.6$ Hz, 1H), 2.11 – 1.89 (m, 2H), 1.35 – 1.15 (m, 2H), 0.88 (t, $J = 7.4$ Hz, 3H).

^{13}C NMR (151 MHz, D_2O) δ 170.0, 135.7, 135.3, 133.8, 133.0, 129.6, 122.6, 51.3, 34.4, 18.6, 12.7.

HRMS (ESI): m/z calculated for $[\text{C}_{11}\text{H}_{15}\text{BrNO}_2]^+ [\text{M}]^+$: 272.0281, found: 272.0277.

2-(1-amino-4-phenylbutyl)-5-methylbenzoic acid hydrochloride (39)



Chemical Formula: $\text{C}_{18}\text{H}_{22}\text{ClNO}_2$

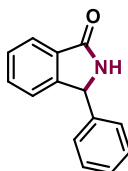
Product **39** was obtained as a white solid (49.0 mg, 77%)

^1H NMR (600 MHz, D_2O) δ 7.63 (d, $J = 1.9$ Hz, 1H), 7.31 (d, $J = 8.0$ Hz, 1H), 7.24 – 7.20 (m, 1H), 6.94 (ddd, $J = 8.6, 6.9, 2.1$ Hz, 2H), 6.88 – 6.80 (m, 3H), 4.87 (t, $J = 6.9$ Hz, 1H), 2.42 (ddd, $J = 14.5, 8.9, 6.0$ Hz, 1H), 2.36 – 2.29 (m, 1H), 2.08 (d, $J = 1.9$ Hz, 3H), 2.00 (p, $J = 7.4$ Hz, 3H), 1.43 – 1.34 (m, 1H), 1.26 – 1.18 (m, 1H).

^{13}C NMR (151 MHz, D_2O) δ 171.0, 141.6, 139.6, 139.5, 133.6, 132.9, 131.9, 130.6, 128.2, 128.2, 125.7, 51.7, 34.4, 31.6, 27.1, 20.1.

HRMS (ESI): m/z calculated for $[\text{C}_{18}\text{H}_{22}\text{NO}_2]^+ [\text{M}]^+$: 284.1645, found: 284.1643.

3-phenylisoindolin-1-one (40)



Chemical Formula: $\text{C}_{14}\text{H}_{11}\text{NO}$

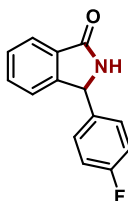
Product **40** was obtained as a white solid (29.7 mg, 71%)

^1H NMR (400 MHz, CDCl_3) δ 7.99 – 7.78 (m, 1H), 7.55 – 7.43 (m, 2H), 7.40 – 7.20 (m, 6H), 6.77 (s, 1H), 5.63 (s, 1H).

^{13}C NMR (101 MHz, CDCl_3) δ 171.0, 148.0, 138.4, 132.3, 130.8, 129.1, 128.6, 128.4, 126.8, 123.9, 123.3, 60.8.

HRMS (ESI): m/z calculated for $[\text{C}_{14}\text{H}_{12}\text{NO}]^+ [\text{M} + \text{H}]^+$: 210.0913, found: 210.0913.

3-(4-fluorophenyl)isoindolin-1-one (41)



Chemical Formula: $\text{C}_{14}\text{H}_{10}\text{FNO}$

Product **41** was obtained as a white solid (34.6 mg, 76%)

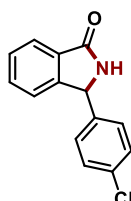
¹H NMR (400 MHz, CDCl₃) δ 7.85 (d, *J* = 7.6 Hz, 2H), 7.60 – 7.38 (m, 2H), 7.32 – 7.16 (m, 3H), 7.02 (t, *J* = 8.6 Hz, 2H), 5.62 (s, 1H).

¹³C NMR (101 MHz, CDCl₃) δ 171.5, 163.9, 161.5, 147.9, 134.2 (d, *J* = 4.04 Hz), 132.4, 130.9, 128.6, 128.5 (d, *J* = 5.05 Hz), 123.5 (d, *J* = 55.55 Hz), 116.0 (d, *J* = 22.22 Hz), 60.2.

¹⁹F NMR (376 MHz, CDCl₃) δ -113.4.

HRMS (ESI): *m/z* calculated for [C₁₄H₁₁FNO]⁺ [M + H]⁺: 250.0639, found: 250.0639.

3-(4-chlorophenyl)isoindolin-1-one (42)



Chemical Formula: C₁₄H₁₀ClNO

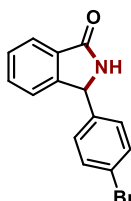
Product **42** was obtained as a white solid (40.8 mg, 84%)

¹H NMR (400 MHz, DMSO-*d*₆) δ 9.11 (s, 1H), 7.72 (d, *J* = 7.4 Hz, 1H), 7.57 – 7.45 (m, 2H), 7.42 (d, *J* = 8.2 Hz, 2H), 7.31 (dd, *J* = 11.2, 7.8 Hz, 3H), 5.77 (s, 1H).

¹³C NMR (101 MHz, DMSO-*d*₆) δ 169.7, 147.8, 138.7, 132.5, 132.0, 131.3, 128.8, 128.5, 128.3, 123.5, 123.0, 58.8.

HRMS (ESI): *m/z* calculated for [C₁₄H₁₁ClNO]⁺ [M + H]⁺: 244.0523, found: 244.0532.

3-(4-bromophenyl)isoindolin-1-one (43)



Chemical Formula: C₁₄H₁₀BrNO

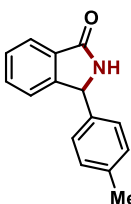
Product **43** was obtained as a white solid (52.1 mg, 89%)

¹H NMR (600 MHz, DMSO-*d*₆) δ 9.10 (s, 1H), 7.71 (d, *J* = 7.5 Hz, 1H), 7.63 – 7.43 (m, 4H), 7.27 (dd, *J* = 21.0, 7.8 Hz, 3H), 5.75 (s, 1H).

¹³C NMR (151 MHz, DMSO-*d*₆) δ 169.7, 147.7, 139.2, 132.0, 131.7, 131.3, 128.8, 128.3, 123.5, 123.0, 121.0, 58.8.

HRMS (ESI): *m/z* calculated for [C₁₄H₁₁BrNO]⁺ [M + H]⁺: 288.0019, found: 288.0023.

3-(*p*-tolyl)isoindolin-1-one (44)



Chemical Formula: C₁₅H₁₃NO

Product **44** was obtained as a white solid (30.0 mg, 67%)

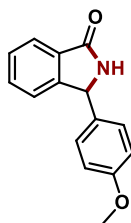
¹H NMR (600 MHz, CDCl₃) δ 7.86 (d, *J* = 7.4 Hz, 1H), 7.51 – 7.42 (m, 2H), 7.32 (s, 1H), 7.23 (d,

$J = 7.5$ Hz, 1H), 7.15 (d, $J = 1.8$ Hz, 4H), 5.60 (s, 1H), 2.33 (s, 3H).

^{13}C NMR (151 MHz, CDCl_3) δ 171.4, 148.2, 138.4, 135.3, 132.3, 130.9, 129.7, 128.3, 126.8, 123.8, 123.3, 60.7, 21.2.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{14}\text{NO}]^+ [\text{M} + \text{H}]^+$: 224.1070, found: 224.1073.

3-(4-methoxyphenyl)isoindolin-1-one (45)



Chemical Formula: $\text{C}_{15}\text{H}_{13}\text{NO}_2$

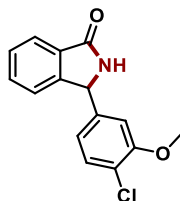
Product **45** was obtained as a white solid (28.5 mg, 60%)

^1H NMR (400 MHz, CDCl_3) δ 7.92 – 7.82 (m, 1H), 7.55 – 7.41 (m, 2H), 7.24 – 7.12 (m, 3H), 6.92 – 6.80 (m, 3H), 5.58 (s, 1H), 3.79 (s, 3H).

^{13}C NMR (101 MHz, CDCl_3) δ 171.0, 159.8, 148.3, 132.3, 130.9, 130.2, 128.3, 128.1, 123.7, 123.3, 114.4, 60.3, 55.4.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{14}\text{NO}]^+ [\text{M} + \text{H}]^+$: 240.1019, found: 240.1023.

3-(4-chloro-3-methoxyphenyl)isoindolin-1-one (46)



Chemical Formula: $\text{C}_{15}\text{H}_{12}\text{ClNO}_2$

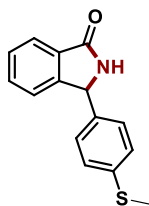
Product **46** was obtained as a white solid (45.6 mg, 83%)

^1H NMR (600 MHz, CDCl_3) δ 7.88 (d, $J = 7.5$ Hz, 1H), 7.51 (dt, $J = 23.4, 7.5$ Hz, 2H), 7.35 (dd, $J = 8.1, 1.7$ Hz, 1H), 7.23 (d, $J = 7.5$ Hz, 1H), 7.04 (s, 1H), 6.88 (d, $J = 8.1$ Hz, 1H), 6.75 (s, 1H), 5.60 (s, 1H), 3.82 (d, $J = 1.7$ Hz, 3H).

^{13}C NMR (151 MHz, CDCl_3) δ 171.2, 155.6, 147.5, 138.5, 132.5, 130.7, 130.6, 128.7, 124.0, 123.2, 122.7, 119.8, 109.8, 60.5, 56.2.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{13}\text{ClNO}_2]^+ [\text{M} + \text{H}]^+$: 274.0629, found: 274.0637.

3-(4-(methylthio)phenyl)isoindolin-1-one (47)



Chemical Formula: $\text{C}_{15}\text{H}_{13}\text{NOS}$

Product **47** was obtained as a white solid (32.4 mg, 63%)

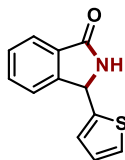
^1H NMR (400 MHz, DMSO-d_6) δ 9.07 (s, 1H), 7.70 (dd, $J = 7.3, 1.3$ Hz, 1H), 7.57 – 7.44 (m,

2H), 7.30 – 7.18 (m, 5H), 5.70 (s, 1H), 2.45 (s, 3H).

^{13}C NMR (101 MHz, DMSO- d_6) δ 169.7, 148.2, 137.8, 136.2, 131.9, 131.4, 128.2, 127.2, 126.2, 123.5, 122.9, 59.1, 14.7.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{14}\text{NOS}]^+ [\text{M} + \text{H}]^+$: 256.0791, found: 256.0798.

3-(thiophen-2-yl)isoindolin-1-one (48)



Chemical Formula: $\text{C}_{12}\text{H}_9\text{NOS}$

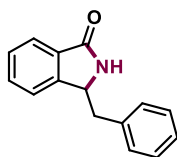
Product **48** was obtained as a white solid (20.5 mg, 47%)

^1H NMR (400 MHz, CDCl_3) δ 7.89 (dt, $J = 7.3, 1.0$ Hz, 1H), 7.61 – 7.47 (m, 2H), 7.38 (dq, $J = 7.6, 1.0$ Hz, 1H), 7.31 – 7.24 (m, 1H), 7.14 (dt, $J = 3.5, 0.9$ Hz, 1H), 7.00 (dd, $J = 5.1, 3.5$ Hz, 1H), 6.91 (s, 1H), 5.95 (s, 1H).

^{13}C NMR (101 MHz, CDCl_3) δ 170.4, 147.1, 141.8, 132.5, 130.6, 128.8, 127.1, 125.9, 125.8, 123.9, 123.4, 56.2.

HRMS (ESI): m/z calculated for $[\text{C}_{12}\text{H}_{10}\text{NOS}]^+ [\text{M} + \text{H}]^+$: 216.0478, found: 216.0486.

3-benzylisoindolin-1-one (49)



Chemical Formula: $\text{C}_{15}\text{H}_{13}\text{NO}$

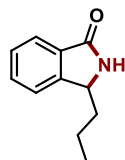
Product **49** was obtained as a white solid (34.4 mg, 77%)

^1H NMR (600 MHz, CDCl_3) δ 7.84 (d, $J = 7.5$ Hz, 1H), 7.58 – 7.51 (m, 1H), 7.47 (t, $J = 7.5$ Hz, 1H), 7.37 – 7.30 (m, 3H), 7.30 – 7.25 (m, 1H), 7.25 – 7.21 (m, 2H), 6.79 (s, 1H), 4.81 (dd, $J = 9.0, 5.4$ Hz, 1H), 3.22 (dd, $J = 13.6, 5.3$ Hz, 1H), 2.83 (dd, $J = 13.6, 8.9$ Hz, 1H).

^{13}C NMR (151 MHz, CDCl_3) δ 170.4, 146.8, 136.9, 131.9, 131.7, 129.2, 128.8, 128.3, 127.1, 123.9, 122.7, 58.0, 41.3.

HRMS (ESI): m/z calculated for $[\text{C}_{15}\text{H}_{14}\text{NO}]^+ [\text{M} + \text{H}]^+$: 224.1070, found: 224.1070.

3-propylisoindolin-1-one (50)



Chemical Formula: $\text{C}_{11}\text{H}_{13}\text{NO}$

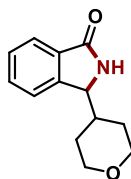
Product **50** was obtained as a white solid (19.6 mg, 56%)

^1H NMR (600 MHz, CDCl_3) δ 7.84 (d, $J = 7.2$ Hz, 1H), 7.56 (t, $J = 7.5$ Hz, 1H), 7.49 – 7.40 (m, 2H), 6.91 (s, 1H), 4.62 (dd, $J = 8.2, 4.3$ Hz, 1H), 1.93 (m, 1H), 1.62 (m, 1H), 1.56 – 1.43 (m, 1H), 1.39 (m, 1H), 0.97 (m, 3H).

^{13}C NMR (151 MHz, CDCl_3) δ 147.8, 131.8, 128.1, 123.8, 122.4, 56.8, 36.8, 19.1, 14.1.

HRMS (ESI): m/z calculated for $[C_{11}H_{14}NO]^+$ $[M + H]^+$: 176.1070, found: 176.1075.

3-(tetrahydro-2H-pyran-4-yl)isoindolin-1-one (51)



Chemical Formula: $C_{13}H_{15}NO_2$

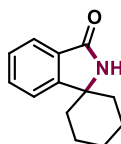
Product **51** was obtained as a white solid (32.6 mg, 75%)

1H NMR (400 MHz, $CDCl_3$) δ 7.86 (d, $J = 7.5$ Hz, 1H), 7.57 (td, $J = 7.5, 1.2$ Hz, 1H), 7.53 – 7.38 (m, 3H), 4.55 (d, $J = 4.1$ Hz, 1H), 4.11 – 3.87 (m, 2H), 3.37 (m, 2H), 2.10 (m, 1H), 1.83 – 1.70 (m, 1H), 1.66 – 1.40 (m, 2H), 1.18 (m, 1H).

^{13}C NMR (101 MHz, $CDCl_3$) δ 171.3, 145.6, 132.5, 131.9, 128.4, 124.0, 122.9, 67.8, 67.7, 61.0, 39.4, 29.9, 26.8.

HRMS (ESI): m/z calculated for $[C_{13}H_{16}NO_2]^+$ $[M + H]^+$: 218.1176, found: 218.1183.

spiro[cyclohexane-1,1'-isoindolin]-3'-one (52)



Chemical Formula: $C_{13}H_{15}NO$

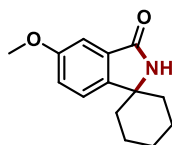
Product **52** was obtained as a white solid (25.2 mg, 63%)

1H NMR (600 MHz, $CDCl_3$) δ 7.83 (dt, $J = 7.5, 1.0$ Hz, 1H), 7.63 (s, 1H), 7.55 (td, $J = 7.5, 1.2$ Hz, 1H), 7.48 – 7.36 (m, 2H), 1.93 – 1.81 (m, 6H), 1.62 – 1.54 (m, 4H), 1.48 – 1.35 (m, 1H).

^{13}C NMR (151 MHz, $CDCl_3$) δ 170.1, 153.0, 131.9, 130.9, 128.1, 124.0, 121.3, 61.7, 36.7, 25.0, 23.4.

HRMS (ESI): m/z calculated for $[C_{13}H_{16}NO]^+$ $[M + H]^+$: 202.1226, found: 202.1226.

5'-methoxyspiro[cyclohexane-1,1'-isoindolin]-3'-one (53)



Chemical Formula: $C_{14}H_{17}NO_2$

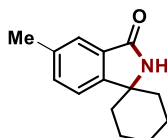
Product **53** was obtained as a white solid (36.1 mg, 78%)

1H NMR (600 MHz, $CDCl_3$) δ 7.61 (s, 1H), 7.33 – 7.27 (m, 2H), 7.10 (dd, $J = 8.4, 2.5$ Hz, 1H), 3.85 (s, 3H), 1.92 – 1.78 (m, 5H), 1.62 – 1.50 (m, 4H), 1.40 (m, 1H).

^{13}C NMR (151 MHz, $CDCl_3$) δ 170.0, 159.9, 145.5, 132.1, 122.2, 120.2, 106.5, 61.4, 55.7, 36.8, 25.0, 23.5.

HRMS (ESI): m/z calculated for $[C_{14}H_{18}NO_2]^+$ $[M + H]^+$: 232.1332, found: 232.1336.

5'-methylspiro[cyclohexane-1,1'-isoindolin]-3'-one (54)



Chemical Formula: C₁₄H₁₇NO

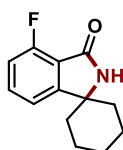
Product **54** was obtained as a white solid (31.8 mg, 74%)

¹H NMR (600 MHz, CDCl₃) δ 7.64 (d, *J* = 9.1 Hz, 2H), 7.36 (dd, *J* = 7.8, 1.6 Hz, 1H), 7.29 (d, *J* = 7.7 Hz, 1H), 2.43 (s, 3H), 1.91 – 1.80 (m, 5H), 1.62 – 1.51 (m, 4H), 1.46 – 1.35 (m, 1H).

¹³C NMR (151 MHz, CDCl₃) δ 170.2, 150.4, 138.0, 132.8, 131.0, 124.2, 121.1, 61.5, 36.7, 25.0, 23.4, 21.3.

HRMS (ESI): *m/z* calculated for [C₁₄H₁₈NO]⁺ [M + H]⁺: 216.1383, found: 216.1389.

4'-fluorospiro[cyclohexane-1,1'-isoindolin]-3'-one (**55**)



Chemical Formula: C₁₃H₁₄FNO

Product **55** was obtained as a white solid (30.1 mg, 69%)

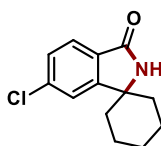
¹H NMR (600 MHz, CDCl₃) δ 7.51 (td, *J* = 7.8, 4.2 Hz, 1H), 7.40 (s, 1H), 7.17 (d, *J* = 7.6 Hz, 1H), 7.06 (t, *J* = 8.8 Hz, 1H), 1.93 – 1.81 (m, 5H), 1.65 – 1.47 (m, 4H), 1.46 – 1.33 (m, 1H).

¹³C NMR (151 MHz, CDCl₃) δ 166.9, 160.0, 158.3, 155.9 (d, *J* = 3.02 Hz), 134.0 (d, *J* = 7.55 Hz), 118.0 (d, *J* = 13.59 Hz), 117.3 (d, *J* = 4.53 Hz), 115.3 (d, *J* = 10.57 Hz), 61.6, 36.7, 24.9, 23.3.

¹⁹F NMR (376 MHz, CDCl₃) δ -114.2.

HRMS (ESI): *m/z* calculated for [C₁₃H₁₅FNO]⁺ [M + H]⁺: 220.1132, found: 220.1142.

6'-chlorospiro[cyclohexane-1,1'-isoindolin]-3'-one (**56**)



Chemical Formula: C₁₃H₁₄ClNO

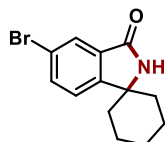
Product **56** was obtained as a white solid (31.3 mg, 66%)

¹H NMR (600 MHz, CDCl₃) δ 8.08 (s, 1H), 7.75 (d, *J* = 8.0 Hz, 1H), 7.50 – 7.32 (m, 2H), 2.01 – 1.76 (m, 5H), 1.72 – 1.51 (m, 4H), 1.40 (m, 1H).

¹³C NMR (151 MHz, CDCl₃) δ 169.1, 154.6, 138.2, 129.5, 128.6, 125.3, 122.0, 61.7, 36.6, 24.9, 23.2.

HRMS (ESI): *m/z* calculated for [C₁₃H₁₅ClNO]⁺ [M + H]⁺: 236.0837, found: 236.0842.

5'-bromospiro[cyclohexane-1,1'-isoindolin]-3'-one (**57**)



Chemical Formula: C₁₃H₁₄BrNO

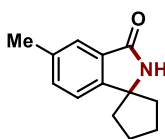
Product **57** was obtained as a white solid (36.2 mg, 65%)

¹H NMR (600 MHz, CDCl₃) δ 8.34 (s, 1H), 7.96 (d, *J* = 2.1 Hz, 1H), 7.65 (dd, *J* = 8.0, 2.1 Hz, 1H), 7.29 (dd, *J* = 8.1, 1.9 Hz, 1H), 1.86 m, 5H), 1.61 (m, 4H), 1.45 – 1.35 (m, 1H).

¹³C NMR (151 MHz, CDCl₃) δ 168.7, 151.7, 134.8, 133.1, 127.1, 123.0, 121.9, 62.0, 36.6, 25.0, 23.2.

HRMS (ESI): *m/z* calculated for [C₁₃H₁₃BrNO]⁺ [M + H]⁺: 280.0332, found: 280.0339.

5'-methylspiro[cyclopentane-1,1'-isoindolin]-3'-one (**58**)



Chemical Formula: C₁₃H₁₅NO

Product **58** was obtained as a white solid (35.0 mg, 87%)

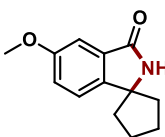
¹H NMR (400 MHz, CDCl₃) δ 7.62 (s, 1H), 7.38 (dd, *J* = 7.8, 1.6 Hz, 1H), 7.31 – 7.25 (m, 1H), 7.08 (s, 1H), 2.44 (s, 3H), 2.15 – 1.87 (m, 8H).

¹³C NMR (151 MHz, CDCl₃) δ 148.8, 137.9, 133.1, 131.4, 123.9, 120.9, 69.2, 39.2, 24.8, 21.3.

HRMS (ESI, *m/z*): [M+H]⁺ Calcd for ; Found.]

HRMS (ESI): *m/z* calculated for [C₁₃H₁₆NO]⁺ [M + H]⁺: 202.1226, found: 202.1233.

5'-methoxyspiro[cyclopentane-1,1'-isoindolin]-3'-one (**59**)



Chemical Formula: C₁₃H₁₅NO₂

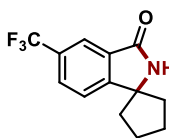
Product **59** was obtained as a white solid (35.3 mg, 81%)

¹H NMR (400 MHz, CDCl₃) δ 7.48 (s, 1H), 7.31 – 7.26 (m, 2H), 7.12 (dd, *J* = 8.4, 2.5 Hz, 1H), 3.86 (s, 3H), 2.13 – 2.04 (m, 2H), 2.04 – 1.88 (m, 6H).

¹³C NMR (101 MHz, CDCl₃) δ 169.9, 159.9, 143.7, 132.6, 122.1, 120.5, 106.2, 69.1, 55.7, 38.9, 24.7.

HRMS (ESI): *m/z* calculated for [C₁₃H₁₆NO₂]⁺ [M + H]⁺: 218.1176, found: 218.1186.

5'-(trifluoromethyl)spiro[cyclopentane-1,1'-isoindolin]-3'-one (**60**)



Chemical Formula: C₁₃H₁₂F₃NO

Product **60** was obtained as a white solid (30.7 mg, 60%)

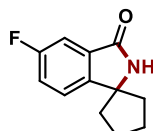
$^1\text{H NMR}$ (600 MHz, CDCl_3) δ 8.09 (s, 1H), 7.98 (s, 1H), 7.82 (d, $J = 8.0$ Hz, 1H), 7.53 (d, $J = 8.0$ Hz, 1H), 2.21 – 1.90 (m, 8H).

$^{13}\text{C NMR}$ (151 MHz, CDCl_3) δ 168.6, 154.9, 132.2, 130.8 (q, $J = 33.22$ Hz), 129.0 (q, $J = 3.02$ Hz), 123.9 (q, $J = 273.31$ Hz), 121.9, 121.1 (q, $J = 4.53$ Hz), 69.7, 39.2, 24.9.

$^{19}\text{F NMR}$ (565 MHz, CDCl_3) δ -62.2.

HRMS (ESI): m/z calculated for $[\text{C}_{13}\text{H}_{13}\text{F}_3\text{NO}]^+ [\text{M} + \text{H}]^+$: 256.0944, found: 256.0954.

5'-fluorospiro[cyclopentane-1,1'-isoindolin]-3'-one (**61**)



Chemical Formula: $\text{C}_{12}\text{H}_{12}\text{FNO}$

Product **61** was obtained as a white solid (31.3 mg, 76%)

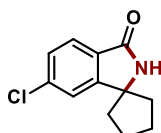
$^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.61 (s, 1H), 7.47 (dd, $J = 7.7, 2.4$ Hz, 1H), 7.36 (dd, $J = 8.4, 4.4$ Hz, 1H), 7.31 – 7.22 (m, 1H), 2.17 – 1.88 (m, 8H).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 168.7, 162.7 (d, $J = 247.45$), 146.9 (d, $J = 2.02$ Hz), 133.4 (d, $J = 8.08$ Hz), 122.7 (d, $J = 8.08$ Hz), 119.5 (d, $J = 23.23$ Hz), 110.3 (d, $J = 23.23$ Hz), 69.3, 39.1, 24.7.

$^{19}\text{F NMR}$ (376 MHz, CDCl_3) δ -113.9.

HRMS (ESI): m/z calculated for $[\text{C}_{12}\text{H}_{13}\text{FNO}]^+ [\text{M} + \text{H}]^+$: 206.0976, found: 206.0983.

6'-chlorospiro[cyclopentane-1,1'-isoindolin]-3'-one (**62**)



Chemical Formula: $\text{C}_{12}\text{H}_{12}\text{ClNO}$

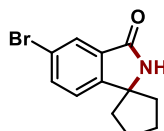
Product **62** was obtained as a white solid (34.2 mg, 77%)

$^1\text{H NMR}$ (600 MHz, CDCl_3) δ 8.06 (s, 1H), 7.72 (dd, $J = 8.0, 1.6$ Hz, 1H), 7.43 – 7.36 (m, 2H), 2.08 (m, 2H), 2.04 – 1.94 (m, 6H).

$^{13}\text{C NMR}$ (151 MHz, CDCl_3) δ 169.1, 153.3, 138.4, 129.9, 128.4, 124.9, 121.8, 69.3, 39.1, 24.8.

HRMS (ESI): m/z calculated for $[\text{C}_{12}\text{H}_{13}\text{ClNO}]^+ [\text{M} + \text{H}]^+$: 222.0680, found: 250. 222.0690.

5'-bromospiro[cyclopentane-1,1'-isoindolin]-3'-one (**63**)



Chemical Formula: $\text{C}_{12}\text{H}_{12}\text{BrNO}$

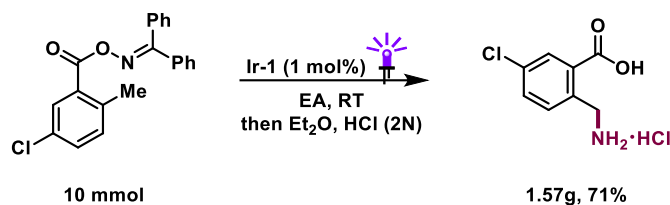
Product **63** was obtained as a white solid (38.5 mg, 72%)

$^1\text{H NMR}$ (600 MHz, CDCl_3) δ 8.10 (s, 1H), 7.96 – 7.90 (m, 1H), 7.69 – 7.63 (m, 1H), 7.28 (d, $J = 8.1$ Hz, 1H), 2.01 (m, 8H).

^{13}C NMR (151 MHz, CDCl_3) δ 168.6, 150.2, 135.0, 133.5, 126.8, 122.9, 121.8, 69.5, 39.0, 24.8.

HRMS (ESI): m/z calculated for $[\text{C}_{12}\text{H}_{13}\text{BrNO}]^+ [\text{M} + \text{H}]^+$: 266.0175, found: 266.0182.

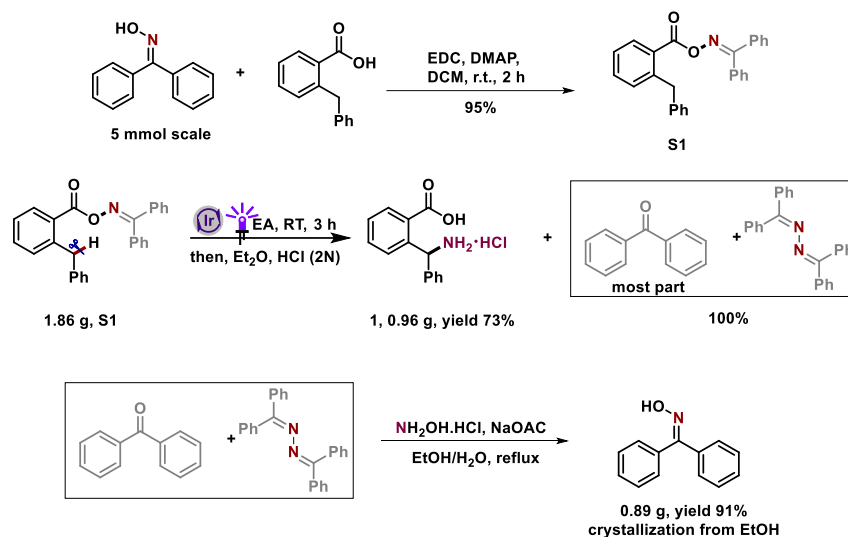
2.7 Gram scale of S23



To an oven-dried glass flask (250 mL) equipped with a magnetic stir bar were added the oxime esters **S23** (10 mmol, 1.0 equiv.), $[\text{Ir}(\text{dFppy})_2(\text{phpzpy})]\text{PF}_6$ (1 mol%), and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 30 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395$ nm) at room temperature for 12 h (1 cm away, with a cooling fan to keep the reaction temperature at 25-30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, acidified by 20 mL HCl (2 N) and stirred in Et₂O (30 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (20 mL) and extracted with Et₂O (30 mL x 3). The water layer was combined, and water was removed *in vacuo*, and the product was obtained as a white solid **23** (1.57g, 71%).

2.8 Recycling of the benzophenone

To verify the recovery of benzophenone byproducts, we also conducted a large-scale experiment on the template substrate **S1** (5 mmol) under the standard procedure.



To a solution of benzophenone oxime (5.0 mmol) and 2-benzylbenzoic acid (5.0 mmol) in CH₂Cl₂ (30 mL), DMAP (10 mol%) and EDCI•HCl (12.5 mmol) was added. The mixture was stirred at room temperature under inert atmosphere until the reaction was complete as observed from TLC monitoring. The mixture was diluted with distilled water (25 mL) and the CH₂Cl₂ layer was separated, dried over anhydrous Na₂SO₄ and concentrated. The crude mass was treated with pentane (3 mL) and sonicated for 15 minutes. The resultant solid was filtered and dried under vacuum to obtain the diphenylmethanone *O*-(2-benzylbenzoyl) oxime (**S1**).

To an oven-dried glass flask (250 mL) equipped with a magnetic stir bar were added the oxime esters **S1** (1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), and EA (0.05 M). Then the flask was bubbled with a stream of argon for 30 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature for 12 h (1 cm away, with a cooling fan to keep the reaction temperature at 25-30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, acidified by 10 mL HCl (2 N) and stirred in Et₂O (30 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (20 mL) and extracted with Et₂O (30 mL x 3). The water layer was combined, and water was removed *in vacuo*, and the product **1** was obtained as a white solid **1** (0.96 g, 73%). And the organic phase was concentrated *in vacuo*, and the mixture of imine self-coupling byproducts and benzophenone was obtained as a yellow solid.

And then, add hydroxylamine hydrochloride (1.5 equiv.), sodium acetate (2 equiv.) and EtOH/H₂O (4/1; 50 mL) to this yellow mixture, and the mixture was stirred at 80 °C overnight. The mixture

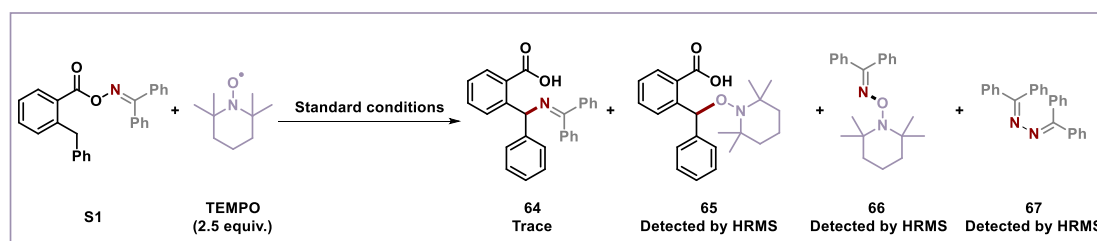
was cooled down to r.t. and EtOH was removed under reduced pressure. H₂O (50 mL) was added, and the resulting mixture was extracted with EtOAc (3x 20 mL). The combined organic layers were washed with H₂O (2x 50 mL) and brine (50 mL) and were dried over Na₂SO₄. the organic phase was concentrated *in vacuo*, and the product was obtained as a white solid. Further crystallization from EtOH gave pure benzophenone oxime as a white solid (0.89 g, 91 %).

3 Supplementary Notes

3.1 Mechanistic Investigations

3.1.1 Trapping Experiments

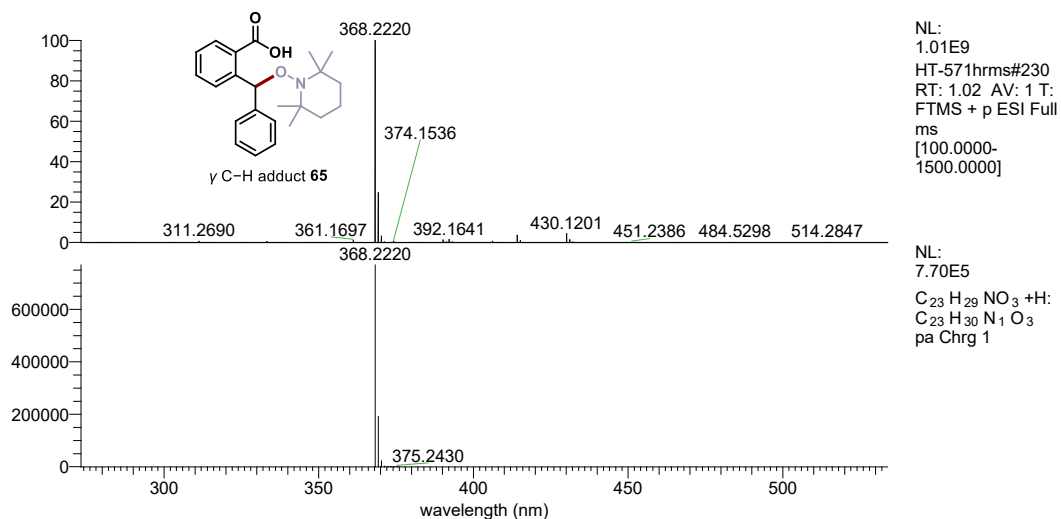
To investigate the intermediacy of radicals which are involved in the presented methodology, a trapping experiment for the intramolecular C–H amination using 2,2,6,6-tetramethylpiperidine-N-oxyl (TEMPO) in (over-)stoichiometric quantities was performed.



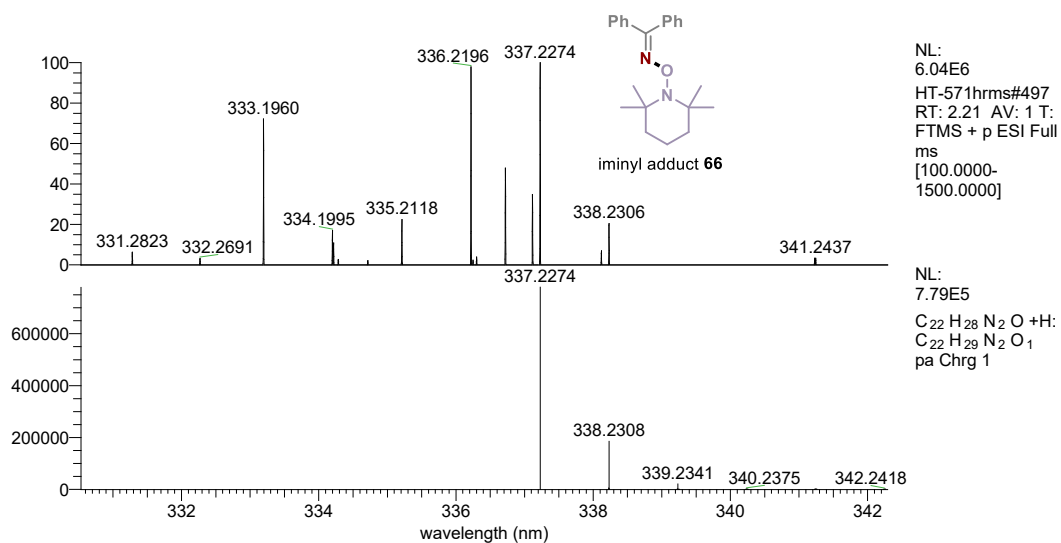
Supplementary Figure 2. TEMPO-trapping experiment for γ C–H amination.

To an oven-dried glass tube (10 mL) equipped with a magnetic stir bar were added the oxime esters **S1** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phzppy)]PF₆ (1 mol%), and TEMPO (2.5 equiv.) in EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395$ nm) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25–30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion, the crude reaction mixture was filtered over basified alumina and washed with EtOAc. Both, the adduct of the transient carboxyl oxygen radical and of the persistent iminyl radical with TEMPO could be observed in HRMS (γ C–H adduct **65**: [M+H]⁺ calc 368.2220, found 368.2220; iminyl adduct **66**: [M+H]⁺ calc 337.2274, found 337.2274; radical–radical self-coupled byproduct **67**: [M+H]⁺ calc

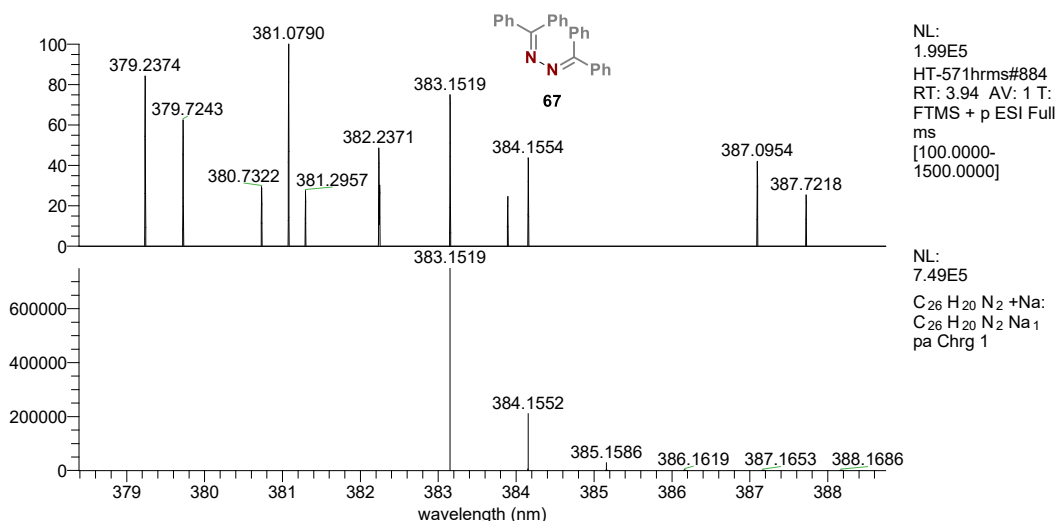
383.1519, found 383.1519). No product formation could be observed by NMR analysis.



Supplementary Figure 3. HRMS of TEMPO-trapping experiment (γ C-H adduct 65)



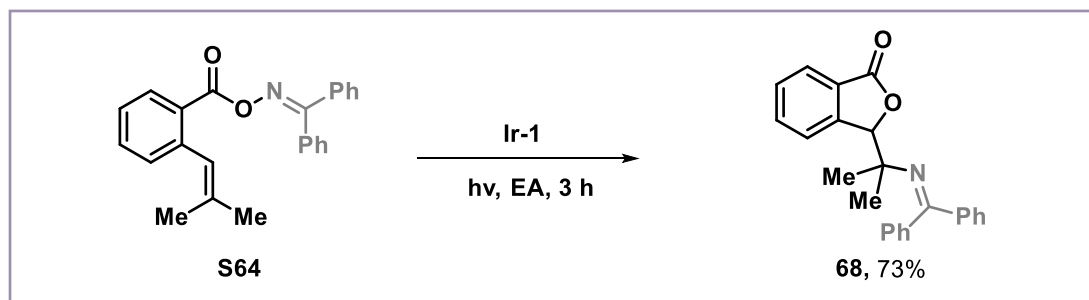
Supplementary Figure 4. HRMS of TEMPO-trapping experiment (iminyl adduct 66)



Supplementary Figure 5. HRMS of TEMPO-trapping experiment (radical–radical self-coupled byproduct 67)

Therefore, the TEMPO trapping experiment supports the intermediacy of both aryloxy radical and persistent iminyl radical and their participation for product formation.

3.1.2 Radical Probe Experiment



Supplementary Figure 6. Radical Probe Experiment of S64

To an oven-dried glass tube (10 mL) equipped with a magnetic stir bar were added the oxime esters **S64** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25–30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, column chromatography on silica gel pre-basified with NEt₃

using pentane: EtOAc mixtures afforded 3-(2-((diphenylmethylene)amino)propan-2-yl)isobenzofuran-1(3*H*)-one (**68**, 51.9 mg, 73 %) as a white solid.

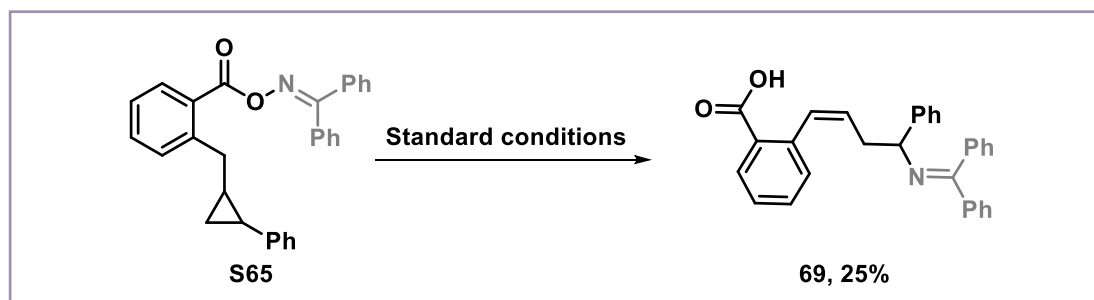
¹H NMR (600 MHz, CDCl₃) δ 7.97 – 7.84 (m, 1H), 7.78 (d, *J* = 7.7 Hz, 1H), 7.66 – 7.56 (m, 3H), 7.51 (t, *J* = 7.5 Hz, 1H), 7.45 – 7.31 (m, 6H), 7.22 – 7.08 (m, 2H), 5.65 (s, 1H), 1.40 (s, 3H), 0.59 (s, 3H).

¹³C NMR (151 MHz, CDCl₃) δ 171.1, 166.1, 148.9, 140.8, 139.1, 133.7, 130.1, 129.0, 128.3, 128.2, 128.1, 128.0, 127.2, 125.3, 124.1, 88.0, 62.7, 26.4, 21.2.

HRMS (ESI): *m/z* calculated for [C₂₄H₂₂NO₂]⁺ [M + H]⁺: 356.1645, found: 356.1645.

This result suggests the involvement of aryloxy radicals through a photo-induced EnT process.

3.1.3 Radical Clock Experiment



Supplementary Figure 7. Radical clock experiment of S64

To an oven-dried glass tube (10 mL) equipped with a magnetic stir bar were added the oxime esters **S65** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25–30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, Column chromatography on basified silica gel (hexane/EA = 5/1 – 1/1) afforded 2-(4-((diphenylmethylene)amino)-4-phenylbut-1-en-1-yl)benzoic acid (**69**, 21.58 mg, 25 %) as a colorless oil.

¹H NMR (600 MHz, CDCl₃) δ 9.35 (s, 1H), 7.98 (d, *J* = 7.8 Hz, 1H), 7.81 (d, *J* = 7.6 Hz, 1H), 7.59 (dd, *J* = 12.5, 7.4 Hz, 2H), 7.53 – 7.36 (m, 5H), 7.34 – 7.17 (m, 9H), 7.07 – 7.02 (m, 1H), 6.90 (d, *J* = 11.7 Hz, 1H), 5.57 (dt, *J* = 11.7, 7.7 Hz, 1H), 4.60 – 4.34 (m, 1H), 2.72 (ddt, *J* = 90.6, 13.7, 7.1

Hz, 2H).

^{13}C NMR (151 MHz, CDCl_3) δ 171.8, 167.1, 144.2, 139.7, 139.5, 137.6, 136.9, 132.4, 132.2, 131.2, 131.0, 130.9, 130.1, 129.9, 128.6, 128.4, 128.3, 128.3, 128.3, 127.9, 127.8, 127.1, 126.8, 126.7, 66.5, 37.9.

HRMS (ESI): m/z calculated for $[\text{C}_{30}\text{H}_{26}\text{NO}_2]^+ [\text{M} + \text{H}]^+$: 432.1958, found: 432.1958.

This result further suggests the involvement of benzyl radicals formed from the γ -position of aryloxy radicals via 1,5-intramolecular HAT.

3.1.4 Quantum Yield Measurement

A solution of ferrioxalate was chosen as an actinometer following the procedure described by the IUPAC (subcommittee on photochemistry).^{11,12} The procedure is based on the decomposition under irradiation of ferric ions to ferrous ions which are complexed by 1,10-phenanthroline. This photochemical transformation has a known quantum yield, and the complexation of Fe^{2+} with 1,10-phenanthroline can be monitored by UV-visible absorption since its extinction coefficient at 510 nm is known ($\epsilon = 11100 \text{ M}^{-1} \text{ cm}^{-1}$).¹³ Therefore, the moles of the iron-phenanthroline complex formed are related to moles of photons absorbed. 0.006 M, 0.012 M, or 0.15 M solutions of ferrioxalate can be used for actinometry. In this case we chose a concentration of 0.15 M. The solutions were prepared and stored in a dark laboratory as follows:

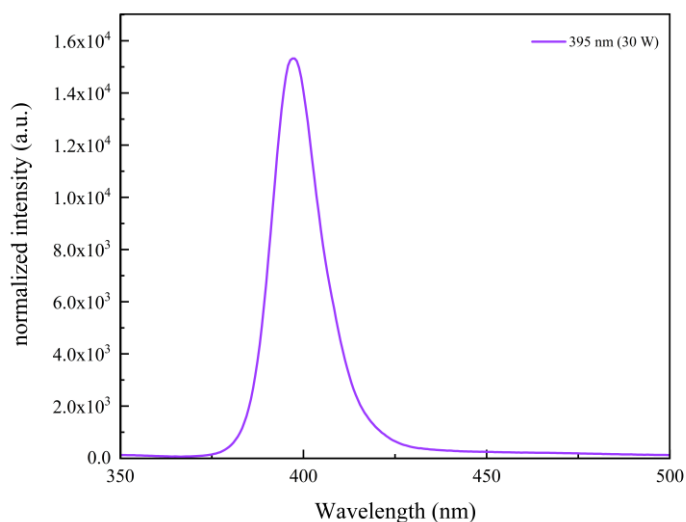
Solution A: 1.84 g of $\text{K}_3[\text{Fe}(\text{C}_2\text{O}_4)_3] \cdot 3\text{H}_2\text{O}$ and 1.75 mL of H_2SO_4 were added into a 25 mL volumetric flask and filled to the mark with ultrapure water.

Solution B: 49.6 mg of 1,10-phenanthroline monohydrate, 10.0 g of NaOAc, and 1.0 mL of H_2SO_4 were added to a 100-mL volumetric flask and filled to the mark with ultrapure water.

Model reaction solution: Oxime ester **S1** (78.2 mg, 0.2 mmol, 1.0 equiv.), and $[\text{Ir}(\text{dFppy})_2(\text{phpzpy})]\text{PF}_6$ (1 mol %), were added to an oven-dried quartz cuvette ($l = 1 \text{ cm}$) equipped with a magnetic stir bar and a plastic plug. The combined materials were dissolved in deoxygenated EA (4.0 mL) under argon atmosphere.

Actinometry procedure: The simultaneous irradiation of both the actinometer solution and model reaction by the self-same reactor setup is not feasible. However, the stability of the irradiation light

was checked through radiometer measurements (recorded on an AVANTES® AvaSpec-ULS2048 spectrometer instrument). Therefore, we assumed that consecutive measurements of both actinometer and model reaction are comparable. We used 30 W LED as a light source, detecting a maximum wavelength of emission of 395 nm (**Supplementary Figure 8**).



Supplementary Figure 8. The Emission spectrum of the photochemical reactor ($\lambda_{\text{max}} = 395 \text{ nm}$).

(1) 2 mL of solution A was added to a quartz cuvette under dark conditions while being stirred. Then, the actinometry solutions were irradiated with a 395 nm LED for specified time intervals (0, 5, 10, 15, 20 s) and a 0.1 mL aliquot was taken.

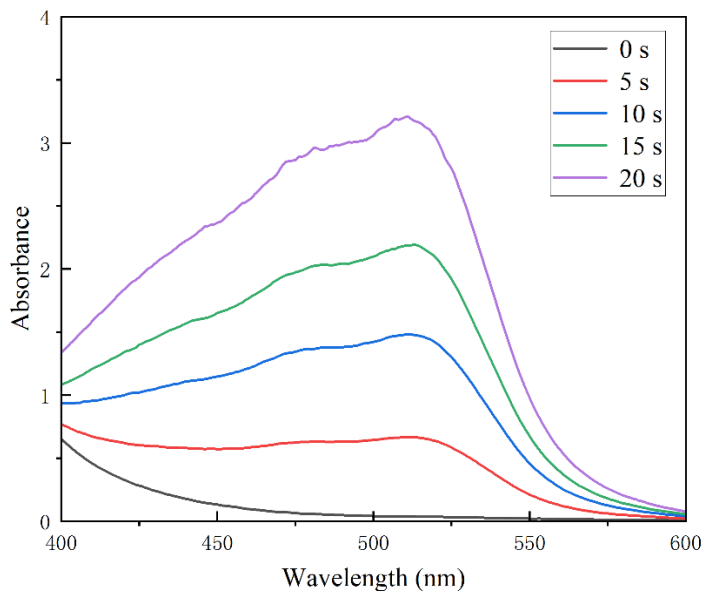
(2) 4 mL solution B was added to each aliquot, and the final volume was raised to 10 mL with ultrapure water. All samples were stored in the dark and stirred for one hour.

(3) The absorbance spectrum of each sample was monitored at 510 nm for each time interval by the UV-2600 spectrometer. The absorbance to each time was related to the photochemically produced Fe^{2+} ions across the Lambert-Beer Law (Equation [1]):

$$\text{moles Fe}^{2+} = \frac{V1 \times V3 \times \Delta A(510 \text{ nm})}{V2 \times 1 \times \epsilon(510 \text{ nm})} \quad [1]$$

Where $V1$ is the irradiated volume (2 mL), $V2$ is the aliquot of the irradiated solution taken for the determination of the ferrous ions (0.1 mL), $V3$ is the final volume after complexation with phenanthroline (10 mL), l is the optical path-length of the irradiation quartz cuvette (1 cm), $\Delta A(510$

nm) the optical difference in absorbance between the irradiated solution and that taken in the dark, $\epsilon(510 \text{ nm})$ is the extinction coefficient of the complex $\text{Fe}(\text{phen})_3^{2+}$ ($11100 \text{ M}^{-1} \text{ cm}^{-1}$).

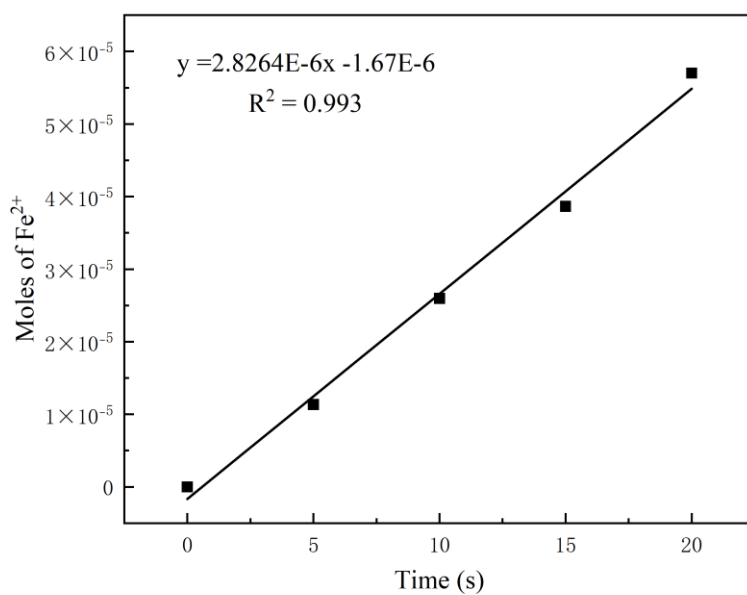


Supplementary Figure 9. The absorbance spectrum of each time interval.

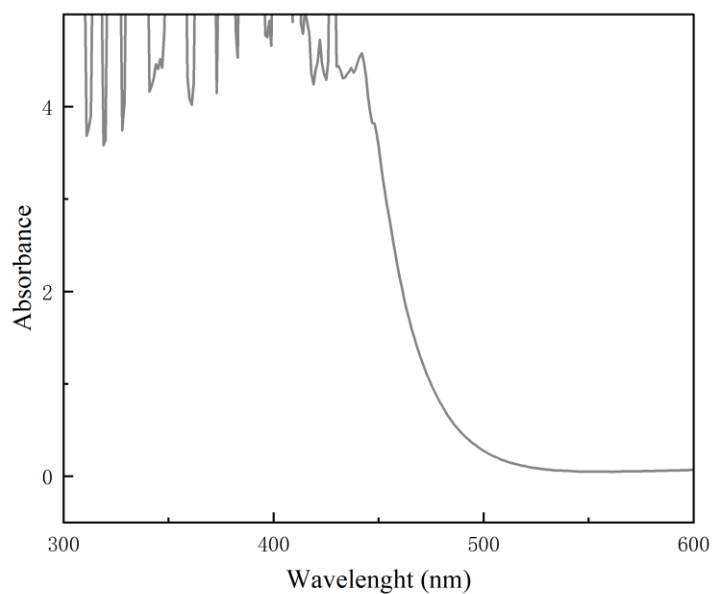
(4) The moles of Fe^{2+} formed (x) are plotted as a function of time (t) (Figure S10.). The slope of the line (dx/dt) was correlated to the moles of incident photons by unit of time ($q_{n,p}$) using the following equation [2]:

$$q_{n,p} = \frac{dx/dt}{\Phi(\lambda) \times f}, \quad f = [1 - 10^{-A(\lambda)}] \quad [2]$$

Where $\Phi(\lambda)$ is the quantum yield of the actinometer reaction at the irradiated wavelength, in this case being 1.13 at 395 nm.^{12,14} The value of the photon flux must be divided by the fraction of absorbed light f at the irradiation wavelength, and $A(\lambda)$ is the absorbance of the actinometer solution (ferrioxalate) at the irradiated wavelength (395 nm) obtaining a value of 6.4, $f = 1$. Therefore, the moles of incident photons by a unit of time were determined as $2.50 \times 10^{-6} \text{ einstein s}^{-1}$.



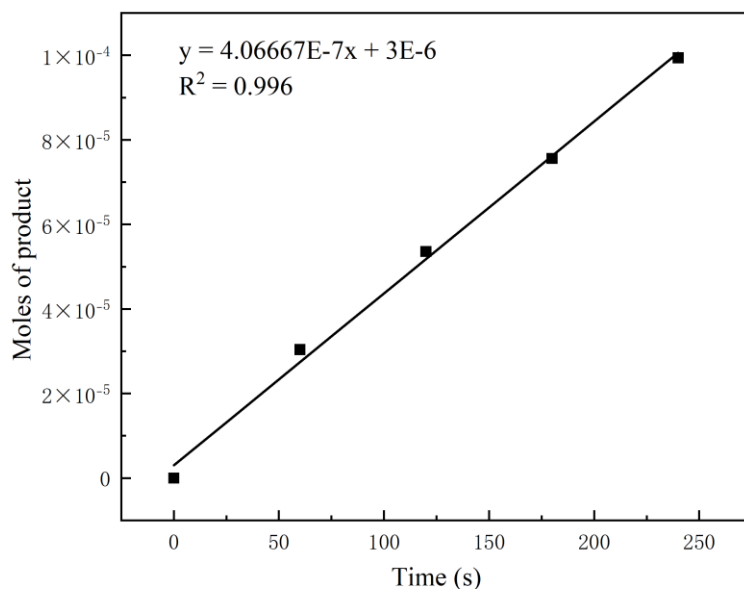
Supplementary Figure 10. The moles of Fe²⁺ formed (x) are plotted as a function of time (t).



Supplementary Figure 11. Absorbance of a 0.15 M solution of K₃[Fe(C₂O₄)₃]•3H₂O in ultrapure water.

The kinetics of the reaction under study were done by irradiating the actinometer solution described above. The model reaction solutions were irradiated using the same spectrometer with consecutive measurements every minute. The moles of products formed were determined by ¹H-NMR spectrum

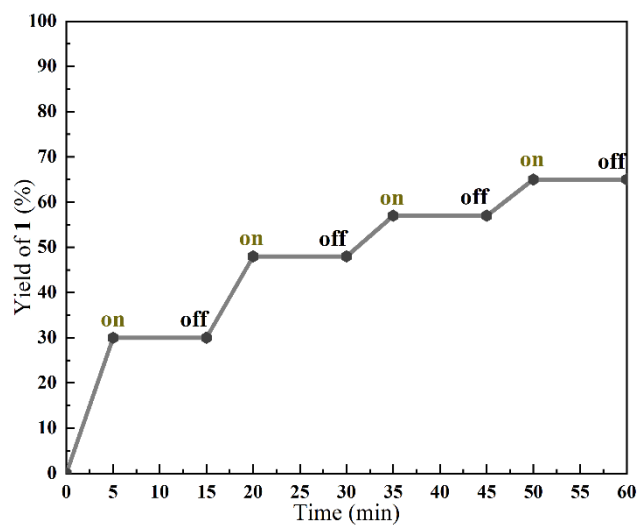
in CDCl_3 . Plotting the moles of product versus the irradiation time, the slope dx/dt can be related to the quantum yield across the equation [2] being equal to $(q_{n,p}) \times \Phi(\lambda) \times (1 - 10^{-A(\lambda)})$. The model reaction solution was added to a 1 cm optical pathway cuvette to measure A (395 nm), The quantum yield at 395 nm of the reaction by $[\text{Ir}(\text{dFppy})_2(\text{phpzpy})]\text{PF}_6$ was calculated as **0.24**.



Supplementary Figure 12. Plotting the moles of product versus the irradiation time

3.1.5 Light On/Off Experiment

Under standard conditions, the light On/Off experiments were carried out where the light was switched on every 5 min and off every 10 min with stirring maintained.

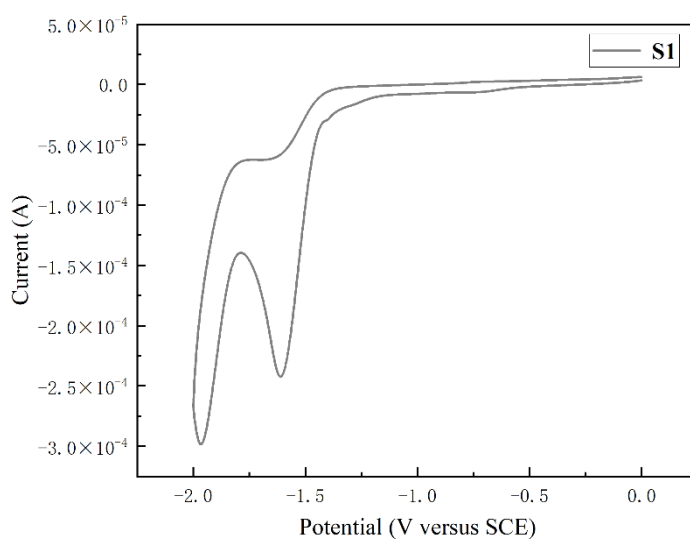


Supplementary Figure 13. Light On/Off experiment

The results clearly show that the reaction only takes place when the light is on.

3.1.6 Cyclic Voltammetry Studies

The experiments were conducted using a cyclic potentiometer with a glassy carbon working electrode polished by aluminum oxide (50 nm), a Pt counter electrode and a saturated calomel electrode (SCE) reference electrode. In the standard procedure, 0.1 mmol of substrate (S1) were dissolved in 10 mL of a 0.1 M $[N(\text{Bu})_4]\text{PF}_6$ electrolyte solution in MeCN. Each measurement was conducted at 50 mV/s at room temperature without stirring.



Supplementary Figure 14. The Cyclic Voltammetry (CV) experiments of substrates **S1** and $[N(\text{Bu})_4]\text{PF}_6$ in MeCN, $[\text{S1}] = 1.00 \times 10^{-2}$ M. $E_{\text{red}} = -1.61$ V vs SCE

Excited Ir-1 not reducing enough to reduce the reagent. No product formation was observed by using more reducing photocatalysts (see section 3.1.6).

3.1.7 Evaluation of C–H amination between yield and photocatalyst triplet energy

To investigate the substrate activation, present in the disclosed photocatalytic manifolds, photocatalysts of different triplet energy (and redox potentials) were evaluated in cross-relationship with product yield for the γ C–H amination. (**Supplementary Table 5**) Triplet energies and redox potentials (vs SCE) were adopted from the following sources.^{15,16}

Entry	PC	E_T (Kcal/mol)	$E_{1/2}^{[PC]^+/[PC]^*}$ (V)	$E_{1/2}^{[PC]^*/[PC]^-}$ (V)	Yield of 1
1 ^a	Thioxanthone	65.5	-1.11	+1.18	73%
2	[Ir(dFppy) ₂ (phpzpy)]PF ₆	62.8	-1.16	+1.21	82%
3 ^b	[Ir(dF(CF ₃)ppy) ₂ (dtbbpy)]PF ₆	61.8	-0.89	+1.21	57%
4 ^b	<i>fac</i> -Ir(ppy) ₃	58.1	-1.73	+0.31	N.D.
5 ^c	4CzIPN	57.1	-1.04	+1.43	58%
6 ^b	Ru(bpy) ₃ Cl ₂	46.5	-0.81	+0.77	N.D.

Supplementary Table 5. Evaluation of cross-relationship between yield and photocatalyst triplet energy.

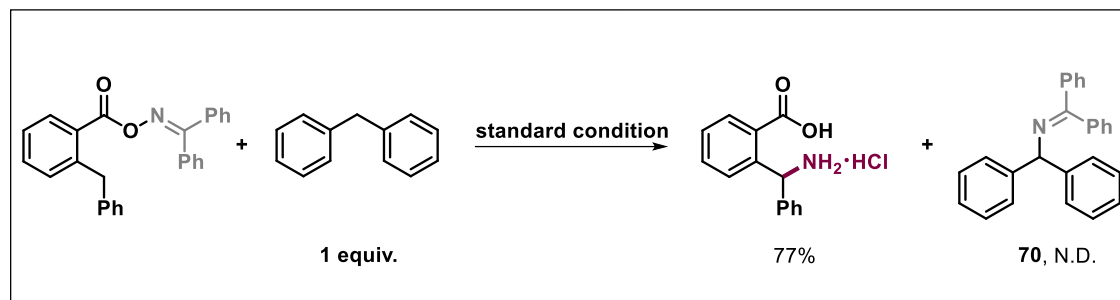
For each entry, to an oven-dried glass tube (10 mL) equipped with a magnetic stir bar were added **S1** (0.2 mmol, 1.0 equiv.), PC (1 mol%), and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{max} = 395$ nm) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25-30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated. Upon completion of the reaction, the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. After consumption of the starting material was confirmed by TLC analysis, the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and the product was obtained as a white solid. ^a5 mol% PC used.

^bThe reaction was irradiated at 450nm. ^c5 mol% PC used, 450 nm.

In summary, the correlation between triplet energy and the yield of the γ C-H amination product is shown. These results implied that an EnT process is likely to have been operational in the reaction.

3.1.8 Regioselective experiment of aroyloxy radicals in HAT process

To demonstrate that this is an intramolecular hydrogen atom transfer (HAT) process rather than an intermolecular process, we artificially added an alkane with a lower bond dissociation energy (BDE) to the reaction system. Fortunately, no intermolecular HAT product **70** was formed, confirming that this is indeed an intramolecular HAT process.



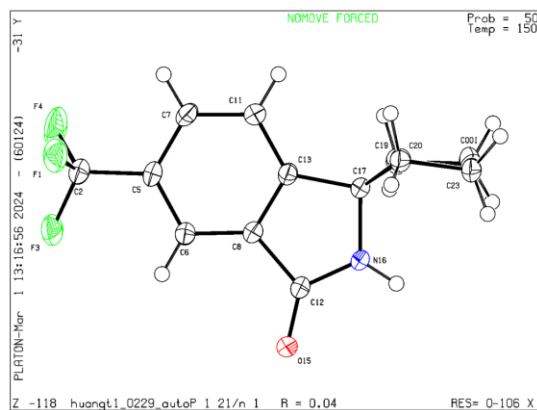
Supplementary Figure 15. Regioselective experiment of aroyloxy radicals in HAT process

To an oven-dried glass tube (10 mL) equipped with a magnetic stir bar were added the oxime esters **S1** (0.2 mmol, 1.0 equiv.), [Ir(dFppy)₂(phpzpy)]PF₆ (1 mol%), diphenylmethane (1 equiv.) and EA (0.05 M). Then the vessel was bubbled with a stream of argon for 20 min via a syringe needle and the tightly sealed tube was irradiated with two 30 W purple LEDs ($\lambda_{\text{max}} = 395 \text{ nm}$) at room temperature for 3 h (1 cm away, with a cooling fan to keep the reaction temperature at 25-30 °C and keeping the reaction region located in the center of LEDs lamp), unless otherwise stated.

After consumption of the starting material was confirmed by TLC analysis, the solvents were removed under reduced pressure, acidified by 2 mL HCl (2 N) and stirred in Et₂O (2 mL) for 2 h at room temperature. the solution was poured into water (2.0 mL) and extracted with Et₂O (10 mL x 3). The water layer was combined, and water was removed *in vacuo*, and the product was obtained as a white solid **1** (40.6 mg, 77%).

3.2 X-Ray Crystallographic Data

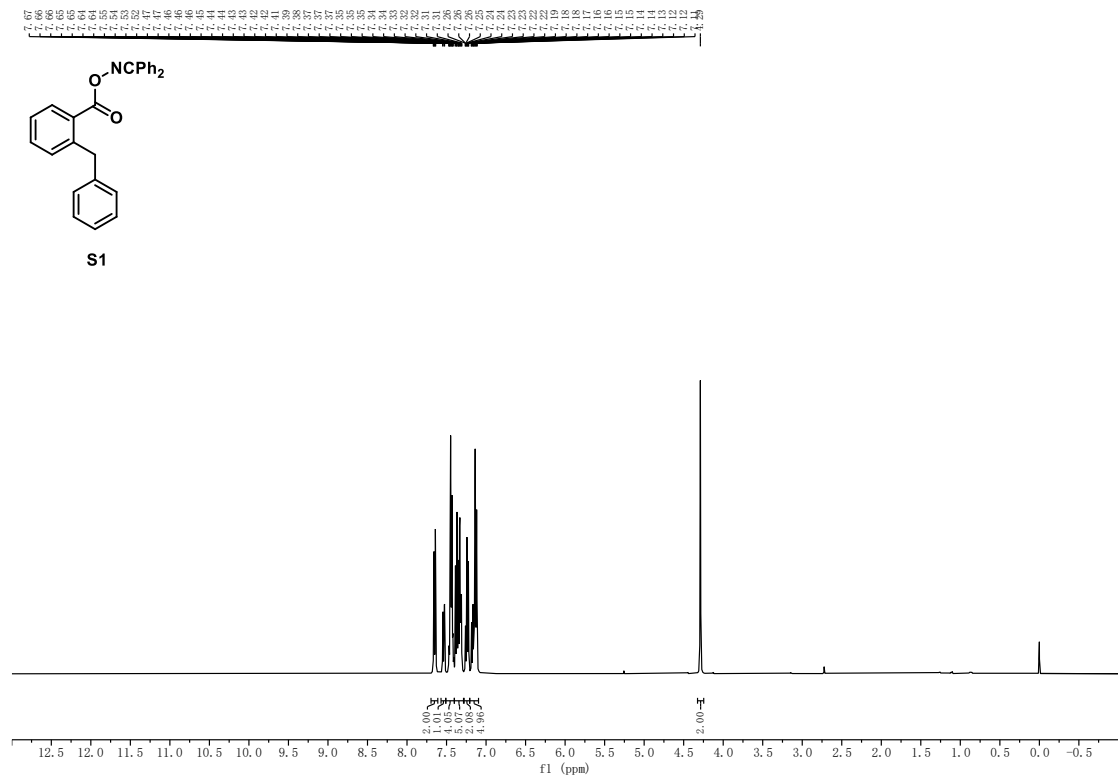
Single crystal of **60** was obtained by recrystallization from a mixed solvent of CH₂Cl₂/ petroleum ether at room temperature (evaporation in air). The X-ray single-crystal determination was performed on a Bruker APEX II X-ray single crystal diffractometer.



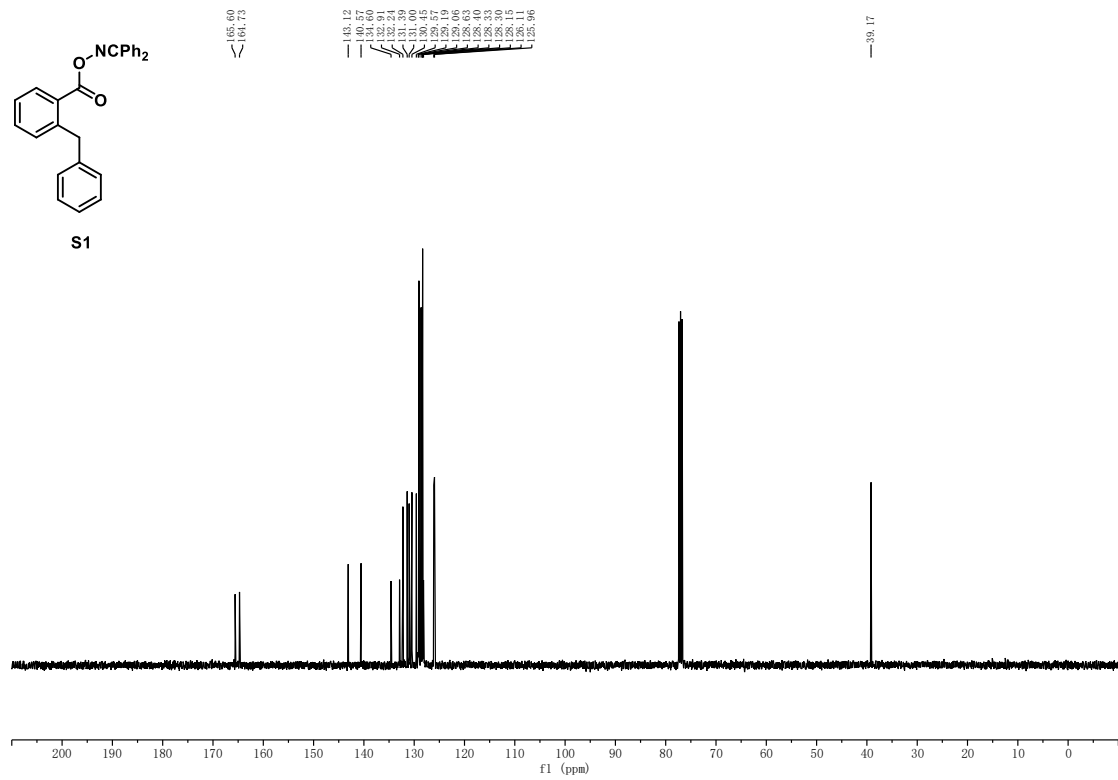
Supplementary Figure 16. X-Ray diffraction of **60** (CCDC-2336872).

Bond precision:	C-C = 0.0019 Å	Wavelength=0.71073	
Cell:	a=10.6674(5) alpha=90	b=10.0950(4) beta=115.445(6)	c=11.8549(6) gamma=90
Temperature:	150 K		
	Calculated	Reported	
Volume	1152.79(11)	1152.79(11)	
Spacegroup	P 21/n	P 1 21/n 1	
Hall group	-P 2yn	-P 2yn	
Moiety formula	C13 H12 F3 N O	C13 H12 F3 N O	
Sum formula	C13 H12 F3 N O	C13 H12 F3 N O	
Mr	255.24	255.24	
Dx, g cm ⁻³	1.471	1.471	
Z	4	4	
Mu (mm ⁻¹)	0.126	0.126	
F000	528.0	528.0	
F000'	528.37		
h, k, lmax	15,14,17	15,14,17	
Nref	3579	2899	
Tmin, Tmax	0.989,0.991	0.948,1.000	
Tmin'	0.989		
Correction method= # Reported T Limits: Tmin=0.948 Tmax=1.000			
AbsCorr = MULTI-SCAN			
Data completeness= 0.810		Theta(max)= 30.707	
R(reflections)= 0.0427 (2438)		wR2(reflections)= 0.1118 (2899)	
S = 1.049	Npar= 163		

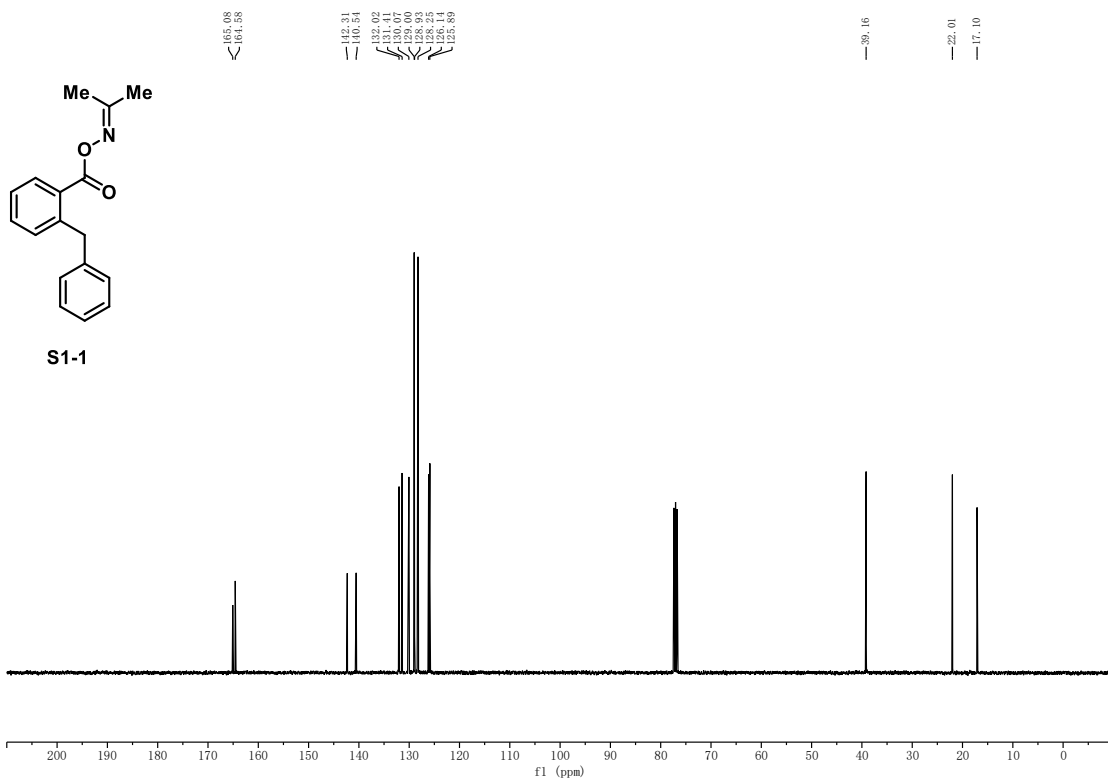
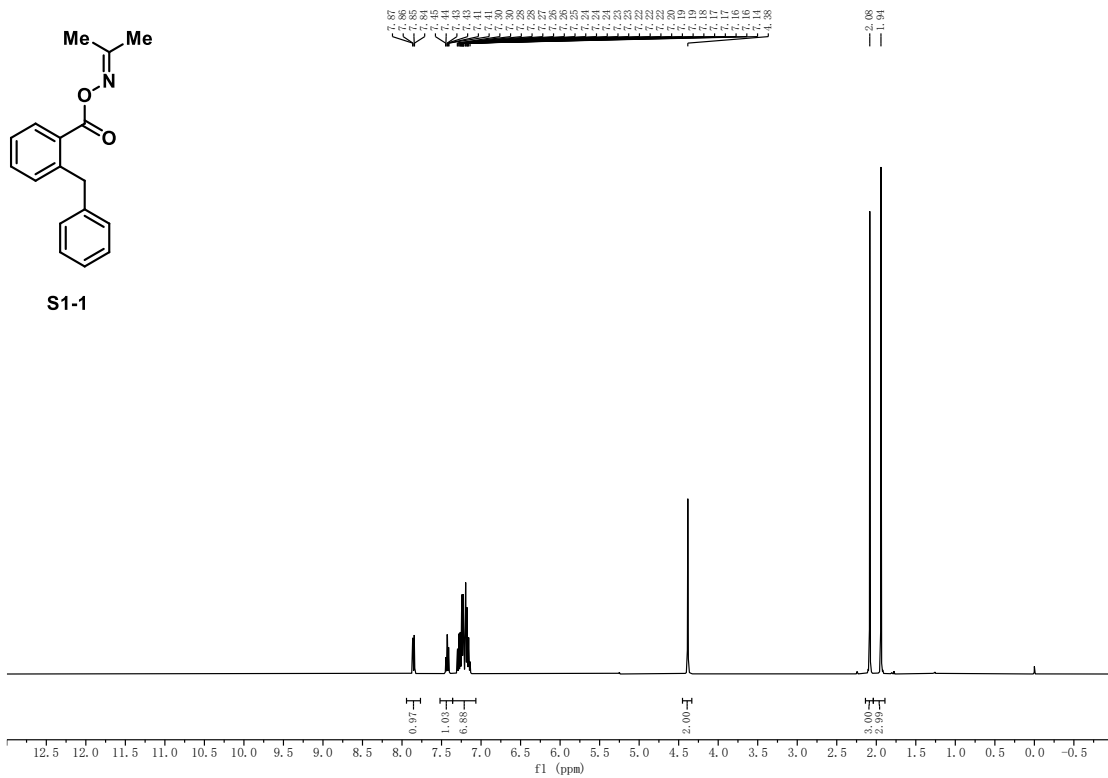
3.3 NMR spectra

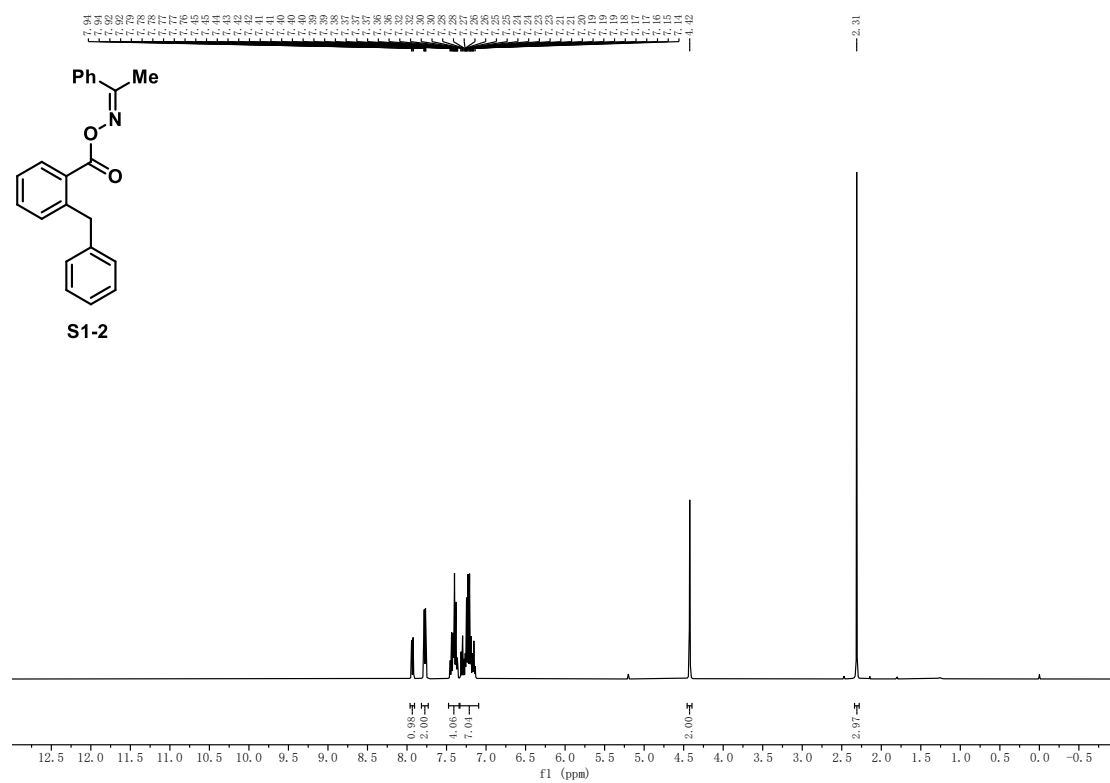


Supplementary Figure 17 ¹H NMR spectrum of S1

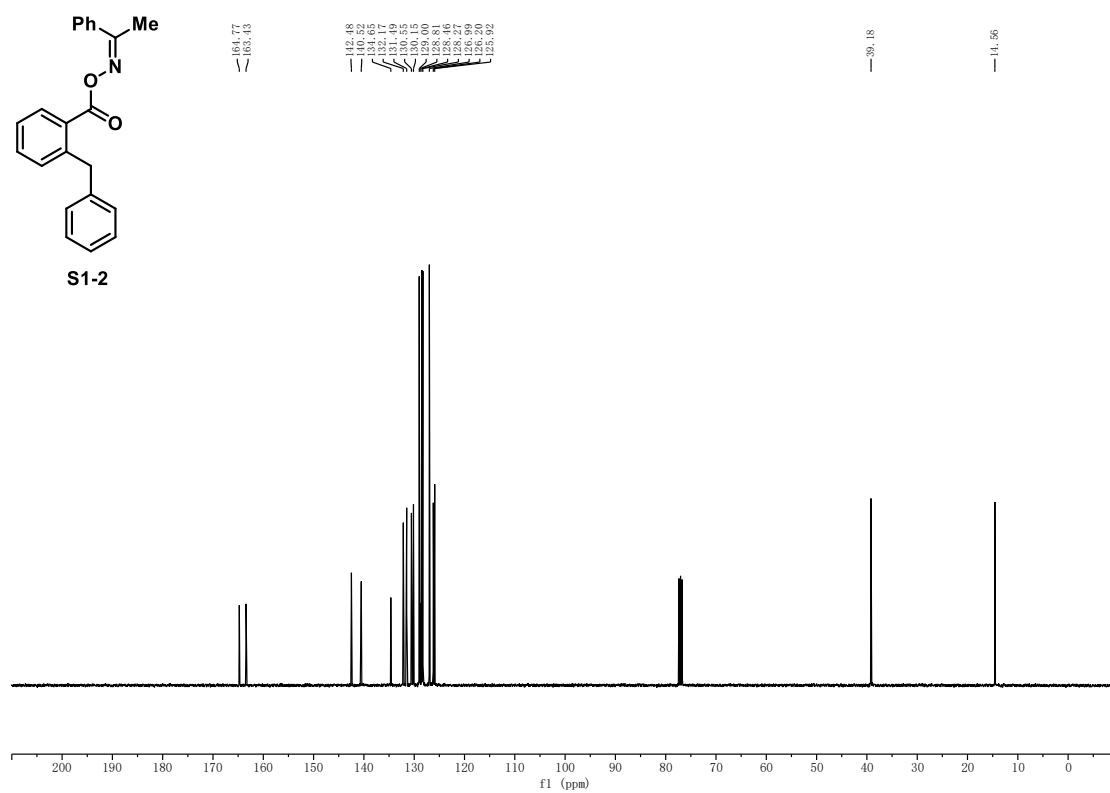


Supplementary Figure 18 ¹³C NMR spectrum of S1

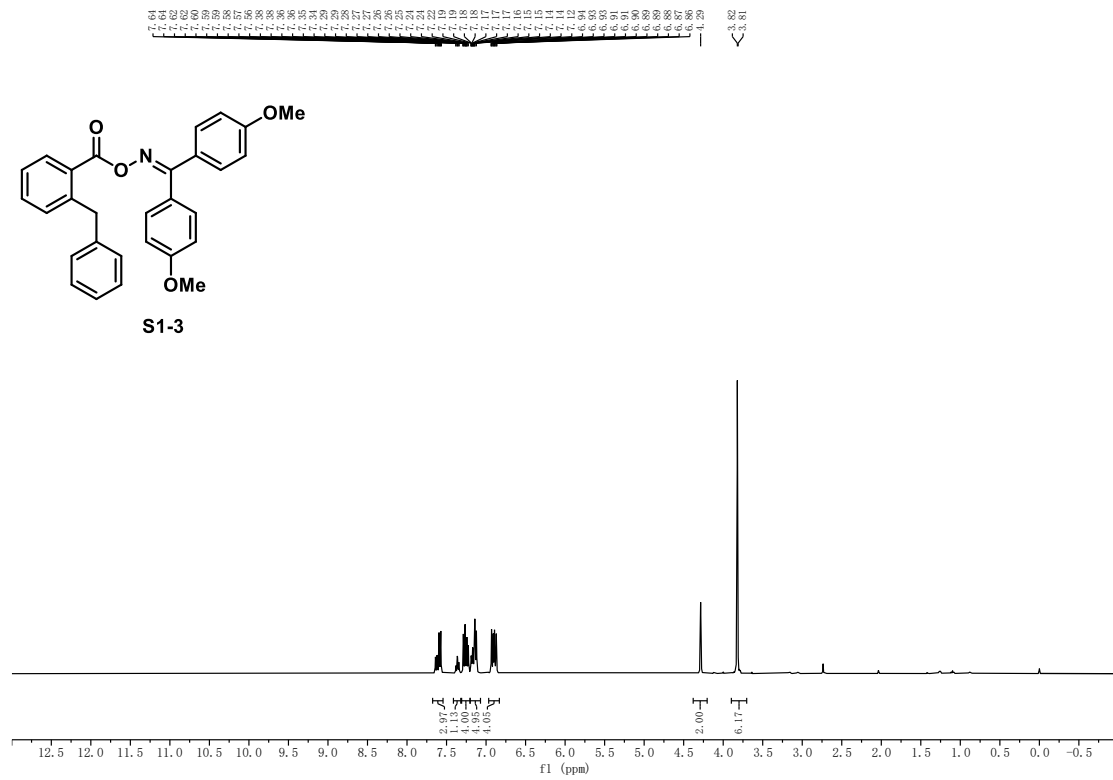




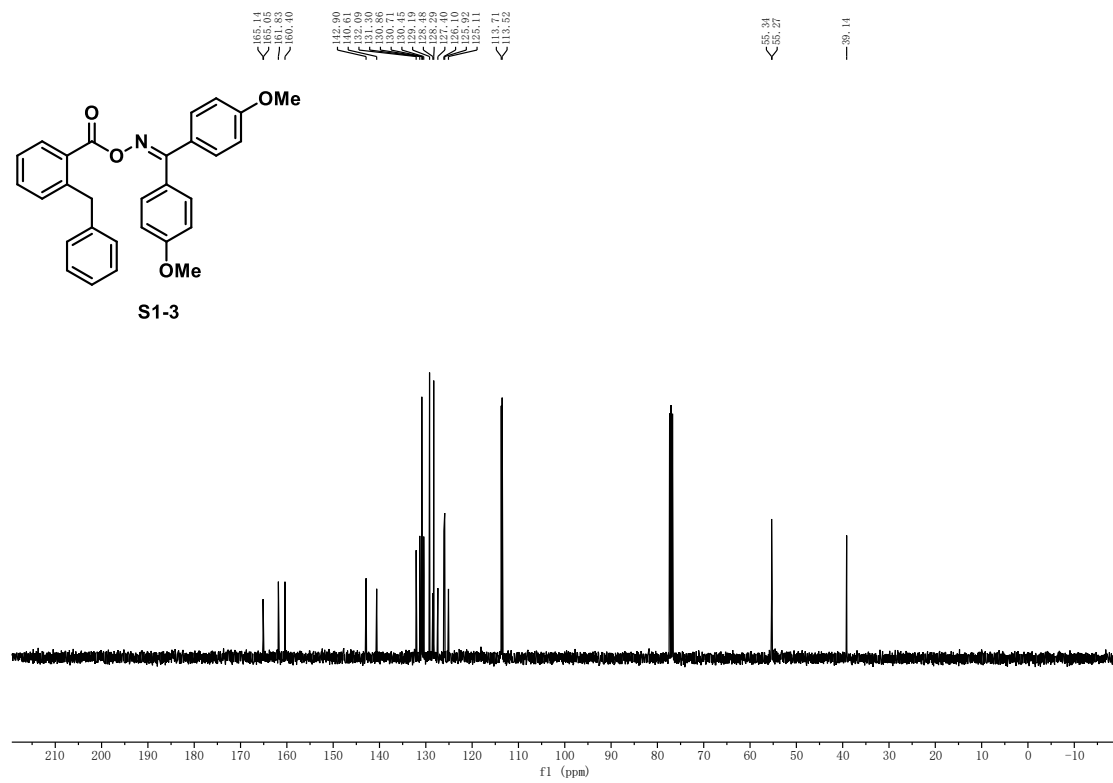
Supplementary Figure 21 ¹H NMR spectrum of S1-2



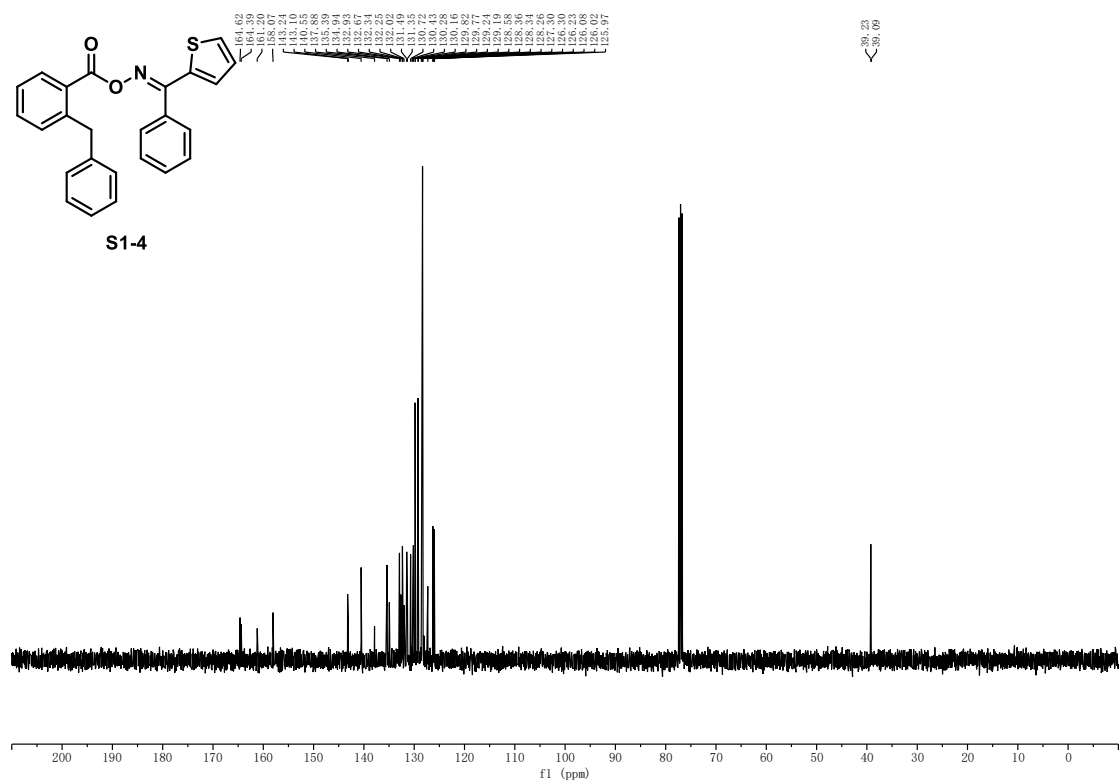
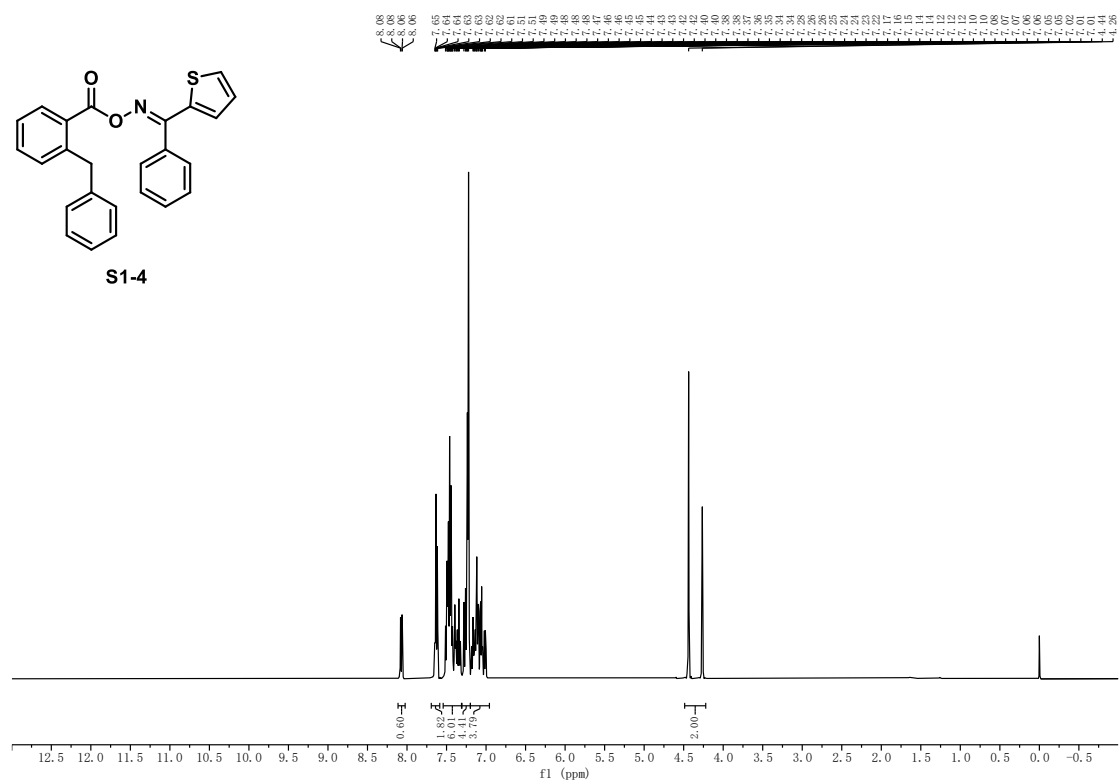
Supplementary Figure 22 ¹³C NMR spectrum of S1-2

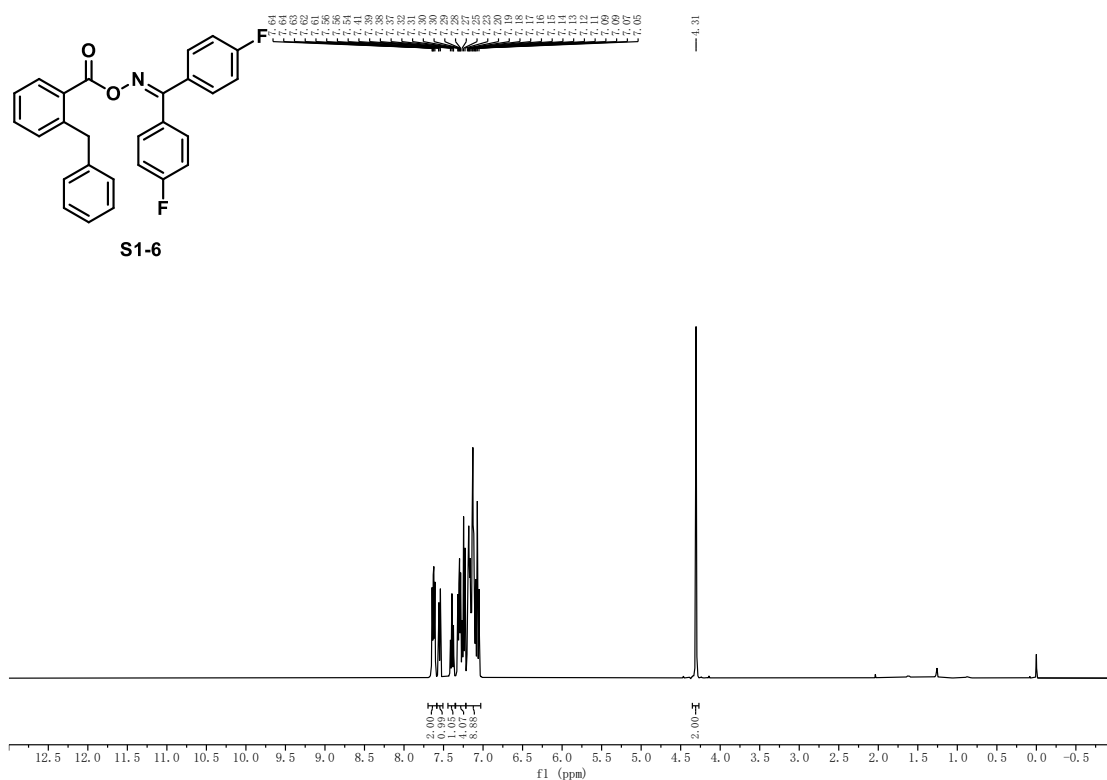


Supplementary Figure 23 ^1H NMR spectrum of S1-3

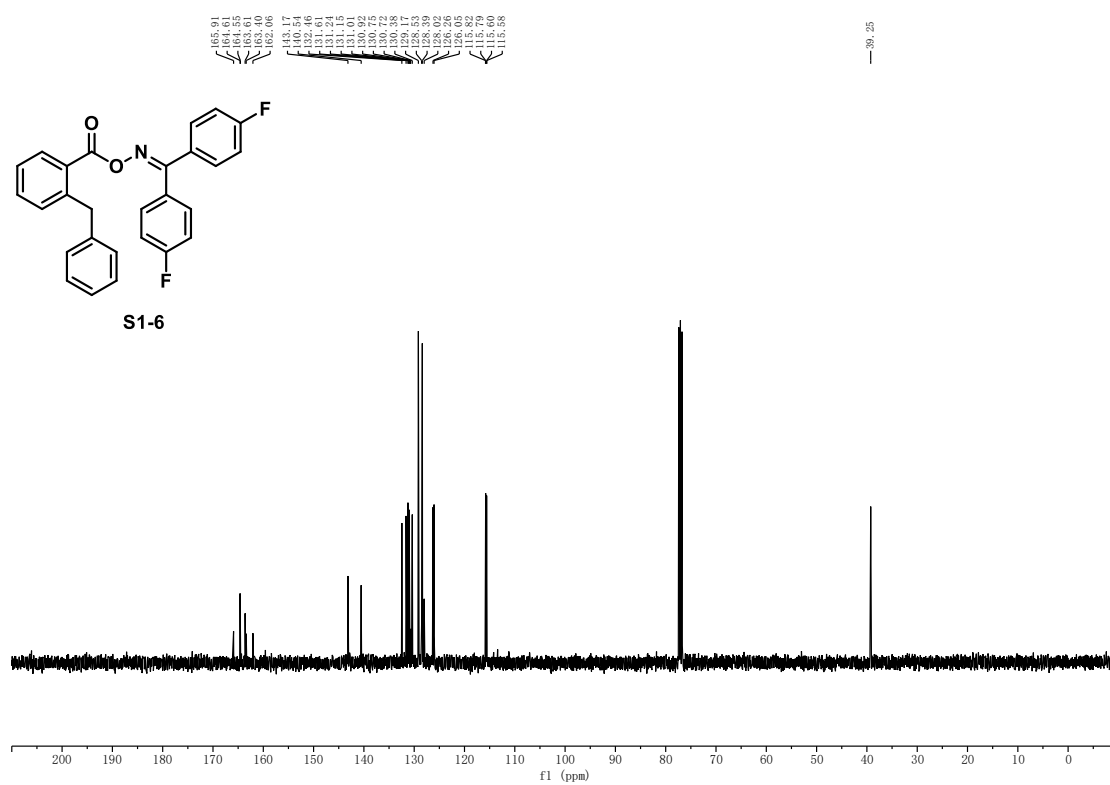


Supplementary Figure 24 ^{13}C NMR spectrum of S1-3

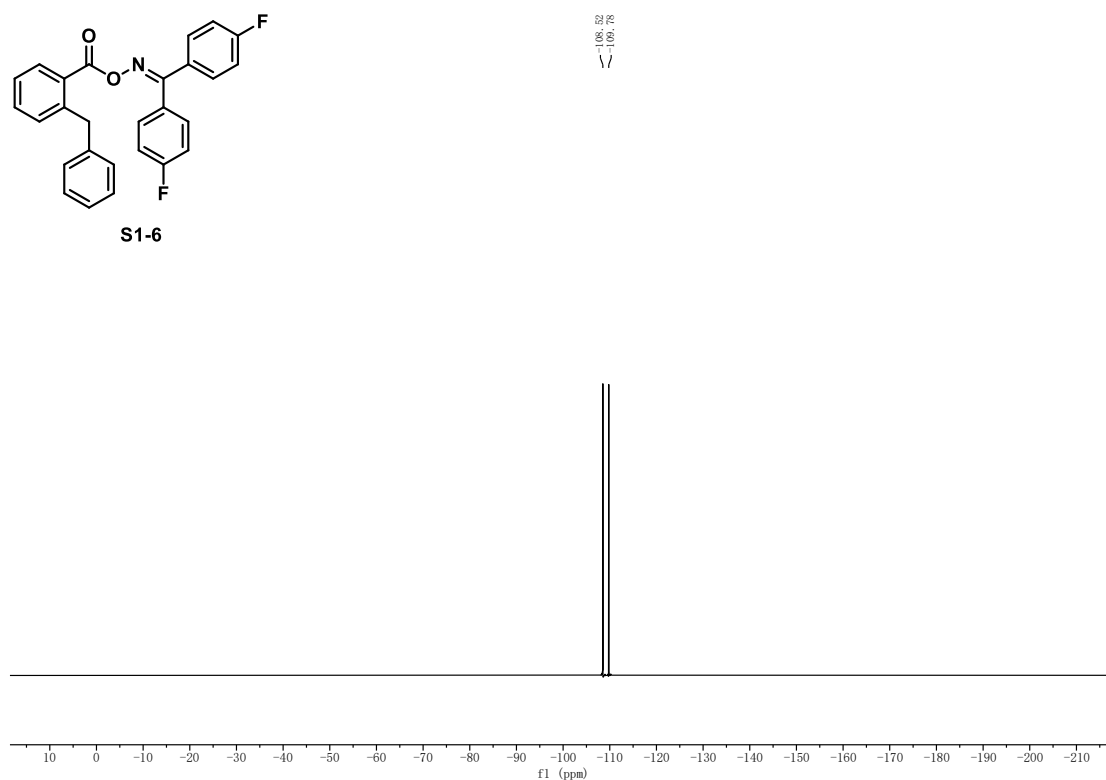
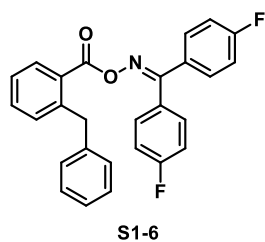




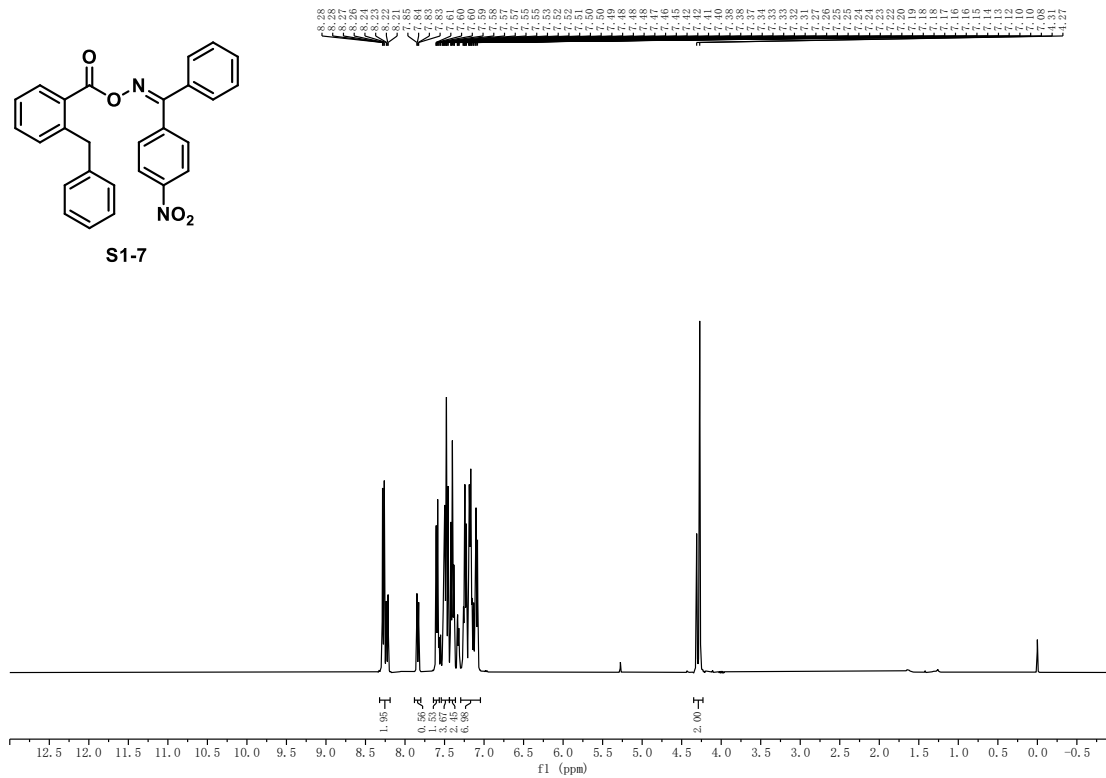
Supplementary Figure 29 ¹H NMR spectrum of S1-6



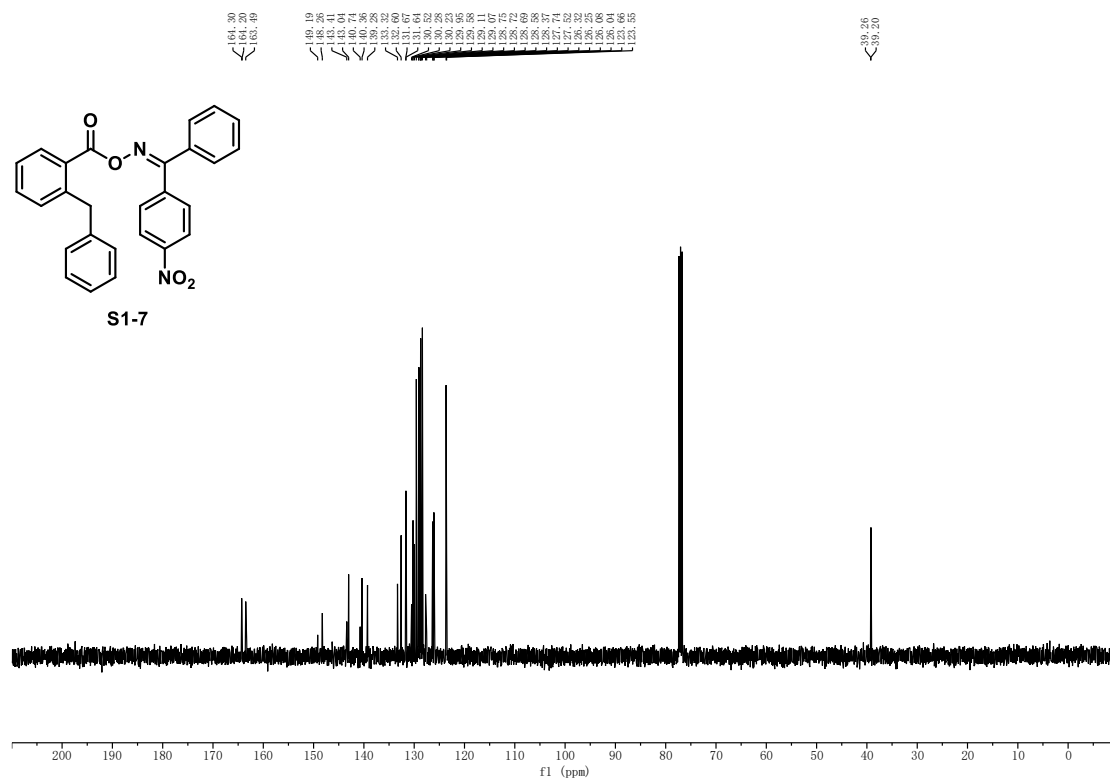
Supplementary Figure 30 ¹³C NMR spectrum of S1-6



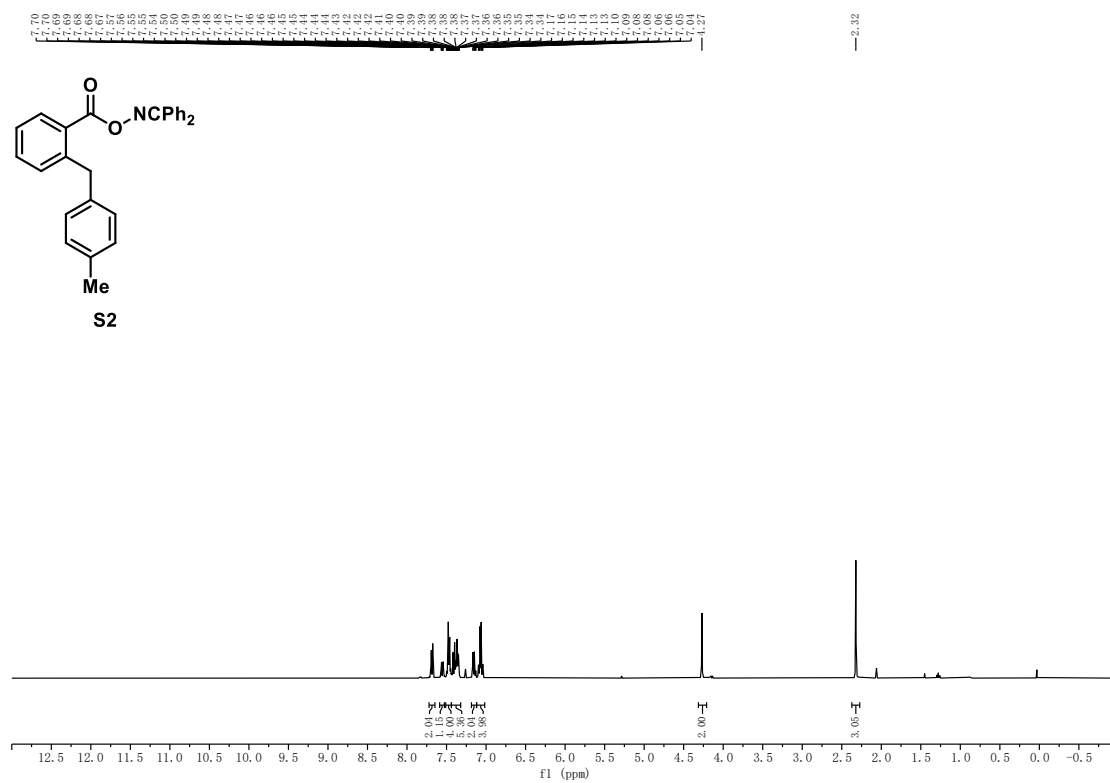
Supplementary Figure 31 ¹⁹F NMR spectrum of S1-6



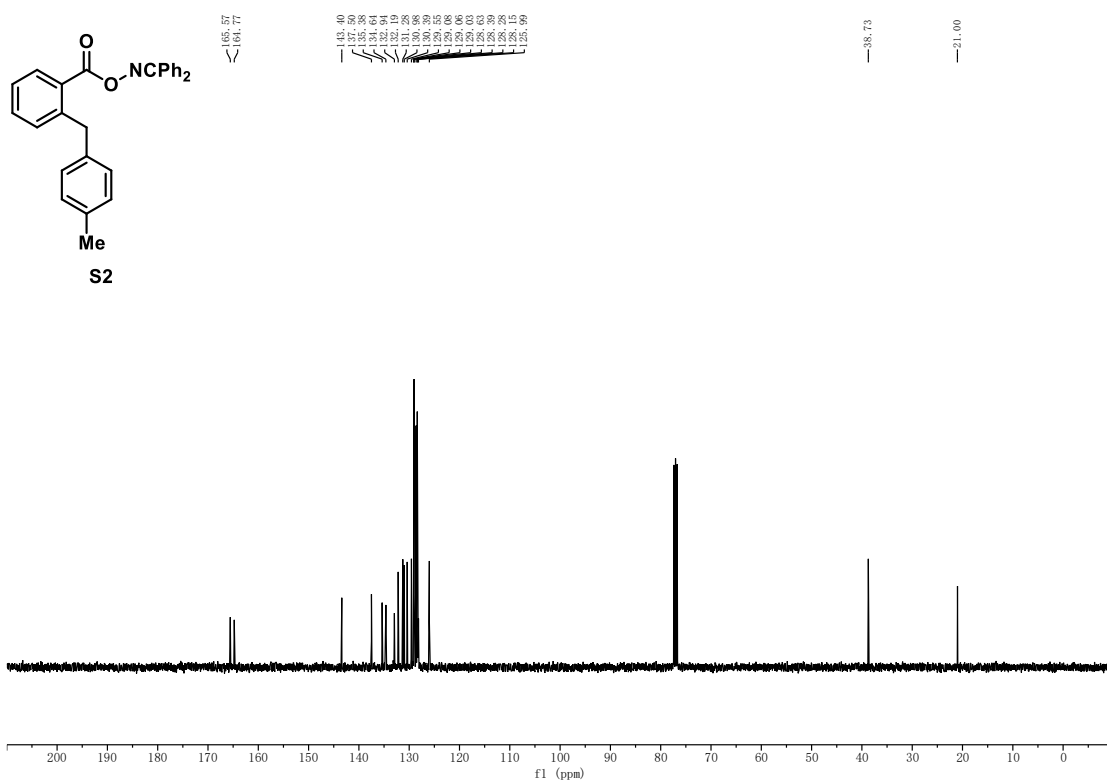
Supplementary Figure 32 ¹H NMR spectrum of S1-7



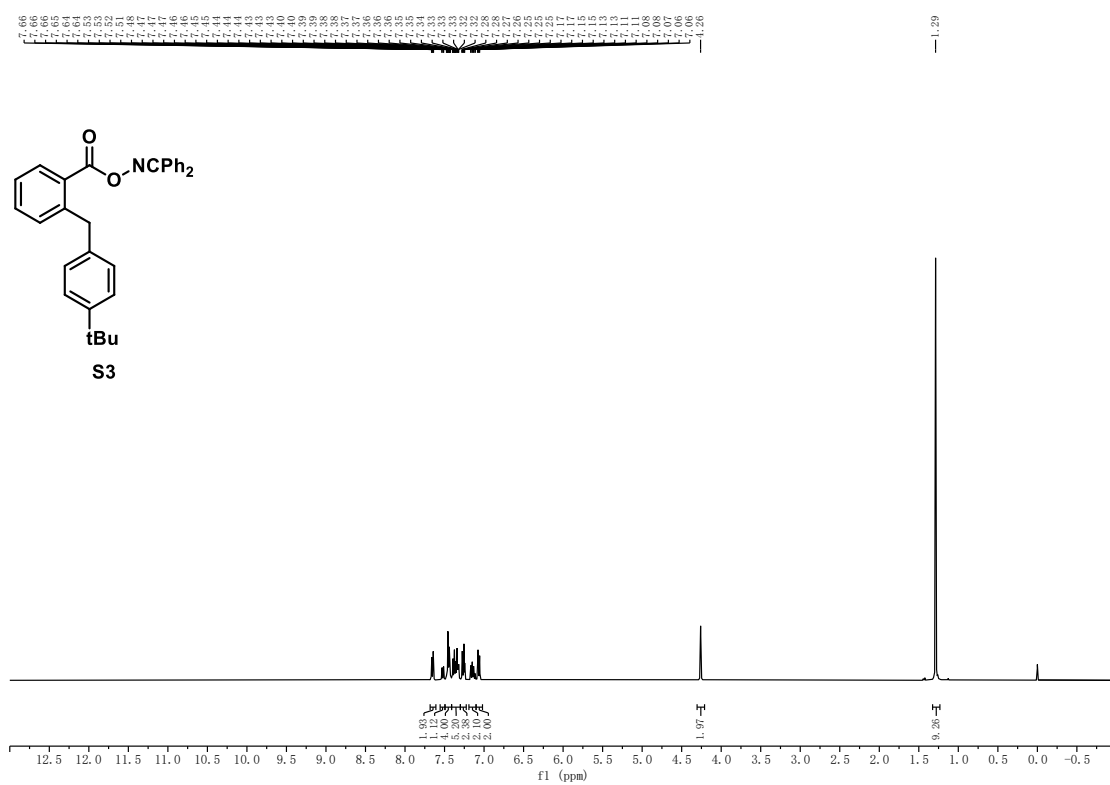
Supplementary Figure 33 ^{13}C NMR spectrum of S1-7



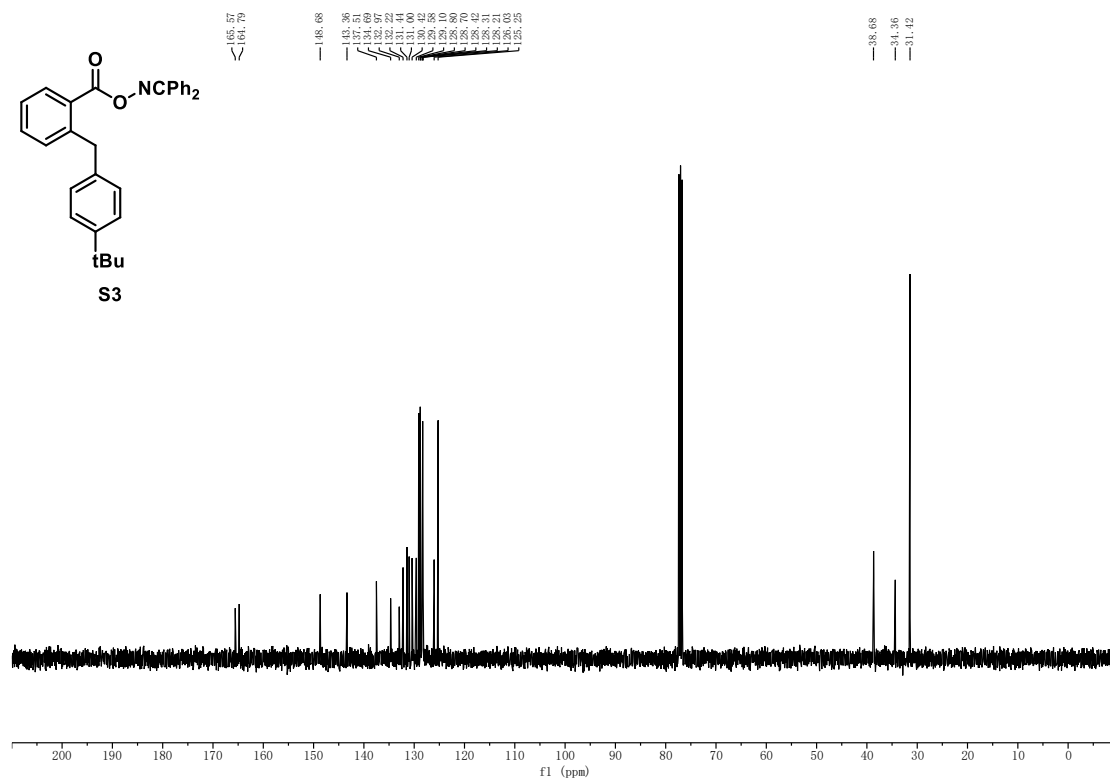
Supplementary Figure 34 ^1H NMR spectrum of S2



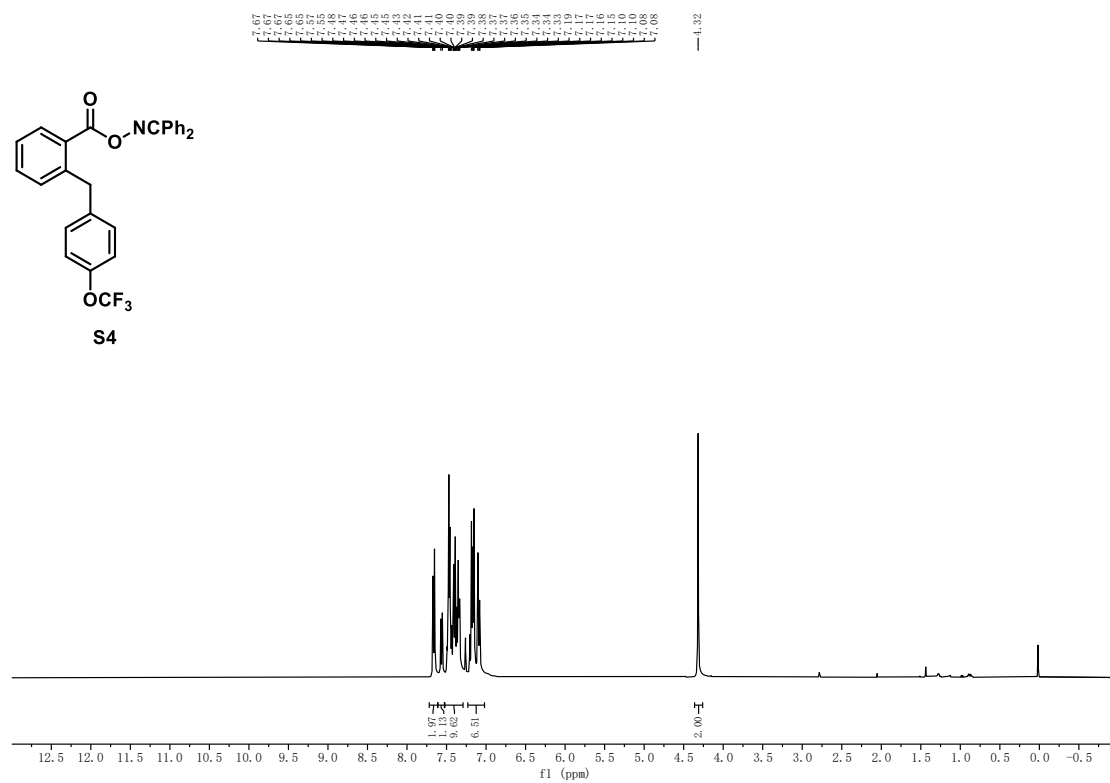
Supplementary Figure 35 ^{13}C NMR spectrum of S2



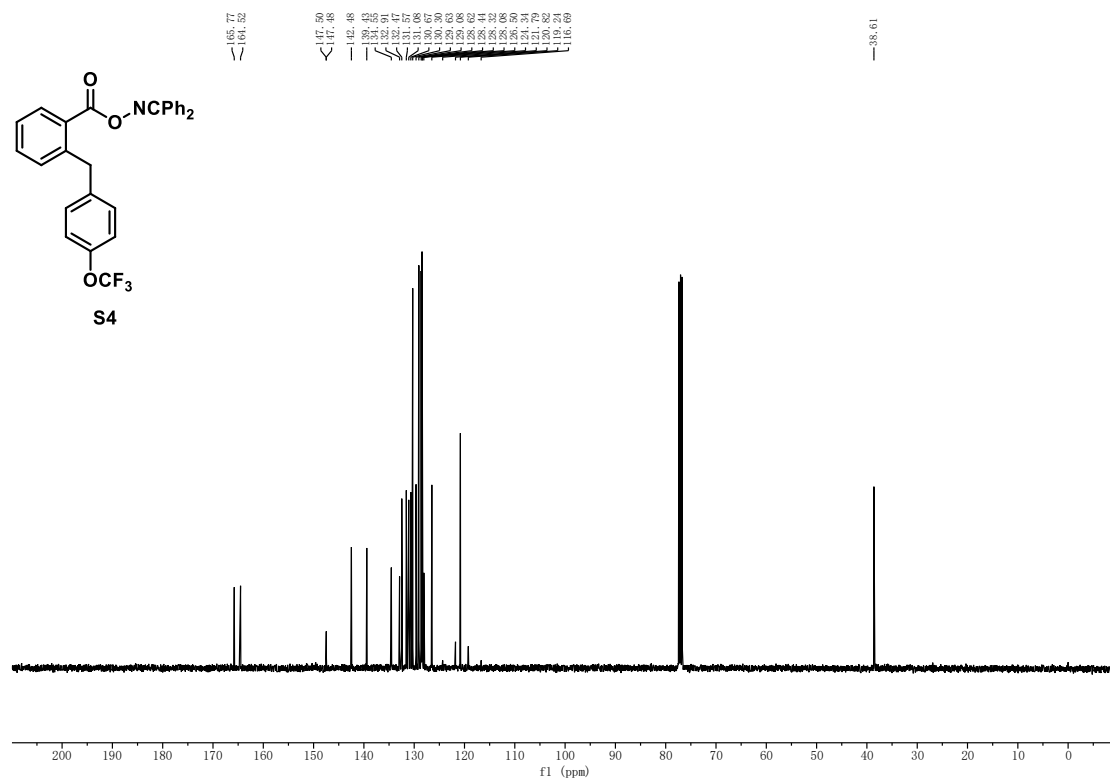
Supplementary Figure 36 ^1H NMR spectrum of S3



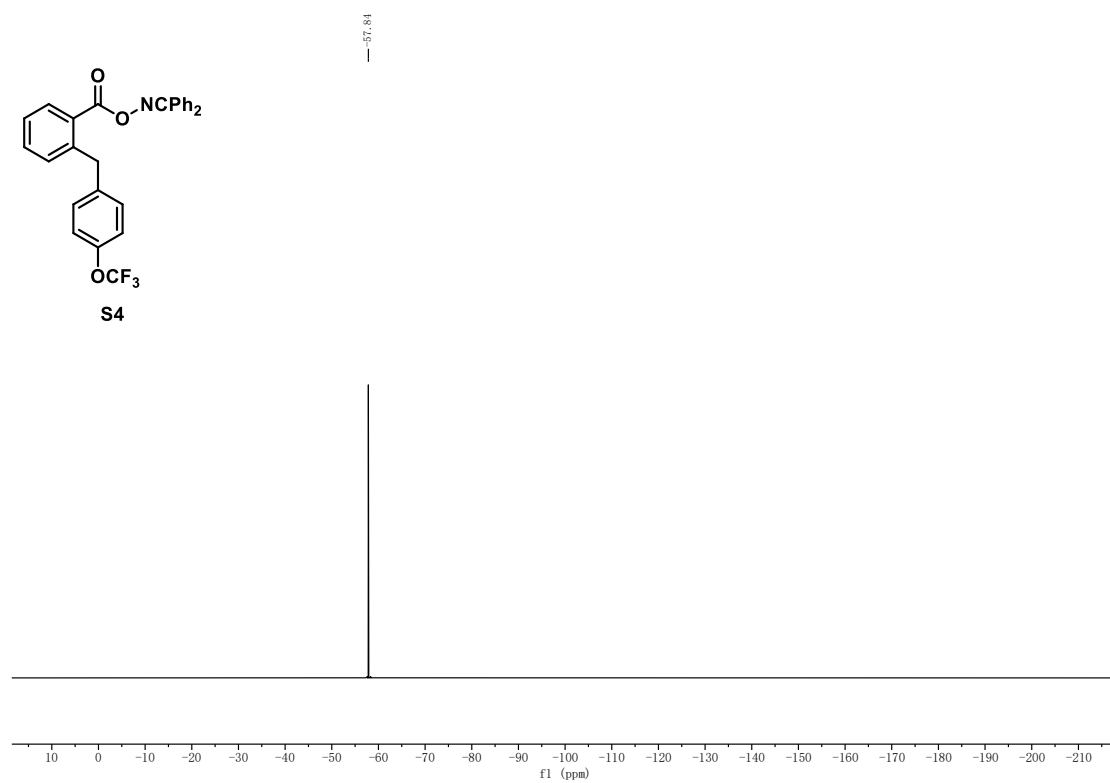
Supplementary Figure 37 ¹³C NMR spectrum of S3



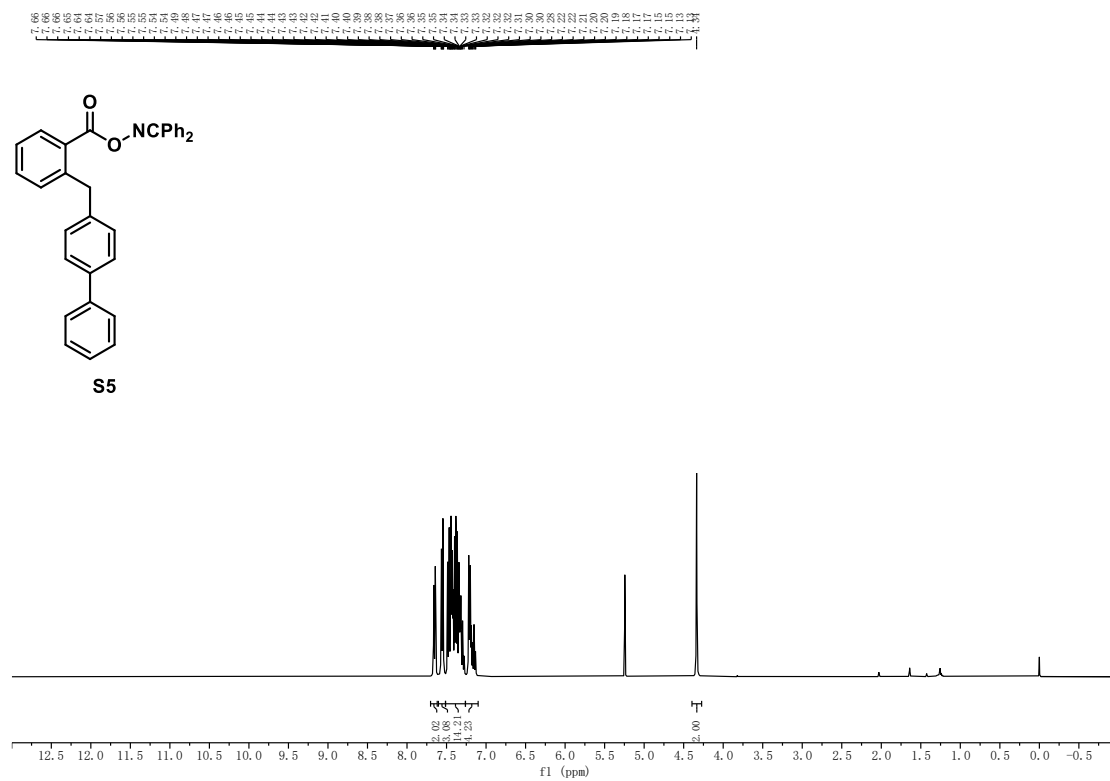
Supplementary Figure 38 ¹H NMR spectrum of S4



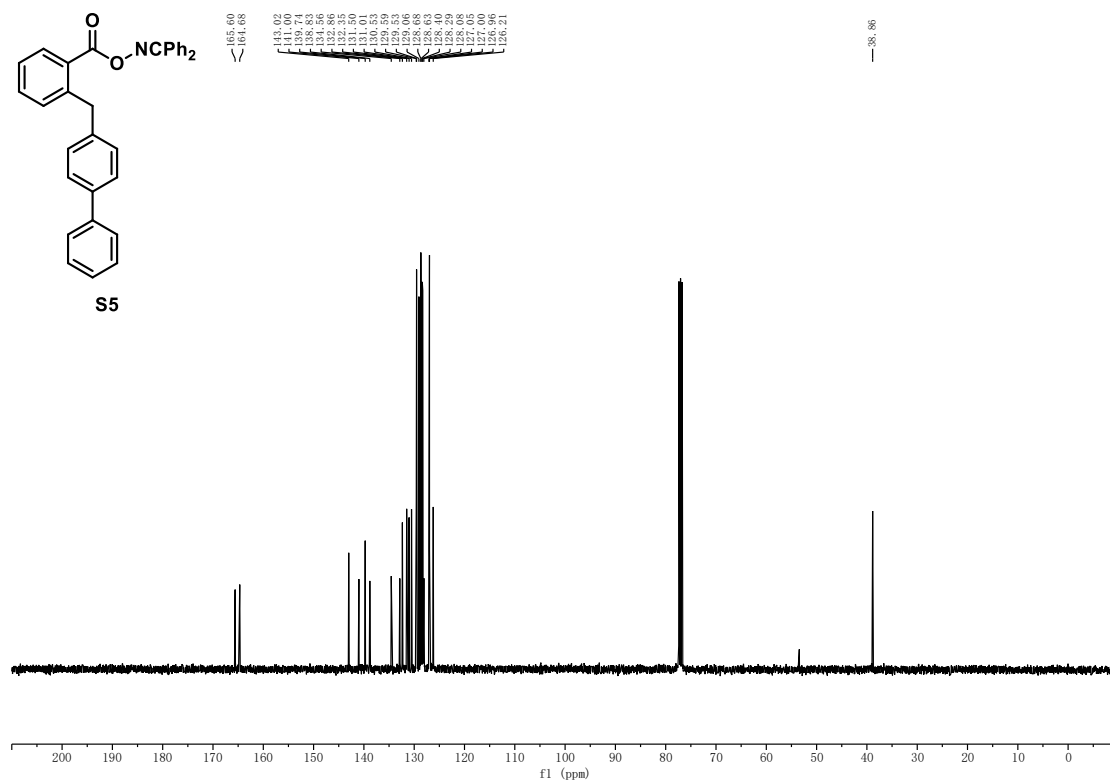
Supplementary Figure 39 ^{13}C NMR spectrum of S4



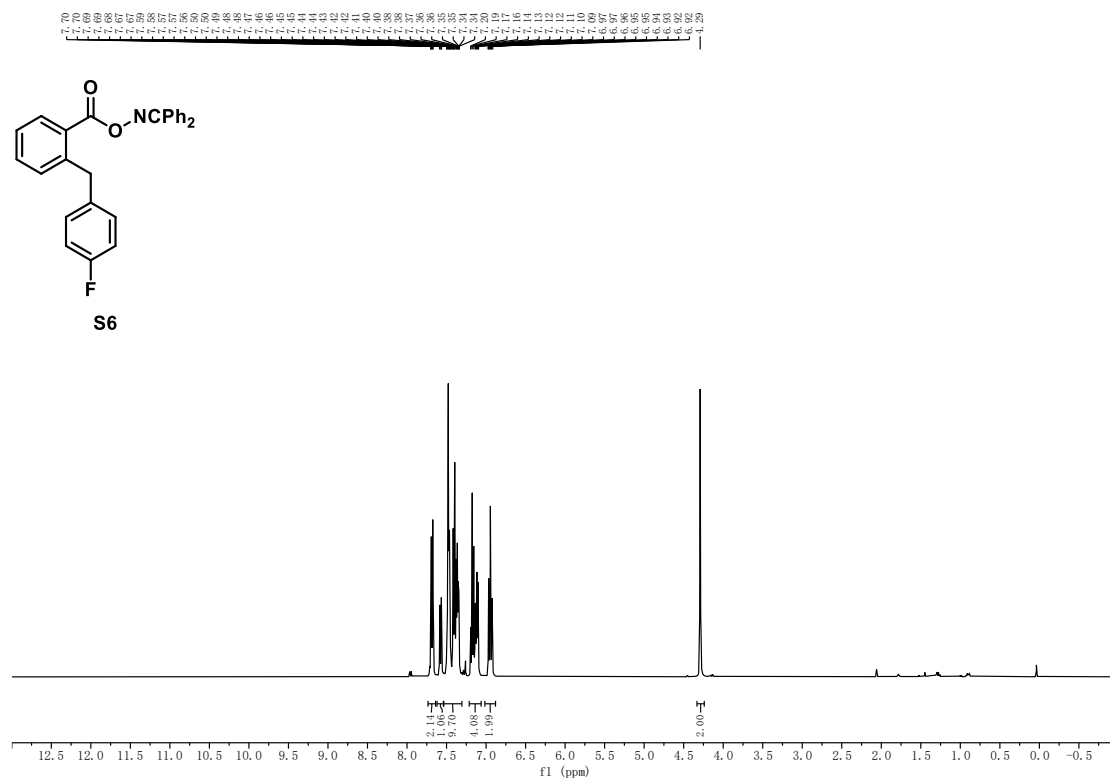
Supplementary Figure 40 ^{19}F NMR spectrum of S4



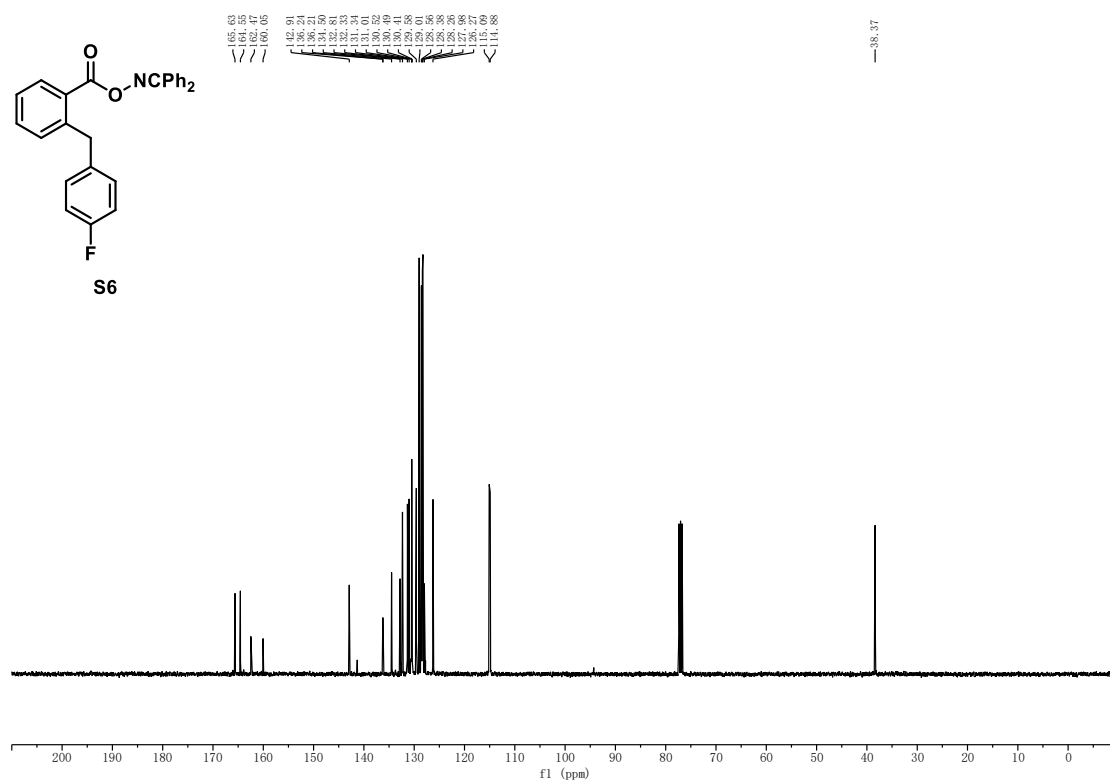
Supplementary Figure 41 ¹H NMR spectrum of S5



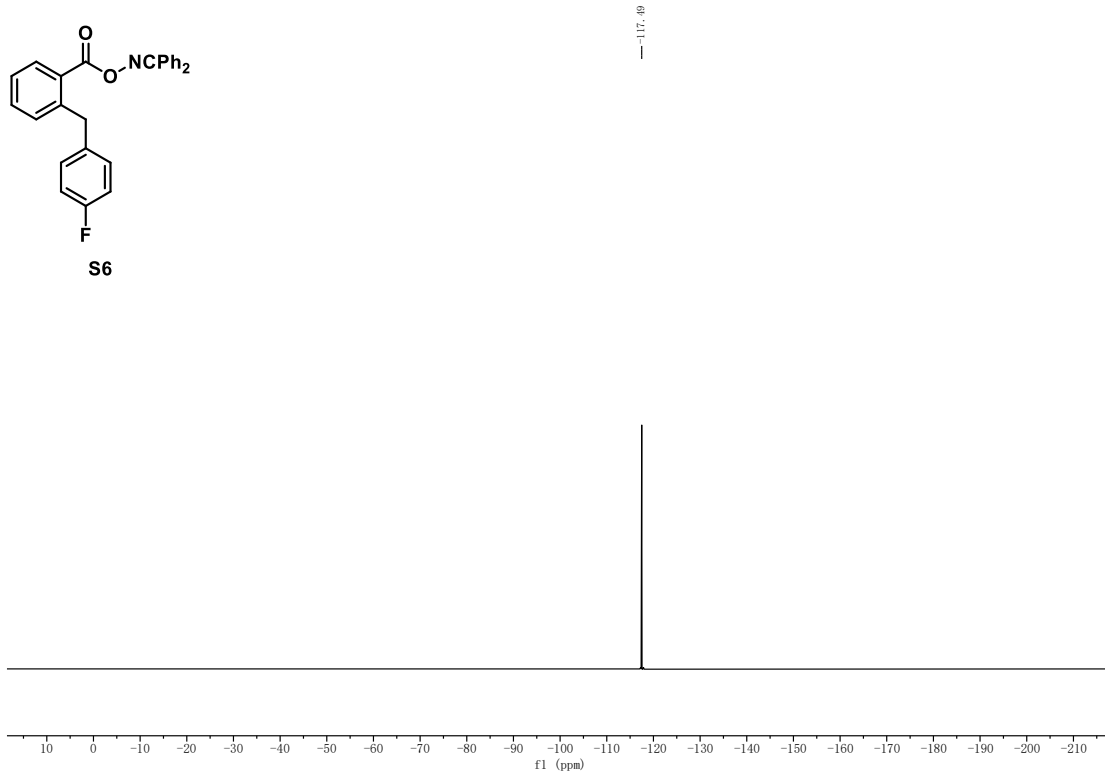
Supplementary Figure 42 ¹³C NMR spectrum of S5



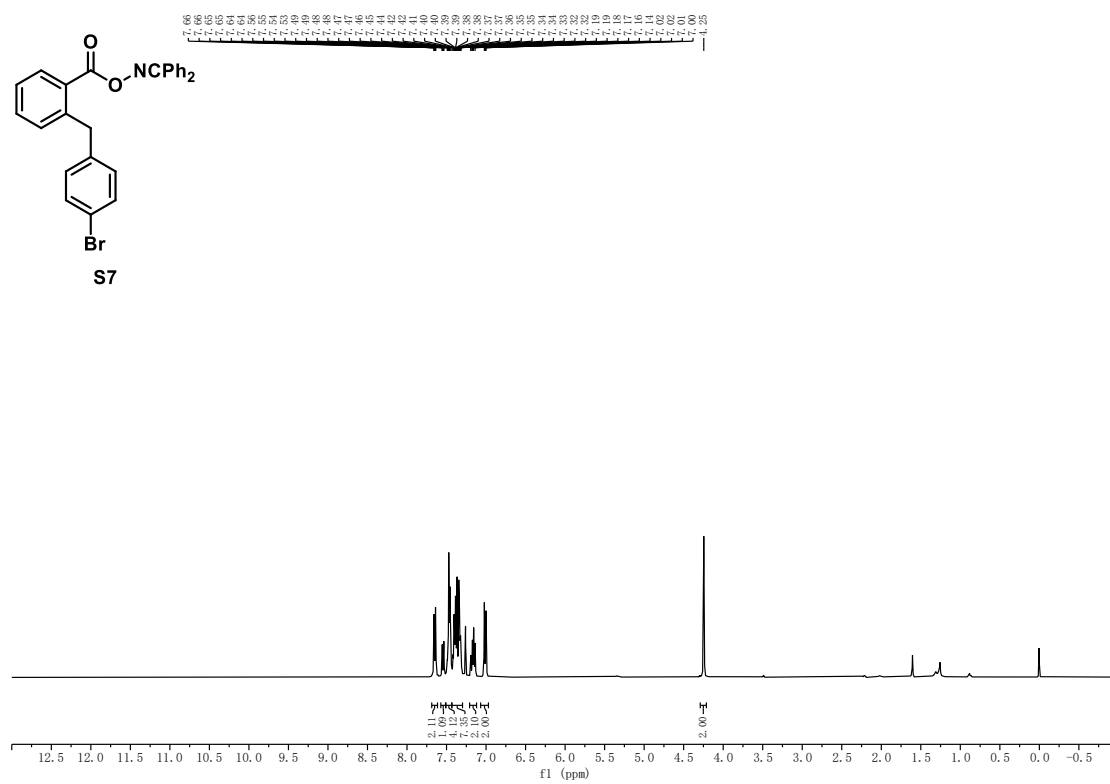
Supplementary Figure 43 ^1H NMR spectrum of S6



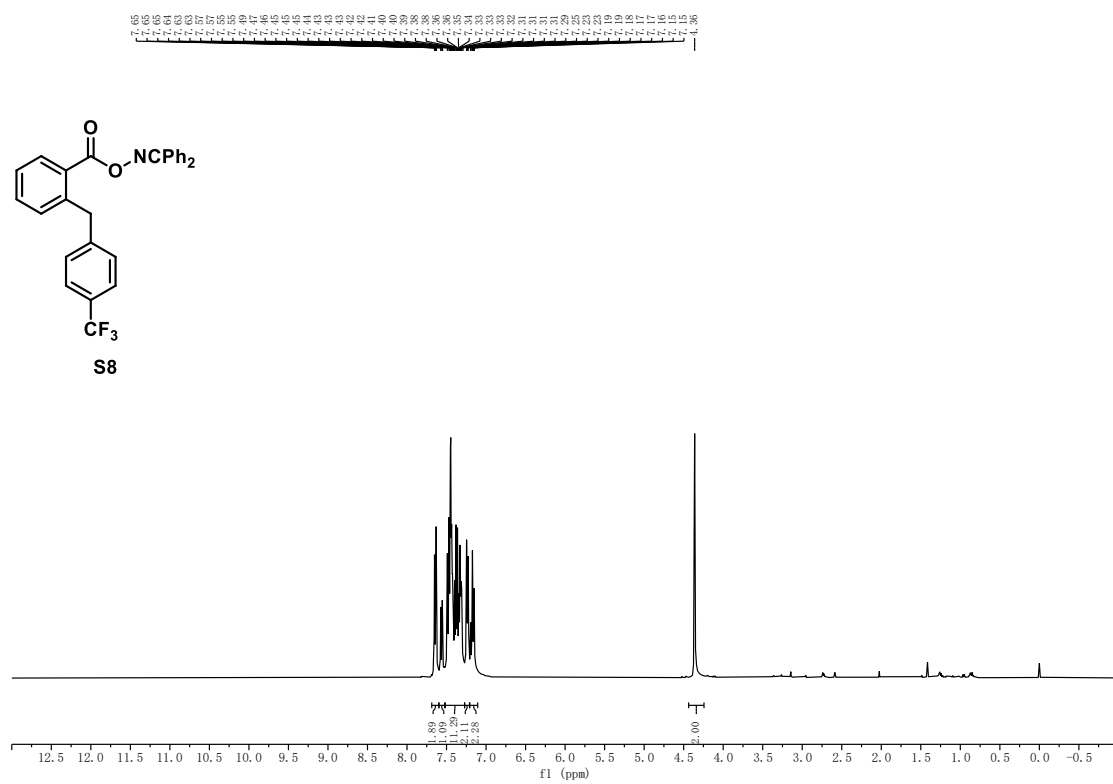
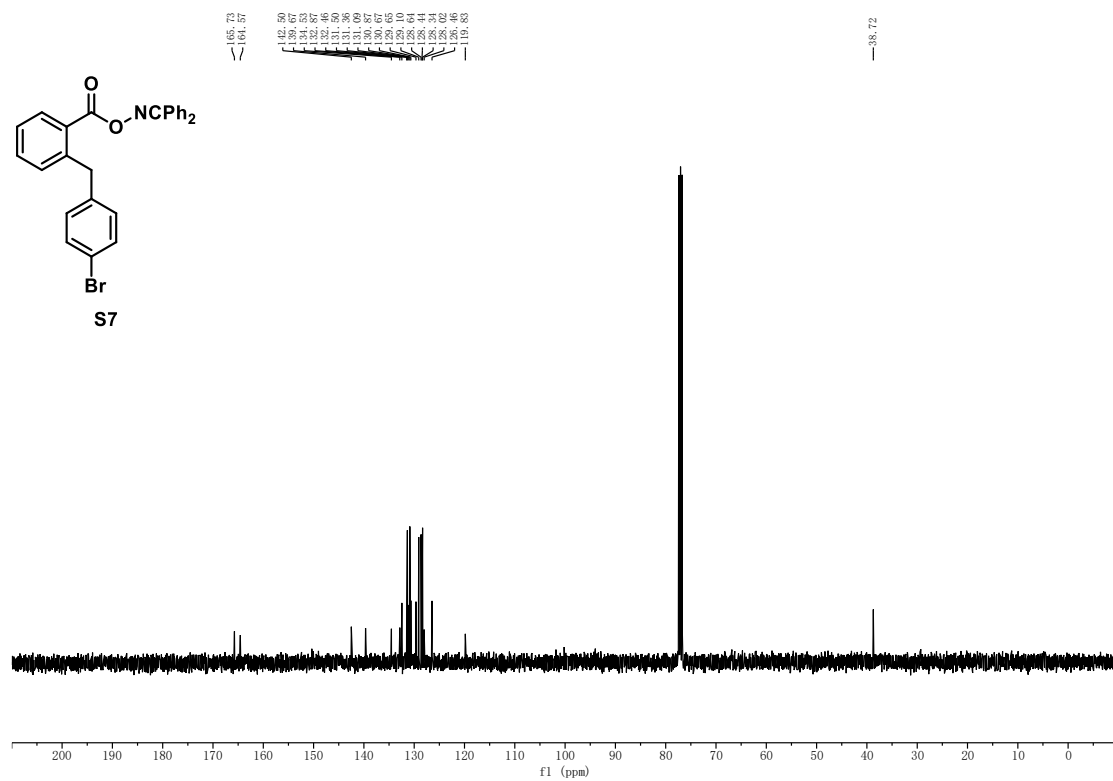
Supplementary Figure 44 ^{13}C NMR spectrum of S6

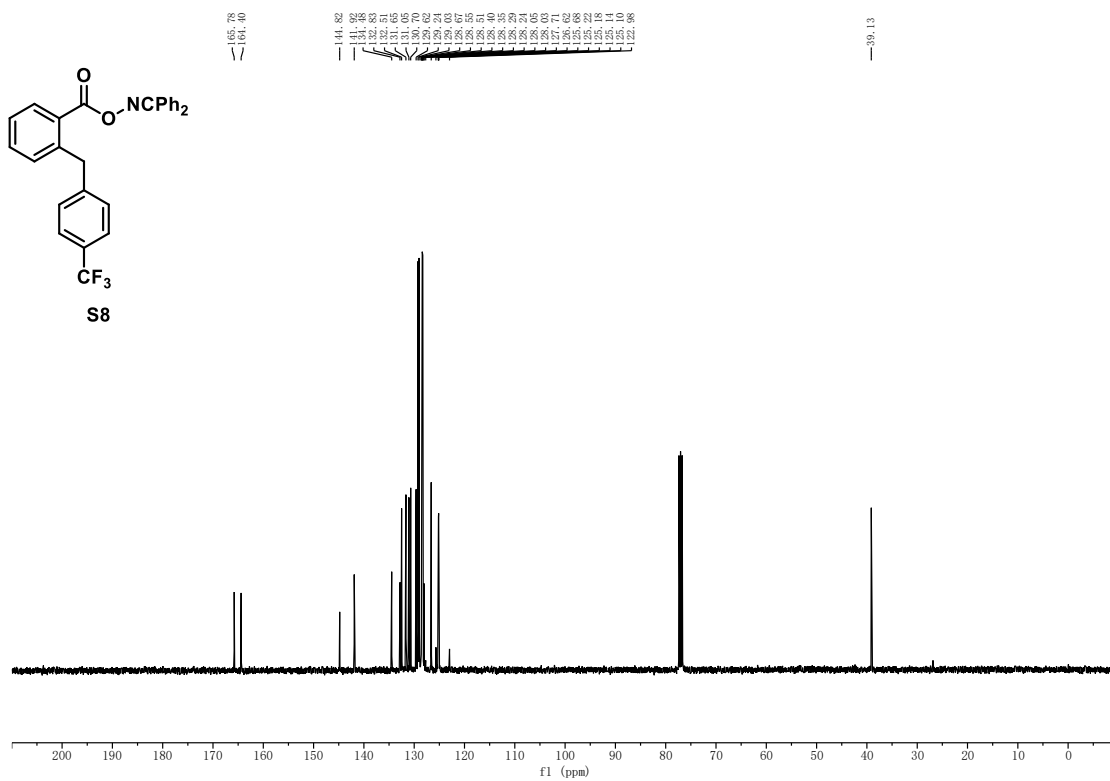


Supplementary Figure 45 ^{19}F NMR spectrum of S6

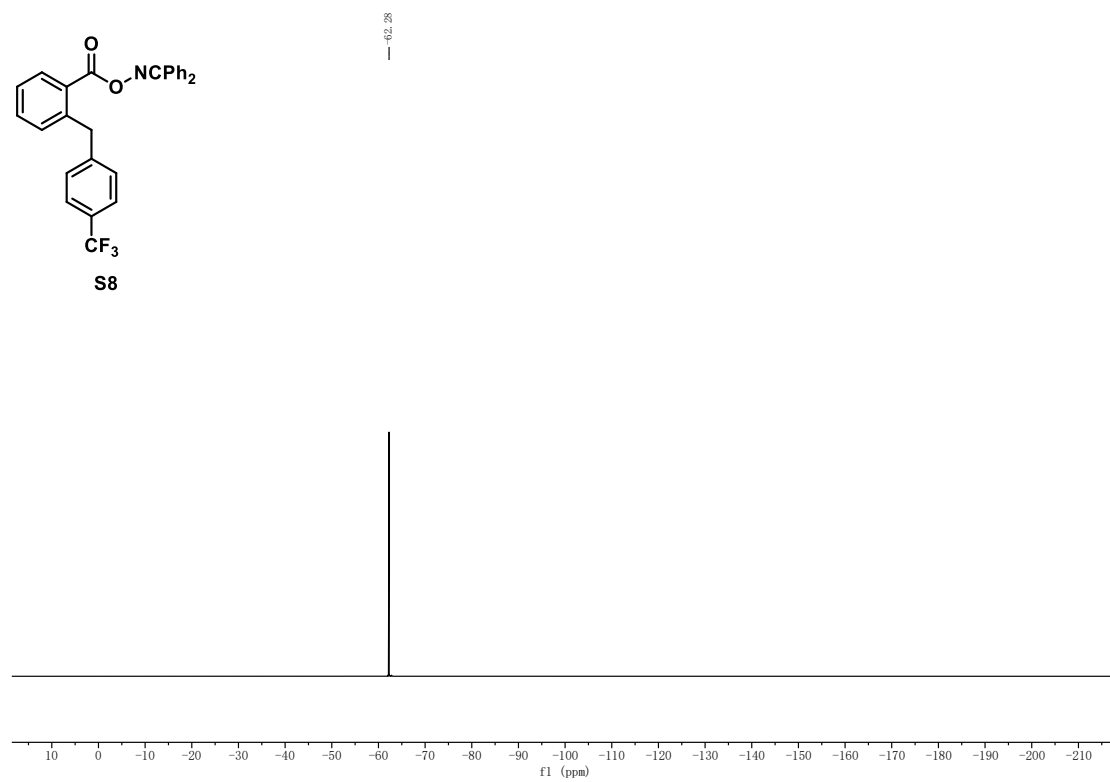


Supplementary Figure 46 ^1H NMR spectrum of S7

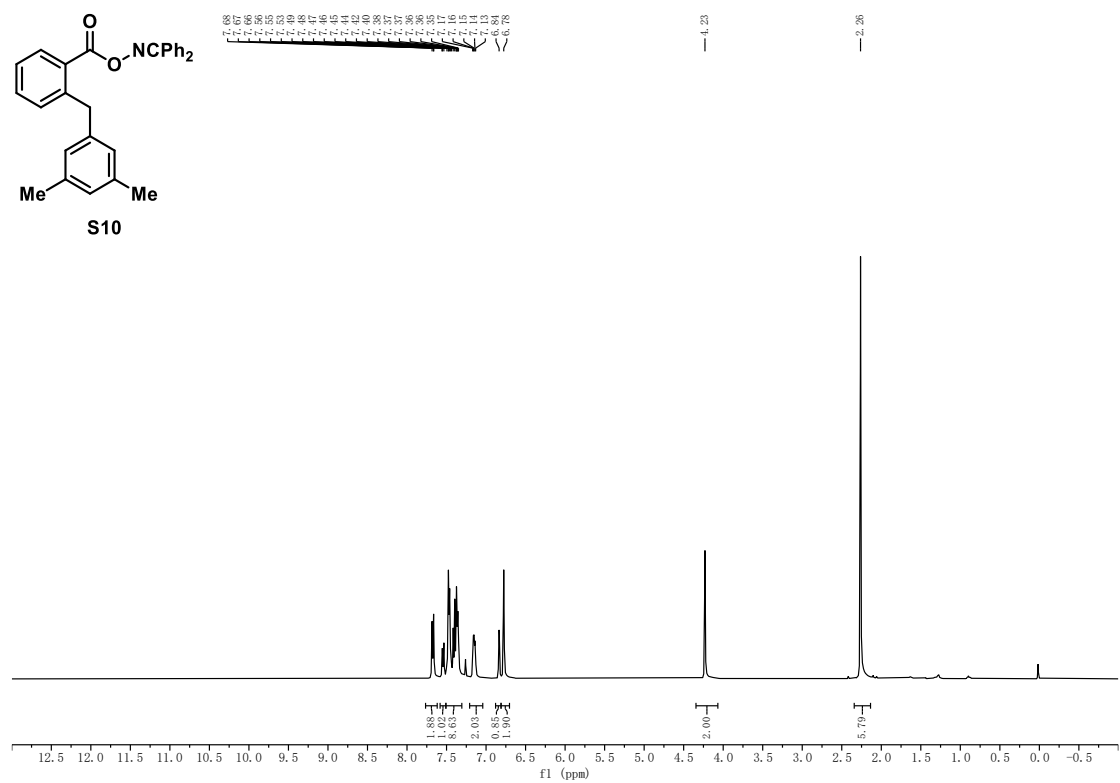




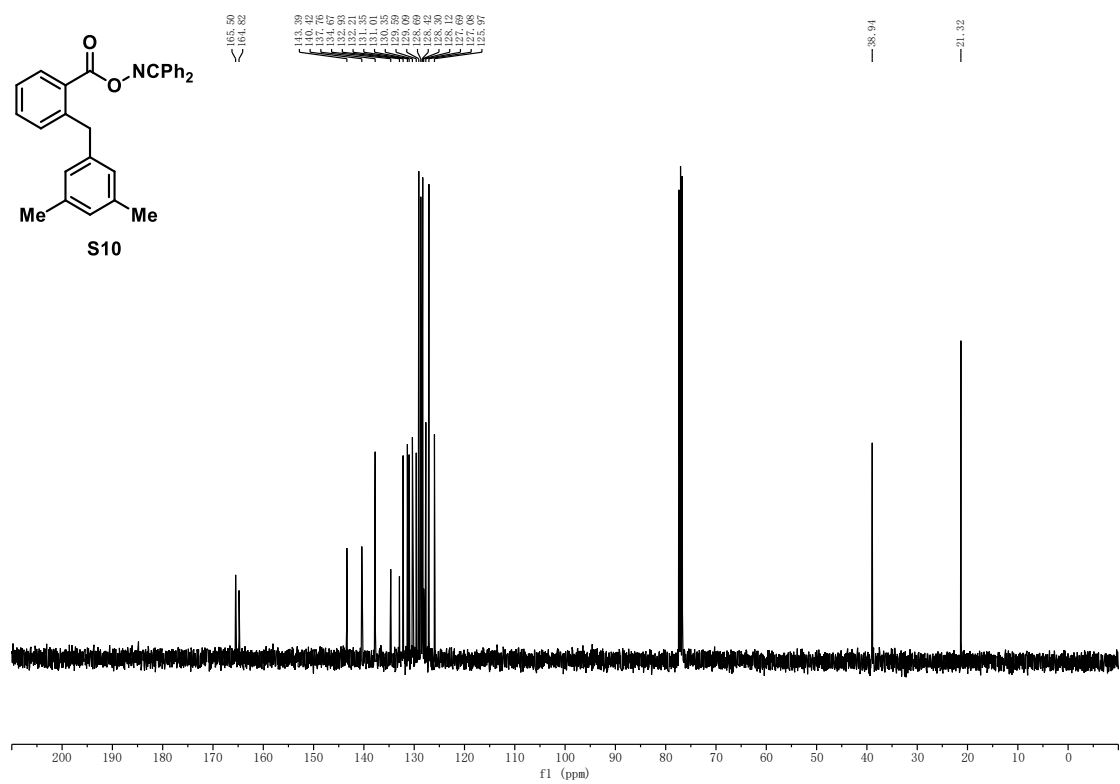
Supplementary Figure 49 ^{13}C NMR spectrum of S8



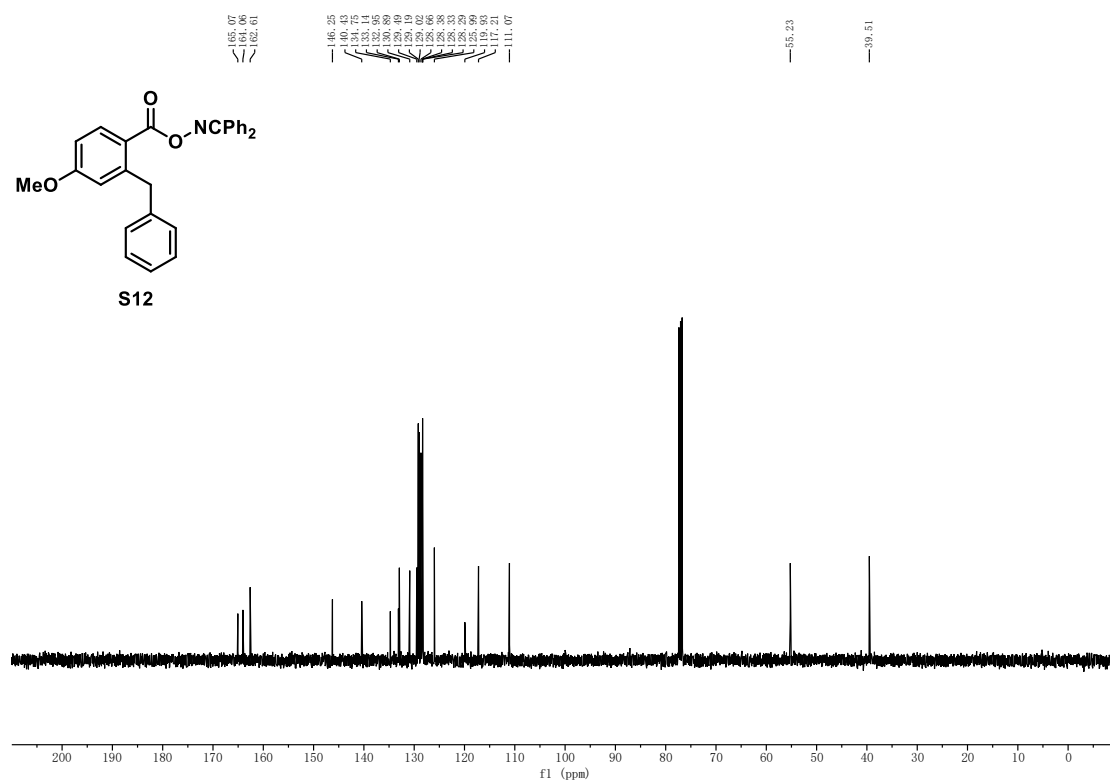
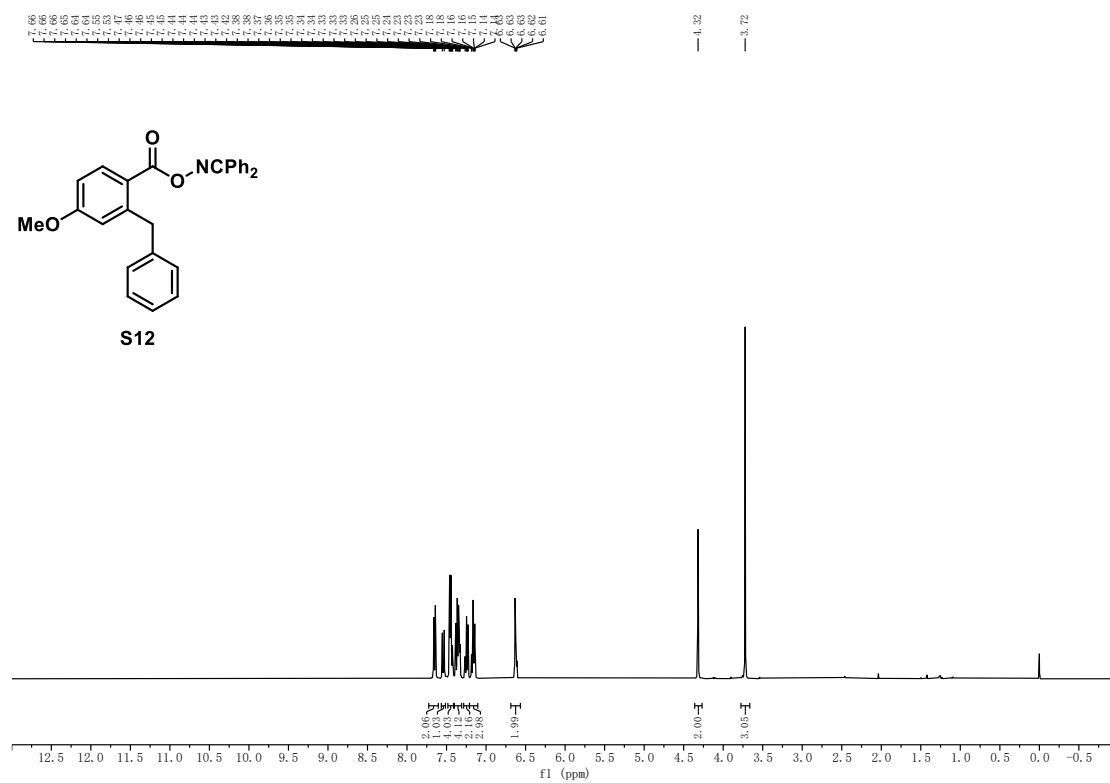
Supplementary Figure 50 ^{19}F NMR spectrum of S8

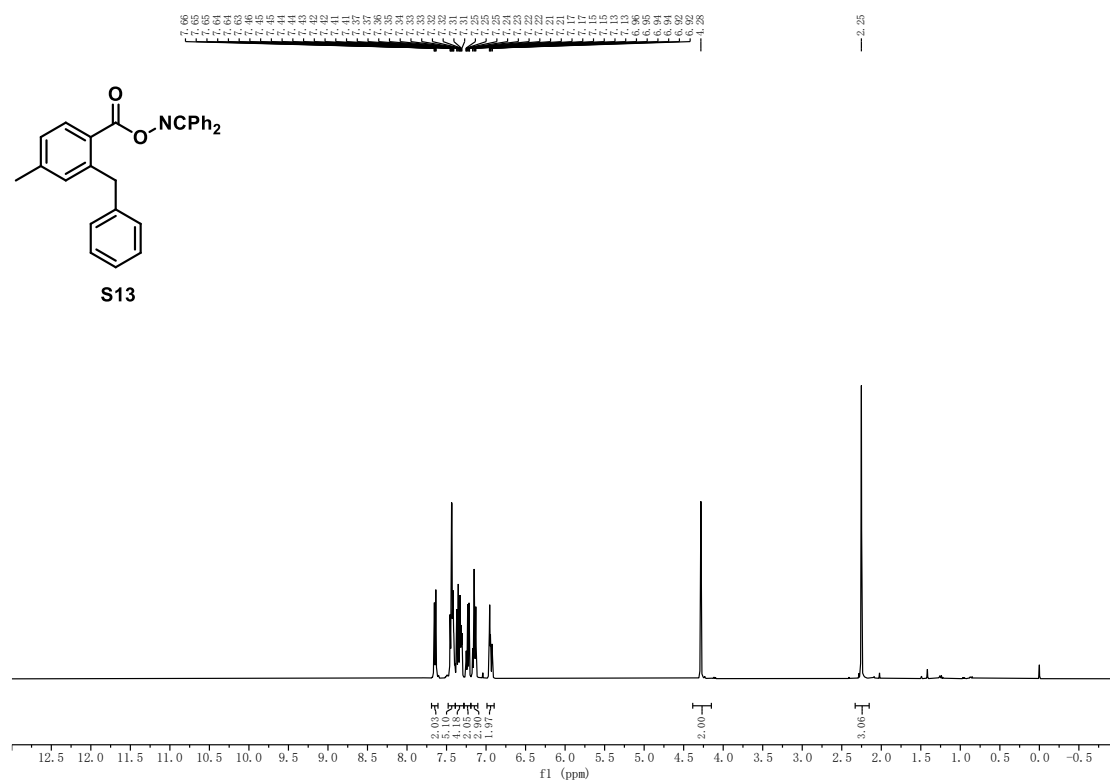


Supplementary Figure 53 ^1H NMR spectrum of S10

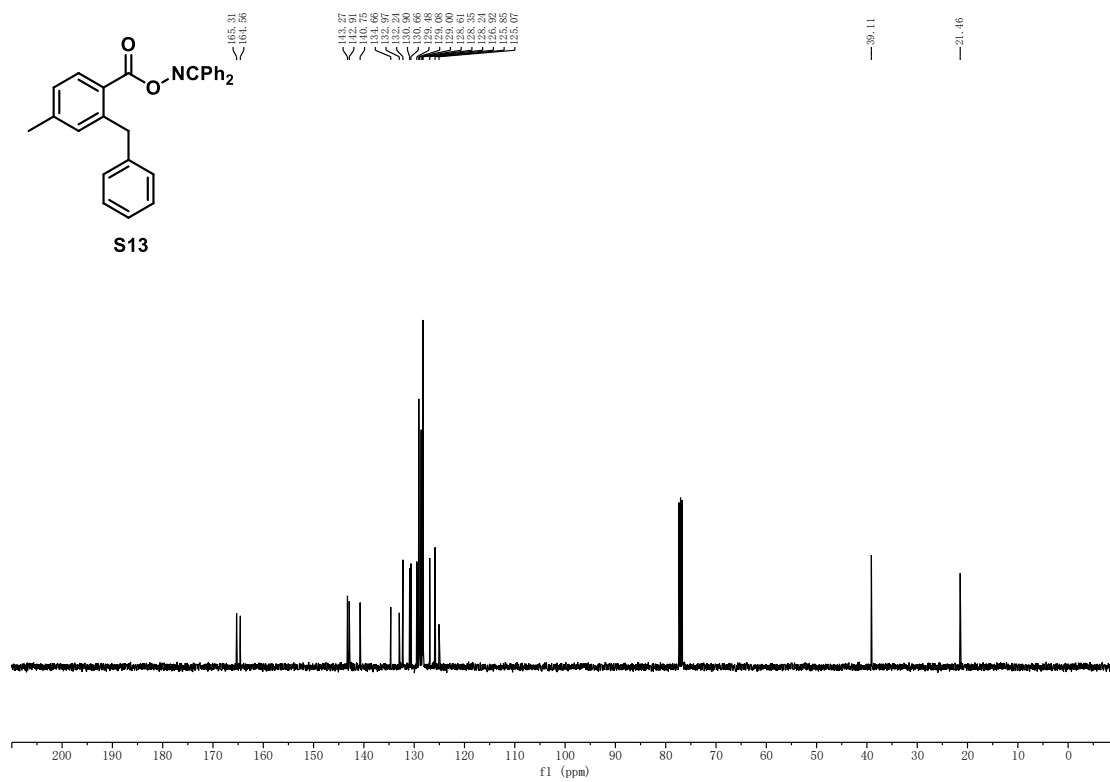


Supplementary Figure 54 ^{13}C NMR spectrum of S10

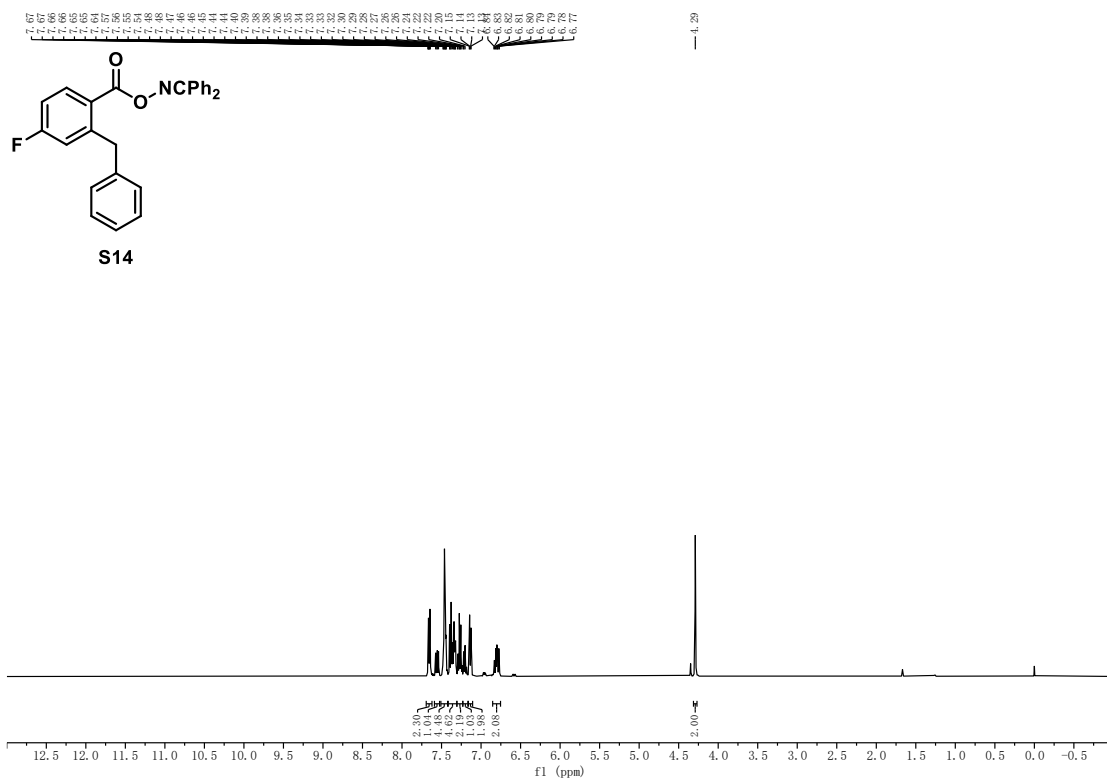




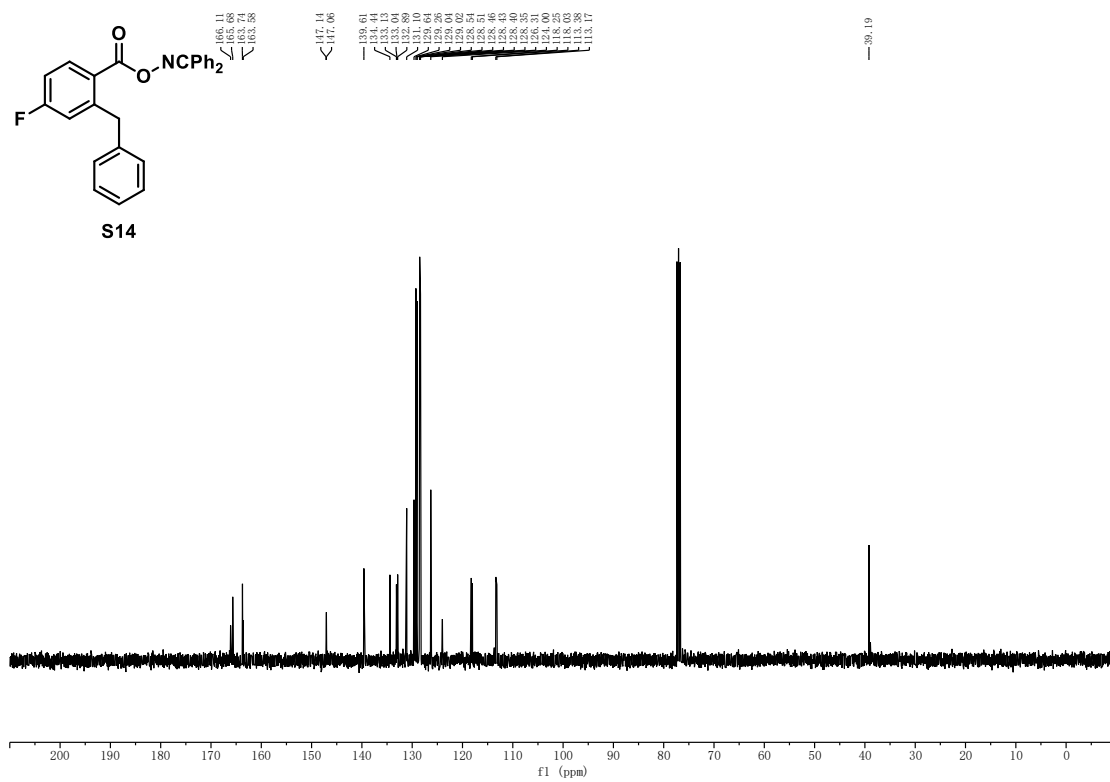
Supplementary Figure 59 ¹H NMR spectrum of S13



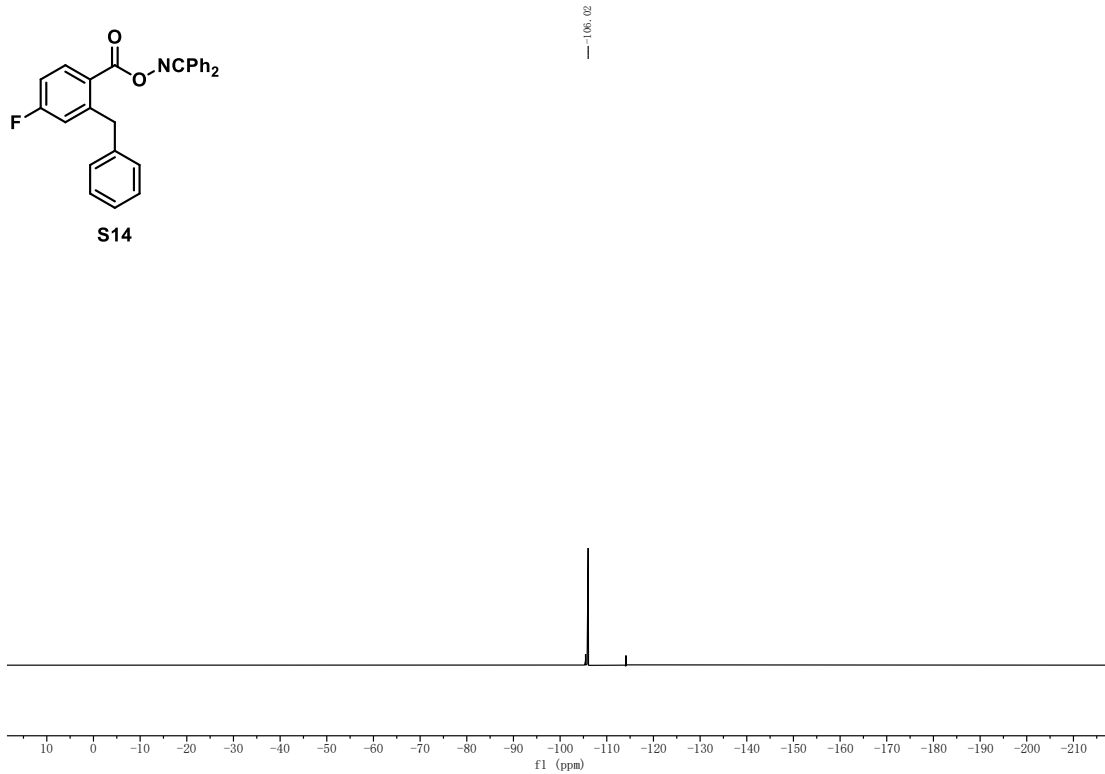
Supplementary Figure 60 ¹³C NMR spectrum of S13



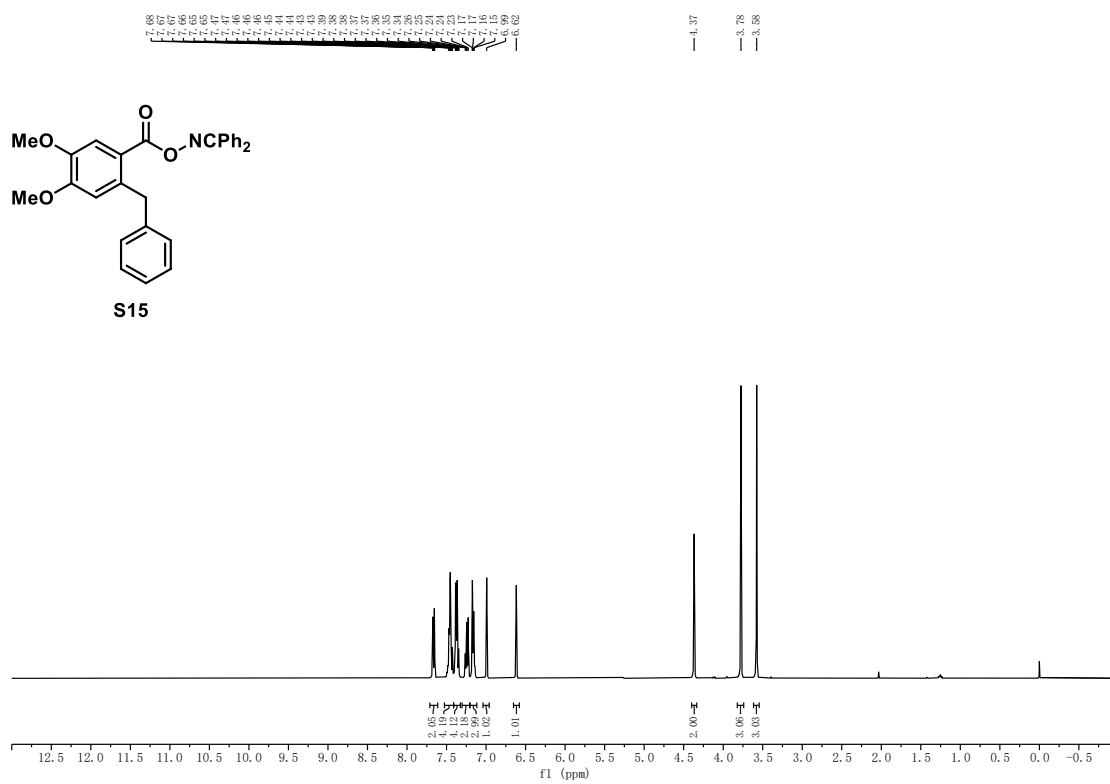
Supplementary Figure 61 ¹H NMR spectrum of S14



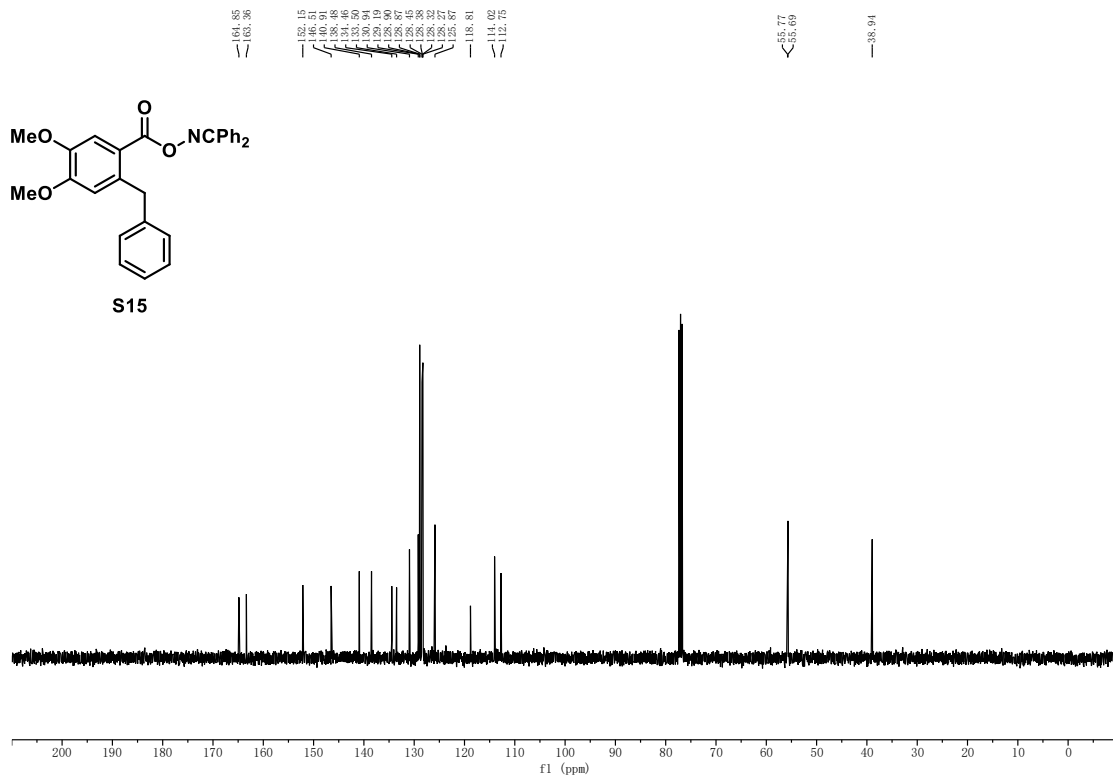
Supplementary Figure 62 ¹³C NMR spectrum of S14



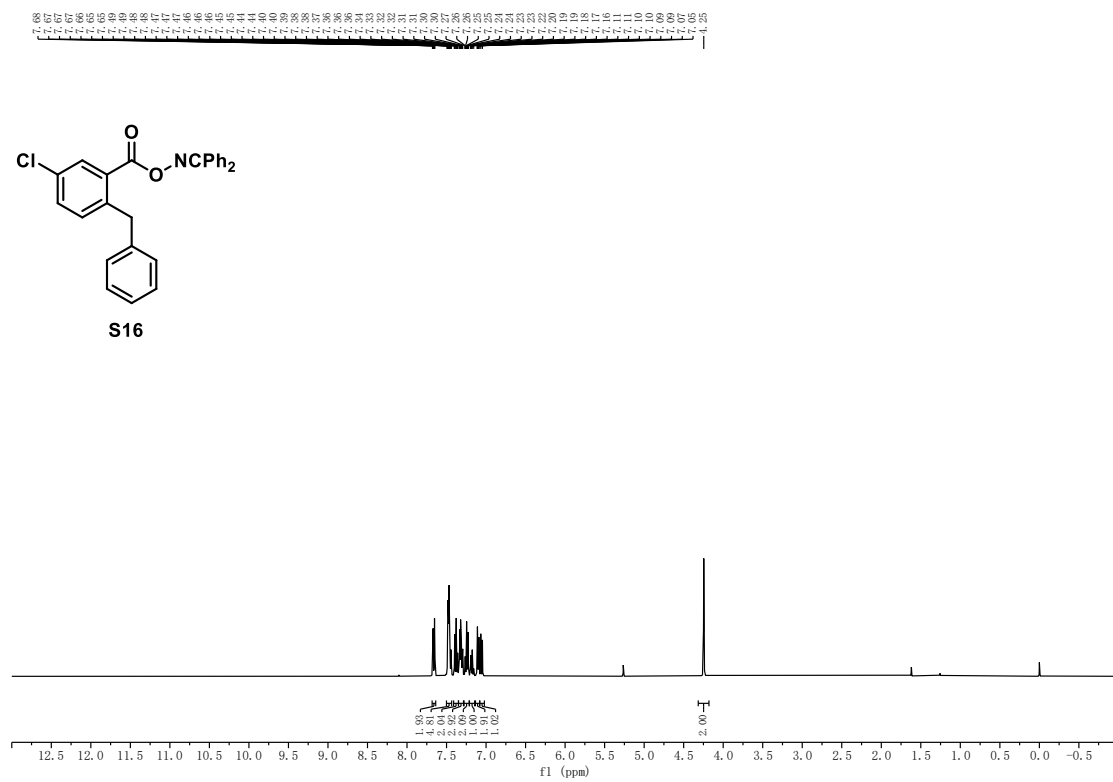
Supplementary Figure 63 ^{19}F NMR spectrum of **S14**



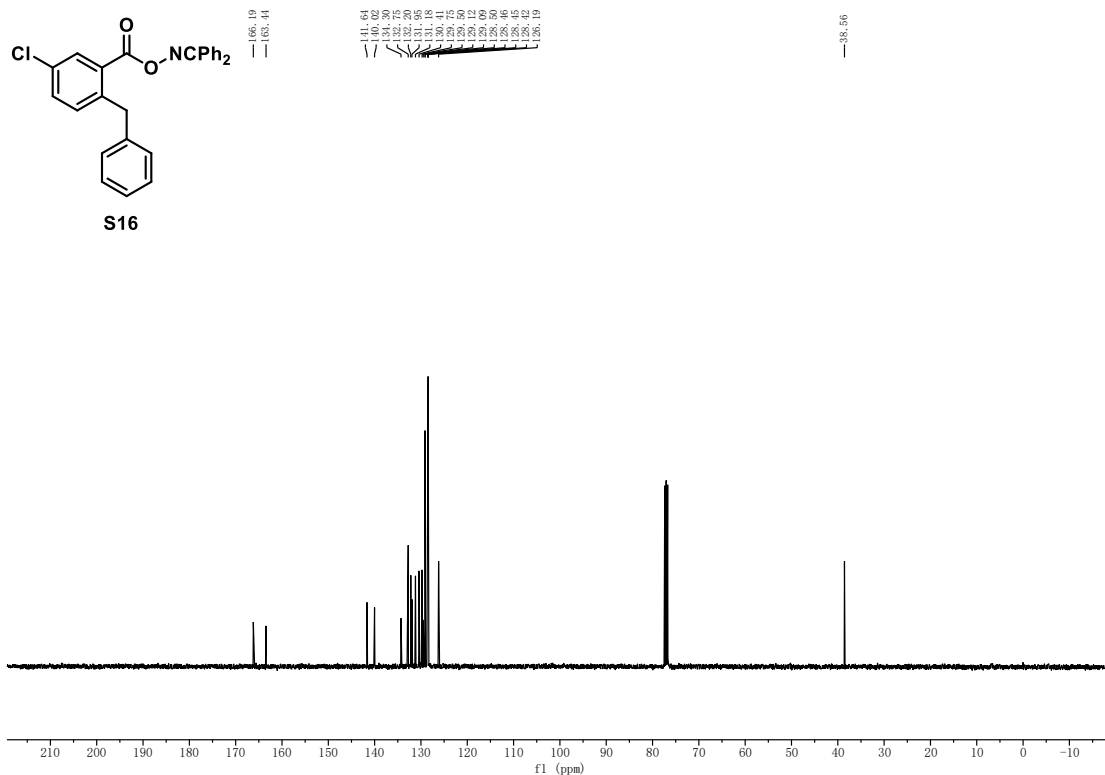
Supplementary Figure 64 ^1H NMR spectrum of **S15**



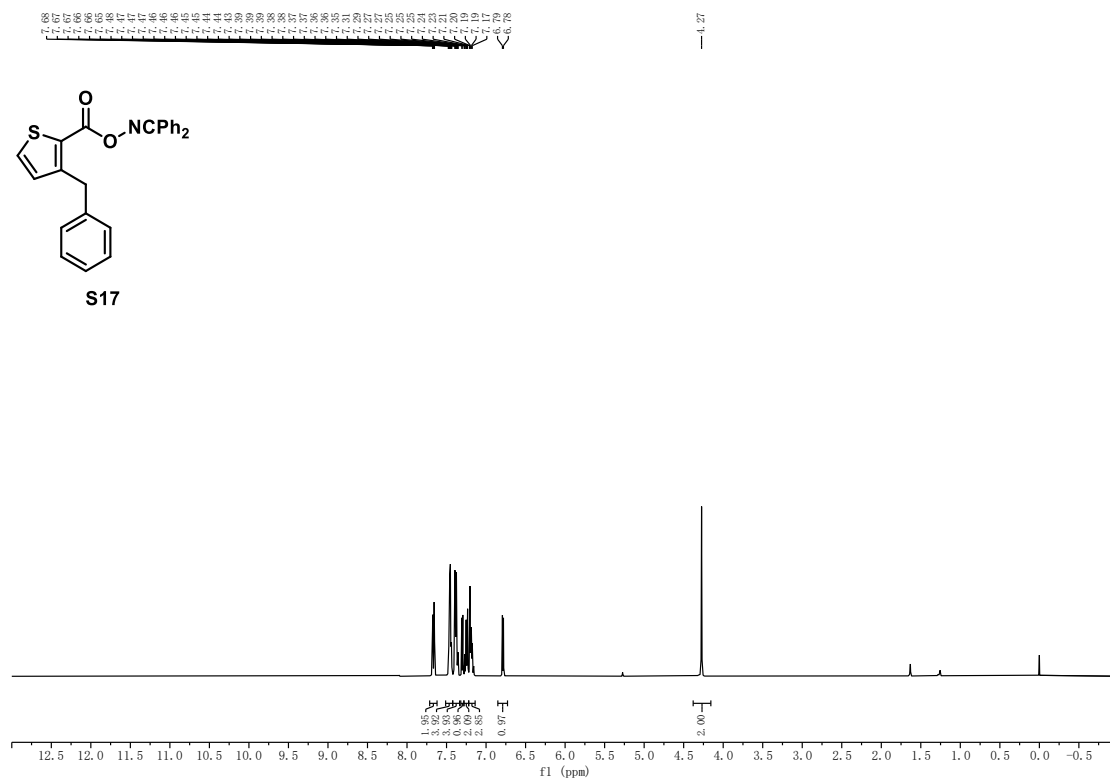
Supplementary Figure 65 ^{13}C NMR spectrum of S15



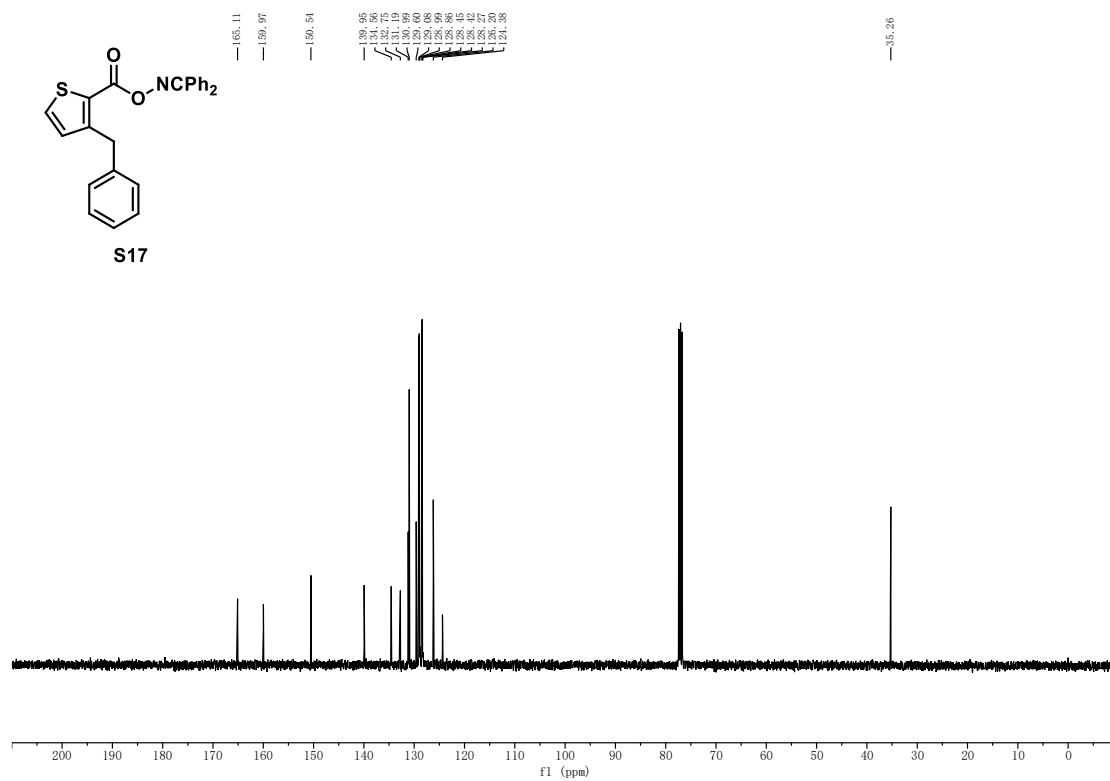
Supplementary Figure 66 ^1H NMR spectrum of S16



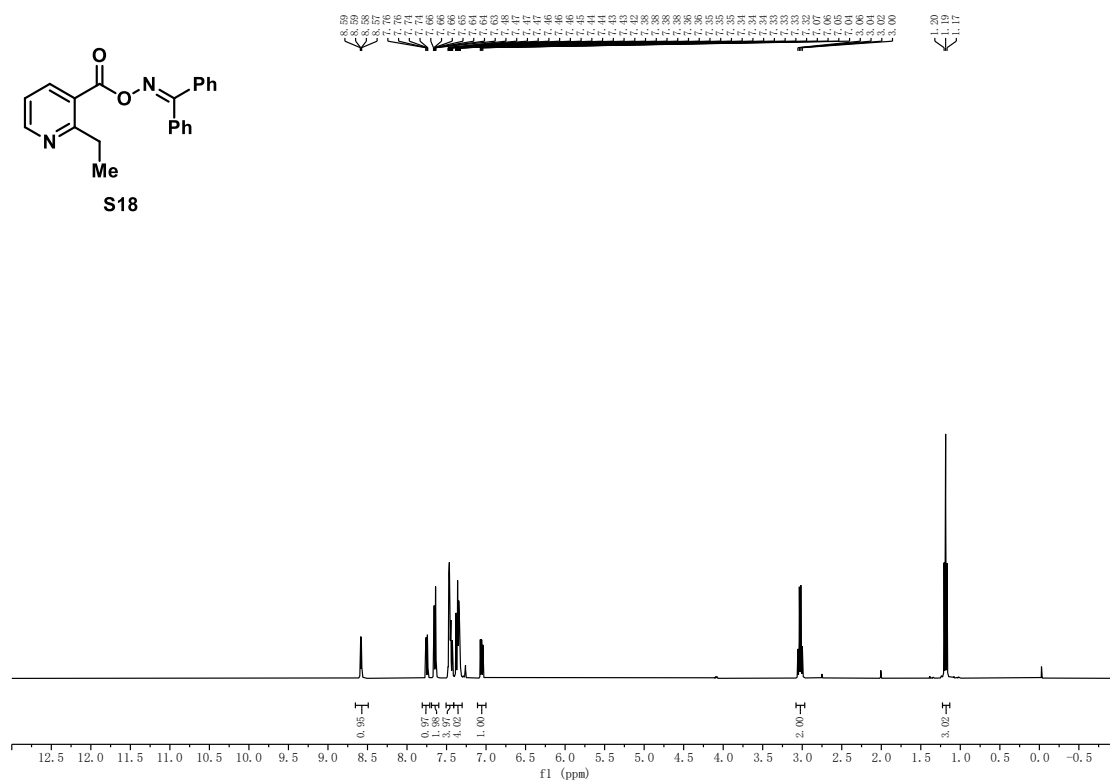
Supplementary Figure 67 ¹³C NMR spectrum of S16



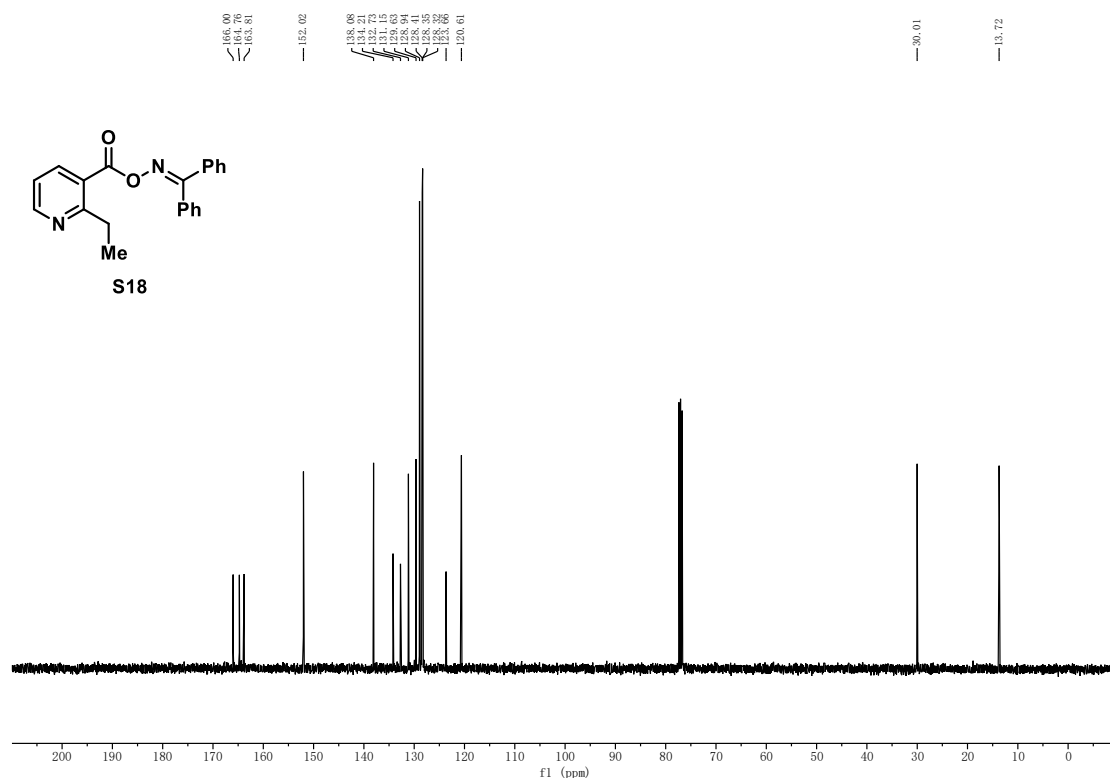
Supplementary Figure 68 ¹H NMR spectrum of S17



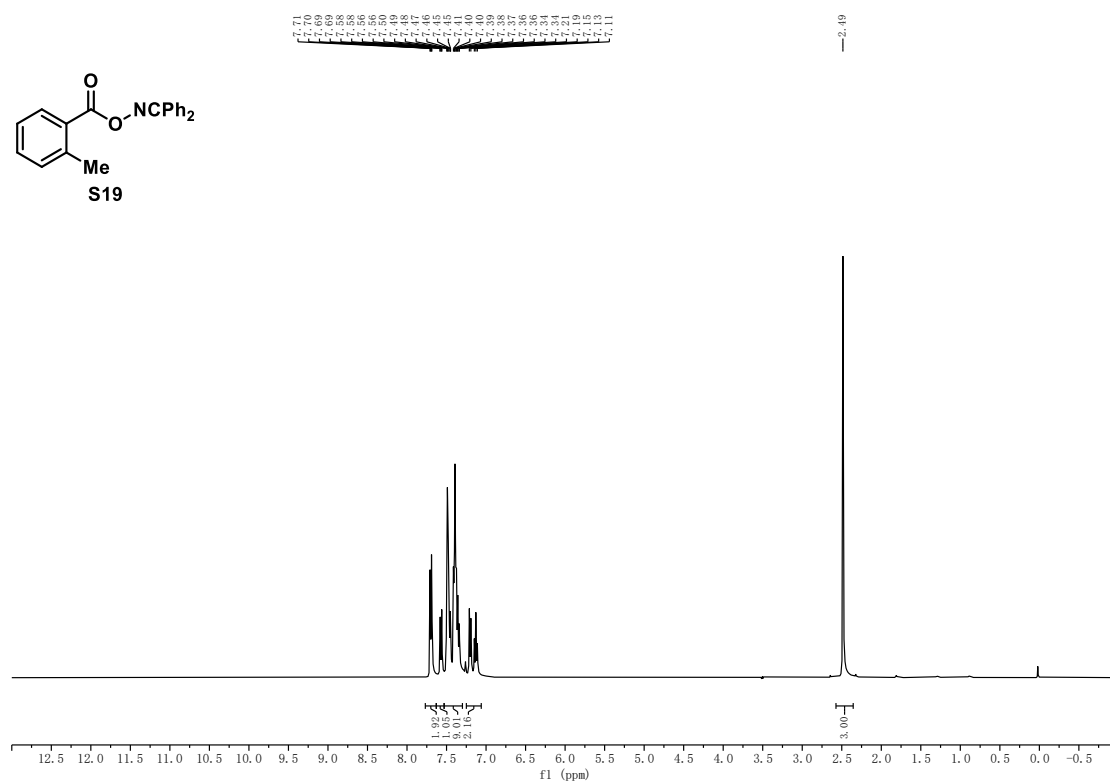
Supplementary Figure 69 ^{13}C NMR spectrum of S17



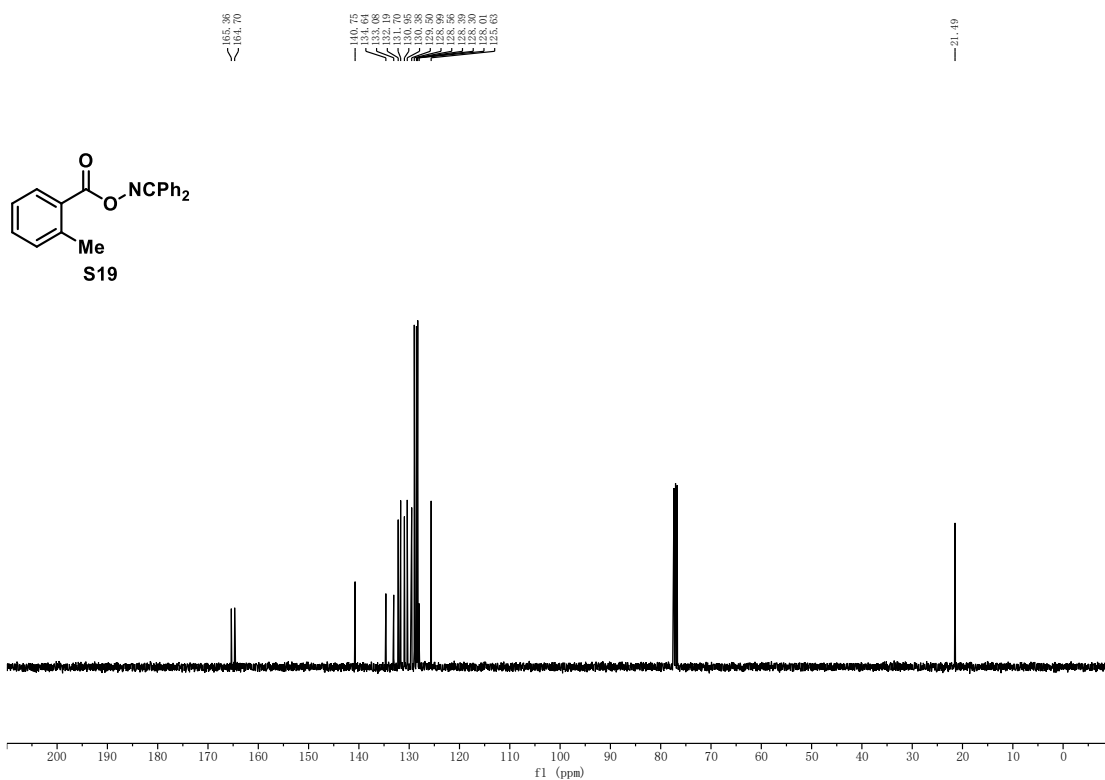
Supplementary Figure 70 ^1H NMR spectrum of S18



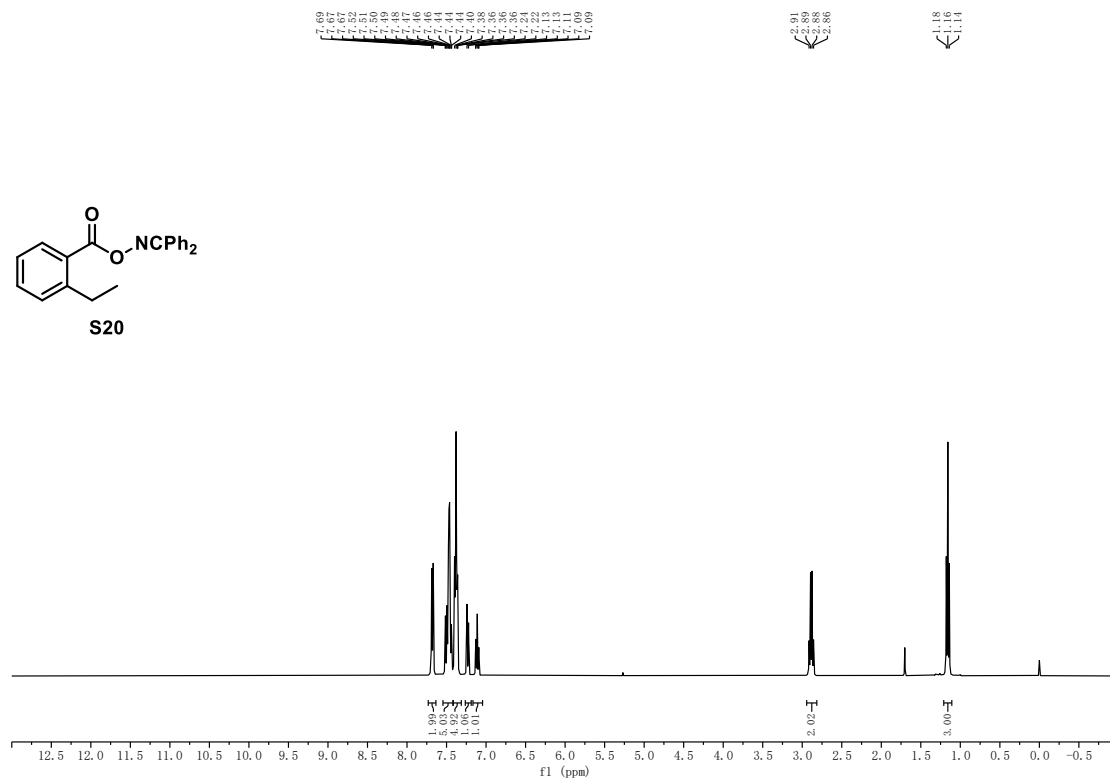
Supplementary Figure 71 ¹³C NMR spectrum of S18



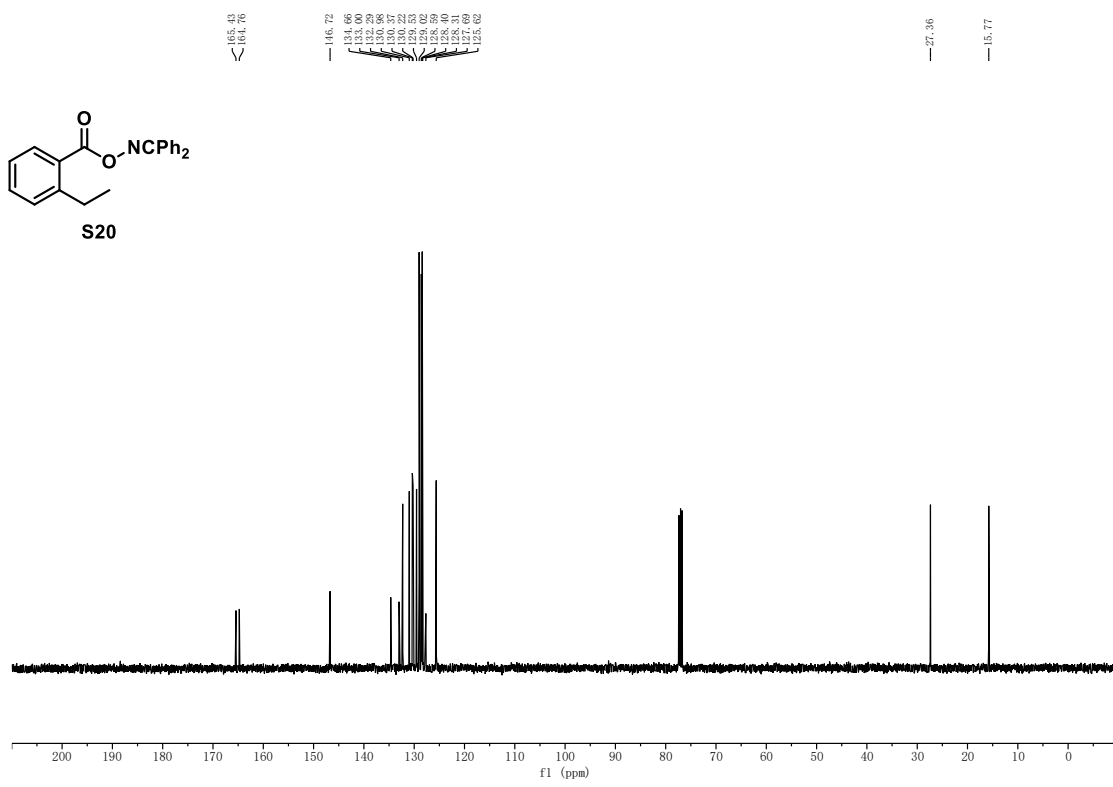
Supplementary Figure 72 ¹H NMR spectrum of S19



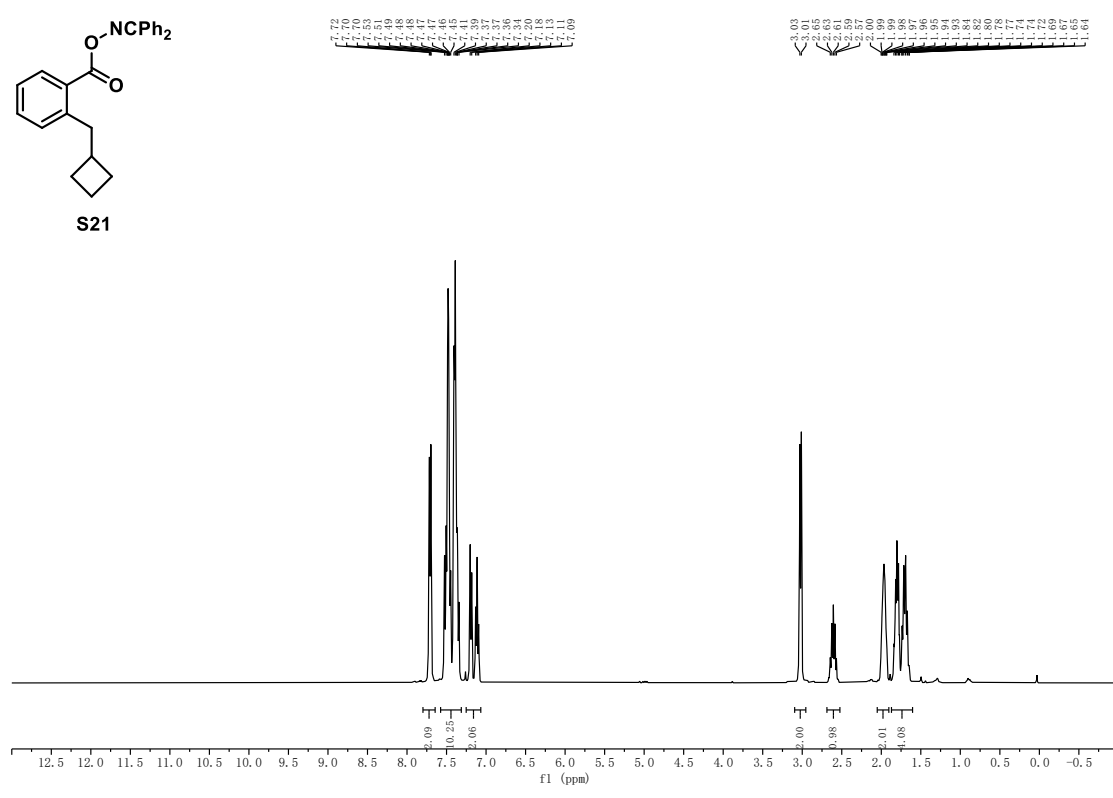
Supplementary Figure 73 ^{13}C NMR spectrum of S19



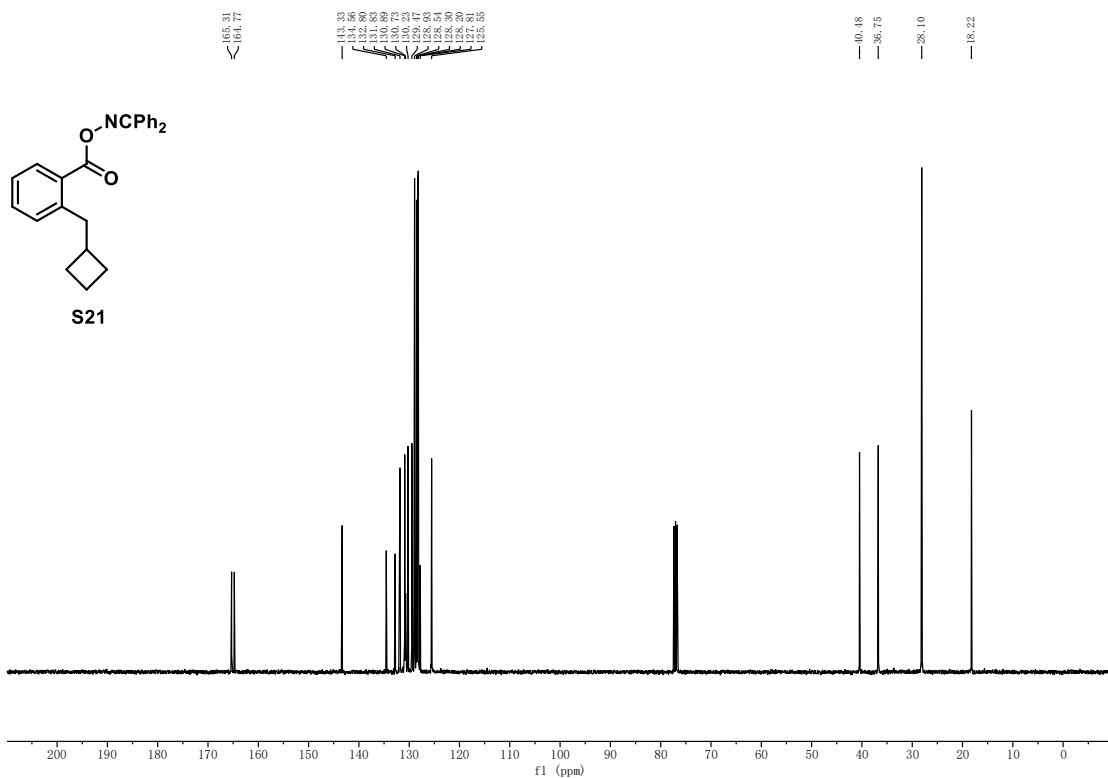
Supplementary Figure 74 ^1H NMR spectrum of S20



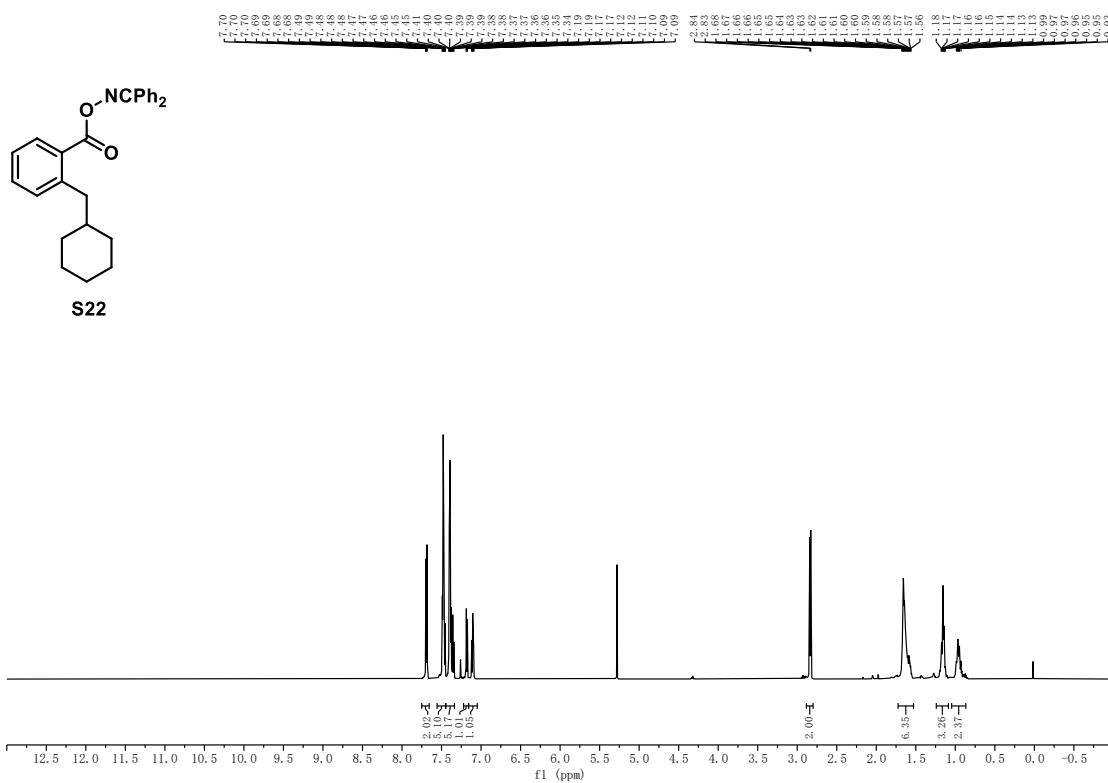
Supplementary Figure 75 ¹³C NMR spectrum of S20



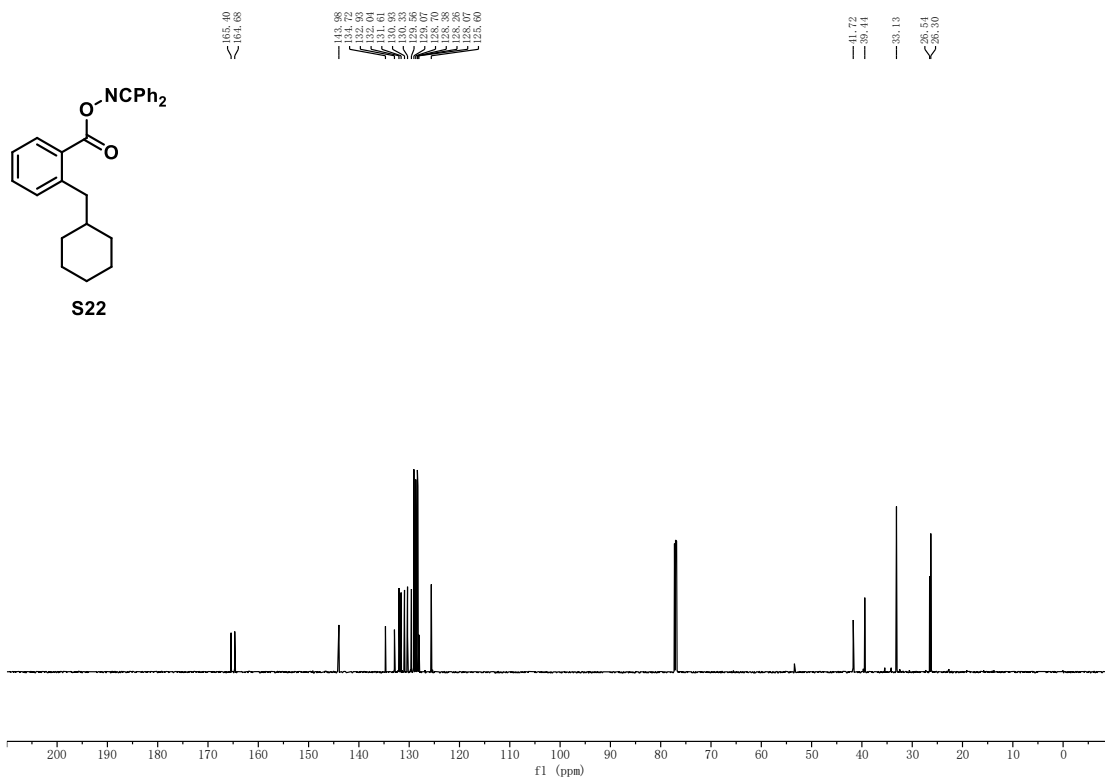
Supplementary Figure 76 ¹H NMR spectrum of S21



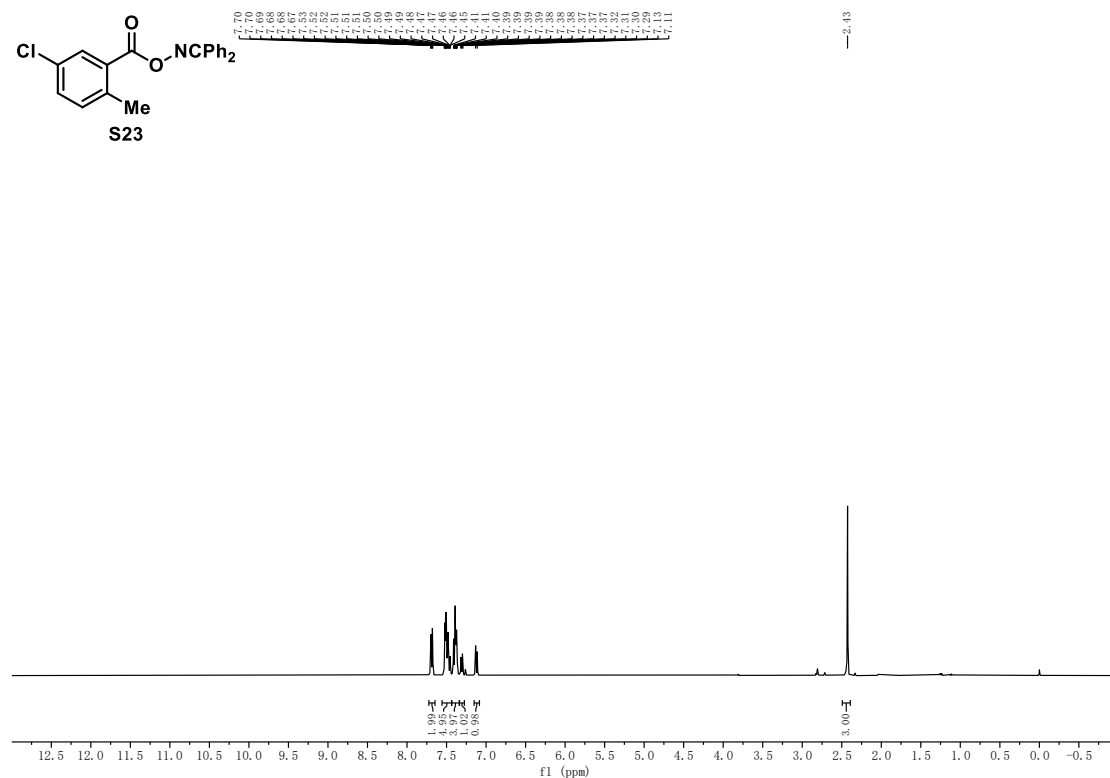
Supplementary Figure 77 ^{13}C NMR spectrum of S21



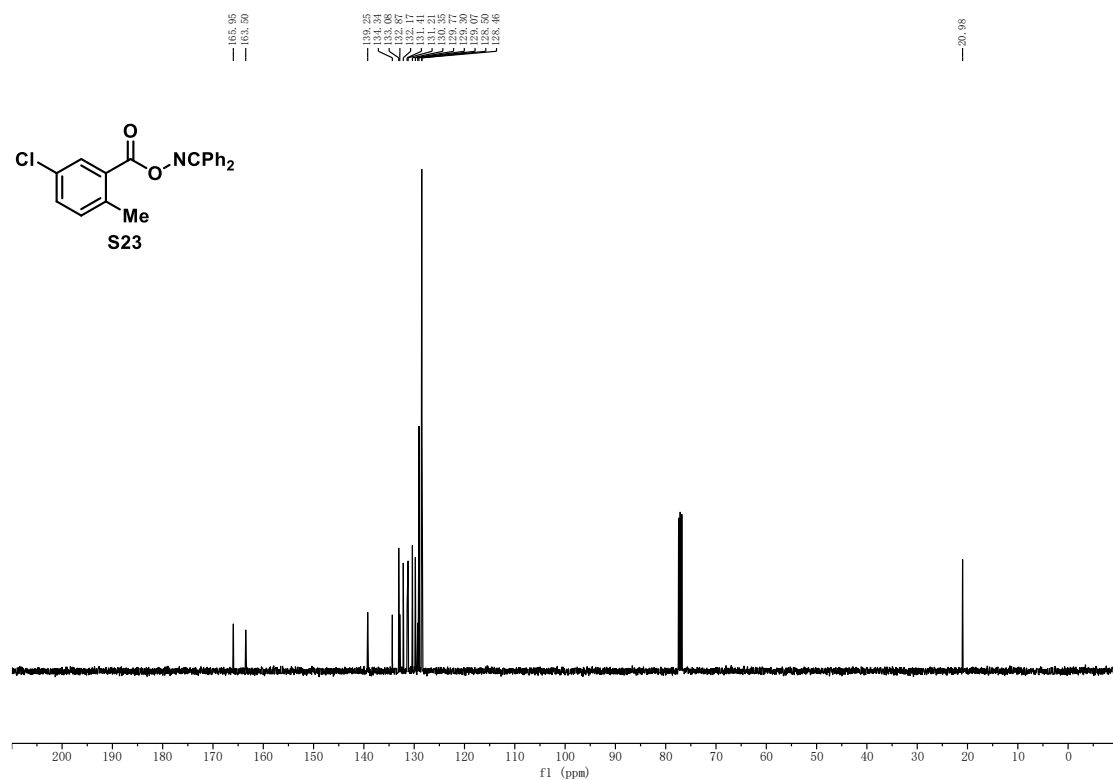
Supplementary Figure 78 ^1H NMR spectrum of S22



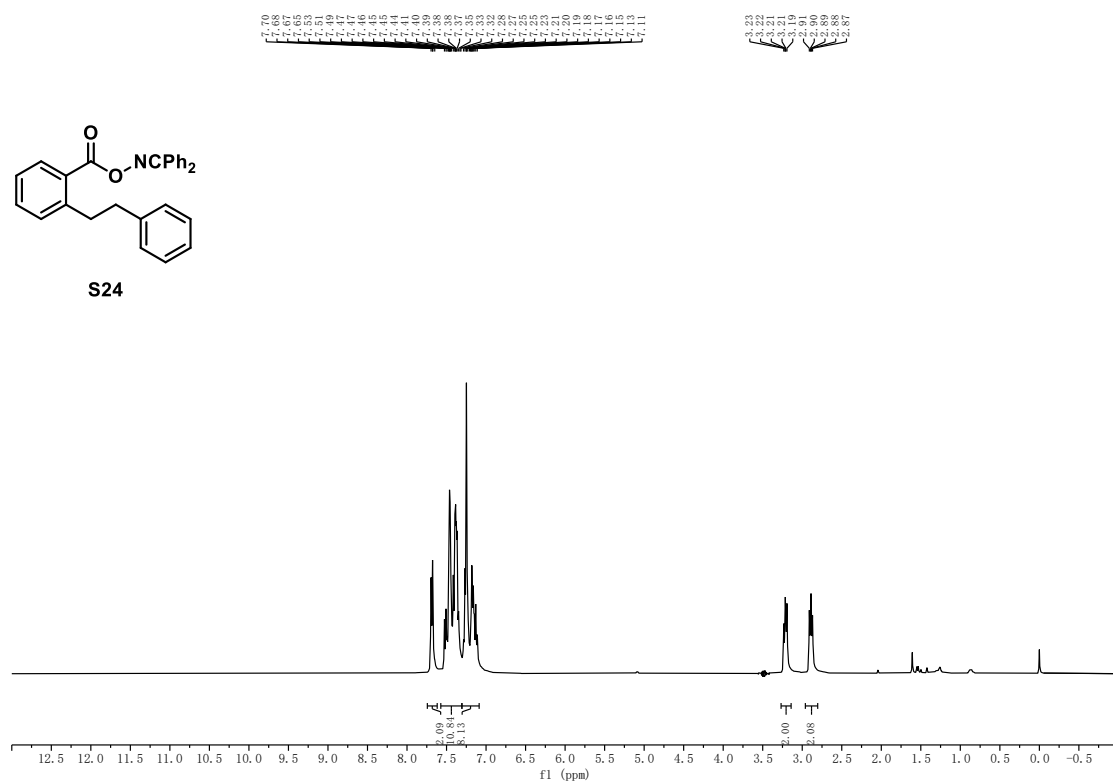
Supplementary Figure 79 ¹³C NMR spectrum of S22



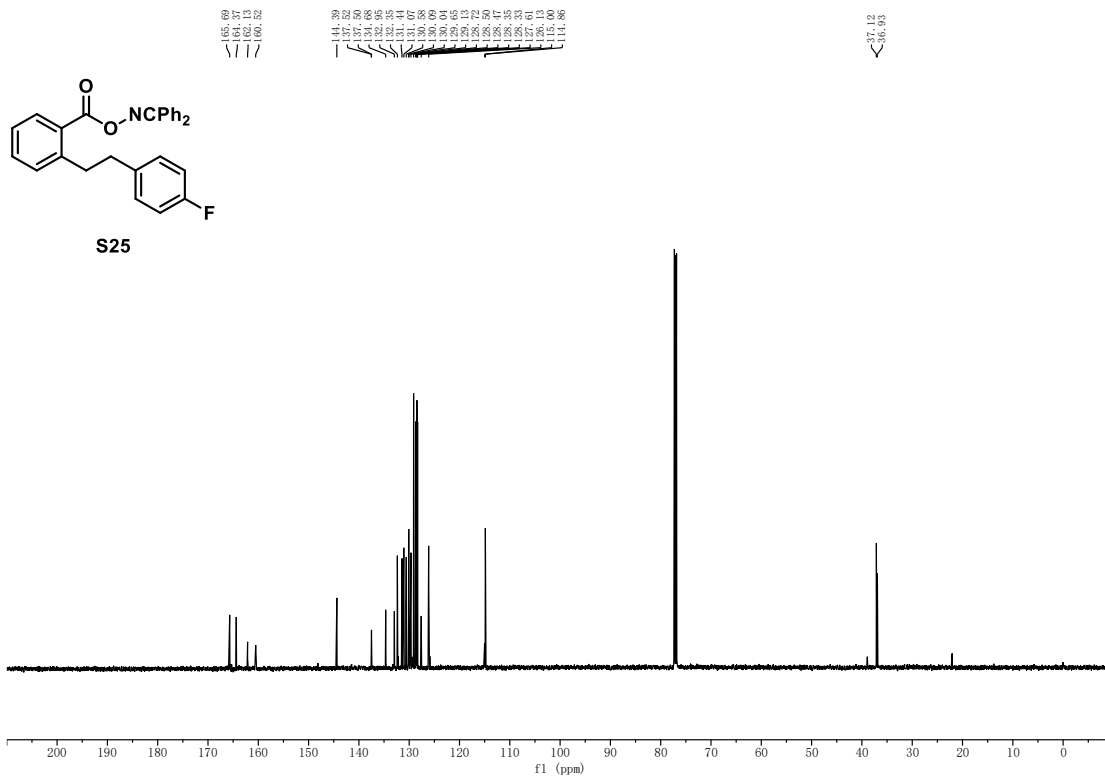
Supplementary Figure 80 ¹H NMR spectrum of S23



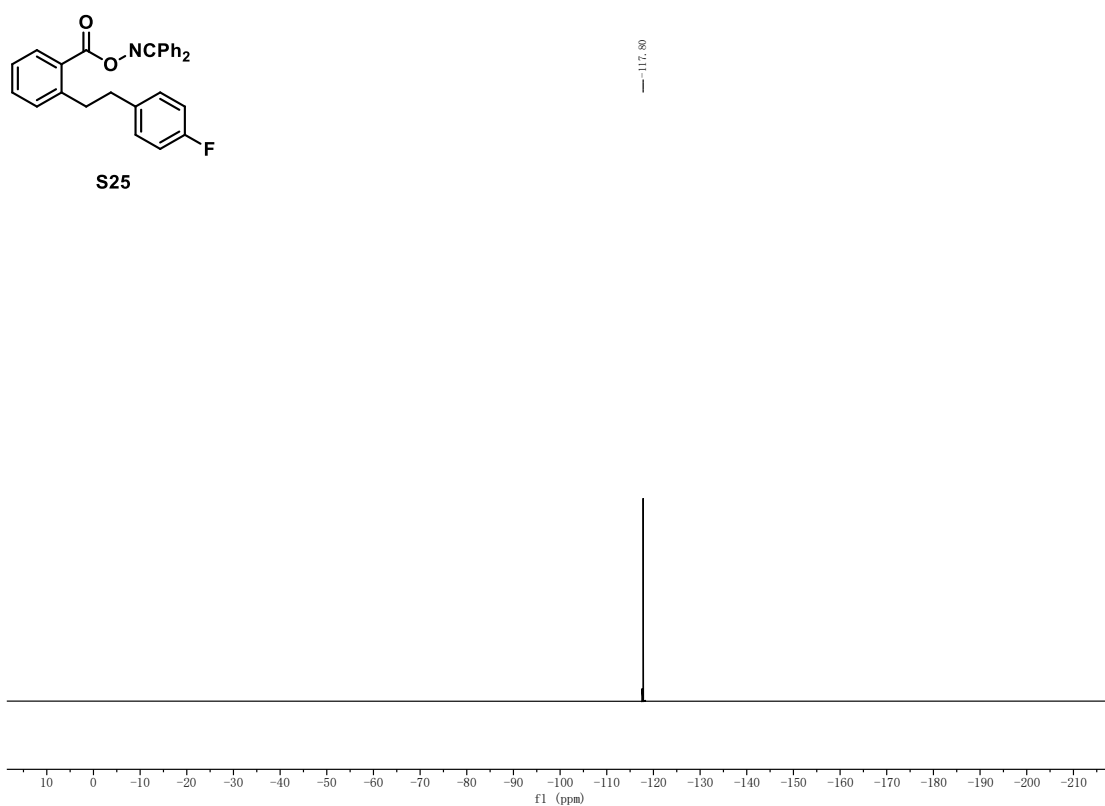
Supplementary Figure 81 ^{13}C NMR spectrum of S23



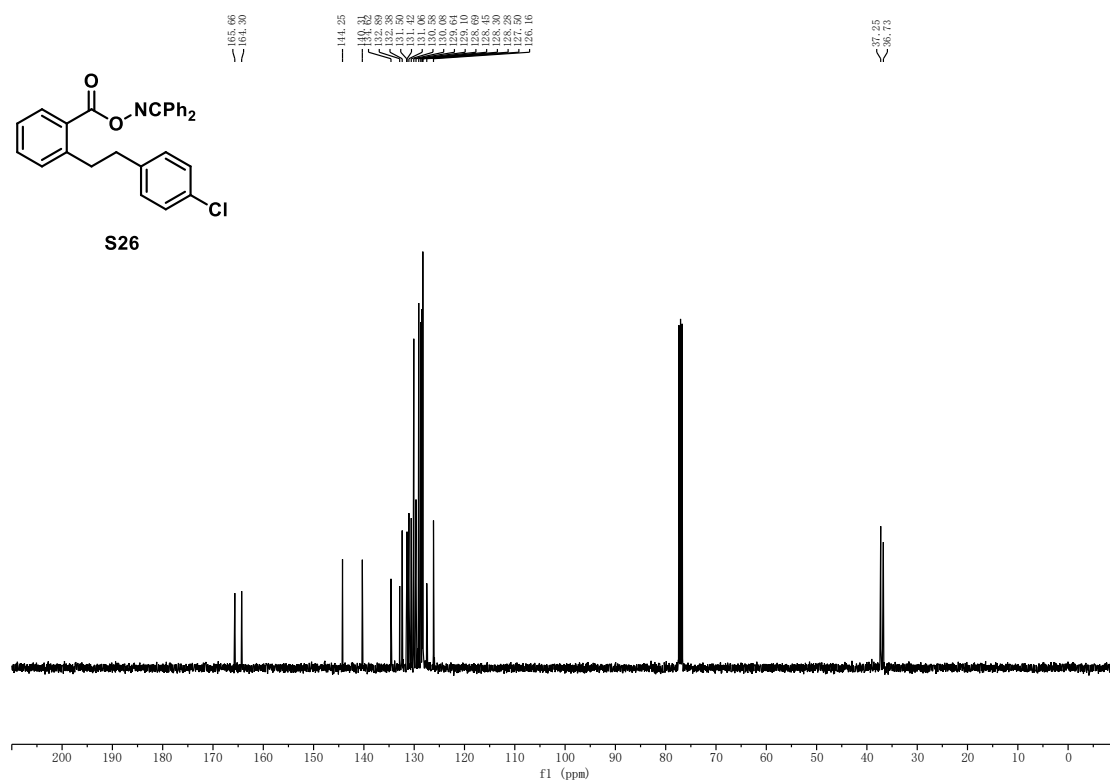
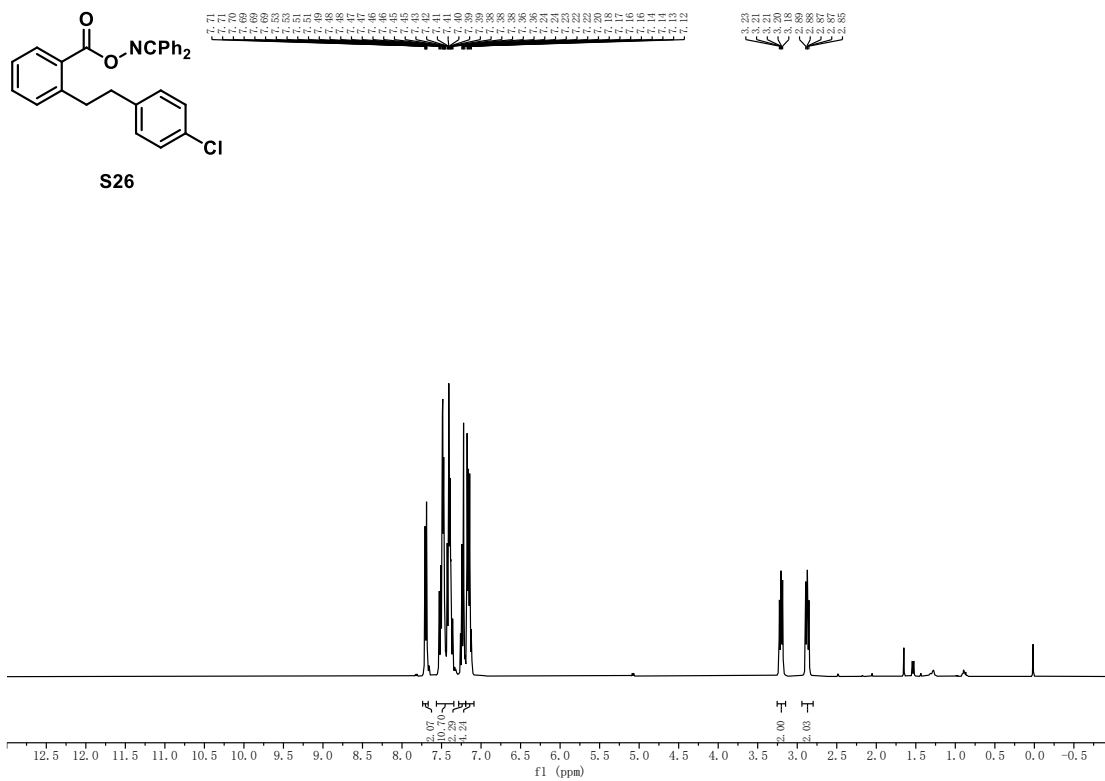
Supplementary Figure 82 ^1H NMR spectrum of S24

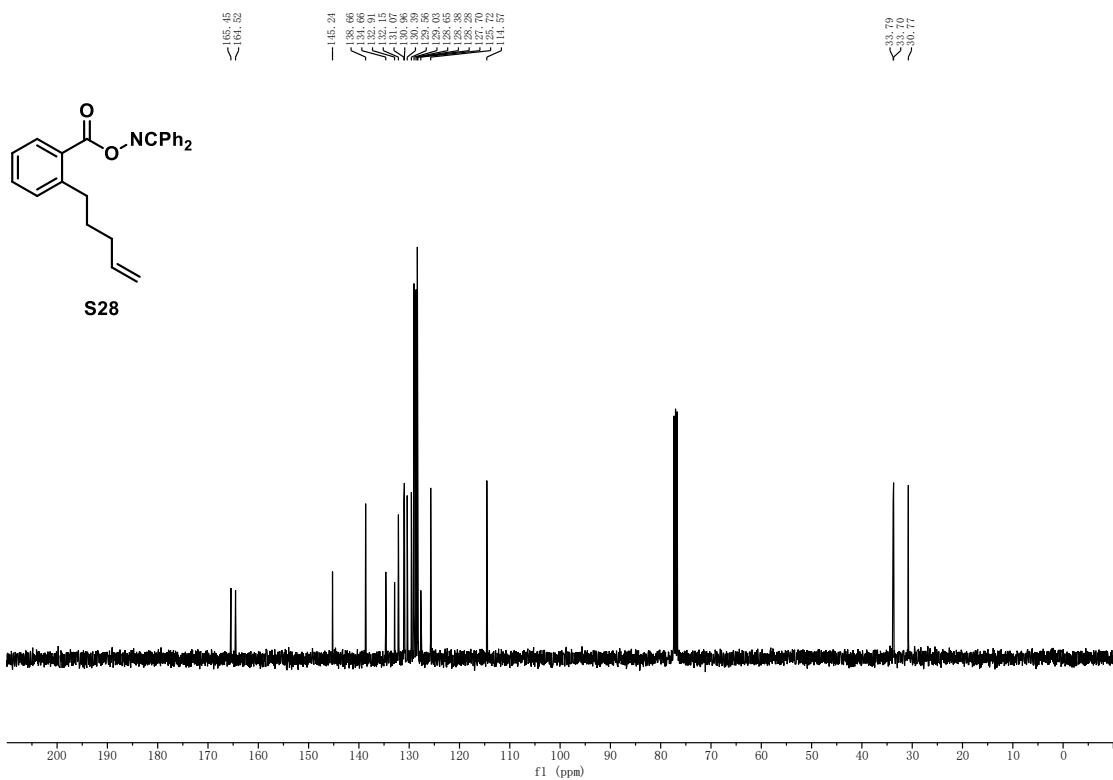
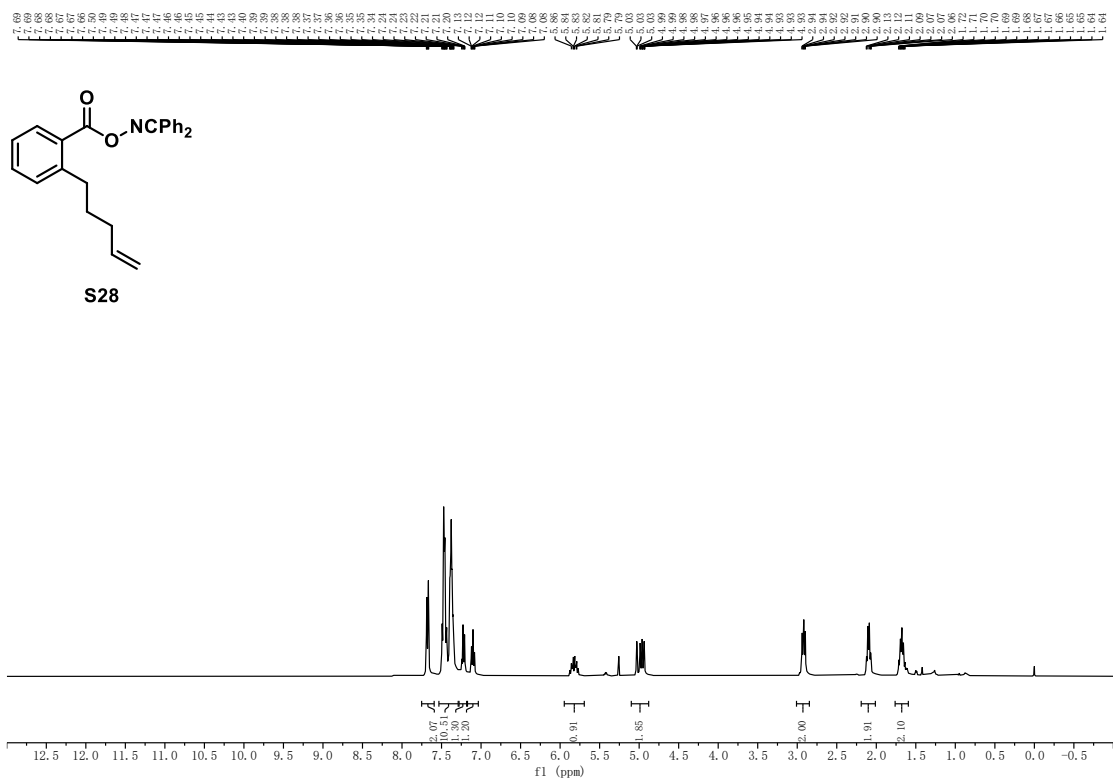


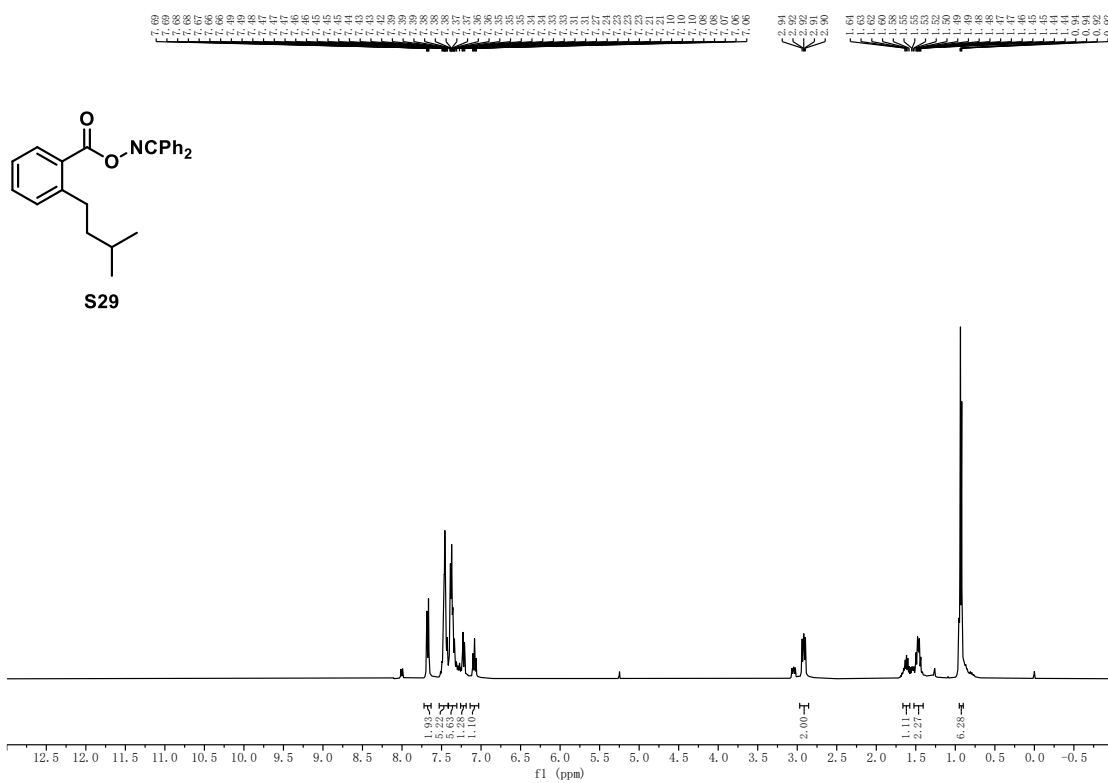
Supplementary Figure 85 ¹³C NMR spectrum of S25



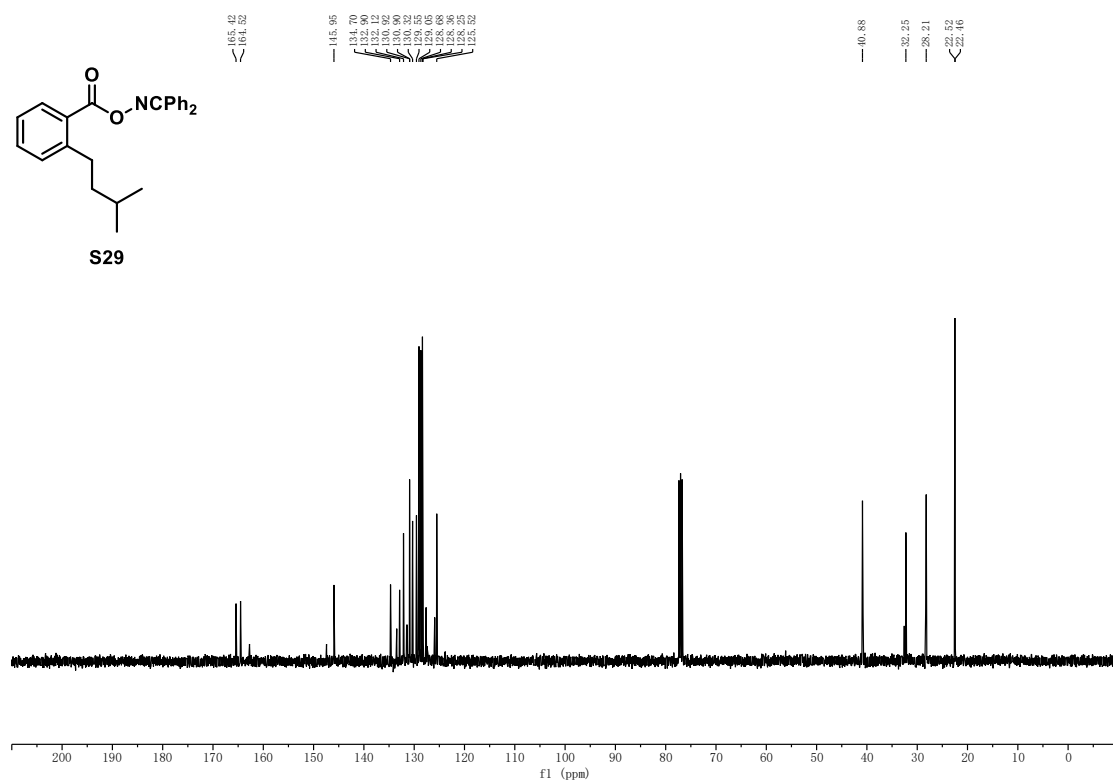
Supplementary Figure 86 ¹⁹F NMR spectrum of S25



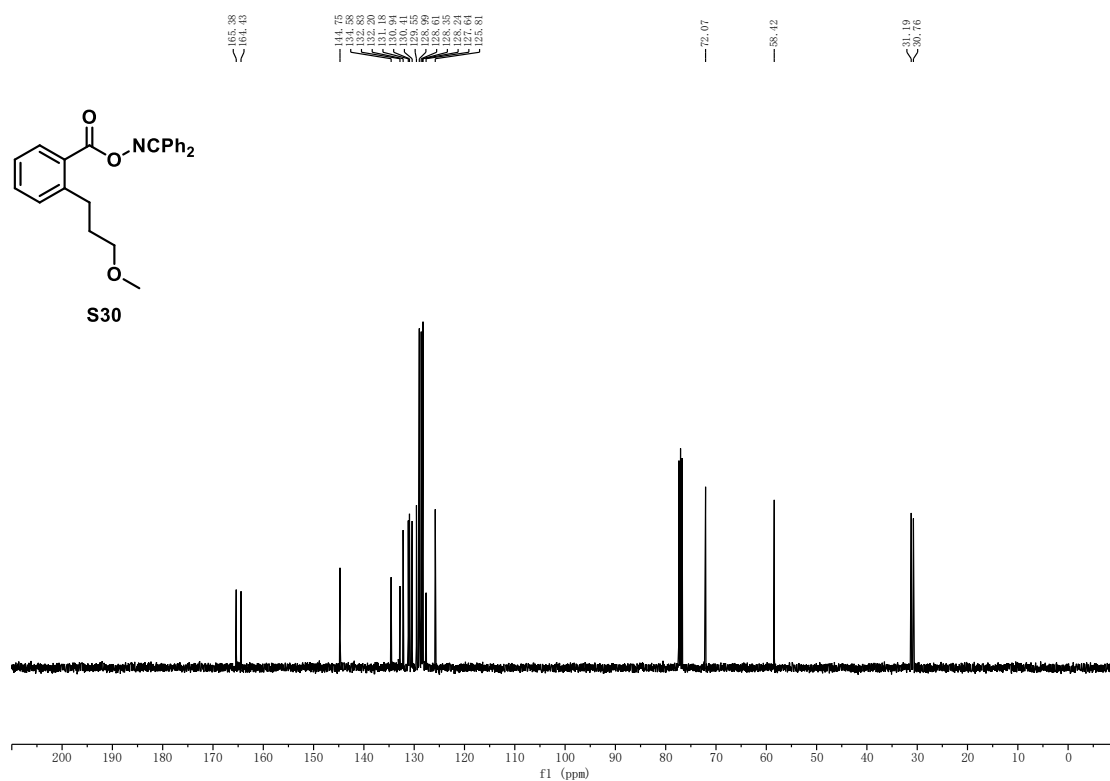
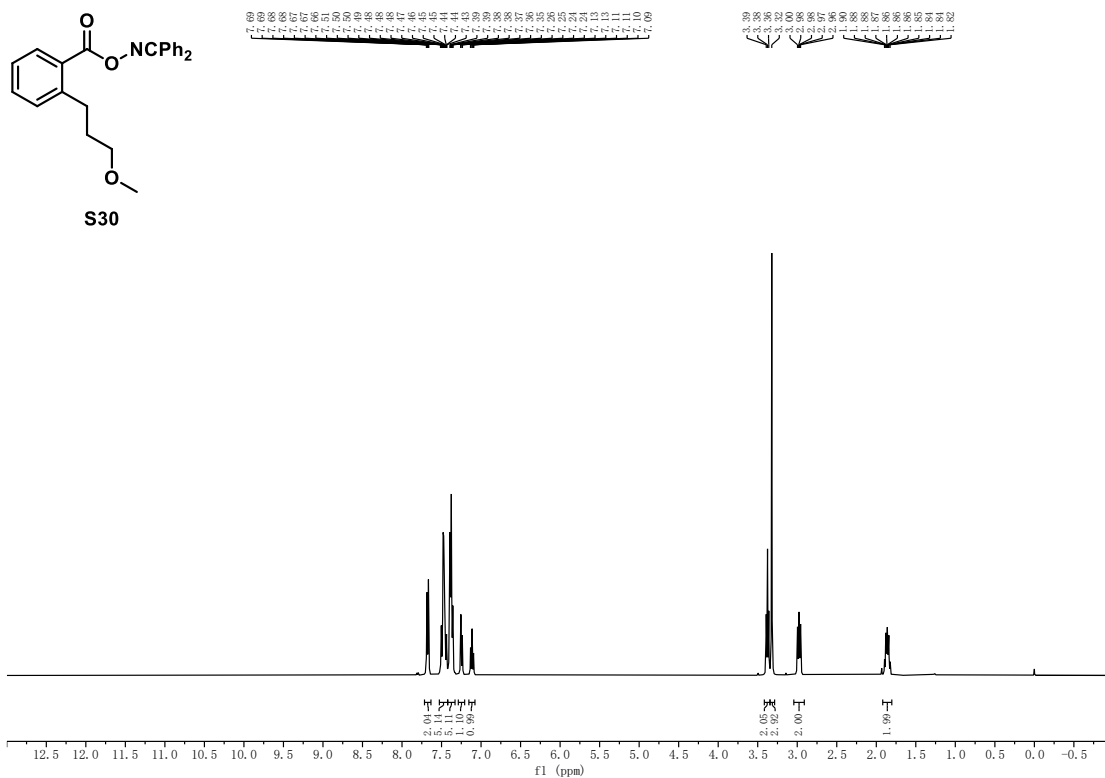


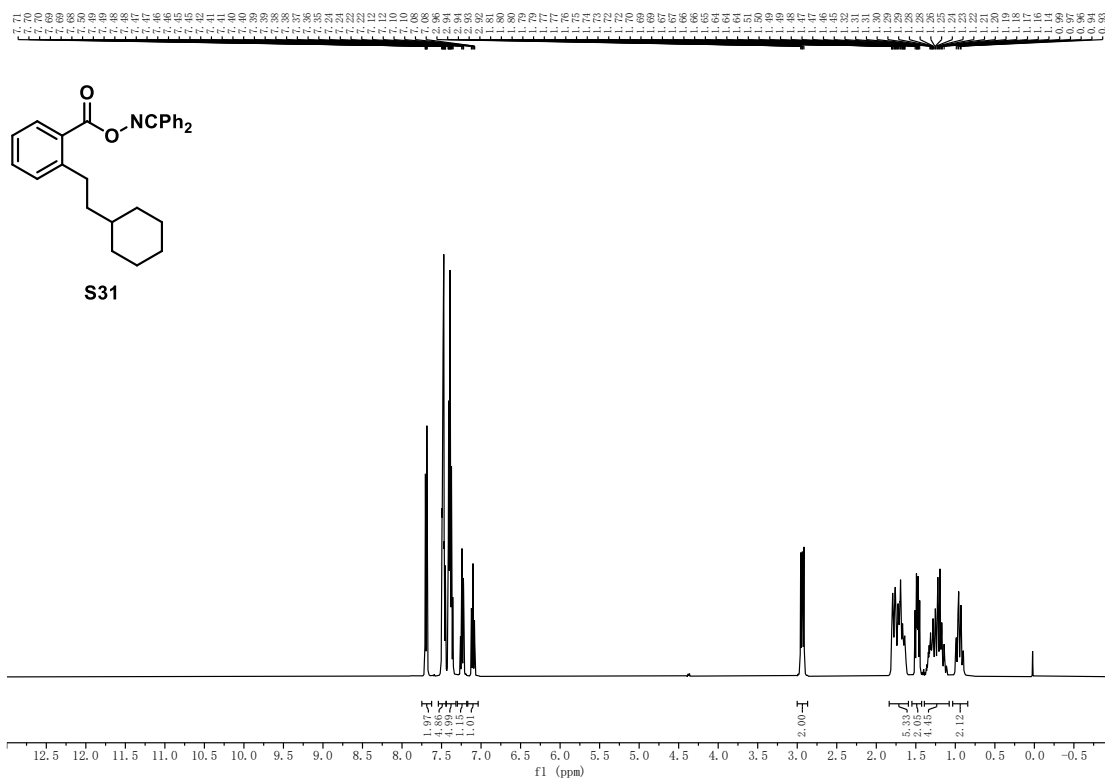


Supplementary Figure 93 ¹H NMR spectrum of S29

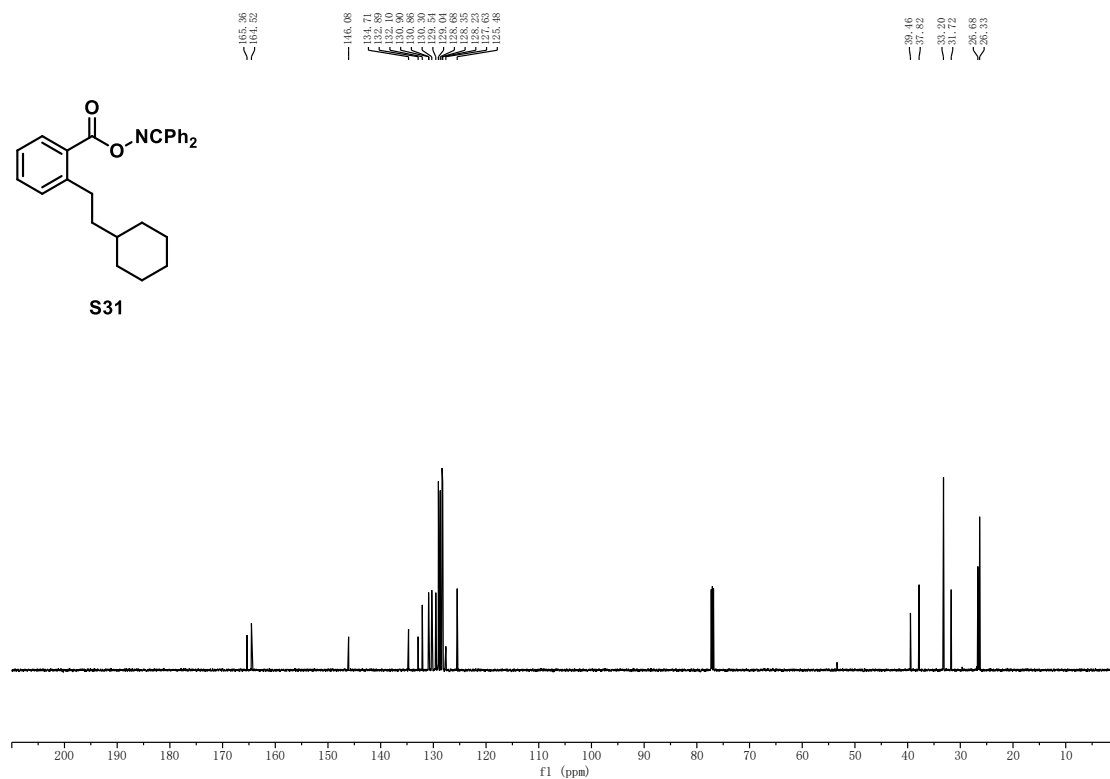


Supplementary Figure 94 ¹³C NMR spectrum of S29

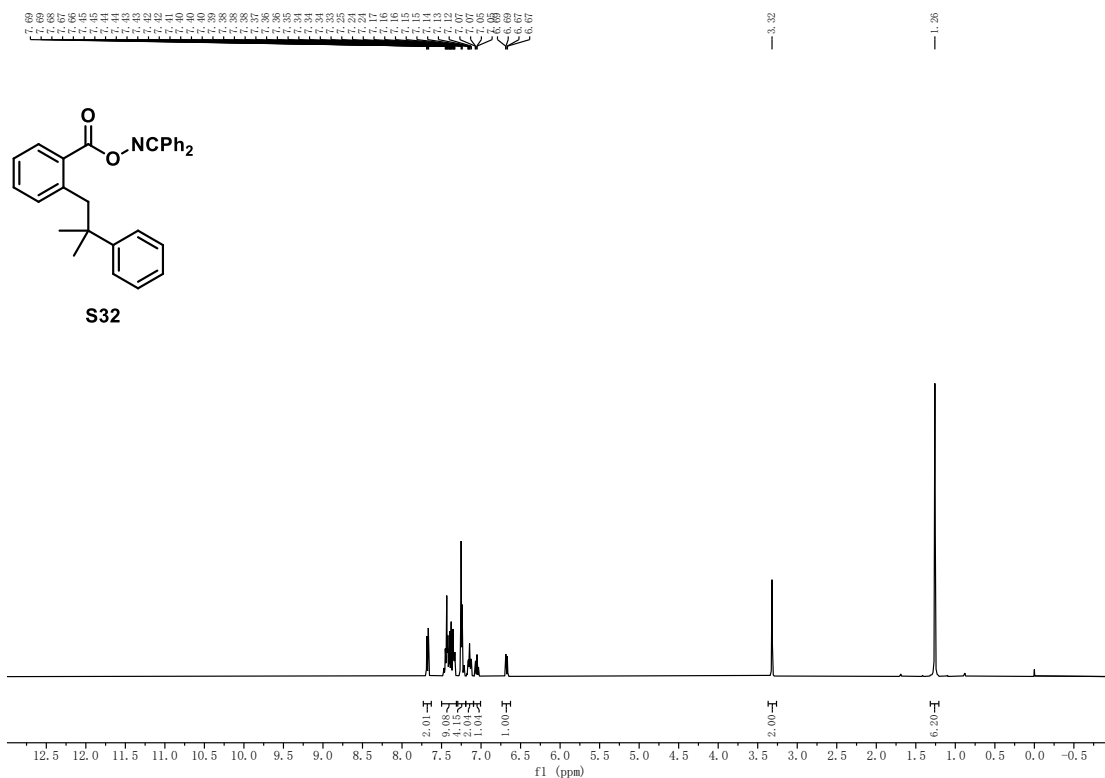




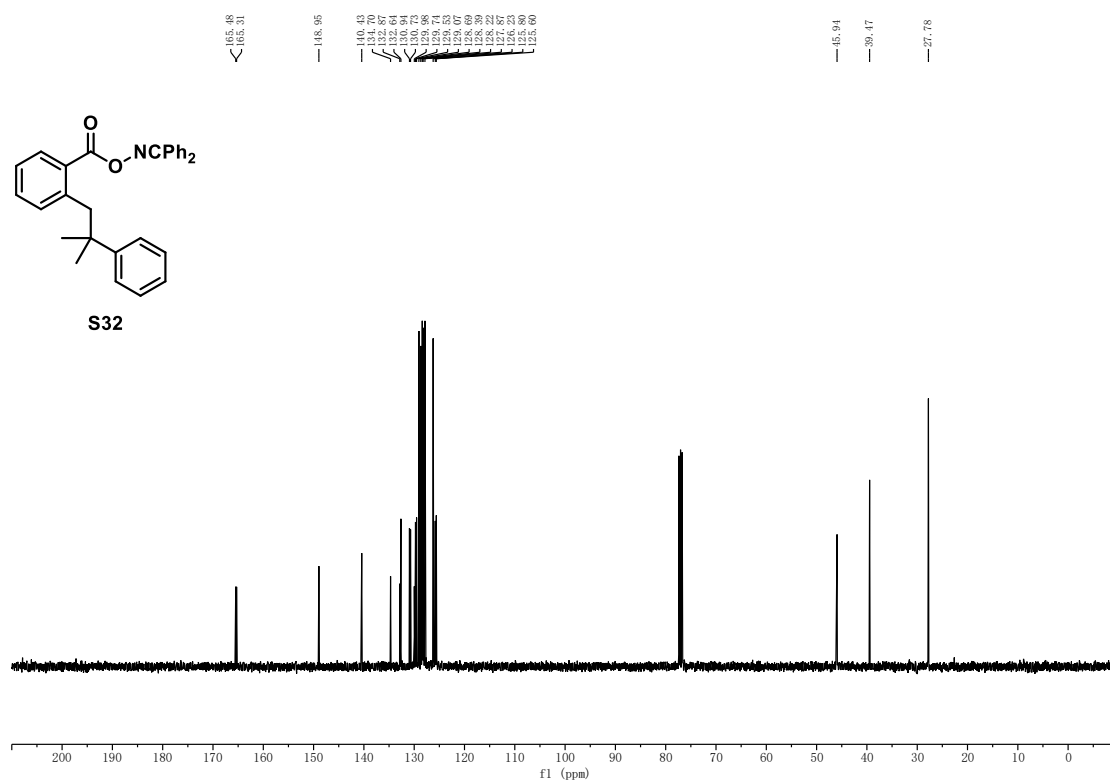
Supplementary Figure 97 ¹H NMR spectrum of S31



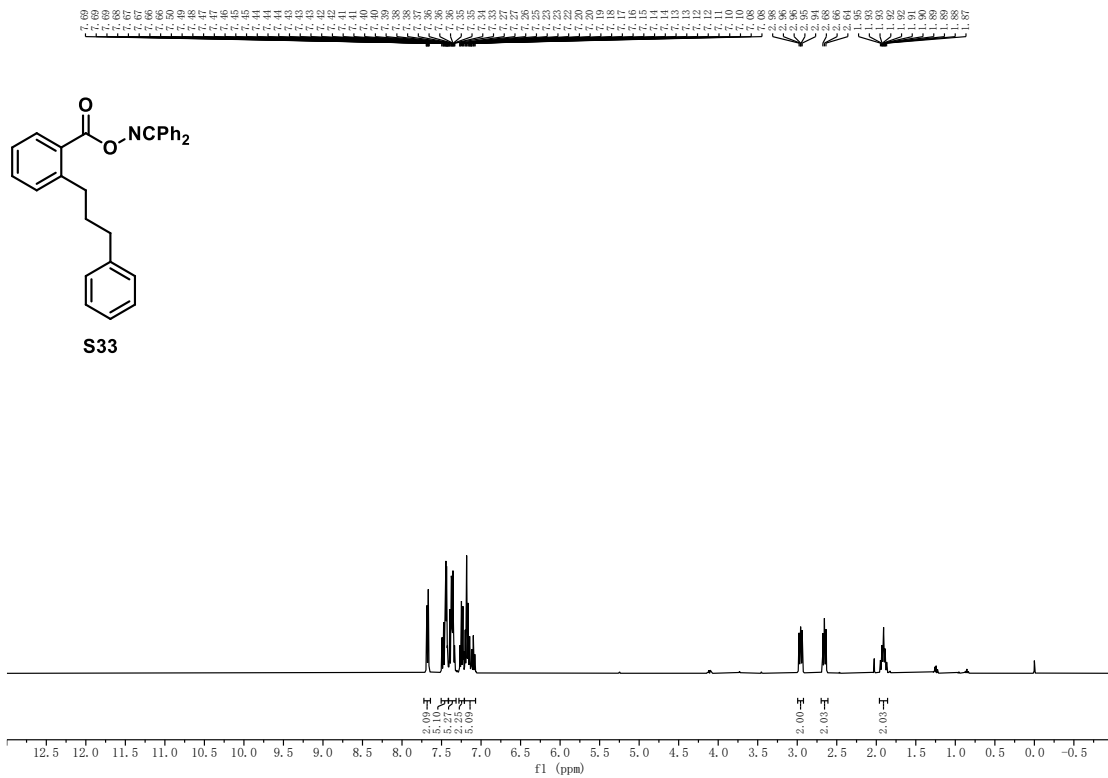
Supplementary Figure 98 ¹³C NMR spectrum of S31



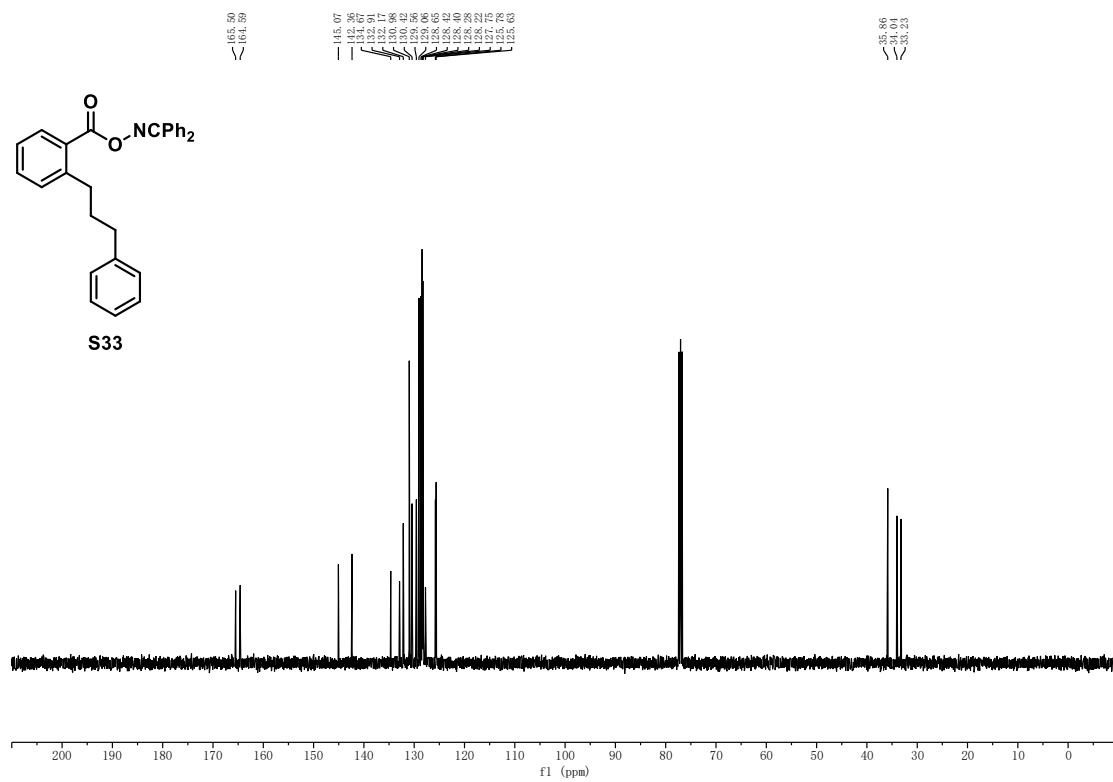
Supplementary Figure 99 ¹H NMR spectrum of S32



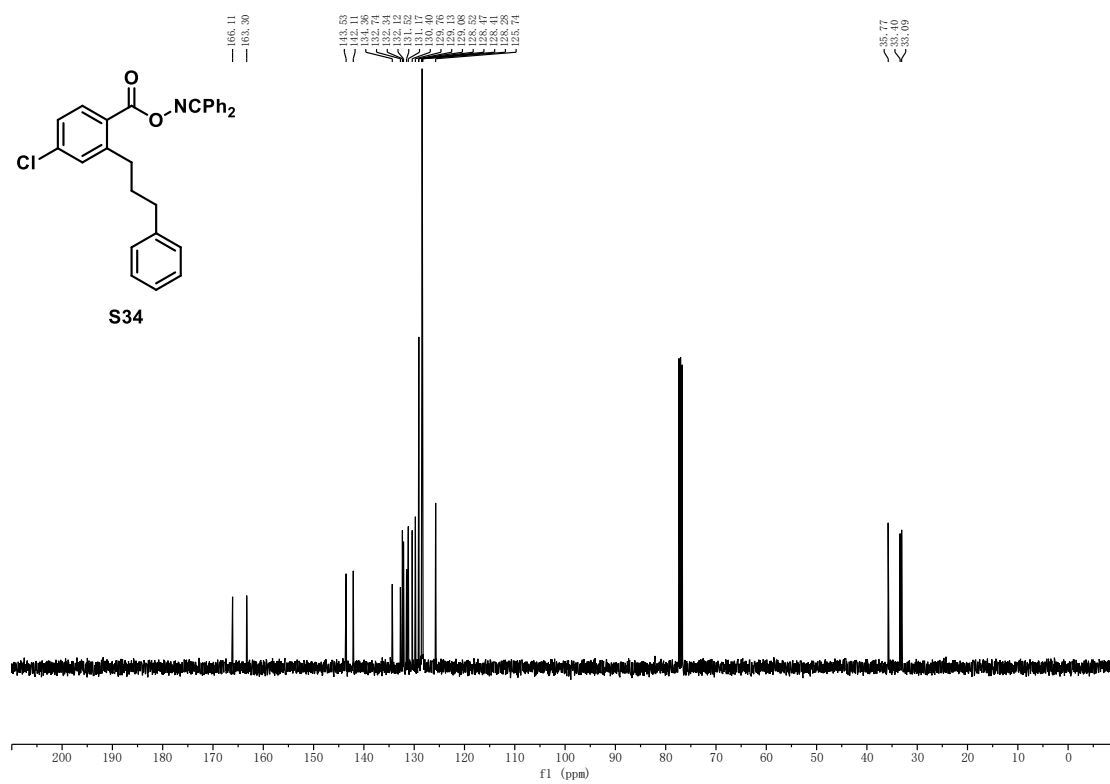
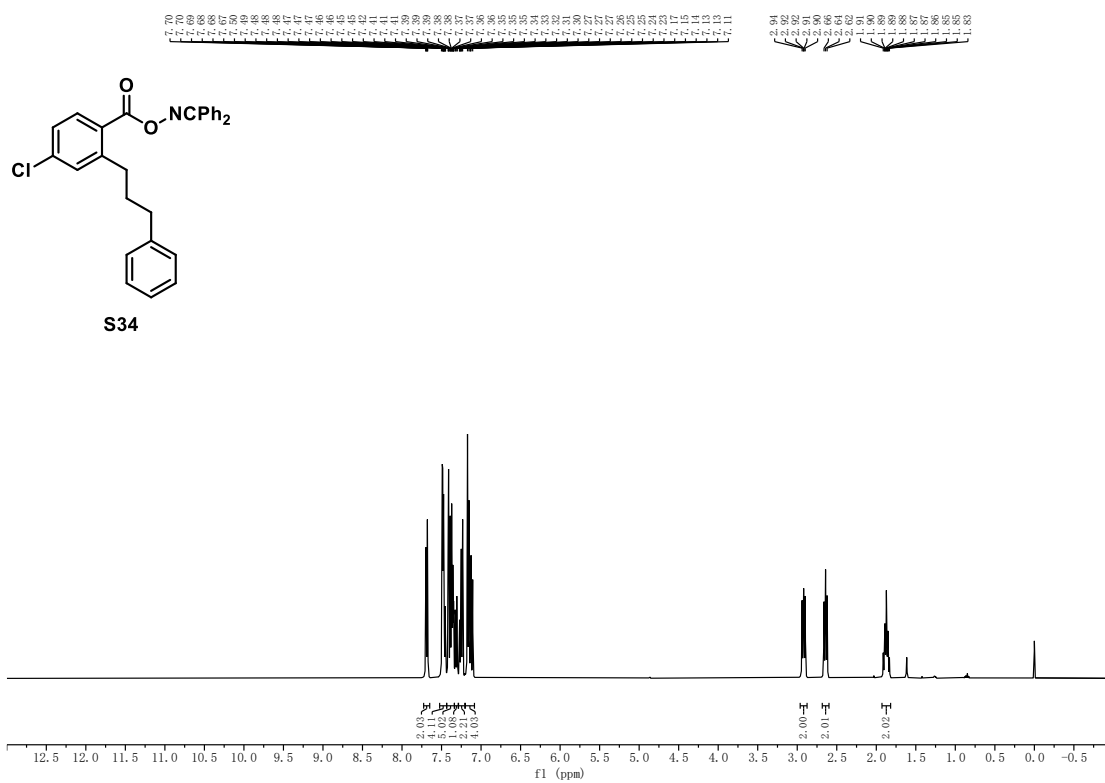
Supplementary Figure 100 ¹³C NMR spectrum of S32

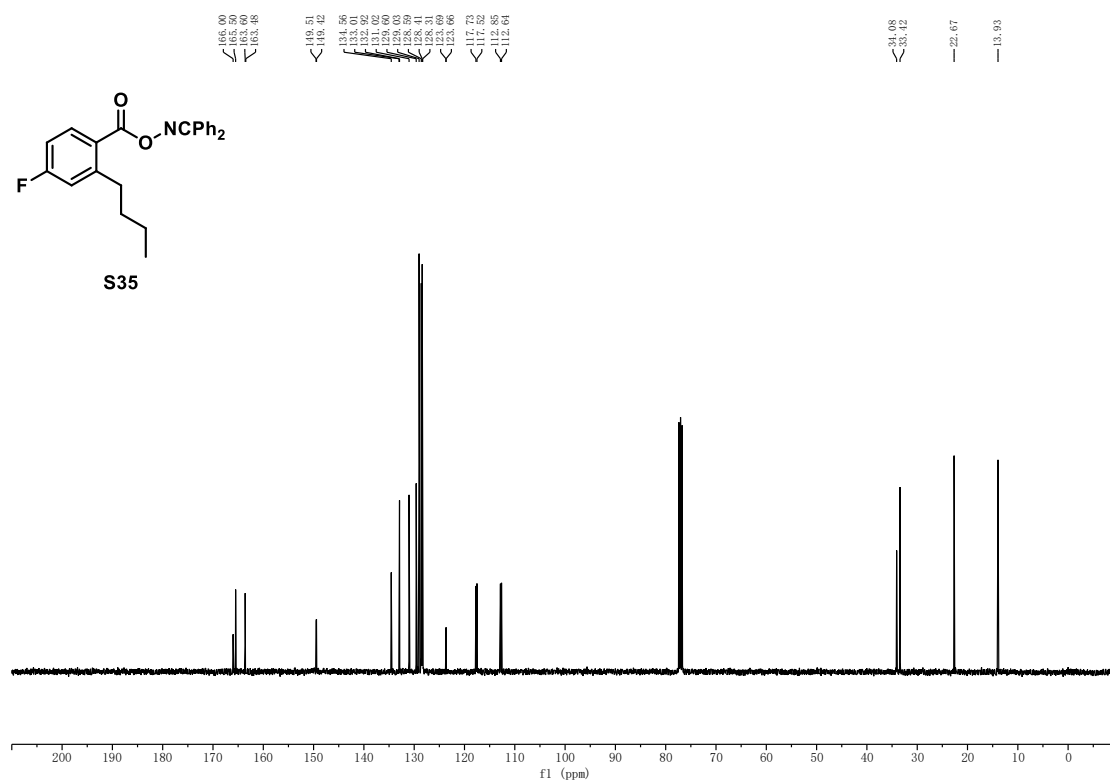
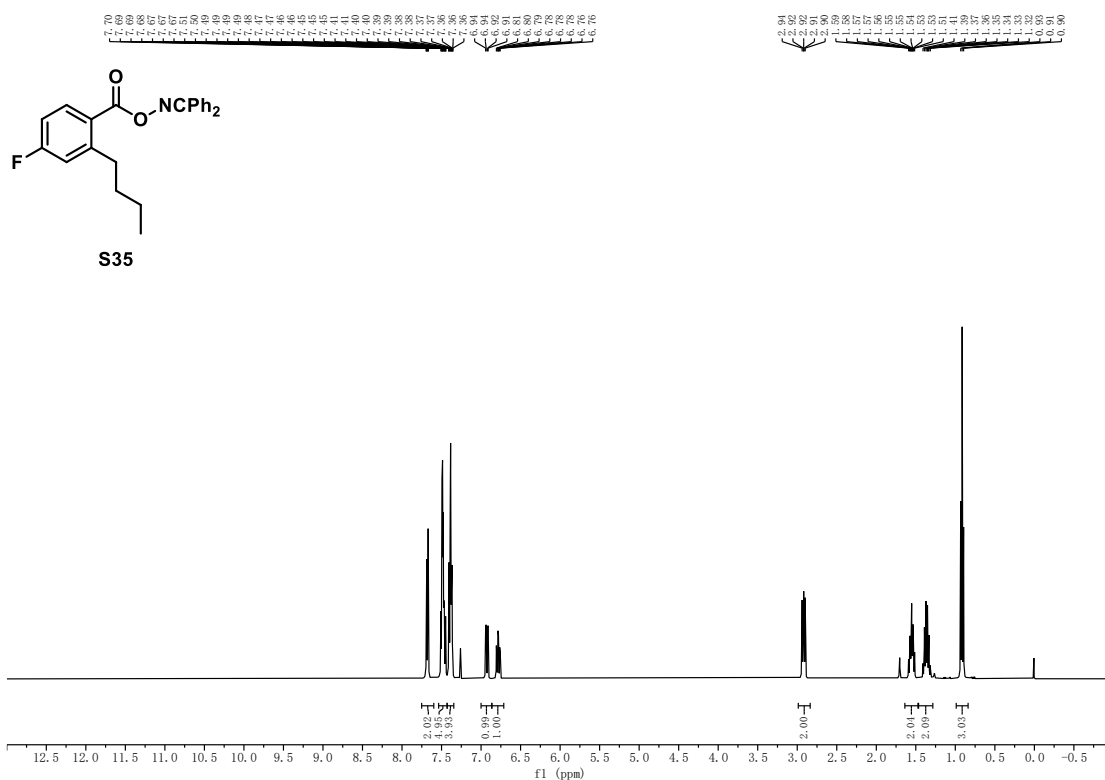


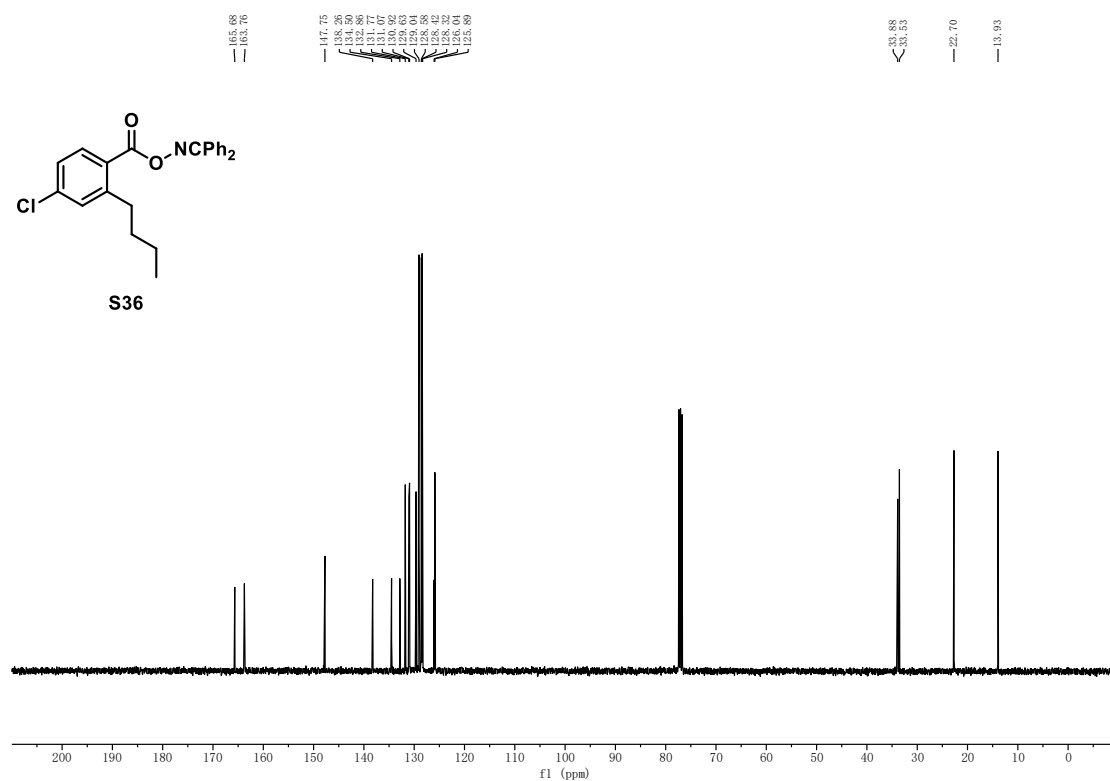
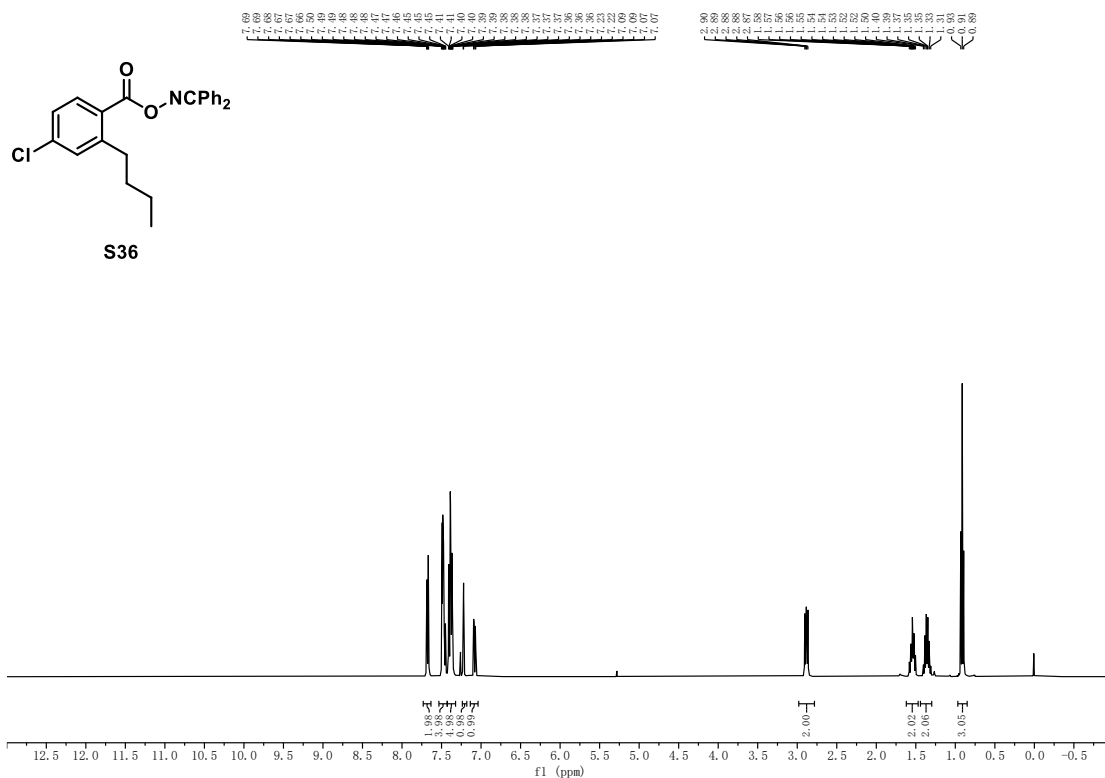
Supplementary Figure 101 ^1H NMR spectrum of S33

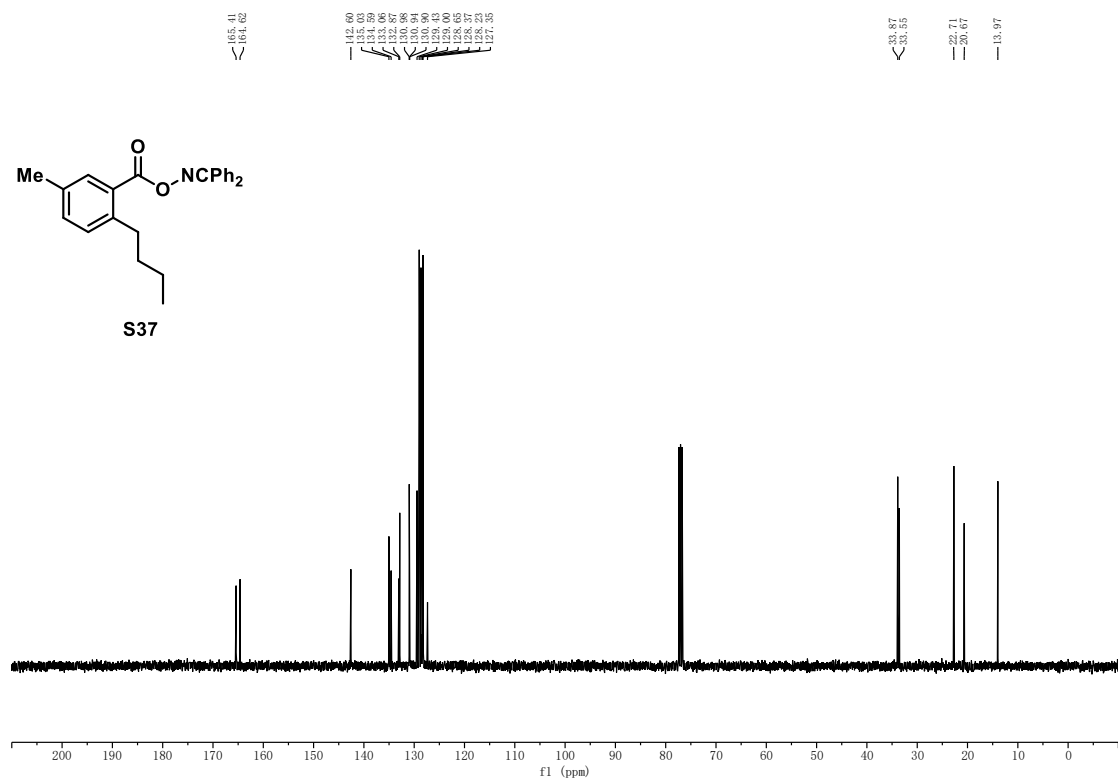
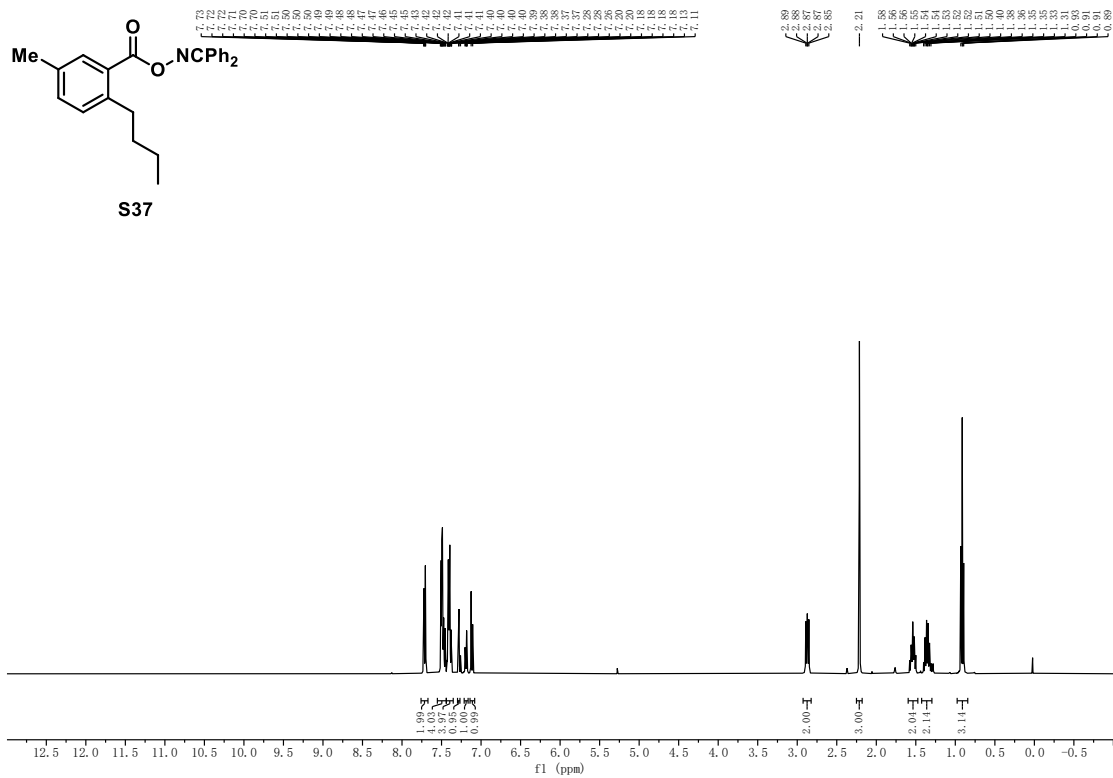


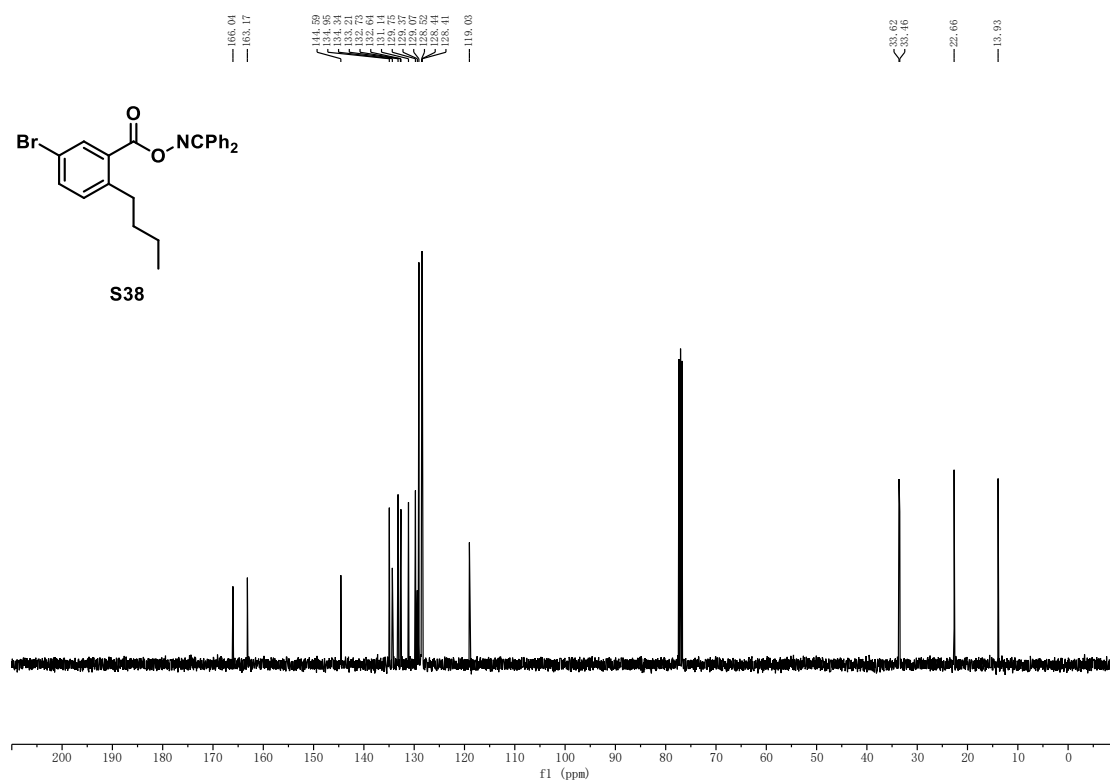
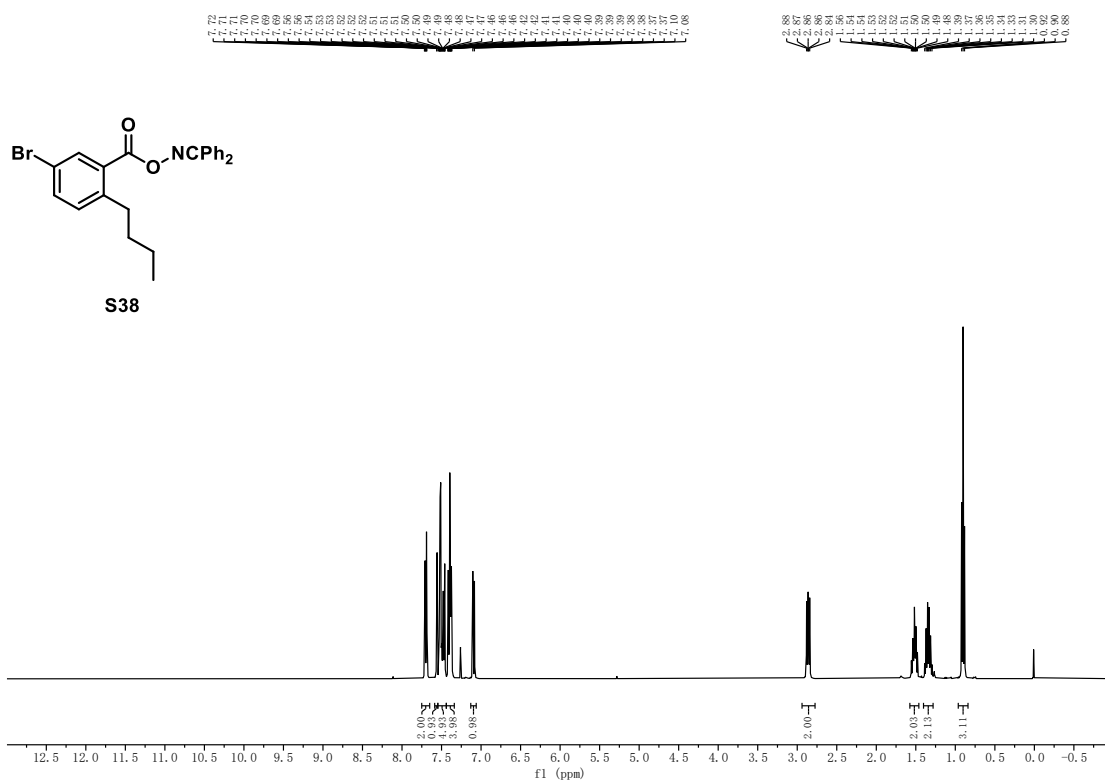
Supplementary Figure 102 ^{13}C NMR spectrum of S33

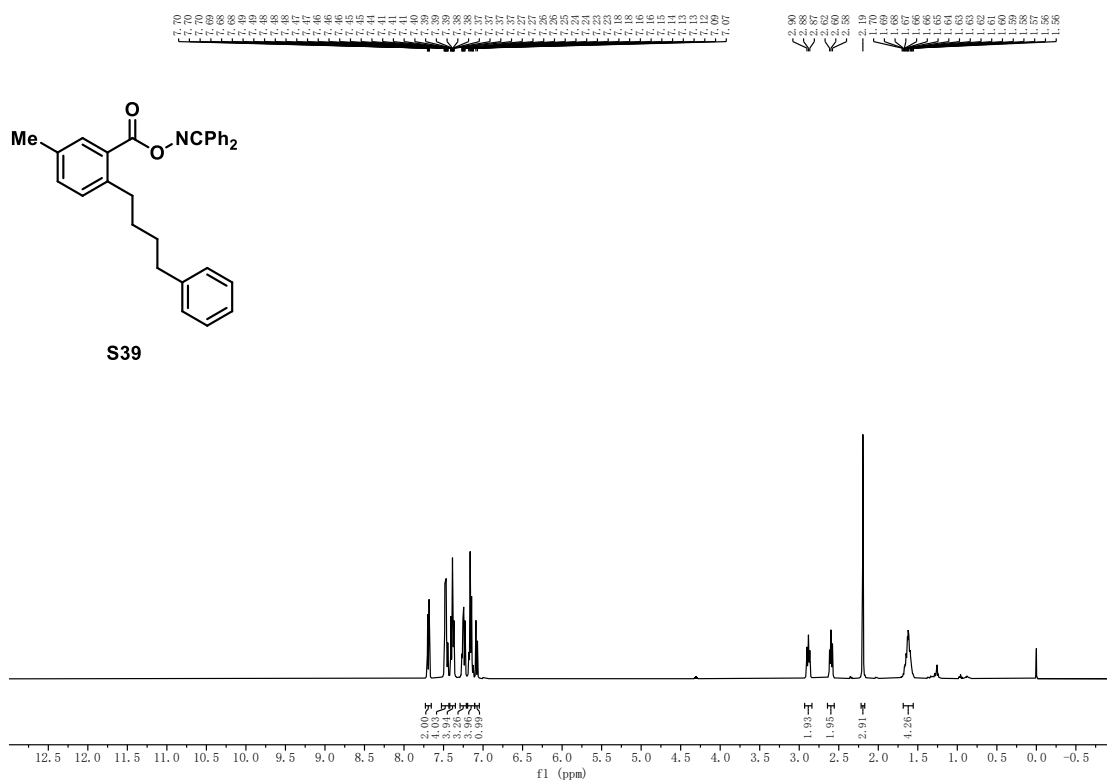




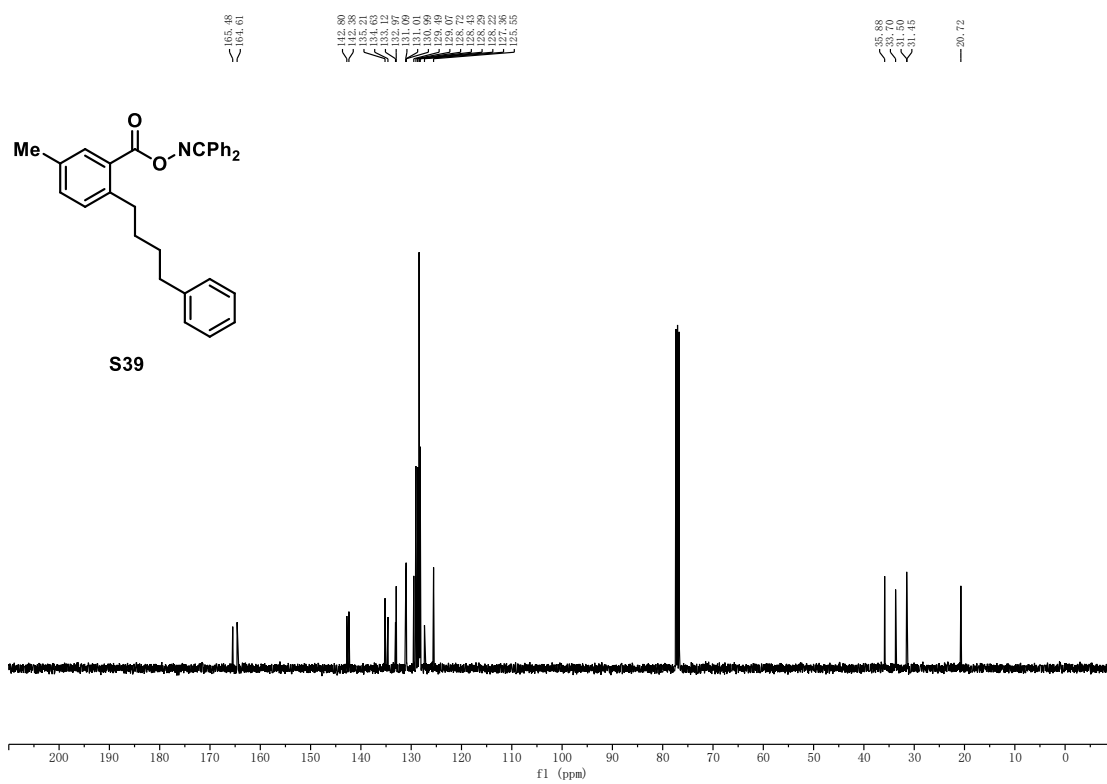




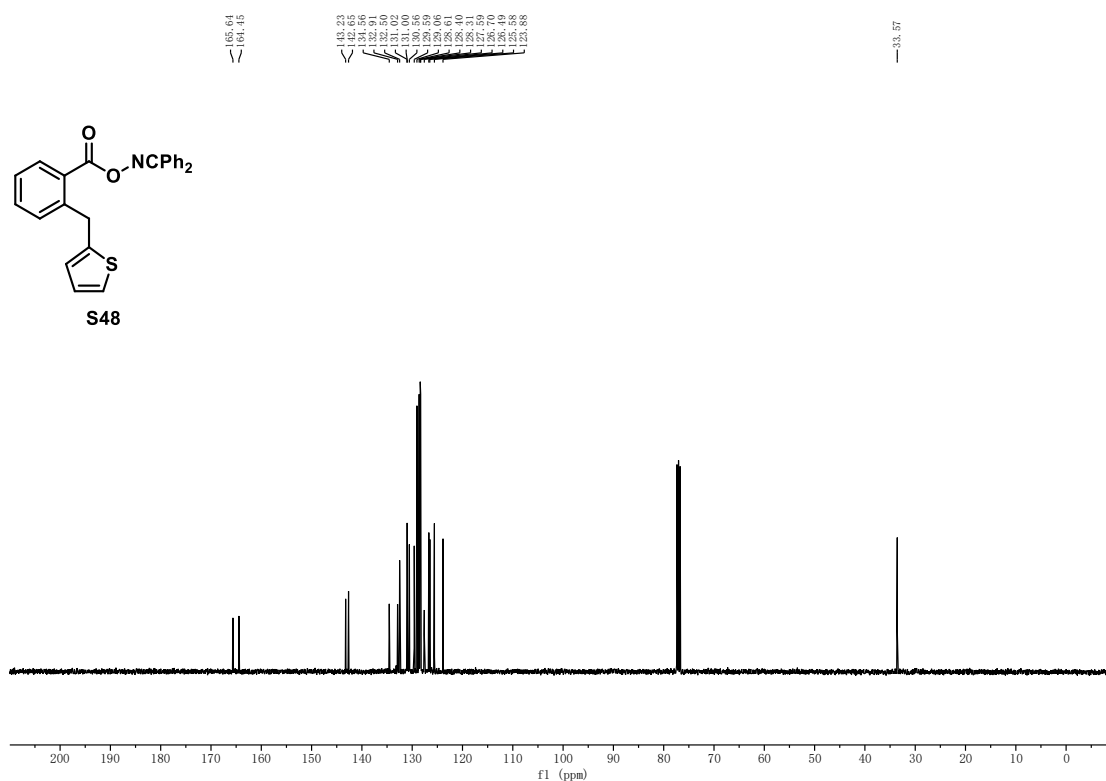
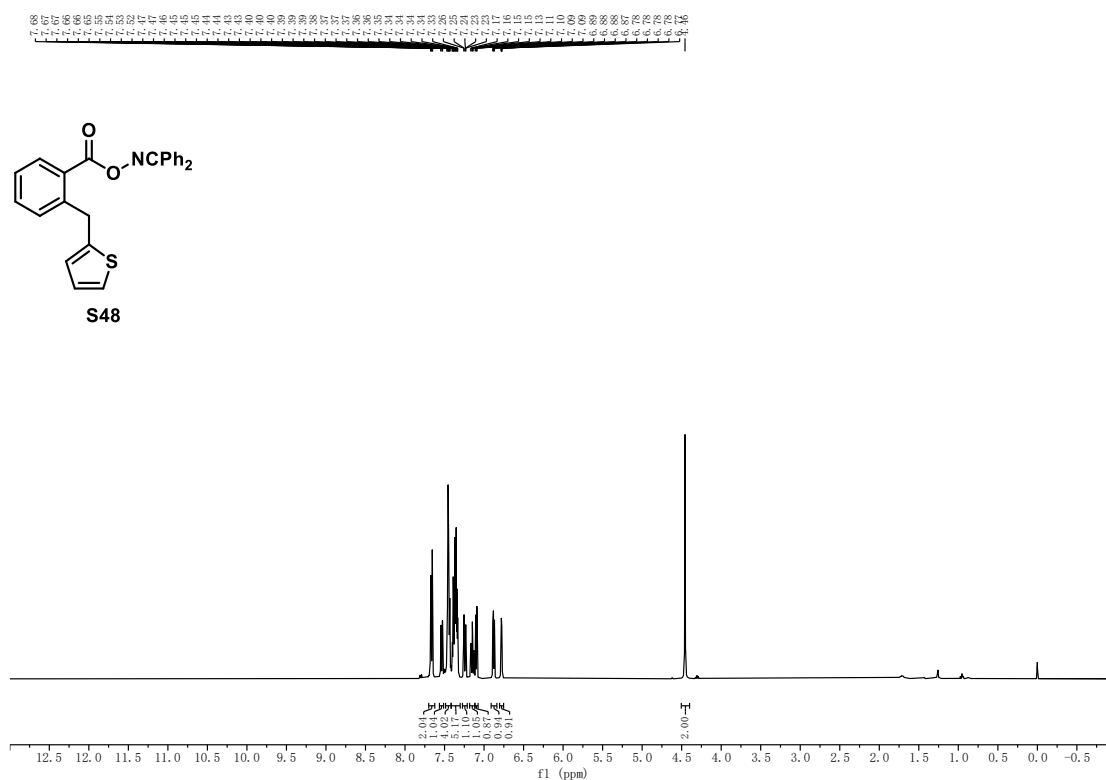


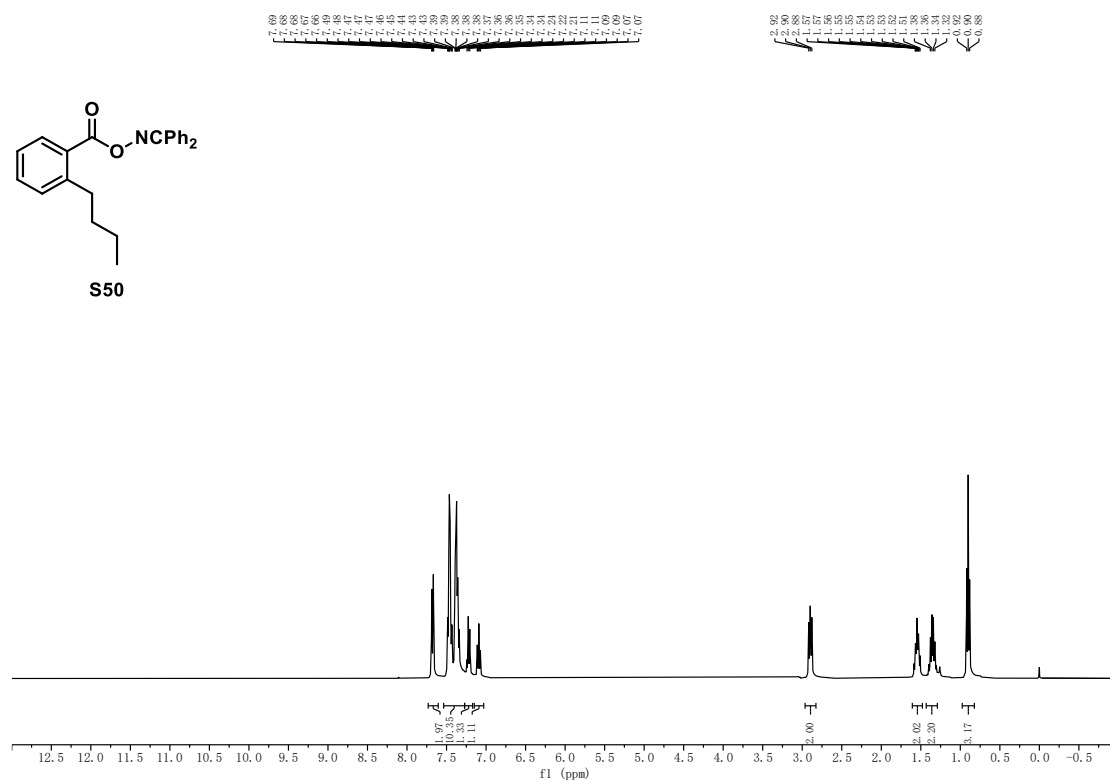
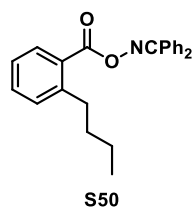


Supplementary Figure 113 ¹H NMR spectrum of S39

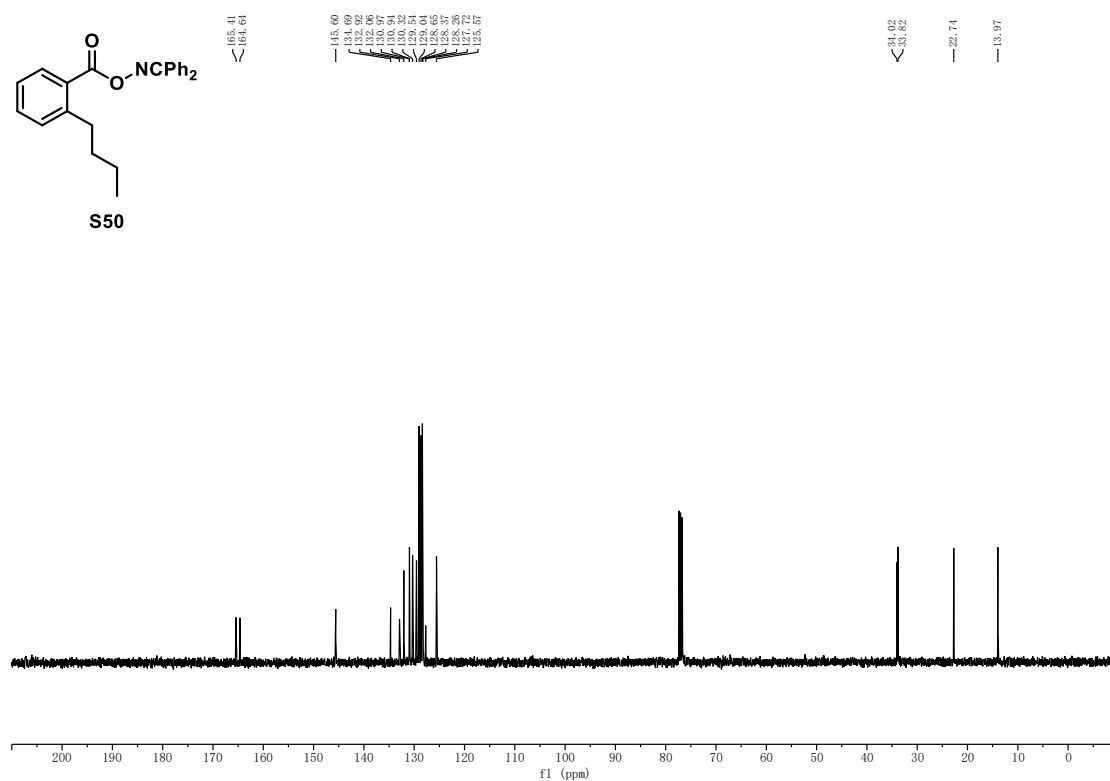


Supplementary Figure 114 ¹³C NMR spectrum of S39

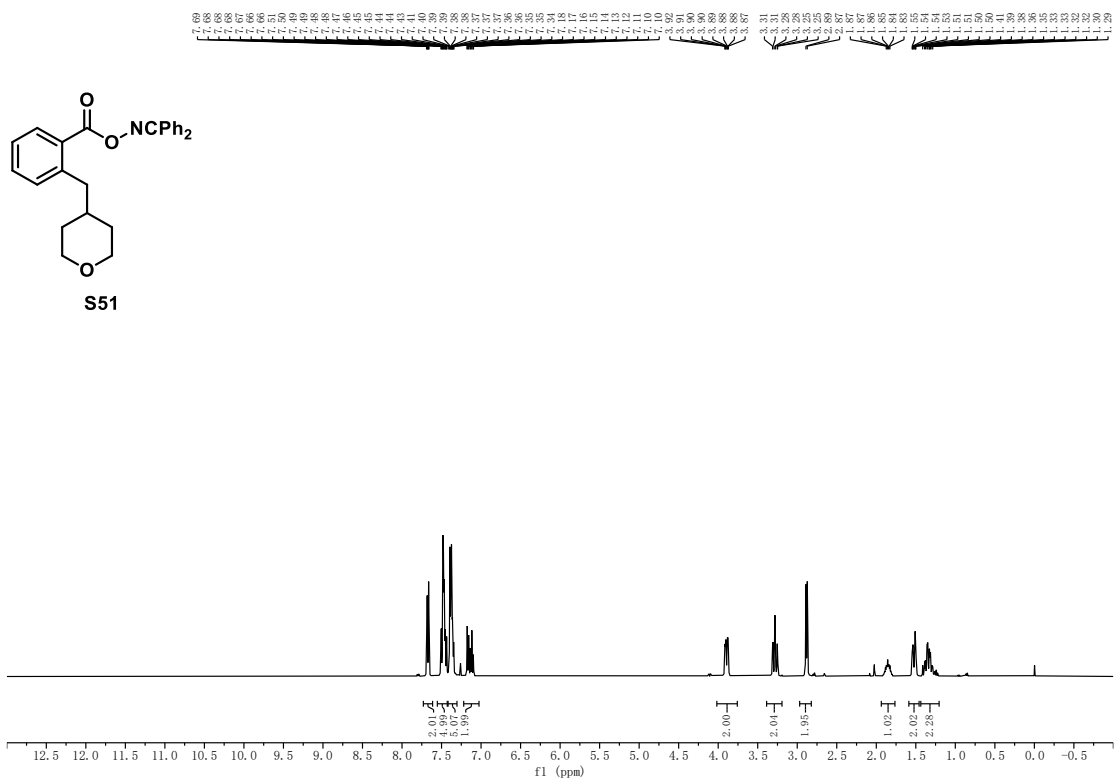




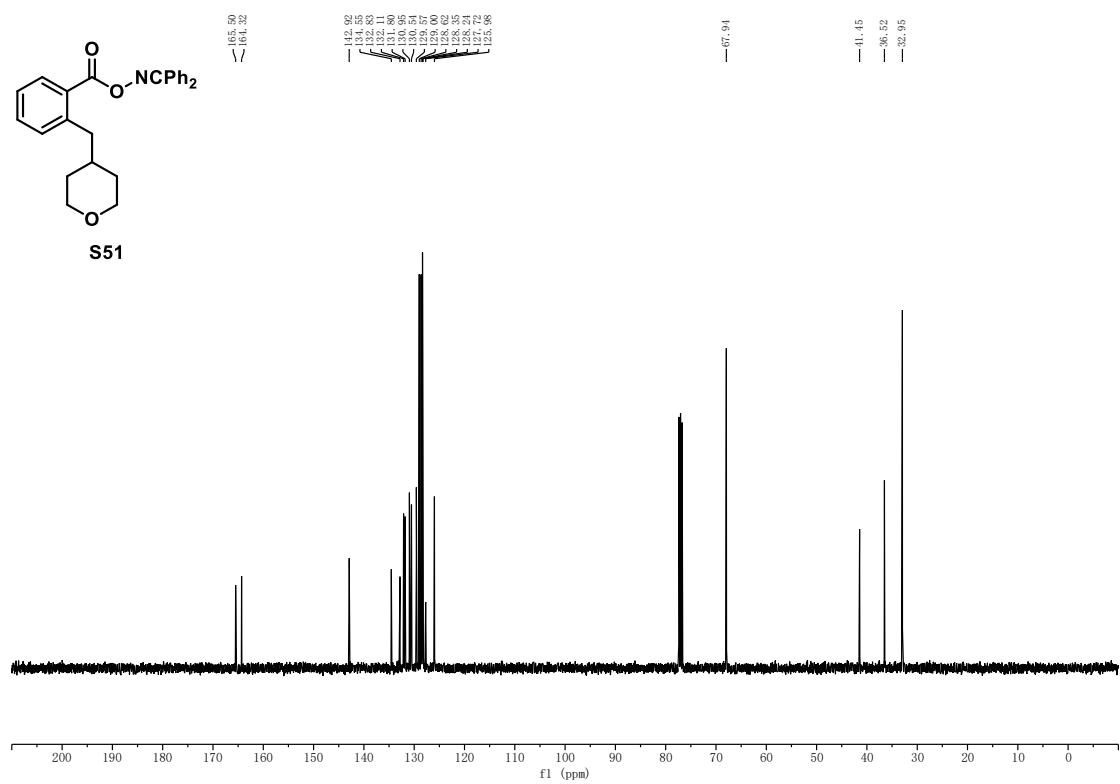
Supplementary Figure 121 ¹H NMR spectrum of S50



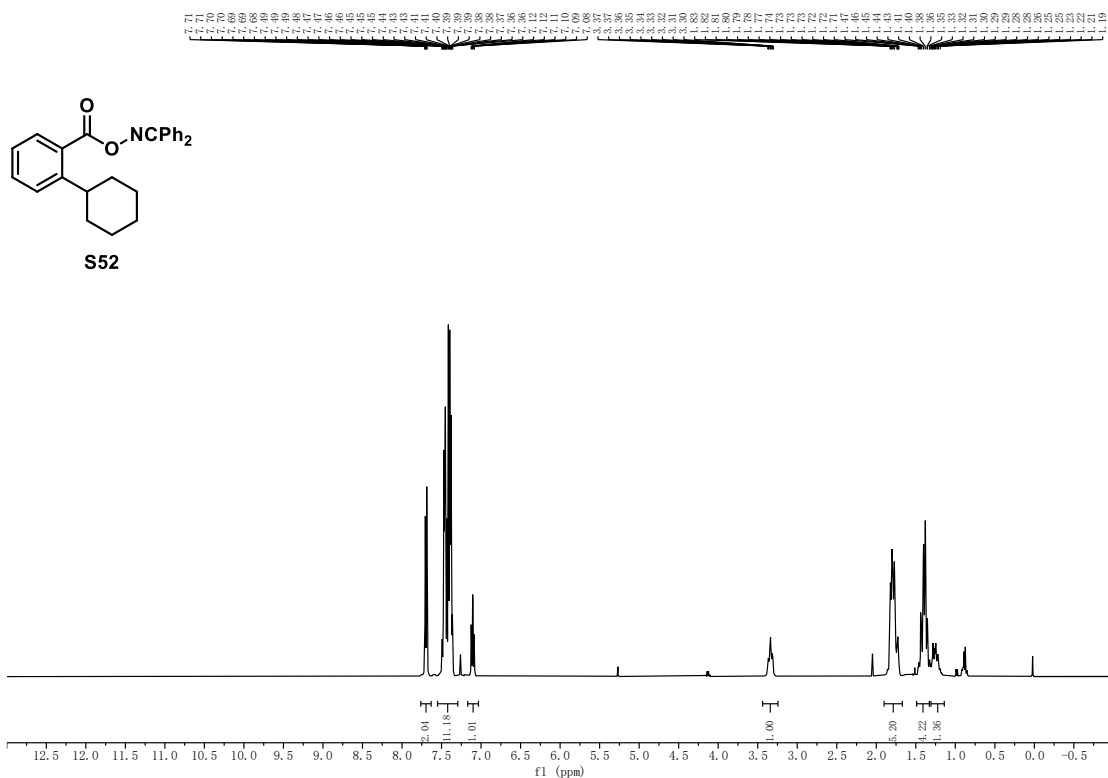
Supplementary Figure 122 ¹³C NMR spectrum of S50



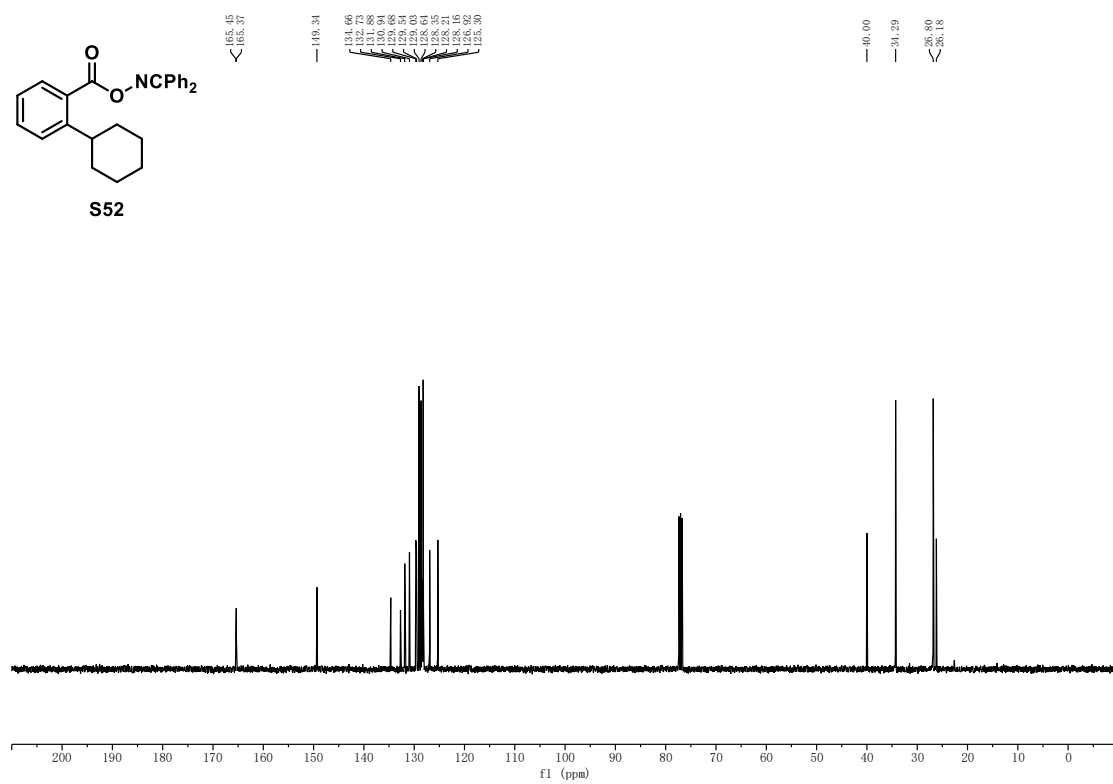
Supplementary Figure 123 ^1H NMR spectrum of S51



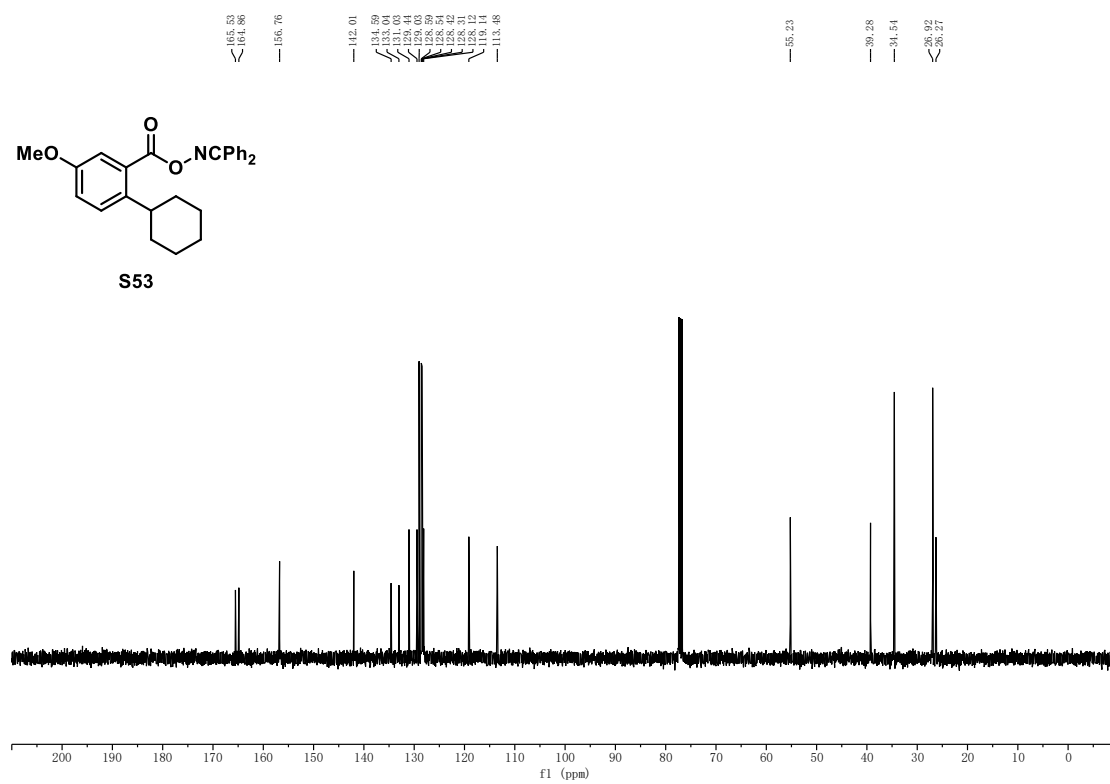
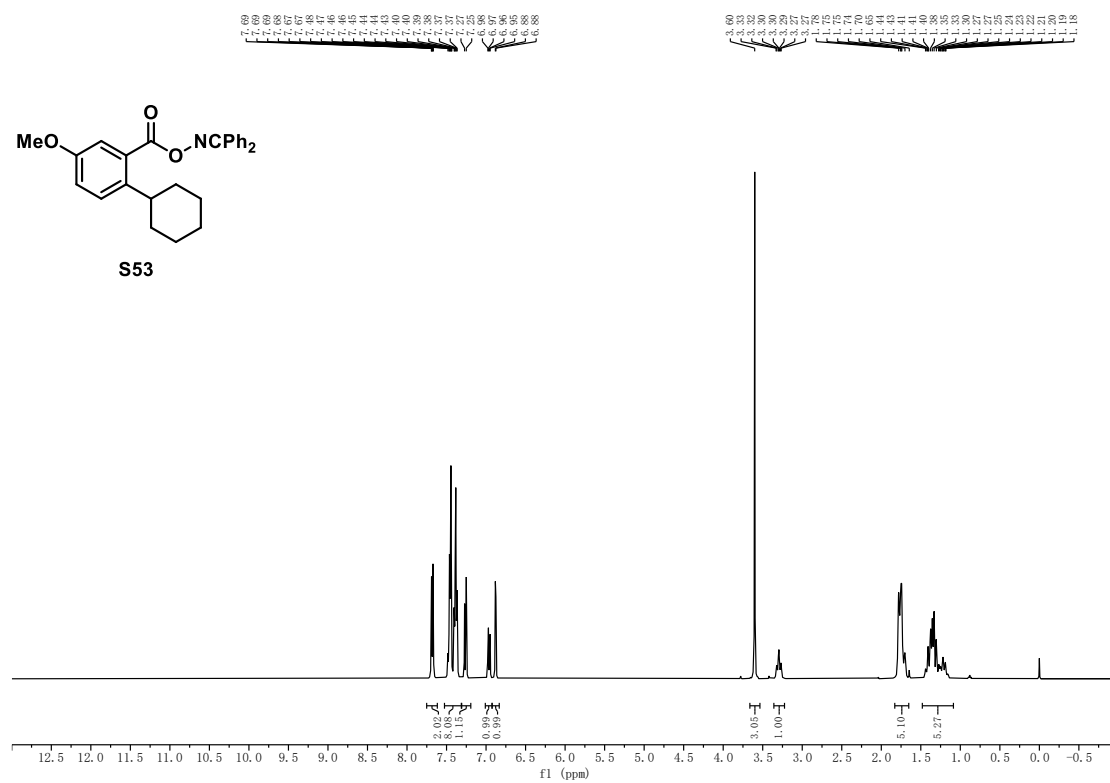
Supplementary Figure 124 ^{13}C NMR spectrum of S51

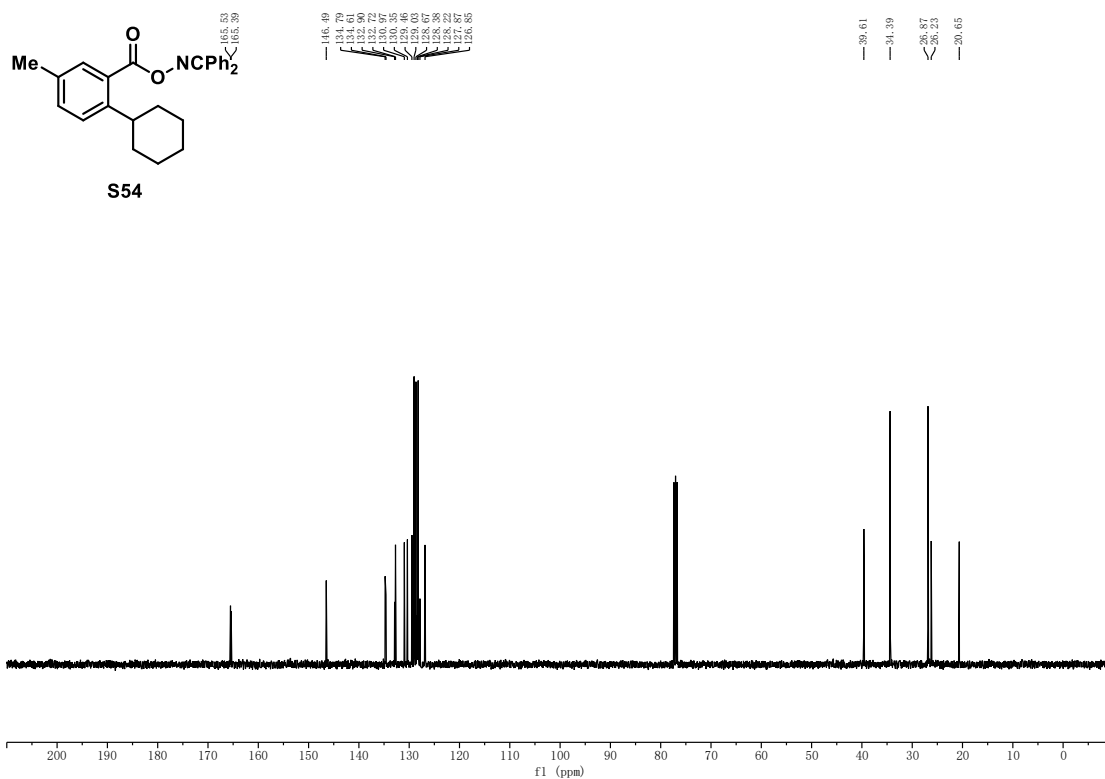
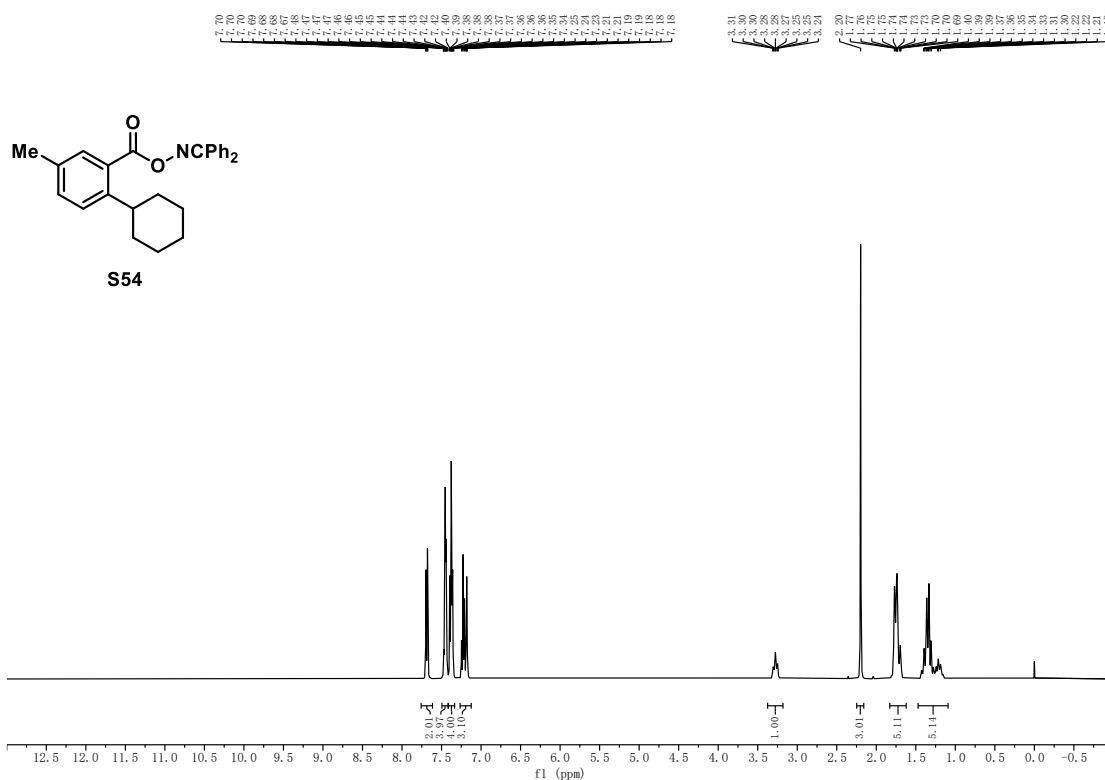


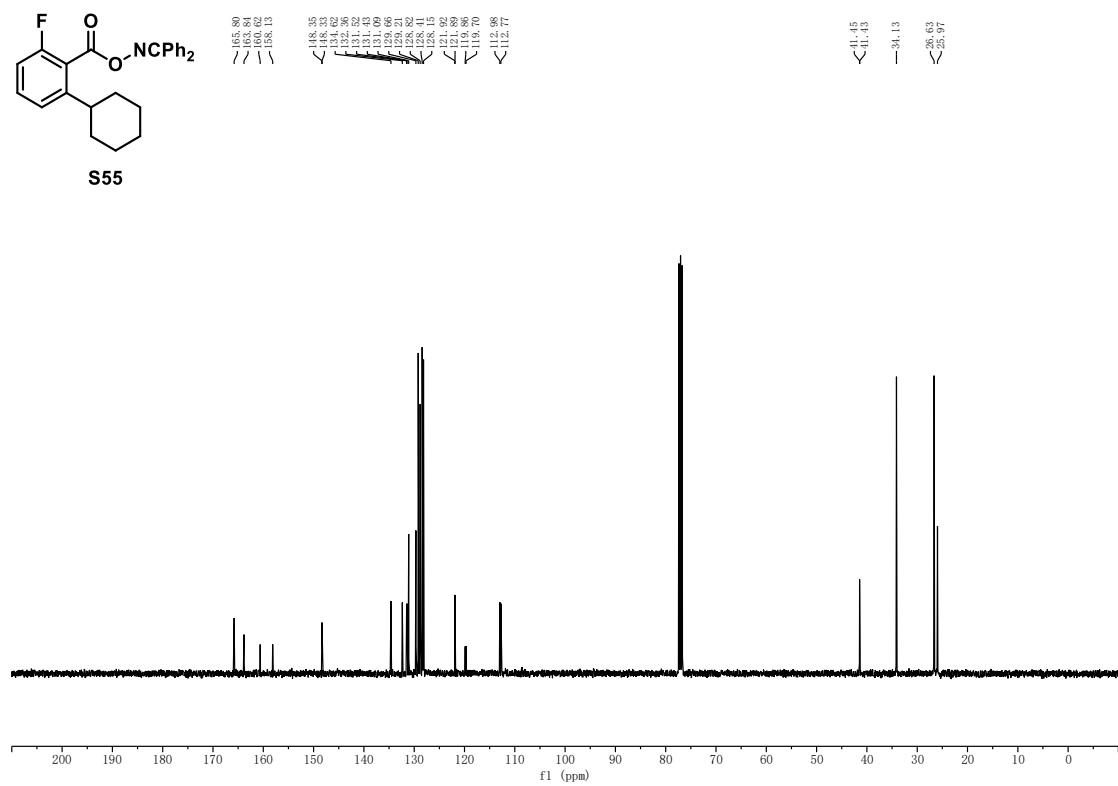
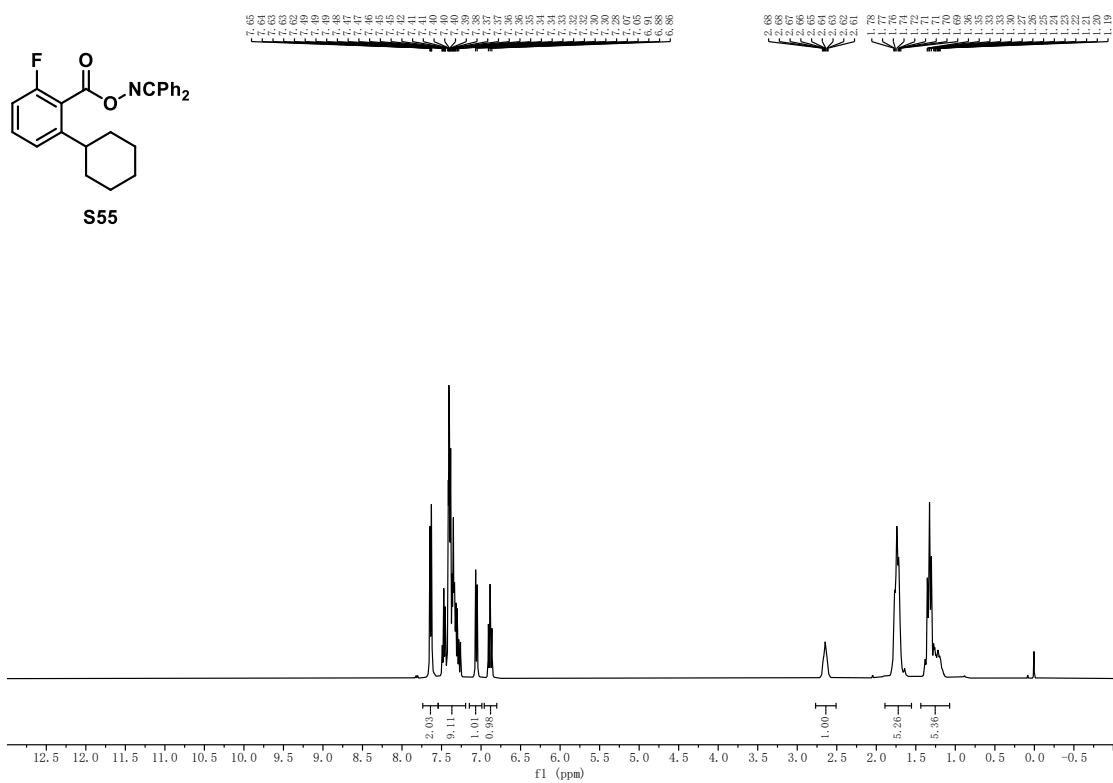
Supplementary Figure 125 ^1H NMR spectrum of S52

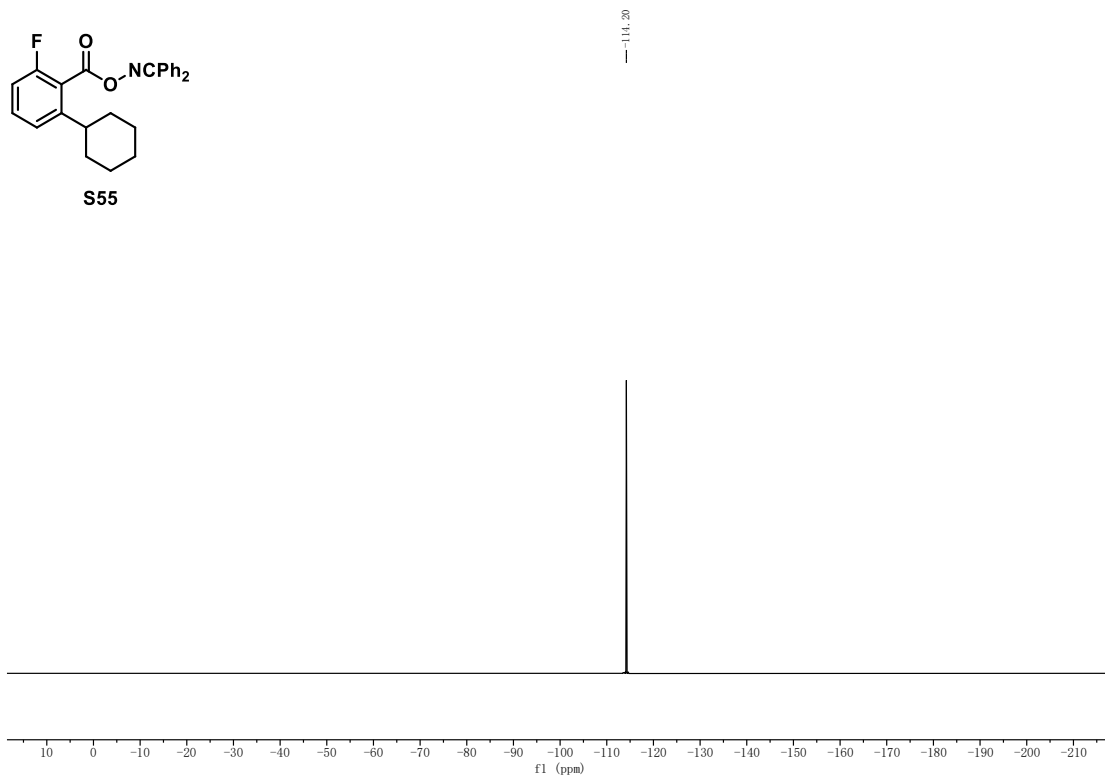


Supplementary Figure 126 ^{13}C NMR spectrum of S52

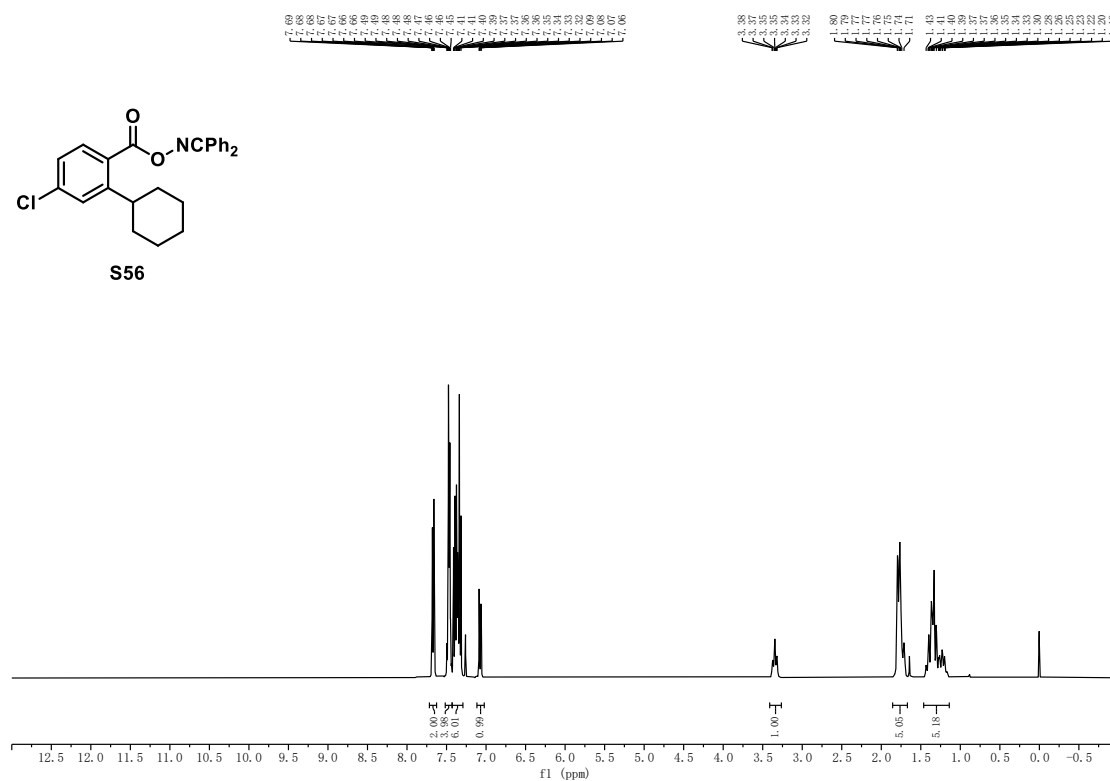




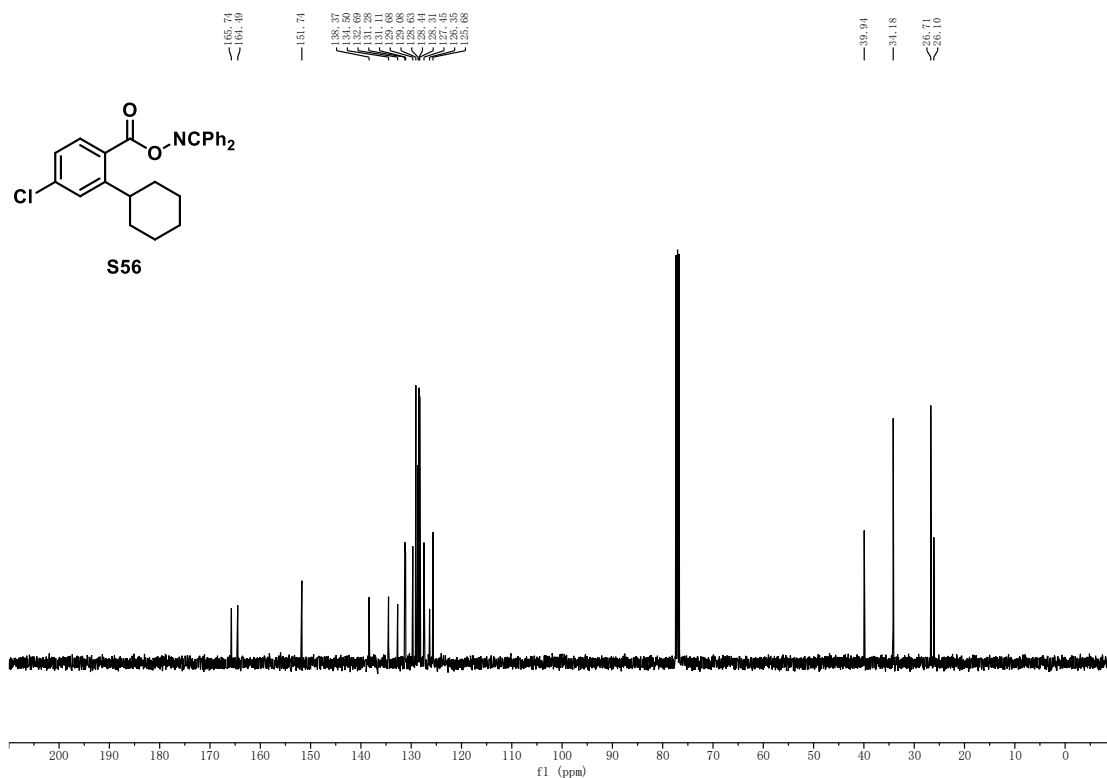




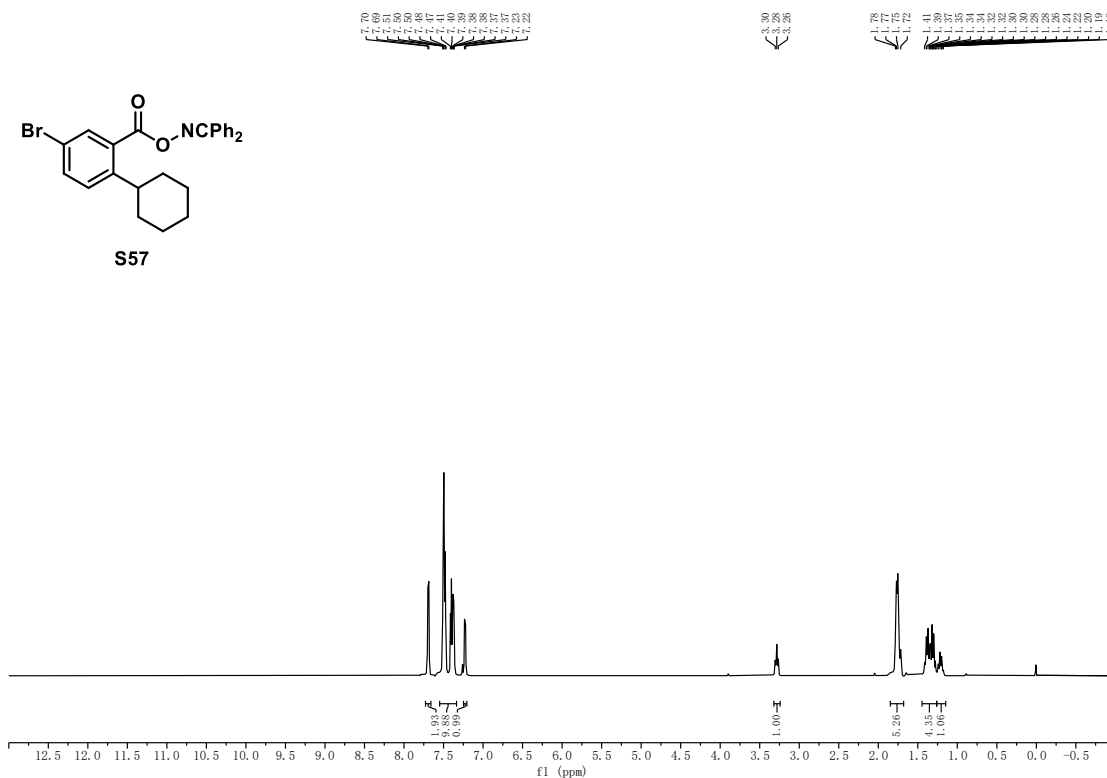
Supplementary Figure 133 ^{19}F NMR spectrum of S55



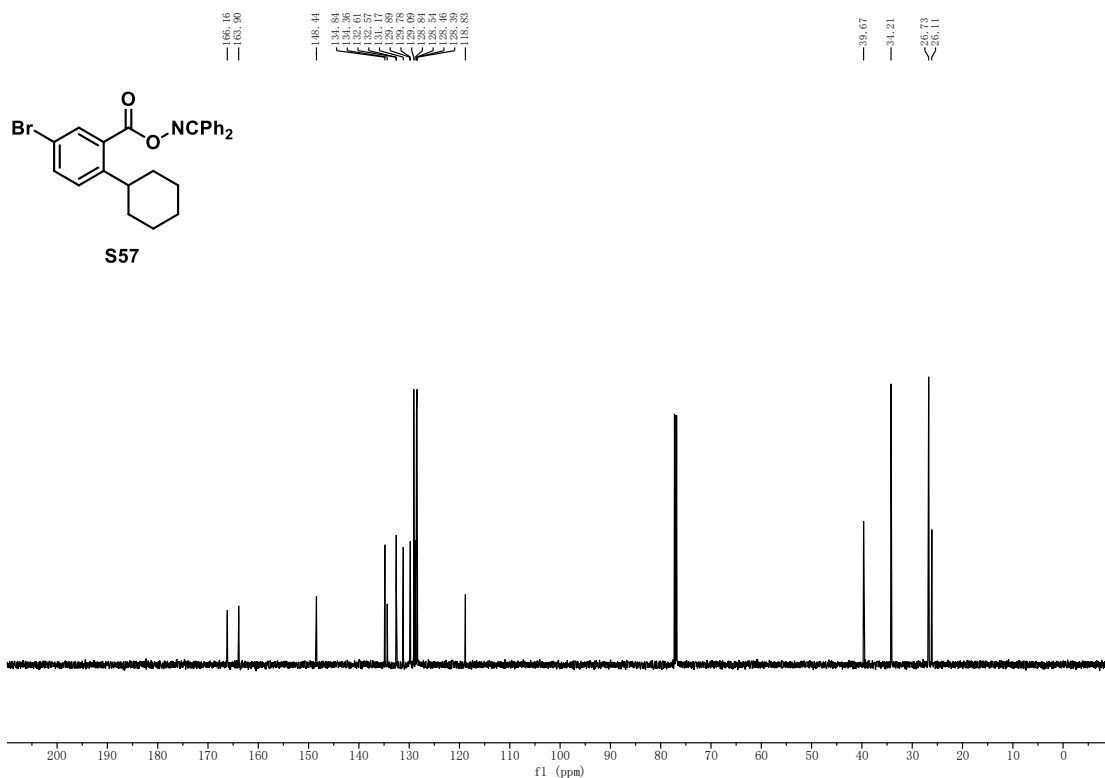
Supplementary Figure 134 ^1H NMR spectrum of S56



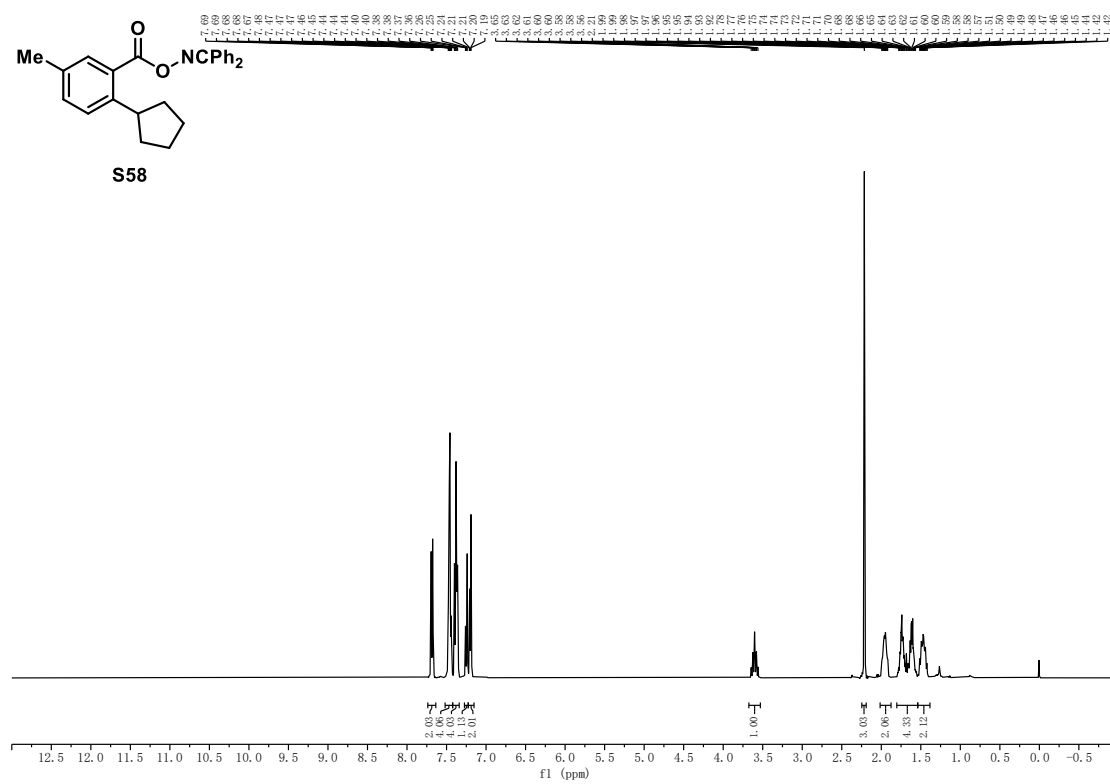
Supplementary Figure 135 ¹³C NMR spectrum of S56



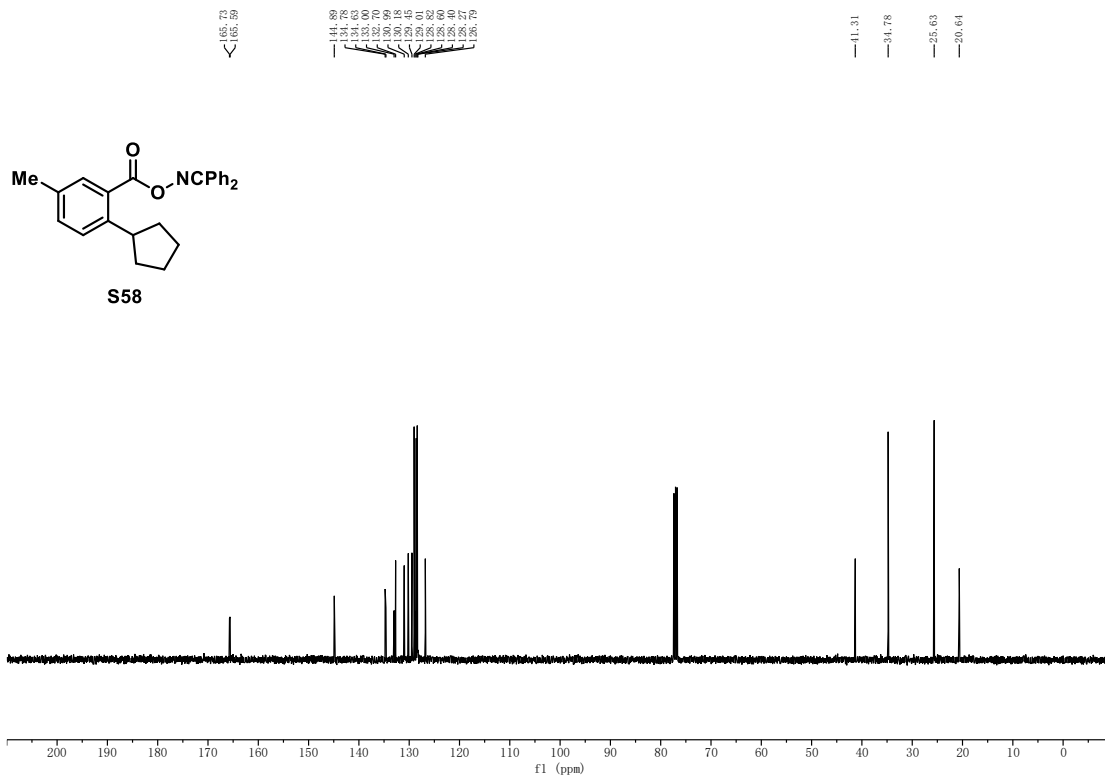
Supplementary Figure 136 ¹H NMR spectrum of S57



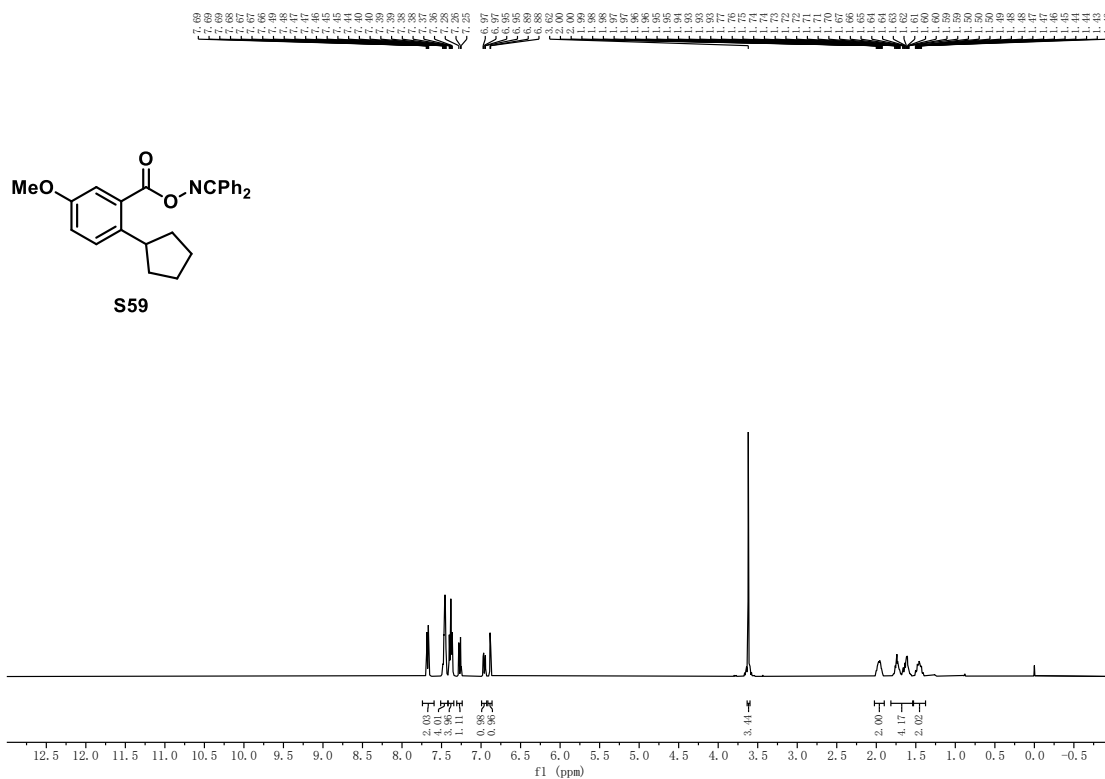
Supplementary Figure 137 ¹³C NMR spectrum of **S57**



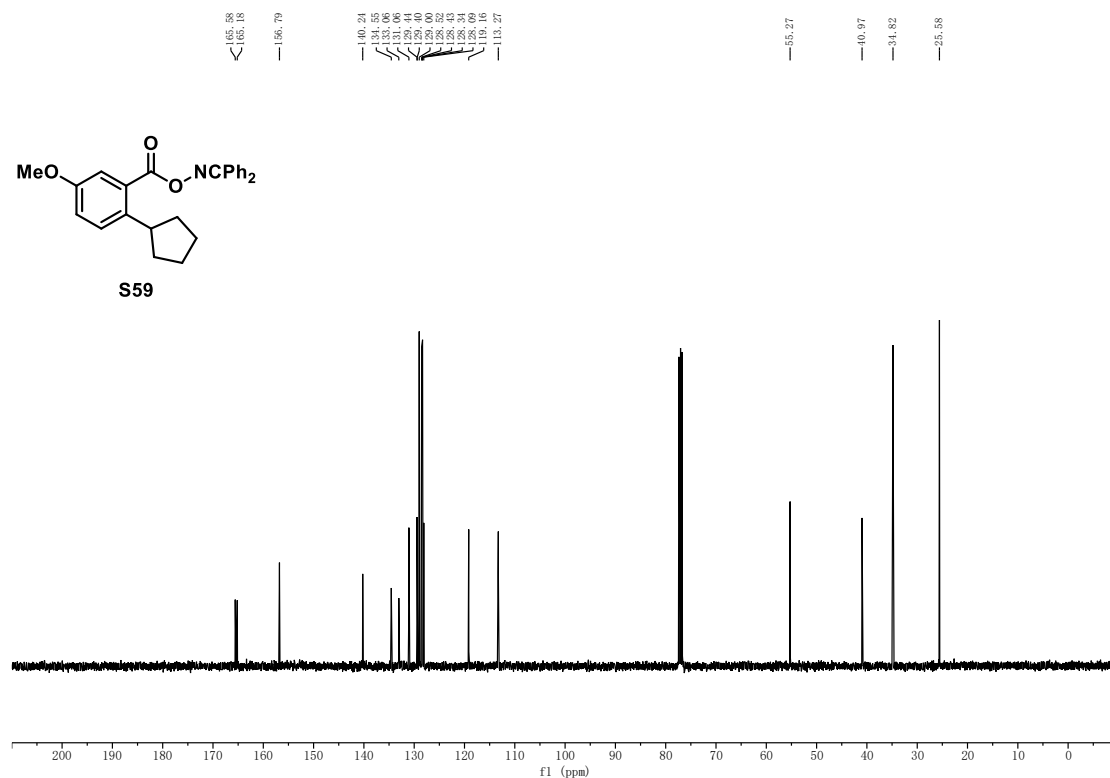
Supplementary Figure 138 ¹H NMR spectrum of **S58**



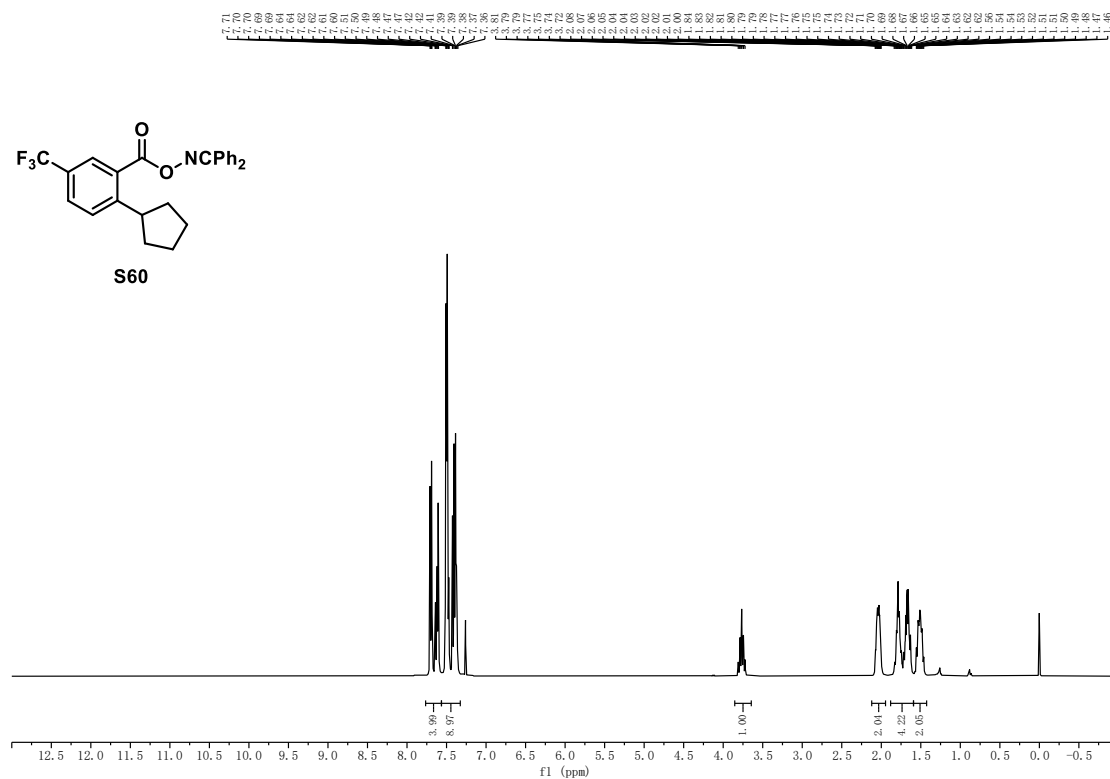
Supplementary Figure 139 ¹³C NMR spectrum of S58



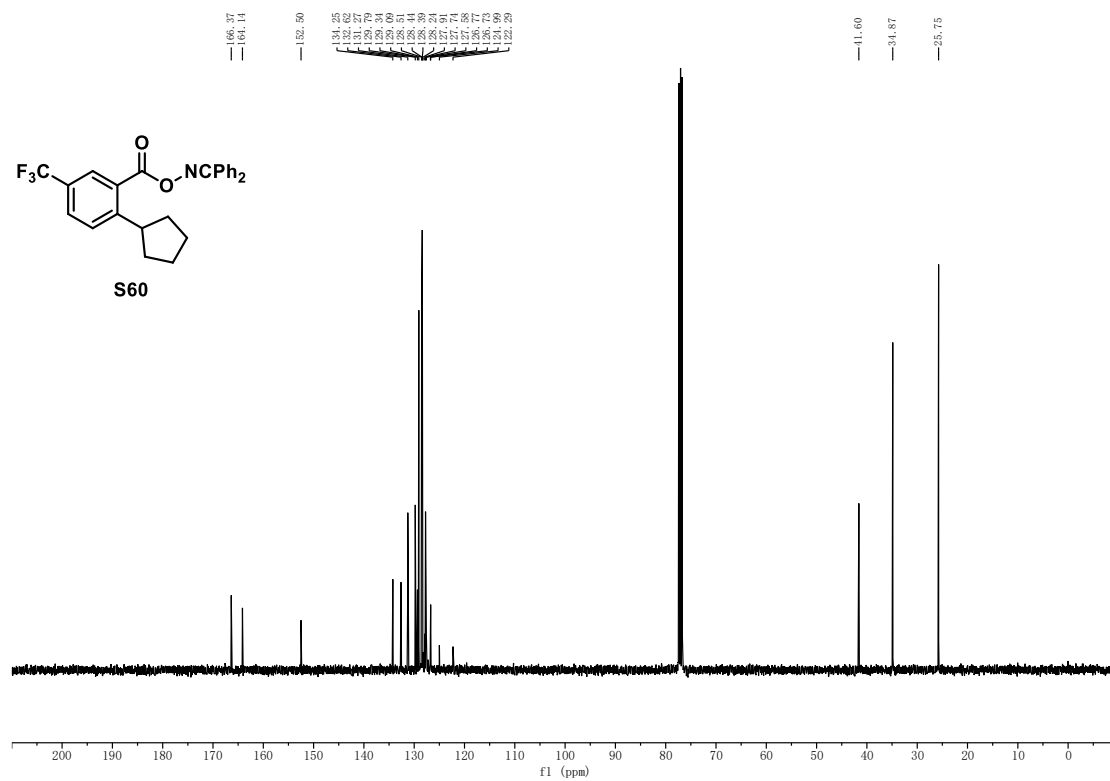
Supplementary Figure 140 ¹H NMR spectrum of S59



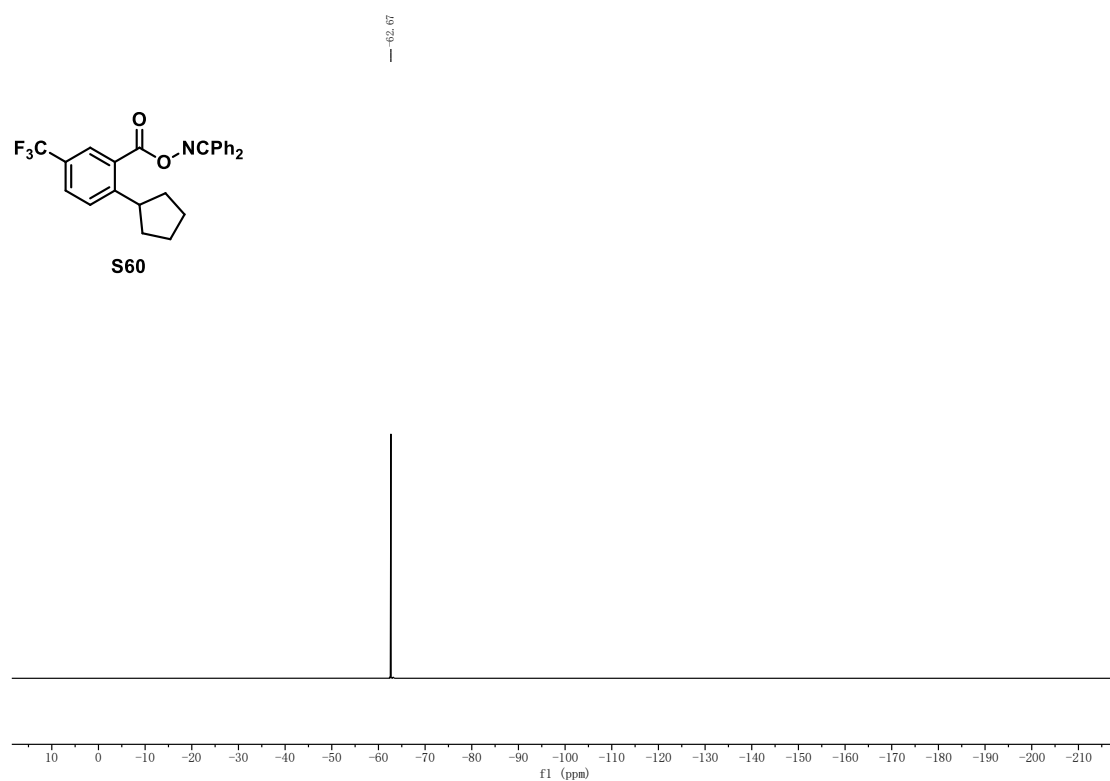
Supplementary Figure 141 ¹³C NMR spectrum of S59



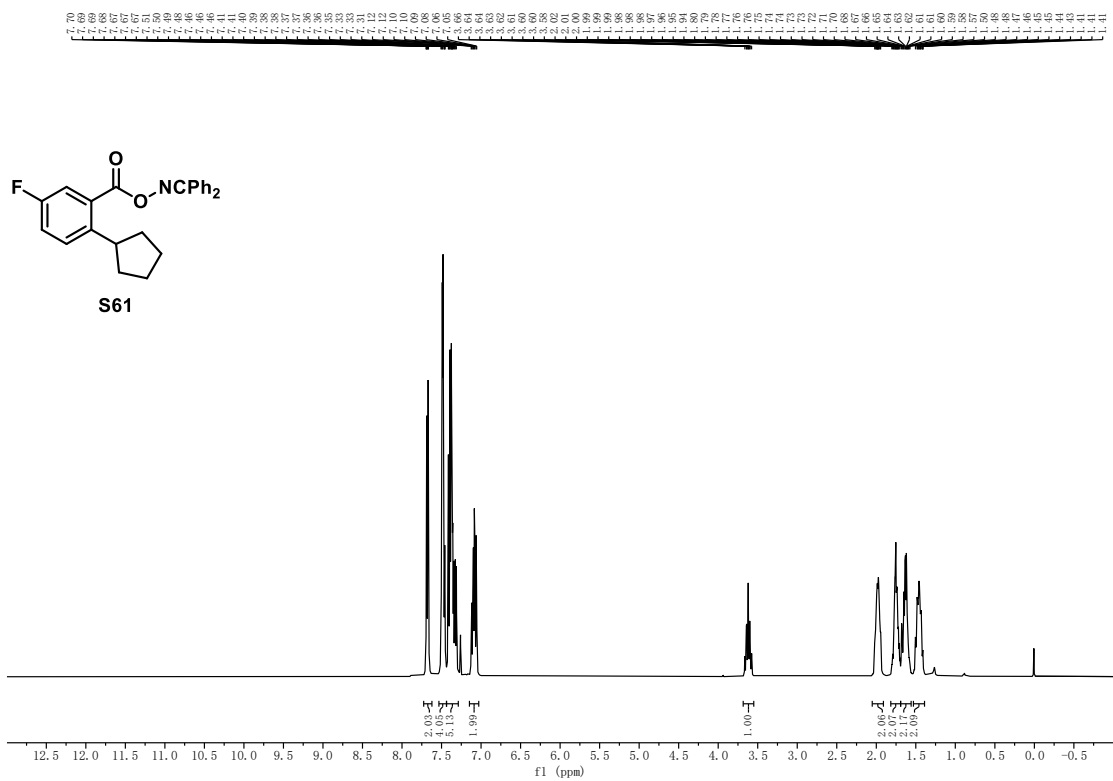
Supplementary Figure 142 ¹H NMR spectrum of S60



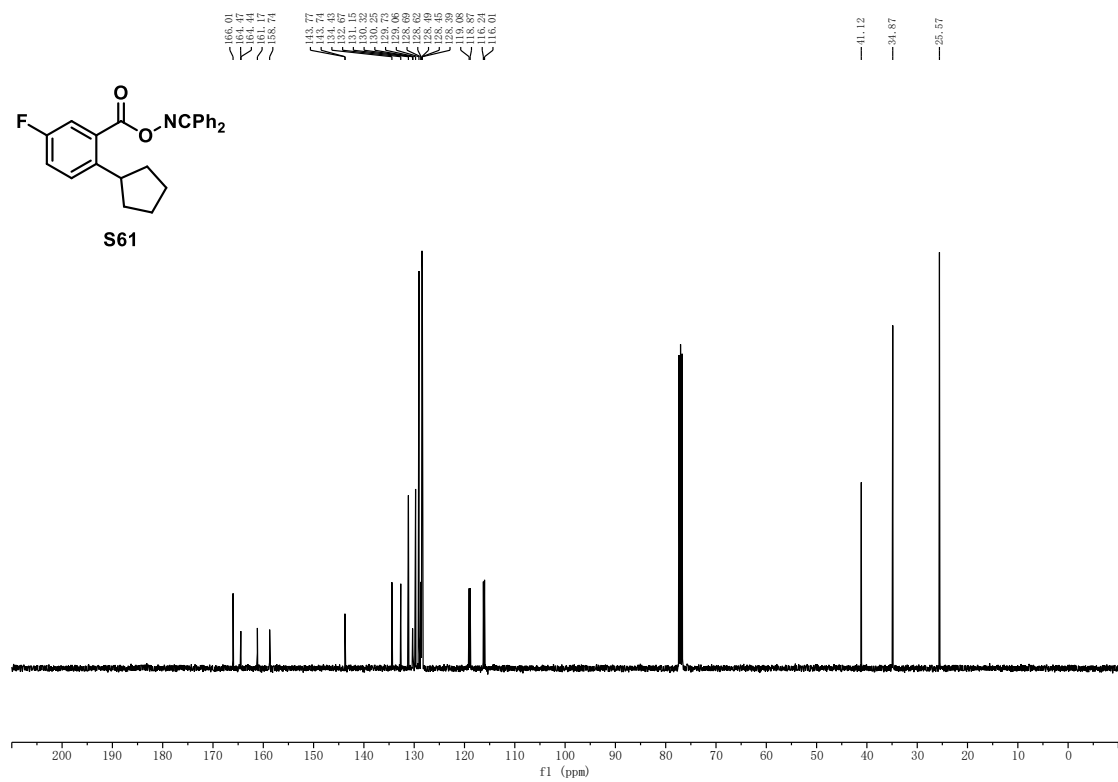
Supplementary Figure 143 ¹³C NMR spectrum of S60



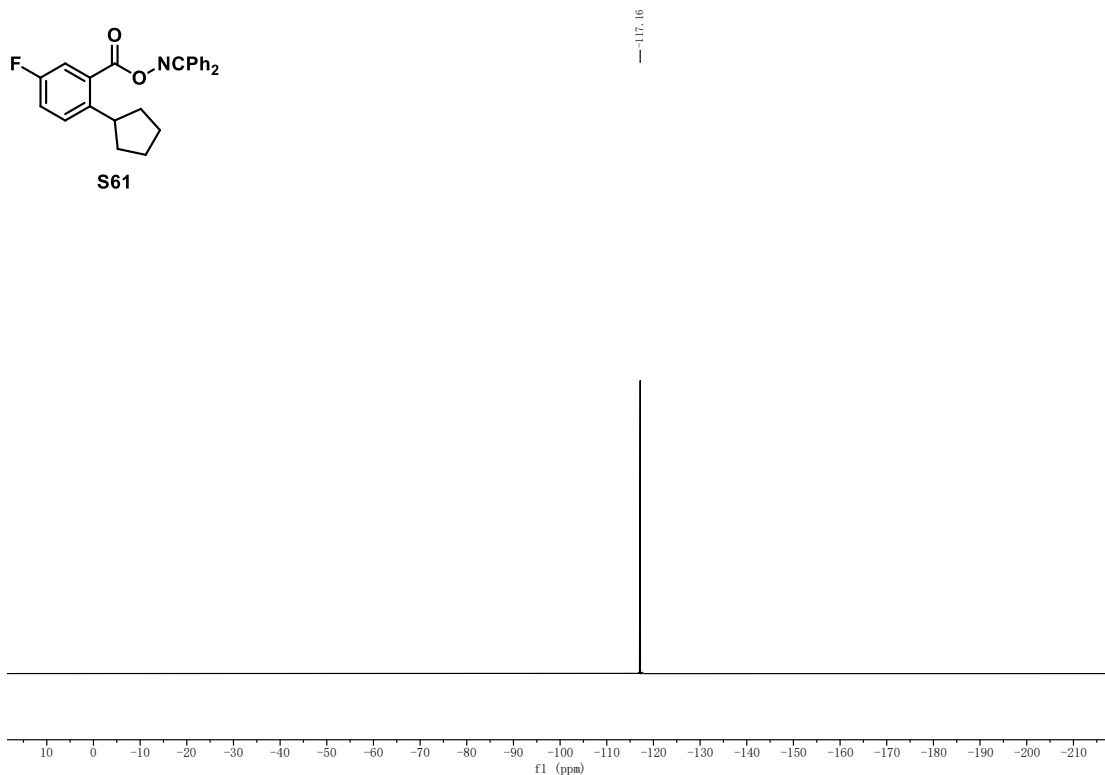
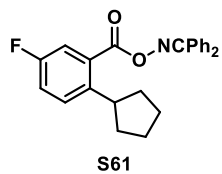
Supplementary Figure 144 ¹⁹F NMR spectrum of S60



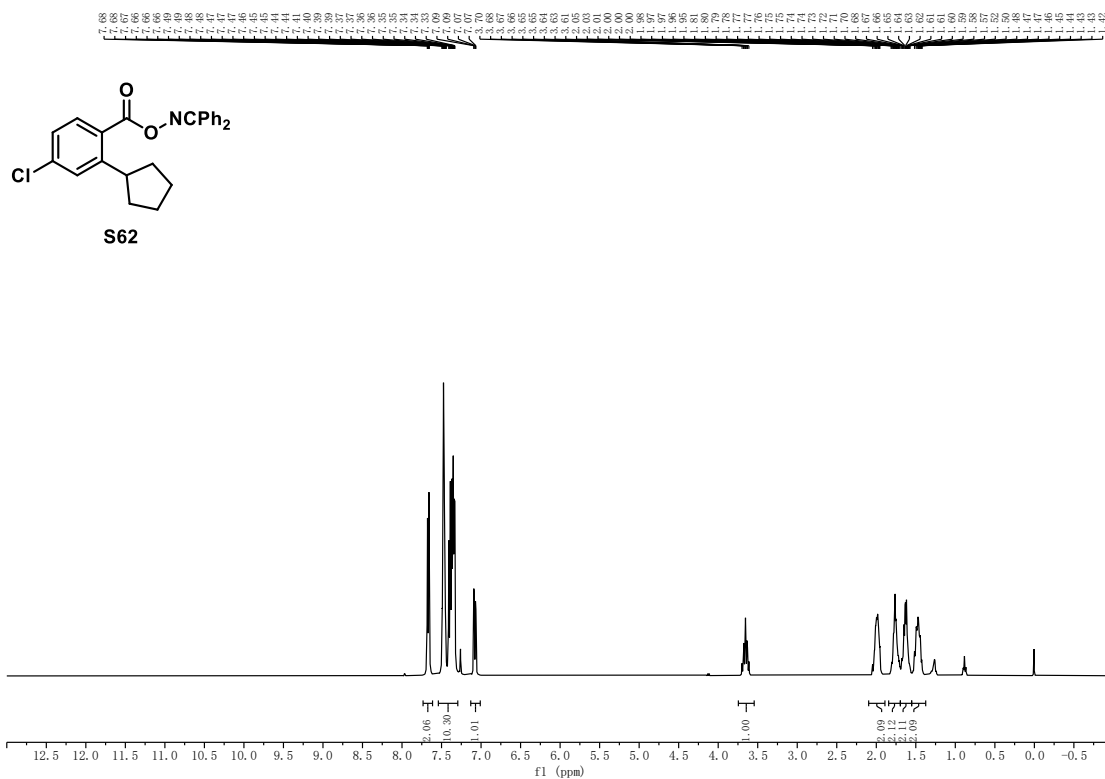
Supplementary Figure 145 ¹H NMR spectrum of S61



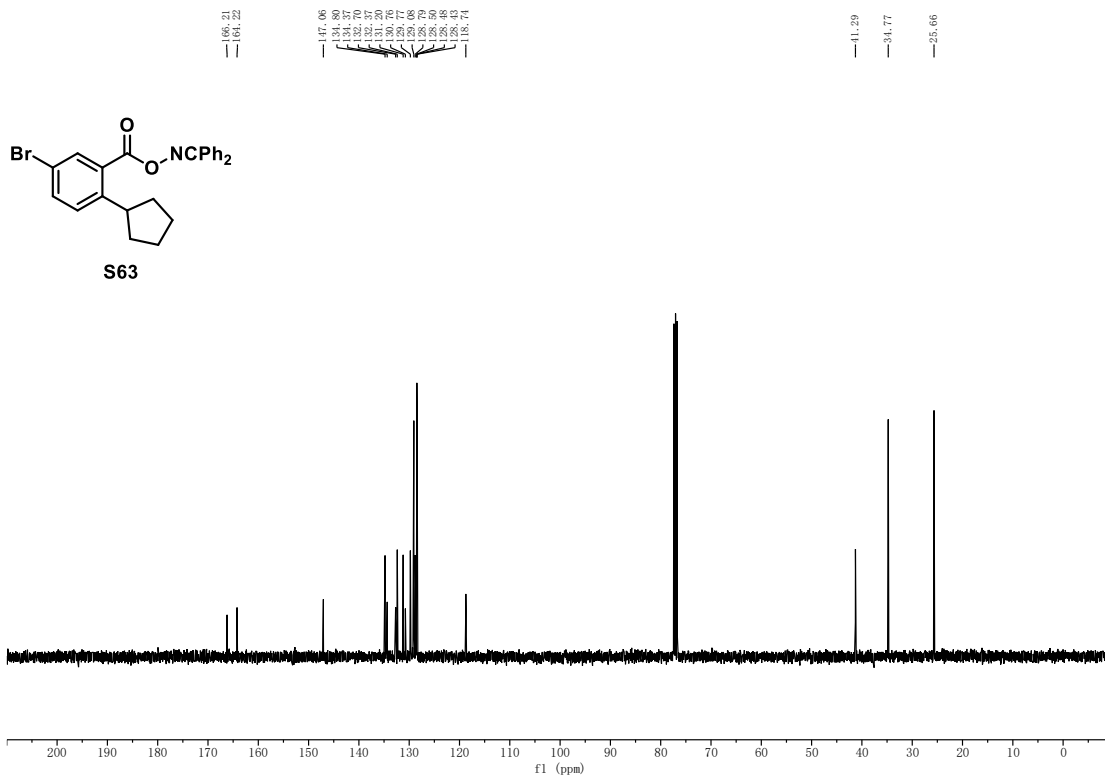
Supplementary Figure 146 ¹³C NMR spectrum of S61



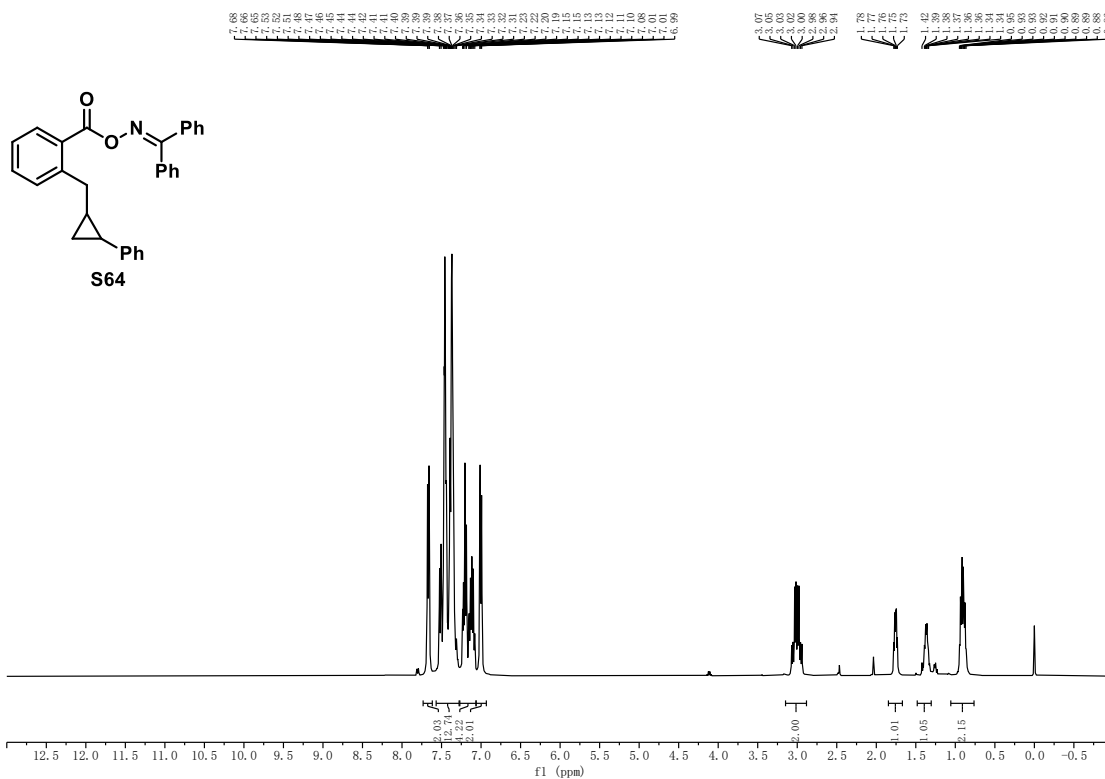
Supplementary Figure 147 ^{19}F NMR spectrum of S61



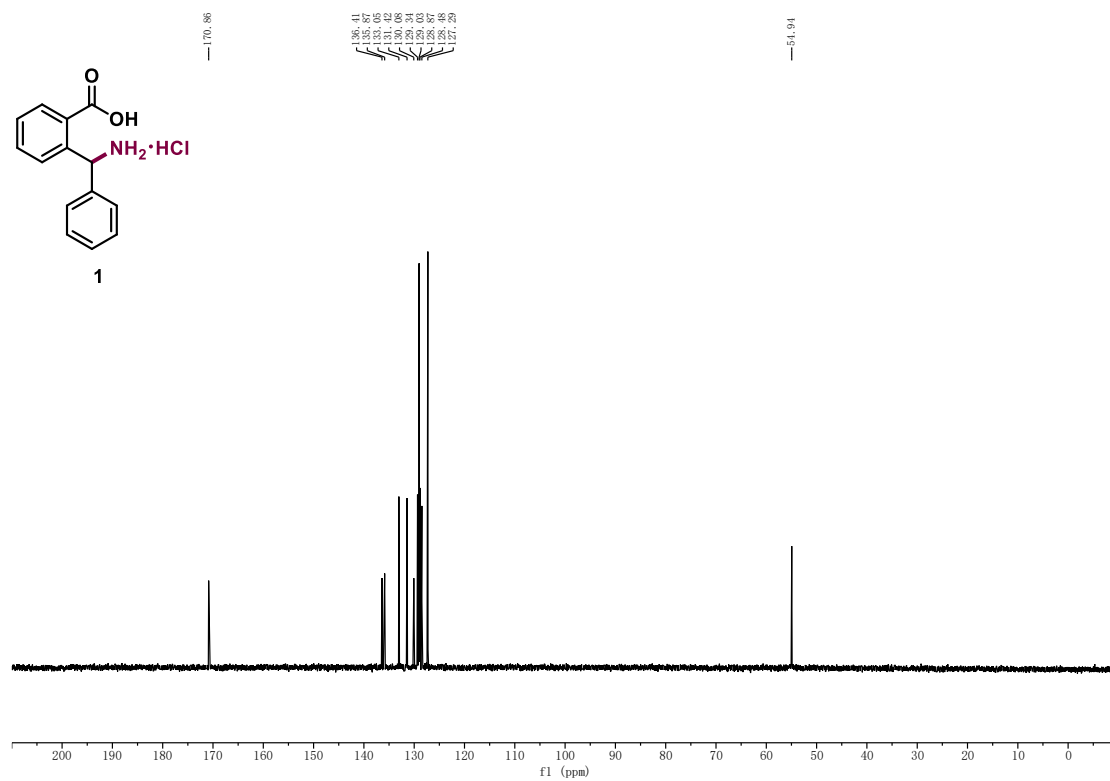
Supplementary Figure 148 ^1H NMR spectrum of S62



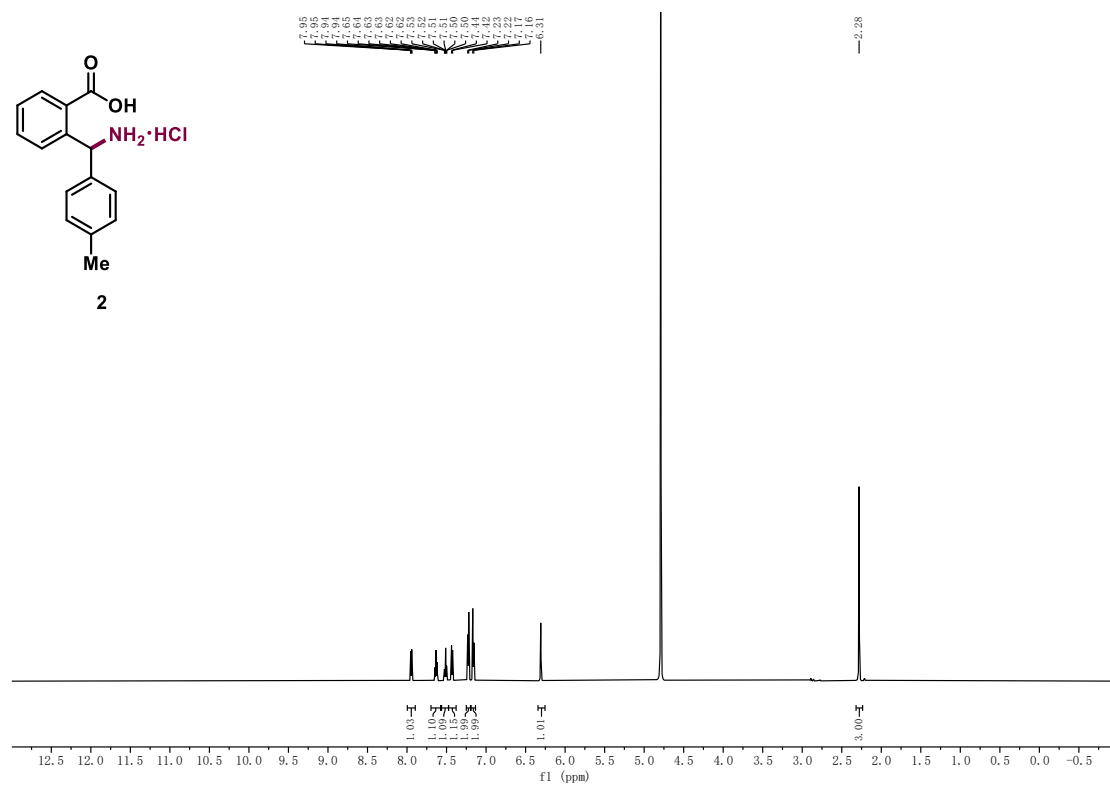
Supplementary Figure 151 ^{13}C NMR spectrum of S63



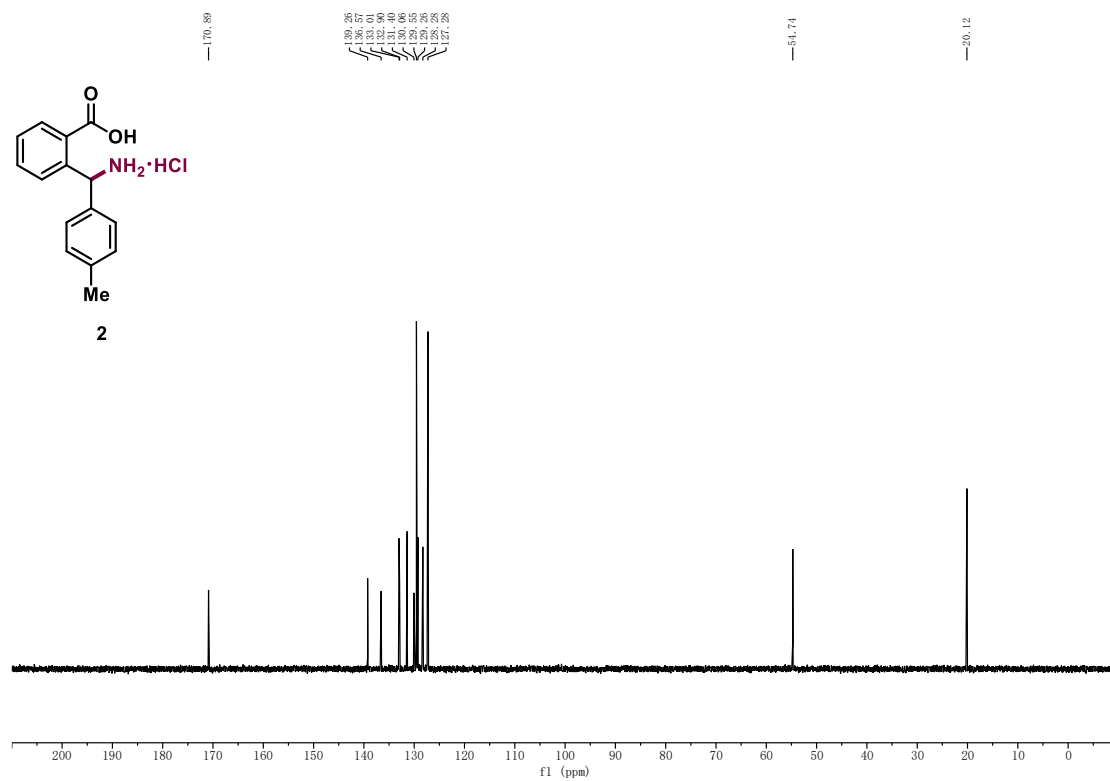
Supplementary Figure 152 ^1H NMR spectrum of S64



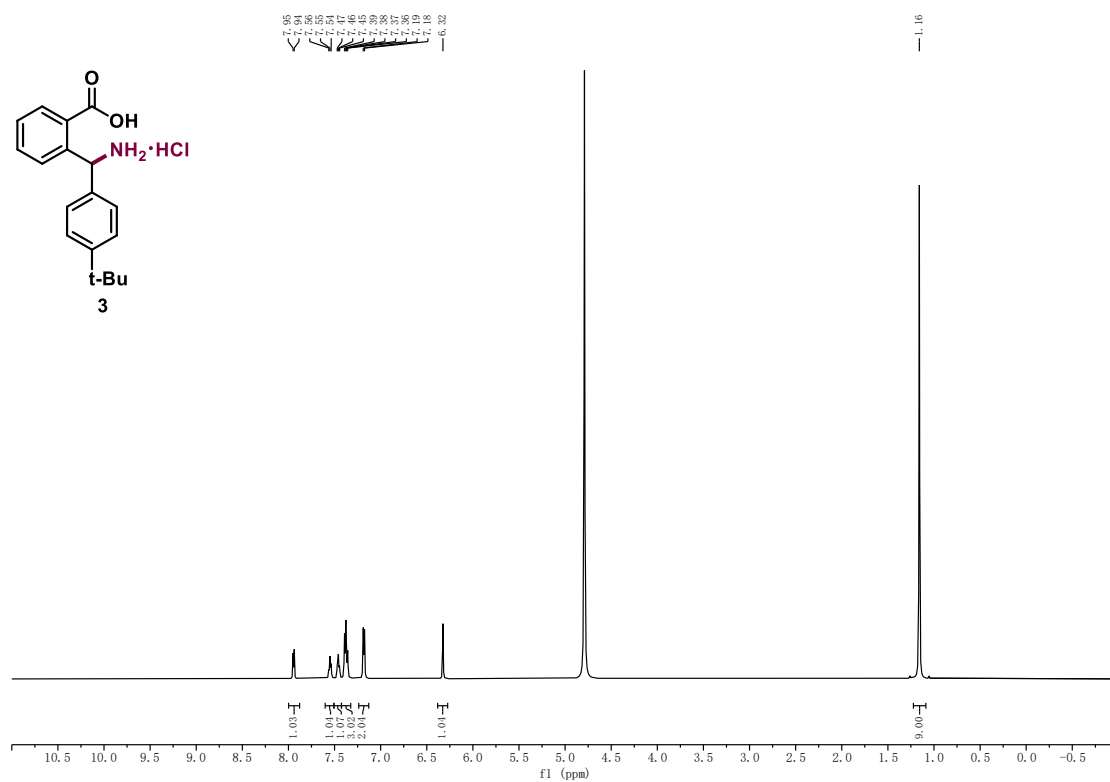
Supplementary Figure 157 ^{13}C NMR spectrum of compound 1



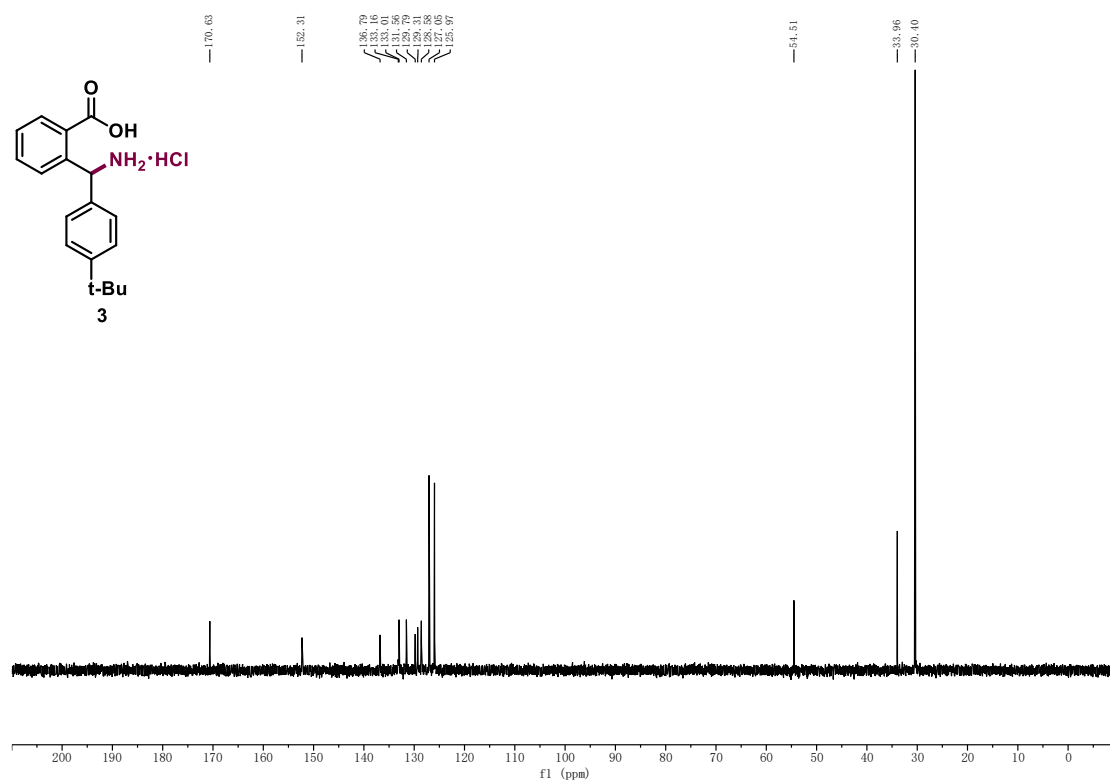
Supplementary Figure 158 ^1H NMR spectrum of compound 2



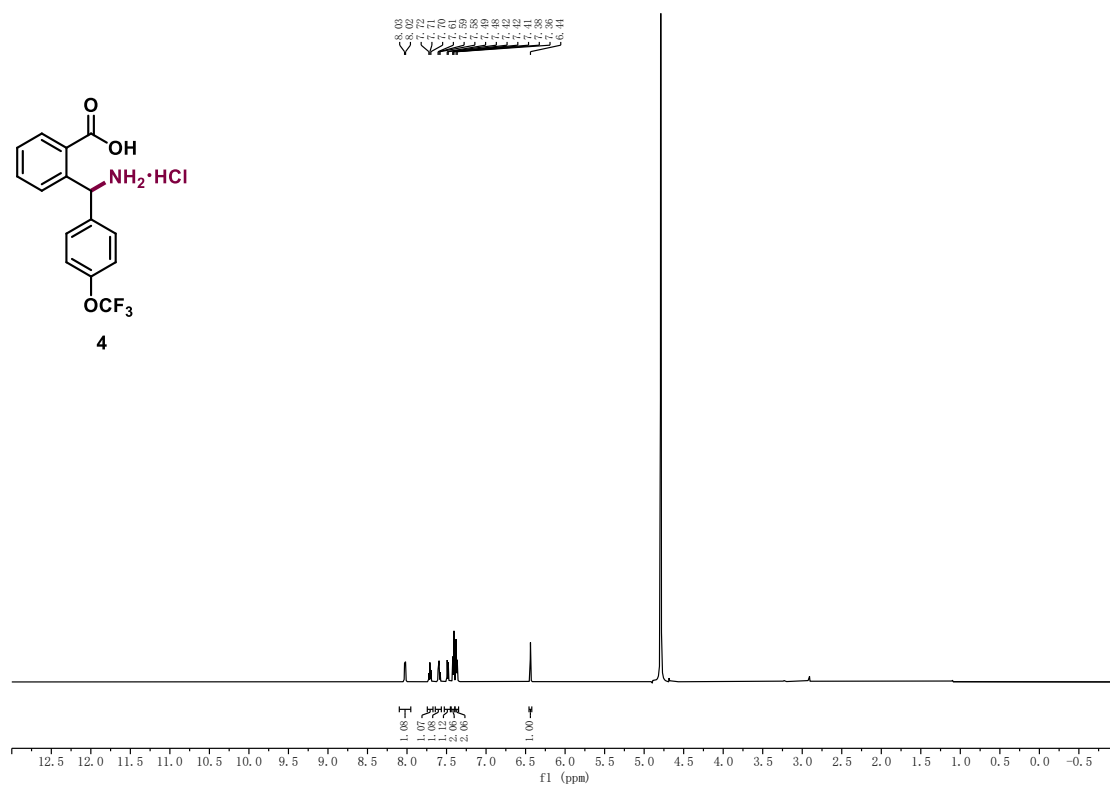
Supplementary Figure 159 ¹³C NMR spectrum of compound 2



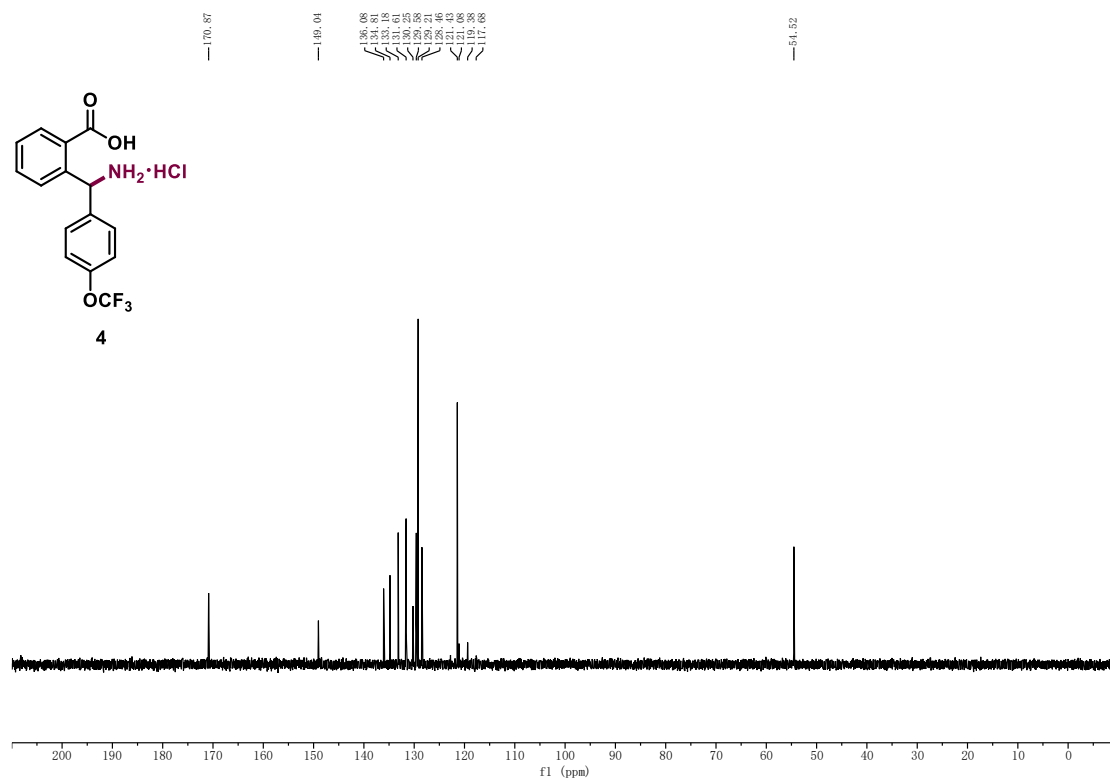
Supplementary Figure 160 ¹H NMR spectrum of compound 3



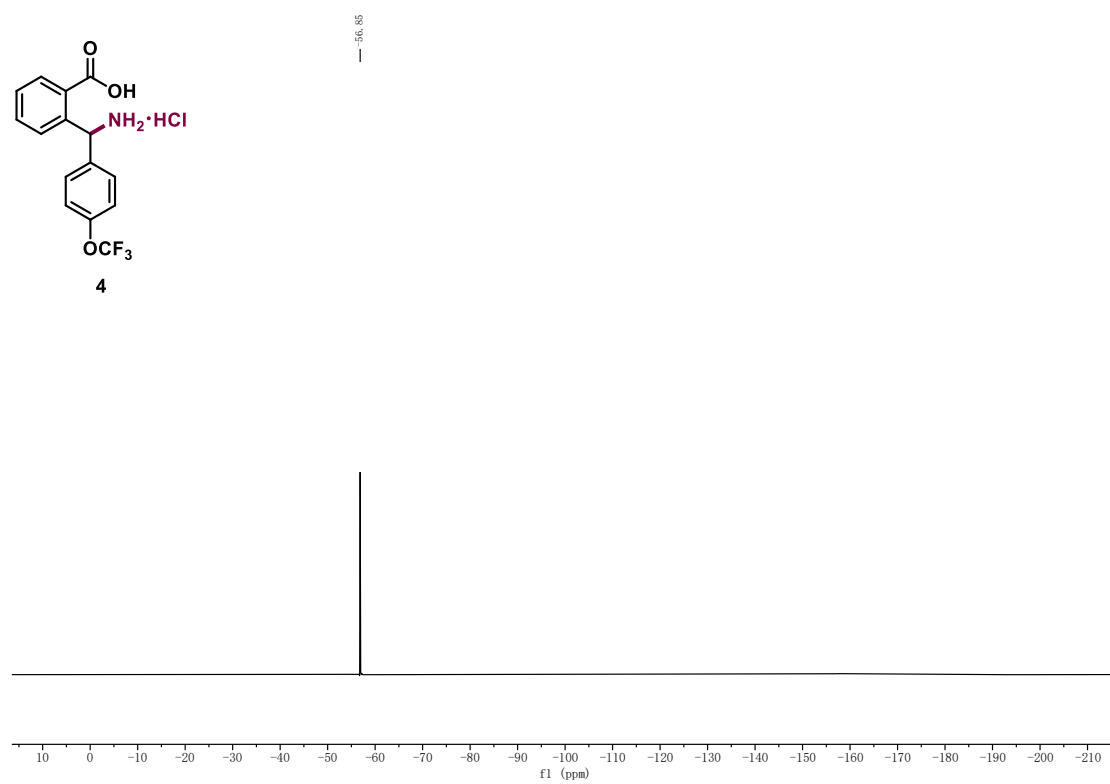
Supplementary Figure 161 ^{13}C NMR spectrum of compound 3



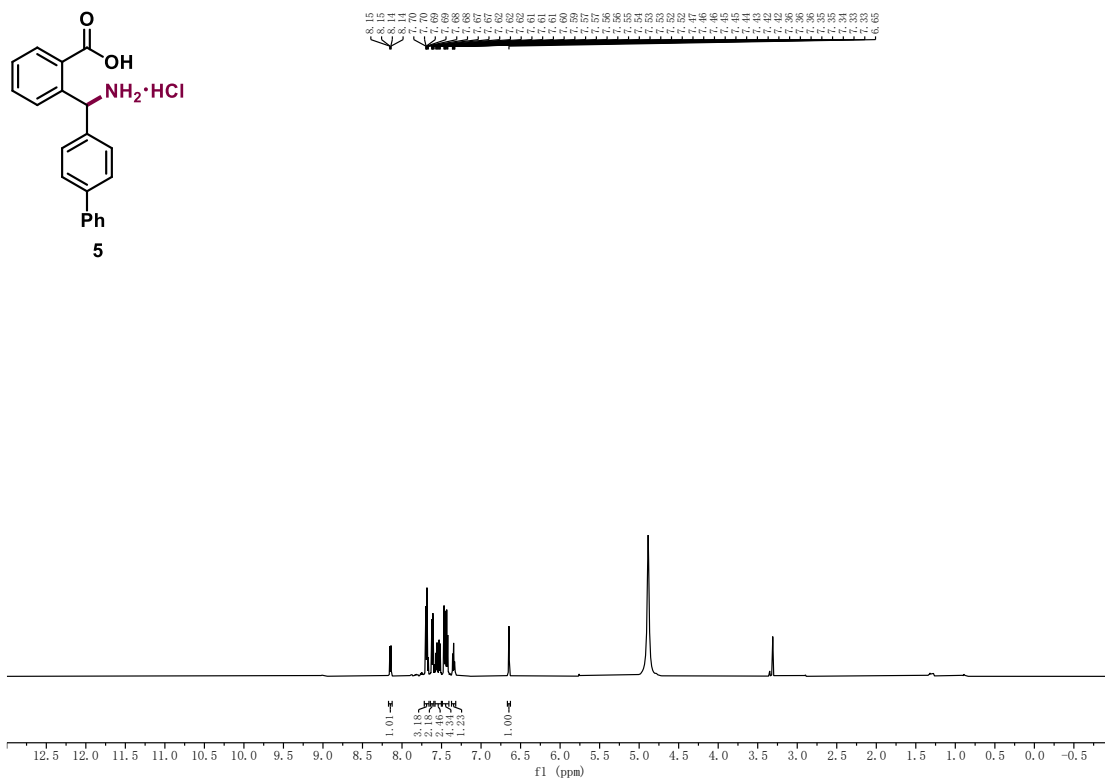
Supplementary Figure 162 ^1H NMR spectrum of compound 4



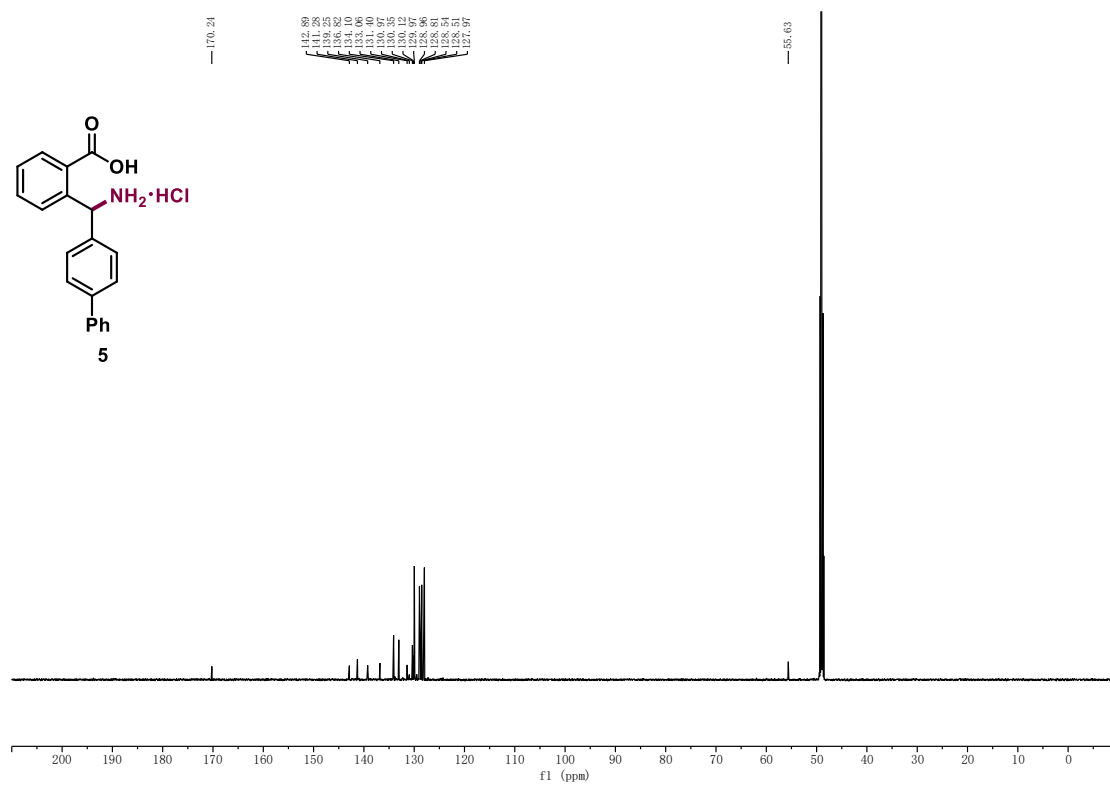
Supplementary Figure 163 ¹³C NMR spectrum of compound 4



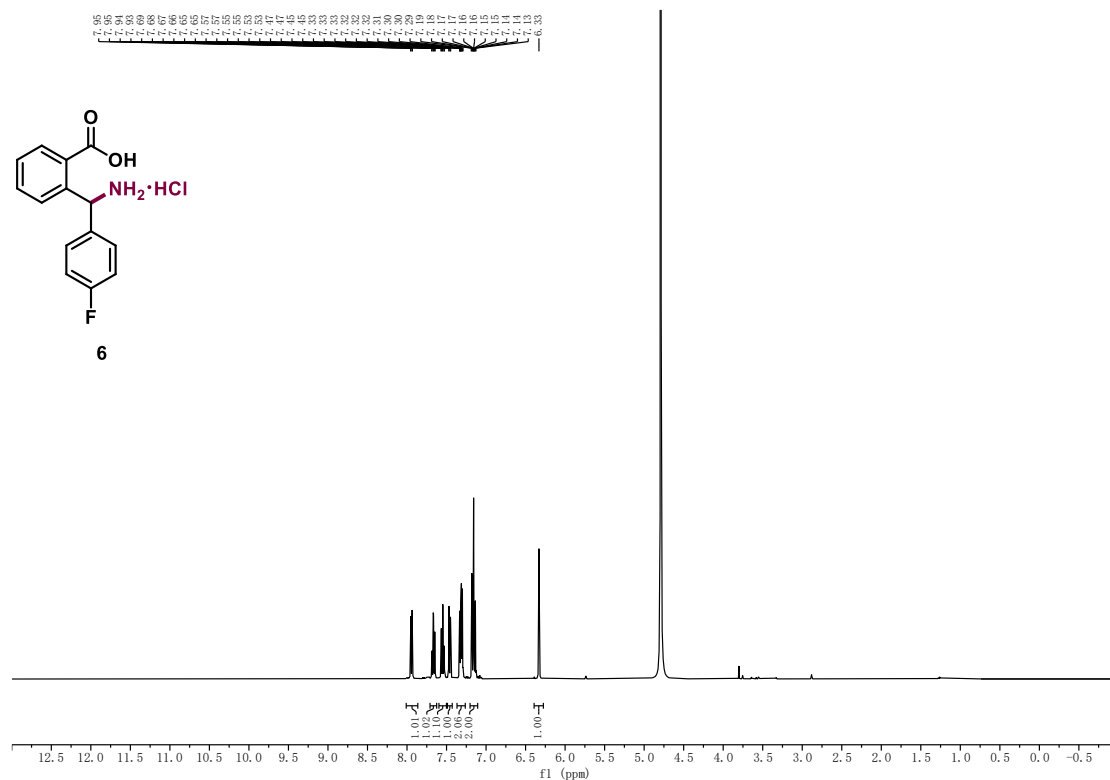
Supplementary Figure 164 ¹⁹F NMR spectrum of compound 4



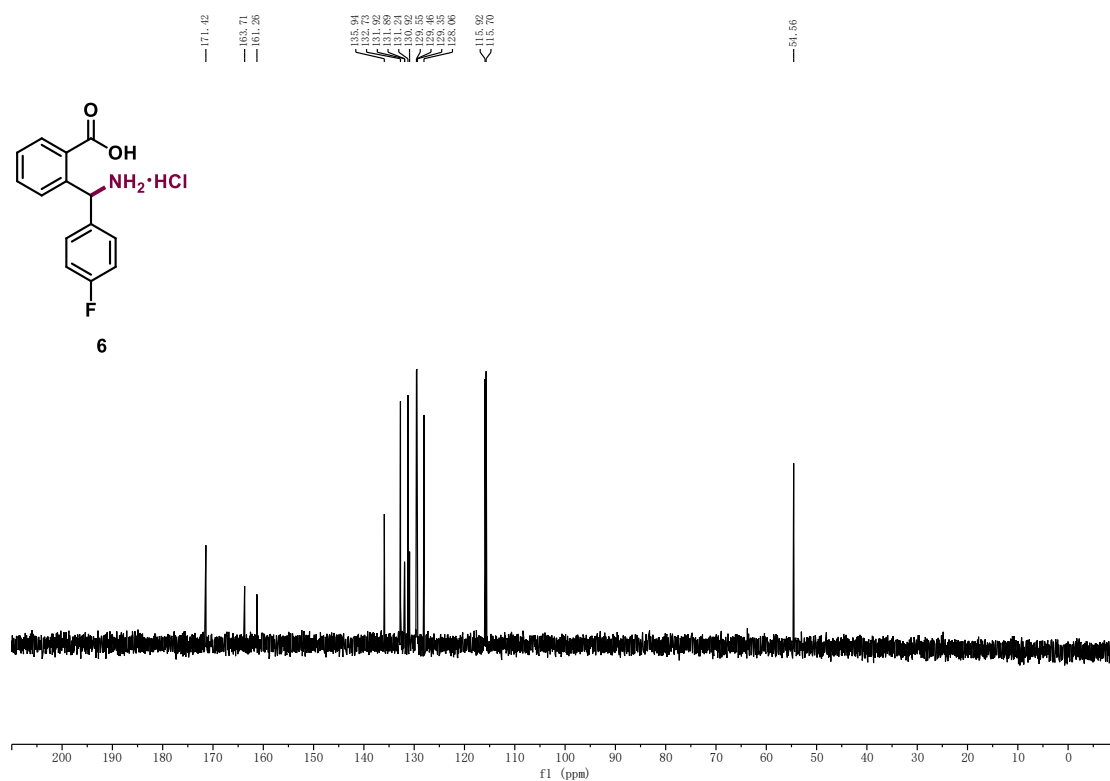
Supplementary Figure 165 ¹H NMR spectrum of compound 5



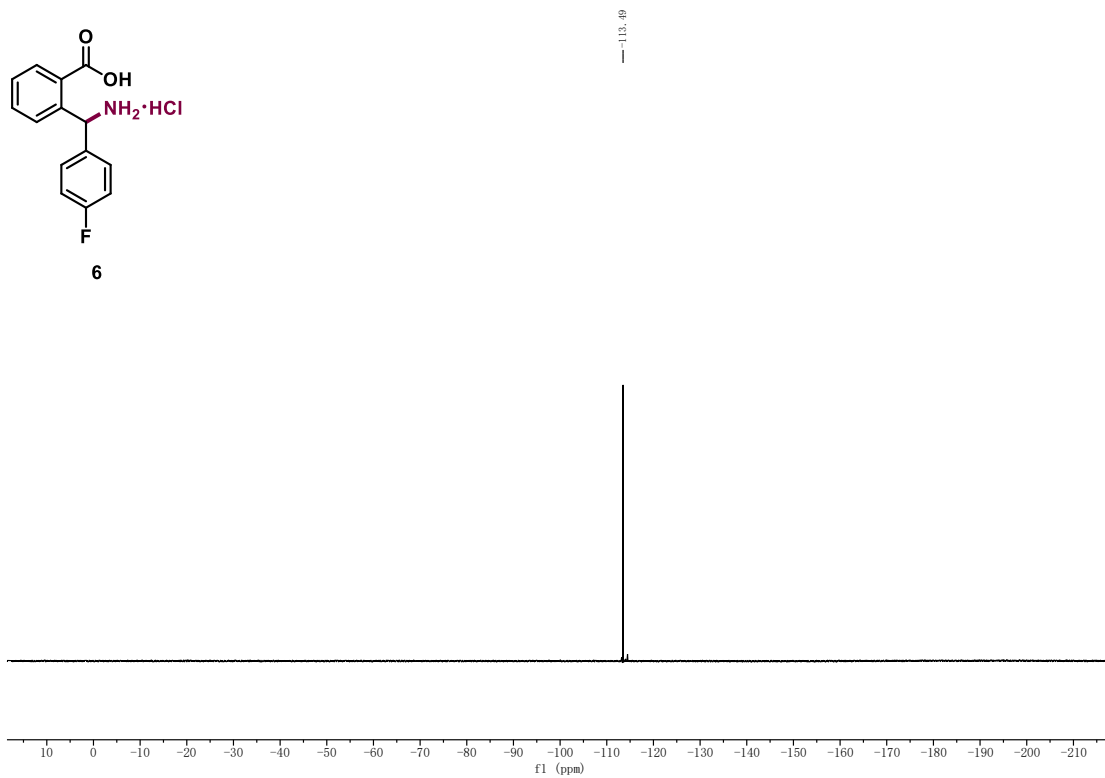
Supplementary Figure 166 ¹³C NMR spectrum of compound 5



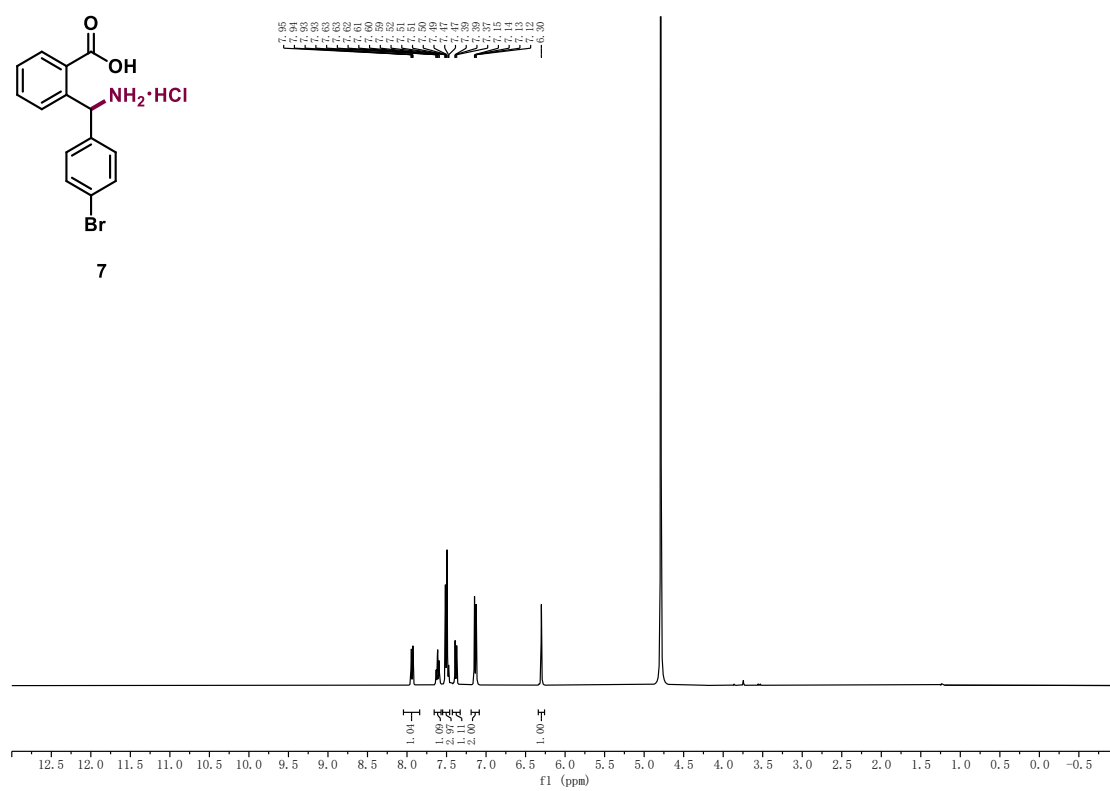
Supplementary Figure 167 ¹H NMR spectrum of compound 6



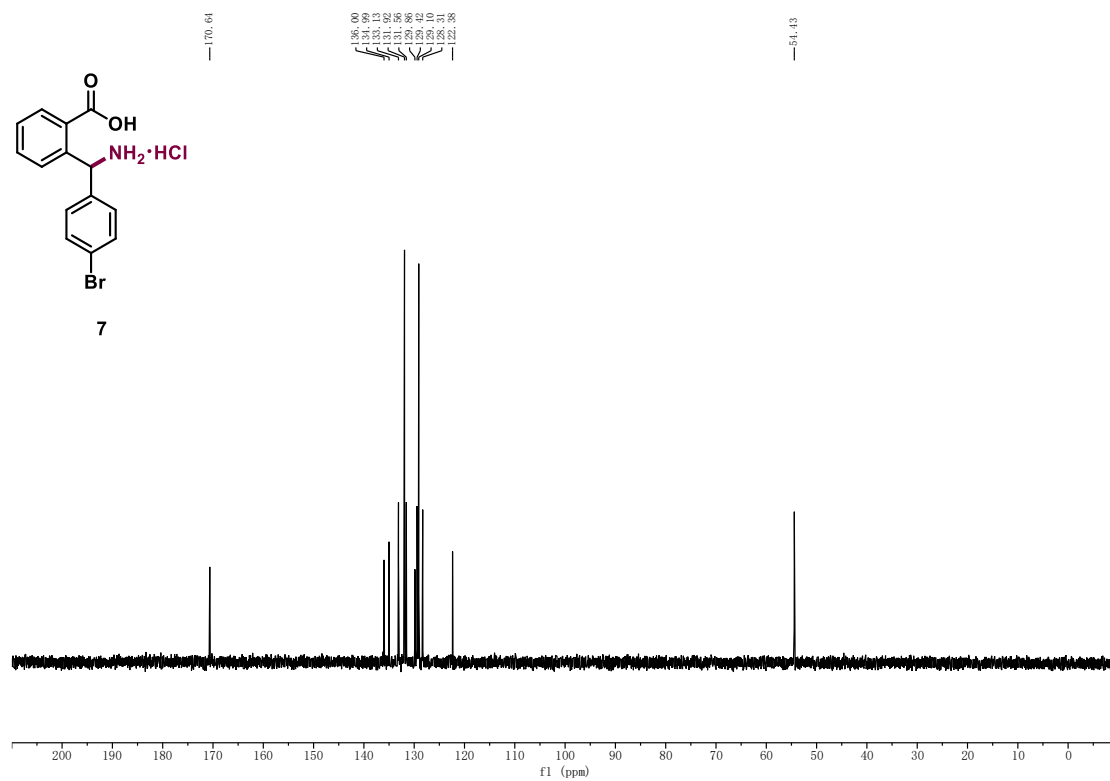
Supplementary Figure 168 ¹³C NMR spectrum of compound 6



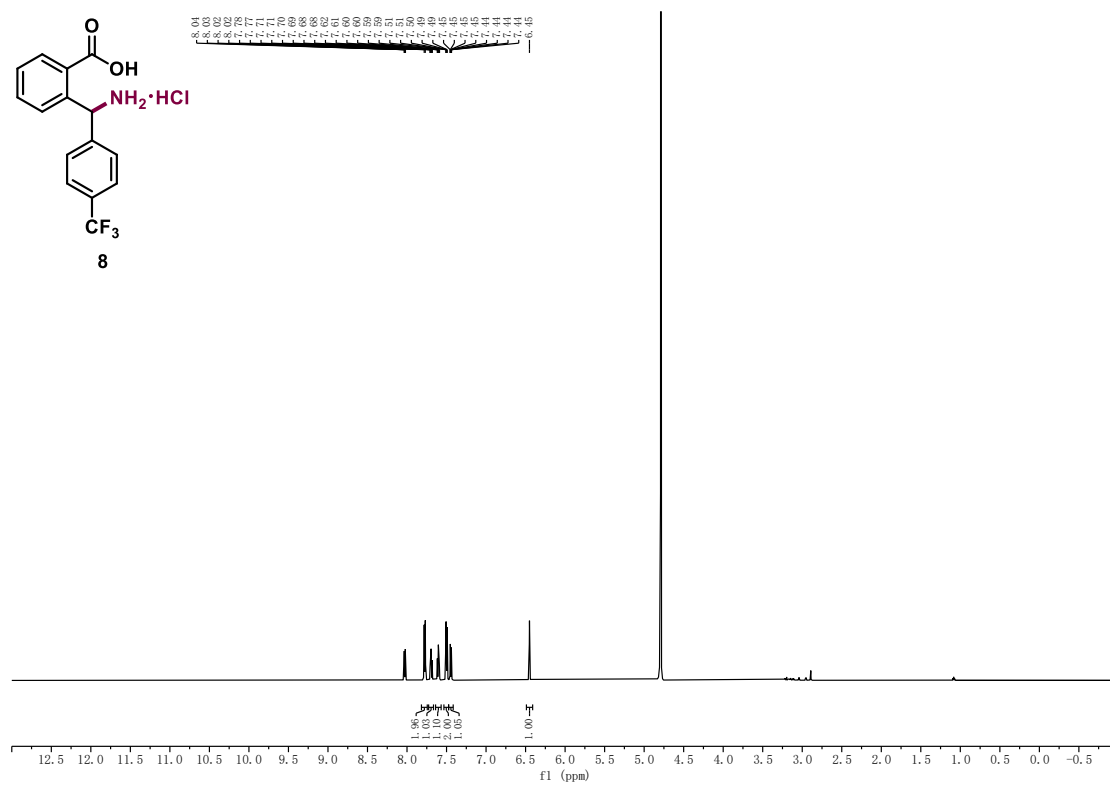
Supplementary Figure 169 ¹⁹F NMR spectrum of compound 6



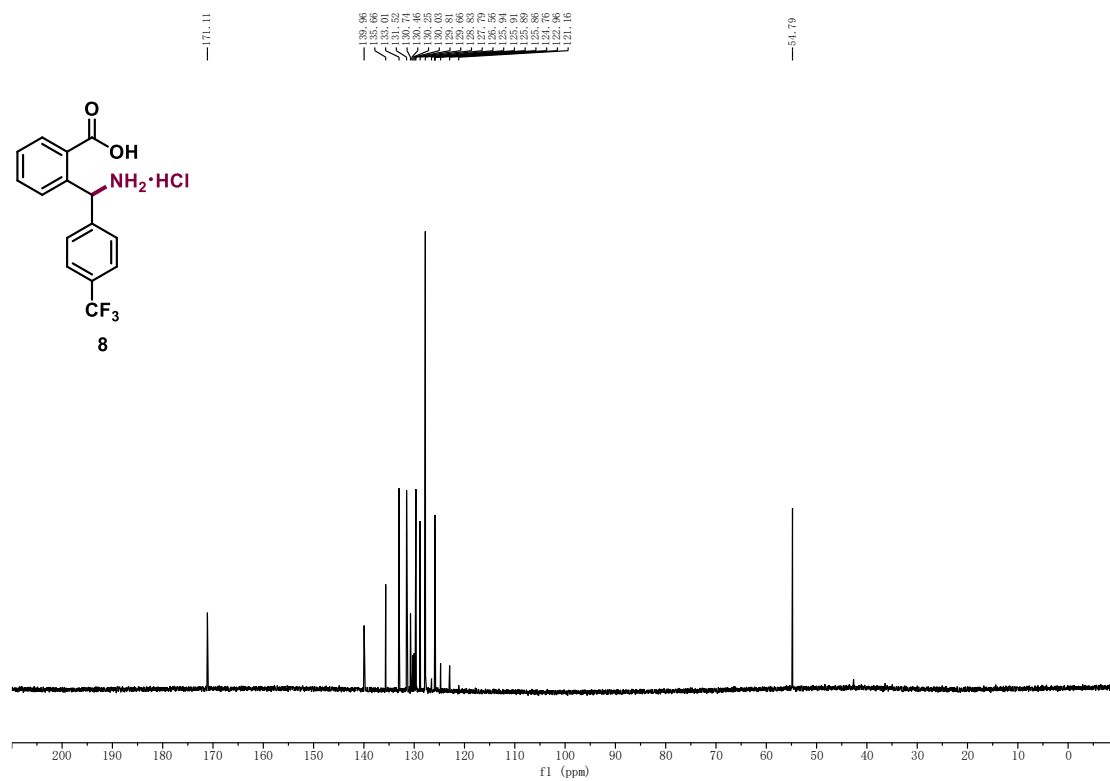
Supplementary Figure 170 ¹H NMR spectrum of compound 7



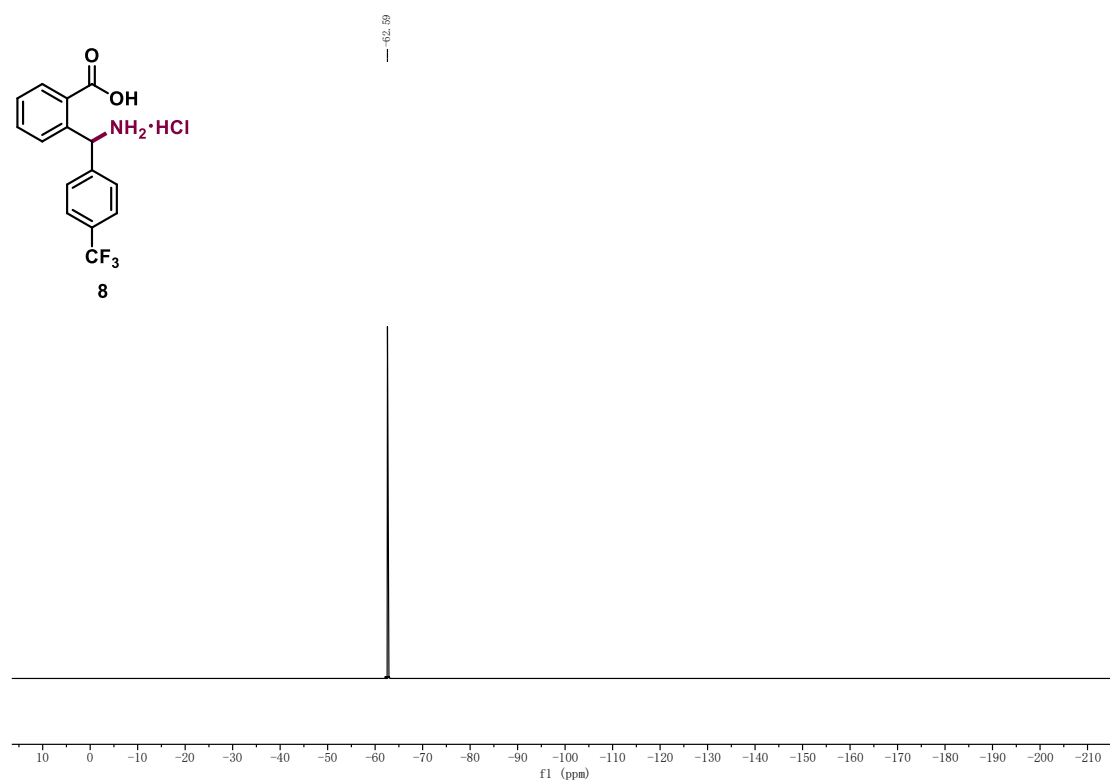
Supplementary Figure 171 ^{13}C NMR spectrum of compound 7



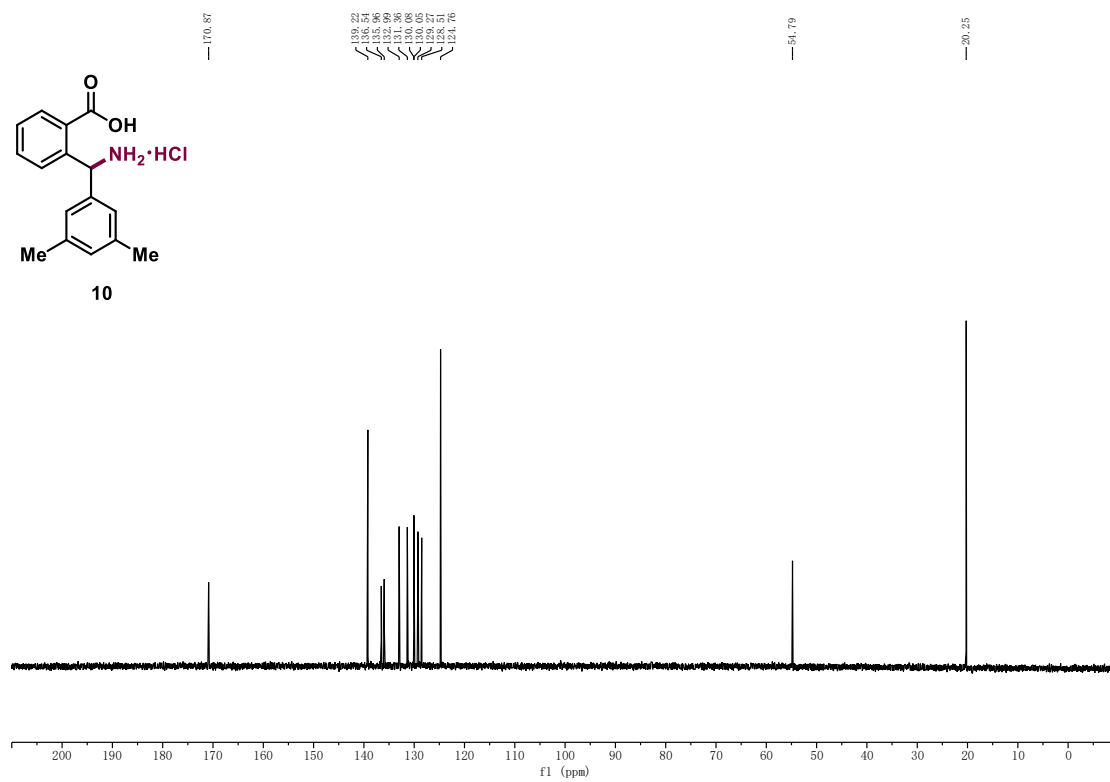
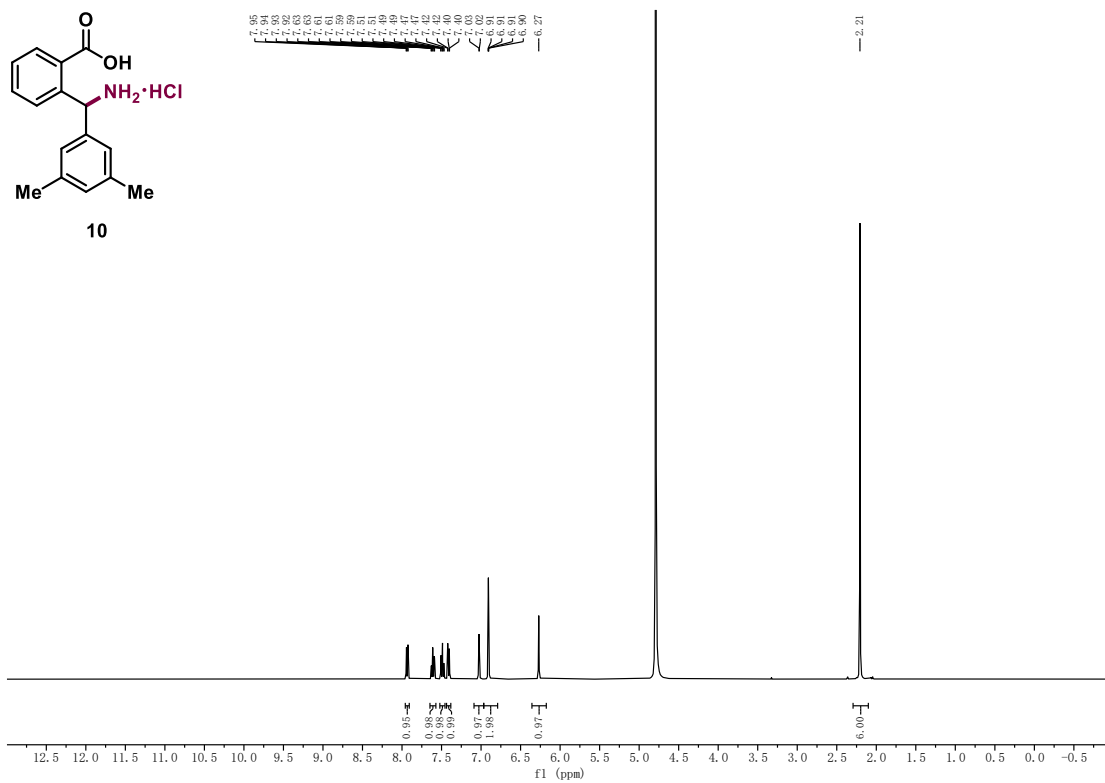
Supplementary Figure 172 ^1H NMR spectrum of compound 8

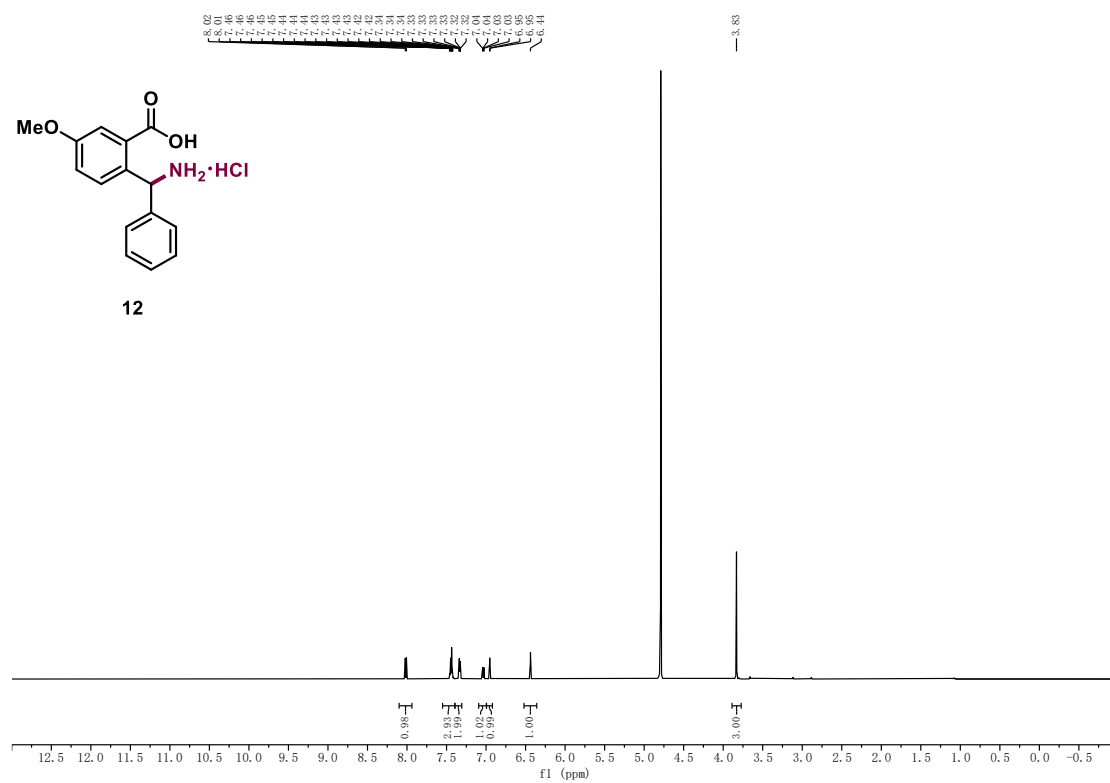


Supplementary Figure 173 ^{13}C NMR spectrum of compound 8

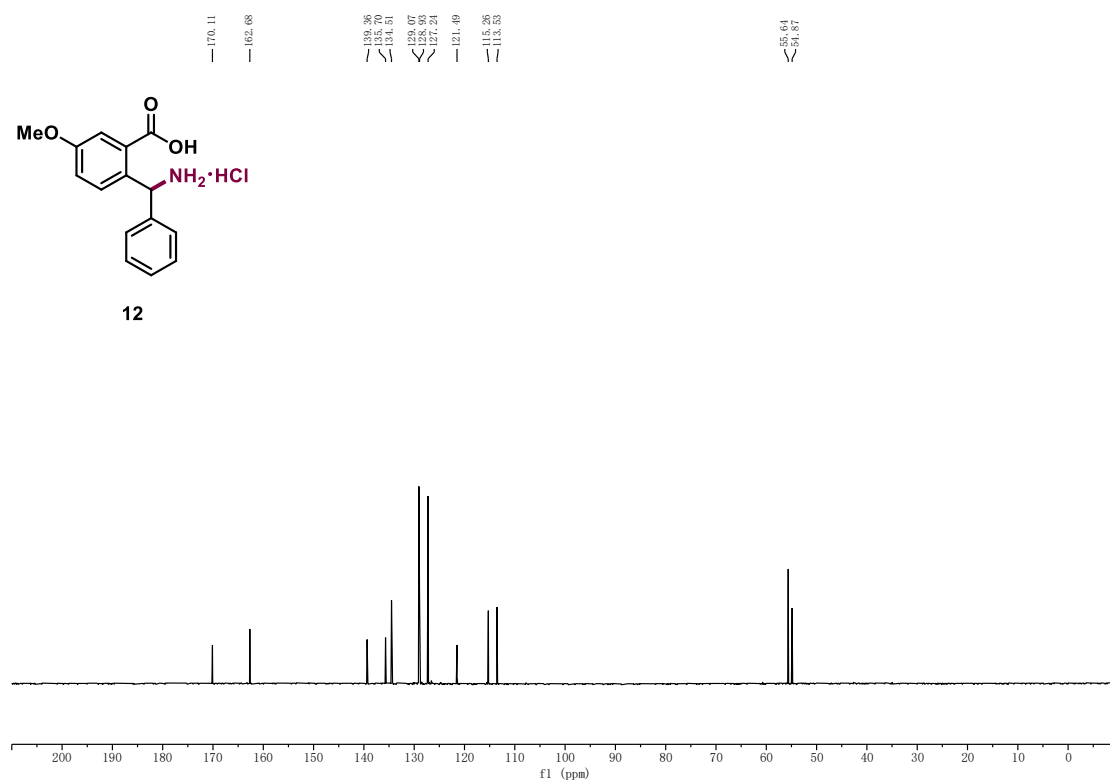


Supplementary Figure 174 ^{19}F NMR spectrum of compound 8

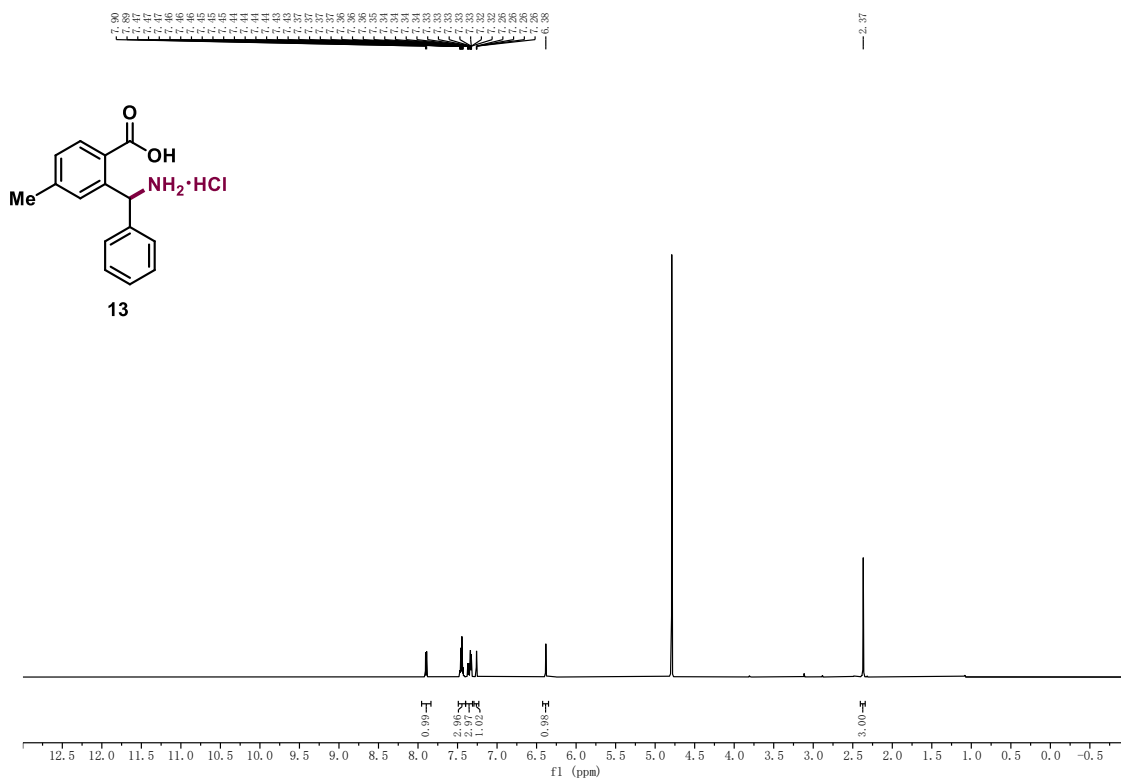




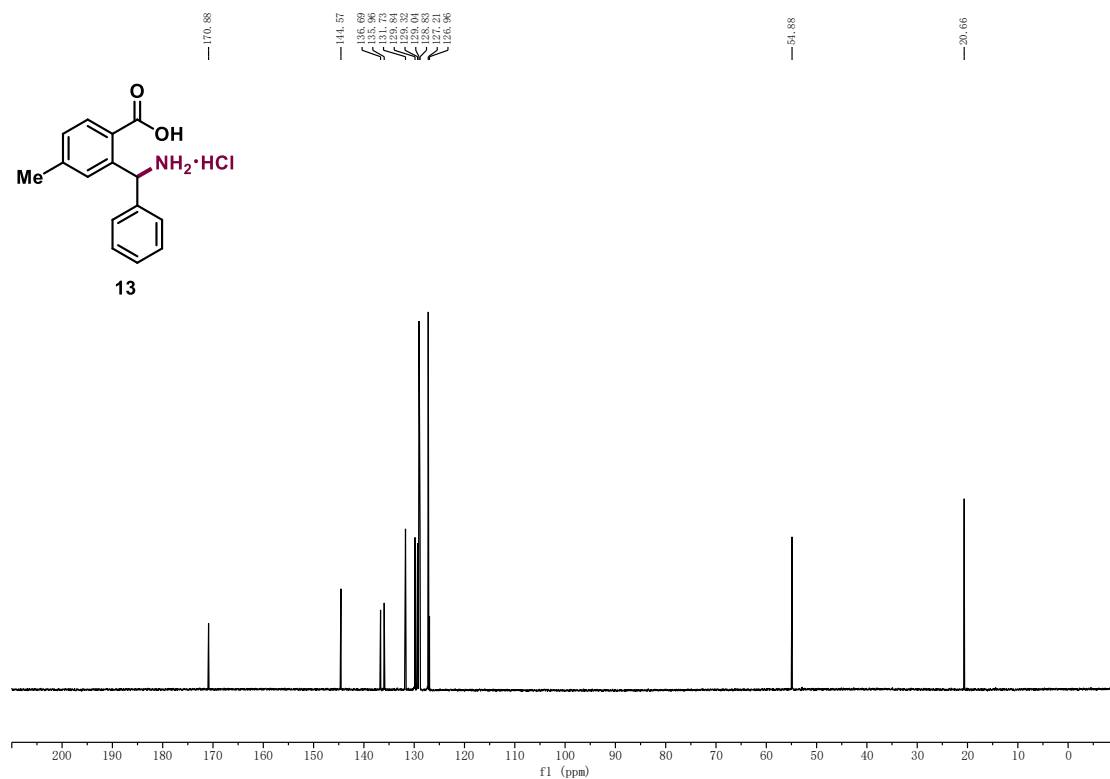
Supplementary Figure 181 ¹H NMR spectrum of compound 12



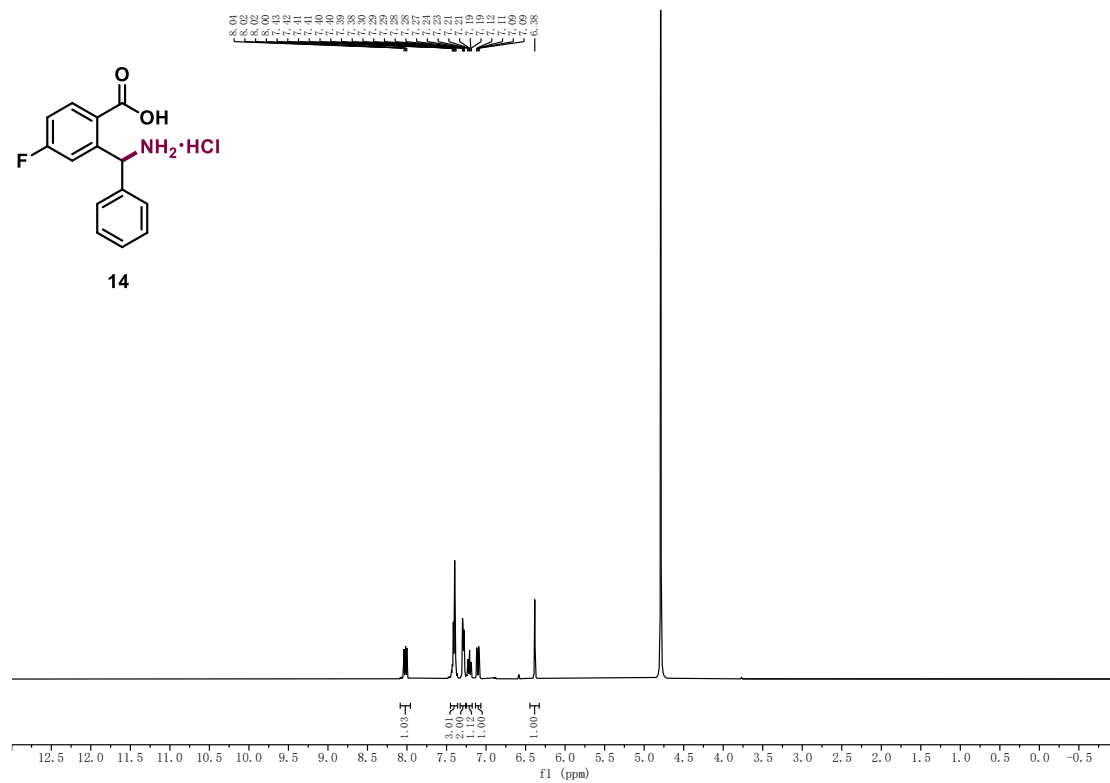
Supplementary Figure 182 ¹³C NMR spectrum of compound 12



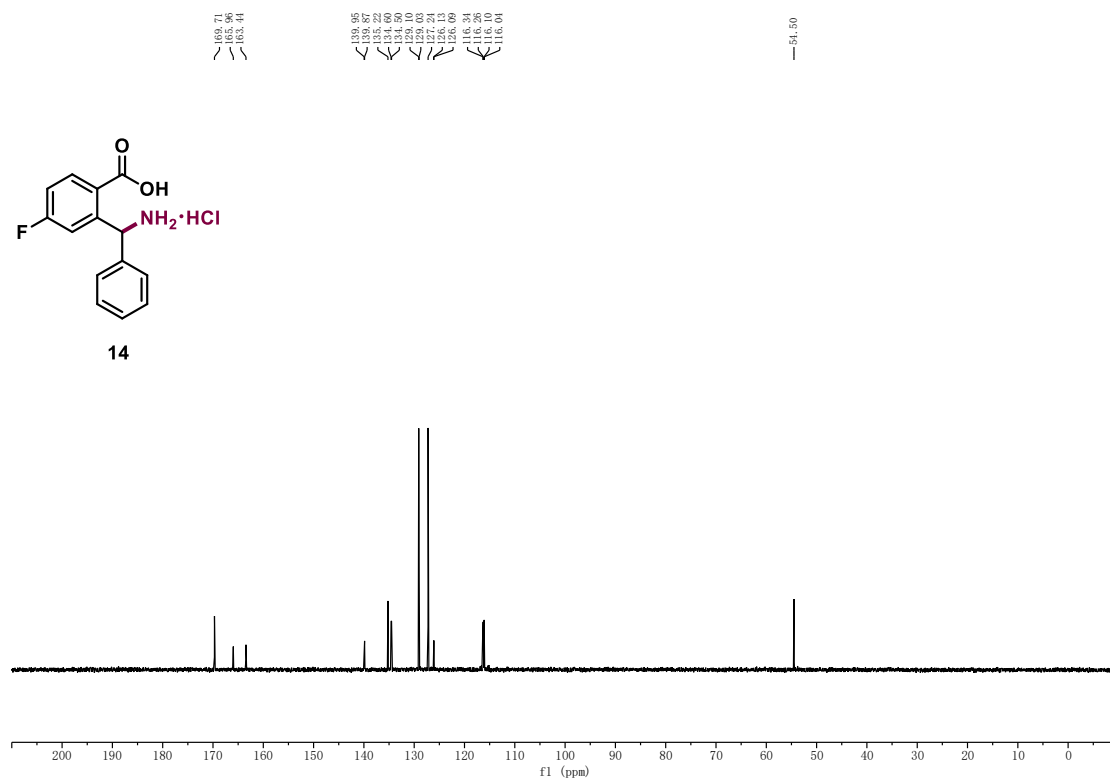
Supplementary Figure 183 ^1H NMR spectrum of compound 13



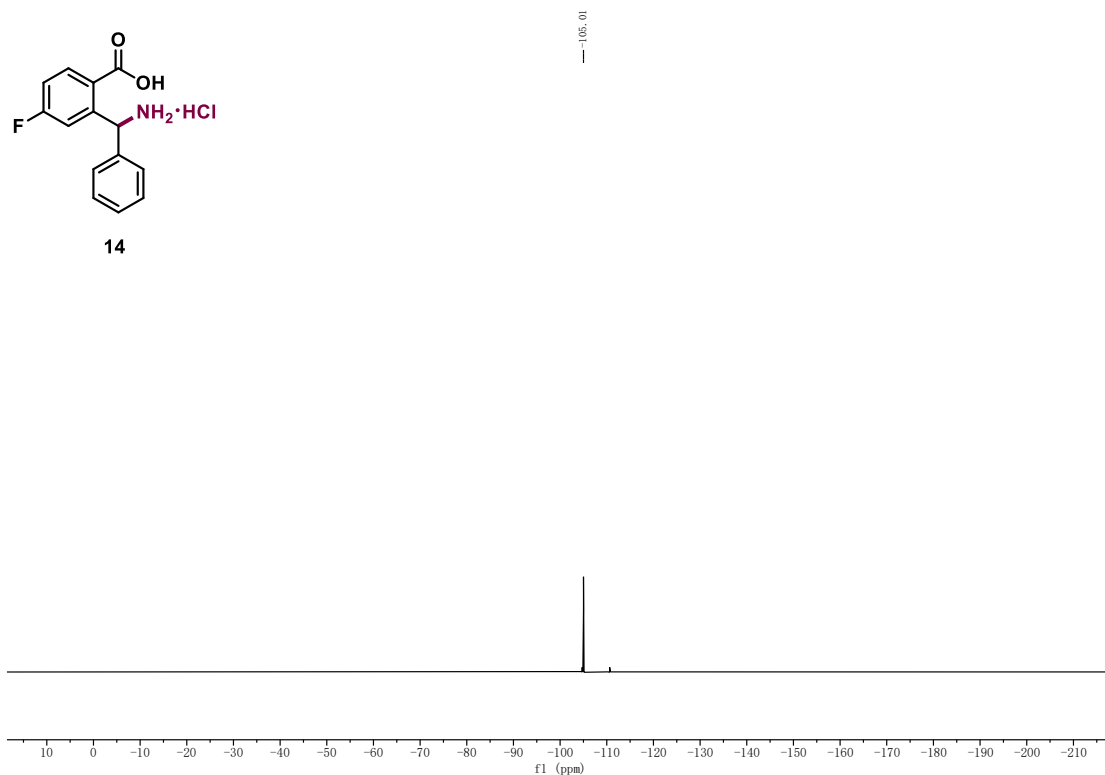
Supplementary Figure 184 ^{13}C NMR spectrum of compound 13



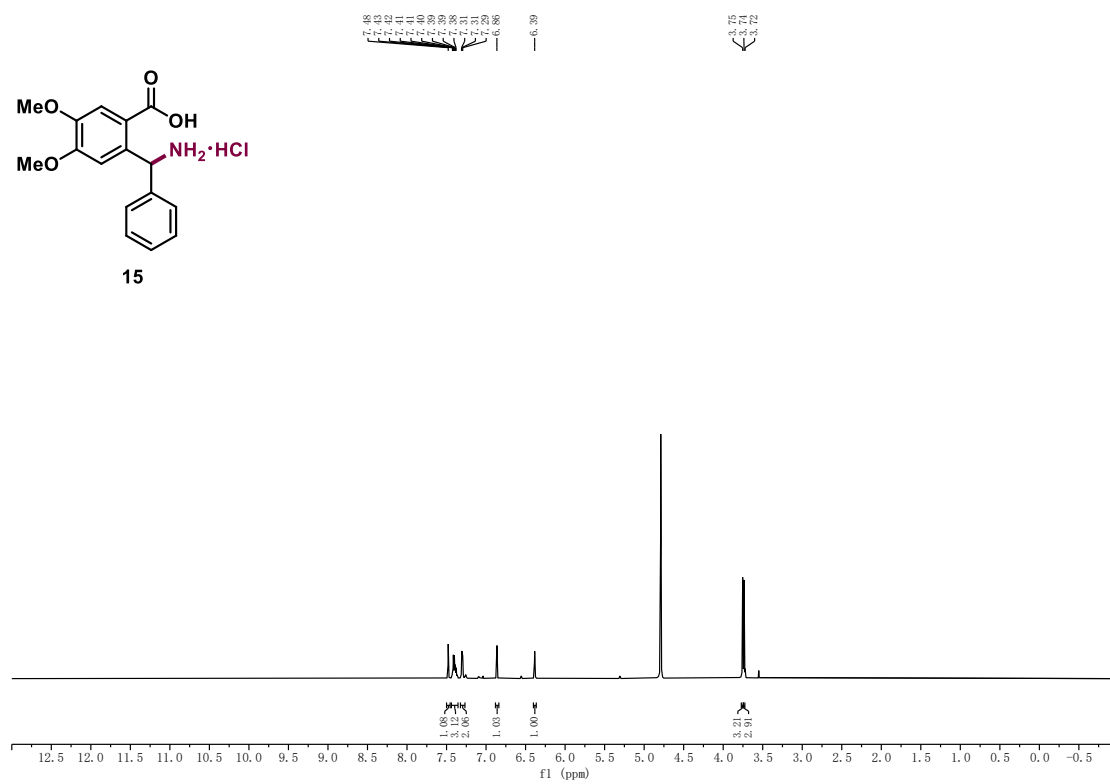
Supplementary Figure 185 ¹H NMR spectrum of compound 14



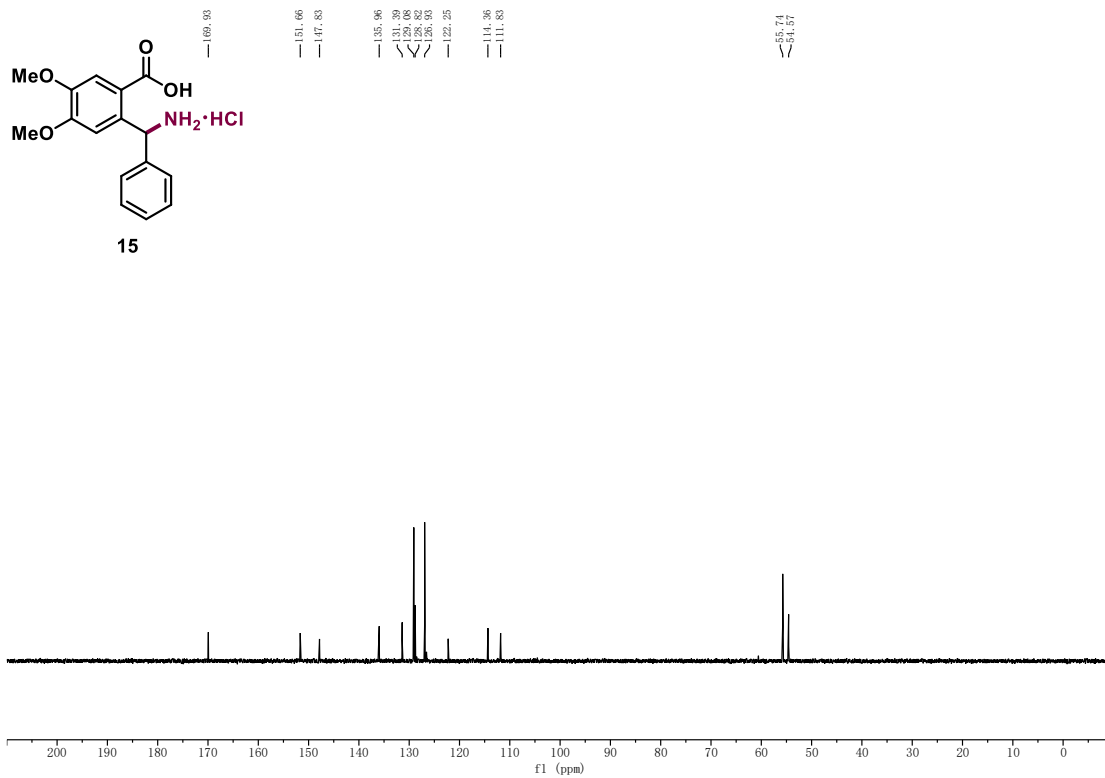
Supplementary Figure 186 ¹³C NMR spectrum of compound 14



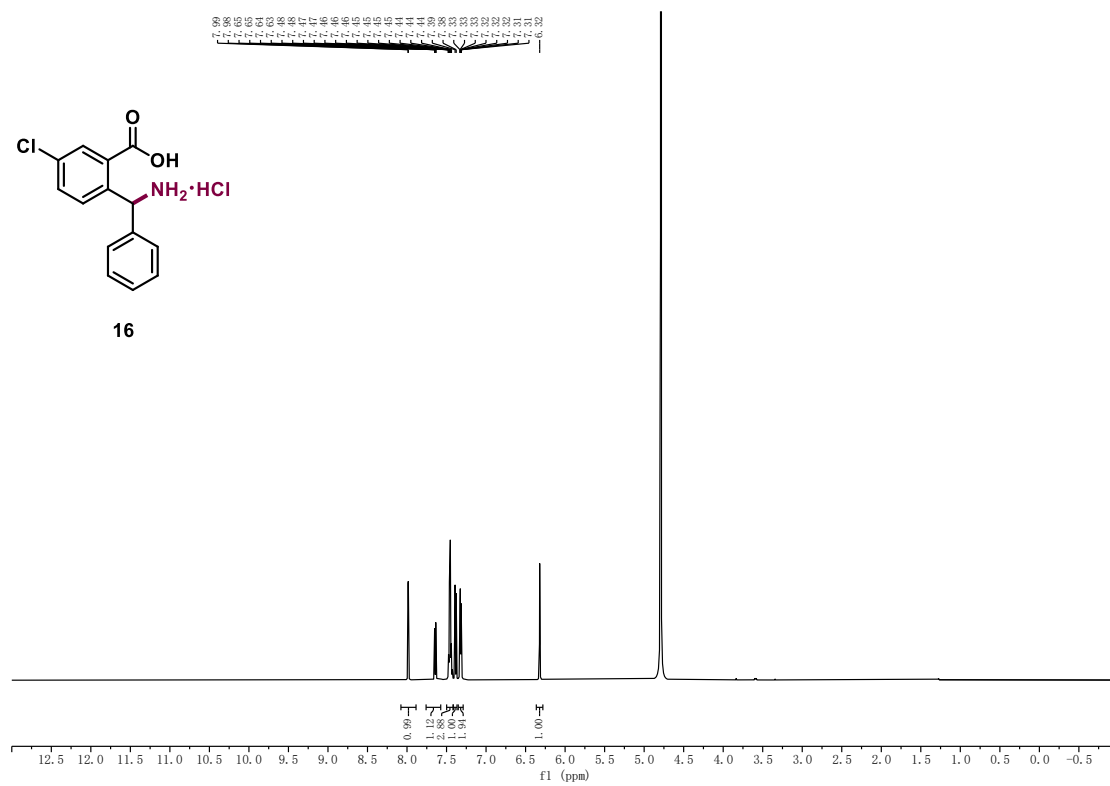
Supplementary Figure 187 ^{19}F NMR spectrum of compound 14



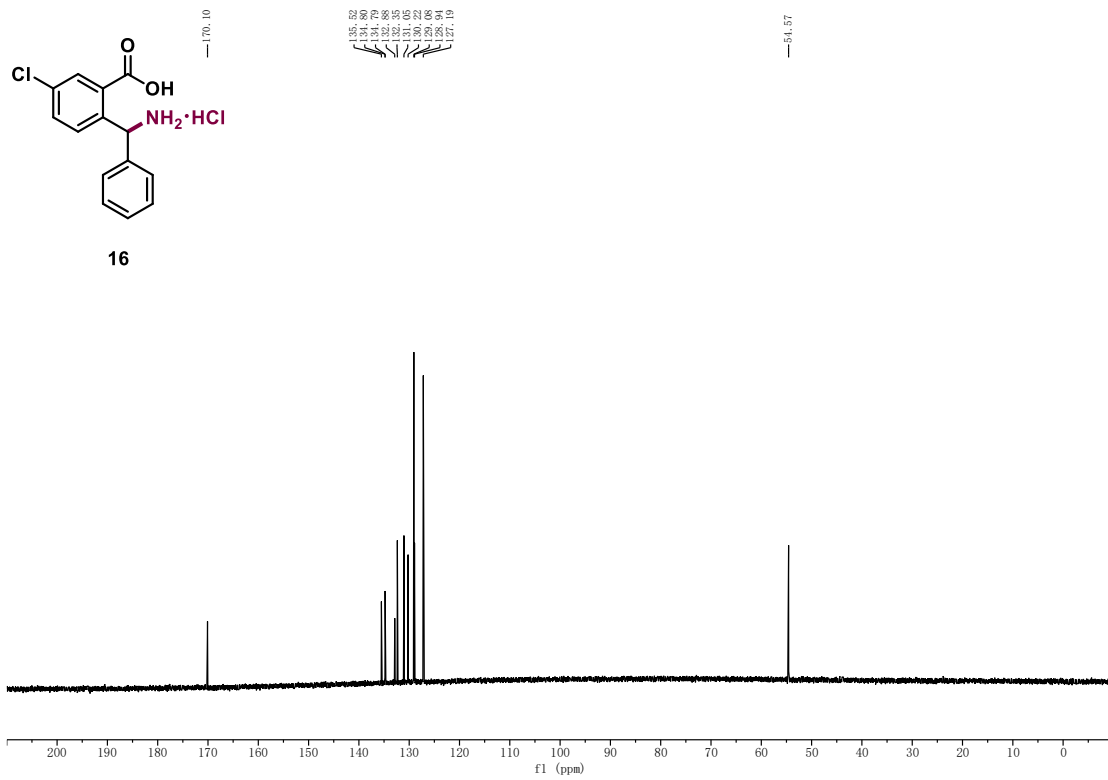
Supplementary Figure 188 ^1H NMR spectrum of compound 15



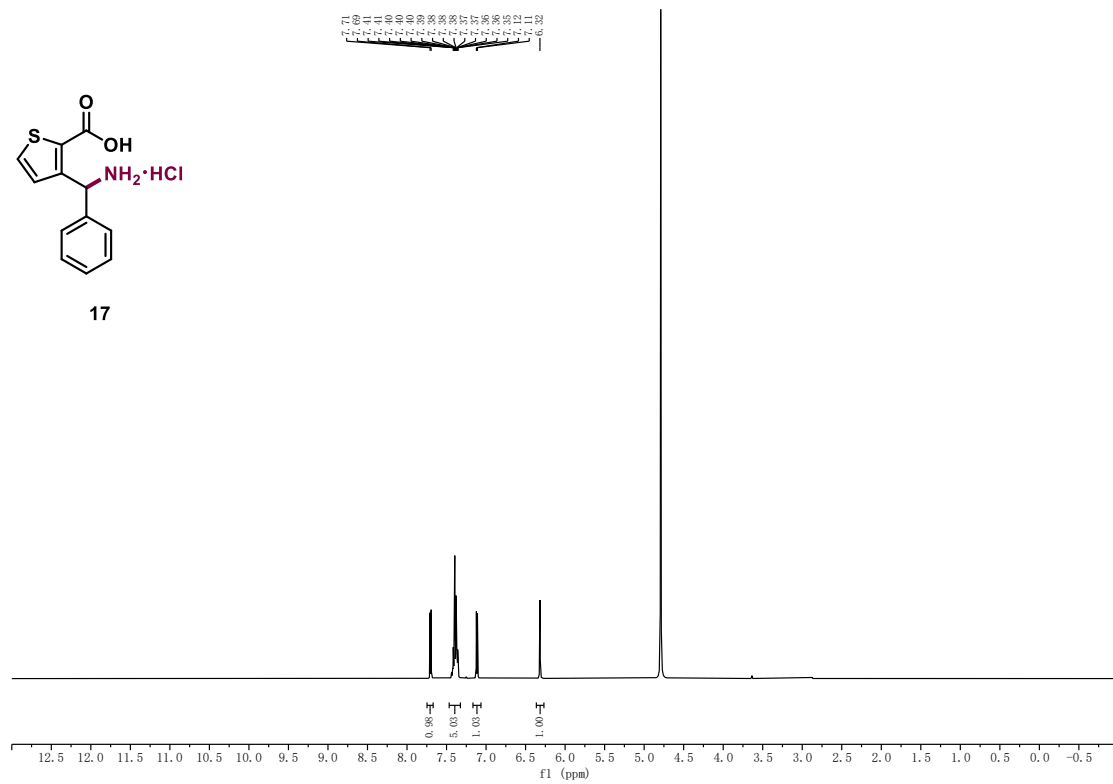
Supplementary Figure 189 ¹³C NMR spectrum of compound 15



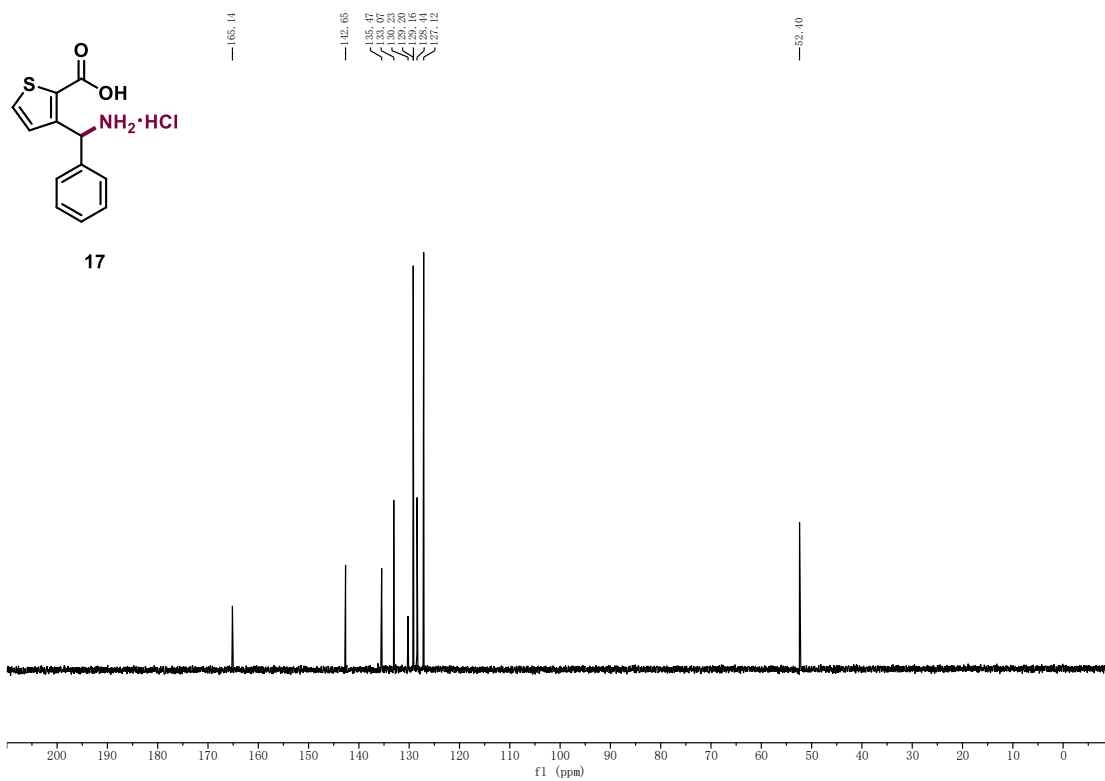
Supplementary Figure 190 ¹H NMR spectrum of compound 16



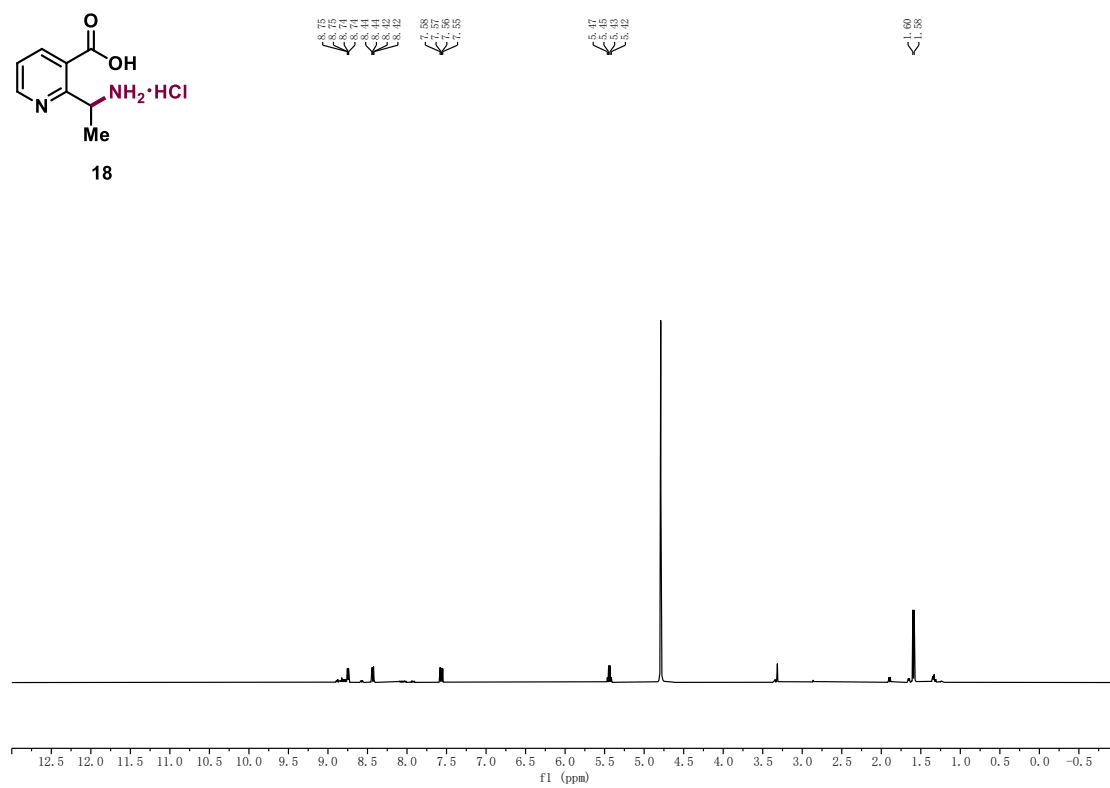
Supplementary Figure 191 ¹³C NMR spectrum of compound 16



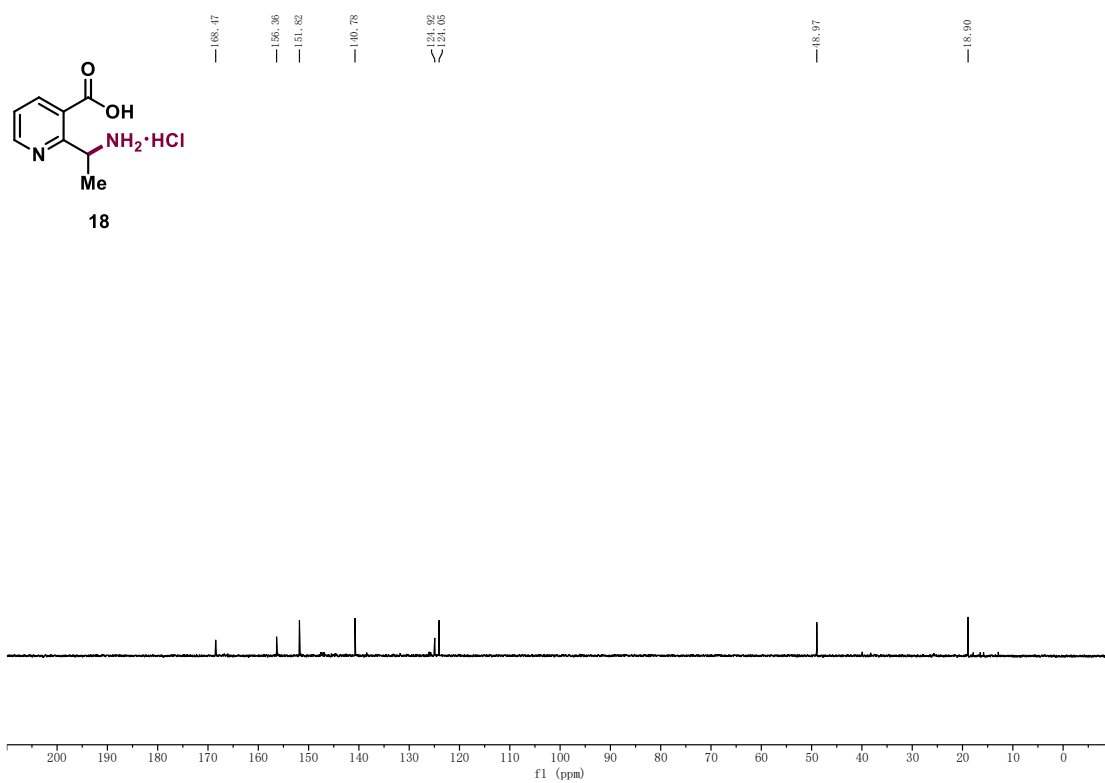
Supplementary Figure 192 ¹H NMR spectrum of compound 17



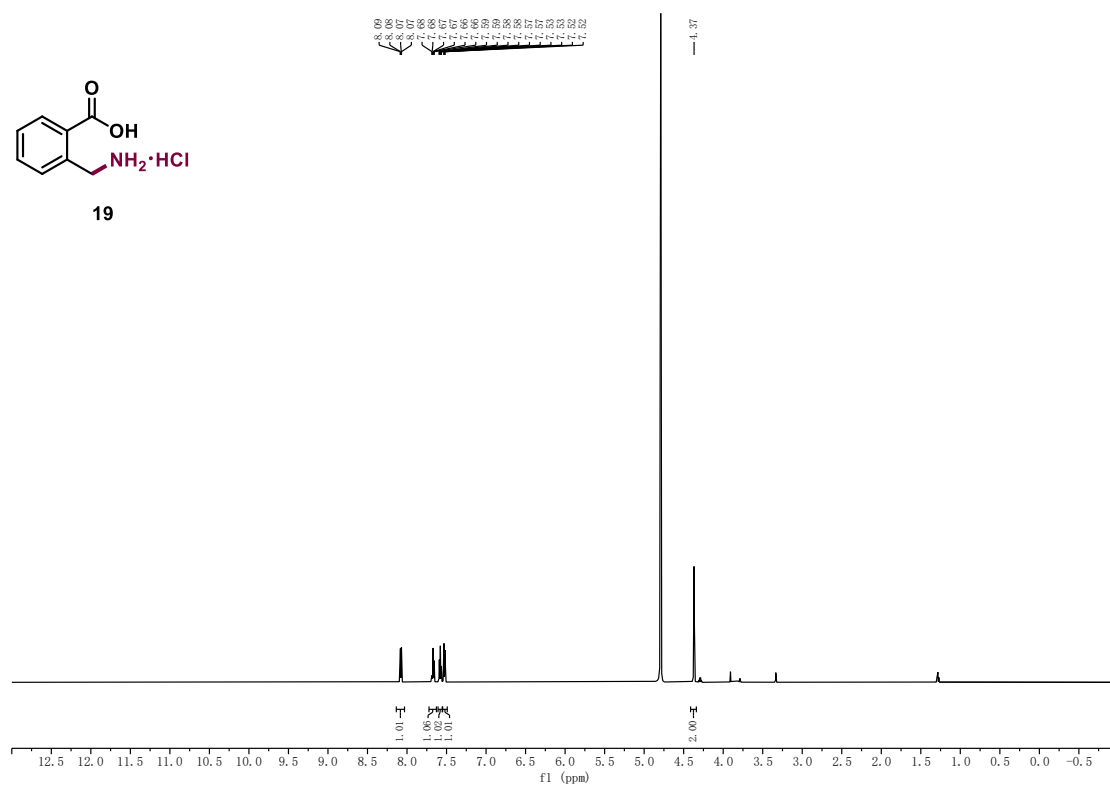
Supplementary Figure 193 ^{13}C NMR spectrum of compound 17



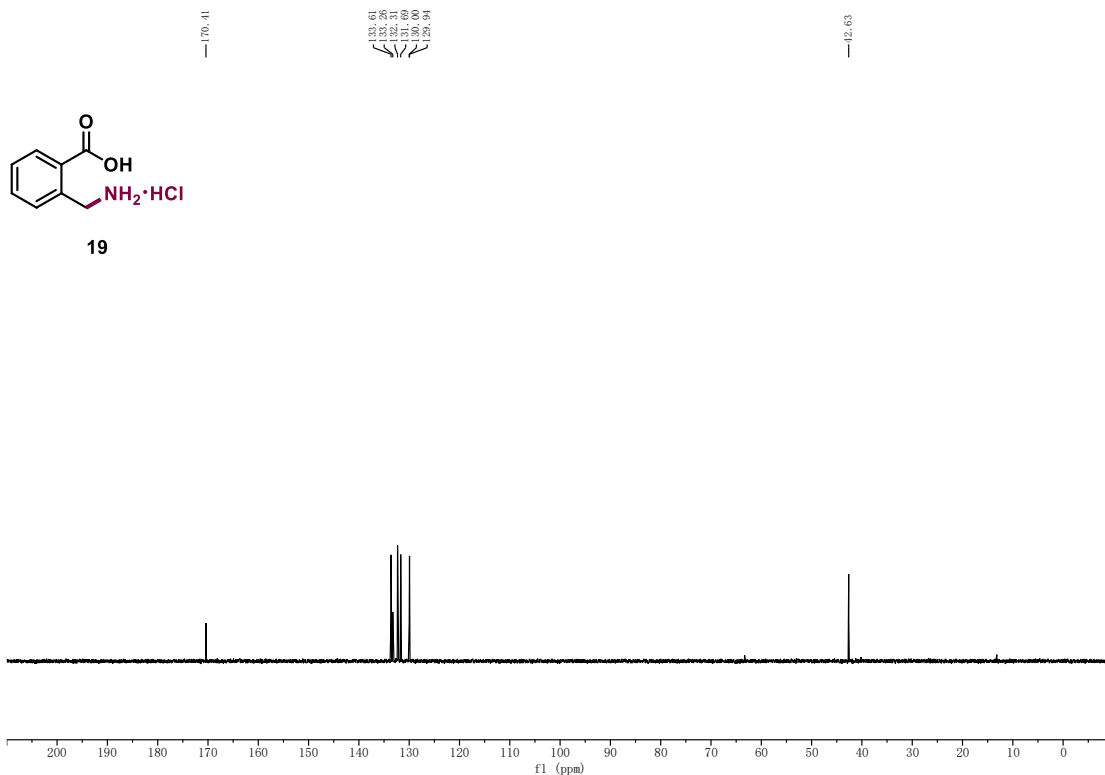
Supplementary Figure 194 ^1H NMR spectrum of compound 18



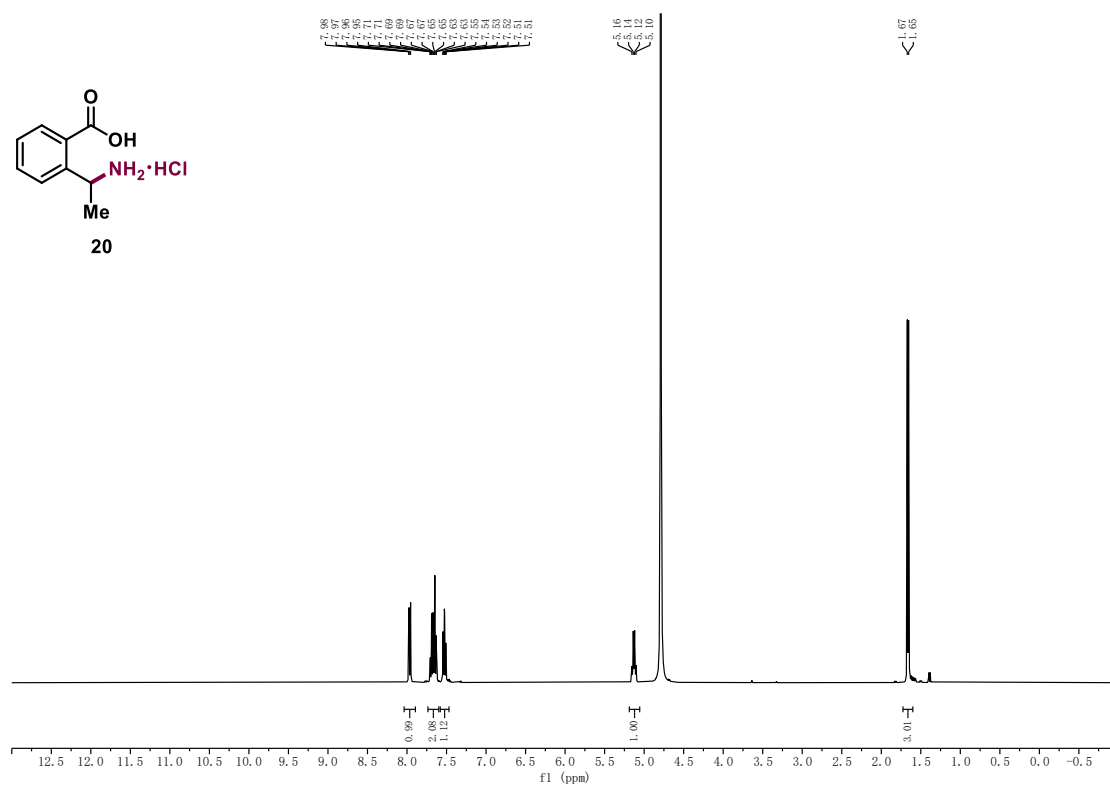
Supplementary Figure 195 ¹³C NMR spectrum of compound 18



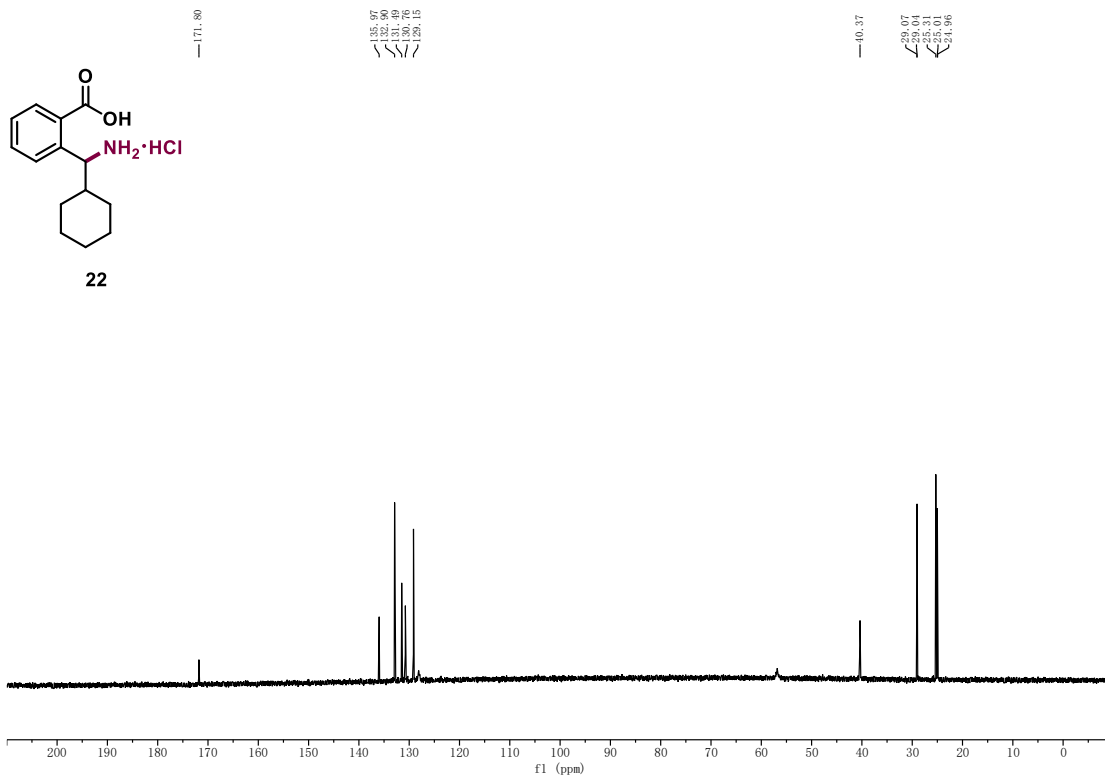
Supplementary Figure 196 ¹H NMR spectrum of compound 19



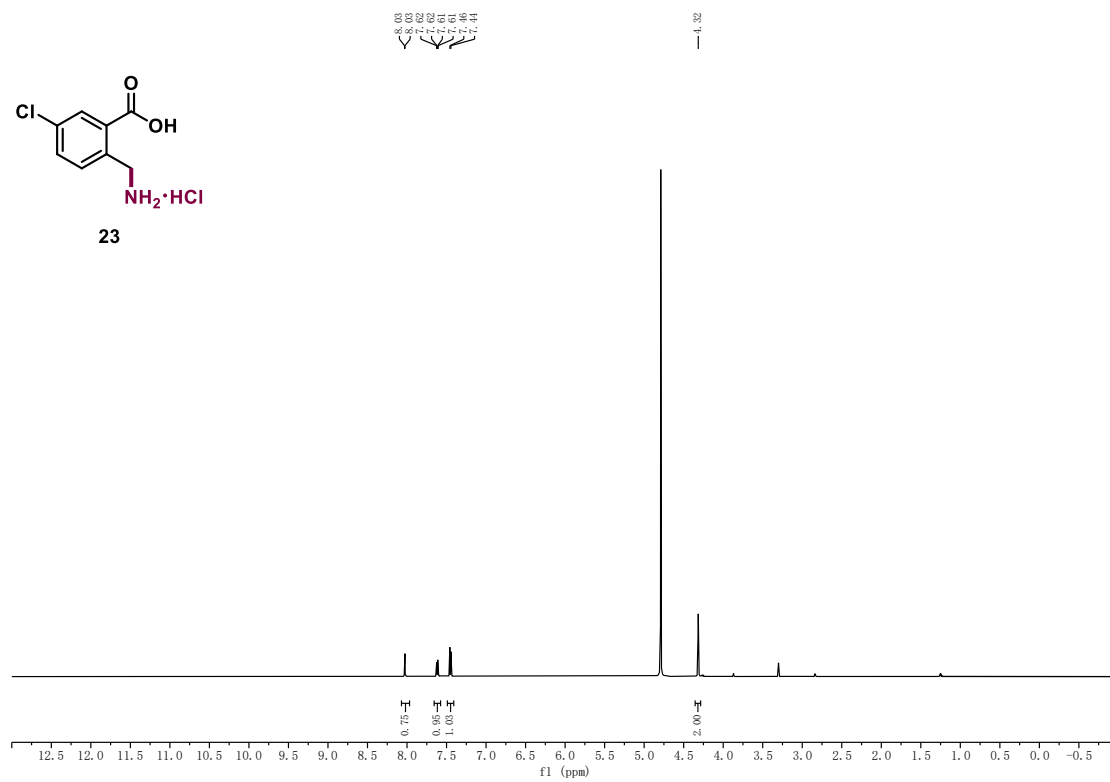
Supplementary Figure 197 ^{13}C NMR spectrum of compound 19



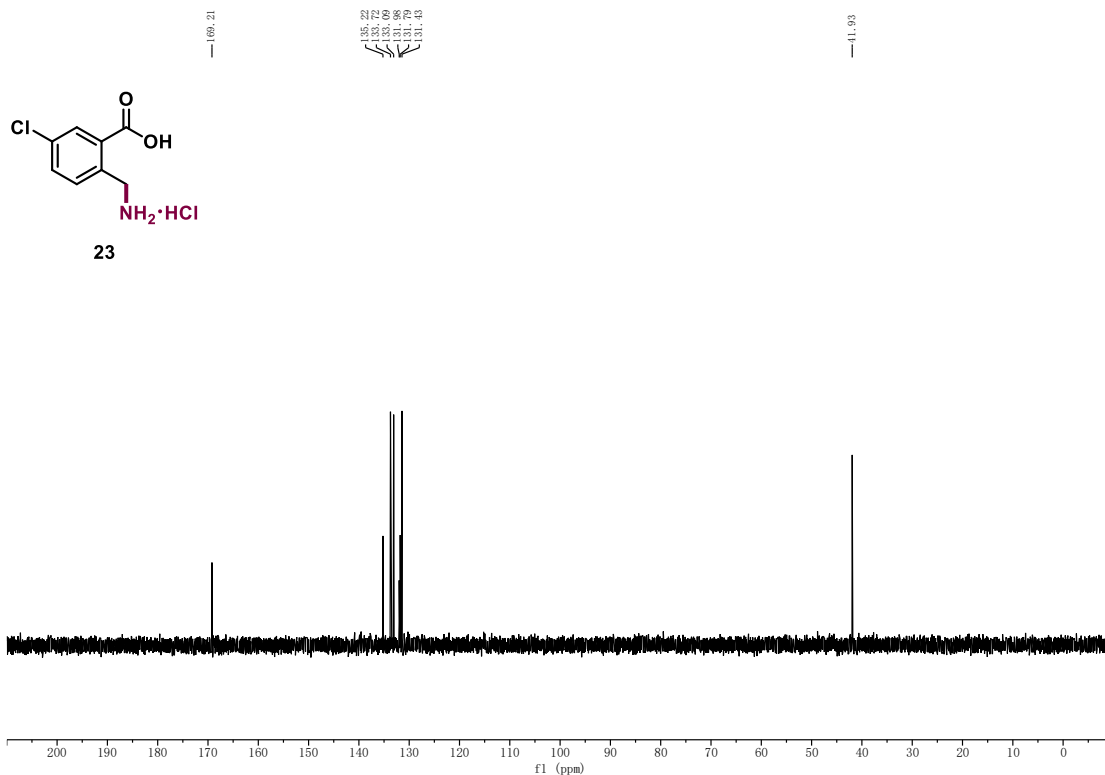
Supplementary Figure 198 ^1H NMR spectrum of compound 20



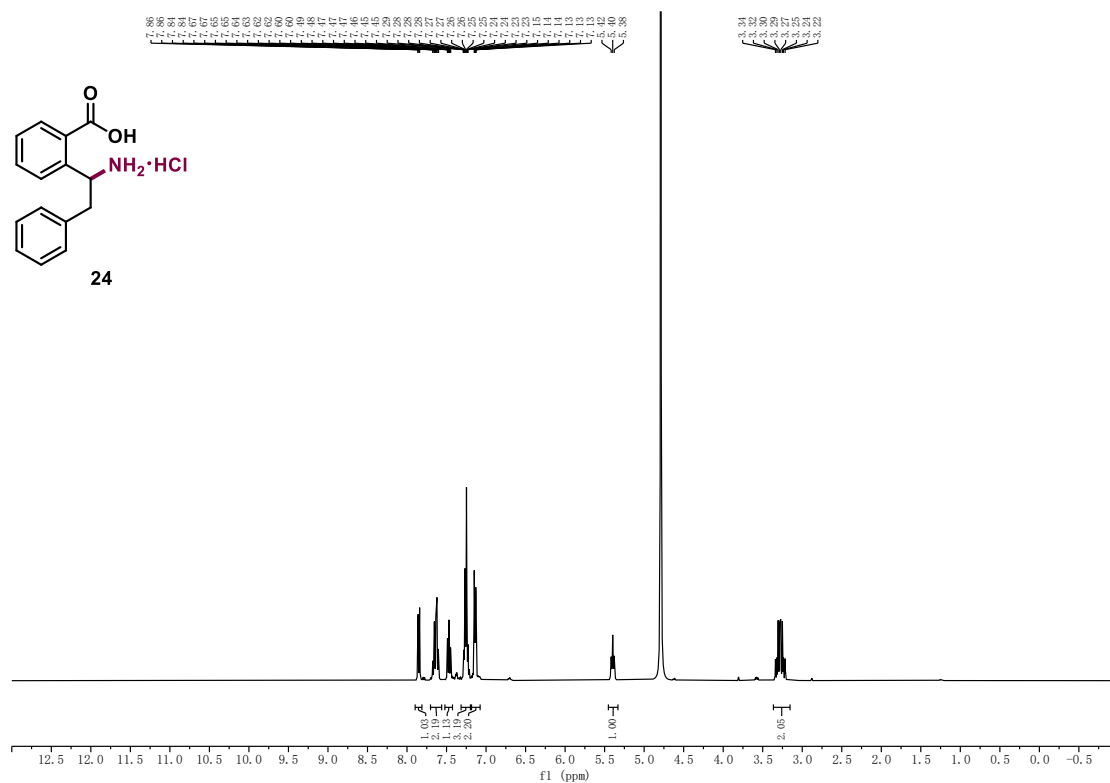
Supplementary Figure 203 ¹³C NMR spectrum of compound 22



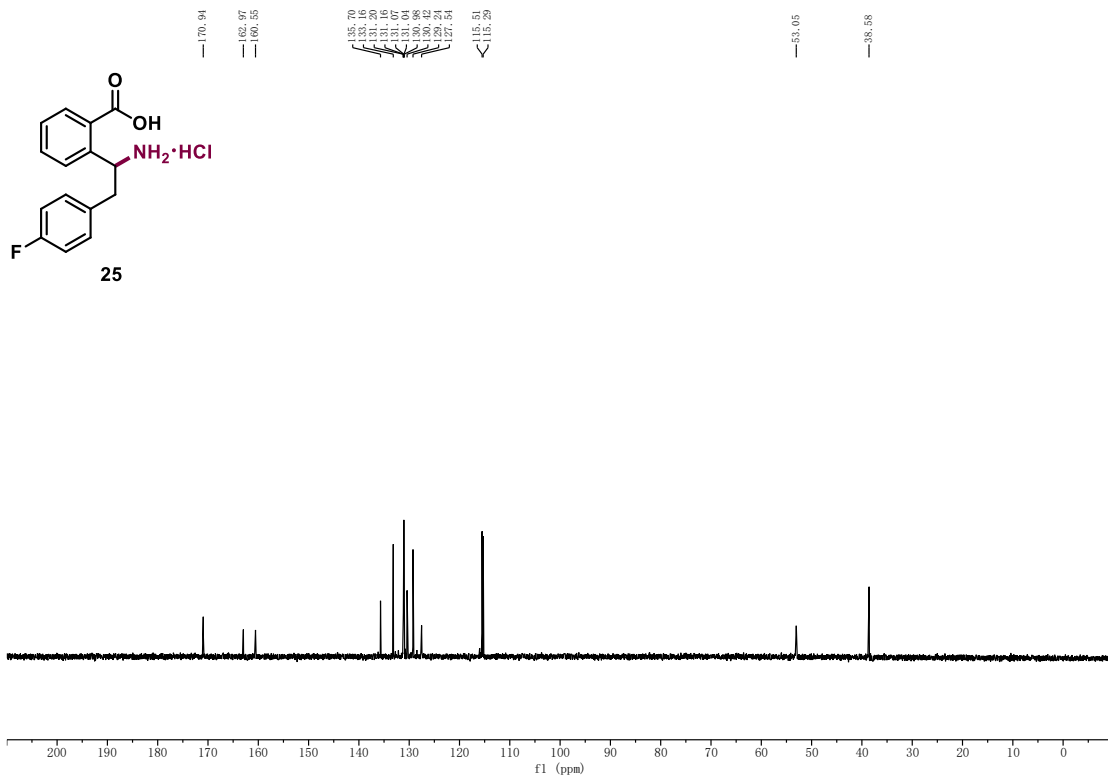
Supplementary Figure 204 ¹H NMR spectrum of compound 23



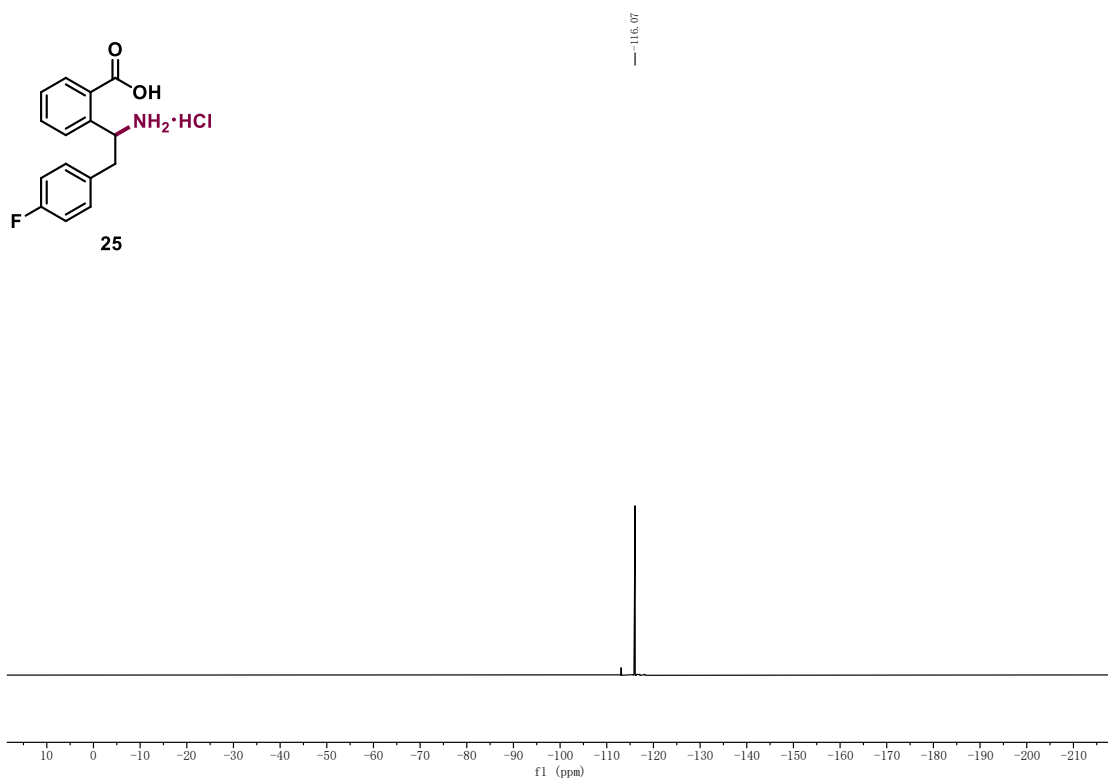
Supplementary Figure 205 ¹³C NMR spectrum of compound 23



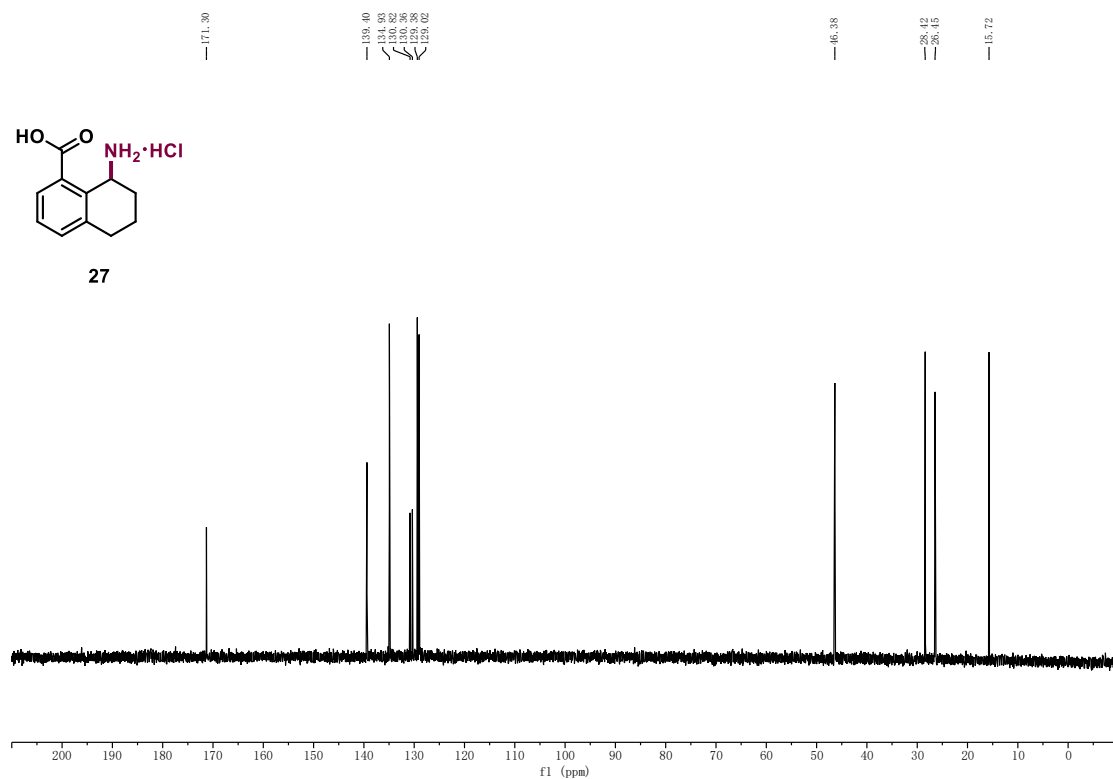
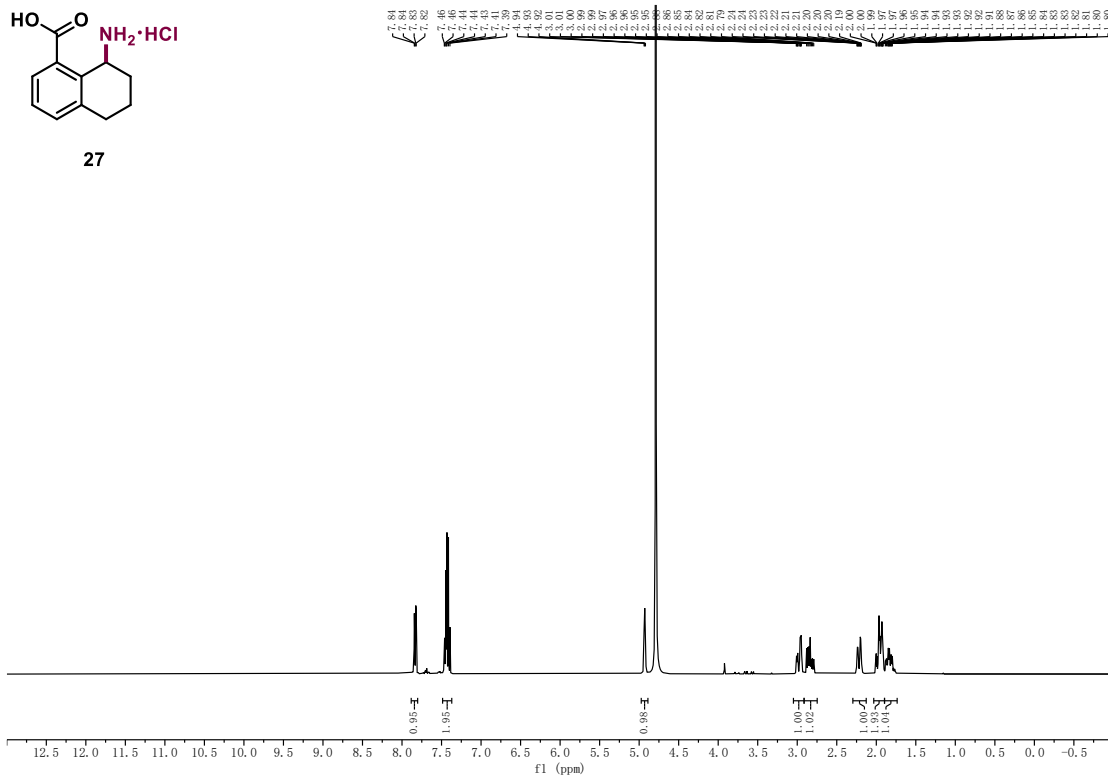
Supplementary Figure 206 ¹H NMR spectrum of compound 24

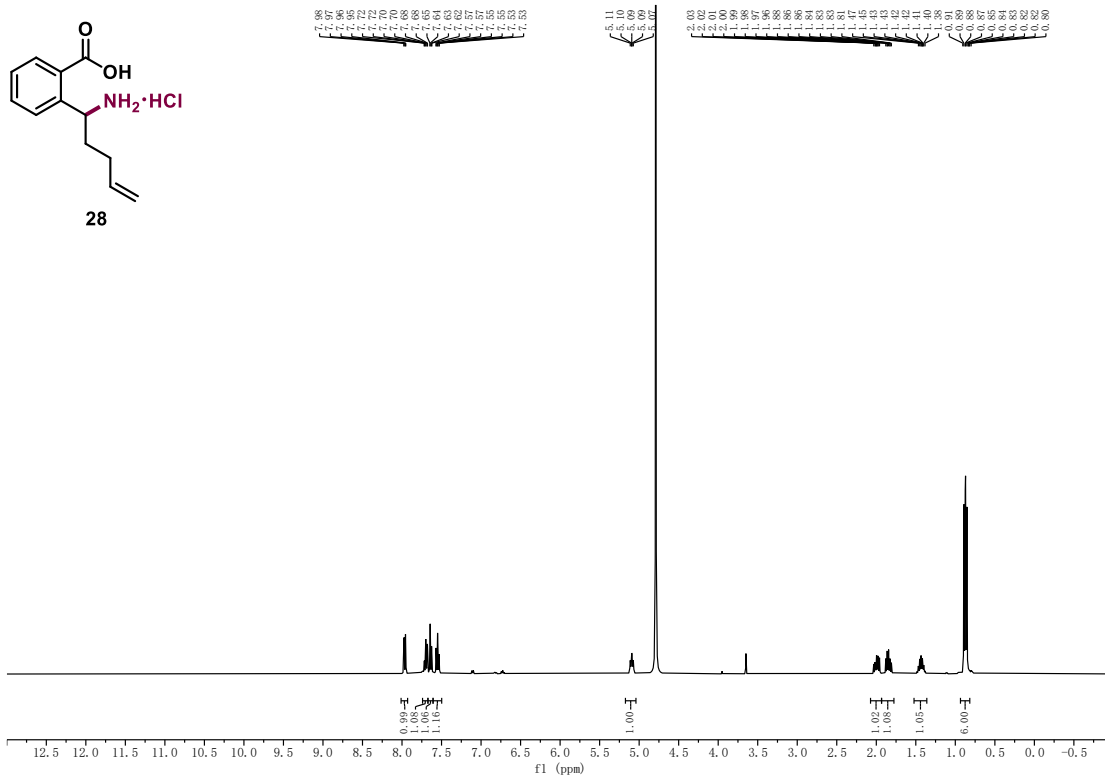


Supplementary Figure 209 ^{13}C NMR spectrum of compound 25

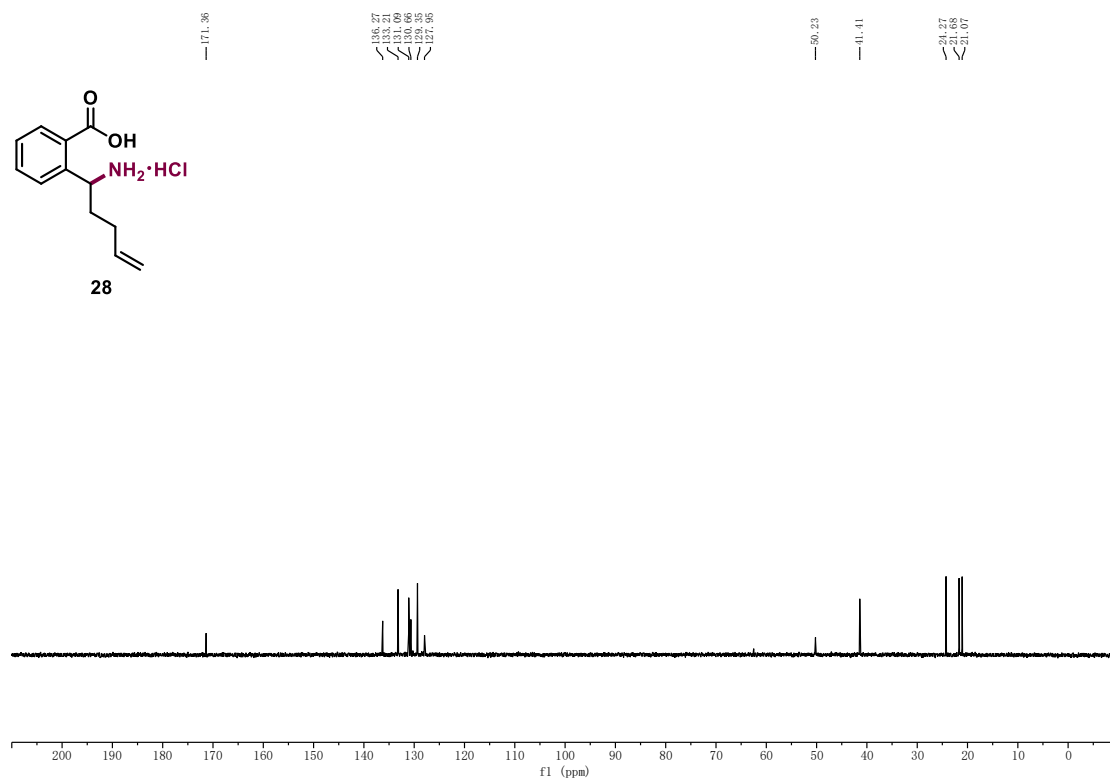


Supplementary Figure 210 ^{19}F NMR spectrum of compound 25

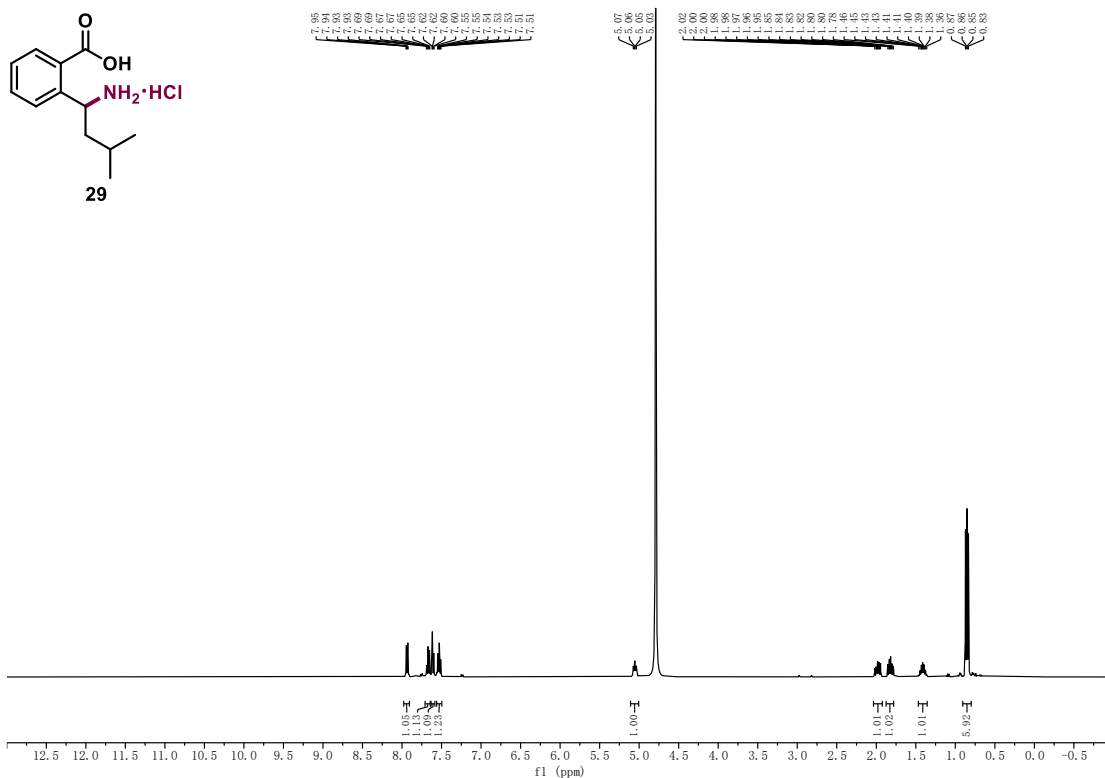




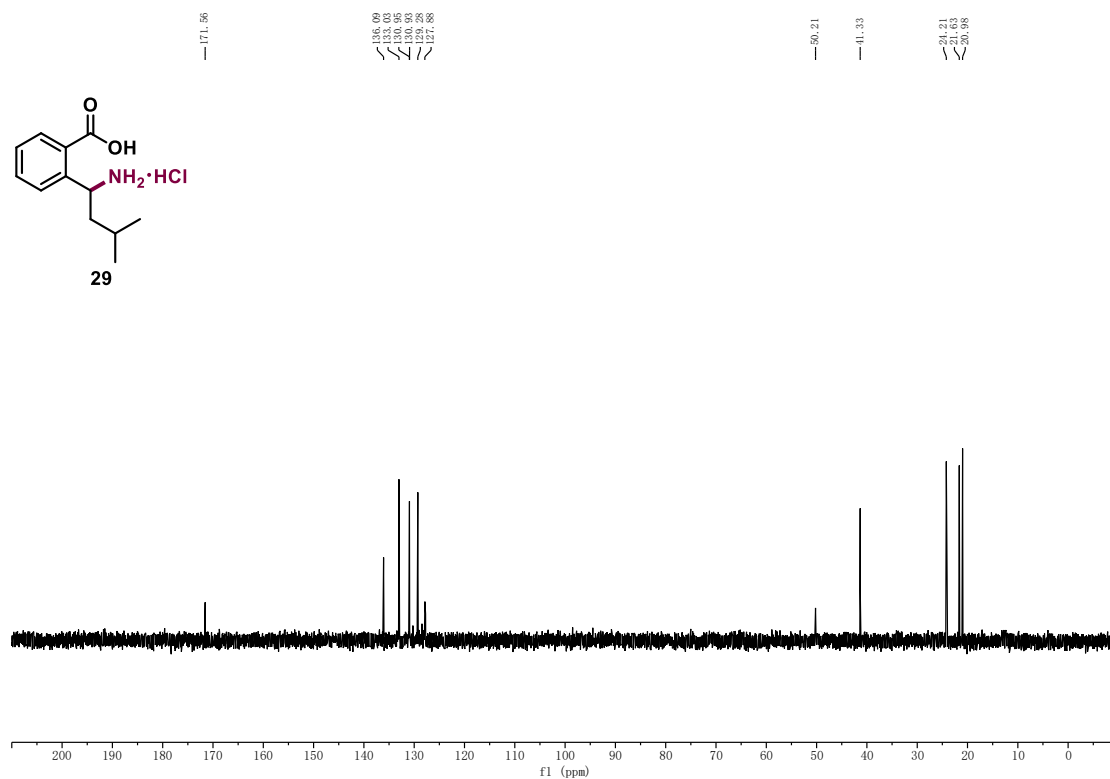
Supplementary Figure 215 ^1H NMR spectrum of compound 28



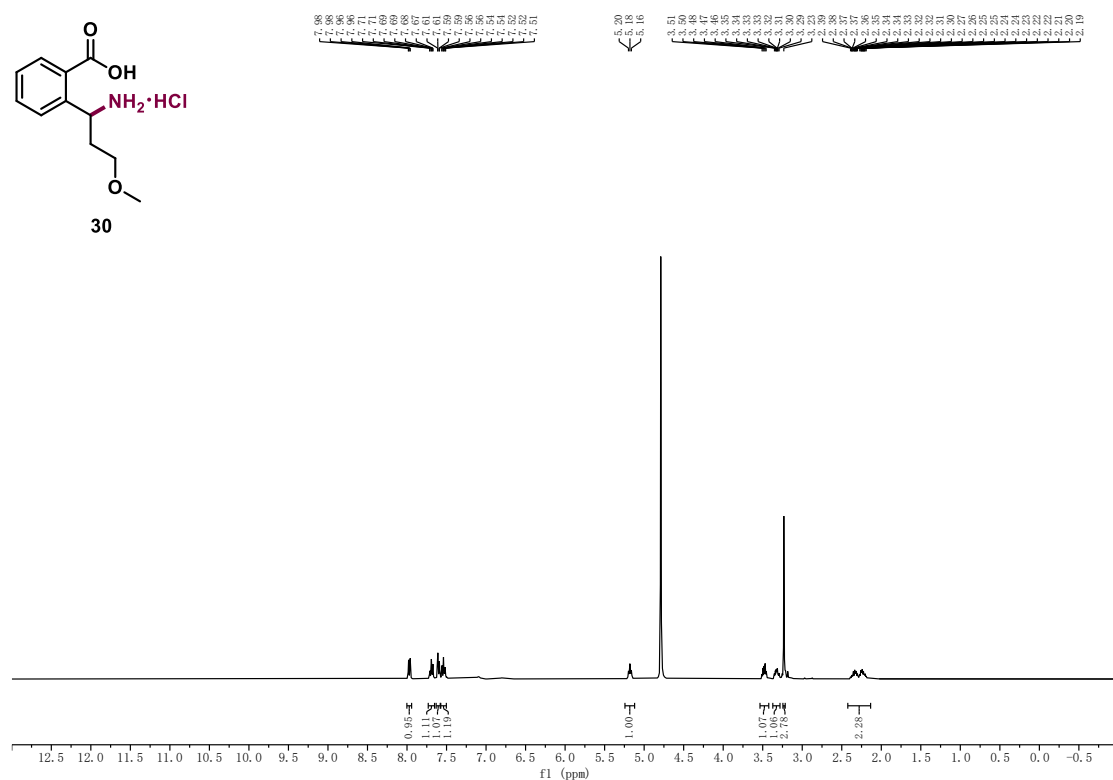
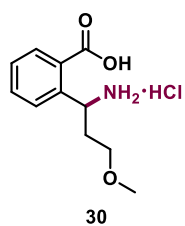
Supplementary Figure 216 ^{13}C NMR spectrum of compound 28



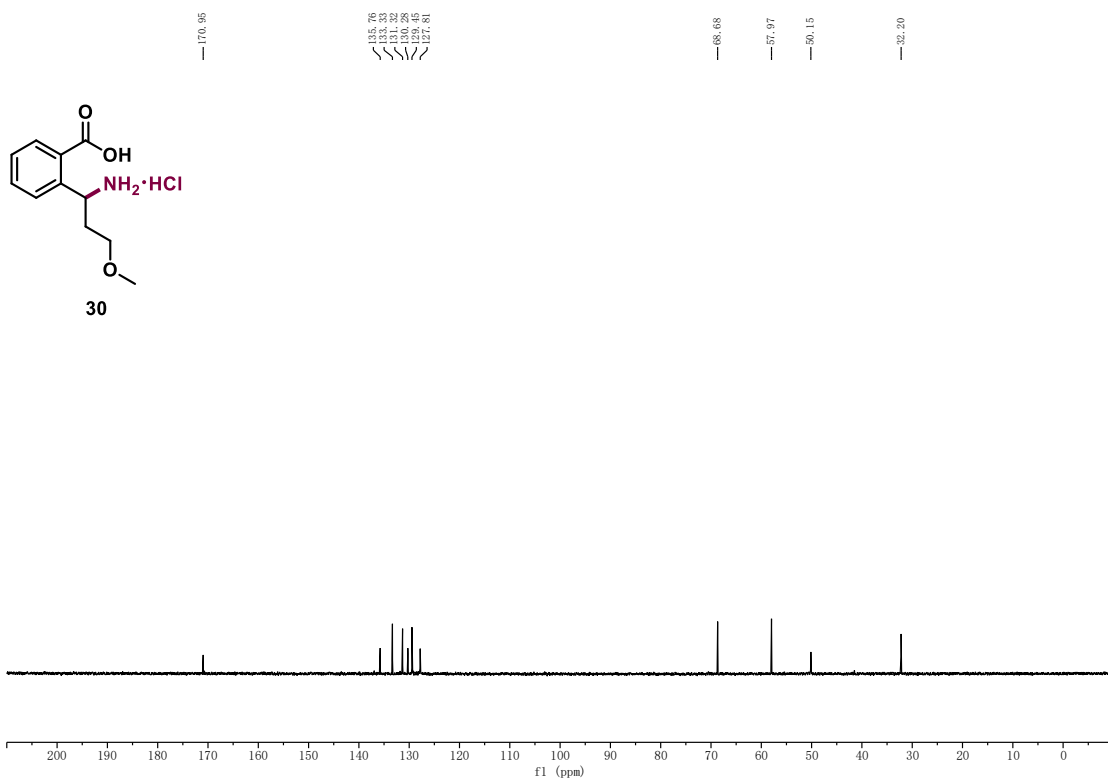
Supplementary Figure 217 ¹H NMR spectrum of compound 29



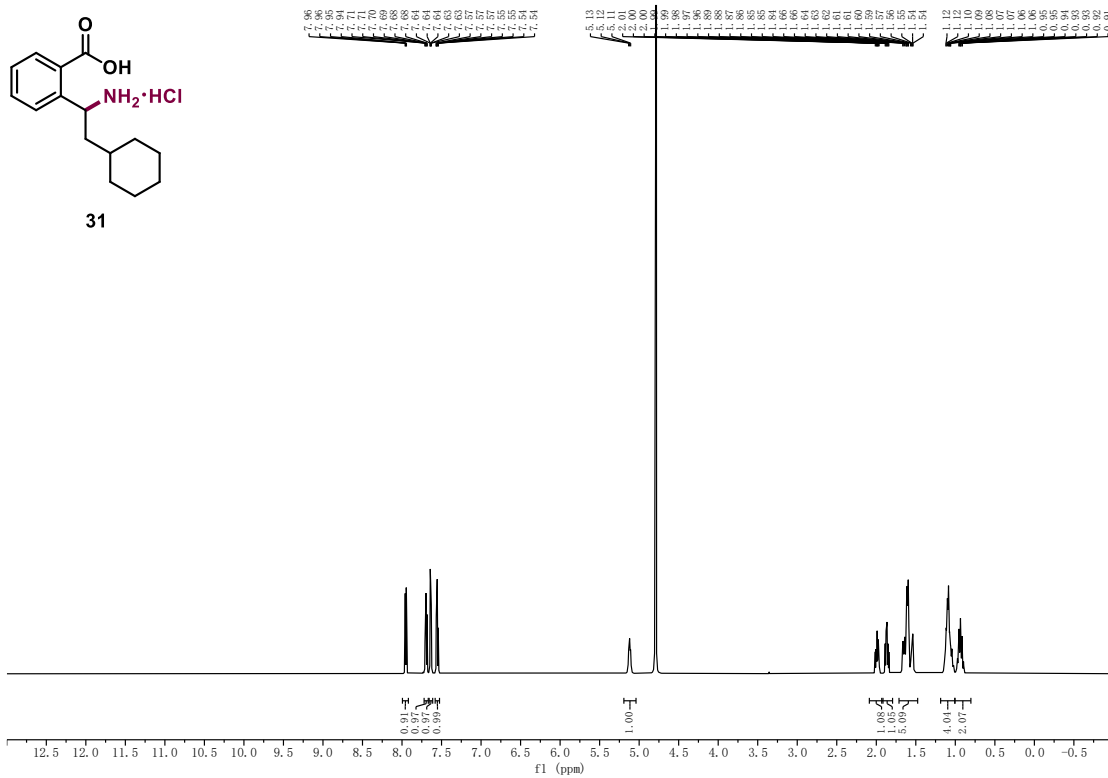
Supplementary Figure 218 ¹³C NMR spectrum of compound 29



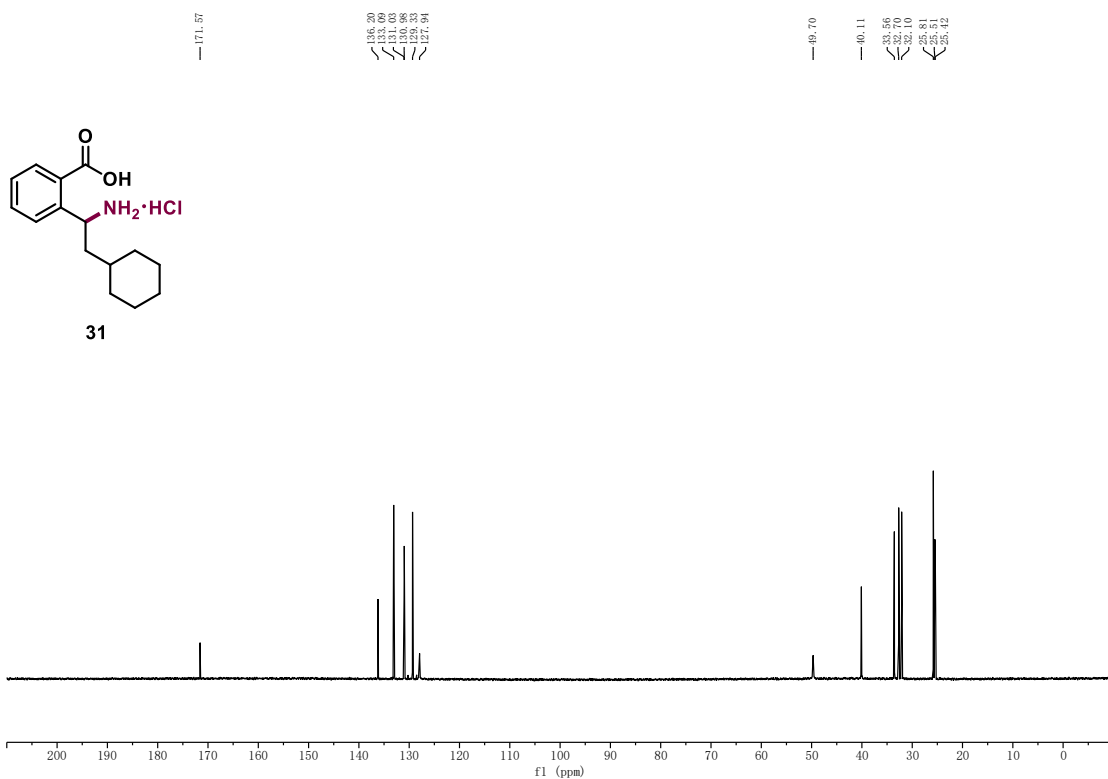
Supplementary Figure 219 ¹H NMR spectrum of compound **30**



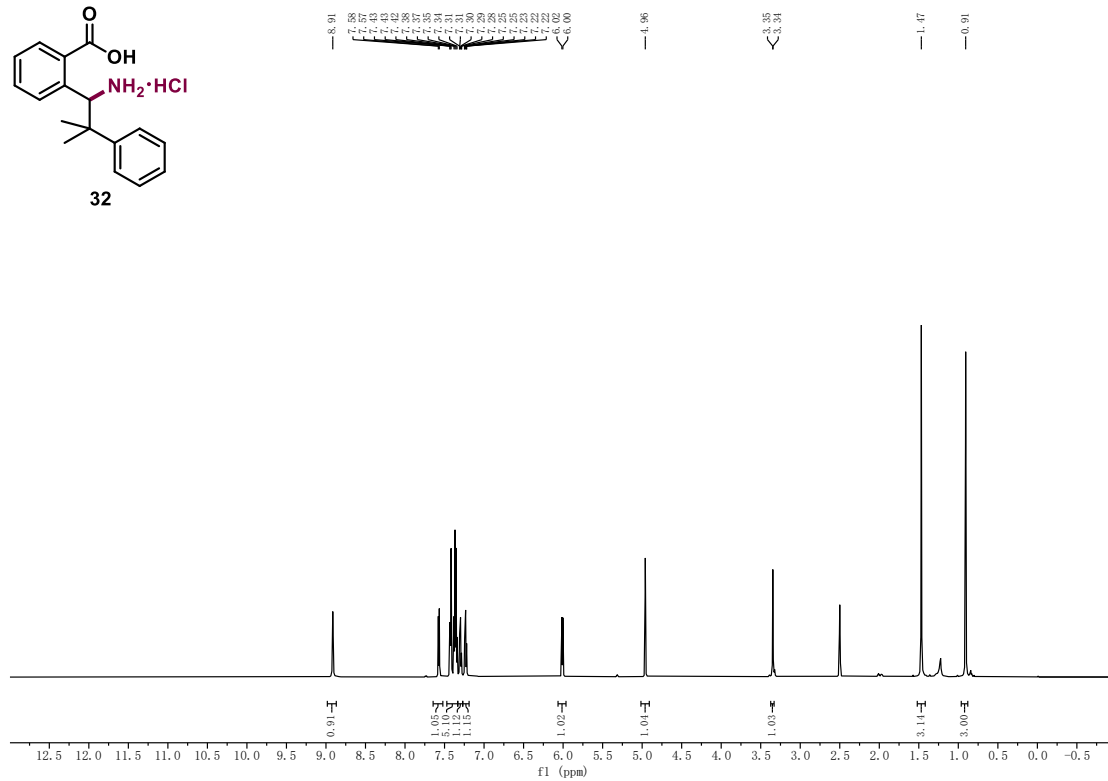
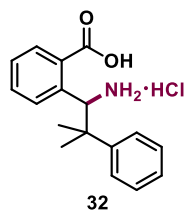
Supplementary Figure 220 ¹³C NMR spectrum of compound **30**



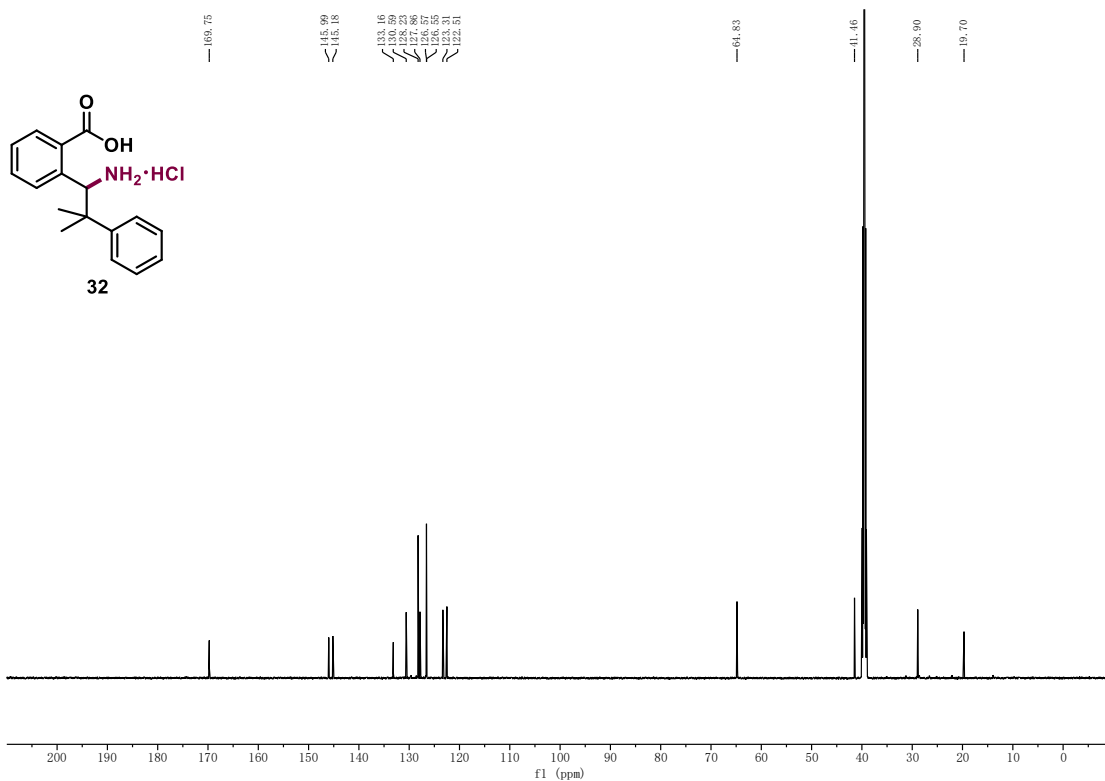
Supplementary Figure 221 ^1H NMR spectrum of compound **31**



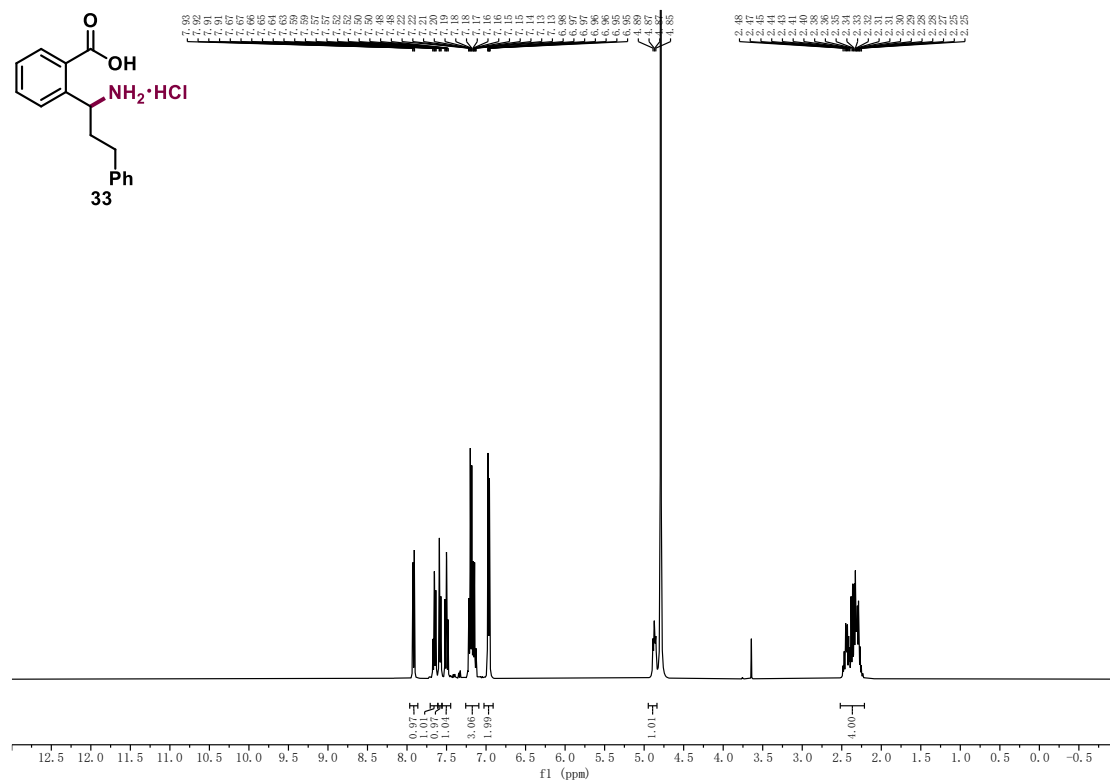
Supplementary Figure 222 ^{13}C NMR spectrum of compound **31**



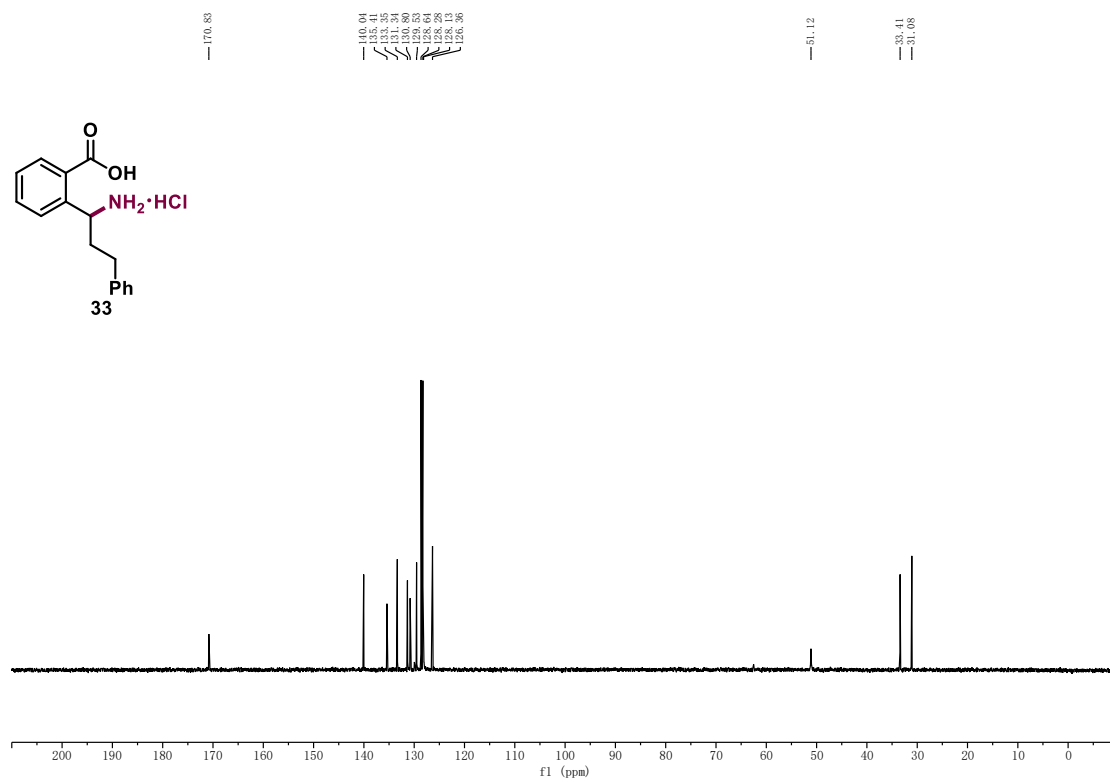
Supplementary Figure 223 ¹H NMR spectrum of compound **32**



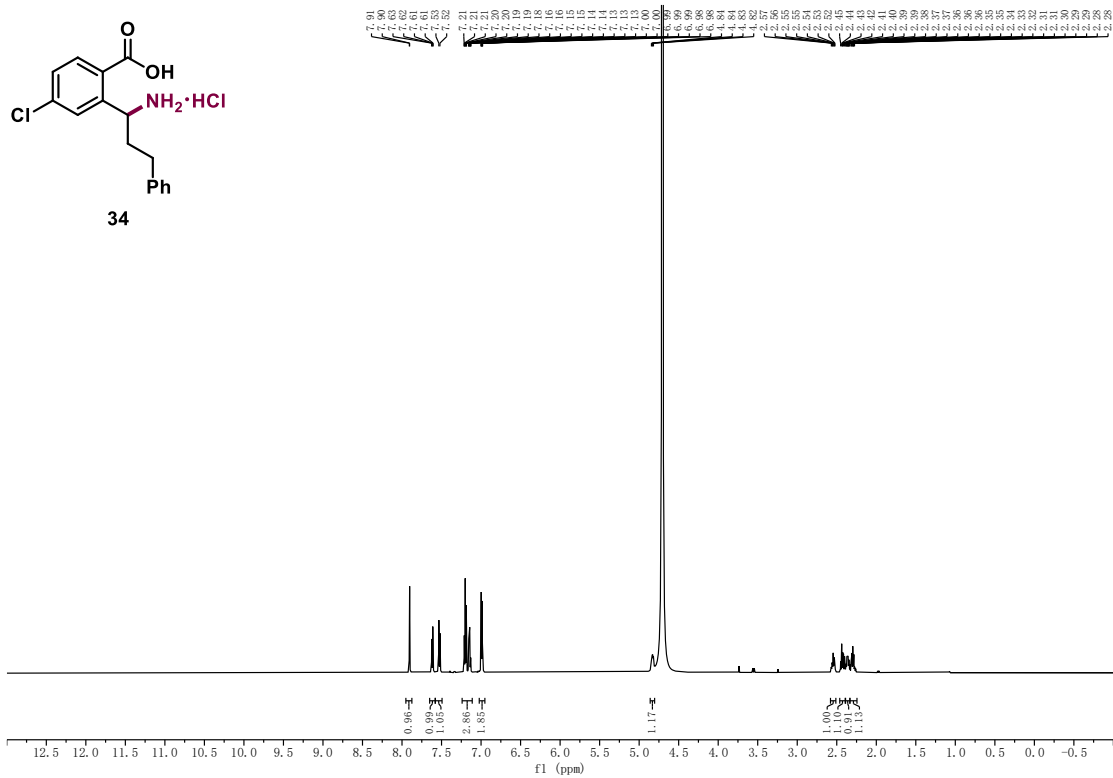
Supplementary Figure 224 ¹³C NMR spectrum of compound **32**



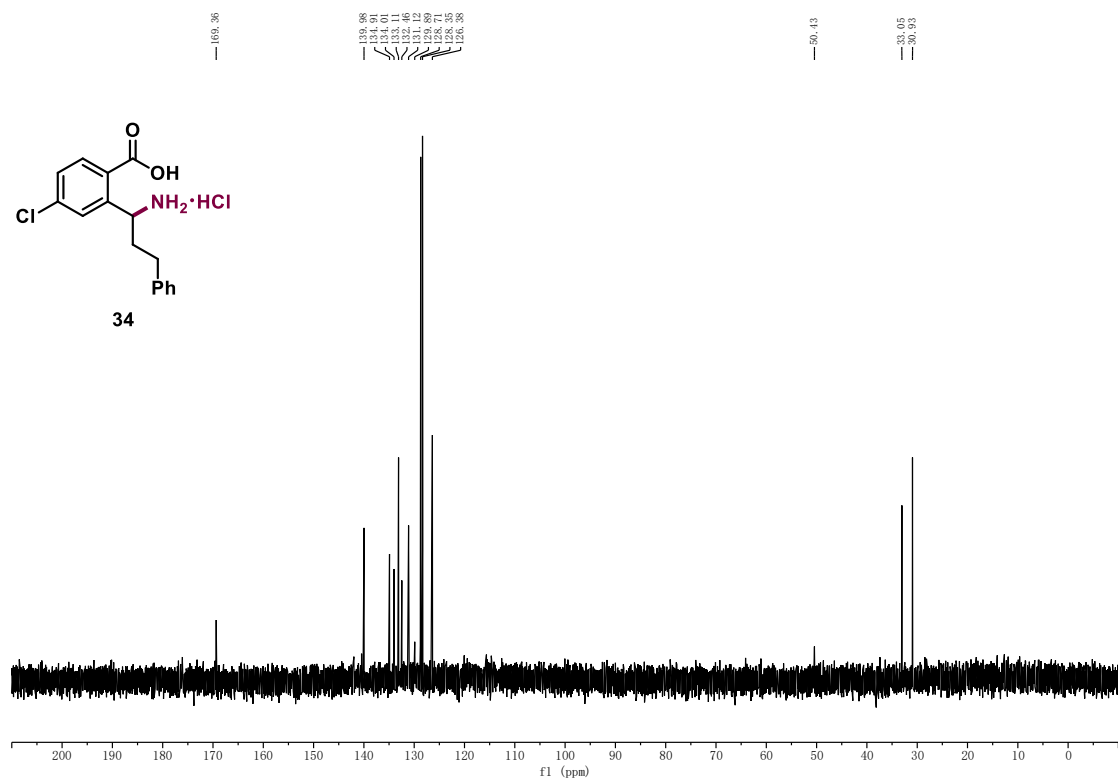
Supplementary Figure 225 ¹H NMR spectrum of compound 33



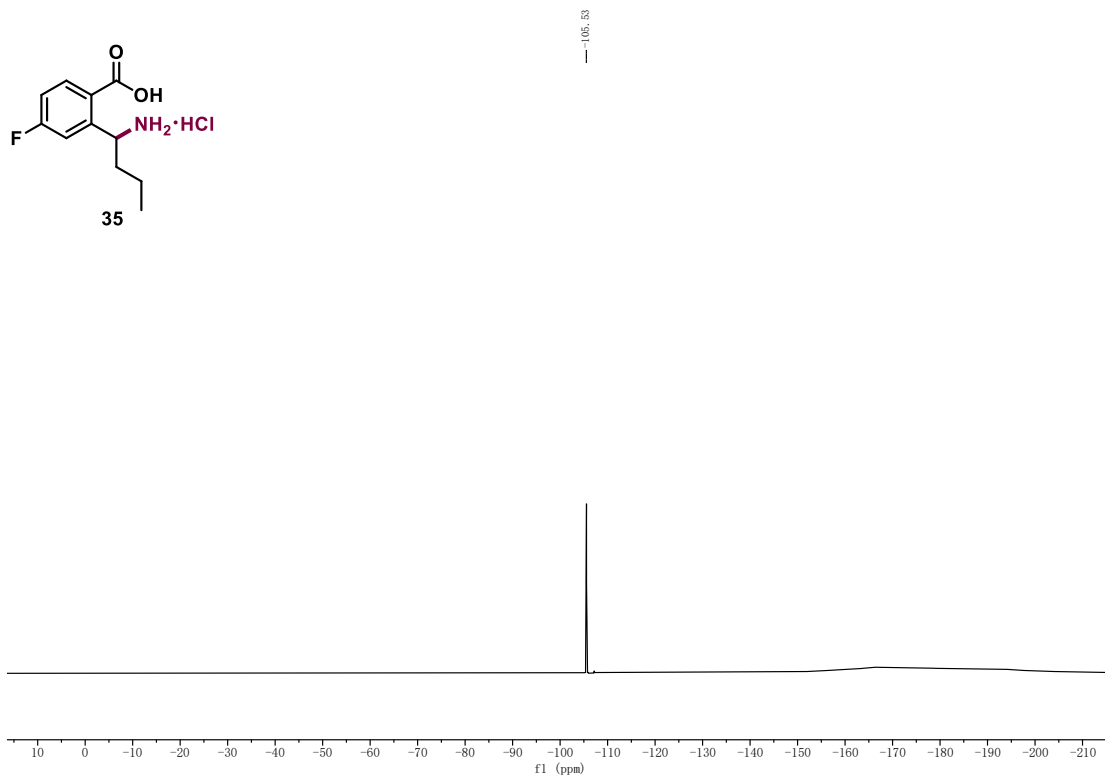
Supplementary Figure 226 ¹³C NMR spectrum of compound 33



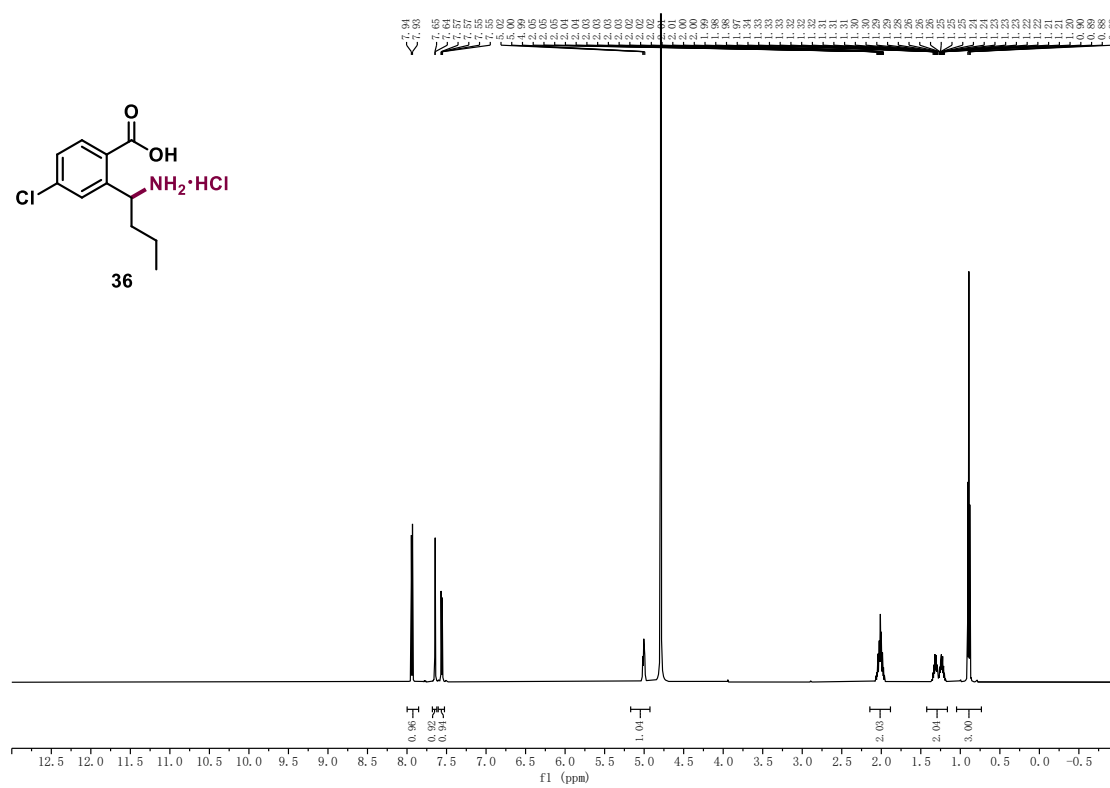
Supplementary Figure 227 ¹H NMR spectrum of compound 34



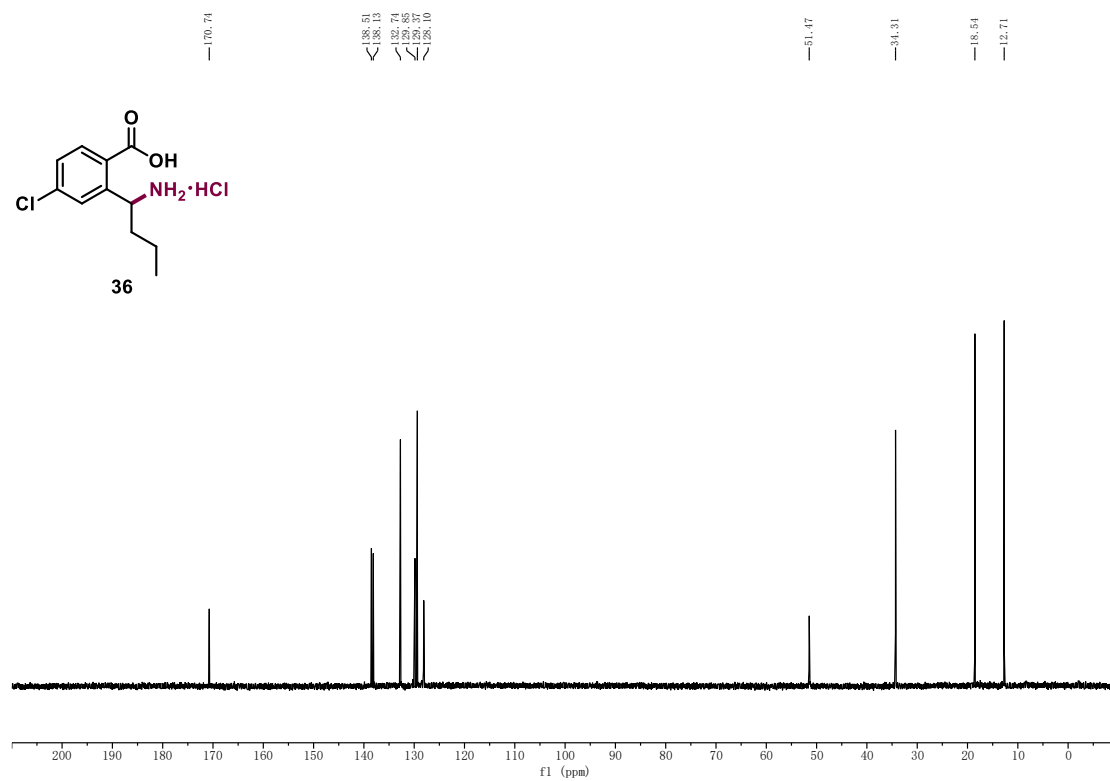
Supplementary Figure 228 ¹³C NMR spectrum of compound 34



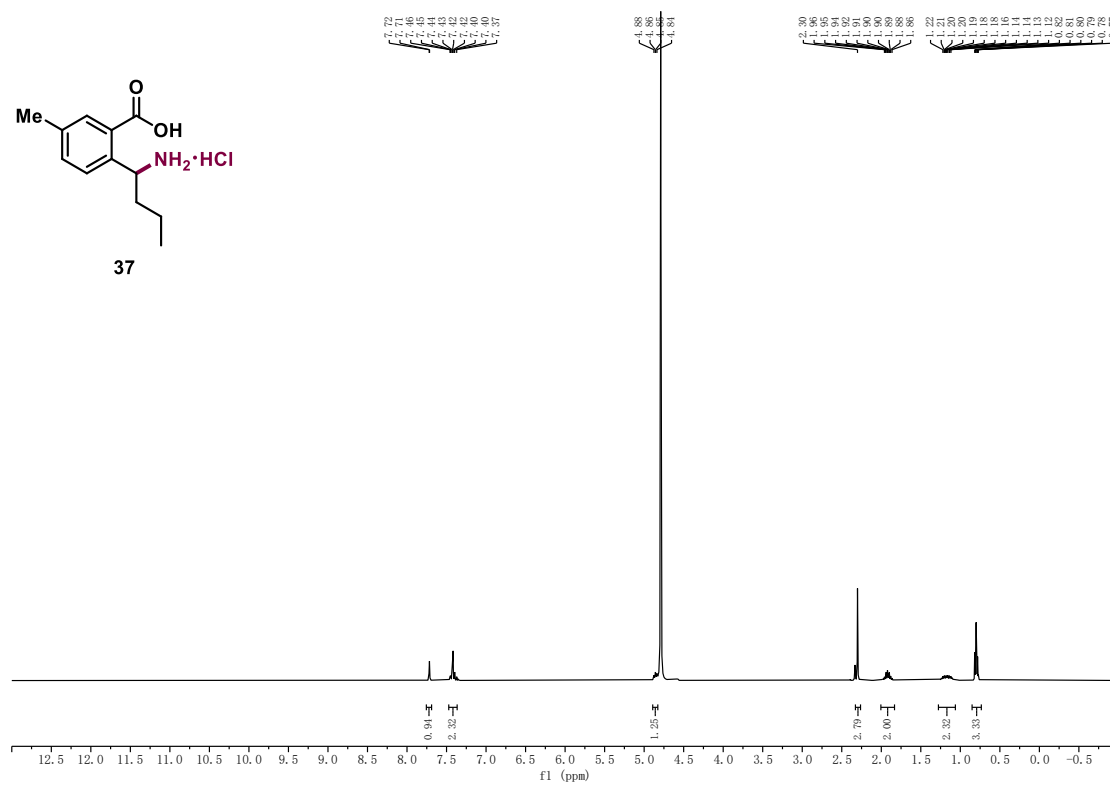
Supplementary Figure 231 ¹⁹F NMR spectrum of compound 35



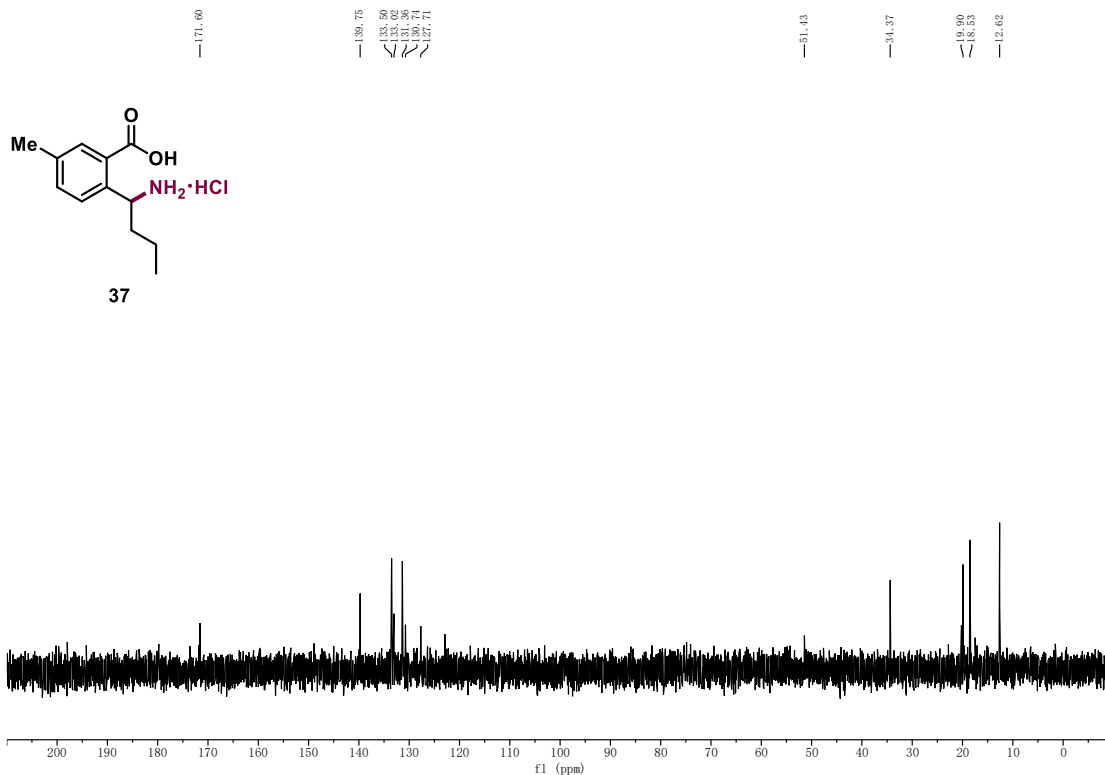
Supplementary Figure 232 ¹H NMR spectrum of compound 36



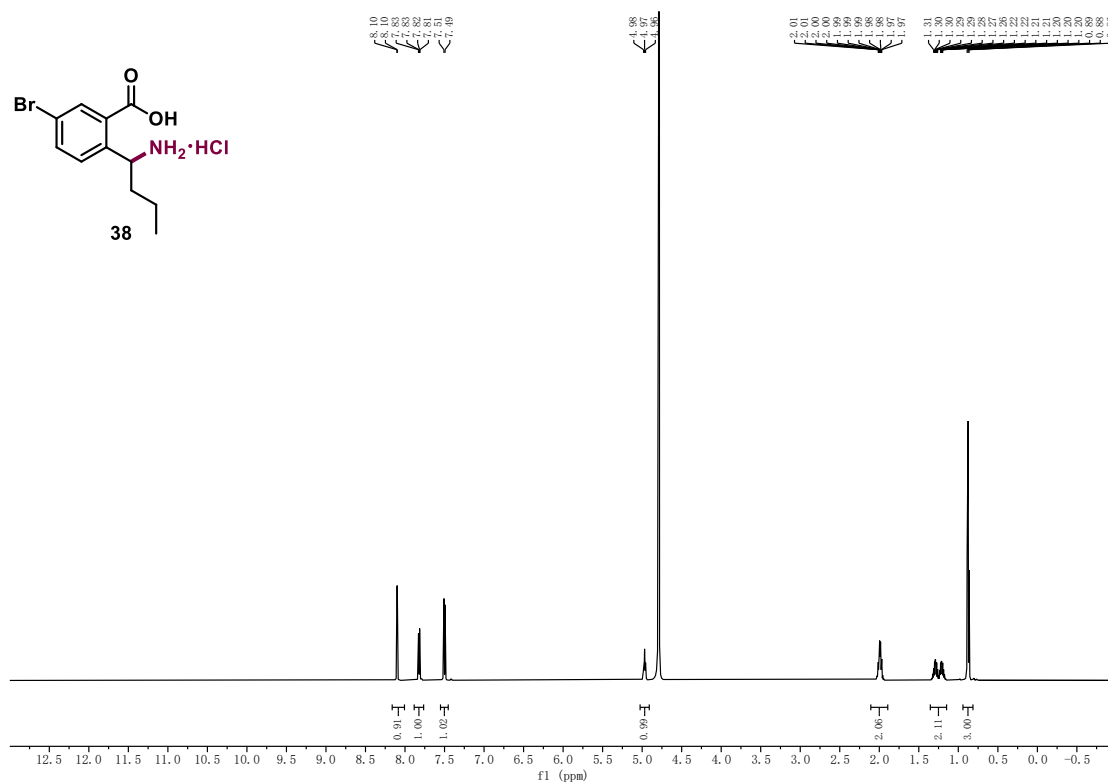
Supplementary Figure 233 ^{13}C NMR spectrum of compound 36



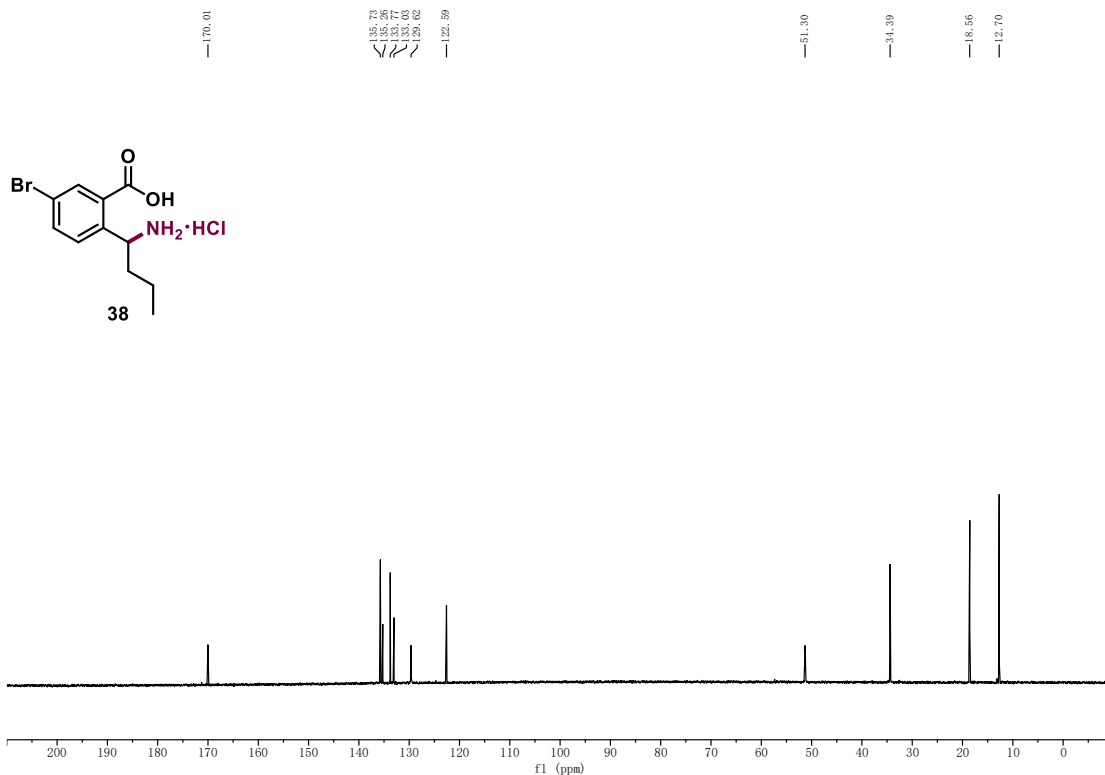
Supplementary Figure 234 ^1H NMR spectrum of compound 37



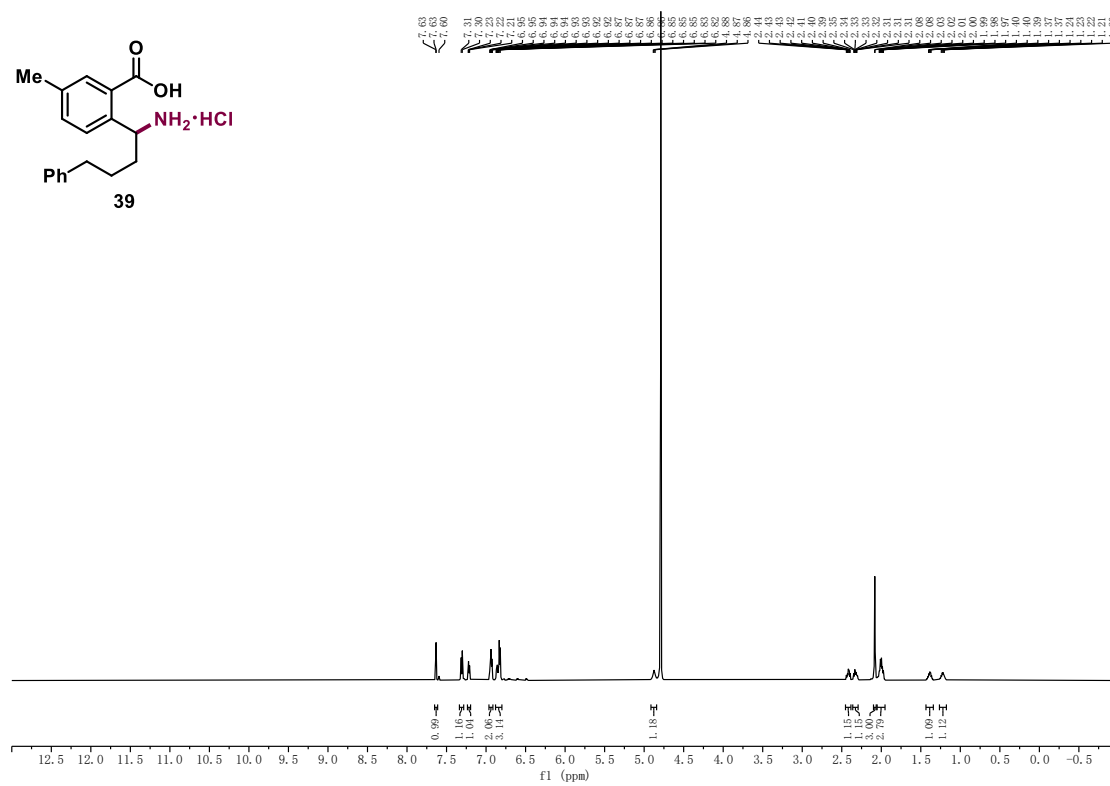
Supplementary Figure 235 ¹³C NMR spectrum of compound 37



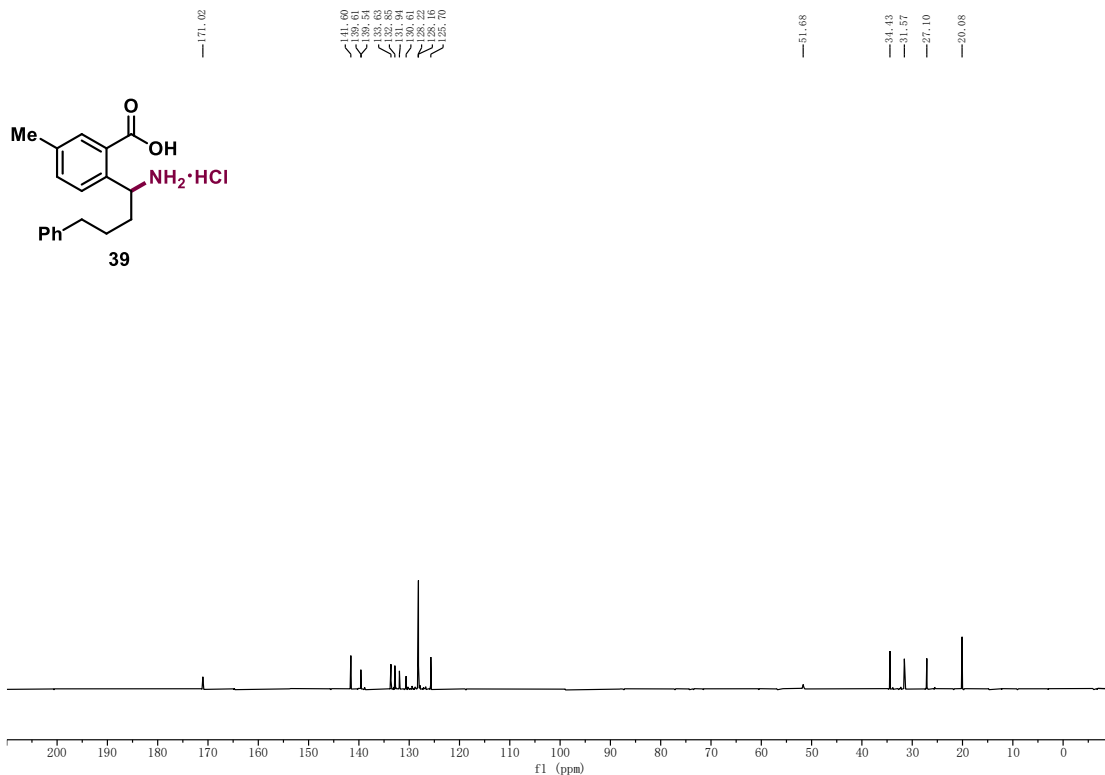
Supplementary Figure 236 ¹H NMR spectrum of compound 38



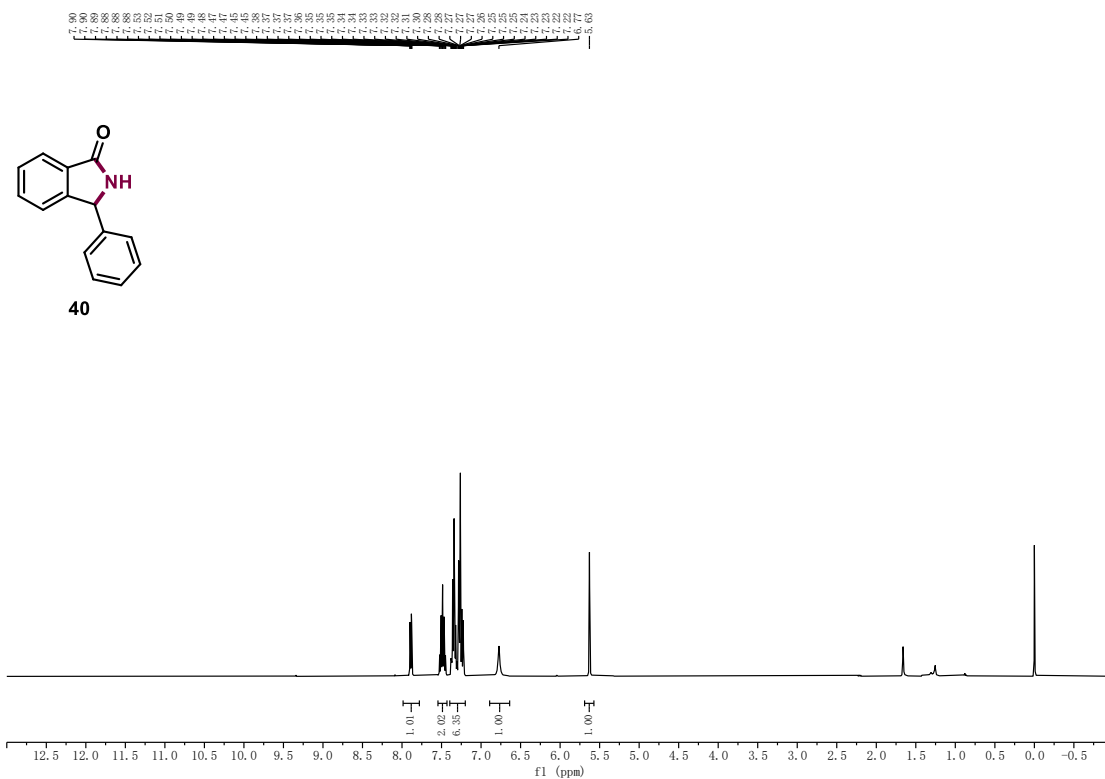
Supplementary Figure 237 ^{13}C NMR spectrum of compound 38



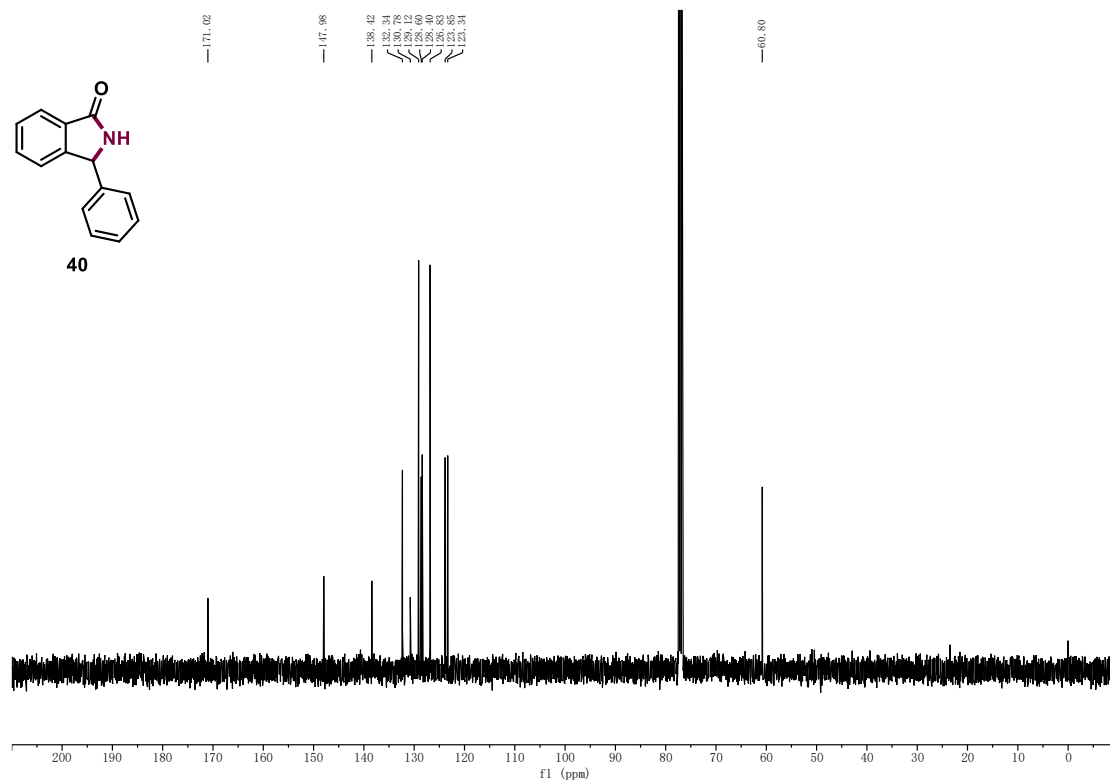
Supplementary Figure 238 ^1H NMR spectrum of compound 39



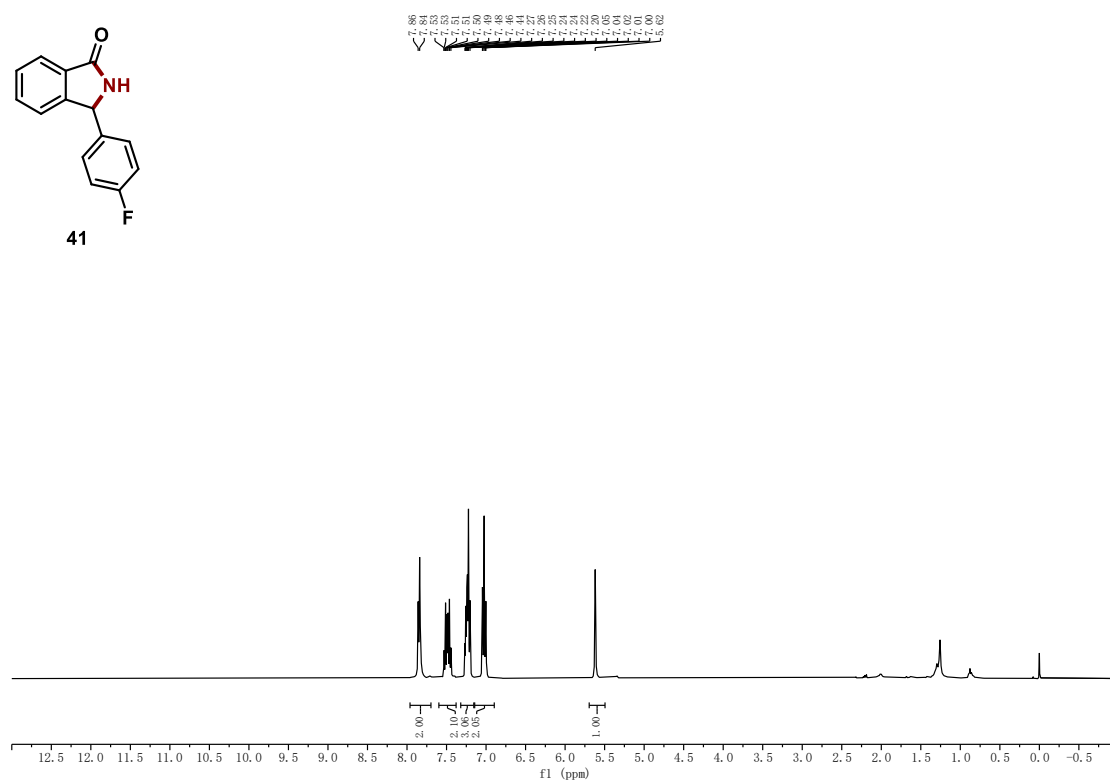
Supplementary Figure 239 ¹³C NMR spectrum of compound 39



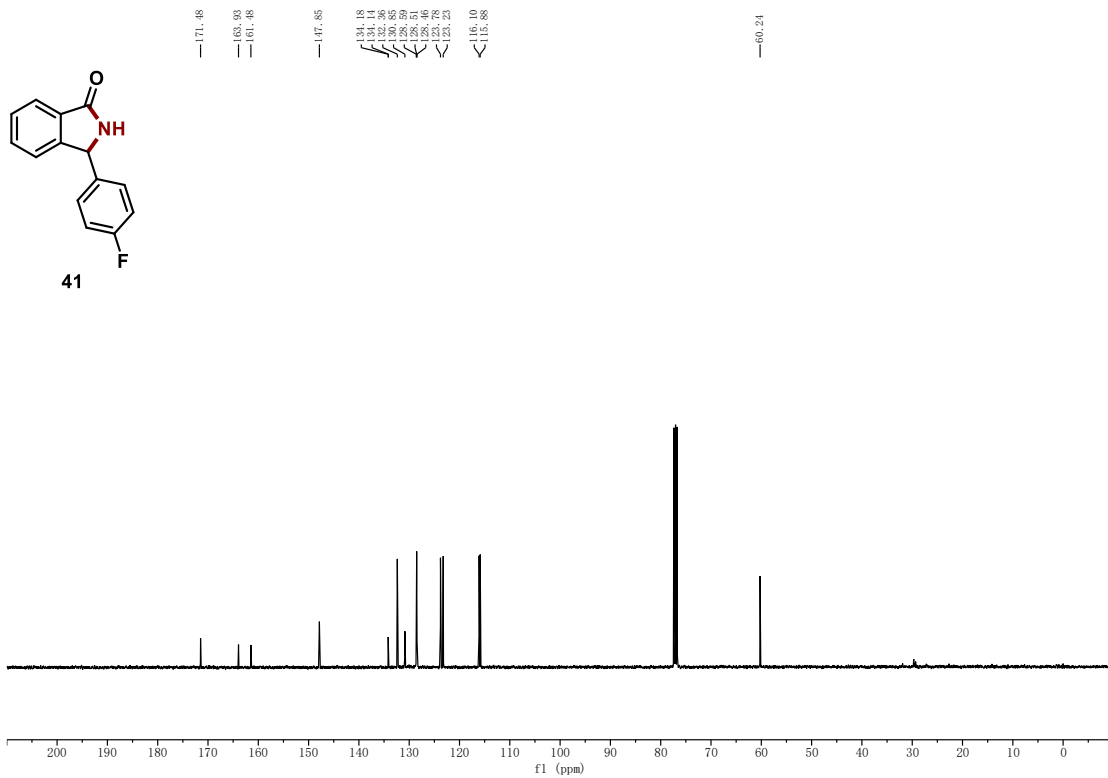
Supplementary Figure 240 ¹H NMR spectrum of compound 40



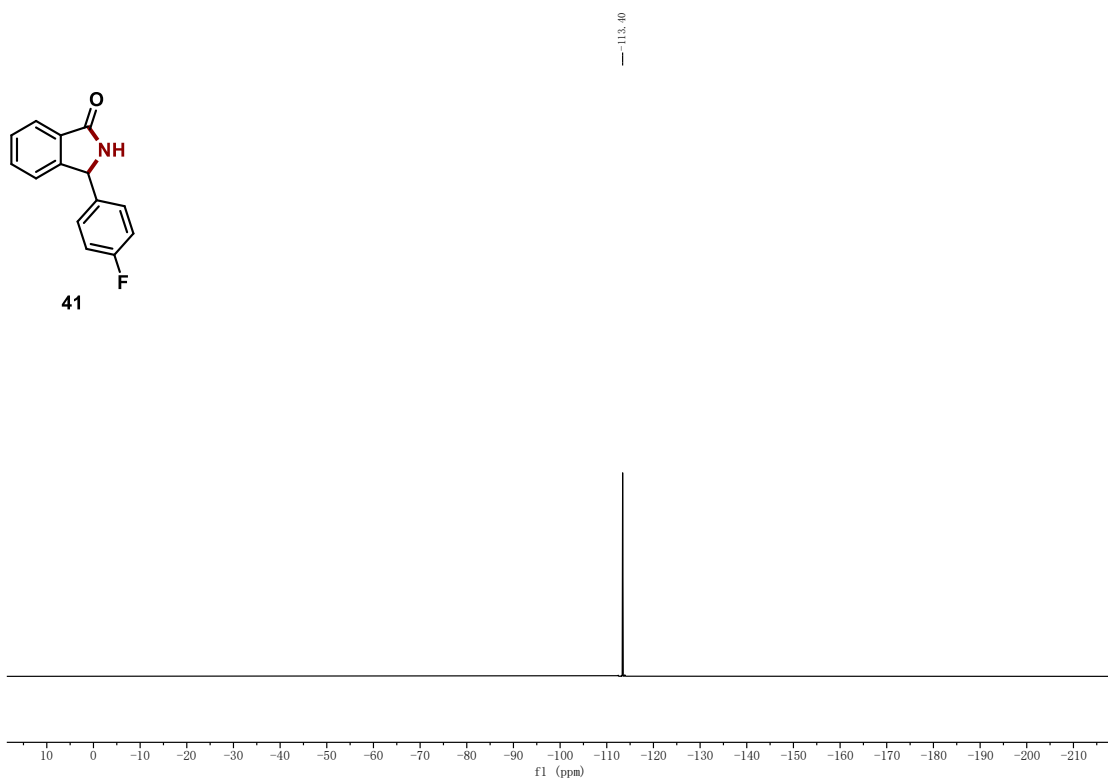
Supplementary Figure 241 ^{13}C NMR spectrum of compound 40



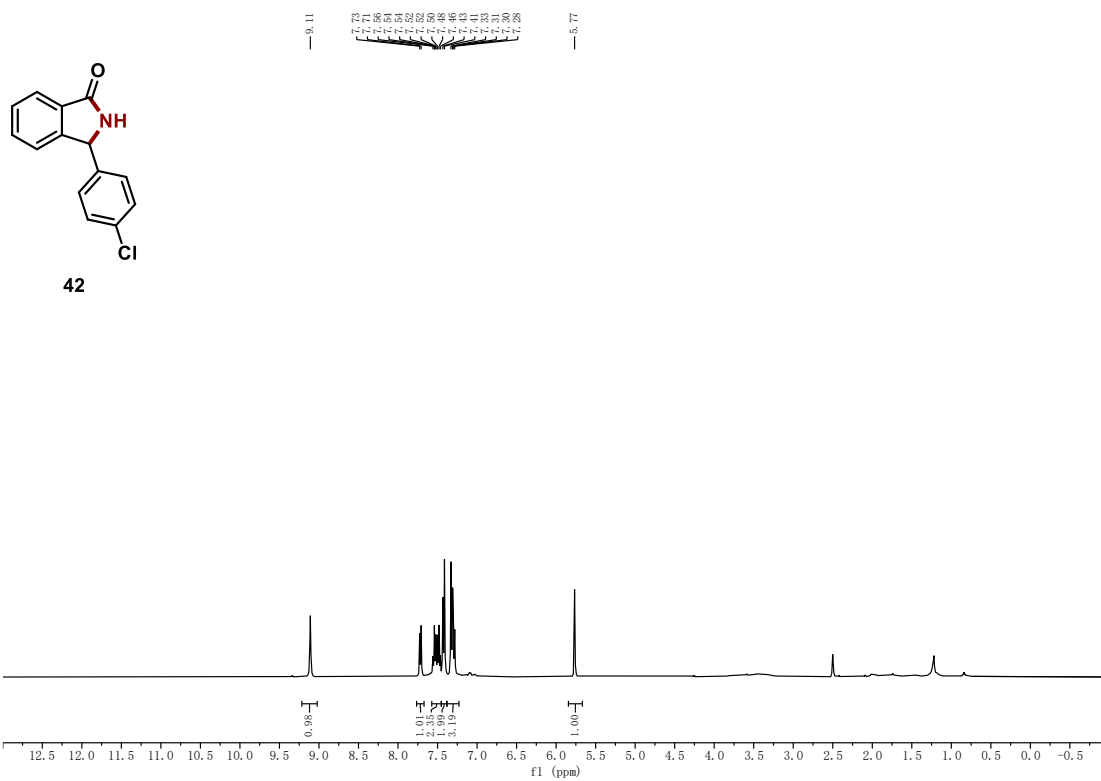
Supplementary Figure 242 ^1H NMR spectrum of compound 41



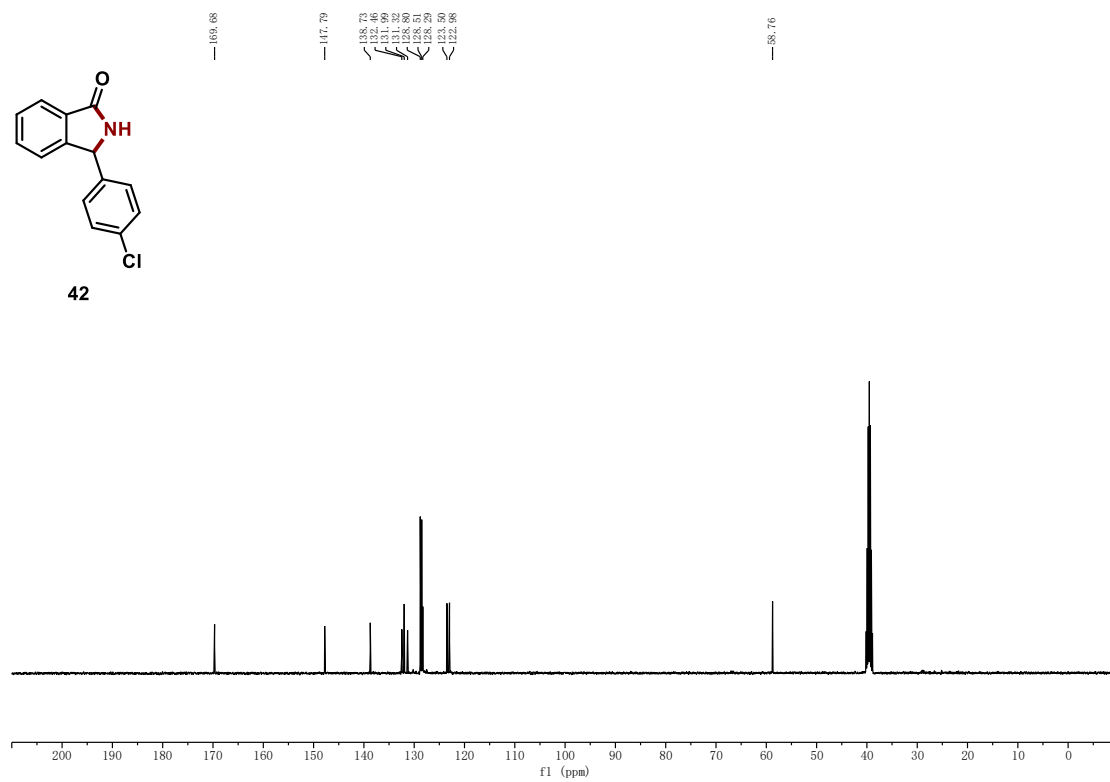
Supplementary Figure 243 ¹³C NMR spectrum of compound 41



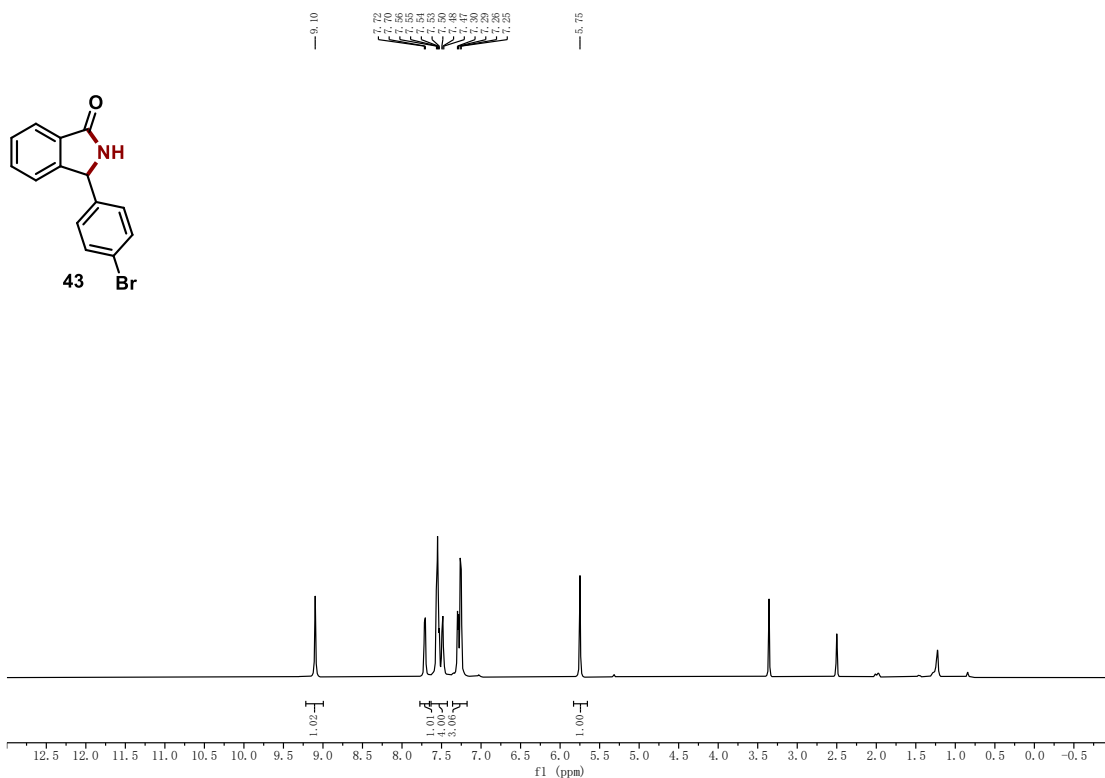
Supplementary Figure 244 ¹⁹F NMR spectrum of compound 41



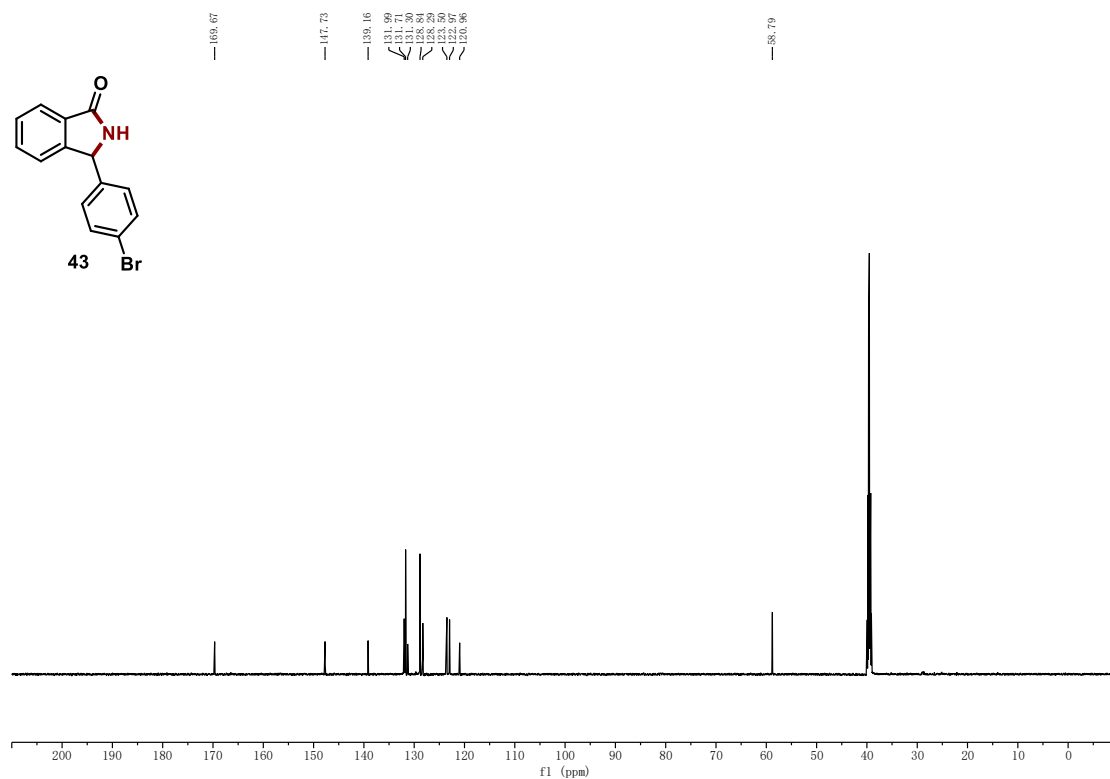
Supplementary Figure 245 ^1H NMR spectrum of compound **42**



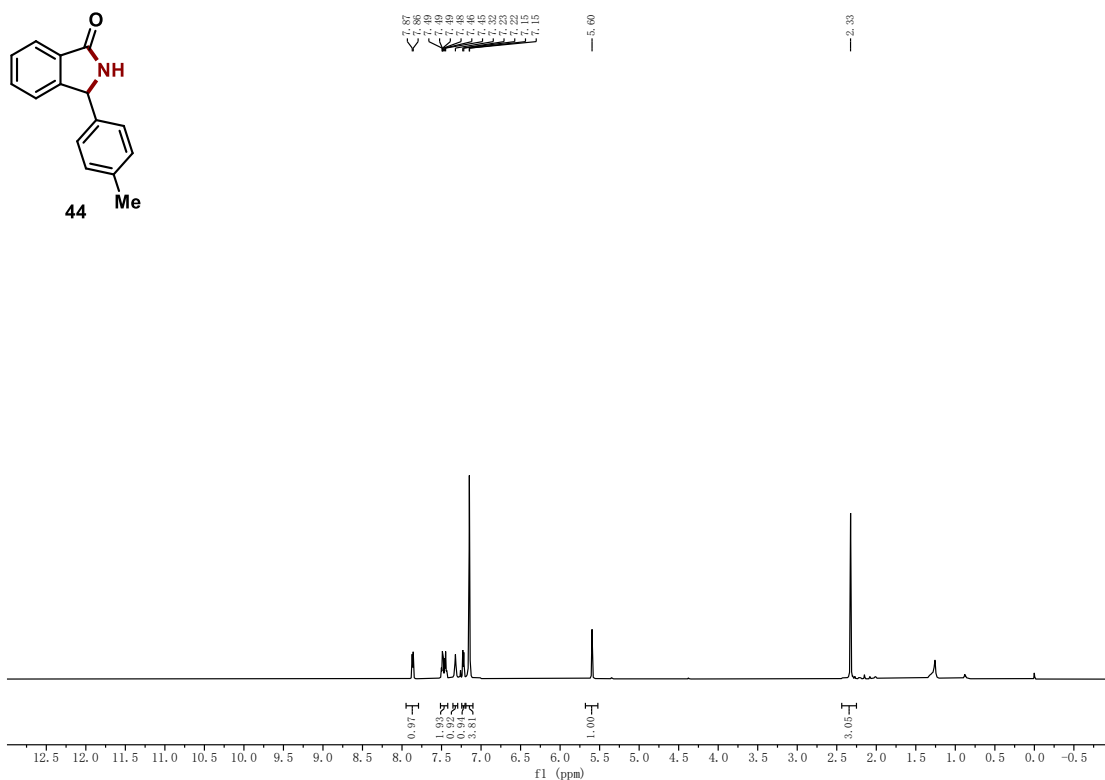
Supplementary Figure 246 ^{13}C NMR spectrum of compound **42**



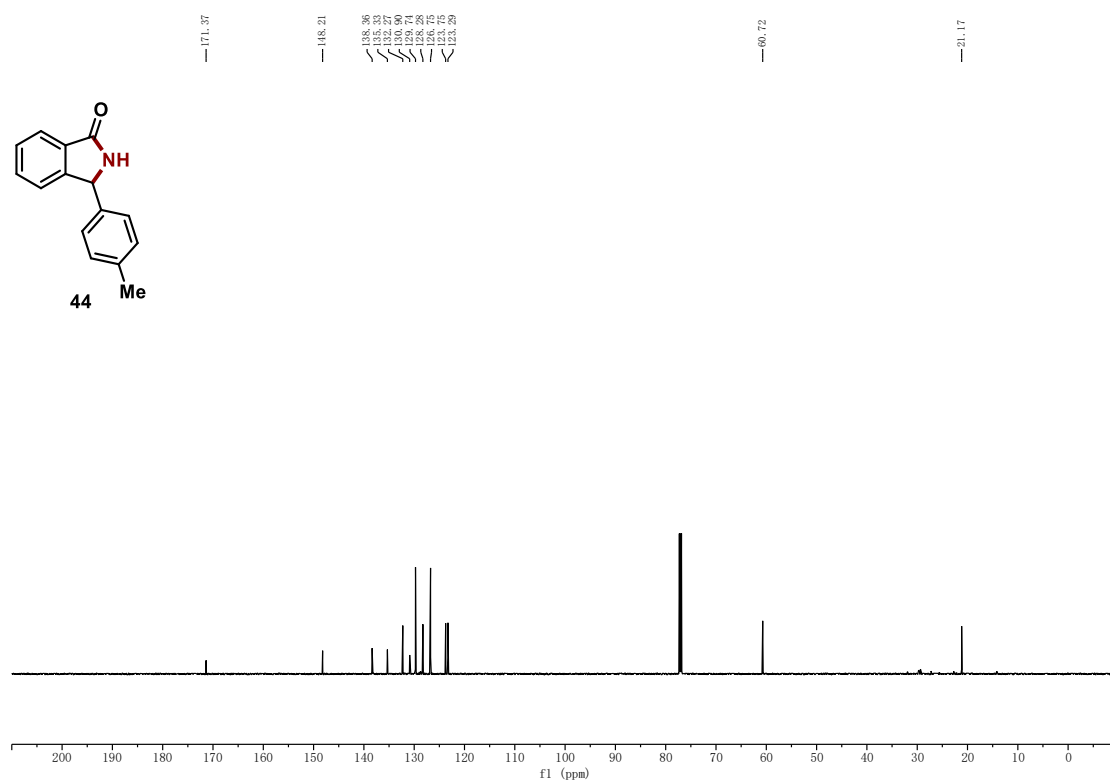
Supplementary Figure 247 ^1H NMR spectrum of compound 43



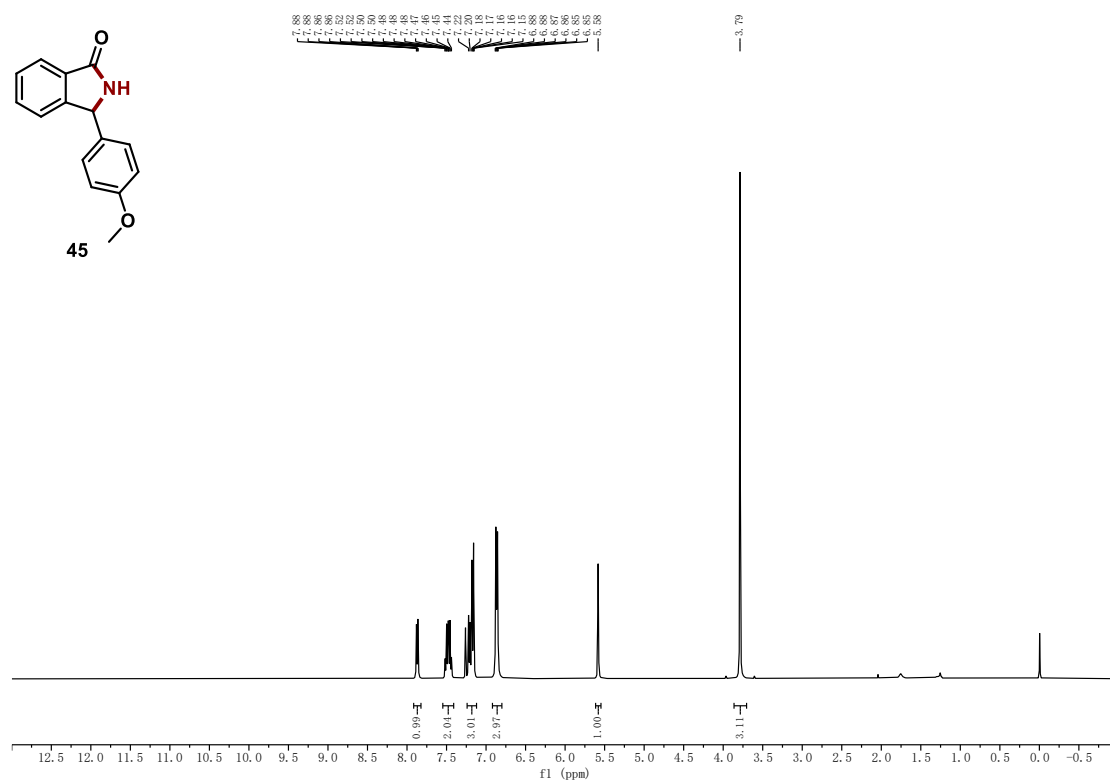
Supplementary Figure 248 ^{13}C NMR spectrum of compound 43



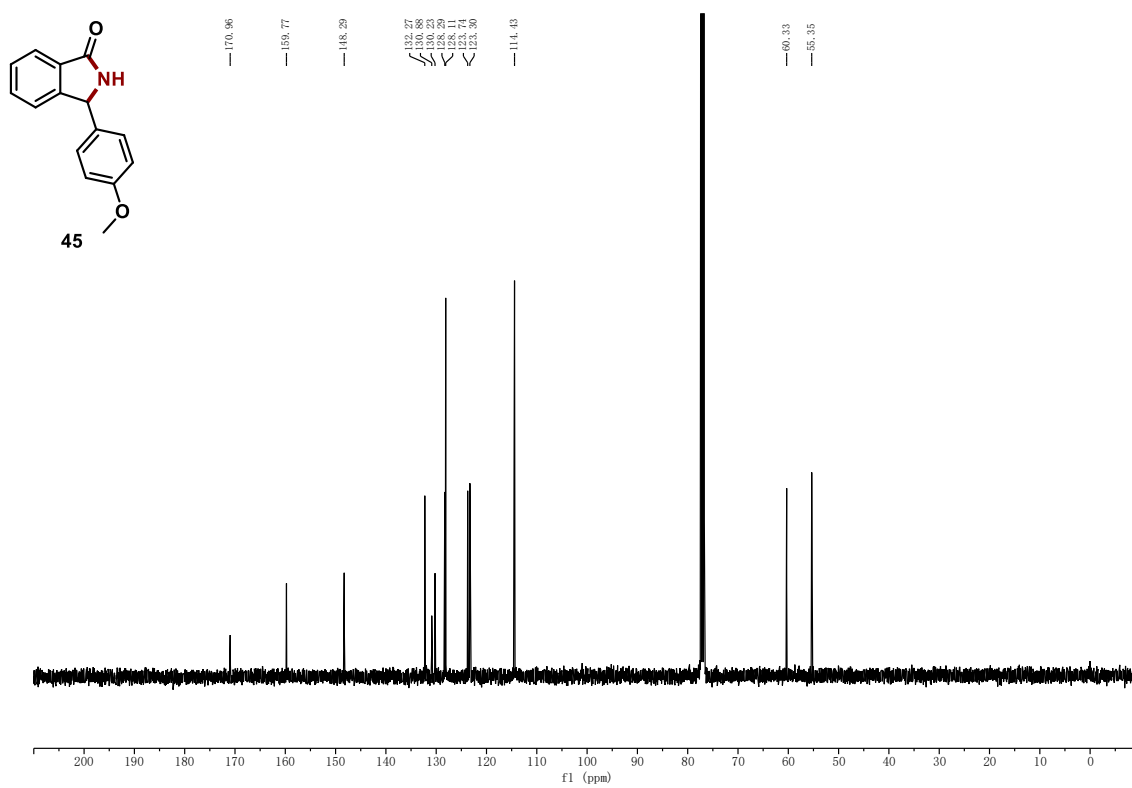
Supplementary Figure 249 ¹H NMR spectrum of compound 44



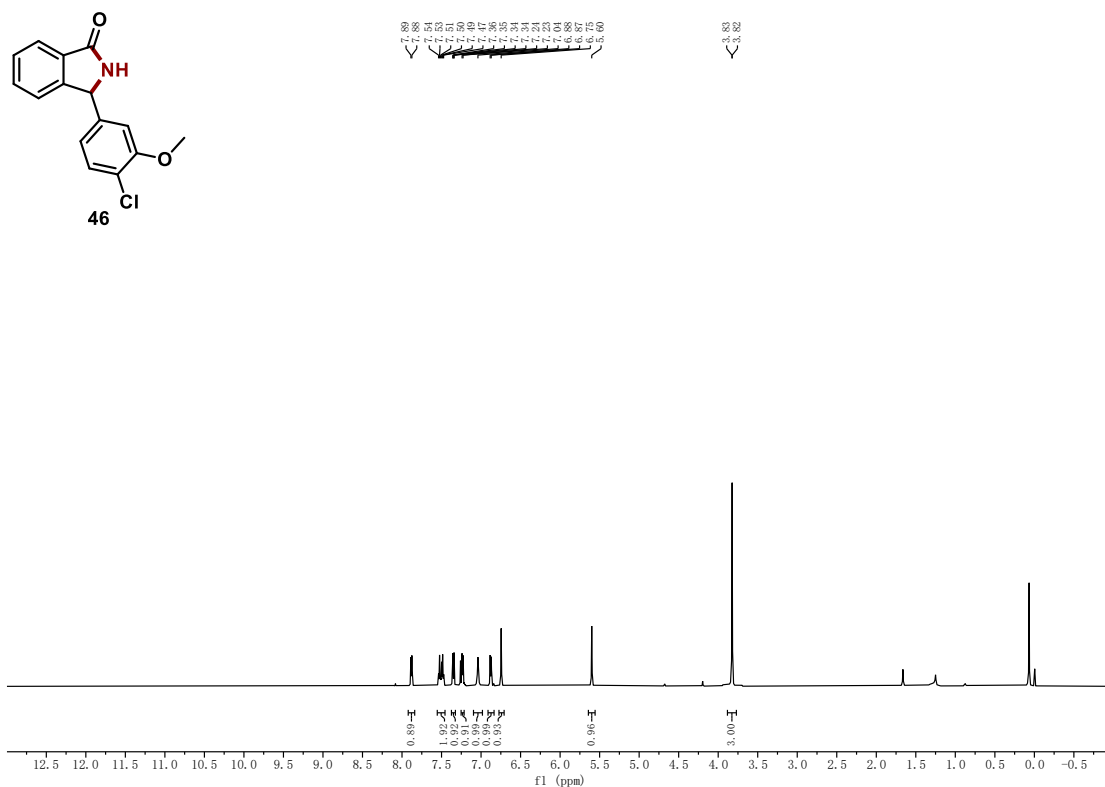
Supplementary Figure 250 ¹³C NMR spectrum of compound 44



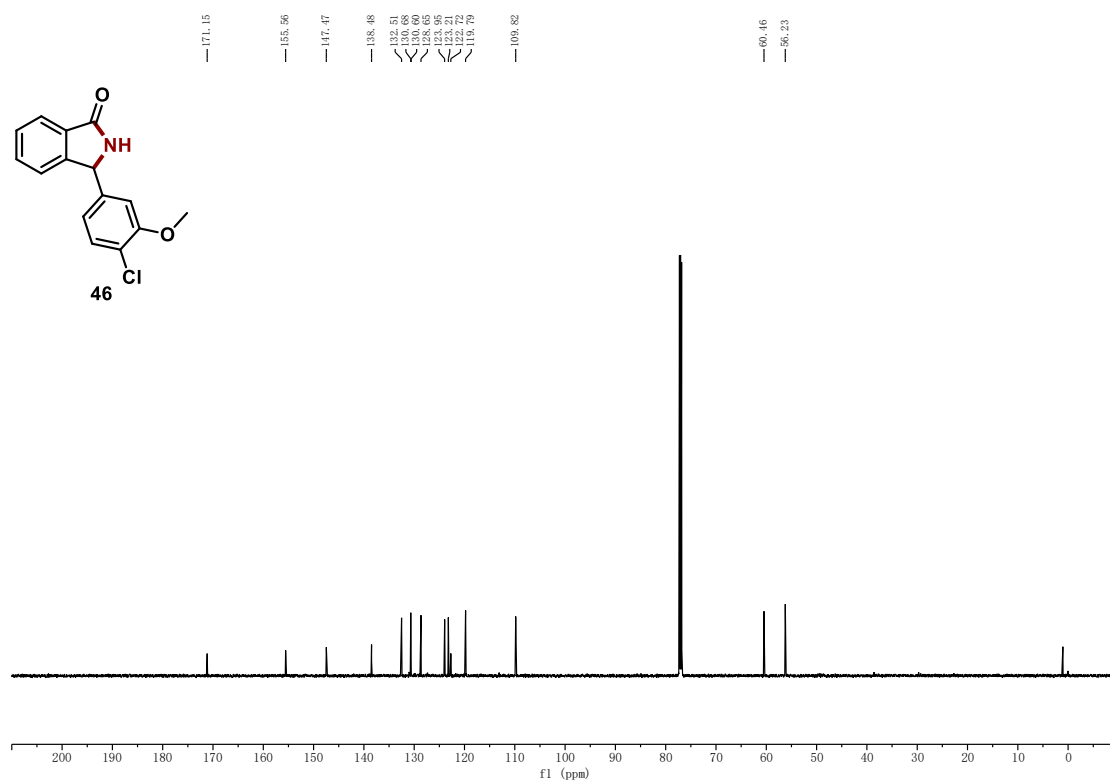
Supplementary Figure 251 ¹H NMR spectrum of compound 45



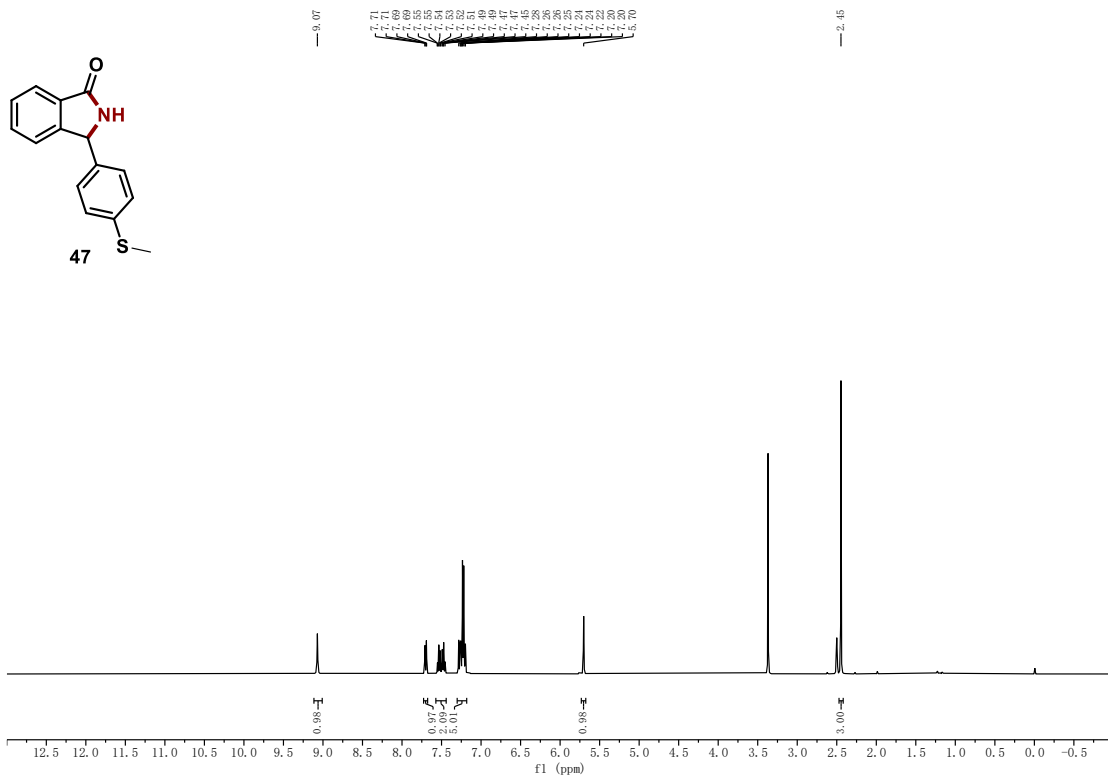
Supplementary Figure 252 ¹³C NMR spectrum of compound 45



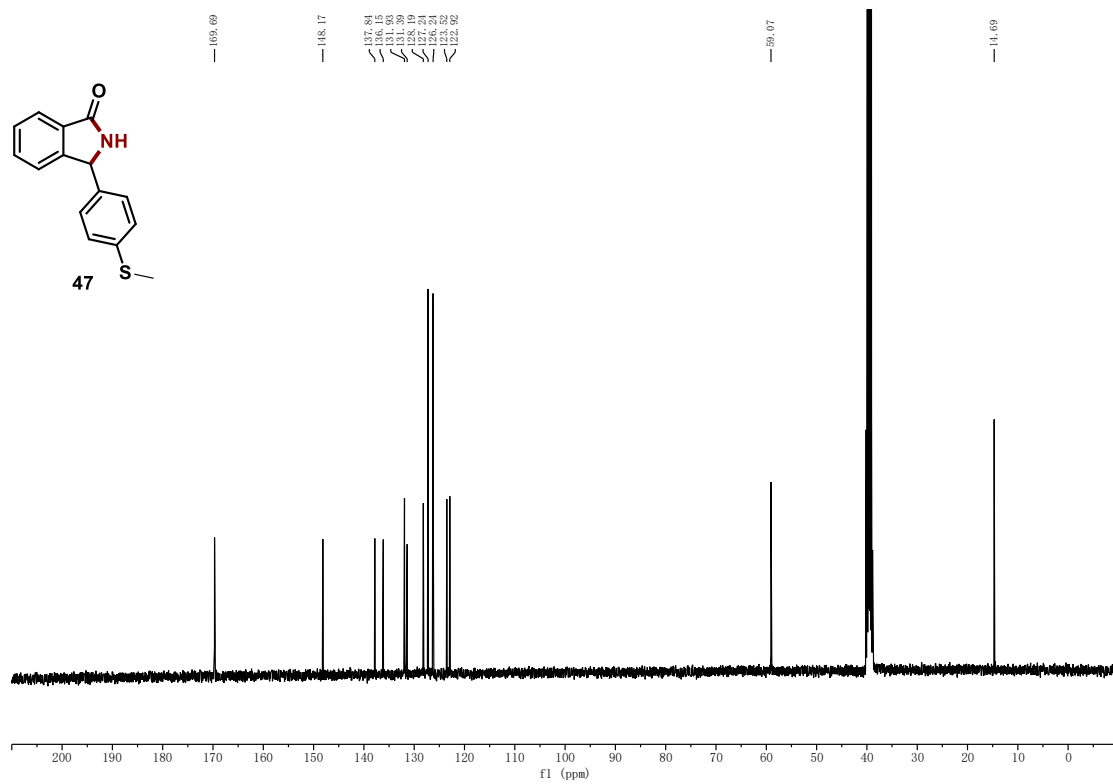
Supplementary Figure 253 ¹H NMR spectrum of compound 46



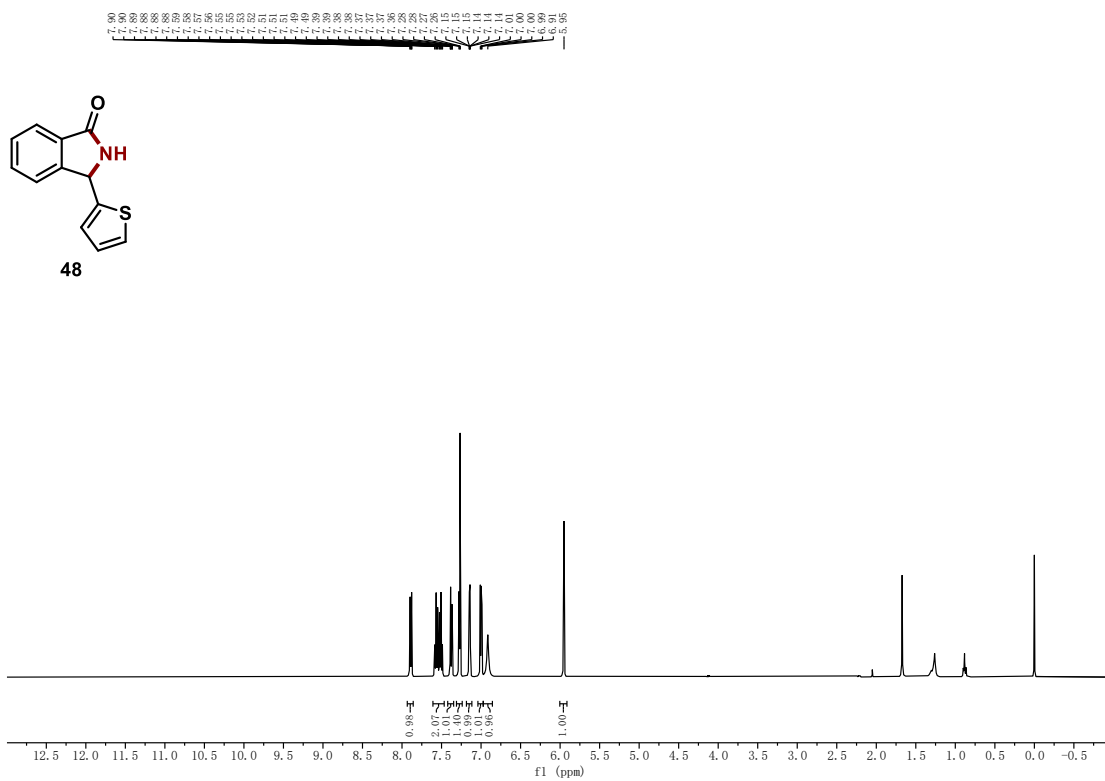
Supplementary Figure 254 ¹³C NMR spectrum of compound 46



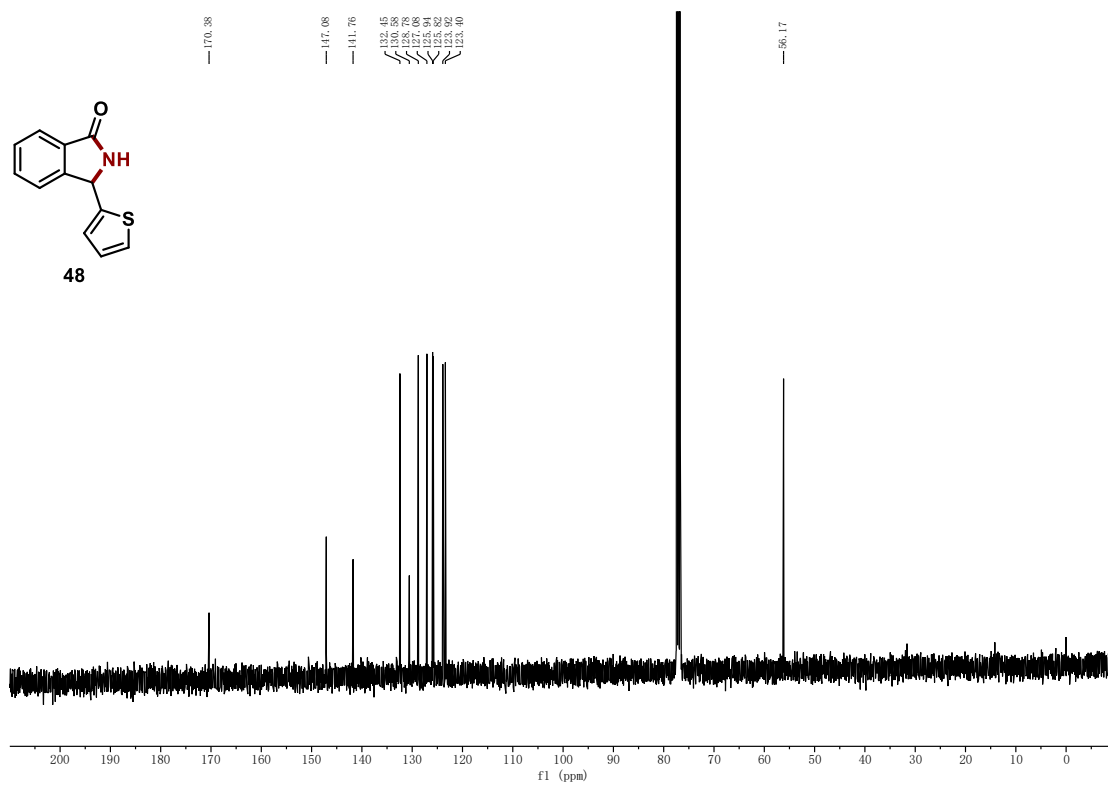
Supplementary Figure 255 ^1H NMR spectrum of compound 47



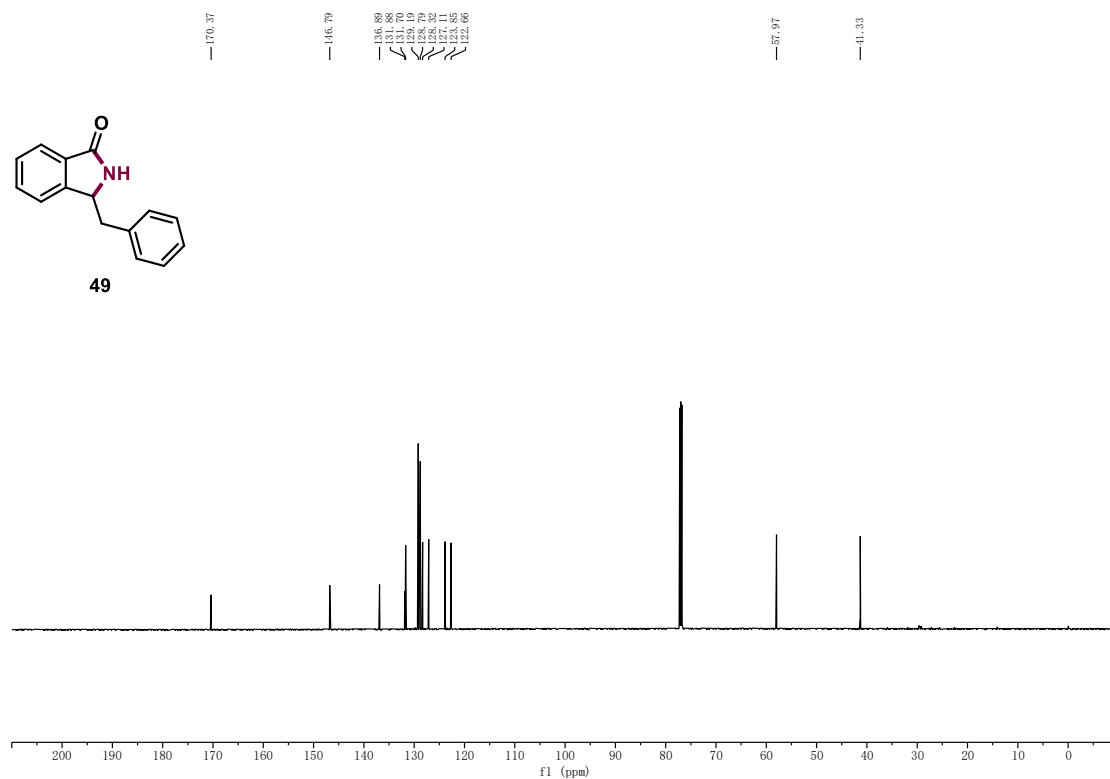
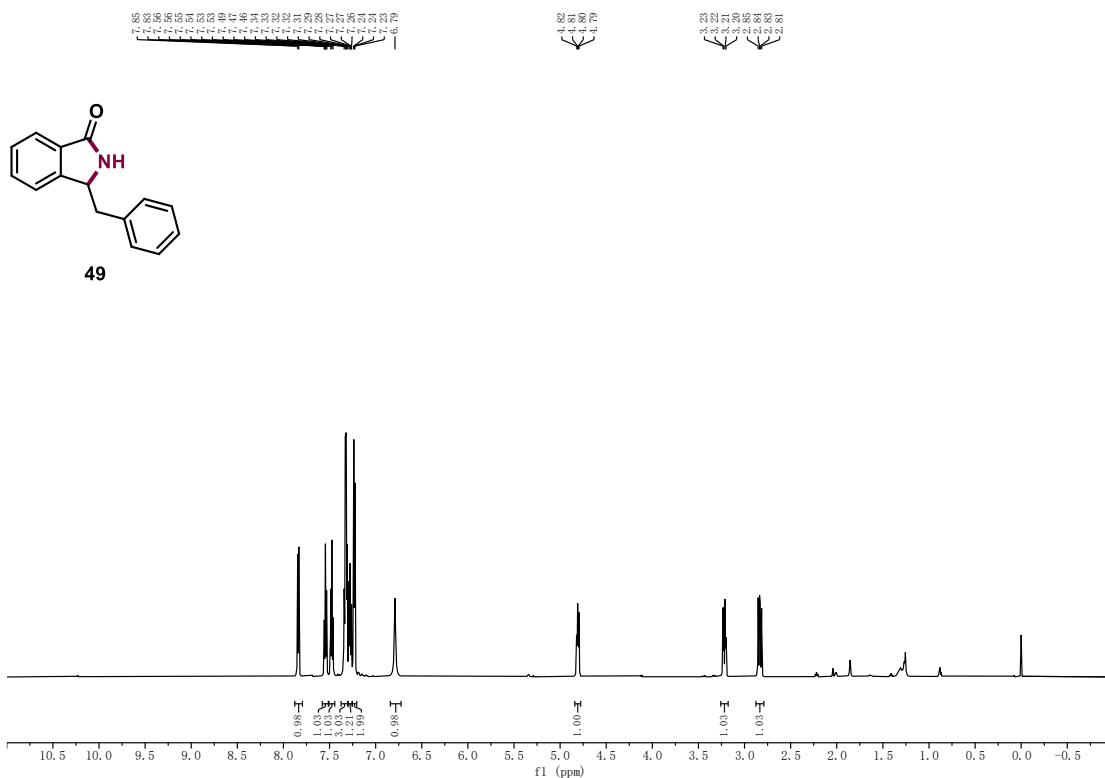
Supplementary Figure 256 ^{13}C NMR spectrum of compound 47

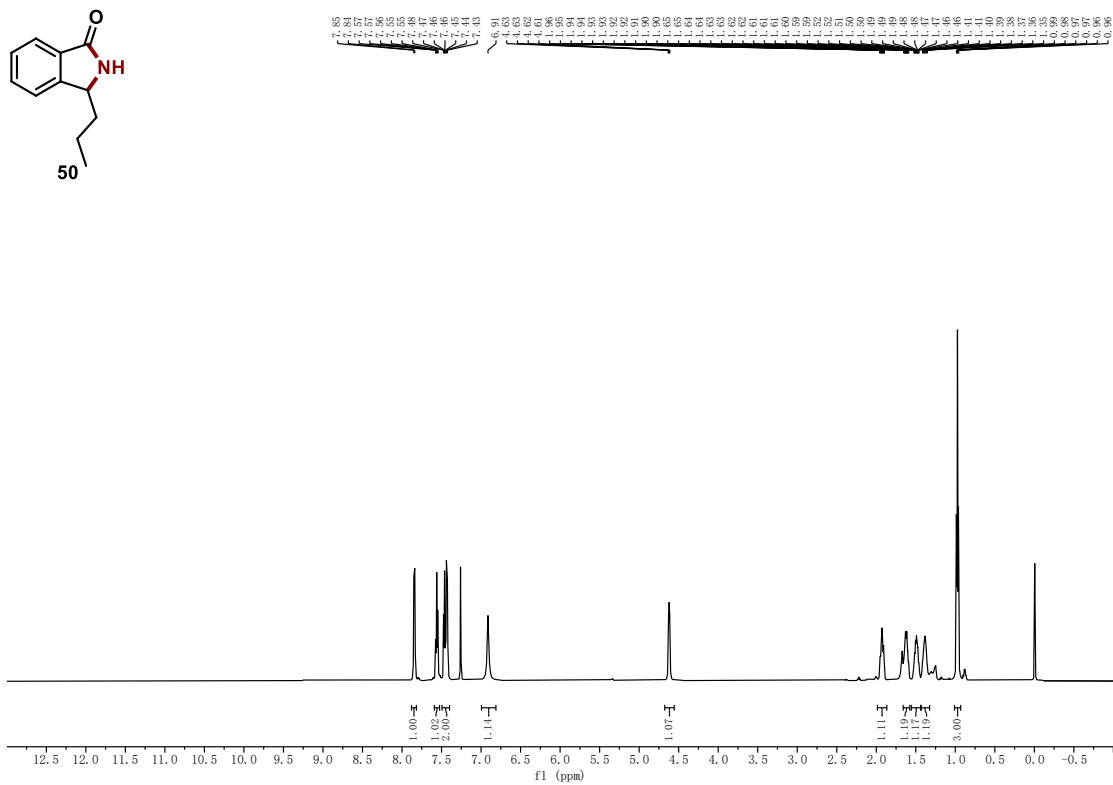


Supplementary Figure 257 ¹H NMR spectrum of compound 48

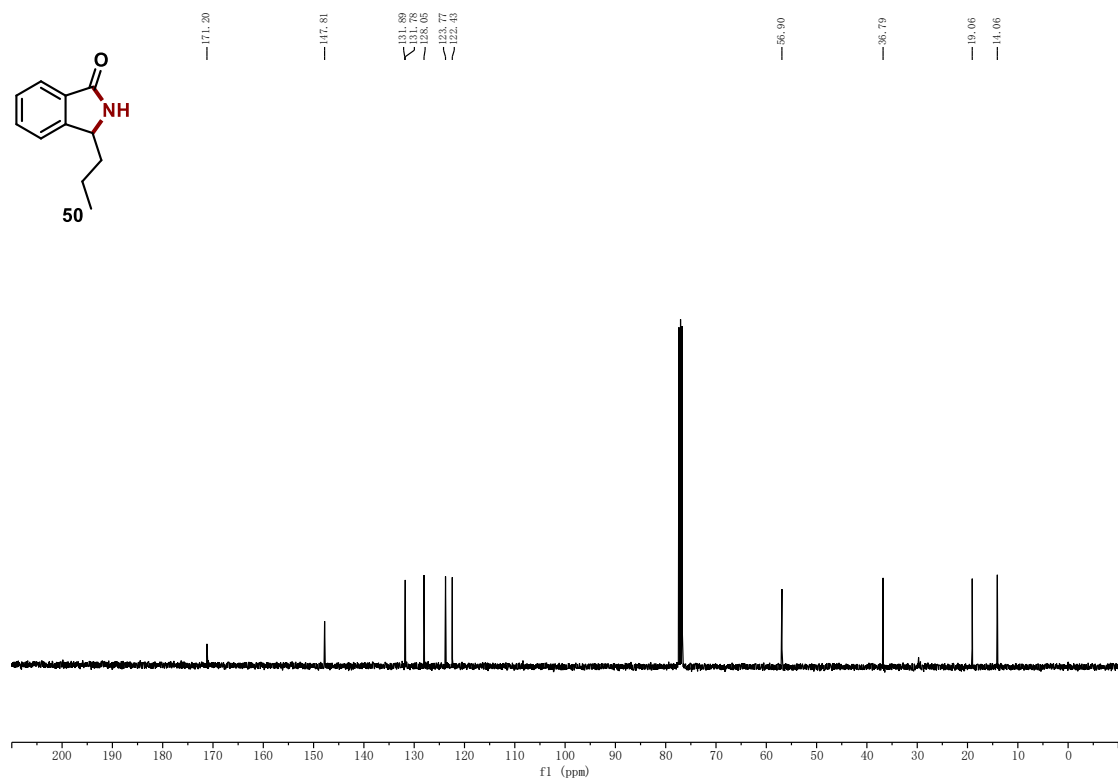


Supplementary Figure 258 ¹³C NMR spectrum of compound 48

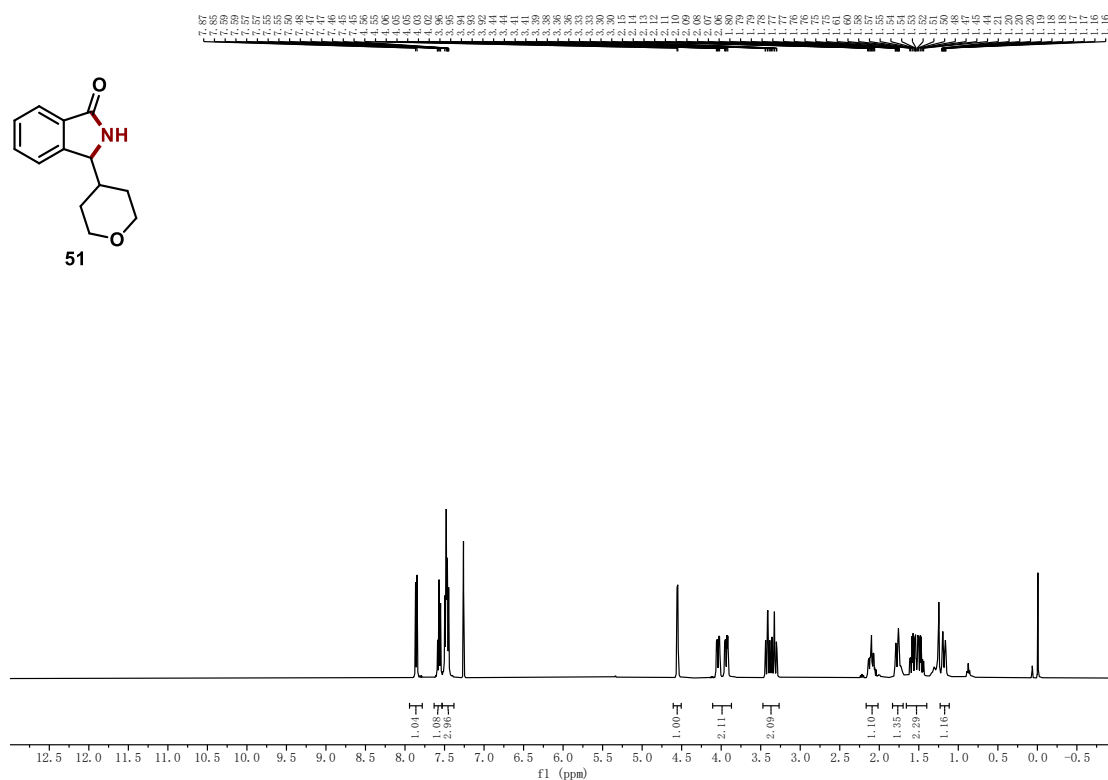
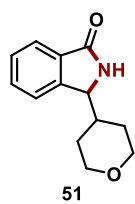




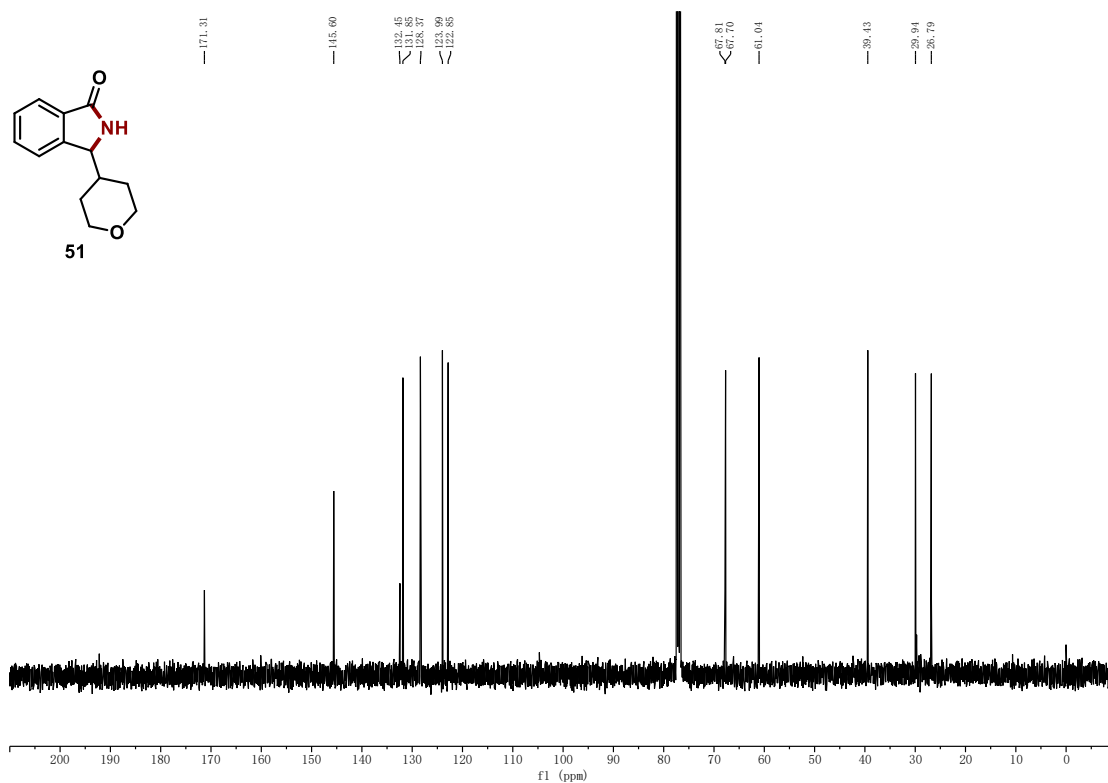
Supplementary Figure 261 ^1H NMR spectrum of compound 50



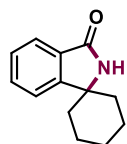
Supplementary Figure 262 ^{13}C NMR spectrum of compound 50



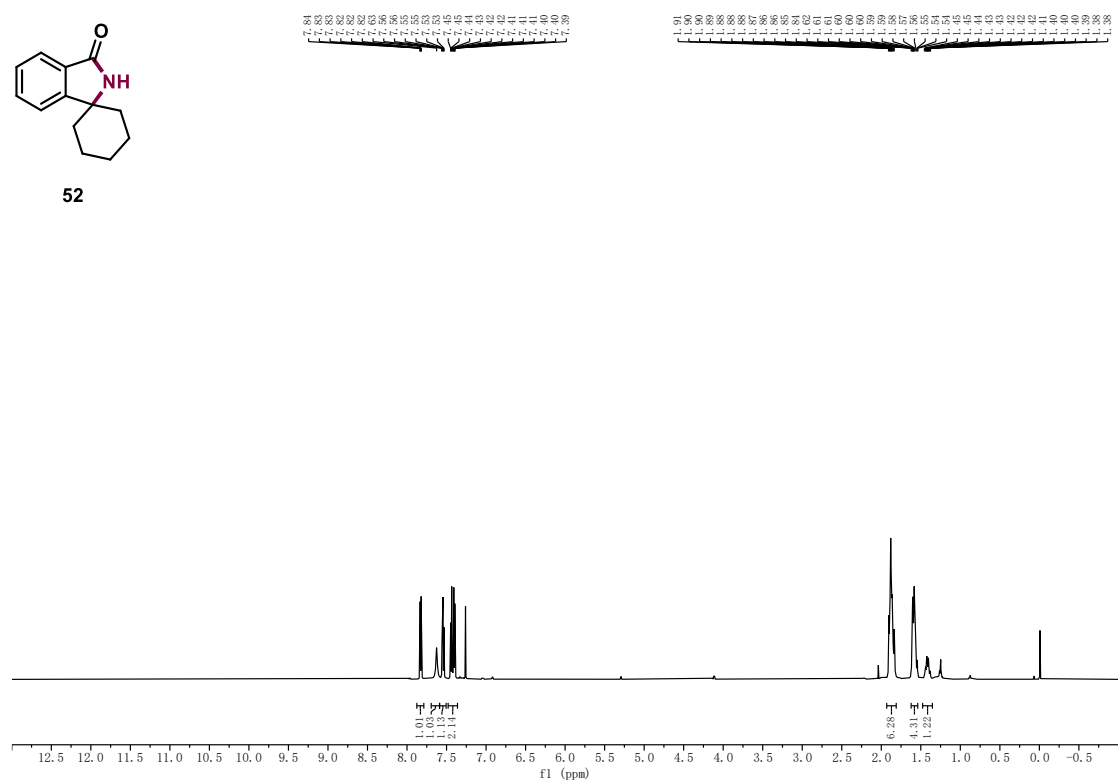
Supplementary Figure 263 ¹H NMR spectrum of compound 51



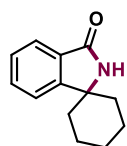
Supplementary Figure 264 ¹³C NMR spectrum of compound 51



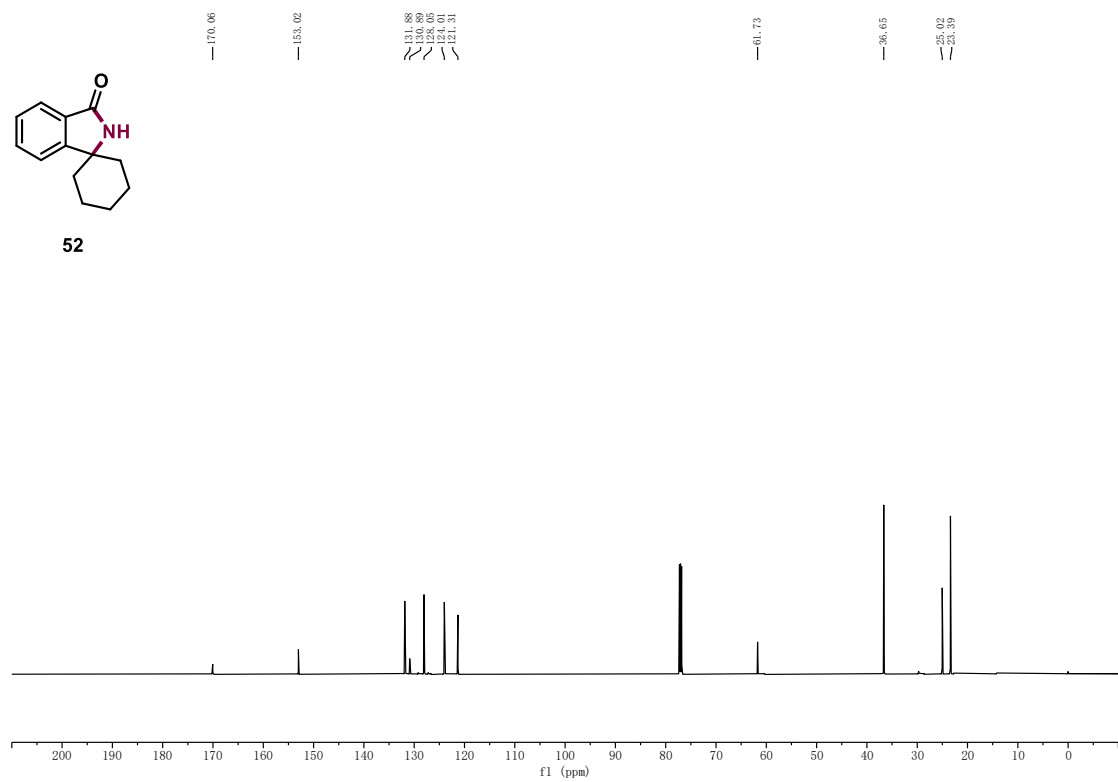
52



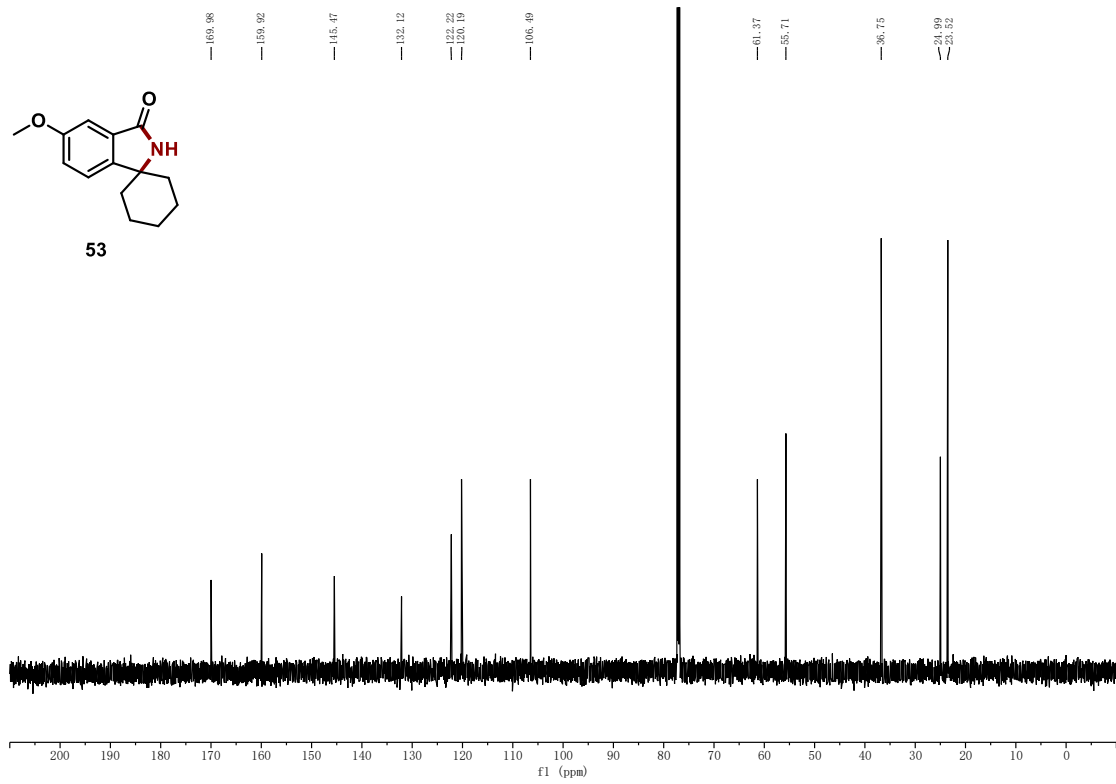
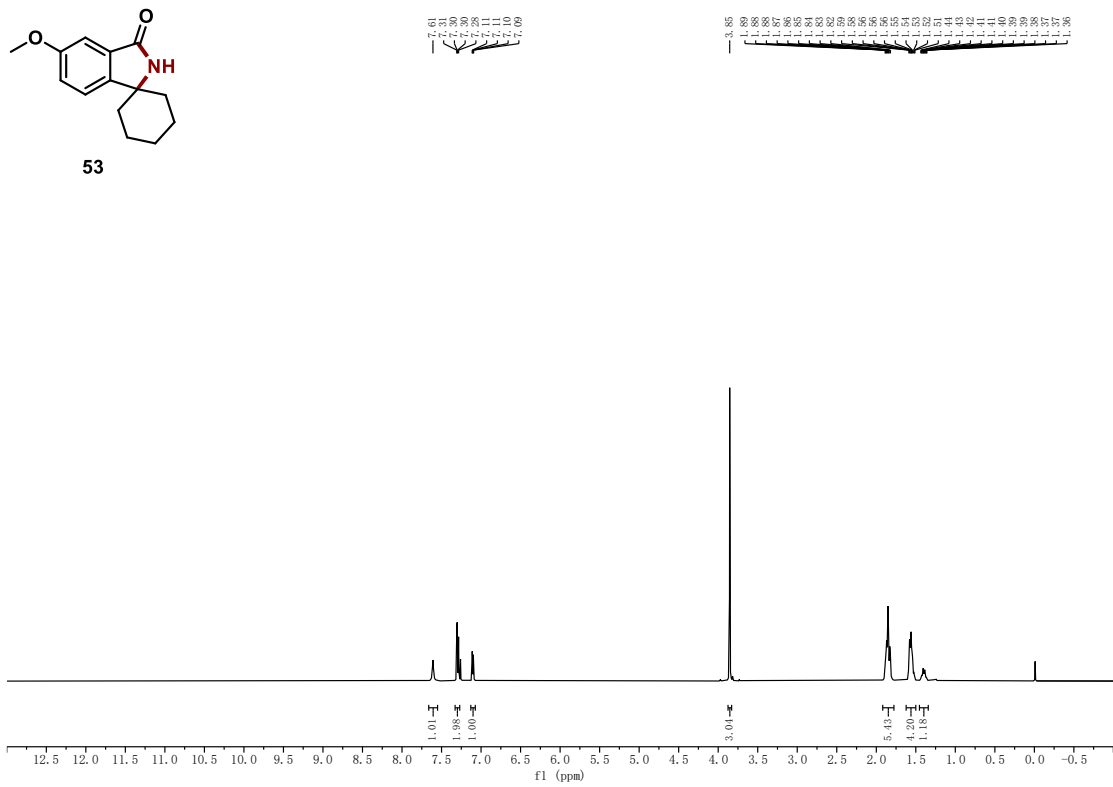
Supplementary Figure 265 ¹H NMR spectrum of compound 52

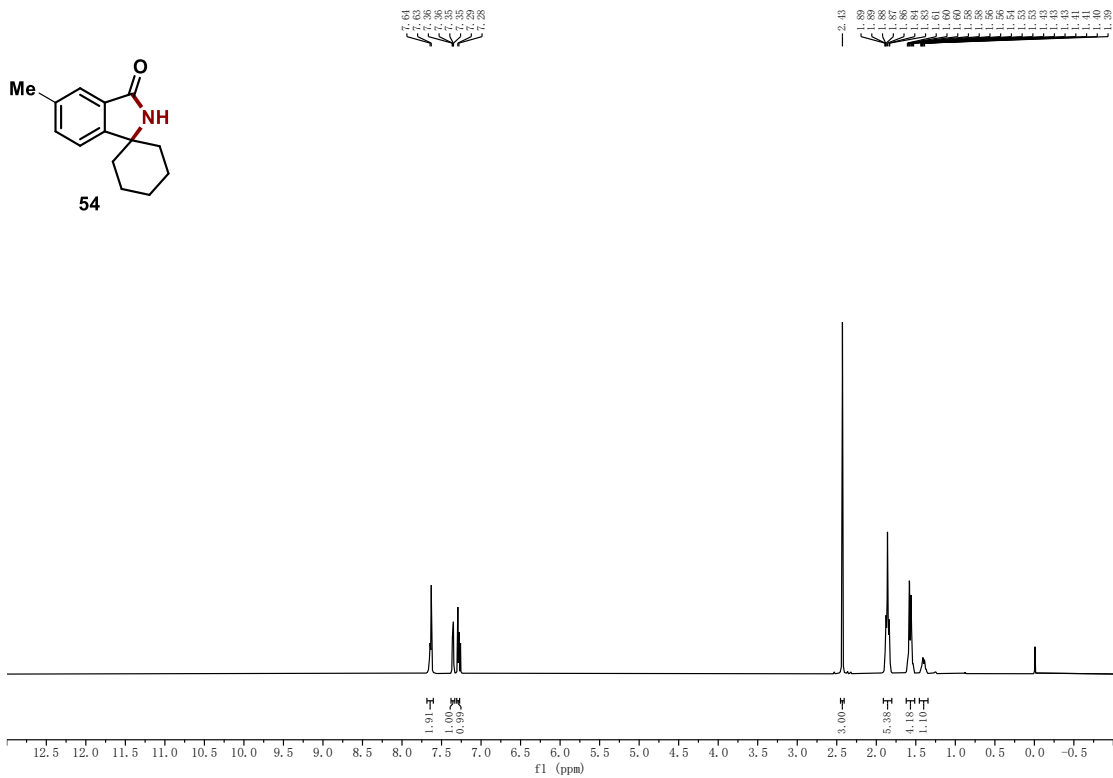


52

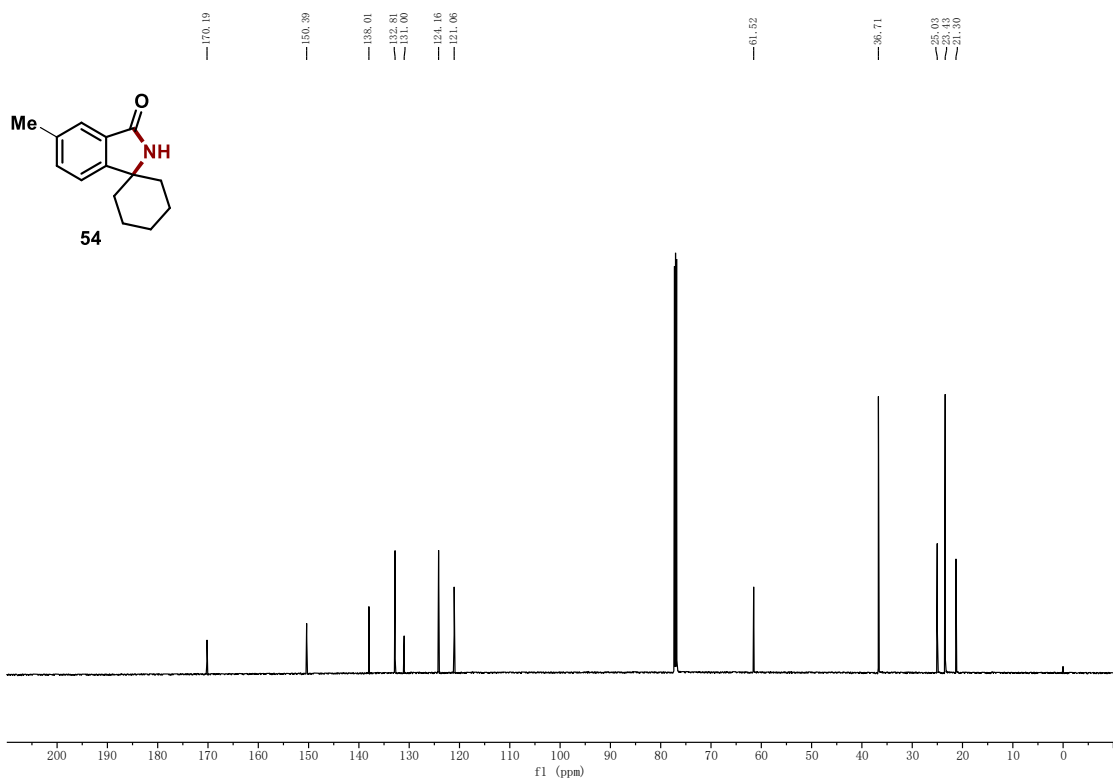


Supplementary Figure 266 ¹³C NMR spectrum of compound 52

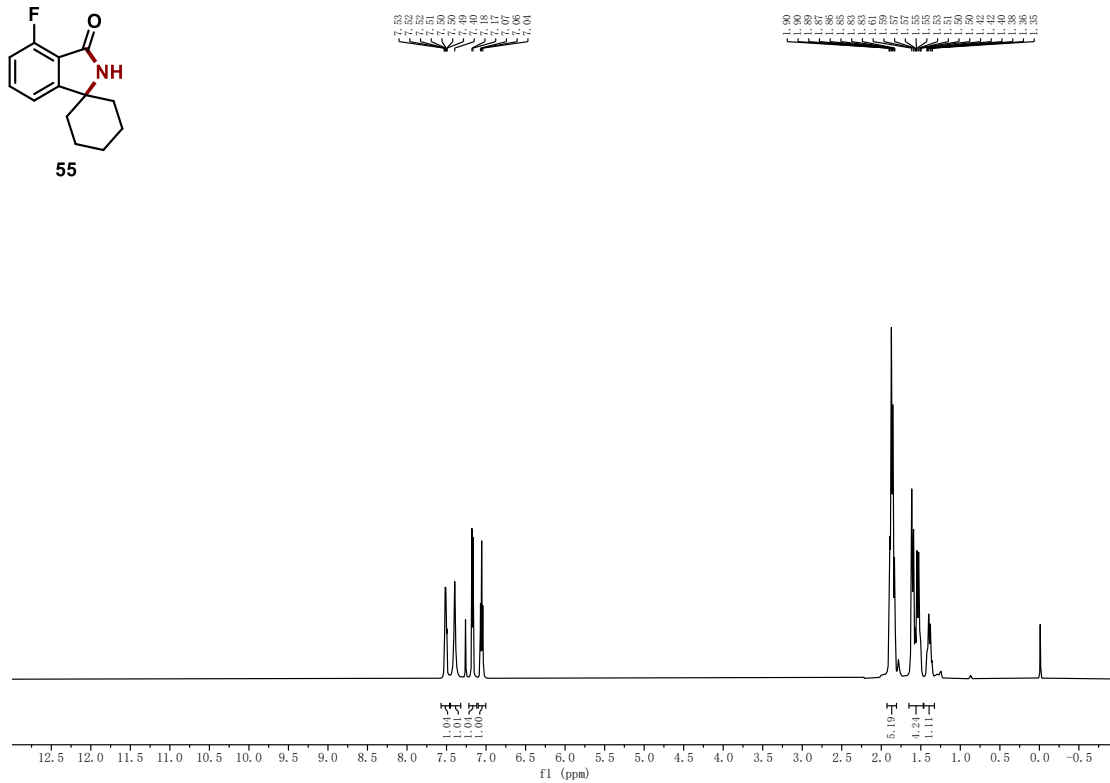
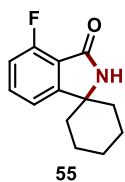




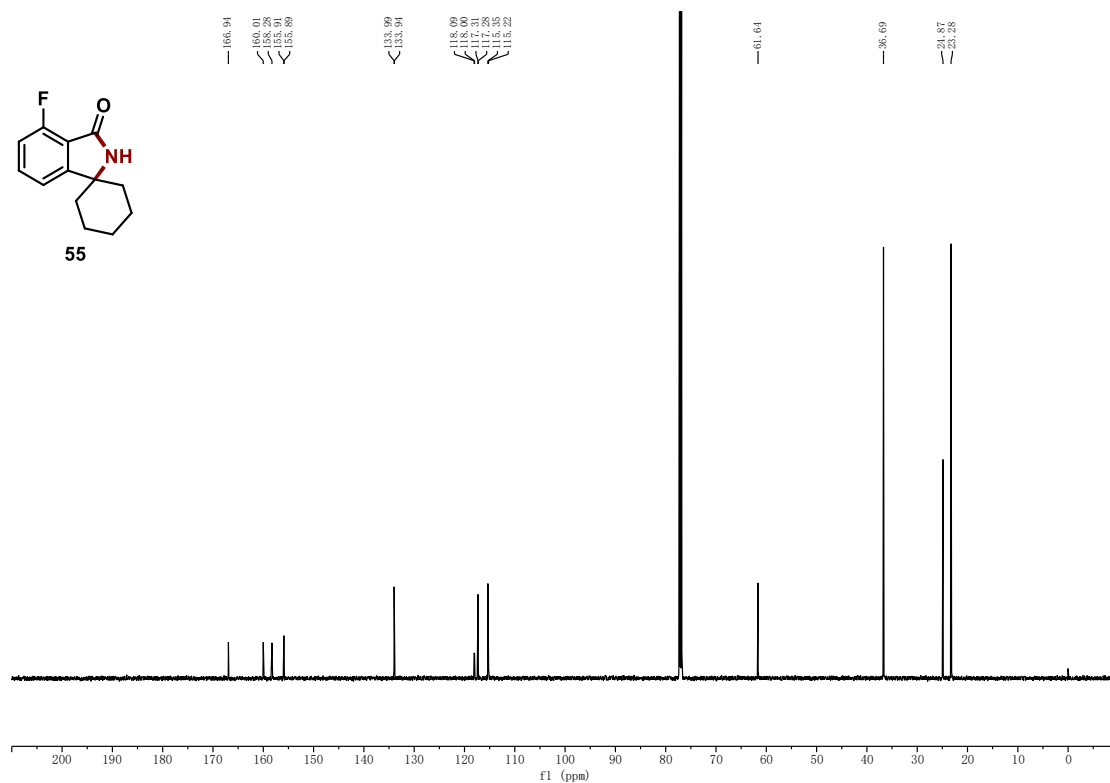
Supplementary Figure 269 ^1H NMR spectrum of compound **54**



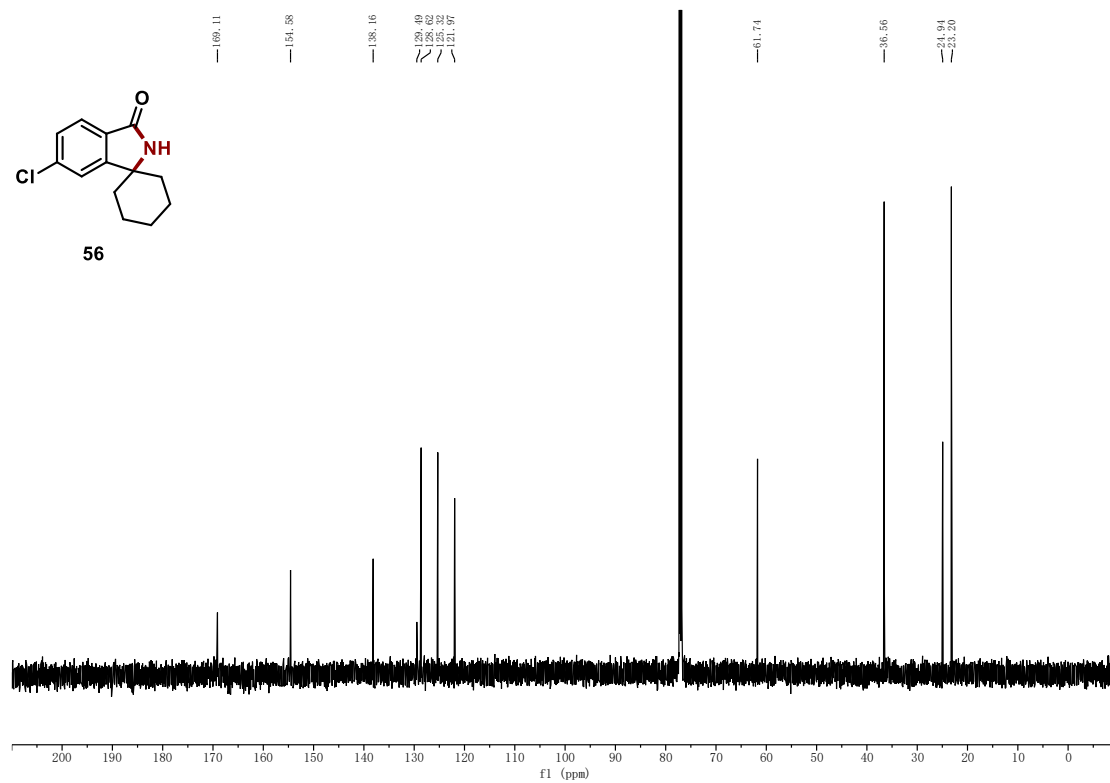
Supplementary Figure 270 ^{13}C NMR spectrum of compound **54**



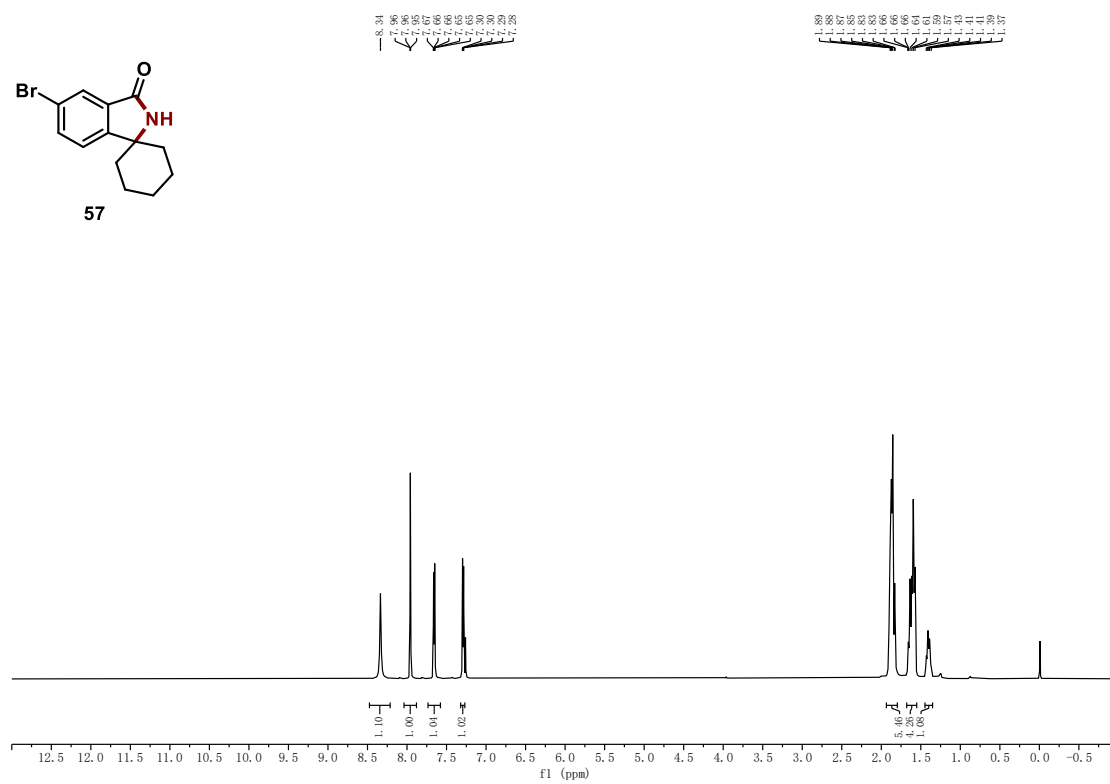
Supplementary Figure 271 ^1H NMR spectrum of compound **55**

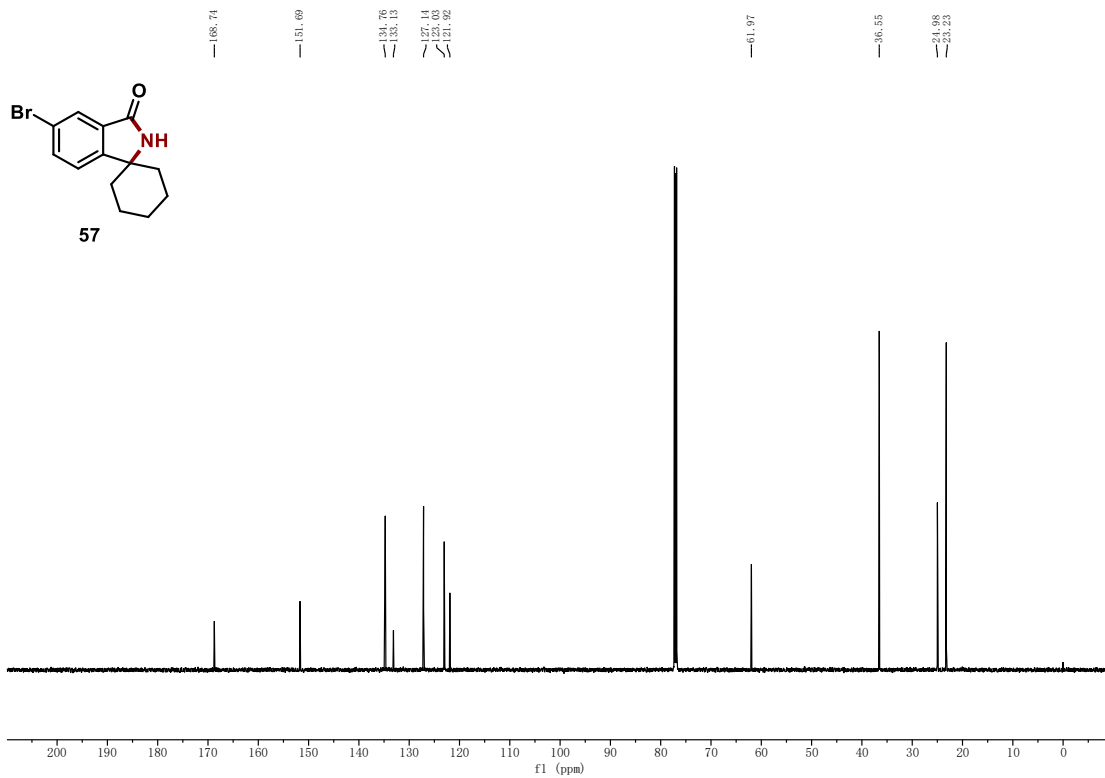


Supplementary Figure 272 ^{13}C NMR spectrum of compound **55**

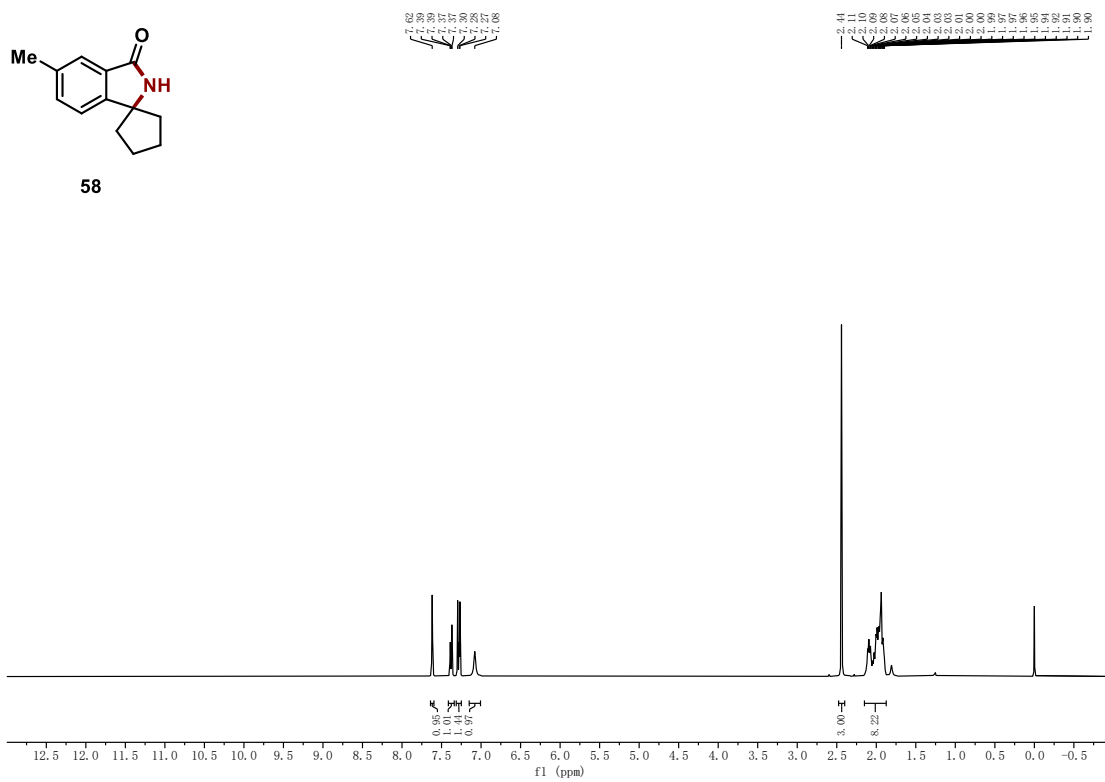


Supplementary Figure 275 ¹³C NMR spectrum of compound 56

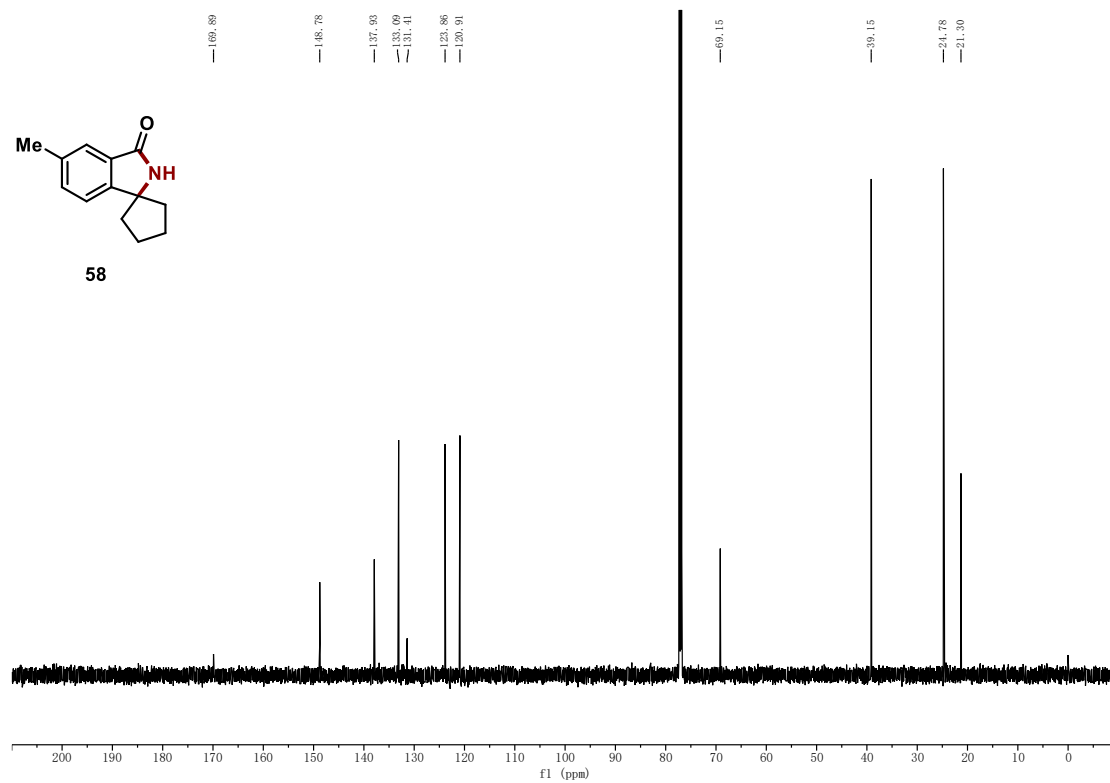




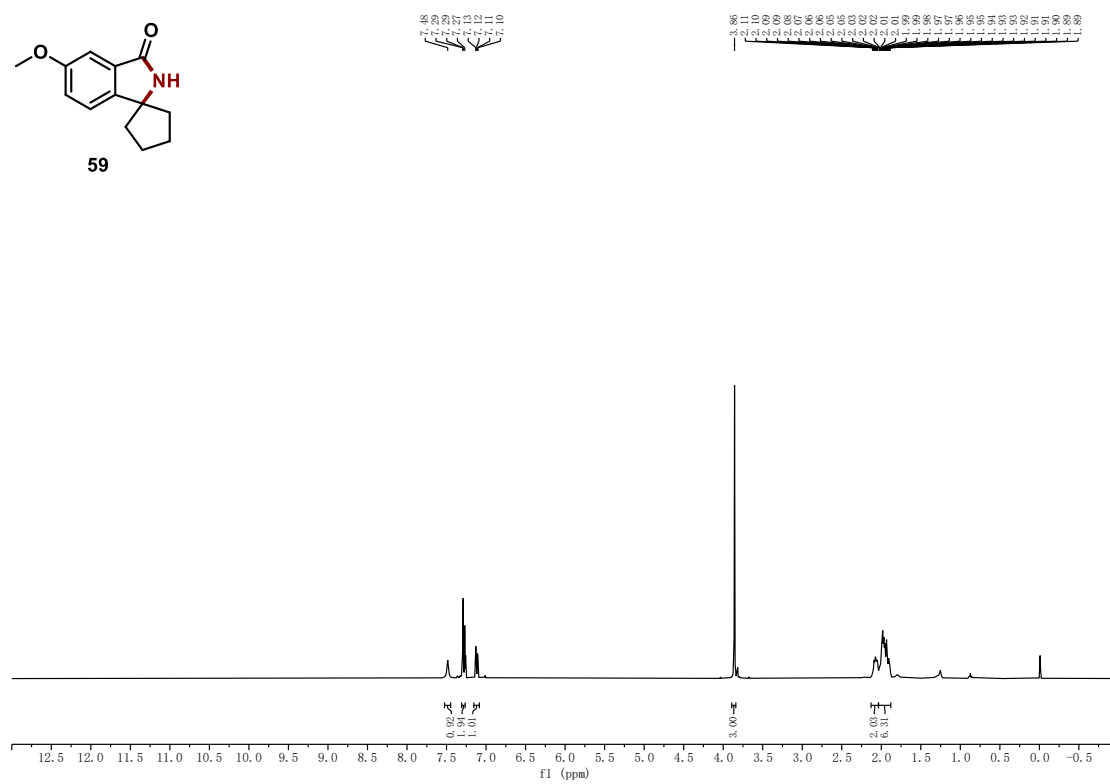
Supplementary Figure 277 ^{13}C NMR spectrum of compound **57**



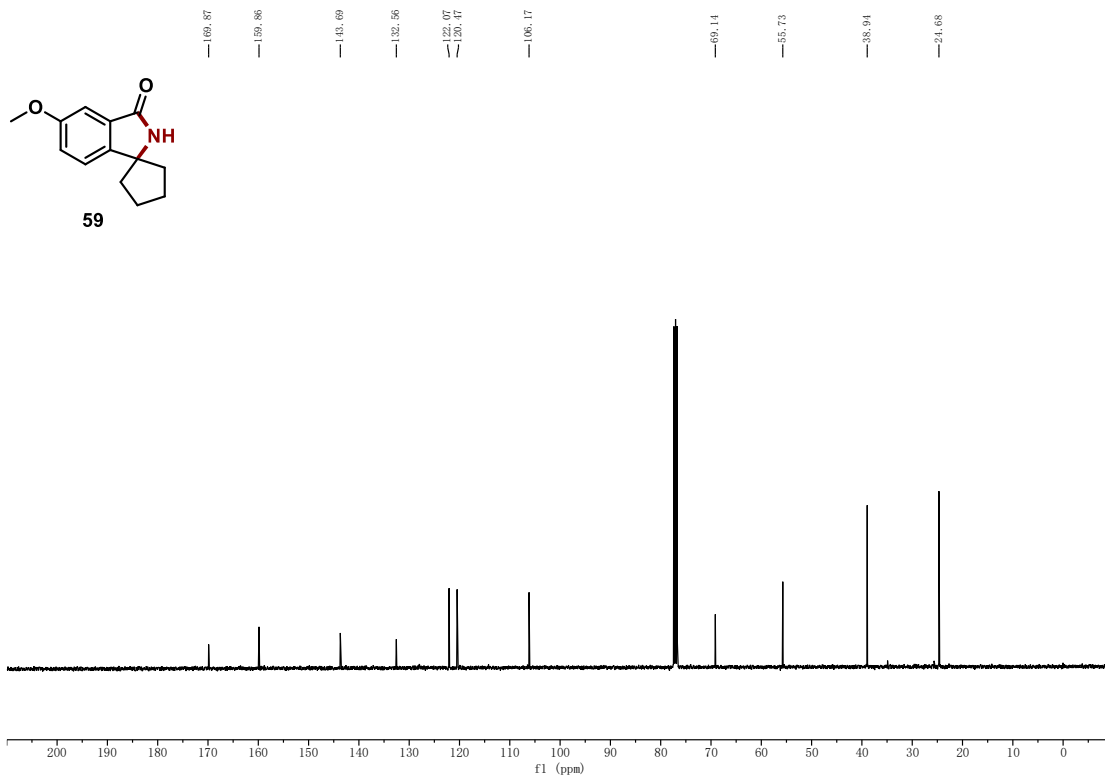
Supplementary Figure 278 ^1H NMR spectrum of compound **58**



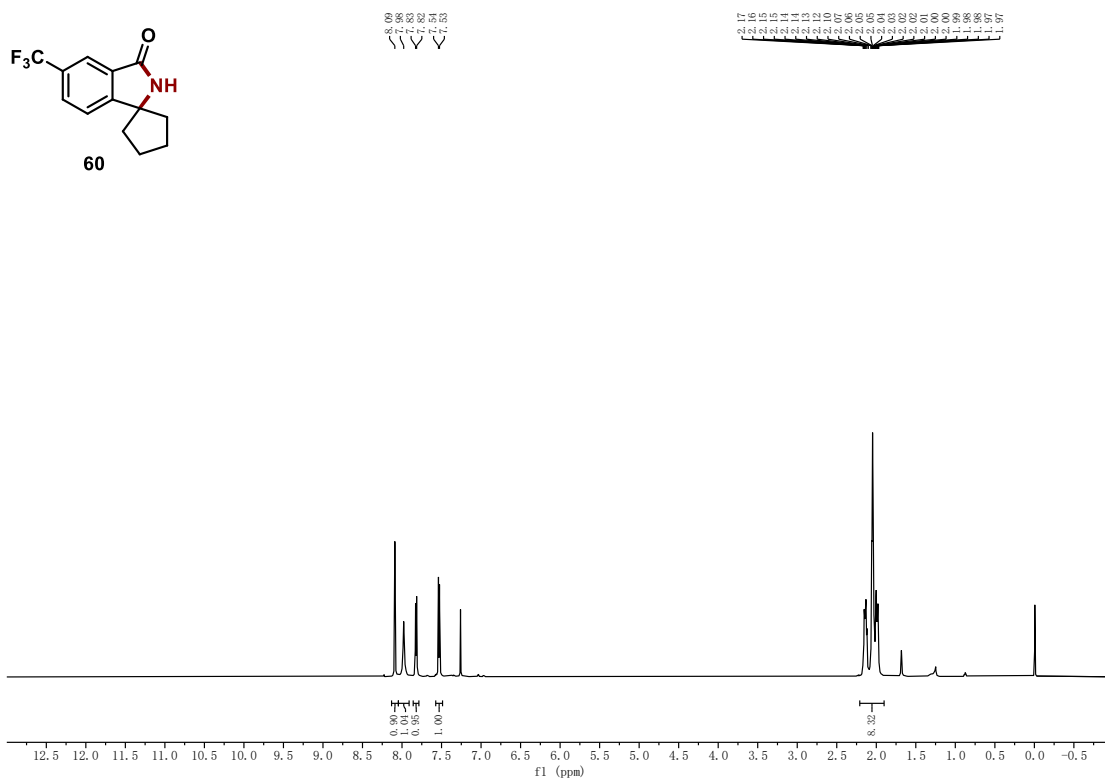
Supplementary Figure 279 ^{13}C NMR spectrum of compound **58**



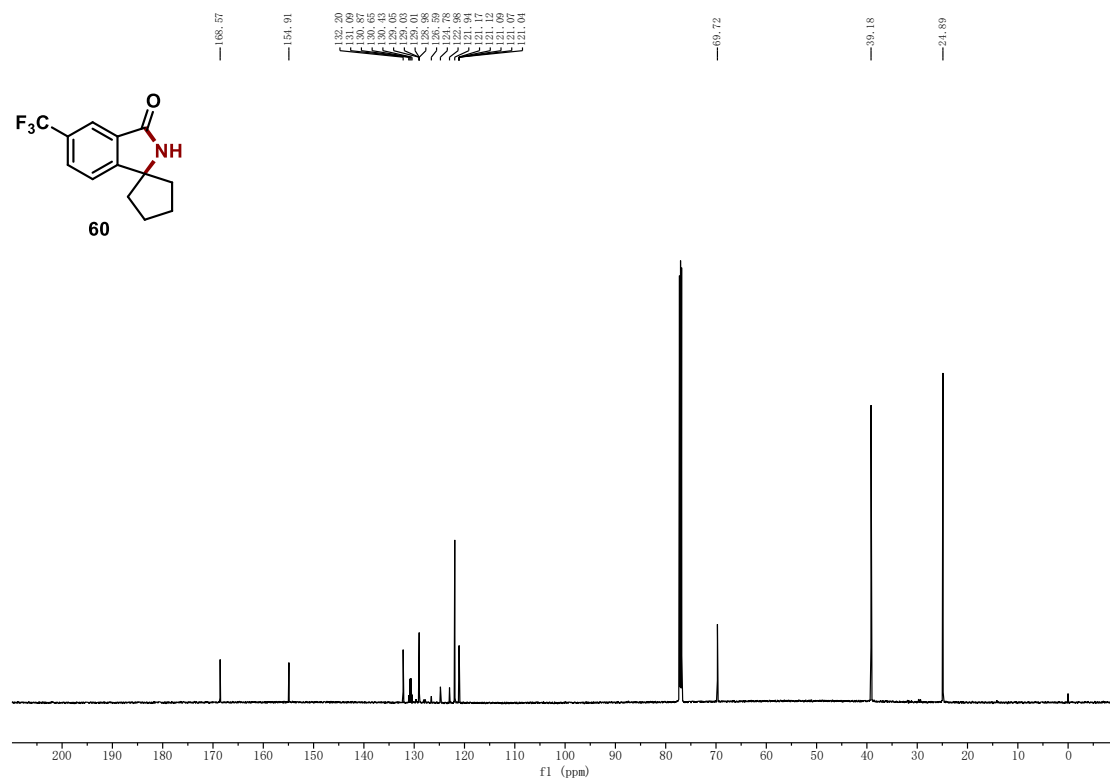
Supplementary Figure 280 ^1H NMR spectrum of compound **59**



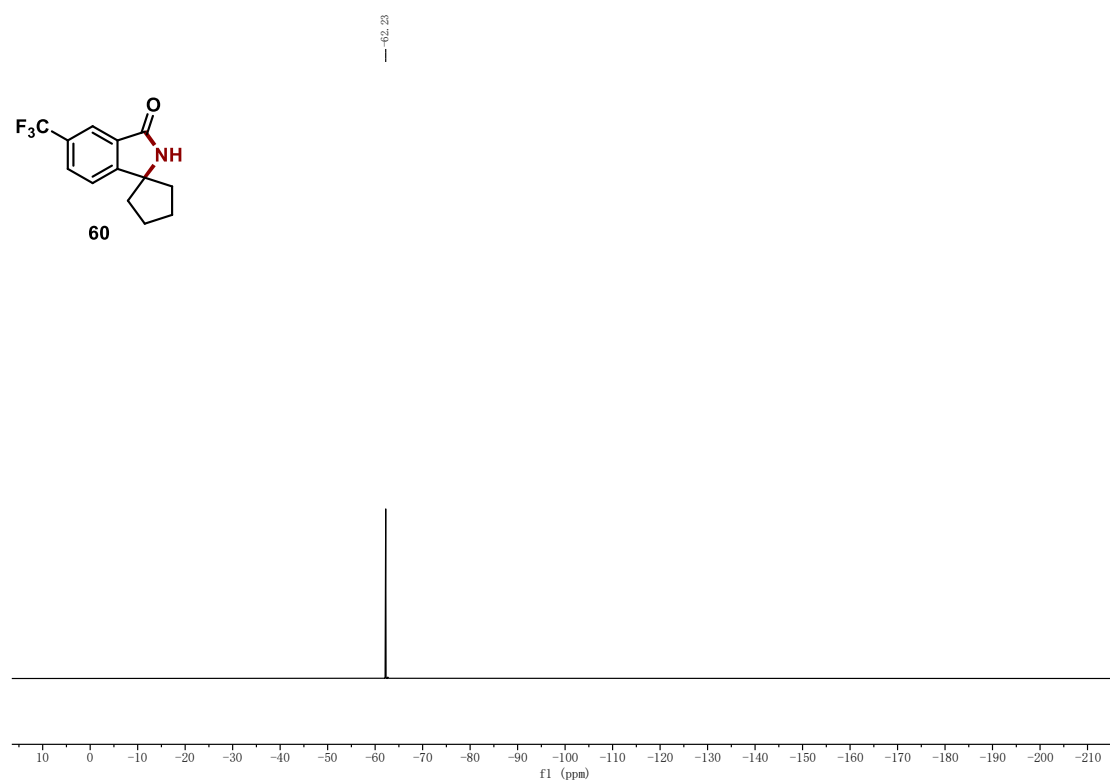
Supplementary Figure 281 ^{13}C NMR spectrum of compound **59**



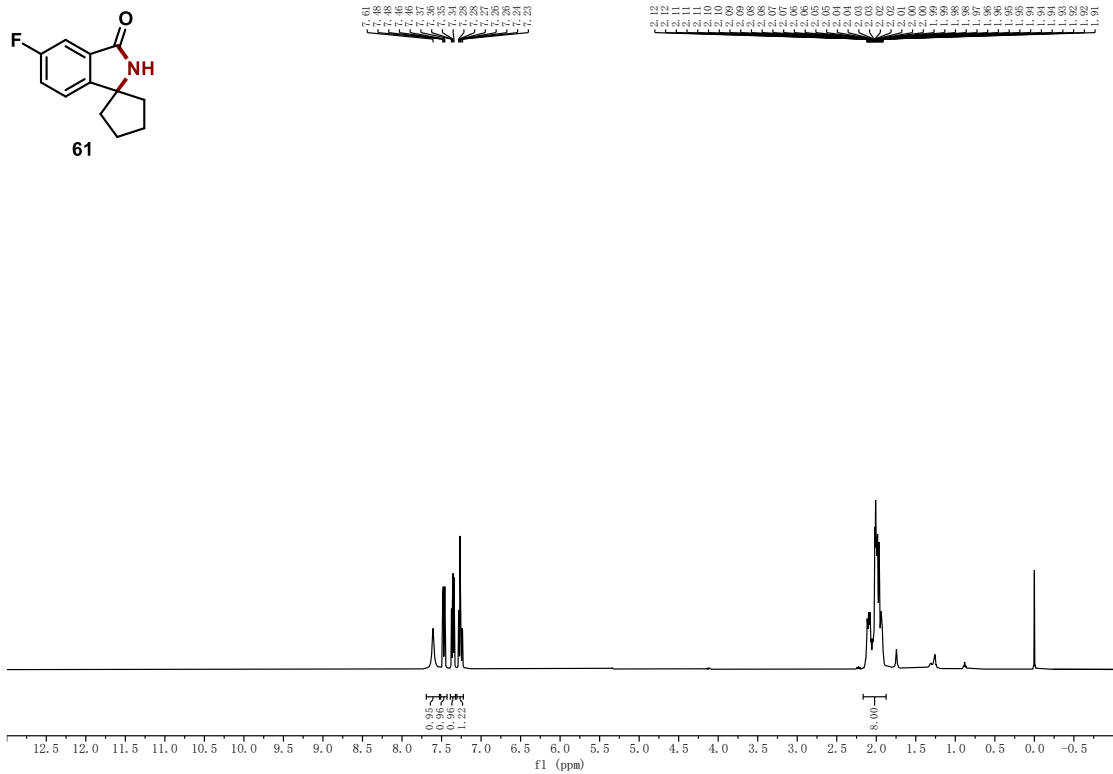
Supplementary Figure 282 ^1H NMR spectrum of compound **60**



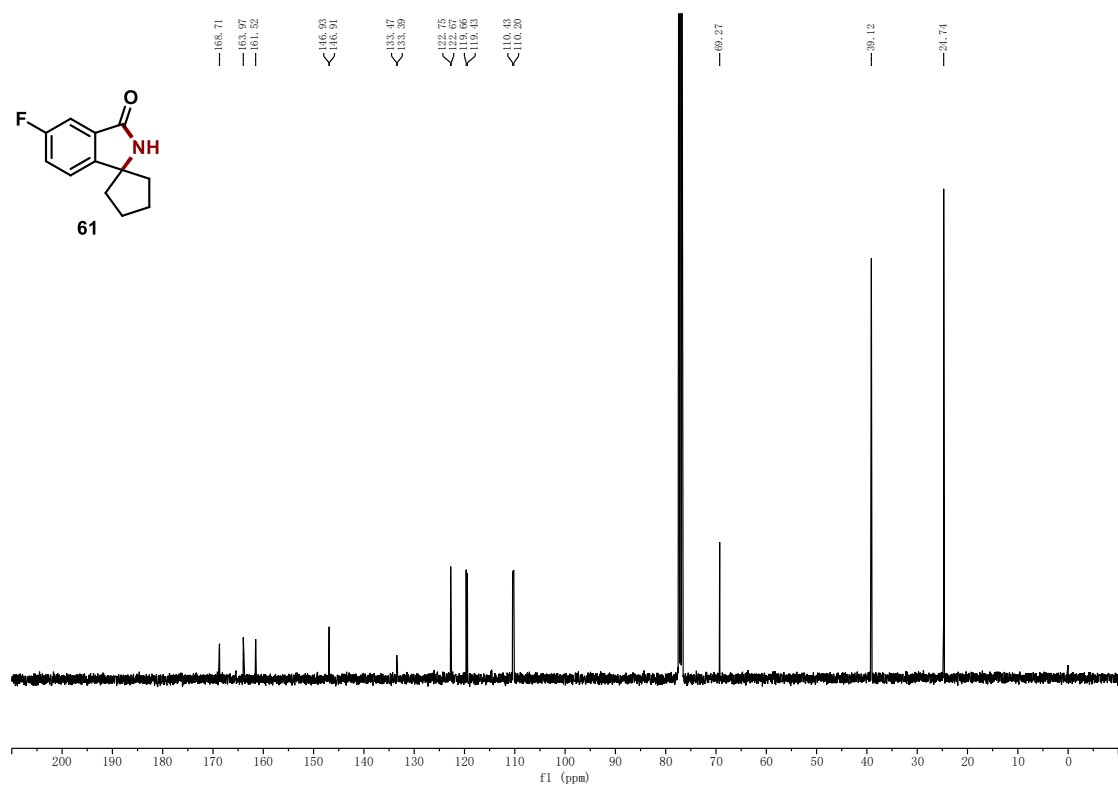
Supplementary Figure 283 ^{13}C NMR spectrum of compound **60**



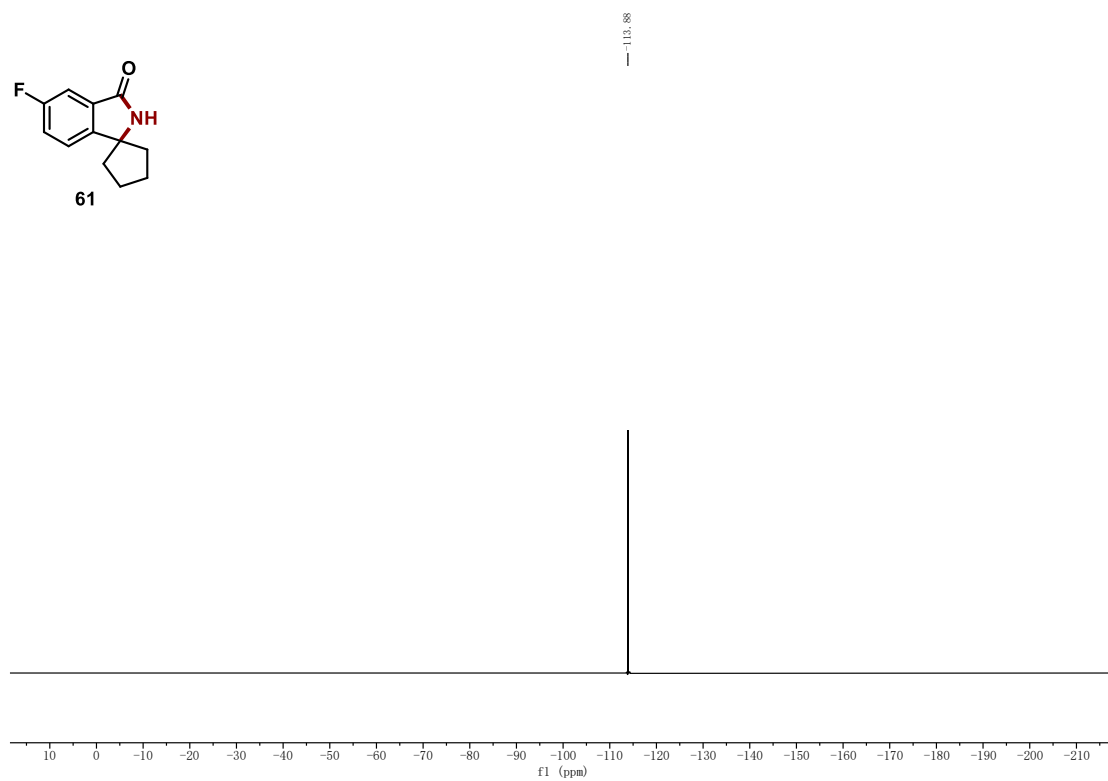
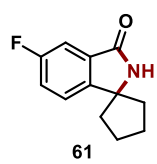
Supplementary Figure 284 ^{19}F NMR spectrum of compound **60**



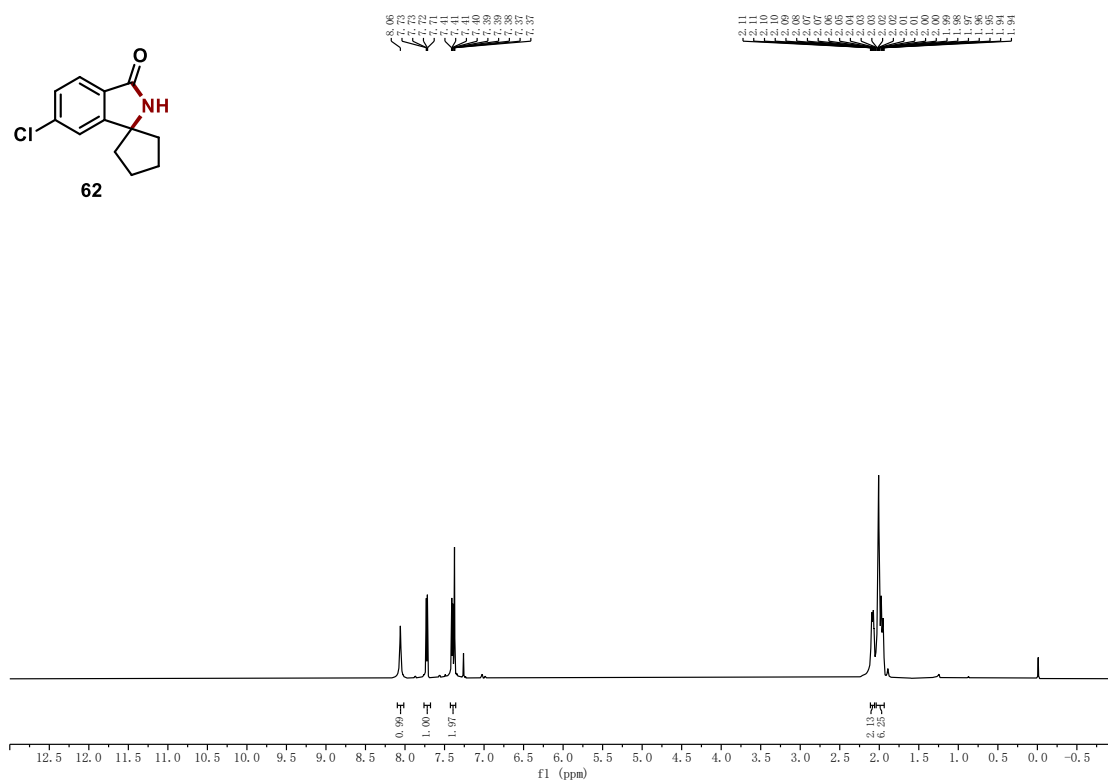
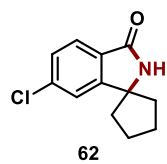
Supplementary Figure 285 ¹H NMR spectrum of compound 61



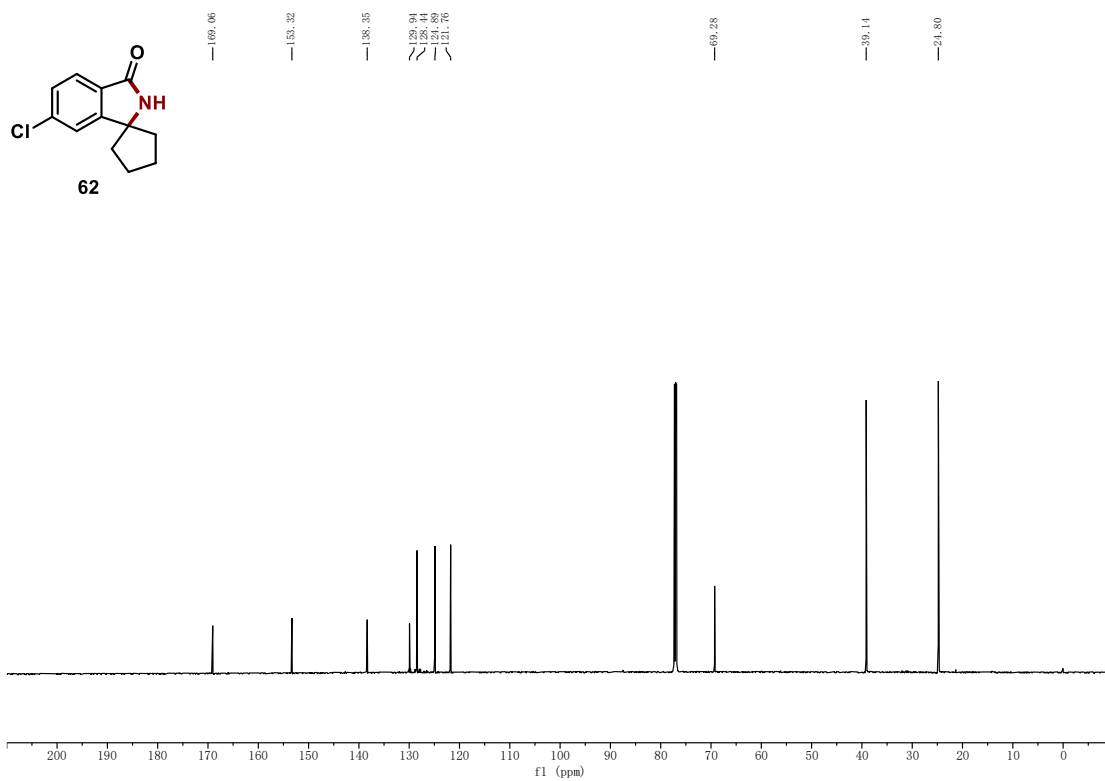
Supplementary Figure 286 ¹³C NMR spectrum of compound 61



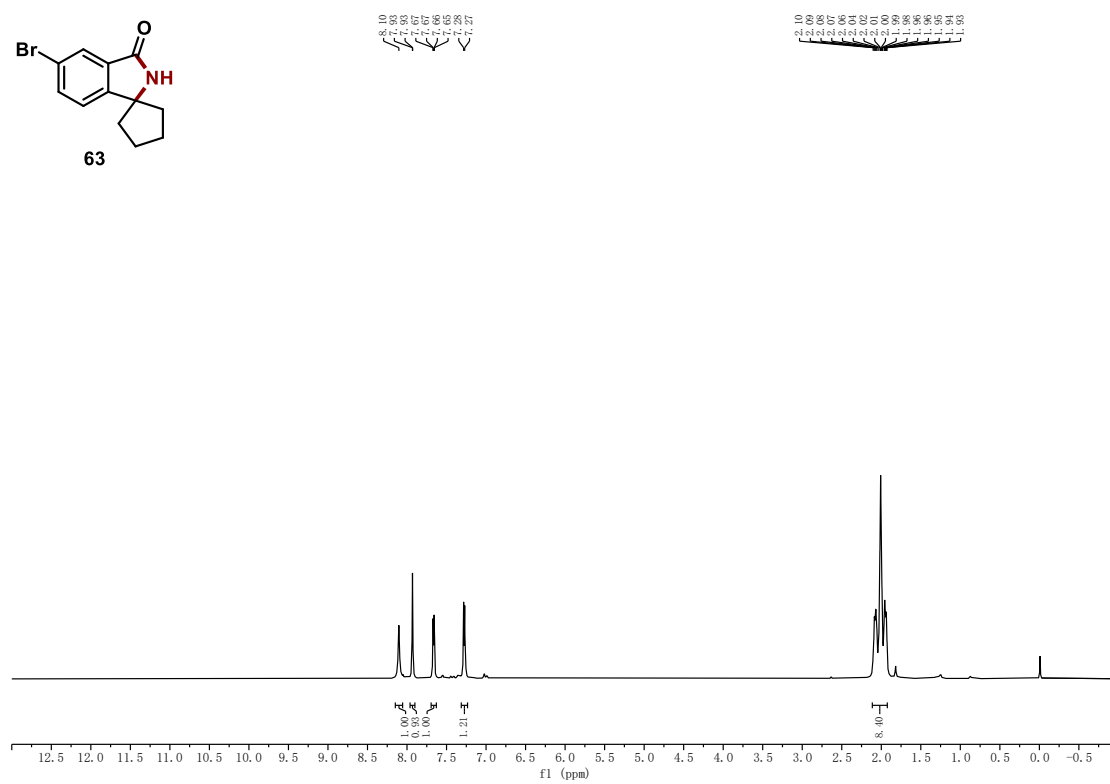
Supplementary Figure 287 ^{19}F NMR spectrum of compound **61**



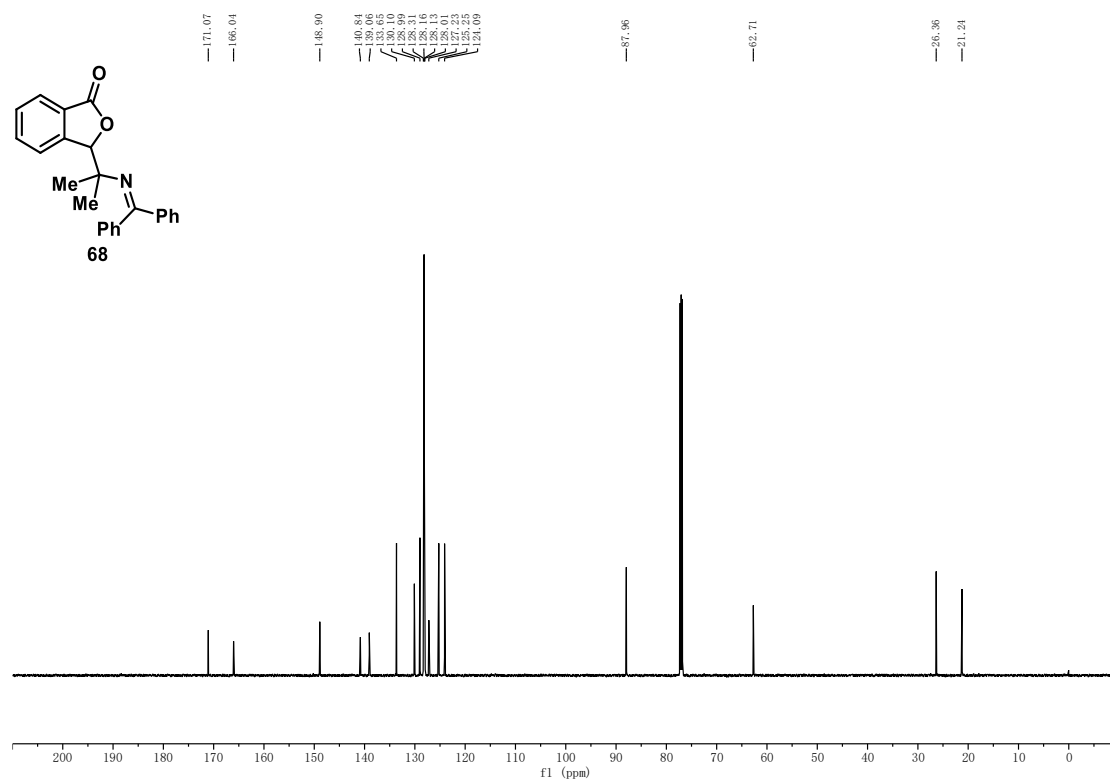
Supplementary Figure 288 ^1H NMR spectrum of compound **62**



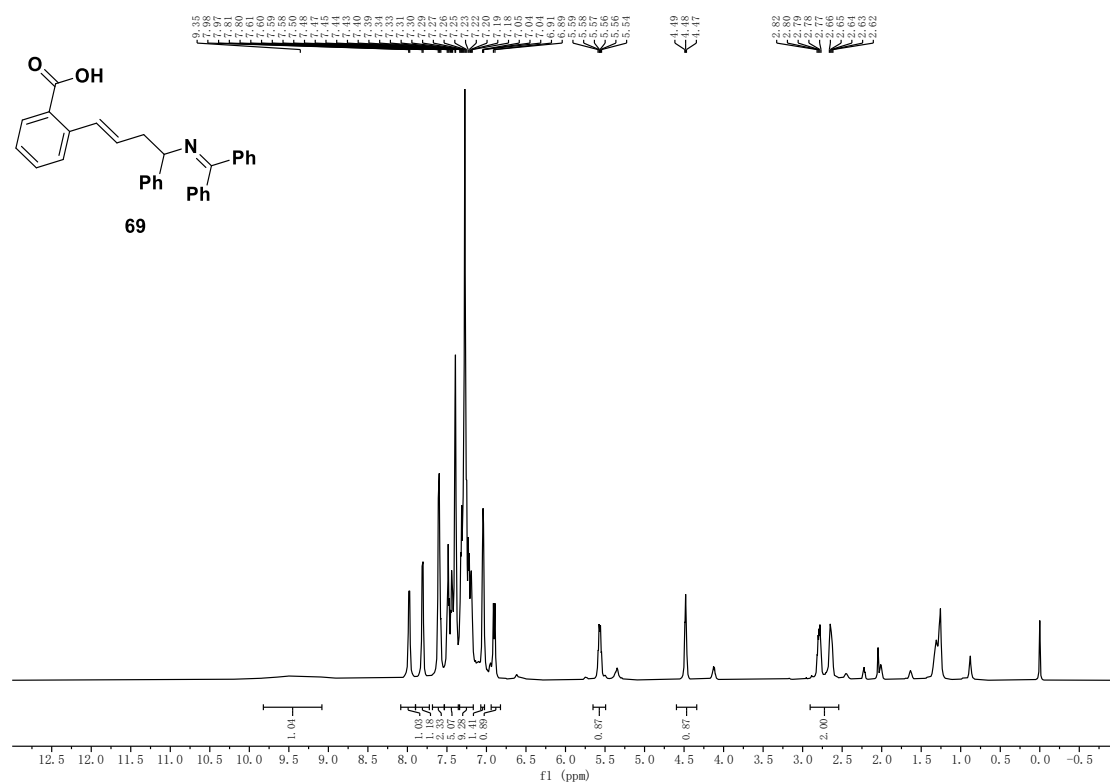
Supplementary Figure 289 ¹³C NMR spectrum of compound **62**



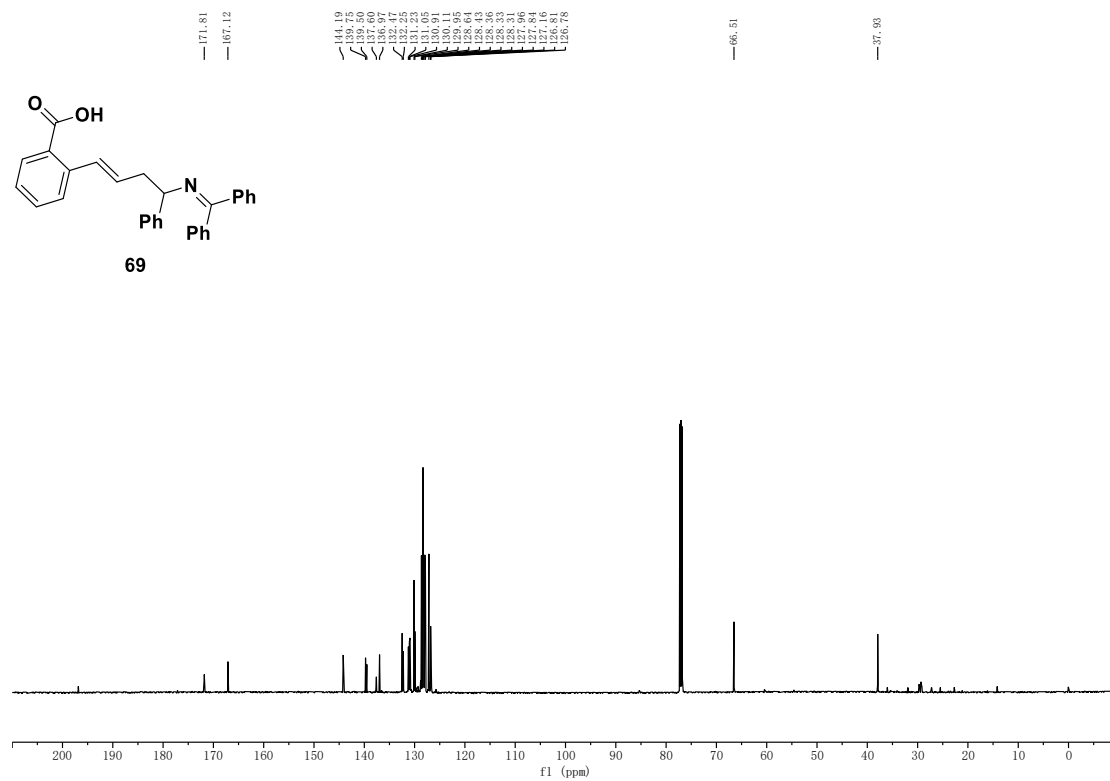
Supplementary Figure 290 ¹H NMR spectrum of compound **63**



Supplementary Figure 293 ¹³C NMR spectrum of compound **68**



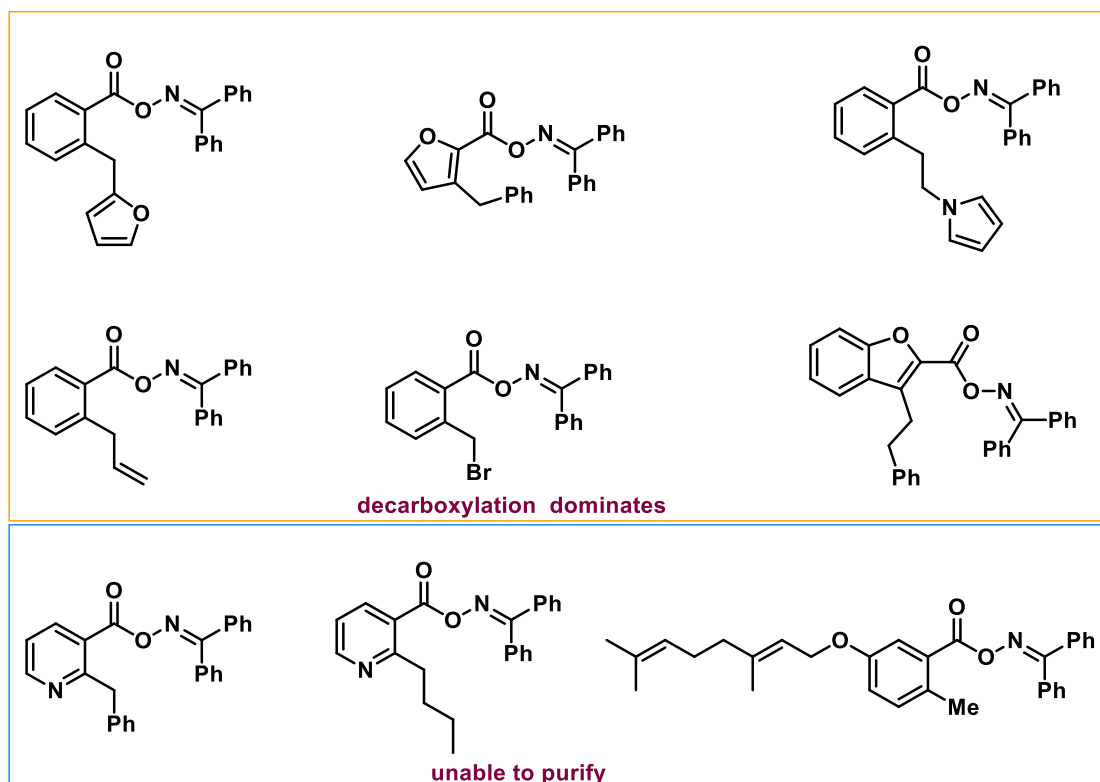
Supplementary Figure 294 ¹H NMR spectrum of compound **69**



Supplementary Figure 295 ^{13}C NMR spectrum of compound **69**

3.4 Unsuccessful Substrates

The unsuccessful substrates for the desired γ C-H Amination reactions are listed in Supplementary Figure 296.



Supplementary Figure 296. Unsuccessful substrates for the desired γ C-H amination reactions.

4 Supplementary References

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