

Hypervalent iodine chemistry with a mechanochemical twist

Sayad Doobary,* Miguel M. de Vries Ibáñez and Berit Olofsson *

Berit.Olofsson@su.se

SUPPORTING INFORMATION

CONTENTS TABLE

1. General experimental	3
2. Mechanochemical setup	4
3. Optimisation of different types of reactions	5
3.1 <i>O</i> -ARYLATION OPTIMISATION	5
3.2 <i>S</i> -ARYLATION OPTIMISATION	8
3.4 <i>C</i> -ARYLATION OPTIMISATION	10
3.5 <i>S</i> -VINYLATION OPTIMISATION	11
3.6 <i>C</i> -VINYLATION OPTIMISATION	12
3.7 CATALYTIC TOSYLOXYLATION OPTIMISATION	13
4. Substrates	15
4.1 IODINE(III) REAGENTS	15
4.2 NUCLEOPHILES	16
5. Products	17
5.1 PRODUCT COMPOUND TABLE	17
5.2 <i>O</i> -ARYLATION	22
5.3 <i>S</i> -ARYLATION	30
5.4 <i>C</i> -ARYLATION	35
5.5 <i>S</i> -VINYLATION	38
5.6 <i>C</i> -VINYLATION	42
5.7 CATALYTIC TOSYLOXYLATION	44
6. Benchmarking studies	47
6.1 SOLUTION-PHASE REACTIONS	47
6.2 SOLUTION-PHASE REACTION CONDITIONS HIGHER SOLVENT CONCENTRATION	48
6.3 MECHANO-CHEMICAL REACTION CONDITIONS WITH MICROWAVE VIALS	48
7. E-factor calculations	51
8. References	73
9. NMR Spectra	76

1. General experimental

The used reagents were bought from commercial suppliers and used as received, unless noted otherwise. Moisture and air sensitive reactions were carried out under argon or nitrogen environment using standard Schlenk techniques. Reactions performed above the boiling point of the solvent(s) were performed in pressure-stable microwave vials or using a standard condenser. Solvents were obtained as P.A. grade and dried using a VAC solvent purification system. *m*-CPBA (Aldrich, 77 % active oxidant) was dried under vacuum for 4 hours, after which the amount of active oxidant was determined through an iodometric titration.¹ Thin layer chromatography (TLC) was performed using TLC Silica gel 60 F254 plates (Merck) and visualized using UV-light and all TLC plates were stained with potassium permanganate stain. Purification of the products was conducted by flash column chromatography on SiO₂ purchased from Aldrich (technical grade, 60 Å pore size, 230-400 mesh). Melting points were measured using a STUART SMP3 and are reported uncorrected. NMR measurements were conducted using a 400 MHz Bruker AVANCE II with a BBO probe at 298 K unless otherwise stated. Chemical shifts (δ) are reported in parts per million (ppm) and referenced CDCl₃ (¹H: 7.26 ppm; ¹³C: 77.0 ppm), DMSO-*d*₆ (¹H: 2.50 ppm; ¹³C: 39.5 ppm) or MeOD-*d*₆ (¹H: 3.31 ppm; ¹³C: 49.0 ppm). Coupling constants (*J*) are given in Hertz (Hz) and refer to apparent multiplicities (s = singlet, d = doublet, t = triplet, q = quartet, pent = pentet, m = multiplet, br = broad signal, and combinations thereof for example, dd = doublet of doublets.) HRMS spectra were measured on a Bruker microTOF with electron spray ionization (ESI). Retsch MM500 vario and Retsch MM2 were utilised for all ball milling reactions. Ball milling vessels were purchased as standard from Retsch. 1-Butyl-3-methylimidazolium tetrafluoroborate abbreviated to BMIM•BF₄. Petroleum ether (40-60) is abbreviated as PE throughout this supporting information.

The frequencies utilised under different protocols vary from 25 Hz to 35 Hz. This is because we aimed to use the lowest possible frequency to conduct our reactions in a timely manner to preserve the lifetime of the equipment.

Organic bases (e.g. DBU and TMG) can be utilised in most of the transformations, but it is established routine in our lab to avoid the use of liquid, organic bases due to the increased reactivity under mechanochemical conditions, often resulting in impure products and more difficult purifications.

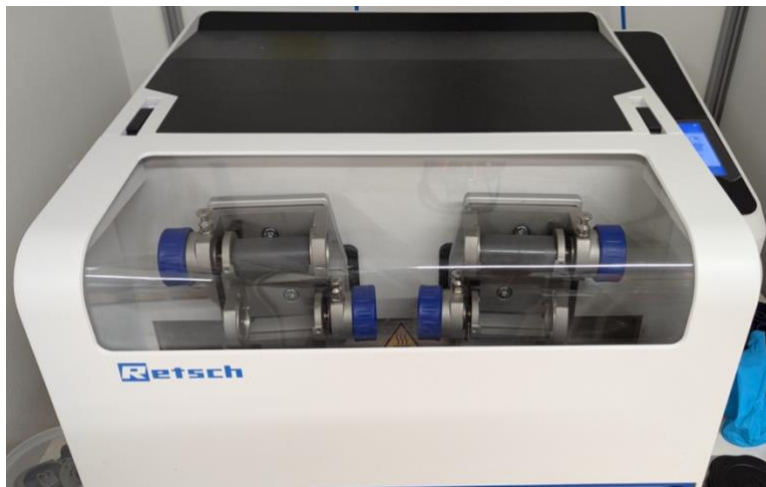
Calculation of LAG amounts:

The amounts of LAG are calculated as $\mu\text{L}/\text{mg}$. This refers to the μL of LAG per mass of all other reagents added to the reaction.

Safety and hazard statement:

In general, care and attention must be considered when attempting to scale up mechanochemical transformations. There are a number of reports in the literature of large exotherms being produced during mechanochemical procedure.²⁻⁵ Whilst this has certainly never been observed under our reaction conditions, it cannot be ruled out. A recent study by Waser and coworkers illustrates the utility of TGA and DSC measurements to characterise the probability of violent decomposition of many key hypervalent iodine intermediates.⁶ Furthermore it is advised to carry out the safety protocol and procedure as described by Browne and coworkers.⁷ HFIP is suspected of damaging fertility and can cause severe eye damage, so need to be used inside a ventilated fumehood.

2. Mechanochemical setup



Retsch MM500 vario used as a ball milling device.



Retsch MM2 used as a ball milling device.



Retsch ball milling vessels.
1.5 mL (left) and 5 mL (right).



Retsch ball milling stainless steel balls. 5 mm (left), 7 mm (middle) and 10 mm (right).

3. Optimisation of different types of reactions

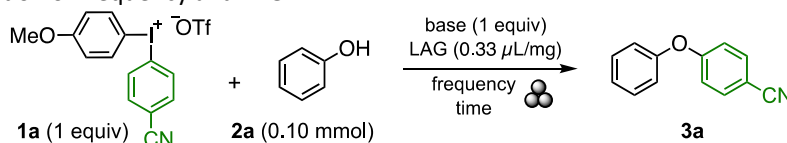
During the course of this optimisation, we found it was possible to carry this reaction out with a phenyl dummy ligand but chose to opt for the anisyl ligand due to consistency throughout the reported works.

3.1 O-Arylation optimisation

General procedure for the optimisation: Sodium carbonate (10.6 mg, 0.10 mmol, 1.0 equiv), phenol (**2a**, 9.4 mg, 0.10 mmol, 1.0 equiv), diaryliodonium salt **1a** (48.8 mg, 0.10 mmol, 1.0 equiv) and one 7 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for a given time at a given frequency. 1,3,5-Trimethoxybenzene (TMB, 1 equiv) was added to the vessel as internal standard, CDCl₃ was added and a crude NMR of the solution was recorded. The yield of **2a** was determined using TMB as the internal standard.

The optimisation details are given in Table S1. We found that the most important variable was the addition of the right liquid additive (LAG), which in general was used in 0.33 μL/mg = 20 μL. The most efficient process had EtOAc as the LAG, in combination with potassium carbonate as base (entry 25).

Table S1. Optimization of frequency and LAG.

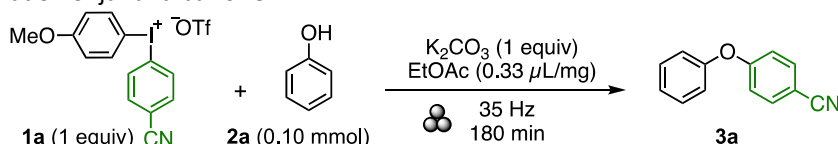


Entry	Time (min)	Frequency (Hz)	Base (1 equiv)	LAG (0.33 μL/mg)	NMR yield (%)
1	30	15	Na ₂ CO ₃	/	4
2	30	25	NaO ^t Bu	/	5
3	30	25	NaOH	/	5
4	30	20	Na ₂ CO ₃	/	6
5	30	25	Na ₂ CO ₃	/	8
6	30	30	Na ₂ CO ₃	/	10
7	30	35	Na ₂ CO ₃	/	15
8	30	30	Na ₂ CO ₃	THF	10
9	30	35	Na ₂ CO ₃	THF	23
10	30	35	Na ₂ CO ₃	Toluene	40
11	30	35	Na ₂ CO ₃	MeCN	35
12	30	35	Na ₂ CO ₃	EtOAc	40
13	30	35	Na ₂ CO ₃	MeOH	39
14	30	35	Na ₂ CO ₃	Cyclopentanone	28
15	30	35	Na ₂ CO ₃	EtOAc (10 μL)	33
16	30	35	Na ₂ CO ₃	EtOAc (30 μL)	41
17	30	35	Na ₂ CO ₃	EtOAc (40 μL)	35
18	30	35	Na ₂ CO ₃	EtOAc (50 μL)	20
19	180	35	Na ₂ CO ₃	EtOAc	60
20	180	35	NaO ^t Bu	EtOAc	55
21	180	35	LiO ^t Bu	EtOAc	25
22	180	35	NaOH	EtOAc	44

23	180	35	KO ^t Bu	EtOAc	69
24	180	35	NEt ₃	EtOAc	40
25	180	35	K ₂ CO ₃	EtOAc	88

The jar and ball size, and number of balls were then varied (Table S2). It was observed that the reaction was not very sensitive to either jar size or ball size. However, when two stainless steel balls were used with either 1.5 mL or 5 mL jars, the yields were markedly lower. This could be due to the balls becoming stuck inside the jars and effectively stopping the milling.

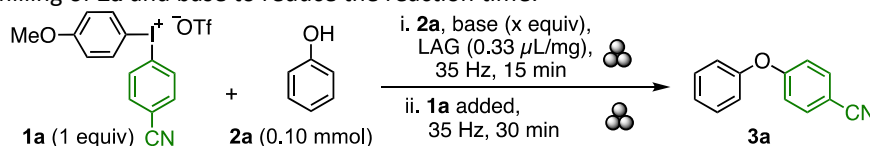
Table S2. Optimization of jar and ball size.



Entry	Jar size (mL)	Ball size (mm)	Number of balls	NMR yield (%)
1	1.5	7	1	88
2	1.5	5	1	87
3	1.5	5	2	63
4	5	10	1	90
5	5	7	1	89
6	5	5	1	90
7	5	5	2	85

In an effort to reduce the reaction time, a pre-mill of **2a** and base was performed to separate the deprotonation from the arylation step (Table S3). With 15 min deprotonation pre-mill and 30 min reaction time, the reaction time was reduced to just 45 min instead of 180 min (entry 1). An increase in the number of equivalents of base, does not increase the reaction yield (entry 2). To ascertain whether the 45 mins reaction time alone contributes to the reaction outcome, the reaction without pre-milling was carried out for 45 mins leading to a lower yield (entry 3). A scale-up from 0.1 mmol to 0.30 mmol resulted in a similarly high yield as previous (entry 4), which was then used as standard conditions in the reaction.

Table S3. Pre-milling of **2a** and base to reduce the reaction time.



Entry	Base (x equiv)	NMR yield (%)	Isolated yield (%)
1	K ₂ CO ₃ (1 equiv)	95	/
2	K ₂ CO ₃ (2 equiv)	95	/
3 ^a	K ₂ CO ₃ (1 equiv)	68	/
4 ^b	K ₂ CO ₃ (1 equiv)	/	91

^a 45 min reaction time without pre-milling of base and phenol; ^b 0.30 mmol scale.

Next, the arylation of 1-pentanol (**2b**) was performed (Table S4). Utilising the above conditions with 60 min milling time led to a yield of 43% (entry 1). By increasing the reaction time to 120 min, the yield increased to quantitative. Furthermore, changing the LAG to acetonitrile or adding everything at once produced similar yields. When the reaction was

increased to 0.3 mmol, a 91% isolated yield was acquired and these were used as standard conditions for the scope with aliphatic alcohols (entry 5).

Table S4. Optimization with aliphatic alcohol **2b**.

Entry	Changes to above conditions	NMR yield (%)	Isolated yield (%)
1	/	>95	/
2	60 mins milling with 1a	43	/
3	MeCN instead of EtOAc	>95	/
4	Everything added at beginning	>95	/
5^a	/	/	91

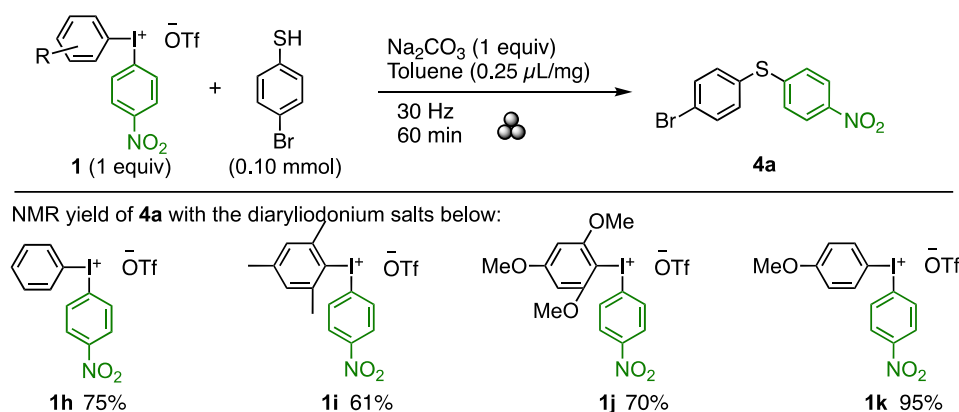
^a Increase in scale from 0.1 mmol to 0.30 mmol.

3.2 S-Arylation optimisation

Early test reactions were carried out with a number of inorganic and organic bases and different types of diaryliodonium salts, which produced good yields. Sodium carbonate was ultimately chosen as a weak and functional group-tolerant base, which is also a solid and cheap to acquire.

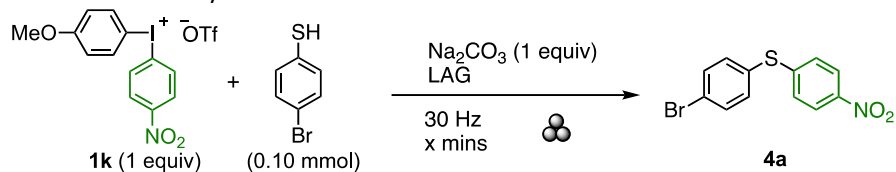
General procedure for the optimisation: Sodium carbonate (10.6 mg, 0.10 mmol, 1.0 equiv), 4-bromothiophenol (18.9 mg, 0.10 mmol, 1.0 equiv), diaryliodonium salt **1** (0.10 mmol, 1.0 equiv), toluene (20 μ L) and one 7 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 60 min at 30 Hz. TMB (1 equiv) was added to the vessel, CDCl_3 was added and a crude NMR was recorded. The NMR yield of **4a** was determined using TMB as the internal standard.

A variety of “dummy” groups were evaluated to ensure complete chemoselectivity and good yields, and the anisyl group was found to be the best (Scheme S1).



Scheme S1. Evaluation of dummy groups for chemoselective and high-yielding S-arylation.

Then the toluene LAG was varied to find a more sustainable additive. Utilising the GSK solvent guide,⁸ several more sustainable solvents were evaluated (Table S5). We found that green solvent additives such as cyclopentanone and dimethyl carbonate provided an efficient reaction environment for the reaction (entries 5, 6). Additionally, the amount of cyclopentanone additive could be reduced to 2 μL (0.025 $\mu\text{L}/\text{mg}$) and the reaction time was reduced to 30 minutes (entries 8, 9). The reaction could be scaled up to 0.30 mmol scale (entry 10) and these conditions were used as standard for the scope evaluation.

Table S5. Optimization of thiol arylation.

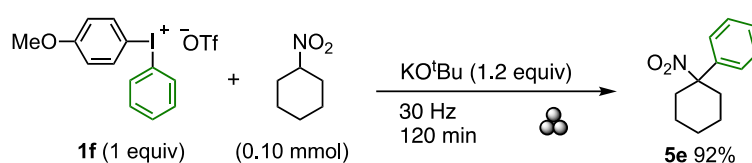
Entry	Time (min)	LAG (x $\mu\text{L}/\text{mg}$)	NMR yield (%) ^a
1	60	Toluene (0.25)	95
2	60	Anisole (0.25)	86
3	60	Ethyl acetate (0.25)	92
4	60	Isopropyl acetate (0.25)	>95
5	60	Dimethyl carbonate (0.25)	>95
6	60	Cyclopentanone (0.25)	>95
7	60	Cyclopentanone (0.125)	>95
8	60	Cyclopentanone (0.025)	>95
9	30	Cyclopentanone (0.025)	>95
10^a	30	Cyclopentanone (0.025)	(91)

^a 0.30 mmol scale, isolated yield.

3.4 C-Arylation optimisation

General procedure for the planned optimisation: Potassium *tert*-butoxide (11.9 mg, 0.12 mmol, 1.2 equiv), nitrocyclohexane (12.7 mg, 0.1 mmol, 1.0 equiv), and one 7 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 30 minutes at 30 Hz and then salt diaryliodonium **1f** (43.4 mg, 0.10 mmol, 1.0 equiv) was added. This mixture was milled for 120 minutes. TMB (1 equiv) was added to the vessel, CDCl₃ was added and a crude NMR was recorded. The NMR yield of **5e** was determined using TMB as the internal standard.

Conditions based on the standard solution phase reaction⁹ produced a yield of 92% (Scheme S2). Based on the excellent reaction outcome, no optimization was performed.

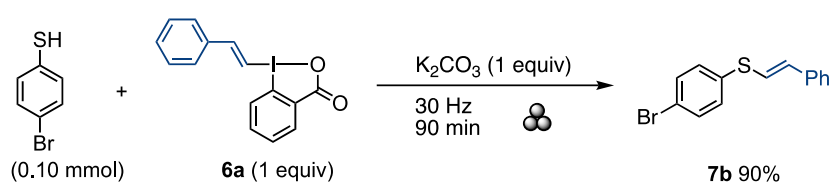


Scheme S2. C-arylation conditions.

3.5 S-vinylation optimisation

General procedure for the planned optimisation: Potassium carbonate (13.2 mg, 0.1 mmol, 1.0 equiv), 4-bromothiophenol (18.9 mg, 0.1 mmol, 1.0 equiv), one 7 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the vessel was milled for 30 minutes at 30 Hz and VBX **6a** (43.4 mg, 0.10 mmol, 1.0 equiv) was added. This mixture was milled for 90 minutes. 1,3,5-Trimethoxybenzene (1 equiv) was added to the vessel, CDCl₃ was added and a crude NMR was recorded. The NMR yield of **7b** was determined using TMB as the internal standard.

Literature conditions from the standard solution phase reaction¹⁰ produced a yield of 90% (Scheme S3). Based on the excellent reaction outcome, no optimization was performed.

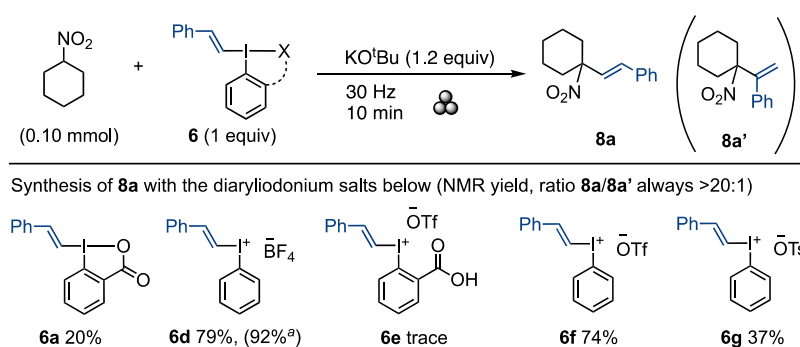


Scheme S3. C-vinylation conditions.

3.6 C-Vinylation optimisation

General procedure for the optimization: Potassium *tert*-butoxide (11.9 mg, 0.12 mmol, 1.2 equiv), nitrocyclohexane (12.7 mg, 0.1 mmol, 1.0 equiv), and one 7 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the vessel was milled for 15 minutes at 30 Hz and **6** (0.10 mmol, 1.0 equiv) was added. This mixture was milled for 10 minutes. TMB (1 equiv) was added to the vessel, CDCl₃ was added and a crude NMR was recorded. The NMR yield of **8a** was determined using TMB as the internal standard.

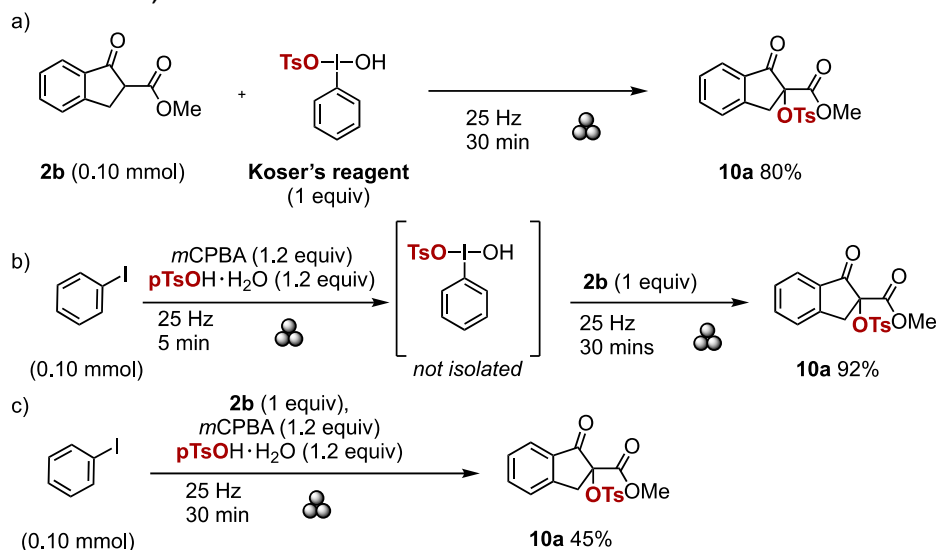
Optimisation began with using nitrocyclohexane as the model substrate since it was used previously by the Olofsson group. The reactions were milled for 10 minutes at 30 Hz. Utilising VBX reagent **6a** produced the internal alkene with excellent regioselectivity but only 20% yield (Scheme S4). Acyclic vinylidonium salts **6d-6g** were then evaluated and reagent **6d** proved superior to the rest (92% yield). Different LAGs and excess equivalents of **6d** were utilised but did not produced higher yields. Further optimization was not performed.



Scheme S4. Evaluation of vinylation reagents for regioselective and high-yielding C-vinylation. ^a 0.30 mmol scale, isolated yield.

3.7 Catalytic tosyloxylation optimisation

Initial investigations used stoichiometric amounts of iodine(III) reagent. The synthesis of **10a** using Koser's reagent led to a yield of 80% (Scheme S5a). We then synthesized Koser's reagent from iodobenzene, followed by in-situ reaction with **2b** to form product **10a** in 92% yield (Scheme S5b). Finally, a true one-pot reaction with all reagents added at once gave the product in the same yield (Scheme S5c). These reactions were the first examples of mechanochemical oxidation of iodine(I) to iodine(III), and quite promising for development of a catalytic reaction, which is discussed below.



Scheme S5. Reactions with stoichiometric amount of iodine(III) reagent.

General procedure for the optimization: **2b** (19 mg, 0.1 mmol, 1.0 equiv), iodoarene (0.02 mmol, 0.2 equiv), *p*TsOH·H₂O (19 mg, 0.1 mmol, 1.0 equiv), *m*CPBA (20.8 mg, 0.1 mmol, 1.0 equiv) and one 10 mm stainless steel ball were added to a 5 mL stainless steel ball milling vessel. The vessel was closed and the vessel was milled for 15 minutes at 25 Hz for the tabulated time. TMB (1 equiv) was added to the vessel, CDCl₃ was added and a crude NMR was recorded. The NMR yields of **10a** and **10a'** were determined using TMB as the internal standard.

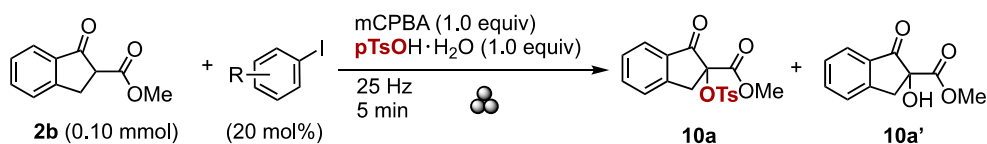
Optimisation began with iodobenzene as catalyst, which produced moderate yield of **10a** with considerable amount of the byproduct **10a'** (Table S6, entries 1-4). Next, a screen of iodoarene catalysts was conducted, which showed that strongly withdrawing and donating groups were unfavourable, and 4-iodotoluene was best (entries 5-9). Investigation of sterics on the iodoarene showed that substitution in the *ortho* position can be favourable (entry 5 vs 10), and 2,4-dimethyliodobenzene was chosen as the best catalyst with multiple substitutions (entries 13-16), giving **10a** in 84% yield, with 24% of **10a'**.

We then screened a range of LAGs (0.25 μL/mg), including water, EtOAc and isopropanol in order to avoid the formation of byproduct **10a'** (entries 15-23). However, we found that HFIP was the only LAG that suppressed the formation of **10a'** (entries 13 vs 24), and the optimal amount was found to be 0.5 μL/mg (entries 24-27).

Finally the amounts of iodoarene and *p*TsOH·H₂O were varied but the original amounts were the best (entry 26). The reaction was then scaled up to 0.3 mmol and these conditions were used as standard for the scope (entry 30).

While HFIP is not a great additive for sustainability reasons, it should be pointed out that only 0.5 $\mu\text{L}/\text{mg}$ of HFIP is added, amounting to about 100 μL for each reaction in Scheme 4. As reactions without HFIP were difficult to purify due to **10a'**, the use of HFIP as LAG is motivated. Note that the synthesis of product **10c** was efficient in absence of HFIP.

Table S6. Optimization of the tosyloxylation.



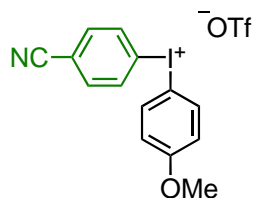
Entry	R	Time (min)	LAG (0.25 $\mu\text{L}/\text{mg}$)	Yield 10a (%)	Yield 10a' (%)
1	H	120	/	56	36
2	H	60	/	61	26
3	H	15	/	62	35
4	H	5	/	66	32
5	4-NO ₂	5	/	0	88
6	4-OMe	5	/	70	22
7	4-Me	5	/	77	28
8	4-OH	5	/	19	20
9	4-N(Me) ₂	5	/	7	65
10	2-NO ₂	5	/	30	65
11	2-CF ₃	5	/	17	70
12	2-Me	5	/	72	24
13	2,4-Me	5	/	84	23
14	2,4-OMe	5	/	46	26
15	2,4,6-Me	5	/	76	27
16	3,5-Me	5	/	82	27
17	2,4-Me	5	H ₂ O	40	38
18	2,4-Me	5	EtOAc	63	32
19	2,4-Me	5	Isopropanol	58	26
20	2,4-Me	5	MeCN	62	31
21	2,4-Me	5	CH ₂ Cl ₂	65	23
22	2,4-Me	5	BMIM•BF ₄	75	20
23	2,4-Me	5	TFE	76	19
24	2,4-Me	5	HFIP	81	14
25	2,4-Me	5	Anhydrous HFIP	82	11
26	2,4-Me	5	HFIP (0.5 $\mu\text{L}/\text{mg}$)	83	11
27	2,4-Me	5	HFIP (0.75 $\mu\text{L}/\text{mg}$)	76	7
28 ^a	2,4-Me	5	HFIP (0.5 $\mu\text{L}/\text{mg}$)	76	23
29 ^b	2,4-Me	5	HFIP (0.5 $\mu\text{L}/\text{mg}$)	2	11
30 ^c	2,4-Me	5	HFIP (0.5 $\mu\text{L}/\text{mg}$)	(87)	Not isolated

1-Butyl-3-methylimidazolium tetrafluoroborate = BMIM BF₄. ^a 10 mol% ArI; ^b 1.2 equiv TsOH; ^c 0.30 mmol scale, isolated yield.

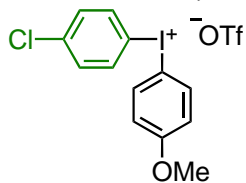
4. Substrates

4.1 Iodine(III) reagents

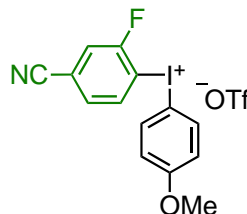
All of the iodine(III) reagents used in this manuscript have been reported previously. Thus, procedures data is not given for each salt, but can be found in the references provided.



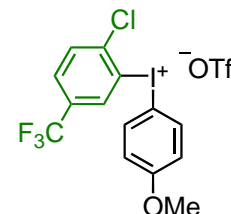
1a, Ref¹¹



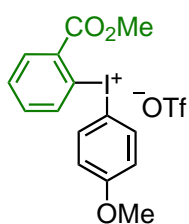
1b, Ref¹¹



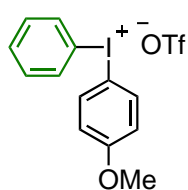
1c, Ref¹¹



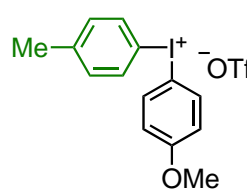
1d, Ref¹¹



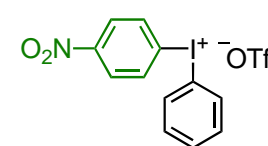
1e, Ref¹¹



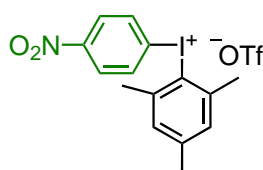
1f, Ref¹¹



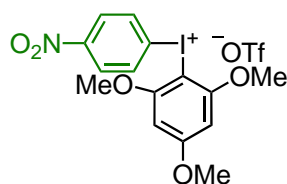
1g, Ref¹¹



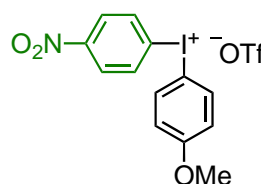
1h, Ref¹¹



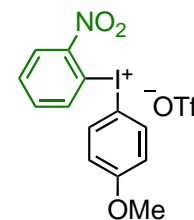
1i, Ref¹¹



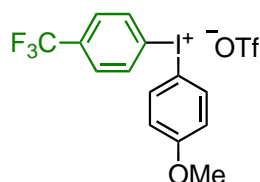
1j, Ref¹²



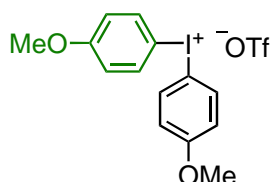
1k, Ref¹¹



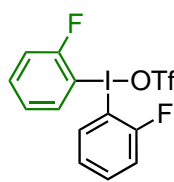
1l, Ref¹¹



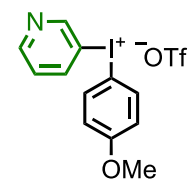
1m, Ref¹¹



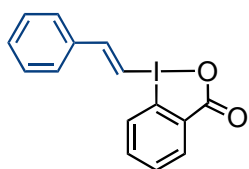
1n, Ref⁹



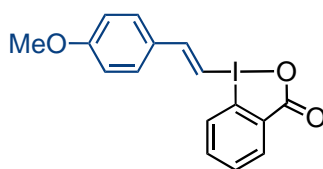
1o, Ref⁹



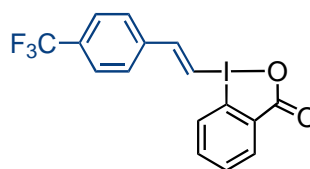
1p, Ref¹³



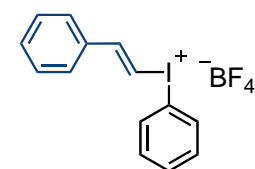
6a, Ref¹⁴



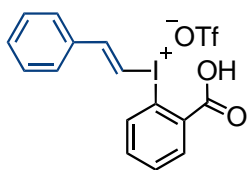
6b, Ref¹⁵



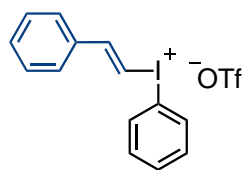
6c, Ref¹⁵



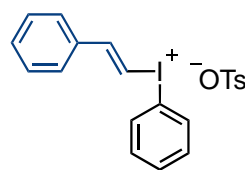
6d, Ref¹⁴



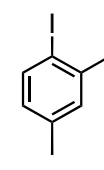
6e, Ref¹⁴



6f, Ref¹⁶



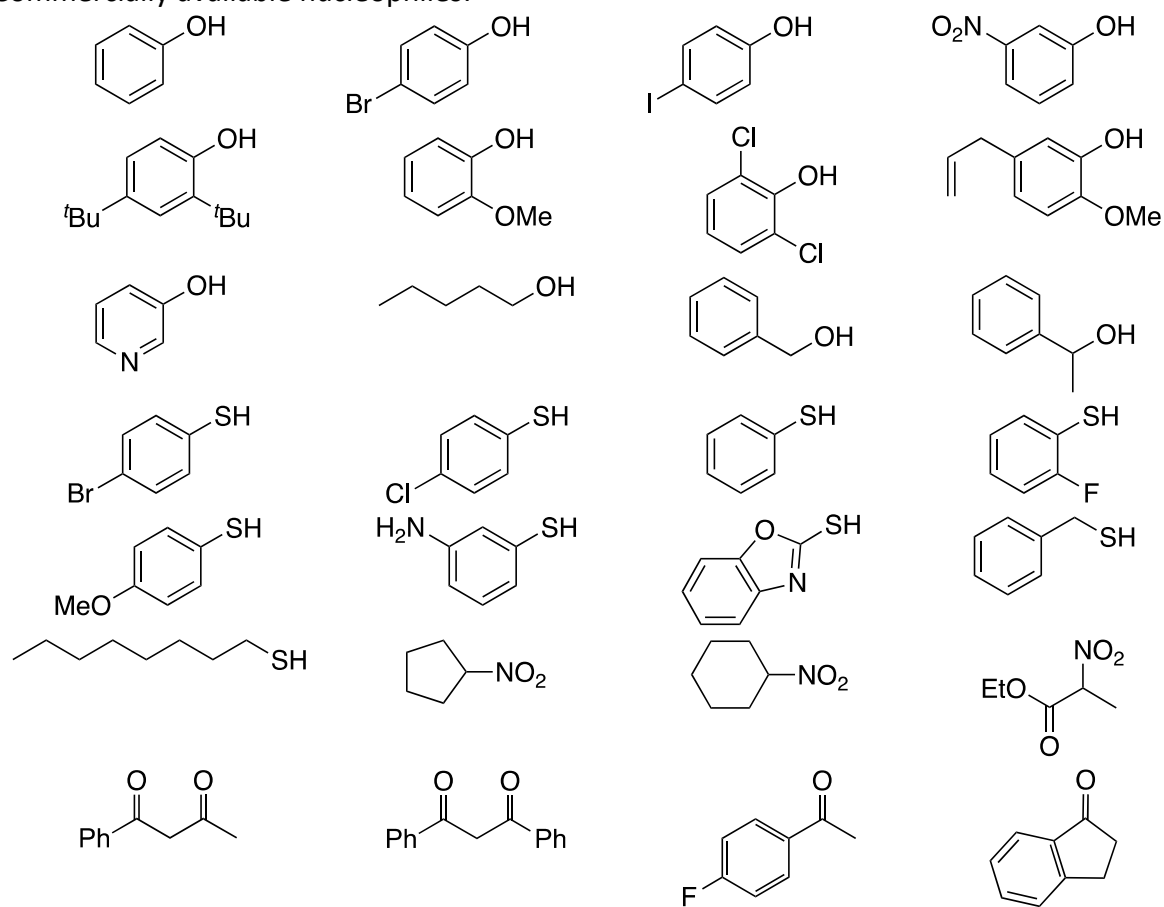
6g, Ref¹⁷



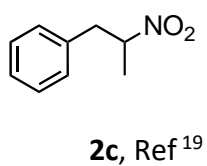
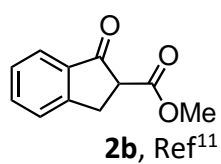
9, Ref¹⁸

4.2 Nucleophiles

Commercially available nucleophiles:



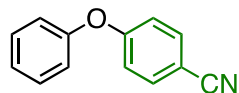
The following were synthesised using the procedures indicated:



5. Products

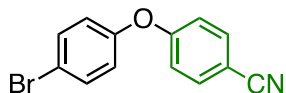
5.1 Product compound table

O-Arylation



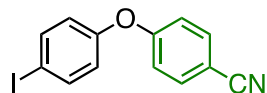
3a

[Characterization](#)
[Spectra](#)



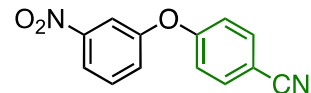
3b

[Characterization](#)
[Spectra](#)



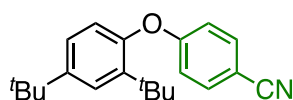
3c

[Characterization](#)
[Spectra](#)



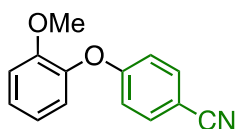
3d

[Characterization](#)
[Spectra](#)



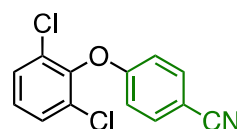
3e

[Characterization](#)
[Spectra](#)



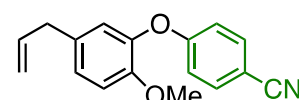
3f

[Characterization](#)
[Spectra](#)



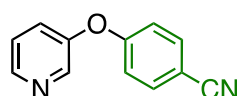
3g

[Characterization](#)
[Spectra](#)



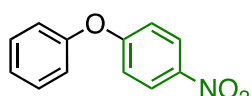
3h

[Characterization](#)
[Spectra](#)



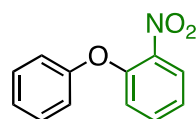
3i

[Characterization](#)
[Spectra](#)



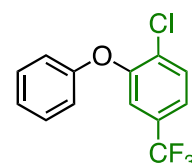
3j

[Characterization](#)
[Spectra](#)



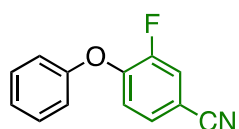
3k

[Characterization](#)
[Spectra](#)



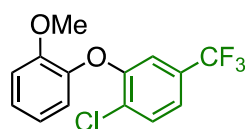
3l

[Characterization](#)
[Spectra](#)



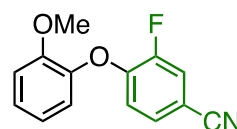
3m

[Characterization](#)
[Spectra](#)



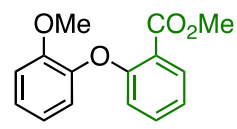
3n

[Characterization](#)
[Spectra](#)



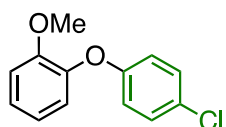
3o

[Characterization](#)
[Spectra](#)



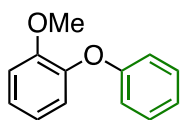
3p

[Characterization](#)
[Spectra](#)



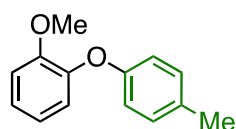
3q

Characterization
Spectra



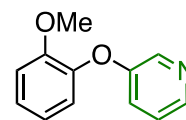
3r

Characterization
Spectra



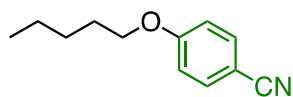
3s

Characterization
Spectra



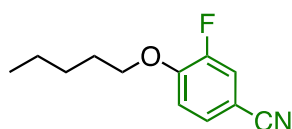
3t

Characterization
Spectra



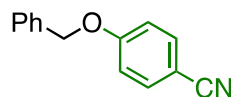
3u

Characterization
Spectra



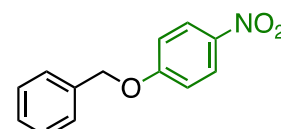
3v

Characterization
Spectra



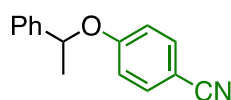
3w

Characterization
Spectra



3x

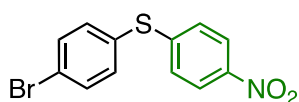
Characterization
Spectra



3y

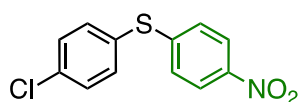
Characterization
Spectra

S-arylation



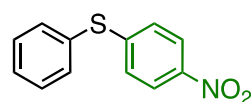
4a

Characterization
Spectra



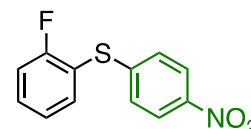
4b

Characterization
Spectra



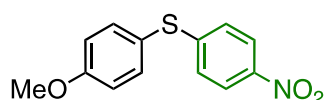
4c

Characterization
Spectra



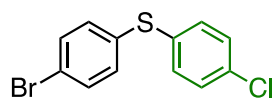
4d

Characterization
Spectra



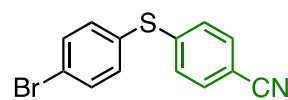
4e

Characterization
Spectra



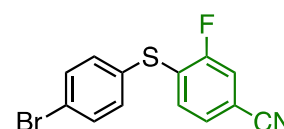
4f

Characterization
Spectra



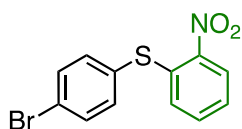
4g

Characterization
Spectra



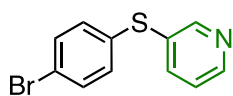
4h

Characterization
Spectra



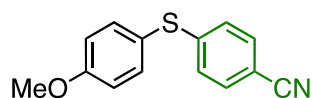
4i

Characterization
Spectra



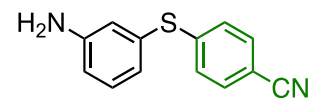
4j

Characterization
Spectra



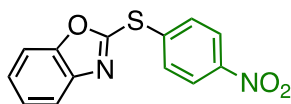
4k

Characterization
Spectra



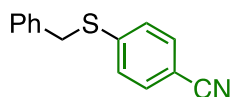
4l

Characterization
Spectra



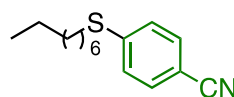
4m

Characterization
Spectra



4n

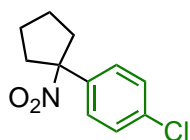
Characterization
Spectra



4o

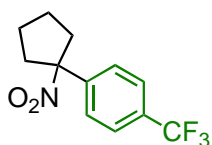
Characterization
Spectra

C-Arylation



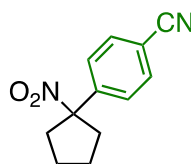
5a

Characterization
Spectra



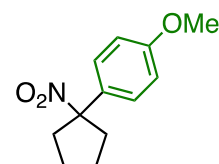
5b

Characterization
Spectra



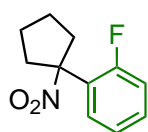
5c

Characterization
Spectra



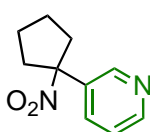
5d

Characterization
Spectra



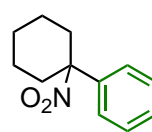
5e

Characterization
Spectra



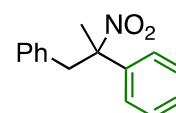
5f

Characterization
Spectra



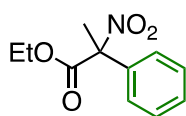
5g

Characterization
Spectra



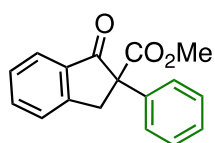
5h

Characterization
Spectra



5i

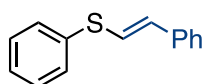
Characterization
Spectra



5j

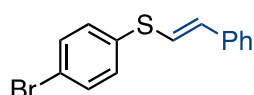
Characterization
Spectra

S-Vinylation



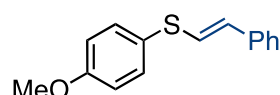
7a

Characterization
Spectra



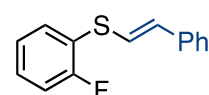
7b

Characterization
Spectra



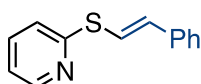
7c

Characterization
Spectra



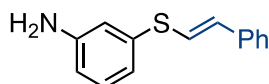
7d

Characterization
Spectra



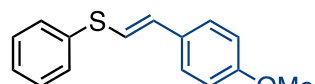
7e

Characterization
Spectra



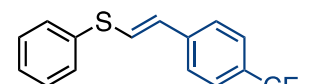
7f

Characterization
Spectra



7g

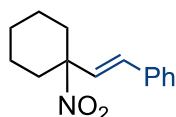
Characterization
Spectra



7h

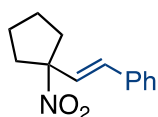
Characterization
Spectra

C-Vinylation



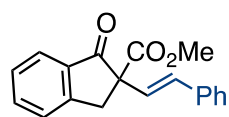
8a

Characterization
Spectra



8b

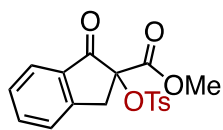
Characterization
Spectra



8c

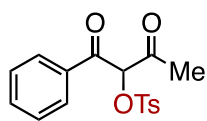
Characterization
Spectra

Catalytic tosyloxylation



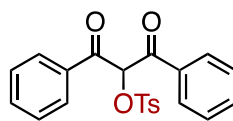
10a

Characterization
Spectra



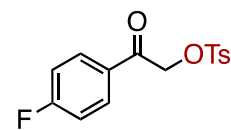
10b

Characterization
Spectra



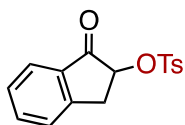
10c

Characterisation
Spectra



10d

Characterization
Spectra



10e

Characterization
Spectra

5.2 O-Arylation

General procedure 1: O-arylation

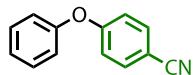
In prose:

Potassium carbonate (41.5 mg, 0.30 mmol, 1 equiv), the nucleophile (0.30 mmol), one 5 mm stainless steel ball and LAG (ethyl acetate or MeCN, 0.33 $\mu\text{L}/\text{mg}$) were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 15 minutes at 35 Hz and then diaryliodonium salt **1** (0.30 mmol, 1 equiv) was added. This mixture was milled for 30 minutes at 35 Hz. The contents of the vessel were washed out using EtOAc (5 x 0.75 mL) and concentrated *in vacuo*. This residue was purified using silica gel flash chromatography to provide product **3**.

In recipe style:

- Potassium carbonate (41.5 mg, 0.30 mmol, 1 equiv), nucleophile (0.30 mmol), one 5 mm stainless steel ball and LAG (ethyl acetate or MeCN, 0.33 $\mu\text{L}/\text{mg}$) were added to a 1.5 mL stainless steel ball milling vessel.
- Mill for 15 minutes at 35 Hz.
- Add diaryliodonium salt **1** (0.30 mmol, 1.0 equiv).
- Mill for 30 minutes at 35 Hz.
- Contents of vessel washed out with EtOAc (5 x 0.75 mL).
- Concentrate *in vacuo*.
- Purify residue using silica gel flash chromatography to provide product **3**.

4-Phenoxybenzonitrile (**3a**)



This compound was synthesised according to **general procedure 1** using phenol (28.2 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10%

EtOAc/pentane) to give **3a** (53.0 mg, 91%) as a colourless solid;

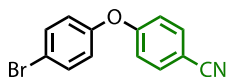
R_f (10% EtOAc/pentane) 0.3;

δ_H (400 MHz, CDCl_3) 7.64 – 7.56 (2H, m), 7.46 – 7.38 (2H, m), 7.26 – 7.20 (1H, m), 7.10 – 7.05 (2H, m), 7.03 – 6.98 (2H, m);

δ_C (101 MHz, CDCl_3) 161.8, 154.9, 134.3, 130.4, 125.3, 120.6, 119.0, 118.1, 106.0.

This data is consistent with literature precedent.²⁰

4-(4-Bromophenoxy)benzonitrile (**3b**)



This compound was synthesised according to **general procedure 1** using 4-bromophenol (51.9 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10%

EtOAc/pentane) to give **3b** (74.0 mg, 90%) as a colourless solid;

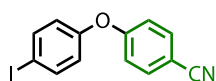
R_f (10% EtOAc/pentane) 0.3;

δ_H (400 MHz, CDCl_3) 7.64 – 7.59 (2H, m), 7.55 – 7.49 (2H, m), 7.03 – 6.97 (2H, m), 6.97 – 6.93 (2H, m);

δ_C (101 MHz, CDCl_3) 161.2, 154.2, 134.4, 133.4, 122.2, 118.8, 118.2, 118.0, 106.5.

This data is consistent with literature precedent.²¹

4-(4-Iodophenoxy)benzonitrile (3c)



This compound was synthesised according to **general procedure X** using 4-iodophenol (66.0 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (5% EtOAc/pentane) to give **3c** (70.30 mg, 73%) as a colourless solid;

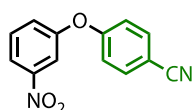
R_f (10% EtOAc/pentane) 0.4;

δ_H (400 MHz, CDCl₃) 7.73 – 7.68 (2H, m), 7.64 – 7.59 (2H, m), 7.04 – 6.98 (2H, m), 6.87 – 6.80 (2H, m);

δ_c (101 MHz, CDCl₃) 161.1, 155.1, 139.4, 134.4, 122.5, 118.8, 118.3, 106.6, 88.6.

This data is consistent with literature precedent.²²

4-(3-Nitrophenoxy)benzonitrile (3d)



This compound was synthesised according to **general procedure 1** using 3-nitrophenol (41.7 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3d** (60.0 mg, 83%) as a colourless solid;

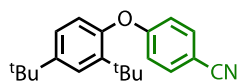
R_f (10% EtOAc/pentane) 0.1;

δ_H (400 MHz, CDCl₃) 8.06 (1H, ddd, *J* = 8.2, 2.1, 1.0 Hz), 7.88 (1H, t, *J* = 2.3 Hz), 7.71 – 7.65 (2H, m), 7.59 (1H, t, *J* = 8.2 Hz), 7.40 (1H, ddd, *J* = 8.2, 2.3, 1.0 Hz), 7.11 – 7.07 (2H, m);

δ_c (101 MHz, CDCl₃) 160.0, 156.1, 149.5, 134.6, 131.0, 125.9, 119.6, 119.1, 118.4, 114.9, 107.8.

This data is consistent with literature precedent.²³

4-(2,4-Di-tert-butylphenoxy)benzonitrile (3e)



This compound was synthesised according to **general procedure 1** using 2,4-bis(*t*-butyl)phenol (61.9 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (2% EtOAc/pentane) to give **3e** (70.7 mg, 77%) as a colourless solid;

R_f (5% EtOAc/pentane) 0.3;

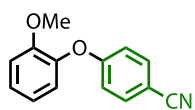
δ_H (400 MHz, CDCl₃) 7.62 – 7.56 (2H, m), 7.45 (1H, d, *J* = 2.4 Hz), 7.20 (1H, dd, *J* = 8.4, 2.4 Hz), 7.03 – 6.97 (2H, m), 6.78 (1H, d, *J* = 8.4 Hz), 1.36 (9H, s), 1.34 (9H, s);

δ_c (101 MHz, CDCl₃) 162.2, 151.2, 147.8, 141.0, 134.2, 124.8, 124.4, 121.1, 119.2, 118.1, 105.3, 35.0, 34.8, 31.7, 30.4;

HRMS (ESI) calculated for C₂₁H₂₅NONa (M+Na⁺): 330.1818; found: 330.1828

Mp 167.4 – 169.1 °C

4-(2-Methoxyphenoxy)benzonitrile (3f)



This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (37.2 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3f** (60.1 mg, 89%) as a colourless solid;

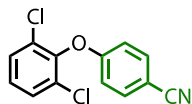
R_f (10% EtOAc/pentane) 0.2;

δ_{H} (400 MHz, CDCl_3) 7.59 – 7.53 (2H, m), 7.24 (1H, ddd, $J = 9.2, 7.1, 1.4$ Hz), 7.10 – 6.96 (3H, m), 6.95 – 6.89 (2H, m), 3.78 (3H, s);

δ_{C} (101 MHz, CDCl_3) 162.0, 151.8, 142.6, 134.0, 126.8, 122.7, 121.5, 119.1, 116.6, 113.2, 105.3, 55.9.

This data is consistent with literature precedent.²⁴

4-(2,6-Dichlorophenoxy)benzonitrile (3g)



This compound was synthesised according to **general procedure 1** using 2,6-dichlorophenol (48.9 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10%

EtOAc/pentane) to give **3g** (69.8 mg, 88%) as a colourless solid;

R_{f} (10% EtOAc/pentane) 0.3;

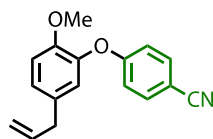
δ_{H} (400 MHz, CDCl_3) 7.64 – 7.58 (2H, m), 7.43 (2H, d, $J = 8.2$ Hz), 7.21 (1H, dd, $J = 8.5, 7.8$ Hz), 6.93 – 6.86 (2H, m);

δ_{C} (101 MHz, CDCl_3) 159.9, 146.2, 134.3, 129.8, 129.5, 127.3, 118.8, 116.0, 106.5;

HRMS (ESI) calculated for $\text{C}_{12}\text{H}_7^{35}\text{Cl}_2\text{NONa}$ ($\text{M}+\text{Na}^+$): 285.9790; found: 285.9797;

Mp 143.4 – 145.1 °C.

4-(5-Allyl-2-methoxyphenoxy)benzonitrile (3h)



This compound was synthesised according to **general procedure 1** using Eugenol (28.5 μL , 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3h** (70.4 mg, 87%) as a colourless oil;

Large scale reaction: Potassium carbonate (414.6 mg, 3.00 mmol, 1 equiv), eugenol (0.467 mL, 3.00 mmol, 1 equiv), one 10 mm stainless steel ball and LAG (0.45 mL) were added to a 5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 15 minutes at 35 Hz and then diaryliodonium salt **1a** (1.455 g, 3.00 mmol, 1 equiv) was added. This mixture was milled for 30 minutes at 35 Hz. The contents of the vessel were washed out using EtOAc (5 x 4 mL) and concentrated *in vacuo*. This residue was purified using silica gel flash chromatography (10% EtOAc/pentane) to provide product **3h** (709.5 mg, 89%) as a colourless oil;

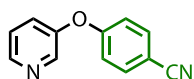
R_{f} (10% EtOAc/pentane) 0.23;

δ_{H} (400 MHz, CDCl_3) 7.57 – 7.52 (2H, m), 6.99 (1H, d, $J = 8.0$ Hz), 6.94 – 6.89 (2H, m), 6.85 (1H, d, $J = 1.9$ Hz), 6.81 (1H, dd, $J = 8.0, 1.3$ Hz), 5.99 (1H, ddt, 17.0, 10.2, 6.7 Hz), 5.18 – 5.09 (2H, m), 3.76 (3H, s), 3.41 (2H, dt, $J = 6.7$ Hz);

δ_{C} (101 MHz, CDCl_3) 162.2, 151.6, 140.8, 139.0, 137.0, 134.0, 122.5, 121.4, 119.2, 116.5, 116.4, 113.4, 105.1, 55.9, 40.1;

HRMS (ESI) calculated for $\text{C}_{17}\text{H}_{15}\text{NO}_2\text{Na}$ ($\text{M}+\text{Na}^+$): 288.0989; found: 288.0995

4-(Pyridin-3-yloxy)benzonitrile (3i)



This compound was synthesised according to **general procedure 1** using 3-hydroxypyridine (28.5 mg, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv)

and EtOAc as LAG. The residue was purified by flash chromatography (80% EtOAc/pentane) to give **3i** (51.2 mg, 86%) as a colourless solid;

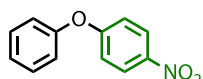
R_f (80% EtOAc/pentane) 0.3;

δ_H (400 MHz, CDCl₃) 8.48 (1H, dd, *J* = 4.4, 1.7 Hz), 8.44 (1H, d, *J* = 2.7 Hz), 7.66 – 7.61 (2H, m), 7.41 – 7.33 (2H, m), 7.05 – 7.00 (2H, m);

δ_C (101 MHz, CDCl₃) 160.8, 151.7, 146.4, 142.7, 134.5, 127.5, 124.6, 118.6, 118.3, 107.0.

This data is consistent with literature precedent.²⁵

1-Nitro-4-phenoxybenzene (3j)



This compound was synthesised according to **general procedure 1** using Phenol (28.2 mg, 0.30 mmol), **1h** (151.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (5% -> 10% EtOAc/Pentane) to give **3j** (44.0 mg, 69%) as a colourless oil;

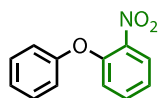
R_f (2% EtOAc/Pentane) 0.4;

δ_H (400 MHz, CDCl₃) 8.23 – 8.17 (2H, m), 7.48 – 7.40 (2H, m), 7.30 – 7.23 (1H, m), 7.12 – 7.07 (2H, m), 7.05 – 6.98 (2H, m);

δ_C (101 MHz, CDCl₃) 163.5, 154.8, 142.8, 130.4, 126.1, 125.5, 120.7, 117.2.

This data is consistent with literature precedent.²⁶

1-Nitro-2-phenoxybenzene (3k)



This compound was synthesised according to **general procedure 1** using phenol (18.8 mg, 0.2 mmol), **1l** (101.0 mg, 0.20 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (5% -> 10% EtOAc/pentane) to give **3k** (21.0 mg, 48%) as an orange oil.

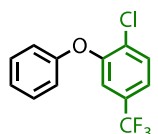
R_f (10% EtOAc/pentane) 0.25;

δ_H (400 MHz, CDCl₃) 7.95 (1H, dd, *J* = 8.2, 1.7 Hz), 7.50 (1H, ddd, *J* = 8.2, 7.4, 1.7 Hz), 7.41 – 7.36 (2H, m), 7.22 – 7.17 (2H, m), 7.08 – 6.98 (3H, m);

δ_C (101 MHz, CDCl₃) 155.9, 150.9, 141.5, 134.2, 130.2, 125.9, 124.7, 123.2, 120.6, 119.4.

This data is consistent with literature precedent.²⁷

1-Chloro-2-phenoxy-4-(trifluoromethyl)benzene (3l)



This compound was synthesised according to **general procedure 1** using phenol (28.2 mg, 0.30 mmol), **1d** (168.8 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (2% EtOAc/pentane) to give **3l** (65.5 mg, 80%) as a colourless oil;

R_f (2% EtOAc/pentane) 0.4;

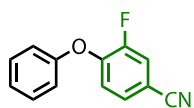
δ_H (400 MHz, CDCl₃) 7.59 (1H, dd, *J* = 8.4, 1.0 Hz), 7.43 – 7.36 (2H, m), 7.33 (1H, ddd, *J* = 8.3, 2.1, 1.0 Hz), 7.22 – 7.16 (2H, m), 7.04 – 6.97 (2H, m);

δ_F (376 MHz, CDCl₃) -62.62 (s);

δ_C (101 MHz, CDCl₃) 156.1, 153.4, 131.5, 130.6 (q, *J* = 33.0 Hz), 130.3, 129.5 (q, *J* = 1.4 Hz), 124.5, 123.3 (q, *J* = 272.5 Hz), 121.0 (q, *J* = 4.3 Hz), 118.7, 116.9 (q, *J* = 3.75 Hz);

HRMS (ESI) calculated for C₁₃H₉³⁵ClF₃O (M+H⁺): 273.0452; found: 273.0473

3-Fluoro-4-phenoxybenzonitrile (3m)



This compound was synthesised according to **general procedure 1** using phenol (28.2 mg, 0.30 mmol), **1c** (151.0 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3m** (44.0 mg, 88%) as a colourless oil;

R_f (10% EtOAc/pentane) 0.3;

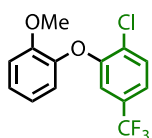
δ_H (400 MHz, CDCl₃) 7.47 (1H, dd, *J* = 10.1, 2.0 Hz), 7.44 – 7.39 (2H, m), 7.37 (1H, ddd, *J* = 8.5, 2.0, 1.3 Hz), 7.26 – 7.20 (1H, m), 7.08 – 7.04 (2H, m), 6.96 (1H, t, *J* = 8.5 Hz);

δ_F (376 MHz, CDCl₃) -129.02 – -129.10 (m);

δ_C (101 MHz, CDCl₃) 155.0, 153.0 (d, *J* = 252.4 Hz), 149.8 (d, *J* = 10.9 Hz), 130.4, 129.5 (d, *J* = 4.0 Hz), 125.3, 120.9 (d, *J* = 21.3 Hz), 119.8, 119.5, 117.7 (d, *J* = 2.6 Hz), 106.8 (d, *J* = 8.2 Hz).

HRMS (ESI) calculated for C₁₃H₈FNONa (M+Na⁺): 236.0482; found: 236.0486.

1-Chloro-2-(2-methoxyphenoxy)-4-(trifluoromethyl)benzene (3n)



This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (37.2 mg, 0.30 mmol), **1d** (168.8 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (0% -> 5% EtOAc/PE) to give **3n** (56.7 mg, 63%) as a colourless oil;

R_f (5% EtOAc/PE) 0.45;

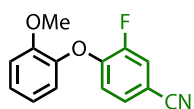
δ_H (400 MHz, CDCl₃) 7.55 (1H, dd, *J* = 10.3, 2.0 Hz), 7.26 – 7.19 (2H, m), 7.06 – 6.95 (3H, m), 6.91 (1H, d, *J* = 1.7 Hz), 3.81 (3H, s);

δ_F (376 MHz, CDCl₃) -62.58 (3H, s);

δ_C (101 MHz, CDCl₃) 154.2, 151.3, 143.5, 131.1, 130.2 (q, *J* = 33.4 Hz), 127.6 (q, *J* = 1.3 Hz), 126.3, 123.6 (q, *J* = 272.4 Hz), 121.5, 121.5, 119.8 (q, *J* = 3.8 Hz), 113.9 (q, *J* = 3.8 Hz), 113.3, 56.1.

HRMS calculated for C₁₄H₁₀³⁵ClF₃O₂Na (M+Na⁺): 325.0211; found: 325.0214.

3-Fluoro-4-(2-methoxyphenoxy)benzonitrile (3o)



This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (33 μL, 0.30 mmol), **1c** (151.0 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (10% EtOAc/PE) to give **3o** (52.4 mg, 72%) as a colourless oil;

R_f (10% EtOAc/PE) 0.30;

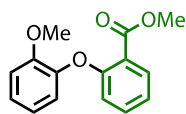
δ_H (400 MHz, CDCl₃) 7.44 (1H, dd, *J* = 10.3, 2.0 Hz), 7.31 – 7.28 (1H, m), 7.25 (1H, ddd, *J* = 8.2, 7.4, 1.7 Hz), 7.10 (1H, dd, *J* = 7.9, 1.7 Hz), 7.03 (1H, dd, *J* = 8.3, 1.4 Hz), 6.99 (1H, td, *J* = 7.7, 1.4 Hz), 6.73 (1H, t, *J* = 8.4 Hz), 3.79 (3H, s);

δ_F (376 MHz, CDCl₃) -130.74 (dd, *J* = 10.1, 8.1 Hz);

δ_C (101 MHz, CDCl₃) 151.8 (d, *J* = 258.6 Hz), 151.4, 150.7 (d, *J* = 17.5 Hz), 142.5, 129.3 (d, *J* = 3.8 Hz), 127.0, 122.2, 121.5, 120.5 (d, *J* = 21.2 Hz), 118.0, (d, *J* = 2.6 Hz), 117.5 (d, *J* = 1.9 Hz), 113.2, 105.6 (d, *J* = 7.5 Hz), 56.0.

HRMS calculated for C₁₄H₁₀FNO₂Na (M+Na⁺): 266.0579; found: 266.0588.

Methyl 2-(2-methoxyphenoxy)benzoate (3p)



This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (37.2 mg, 0.30 mmol), **1e** (155.5 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (0% -> 5% EtOAc/PE) to give **3p** (47.9 mg, 62%) as a colourless oil;

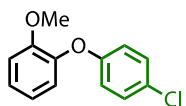
R_f (5% EtOAc/PE) 0.30;

δ_H (400 MHz, CDCl₃) 7.90 (1H, dd, *J* = 7.8, 1.8 Hz), 7.39 (1H, ddd, *J* = 8.3, 7.3, 1.8 Hz), 7.15 – 7.07 (2H, m), 7.01 – 6.98 (1H, m), 6.91 – 6.87 (2H, m), 6.83 (1H, dd, *J* = 8.3, 1.1 Hz), 3.84 (3H, s), 3.84 (3H, s);

δ_C (101 MHz, CDCl₃) 166.5, 157.3, 151.2, 145.8, 133.5, 131.9, 124.8, 122.7, 121.9, 121.3, 120.3, 118.8, 113.1, 56.2, 52.3;

HRMS calculated for C₁₅H₁₄O₄Na (M+Na⁺): 281.0777; found: 281.0784.

1-(4-Chlorophenoxy)-2-methoxybenzene (3q)



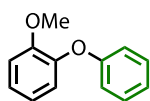
This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (33.0 μL, 0.30 mmol), **1b** (155.0 mg, 0.30 mmol, 1.0 equiv) and MeCN as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (PE) to give **3q** (54.7 mg, 81%) as a colourless oil;

R_f (PE) 0.45;

δ_H (400 MHz, CDCl₃) 7.60 – 7.55 (2H, m), 7.48 – 7.42 (2H, m), 7.39 – 7.32 (2H, m), 7.10 – 6.99 (2H, m), 3.84 (3H, s);

δ_C (101 MHz, CDCl₃) 156.6, 138.7, 131.0, 130.9, 129.7, 128.7, 128.1, 127.0, 121.0, 111.4, 55.7. This data is consistent with literature precedent.²⁸

1-Methoxy-2-phenoxybenzene (3r)



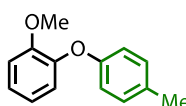
This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (33.0 μL, 0.30 mmol), **1f** (155.0 mg, 0.30 mmol, 1.0 equiv) and MeCN as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (PE) to give **3r** (47.2 mg, 76%) as a colourless oil;

R_f (PE) 0.45;

δ_H (400 MHz, CDCl₃) 7.34 – 7.25 (2H, m), 7.14 (1H, ddd, *J* = 8.2, 7.2, 1.8 Hz), 7.08 – 6.90 (6H, m), 3.85 (3H, m);

δ_C (101 MHz, CDCl₃) 158.1, 151.6, 145.2, 129.6, 124.9, 122.6, 121.2, 121.2, 117.3, 113.0, 56.1. This data is consistent with literature precedent.²⁹

1-Methoxy-2-(p-tolyloxy)benzene (3s)



This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (33.0 μL, 0.30 mmol), **1g** (155.0 mg, 0.30 mmol, 1.0 equiv) and MeCN as LAG. After addition of iodonium salt, the reaction was milled

for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (PE) to give **3s** (26.9 mg, 47%) as a colourless oil;

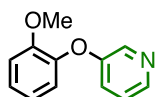
R_f (PE) 0.45;

δ_H (400 MHz, $CDCl_3$) 7.14 – 7.08 (3H, m), 7.01 (1H, d, $J = 8.7$ Hz), 6.97 – 6.85 (4H, m), 3.86 (3H, s), 2.33 (3H, s).

δ_C (101 MHz, $CDCl_3$) 155.6, 151.3, 145.9, 131.2, 130.1, 124.4, 121.1, 120.5, 117.6, 112.8, 56.1, 20.7.

This data is consistent with literature precedent.³⁰

3-(2-methoxyphenoxy)pyridine (**3t**)



This compound was synthesised according to **general procedure 1** using 2-methoxyphenol (33.0 μ L, 0.30 mmol), **1p** (138.4 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (30% EtOAc/pentane) to give **3t** (63.1 mg, 79%) as a colourless oil;

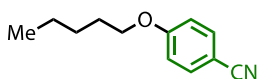
R_f (30% EtOAc/pentane) 0.4;

δ_H (400 MHz, $CDCl_3$) 8.34 (1H, dd, $J = 2.7, 0.9$ Hz), 8.29 (1H, dd, $J = 4.3, 1.7$ Hz), 7.23 – 7.14 (3H, m), 7.02 (2H, ddd, $J = 7.9, 5.9, 1.7$ Hz), 6.99 – 6.92 (1H, m), 3.81 (3H, s);

δ_C (101 MHz, $CDCl_3$) 154.7, 151.6, 143.9, 143.6, 139.9, 125.9, 123.9, 123.5, 121.6, 121.4, 113.0, 56.0;

HRMS (ESI) calculated for $C_{12}H_{12}NO_2$ ($M+H^+$): 202.0863; found: 202.0849.

4-(Pentyloxy)benzonitrile (**3u**)



This compound was synthesised according to **general procedure 1** using 1-pentanol (33.0 μ L, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and MeCN as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3u** (51.7 mg, 91%) as a colourless oil;

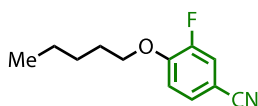
R_f (10% EtOAc/pentane) 0.40;

δ_H (400 MHz, $CDCl_3$) 7.59 – 7.54 (2H, m), 6.96 – 6.89 (2H, m), 3.99 (2H, t, $J = 6.6$ Hz), 1.80 (2H, dq, $J = 8.0, 6.5$ Hz), 1.50 – 1.32 (4H, m), 0.93 (3H, t, $J = 7.1$ Hz);

δ_C (101 MHz, $CDCl_3$) 162.6, 134.1, 119.5, 115.3, 103.8, 68.5, 28.8, 28.2, 22.5, 14.1.

This data is consistent with literature precedent.³¹

3-Fluoro-4-(pentyloxy)benzonitrile (**3v**)



This compound was synthesised according to **general procedure 1** using 1-pentanol (33.0 μ L, 0.30 mmol), **1c** (151.0 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. After addition of iodonium salt, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (5% EtOAc/pentane) to give **3v** (34.1 mg, 55%) as a colourless oil;

R_f (10% EtOAc/pentane) 0.45;

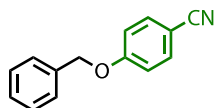
δ_H (400 MHz, $CDCl_3$) 7.40 (1H, ddd, $J = 8.5, 2.0, 1.3$ Hz), 7.35 (1H, dd, $J = 10.6, 2.0$ Hz), 6.99 (1H, t, $J = 8.5$ Hz), 4.08 (2H, t, $J = 6.6$ Hz), 1.85 (2H, dq, $J = 8.1, 6.6$ Hz), 1.52 – 1.33 (4H, m), 0.94 (3H, t, $J = 7.1$ Hz);

δ_F (376 MHz, $CDCl_3$) -131.47 – -131.57 (m);

δ_C (101 MHz, $CDCl_3$) 152.1 (d, $J = 248.4$ Hz), 151.6 (d, $J = 10.2$ Hz), 129.8 (d, $J = 3.9$ Hz), 119.7 (d, $J = 21.4$ Hz), 118.3 (d, $J = 2.5$ Hz), 114.6 (d, $J = 2.4$ Hz), 103.7 (d, $J = 7.9$ Hz), 68.5, 28.8, 28.2, 22.5, 14.1.

HRMS (ESI) calculated for $C_{12}H_{14}FNNa$ ($M+Na^+$): 230.0952; found: 230.0945.

4-(Benzyloxy)benzonitrile (3w)



This compound was synthesised according to **general procedure 1** using benzyl alcohol (31.0 μ L, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3w** (55.0 mg, 88%) as a colourless oil;

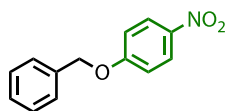
R_f (10% EtOAc/pentane) 0.20;

δ_H (400 MHz, $CDCl_3$) 7.62 – 7.56 (2H, m), 7.45 – 7.32 (5H, m), 7.05 – 6.99 (2H, m), 5.12 (2H, s);

δ_C (101 MHz, $CDCl_3$) 162.1, 135.8, 134.1, 128.9, 128.5, 127.6, 119.3, 115.7, 104.3, 70.4.

This data is consistent with literature precedent.³²

1-(Benzyloxy)-4-nitrobenzene (3x)



This compound was synthesised according to **general procedure 1** using benzyl alcohol (31.0 μ L, 0.30 mmol), **1h** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (5% EtOAc/pentane) to give **3x** (50.3 mg, 80%) as a colourless oil;

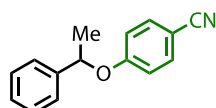
R_f (10% EtOAc/pentane) 0.4;

δ_H (400 MHz, $CDCl_3$) 8.24 – 8.17 (2H, m), 7.47 – 7.34 (5H, m), 7.06 – 7.00 (2H, m), 5.17 (2H, s);

δ_C (101 MHz, $CDCl_3$) 163.8, 141.8, 135.6, 128.9, 128.6, 127.6, 126.1, 115.0, 70.8.

This data is consistent with literature precedent.³³

4-(1-Phenylethoxy)benzonitrile (3y)



This compound was synthesised according to **general procedure 1** using 1-phenyl ethanol (36.0 μ L, 0.30 mmol), **1a** (145.6 mg, 0.30 mmol, 1.0 equiv) and EtOAc as LAG. The residue was purified by flash chromatography (10% EtOAc/pentane) to give **3y** (39.8 mg, 59%) as a colourless oil;

R_f (10% EtOAc/pentane) 0.2;

δ_H (400 MHz, $CDCl_3$) 7.51 – 7.45 (2H, m), 7.38 – 7.27 (5H, m), 6.92 – 6.87 (2H, m), 5.35 (1H, q, $J = 6.4$ Hz), 1.66 (3H, d, $J = 6.4$ Hz);

δ_C (101 MHz, $CDCl_3$) 161.4, 142.0, 134.0, 129.0, 128.0, 125.5, 119.3, 116.6, 103.9, 76.7, 24.5.

This data is consistent with literature precedent.³⁴

5.3 S-Arylation

General procedure 2: S-arylation

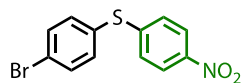
In prose:

Sodium carbonate (31.8 mg, 0.30 mmol, 1 equiv), nucleophile (0.30 mmol), diaryliodonium salt **1** (0.30 mmol, 1 equiv), cyclopentanone (0.025 $\mu\text{L}/\text{mg}$) and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 30 minutes at 30 Hz. The contents of the vessel were washed out using EtOAc (5 x 0.75 mL) and concentrated *in vacuo*. This residue was purified using silica gel flash chromatography to provide product **4**.

In recipe style:

- Sodium carbonate (31.8 mg, 0.30 mmol, 1 equiv), nucleophile (0.30 mmol), diaryliodonium salt **1** (0.30 mmol, 1 equiv), cyclopentanone (0.025 $\mu\text{L}/\text{mg}$), and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel.
- Mill for 30 minutes at 30 Hz.
- Contents of vessel washed out with EtOAc (5 x 0.75 mL).
- Concentrate *in vacuo*.
- Purify residue using silica gel flash chromatography to provide products **5**.

(4-bromophenyl)(4-nitrophenyl)sulfane (**4a**)



This compound was synthesised according to **general procedure 2** using 4-bromothiophenol (56.7 mg, 0.30 mmol) and **1h** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash column chromatography (2% EtOAc/pentane) to give **4a** (84.0 mg, 91%) as a yellow solid;

Large scale reaction: Sodium carbonate (318 mg, 3.00 mmol, 1 equiv), 4-bromothiophenol (567 mg, 3.00 mmol, 1 equiv), one 10 mm stainless steel ball and cyclopentanone (0.06 mL), **1h** (1.516 g, 3.00 mmol, 1 equiv) were added to a 5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 30 minutes at 30 Hz. The contents of the vessel were washed out using EtOAc (5 x 4 mL) and concentrated *in vacuo*. This residue was purified using silica gel flash chromatography (2% EtOAc/pentane) to provide product **4a** (724 mg, 78%) as a yellow solid;

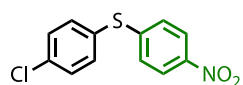
R_f (2% EtOAc/pentane) 0.33;

δ_H (400 MHz, CDCl_3) 8.11 – 8.06 (2H, m), 7.60 – 7.55 (2H, m), 7.42 – 7.37 (2H, m), 7.22 – 7.18 (2H, m);

δ_C (101 MHz, CDCl_3) 147.5, 145.8, 138.3, 136.1, 133.4, 130.1, 127.3, 124.3.

This data is consistent with literature precedent.³⁵

(4-Chlorophenyl)(4-nitrophenyl)sulfane (**4b**)



This compound was synthesised according to **general procedure 2** using 4-chlorothiophenol (43.4 mg, 0.30 mmol) and **1h** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **4g** (70.4 mg, 88%) as a yellow solid.

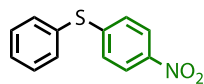
R_f (2% EtOAc/pentane) 0.35;

δ_{H} (400 MHz, CDCl_3) 8.12 – 8.06 (2H, m), 7.50 – 7.40 (4H, m), 7.22 – 7.16 (2H, m);

δ_{C} (101 MHz, CDCl_3) 147.7, 145.8, 136.2, 136.0, 130.4, 129.3, 127.1, 124.3.

This data is consistent with literature precedent.³⁵

(4-Nitrophenyl)(phenyl)sulfane (4c)



This compound was synthesised according to **general procedure 2** using thiophenol (33.1 mg, 0.30 mmol) and **1h** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **4c** (51.9 mg, 75%) as a yellow solid.

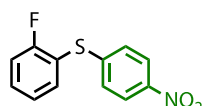
R_{f} (2% EtOAc/pentane) 0.55;

δ_{H} (400 MHz, CDCl_3) 8.10 – 8.03 (2H, m), 7.57 – 7.52 (2H, m), 7.49 – 7.43 (3H, m), 7.21 – 7.15 (2H, m);

δ_{C} (101 MHz, CDCl_3) 148.7, 145.5, 134.9, 130.6, 130.2, 129.8, 126.9, 124.2.

This data is consistent with literature precedent.³⁶

(2-Fluorophenyl)(4-nitrophenyl)sulfane (4d)



This compound was synthesised according to **general procedure 2** using 2-fluoro thiophenol (38.4 mg, 0.30 mmol) and **1h** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **4d** (61.3 mg, 82%) as a yellow solid;

R_{f} (2% EtOAc/pentane) 0.45;

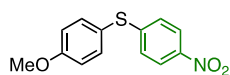
δ_{H} (400 MHz, CDCl_3) 8.11 – 8.07 (2H, m), 7.57 (1H, ddd, $J = 7.8, 7.1, 1.9$ Hz), 7.50 (1H, dddd, $J = 8.3, 7.8, 5.1, 1.9$ Hz), 7.27 – 7.17 (4H, m);

δ_{F} (376 MHz, CDCl_3) -105.59 – -105.70 (m);

δ_{C} (101 MHz, CDCl_3) 162.7 (d, $J = 251.0$ Hz), 146.5, 145.8, 137.0, 132.5 (d, $J = 7.8$ Hz), 126.8, 125.6 (d, $J = 3.9$ Hz), 124.3, 117.8 (d, $J = 18.1$ Hz), 117.0 (d, $J = 22.7$ Hz).

This data is consistent with literature precedent.³⁷

(4-Methoxyphenyl)(4-nitrophenyl)sulfane (4e)



This compound was synthesised according to **general procedure 2** using 4-methoxythiophenol (42.1 mg, 0.30 mmol) and **1h** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **4e** (64.4 mg, 82%) as a yellow solid;

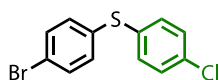
R_{f} (5% EtOAc/pentane) 0.35;

δ_{H} (400 MHz, CDCl_3) 8.04 – 8.00 (2H, m), 7.51 – 7.45 (2H, m), 7.09 – 7.06 (2H, m), 7.01 – 6.97 (2H, m), 3.87 (3H, s);

δ_{C} (101 MHz, CDCl_3) 161.2, 150.2, 137.3, 132.8, 125.7, 124.1, 120.2, 115.8, 55.6.

This data is consistent with literature precedent.³⁸

(4-Bromophenyl)(4-chlorophenyl)sulfane (4f)



This compound was synthesised according to **general procedure 2** using 4-bromothiophenol (56.7 mg, 0.30 mmol) and **1b** (148.4 mg, 0.30 mmol, 1.0

equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **4f** (75.4 mg, 84%) as a colourless solid.

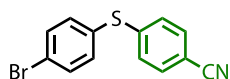
R_f (2% EtOAc/pentane) 0.40;

δ_H (400 MHz, $CDCl_3$) 7.47 – 7.43 (2H, m), 7.33 – 7.27 (4H, m), 7.22 – 6.17 (2H, m);

δ_C (101 MHz, $CDCl_3$) 134.9, 133.8, 133.7, 132.7, 132.5, 132.5, 129.7, 121.5.

This data is consistent with literature precedent.³⁹

4-((4-Bromophenyl)thio)benzonitrile (4g)



This compound was synthesised according to **general procedure 2** using 4-bromothiophenol (56.7 mg, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2%

EtOAc/pentane) to give **4g** (72.2 mg, 83%) as a colourless solid.

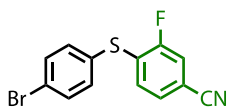
R_f (2% EtOAc/pentane) 0.25;

δ_H (400 MHz, $CDCl_3$) 7.57 – 7.53 (2H, m), 7.52 – 7.47 (2H, m), 7.38 – 7.33 (2H, m), 7.21 – 7.16 (2H, m);

δ_C (101 MHz, $CDCl_3$) 144.8, 135.8, 133.2, 132.6, 130.5, 127.8, 124.0, 118.7, 109.4.

This data is consistent with literature precedent.⁴⁰

4-((4-Bromophenyl)thio)-3-fluorobenzonitrile (4h)



This compound was synthesised according to **general procedure 2** using 4-bromothiophenol (56.7 mg, 0.30 mmol) and **1c** (157.0 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (5%

EtOAc/pentane) to give **4h** (91.2 mg, 92%) as a colourless solid;

R_f (5% EtOAc/pentane) 0.35;

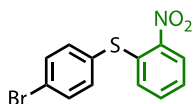
δ_H (400 MHz, $CDCl_3$) 7.60 – 7.54 (2H, m), 7.39 – 7.25 (4H, m), 6.95 (1H, dd, $J = 8.2, 7.3$ Hz);

δ_F (376 MHz, $CDCl_3$) -108.09 (dd, $J = 9.1, 7.2$ Hz);

δ_C (101 MHz, $CDCl_3$) 158.6 (d, $J = 249.1$ Hz), 136.0, 133.4, 132.9 (d, $J = 16.9$ Hz), 129.8 (d, $J = 2.7$ Hz), 128.7 (d, $J = 1.8$ Hz), 128.6 (d, $J = 3.9$ Hz), 124.4, 118.9 (d, $J = 24.8$ Hz), 117.6 (d, $J = 2.8$ Hz), 110.7 (d, $J = 9.0$ Hz).

This data is consistent with literature precedent.⁴⁰

(4-Bromophenyl)(2-nitrophenyl)sulfane (4i)



This compound was synthesised according to **general procedure 2** using 4-bromothiophenol (56.7 mg, 0.30 mmol) and **1l** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2%

EtOAc/pentane) to give **4i** (70.7 mg, 76%) as a colourless solid.

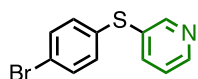
R_f (2% EtOAc/pentane) 0.3;

δ_H (400 MHz, $CDCl_3$) 8.22 (1H, dd, $J = 8.2, 1.5$ Hz), 7.63 – 7.59 (2H, m), 7.47 – 7.42 (2H, m), 7.37 (1H, ddd, $J = 8.2, 7.2, 1.5$ Hz), 7.24 (1H, dd, $J = 8.2, 7.2, 1.3$ Hz), 6.86 (1H, dd, $J = 8.2, 1.5$ Hz);

δ_C (101 MHz, $CDCl_3$) 145.3, 138.7, 137.4, 133.7, 133.5, 130.4, 128.4, 125.9, 125.4, 124.9.

This data is consistent with literature precedent.⁴¹

3-((4-Bromophenyl)thio)pyridine (4j)



This compound was synthesised according to **general procedure 2** using 4-bromothiophenol (56.7 μ L, 0.30 mmol) and **1p** (138.4 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (20% \rightarrow 30% EtOAc/pentane) to give **4j** (69.7 mg, 87%) as a colourless oil.

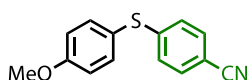
R_f (30% EtOAc/ pentane) 0.33;

δ_H (400 MHz, $CDCl_3$) 8.57 (1H, d, $J = 2.4$ Hz), 8.49 (1H, dd, $J = 4.8, 1.6$ Hz), 7.61 (1H, dt, $J = 8.0, 2.0$ Hz), 7.48 – 7.41 (2H, m), 7.25 – 7.16 (3H, m);

δ_C (101 MHz, $CDCl_3$) 151.6, 148.5, 138.5, 133.7, 133.0, 132.8, 132.7, 124.2, 122.0.

This data is consistent with literature precedent.⁴²

4-((4-Methoxyphenyl)thio)benzonitrile (4k)



This compound was synthesised according to **general procedure 2** using 4-methoxythiophenol (37 μ L, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (5% EtOAc/pentane) to give **4k** (51.0 mg, 70%) as a colourless solid;

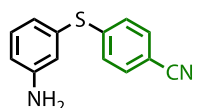
R_f (5% EtOAc/pentane) 0.20;

δ_H (400 MHz, $CDCl_3$) 7.51 – 7.41 (4H, m), 7.10 – 7.04 (2H, m), 7.00 – 6.94 (2H, m), 3.86 (3H, s);

δ_C (101 MHz, $CDCl_3$) 161.1, 147.5, 137.2, 132.4, 126.2, 120.5, 119.1, 115.7, 108.2, 55.6.

This data is consistent with literature precedent.³⁶

4-((3-Aminophenyl)thio)benzonitrile (4l)



This compound was synthesised according to **general procedure 2** using 3-aminothiophenol (37.6 mg, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv). Additionally, instead of cyclopentanone, ethyl acetate (60 μ L) was used as LAG. The residue was purified by flash chromatography (35% EtOAc/pentane) to give **4l** (45.4 mg, 67%) as a colourless solid;

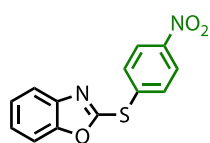
R_f (35% EtOAc/pentane) 0.45;

δ_H (400 MHz, $CDCl_3$) 7.49 – 7.43 (2H, m), 7.22 – 7.16 (3H, m), 6.87 (1H, ddd, $J = 7.6, 1.7, 1.0$ Hz), 6.82 (1H, t, $J = 2.4$ Hz), 6.72 (1H, ddd, $J = 7.6, 2.4, 1.0$ Hz), 3.78 (2H, s);

δ_C (101 MHz, $CDCl_3$) 147.9, 146.1, 132.4, 131.4, 130.8, 127.4, 124.3, 120.4, 119.0, 116.1, 108.6.

This data is consistent with literature precedent.⁴³

2-((4-Nitrophenyl)thio)benzo[d]oxazole (4m)



This compound was synthesised according to **general procedure 2** using 2-mercaptobenzoxazole (45.4 mg, 0.30 mmol) and **1h** (151.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **4m** (62.6 mg, 76%) as a colourless solid;

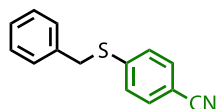
R_f (2% EtOAc/pentane) 0.20;

δ_H (400 MHz, $CDCl_3$) 8.31 – 8.25 (2H, m), 7.88 – 7.83 (2H, m), 7.69 – 7.63 (1H, m), 7.50 – 7.45 (1H, m), 7.36 – 7.30 (2H, m);

δ_c (101 MHz, $CDCl_3$) 160.5, 152.0, 148.2, 141.7, 136.8, 133.2, 125.3, 125.0, 124.6, 119.6, 110.4.

This data is consistent with literature precedent.⁴⁴

4-(Benzylthio)benzonitrile (4n)



This compound was synthesised according to **general procedure 2** using benzylmercaptan (37.3 mg, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv). After addition of **1a**, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (5% -> 10% EtOAc/pentane) to give **4n** (33.0 mg, 49%) as a yellow oil.

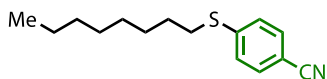
R_f (10% EtOAc/ pentane) 0.4;

δ_H (400 MHz, $CDCl_3$) 7.52 – 7.48 (2H, m), 7.38 – 7.27 (7H, m), 4.20 (2H, s);

δ_c (101 MHz, $CDCl_3$) 144.6, 135.9, 132.4, 128.9, 128.8, 127.8, 127.5, 118.9, 108.7, 37.2.

This data is consistent with literature precedent.⁴⁵

4-(Octylthio)benzonitrile (4o)



This compound was synthesised according to **general procedure 2** using octanethiol (52.0 μ L, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv). After addition of **1a**, the reaction was milled for 120 minutes instead of 30 minutes. The residue was purified by flash chromatography (2% -> 5% EtOAc/pentane) to give **4p** (37.5 mg, 51%) as a colourless oil.

R_f (5% EtOAc/ pentane) 0.6;

δ_H (400 MHz, $CDCl_3$) 7.54 – 7.49 (2H, m), 7.31 – 7.27 (2H, m), 2.97 (2H, t, $J = 7.1$ Hz), 1.69 (2H, p, $J = 7.6$ Hz), 1.49 – 1.40 (m, 2H), 1.36 – 1.22 (m, 8H), (3H, t, $J = 7.1$ Hz);

δ_c (101 MHz, $CDCl_3$) 145.5, 132.3, 126.8, 119.1, 108.0, 32.0, 31.9, 29.3, 29.2, 29.0, 28.7, 22.8, 14.2.

This data is consistent with literature precedent.⁴⁶

5.4 C-Arylation

General procedure 3: C-arylation

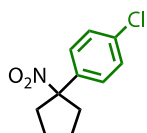
In prose:

Potassium *tert*-butoxide (43.0 mg, 0.36 mmol, 1.2 equiv), nucleophile (0.30 mmol), and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 30 minutes at 30 Hz. Then diaryliodonium salt **1** (0.30 mmol, 1 equiv) was added. This mixture was milled for 120 minutes. The contents of the vessel were washed out using EtOAc (5 x 0.75 mL) and concentrated *in vacuo*. This residue was purified using silica gel flash chromatography to provide products **5**.

In recipe style:

- Potassium *tert*-butoxide (43.0 mg, 0.36 mmol, 1.2 equiv), nucleophile (0.30 mmol), and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel.
- Mill for 30 minutes at 30 Hz.
- Add diaryliodonium salt **1** (0.30 mmol, 1.0 equiv).
- Mill for 120 minutes at 30 Hz.
- Contents of vessel washed out with EtOAc (5 x 0.75 mL).
- Concentrate *in vacuo*.
- Purify residue using silica gel flash chromatography to provide products **5**.

1-Chloro-4-(1-nitrocyclopentyl)benzene (5a)



This compound was synthesised according to **general procedure 3** using nitrocyclopentane (32.0 μ L, 0.30 mmol) and **1b** (148.4 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (1% Et₂O/pentane) to give **5a** (62.1 mg, 99%) as a colourless oil;

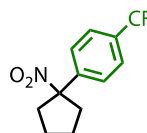
R_f (1% Et₂O/pentane) 0.40;

δ_{H} (400 MHz, CDCl₃) 7.47 – 7.41 (2H, m), 7.37 – 7.32 (2H, m), 3.24 – 3.13 (2H, m), 2.21 – 2.07 (2H, m), 1.94 – 1.77 (4H, m);

δ_{C} (101 MHz, CDCl₃) 137.2, 135.4, 129.0, 128.5, 101.5, 37.0, 22.9.

This data is consistent with literature precedent.⁹

1-(1-Nitrocyclopentyl)-4-(trifluoromethyl)benzene (5b)



This compound was synthesised according to **general procedure 3** using nitrocyclopentane (32.0 μ L, 0.30 mmol) and **1m** (158.5 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (1% Et₂O/pentane) to give **5b** (75.2 mg, 96%) as a colourless oil;

R_f (1% EtOAc/pentane) 0.50;

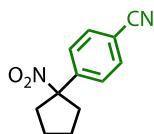
δ_{H} (400 MHz, CDCl₃) 7.68 – 7.59 (4H, m), 3.27 – 3.18 (2H, m), 2.25 – 2.12 (2H, m), 1.94 – 1.83 (4H, m);

δ_{F} (376 MHz, CDCl₃) -62.88;

δ_{C} (101 MHz, CDCl₃) 142.3, 131.5 (q, *J* = 32.3 Hz), 127.9, 125.8 (q, *J* = 3.3 Hz), 123.8 (q, *J* = 270.6 Hz), 101.6, 37.1, 22.9.

This data is consistent with literature precedent.⁹

4-(1-Nitrocyclopentyl)benzonitrile (5c)



This compound was synthesised according to **general procedure 3** using nitrocyclopentane (32.0 μ L, 0.30 mmol) and **1a** (145.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (5% EtOAc/pentane) to give **5c** (62.0 mg, 96%) as a colourless oil.

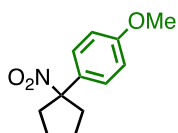
R_f (5% EtOAc/pentane) 0.30;

δ_{H} (400 MHz, CDCl_3) 7.74 – 7.66 (2H, m), 7.63 – 7.58 (2H, m), 3.22 – 3.15 (2H, m), 2.22 – 2.11 (2H, m), 1.97 – 1.80 (4H, m);

δ_{C} (101 MHz, CDCl_3) 143.2, 132.6, 127.8, 118.2, 113.3, 101.5, 37.1, 22.9;

HRMS (ESI) calculated for $\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}$ ($\text{M}+\text{Na}^+$): 239.0815; found: 239.0791.

1-methoxy-4-(1-nitrocyclopentyl)benzene (5d)



This compound was synthesised according to **general procedure 3** using nitrocyclopentane (32.0 μ L, 0.30 mmol) and **1n** (153.4 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane \rightarrow 5% EtOAc/pentane) to give **5d** (55.0 mg, 83%) as a colourless oil.

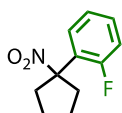
R_f (2% EtOAc/pentane) 0.30;

δ_{H} (400 MHz, CDCl_3) 7.48 – 7.42 (2H, m), 6.91 – 6.84 (2H, m), 3.80 (3H, s), 3.24 – 3.15 (2H, m), 2.19 – 2.09 (2H, m), 1.92 – 1.75 (4H, m);

δ_{C} (101 MHz, CDCl_3) 159.9, 130.7, 128.1, 113.7, 101.5, 55.1, 36.6, 22.6;

This data is consistent with literature precedent.⁹

1-Fluoro-2-(1-nitrocyclopentyl)benzene (5e)



This compound was synthesised according to **general procedure 1** using nitrocyclopentane (32 μ L, 0.30 mmol) and **1o** (139.8 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (1% Et_2O /pentane) to give **5e** (46.8 mg, 76%) as a colourless oil;

R_f (1% EtOAc/pentane) 0.45;

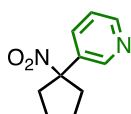
δ_{H} (400 MHz, CDCl_3) 7.47 (1H, td, $J = 7.8, 1.7$ Hz), 7.37 (1H, tdd, $J = 7.8, 5.1, 1.7$ Hz), 7.18 (1H, td, $J = 7.6, 1.7$ Hz), 7.08 (1H, ddd, $J = 11.4, 7.8, 1.7$ Hz), 3.12 – 3.02 (2H, m), 2.33 – 2.20 (2H, m), 2.00 – 1.81 (4H, m);

δ_{F} (376 MHz, CDCl_3) -110.55 - -110.66 (m);

δ_{C} (101 MHz, CDCl_3) 160.9 (d, $J = 249.6$ Hz), 131.0 (d, $J = 8.7$ Hz), 128.3 (d, $J = 3.7$ Hz), 126.8 (d, $J = 13.5$ Hz), 124.3 (d, $J = 3.5$ Hz), 116.5 (d, $J = 22.4$ Hz), 98.9, 37.6 (d, $J = 2.1$ Hz), 23.4.

This data is consistent with literature precedent.⁹

3-(1-nitrocyclopentyl)pyridine (5f)



This compound was synthesised according to **general procedure 3** using nitrocyclopentane (32.0 μ L, 0.30 mmol) and **1p** (138.4 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (40% EtOAc/pentane) to give **5f** (53.1 mg, 92%) as a colourless oil.

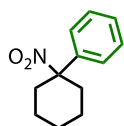
R_f (40% EtOAc/pentane) 0.25;

δ_H (400 MHz, CDCl₃) 8.77 (1H, dd, *J* = 2.5, 0.9 Hz), 8.60 (1H, dd, *J* = 4.8, 1.6 Hz), 7.80 (1H, ddd, *J* = 8.0, 2.5, 1.6 Hz), 7.30 (1H, ddd, *J* = 8.1, 4.8, 0.9 Hz), 3.34 – 3.12 (2H, m), 2.27 – 2.10 (2H, m), 1.93 – 1.67 (4H, m);

δ_C (101 MHz, CDCl₃) 150.4, 148.4, 134.8, 134.3, 123.5, 100.3, 36.8, 22.8.

This data is consistent with literature precedent.⁹

(1-Nitrocyclohexyl)benzene (5g)



This compound was synthesised according to **general procedure 3** using nitrocyclohexane (38.7 mg, 0.30 mmol) and **1f** (145.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (0% → 5% EtOAc/pentane) to give **5g** (57.3 mg, 92%) as a colourless oil;

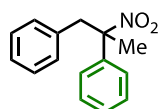
R_f (5% EtOAc/pentane) 0.60;

δ_H (400 MHz, CDCl₃) 7.52 – 7.46 (2H, m), 7.41 – 7.32 (3H, m), 2.89 – 2.80 (2H, m), 2.12 (2H, ddd, *J* = 14.6, 10.9, 3.9 Hz), 1.78 – 1.49 (5H, m), 1.42 – 1.31 (1H, m);

δ_C (101 MHz, CDCl₃) 139.6, 129.1, 129.0, 125.5, 92.8, 35.0, 24.7, 22.9.

This data is consistent with literature precedent.⁹

(2-Nitropropyl)benzene (5h)



This compound was synthesised according to **general procedure 3** using (2-nitropropyl)benzene **2c** (49.6 mg, 0.30 mmol) and **1f** (145.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (0% → 5%

EtOAc/pentane) to give **5h** (66.9 mg, 95%) as a colourless oil;

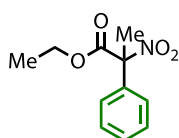
R_f (5% EtOAc/pentane) 0.60;

δ_H (400 MHz, CDCl₃) 7.42 – 4.38 (5H, m), 7.25 – 7.19 (3H, m), 6.98 – 6.93 (2H, m), 3.87 (1H, d, *J* = 13.8 Hz), 3.52 (1H, d, *J* = 13.8 Hz), 1.86 (3H, s);

δ_C (101 MHz, CDCl₃) 139.8, 134.8, 130.6, 129.1, 128.9, 128.5, 127.5, 125.8, 93.7, 45.7, 23.6.

This data is consistent with literature precedent.⁹

Ethyl 2-nitro-2-phenylpropanoate (5i)



This compound was synthesised according to **general procedure 3** using ethyl 2-nitropropanoate (44.1 mg, 0.30 mmol) and **1f** (145.6 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (0% → 10% EtOAc/pentane) to give **5i** (24.8 mg, 39%) as a yellow oil;

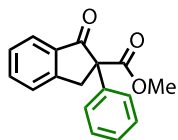
R_f (5% EtOAc/pentane) 0.40;

δ_H (400 MHz, CDCl₃) 7.48 – 7.40 (5H, m), 4.34 (2H, q, *J* = 7.1 Hz), 2.26 (3H, s), 1.31 (3H, t, *J* = 7.1 Hz);

δ_C (101 MHz, CDCl₃) 167.3, 134.2, 130.0, 128.7, 127.6, 95.2, 63.4, 23.3, 14.0.

This data is consistent with literature precedent.⁹

Methyl 1-oxo-2-phenyl-2,3-dihydro-1H-indene-2-carboxylate (5j)



This compound was synthesised according to **general procedure 3** using methyl 1-oxo-2,3-dihydro-1H-indene-2-carboxylate (**2b**, 57.1 mg, 0.30 mmol) and **1j** (138 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (5% → 10% EtOAc/pentane) to give **5i** (60.1 mg, 78%) as a

colourless oil;

R_f (10% EtOAc/pentane) 0.30;

δ_H (400 MHz, $CDCl_3$) 7.87 – 7.82 (1H, m), 7.65 (1H, td, $J = 7.6, 1.1$ Hz), 7.49 (1H, dp, $J = 7.6, 0.9$ Hz), 7.45 – 7.24 (6H, m), 4.23 (1H, d, $J = 17.2$ Hz), 3.75 (3H, s), 3.57 (1H, d, $J = 17.2$ Hz);

δ_C (101 MHz, $CDCl_3$) 200.3, 171.2, 152.2, 138.8, 135.8, 135.2, 128.8, 128.1, 127.7, 127.4, 126.3, 125.2, 65.5, 53.4, 41.0.

This data is consistent with literature precedent.⁴⁷

5.5 S-Vinylation

General procedure 4

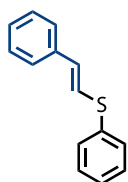
In prose:

Thiol (0.30 mmol) and potassium carbonate (41.5 mg, 0.30 mmol, 1.0 equiv), and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 30 minutes at 30 Hz. Then VBX reagent **6** (0.30 mmol, 1.0 equiv) was added. This mixture was milled for 90 minutes at 30 Hz. The contents of the vessel were washed out using EtOAc (5 x 0.75 mL) and concentrated *in vacuo*. This residue was purified using silica gel flash chromatography to provide product **7** as the only regioisomer according to NMR analysis.

In recipe style:

- Thiol (0.30 mmol) and potassium carbonate (41.5 mg, 0.30 mmol, 1.0 equiv), and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel.
- Mill for 30 minutes at 30 Hz.
- Then reagent **6** (0.30 mmol, 1.0 equiv) was added.
- Mill for 90 minutes at 30 Hz.
- Contents of vessel washed out with EtOAc (5 x 0.75 mL).
- Concentrate *in vacuo*.
- Purify residue using silica gel flash chromatography to provide products **7**.

(E)-Phenyl(styryl)sulfane (7a)



This compound was synthesised according to **general procedure 4** using thiophenol (31.0 μ L, 0.30 mmol) and **6a** (105 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (0% → 5% EtOAc/pentane) to give **7a** (49.1 mg, 77%) as a colourless oil;

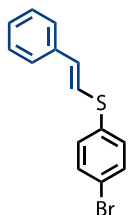
R_f (2% EtOAc/pentane) 0.35;

δ_H (400 MHz, $CDCl_3$) 7.45 – 7.40 (2H, m), 7.38 – 7.21 (8H, m), 6.90 (1H, d, $J = 15.2$ Hz), 6.74 (1H, d, $J = 15.2$ Hz);

δ_C (101 MHz, $CDCl_3$) 136.7, 135.4, 131.9, 130.0, 129.3, 128.8, 127.7, 127.1, 126.2, 123.5.

This data is consistent with literature precedent.⁴⁸

(E)-(4-Bromophenyl)(styryl)sulfane (7b)



This compound was synthesised according to **general procedure 4** using 4-bromothiophenol (56.7 mg, 0.30 mmol) and **6a** (105 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **7b** (53.2 mg, 90%) as a colourless oil.

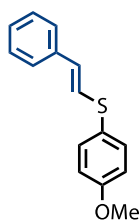
R_f (2% EtOAc/pentane) 0.40;

δ_H (400 MHz, $CDCl_3$) 7.52 – 7.21 (9H, m), 6.80 (1H, d, $J = 16.3$ Hz), 6.75 (1H, d, $J = 16.3$ Hz);

δ_C (101 MHz, $CDCl_3$) 136.4, 134.8, 133.2, 132.4, 131.3, 128.9, 128.0, 126.3, 122.4, 121.0.

This data is consistent with literature precedent.⁴⁸

(E)-(4-Methoxyphenyl)(styryl)sulfane (7c)



This compound was synthesised according to **general procedure 4** using 4-methoxythiophenol (38.0 μ L, 0.30 mmol) and **6a** (105 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **7c** (53.7 mg, 73%) as a colourless oil;

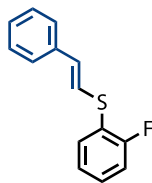
R_f (2% EtOAc/pentane) 0.34;

δ_H (400 MHz, $CDCl_3$) 7.44 – 7.39 (2H, m), 7.32 – 7.17 (5H, m), 6.94 – 6.88 (2H, m), 6.84 (1H, d, $J = 16.1$ Hz), 6.51 (1H, d, $J = 16.1$ Hz), 3.83 (3H, s);

δ_C (101 MHz, $CDCl_3$) 159.7, 136.9, 133.6, 129.1, 128.8, 127.3, 125.9, 125.8, 124.6, 115.0, 55.5.

This data is consistent with literature precedent.⁴⁸

(E)-(2-Fluorophenyl)(styryl)sulfane (7d)



This compound was synthesised according to **general procedure 4** using 2-fluorothiophenol (38.4 mg, 0.30 mmol) and **6a** (105 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (2% EtOAc/pentane) to give **7d** (49.1 mg, 90%) as a colourless oil.

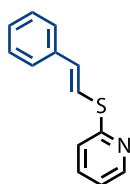
R_f (2% EtOAc/pentane) 0.33;

δ_H (400 MHz, $CDCl_3$) 7.46 (1H, td, $J = 7.5, 1.7$ Hz), 7.40 – 7.25 (4H, m), 7.20 – 7.03 (3H, m), 6.84 (1H, d, $J = 15.8$ Hz), 6.77 (1H, d, $J = 15.8$ Hz);

δ_C (101 MHz, $CDCl_3$) 160.9 (d, $J = 247.6$ Hz), 136.5, 132.8, 132.2 (d, $J = 1.4$ Hz), 131.4, 129.2 (d, $J = 8.0$ Hz), 128.8, 127.9, 126.2, 124.9 (d, $J = 4.1$ Hz), 121.7 (d, $J = 1.9$ Hz), 116.0 (d, $J = 21.9$ Hz).

This data is consistent with literature precedent.⁴⁸

(E)-3-(Styrylthio)pyridine (7e)



This compound was synthesised according to **general procedure 4** using 2-mercaptopyridine (33.3 mg, 0.30 mmol) and **6a** (105 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (0 -> 20% EtOAc/Pentane) to give **7e** (50.7 mg, 77%, E/Z 12:1) as a colourless oil;

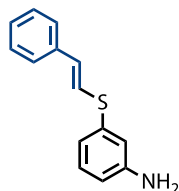
R_f (10% EtOAc/Pentane) 0.45;

δ_{H} (400 MHz, CDCl_3) 8.49 (1H, ddd, $J = 4.9, 1.9, 0.9$ Hz), 7.55 (1H, ddd, $J = 8.1, 7.4, 1.9$ Hz), 7.50 – 7.42 (3H, m), 7.38 – 7.30 (2H, m), 7.28 – 7.23 (2H, m), 7.06 (1H, ddd, $J = 7.4, 4.9, 1.9$ Hz), 6.89 (1H, d, $J = 15.8$ Hz);

δ_{C} (101 MHz, CDCl_3) 158.1, 150.0, 136.7, 136.6, 132.1, 128.8, 127.9, 126.4, 122.2, 120.4, 119.8.

This data is consistent with literature precedent.⁴⁸

(*E*)-3-(Styrylthio)aniline (7f)



This compound was synthesised according to **general procedure 4** using 3-aminothiophenol (32.0 μL , 0.30 mmol) and **6a** (105 mg, 0.30 mmol, 1.0 equiv). The residue was purified by flash chromatography (40% EtOAc/Pentane) to give **7f** (58.0 mg, 85%) as a yellow oil;

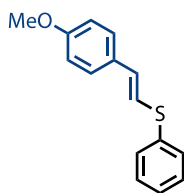
R_{f} (30% EtOAc/Pentane) 0.25;

δ_{H} (400 MHz, CDCl_3) 7.39 – 7.30 (4H, m), 7.28 – 7.21 (1H, m), 7.13 (1H, t, $J = 7.3$ Hz), 6.91 (1H, d, $J = 16.1$ Hz), 6.81 (1H, ddd, $J = 7.7, 1.8, 1.0$ Hz), 6.79 – 6.71 (2H, m), 6.58 (1H, ddd, $J = 8.0, 2.3, 1.0$ Hz), 3.71 (2H, br.s);

δ_{C} (101 MHz, CDCl_3) 147.2, 136.7, 136.3, 131.8, 130.1, 128.8, 127.7, 126.1, 123.6, 119.8, 115.9, 113.9.

This data is consistent with literature precedent.⁴⁸

(*E*)-(4-Methoxystyryl)(phenyl)sulfane (7g)



This compound was synthesised according to **general procedure 4** using thiophenol (20.0 μL , 0.20 mmol), potassium *tert*-butoxide (33.7 mg, 0.20 mmol, 1.0 equiv) and THF (50 μL). This was initially milled for 15 minutes. then **6b** (76.0 mg, 0.20 mmol, 1.0 equiv) was added and the mixture was milled for 120 minutes. The residue in the ball milling vessel was washed with $\text{Et}_2\text{O}:\text{PE}$ (1:1, 6 mL) and this organic extract was concentrated *in vacuo* to produce **7g** as a colourless solid (40.9 mg, 85%).

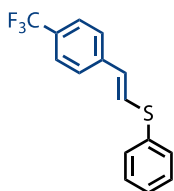
R_{f} (PE) 0.40;

δ_{H} (400 MHz, CDCl_3) 7.41 – 7.38 (2H, m), 7.36 – 7.28 (4H, m), 7.26 – 7.21 (1H, m), 6.88 – 6.84 (2H, m), 6.76 (1H, d, $J = 15.4$ Hz), 6.71 (1H, d, $J = 15.4$ Hz), 3.82 (3H, s);

δ_{C} (101 MHz, CDCl_3) 159.5, 136.1, 132.9, 129.5, 129.4, 129.2, 127.5, 126.7, 120.2, 114.3, 55.5.

This data is consistent with literature precedent.¹⁰

(*E*)-Phenyl(4-(trifluoromethyl)styryl)sulfane (7h)



This compound was synthesised according to **general procedure 4** using thiophenol (20.0 μL , 0.20 mmol), potassium *tert*-butoxide (33.7 mg, 0.20 mmol, 1.0 equiv) and THF (50 μL). This was initially milled for 15 minutes, then **6c** (83.6 mg, 0.20 mmol, 1.0 equiv) was added and the mixture was milled for 120 minutes. The residue in the ball milling vessel was washed with $\text{Et}_2\text{O}:\text{PE}$ (1:1, 6 mL) and this organic extract was concentrated *en vacuo* to produce **7h** as a colourless solid (52.3 mg, 93%);

R_{f} (PE) 0.45;

δ_{H} (400 MHz, CDCl_3) 7.58 – 7.53 (2H, m), 7.48 – 7.28 (7H, m), 7.03 (1H, d, $J = 15.4$ Hz), 6.65 (1H, d, $J = 15.4$ Hz);

δ_F (376 MHz, $CDCl_3$) -62.45;

δ_C (101 MHz, $CDCl_3$) 140.1, 134.1, 130.9, 129.5, 129.2 (q, $J = 31.7$ Hz), 128.5, 127.8, 127.8, 127.7, 126.1, 125.8 (q, $J = 3.9$ Hz), 124.6 (q, $J = 271.2$ Hz).

This data is consistent with literature precedent.¹⁰

5.6 C-Vinylation

General procedure 5: C-vinylation

In prose:

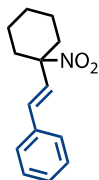
Potassium *tert*-butoxide (40.3 mg, 0.36 mmol, 1.2 equiv), the nucleophile (0.30 mmol), and one 5 mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 15 minutes at 30 Hz and then vinyliodonium salt **6d** (118 mg, 0.30 mmol, 1 equiv) was added. This mixture was milled for 10 minutes at 30 Hz. The contents of the vessel were washed out using EtOAc (5 x 0.75 mL) and concentrated *in vacuo* to provide product **8** as the only regioisomer according to NMR analysis.

No flash chromatography needed.

In recipe style:

- Potassium *tert*-butoxide (43.2 mg, 0.36 mmol, 1.2 equiv), nucleophile (0.30 mmol), and one 5mm stainless steel ball were added to a 1.5 mL stainless steel ball milling vessel.
- Mill for 15 minutes at 30 Hz.
- Add vinyliodonium salt **6d** (0.30 mmol, 1.0 equiv).
- Mill for 10 minutes at 30 Hz.
- Wash out the contents of vessel with EtOAc (5 x 0.75 mL).
- Concentrate *in vacuo* to provide products **8**. *No flash chromatography needed.*

(*E*)-(2-(1-nitrocyclohexyl)vinyl)benzene (**8a**)



This compound was synthesised according to **general procedure 5** using nitrocyclohexane (35.6 μ L, 0.30 mmol). After evaporation of the organic extract *in vacuo*, a yellow oil **8a** (65 mg, 92%) was isolated;

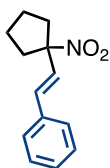
Large scale reaction: Potassium *tert*-butoxide (403 mg, 3.6 mmol, 1.2 equiv), nitrocyclohexane, and one 10 mm stainless steel ball were added to a 5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 15 minutes at 30 Hz and then vinyliodonium salt **6d** (1.181 g, 3.00 mmol, 1.0 equiv) was added. This mixture was milled for 10 minutes at 30 Hz. The contents of the vessel were washed out using EtOAc (5 x 4 mL) and concentrated *in vacuo*. The residue was purified by flash chromatography (10% EtOAc/Pentane) to give **8a** (492 mg, 71%) as a yellow oil;

δ_{H} (400 MHz, CDCl_3) 7.41 – 7.27 (5H, m), 6.71 (1H, d, $J = 16.3$ Hz), 6.26 (1H, d, $J = 16.3$ Hz), 2.61 – 2.48 (2H, m), 2.06 – 1.95 (2H, m), 1.72 – 1.61 (2H, m), 1.60 – 1.47 (3H, m), 1.46 – 1.35 (1H, m);

δ_{C} (101 MHz, CDCl_3) 135.6, 133.2, 129.2, 128.9, 128.8, 127.0, 91.1, 34.6, 24.8, 22.7.

This data is consistent with literature precedent.⁴⁹

(*E*)-(2-(1-nitrocyclopentyl)vinyl)benzene (**8b**)



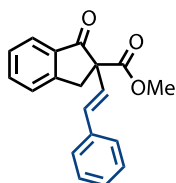
This compound was synthesised according to **general procedure 5** using nitrocyclopentane (32.0 μ L, 0.30 mmol). After evaporation of the organic extract *in vacuo*, a yellow oil **8b** (58.0 mg, 89%) was isolated.

δ_{H} (400 MHz, CDCl_3) 7.44 – 7.25 (5H, m), 6.68 (1H, d, $J = 15.9$ Hz), 6.55 (1H, d, $J = 15.9$ Hz), 2.85 – 2.76 (2H, m), 2.17 – 2.03 (2H, m), 1.86 – 1.79 (4H, m);

δ_{C} (101 MHz, CDCl_3) 135.6, 132.4, 128.8, 128.8, 128.7, 127.6, 126.9, 99.7, 37.0, 23.4;

HRMS calculated for $\text{C}_{13}\text{H}_{15}\text{NO}_2$ ($\text{M}+\text{Na}^+$): 240.0995; found: 240.0986.

Methyl (*E*)-1-oxo-2-styryl-2,3-dihydro-1H-indene-2-carboxylate (**8c**)



This compound was synthesised according to **general procedure 5** using **1b** (57.1 mg, 0.30 mmol). After addition of iodonium salt **6d**, the mixture was milled for 30 minutes instead of 10 minutes. The residue was purified by flash chromatography (20% EtOAc/pentane) to give **8c** (55.0 mg, 55%) as a yellow oil.

R_{f} (20% EtOAc/pentane) 0.33;

δ_{H} (400 MHz, CDCl_3) 7.83 – 7.80 (1H, m), 7.65 (1H, dt, $J = 7.5, 1.2$ Hz), 7.55 – 7.52 (1H, m), 7.45 – 7.35 (3H, m), 7.32 – 7.19 (3H, m), 6.76 (1H, d, $J = 16.7$ Hz), 6.52 (1H, d, $J = 16.7$ Hz), 3.95 (1H, d, $J = 17.1$ Hz), 3.76 (3H, s), 3.49 (1H, d, $J = 17.1$ Hz);

δ_{C} (101 MHz, CDCl_3) 200.1, 170.9, 152.6, 136.5, 135.7, 134.4, 131.2, 128.9, 128.7, 128.1, 128.1, 126.7, 126.5, 125.5, 63.0, 53.3, 37.9.

This data is consistent with literature precedent.⁵⁰

5.7 Catalytic tosyloxylation

Meta-chloroperbenzoic acid is known to be explosive when completely pure, and is hence used as a commercial mixture with *m*-chlorobenzoic acid. However, care should be taken when upscaling this reaction. HFIP is suspected of damaging fertility and can cause severe eye damage, so needs to be used inside a ventilated fumehood.

General procedure 6

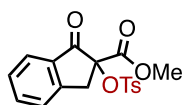
In prose:

The nucleophile (0.30 mmol, 1 equiv), 4-iodo-*m*-xylene (**9**, 9 μ L, 0.30 mmol, 0.2 equiv), *m*CPBA (92% purity) (62.4 mg, 0.30 mmol, 1 equiv), *p*TsOH·H₂O (57.1 mg, 0.30 mmol, 1 equiv), HFIP (0.5 μ L/mg) and one 10 mm stainless steel ball were added to a 5 mL stainless steel ball milling vessel. The vessel was closed and the mixture was milled for 5-30 minutes at 25 Hz. The contents of the vessel were washed out using EtOAc (5 x 0.75 mL) and washed with a saturated aqueous solution of NaHCO₃. The aqueous layer was washed with more EtOAc (3 x 5 mL). The organic layers were collected and dried over Na₂SO₄. The residue was purified by silica gel column chromatography to provide products **10**.

In recipe style:

- Add nucleophile (0.30 mmol, 1 equiv), 4-iodo-*m*-xylene (**9**, 9 μ L, 0.30 mmol, 0.2 equiv), *m*CPBA (62.4 mg, 0.30 mmol, 1 equiv), *p*TsOH·H₂O (57.1 mg, 0.30 mmol, 1 equiv), HFIP (0.5 μ L/mg) and one 10 mm stainless steel ball to a 5 mL stainless steel ball milling vessel.
- Mill for 15 minutes at 35 Hz.
- Wash out the contents of the vessel using EtOAc (5 x 0.75 mL) and wash with a saturated aqueous solution of NaHCO₃.
- Wash the aqueous layer with more EtOAc (3 x 5 mL).
- Collect the organic layers and dry over Na₂SO₄.
- Purify the residue by silica gel column chromatography to provide products **10**.

Methyl 1-oxo-2-(tosyloxy)-2,3-dihydro-1H-indene-2-carboxylate (**10a**)



This compound was synthesised according to **general procedure 6** using **2b** (57.1 mg, 0.3 mmol). The residue was purified by flash chromatography (10% EtOAc/PE) to give **10a** (94 mg, 87%) as a white solid;

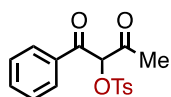
R_f (10% EtOAc/pentane) 0.15;

δ_{H} (400 MHz, CDCl₃) δ 7.91 (2H, d, *J* = 8.2 Hz), 7.80 (1H, d, *J* = 7.7 Hz), 7.69 (1H, tt, *J* = 7.7, 1.2 Hz), 7.52 (1H, d, *J* = 7.7 Hz), 7.44 (1H, t, *J* = 7.5 Hz), 7.36 (2H, d, *J* = 8.2 Hz), 4.16 (1H, d, *J* = 17.6 Hz), 3.89 (1H, d, *J* = 17.6 Hz), 3.74 (3H, s), 2.45 (3H, s).

δ_{C} (101 MHz, CDCl₃) δ 193.8, 167.0, 151.8, 145.7, 137.2, 134.8, 133.1, 130.2, 128.9, 128.6, 126.9, 126.1, 87.7, 54.1, 39.0, 22.2.

This data is consistent with literature precedent.⁵¹

1,3-Dioxo-1-phenylbutan-2-yl 4-methylbenzenesulfonate (10b)



This compound was synthesised according to **general procedure 6** using 1-phenyl-1,3-propandione (48.7 mg, 0.3 mmol) and the reaction was milled for 15 minutes. The residue was purified by flash chromatography (10% EtOAc/PE)

to give **10b** (69 mg, 69%) as a white solid;

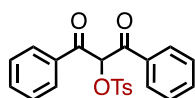
R_f (10% EtOAc/pentane) 0.2;

δ_H (400 MHz, $CDCl_3$) 15.17 (1H, s), 7.44 – 7.41 (2H, m), 7.36 – 7.32 (3H, m), 7.22 – 7.18 (2H, m), 6.95 (2H, d, $J = 8.0$ Hz), 2.39 (3H, s), 2.32 (3H, s).

δ_C (101 MHz, $CDCl_3$) δ 192.0, 180.2, 145.8, 133.8, 131.9, 131.5, 129.9, 129.2, 129.0, 128.5, 128.3, 23.1, 22.0.

This data is consistent with literature precedent.⁵²

Tosyloxydibenzoylmethane (10c):



This compound was synthesised according to **general procedure 6** using 1,3-diphenyl-1,3-propandione (67.3 mg, 0.3 mmol, 1.0 equiv) but no HFIP was used. The reaction was milled for 15 minutes. The residue was purified by

flash chromatography (10% EtOAc/PE) to give **10c** (90.7 mg, 77%) as a white solid;

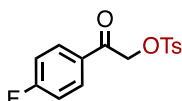
R_f (10% EtOAc/pentane) 0.2;

δ_H (400 MHz, $CDCl_3$) δ 15.62 (1H, s), 7.97 – 7.94 (4H, m), 7.79 – 7.76 (4H, m), 7.69 (2H, d, $J = 8.1$ Hz), 7.59 – 7.55 (2H, m), 7.49 – 7.35 (10H, m), 7.27 – 7.24 (2H, m), 7.20 (2H, d, $J = 8.4$ Hz), 6.87 (2H, d, $J = 8.1$ Hz), 6.67 (1H, s), 2.39 (3H, s), 2.29 (3H, s).

δ_C (101 MHz, $CDCl_3$) δ 190.1, 183.6, 145.7, 145.2, 134.5, 134.2, 133.8, 132.6, 132.1, 131.0, 129.9, 129.8, 129.5, 129.3, 128.8, 128.6, 128.4, 127.1, 84.0, 21.8, 21.7.

This data is consistent with literature precedent.⁵²

1-(4-fluorophenyl)-2-(p-tolylsulfonyloxy)ethanone (10d)



This compound was synthesised according to **general procedure 6** with modifications: 4-fluoroacetophenone (36 μL , 0.3 mmol), *m*CPBA (93.6 mg, 0.45 mmol, 1.5 equiv) and *p*TsOH·H₂O (85.6 mg, 0.45 mmol, 1.5 equiv) were

added to a vessel which had been warmed in a 60 °C oven overnight. The residue was purified by flash chromatography (10% EtOAc/PE) to give **10d** (68 mg, 74%) as a white solid;

R_f (10% EtOAc/pentane) 0.2;

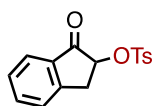
δ_H (400 MHz, $CDCl_3$) δ 7.91 – 7.87 (2H, m), 7.84 (2H, d, $J = 8.4$ Hz), 7.35 (2H, d, $J = 8.2$ Hz), 7.18 – 7.12 (2H, m), 5.21 (2H, s), 2.45 (3H, s).

δ_C (101 MHz, $CDCl_3$) δ 189.0, 167.6, 165.0, 145.4, 132.6, 130.9, 130.8, 130.3, 130.3, 129.9, 128.2, 116.3, 116.1, 69.8, 21.7.

δ_F (377 MHz, $CDCl_3$) δ -102.43 – -102.61 (m).

This data is consistent with literature precedent.⁵³

1-oxo-2,3-dihydro-1H-inden-2-yl 4-methylbenzenesulfonate (10e)



This compound was synthesised according to **general procedure 6** with modifications: 1-indanone (39.6 mg, 0.3 mmol), *m*CPBA (93.6 mg, 0.45 mmol, 1.5 equiv) and *p*TsOH·H₂O (85.6 mg, 0.45 mmol, 1.5 equiv) were added to a vessel which had been warmed in a 60°C oven overnight. The residue was purified by flash chromatography (10% EtOAc/PE) to give **10e** (37 mg, 41%) as a white solid;

R_f (10% EtOAc/pentane) 0.2;

δ_H (400 MHz, CDCl₃) δ 7.92 (2H, d, *J* = 8.3 Hz), 7.72 (1H, d, *J* = 7.7 Hz), 7.64 (1H, td, *J* = 7.5, 1.2 Hz), 7.43 (1H, dq, *J* = 7.9, 1.0 Hz), 7.40 – 7.37 (3H, m), 5.12 (1H, dd, *J* = 8.0, 4.8 Hz), 3.65 (1H, dd, *J* = 17.2, 8.0 Hz), 3.26 (1H, ddt, *J* = 17.3, 4.7, 1.1 Hz), 2.46 (3H, s).

δ_C (101 MHz, CDCl₃) δ 197.7, 150.1, 145.3, 136.5, 133.8, 133.4, 130.0, 128.6, 128.4, 126.8, 124.8, 78.4, 34.0, 21.9.

This data is consistent with literature precedent.⁵⁴

6. Benchmarking studies

6.1 Solution-phase reactions

Literature methods for solution phase arylation with diaryliodonium salts **1**, vinylation with vinylating reagents **6** and tosyloxylations with **9** were performed to evaluate the reaction efficiency compared to the mechanochemical reactions. The methods are described below, and the results are given in Table S7.

BM1: O-arylation using KO^tBu as base⁵⁵

To a solution of KO^tBu (37.0 mg, 0.33 mmol, 1.1 equiv) in dry THF (1.5 mL) under nitrogen atmosphere, was added the alcohol (0.30 mmol, 1 equiv). This mixture was stirred for 5 min at room temperature. Salt **1** (0.30 mmol) was added at 0 °C, and the solution was stirred for 1 h at 40 °C. The reaction mixture was quenched with H₂O (10 mL) and extracted with EtOAc, dried over Na₂SO₄ and concentrated *in vacuo*. The crude material was purified with flash chromatography, to yield ether **3**.

BM2: O-arylation using NaOH as base⁵⁶

To a solution of NaOH (24 mg, 0.6 mmol, 2 equiv) in distilled water (0.75 mL) was added alcohol (0.30 mmol) at rt and the mixture was stirred for a few minutes. Diaryliodonium salt **1** (0.36 mmol, 1.2 equiv) was added in one portion and the reaction was vigorously stirred at 60 °C for 3 h. The reaction was cooled to room temperature and extracted with EtOAc, dried over Na₂SO₄ and concentrated *in vacuo*. The crude material was purified with flash chromatography, to yield ether **3**.

BM3: S-arylation⁵⁷

To a solution of thiol (0.3 mmol) and diaryliodonium salt **1** (0.3 mmol, 1 equiv) was added MeCN (3 mL) and DBU (43 μL, 0.33 mmol, 1.1 equiv). This mixture was then stirred at 80 °C for 1.5 h. The reaction was extracted with EtOAc (3 x 5 mL), then the organic layer was washed with brine (5 x 10 mL), dried over Na₂SO₄ and concentrated *in vacuo*. The crude material was purified with flash chromatography, to yield ether **4**.

BM3: C-vinylation⁴⁹

To an oven-dried microwave vial was added KO^tBu (37.0 mg, 0.33 mmol, 1.1 equiv). After evacuating the vial and back-filling with nitrogen, anhydrous THF (12.5) was added, followed by addition of the carbon nucleophile (57.1 mg, 0.3 mmol). This mixture was allowed to stir for one hour at room temperature, then vinylidonium salt **6d** and additional anhydrous THF (7.5 mL) were added. The mixture was allowed to stir for 18 h and then the reaction was quenched with water (10 mL), extracted with CH₂Cl₂, dried over Na₂SO₄ and concentrated *in vacuo*. The crude material was purified with flash chromatography, to yield alkane **8**. THF (0.024 M

The C-arylation, S-vinylation, C-vinylation and tosyloxylation data were taken from literature.

6.2 Solution-phase reaction conditions higher solvent concentration

The following reactions were conducted using solution-phase conditions with the exception of using the same amount of solvent as the LAG in the mechanochemical reactions. This allows for a conclusion on whether solution-phase reactions can just be conducted at a much higher concentration in order to optimise further. The conditions are described below, and the results are given in Table S7.

BM5: O-arylation using KO^tBu as base⁵⁵

The same as BM1 but with THF (60 μ L).

BM6: O-arylation using NaOH as base⁵⁶

The same as BM2 but with distilled water (60 μ L).

BM7: S-arylation⁵⁷

The same as BM2 but with MeCN (6 μ L).

BM8: C-vinylation⁴⁹

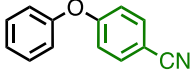
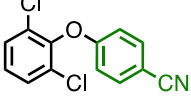
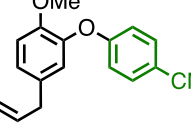
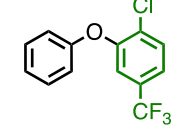
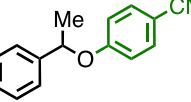
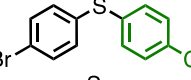
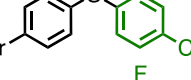
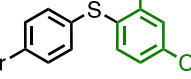
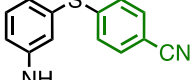
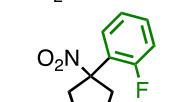
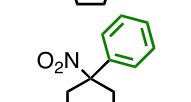
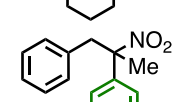
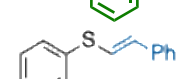
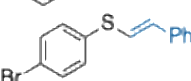
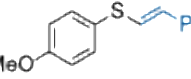
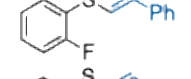

To an oven-dried microwave vial was added KO^tBu (37.0 mg, 0.33 mmol, 1.1 equiv). After evacuating the vial and back-filling with nitrogen, anhydrous THF (30 μ L) was added, followed by addition of the carbon nucleophile (57.1 mg, 0.3 mmol). This mixture was allowed to stir for one hour at room temperature, then vinyliodonium salt **6d** and additional anhydrous THF (7.5 mL) were added. The mixture was allowed to stir for 18 h and then the reaction was quenched with water (10 mL), extracted with CH₂Cl₂, dried over Na₂SO₄ and concentrated *in vacuo*. The crude material was purified with flash chromatography, to yield alkane **8**.

6.3 Mechanochemical reaction conditions with microwave vials

Several mechanochemical reactions were performed with the exact same experimental procedures as the aforementioned mechanochemical conditions, apart from that the reagents were added to a small microwave vial with stirring bar and stirred on a stirrer plate, instead of being added to a ball milling vessel and then milled in a ball miller.

This was done to ascertain whether the ball milling vessel/ ball miller itself was required for activation of the reactions. The results are given in Table S7.

Table S7. Comparison of mechanochemical and solution-based methods.

Product	Yield obtained with various methods (%)			
	In solution	Higher conc	Microwave vials	Ball-milling
	3a BM1: 95 BM2: 69	BM5: 85	28	91
	3g BM1: 80 BM2: 37	BM5: 39	51	88
	3h BM1: 87 BM2: 25	BM5: 41	18	87
	3l BM1: 79 BM2: 63	BM5: 58	16	80
	3y BM1: 63 BM2: 37	BM6: 42	Trace	59
	4f BM3: 71	BM7: 41	28	84
	4g BM3: 74	BM7: 39	22	83
	4h BM3: 71	BM7: 37	39	92
	4l BM3: 50	BM7: 43	49	67
	5d Ref ⁹ : 91	-	-	76
	5g Ref ⁹ : 89	-	-	92
	5h Ref ⁹ : 73	-	-	95
	7a Ref ⁴⁸ : 81 (<i>E:Z</i> >20:1)	-	-	77 (<i>E:Z</i> >20:1)
	7b Ref ¹⁵ : 75 (<i>E:Z</i> >20:1)	-	-	90 (<i>E:Z</i> >20:1)
	7c Ref ¹⁵ : 73 (<i>E:Z</i> >20:1)	-	-	73 (<i>E:Z</i> >20:1)
	7d Ref ¹⁵ : 52 (<i>E:Z</i> >20:1)	-	-	90 (<i>E:Z</i> >20:1)
	7e Ref ¹⁵ : 60 (<i>E:Z</i> 5:1)	-	-	77 (<i>E:Z</i> 12:1)

	7f	Ref ¹⁵ : 37 (<i>E:Z</i> >20:1)	-	-	85 (<i>E:Z</i> >20:1)
	7h	Ref ¹⁵ : 72 (<i>E:Z</i> >20:1)	-	-	93 (<i>E:Z</i> >20:1)
	7g	Ref ¹⁵ : 51 (<i>E:Z</i> 4:1)	-	-	85 (<i>E:Z</i> >20:1)
	8a	Ref ¹⁴ : 67 (regiosel 4:1)	BM8: 33 (regiosel >20:1)	33 (regiosel >20:1)	92 (regiosel >20:1)
	8c	BM4: 11 (regiosel >20:1)	-	-	55 (regiosel >20:1)
	10a	59	-	-	87
	10d	Ref ⁵³ : 85	-	-	74

The different regioselectivities observed in vinylation to **8a** are interesting. While high concentration is sufficient to reach excellent regioselectivity, the yields are low in absence of ball milling (Table S8).

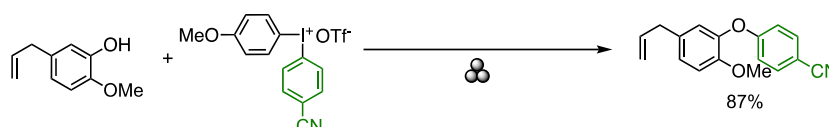
Table S8. Regioselectivity investigation.

6d + $\xrightarrow[\text{THF (x M), rt, 18 h}]{\text{KO}^t\text{Bu}}$ 8a + 8a'			
concentration (M)	NMR yield 8a (%)	NMR yield 8a' (%)	Ratio
0.024	55	12	5:1
0.05	59	10	6:1
0.2	59	6	10:1
1	48	trace	>20:1
10	36	trace	>20:1

7. E-factor calculations

$$E - \text{factor} = \frac{\text{waste mass (g)}}{\text{product (g)}} = \frac{\text{total mass of everything used (g)} - \text{product (g)}}{\text{product (g)}}$$

NOTE on purification techniques: At the 0.3 mmol scale, the same amount of silica and column solvents was always used. In the case for reactions whose yields come directly from the literature, purification techniques are taken from their respective experimental and the same amount of silica and column solvents are used as in our experimental.



Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	-------------------	-------------

E-factor not including purifications below

C ₁₅ H ₁₁ INF ₃ O ₄ S	485	0.3	1	/	0.145
C ₁₀ H ₁₂ O ₂	164	0.3	1	1.06	0.052
K ₂ CO ₃	138	0.3	1	/	0.042
C ₄ H ₈ O ₂		0.61	2	0.902	0.054
Total mass of all substrates (g)					0.293
C ₁₇ H ₁₅ NO ₂	265	0.27		/	0.070
Waste mass (g)					0.223
E-factor					3.2

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	18
Pentane: column	/	/	/	0.604	108
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					185.2
E-factor including purification					2646



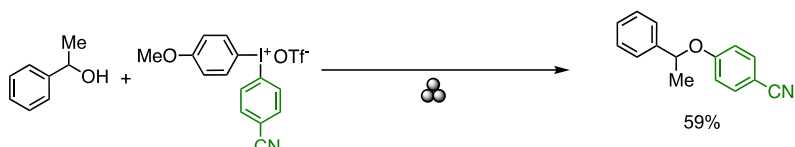
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₁ IN ₃ O ₄ S	485	0.30	1	/	0.146
C ₁₀ H ₁₂ O ₂	164	0.3	1	/	0.049
KOC ₄ H ₉	112	0.33	1.1	/	0.037
C ₄ H ₈ O	/	/	/	0.867	1.730
Total mass of all substrates (g)					1.962
C ₁₇ H ₁₅ NO ₂	265	0.168		/	0.069
Waste mass (g)					1.893
E-factor					27.4

E-factor including purification below

EtOAc: column	/	/	/	0.902	18
Pentane: column	/	/	/	0.604	108
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					177.893
E-factor including purification					2578.2



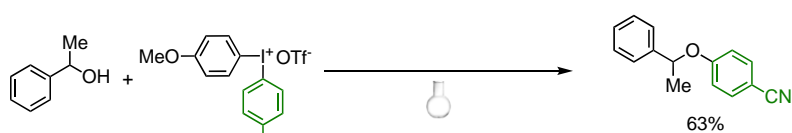
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₁ IN ₃ O ₄ S	485	0.3	1	/	0.145
C ₈ H ₁₀ O	122	0.3	1	1.102	0.036
K ₂ CO ₃	138	0.3	1	/	0.042
C ₄ H ₈ O ₂		0.61	2	0.902	0.054
Total mass of all substrates (g)					0.277
C ₁₅ H ₁₃ NO	223	0.18		/	0.040
Waste mass (g)					0.237
E-factor					5.9

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	18
Pentane: column	/	/	/	0.604	108
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					185.237
E-factor including purification					4630.9



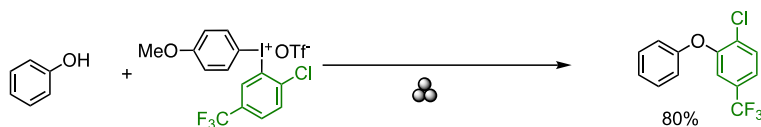
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₁ IN ₃ O ₄ S	485	0.3	1	/	0.146
C ₈ H ₁₀ O	122	0.3	1	/	0.036
KOC ₄ H ₉	96	0.33	1.1	/	0.037
C ₄ H ₈ O	/	/	/	0.867	1.730
Total mass of all substrates (g)					1.949
C ₁₅ H ₁₃ NO	223	0.042		/	0.042
Waste mass (g)					1.907
E-factor					45.4

E-factor including purification below

EtOAc: column	/	/	/	0.902	18
Pentane: column	/	/	/	0.604	108
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					177.907
E-factor including purification					4235.9



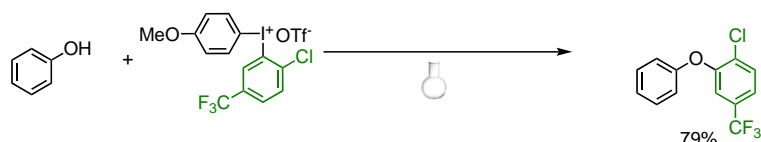
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₀ ClF ₆ IO ₄ S	562	0.3	1	/	0.169
C ₆ H ₆ O	94	0.3	1	/	0.028
K ₂ CO ₃	138	0.3	1	/	0.042
C ₄ H ₈ O ₂		0.61	2	0.902	0.054
Total mass of all substrates (g)					0.293
C ₁₃ H ₇ ClF ₃ O ₂	272	0.24		/	0.066
Waste mass (g)					0.227
E-factor					3.4

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	18
Pentane: column	/	/	/	0.604	108
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					185.227
E-factor including purification					2806.5



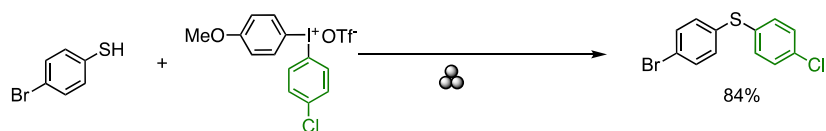
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₀ ClF ₆ IO ₄ S	562	0.3	1	/	0.168
C ₆ H ₆ O	94	0.3	1	1.129	0.028
KOC ₄ H ₉	96	0.33	1.1	/	0.037
C ₄ H ₈ O	/	/	/	0.867	1.730
Total mass of all substrates (g)					1.963
C ₁₄ H ₁₀ ClF ₃ O ₂	272	0.27		/	0.064
Waste mass (g)					1.899
E-factor					29.7

E-factor including purification below

EtOAc: column	/	/	/	0.902	18
Pentane: column	/	/	/	0.604	108
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					177.899
E-factor including purification					2779.7



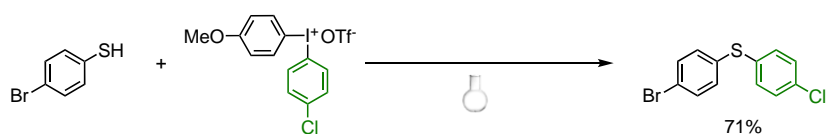
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₄ H ₁₁ ClF ₃ O ₄ S	494	0.3	1	/	0.148
C ₆ H ₅ BrS	189	0.3	1	/	0.057
Na ₂ CO ₃	105	0.3	1	/	0.032
C ₅ H ₈ O		0.068	0.226	0.951	0.006
Total mass of all substrates (g)					0.243
C ₁₂ H ₈ BrClS	224	0.25	/	/	0.075
Waste mass (g)					0.168
E-factor					2.2

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	3.6
Pentane: column	/	/	/	0.604	118.3
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					181.1
E-factor including purification					2417



Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

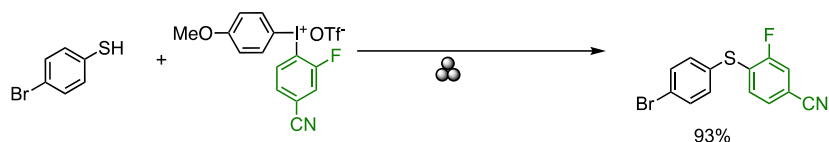
E-factor not including purifications below

C ₁₄ H ₁₁ ClF ₃ O ₄ S	494	0.3	1	/	0.148
C ₆ H ₅ BrS	189	0.3	1	/	0.057
C ₉ H ₁₆ N ₂	24	0.33	1.1	/	0.050
C ₂ H ₃ N		/	/	0.782	3.836
Total mass of all substrates (g)					4.091
C ₁₂ H ₈ BrClS	224	0.024	/	/	0.064
Waste mass (g)					4.027

E-factor	62.9
----------	------

E-factor including purification below

H ₂ O wash	/	/	/	1	25
EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	3.6
Pentane: column	/	/	/	0.604	118.3
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					209.927
E-factor including purification					3280.1



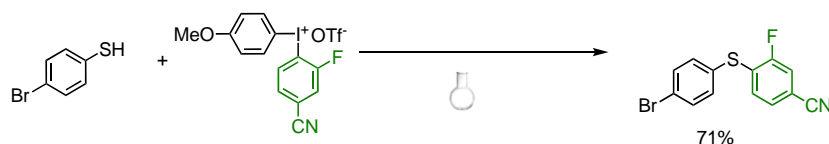
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₀ IF ₄ O ₄ SN	503	0.3	1	/	0.157
C ₆ H ₅ BrS	189	0.3	1	/	0.057
Na ₂ CO ₃	105	0.3	1.2	/	0.032
C ₅ H ₈ O		0.068	0.226	0.951	0.006
Total mass of all substrates (g)					0.252
C ₁₃ H ₇ BrFNS	308	0.28	/	/	0.091
Waste mass (g)					0.161
E-factor					1.8

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	9
Pentane: column	/	/	/	0.604	114.8
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					183.0
E-factor including purification					2010.6



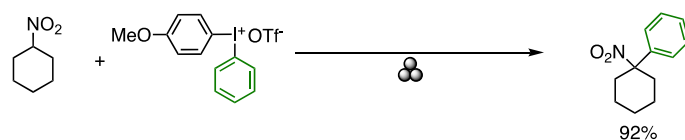
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₀ IF ₄ O ₄ SN	503	0.3	1	/	0.151
C ₆ H ₅ BrS	189	0.3	1	/	0.057
C ₉ H ₁₆ N ₂	24	0.33	1.1	/	0.050
C ₂ H ₃ N		/	/	0.782	3.836
Total mass of all substrates (g)					4.094
C ₁₃ H ₇ BrFNS	308	0.020	/	/	0.066
Waste mass (g)					4.028
E-factor					61.0

E-factor including purification below

H ₂ O wash	/	/	/	1	25
EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	3.6
Pentane: column	/	/	/	0.604	118.3
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					209.928
E-factor including purification					3180.7



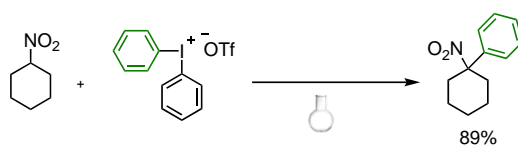
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₄ H ₁₂ IF ₃ O ₄ S	460	0.3	1	/	0.138
C ₆ H ₁₁ NO ₂	129	0.3	1	1.06	0.038
KOC ₄ H ₉	112	0.33	1.1	/	0.032
Total mass of all substrates (g)					0.208
C ₁₂ H ₁₅ NO ₂	205	0.27	/	/	0.055
Waste mass (g)					0.153
E-factor					2.8

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	9
Pentane: column	/	/	/	0.604	114.8
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					183.0
E-factor including purification					3326.4



NOTE: this data was taken from the literature.⁹

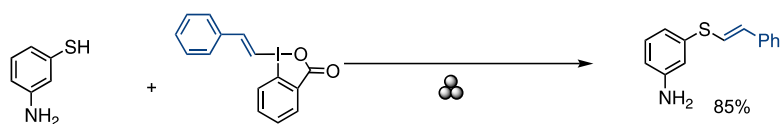
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₄ H ₁₂ IF ₃ O ₄ S	460	0.4	1	/	0.138
C ₆ H ₁₁ NO ₂	129	0.4	1	1.06	0.038
KOC ₄ H ₉	112	0.48	1.2	/	0.032
C ₄ H ₁₀ O ₂	/	/	/	0.867	2.601
Total mass of all substrates (g)					2.809
C ₁₂ H ₁₅ NO ₂	205	0.36	/	/	0.072
Waste mass (g)					2.737
E-factor					38.0

E-factor including purification below

Brine quench	/	/	/	/	7
EtOAc wash	/	/	/	0.902	34.1
EtOAc: column	/	/	/	0.902	11.7
Petroleum/Ether: column	/	/	/	0.604	152.8
Silica	/	/	/	/	75
Total waste of all substrates including purification (g)					283.3
E-factor including purification					3935.2



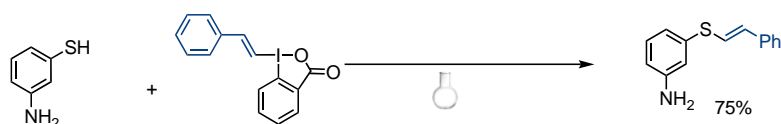
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₁ IO ₂	350	0.3	1	/	0.105
C ₆ H ₇ NS	125	0.3	1	1.17	0.037
K ₂ CO ₃	138	0.3	1	/	0.042
Total mass of all substrates (g)					0.184
C ₁₄ H ₁₃ NS	227	0.26	/	/	0.058
Waste mass (g)					0.126
E-factor					2.2

E-factor including purification below

EtOAc wash	/	/	/	0.902	9
EtOAc: column	/	/	/	0.902	72
Pentane: column	/	/	/	0.604	72
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					203.1
E-factor including purification					3502.2



NOTE: this data was taken from the literature.¹⁵

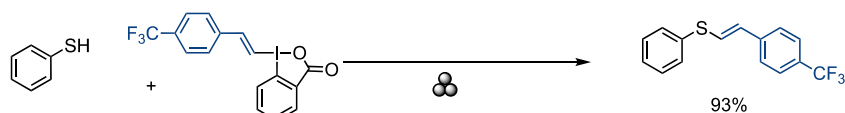
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₅ H ₁₁ IO ₂	350	0.36	1.2	/	0.126
C ₆ H ₇ NS	125	0.3	1	1.17	0.038
KOC ₄ H ₉	112	0.3	1	/	0.034
C ₄ H ₈ O	/	/	/	0.88	2.67
Total mass of all substrates (g)					2.868
C ₁₄ H ₁₃ NS	227	0.11	/	/	0.022
Waste mass (g)					2.846
E-factor					129.4

E-factor including purification below

Water quench	/	/	/	1	2
DCM extraction	/	/	/	0.902	26
EtOAc: column	/	/	/	0.902	14
Pentane: column	/	/	/	0.604	112
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					206.846
E-factor including purification					9402.1



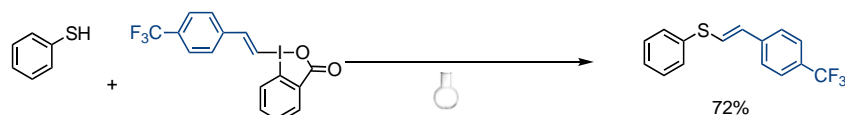
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₆ H ₁₀ IO ₂ F ₃	418	0.2	1	/	0.084
C ₆ H ₆ S	110	0.2	1	1.08	0.022
KOC ₄ H ₉	112	0.2	1	/	0.022
C ₄ H ₈ O	72	0.62	2.1	0.88	0.044
Total mass of all substrates (g)					0.172
C ₁₅ H ₁₁ F ₃ S	280	0.18	/	/	0.052
Waste mass (g)					0.120
E-factor					2.3

E-factor including purification below

Et ₂ O wash	/	/	/	0.713	2.1
Pentane wash	/	/	/	0.604	1.8
Total waste of all substrates including purification (g)					4.020
E-factor including purification					77.3



NOTE: this data was taken from the literature.¹⁵

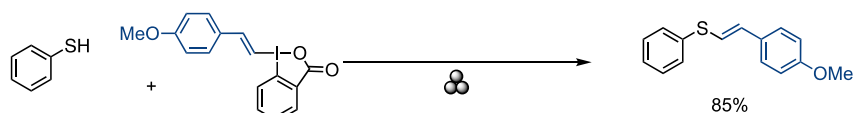
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₆ H ₁₀ IO ₂ F ₃	418	0.11	1.1	/	0.046
C ₆ H ₆ S	110	0.1	1	1.08	0.011
KOC ₄ H ₉	112	0.1	1	/	0.011
C ₄ H ₈ O	/	/	/	0.88	2.667
Total mass of all substrates (g)					2.735
C ₁₅ H ₁₁ F ₃ S	280	0.07	/	/	0.015
Waste mass (g)					2.72
E-factor					181.3

E-factor including purification below

Water quench	/	/	/	1	2
DCM extraction	/	/	/	0.902	26
Pentane: column	/	/	/	0.626	125.2
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					205.9
E-factor including purification					13728



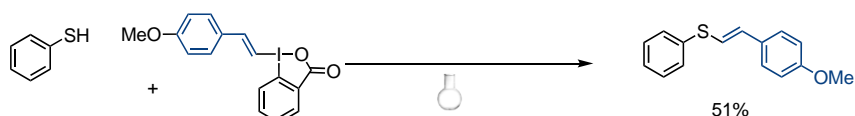
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₆ H ₁₃ IO ₃	380	0.2	1	/	0.076
C ₆ H ₆ S	110	0.2	1	1.08	0.022
KOC ₄ H ₉	112	0.2	1	/	0.022
C ₄ H ₈ O	72	0.62	2.1	0.88	0.044
Total mass of all substrates (g)					0.164
C ₁₅ H ₁₁ F ₃ S	280	0.18	/	/	0.041
Waste mass (g)					0.123
E-factor					3.0

E-factor including purification below

Et ₂ O wash	/	/	/	0.713	2.1
Pentane wash	/	/	/	0.604	1.8
Total waste of all substrates including purification (g)					4.023
E-factor including purification					98.1



NOTE: this data was taken from the literature.¹⁵

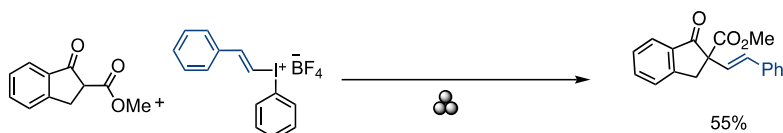
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₆ H ₁₃ IO ₃	380	0.11	1.1	/	0.039
C ₆ H ₆ S	110	0.1	1	1.08	0.011
KOC ₄ H ₉	112	0.1	1	/	0.011
C ₄ H ₈ O	/	/	/	0.88	2.667
Total mass of all substrates (g)					2.728
C ₁₅ H ₁₁ F ₃ S	280	0.07	/	/	0.012
Waste mass (g)					2.716
E-factor					226.3

E-factor including purification below

Water quench	/	/	/	1	2
DCM extraction	/	/	/	0.902	26
Pentane: column	/	/	/	0.626	125.2
Silica	/	/	/	/	50
Total waste of all substrates including purification (g)					205.9
E-factor including purification					17159.7



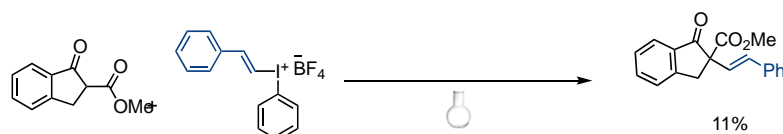
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₁₄ H ₁₂ IBF ₄	394	0.3	1	/	0.118
C ₁₁ H ₁₄ O ₃	190	0.3	1	/	0.057
KOC ₄ H ₉	112	0.33	1.1	/	0.040
Total mass of all substrates (g)					0.215
C ₁₉ H ₂₁ O ₃	291	0.19	/	/	0.055
Waste mass (g)					0.160
E-factor					2.9

E-factor including purification below

EtOAc wash	/	/	/	0.713	27
Pentane wash	/	/	/	0.604	102
Total waste of all substrates including purification (g)					129.160
E-factor including purification					2348.4



Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

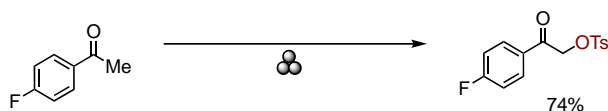
E-factor not including purifications below

C ₁₄ H ₁₂ IBF ₄	394	0.33	1.1	/	0.129
C ₁₁ H ₁₄ O ₃	190	0.3	1	/	0.057
KOC ₄ H ₉	112	0.33	1.1	/	0.037
C ₄ H ₈ O	/	/	/	0.88	11.11
Total mass of all substrates (g)					11.33
C ₁₉ H ₂₁ O ₃	291	0.19	/	/	0.010
Waste mass (g)					11.32
E-factor					1132.0

E-factor including purification below

Water wash	/	/	/	1	20
------------	---	---	---	---	----

Brine wash	/	/	/	1	20
EtOAc: column	/	/	/	0.902	26.6
Pentane: column	/	/	/	0.604	102.7
Silica	/	/	/		50
Total waste of all substrates including purification (g)					230.6
E-factor including purification					23062



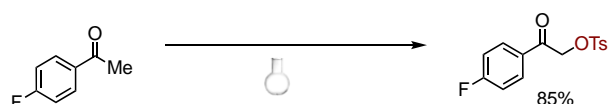
Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₆ H ₇ FO	138	0.3	1	/	0.041
C ₈ H ₉ I	232	0.06	0.2	/	0.014
C ₇ H ₅ ClO ₃	172	0.45	1.5	/	0.094
C ₇ H ₁₀ SO ₄	190	0.45	1.5	/	0.086
C ₃ H ₂ OF ₆	168	0.83	2.7	1.596	0.139
Total mass of all substrates (g)					0.374
C ₁₅ H ₁₃ FO ₄ S	308	0.22	/	/	0.068
Waste mass (g)					0.306
E-factor					4.5

E-factor including purification below

NaHCO ₃ quench	/	/	/	3.66	18.3
EtOAc: extraction	/	/	/	0.902	13.5
EtOAc: column	/	/	/	0.902	27.1
Pentane: column	/	/	/	0.604	102.7
Total waste of all substrates including purification (g)					161.906
E-factor including purification					2381.0



NOTE: this data was taken from the literature. ⁵³

Compound	M _r (g/mol)	mmol	equiv	density (g/mL)	mass (g)
----------	---------------------------	------	-------	----------------	----------

E-factor not including purifications below

C ₆ H ₇ FO	138	0.3	1	/	0.041
C ₆ H ₅ I	232	0.03	0.1	/	0.007
C ₇ H ₅ ClO ₃	172	0.33	1.1	/	0.069
C ₇ H ₁₀ SO ₄	190	0.33	1.1	/	0.063
C ₂ H ₃ N	/	/	/	0.786	1.179
Total mass of all substrates (g)					1.359

C ₁₅ H ₁₃ FO ₄ S	308	0.28	/	/	0.082
Waste mass (g)					1.277
E-factor					15.6

E-factor including purification below

NaHCO ₃ quench	/	/	/	3.66	18.3
CHCl ₃ : extraction	/	/	/	1.49	44.7
EtOAc: column	/	/	/	0.902	27.06
Pentane: column	/	/	/	0.604	102.68
Total waste of all substrates including purification (g)					194.016
E-factor including purification					2366.1

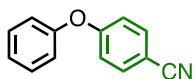
8. References

- 1 A. I. Vogel, B. S. Furniss, A. J. Hannaford, V. Rogers, P. W. G. Smith and A. R. Tatchell, *Vogel's Textbook of Practical Organic Chemistry*, Prentice Hall, Harlow, 5th edn., 1996.
- 2 P. Kumar, Meenakshi, S. Kumar, A. Kumar, K. Hussain and S. Kumar, *J. Hetero. Chem.*, 2012, **49**, 1243–1249.
- 3 P. Kumar, K. Hussain and A. Kumar, *Curr. Chem. Lett.*, 2014, **3**, 75–84.
- 4 V. S. Rao and K. V. G. Chandra Sekhar, *Synth. Commun.*, 2004, **34**, 2153–2157.
- 5 T. Kumar Achar and P. Mal, *Adv. Synth. Catal.*, 2015, **357**, 3977–3985.
- 6 R. Obermüller, H. Tobisch, L. Stockhammer and M. Waser, *Org. Process. Res. Dev.*, 2024, **28**, 3735–3744.
- 7 I. Priestley, C. Battilocchio, A. V. Iosub, F. Barreteau, G. W. Bluck, K. B. Ling, K. Ingram, M. Ciaccia, J. A. Leitch and D. L. Browne, *Org. Process. Res. Dev.*, 2023, **27**, 269–275.
- 8 C. M. Alder, J. D. Hayler, R. K. Henderson, A. M. Redman, L. Shukla, L. E. Shuster and H. F. Sneddon, *Green Chem.*, 2016, **18**, 3879–3890.
- 9 C. Dey, E. Lindstedt and B. Olofsson, *Org. Lett.*, 2015, **17**, 4554–4557.
- 10 D. G. Bachmann, C. C. Wittwer and D. G. Gillingham, *Adv. Synth. Catal.*, 2013, **355**, 3703–3707.
- 11 G. Kervefors, L. Kersting and B. Olofsson, *Chem. – A Eur. J.*, 2021, **27**, 5790–5795.
- 12 N. S. Soldatova, A. V. Semenov, K. K. Geyl, S. V. Baykov, A. A. Shetnev, A. S. Konstantinova, M. M. Korsakov, M. S. Yusubov and P. S. Postnikov, *Adv. Synth. Catal.*, 2021, **363**, 3566–3576.
- 13 M. Bielawski, J. Malmgren, L. M. Pardo, Y. Wikmark and B. Olofsson, *ChemistryOpen*, 2014, **3**, 19–22.
- 14 E. Stridfeldt, A. Seemann, M. J. Bouma, C. Dey, A. Ertan and B. Olofsson, *Chem. - A Eur. J.*, 2016, **22**, 16066–16070.
- 15 L. Castoldi, E. M. Di Tommaso, M. Reitti, B. Gräfen and B. Olofsson, *Angew. Chem. Int. Ed.*, 2020, **59**, 15512–15516.
- 16 A. A. Rajkiewicz and M. Kalek, *Org. Lett.*, 2018, **20**, 1906–1909.
- 17 S. Altomonte, S. Telu, S. Lu and V. W. Pike, *J. Org. Chem.*, 2017, **82**, 11925–11932.
- 18 A.-S. Castanet, F. Colobert and P.-E. Broutin, *Tet. Lett.*, 2002, **43**, 5047–5048.
- 19 S. Li, K. Huang and X. Zhang, *Chem. Comm.*, 2014, **50**, 8878–8881.
- 20 Q. Elliott and I. V. Alabugin, *J. Org. Chem.*, 2023, **88**, 2648–2654.
- 21 D. Y. Wang, Z. K. Yang, C. Wang, A. Zhang and M. Uchiyama, *Angew. Chem. Int. Ed.*, 2018, **57**, 3641–3645.
- 22 A. Nagaki, Y. Ashikari, T. Kawaguchi, K. Mandai and Y. Aizawa, *J. Am. Chem. Soc.*, 2020, **142**, 17039–17047.
- 23 J. Hwang, P. Li, W. R. Carroll, M. D. Smith, P. J. Pellechia and K. D. Shimizu, *J. Am. Chem. Soc.*, 2014, **136**, 14060–14067.
- 24 S. D. Rychnovsky and K. Hwang, *J. Org. Chem.*, 1994, **59**, 5414–5418.
- 25 L. Salvi, N. R. Davis, S. Z. Ali and S. L. Buchwald, *Org. Lett.*, 2012, **14**, 170–173.
- 26 P. Oswal, A. Arora, S. Purohit, A. Bahuguna, P. Sharma, J. Roy and A. Kumar, *New J. Chem.*, 2023, **47**, 4346–4354.
- 27 X. Zhang, G. P. Lu and C. Cai, *Green Chem.*, 2016, **18**, 5580–5585.
- 28 S. Zhang, *Synlett.*, 2010, **3**, 5–10.
- 29 R. T. Gallagher, S. Basu and D. R. Stuart, *Adv. Synth. Catal.*, **362**, 2020, 320–325.

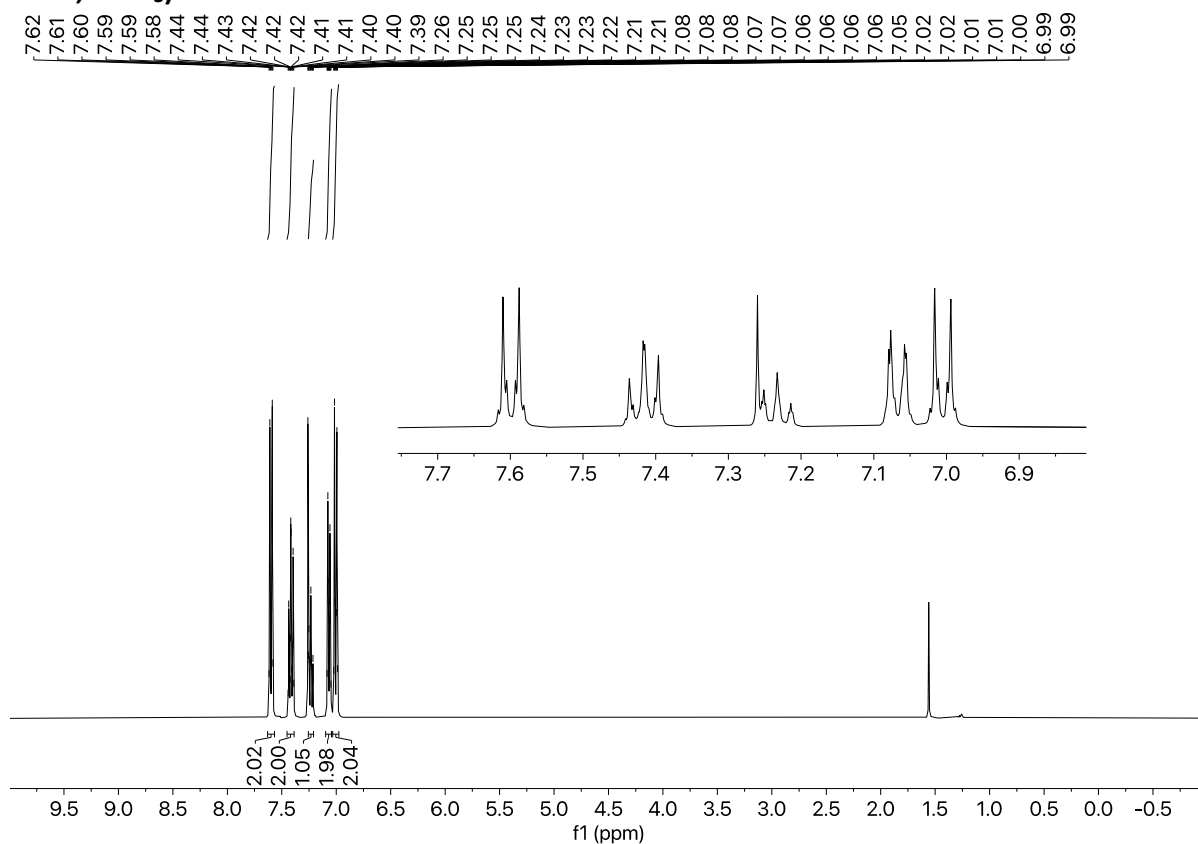
- 30 N. Matsushita, M. Kashihara, M. Formica and Y. Nakao, *Organometallics*, 2021, **40**, 2209–2214.
- 31 X. Zong, Q. Z. Zheng and N. Jiao, *Org. Biomol. Chem.*, 2014, **12**, 1198–1202.
- 32 J. Kim, J. Choi, K. Shin and S. Chang, *J. Am. Chem. Soc.*, 2012, **134**, 2528–2531.
- 33 J. Tang, J. Kong, H. Xu, Z. J. Jiang, Y. She, J. Bai, B. Tang, J. Chen, Z. Gao and K. Gao, *J. Org. Chem.*, 2023, **88**, 1560–1567.
- 34 P. M. MacQueen, J. P. Tassone, C. Diaz and M. Stradiotto, *J. Am. Chem. Soc.*, 2018, **140**, 5023–5027.
- 35 C. Wang, Z. Zhang, Y. Tu, Y. Li, J. Wu and J. Zhao, *J. Org. Chem.*, 2018, **83**, 2389–2394.
- 36 T. T. Wang, F. L. Yang and S. K. Tian, *Adv. Synth. Catal.*, 2015, **357**, 928–932.
- 37 G. Kibriya, S. Mondal and A. Hajra, *Org. Lett.*, 2018, **20**, 7740–7743.
- 38 M. Arisawa, T. Suzuki, T. Ishikawa and M. Yamaguchi, *J. Am. Chem. Soc.*, 2008, **130**, 12214–12215.
- 39 A. Bhowmik, M. Yadav and R. A. Fernandes, *Org. Biomol. Chem.*, 2020, **18**, 2447–2458.
- 40 J. M. Fernández-Hernández, V. Cámara and J. Vicente, *Tetrahedron*, 2015, **71**, 5506–5512.
- 41 J. Chen, K. Zhang, Y. Zhao and S. Pu, *Synth. Commun.*, 2018, **48**, 1316–1323.
- 42 Y. M. Tian, E. Hofmann, W. Silva, X. Pu, D. Touraud, R. M. Gschwind, W. Kunz and B. König, *Angew. Chem. Int. Ed.*, **2023**, 62, e202218775.
- 43 W. Lin, L. Chen and P. Knochel, *Tetrahedron*, 2007, **63**, 2787–2797.
- 44 S. Sarkar, N. Wojciechowska, A. A. Rajkiewicz and M. Kalek, *Eur. J. Org. Chem.*, 2022, e202101408.
- 45 Y. Li, J. Pu and X. Jiang, *Org. Lett.*, 2014, **16**, 2692–2695.
- 46 C. Pezzetta, A. Folli, O. Matuszewska, D. Murphy, R. W. M. Davidson and D. Bonifazi, *Adv. Synth. Catal.*, 2021, **363**, 4740–4753.
- 47 M. Nakajima, K. Miyamoto, K. Hirano and M. Uchiyama, *J. Am. Chem. Soc.*, 2019, **141**, 6499–6503.
- 48 L. Castoldi, E. M. Di Tommaso, M. Reitti, B. Gräfen and B. Olofsson, *Angew. Chem. Int. Ed.*, 2020, **59**, 15512–15516.
- 49 C. J. Parkinson, J. T. Pinhey and M. J. Stoermer, *J. Chem. Soc. Perkin 1*, 1992, 1911–1915.
- 50 T. Y. Ko and S. W. Youn, *Adv. Synth. Catal.*, 2016, **358**, 1934–1941.
- 51 R. Liu, J. Wang, W. Hu, X. Zhang and Y. Xiong, *Synth. Commun.*, 2018, **48**, 1957–1965.
- 52 Y. Jun, T. Jun and Z. Chi, *Adv. Synth. Catal.*, 2010, **352**, 531–546.
- 53 B. D. Rupanawar, K. D. Mane and G. Suryavanshi, *New J. Chem.*, 2022, **46**, 16832–16839.
- 54 A. Boelke and B. J. Nachtsheim, *Adv. Synth. Catal.*, 2020, **362**, 184–191.
- 55 N. Jalalian, E. E. Ishikawa, L. F. Silva and B. Olofsson, *Org. Lett.*, 2011, **13**, 1552–1555.
- 56 E. Lindstedt, R. Ghosh and B. Olofsson, *Org. Lett.*, 2013, **15**, 6070–6073.
- 57 S. Sarkar, N. Wojciechowska, A. A. Rajkiewicz and M. Kalek, *Eur. J. Org. Chem.*, 2022, e202101408.

9. NMR Spectra

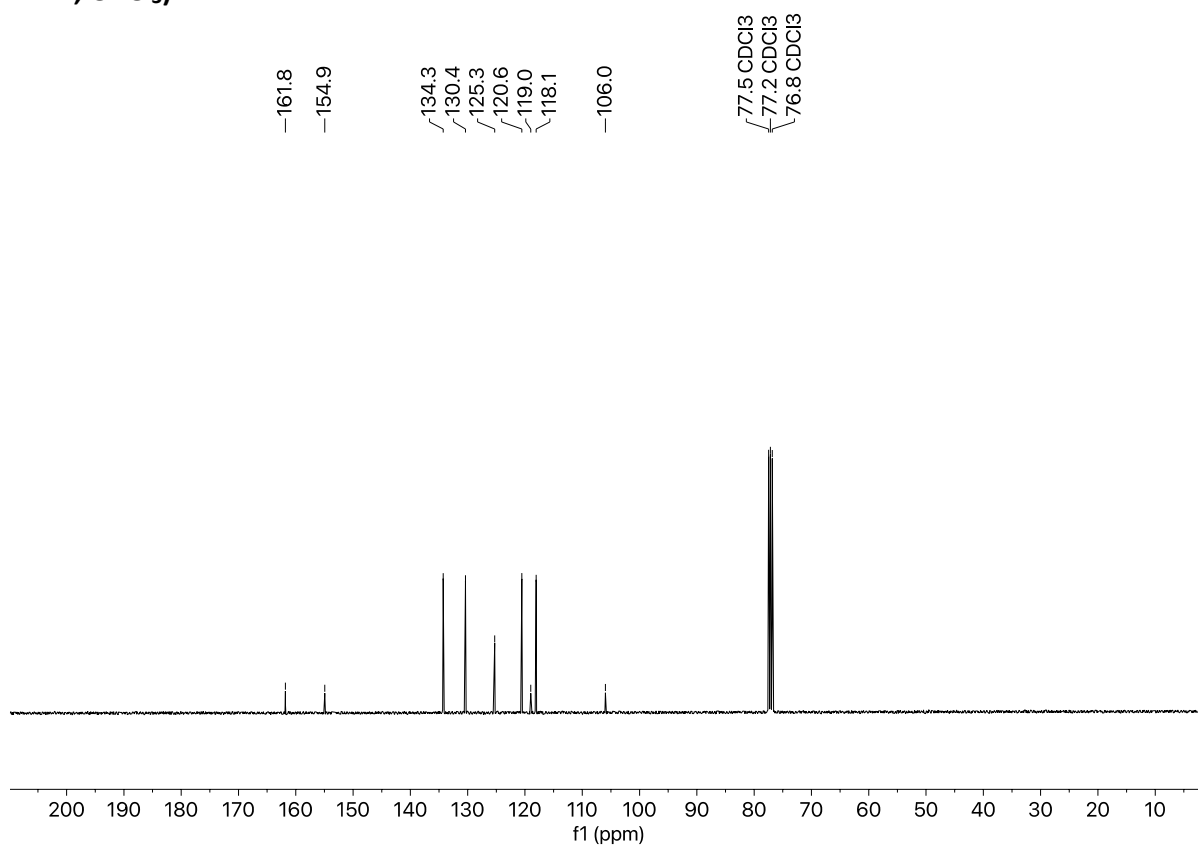
4-Phenoxybenzonitrile (3a)



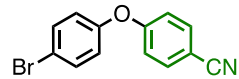
δ_H (400 MHz, $CDCl_3$)



δ_C (101 MHz, $CDCl_3$)

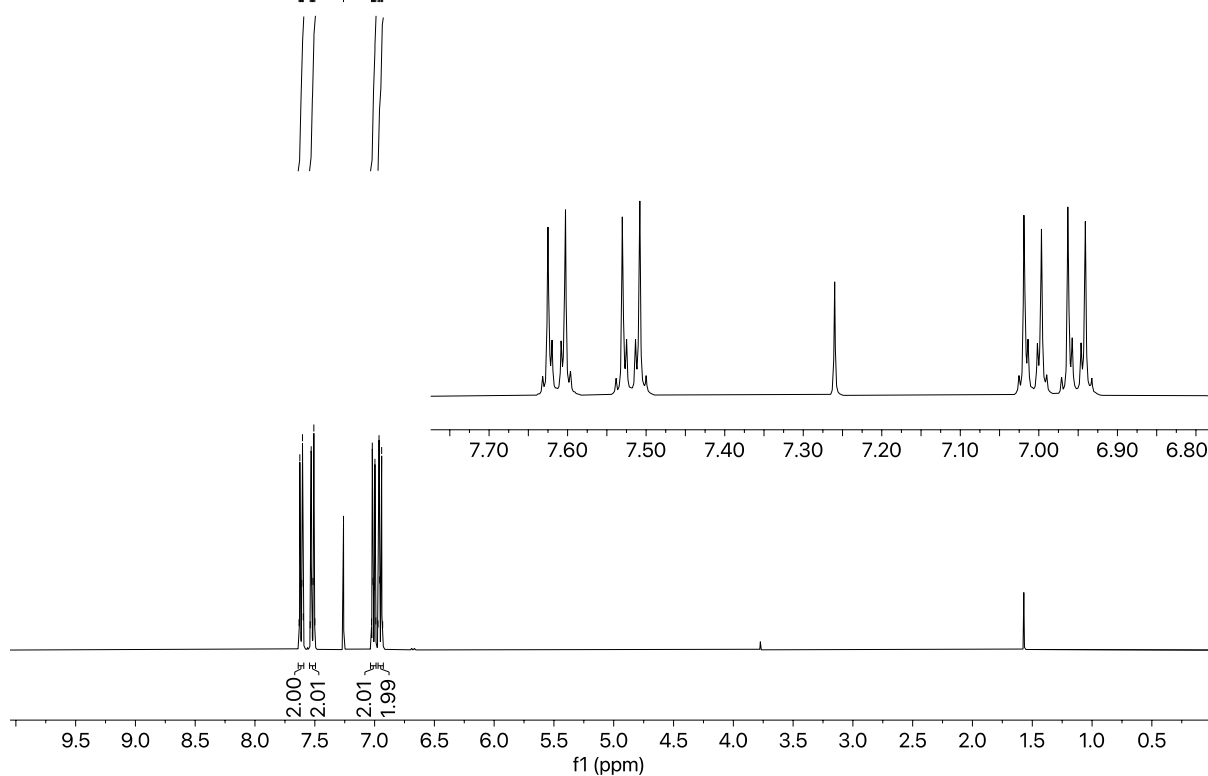


4-(4-Bromophenoxy)benzonitrile (3b)



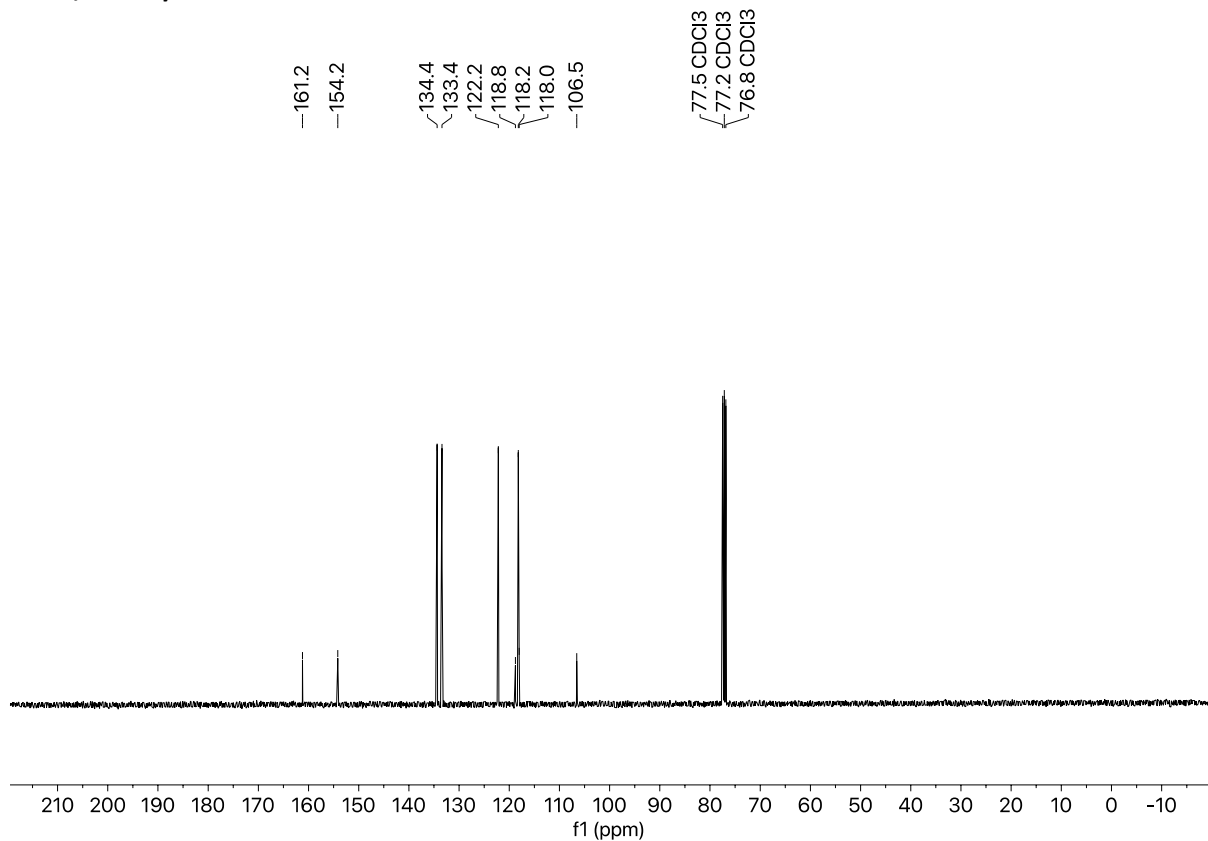
δ_H (400 MHz, $CDCl_3$)

7.63
7.63
7.62
7.61
7.61
7.60
7.60
7.54
7.53
7.52
7.52
7.51
7.51
7.50
7.26
7.03
7.02
7.01
7.01
7.00
7.00
6.99
6.97
6.96
6.95
6.94
6.93

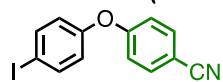


δ_C (101 MHz, $CDCl_3$)

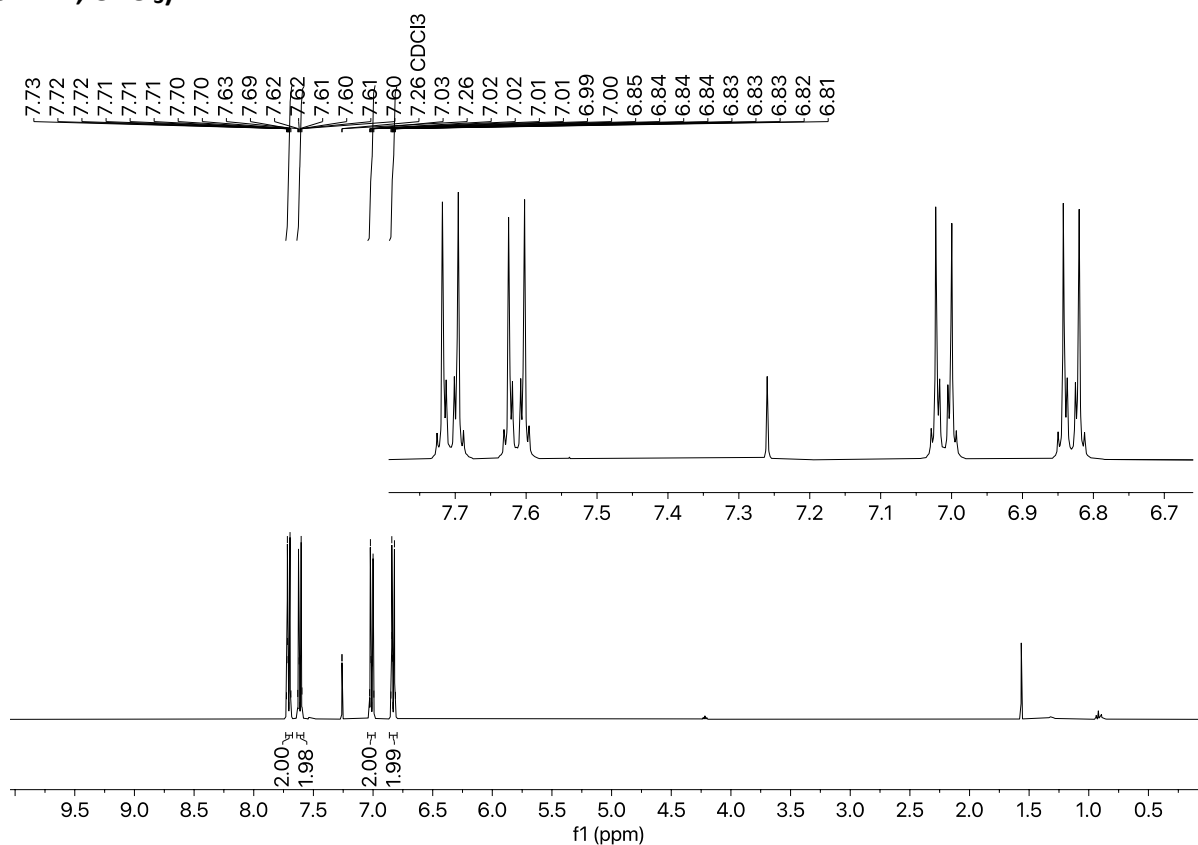
161.2
154.2
134.4
133.4
122.2
118.8
118.2
118.0
106.5
77.5 $CDCl_3$
77.2 $CDCl_3$
76.8 $CDCl_3$



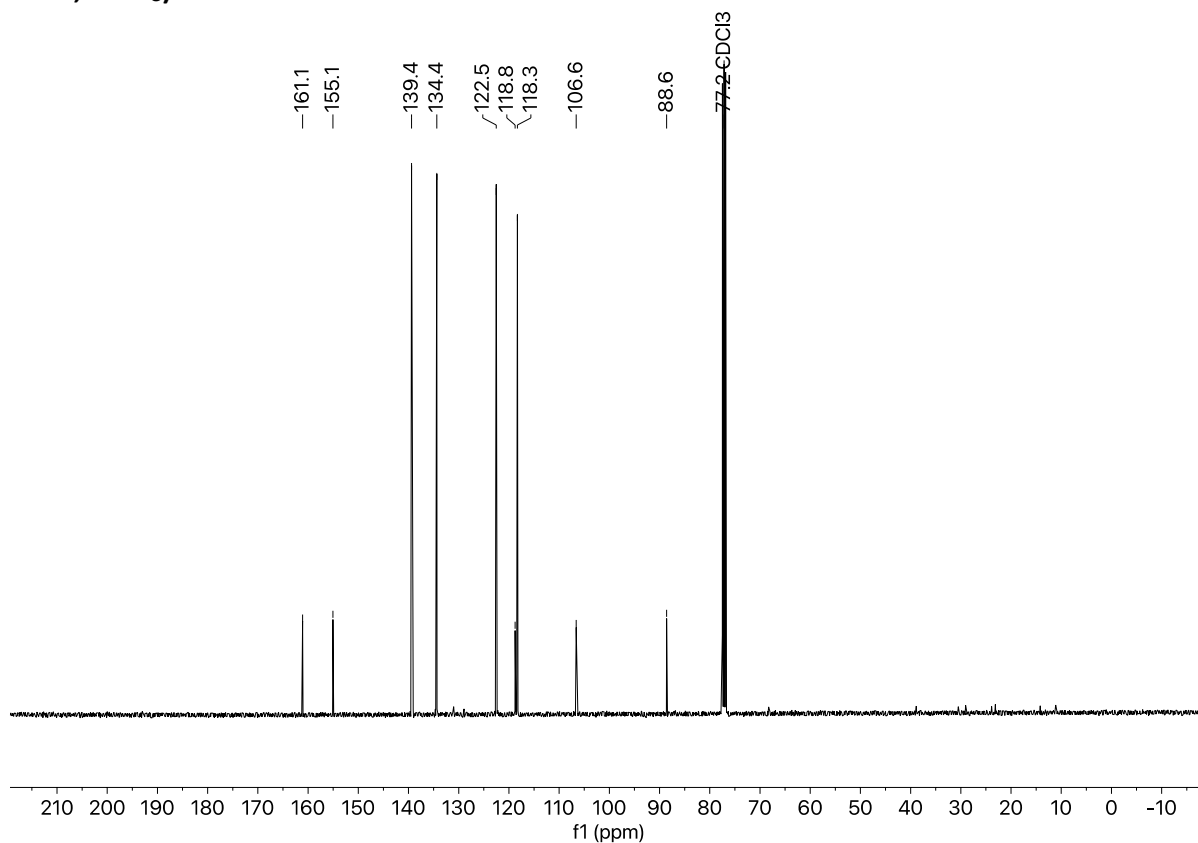
4-(4-Iodophenoxy)benzonitrile (3c)



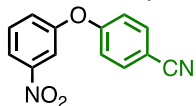
δ_H (400 MHz, $CDCl_3$)



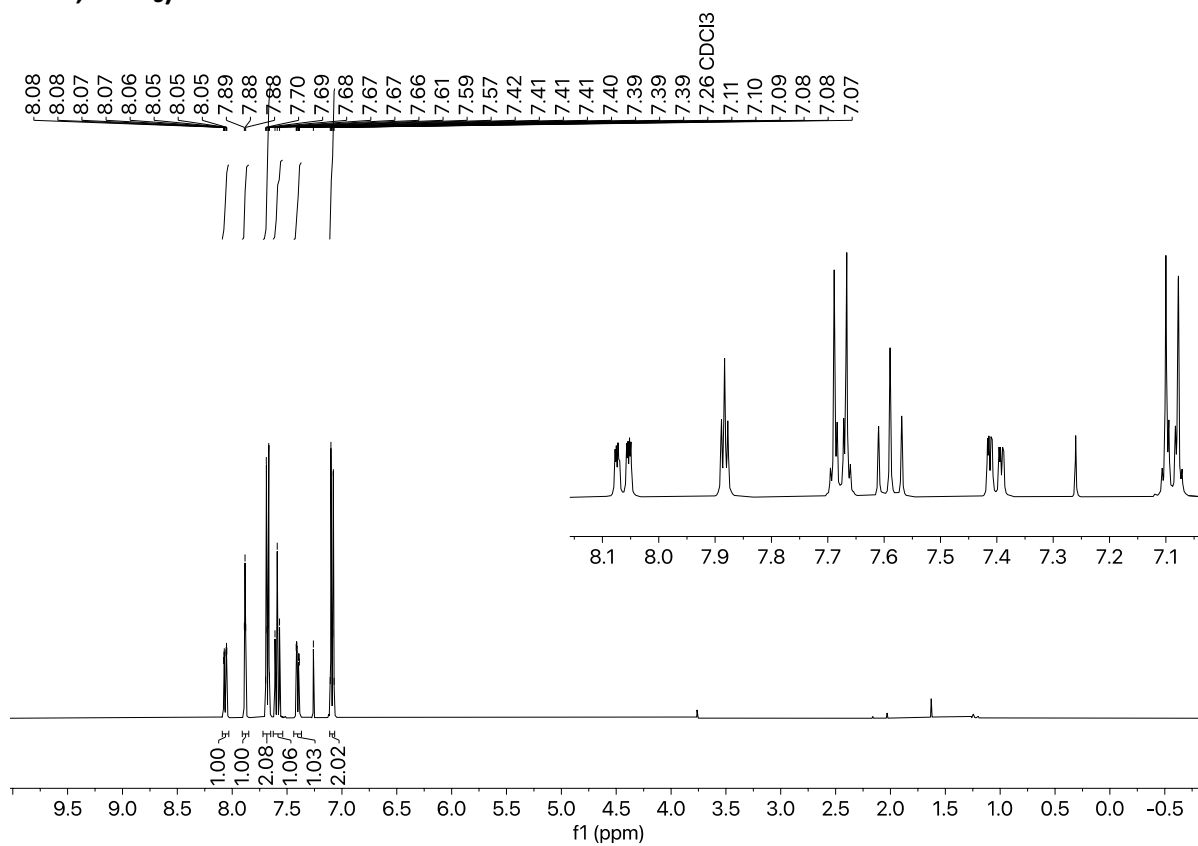
δ_C (101 MHz, $CDCl_3$)



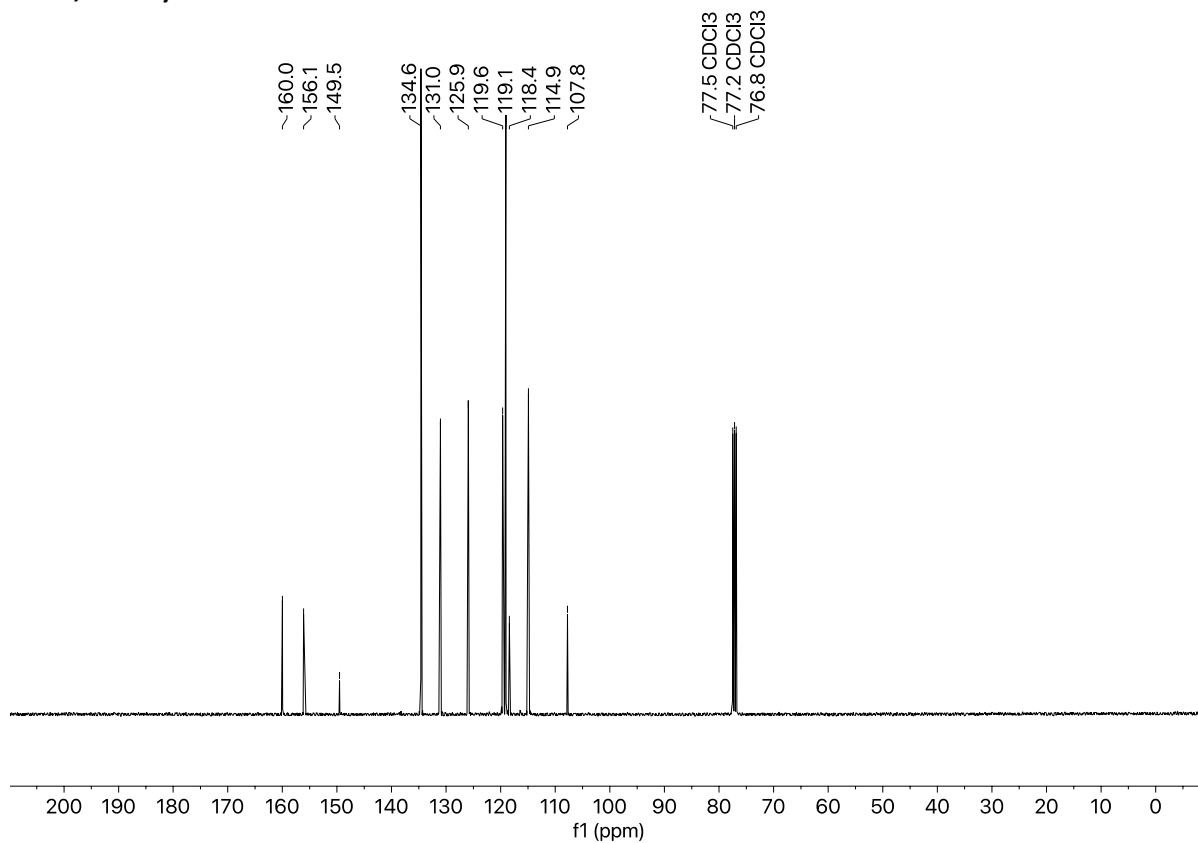
4-(3-Nitrophenoxy)benzonitrile (3d)



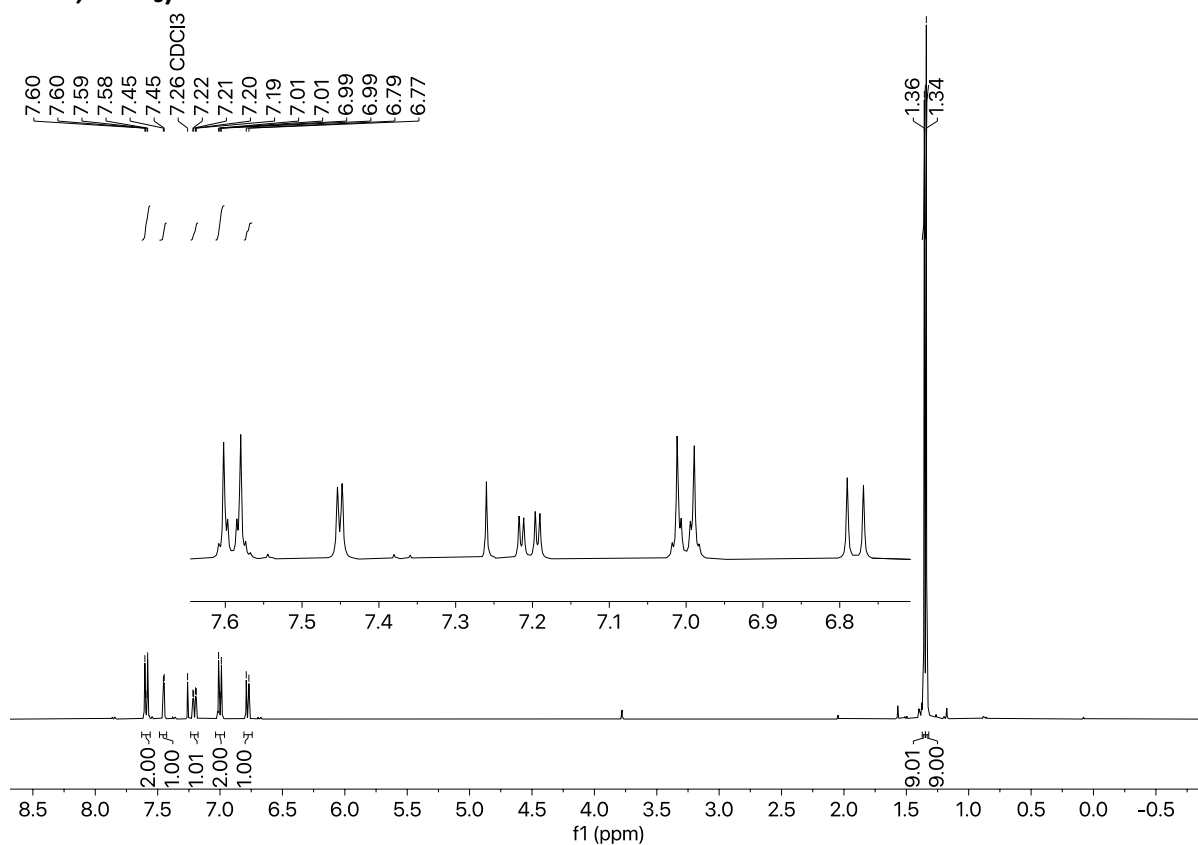
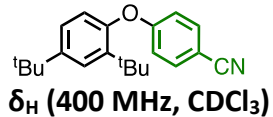
δ_H (400 MHz, $CDCl_3$)



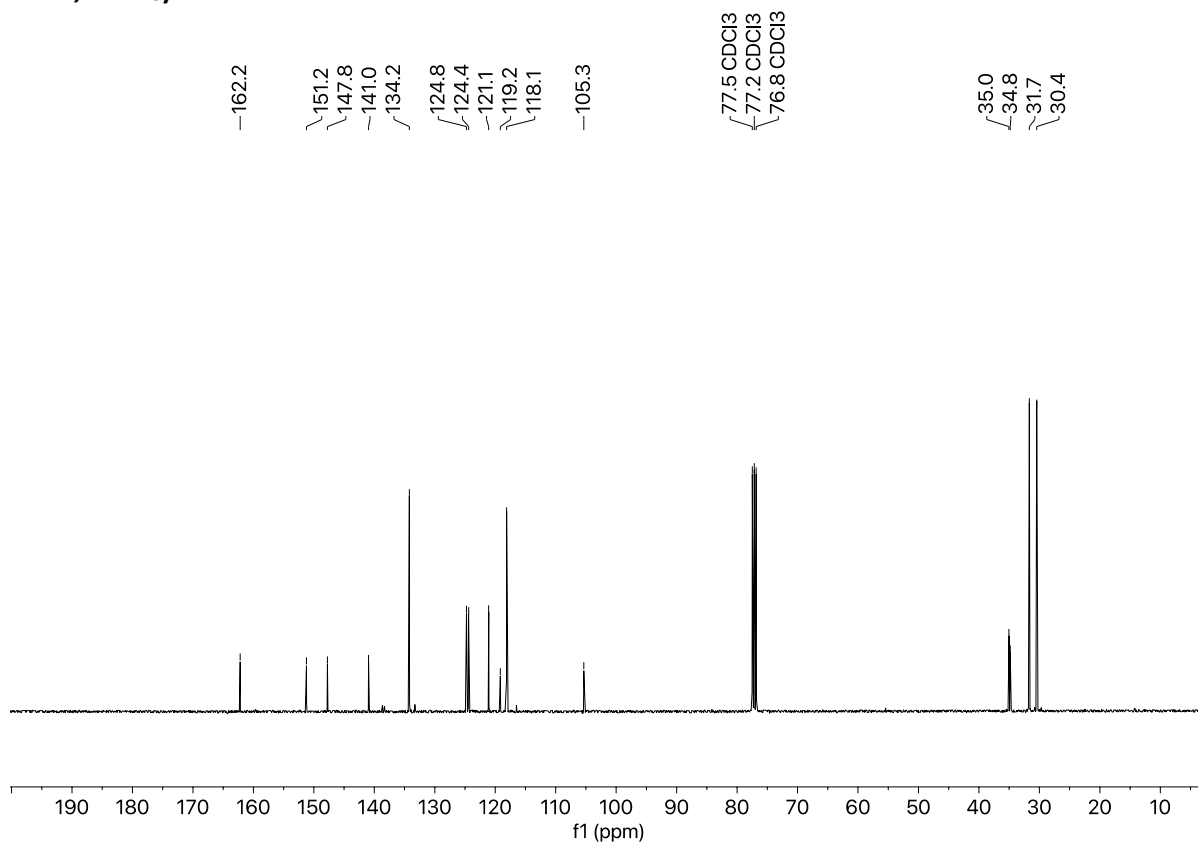
δ_C (101 MHz, $CDCl_3$)



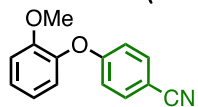
4-(2,4-Di-*tert*-butylphenoxy)benzonitrile. (3e)



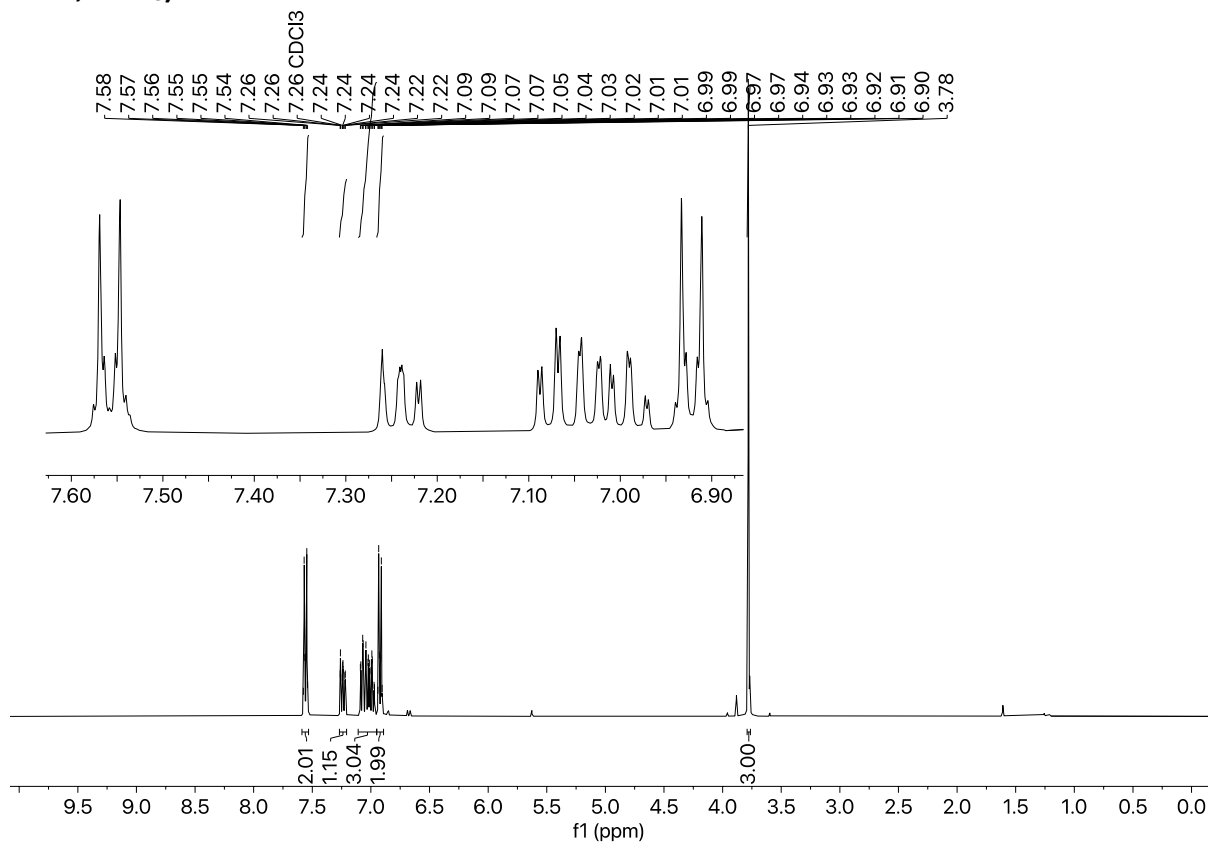
δ_C (101 MHz, $CDCl_3$)



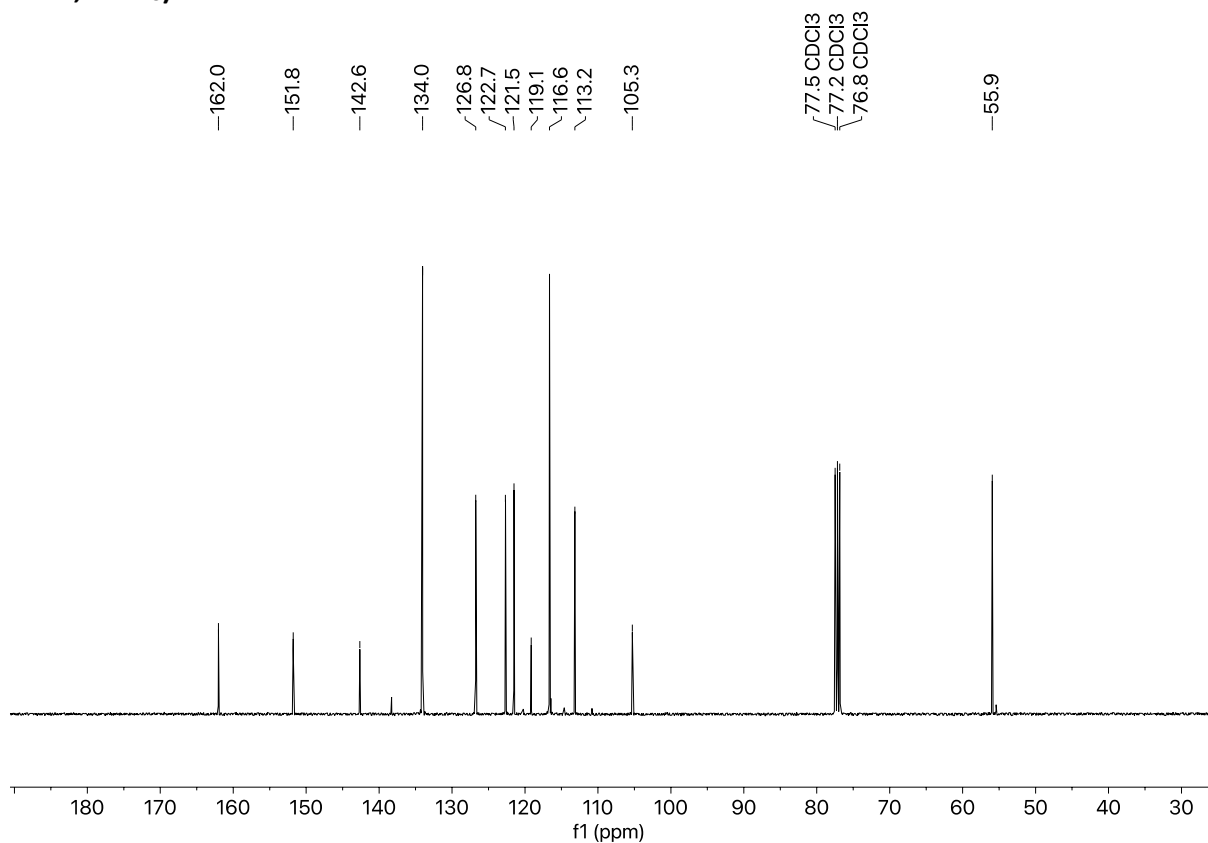
4-(2-Methoxyphenoxy)benzonitrile (3f)



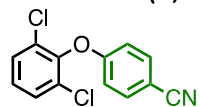
δ_H (400 MHz, $CDCl_3$)



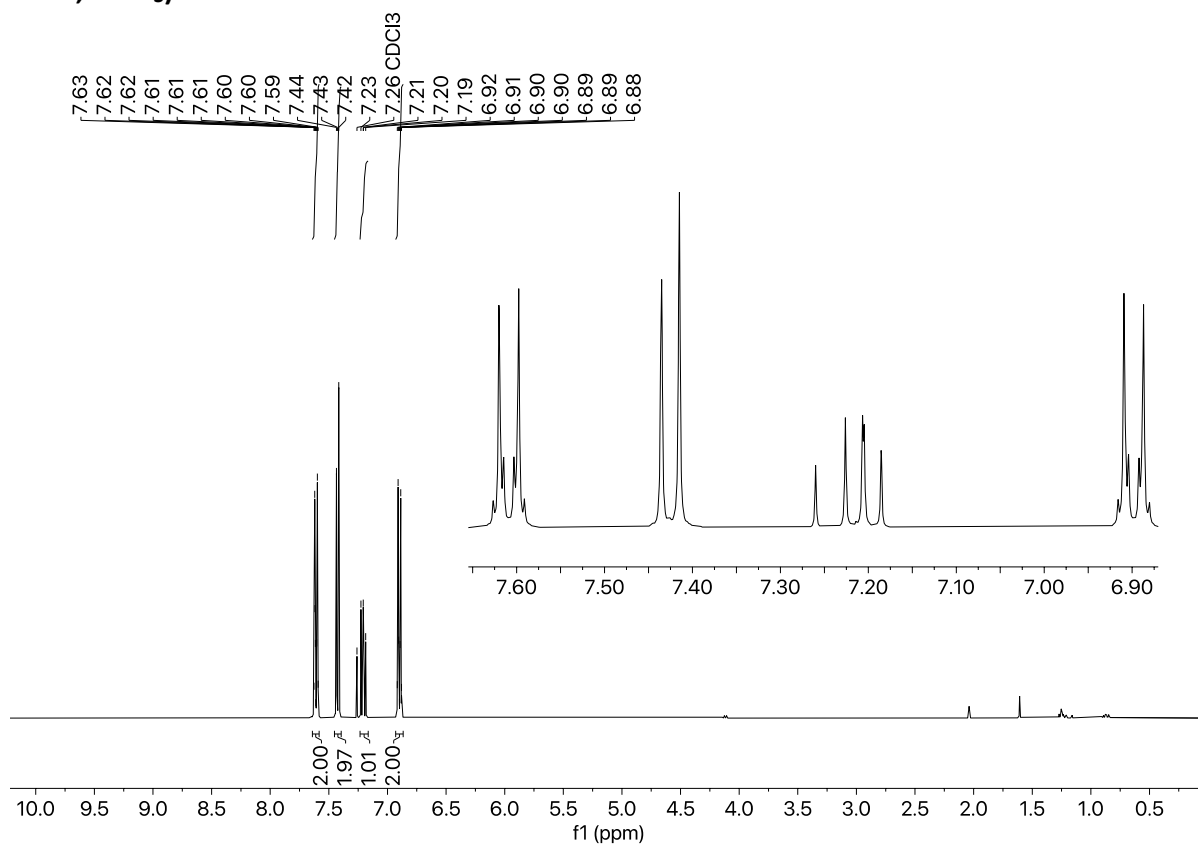
δ_C (101 MHz, $CDCl_3$)



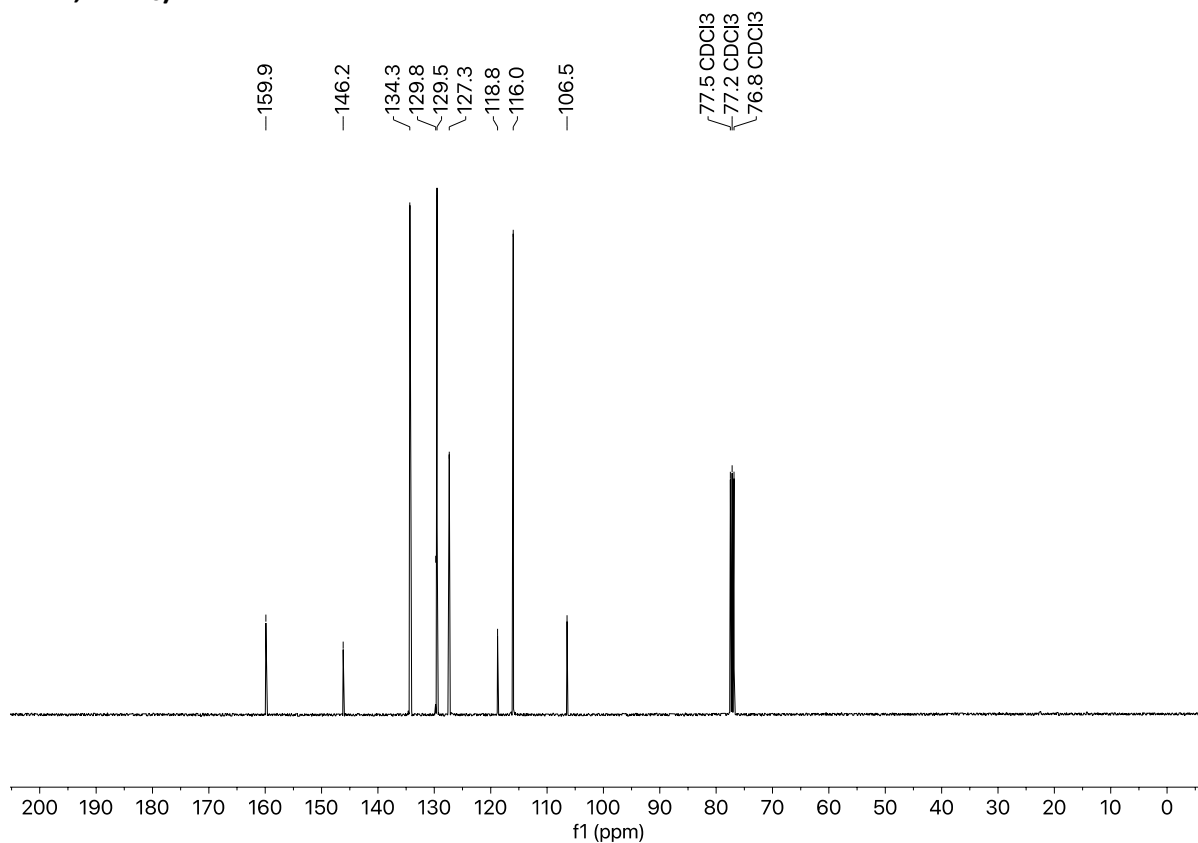
4-(2,6-Dichlorophenoxy)benzonitrile. (3g)



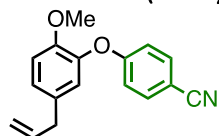
δ_H (400 MHz, $CDCl_3$)



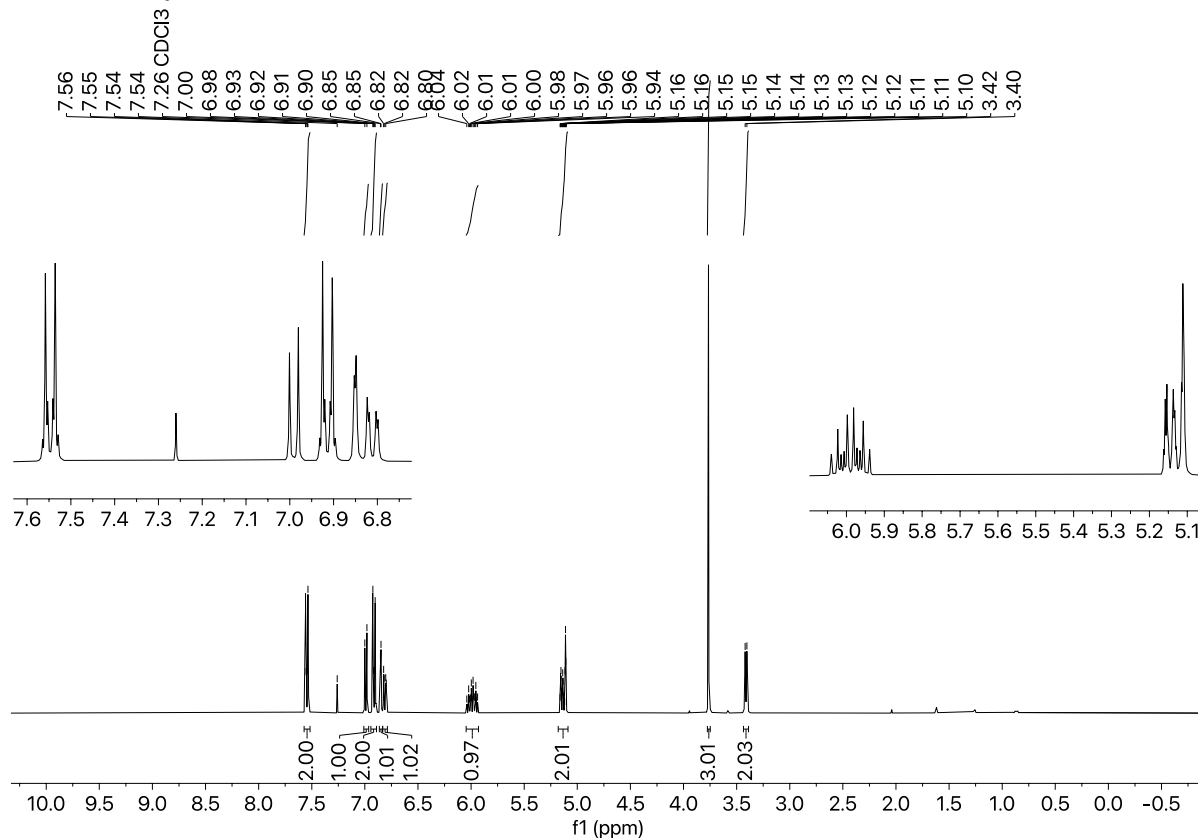
δ_C (101 MHz, $CDCl_3$)



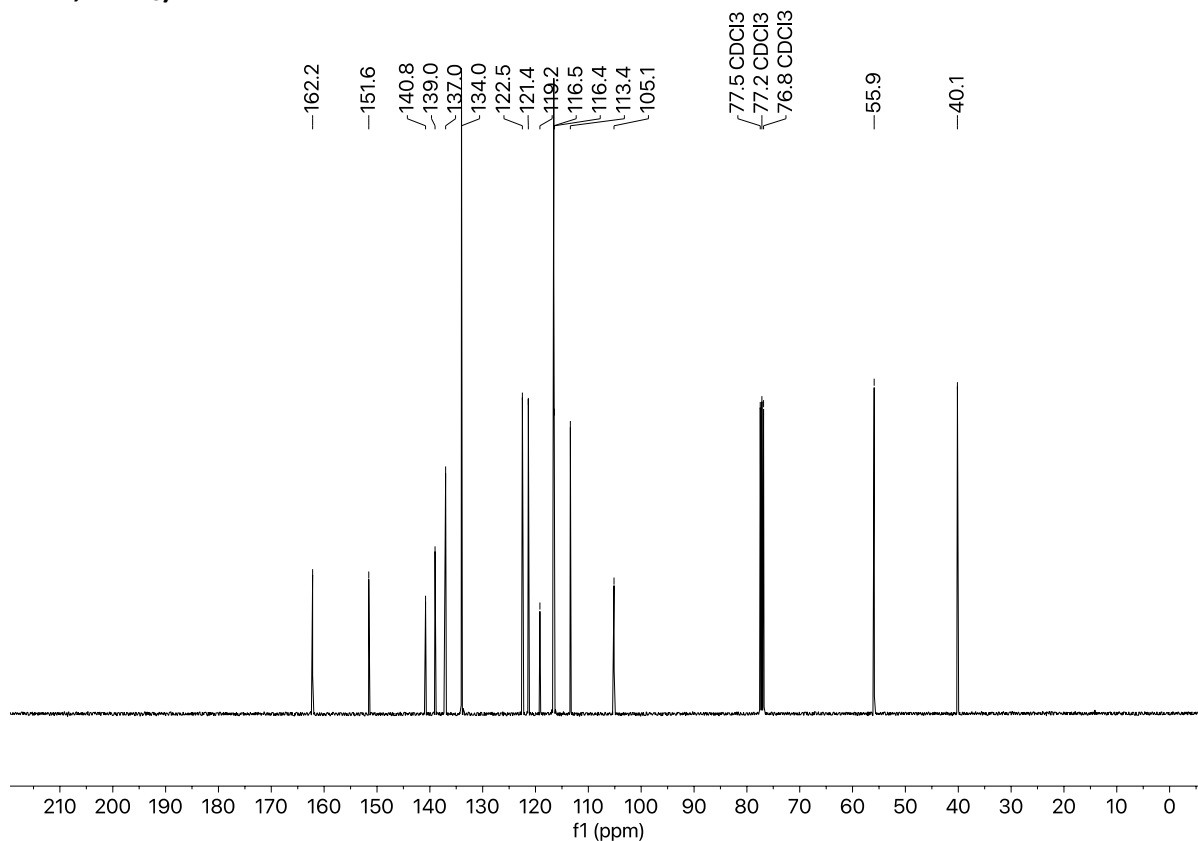
4-(5-Allyl-2-methoxyphenoxy)benzonitrile (3h)



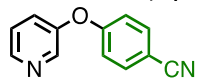
δ_H (400 MHz, $CDCl_3$)



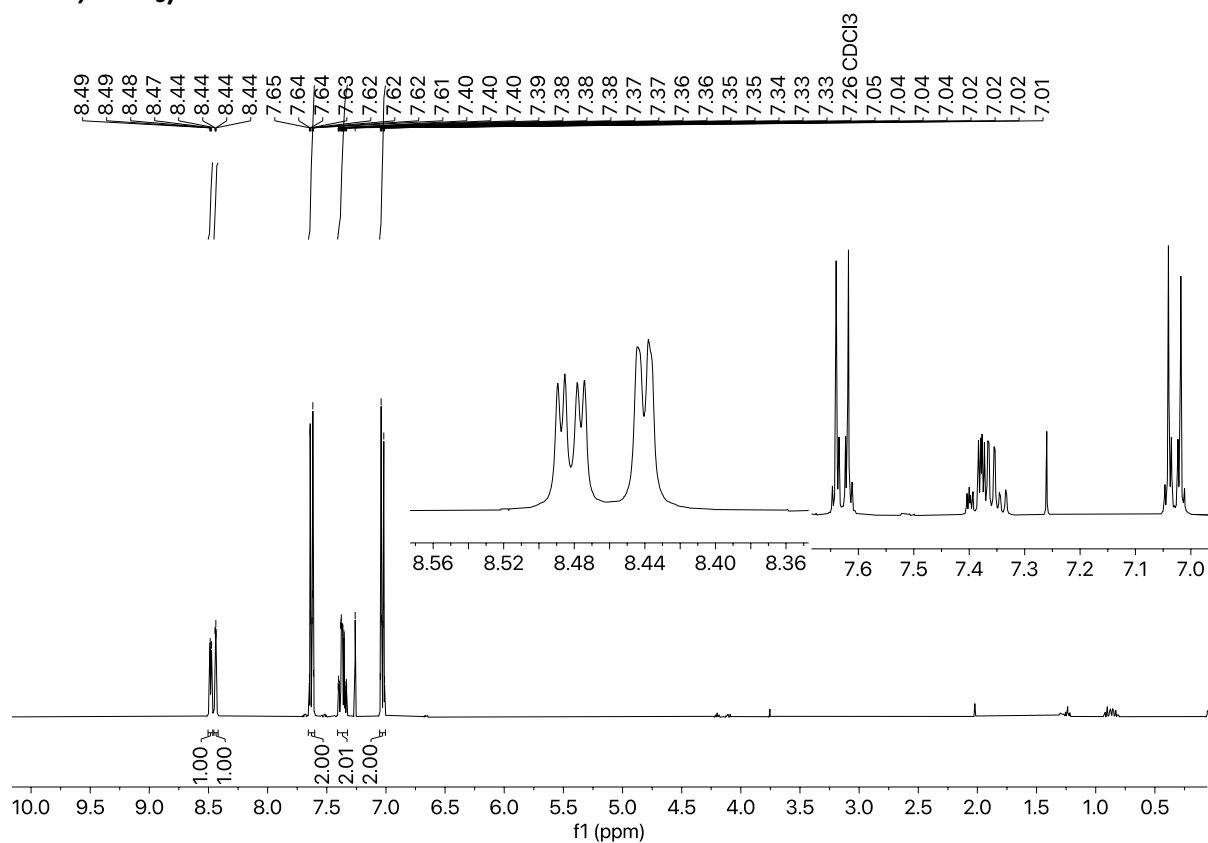
δ_C (101 MHz, $CDCl_3$)



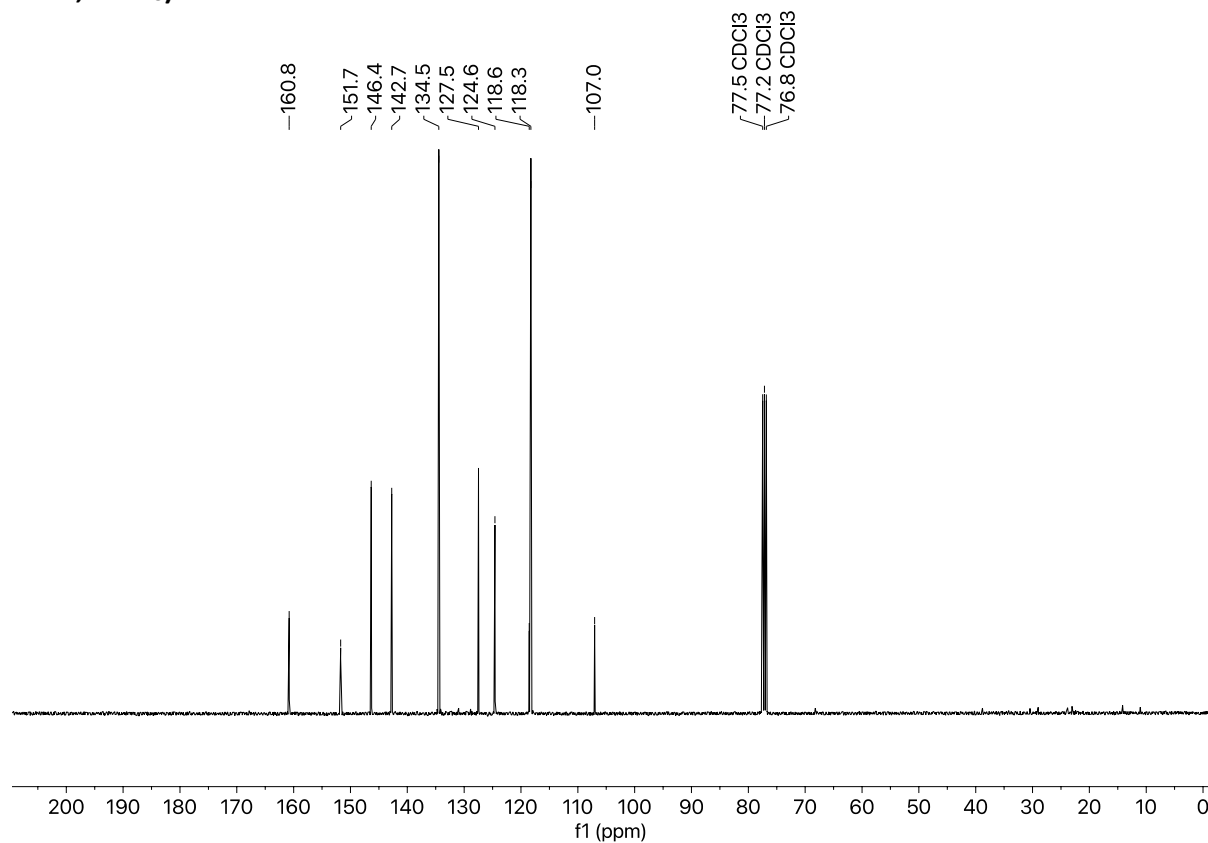
4-(Pyridin-3-yloxy)benzonitrile (3i)



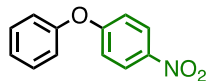
δ_H (400 MHz, $CDCl_3$)



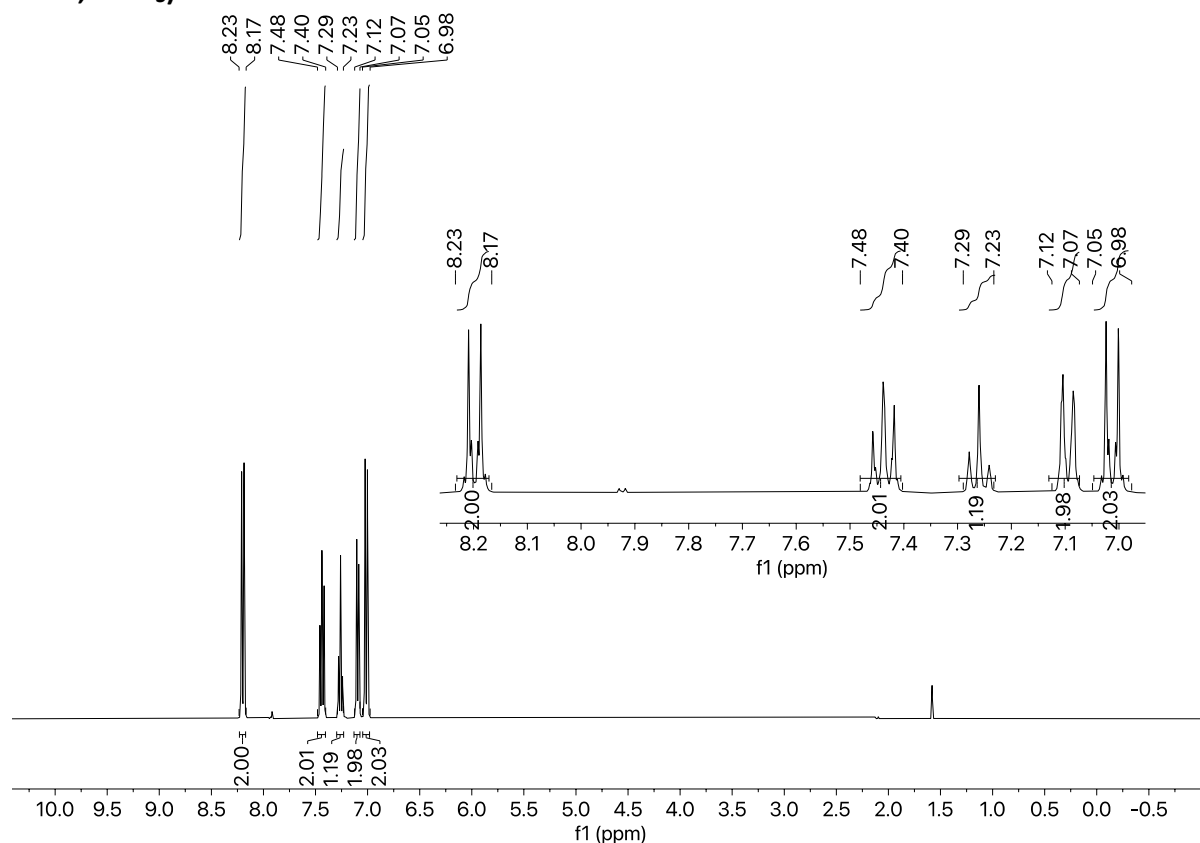
δ_C (101 MHz, $CDCl_3$)



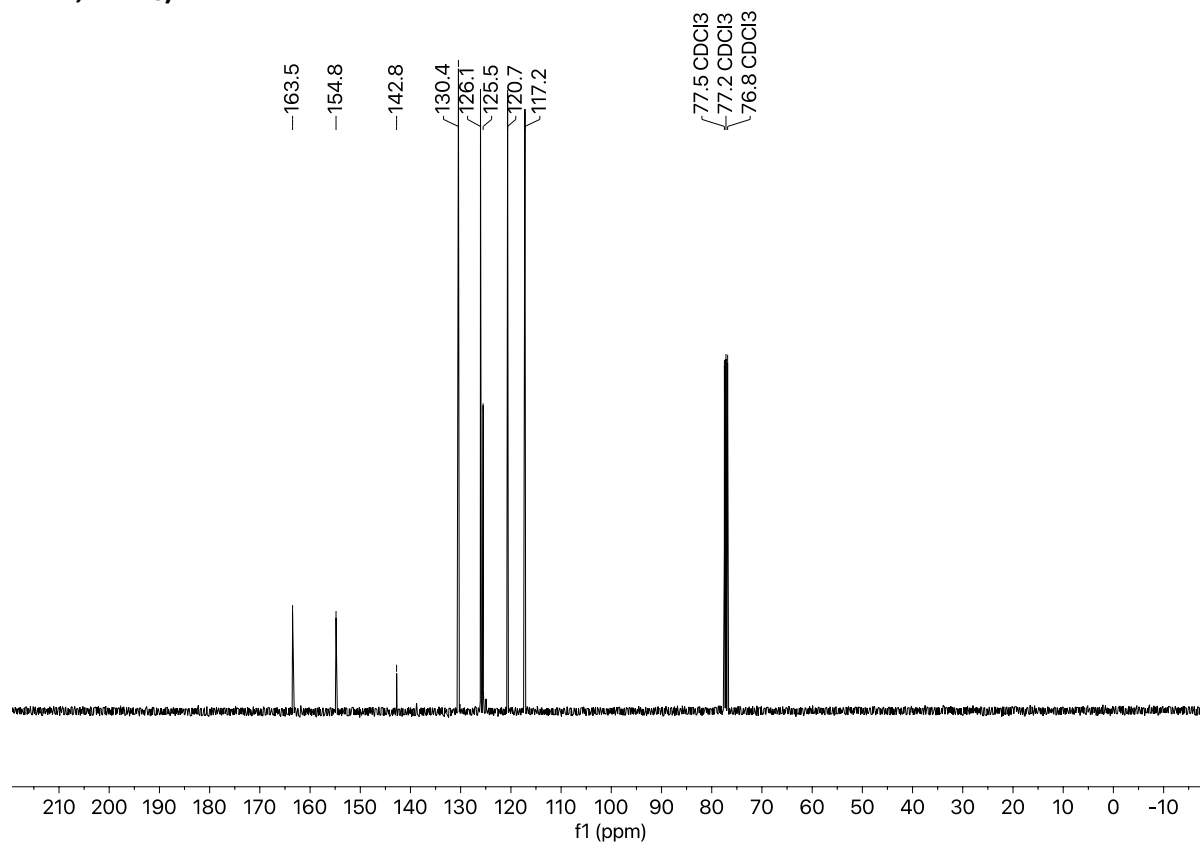
1-Nitro-4-phenoxybenzene (3j)



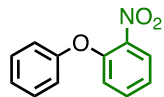
δ_H (400 MHz, $CDCl_3$)



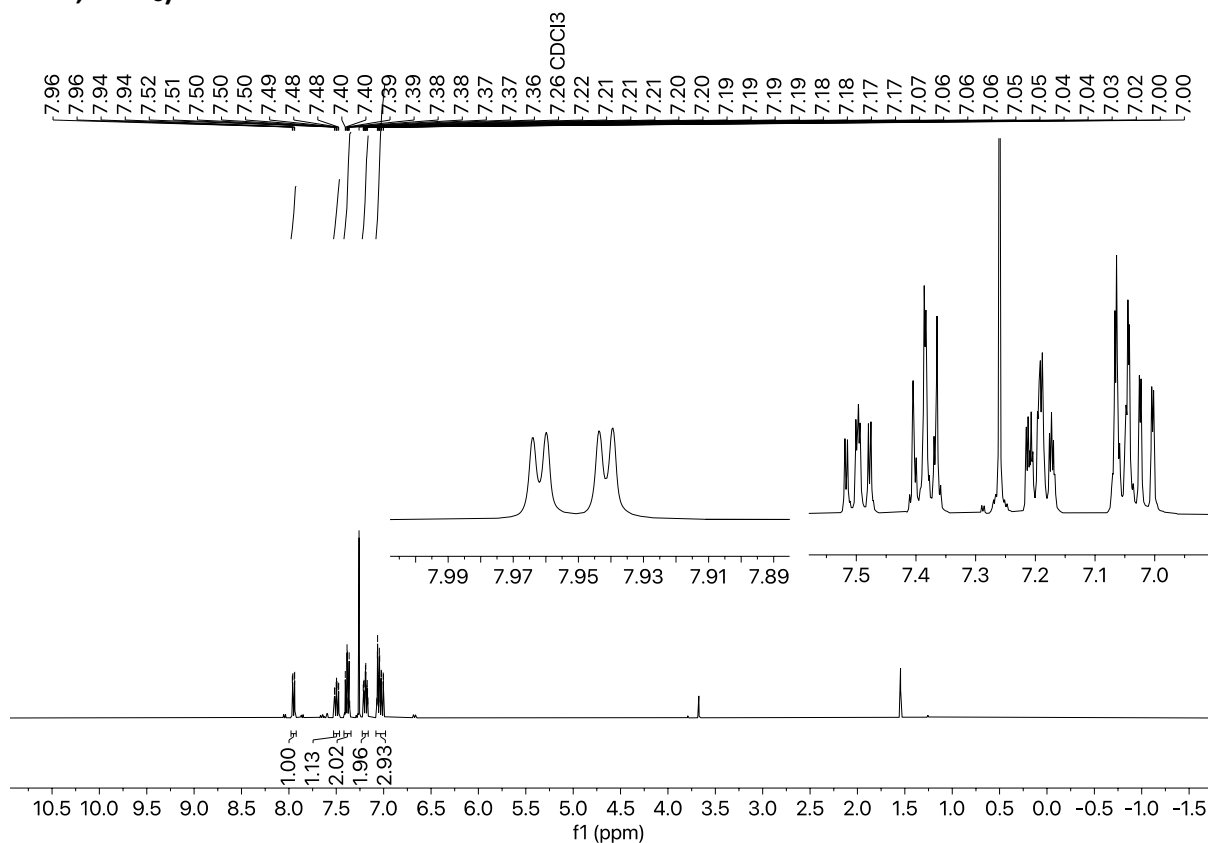
δ_C (101 MHz, $CDCl_3$)



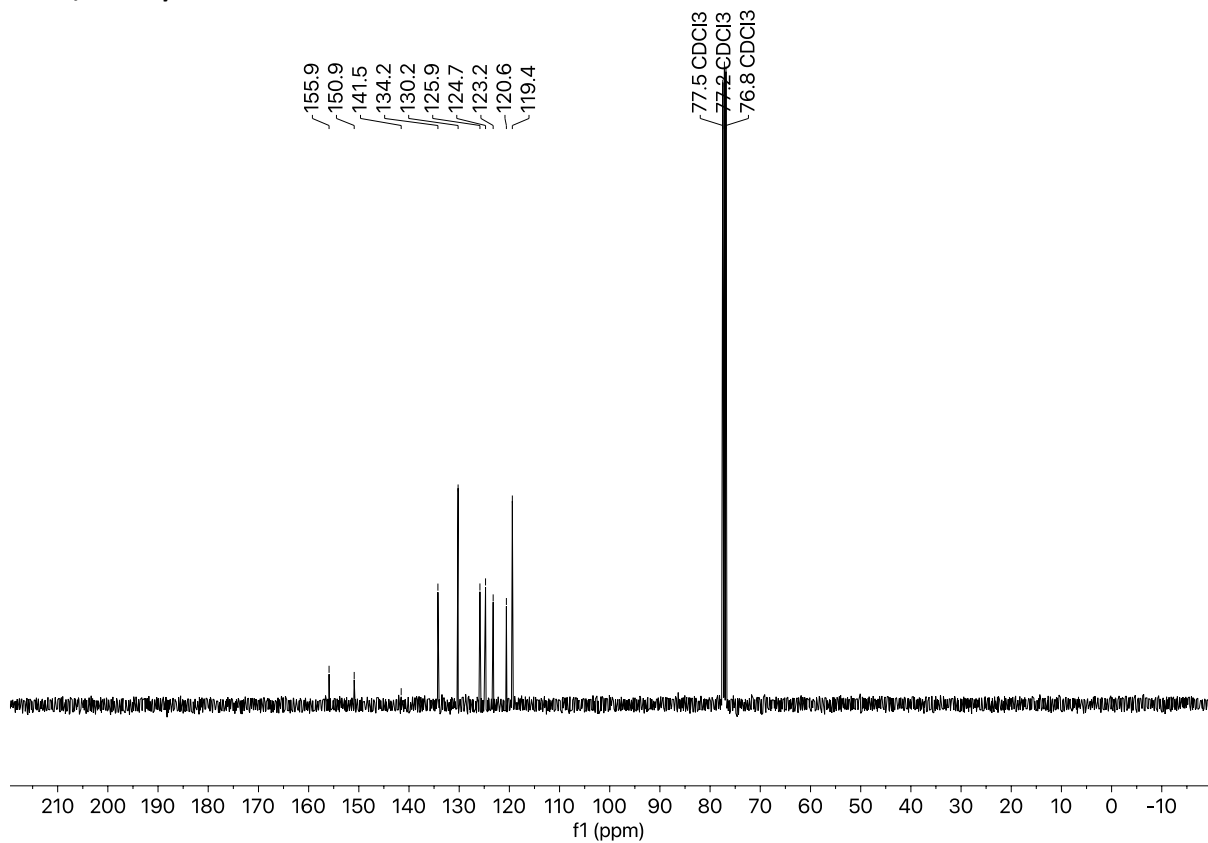
1-Nitro-2-phenoxybenzene (3k)



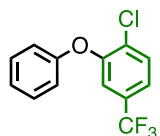
δ_H (400 MHz, $CDCl_3$)



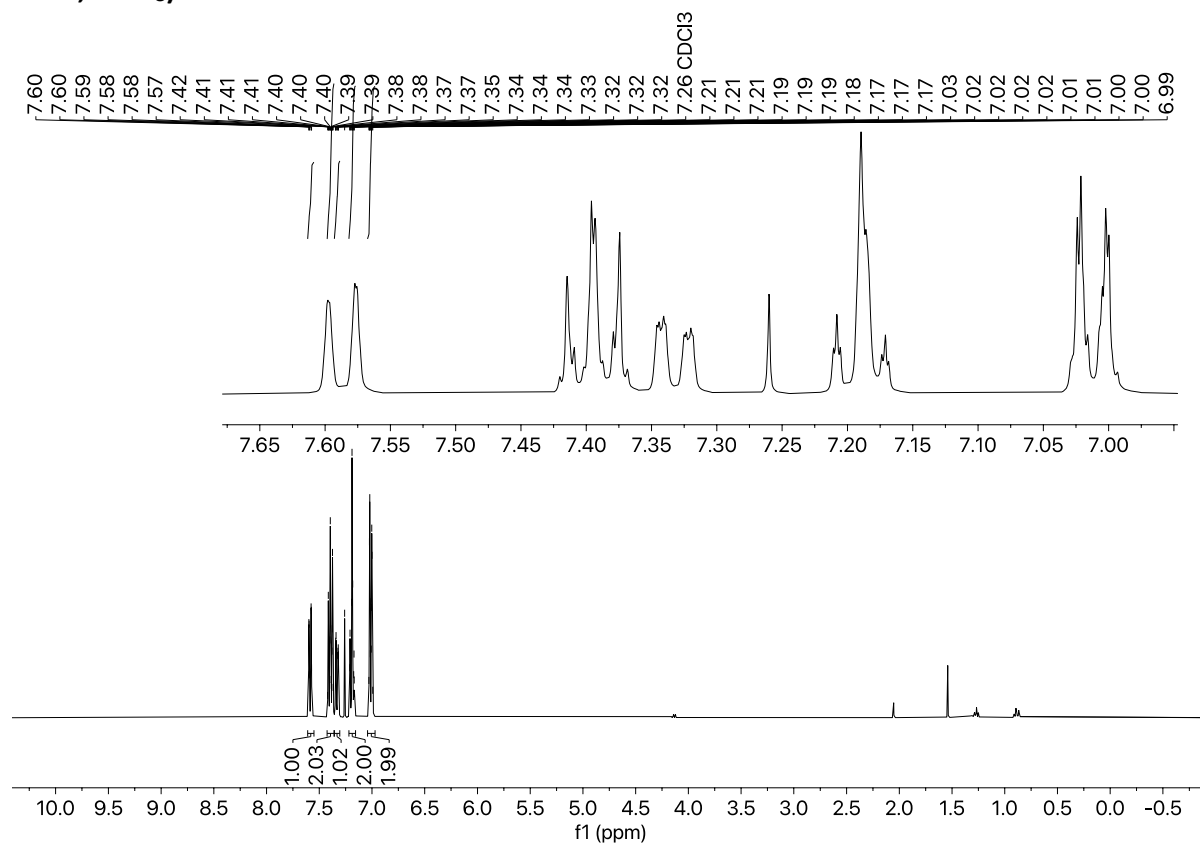
δ_C (101 MHz, $CDCl_3$)



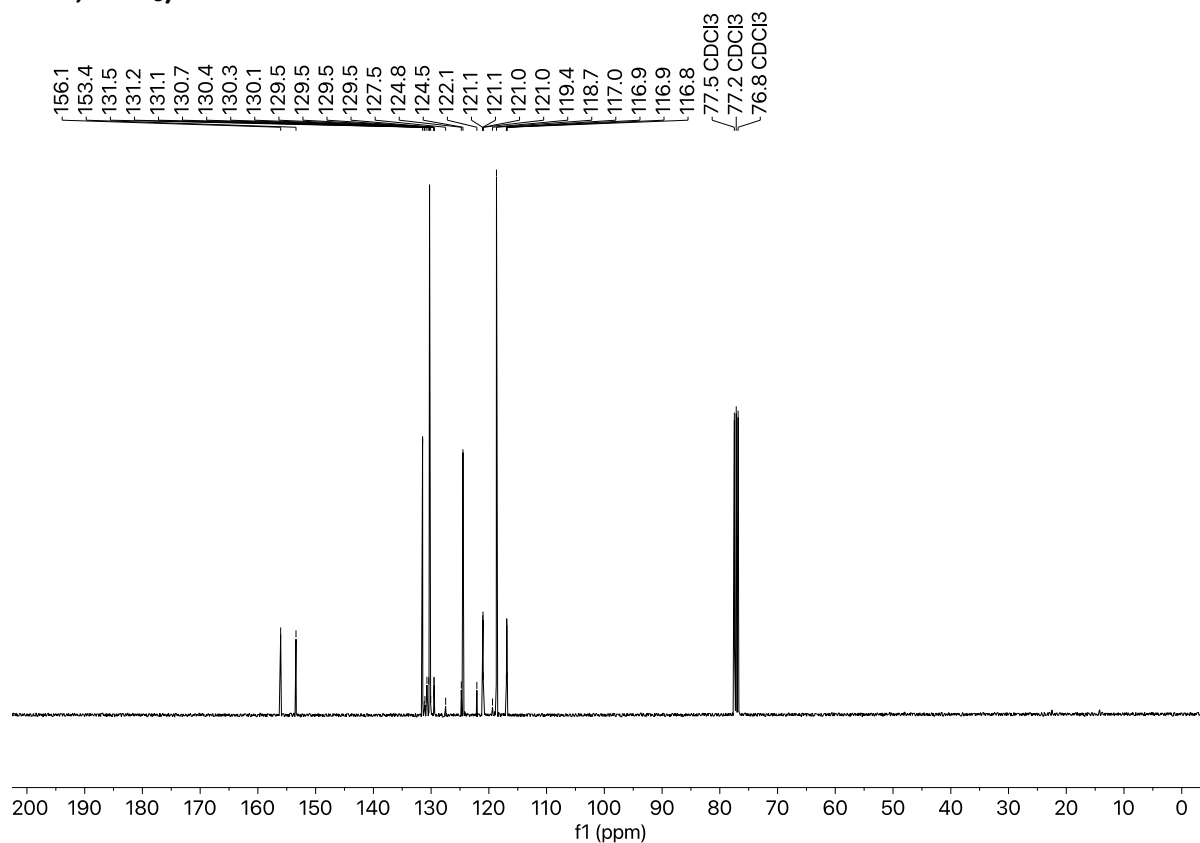
1-Chloro-2-phenoxy-4-(trifluoromethyl)benzene (3l)



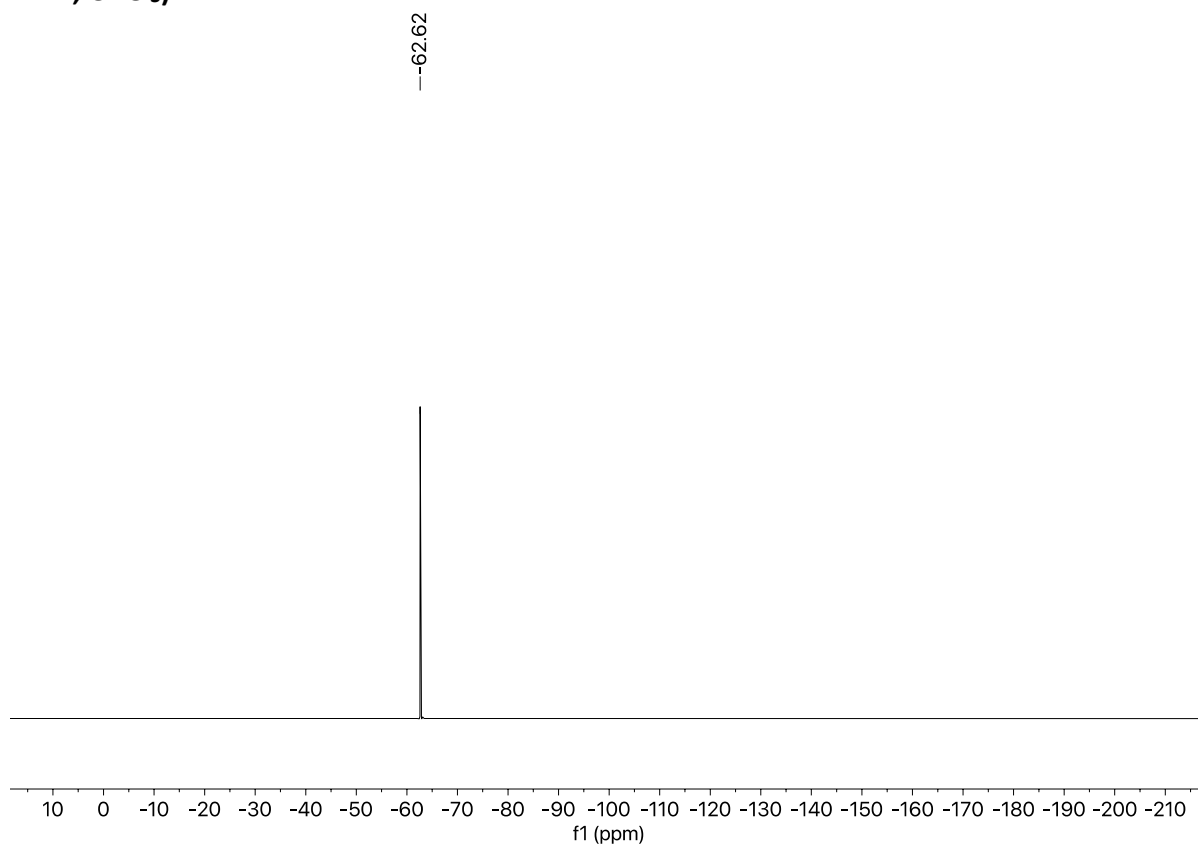
δ_H (400 MHz, $CDCl_3$)



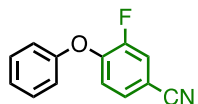
δ_C (101 MHz, $CDCl_3$)



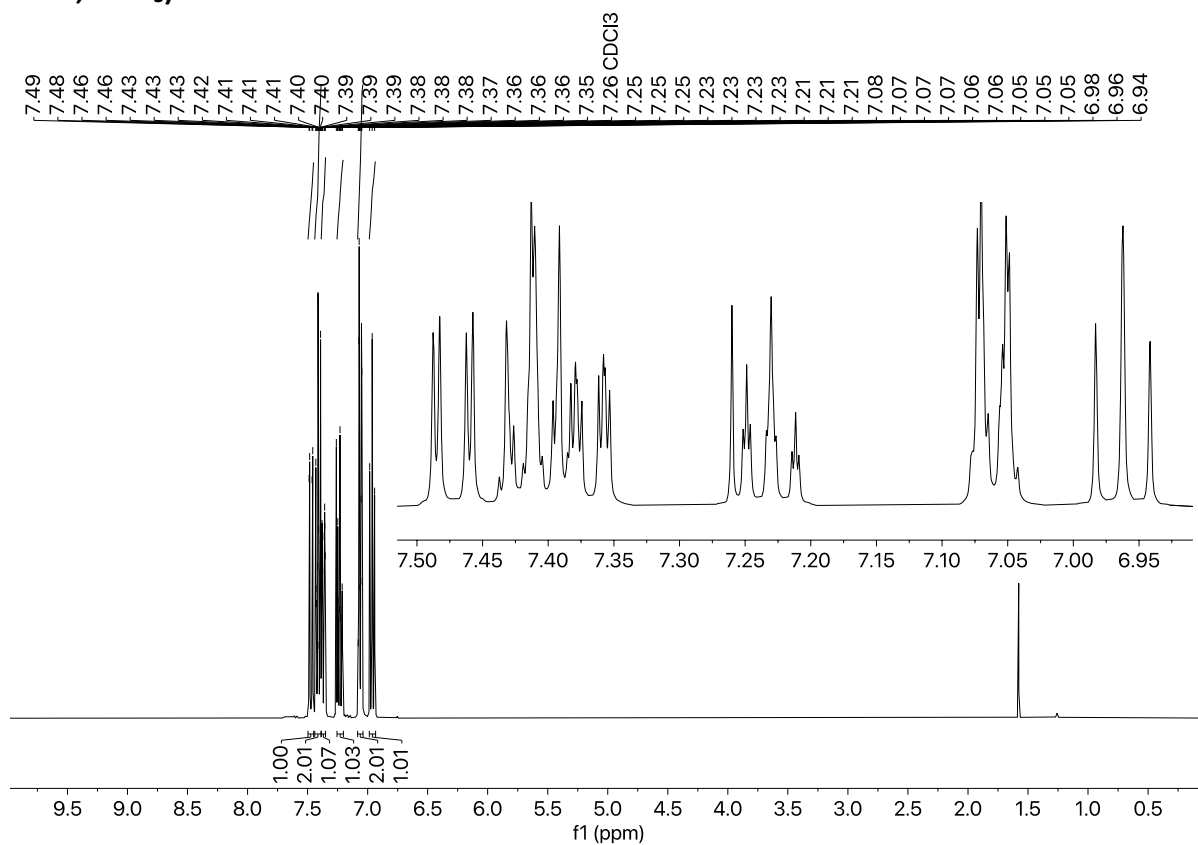
δ_F (376 MHz, $CDCl_3$)



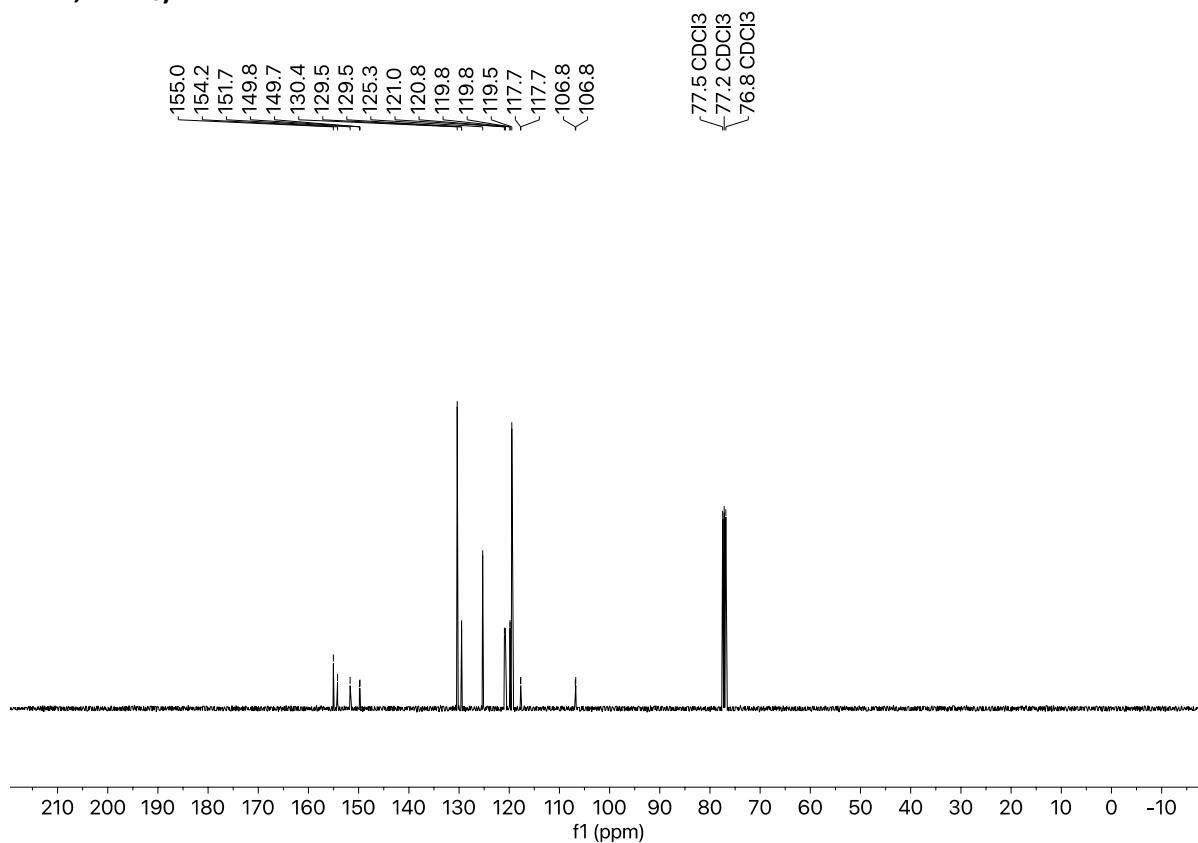
3-Fluoro-4-phenoxybenzonitrile (3m)



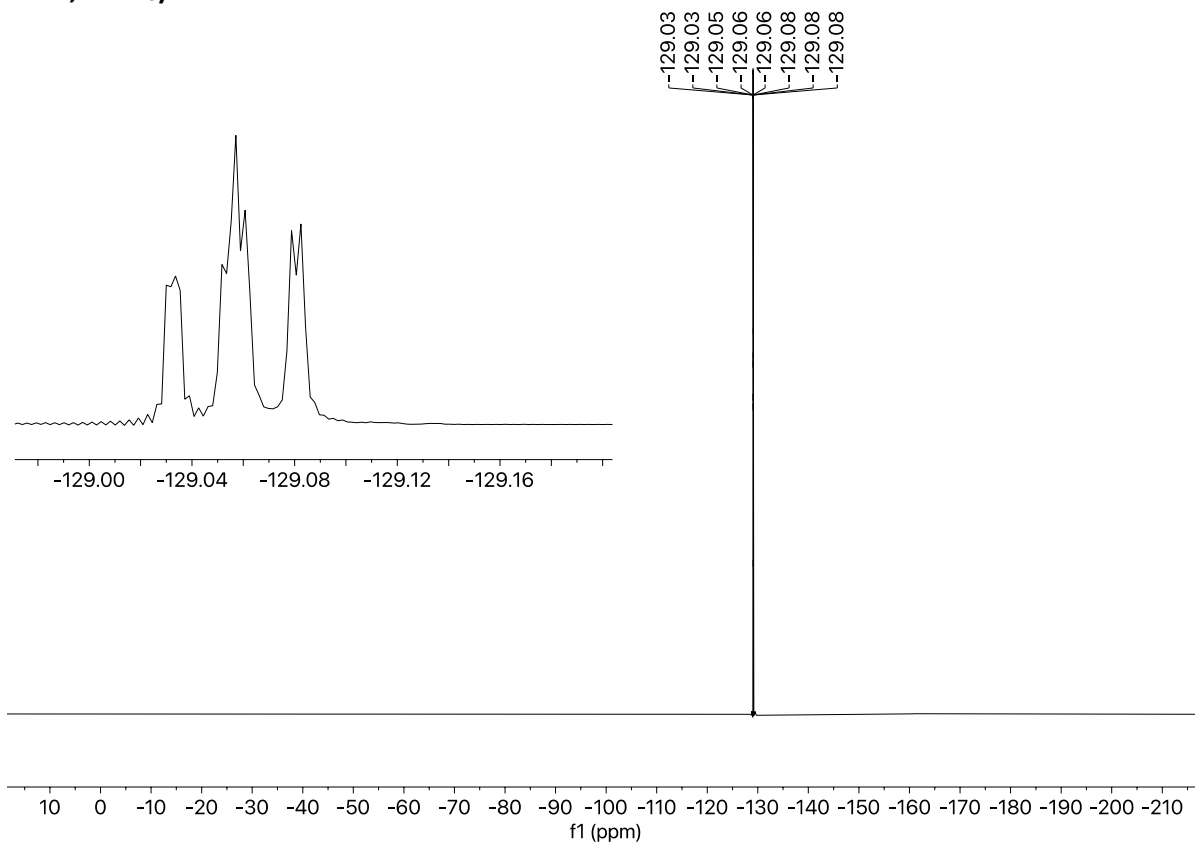
δ_H (400 MHz, $CDCl_3$)



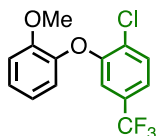
δ_C (101 MHz, $CDCl_3$)



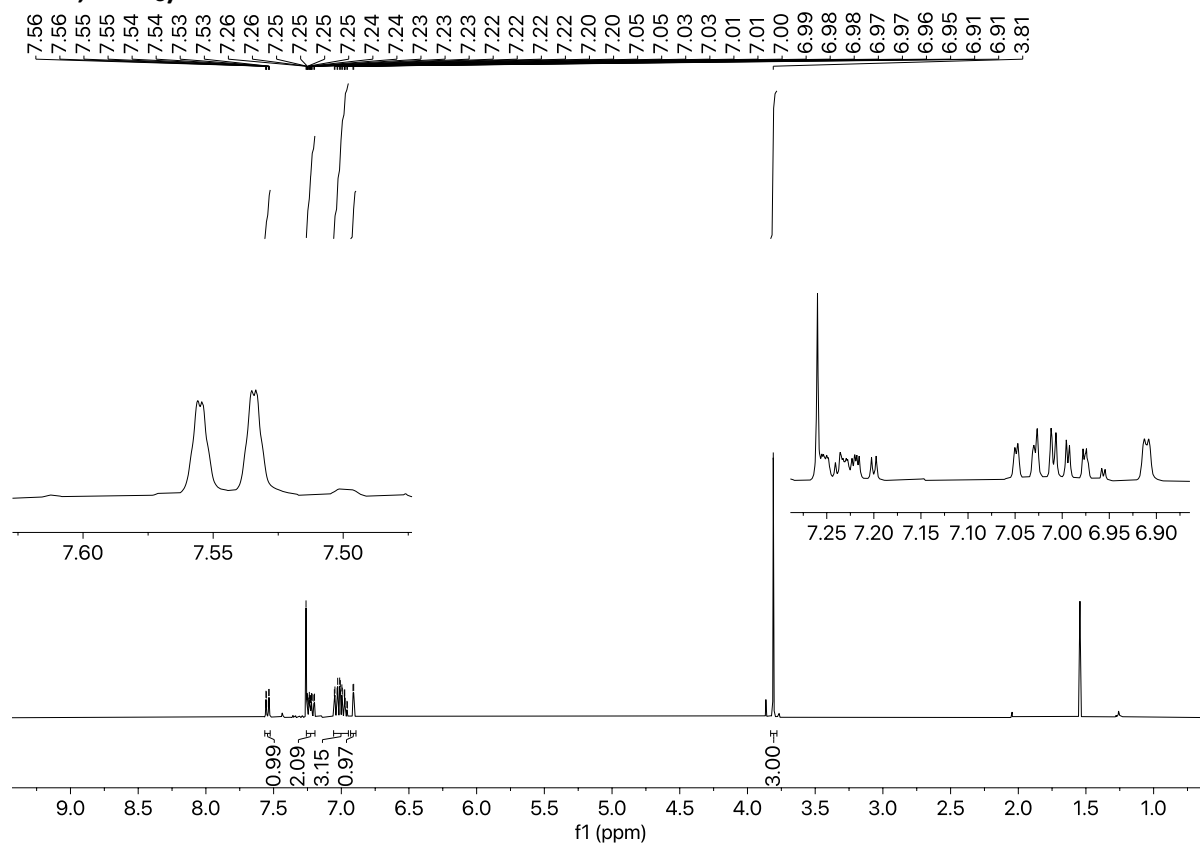
δ_F (376 MHz, $CDCl_3$)



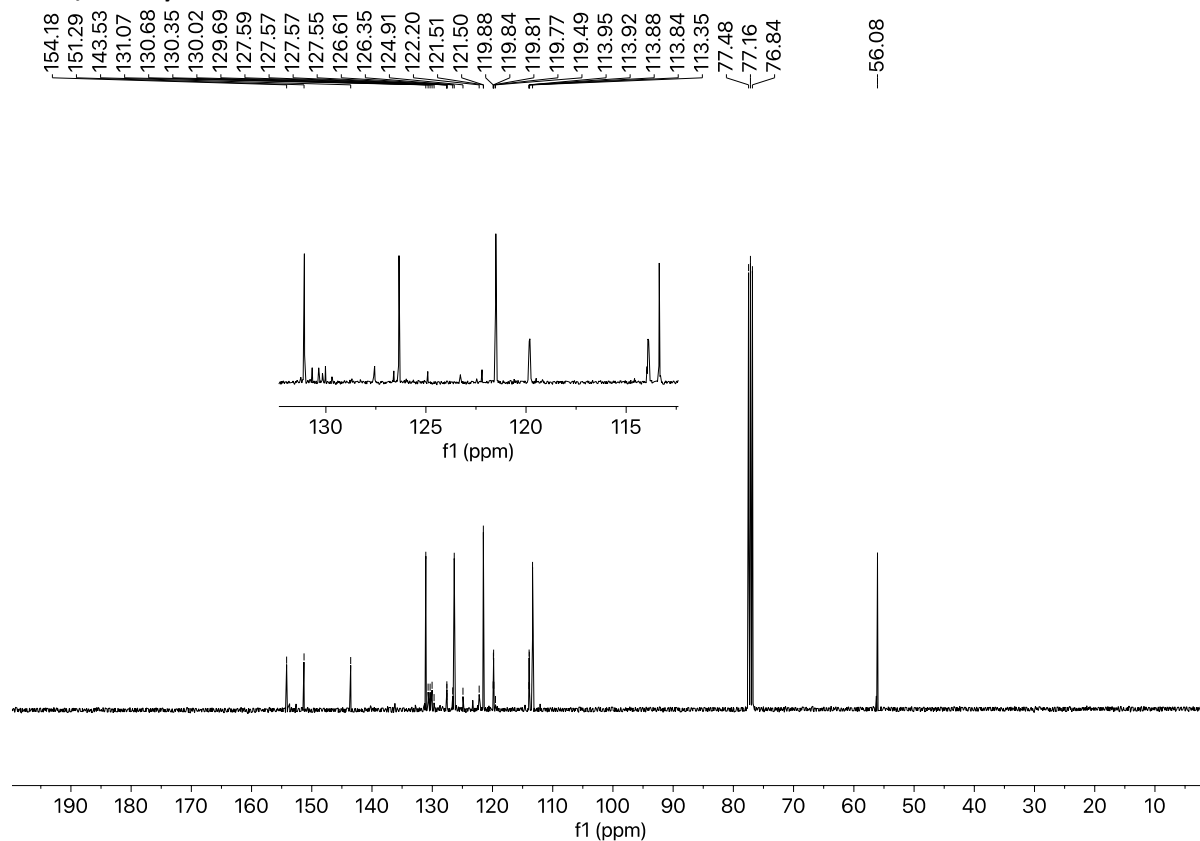
1-Chloro-2-(2-methoxyphenoxy)-4-(trifluoromethyl)benzene (3n)



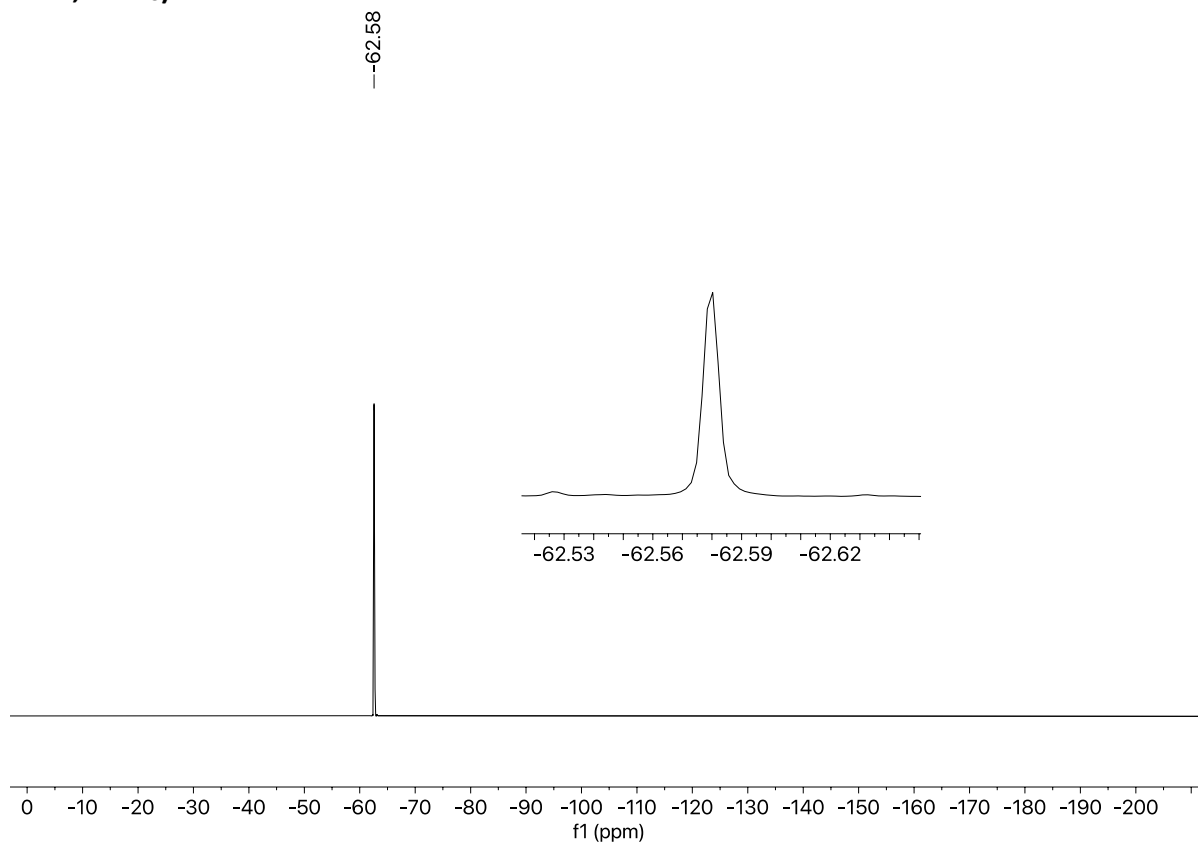
δ_H (400 MHz, $CDCl_3$)



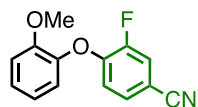
δ_C (101 MHz, $CDCl_3$)



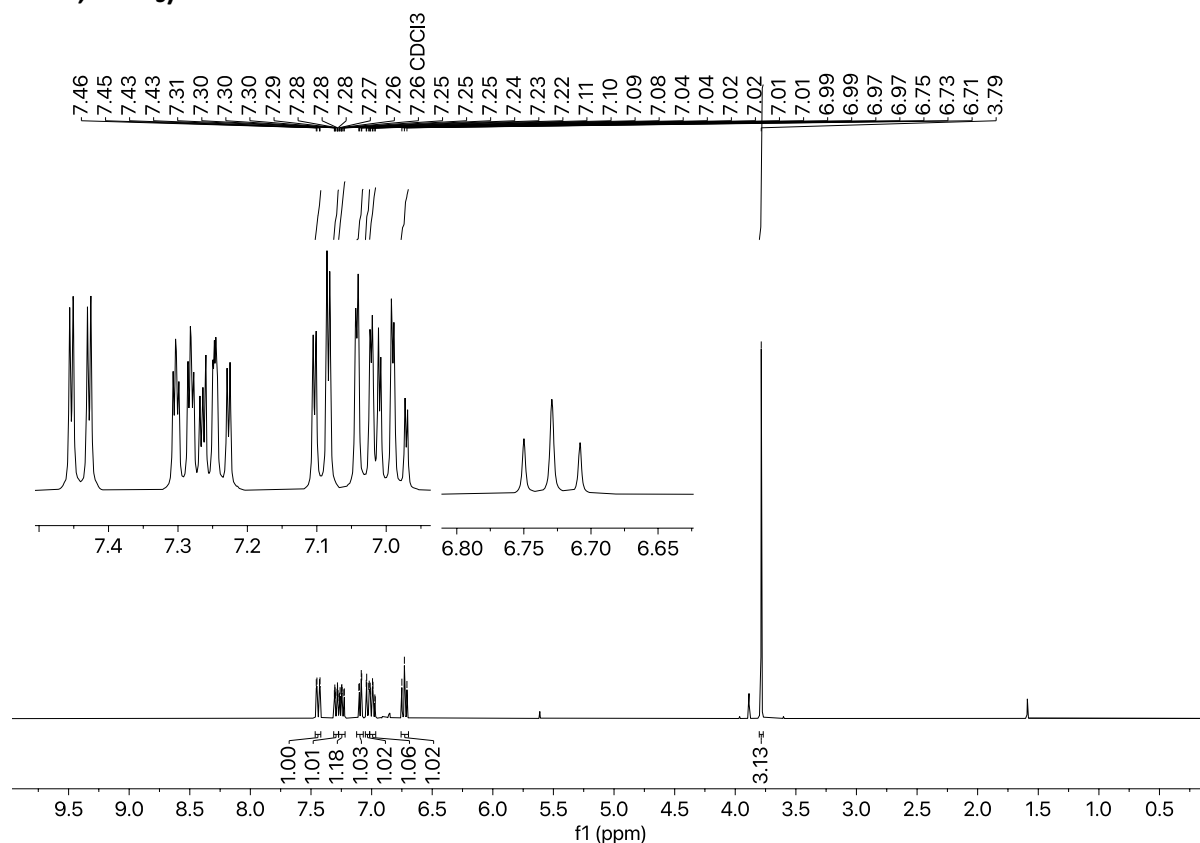
δ_F (376 MHz, $CDCl_3$)



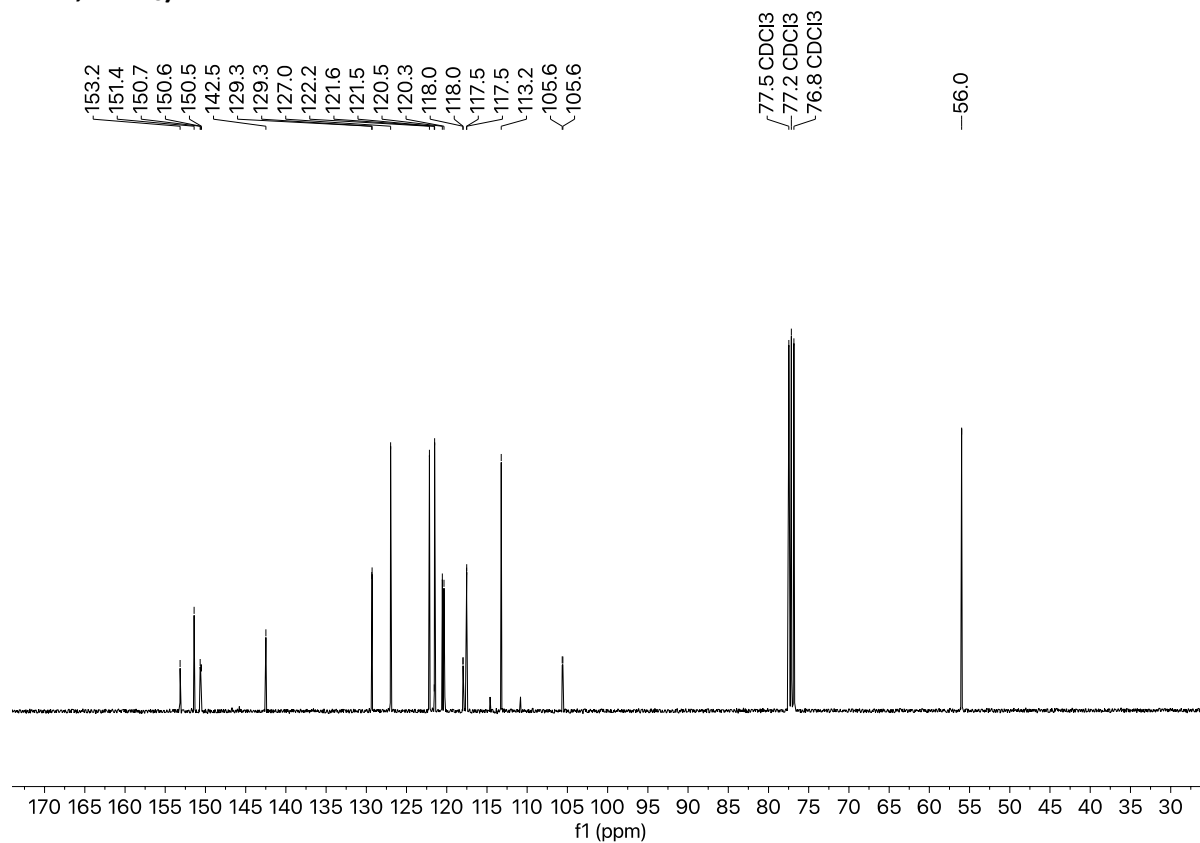
3-Fluoro-4-(2-methoxyphenoxy)benzonitrile (3o)



δ_H (400 MHz, $CDCl_3$)

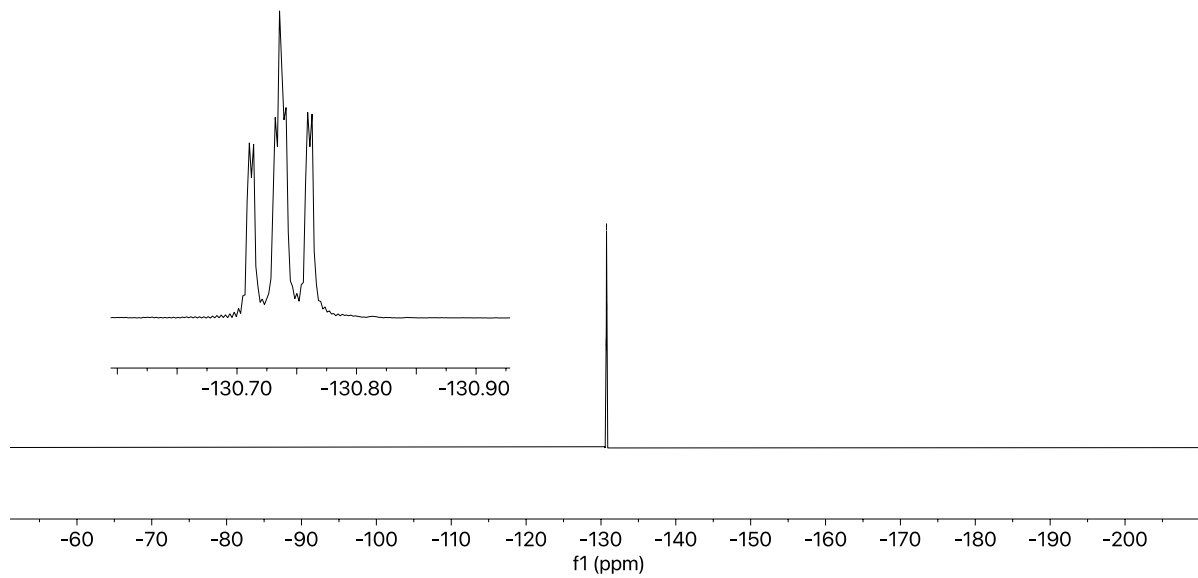


δ_C (101 MHz, $CDCl_3$)

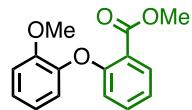


δ_F (376 MHz, $CDCl_3$)

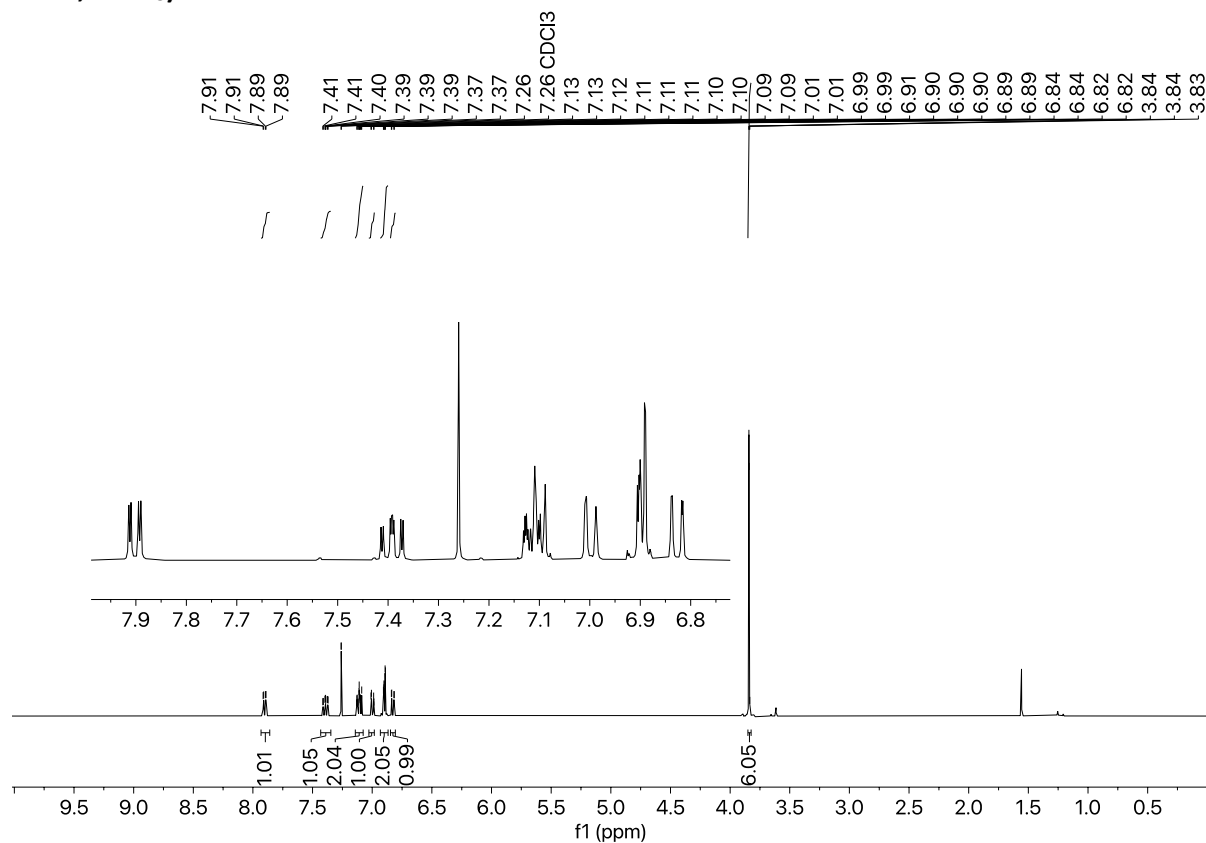
-130.71
-130.73
-130.74
-130.74
-130.76



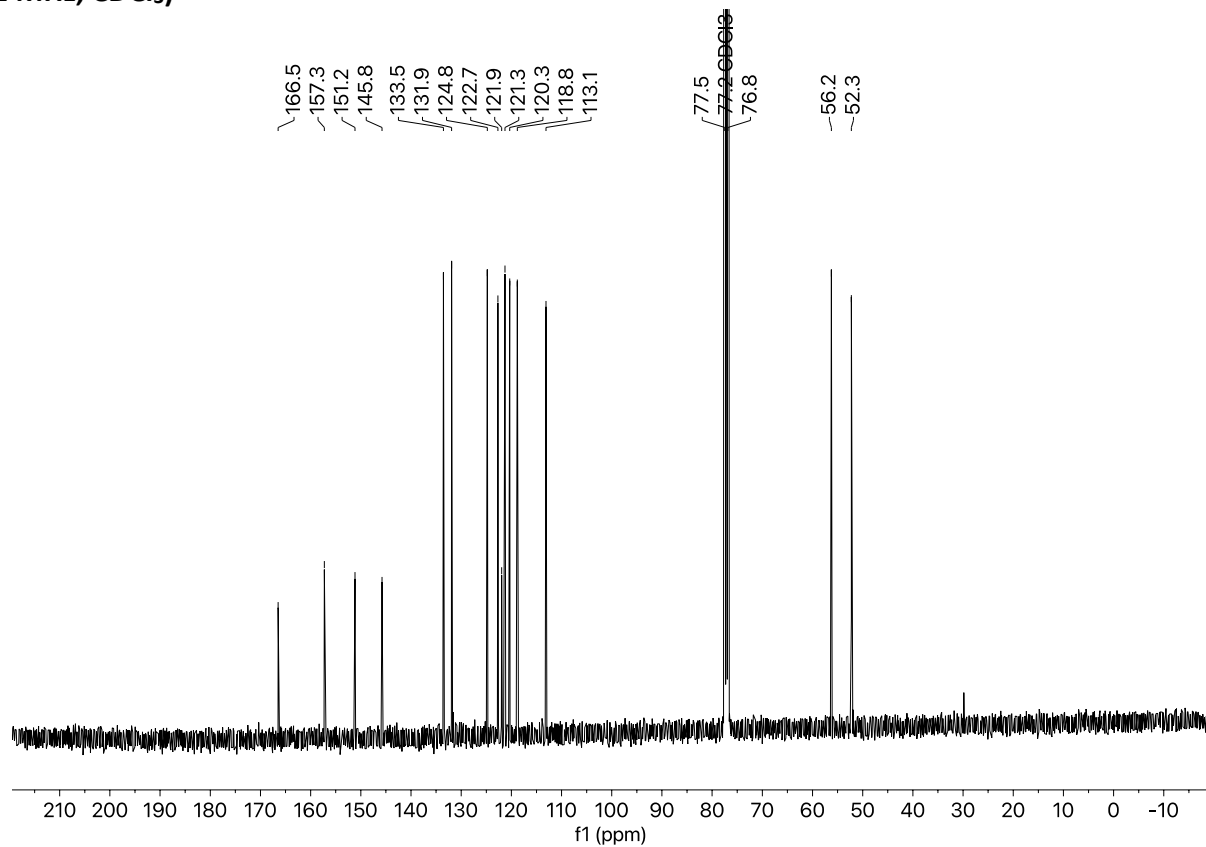
Methyl 2-(2-methoxyphenoxy)benzoate (3p)



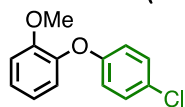
δ_H (400 MHz, $CDCl_3$)



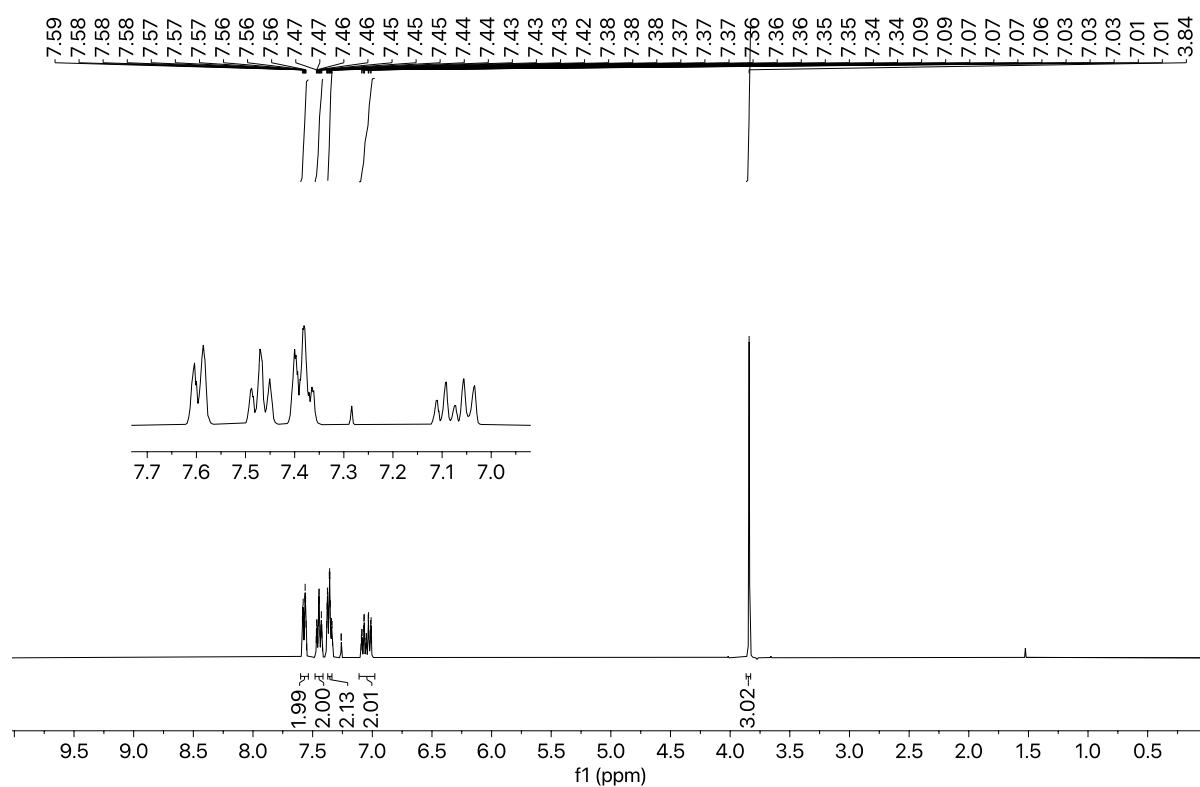
δ_C (101 MHz, $CDCl_3$)



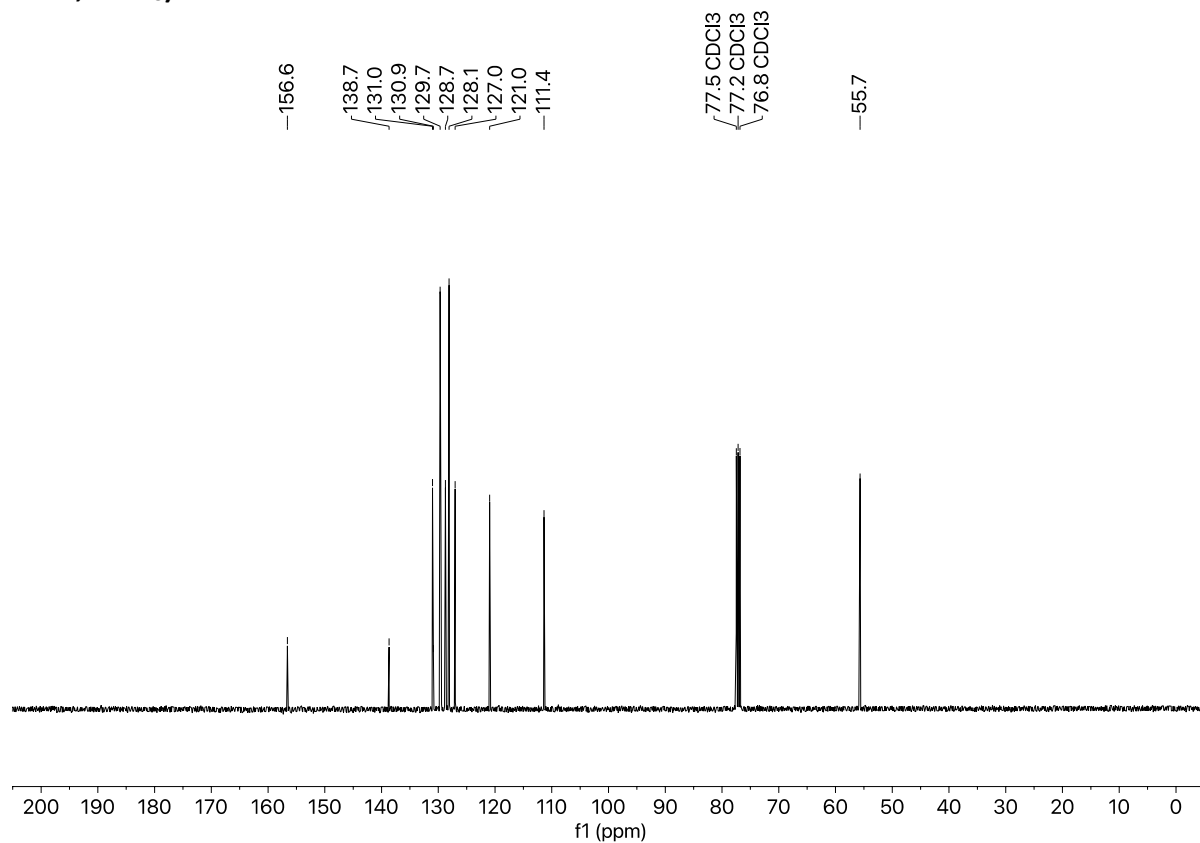
1-(4-Chlorophenoxy)-2-methoxybenzene (3q)



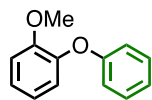
δ_H (400 MHz, $CDCl_3$)



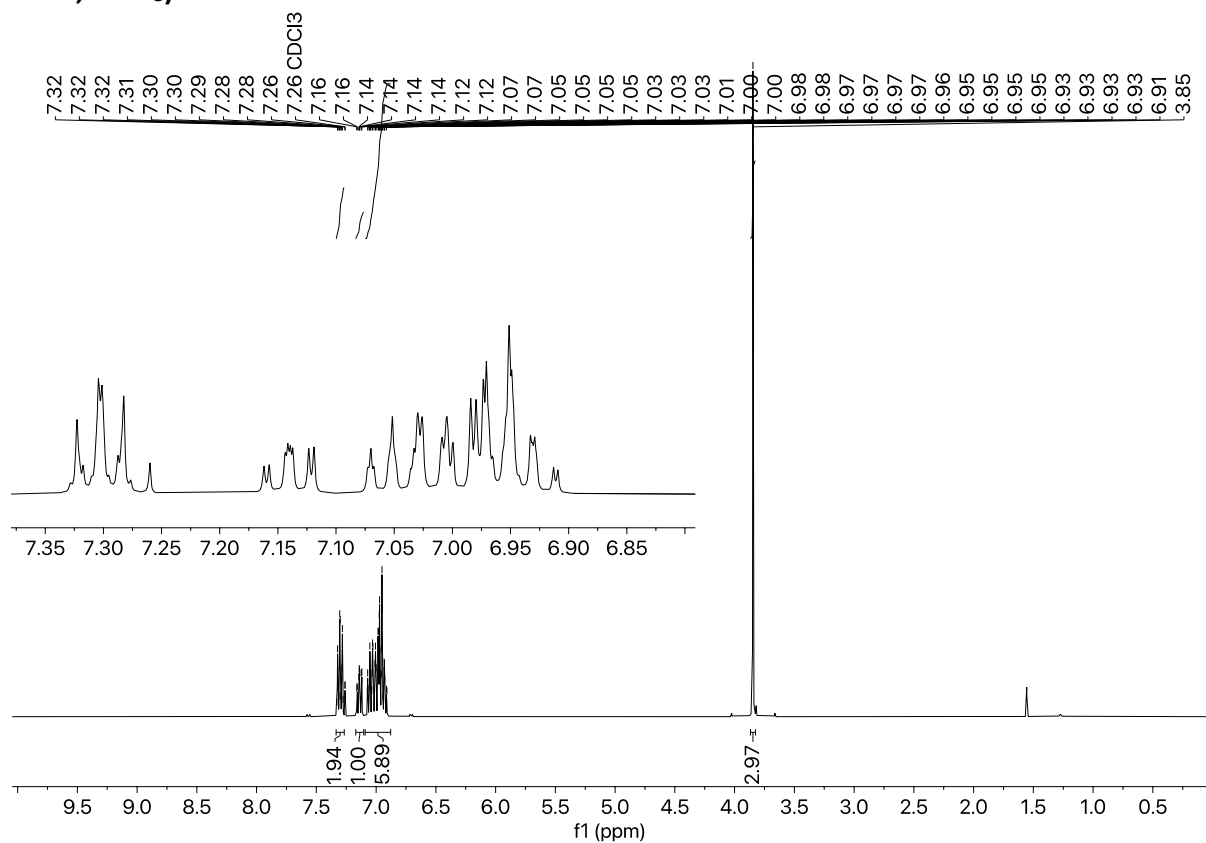
δ_C (101 MHz, $CDCl_3$)



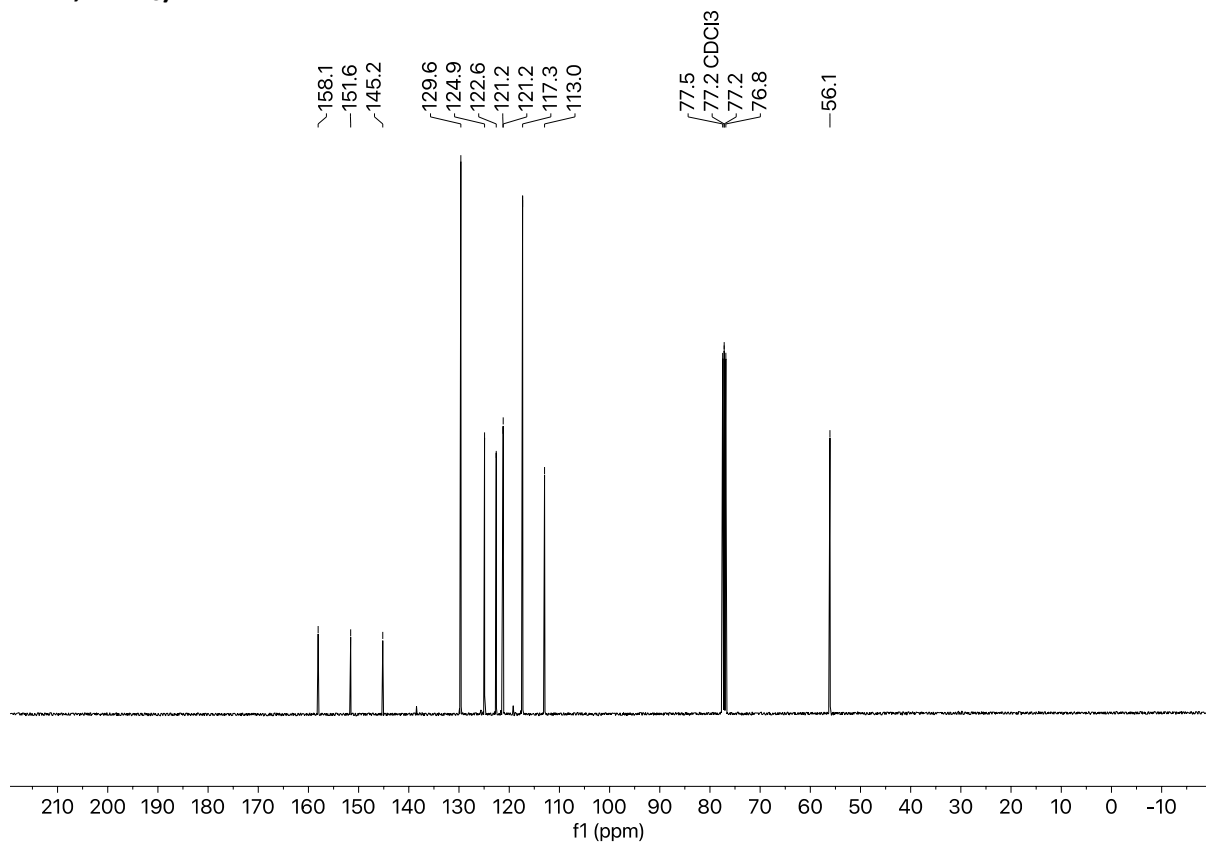
1-Methoxy-2-phenoxybenzene (3r)



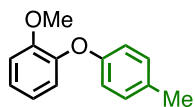
δ_H (400 MHz, $CDCl_3$)



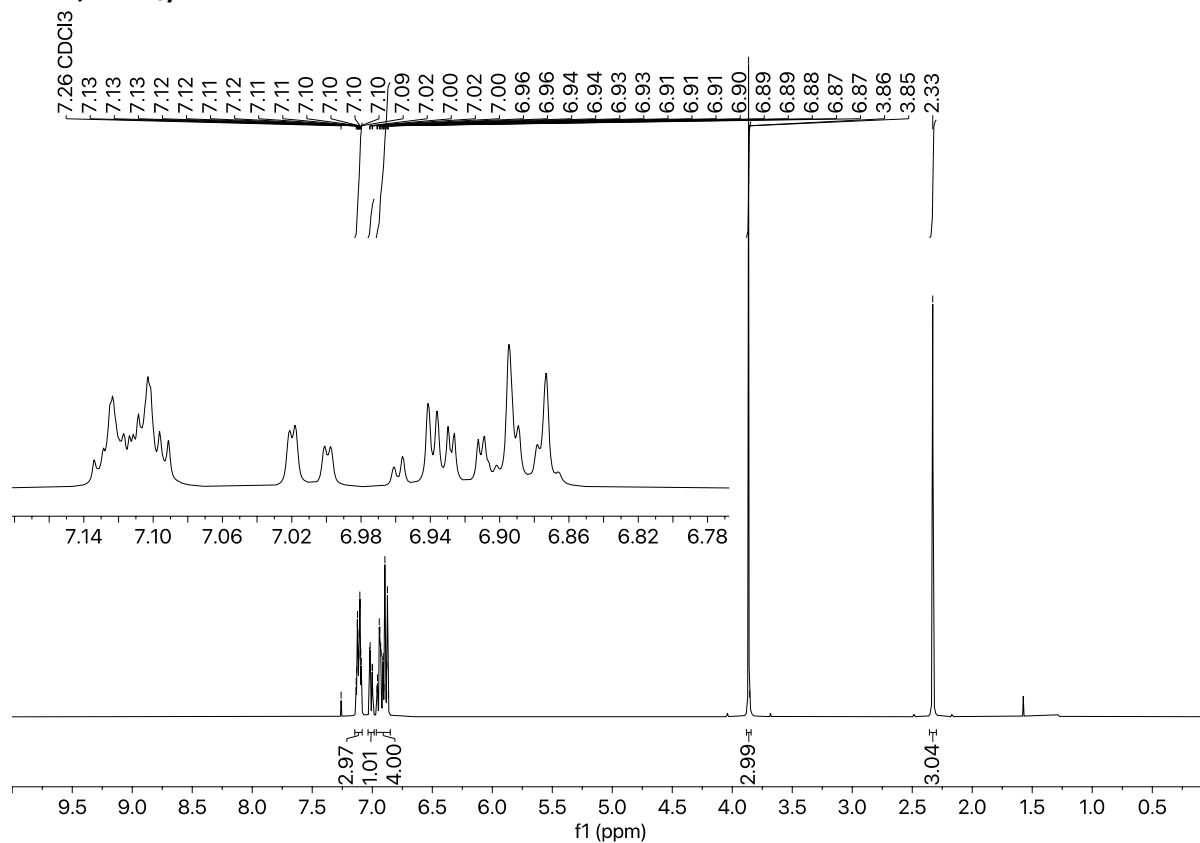
δ_C (101 MHz, $CDCl_3$)



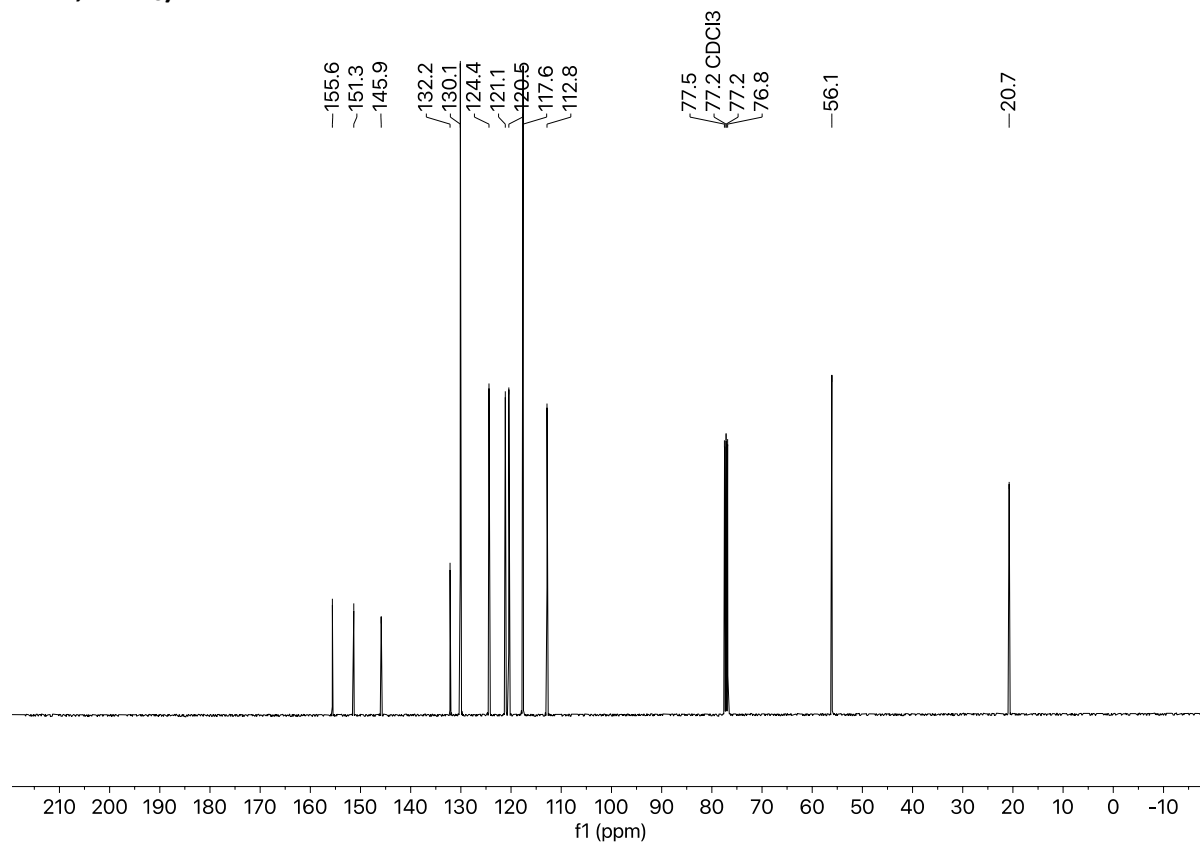
1-Methoxy-2-(*p*-toloxy)benzene (3s)



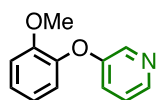
δ_H (400 MHz, $CDCl_3$)



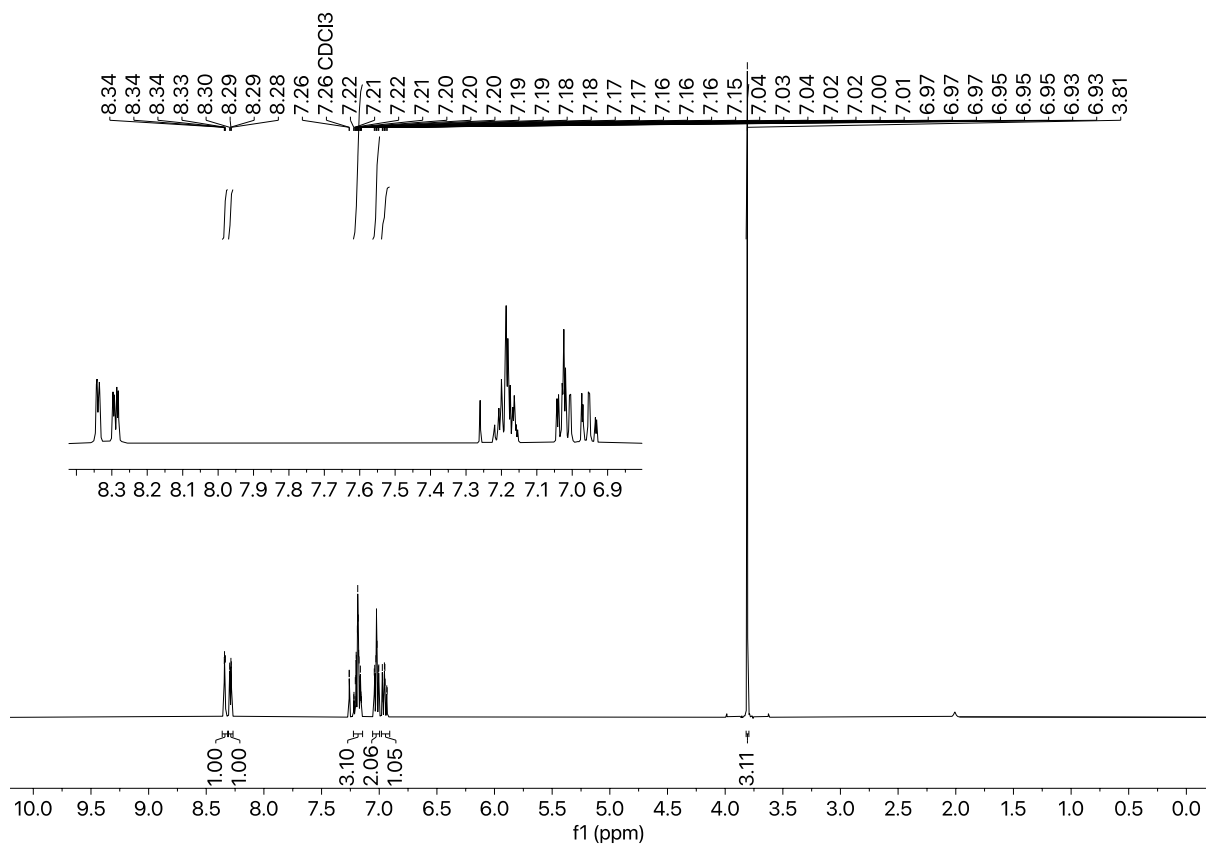
δ_C (101 MHz, $CDCl_3$)



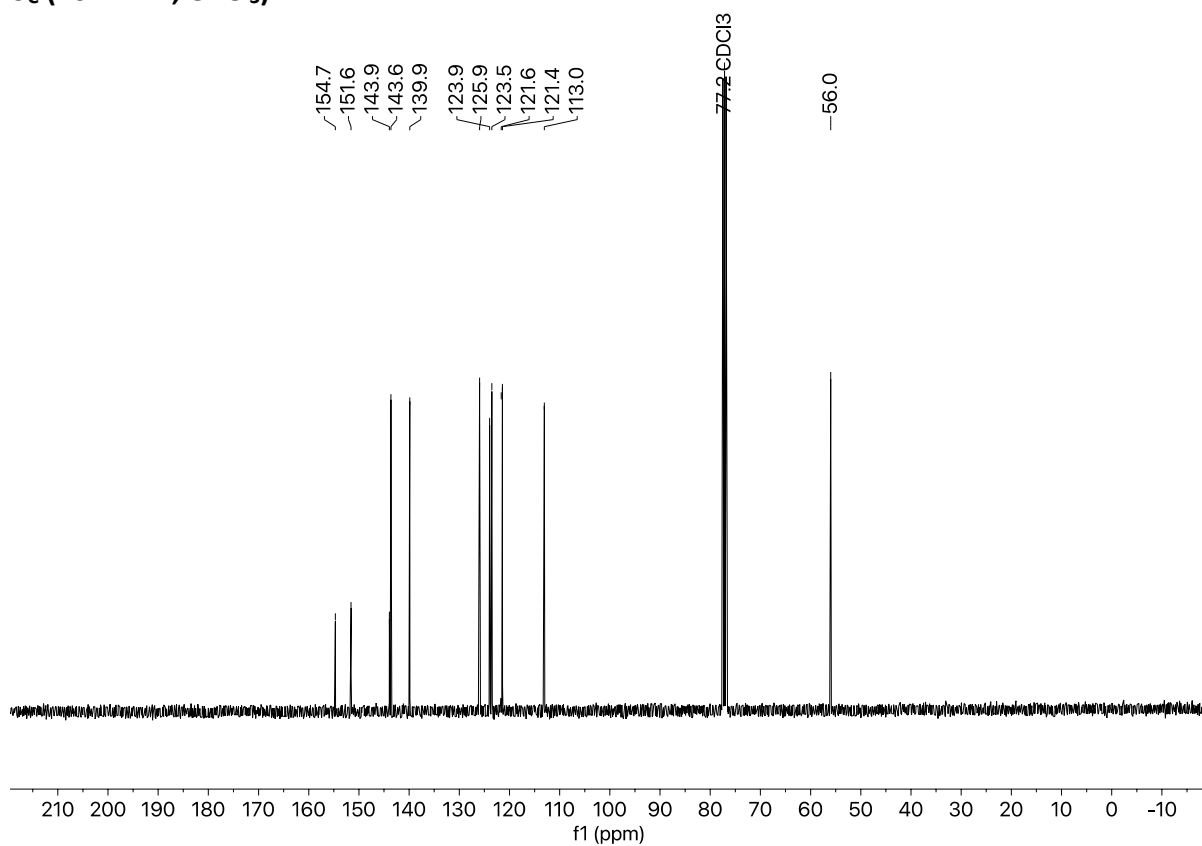
3-(2-methoxyphenoxy)pyridine (3t)

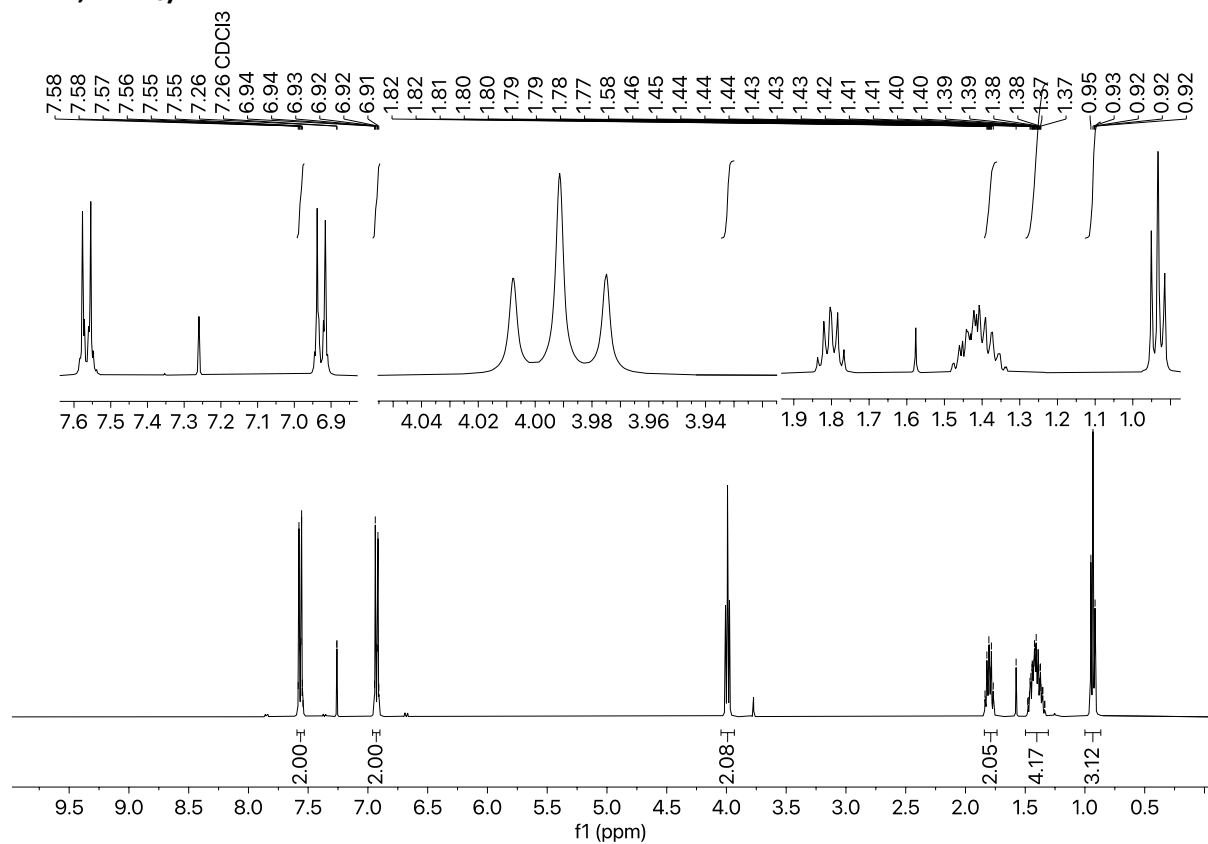
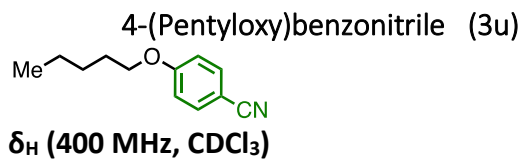


δ_H (400 MHz, $CDCl_3$)

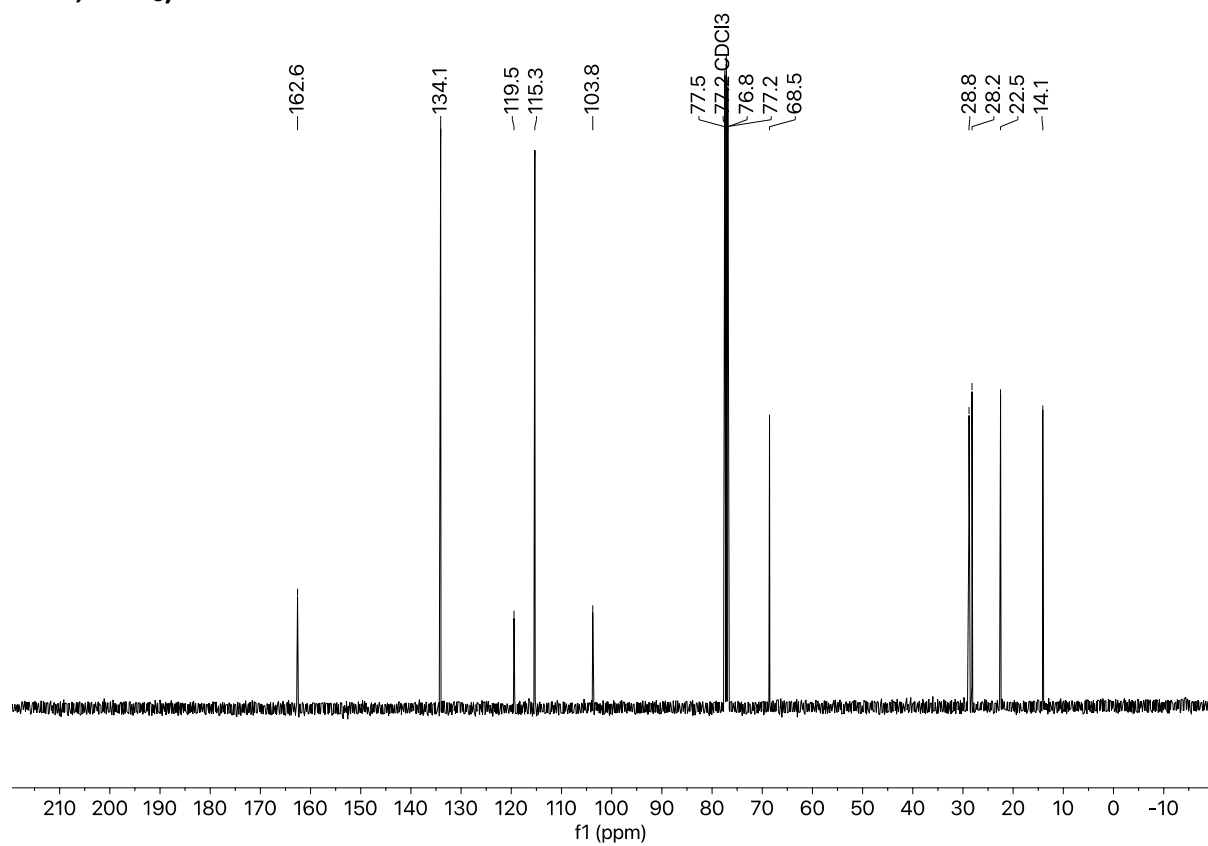


δ_C (101 MHz, $CDCl_3$)

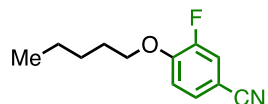




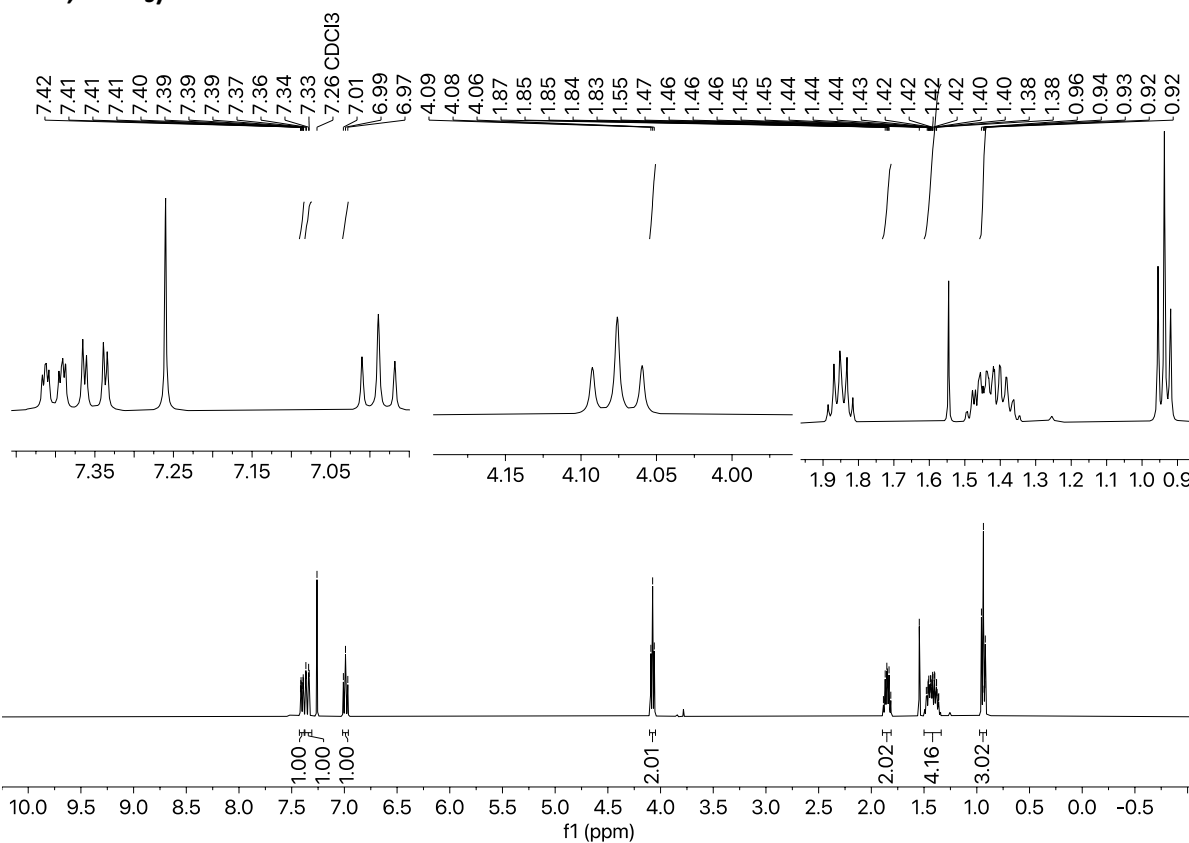
δ_C (101 MHz, $CDCl_3$)



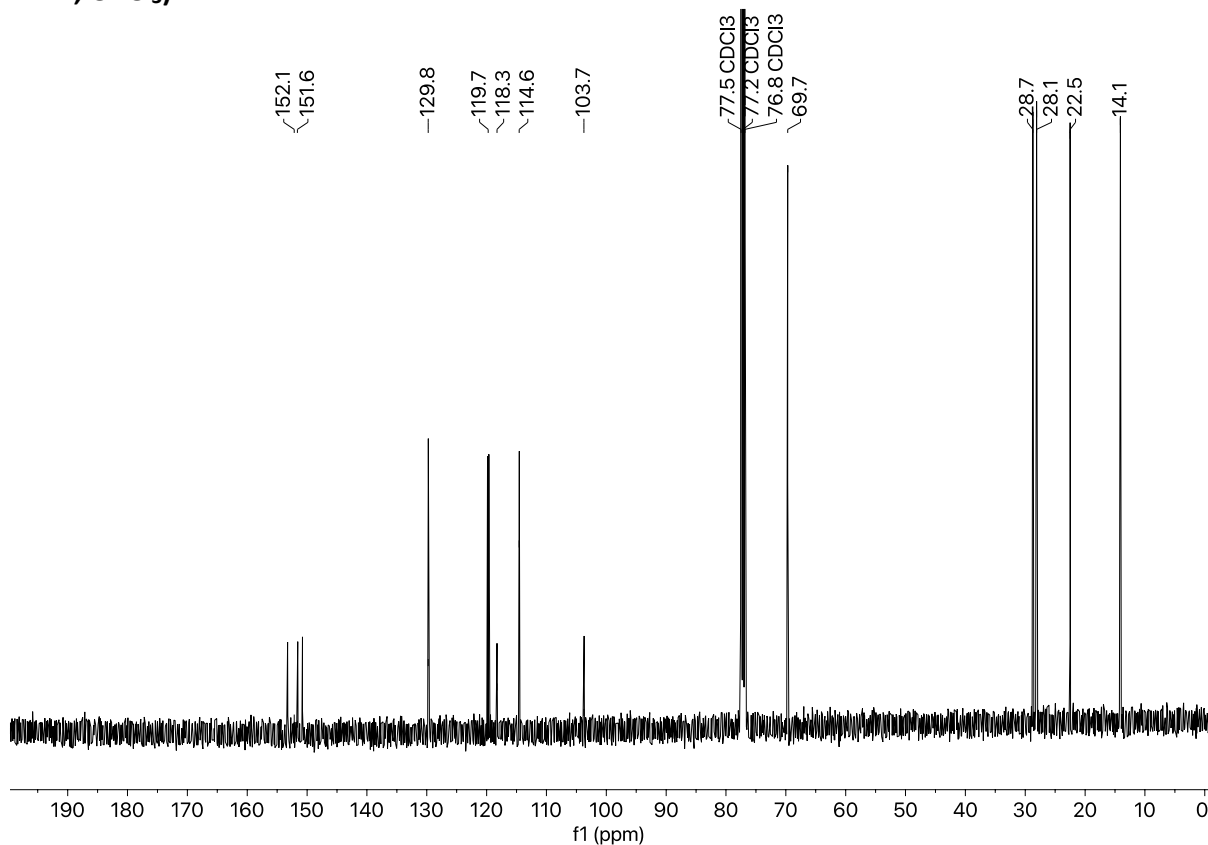
3-Fluoro-4-(pentyloxy)benzonitrile (3v)



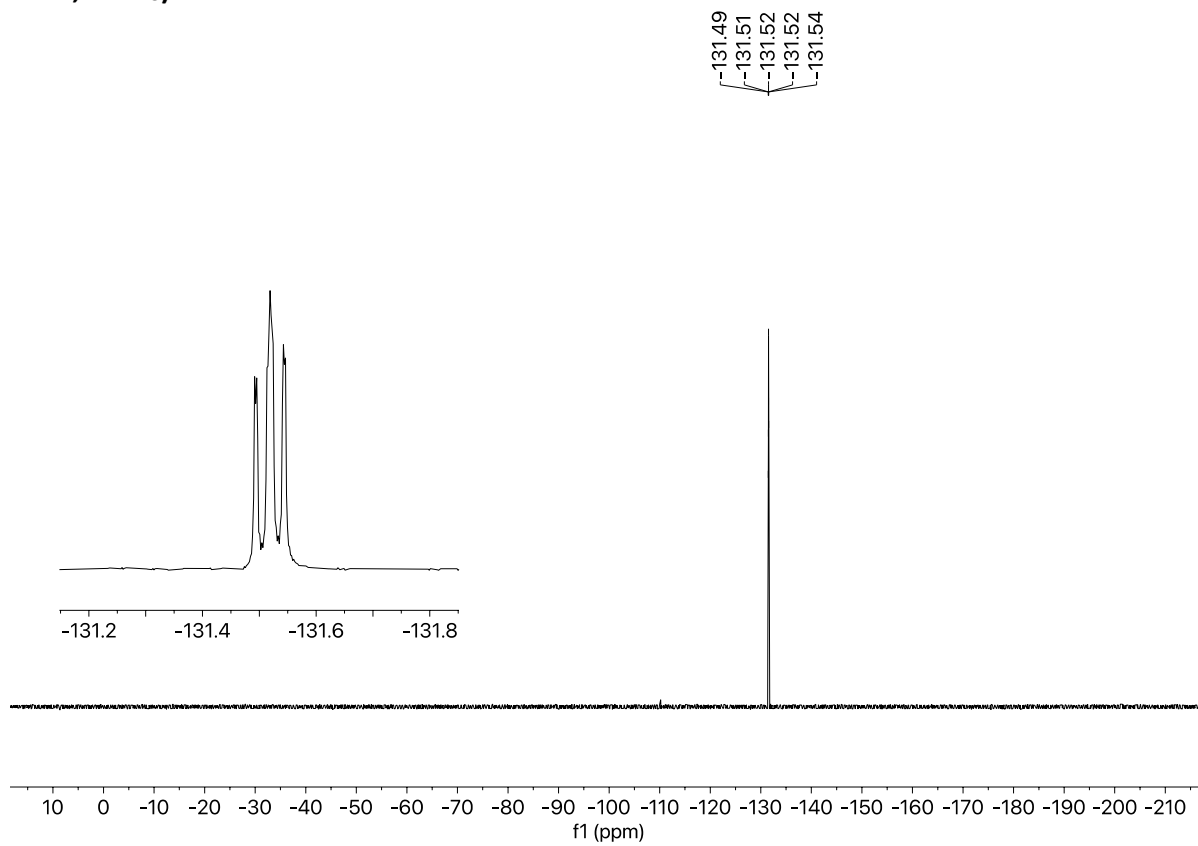
δ_H (400 MHz, $CDCl_3$)



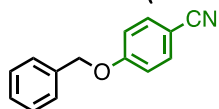
δ_C (101 MHz, $CDCl_3$)



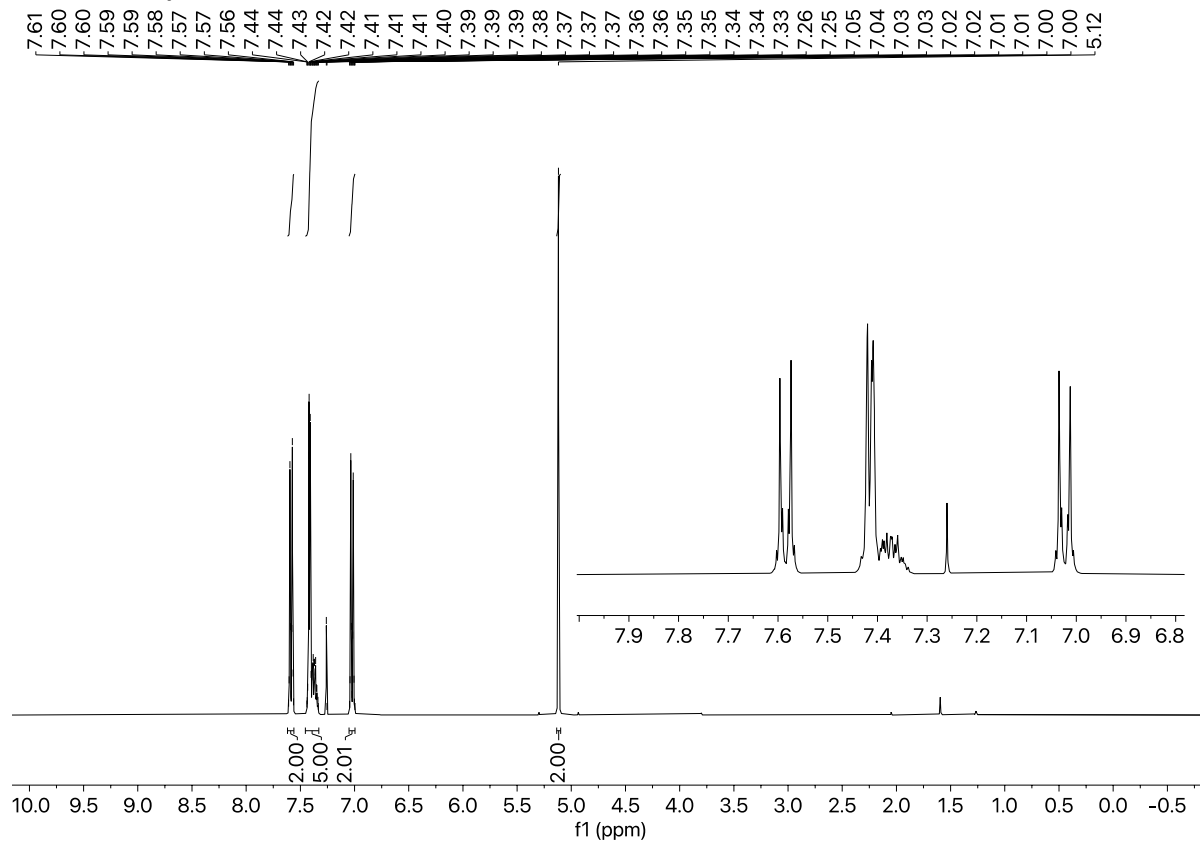
δ_F (376 MHz, $CDCl_3$)



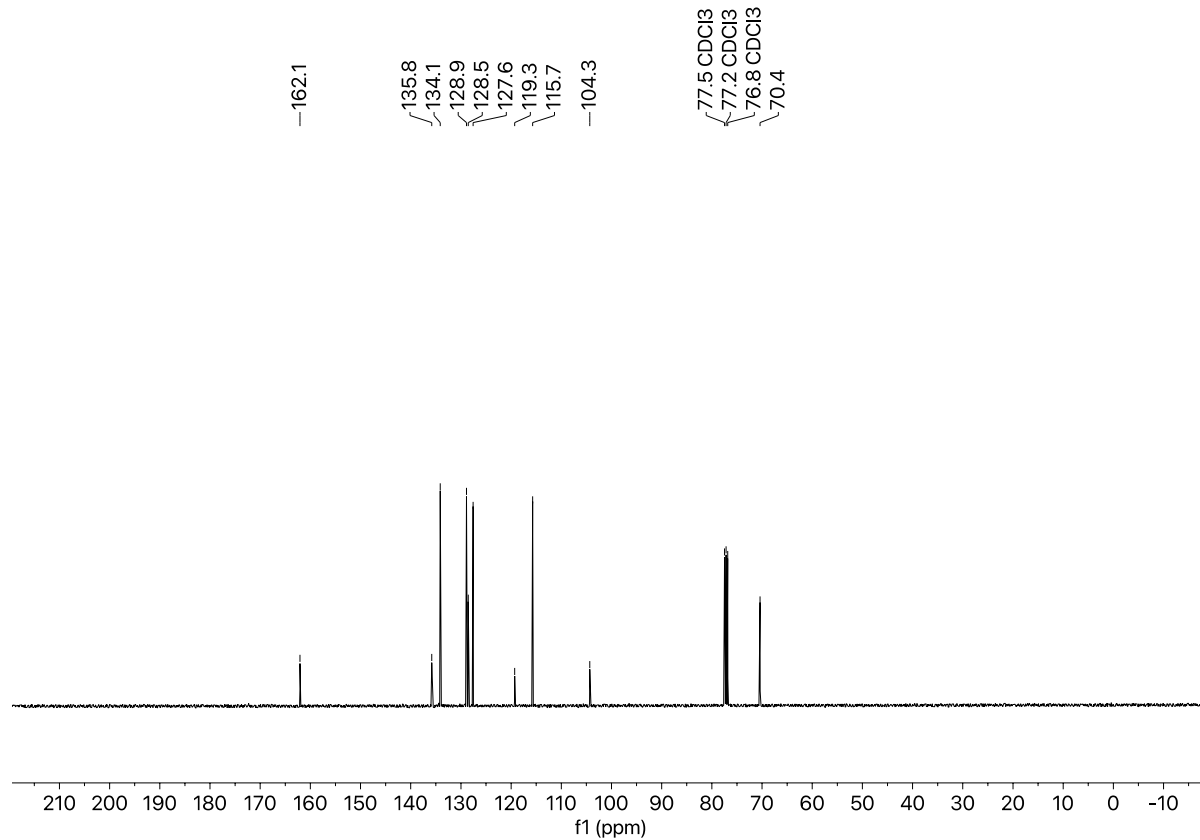
4-(Benzyloxy)benzonitrile (3w)



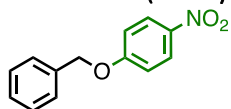
δ_H (400 MHz, $CDCl_3$)



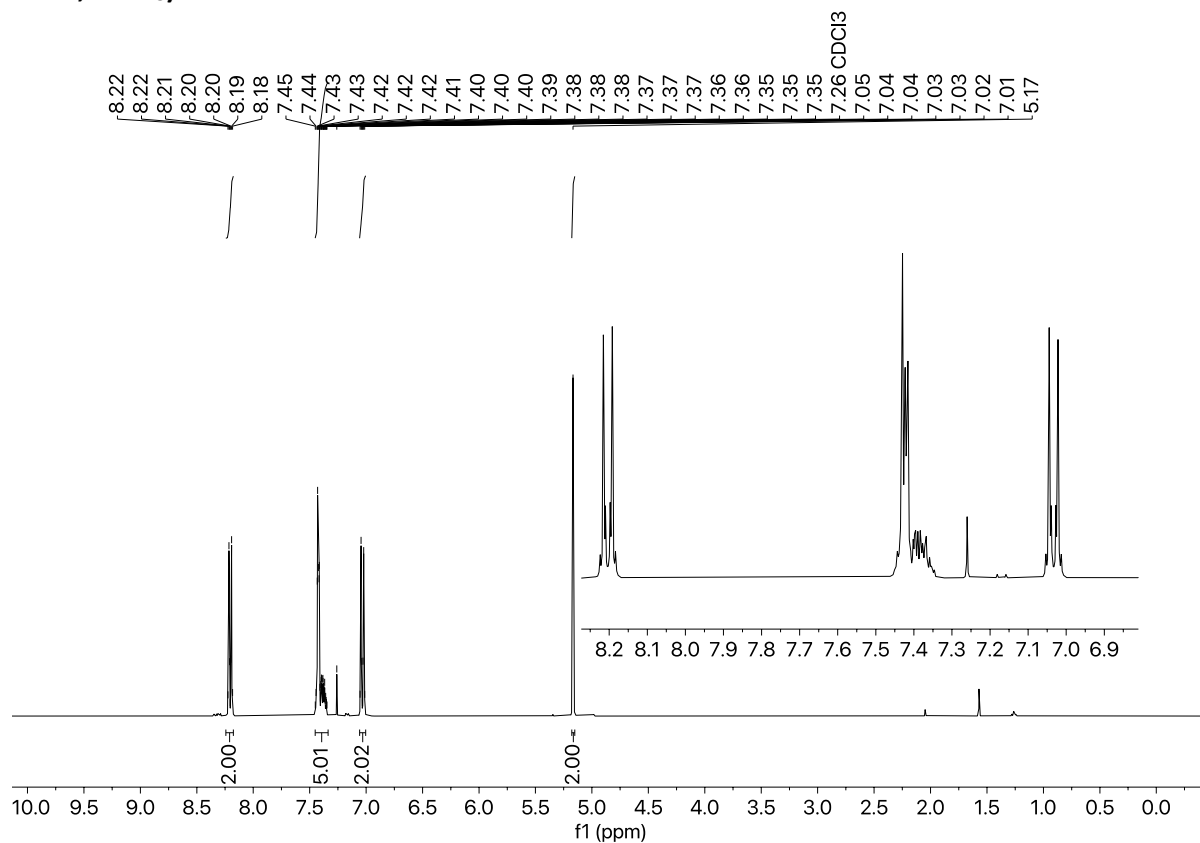
δ_C (101 MHz, $CDCl_3$)



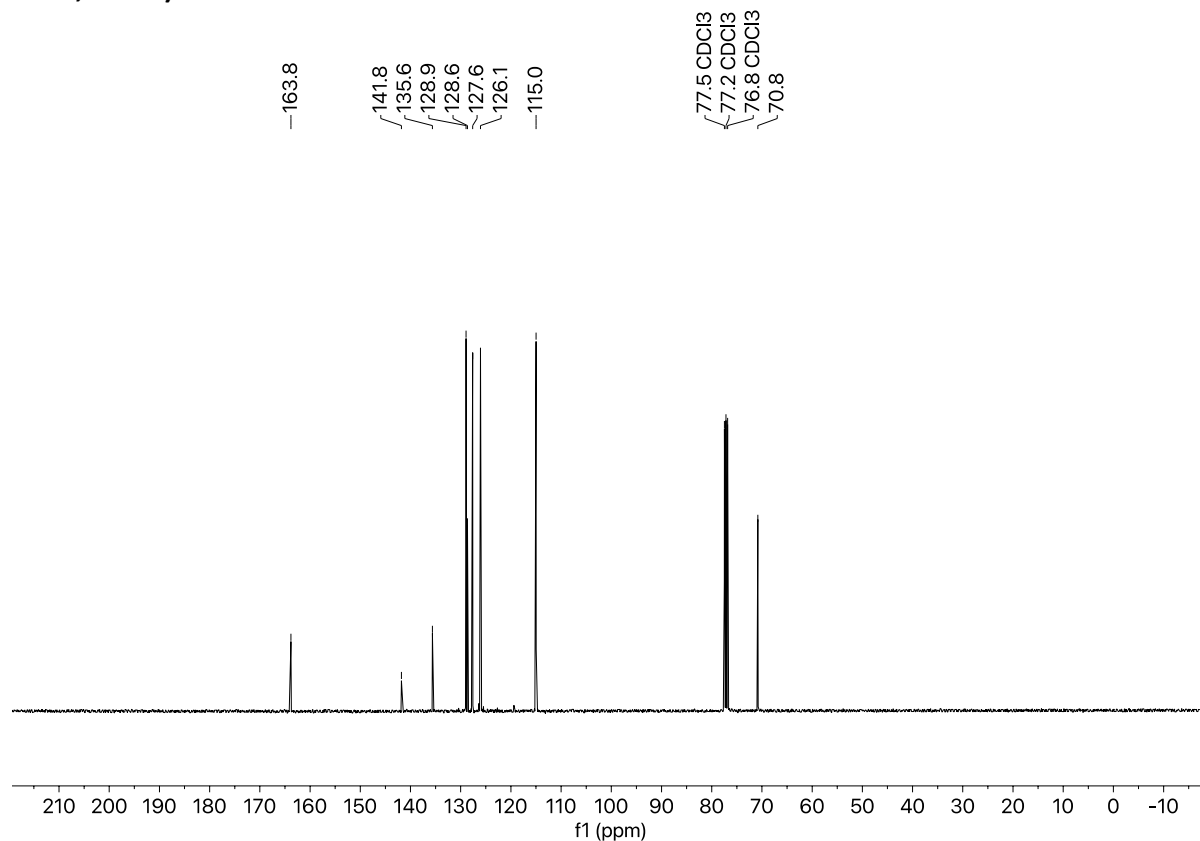
1-(Benzyloxy)-4-nitrobenzene (3x)



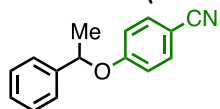
δ_H (400 MHz, $CDCl_3$)



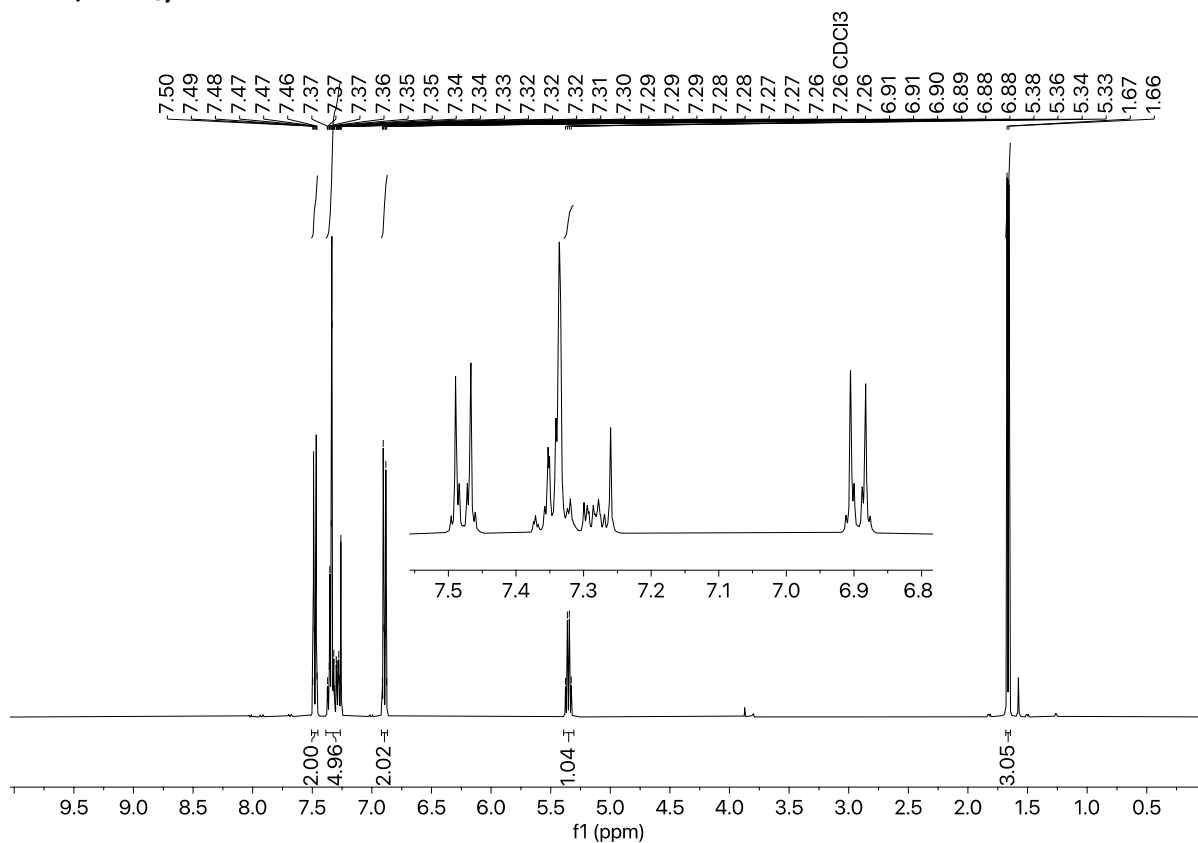
δ_C (101 MHz, $CDCl_3$)



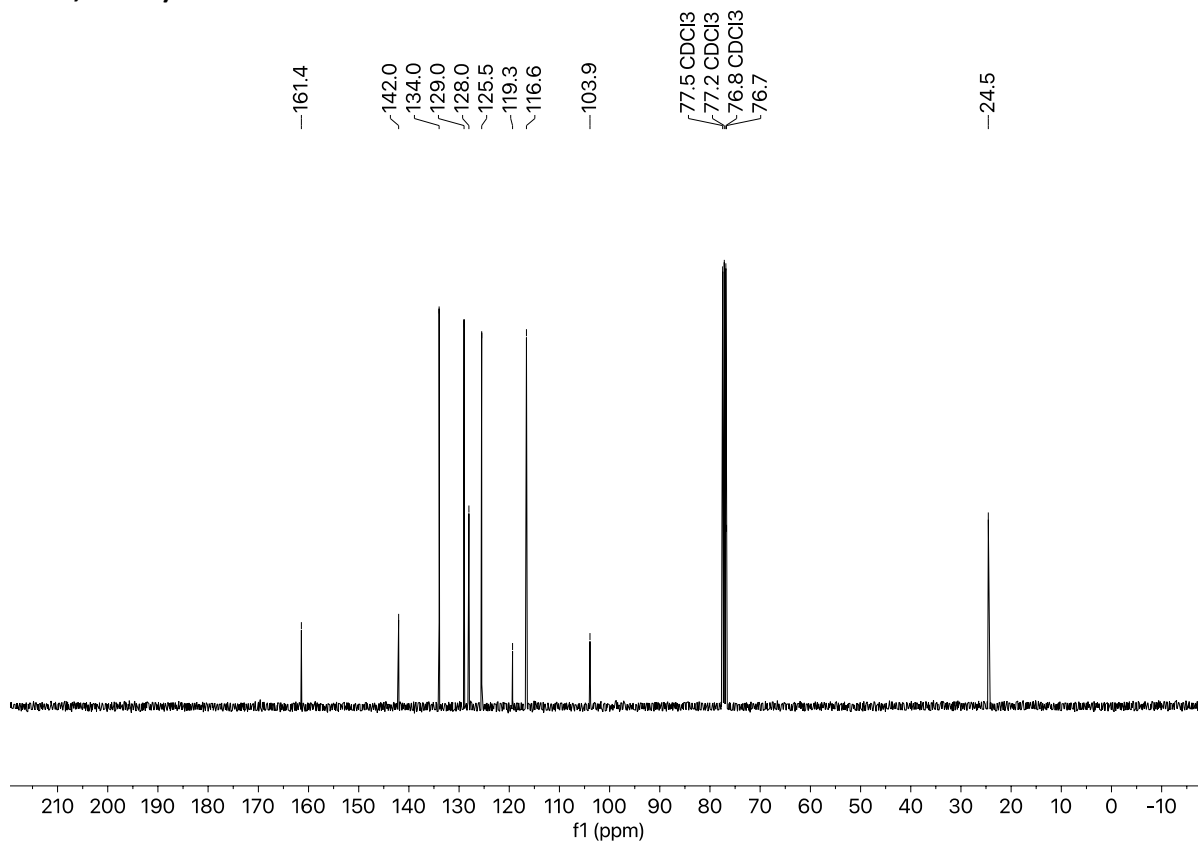
4-(1-Phenylethoxy)benzonitrile (3y)



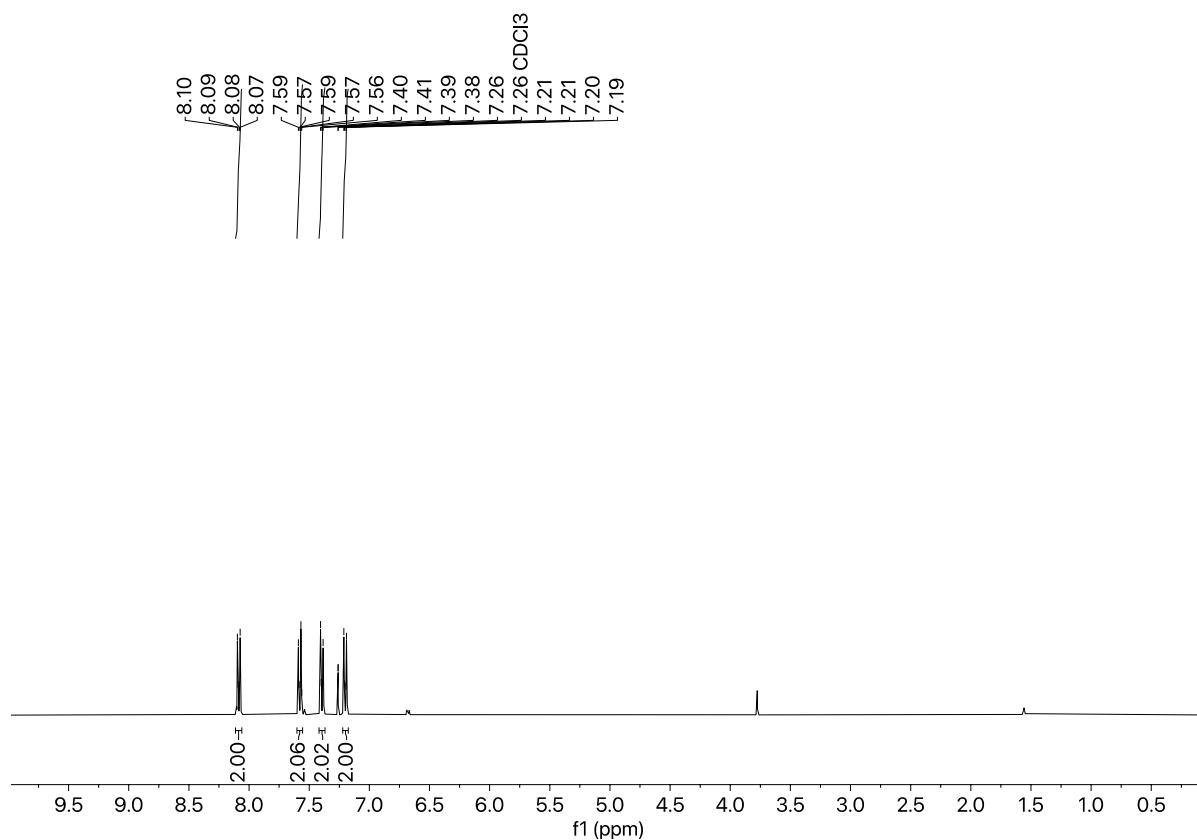
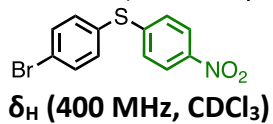
δ_H (400 MHz, $CDCl_3$)



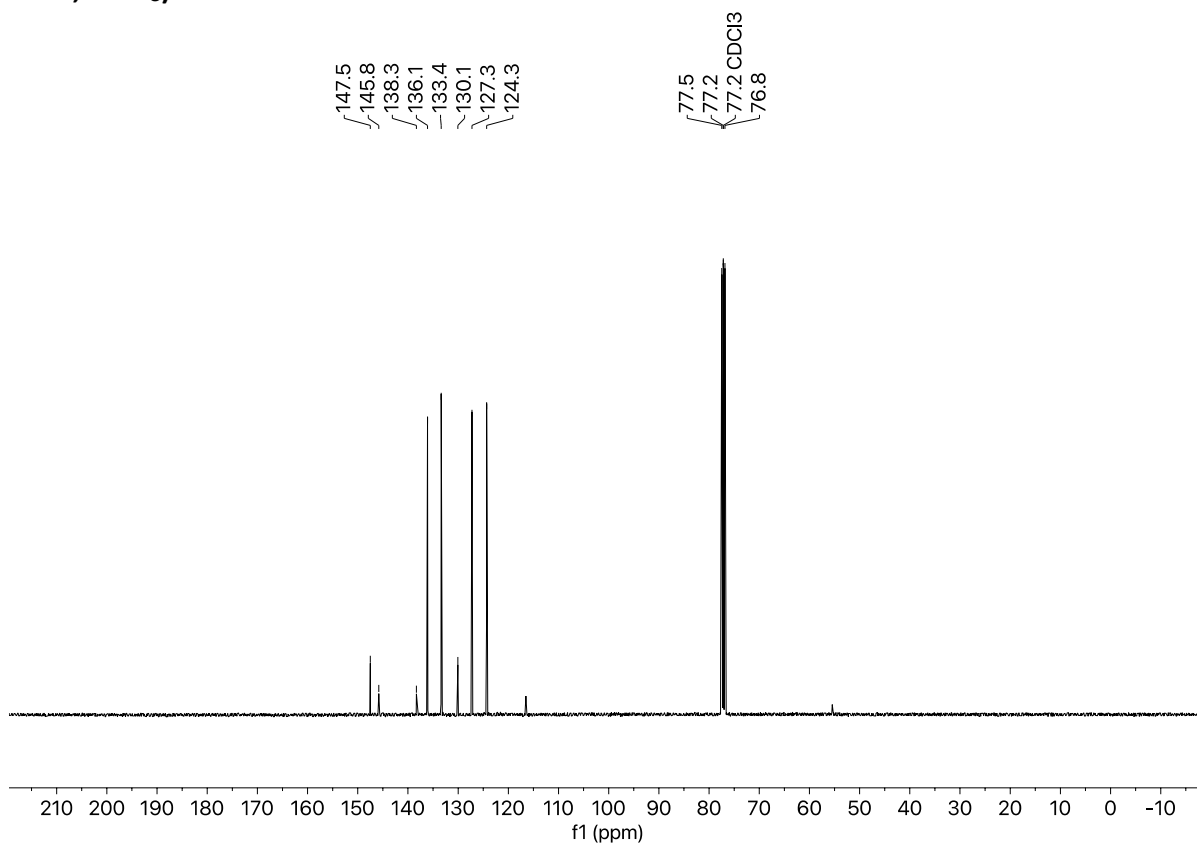
δ_C (101 MHz, $CDCl_3$)



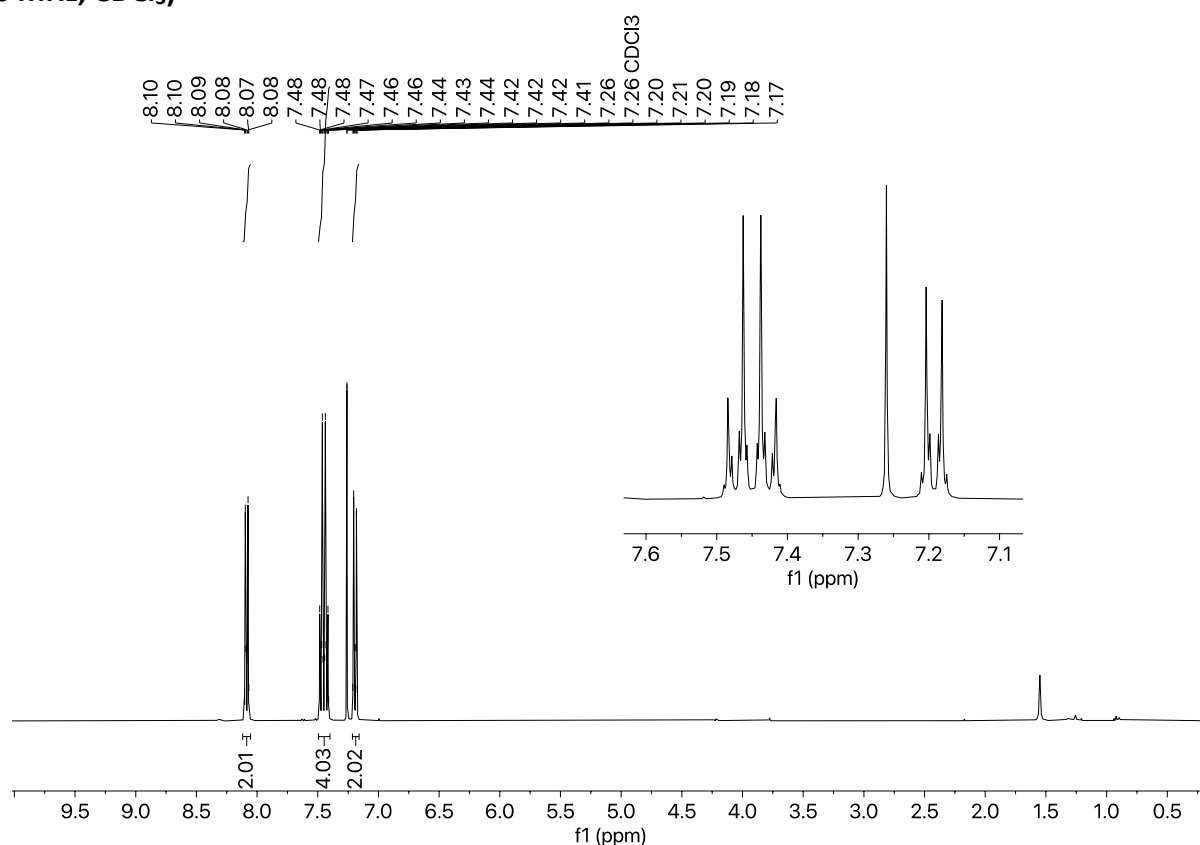
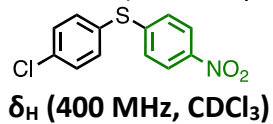
(4-Bromophenyl)(4-nitrophenyl)sulfane (4a)



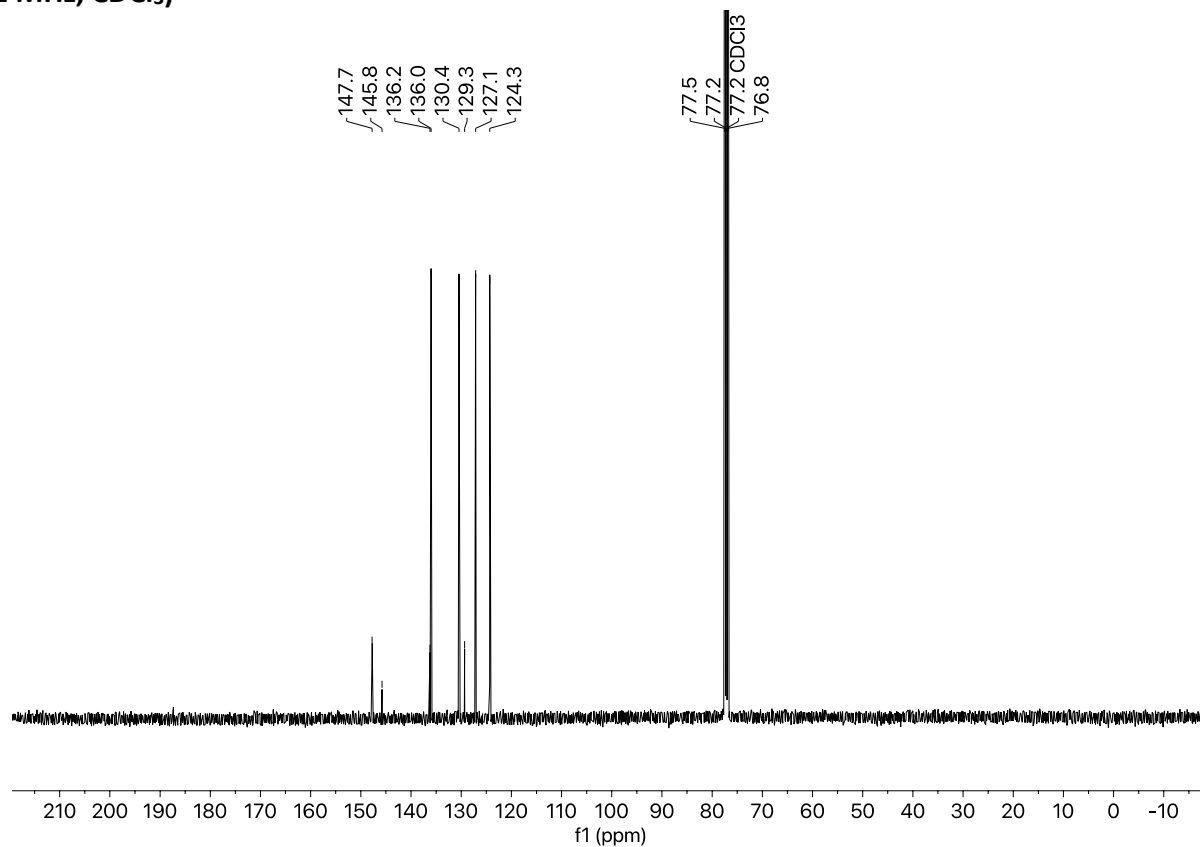
δ_C (101 MHz, $CDCl_3$)



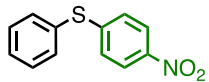
(4-Chlorophenyl)(4-nitrophenyl)sulfane (4b)



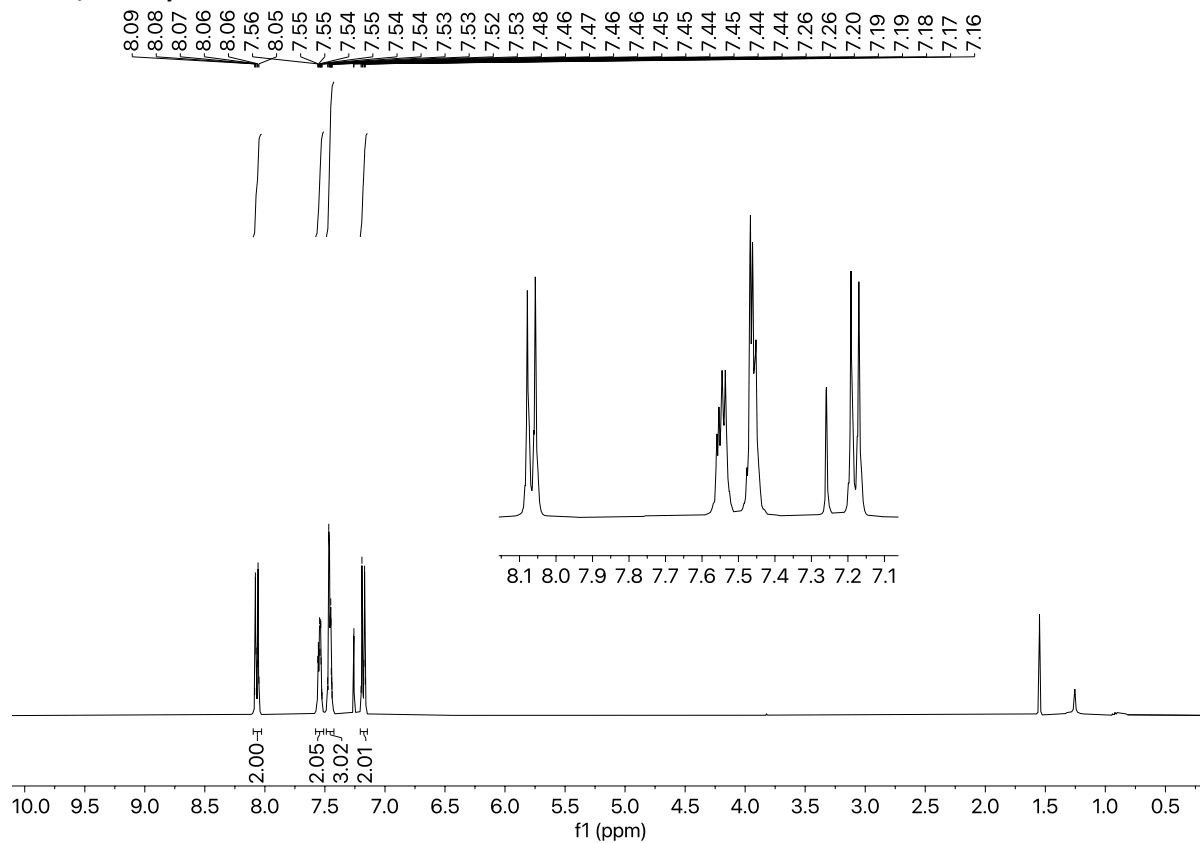
δ_{C} (101 MHz, CDCl_3)



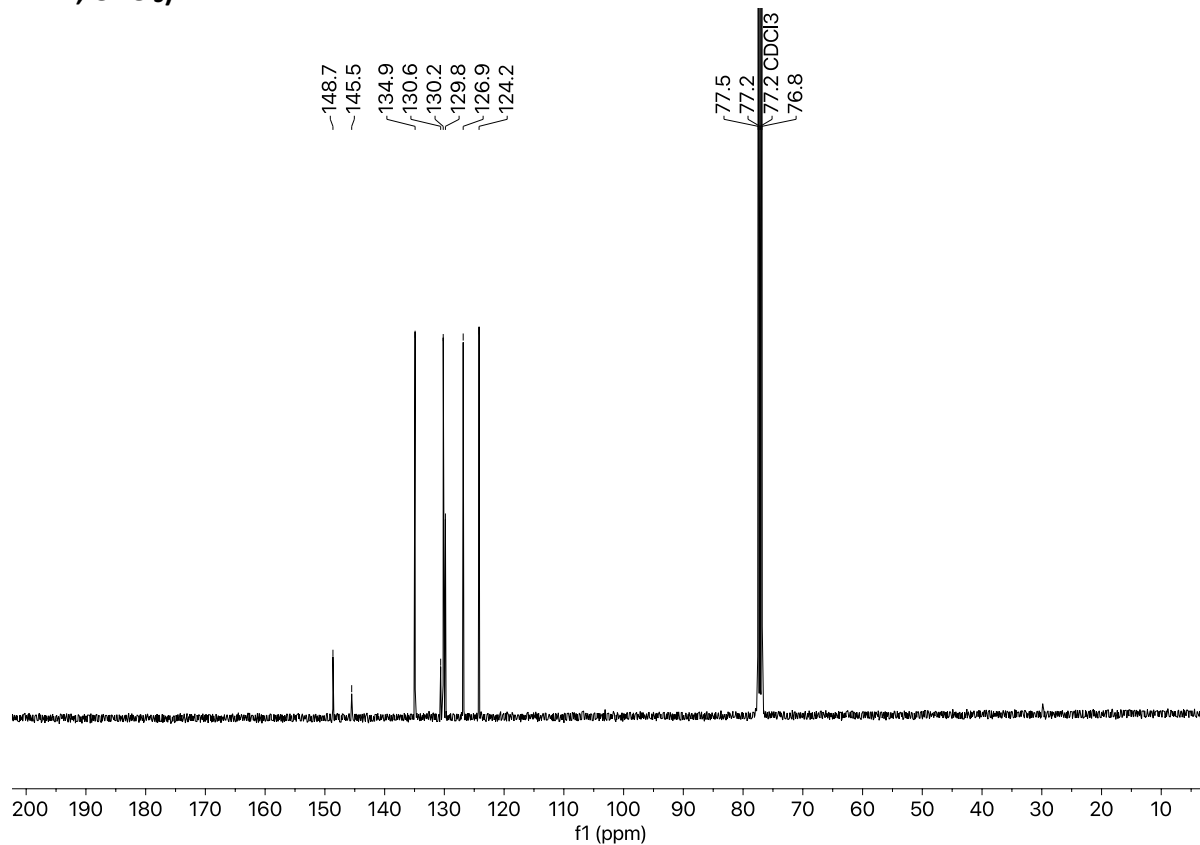
(4-Nitrophenyl)(phenyl)sulfane (4c)



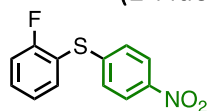
δ_H (400 MHz, $CDCl_3$)



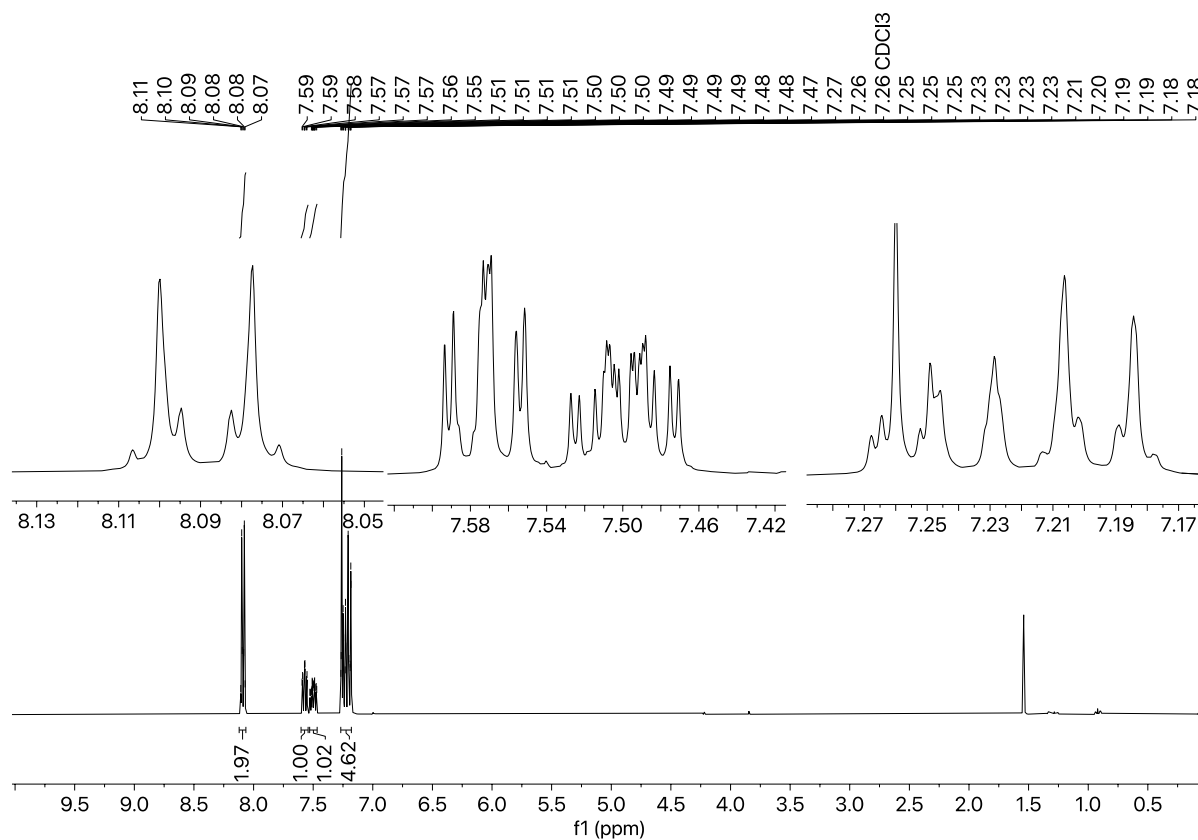
δ_C (101 MHz, $CDCl_3$)



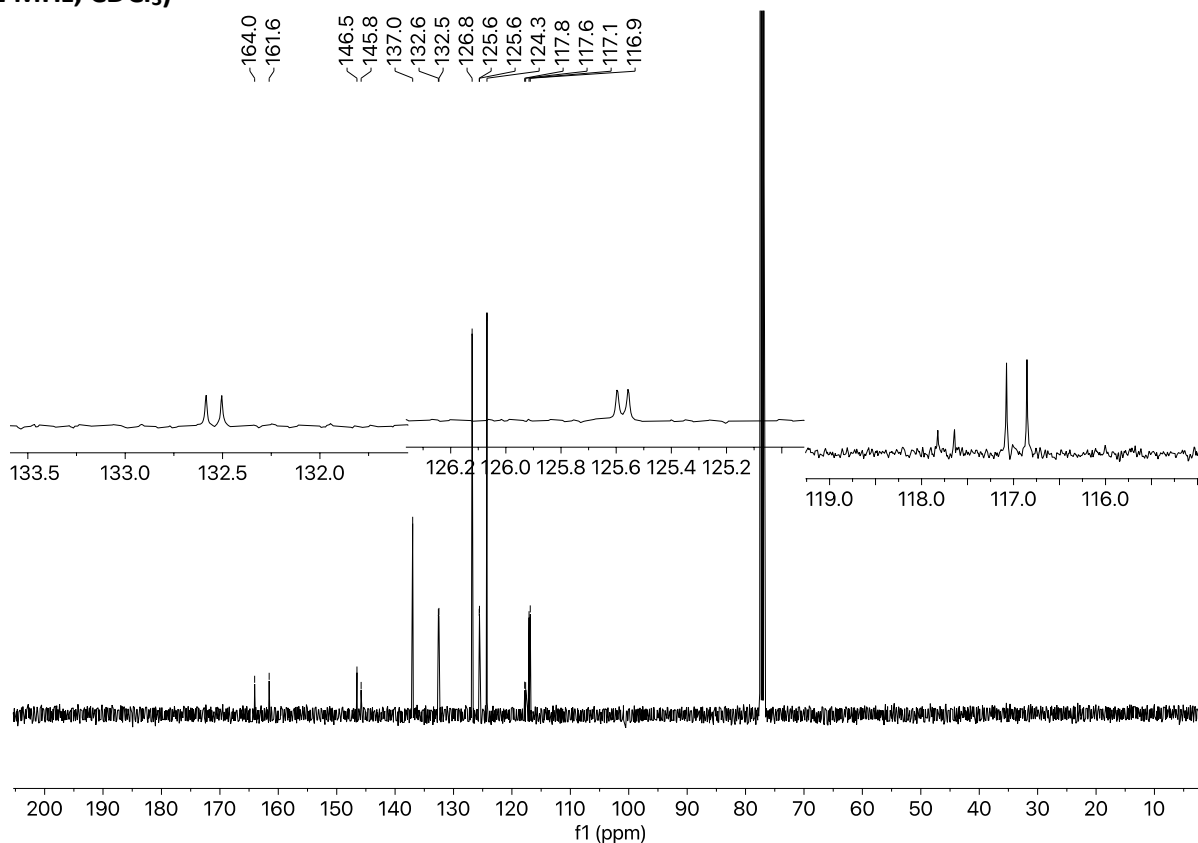
(2-Fluorophenyl)(4-nitrophenyl)sulfane (4d)



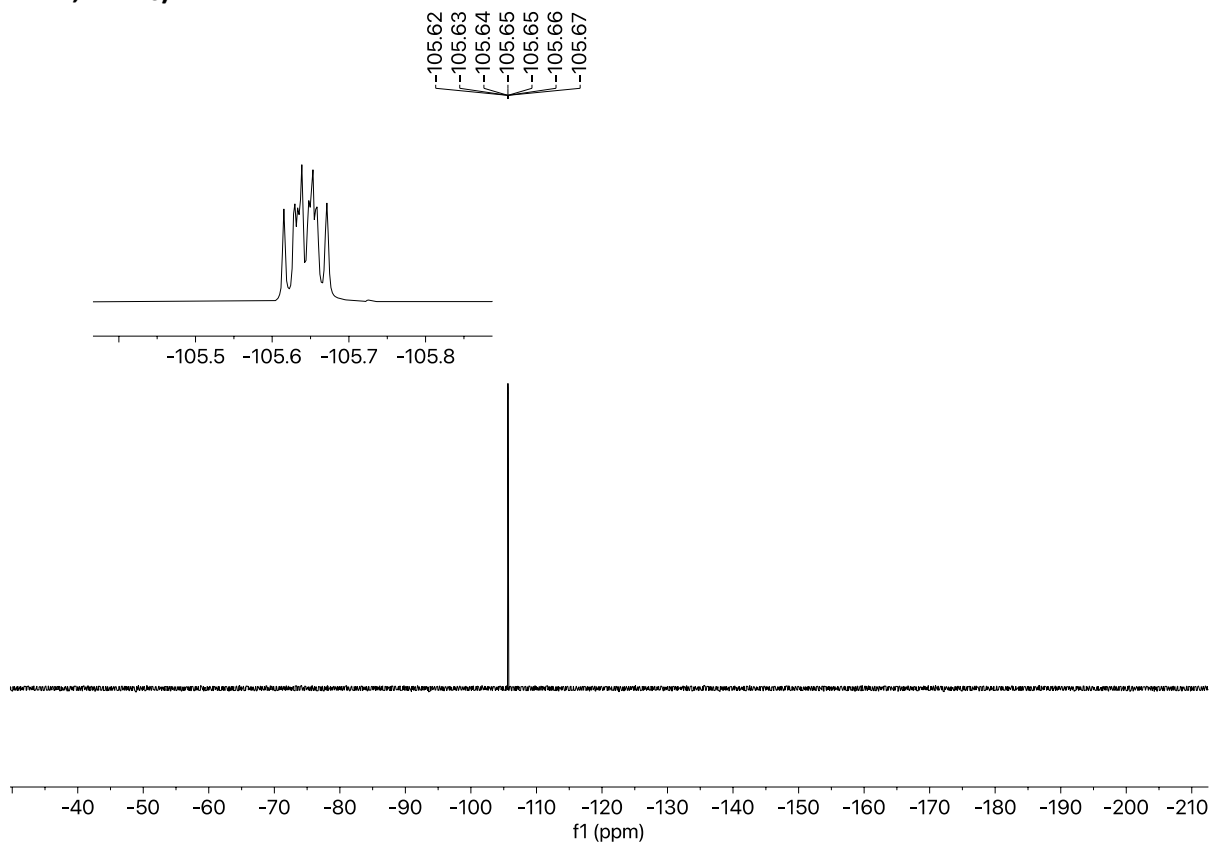
δ_H (400 MHz, $CDCl_3$)



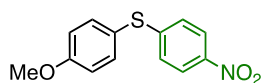
δ_C (101 MHz, $CDCl_3$)



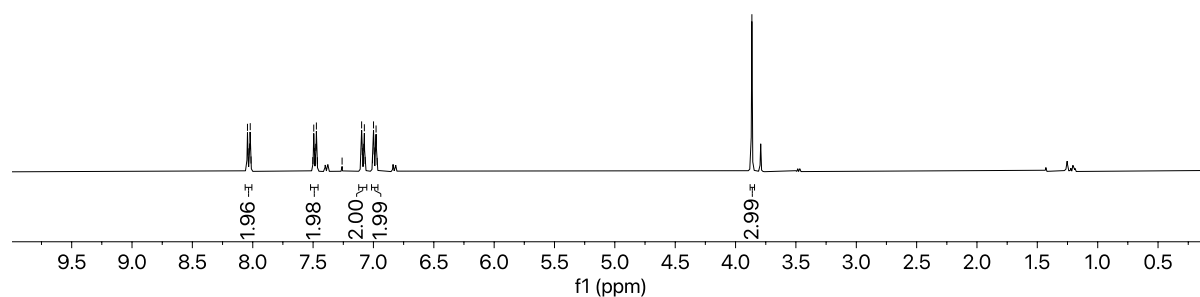
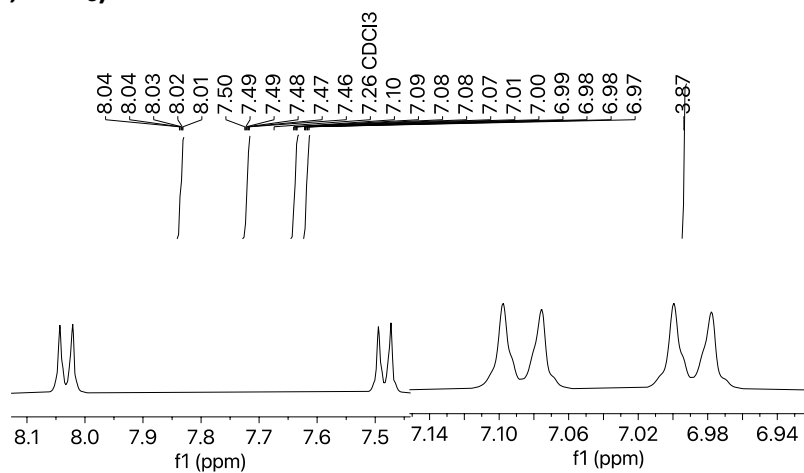
δ_F (376 MHz, $CDCl_3$)



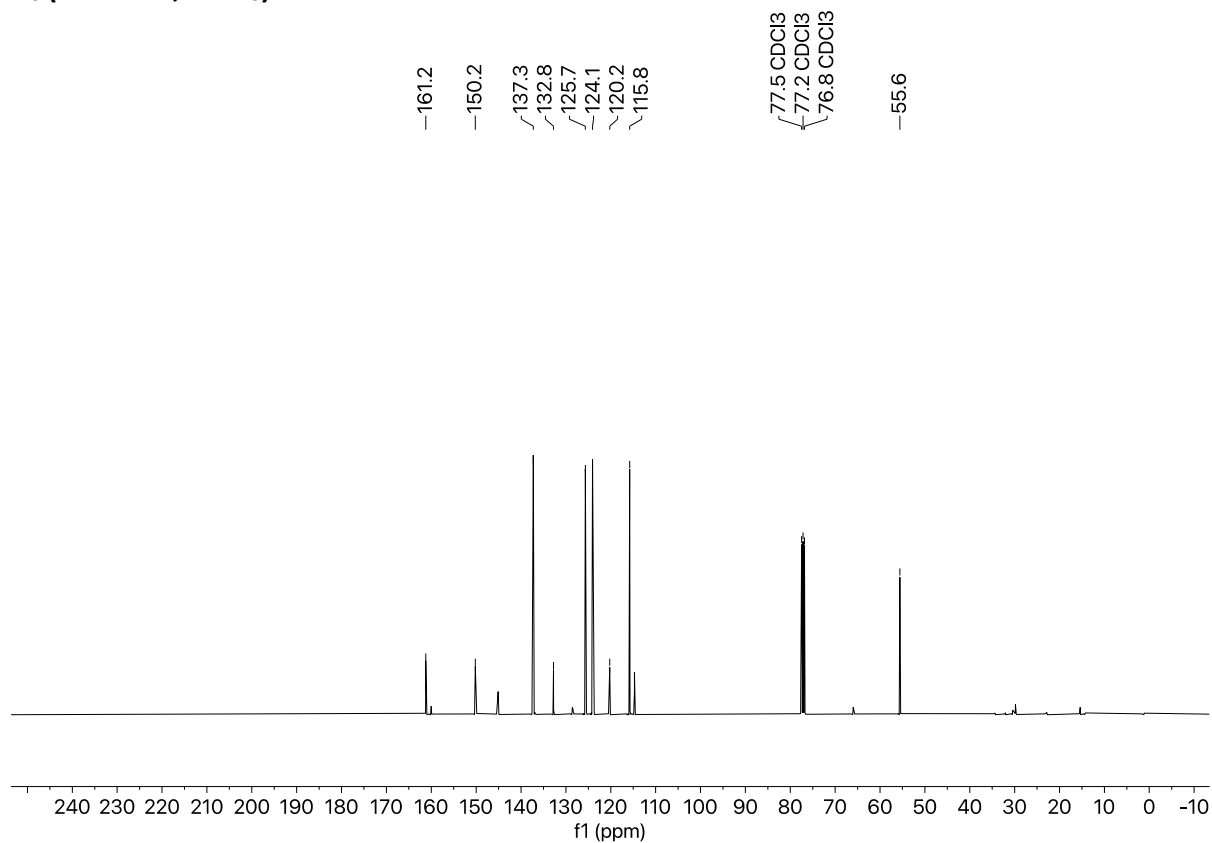
(4-Methoxyphenyl)(4-nitrophenyl)sulfane (4e)



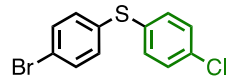
δ_H (400 MHz, $CDCl_3$)



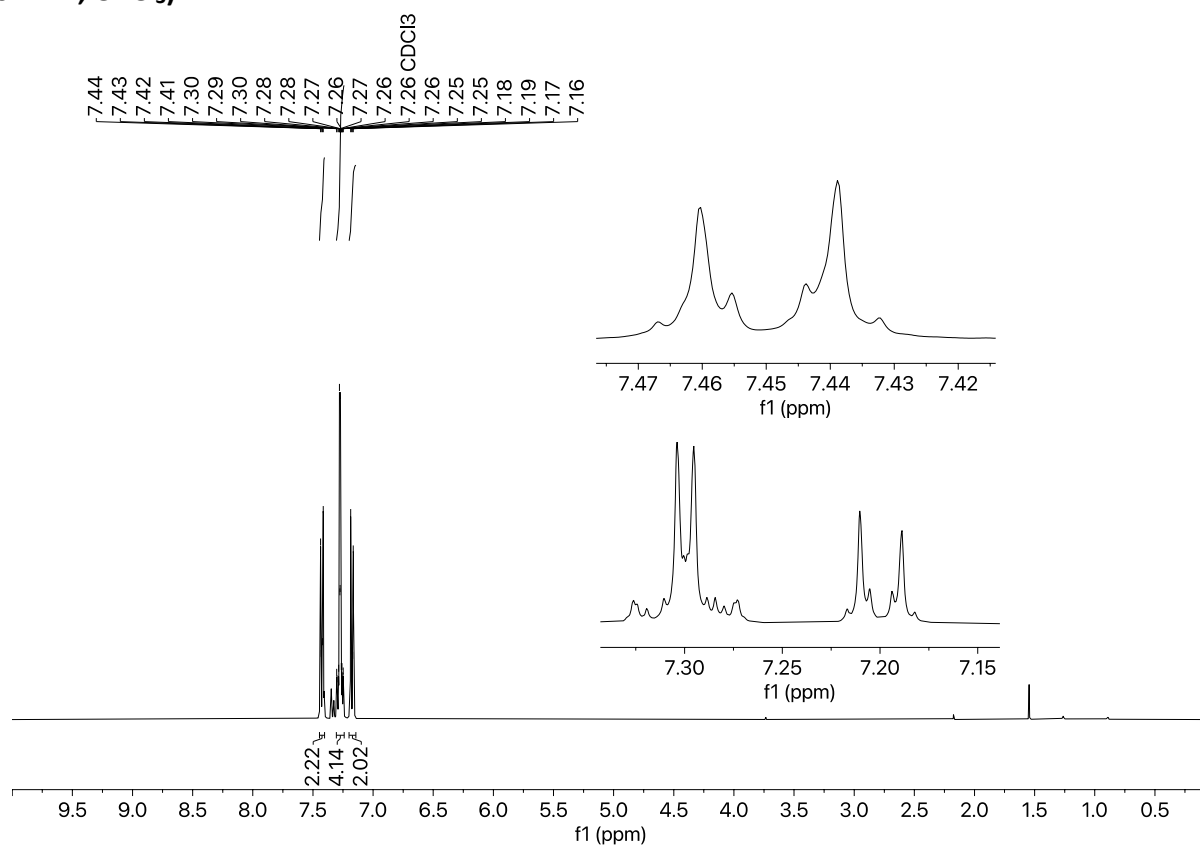
δ_C (101 MHz, $CDCl_3$)



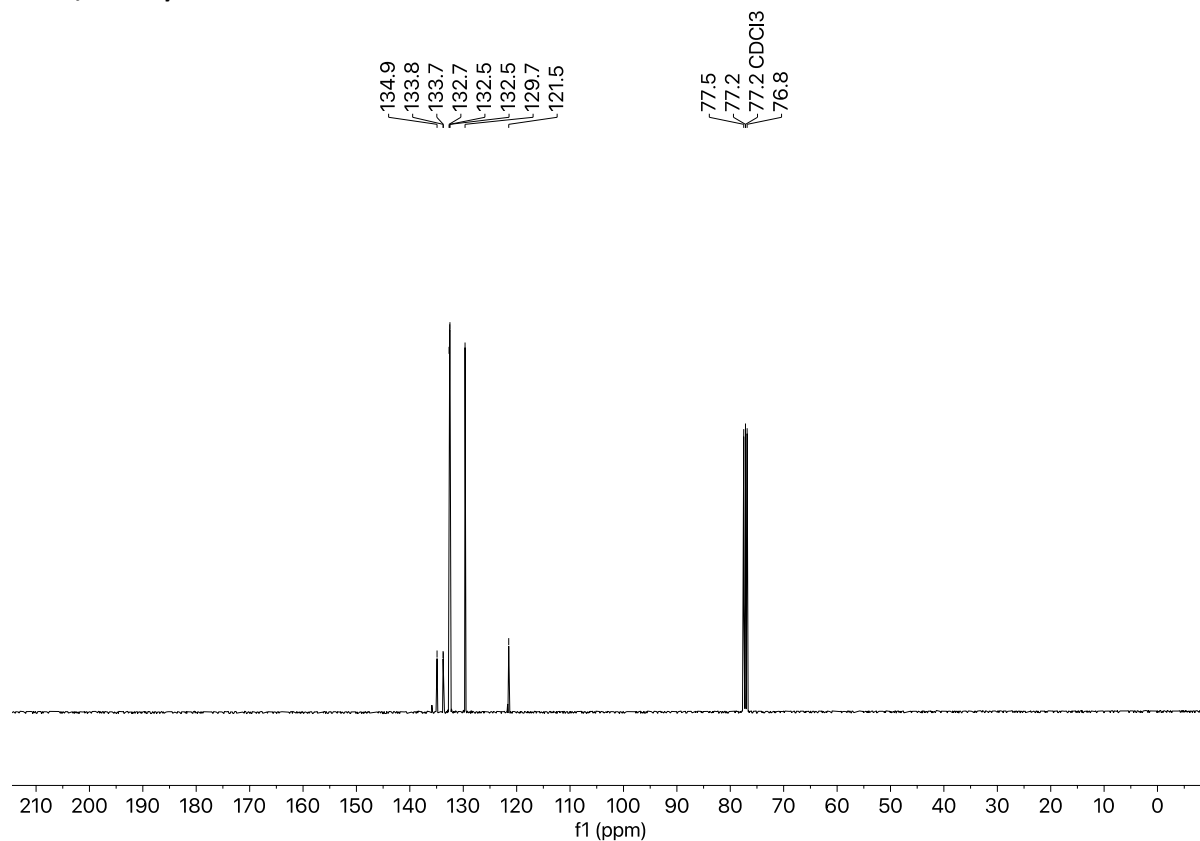
(4-Bromophenyl)(4-chlorophenyl)sulfane (4f)



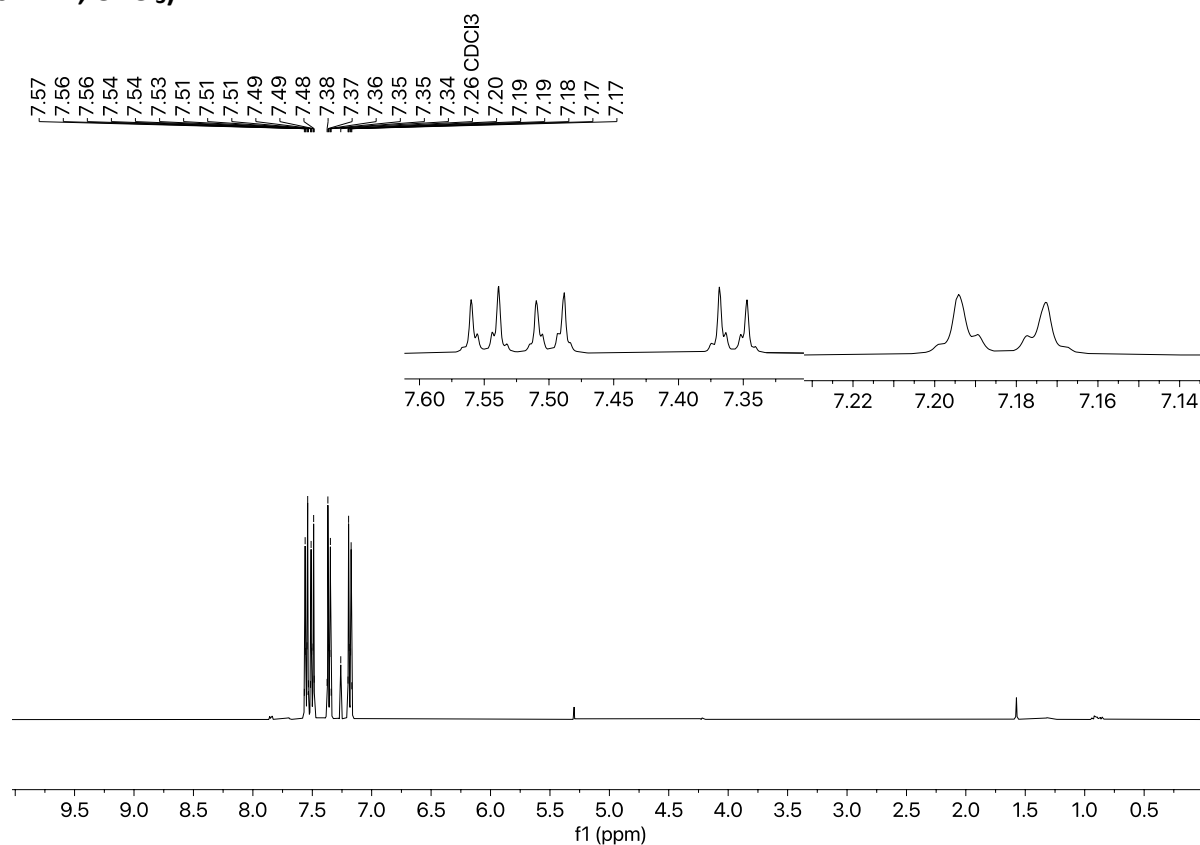
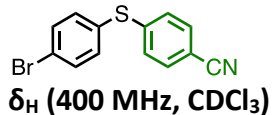
δ_H (400 MHz, $CDCl_3$)



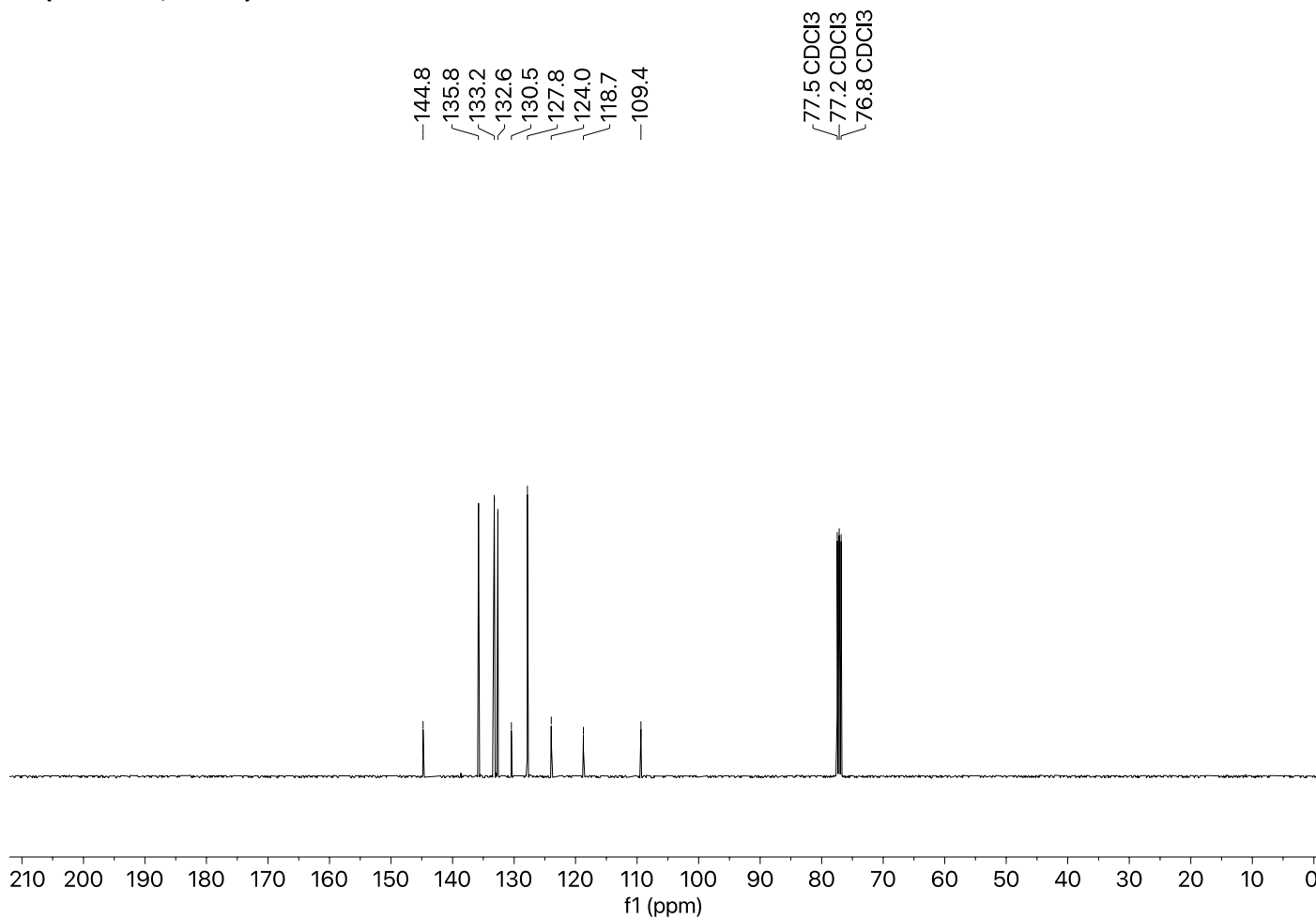
δ_C (101 MHz, $CDCl_3$)



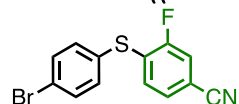
4-((4-Bromophenyl)thio)benzonitrile (4g)



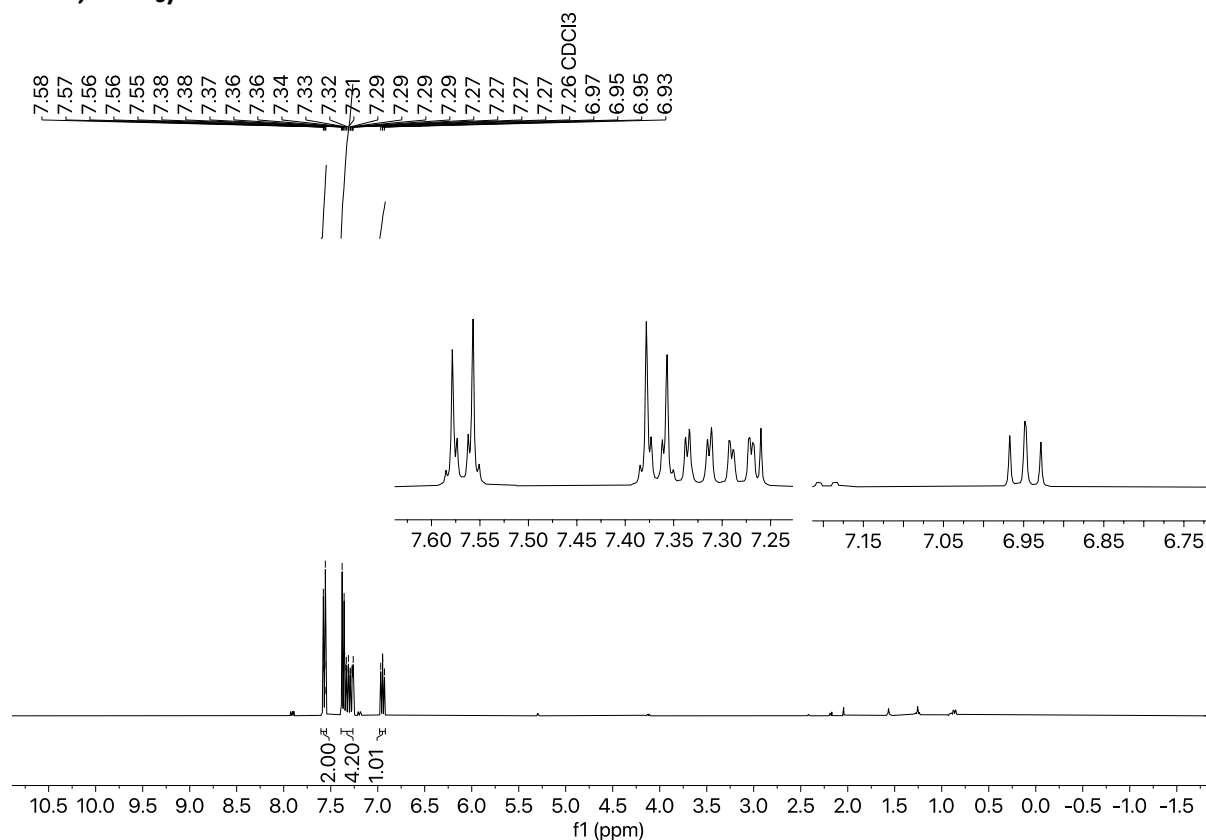
δ_C (101 MHz, $CDCl_3$)



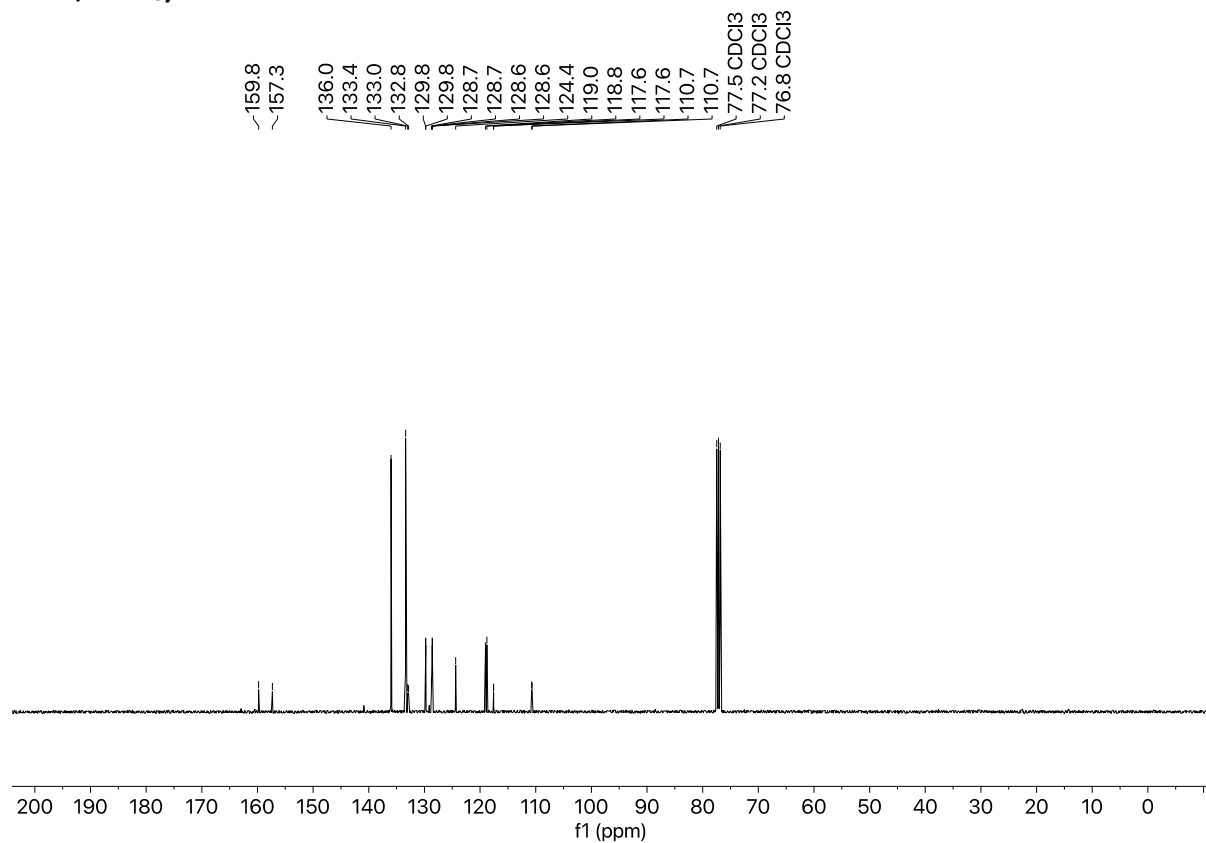
4-((4-Bromophenyl)thio)-3-fluorobenzonitrile (4h)



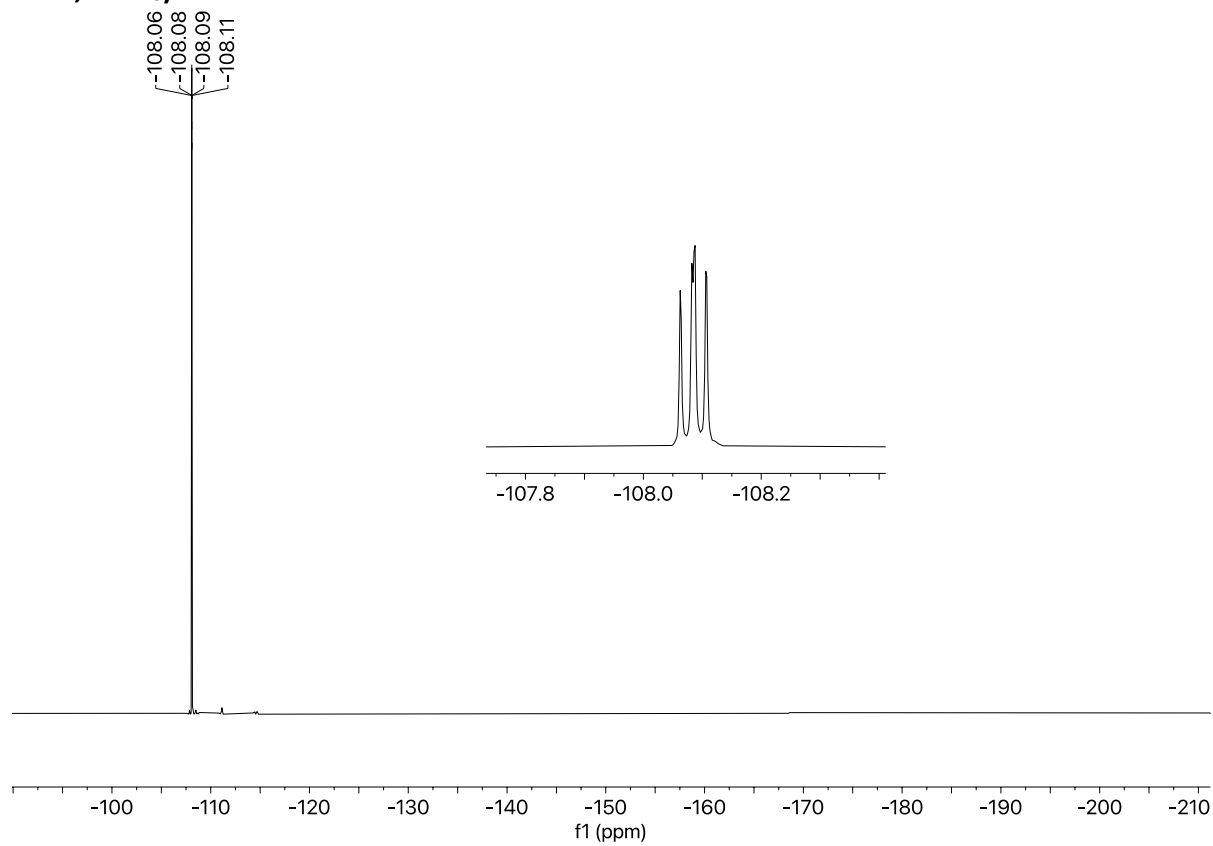
δ_H (400 MHz, $CDCl_3$)



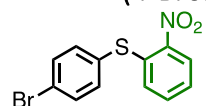
δ_C (101 MHz, $CDCl_3$)



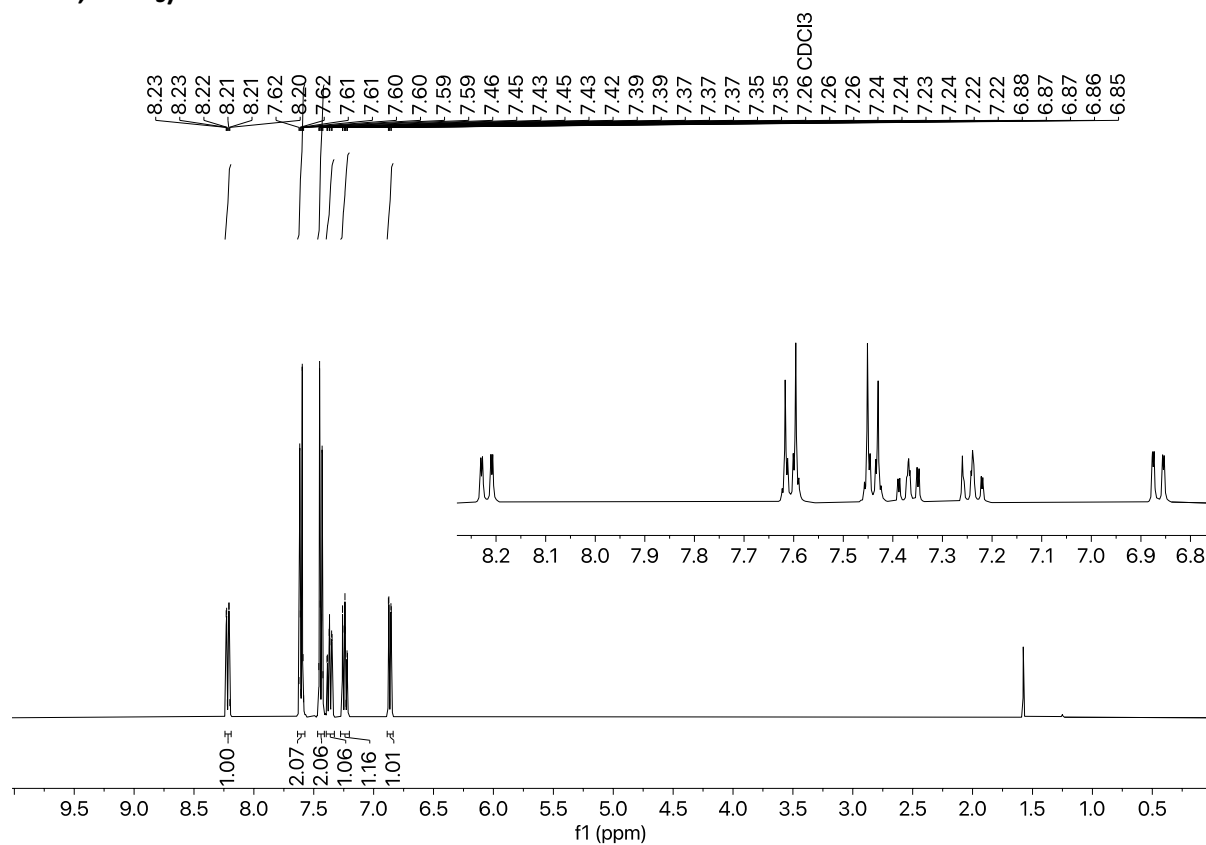
δ_F (376 MHz, $CDCl_3$)



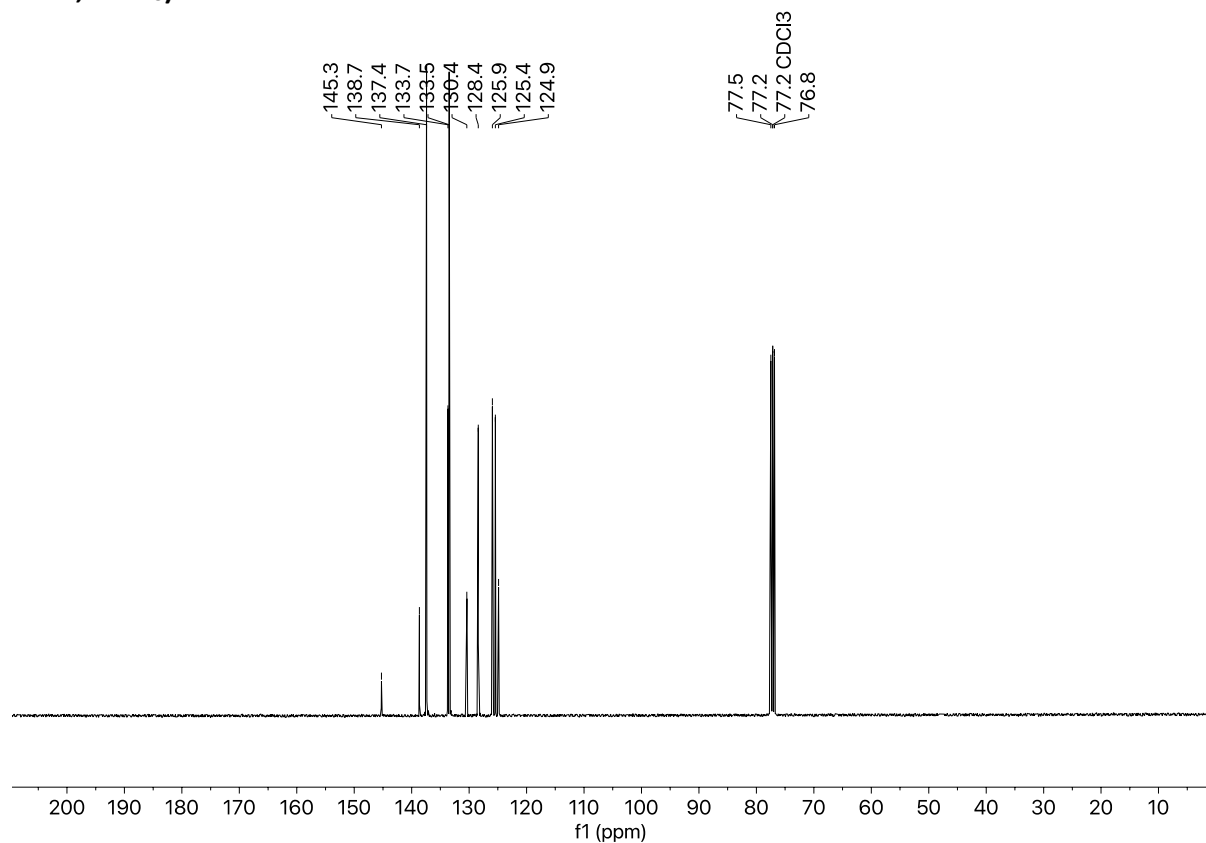
(4-Bromophenyl)(2-nitrophenyl)sulfane (4i)



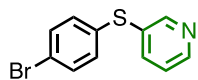
δ_H (400 MHz, $CDCl_3$)



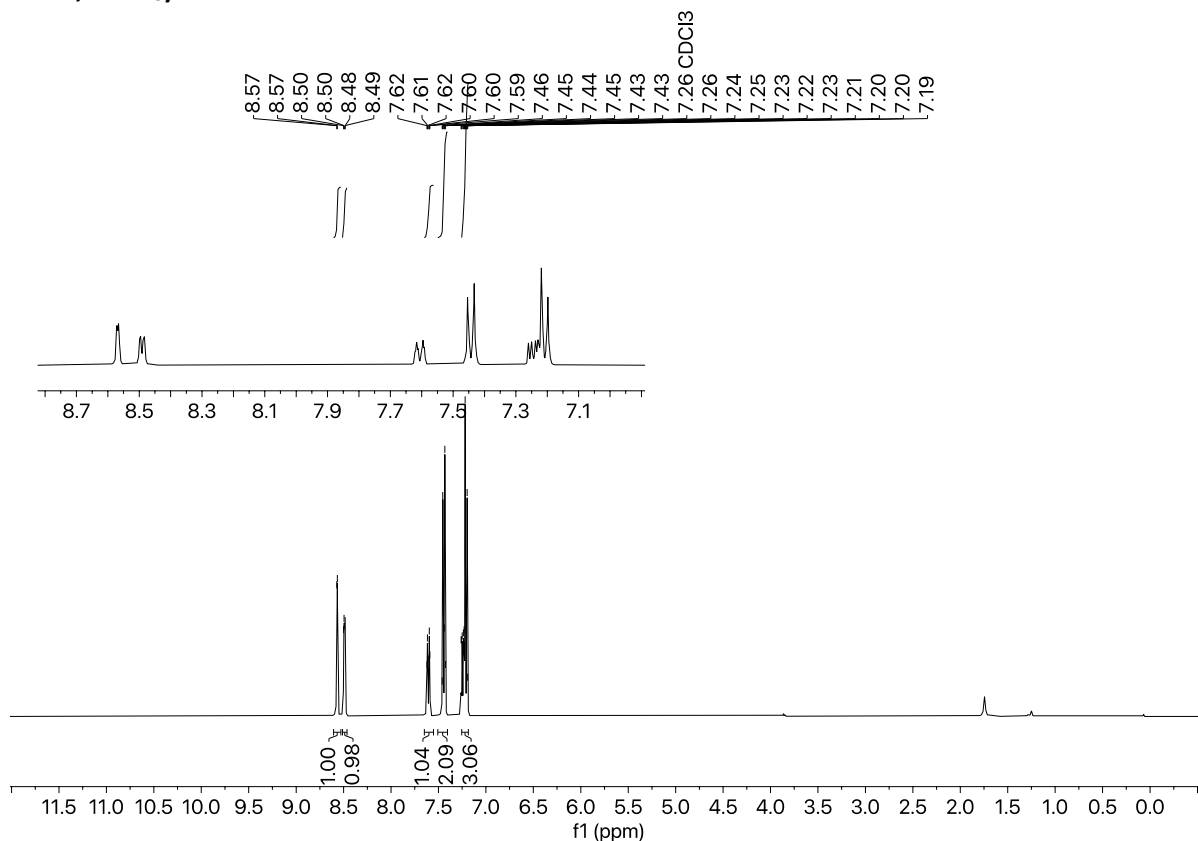
δ_C (101 MHz, $CDCl_3$)



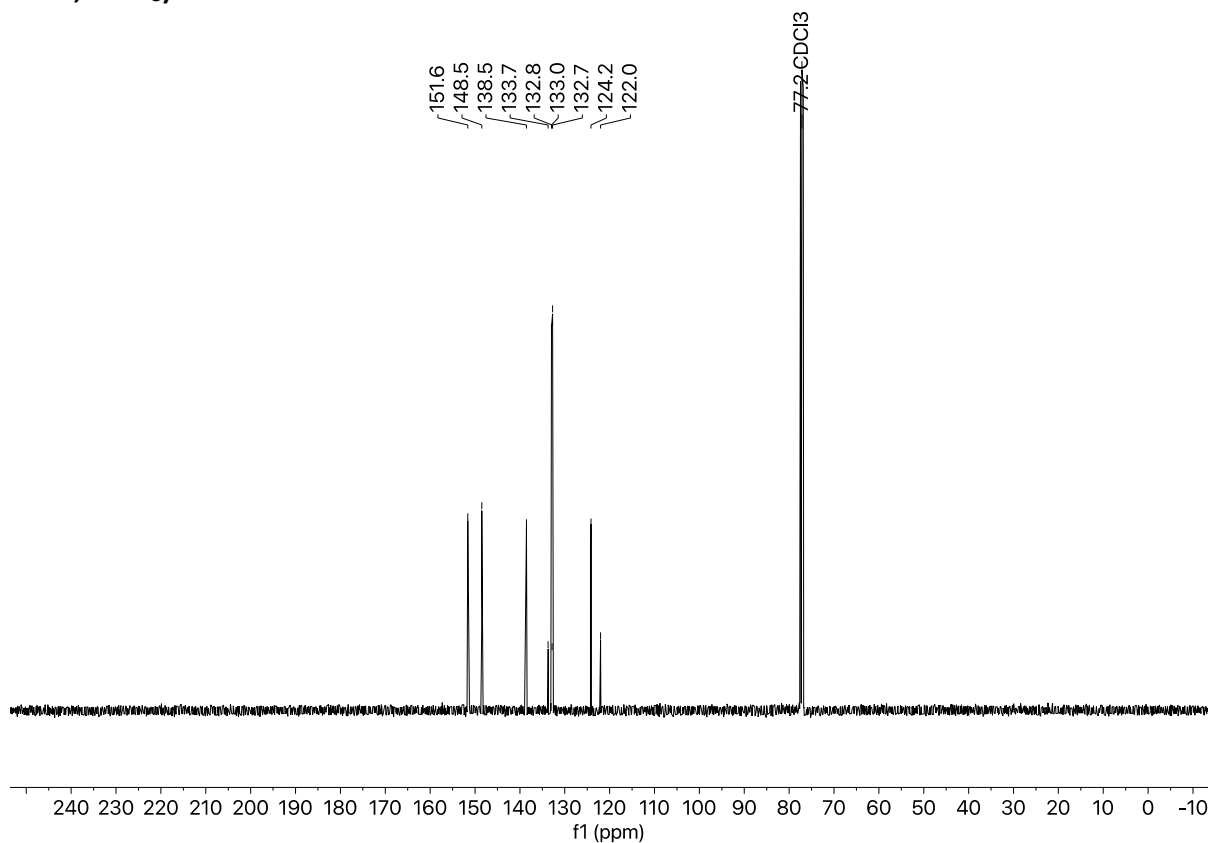
3-((4-Bromophenyl)thio)pyridine (4j)



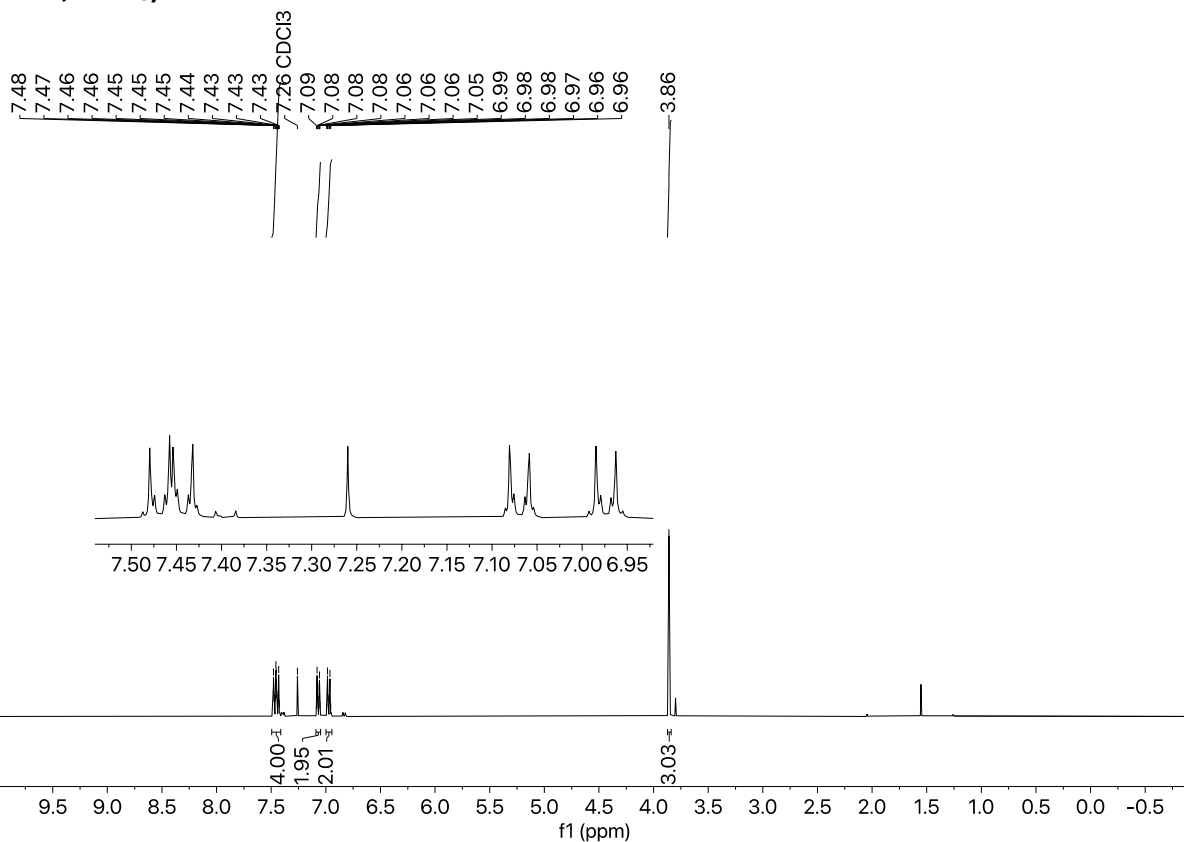
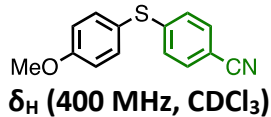
δ_H (400 MHz, $CDCl_3$)



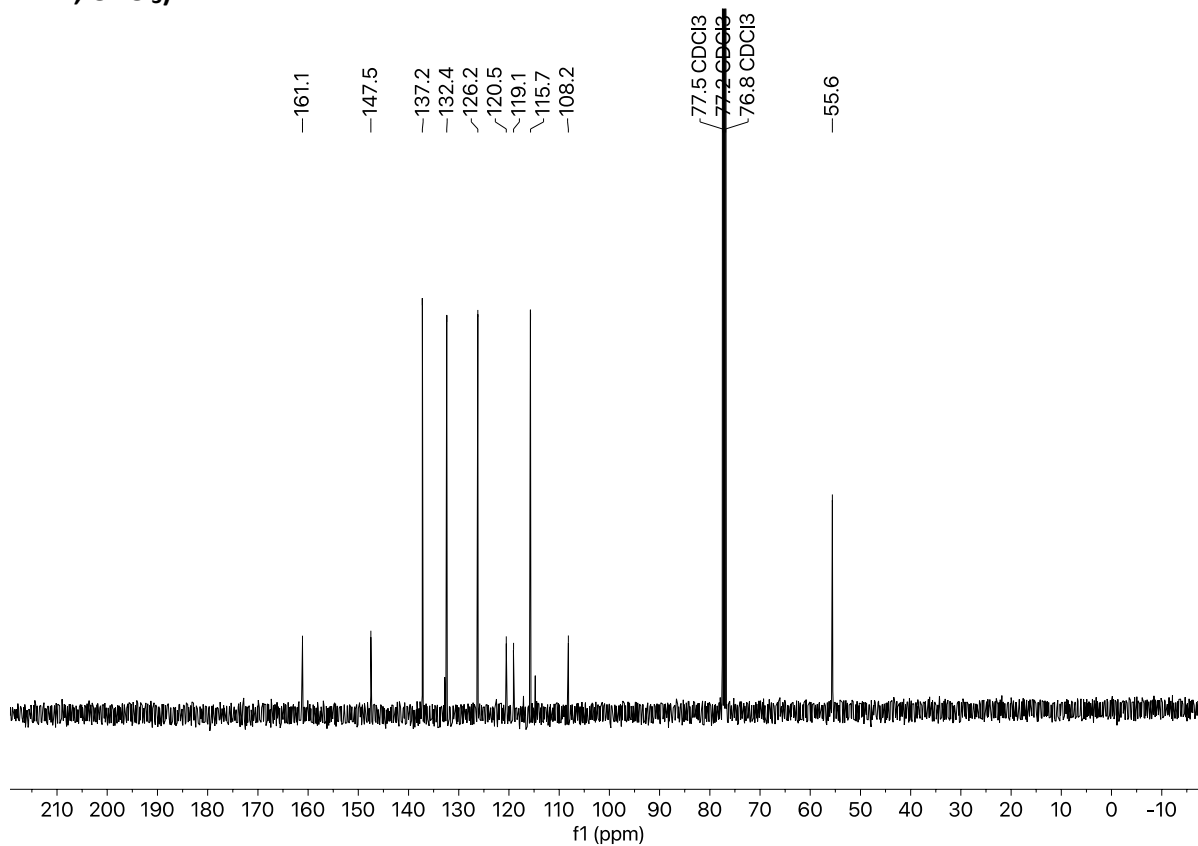
δ_C (101 MHz, $CDCl_3$)



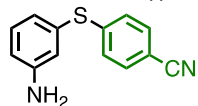
4-((4-Methoxyphenyl)thio)benzotrile (4k)



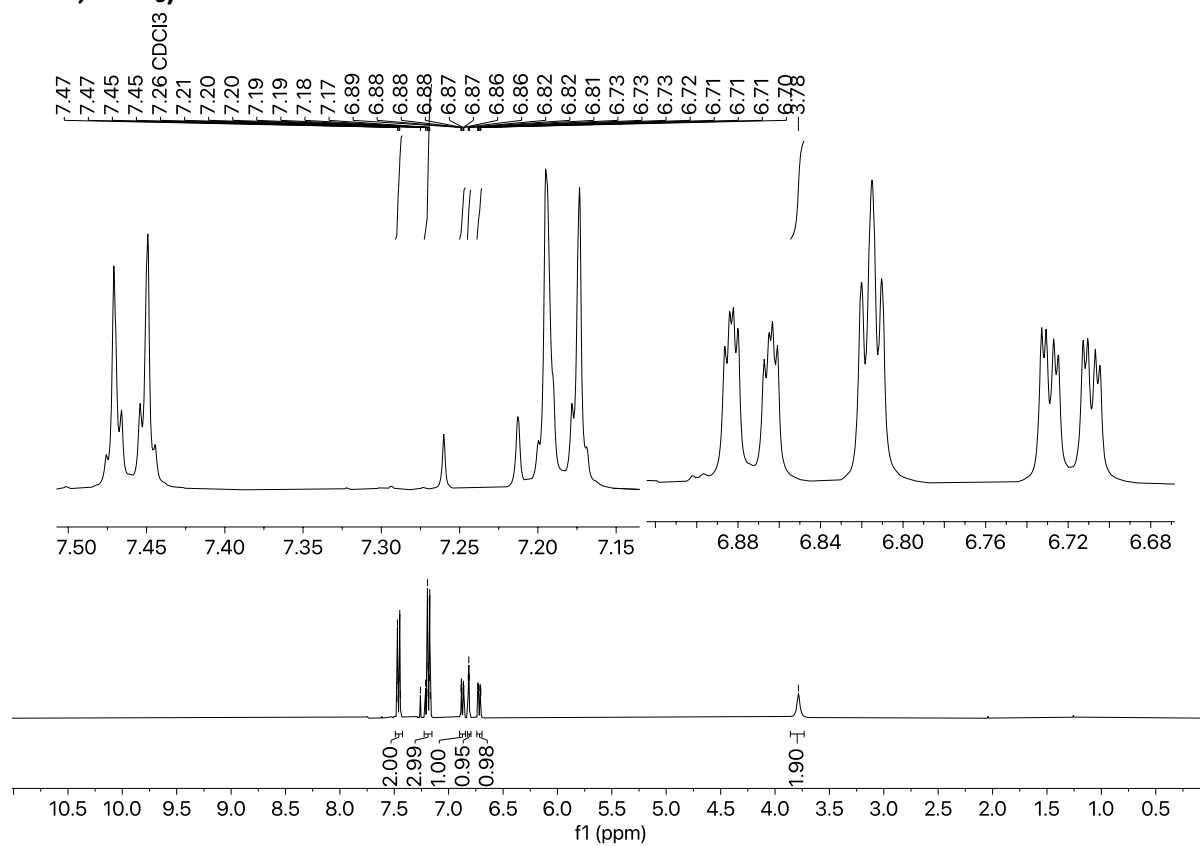
δ_C (101 MHz, $CDCl_3$)



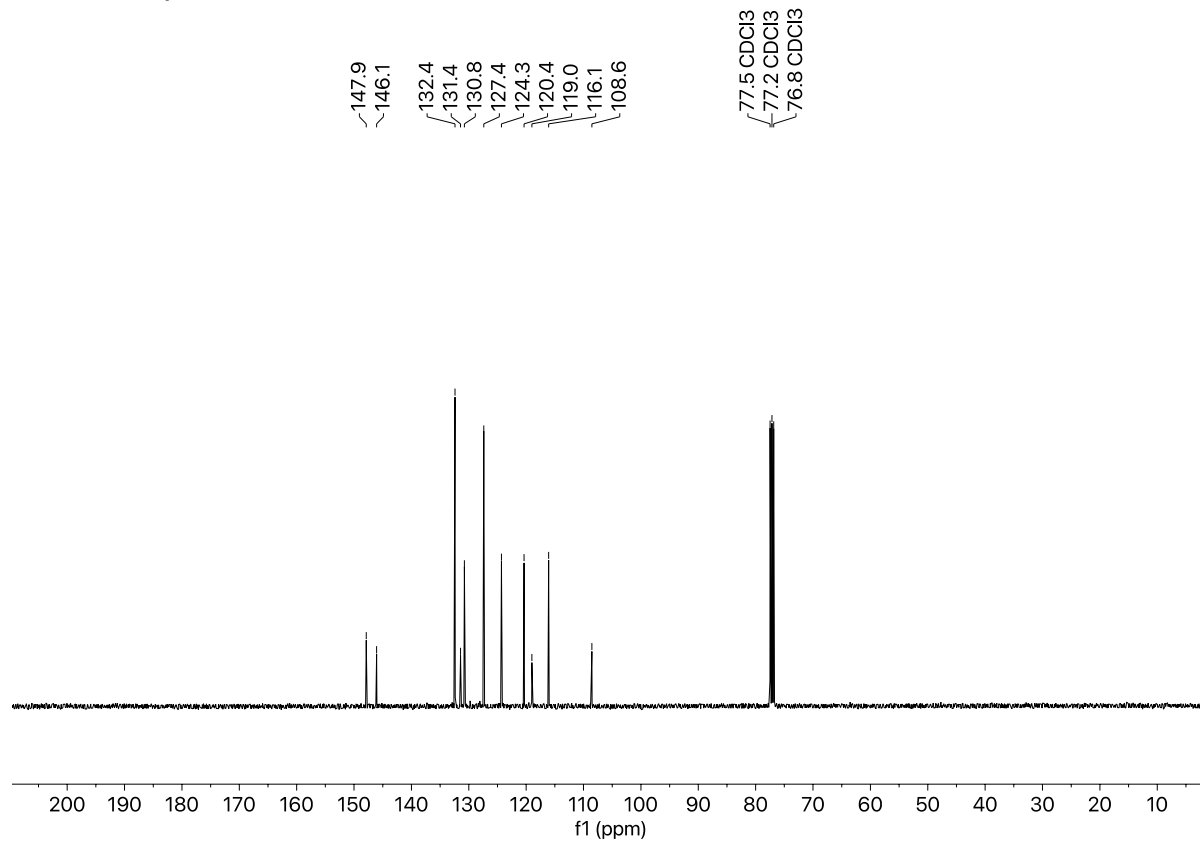
4-((3-Aminophenyl)thio)benzonitrile (4l)



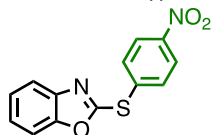
δ_H (400 MHz, $CDCl_3$)



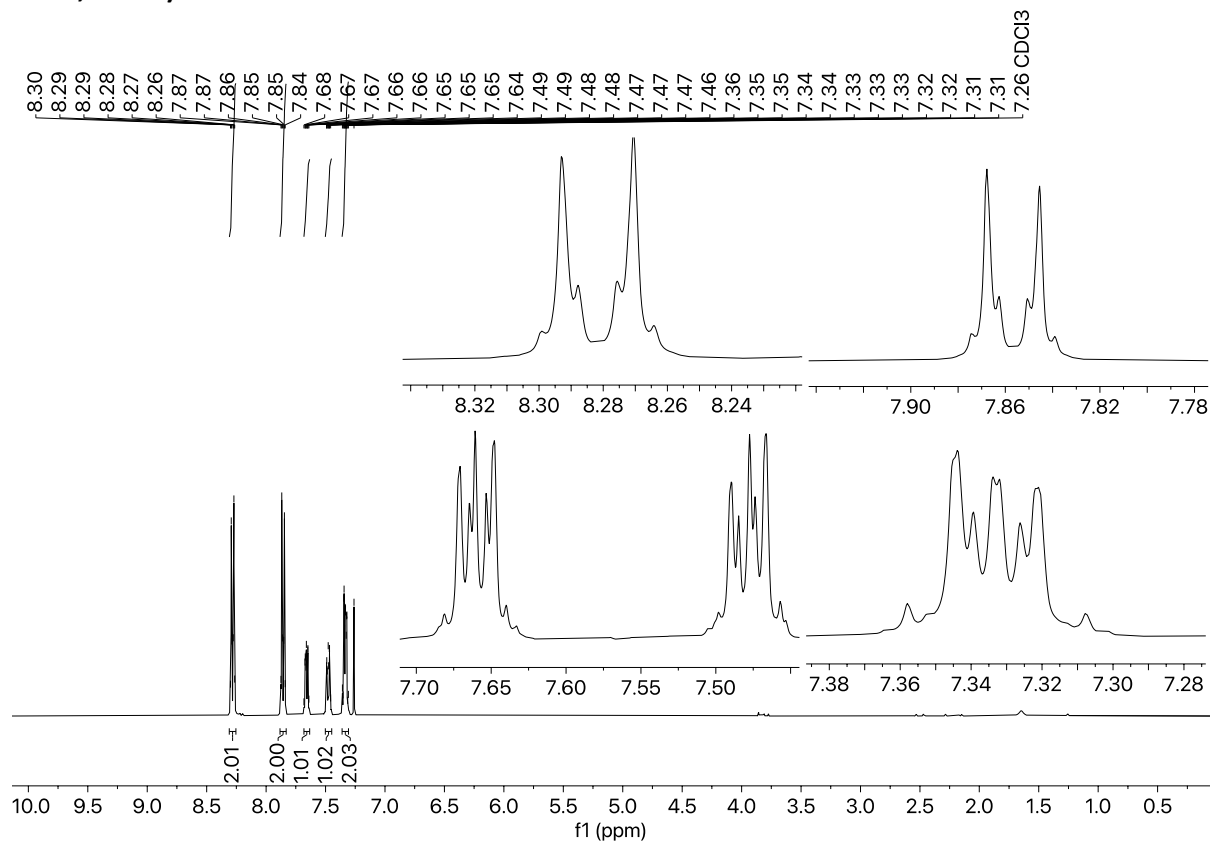
δ_C (101 MHz, $CDCl_3$)



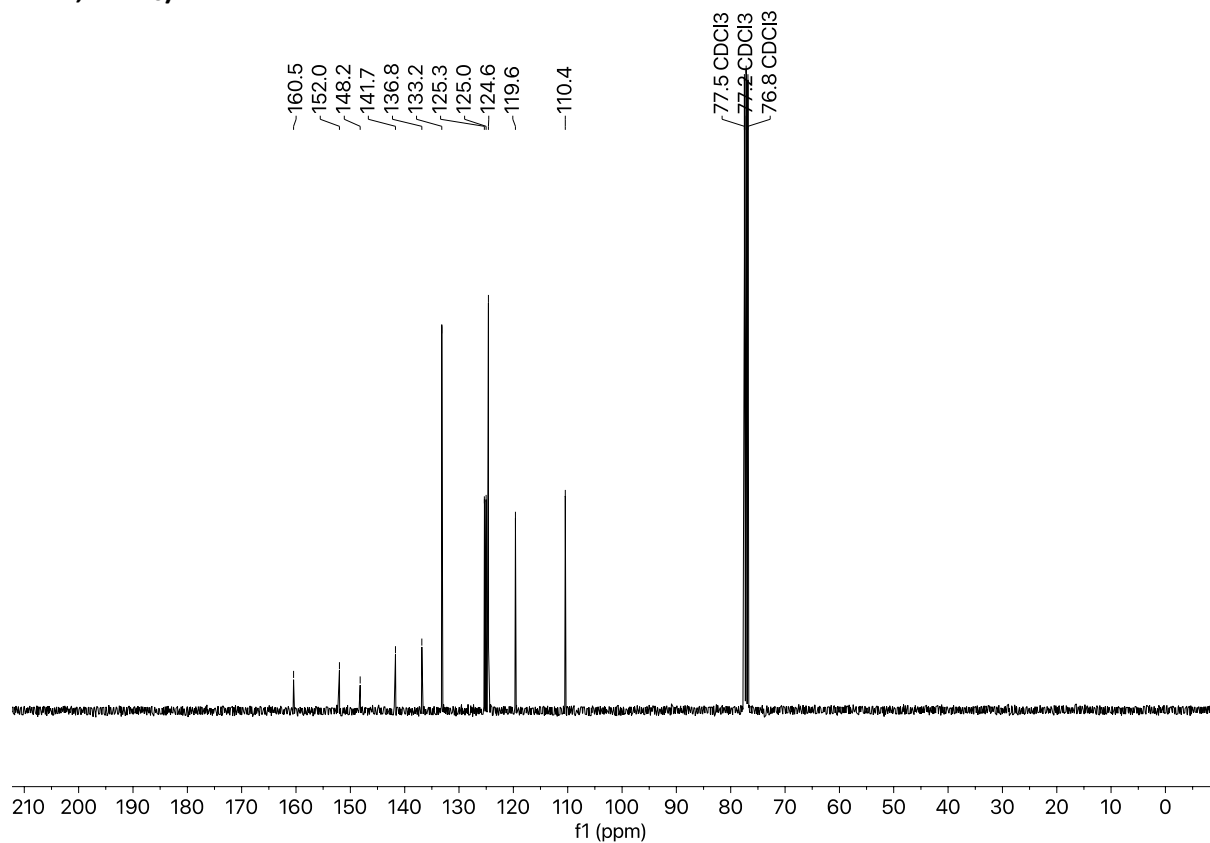
2-((4-Nitrophenyl)thio)benzo[d]oxazole (4m)



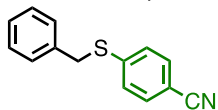
δ_H (400 MHz, $CDCl_3$)



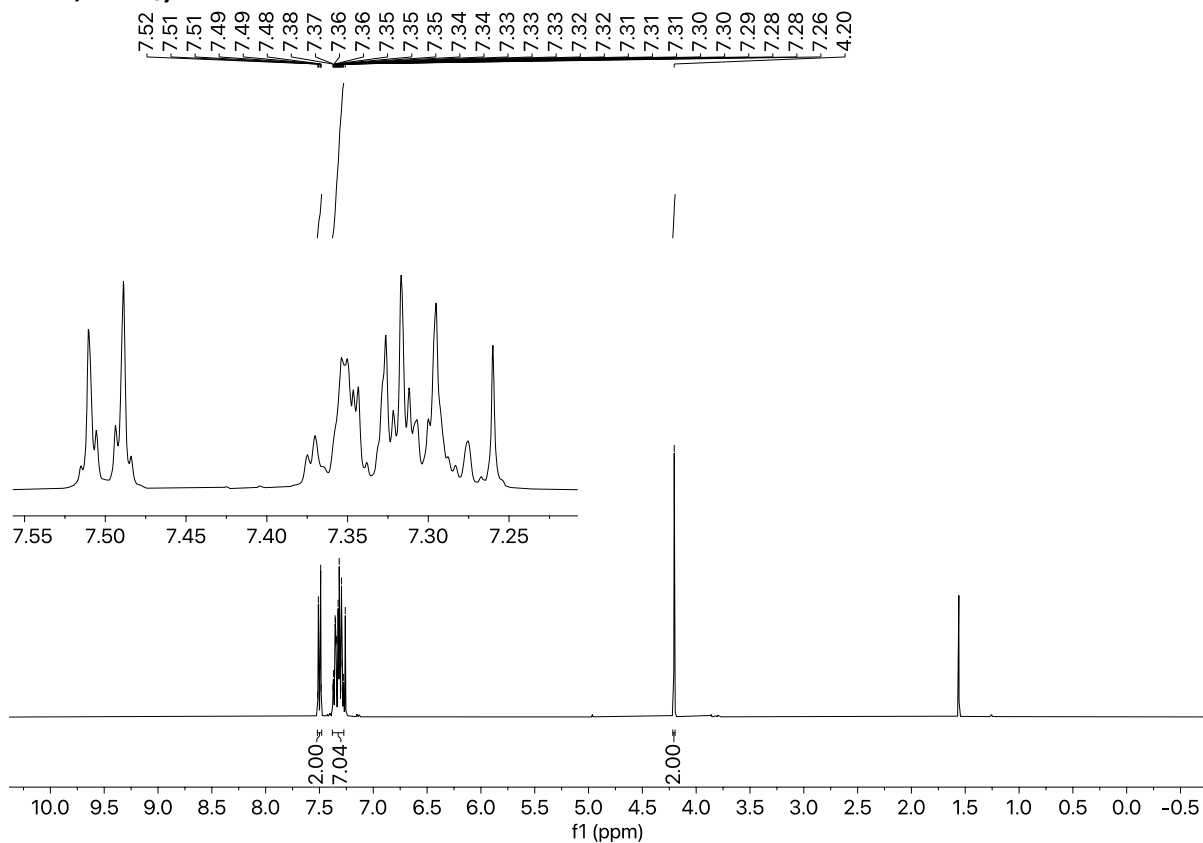
δ_C (101 MHz, $CDCl_3$)



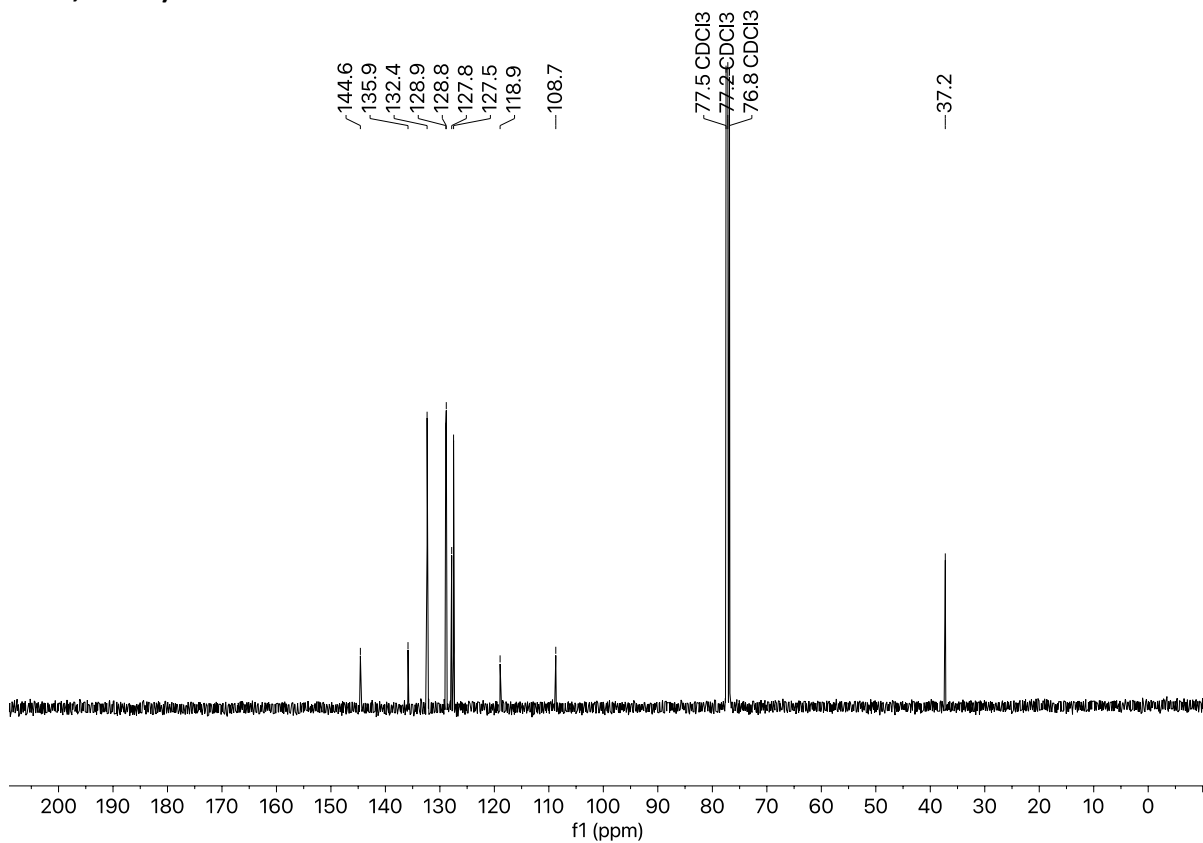
4-(Benzylthio)benzonitrile (4n)

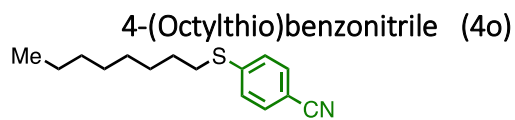


δ_H (400 MHz, $CDCl_3$)

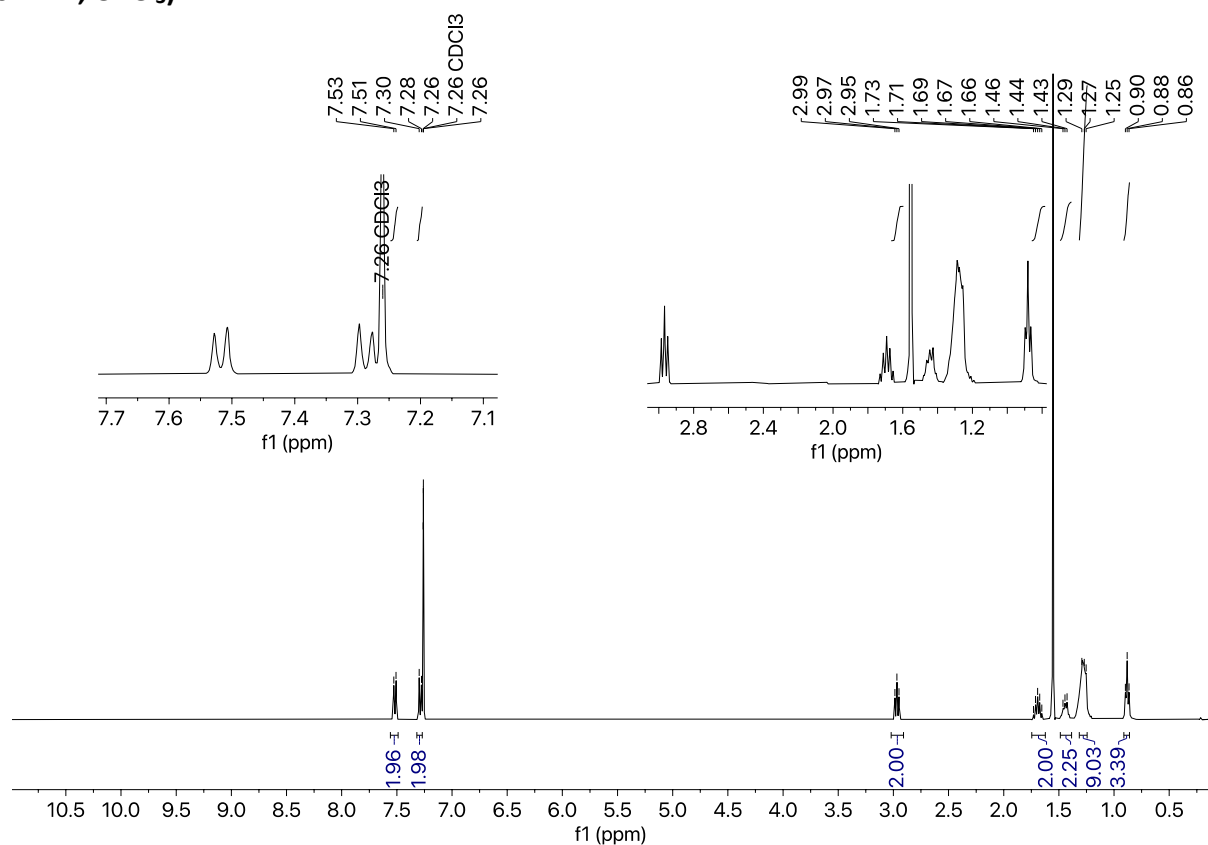


δ_C (101 MHz, $CDCl_3$)

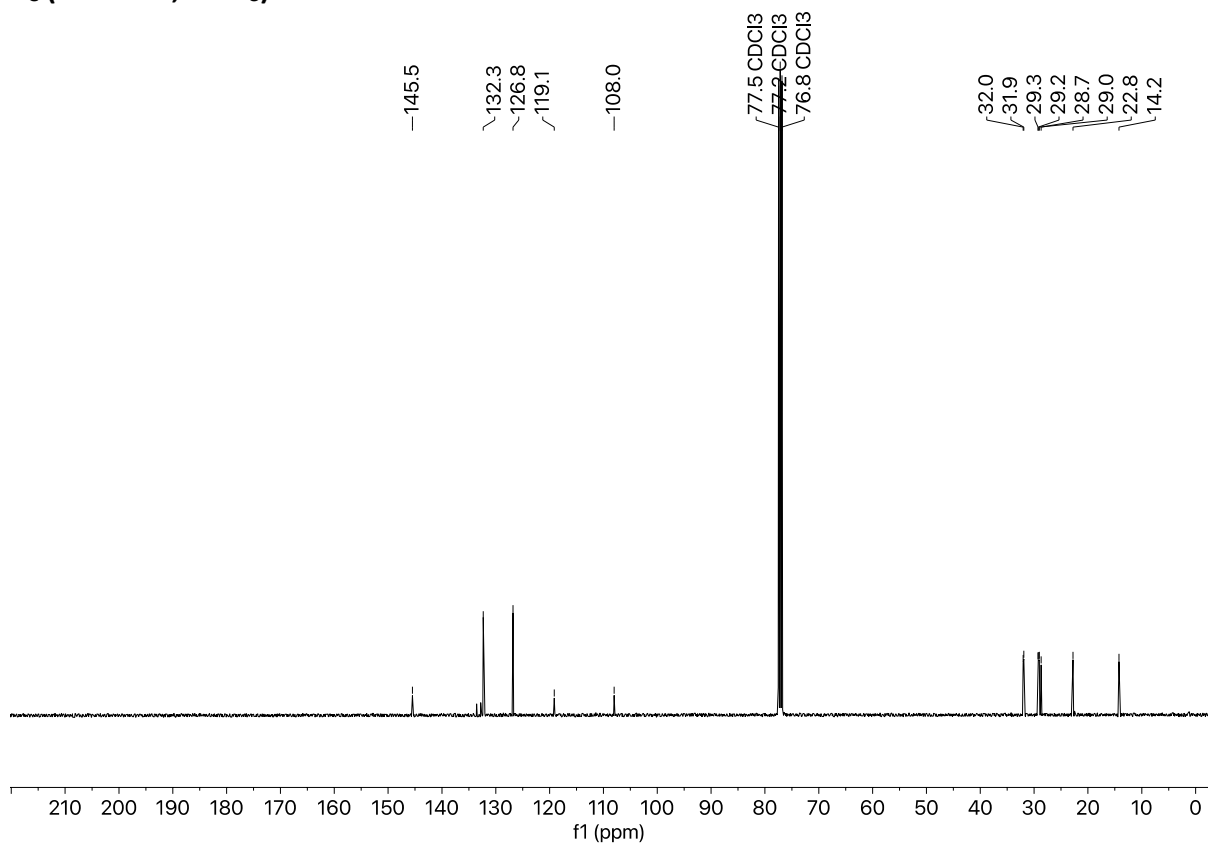




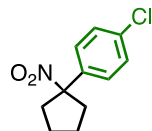
δ_H (400 MHz, $CDCl_3$)



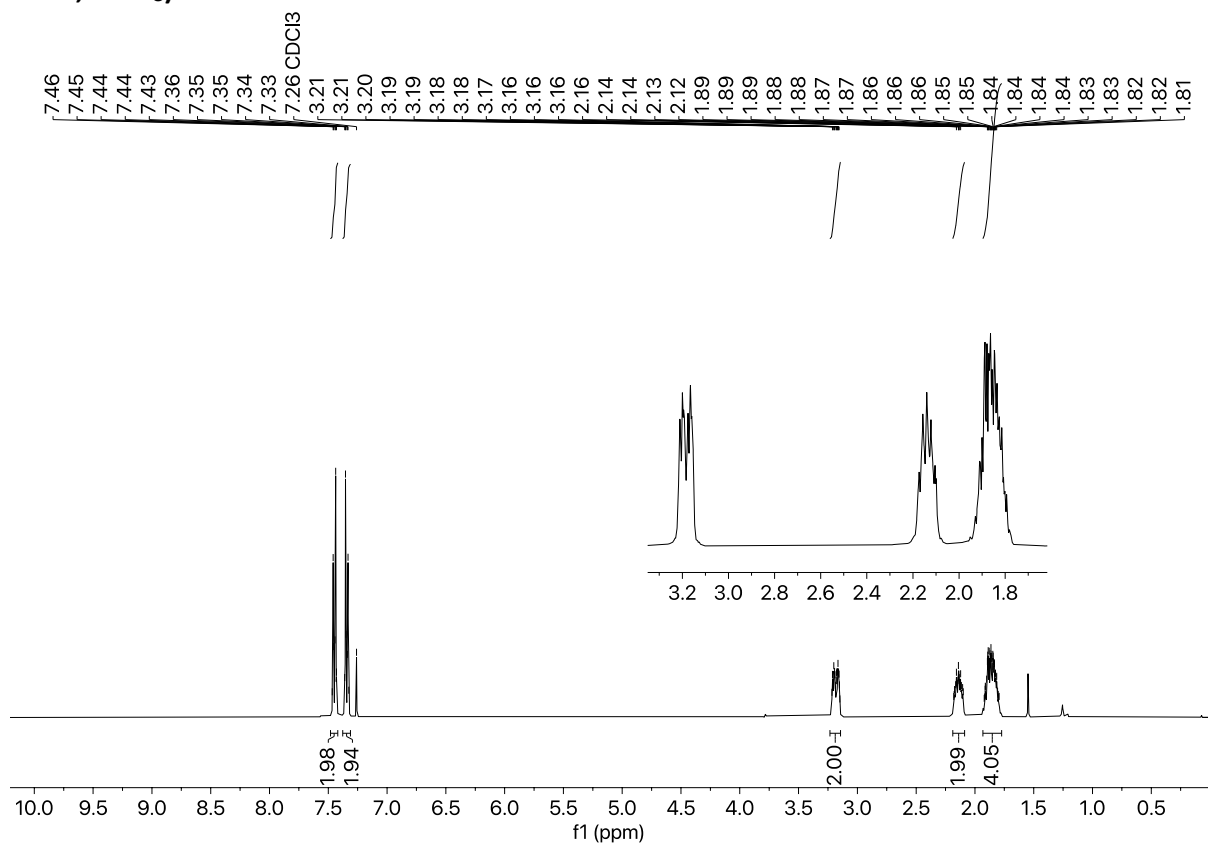
δ_C (101 MHz, $CDCl_3$)



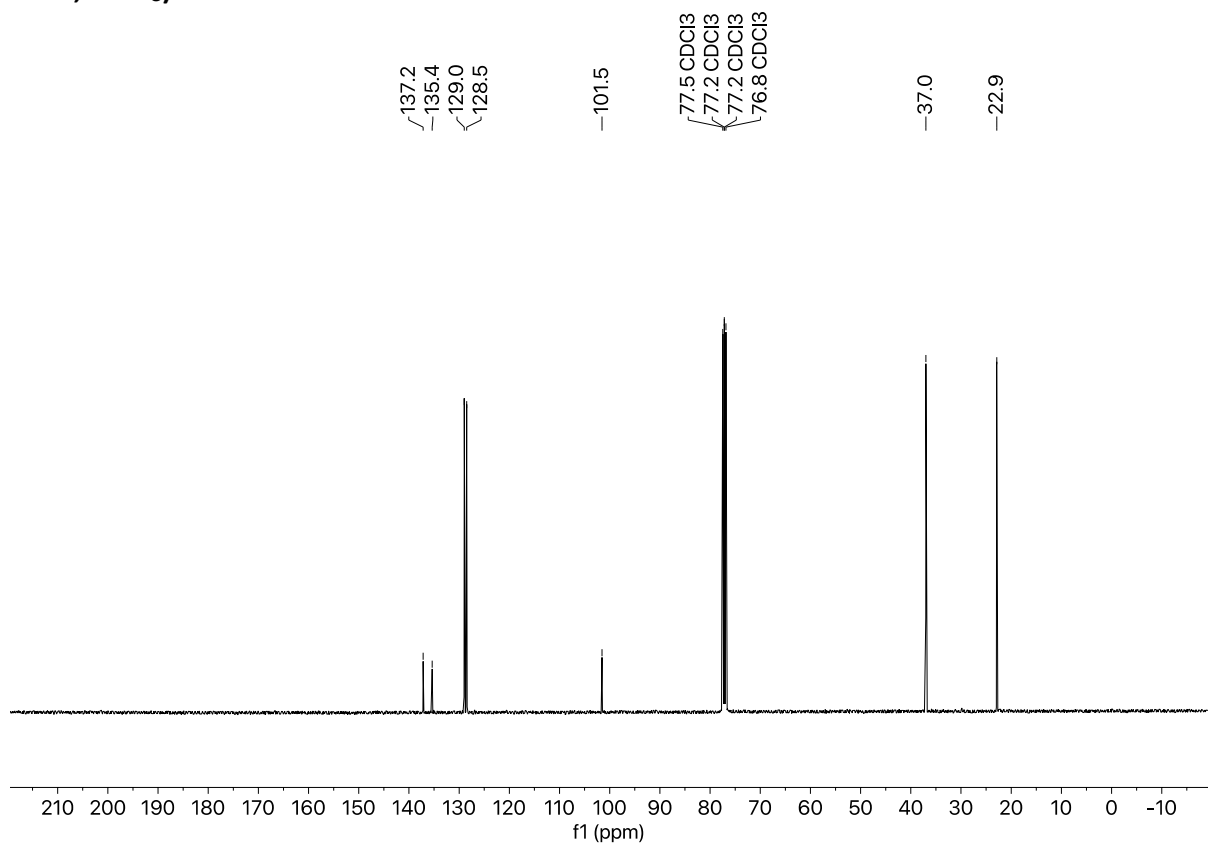
1-Chloro-4-(1-nitrocyclopentyl)benzene (5a)



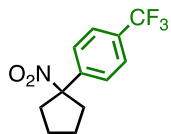
δ_H (400 MHz, $CDCl_3$)



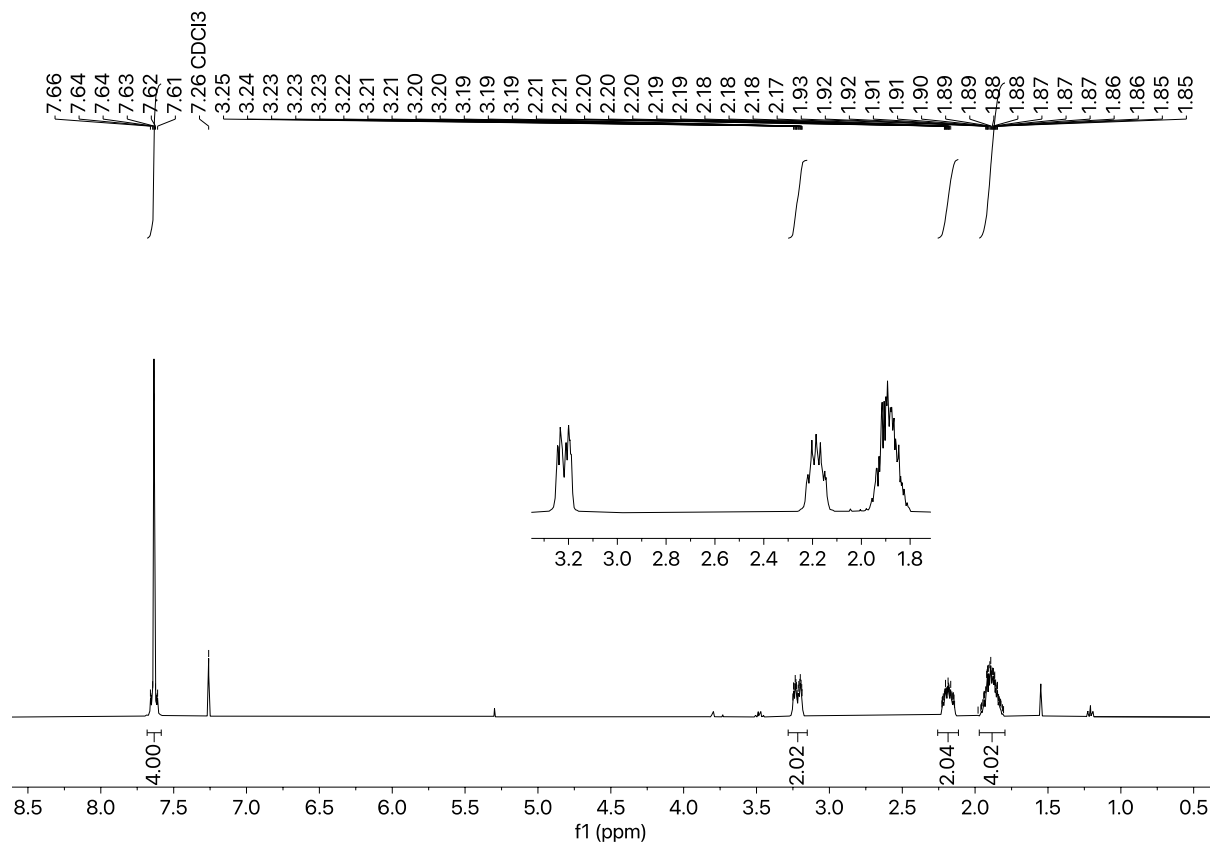
δ_C (101 MHz, $CDCl_3$)



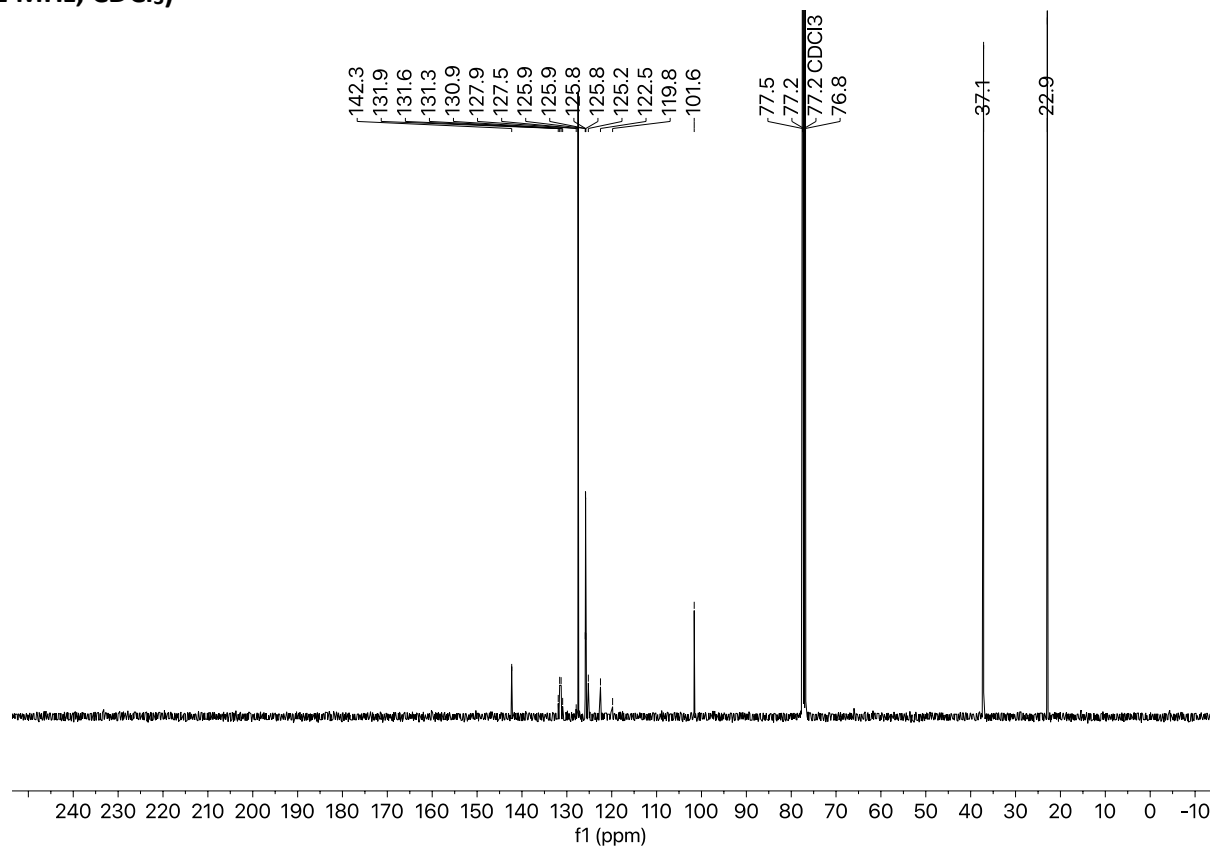
1-Trifluoromethyl-4-(1-nitrocyclopentyl)benzene (5b)



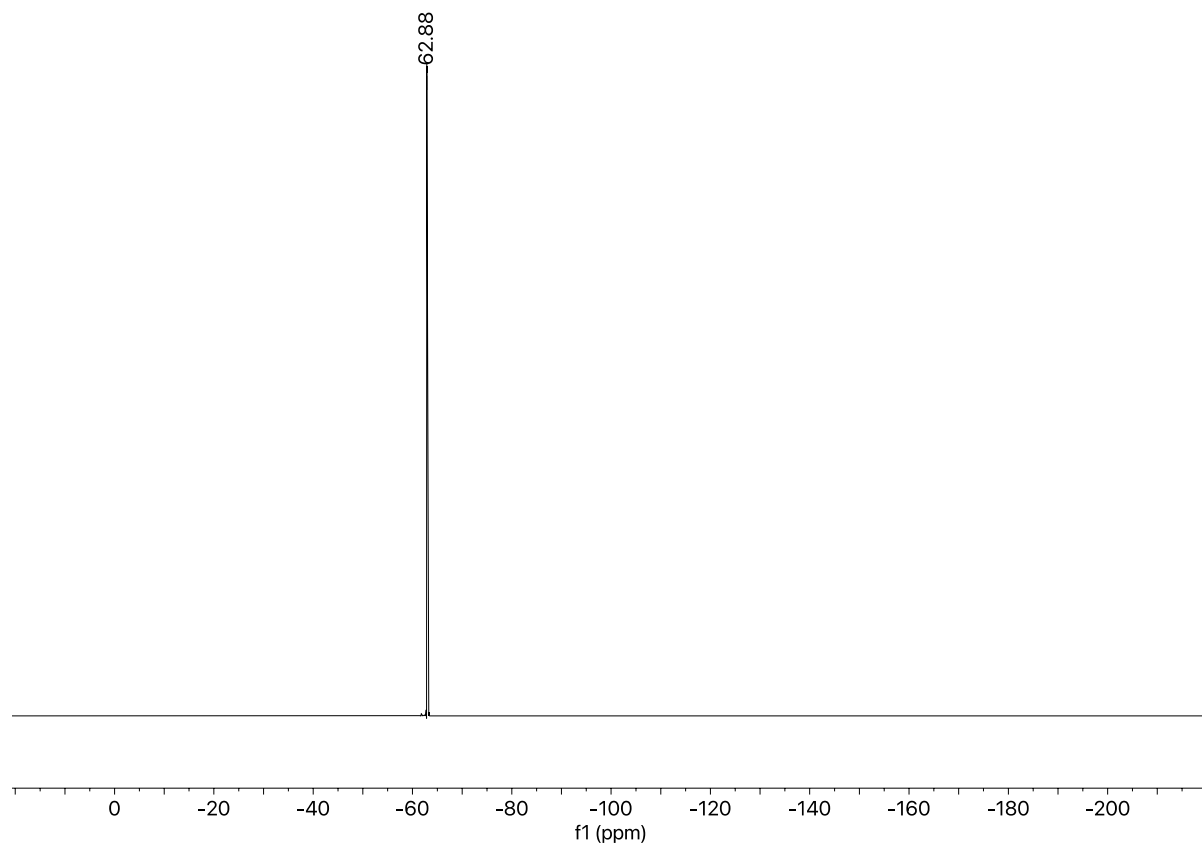
δ_H (400 MHz, $CDCl_3$)



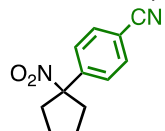
δ_C (101 MHz, $CDCl_3$)



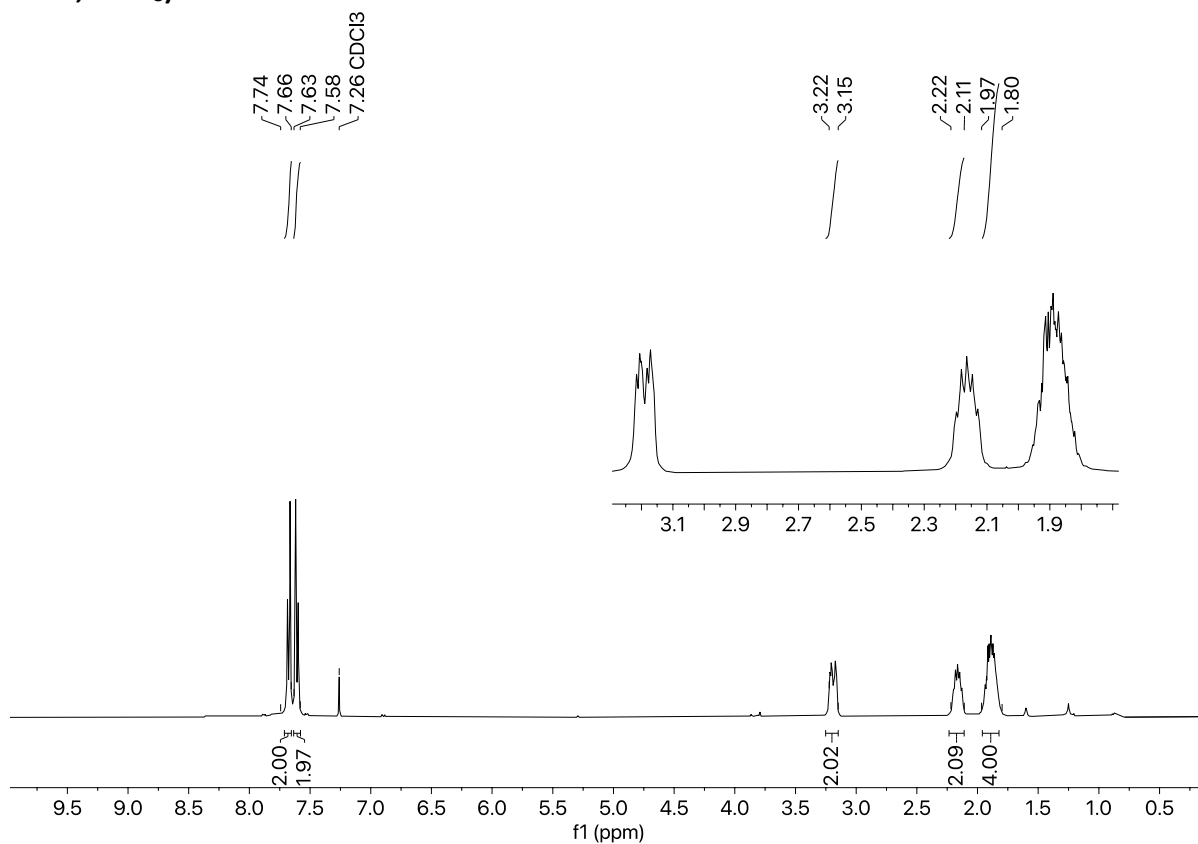
δ_F (376 MHz, $CDCl_3$)



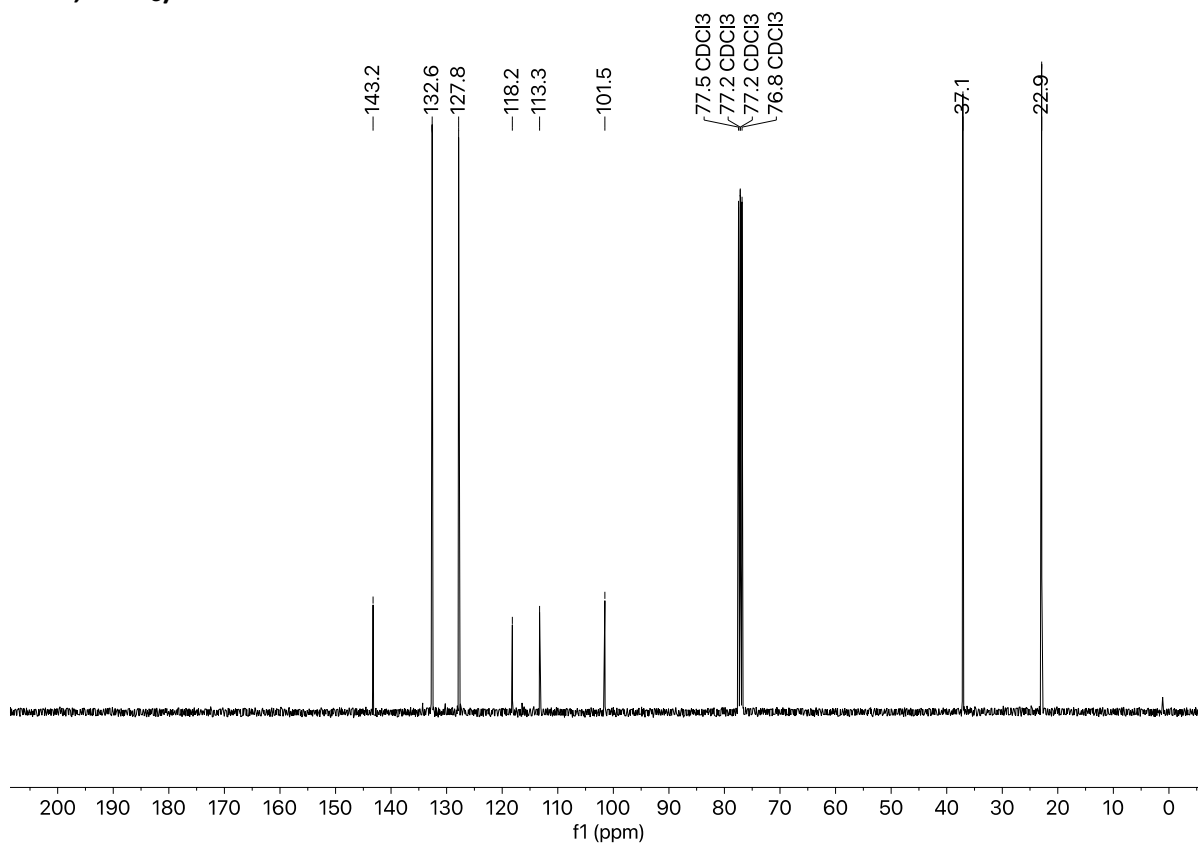
4-(1-Nitrocyclopentyl)benzonitrile (5c)



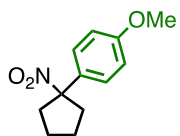
δ_H (400 MHz, $CDCl_3$)



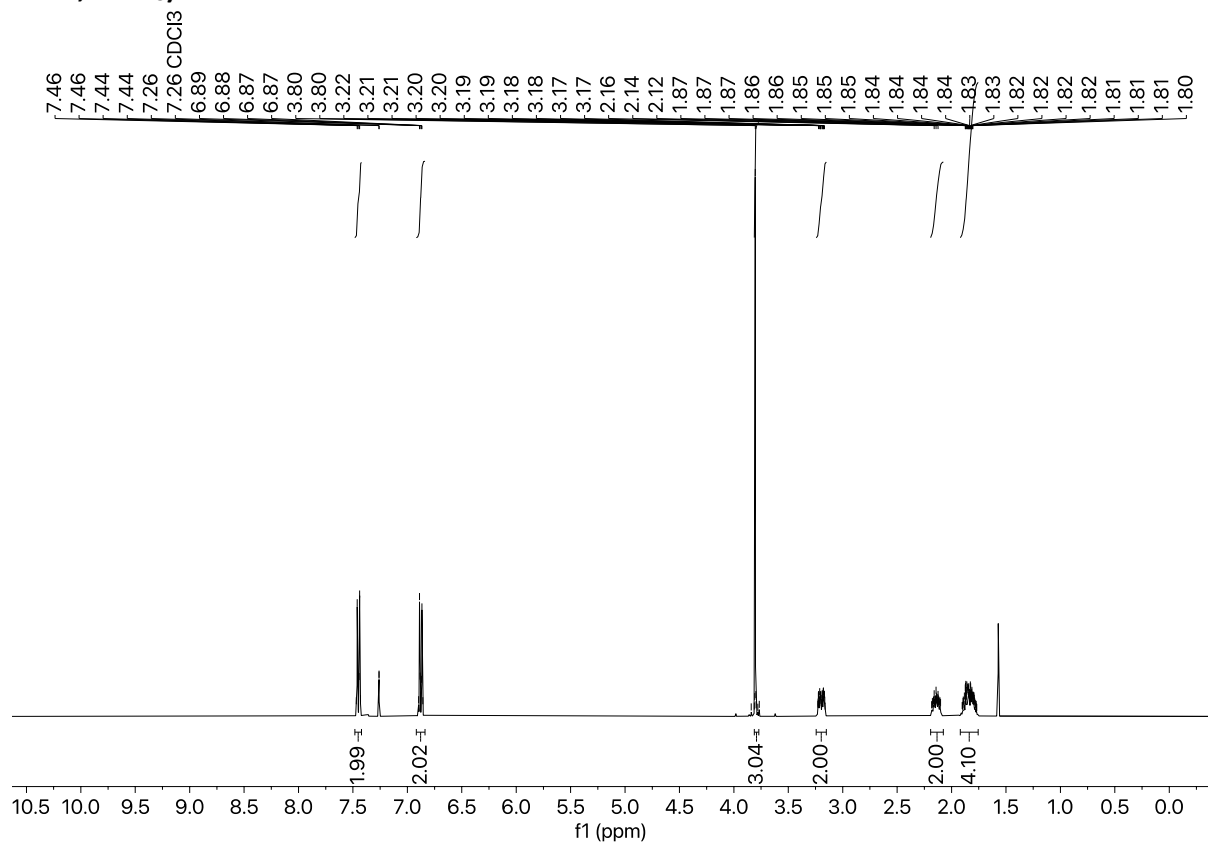
δ_C (101 MHz, $CDCl_3$)



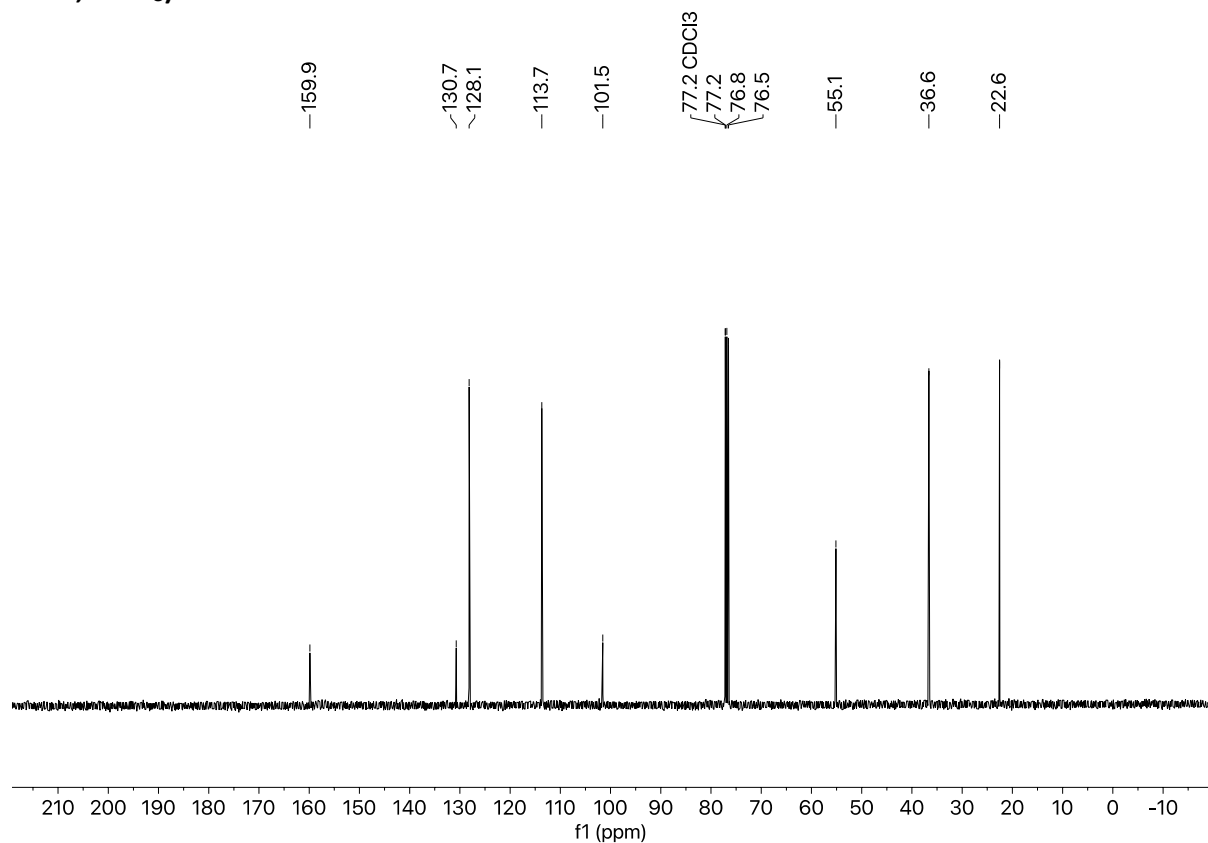
1-methoxy-4-(1-nitrocyclopentyl)benzene (5d)



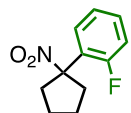
δ_H (400 MHz, $CDCl_3$)



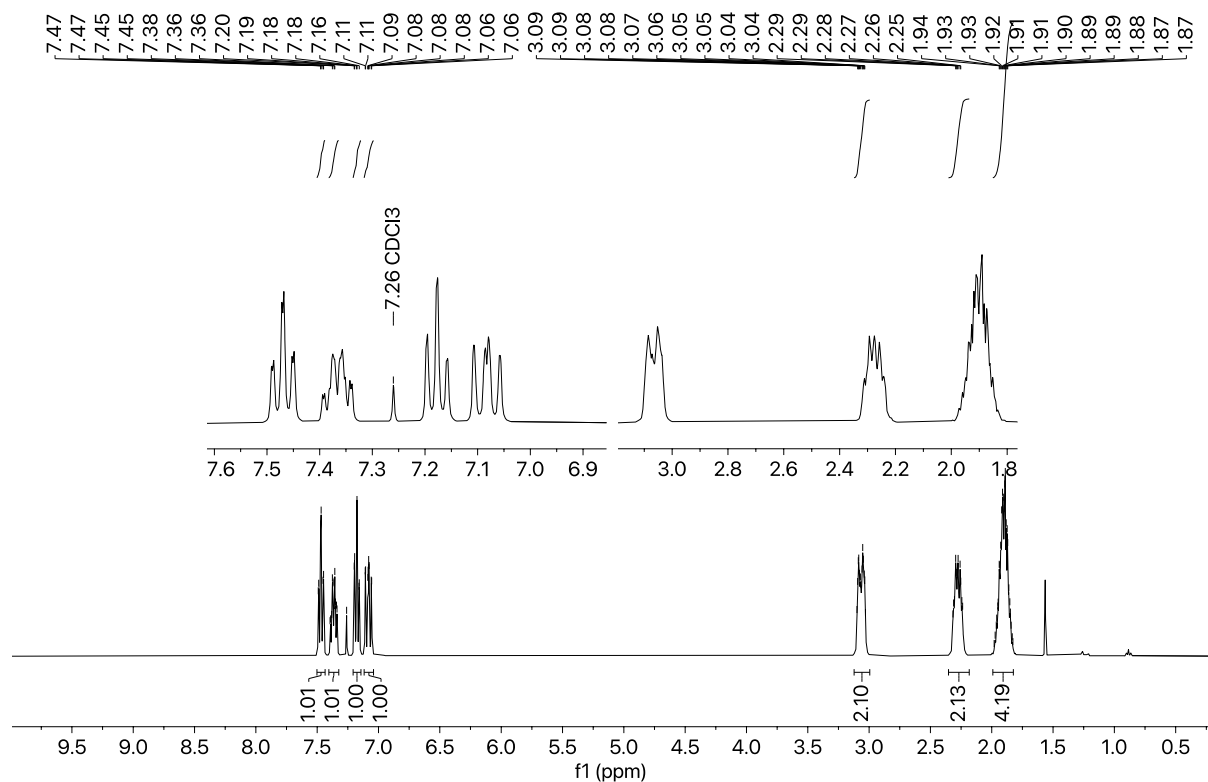
δ_C (101 MHz, $CDCl_3$)



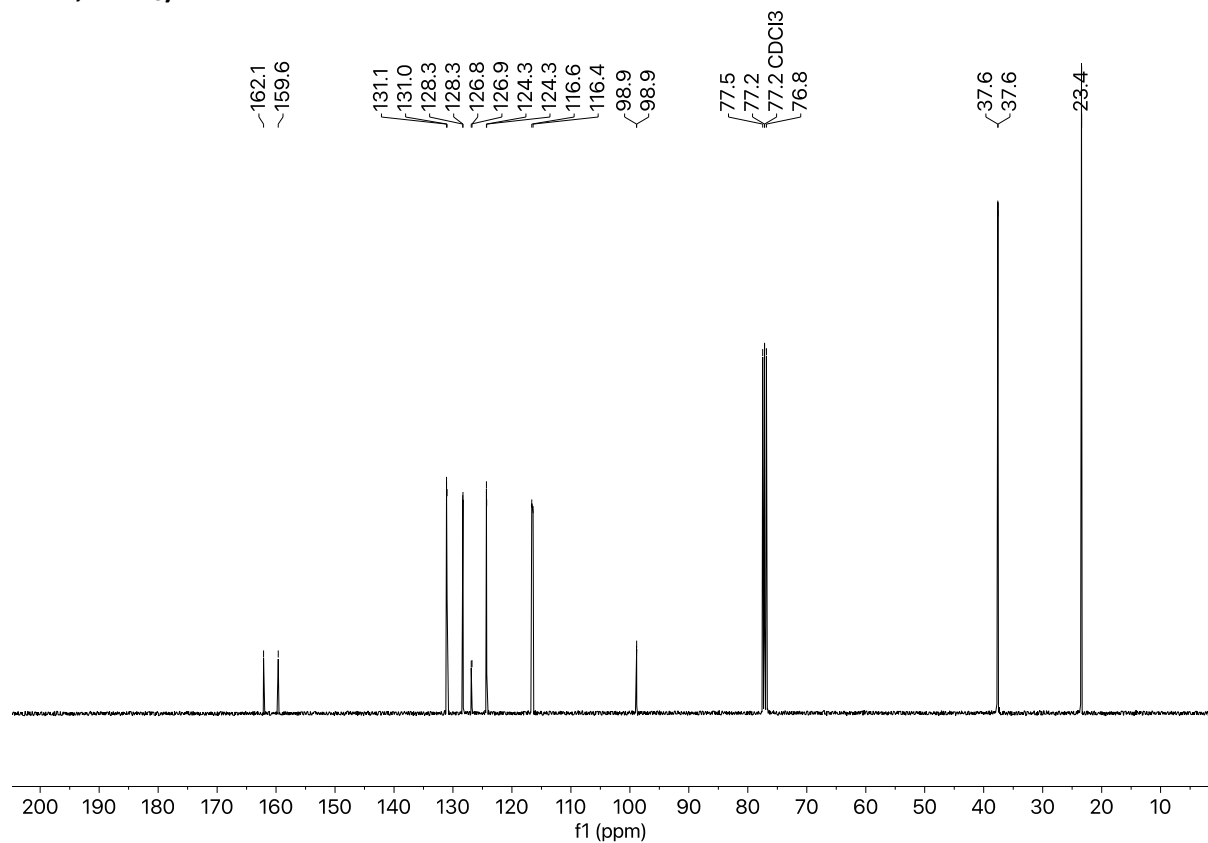
1-Fluoro-2-(1-nitrocyclopentyl)benzene (5e)



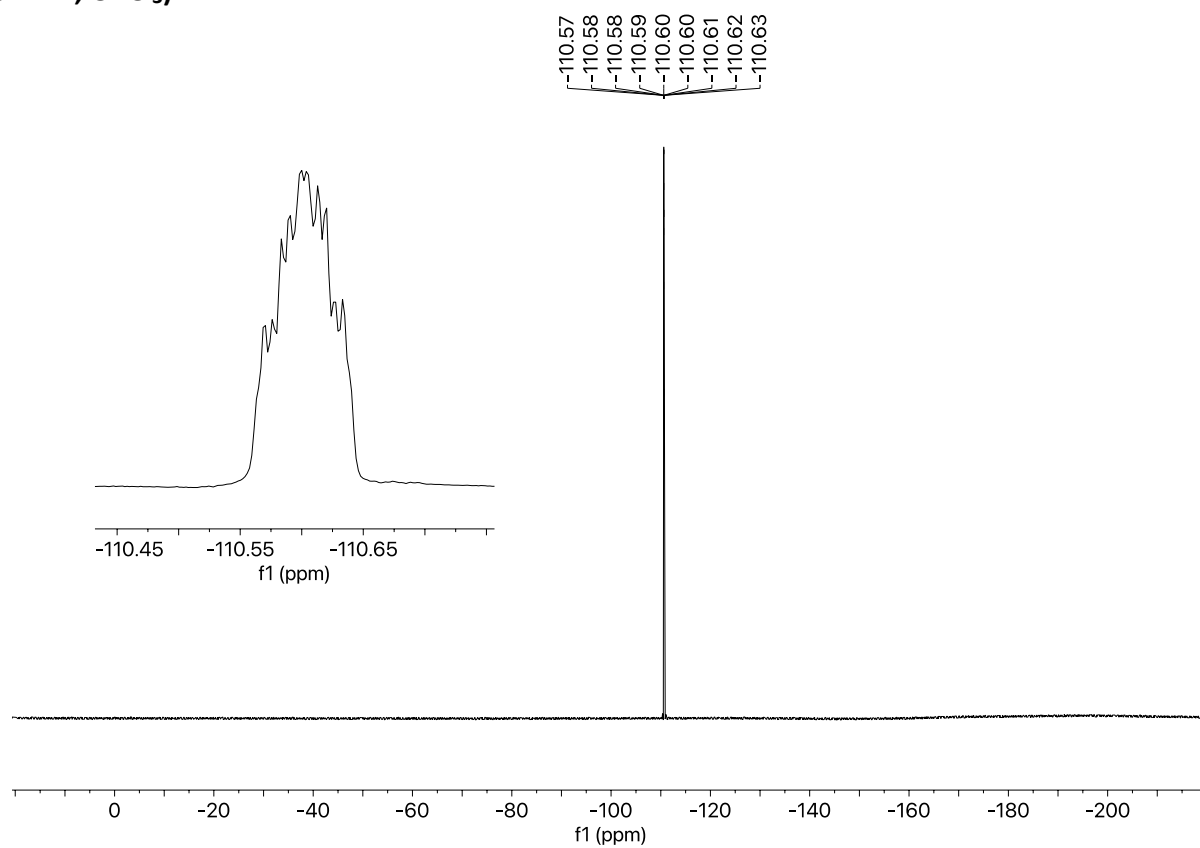
δ_H (400 MHz, $CDCl_3$)



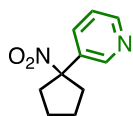
δ_C (101 MHz, $CDCl_3$)



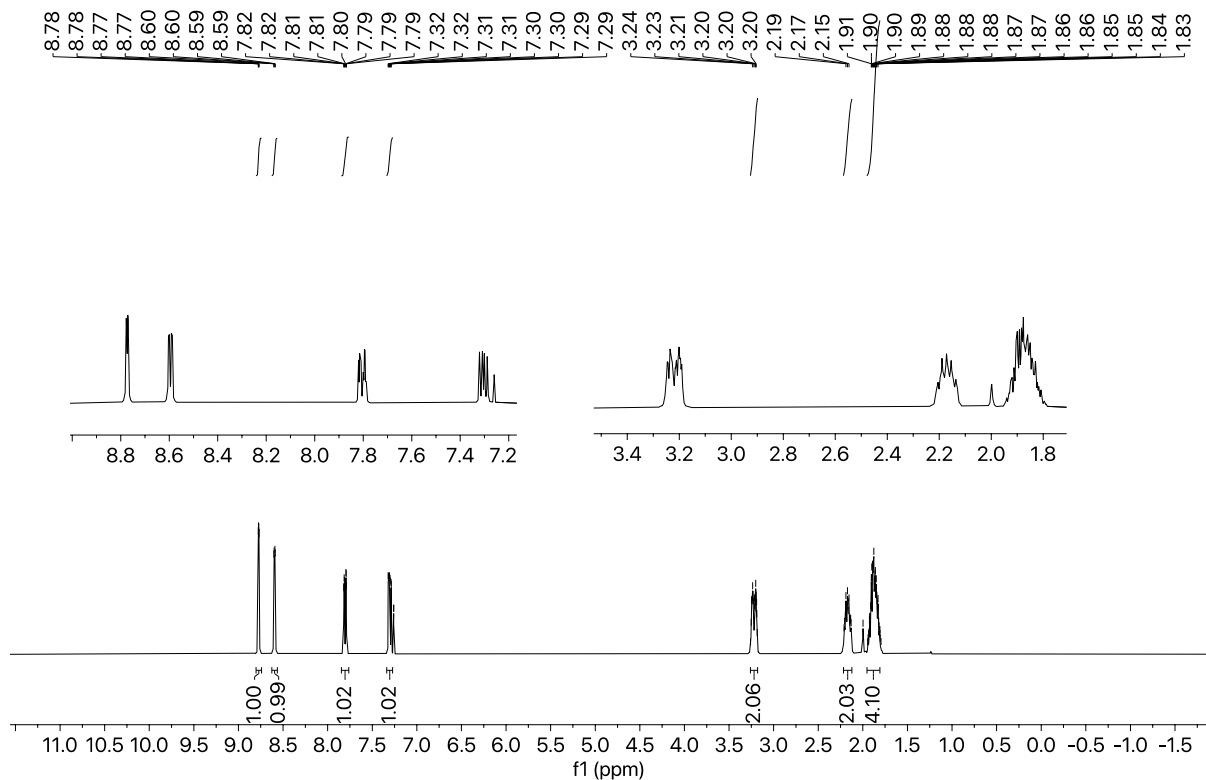
δ_F (376 MHz, $CDCl_3$)



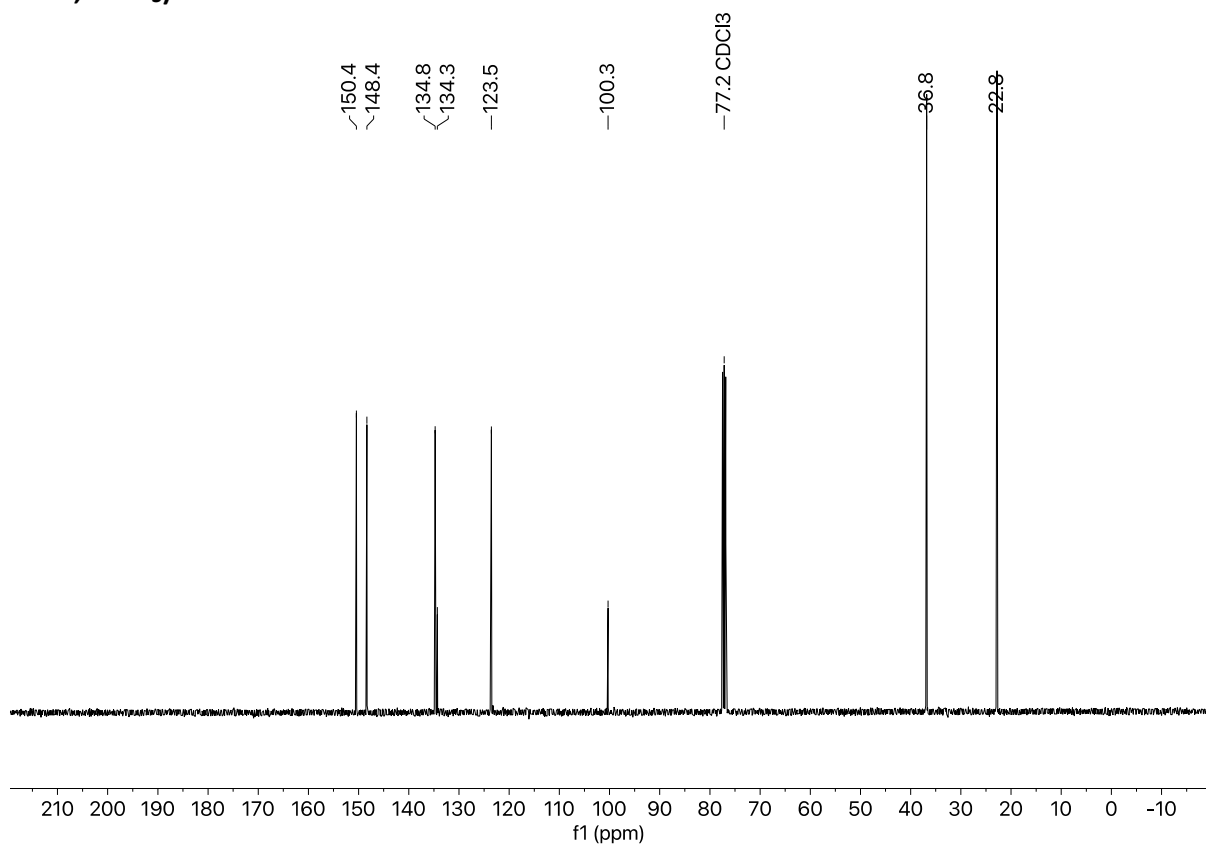
3-(1-nitrocyclopentyl)pyridine (5f)



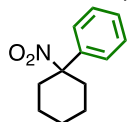
δ_H (400 MHz, $CDCl_3$)



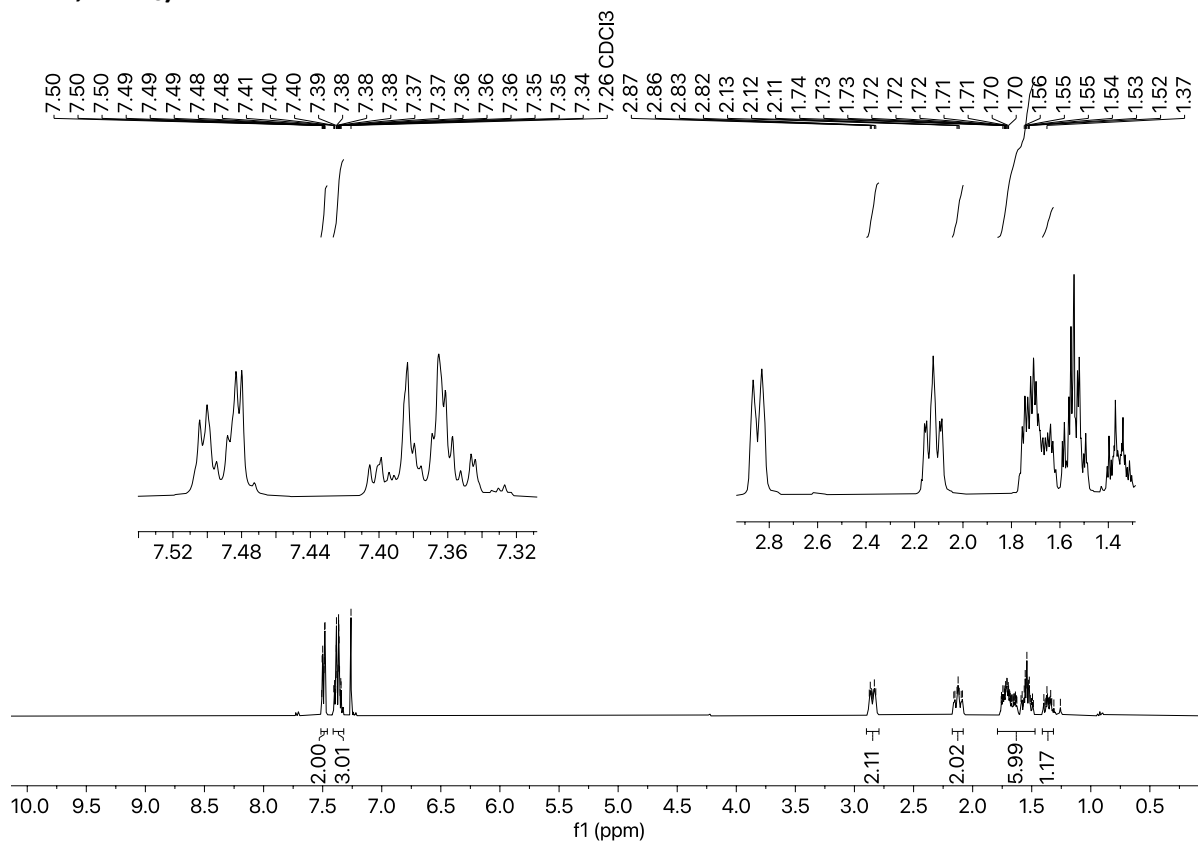
δ_C (101 MHz, $CDCl_3$)



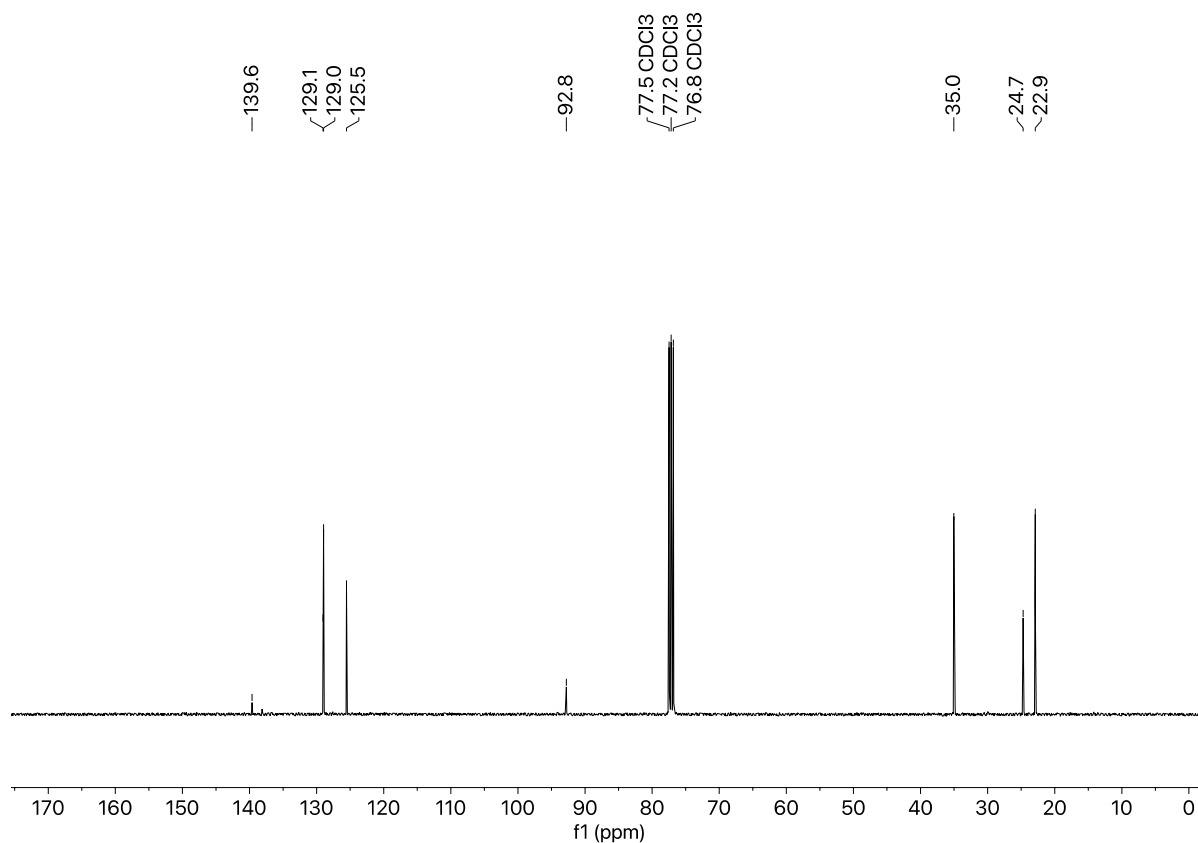
(1-Nitrocyclohexyl)benzene (5g)



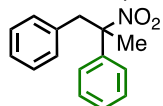
δ_H (400 MHz, $CDCl_3$)



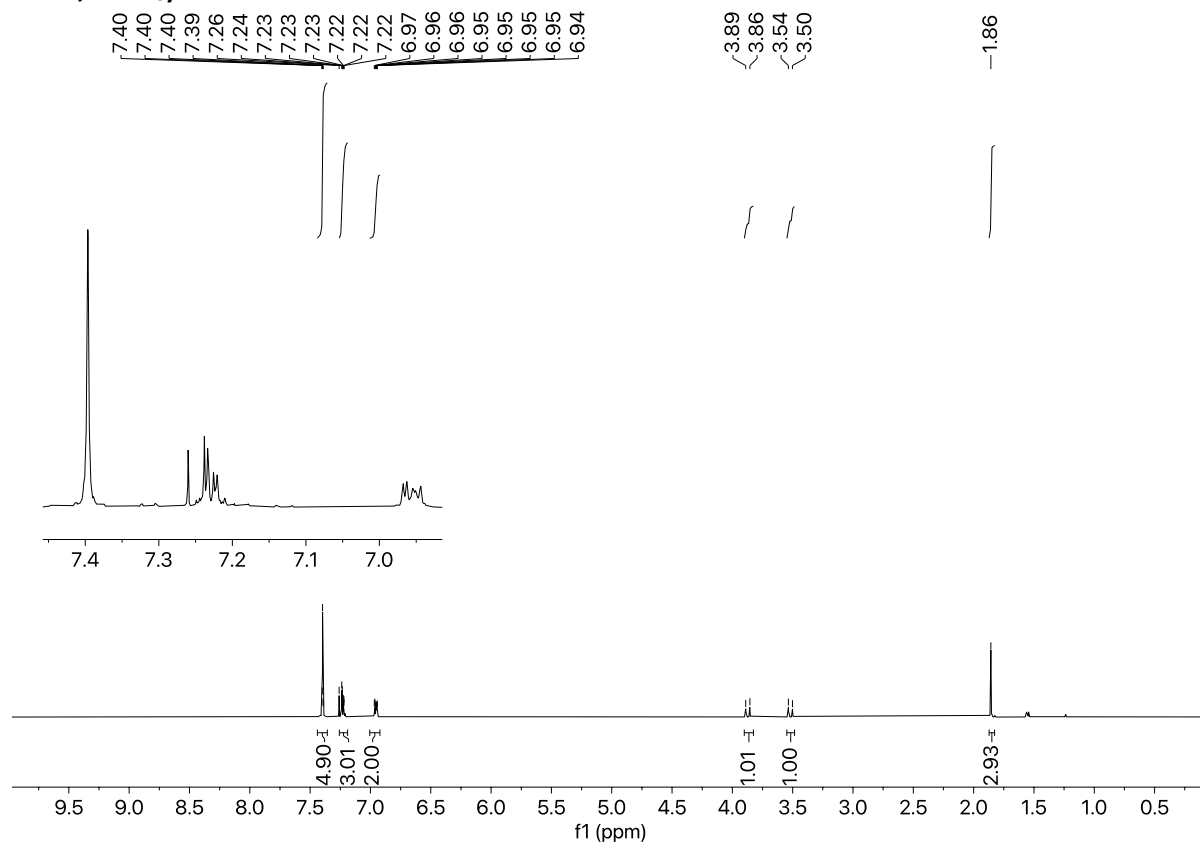
δ_C (101 MHz, $CDCl_3$)



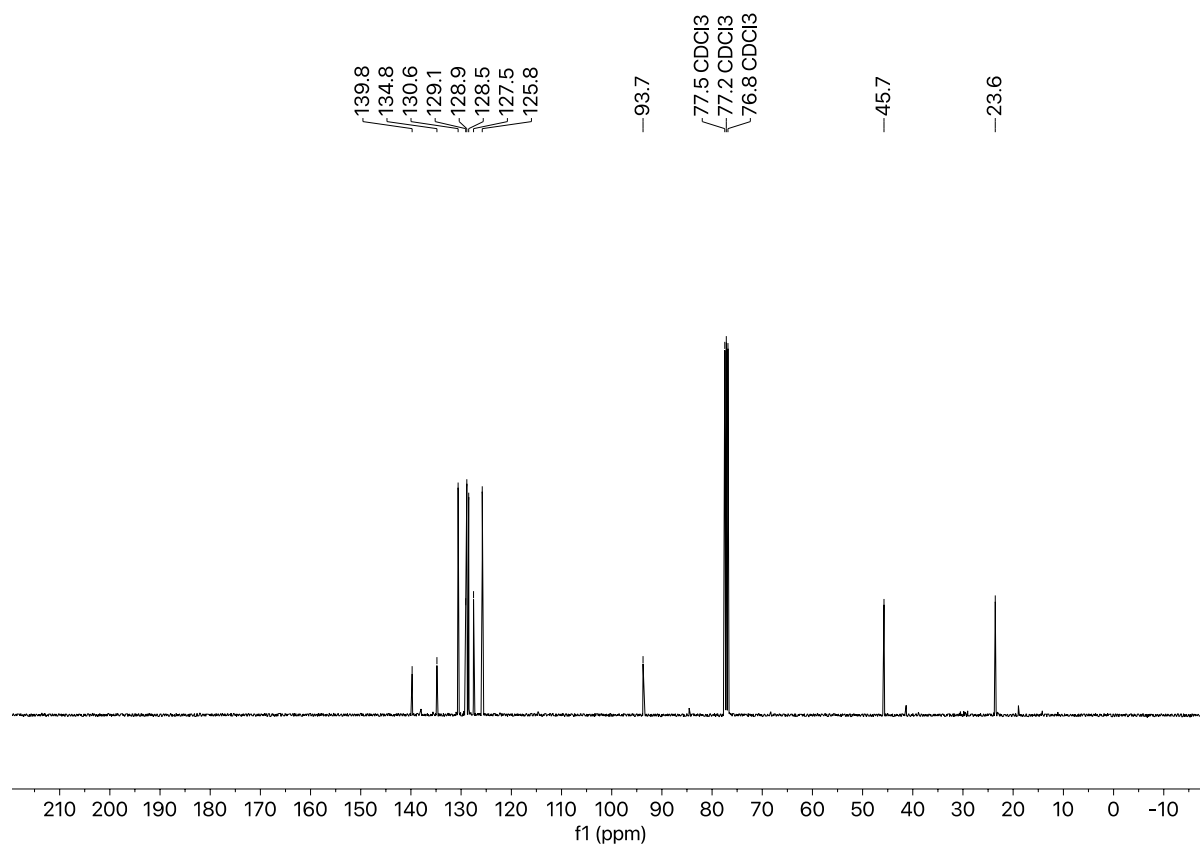
(2-nitropropane-1,2-diyl)dibenzene (5h)



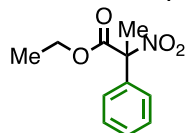
δ_H (400 MHz, $CDCl_3$)



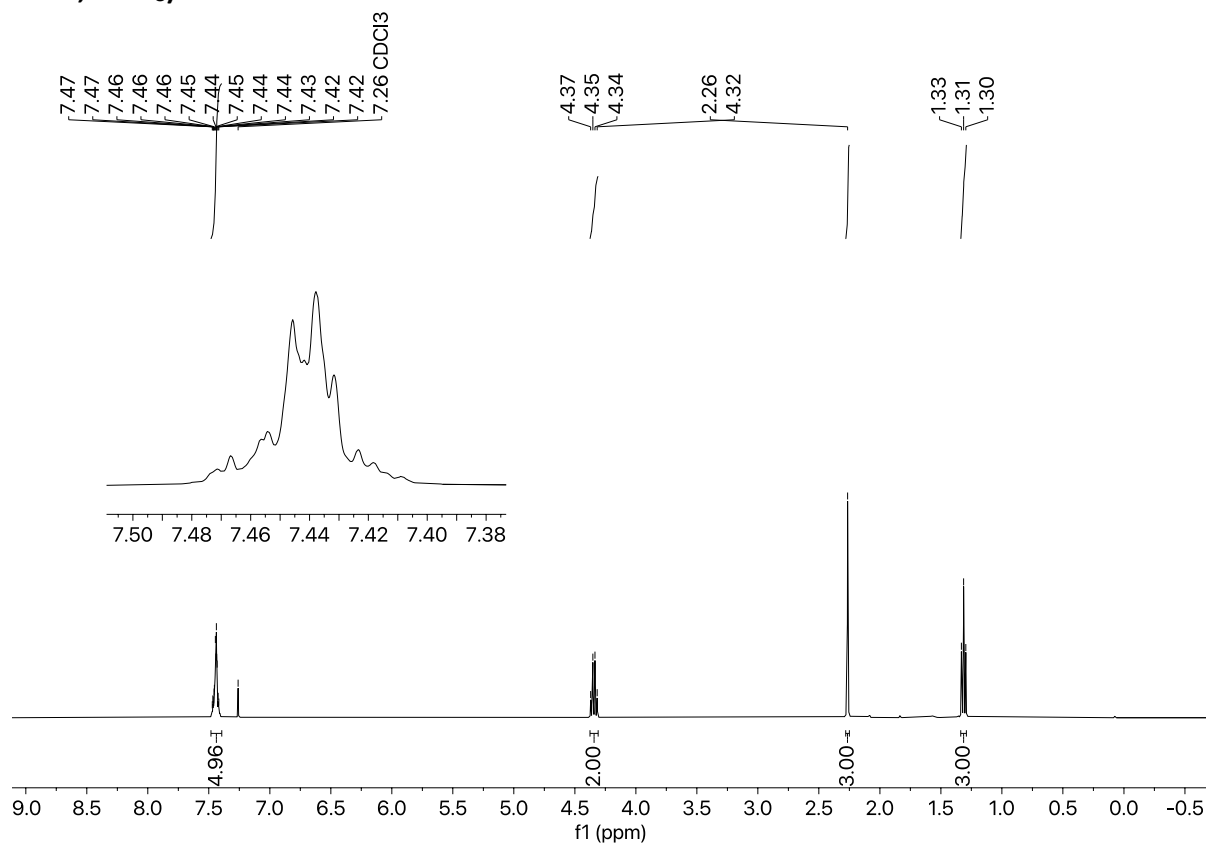
δ_C (101 MHz, $CDCl_3$)



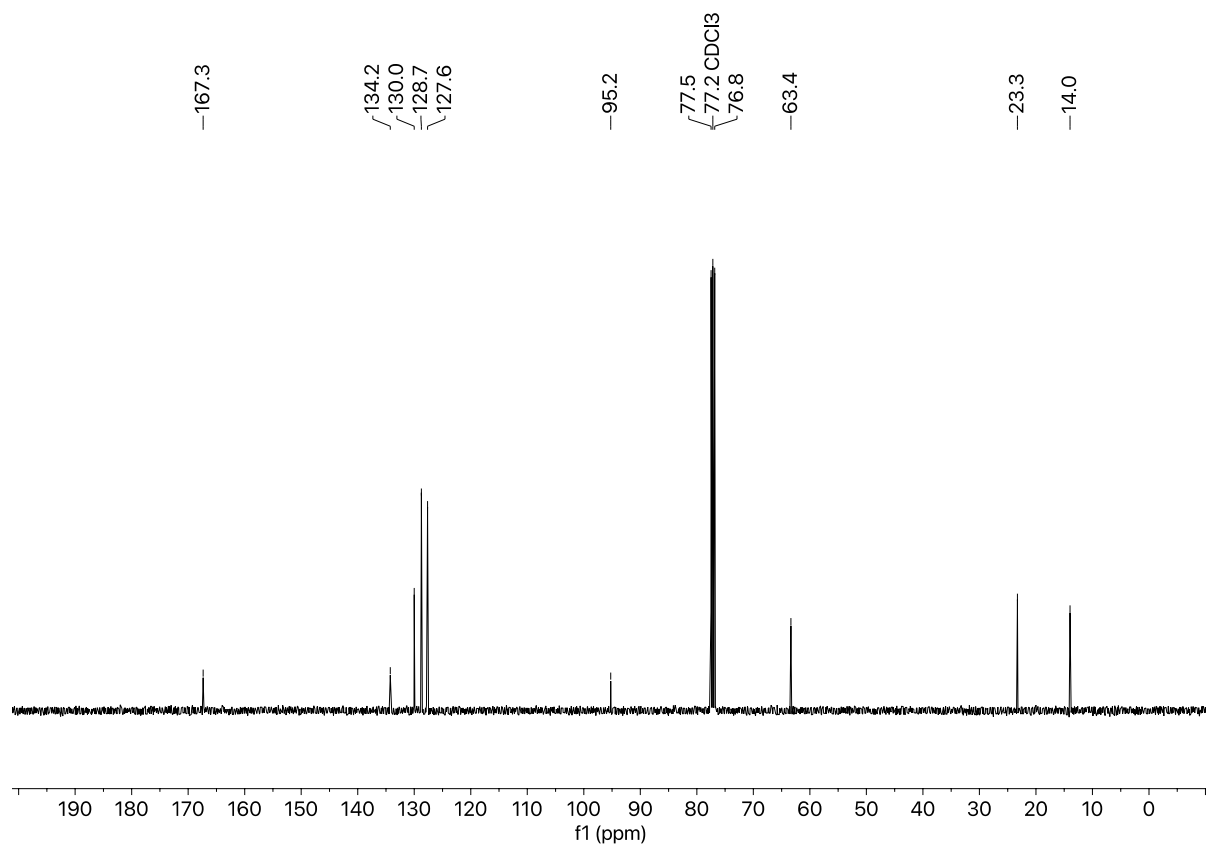
Ethyl 2-nitro-2-phenylpropanoate (5i)



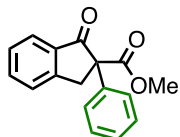
δ_H (400 MHz, $CDCl_3$)



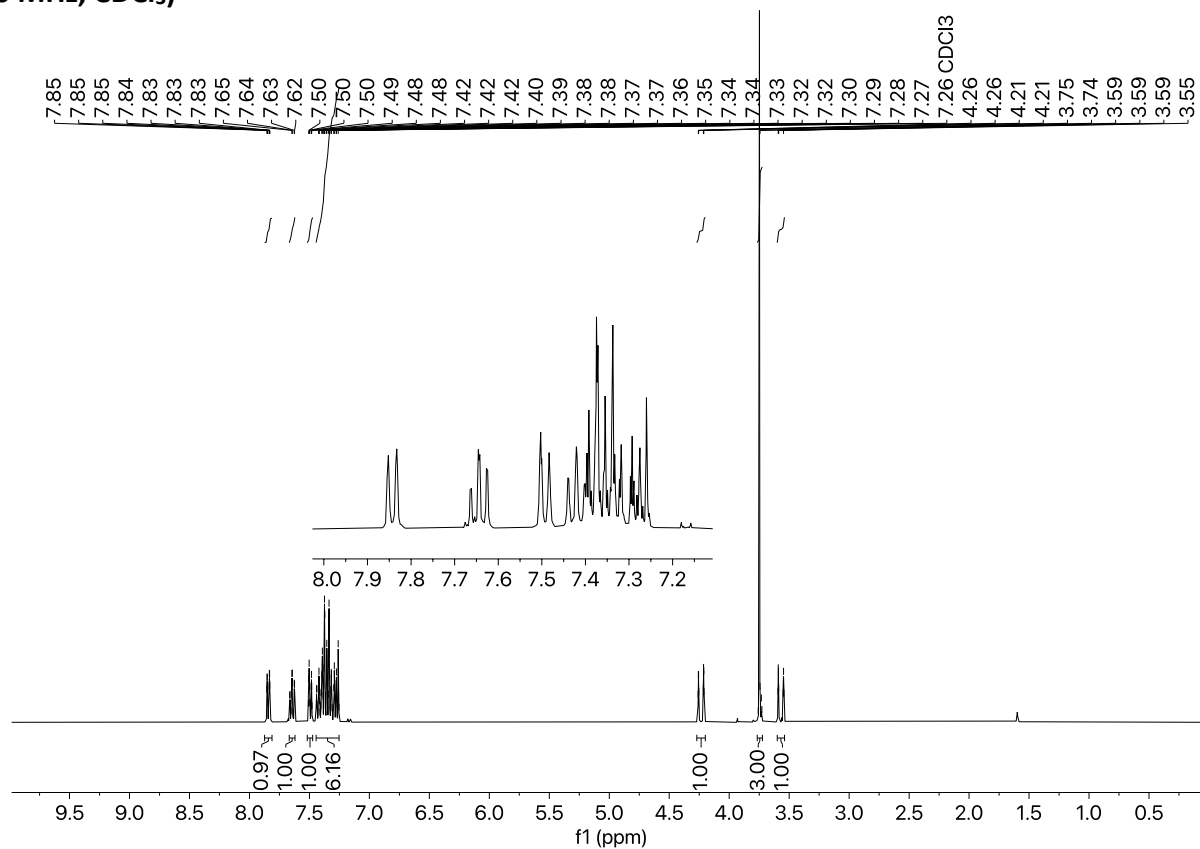
δ_C (101 MHz, $CDCl_3$)



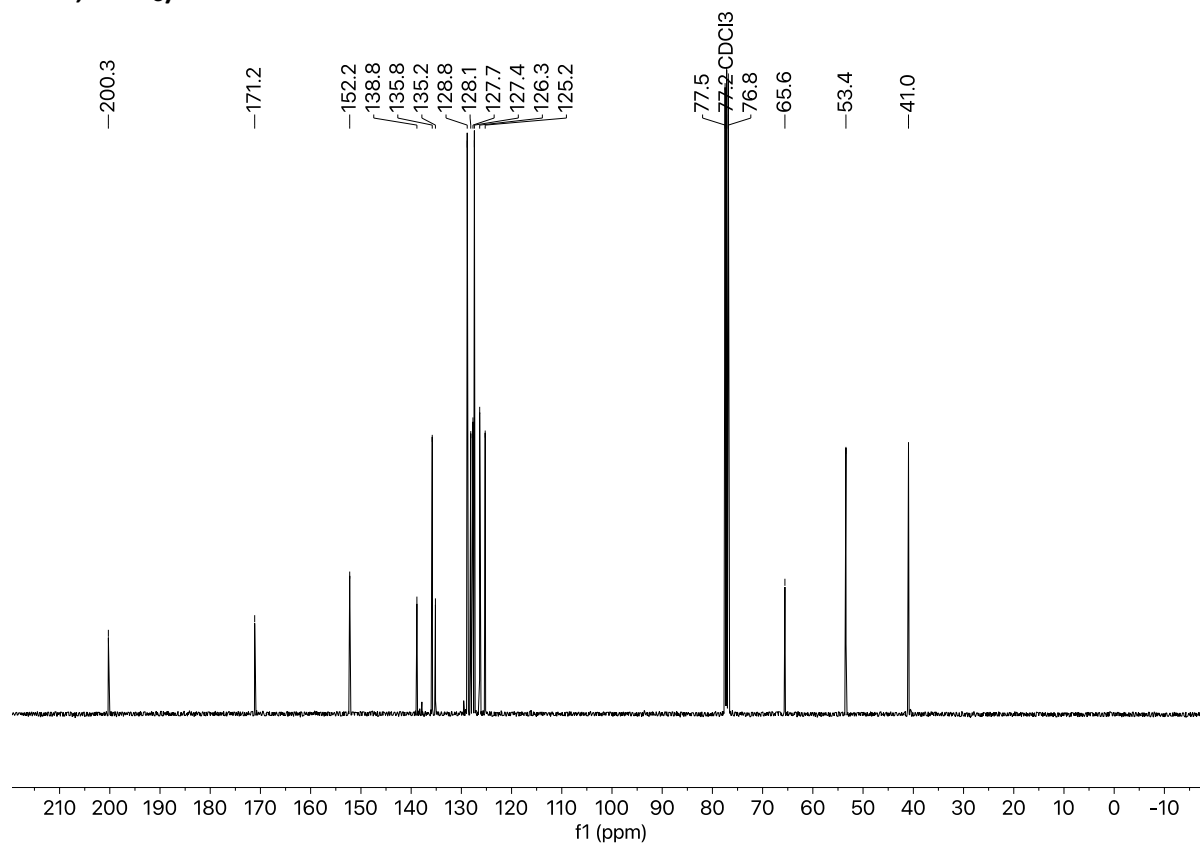
Methyl 1-oxo-2-phenyl-2,3-dihydro-1H-indene-2-carboxylate (5j)



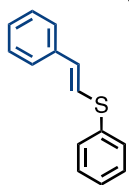
δ_H (400 MHz, $CDCl_3$)



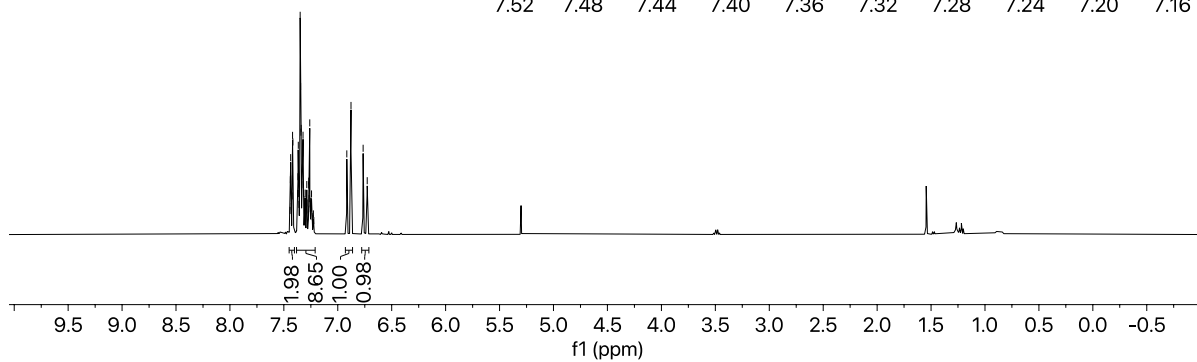
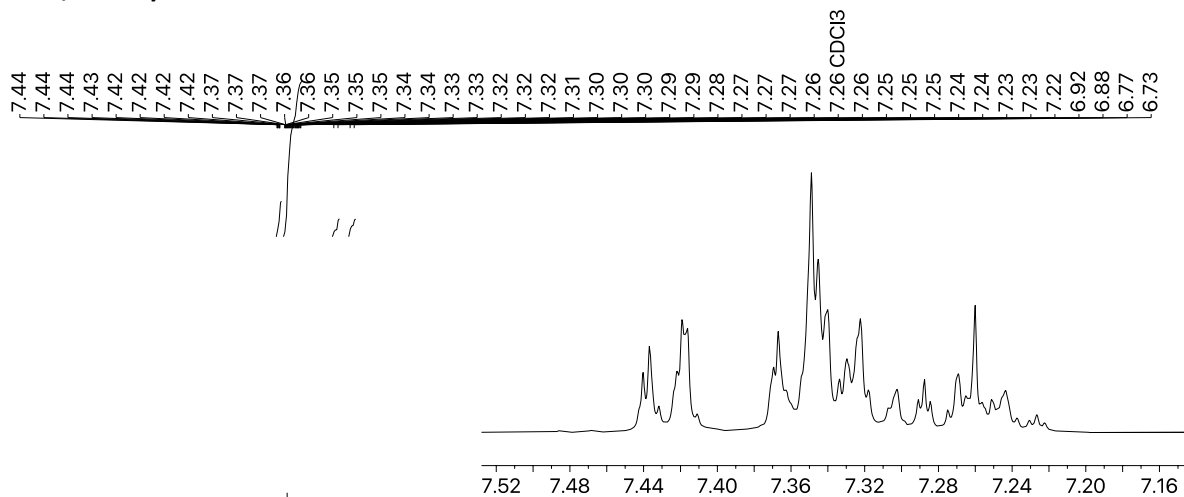
δ_C (101 MHz, $CDCl_3$)



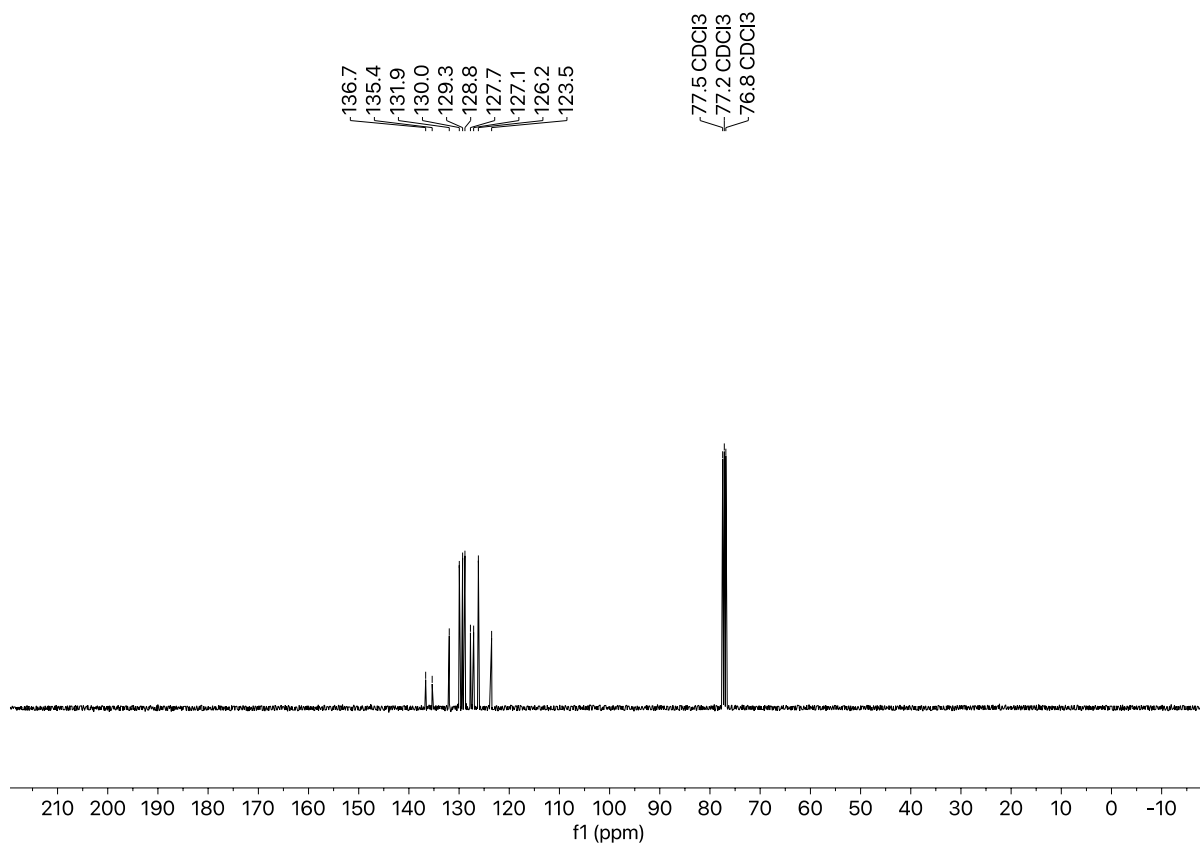
(E)-Phenyl(styryl)sulfane (7a)



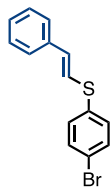
δ_H (400 MHz, $CDCl_3$)



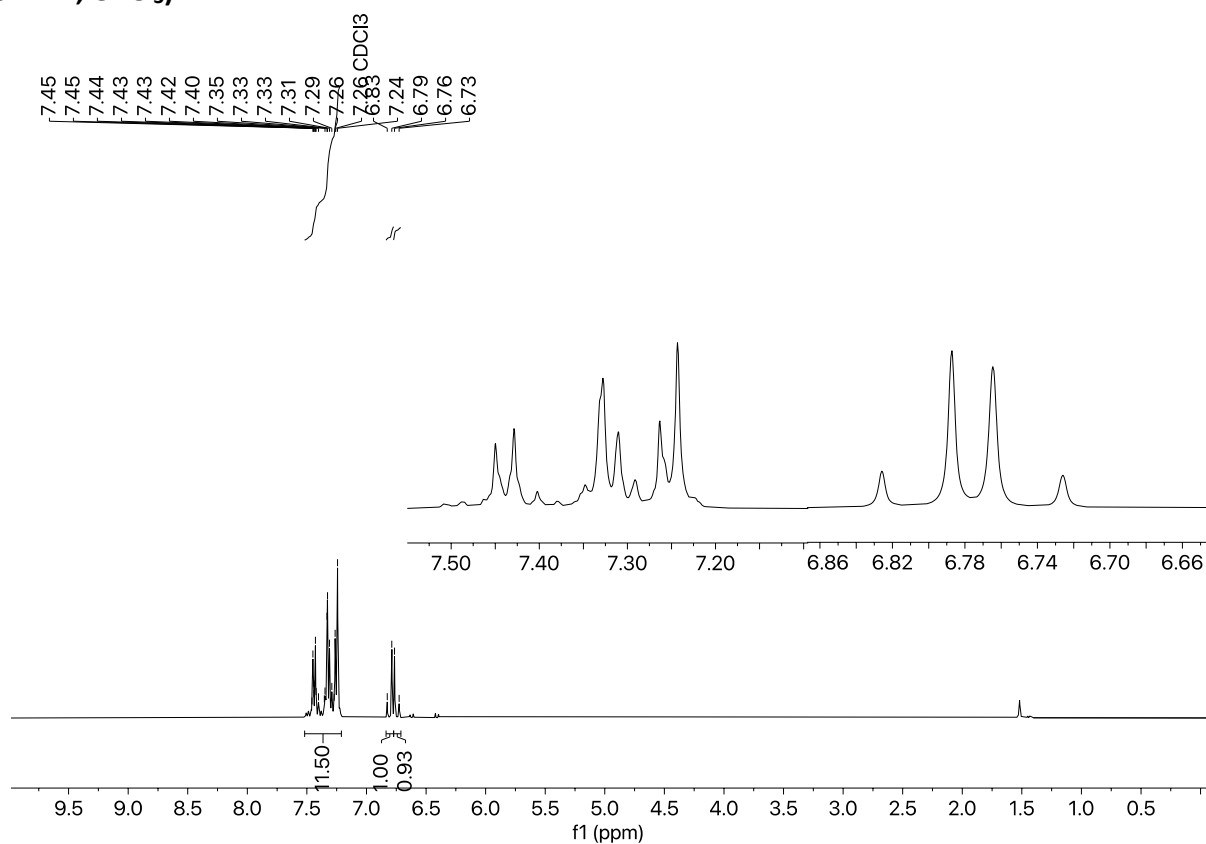
δ_C (101 MHz, $CDCl_3$)



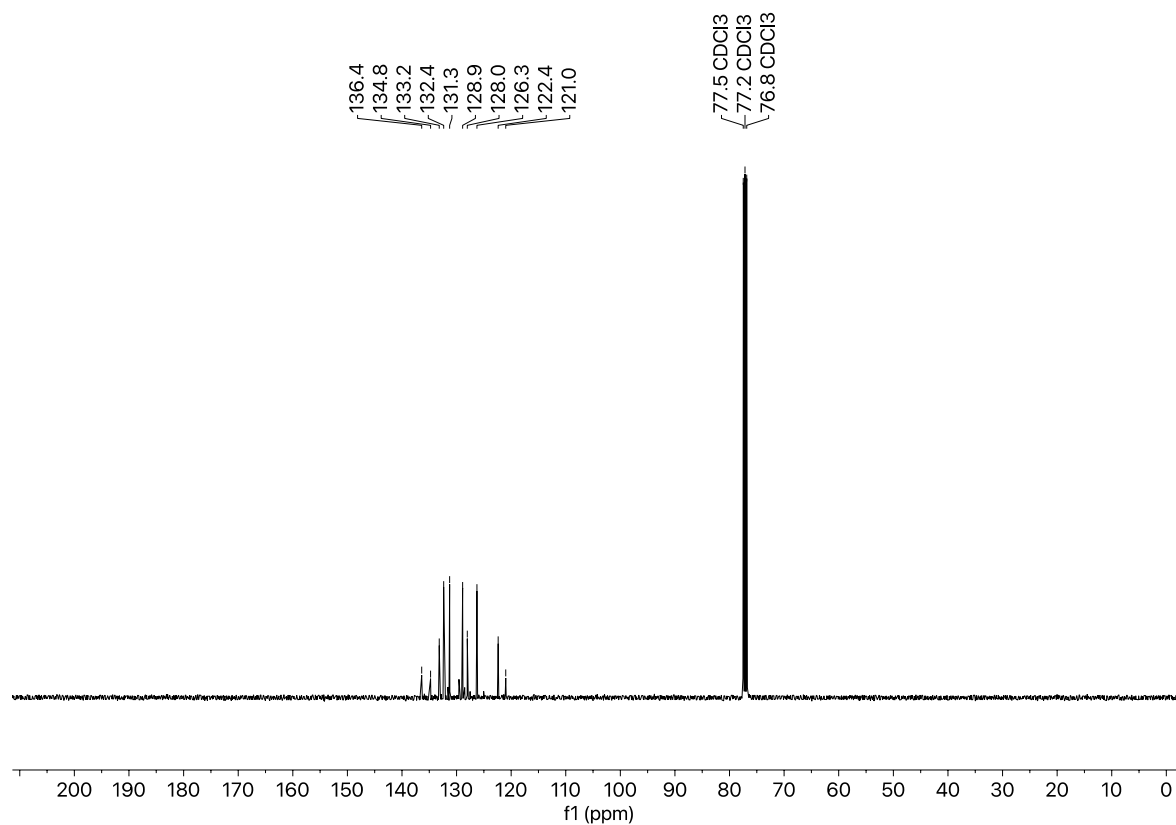
(E)-(4-Bromophenyl)(styryl)sulfane (7b)



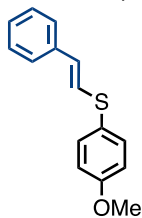
δ_H (400 MHz, $CDCl_3$)



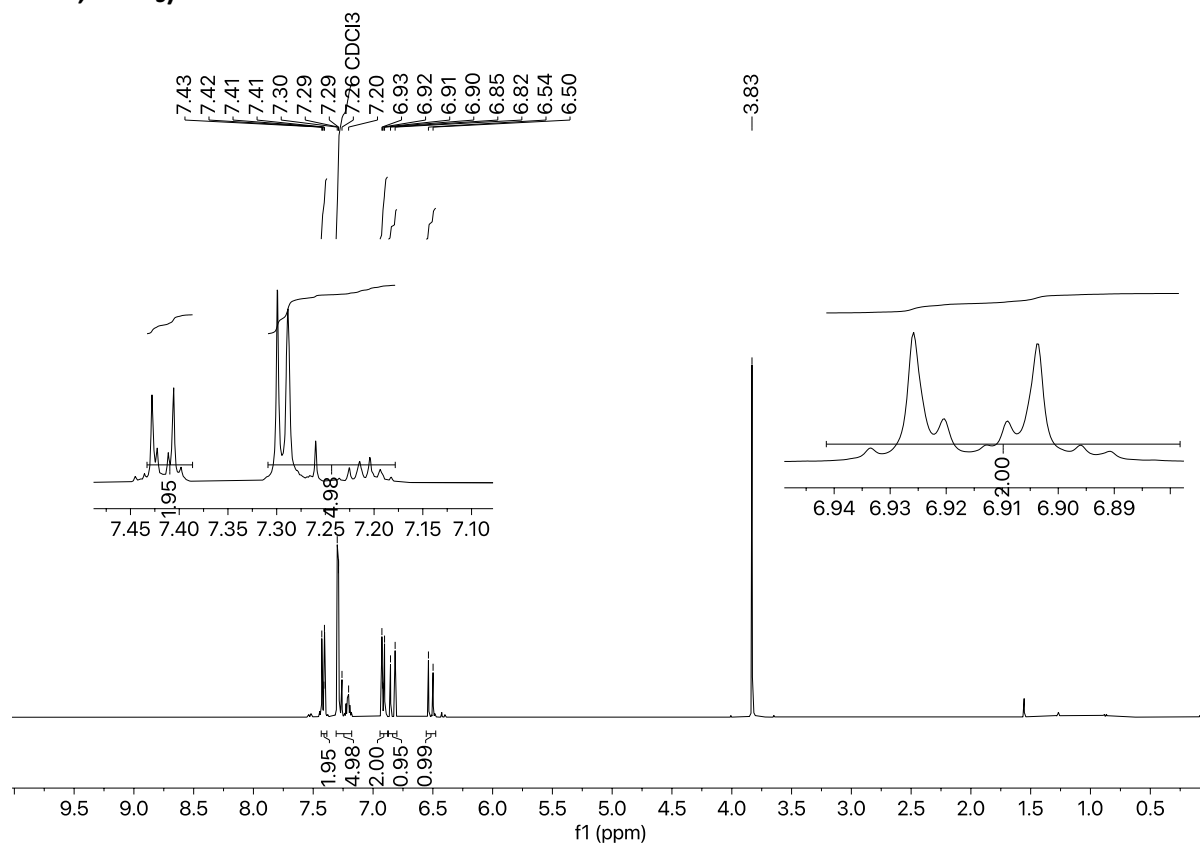
δ_C (101 MHz, $CDCl_3$)



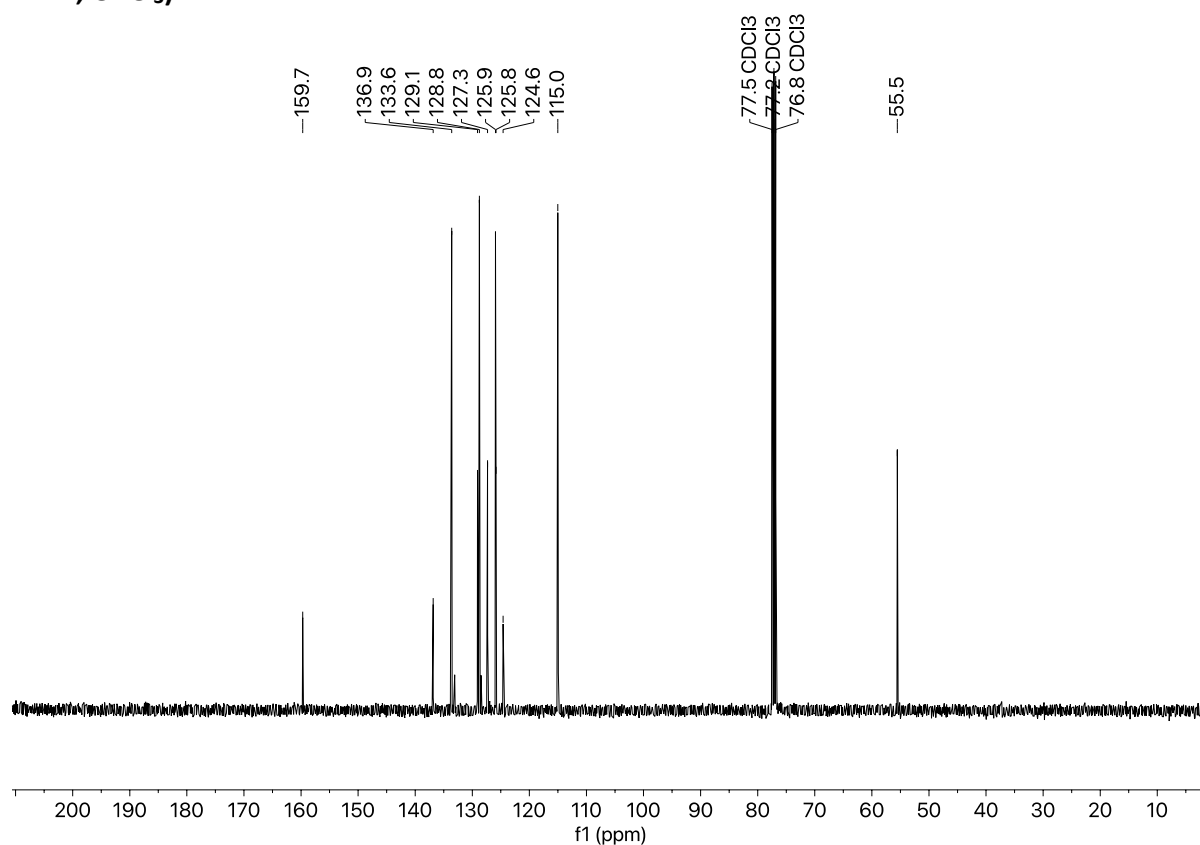
(E)-(4-Methoxyphenyl)(styryl)sulfane (7c)



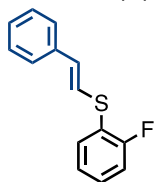
δ_H (400 MHz, $CDCl_3$)



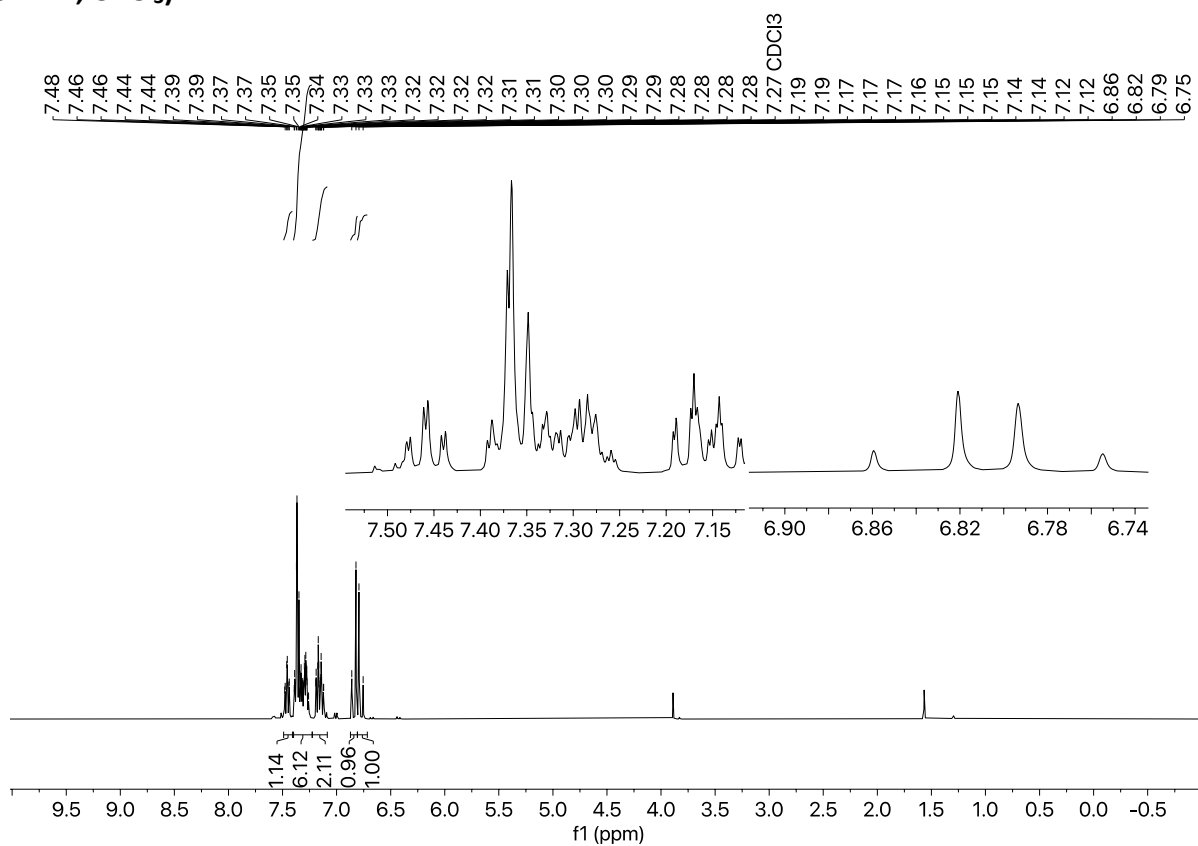
δ_C (101 MHz, $CDCl_3$)



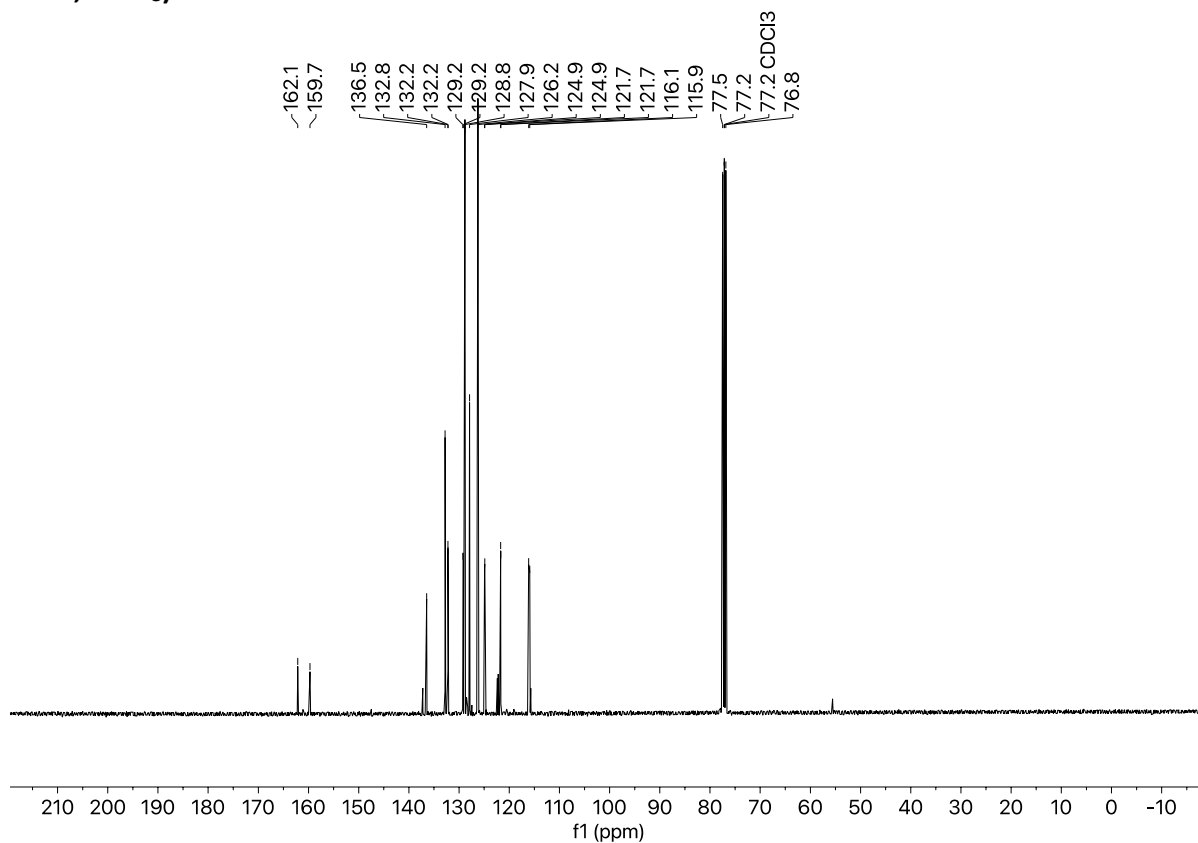
(E)-(2-Fluorophenyl)(styryl)sulfane (7d)



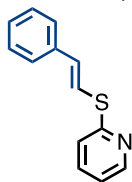
δ_H (400 MHz, CDCl₃)



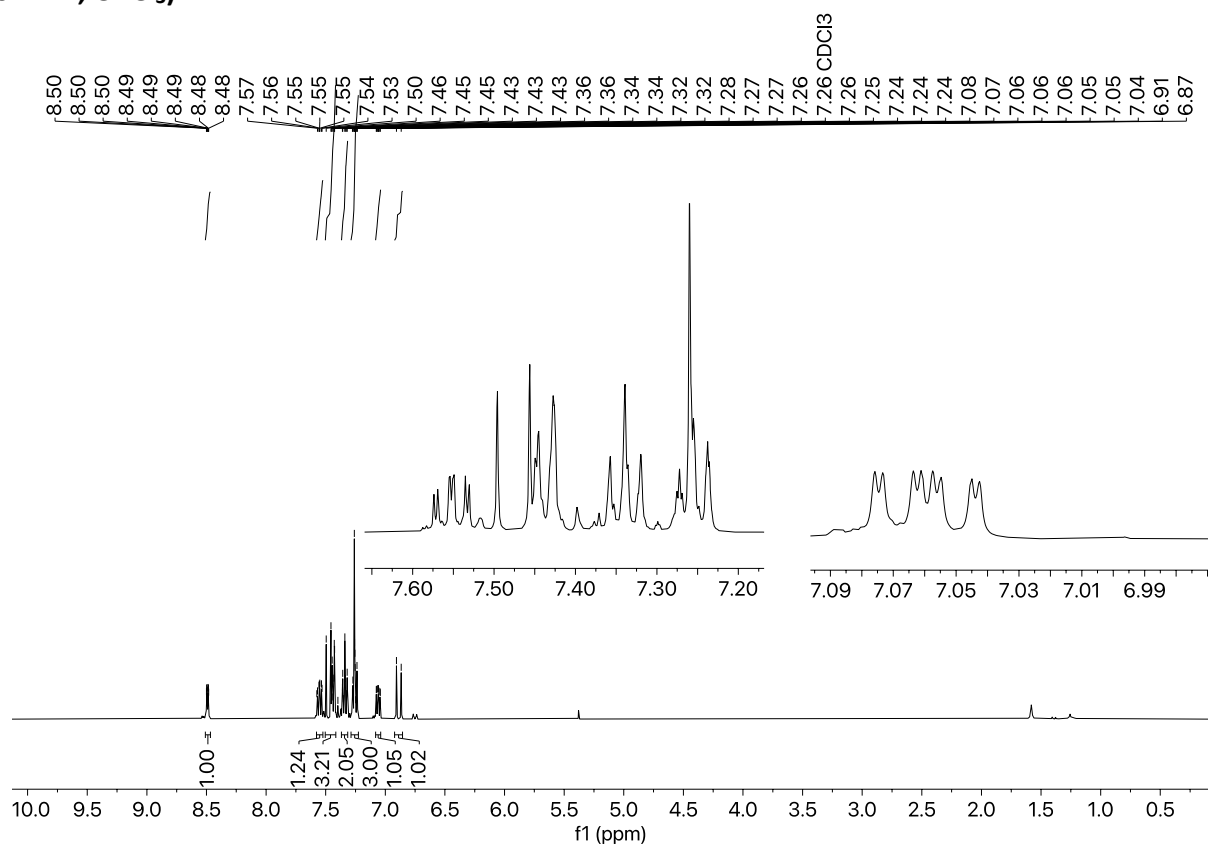
δ_C (101 MHz, CDCl₃)



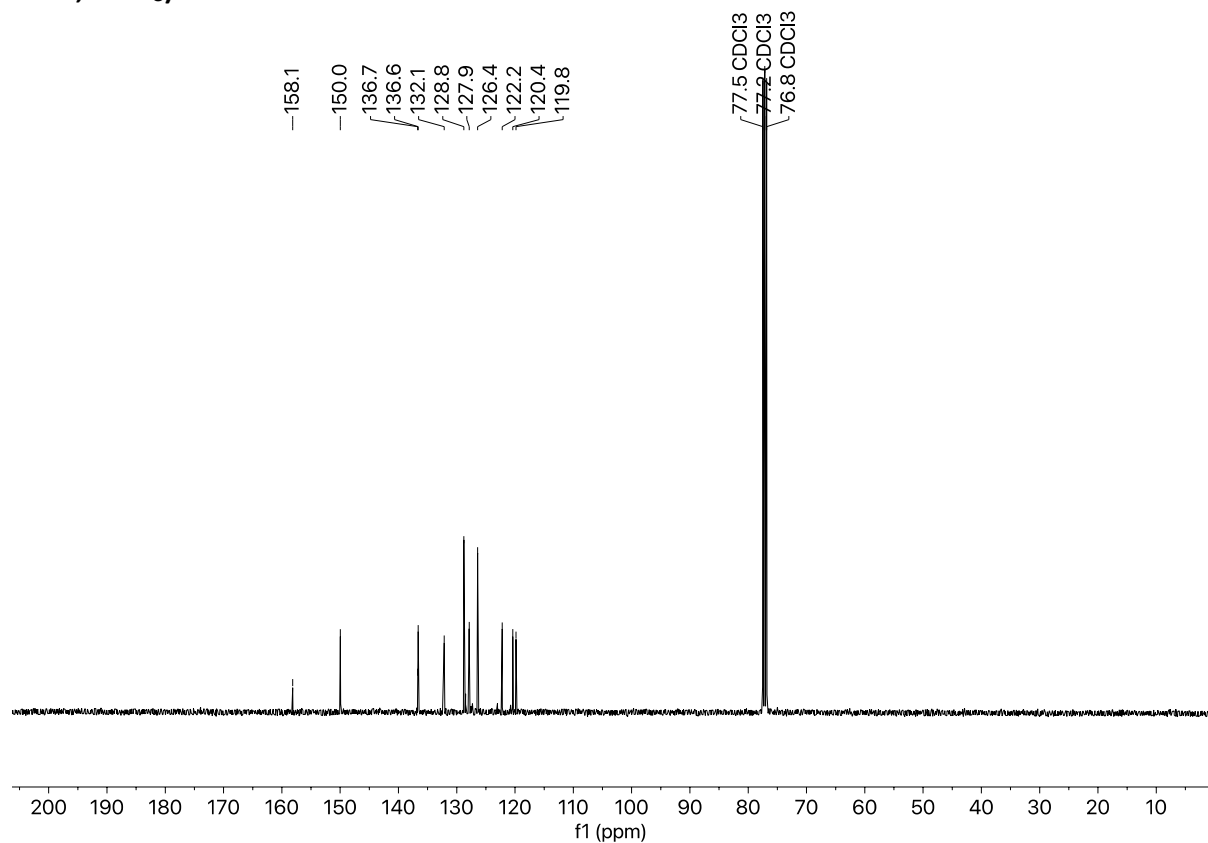
(E)-2-(Styrylthio)pyridine (7e)



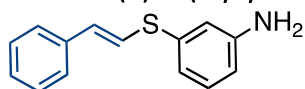
δ_H (400 MHz, CDCl₃)



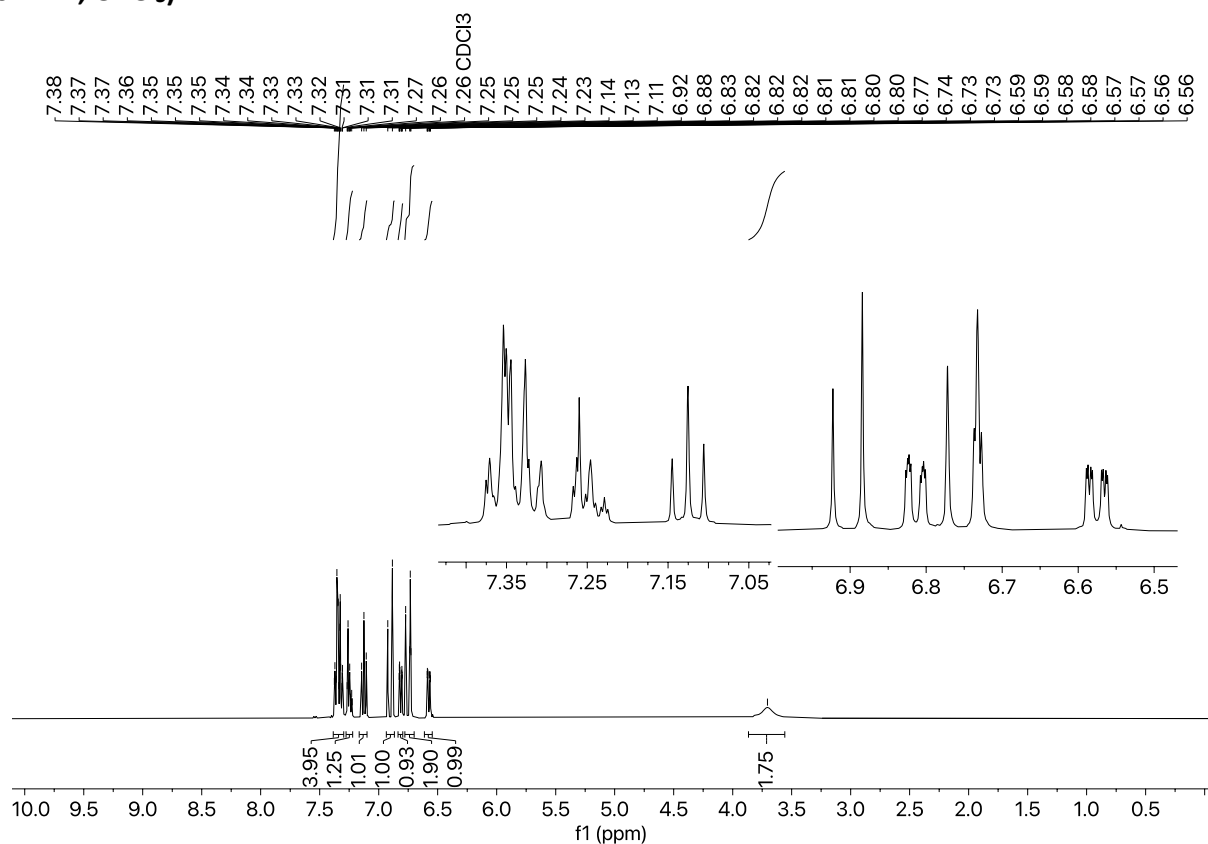
δ_C (101 MHz, CDCl₃)



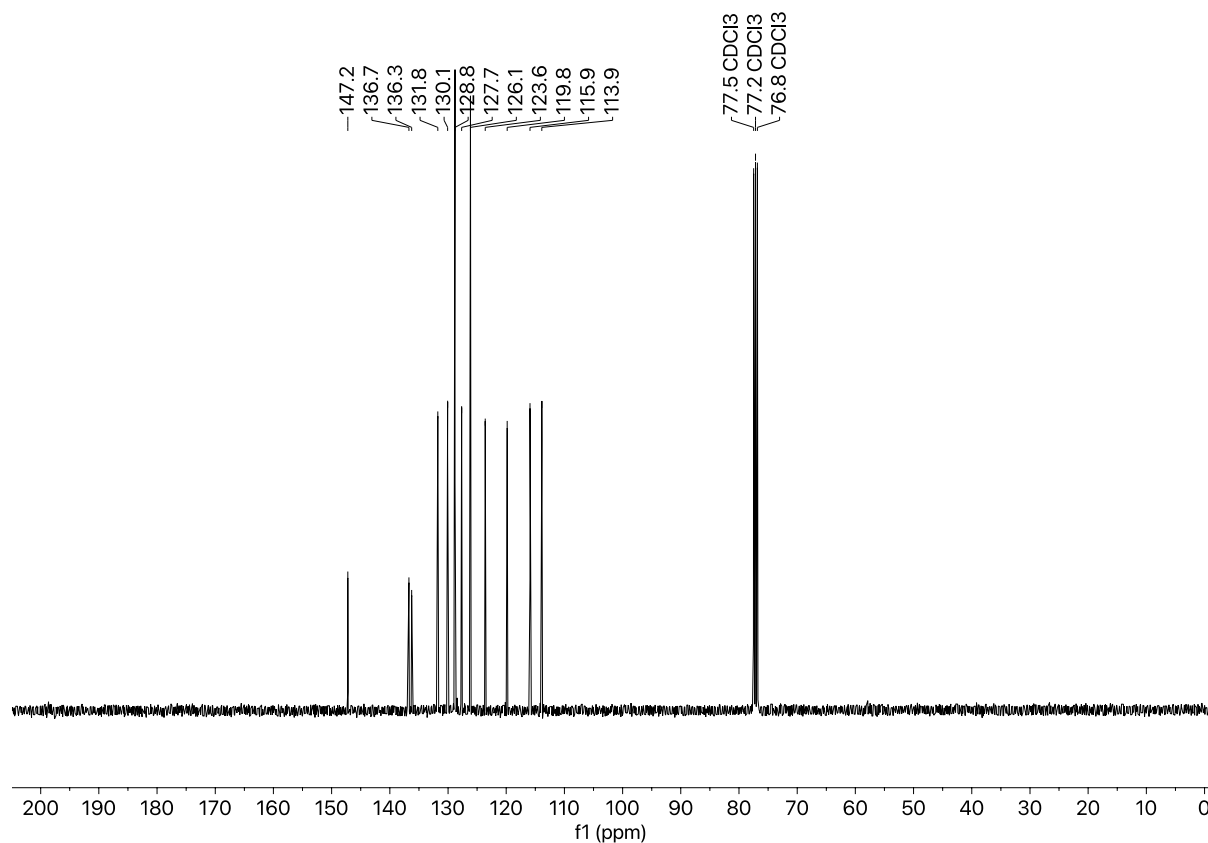
(E)-3-(styrylthio)aniline (7f)



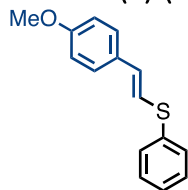
δ_H (400 MHz, CDCl₃)



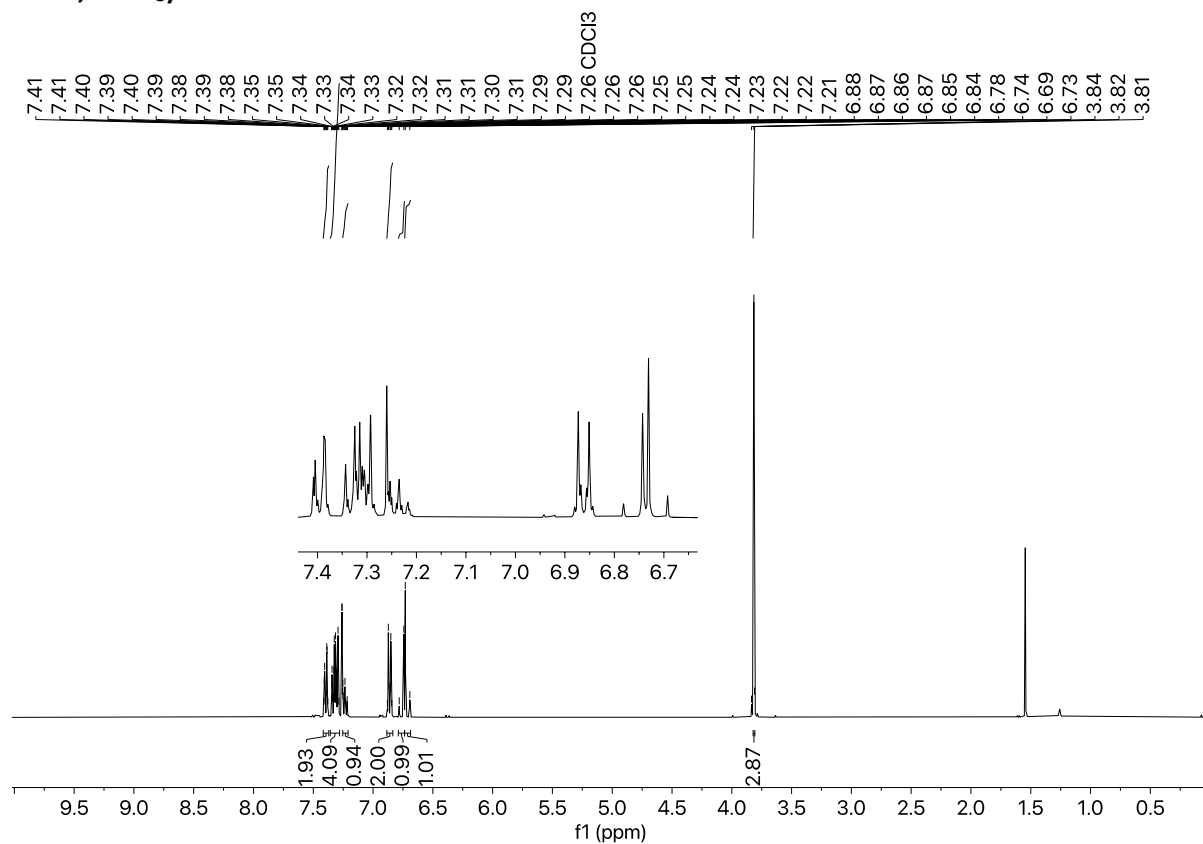
δ_C (101 MHz, CDCl₃)



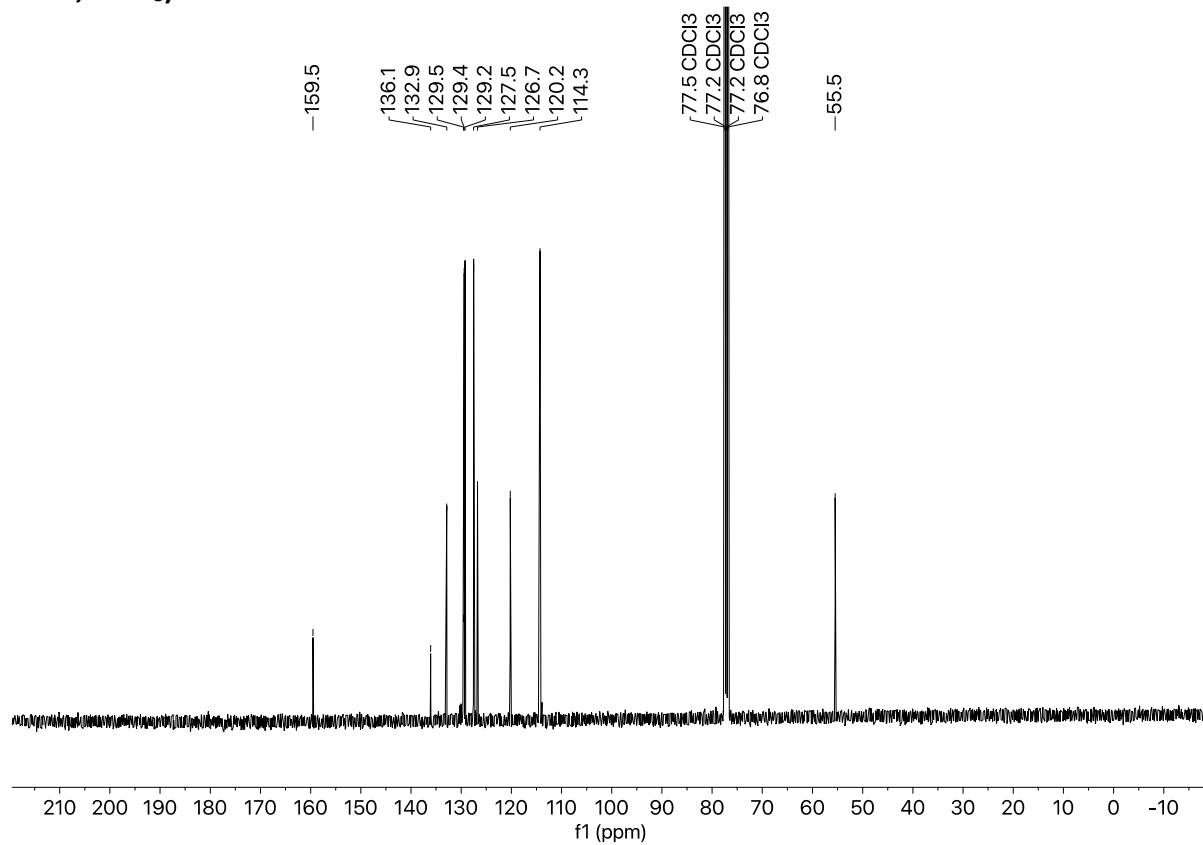
(E)-(4-Methoxystyryl)(phenyl)sulfane (7g)



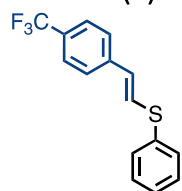
δ_H (400 MHz, $CDCl_3$)



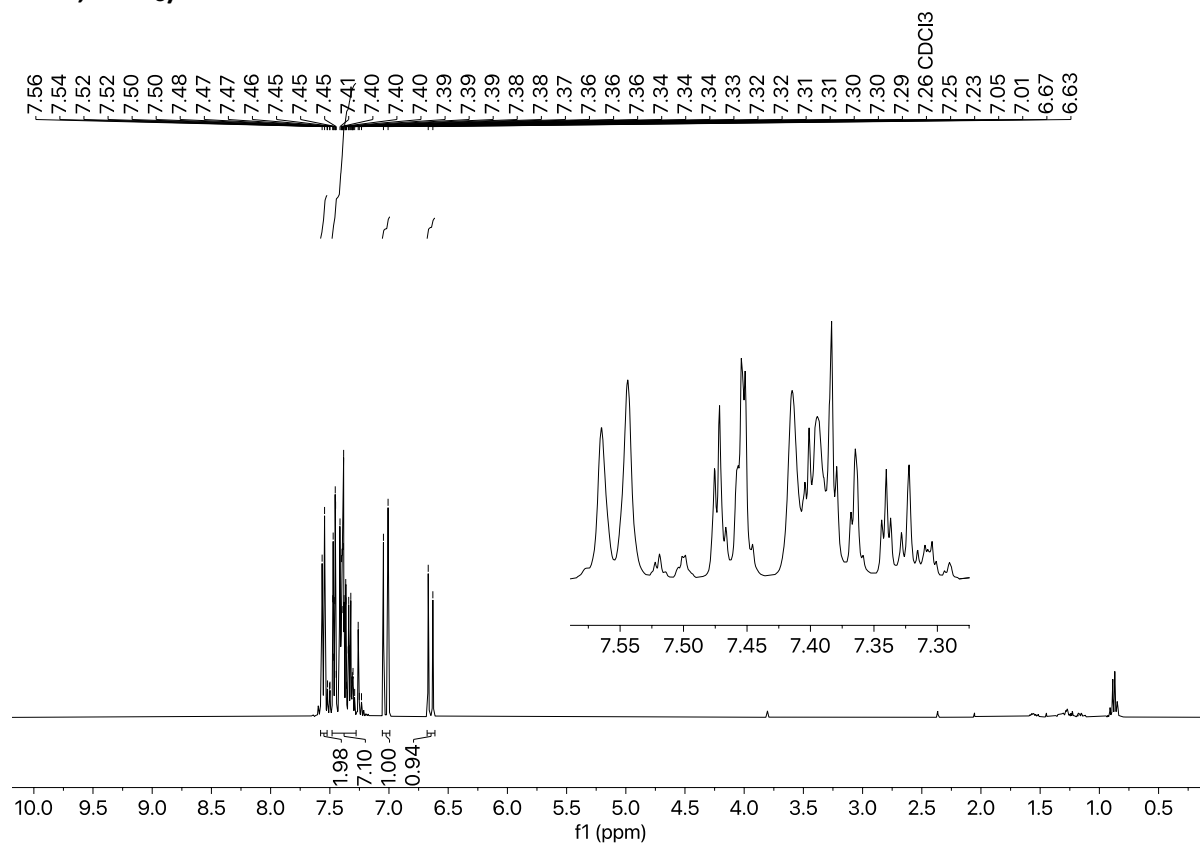
δ_C (101 MHz, $CDCl_3$)



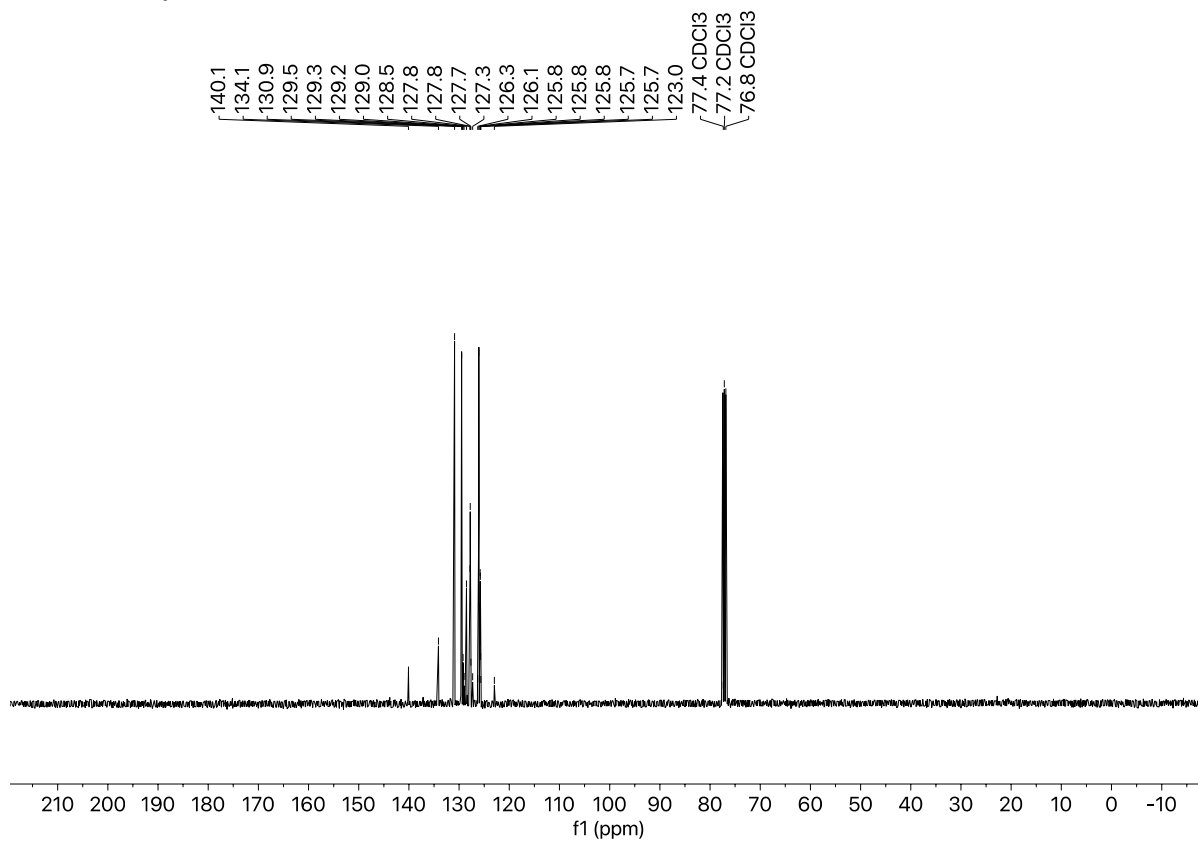
(E)-Phenyl(4-(trifluoromethyl)styryl)sulfane (7h)



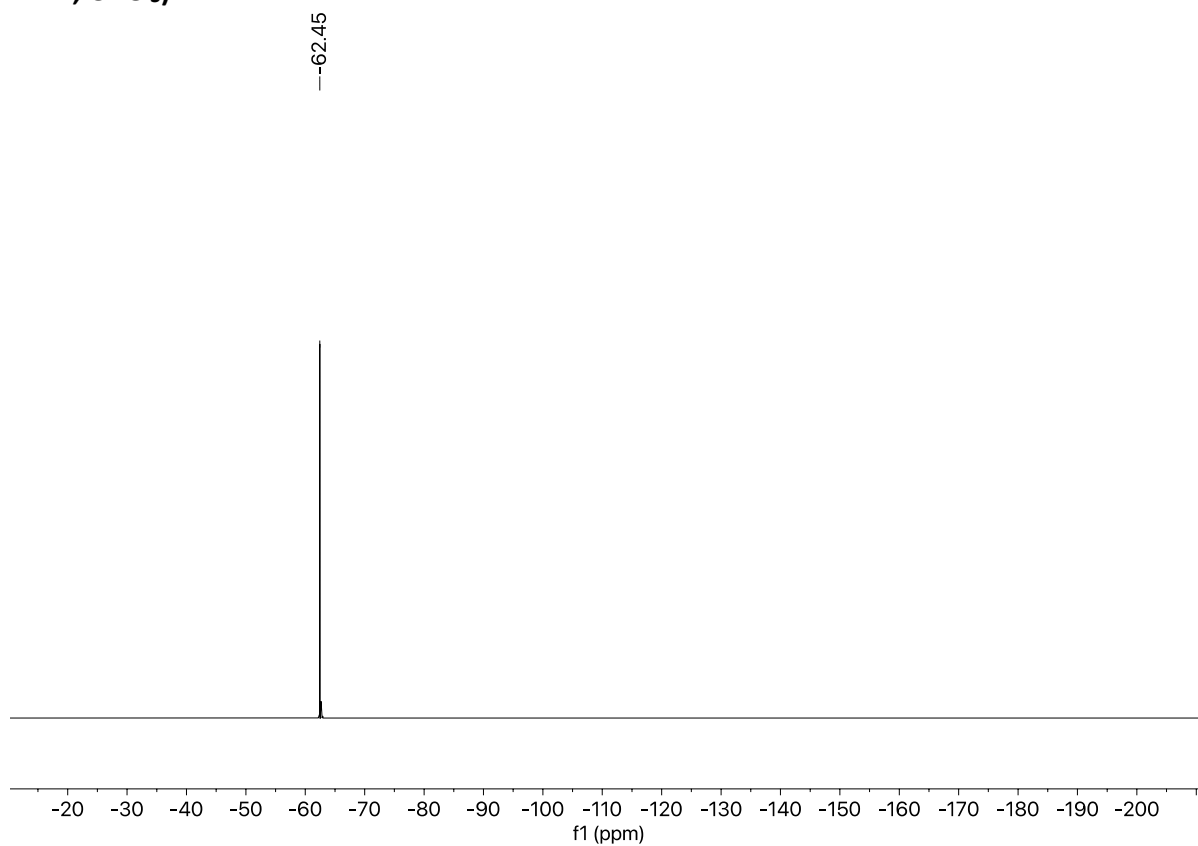
δ_H (400 MHz, CDCl₃)



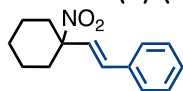
δ_c (101 MHz, $CDCl_3$)



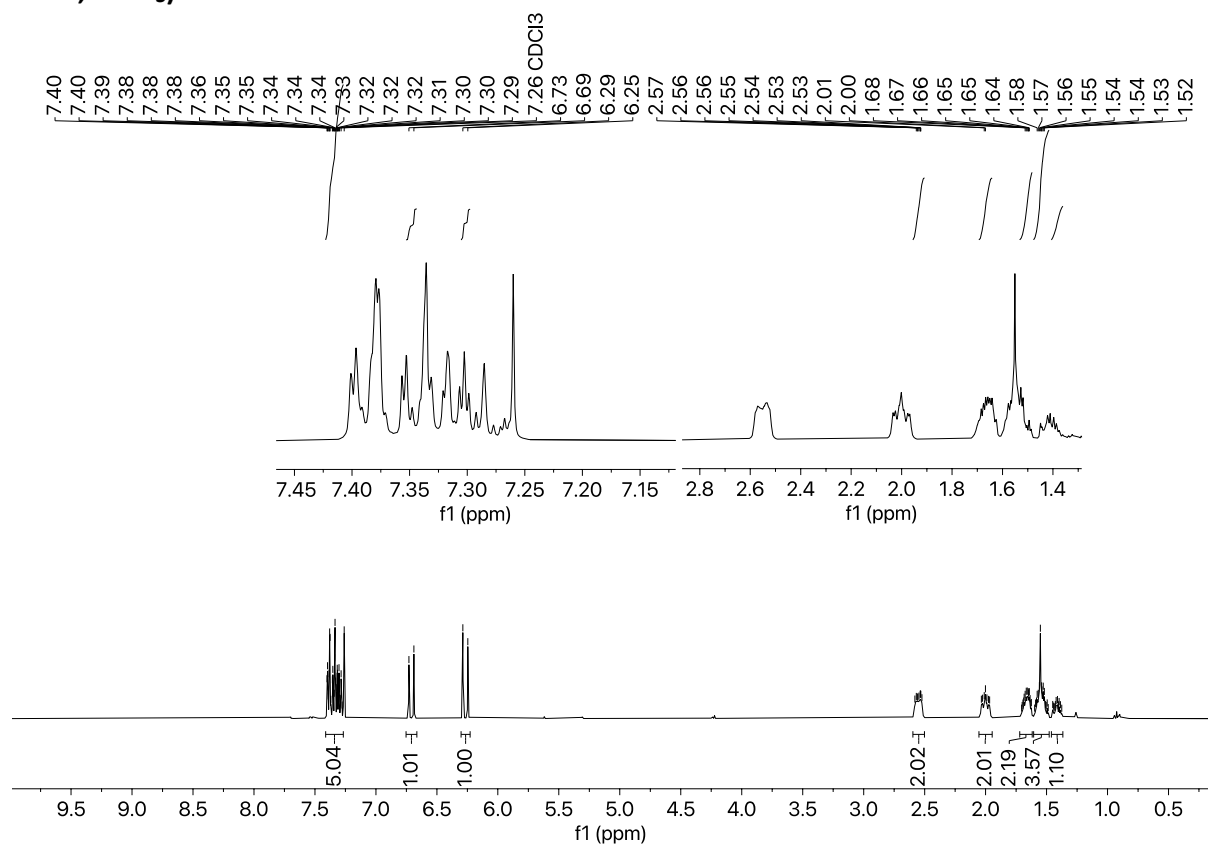
δ_F (376 MHz, $CDCl_3$)



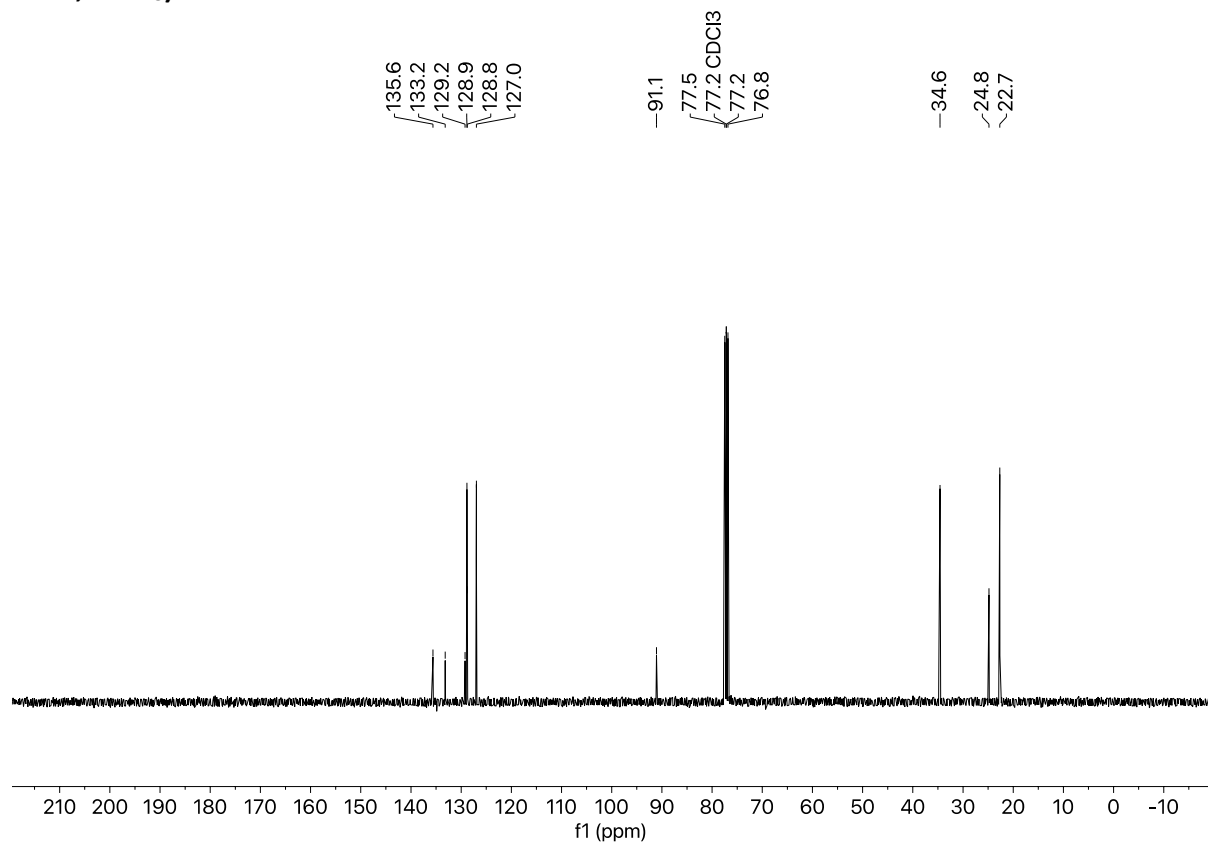
(E)-2-(1-nitrocyclohexyl)vinyl)benzene (8a)



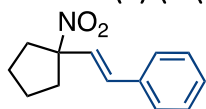
δ_H (400 MHz, CDCl₃)



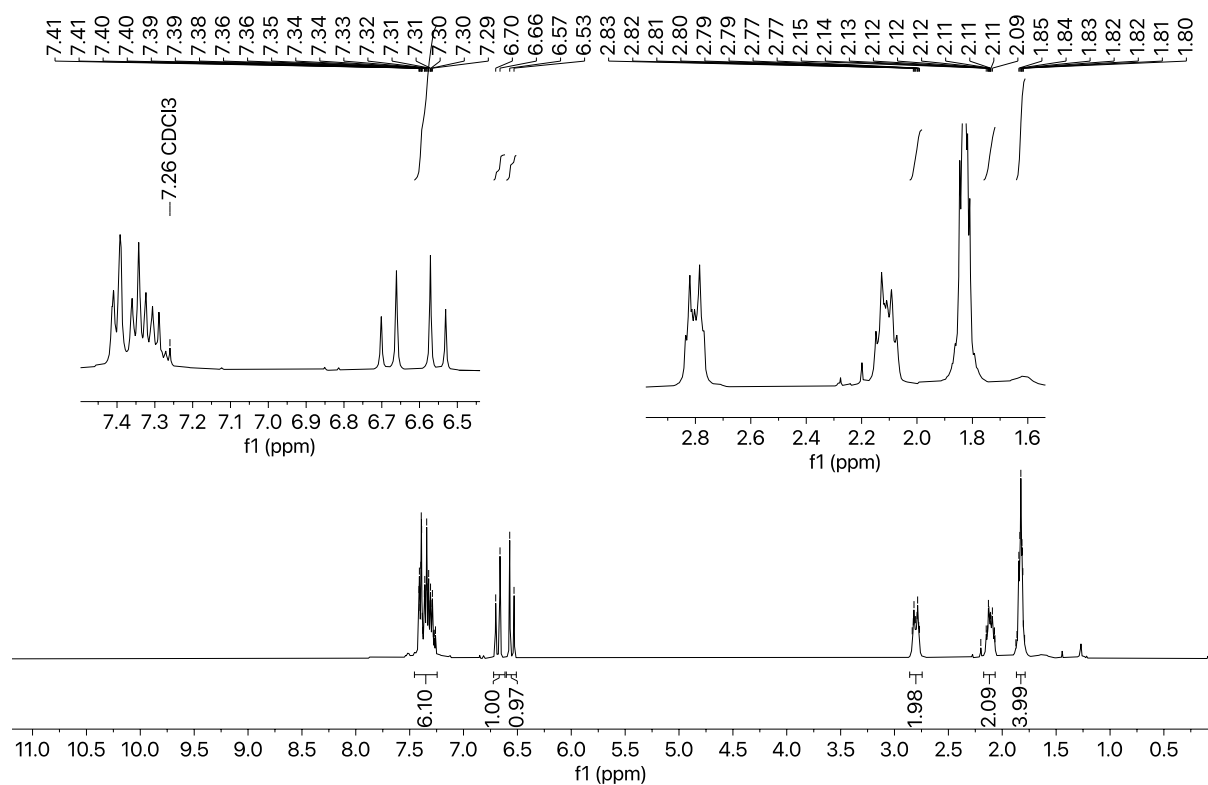
δ_C (101 MHz, CDCl₃)



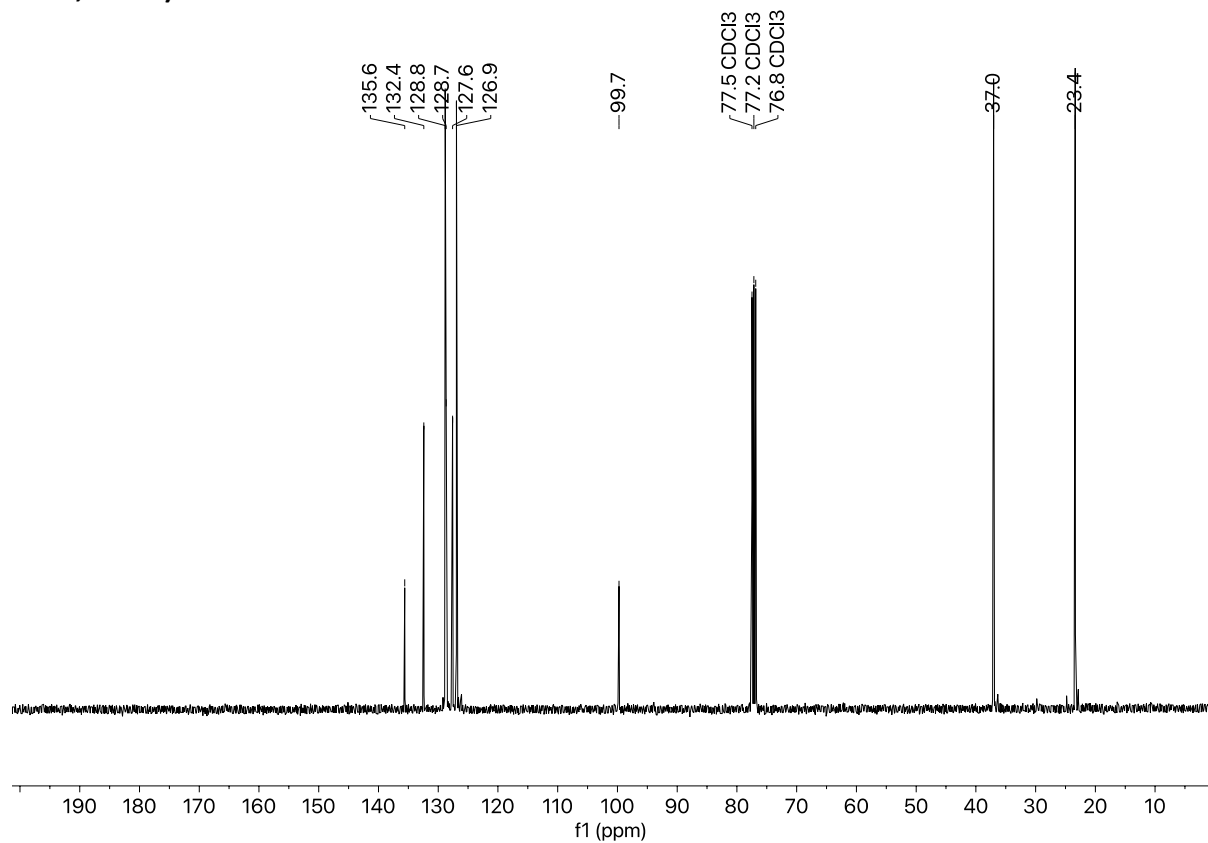
(E)-2-(1-nitrocyclopentyl)vinyl)benzene (8b)



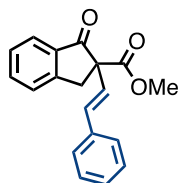
δ_H (400 MHz, $CDCl_3$)



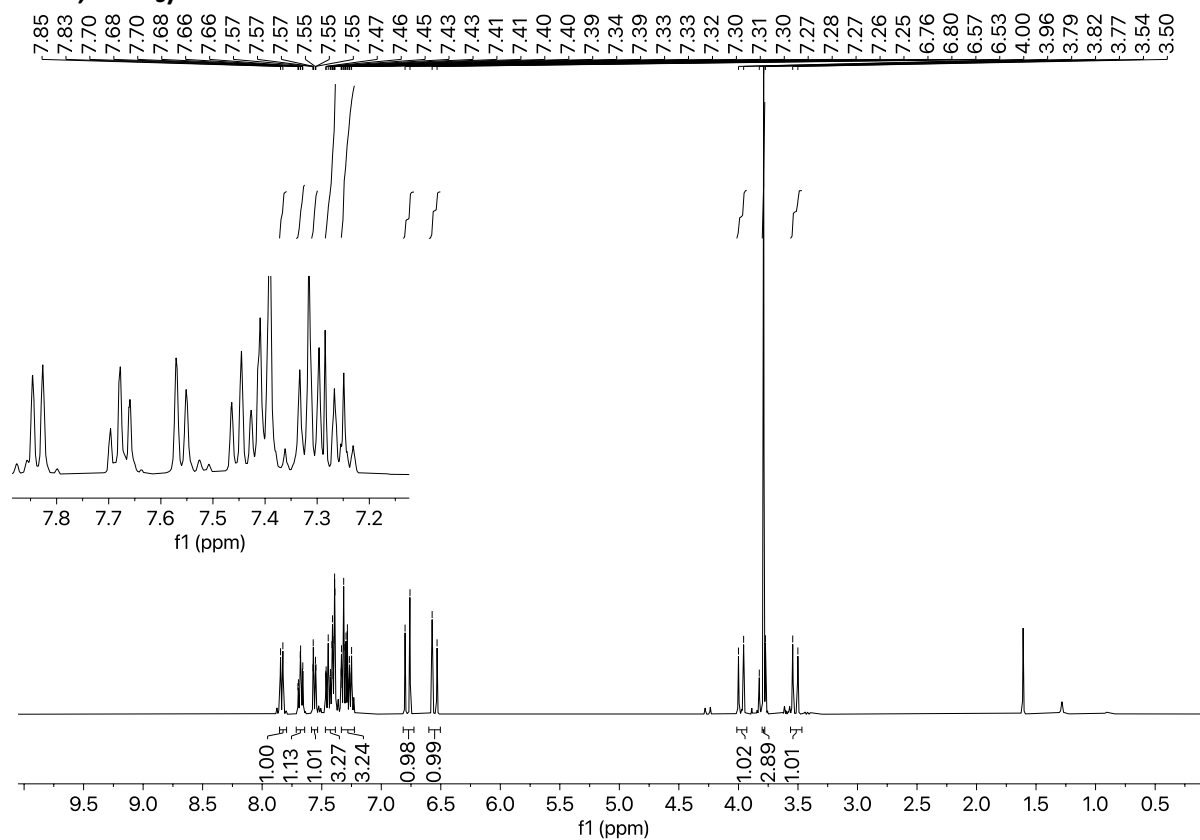
δ_C (101 MHz, $CDCl_3$)



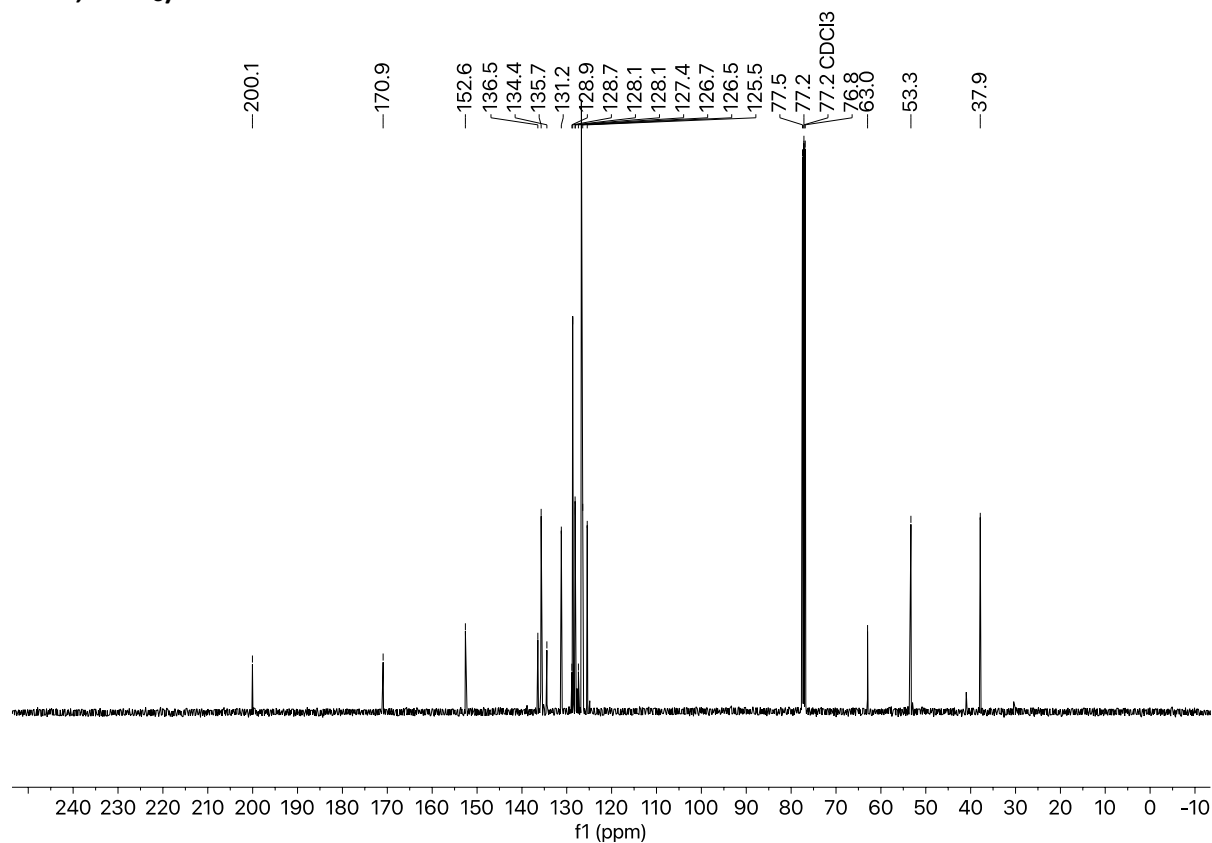
Methyl (*E*)-1-oxo-2-styryl-2,3-dihydro-1*H*-indene-2-carboxylate (**8c**)



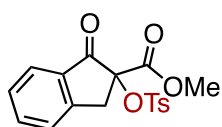
δ_H (400 MHz, $CDCl_3$)



δ_C (101 MHz, $CDCl_3$)

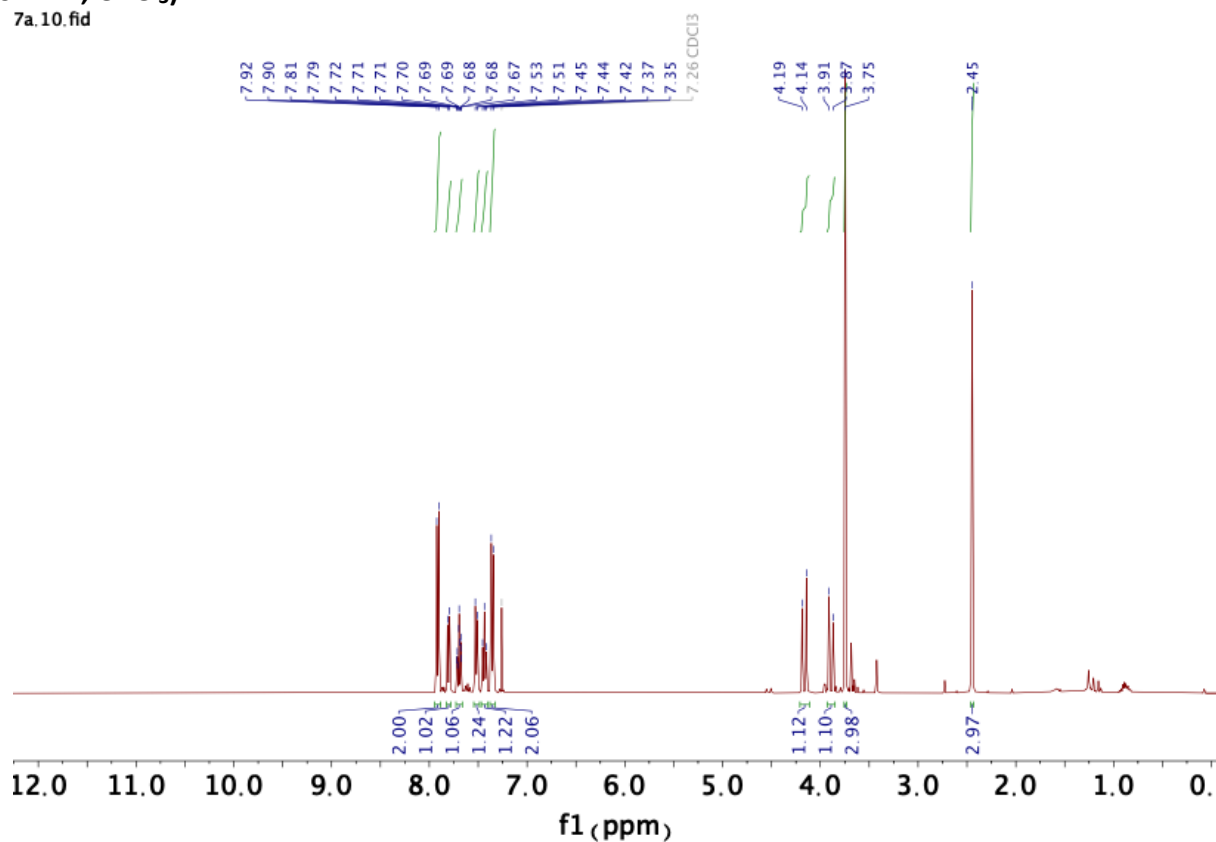


Methyl 1-oxo-2-(tosyloxy)-2,3-dihydro-1H-indene-2-carboxylate (10a)



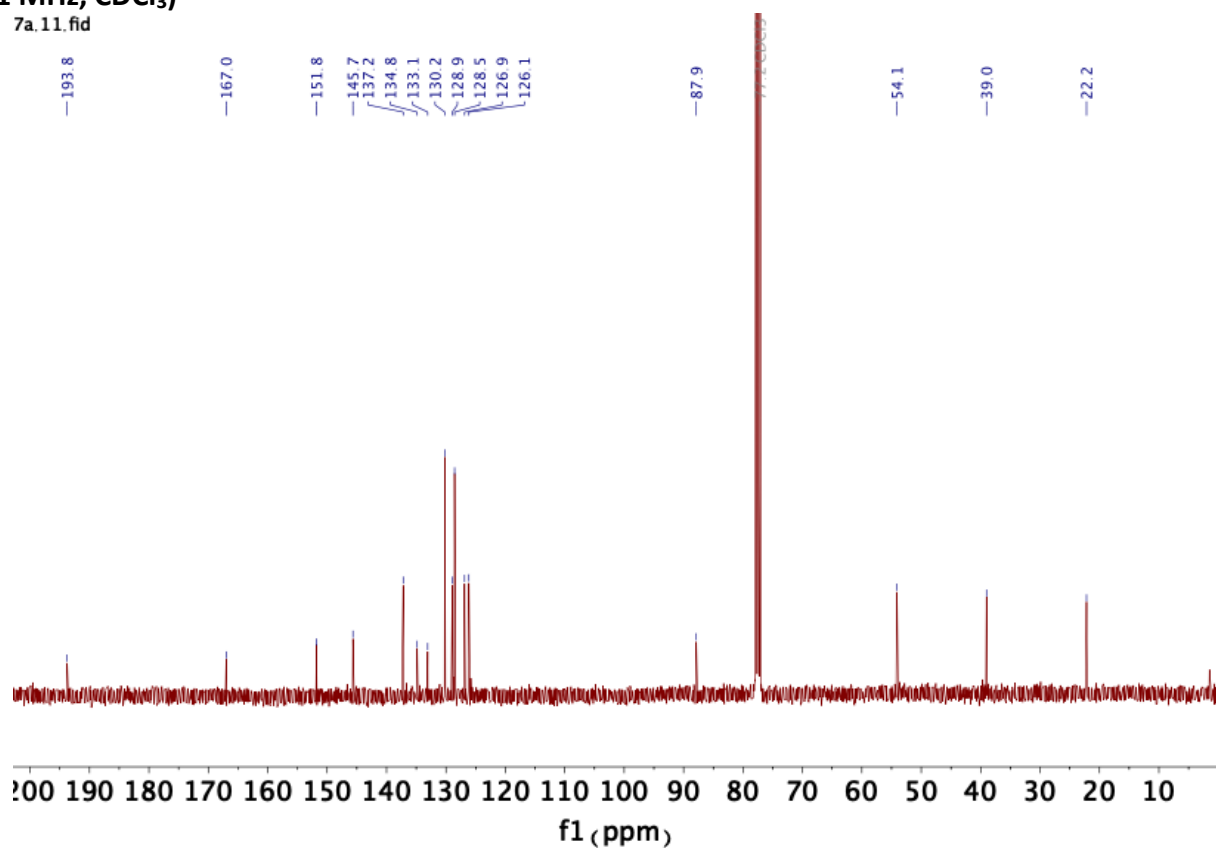
δ_H (400 MHz, $CDCl_3$)

7a.10.fid

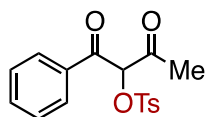


δ_C (101 MHz, $CDCl_3$)

7a.11.fid

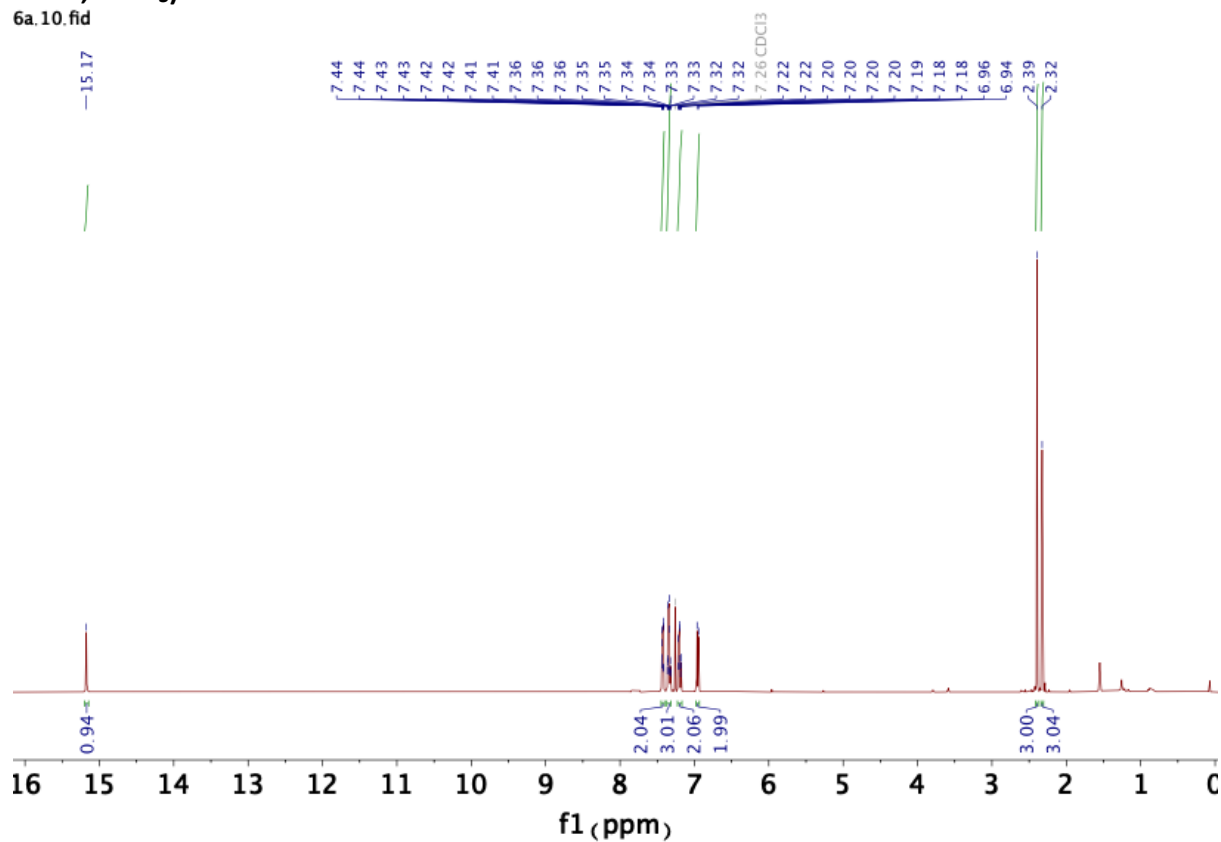


1,3-Dioxo-1-phenylbutan-2-yl 4-methylbenzenesulfonate (10b)



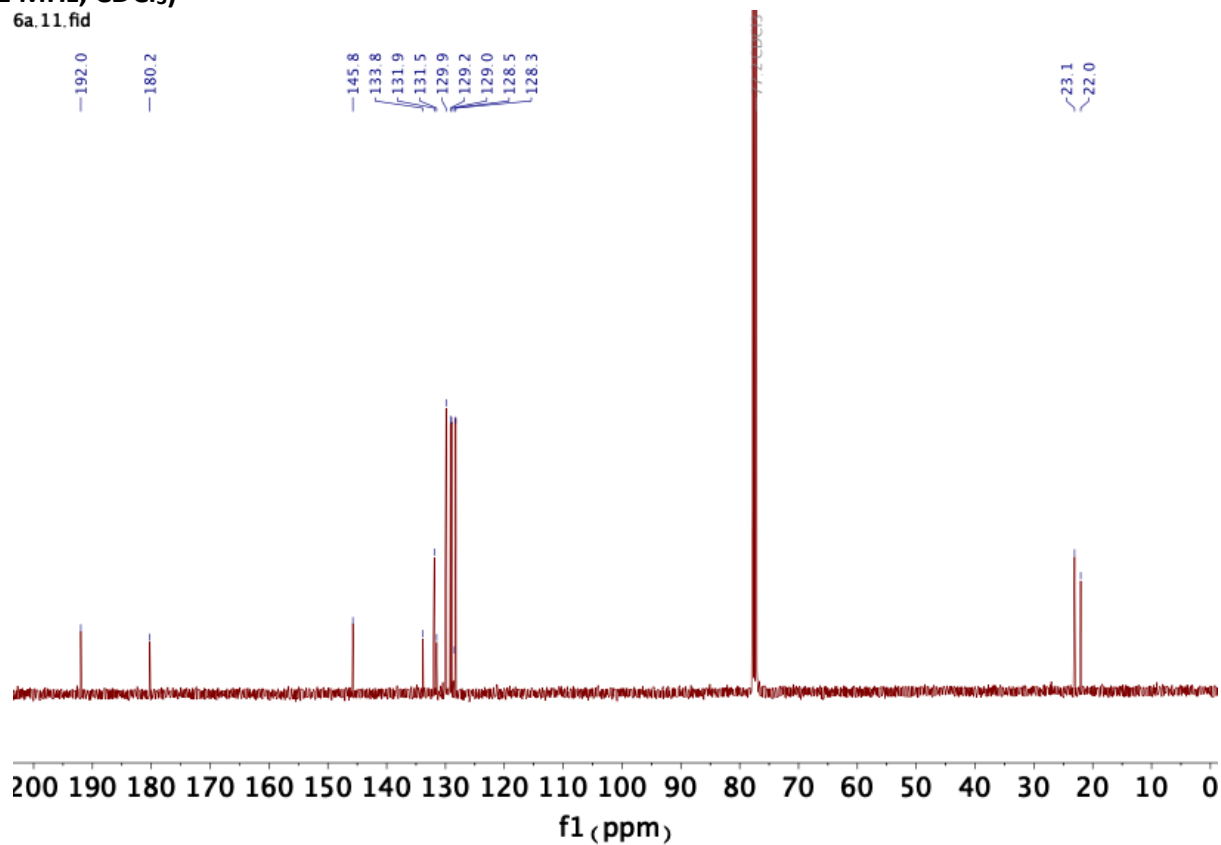
δ_H (400 MHz, $CDCl_3$)

6a.10.fid

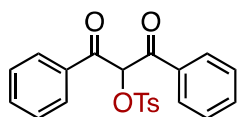


δ_C (101 MHz, $CDCl_3$)

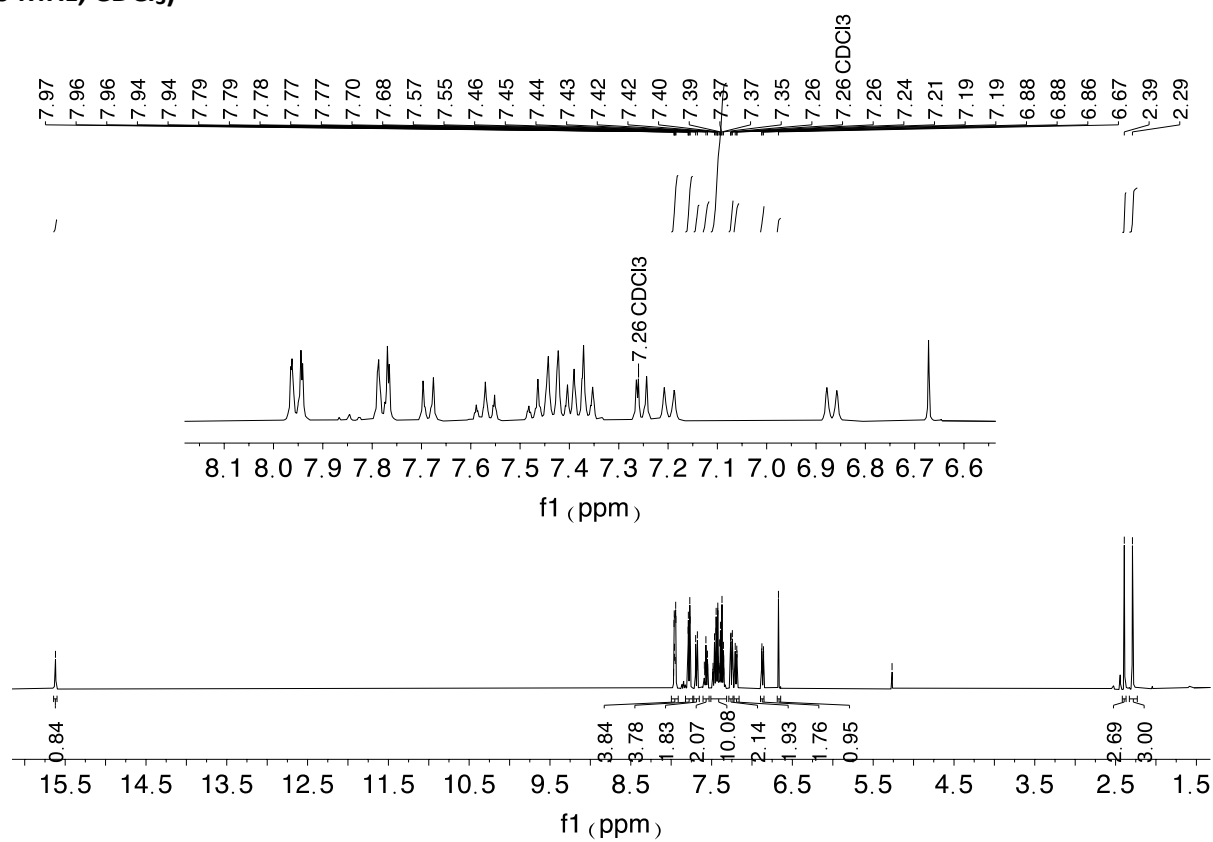
6a.11.fid



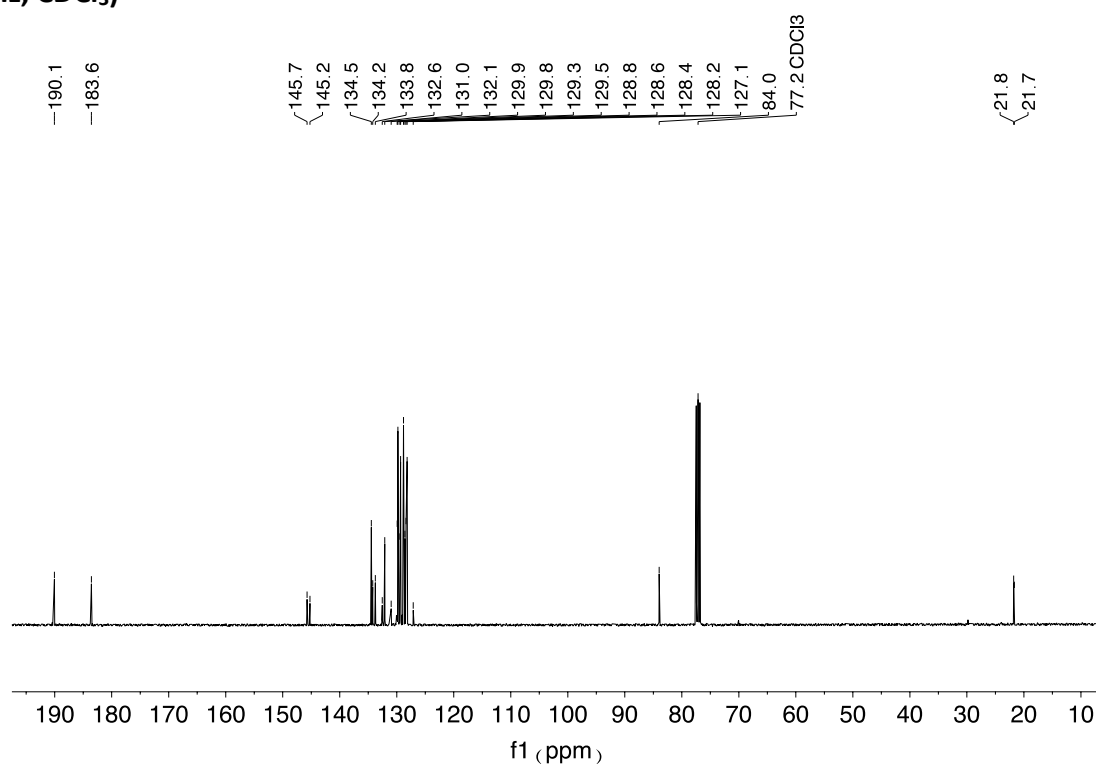
Tosyloxidibenzoylmethane (10c)



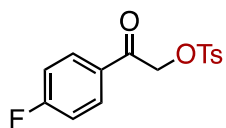
δ_H (400 MHz, $CDCl_3$)



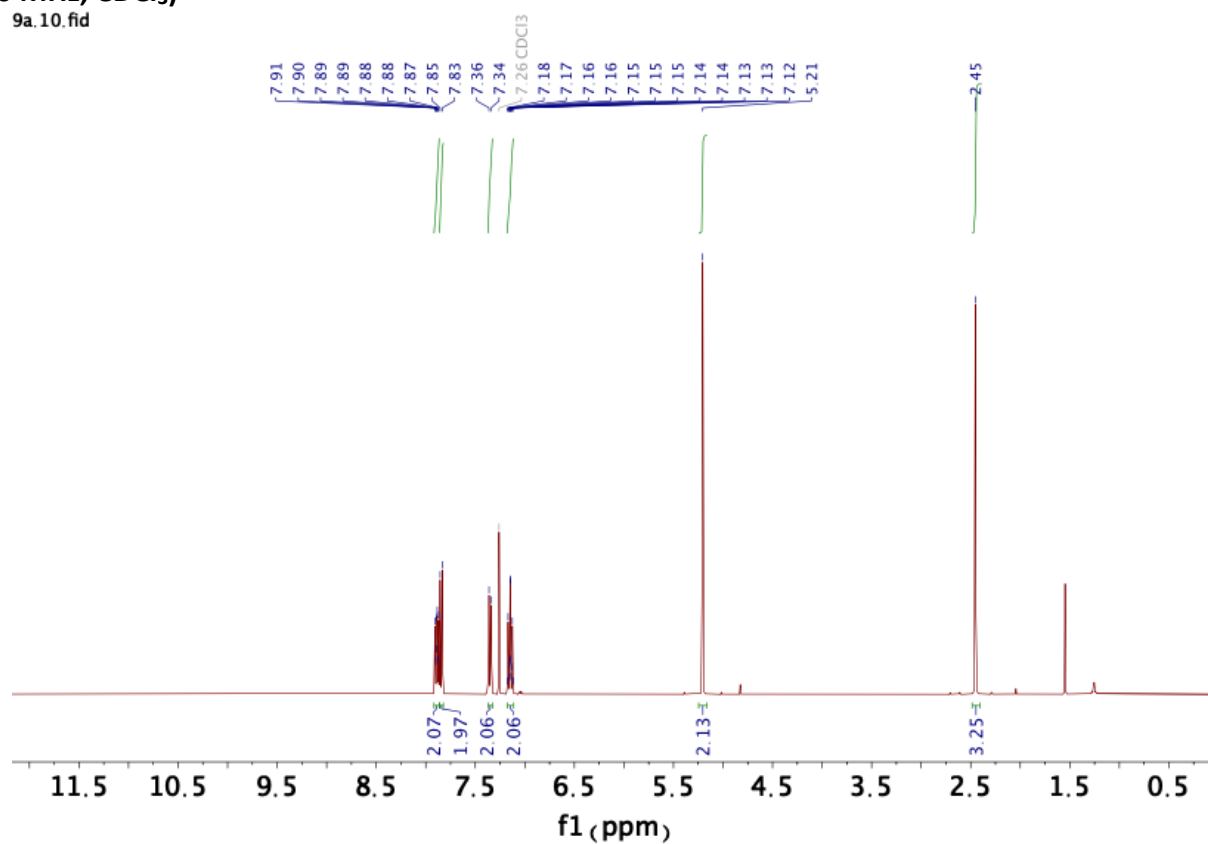
δ_C (101 MHz, $CDCl_3$)



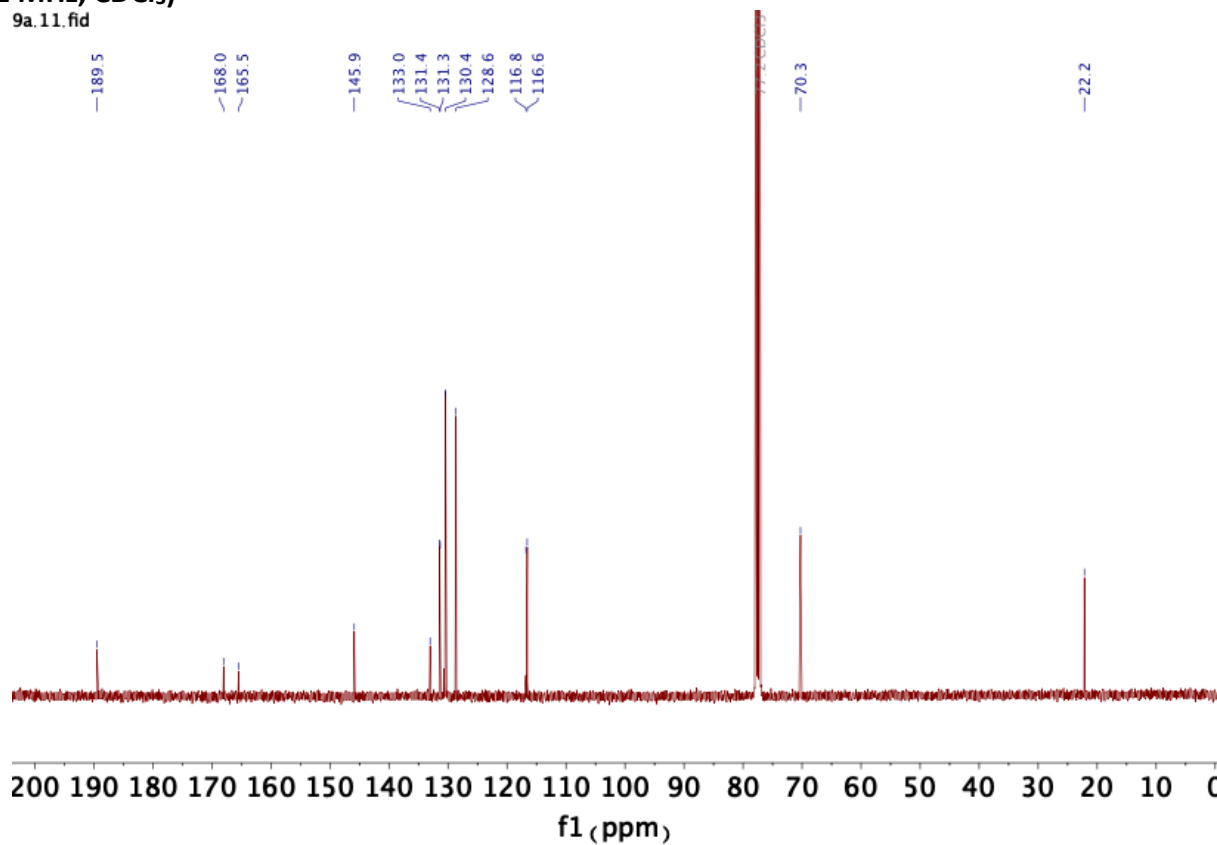
1-(4-fluorophenyl)-2-(p-tolylsulfonyloxy)ethanone (10d)



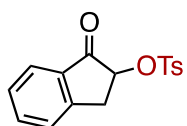
δ_H (400 MHz, $CDCl_3$)
9a.10.fid



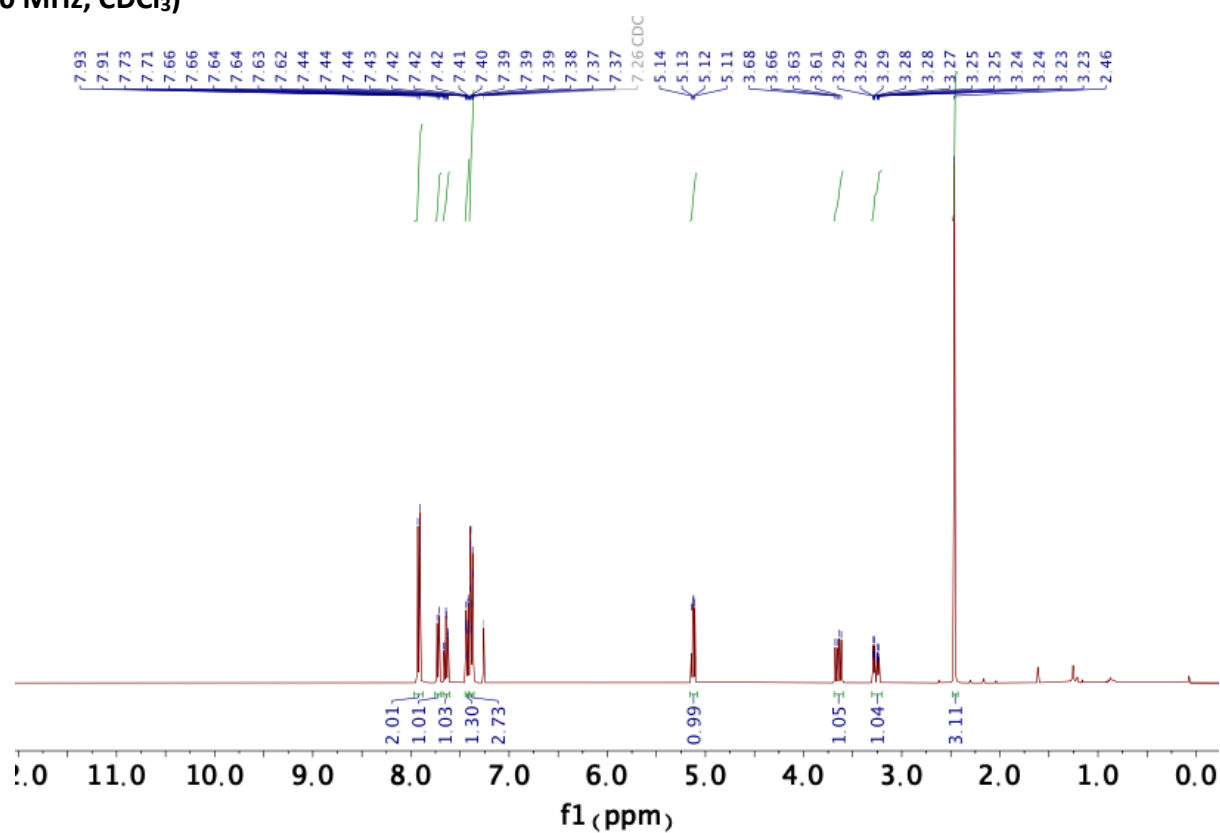
δ_C (101 MHz, $CDCl_3$)
9a.11.fid



1-oxo-2,3-dihydro-1H-inden-2-yl 4-methylbenzenesulfonate (10e)



δ_H (400 MHz, $CDCl_3$)



δ_C (101 MHz, $CDCl_3$)

