

Supporting information

Design of halogen-free hyper-crosslinked porous ionic polymers for efficient CO₂ capture and conversion

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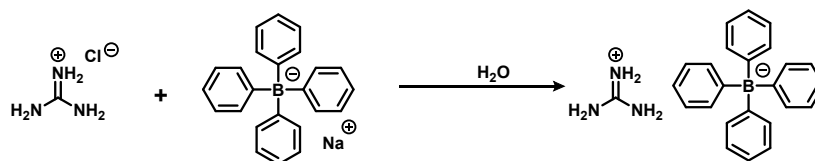
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Preparation of [Gua][Bph₄]:



Generally, 0.05 mol of [Gua][Cl] and [Na][BPh₄] were dissolved in 50 mL of water in a 250 mL beaker, followed by a slow dropwise addition of the aqueous [Na][BPh₄] solution to the [Gua][Cl] at room temperature for 24 h. After the reaction, direct filtration and rinsing with large amounts of water to obtain a white solid. The resulting solid was dried under vacuum to obtain the [Gua][BPh₄] (yield: 95%).

¹H NMR (400 MHz, DMSO-*d*₆) δ 7.22 – 7.15 (m, 8H), 6.98 – 6.87 (m, 13H), 6.80 (t, *J* = 7.2 Hz, 4H) ¹³C NMR (101 MHz, DMSO-*d*₆) δ 164.57, 164.08, 163.59, 163.10, 158.31, 136.00, 125.82, 125.79, 125.77, 125.74, 122.00.

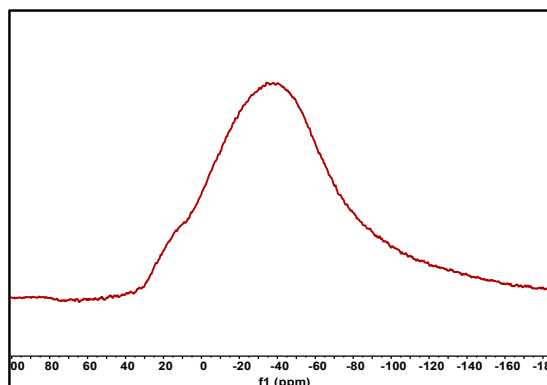


Fig. S1 Solid-state ¹¹B NMR spectrum of Gua-HCPIP-4.

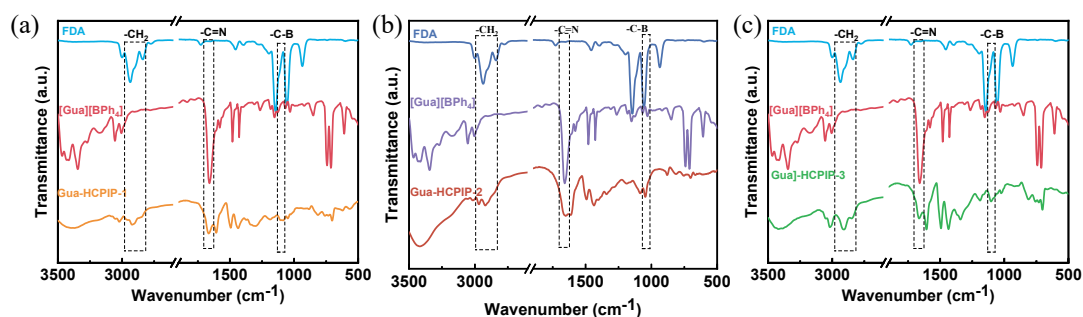


Fig. S2. FT-IR spectra of FDA, [Gua][BPh₄] and Gua-HCPIP-*x*.

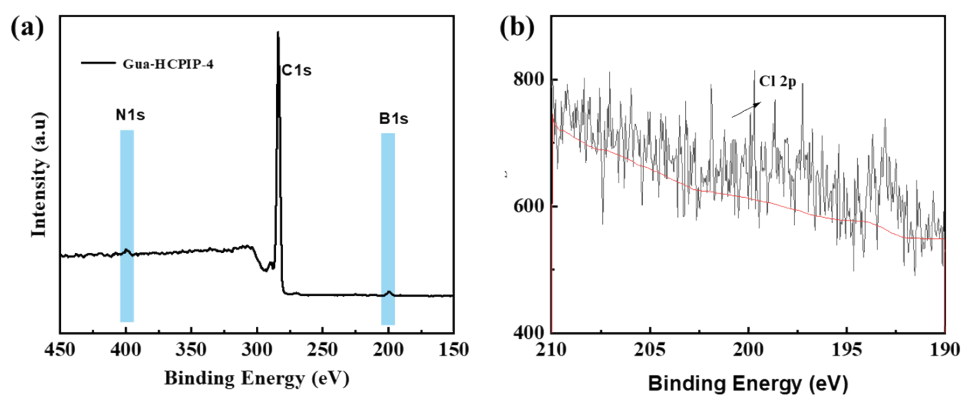


Fig. S3 (a) Survey XPS and (b) high resolution Cl 2p spectra of Gua-HCPIP-4.

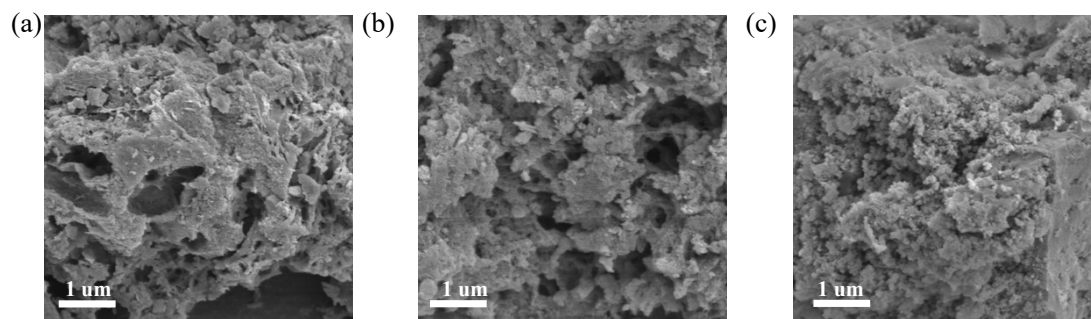


Fig. S4. (a) SEM images of Gua-HCPIP-1, (b)Gua-HCPIP-2, and (c) Gua-HCPIP-3.

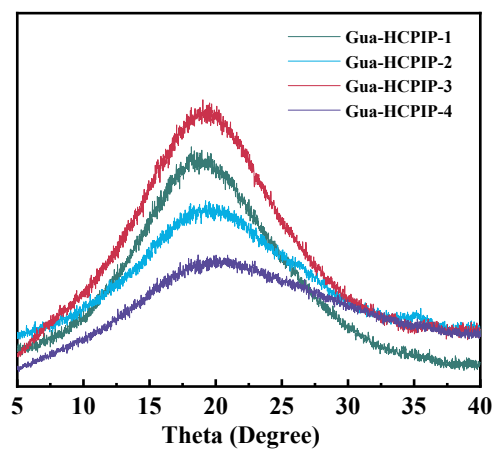


Fig. S5. The XRD patterns of HCPIPs.

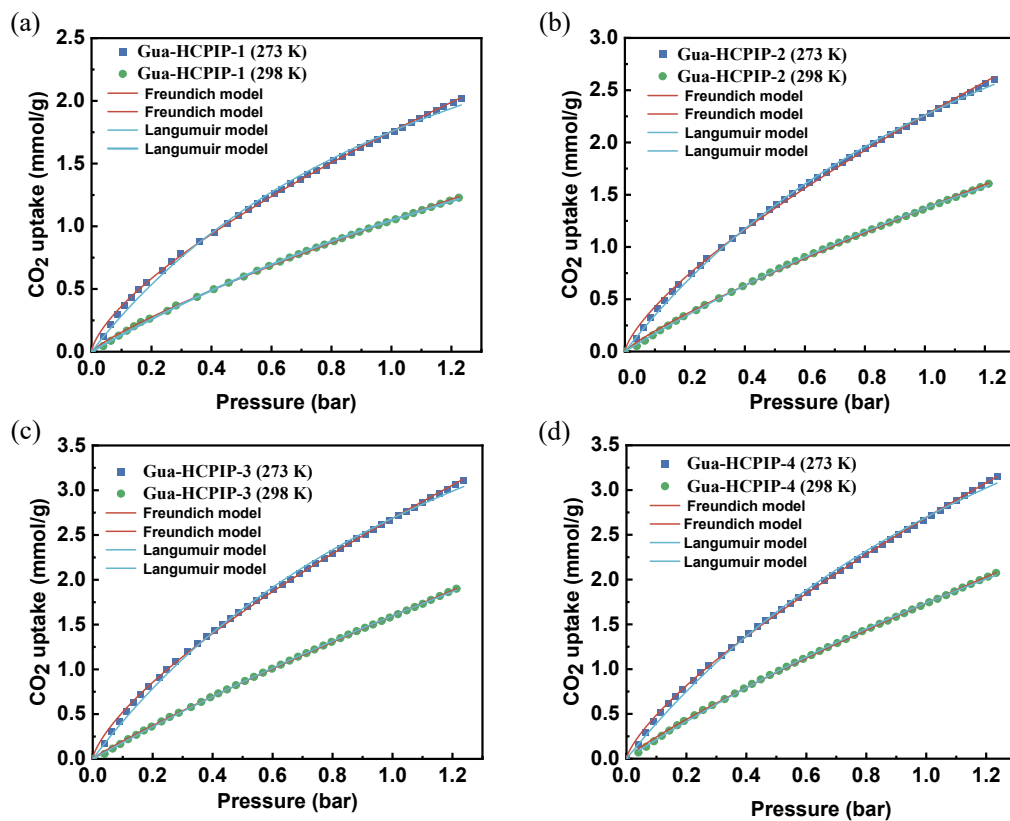


Fig. S6. CO₂ adsorption isotherms for HCPIPs at different temperatures.

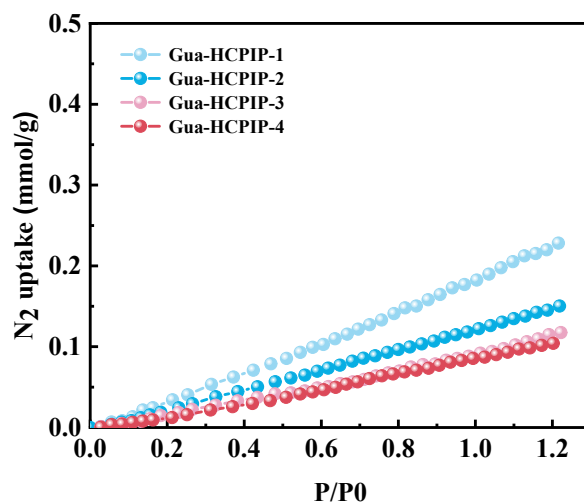


Fig. S7. N₂ adsorption isotherms for HCPIP at 273 K.

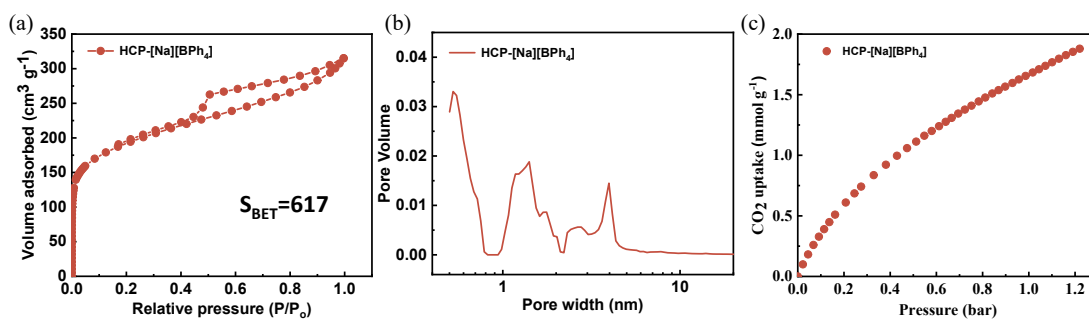


Fig. S8. (a) N_2 adsorption–desorption isotherm and (b) DFT pore distribution of HCP-[Na][BPh₄]; (c) CO_2 adsorption isotherm of HCP-[Na][BPh₄] at 273 K.

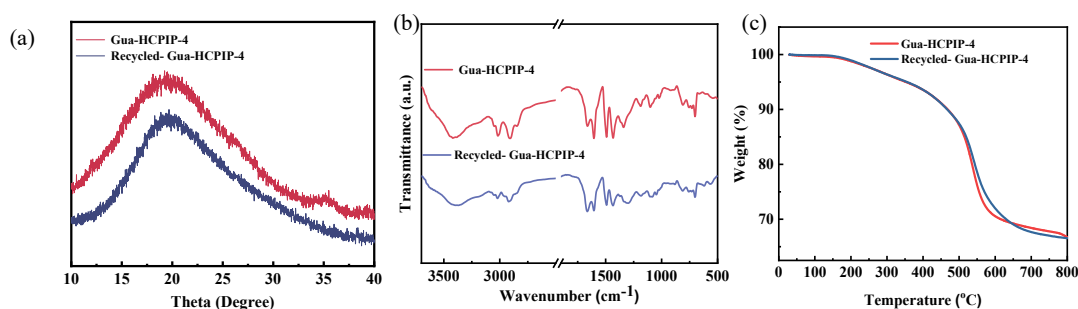


Fig. S9. (a) XRD patterns; (b) FT-IR spectra; and (c) TGA curves of fresh and recycled Gua-HCPIP-4.

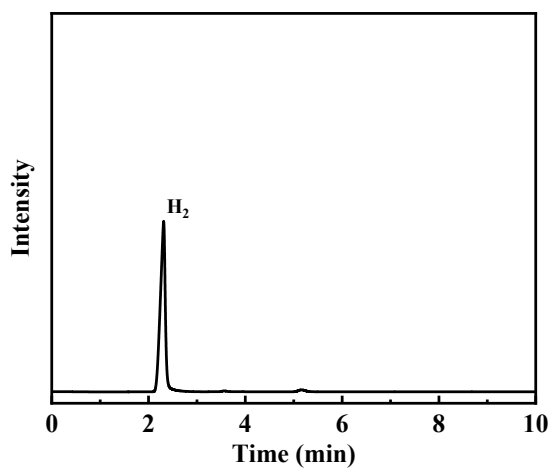
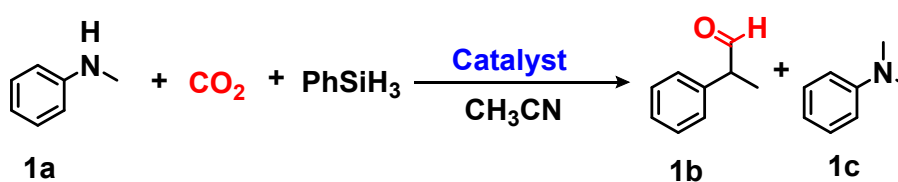


Fig. S10. Gas chromatography chart of reaction exhaust gas.



Scheme S1. N-methylaniline (1a) as a substrate and PhSiH₃ as a reducing agent for CO₂ N-formylation in CH₃CN to form the pharmaceutical intermediate N-methylformanilide (1b)

Table S1. Synthesis conditions, and S_{BET} of the HCPIPs.^a

Sample	FDA/IL	FeCl ₃ (mmol)	Solvent (mL)	S _{BET} (m ² g ⁻¹)
Gua-HCPIP-1	10	20	20	198
Gua-HCPIP-2	20	20	20	392
Gua-HCPIP-3	30	20	20	551
Gua-HCPIP-4	40	20	20	598

^aHyper crosslinking conditions: N₂ protection, 80 °C, 24 h. Solvent: 1,2 dichloroethane

Table S2. Textural parameters of the HCPIP.

Polymers	FDA/ IL	S _{BET} (m ² g ⁻¹)	S _{micro} (m ² g ⁻¹)	S _{meso} (m ² g ⁻¹)	V _{total} (cm ³ g ⁻¹)	V _{micro} (cm ³ g ⁻¹)	N content (%)	IL content (mmol g ⁻¹)
Gua-HCPIP-1	10	196	84	112	0.196	0.083	1.50	0.357
Gua-HCPIP-2	20	392	203	189	0.376	0.177	1.35	0.321
Gua-HCPIP-3	30	522	268	254	0.588	0.228	1.21	0.288
Gua-HCPIP-4	40	598	311	287	0.457	0.263	1.02	0.243

Table S3. CO₂ adsorption Langmuir, Freundlich parameters isotherms.

Polymers	T(K)	Q _{exp}	Langumuir	Freundlich
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			q_{mod}	Q_{max}	kl	R^2	q_{mod}	K_f	n	R^2
Gua-	273.15	2.022	1.968	4.081	0.755	0.997	2.022	1.755	1.462	0.999
HCPIP-1	298.15	1.261	1.217	4.376	0.315	0.999	1.237	1.046	1.207	0.999
Gua-	273.15	2.601	2.556	5.895	0.622	0.999	2.626	2.261	1.383	0.999
HCPIP-2	298.15	1.605	1.587	6.195	0.284	0.999	1.605	1.370	1.184	0.999
	273.15	3.114	3.041	6.805	0.653	0.998	3.125	2.686	1.406	0.999
Gua-				11.75						
HCPIP-3	298.15	1.901	1.884	2	0.157	0.999	1.900	1.596	1.106	0.999
	273.15	3.154	3.078	7.669	0.542	0.998	3.154	1.737	1.171	0.999
Gua-										
HCPIP-4	298.15	2.074	2.047	8.599	0.254	0.999	2.074	2.692	1.349	0.999

Table S4. CO₂ adsorption performance of previous materials.

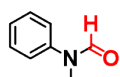
Sample	CO ₂ uptake (mmol g ⁻¹) (273 K, 1bar)	CO ₂ /N ₂ selectivity		Ref.
		Method	(273 K, 1 bar)	
Gua-HCPIP-4	2.72	IAST	332	This work
PIPs-6	2.35	IAST	256	1
COP-222	1.08	-	-	2
ILs-POF _{0.5}	2.14	-	-	3
PAF-167	1.68	IAST	49.7	4
PCP-Cl	2.31	IAST	48	5

HPILs-Cl-1	3.25	IAST	44	6
SCHPP-3	1.32	-	-	7
PDI-HIP-1	2.11	-	-	8
PTPIM-IL	2.69	-	-	9
HIP-Br-His	2.90	IAST	53	10
HBIM(2)@QA	2.97	-	-	11
HPMBr0.5	1.83	-	-	12
[HBIM- 2]Br- DCX (3)	1.80	-	-	13
Al-HIP-3	2.10	-	-	14
IHCP-2	2.87	IAST	24.5	15

Table S5. Comparison of catalysts for CO₂ N-formylation reaction

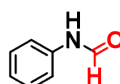
Sample	Solvent	Temperature (°C) / Time (h)	Pressure (bar)	Yield (%)	Ref.
ZIF-90-C	CH ₃ CN	50/24	1	87	16
Zn(OAc) ₂ /phen	CH ₃ CN	25/72	5	92	17
Pd/Bi-ZnOx	CH ₃ CN	30/24	5	29.6	18
(C ₅ H ₇ N ₂) ₅ [CuW ₁₂ O ₄₀]	CH ₃ CN	25/24	1	94	19
[Fe(acac) ₂]+pp ₃	THF	25/18	1	45	20
[Ph ₃ PMe]CO ₃ Me	THF	70/16	1	72	20

NSC1	DMF	50/24	1	77	21
43# ILHCP	-	35/16	5	99	22
TB-G/H-COF	CH ₃ CN	80/7	1	97	23
Gua-HCPIP-4	CH ₃ CN	80/24	3	94	This work



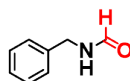
1b

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.53 (s, 1H), 7.42 (t, *J* = 8.0 Hz, 2H), 7.32 (d, *J* = 7.4 Hz, 2H), 7.25 (t, *J* = 7.4 Hz, 1H), 3.21 (s, 3H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 162.48, 142.53, 129.86, 126.04, 121.93, 31.43.



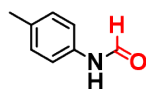
2b

¹H NMR (400 MHz, DMSO-*d*₆) δ 10.22 (s, 1H), 8.88 – 8.29 (m, 1H), 7.65 (d, *J* = 8.0 Hz, 2H), 7.33 (t, *J* = 7.9 Hz, 2H), 7.22 (d, *J* = 8.0 Hz, 1H), 7.09 (d, *J* = 7.0 Hz, 1H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 162.99, 160.06, 138.73, 129.84, 129.31, 124.06, 119.63, 117.99.



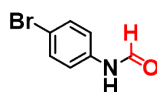
3b

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.52 (s, 1H), 8.15 (s, 1H), 7.41 – 7.19 (m, 5H), 4.31 (d, *J* = 6.2 Hz, 2H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 165.40, 161.50, 139.46, 128.81, 127.77, 127.35, 45.01, 41.18.



4b

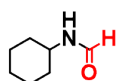
¹H NMR (400 MHz, DMSO-*d*₆) δ 10.12 (s, 1H), 8.25 (s, 1H), 7.49 (d, *J* = 8.2 Hz, 2H), 7.11 (t, *J* = 7.1 Hz, 2H), 2.25 (s, 3H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 162.92, 159.79, 136.26, 136.23, 133.17, 132.98, 130.24, 129.67, 119.56, 118.15, 20.90, 20.76.



5b

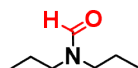
¹H NMR (400 MHz, DMSO-*d*₆) δ 10.29 (d, *J* = 46.1 Hz, 1H), 8.89 – 8.22 (m, 1H), 7.58 (d, *J* = 8.8 Hz, 2H), 7.51 (d, *J* = 8.5 Hz, 2H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 162.97, 160.23,

138.32, 138.05, 132.59, 132.15, 121.54, 119.79, 115.64.



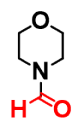
6b

^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ 7.92 (s, 1H), 3.68 – 3.53 (m, 1H), 1.81 – 1.46 (m, 4H), 1.19 (dt, J = 42.6, 12.1 Hz, 4H). ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 160.32, 50.67, 46.51, 34.64, 32.75, 25.61, 24.81.



7b

^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ 7.77 (s, 1H), 2.98 – 2.83 (m, 4H), 1.23 (dh, J = 14.9, 7.4 Hz, 4H), 0.57 (td, J = 7.5, 2.5 Hz, 6H). ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 163.00, 48.43, 43.22, 21.80, 20.59, 11.58, 11.08.



8b

^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ 8.02 (s, 1H), 3.58 (t, J = 4.9 Hz, 2H), 3.53 (t, J = 5.0 Hz, 2H), 3.39 (q, J = 5.6 Hz, 4H). ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 161.36, 67.18, 66.21, 45.54.

REFERENCES

1. K. Cai, P. Liu, Z. Chen, P. Chen, F. Liu, T. Zhao and D.-J. Tao, *Chem. Eng. J.*, 2023, **451**, 138946.
2. S. Subramanian, J. Oppenheim, D. Kim, T. S. Nguyen, W. M. H. Silo, B. Kim, W. A. Goddard and C. T. Yavuz, *Chem*, 2019, **5**, 3232-3242.
3. Y. Yang, Y. Guo, J. Yuan, H. Xie, C. Gao, T. Zhao and Q. Zheng, *ACS Sustainable Chem. Eng.*, 2022, **10**, 7990-8001.
4. T. Wang, Y. Du, Y. Yang, X. Jing and G. Zhu, *Ind. Eng. Chem. Res.*, 2022, **61**, 7284-7291.
5. O. Buyukcakil, S. H. Je, D. S. Choi, S. N. Talapaneni, Y. Seo, Y. Jung, K. Polychronopoulou and A. Coskun, *Chem. Commun.*, 2015, **52**, 934-937.
6. Y. Sang and J. Huang, *Chem. Eng. J.*, 2020, **385**, 123973.
7. X. Meng, Y. Liu, S. Wang, J. Du, Y. Ye, X. Song and Z. Liang, *ACS Appl. Polym. Mater.*, 2020, **2**, 189-197.
8. S. H. Goudar, D. S. Ingle, R. Sahu, S. Kotha, S. K. Reddy, D. J. Babu and V. R. Kotagiri, *ACS Appl. Polym. Mater.*, 2023, **5**, 2097-2104.
9. G. Feng, M. Yang, H. Chen, B. Liu, Y. Liu and H. Li, *Sep. Purif. Technol.*, 2023, **323**, 124484.
10. C. Guo, G. Chen, N. Wang, S. Wang, Y. Gao, J. Dong, Q. Lu and F. Gao, *Sep. Purif. Technol.*, 2023, **312**, 123375.
11. X. Liao, Z. Wang, L. Kong, X. Gao, J. He, D. Huang and J. Lin, *Mol. Catal.*, 2023, **535**, 112834.
12. H. Lyu, X. Wang, W. Sun, E. Xu, Y. She, A. Liu, D. Gao, M. Hu, J. Guo, K. Hu, J. Cheng, Z. Long, Y. Liu and P. Zhang, *Green Chem.*, 2023, **25**, 3592-3605.
13. X. Liao, Z. Wang, Z. Li, L. Kong, W. Tang, Z. Qin and J. Lin, *Chem. Eng. J.*, 2023, **471**, 144455.
14. W. Xu, M. Chen, Y. Yang, K. Chen, Y. Li, Z. Zhang and R. Luo, *ChemCatChem*, 2023, **15**, e202201441.
15. J. Gu, Y. Yuan, T. Zhao, F. Liu, Y. Xu and D.-J. Tao, *Sep. Purif. Technol.*, 2022, **301**, 121971.
16. K. Zhu, Y. Li, Z. Li, Y. Liu, H. Wu and H. Li, *Chem. Commun.*, 2022, **58**, 12712-12715.
17. Q. Zhang, X.-T. Lin, N. Fukaya, T. Fujitani, K. Sato and J.-C. Choi, *Green Chem.*, 2020, **22**, 8414-8422.
18. P. Bai, Y. Zhao and Y. Li, *Catal. Today*, 2024, **430**, 114551.
19. P. Sood, S. Bhatt, H. Bagdwal, A. Joshi, A. Singh, S. L. Jain and n. Monika Singh, *J. Mater. Chem. A*, 2024, **12**, 19168-19175.
20. Y. Li, X. Cui, K. Dong, K. Junge and M. Beller, *ACS Catal.*, 2017, **7**, 1077-1086.
21. F. D. Bobbink, S. Das and P. J. Dyson, *Nature Protoc.*, 2017, **12**, 417-428.
22. Q. Ren, Y. Chen, Y. Qiu, L. Tao and H. Ji, *Catal. Lett.*, 2021, **151**, 2919-2927.
23. Z. Mu, Y. Zhu, Y. Zhang, A. Dong, C. Xing, Z. Niu, B. Wang and X. Feng, *Angew. Chem. Int. Ed.*, 2023, **62**, e202300373.

